



UNIVERSITI PUTRA MALAYSIA

***HYDROGEN SENSORS USING TAPERED OPTICAL FIBER COATED
WITH METAL OXIDE NANOSTRUCTURES SYNTHESIZED VIA
CHEMICAL BATH DEPOSITION TECHNIQUE***

NOR AKMAR BINTI MOHD YAHYA

FK 2018 98



**HYDROGEN SENSORS USING TAPERED OPTICAL FIBER COATED
WITH METAL OXIDE NANOSTRUCTURES SYNTHESIZED VIA
CHEMICAL BATH DEPOSITION TECHNIQUE**

By

NOR AKMAR BINTI MOHD YAHYA

**Thesis Submitted to the School of Graduate Studies, Universiti Putra
Malaysia, in Fulfillment of the Requirements for the Degree of
Doctor of Philosophy**

July 2018

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs, and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright © Universiti Putra Malaysia



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of the requirement for the degree of Doctor of Philosophy

HYDROGEN SENSORS USING TAPERED OPTICAL FIBER COATED WITH METAL OXIDE NANOSTRUCTURES SYNTHESIZED VIA CHEMICAL BATH DEPOSITION TECHNIQUE

By

NOR AKMAR BINTI MOHD YAHYA

July 2018

Chairman : Mohd Hanif Yaacob, PhD
Faculty : Engineering

In this thesis, novel optical hydrogen (H_2) sensors based on manganese dioxide (MnO_2), zinc oxide (ZnO) and molybdenum trioxide (MoO_3) nanostructures coated on tapered multimode fiber (MMF) via chemical bath deposition (CBD) were developed and investigated. The use of H_2 as a clean fuel in various application requires practical and robust sensors as to minimize the risk of explosions associated with its volatile properties. Semiconducting metal oxides (SMO) has been widely used for decades in H_2 sensing purpose due to its simplicity in fabrication, low cost and high sensitivity. Nanostructures SMO thin films as sensing layer has been reported to enhance the sensitivity of the sensors due to its high surface area to increase the gas molecules-sensing layer interaction. Typical SMO gas sensors are electrical based in which conductivity changes as it reacts to H_2 gas. However, it has certain limitations such as easily affected by electromagnetic interference (EMI) thus compromise the signal response and small sparks could ignite massive explosion if the H_2 concentration leaks is more than 4% in the environment. On the other hand, optical sensor which has yet well explored, offers advantages in term of size, light weight, resistant to EMI and resilient in high temperature environment. By integrating the optical transducer with SMO material, it can be employed as a hydrogen gas sensor. There are various methods of producing SMO material such as chemical and physical vapor deposition, RF sputtering, electrochemical deposition and thermal evaporation. These techniques require complicated setup with high operating temperature along with carrier gas during the process and need conductive substrate to perform the procedure. These techniques were also difficult to be implemented on optical fiber. Alternatively, chemical bath deposition method provides simple and easy

setup, low operating temperature, low cost and environmental friendly. Therefore the author opted this method to fabricate H₂ sensor using tapered optical fiber coated with selected SMO incorporated with palladium (Pd) as a catalyst to enhance the optical responses.

In this study, the fabricated sensor is comprised of tapered multimode silica fiber (MMF) as the transducing platform. The tapering process is essential as to enhance the sensitivity to the environment through the interaction of evanescent field on the tapered surface area. The tapered region is then coated with sensing layer which is also important factors that influence the performance of the sensor. For this work, the author focused on a few kinds of SMO material well-known for their electrochromic properties which are manganese dioxide (MnO₂), zinc oxide (ZnO) and molybdenum trioxide (MoO₃), combined with Pd as the catalytic layer. The SMOs were grown via chemical bath technique and in-situ deposited onto the tapered optical fiber. The morphology of MnO₂, ZnO and MoO₃ synthesized and deposited on optical fiber were found to be nanograins, nanoflowers and nanogranules which were well distributed over the cylindrical shaped of the tapered optical fiber. The absorbance response of these sensors was characterized in terms of response and recovery times, sensitivity, repeatability and selectivity. It was discovered that the optimum thickness where the sensors of MnO₂, ZnO and MoO₃ exhibited maximum absorbance response are 300 nm, 280 nm and 250 nm respectively. It was revealed that the annealed sensor demonstrated higher sensitivity compared to as-prepared sensor. It was discovered that annealed Pd/MoO₃ coated on tapered optical fiber sensor exhibited highest absorbance increase of 3.80 when exposed to 1% H₂ at low operating temperature of 150°C as compared to other metal oxides nanostructures. The response and recovery times recorded were 1.2 min and 3.0 min. The developed MnO₂, ZnO and MoO₃ nanostructures coated on tapered optical fiber sensor for H₂ using CBD technique are the first of its kind according to the author's knowledge.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra
Malaysia sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

**PENDERIA HIDROGEN MENGGUNAKAN GENTIAN OPTIK TIRUS
DISALUT DENGAN OKSIDA LOGAM BERSTRUKTUR NANO MELALUI
TEKNIK PEMENDAPAN MANDIAN KIMIA**

Oleh

NOR AKMAR BINTI MOHD YAHYA

Julai 2018

Pengerusi : Mohd Hanif bin Yaacob, PhD
Fakulti : Kejuruteraan

Dalam tesis ini, penderia hidrogen optik baru (H_2) yang berasaskan kepada nanostructures mangan dioksida (MnO_2), zink oksida (ZnO) dan molibdenum trioksida (MoO_3) yang disalut ke atas gentian optik multimodal tirus (MMF) melalui pemendapan mandi kimia (CBD). Penggunaan H_2 sebagai bahan bakar bersih dalam pelbagai aplikasi memerlukan pengesanan praktikal dan tahan lama untuk meminimumkan risiko letupan yang berkaitan dengan sifat hidrogen yang tidak menentu. Semikonduktor logam oksida (SMO) telah digunakan secara meluas selama beberapa dekad dalam tujuan sensing H_2 kerana mudah dalam fabrikasi, berkos rendah dan mempunyai kepekaan yang tinggi. SMO filem-filem nipis berstruktur nano yang digunakan sebagai lapisan penderia telah banyak dilaporkan mengenai peningkatan kepekaan mengesan disebabkan oleh permukaan kawasan yang tinggi untuk meningkatkan interaksi lapisan penderia-molekul gas. Penderia gas SMO yang sering digunapakai adalah jenis elektrik yang berasaskan perubahan konduktiviti apabila ia bertindak balas terhadap gas H_2 . Walau bagaimanapun, ia mempunyai batasan tertentu seperti mudah terjejas oleh gangguan elektromagnetik (EMI) dengan itu menjejaskan tindak balas isyarat dan percikan api kecil boleh menyalakan letupan besar jika kebocoran kepekatan H_2 melebihi daripada 4% dalam persekitaran. Di sisi lain, penderia optik yang masih belum diterokai dengan baik, menawarkan kelebihan dari segi saiz, ringan, tahan terhadap EMI dan tahan lama dalam persekitaran suhu tinggi. Dengan mengintegrasikan transduser optik dengan bahan SMO, ia boleh digunakan sebagai penderia gas hidrogen. Terdapat pelbagai kaedah untuk menghasilkan bahan SMO seperti pemendapan wap kimia dan fizikal, semburan gelombang radio, pemendapan elektrokimia dan penyejatan haba. Teknik ini memerlukan persediaan rumit dengan suhu

operasi yang tinggi bersama-sama dengan gas pembawa semasa proses dan memerlukan substrat konduktif untuk melaksanakan prosedur. Teknik-teknik ini juga sukar untuk dilaksanakan pada gentian optik. Secara alternatif, kaedah pemendapan mandi kimia menyediakan persediaan mudah, suhu operasi yang rendah, kos rendah dan mesra alam sekitar. Oleh itu, pengarang memilih kaedah ini untuk mengesan H₂ menggunakan gentian optik tirus yang disalut dengan SMO terpilih yang digabungkan dengan palladium (Pd) sebagai pemangkin untuk meningkatkan tindak balas optik.

Dalam kajian ini, penderia yang direka terdiri daripada gentian kaca multimodal (MMF) sebagai asas transduser. Proses penirusan adalah penting untuk meningkatkan kepekaan terhadap alam sekitar melalui interaksi gelombang evanesen di kawasan permukaan tirus. Rantau tirus kemudian dilapisi dengan lapisan penderia yang juga merupakan faktor penting yang mempengaruhi prestasi penderia tersebut. Untuk karya ini, pengarang memberi tumpuan kepada beberapa jenis bahan SMO yang terkenal dengan sifat elektrokromiknya iaitu mangan dioksida (MnO₂), zink oksida (ZnO) dan molibdenum trioksida (MoO₃), digabungkan dengan Pd sebagai lapisan pemangkin. SMOs ditumbuh melalui teknik mandi kimia secara in-situ dan didepositkan atas gentian optik tirus. Morfologi MnO₂, ZnO dan MoO₃ yang disintesis dan didepositkan pada gentian optik didapati berbentuk *nanograins*, *nanoflowers* dan *nanogranules* yang menyelaputi dengan baik dan sekata di atas gentian optik tirus berbentuk silinder. Respon penyerapan penderia ini dicirikan dari segi masa tindak balas dan pemulihan, kepekaan, pengulangan dan selektiviti. Telah ditemui bahawa ketebalan optimum di mana sensor MnO₂, ZnO dan MoO₃ menunjukkan tindak balas penyerapan maksimum adalah 300 nm, 280 nm dan 250 nm masing-masing. Telah didedahkan bahawa penderia yang *anneal* menunjukkan kepekaan yang lebih tinggi berbanding penderia yang tidak di *anneal*. Telah didapati bahawa Pd/MoO₃ yang disalut pada penderia gentian optik tirus menunjukkan peningkatan penyerapan tertinggi sebanyak 3.80 apabila terdedah kepada 1% H₂ pada suhu operasi yang rendah iaitu 150°C berbanding dengan struktur nano oksida yang lain. Masa tindak balas dan pemulihan yang direkodkan adalah 1.2 min dan 3.0 min. MnO₂, ZnO dan MoO₃ berstruktur nano yang bersalut pada penderia gentian optik tirus untuk H₂ menggunakan teknik CBD adalah yang pertama seumpamanya menurut pengetahuan pengarang.

