

Microstructure, Optical and Dielectric properties of Cerium oxide Thin films Prepared by Pulsed laser deposition

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Abstract

Cerium oxide (CeO₂) thin films were deposited on Pt (111)/Ti/SiO₂/Si(100) substrates using pulsed laser deposition (PLD) method at different temperatures such as, 300 K, 573 K and 873 K with 3×10^{-2} mbar oxygen partial pressure. The prepared films were systematically investigated using X-ray diffraction (XRD), atomic force microscopy (AFM), photoluminescence (PL) and electrical measurement system. XRD analysis clearly shows improved crystallinity of CeO₂ films prepared at 573 and 873 K substrate temperatures. The AFM analysis indicated the uniform distribution of the nanocrystallites and dense structure with the roughness (RMS) of ~ 2.1-3.6 nm. The photoluminescence (PL) studies of the films showed a broad peak at ~ 366-368 nm, indicating the optical bandgap of 3.37-3.38 eV. The electrical property study showed minimum leakage current density of 2.0×10^{-7} A/cm² at 873 K, which was measured at 100 kV and this value was much lower than that of the CeO₂ film deposited at 300 K. The dielectric constants are increased and dielectric loss values decreased for the films with increasing substrate temperature.

Key words: X-Ray Diffraction, Thin films, Cerium oxide, Dielectric properties, Pulsed laser deposition, Electrical properties, Photoluminescence and Atomic force microscopy

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1. Introduction

High dielectric materials (high- k) are considered as a potential in the fabrication of electronic devices. There are different dielectric materials currently being used in metal oxide semiconductor field-effect transistors (MOSFETs), as gate oxide materials, as capacitor in memory devices and insulator in back-end interconnects [1,2]. The silicon dioxide (SiO_2) is highly compatible with silicon technology and exhibits high interface quality. However, direct tunneling of charge carriers occurs via the potential wall as the SiO_2 layer thickness decreases, which in turn leads to increase in leakage current through gate dielectrics [3]. For thin SiO_2 film, the leakage current from electron tunneling through the dielectrics is a problem [4].

There are series of important high- k oxide materials, such as cerium oxide (CeO_2) [5,6], hafnium oxide (HfO_2) [7], zirconium oxide (ZrO_2) [8] and yttria stabilized zirconium oxide (YSZ) [9] are strongly recommended to replace SiO_2 . Among all, CeO_2 is emerging as highly unique dielectric material due to its superior physical properties with chemical and thermal stability [10-14]. Typically, the CeO_2 exhibits cubic structure having lattice parameter $a = 0.541$ nm, which matches with silicon lattice parameter $a = 0.543$ nm. This lattice matching property is highly interesting for epitaxial growth. These properties play an important role in the MIS, MIM and MOS capacitor applications [15-18]. In addition, using CeO_2 thin films wide variety of applications, which includes ultra-thin gate oxide in CMOS devices, capacitors, buffer layers, optical devices, electro-chromic devices, oxygen sensors, fuel cells and corrosion resistant coatings were proposed. The unique physical properties of CeO_2 , which include high dielectric constant, wide bandgap and low interface state density makes it as a potential candidate in gate dielectric applications [19-21].

CeO_2 films are deposited using spray pyrolysis [25,26], sol-