ACKNOWLEDGEMENTS

First and foremost, I would like to express my sincerest gratitude to my advisor, Dr. Mohd Hanif Yaacob for the continuous support of my PhD study and related research, for his patience, motivation and immense knowledge. His guidance helped me constantly throughout this PhD journey. Many thanks also to my co-supervisor: Prof. Adzir, Dr. Norizah and Dr. Ong Boon Hoong, for their insightful comments, encouragement and assistance whenever I need in difficult time.

I thank my fellow lab mates (UM & UPM) for the stimulating discussions, the encouragement, support and inspiration while cracking our heads trying to find solutions and answers for this PhD work. All the fun we had together, I will cherish them all.

Special thanks to my beloved father, Dato' Dr. Mohd Yahya Nordin and mother, Datin Norhayati Shaari for their non-stop prayers, support and guidance in my life. They always boost my spirit and energy whenever I feel down and helpless. Not to forget my sister, Nur Syahira who helped me in writing my thesis, thank you! Last but not least, to my loving husband, Mohd Faizul and my lovely kids, Najmi and Aqil, I cannot thank you enough for your support, love, care, and understanding for my PhD study. I am so grateful to have all of you in my life.

I certify that a Thesis Examination Committee has met on 24 July 2018 to conduct the final examination of Nor Akmar binti Mohd Yahya on her thesis entitled "Hydrogen Sensors Using Tapered Optical Fiber Coated with Metal Oxide Nanostructures Synthesized via Chemical Bath Deposition Technique" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

Members of the Thesis Examination Committee were as follows:

Ahmad Shukri bin Muhammad Noor, PhD

Associate Professor
Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Siti Barirah binti Ahmad Anas, PhD

Associate Professor
Faculty of Engineering
Universiti Putra Malaysia
(Internal Examiner)

Mohd Nizar b Hamidon, PhD

Associate Professor
Faculty of Engineering
Universiti Putra Malaysia
(Internal Examiner)

Md. Zaini Jamaludin, PhD

Professor
Universiti Tenaga Nasional
Malaysia
(External Examiner)



RUSLI HAJI ABDULLAH, PhD

Professor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

Date: 30 August 2018

This thesis was submitted to the Senate of the Universiti Putra Malaysia and has been accepted as fulfilment of the requirement for the degree of Doctor of Philosophy. The members of the Supervisory Committee were as follows:

Mohd Hanif Yaacob, PhD

Senior Lecturer
Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Mohd Adzir bin Mahdi, PhD

Professor
Faculty of Engineering
Universiti Putra Malaysia
(Member)

Norizah binti Abdul Rahman, PhD

Senior Lecturer
Faculty of Science
Universiti Putra Malaysia
(Member)

Ong Boon Hoong, PhD

Associate Professor
Nanotechnology and Catalysis Research Centre
Universiti Malaya
(Member)

ROBIAH BINTI YUNUS, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software

Signature: _____ Date: _____

Name and Matric No.: Nor Akmar binti Mohd Yahya, GS 40345

Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) were adhered to.

Signature: _____
Name of Chairman
of Supervisory
Committee: Dr. Mohd Hanif Yaacob

Signature: _____
Name of Member
of Supervisory
Committee: Professor Dr. Mohd Adzir bin Mahdi

Signature: _____
Name of Member
of Supervisory
Committee: Dr. Norizah binti Abdul Rahman

Signature: _____
Name of Member
of Supervisory
Committee: Associate Professor Dr. Ong Boon Hoong

TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xiv
LIST OF FIGURES	xv
LIST OF ABBREVIATIONS	xxii
CHAPTER	
1 INTRODUCTION	1
1.1 Overview	1
1.2 Problem Statement	2
1.3 Objectives	3
1.4 Scope of Work and Limitation	5
1.5 Thesis Organisation	5
2 LITERATURE REVIEW AND RESEARCH RATIONALES	7
2.1 Introduction	7
2.2 Optical Fiber Sensor	7
2.3 Optical Measurement Techniques in Gas Sensing	12
2.4 Gas Sensing Mechanism for Optical Sensors Based on Metal Oxide Nanostructured Thin Films	14
2.5 Review of Nanostructured Metal Oxide in Optical Sensors for Hydrogen Sensing Application	15
2.5.1 Manganese Dioxide (MnO ₂)	19
2.5.2 Zinc Oxide (ZnO)	21
2.5.3 Molybdenum Trioxide (MoO ₃)	23
2.5.4 Catalytic Metal of Palladium (Pd) in Optical Hydrogen Gas Sensors	25
2.6 Reviews on SMO Nanomaterials Deposition Techniques	26
2.7 Summary	27
3 OPTICAL FIBER SENSOR MODIFICATION AND SYNTHESIS OF METAL OXIDE NANOSTRUCTURES	28
3.1 Modification of optical fiber via tapering process	28
3.2 Synthesis and deposition process of MnO ₂ , ZnO and MoO ₃ nanostructures via chemical bath	30

3.2.1	Synthesis and Deposition of MnO ₂ Nanograins	31
3.2.2	Synthesis and Deposition of ZnO Nanoflowers	34
3.2.3	Synthesis and Deposition of MoO ₃ Nanogranules	36
3.3	Deposition of Catalytic Metal Layer of Palladium	38
3.4	Design and synthesis parameter of H ₂ fiber sensor	39
3.5	Summary	40
4	MICRO-CHARACTERIZATION TECHNIQUES AND GAS SENSING MEASUREMENT SETUP	42
4.1	Introduction	42
4.2	Micro-Characterization Techniques	42
4.2.1	Field Emission Scanning Electron Microscope (FESEM)	42
4.2.2	Energy Dispersive Spectroscopy (EDS)	44
4.2.3	X-Ray Diffraction (XRD)	45
4.2.4	Raman Spectroscopy	46
4.2.5	Thermogravimetric Analysis (TGA)	47
4.2.6	UV-Visible Spectroscopy	48
4.3	Gas Sensing Measurement Setup	49
4.3.1	Absorbance Measurement Setup	49
4.3.2	Gas Chamber Construction for Optical Fiber Sensors	51
4.3.3	Gas Testing System and Procedures	52
4.4	Optical sensing parameters	53
4.5	Summary	54
5	MICRO-CHARACTERIZATION PROPERTIES OF SYNTHEZIZED NANOMATERIALS AND SENSOR RESPONSES TOWARDS HYDROGEN GAS	55
5.1	Introduction	55
5.2	MnO ₂ Nanostructures H ₂ Gas Optical Sensors	55
5.2.1	Micro-Characterization of MnO ₂ Layer Coated on Tapered Optical Fiber	56
5.2.1.1	Field Emission Scanning Electron Microscope (FESEM)	56
5.2.1.2	Energy Dispersive Spectroscopy (EDS)	58
5.2.1.3	X-ray Diffraction (XRD)	59
5.2.1.4	Raman Spectroscopy	59
5.2.1.5	Thermogravimetric Analysis (TGA)	60
5.2.1.6	Ultraviolet-Visible Spectroscopy (Uv-Vis)	61
5.2.2	Absorbance Measurements of H ₂ Sensing	62
5.2.2.1	Untapered and Tapered Blank Optical Fiber Sensing Properties	62
5.2.2.2	Tapered Optical Fiber Coated with MnO ₂ nanograins thin films Sensing Properties	64

5.2.2.3	Sensing Performance of Tapered Optical Fiber coated with Pd/MnO ₂ nanograins	68
5.2.2.4	Pd/MnO ₂ nanograins coated on tapered optical fiber for selectivity test with NH ₃ and CH ₄	79
5.3	ZnO Nanostructures H ₂ Gas Optical Sensors	81
5.3.1	Micro-Characterization of ZnO Layer Coated on Tapered Optical Fiber	81
5.3.1.1	Field Emission Scanning Electron Microscope (FESEM)	81
5.3.1.2	Energy Dispersive Spectroscopy (EDS)	85
5.3.1.3	X-ray Diffraction (XRD)	85
5.3.1.4	Raman Spectroscopy	86
5.3.1.5	Thermogravimetric Analysis (TGA)	87
5.3.1.6	Ultraviolet-Visible Spectroscopy (Uv-Vis)	88
5.3.2	Absorbance Measurements of H ₂ Sensing	89
5.3.2.1	ZnO nanostructures Coated on Tapered Optical Fiber	89
5.3.2.2	Pd/ZnO nanostructures Coated on Tapered Optical Fiber	90
5.3.2.3	Pd/ZnO nanostructures Coated on Tapered Optical Fiber for Selectivity Test with NH ₃ and CH ₄	101
5.4	MoO ₃ nanostructures H ₂ Gas Optical Sensors	103
5.4.1	Micro-Characterization of MoO ₃ Layer Coated on Tapered Optical Fiber	103
5.4.1.1	Field Emission Scanning Electron Microscope (FESEM)	103
5.4.1.2	Energy Dispersive Spectroscopy (EDS)	105
5.4.1.3	X-ray Diffraction (XRD)	106
5.4.1.4	Raman Spectroscopy	107
5.4.1.5	Thermogravimetric Analysis (TGA)	108
5.4.1.6	Ultraviolet-Visible Spectroscopy (Uv-Vis)	108
5.4.2	Absorbance Measurements of H ₂ Sensing	109
5.4.2.1	MoO ₃ nanogranules Coated on Tapered Optical Fiber	109
5.4.2.2	Pd/MoO ₃ nanogranules Coated on Tapered Optical Fiber	110
5.4.2.3	Pd/MoO ₃ nanogranules Coated on Tapered Optical Fiber for Selectivity Test with NH ₃ and CH ₄	123
5.5	Sensing Mechanism of the Developed SMO Coated Tapered Optical Fiber Sensor	124
5.6	Summary	126

6	CONCLUSIONS AND FUTURE WORKS	129
6.1	Conclusions	129
6.2	Future works	130
6.3	Outcome and Author's Achievement	131
	REFERENCES	132
	BIODATA OF STUDENT	144
	LIST OF PUBLICATIONS	145



LIST OF TABLES

Table		Page
2.1	Electrical based hydrogen sensor using several metal oxides reported in [58][39][51][16][44][57][56]	18
2.2	Optical based hydrogen sensor using several metal oxides reported in [38][61][62][63][64]	19
5.1	Response, recovery time, absorbance increment and sensitivity of MnO ₂ optical sensors when exposed to 1% H ₂	78
5.2	Response and recovery time, absorbance increment and sensitivity for different thicknesses of Pd/ZnO optical sensors induced by 1% of H ₂	97
5.3	Response and recovery time, absorbance increment and sensitivity of Pd/ZnO optical sensors induced by 1% of H ₂	101
5.4	Response and recovery time, absorbance increment and sensitivity for different thicknesses of Pd/MoO ₃ optical sensors induced by 1% of H ₂	119
5.5	Response and recovery time, absorbance increment and sensitivity of Pd/MoO ₃ optical sensors induced by 1% of H ₂	123
5.6	Summary of micro-characterization of synthesized nanomaterial	127
5.7	Summary of optical sensing performances of metal oxides coated with Pd towards 1% H ₂	127

LIST OF FIGURES

Figure		Page
1.1	Research design of the project	5
2.1	Fiber optic components [19]	7
2.2	Total internal reflection (TIR) in optical fiber [20]	8
2.3	Microscope image of etched optical fiber taken from [23]	9
2.4	D-shaped optical fiber cross section [26]	10
2.5	Schematic diagram of the side polished optical fiber embedded in a quartz block (cross-section and side view) [28]	10
2.6	Tapered optical fiber [30]	11
2.7	Spectral region that can be utilized for optical sensors	12
2.8	Illustration of absorption spectroscopy	13
2.9	Schematic structures of one-dimensional tunnel of MnO_2	20
2.10	a) ZnO unit cell with wurtzite structure and b) Various crystal planes of ZnO wurtzite structure [50]	22
2.11	Crystal structures of MoO_3 a) $\alpha\text{-MoO}_3$, b) $\beta\text{-MoO}_3$ and c) $h\text{-MoO}_3$ [78]	24
2.12	Schematic representation of Pd nanoparticles thin film on substrate [85]	25
3.1	Vytran GPX-3400 Optical Glass Fiber Processor in Lab Photonics UPM	29
3.2	SEM image of (a) multimode fiber (MMF), (b) transition region of tapered MMF	30
3.3	Chemical Bath Memmert brand	31
3.4	Tapered optical fiber immersed in the activation solution in chemical bath	32
3.5	Activated tapered optical fiber immersed in the MnO_2 solution in chemical bath	33
3.6	Samples of tapered optical fiber coated MnO_2 nanostructures	33

3.7	MnO ₂ nanostructured synthesized via CBD	33
3.8	MnO ₂ in powder form	34
3.9	Samples of tapered optical fiber coated ZnO nanostructured	35
3.10	ZnO nanostructured synthesized via CBD	36
3.11	ZnO in powder form	36
3.12	Samples of tapered optical fiber coated MoO ₃ nanostructures	37
3.13	MoO ₃ nanogranules synthesized via CBD	38
3.14	MoO ₃ in powder form	38
3.15	Turbo Sputter Coater Systems modelled K575X	39
3.16	Catalytic Pd target	39
3.17	Summary of design and fabrication of tapered optical fiber sensor coated with nanostructures MnO ₂ , ZnO and MoO ₃	41
4.1	Schematic Diagram of SEM [22]	43
4.2	FESEM Equipment a) Hitachi S4500 at Quasi S-Technology, b) Hitachi SU8030 at MIMOS Sdn. Bhd	44
4.3	XRD PANalytical EMPYREAN	46
4.4	Raman Spectroscopy	47
4.5	TGA/SDTA851 located at Nanotechnology and Catalyst UM	48
4.6	UV-VIS at Chemistry, UM	48
4.7	Absorbance measurement gas testing setup	50
4.8	Gas chamber for optical fiber	51
4.9	Light propagates into the tapered optical fiber sensor	52
4.10	Illustration of response and recovery time	53
5.1	FESEM image of (a) tapered MMF coated with MnO ₂ nanograins, (b) MnO ₂ nanograins as-prepared, (c) annealed at 200 °C	56
5.2	FESEM image of cross-section MnO ₂ coated on bare optical fiber prepared by chemical bath deposition technique for (a) 5 min, (b) 10 min and (c) 15 min of deposition time	57

5.3	Thickness layer of MnO ₂ versus deposition time	58
5.4	EDS measurement of MnO ₂ thin film	58
5.5	XRD pattern for both samples as-prepared and annealed MnO ₂ powder	59
5.6	Raman spectra of the as-prepared and annealed MnO ₂	60
5.7	TG curves of the MnO ₂ powder	61
5.8	Estimated optical band gap of MnO ₂ nanograins (3.50 eV)	62
5.9	Absorbance response versus wavelength of uncoated (a) untapered (125 μm), (b) tapered (20 μm) multimode optical fiber when exposed to 1% H ₂ at room temperature	63
5.10	Absorbance response versus wavelength of uncoated (a) untapered (125 μm), (b) tapered (20 μm) multimode optical fiber when exposed to 1% H ₂ at 200°C	64
5.11	Dynamic response of a) 20 μm, b) 30 μm and c) 40 μm of tapered optical fiber coated with MnO ₂ when exposed to 0.125% - 1.0% H ₂ at 240°C	66
5.12	Cumulative absorbance change as a function of H ₂ concentration for 20 μm, 30 μm and 40 μm tapered optical fiber	68
5.13	Change of cumulative absorbance versus operating temperature for MnO ₂ and Pd/MnO ₂ coated on tapered optical fiber when exposed to H ₂ of 1.0% in air	69
5.14	Absorbance versus optical wavelength for Pd/MnO ₂ coated tapered optical fiber exposed to 1% H ₂ at 240°C	70
5.15	Dynamic response of Pd/MnO ₂ coated tapered optical fiber exposed to 1% H ₂ at 240°C	70
5.16	Absorbance versus wavelength of Pd/MnO ₂ coated optical fiber for three deposition time of 5 min, 10 min and 15 min when exposed to 1% H ₂ at 240°C	71
5.17	Cumulative absorbance of Pd/MnO ₂ coated optical fiber for three deposition time of 5 min, 10 min and 15 min when exposed to 1% H ₂ at 240°C	72
5.18	Absorbance versus optical wavelength as-prepared and annealed MnO ₂ coated tapered optical fiber sensor towards synthetic air and 1% H ₂ at 240°C	73

5.19	Absorbance versus optical wavelength of Pd/MnO ₂ sensor for as-prepared and annealed samples exposed to 1% concentration of H ₂ in synthetic air at 240°C	74
5.20	Dynamic response of tapered optical fiber coated with (a) MnO ₂ as-prepared, (b) MnO ₂ annealed, towards different concentrations of H ₂ in synthetic air at 240°C	75
5.21	Dynamic absorbance response of optical fiber coated with (a) Pd/MnO ₂ as-prepared, (b) Pd/MnO ₂ annealed, towards different concentrations of H ₂ in synthetic air at 240°C	76
5.22	Absorbance response for different concentrations of H ₂ for annealed Pd/MnO ₂ coated tapered optical fiber at 240°C	77
5.23	Error bar plot of cumulative absorbance change as a function of H ₂ concentration for sensor (a) as-prepared MnO ₂ , (b) annealed MnO ₂ , (c) as-prepared Pd/MnO ₂ , (d) annealed Pd/MnO ₂	79
5.24	(a) Cumulative absorbance (500-1000nm) versus time for annealed Pd/MnO ₂ when exposed to 1% concentration of H ₂ , NH ₃ and CH ₄ at 240°C, b) Comparison bar graph for selectivity test	80
5.25	FESEM Image of (a) ZnO nanoflowers coated on tapered optical fiber, (b) ZnO nanoflowers as-prepared and (c) annealed at 200°C	82
5.26	Morphologies and cross-section view of synthesized ZnO for different deposition time of (a-b) 5 min, (c-d) 10 min, (e-f) 20 min, (g-h) 60 min	83
5.27	Thickness layer of ZnO versus deposition time	84
5.28	EDS measurement of synthesized ZnO	85
5.29	XRD pattern for both sample as-prepared and annealed ZnO powder	86
5.30	Raman spectra of synthesized ZnO, excited with a 532 nm laser	87
5.31	TG curves of the ZnO powder	88
5.32	Estimated optical band gap of ZnO nanoflowers (3.1 eV)	89
5.33	Absorbance versus optical wavelength for ZnO coated on tapered optical fiber towards synthetic air and 1% H ₂ at 200°C	90

5.34	Change of cumulative absorbance versus operating temperature for ZnO and Pd/ZnO coated on tapered optical fiber when exposed to H ₂ of 1.0% in air	90
5.35	(a) Absorbance response versus wavelength and (b) dynamic response, of Pd/ZnO coated tapered optical fiber exposed to 1% H ₂ at 180°C	91
5.36	(a) Absorbance response versus wavelength and (b) dynamic response, of Pd/ZnO coated optical fiber for 5 min deposition time (170 nm thickness) at 180°C	93
5.37	(a) Absorbance response versus wavelength and (b) dynamic response, of Pd/ZnO coated optical fiber for 10 min deposition time (210 nm thickness) at 180°C	94
5.38	(a) Absorbance response versus wavelength and (b) dynamic response, of Pd/ZnO coated optical fiber for 20 min deposition time (280 nm thickness) at 180°C	95
5.39	(a) Absorbance response versus wavelength and (b) dynamic response, of Pd/ZnO coated optical fiber for 60 min deposition time (580 nm thickness) at 180°C	96
5.40	Cumulative absorbance change as a function of H ₂ concentration for sensor Pd/ZnO with thickness of (a) 170 nm, (b) 210 nm, (c) 280 nm and (d) 580 nm	97
5.41	Absorbance response versus optical wavelength of Pd/ZnO sensor for as-prepared and annealed samples exposed to 1% concentration of H ₂ in synthetic air at 180°C	98
5.42	Dynamic response of tapered optical fiber coated with (a) Pd/ZnO as-prepared, (b) Pd/ZnO annealed, towards different concentrations of H ₂ in synthetic air at 180°C	99
5.43	Error bar plot of cumulative absorbance change as a function of H ₂ concentration for sensor (a) as-prepared Pd/ZnO and (b) annealed Pd/ZnO	100
5.44	(a) Cumulative absorbance (500-1000nm) versus time for annealed Pd/ZnO when exposed to 1% concentration of H ₂ , NH ₃ and CH ₄ at 180°C, (b) Comparison bar graph for selectivity test	102
5.45	FESEM image of (a) tapered MMF coated with MoO ₃ nanograins, (b) MoO ₃ as-prepared, (c) annealed at 200°C	103

5.46	FESEM image of cross-section MoO ₃ coated on optical fiber prepared by CBD technique for (a) 5 min, (b) 10 min, (c) 15 min and (d) 20 min of time deposition	104
5.47	Thickness of MoO ₃ towards deposition time	105
5.48	EDS measurement of MoO ₃ thin film	106
5.49	XRD pattern for both samples as-prepared and annealed MoO ₃	107
5.50	Raman spectra of synthesized MoO ₃ excited with a 532 nm laser for both as-prepared and annealed	107
5.51	TG curve of the synthesized MoO ₃ powder	108
5.52	Estimated optical band gap of MoO ₃ nanogranules (2.79 eV)	109
5.53	Absorbance response against optical wavelength for MoO ₃ coated on tapered optical fiber towards synthetic air and 1% H ₂ at (a) room temperature and (b) 150°C	110
5.54	Change of cumulative absorbance versus operating temperature for MoO ₃ and Pd/MoO ₃ coated on tapered optical fiber when exposed to 1% H ₂ in pure synthetic air	111
5.55	(a) Absorbance response against wavelength and (b) dynamic response, of Pd/MoO ₃ coated tapered optical fiber exposed to 1% H ₂ at 150°C	112
5.56	(a) Absorbance response against wavelength and (b) dynamic response, of Pd/MoO ₃ coated optical fiber for 5 min deposition time (average (average: 90 nm thickness) at 150°C	114
5.57	(a) Absorbance response against wavelength and (b) dynamic response, of Pd/MoO ₃ coated optical fiber for 10 min deposition time (average: 250 nm thickness) at 150°C	115
5.58	(a) Absorbance response against wavelength and (b) dynamic response, of Pd/MoO ₃ coated optical fiber for 15 min deposition time (average: 450 nm thickness) at 150°C	116
5.59	(a) Absorbance response against wavelength and (b) dynamic response, of Pd/MoO ₃ coated optical fiber for 20 min deposition time (average: 750 nm thickness) at 150°C	117
5.60	Cumulative absorbance change as a function of H ₂ concentration for sensor Pd/MoO ₃ with thickness of (a) 90 nm, (b) 250, (c) 450 nm and (d) 750 nm	118

5.61	Absorbance response versus optical wavelength of Pd/MoO ₃ sensor for as-prepared and annealed samples exposed to 1% H ₂ in synthetic air at 150°C	119
5.62	Dynamic response of tapered optical fiber coated with Pd/MoO ₃ (a) as-prepared and (b) annealed, towards different concentrations of H ₂ in synthetic air at 150°C	121
5.63	Error bar plot of cumulative absorbance change as a function of H ₂ concentration for sensor (a) as-prepared Pd/MoO ₃ and (b) annealed Pd/MoO ₃	122
5.64	(a) Cumulative absorbance (800-1000 nm) versus time for annealed Pd/MoO ₃ when exposed to 1% of H ₂ , NH ₃ and CH ₄ at 150°C, (b) comparison bar graph for selectivity test	124
5.65	Illustration of gas sensing mechanism on tapered optical fiber	125
5.66	Diagram of sensing mechanism of SMO towards H ₂ gas and its recovery process	126

LIST OF ABBREVIATIONS

Al_2O_3	Aluminium oxide
CBD	Chemical bath deposition
CH_4	Methane
CO_2	Carbon dioxide
CuO	Copper oxide
DI water	Deionized water
EMI	Electromagnetic Interference
Fe_2O_3	Ferum oxide
H_2	Hydrogen
H_2O	Water molecules
MMF	Multimode fiber
MnO_2	Manganese oxide
MoO_3	Molybdenum oxide
Nb_2O_5	Niobium oxide
NH_3	Ammonia
NiO	Nickle oxide
Pd	Palladium
RF	Radio frequency
SiO_2	Silica oxide
SMF	Single mode fiber
SMO	Semiconducting metal oxides
SnO_2	Tin oxide
TiO_2	Titanium oxide
TIR	Total internal reflection
Vis-NIR	Visible near-infrared
WO_3	Tungsten oxide
ZnO	Zinc oxide

CHAPTER 1

INTRODUCTION

1.1 Overview

Hydrogen (H₂) is a colourless, odourless, nontoxic, highly volatile and inflammable. The use of H₂ as a clean source of energy in various applications such as automobiles, aircraft, fuel cells, chemical industries and food processing has drawn much attention to the safety and health concerns due to its volatile properties [1]. The gas is highly flammable and burnable in air at a very wide range of 4% to 75% by volume [2]. The leaking of H₂ with high concentration mixing with oxygen can cause explosion which is a threat to environment that includes the lives of human being. The explosive reactions can be triggered by heat, spark or even sunlight. The H₂ auto-ignition temperature (spontaneous ignition in air) is reported to be at 500 °C [3]. In order to avoid such devastated state, research and investigation on developing suitable H₂ gas sensors prior to its purpose and application has been carried out over decades.

There are four main types of hydrogen sensors which are chemiresistor, surface acoustic wave, microelectronic and optical based sensors. The most common sensors widely used are electrical sensors (chemiresistor/microelectronic) due to its low cost and high sensitivity towards gases. Even so, this type of sensor is susceptible by electromagnetic interference (EMI) thus compromising the signal response. On the other hand, optical sensors using optical fiber offers other valuable characteristic such as their small size, light weight, immune to electromagnetic interference (EMI), non-inductive with low attenuation and resilient in ruggedness with high temperature environment [4]. By manipulating the core or cladding of the optical fiber, the sensing response can be monitored through absorbance, reflectance or transmission.

W. Jin et al. [5] has reviewed on the recent development for gas detection using micro/nano-engineered optical fibers such as tapered optical fibers, fiber-tip micro-cavities, hollow-cores fibers, and suspended-core fibers. They also discussed on the detection schemes which are direct absorption (evanescent wave) and photoacoustic spectroscopy that can be applied depending on the preferred sensing method. The gas sensing optical fiber based using absorbance measurement is resulted from evanescent wave that changes to its surrounding. The light propagating in the fiber core produces evanescent field which radiates at the boundary of the core into the cladding of the optical fiber. By modifying the fiber cladding and coating it with a gas sensitive layer, the evanescent field changes when the layer

interacts with the gas molecules. Thus, the light in the fiber core alters its properties upon exposure to different gas concentrations [6].

One of the ways to modify the fiber cladding is to taper the optical fiber using the heat and pull technique. As of now, the research on tapered optical fiber has drawn much attention thanks to its advantages on strong evanescent wave and simplicity of production. There are various papers reported on tapered optical fiber as a sensor including strain, humidity, temperature, refractive index, chemical (liquid/gas) and biological sensors [7]–[14]. This has portrayed that tapered optical fiber has great potential to be developed in sensors application.

There are two main approaches to detect chemical substance using fiber optic. One is by measuring the intrinsic optical properties of the target/analyte (eg. refractive index) and another is by monitoring optical properties change of the fixed indicator such as sensing layer coated on the fiber optic [15]. The later approach usually incorporates with sensing layer that is sensitive to react with target/analyte. There are many type of sensing layers that are sensitive towards hydrogen gas. Organic (polymer) and inorganic materials (semiconductor metal oxides) and composite materials have been extensively studied on its electronic, chemical and optical properties. Semiconductor metal oxides (SMO) are popular to exhibit fast response, high sensitivity, long term stability, low cost and simplicity in fabrication [16]. Materials like ZnO, NiO, SnO₂, CuO, MoO₃, TiO₂, WO₃, and Fe₂O₃ are recognized to exhibit strong gas response with conductivity change [17].

Various techniques have been studied to synthesis SMO with nanostructures which offers high surface areas to promote more gas molecules-sensing layer interaction. These techniques include electrochemical deposition, sol-gel method, thermal plasma, hydrothermal, chemical bath deposition, RF sputter, chemical/physical vapor deposition and flame spray pyrolysis [2]. With development of nanostructures SMO with easy fabrication and deposition methods, it is appealing to develop a sensitive and reliable hydrogen sensor which is able to detect leakage instantly. By combining the advantages of optical fiber sensors with sensitive nanostructures sensing layer for hydrogen gas sensing application, it is an interesting research direction to be explored.

1.2 Problem Statement

Current H₂ sensors widely used is electrical based sensors. As mentioned previously, these type of sensors have major drawbacks which are vulnerable towards EMI and small sparks could ignite massive explosion if

the H₂ concentration leakage is more than 4% in the environment. This could be a threat to the human lives. Meanwhile, optical fiber sensor offers features that can overcome disadvantages of electrical sensor. By integrating this optical fiber with nanostructures metal oxides, H₂ sensors can be developed. It is important to detect H₂ concentration lower of its explosive threshold limit with fast response and high sensitivity.

Apart from that, the common nanomaterial deposition for optical fiber is based on physical deposition techniques for instance sputtering, dip-coating and drop casting methods. Sputtering technique could provide uniform coating but not quite suitable for cylindrical shape of optical fiber that could yield different thickness at certain area. Same goes with dip-coating and drop casting methods with easy techniques but hardly produce controlled thickness and the sensing layer attachment on optical fiber is poor. On the contrary, chemical bath deposition (CBD) method could offer better solution for sensing layer coating problem. This technique exhibits homogenous and uniform coating towards cylindrical shape of optical fiber as well as controlled thickness. Furthermore, the sensing layer coating is chemically bonded which makes the adherent strong. This technique is also simple and easy to perform. To coat SMO on optical fiber using CBD technique is still not establish. Therefore, different SMO are required to produce high viability for optical fiber coating.

1.3 Objectives

This thesis focuses on the development of tapered optical fiber sensors coated with metal oxide nanostructures for H₂ sensing application via chemical bath deposition. The objectives of this research are as follows:

- To design and develop hydrogen gas sensors based on metal oxides nanostructures coated on tapered optical fiber via chemical bath deposition method.
- To micro-characterize the synthesized metal oxides nanostructures.
- To evaluate the optical fiber sensor performance (sensitivity, response & recovery time, repeatability, and selectivity) based on absorbance measurement.
- To discuss the sensing mechanism of gas molecules-sensing layer interaction of tapered optical fiber sensor.

In order to achieve these goals, the following research questions are outlined accordingly:

- What are the semiconductor metal oxides that are sensitive and change its optical properties when interact with H₂ gas?
- How these materials can be synthesized and coated onto tapered optical fiber?
- Which optical measurement can be used to investigate the response of the developed sensors toward H₂ gas?
- How different are the sensing performance with different waist diameter of tapered optical fiber?
- How different are the sensing performance of semiconductor metal oxides with different thicknesses and morphologies?

With these research questions, the investigation focused on a few type of semiconductor metal oxides (SMO) which are known for their sensing properties towards H₂ gas. In addition to that, nanostructured SMO has the ability to produce outstanding gasochromic properties as suggested in literature review. Based on the above reference, the author has developed optical fiber sensors based on manganese dioxide (MnO₂), zinc oxide (ZnO) and molybdenum trioxide (MoO₃) nanostructures incorporate with nobel metal catalyst of palladium (Pd). The nanostructures of these SMO were synthesized and deposited via chemical bath depositon (CBD) technique onto tapered optical fiber which was then coated with very thin layer of Pd (5nm). To the best of the author's knowledge, none of these SMO synthesized via CBD coated on tapered multimode optical fiber for H₂ detection have been reported or published. The optical responses were analysed in term of their gas sensing performance and micro-characterization that has helped the author to understand more on the gas interaction mechanism in optical sensors. The description of the research project is illustrated in Figure 1.1.

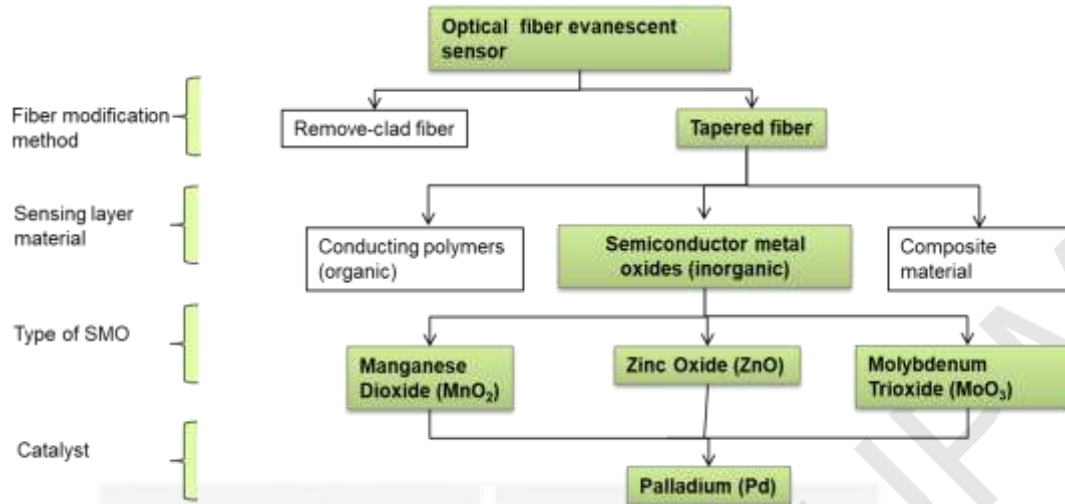


Figure 1.1 : Research design of the project

1.4 Scope of Work and Limitation

In this research project, the author focuses on the synthesis of selected SMO via chemical bath deposition technique onto tapered optical fiber. The dimension profile of waist length is fixed to 10 mm and the up/down taper is 2 mm. This profile is well-established and adequate to provide sensitivity suitable for gas sensing and easy to handle [18]. The deposition parameters are varied to study on the thicknesses and morphologies of the sensing layer. The as-prepared and anneal sensors were prepared to test on their H₂ sensing performance. The investigation was further discussed on the effect of with/without Pd catalyst towards sensing response. The optimum operating temperature for H₂ sensing was also tested so the largest response can be obtained. Sensitivity and repeatability of the fabricated sensors were determined as well as selectivity towards other gases was also measured. Throughout this project, a very thin layer of palladium is DC sputtered on top of the sensing layer. Although the research work focuses on chemical deposition method of SMO that offers better coating and adherent, DC sputtering is used for Pd due to its function is solely as a catalyst and only complementing the actual sensing layer.

1.5 Thesis Organisation

This thesis consists of 6 chapters. Chapter 1 basically touches on the overview of the research work, problem statements, objective and thesis outline. The background of previous works and literature reviews related to the project were discussed and presented in Chapter 2. Methods of developing and fabricating the sensors including how tapered optical fiber was produced as well as SMO synthesization and deposition techniques via

chemical bath were clarified in Chapter 3. Description on equipment used for micro-characterization measurement plus gas sensing setup and measurement were reviewed in Chapter 4. Chapter 5 mainly discusses on the results of each of the sensing layer of SMO sensors fabricated in terms of its micro-characterization and sensing response towards H₂ gas. Finally, Chapter 6 concludes all the work done and summarises some future research suggestion.



REFERENCES

- [1] I. H. Kadhim, H. Abu Hassan, and Q. N. Abdullah, "Hydrogen gas sensor based on nanocrystalline SnO₂ Thin Film grown on bare Si substrates," *Nano-Micro Lett.*, vol. 8, no. 1, pp. 20–28, 2016.
- [2] W. Chomkitichai, H. Ninsonthi, C. Liewhiran, A. Wisitsoraat, S. Sriwichai, and S. Phanichphant, "Flame-Made Pt-Loaded TiO₂ Thin Films and Their Application as H₂ Gas Sensors," vol. 2013, 2013.
- [3] P. Patnaik, *A Comprehensive Guide to the Hazardous Properties of Chemical Substances*. 2007.
- [4] B. Lee, "Review of the present status of optical fiber sensors," vol. 9, pp. 57–79, 2003.
- [5] W. Jin, H. L. Ho, Y. C. Cao, J. Ju, and L. F. Qi, "Optical Fiber Technology Gas detection with micro- and nano-engineered optical fibers q," vol. 19, pp. 741–759, 2013.
- [6] N. A. M. Yahya, M. R. Y. Hamid, S. A. Ibrahim, B. H. Ong, N. A. Rahman, A. R. M. Zain, M. A. Mahdi, and M. H. Yaacob, "H₂ sensor based on tapered optical fiber coated with MnO₂ nanostructures," *Sensors Actuators B Chem.*, vol. 246, pp. 421–427, 2017.
- [7] J. M. Corres, F. J. Arregui, A. Member, I. R. Matias, and S. Member, "Design of Humidity Sensors Based on Tapered Optical Fibers," vol. 24, no. 11, pp. 4329–4336, 2006.
- [8] Z. Yu, L. Jin, L. Sun, J. Li, Y. Ran, and B. O. Guan, "Highly Sensitive Fiber Taper Interferometric Hydrogen Sensors," *IEEE Photonics J.*, vol. 8, no. 1, 2016.
- [9] M. Irigoyen, J. A. Sánchez-Martin, E. Bernabeu, and A. Zamora, "Tapered optical fiber sensor for chemical pollutants detection in seawater," *Meas. Sci. Technol.*, vol. 28, no. 4, p. 45802, 2017.
- [10] H. Latifi, M. I. Zibaii, S. M. Hosseini, and P. Jorge, "Nonadiabatic tapered optical fiber for biosensor applications," *Photonic Sensors*, vol. 2, no. 4, pp. 340–356, 2012.
- [11] H. F. Lima, P. F. Antunes, J. De Lemos Pinto, and R. N. Nogueira, "Simultaneous measurement of strain and temperature with a single fiber Bragg grating written in a tapered optical fiber," *IEEE Sens. J.*, vol. 10, no. 2, pp. 269–273, 2010.
- [12] Y. Tian, W. Wang, N. Wu, X. Zou, and X. Wang, "Tapered optical fiber sensor for label-free detection of biomolecules," *Sensors*, vol. 11, no. 4, pp. 3780–3790, 2011.
- [13] M. Lo, "Tapered optical-fiber-based pressure sensor," pp. 2241–2247, 2013.

- [14] S. Zhu, F. Pang, and T. Wang, "Single-mode tapered optical fiber for temperature sensor based on multimode interference," vol. 8311, p. 83112B, 2011.
- [15] N. Xue, Q. Zhang, S. Zhang, P. Zong, and F. Yang, "Highly Sensitive and Selective Hydrogen Gas Sensor Using the Mesoporous SnO₂ Modified Layers," *Sensors*, vol. 17, no. 10, p. 2351, 2017.
- [16] H. Gu, Z. Wang, and Y. Hu, *Hydrogen Gas Sensors Based on Semiconductor Oxide Nanostructures*, vol. 12, no. 12. 2012.
- [17] C. Wang, L. Yin, L. Zhang, D. Xiang, and R. Gao, "Metal oxide gas sensors: sensitivity and influencing factors.," *Sensors (Basel)*, vol. 10, no. 3, pp. 2088–106, Jan. 2010.
- [18] S. A. Ibrahim, "FEASIBILITY STUDIES OF POLYANILINE NANOSTRUCTURES COATED ON TAPERED OPTICAL FIBER FOR AMMONIA SENSING By SITI AZLIDA BINTI IBRAHIM @ GHAZALI," no. January, 2017.
- [19] "Fiber Optics Communication." [Online]. Available: <http://www.iasscore.in/upsc-prelims/fiber-optic-communication>.
- [20] B&W Tek, "No Title," 2017. [Online]. Available: <http://bwtek.com/spectrometer-part-6-choosing-a-fiber-optic/>.
- [21] B. Renganathan, D. Sastikumar, G. Gobi, N. R. Yogamalar, and A. C. Bose, "Optics & Laser Technology Nanocrystalline ZnO coated fiber optic sensor for ammonia gas detection," *Opt. Laser Technol.*, vol. 43, no. 8, pp. 1398–1404, 2011.
- [22] M. H. Yaacob, "Investigation of Metal Oxide Nanostructured Thin Films Based Optical Hydrogen Sensors," no. March, 2012.
- [23] S. Ko, J. Lee, J. Koo, B. S. Joo, M. Gu, and J. H. Lee, "Chemical wet etching of an optical fiber using a hydrogen fluoride-free solution for a saturable absorber based on the evanescent field interaction," *J. Light. Technol.*, vol. 34, no. 16, pp. 3776–3784, 2016.
- [24] X. Chong, K. J. Kim, P. R. Ohodnicki, E. Li, C. H. Chang, and A. X. Wang, "Ultrashort Near-Infrared Fiber-Optic Sensors for Carbon Dioxide Detection," *IEEE Sens. J.*, vol. 15, no. 9, pp. 5327–5332, 2015.
- [25] Y.-T. Luo, H.-B. Wang, G.-M. Ma, H.-T. Song, C. Li, and J. Jiang, "Research on High Sensitive D-Shaped FBG Hydrogen Sensors in Power Transformer Oil," *Sensors*, vol. 16, no. 10, p. 1641, 2016.
- [26] J. D. Gordon, "Using and Optical D-Fiber as a chemical sensor," no. April, 2007.

- [27] J. Dai, M. Yang, Y. Chen, K. Cao, H. Liao, and P. Zhang, "Side-polished fiber Bragg grating hydrogen sensor with WO₃-Pd composite film as sensing materials," *Opt. Express*, vol. 19, no. 7, p. 6141, 2011.
- [28] M. R. R. Khan and S. W. Kang, "A high sensitivity and wide dynamic range fiber-optic sensor for low-concentration VOC gas detection," *Sensors (Switzerland)*, vol. 14, no. 12, pp. 23321–23336, 2014.
- [29] W. Jin, H. L. Ho, Y. C. Cao, J. Ju, and L. F. Qi, "Optical Fiber Technology Gas detection with micro- and nano-engineered optical fibers q," vol. 19, pp. 741–759, 2013.
- [30] B. Lee, S. Roh, and J. Park, "Current status of micro- and nano-structured optical fiber sensors," *Opt. Fiber Technol.*, vol. 15, no. 3, pp. 209–221, Jun. 2009.
- [31] G. Brambilla, V. Finazzi, and D. Richardson, "Ultra-low-loss optical fiber nanotapers," *Opt. Express*, vol. 12, no. 10, pp. 2258–2263, 2004.
- [32] W. Bin Ji, H. H. Liu, S. C. Tjin, K. K. Chow, and A. Lim, "Ultrahigh Sensitivity Refractive Index Sensor Based on Optical Microfiber," vol. 24, no. 20, pp. 1872–1874, 2012.
- [33] N. E. González-Sierra, L. del C. Gómez-Pavón, G. F. Pérez-Sánchez, A. Luis-Ramos, P. Zaca-Morán, J. M. Muñoz-Pacheco, and F. Chávez-Ramírez, "Tapered optical fiber functionalized with palladium nanoparticles by drop casting and laser radiation for H₂ and volatile organic compounds sensing purposes," *Sensors (Switzerland)*, vol. 17, no. 9, 2017.
- [34] S. Svanberg, *Atomic and Molecular Spectroscopy*. Springer-Verlag Berlin Heidelberg, 2004.
- [35] E. H. Korteb, "IR Specular Reflectance Studies on Rough-Sudan Solids and on Layered Systems with Nonuniform Interfaces: Optical Simulations," vol. 293, pp. 245–248, 1993.
- [36] S. Basu and P. K. Basu, "Nanocrystalline metal oxides for methane sensors: Role of noble metals," *J. Sensors*, vol. 2009, 2009.
- [37] P. Shankar, J. Bosco, and B. Rayappan, "Gas sensing mechanism of metal oxides: The role of ambient atmosphere, type of semiconductor and gases - A review ScienceJet," *Sci. Jet*, vol. 4, no. JANUARY 2015, p. 126, 2015.
- [38] M. H. Yaacob, J. Yu, K. Latham, K. Kalantar-Zadeh, and W. Wlodarski, "Optical hydrogen sensing properties of nanostructured Pd/MoO₃ films," *Sens. Lett.*, vol. 9, no. 1, pp. 16–20, 2011.

- [39] H. Koyanaka, Y. Ueda, K. Takeuchi, and A. I. Kolesnikov, "Effect of crystal structure of manganese dioxide on response for electrolyte of a hydrogen sensor operative at room temperature," *Sensors Actuators B Chem.*, vol. 183, pp. 641–647, Jul. 2013.
- [40] A. Sanger, A. Kumar, A. Kumar, and R. Chandra, "Highly sensitive and selective hydrogen gas sensor using sputtered grown Pd decorated MnO₂ nanowalls," *Sensors Actuators B Chem.*, vol. 234, pp. 8–14, 2016.
- [41] M. Angiola, M. M. Y. A. Alsaif, K. Kalantar-Zadeh, A. Wisitsoraat, W. Wlodarski, and A. Martucci, "Optical hydrogen sensing based on hybrid 2D MoO₃/Au nanoparticles," *Procedia Eng.*, vol. 120, pp. 1141–1144, 2015.
- [42] S. Ozturk, N. Kilinc, and Z. Z. Ozturk, "The effects of annealing on gas sensing properties of ZnO nanorod sensors coated with Pd and Pt," *Procedia Eng.*, vol. 47, pp. 434–437, 2012.
- [43] L. Rajan, C. Periasamy, and V. Sahula, "Comprehensive Study on Electrical and Hydrogen Gas Sensing Characteristics of Pt/ZnO Thin Film Based Schottky Diodes Grown on n-Si Substrates by RF sputtering," *IEEE Trans. Nanotechnol.*, vol. PP, no. 99, pp. 1–1, 2016.
- [44] S. Phanichphant, "Semiconductor Metal Oxides as Hydrogen Gas Sensors," *Procedia Eng.*, vol. 87, pp. 795–802, 2014.
- [45] J. A. Rodriguez, T. Jirsak, and S. Chaturvedi, "Reaction of H₂S with MgO(100) and Cu/MgO(100) surfaces: Band-gap size and chemical reactivity," *J. Chem. Phys.*, vol. 111, no. 17, pp. 8077–8087, 1999.
- [46] R. M. Pasquarelli, D. S. Ginley, and R. O'Hayre, "Solution processing of transparent conductors: From flask to film," *Chem. Soc. Rev.*, vol. 40, no. 11, pp. 5406–5441, 2011.
- [47] C. Kittel and H. Y. Fan, "Introduction to Solid State Physics," *Am. J. Phys.*, vol. 25, no. 5, pp. 330–330, 1957.
- [48] Q. P. Zhang, X. N. Xu, Y. T. Liu, M. Xu, S. H. Deng, Y. Chen, H. Yuan, F. Yu, Y. Huang, K. Zhao, S. Xu, and G. Xiong, "A feasible strategy to balance the crystallinity and specific surface area of metal oxide nanocrystals," *Sci. Rep.*, vol. 7, pp. 1–12, 2017.
- [49] S. Pati, S. B. Majumder, and P. Banerji, "Role of oxygen vacancy in optical and gas sensing characteristics of ZnO thin films," *J. Alloys Compd.*, vol. 541, pp. 376–379, 2012.
- [50] R. Kumar, O. Al-Dossary, G. Kumar, and A. Umar, "Zinc oxide nanostructures for NO₂ gas-sensor applications: A review," *Nano-Micro Lett.*, vol. 7, no. 2, pp. 97–120, 2015.
- [51] J. Lv, W. Gong, K. Huang, J. Zhu, F. Meng, X. Song, and Z. Sun, "Ww W M . Ww W M .," vol. 50, pp. 98–106, 2011.

- [52] K. A. M. Ahmed, "Exploitation of KMnO₄ material as precursors for the fabrication of manganese oxide nanomaterials," *J. Taibah Univ. Sci.*, Aug. 2015.
- [53] M. A. Haque and S. Mahalakshmi, "Effect of Annealing on Structure and Morphology of Cadmium Sulphide Thin Film Prepared by Chemical Bath Deposition," *J. Adv. Phys.*, vol. 3, no. 2, pp. 159–162, 2014.
- [54] A. Dhara, G. Hodes, and S. K. Sarkar, "Two stage chemical bath deposition of MoO₃ nanorod films," *RSC Adv.*, vol. 4, no. 96, pp. 53694–53700, 2014.
- [55] M. Kokotov and G. Hodes, "Reliable chemical bath deposition of ZnO films with controllable morphology from ethanalamine-based solutions using KMnO₄ substrate activation," *J. Mater. Chem.*, vol. 19, no. 23, p. 3847, 2009.
- [56] R. H. Bari and S. B. Patil, "Studies on Spray Pyrolysed Nanostructured SnO₂ Thin Films for H₂ Gas Sensing Application," *Int. Lett. Chem. Phys. Astron.*, vol. 36, pp. 125–141, 2014.
- [57] G. Gasparotto, T. Mazon, G. Gasparotto, M. A. Zaghete, L. A. Perazolli, and J. A. Varela, "Gas Sensor Properties of ZnO Nanorods Grown by Chemical Bath Deposition," *Adv. Mater. Res.*, vol. 975, pp. 189–193, 2014.
- [58] N. Banerjee, S. Roy, C. K. Sarkar, and S. Member, "Pd Modified ZnO Nanorod based High Dynamic Range Hydrogen Sensor," pp. 682–685, 2013.
- [59] M. B. Rahmani, S. H. Keshmiri, J. Yu, A. Z. Sadek, L. Al-Mashat, A. Moafi, K. Latham, Y. X. Li, W. Wlodarski, and K. Kalantar-zadeh, "Gas sensing properties of thermally evaporated lamellar MoO₃," *Sensors Actuators, B Chem.*, vol. 145, no. 1, pp. 13–19, 2010.
- [60] G. Korotcenkov, "Metal oxides for solid-state gas sensors: What determines our choice?," vol. 139, pp. 1–23, 2007.
- [61] M. H. Yaacob, M. Breedon, and W. Wlodarski, "Sensors and Actuators B: Chemical Absorption spectral response of nanotextured WO₃ thin films with Pt catalyst towards H₂ &," vol. 137, pp. 115–120, 2009.
- [62] J. Ou, M. H. Yaacob, J. L. Campbell, K. Kalantar-zadeh, and W. Wlodarski, "H₂ sensing performance of optical fiber coated with nanoplatelet WO₃ film," *Procedia Eng.*, vol. 5, pp. 1204–1207, 2010.
- [63] P. R. Ohodnicki, M. Andio, and C. Wang, "Optical gas sensing responses in transparent conducting oxides with large free carrier density," *J. Appl. Phys.*, vol. 116, no. 2, 2014.

- [64] Q. Yan, S. Tao, and H. Toghiani, "Optical fiber evanescent wave absorption spectrometry of nanocrystalline tin oxide thin films for selective hydrogen sensing in high temperature gas samples.," *Talanta*, vol. 77, no. 3, pp. 953–61, Jan. 2009.
- [65] S. Pishdadian and a. M. Shariati Ghaleno, "Influences of Annealing Temperature on the Optical and Structural Properties of Manganese Oxide Thin Film by Zn Doping from Sol-Gel Technique," *Acta Phys. Pol. A*, vol. 123, no. 4, pp. 741–745, Apr. 2013.
- [66] M. Wang and Y. Feng, "Palladium-silver thin film for hydrogen sensing," *Sensors Actuators, B Chem.*, vol. 123, no. 1, pp. 101–106, 2007.
- [67] H. and K. O. Feng, Q., Kanoh, "Manganese Oxide Porous Crystal," *J. Mater*, vol. 9, pp. 319–33, 1999.
- [68] B. Renganathan and A. R. Ganesan, "Optical Fiber Technology Fiber optic gas sensor with nanocrystalline ZnO," *Opt. Fiber Technol.*, vol. 20, no. 1, pp. 48–52, 2014.
- [69] A. A. Hendi and R. H. Alorainy, "New fabrication of zinc oxide nanostructure thin film gas sensors," *Superlattices Microstruct.*, vol. 66, pp. 23–32, 2014.
- [70] R. Wahab, F. Khan, N. Ahmad, H.-S. Shin, J. Musarrat, and A. A. Al-Khedhairy, "Hydrogen adsorption properties of nano- and microstructures of ZnO," *J. Nanomater.*, vol. 2013, 2013.
- [71] S. Roy, H. Saha, and C. K. Sarkar, "HIGH SENSITIVITY METHANE SENSOR BY CHEMICALLY," vol. 3, no. 4, pp. 605–620, 2010.
- [72] D. E. Motaung, G. H. Mhlongo, I. Kortidis, S. S. Nkosi, G. F. Malgas, B. W. Mwakikunga, S. S. Ray, and G. Kiriakidis, "Structural and optical properties of ZnO nanostructures grown by aerosol spray pyrolysis: Candidates for room temperature methane and hydrogen gas sensing," *Appl. Surf. Sci.*, vol. 279, pp. 142–149, 2013.
- [73] S. N. Das, J. P. Kar, J.-H. Choi, T. Il Lee, K.-J. Moon, and J.-M. Myoung, "Fabrication and Characterization of ZnO Single Nanowire-Based Hydrogen Sensor," *J. Phys. Chem. C*, vol. 114, no. 3, pp. 1689–1693, 2010.
- [74] H. T. Wang, B. S. Kang, F. Ren, L. C. Tien, P. W. Sadik, D. P. Norton, S. J. Pearton, and J. Lin, "Detection of hydrogen at room temperature with catalyst-coated multiple ZnO nanorods," *Appl. Phys. A Mater. Sci. Process.*, vol. 81, no. 6, pp. 1117–1119, 2005.

- [75] P. Pimpang, A. S. Zoolfakar, R. A. Rani, R. A. Kadir, D. Wongratanaphisan, A. Gardchareon, K. Kalantar-zadeh, and S. Choopun, "Hydrogen sensors based on gold nanoclusters assembled onto ZnO nanostructures at low operating temperature," *Ceram. Int.*, vol. 43, pp. S511–S515, 2017.
- [76] A. O. Dikovska, G. B. Atanasova, N. N. Nedyalkov, P. K. Stefanov, P. A. Atanasov, E. I. Karakoleva, and A. T. Andreev, "Optical sensing of ammonia using ZnO nanostructure grown on a side-polished optical-fiber," *Sensors Actuators, B Chem.*, vol. 146, no. 1, pp. 331–336, 2010.
- [77] R. Tabassum and B. D. Gupta, "Surface plasmon resonance-based fiber-optic hydrogen gas sensor utilizing palladium supported zinc oxide multilayers and their nanocomposite," *Appl. Opt.*, vol. 54, no. 5, p. 1032, 2015.
- [78] A. Chithambararaj, N. Rajeswari Yogamalar, and A. C. Bose, "Hydrothermally Synthesized h-MoO₃ and α-MoO₃ Nanocrystals: New Findings on Crystal-Structure-Dependent Charge Transport," *Cryst. Growth Des.*, vol. 16, no. 4, pp. 1984–1995, 2016.
- [79] J. Zhou, N. Lin, L. Wang, K. Zhang, Y. Zhu, and Y. Qian, "Synthesis of hexagonal MoO₃ nanorods and a study of their electrochemical performance as anode materials for lithium-ion batteries," *J. Mater. Chem. A*, vol. 3, no. 14, pp. 7463–7468, 2015.
- [80] A. Chithambararaj, N. S. Sanjini, S. Velmathi, and A. Chandra Bose, "Preparation of h-MoO₃ and α-MoO₃ nanocrystals: comparative study on photocatalytic degradation of methylene blue under visible light irradiation," *Phys. Chem. Chem. Phys.*, vol. 15, no. 35, p. 14761, 2013.
- [81] Y. Xu, L. Xie, Y. Zhang, and X. Cao, "Hydrothermal synthesis of hexagonal MoO₃ and its reversible electrochemical behavior as a cathode for Li-ion batteries," *Electron. Mater. Lett.*, vol. 9, no. 5, pp. 693–696, 2013.
- [82] J. Song, X. Ni, L. Gao, and H. Zheng, "Synthesis of metastable h-MoO₃ by simple chemical precipitation," *Mater. Chem. Phys.*, vol. 102, no. 2–3, pp. 245–248, 2007.
- [83] N. Illyaskutty, H. Kohler, T. Trautmann, M. Schwotzer, and V. P. Mahadevan Pillai, "Hydrogen and ethanol sensing properties of molybdenum oxide nanorods based thin films: Effect of electrode metallization and humid ambience," *Sensors Actuators, B Chem.*, vol. 187, pp. 611–621, 2013.

- [84] H. Shanak, H. Schmitt, J. Nowoczin, and C. Ziebert, "Effect of Pt-catalyst on gasochromic WO₃ films: Optical, electrical and AFM investigations," *Solid State Ionics*, vol. 171, no. 1–2, pp. 99–106, 2004.
- [85] J. S. Noh, J. M. Lee, and W. Lee, "Low-dimensional palladium nanostructures for fast and reliable hydrogen gas detection," *Sensors*, vol. 11, no. 1, pp. 825–851, 2011.
- [86] B. D. Adams and A. Chen, "The role of palladium in a hydrogen economy," *Mater. Today*, vol. 14, no. 6, pp. 282–289, 2011.
- [87] F. D. Manchester, A. San-Martin, and J. M. Pitre, "The H-Pd (hydrogen-palladium) System," *J. Phase Equilibria*, vol. 15, no. 1, pp. 62–83, 1994.
- [88] P. Soundarrajan and F. Schweighardt, "Hydrogen Sensing and Detection," 2009.
- [89] J. Villatoro, D. Luna-Moreno, and D. Monzón-Hernández, "Optical fiber hydrogen sensor for concentrations below the lower explosive limit," *Sensors Actuators, B Chem.*, vol. 110, no. 1, pp. 23–27, 2005.
- [90] A. B. Socorro, I. Del Villar, J. M. Corres, I. R. Matias, and F. J. Arregui, "Lossy mode resonances dependence on the geometry of a tapered monomode optical fiber," *Sensors Actuators, A Phys.*, vol. 180, pp. 25–31, 2012.
- [91] S. A. Ibrahim, N. A. Rahman, M. H. Abu Bakar, S. H. Girei, M. H. Yaacob, H. Ahmad, and M. A. Mahdi, "Room temperature ammonia sensing using tapered multimode fiber coated with polyaniline nanofibers," *Opt. Express*, vol. 23, no. 3, p. 2837, 2015.
- [92] L. F. Koao, F. B. Dejene, and H. C. Swart, "Synthesis of pbs nanostructures by chemical bath deposition method," *Int. J. Electrochem. Sci.*, vol. 9, no. 4, pp. 1747–1757, 2014.
- [93] N. R. Chodankar, G. S. Gund, D. P. Dubal, and C. D. Lokhande, "Alcohol mediated growth of α -MnO₂ thin films from KMnO₄ precursor for high performance supercapacitors," *RSC Adv.*, vol. 4, no. 106, pp. 61503–61513, Nov. 2014.
- [94] W. Song, "Growth and Characterization of Zn_xCd_{1-x}S Films Prepared by Using Chemical Bath Deposition for Photovoltaic Devices," vol. 54, no. 4, pp. 1660–1665, 2009.
- [95] C. D. Lokhande, E. H. Lee, K. D. Jung, and O. S. Joo, "Ammonia-free chemical bath method for deposition of microcrystalline cadmium selenide films," *Mater. Chem. Phys.*, vol. 91, no. 1, pp. 200–204, 2005.

- [96] N. Rathore, D. V. S. Rao, and S. K. Sarkar, "Growth of a polarity controlled ZnO nanorod array on a glass/FTO substrate by chemical bath deposition," *RSC Adv.*, vol. 5, no. 36, pp. 28251–28257, 2015.
- [97] G.-J. Janssen, "Information on the FESEM (Field-emission Scanning Electron Microscope)," *Radboud Univ. Nijmegen*, pp. 1–5, 2015.
- [98] University of California, "Introduction to Energy Dispersive X-ray Spectrometry (EDS)," pp. 1–12, 2015.
- [99] T. A. X. Company, "The multi-purpose solution for your analytical needs The Analytical X-ray Company Pioneering X-ray diffraction," pp. 1–20, 2000.
- [100] T. Thermogravimetric and I. Family, "Thermogravimetric Analysis (TGA) A Beginner's Guide," 1960.
- [101] B. D. Vriezicke, S. Patel, B. E. Davis, and D. P. Birnie, "Evaluation of the Tauc method for optical absorption edge determination: ZnO thin films as a model system," *Phys. Status Solidi*, vol. 252, no. 8, pp. 1700–1710, 2015.
- [102] B. F. Kalantar-Zadeh, Kourosh, *Nanotechnology-Enabled Sensor*. Springer US, 2008.
- [103] V. E. Bochenkov and G. B. Sergeev, "Sensitivity , Selectivity , and Stability of Gas-Sensitive Metal-Oxide Nanostructures," vol. 3, pp. 31–52, 2010.
- [104] A. Aziz, H. N. Lim, S. H. Girei, M. H. Yaacob, M. A. Mahdi, N. M. Huang, and A. Pandikumar, "Sensors and Actuators B: Chemical Silver / graphene nanocomposite-modified optical fiber sensor platform for ethanol detection in water medium," *Sensors Actuators B. Chem.*, vol. 206, pp. 119–125, 2015.
- [105] I. Jiménez, J. Arbiol, G. Dezanneau, A. Cornet, and J. R. Morante, "Crystalline structure, defects and gas sensor response to NO₂ and H₂S of tungsten trioxide nanopowders," *Sensors Actuators, B Chem.*, vol. 93, no. 1–3, pp. 475–485, 2003.
- [106] C. Julien, M. Massot, S. Rangan, M. Lemal, and D. Guyomard, "Study of structural defects in ??-MnO₂ by Raman spectroscopy," *J. Raman Spectrosc.*, vol. 33, no. 4, pp. 223–228, 2002.
- [107] S. M. Baschenko and L. S. Marchenko, "On Raman spectra of water , its structure and dependence on temperature," *Semicond. Physics, Quantum Electron. Optoelectron.*, vol. 14, no. 1, pp. 77–79, 2011.
- [108] Y. Hu, H. Zhu, J. Wang, and Z. Chen, "Synthesis of layered birnessite-type manganese oxide thin films on plastic substrates by chemical bath deposition for flexible transparent supercapacitors," *J. Alloys Compd.*, vol. 509, no. 42, pp. 10234–10240, 2011.

- [109] N. Rajamanickam, P. Ganesan, S. Rajashabala, and K. Ramachandran, "Structural and optical properties of MnO_2 nanowires and MnO_2 nanorods," *AIP Conf. Proc.*, vol. 1591, pp. 267–269, 2014.
- [110] N. Minh, H. Nhat, H. Yi, D. Kim, Y. Han, and M. Kim, "Sensors and Actuators B: Chemical Ni_2O_3 -decorated SnO_2 particulate films for methane gas sensors," *Sensors Actuators B. Chem.*, vol. 192, pp. 327–333, 2014.
- [111] Y. Zhang, L. Zhang, J. Deng, S. Xie, and H. Yang, "Synthesis, Characterization, and Catalytic Properties of MnO_x / SBA-16 for Toluene Oxidation," pp. 154–167.
- [112] H. Shanak and H. Schmitt, "Fast coloration in sputtered gasochromic tungsten oxide films," *Phys. Status Solidi Appl. Mater. Sci.*, vol. 203, no. 15, pp. 3748–3753, 2006.
- [113] J. Z. Ou, M. H. Yaacob, J. L. Campbell, M. Breedon, K. Kalantar-zadeh, and W. Wlodarski, "Sensors and Actuators B: Chemical H₂ sensing performance of optical fiber coated with nano-platelet WO_3 film &," vol. 167, pp. 1–6, 2012.
- [114] J. Z. Ou, M. H. Yaacob, M. Breedon, H. D. Zheng, J. L. Campbell, K. Latham, J. du Plessis, W. Wlodarski, and K. Kalantar-zadeh, "In situ Raman spectroscopy of H₂ interaction with WO_3 films," *Phys. Chem. Chem. Phys.*, vol. 13, no. 16, p. 7330, 2011.
- [115] M. H. Yaacob, A. Z. Sadek, K. Latham, K. Kalantar-zadeh, and W. Wlodarski, "Procedia Chemistry Optical H₂ Sensing Performance of Anodized Nanoporous TiO_2 Thin Films," *PROCHE*, vol. 1, no. 1, pp. 951–954, 2009.
- [116] S. M. Kim, H. Lee, and J. Y. Park, "Charge Transport in Metal/Oxide Interfaces: Genesis and Detection of Hot Electron Flow and Its Role in Heterogeneous Catalysis," *Catal. Letters*, pp. 299–308, 2014.
- [117] S. Bai, T. Guo, D. Li, R. Luo, A. Chen, and C. C. Liu, "Intrinsic sensing properties of the flower-like ZnO nanostructures," *Sensors Actuators, B Chem.*, vol. 182, pp. 747–754, 2013.
- [118] R. Zhang, P. G. Yin, N. Wang, and L. Guo, "Photoluminescence and Raman scattering of ZnO nanorods," *Solid State Sci.*, vol. 11, no. 4, pp. 865–869, 2009.
- [119] R. P. Wang, G. Xu, and P. Jin, "Size dependence of electron-phonon coupling in ZnO nanowires," *Phys. Rev. B*, vol. 69, no. 11, p. 113303, 2004.

- [120] A. H. Moharram, S. A. Mansour, M. A. Hussein, and M. Rashad, "Direct precipitation and characterization of ZnO nanoparticles," *J. Nanomater.*, vol. 2014, 2014.
- [121] I. U. Abhulimen, X. B. Chen, J. L. Morrison, V. K. Rangari, L. Bergman, K. Das, S. Bagheri, K. G. Chandrappa, S. B. A. Hamid, I. E. Characterization, C. References, M. I. Khalil, M. M. Al-Qunaibit, A. M. Al-zahem, J. P. Labis, A. Kołodziejczak-Radzimska, T. Jesionowski, M. Thirumavalavan, K.-L. Huang, and J.-F. Lee, "Synthesis and characterization of ZnO nanoparticles by thermal decomposition of a curcumin zinc complex," *Materials (Basel)*., vol. 7, no. 3, pp. 4198–4212, 2013.
- [122] V. Srikant and D. R. Clarke, "On the optical band gap of zinc oxide," *J. Appl. Phys.*, vol. 83, no. 10, pp. 5447–5451, 1998.
- [123] G. Tang, H. Liu, and W. Zhang, "The variation of optical band gap for ZnO:In films prepared by sol-gel technique," *Adv. Mater. Sci. Eng.*, vol. 2013, pp. 1–5, 2013.
- [124] C. H. Tan, S. T. Tan, H. B. Lee, C. C. Yap, and M. Yahaya, "Growth concentration effect on oxygen vacancy induced band gap narrowing and optical CO gas sensing properties of ZnO nanorods," *AIP Conf. Proc.*, vol. 1784, 2016.
- [125] S. Di and M. Falasconi, "Drift Correction Methods for Gas Chemical Sensors in Artificial Olfaction Systems: Techniques and Challenges," *Adv. Chem. Sensors*, pp. 305–326, 2012.
- [126] A. Arasu and V. Williams, "Effect of Annealing Temperature on the Dispersion Energy Parameters of Sol-Gel Routed Molybdenum Oxide Thin Film," vol. 22, no. 4, pp. 1–8, 2015.
- [127] G. Huyberechts and P. Szeco, "Simultaneous quantification of carbon monoxide and methane in humid air using a sensor array and an artificial neural network," vol. 45, pp. 123–130, 1997.
- [128] Y. Li, H. Yu, X. Huang, Z. Wu, and M. Chen, "A simple synthesis method to prepare a molybdenum oxide hole-transporting layer for efficient polymer solar cells," *RSC Adv.*, vol. 7, no. 13, pp. 7890–7900, 2017.
- [129] A. Chithambararaj and A. C. Bose, "Hydrothermal synthesis of molybdenum oxide microbelts," *AIP Conf. Proc.*, vol. 1447, no. 1, pp. 311–312, 2012.
- [130] V. A. Doss, A. Chithambararaj, and A. C. Bose, "Effect of reaction atmosphere on structural and optical properties of hexagonal molybdenum oxide (h-MoO₃)," vol. 50049, p. 50049, 2016.

- [131] J. Hamagami, Y. Oh, Y. Watanabe, and M. Takata, "Preparation and characterization of an optically detectable sensor consisting of Pd/MoO₃ thin films," vol. 14, pp. 281–283, 1993.

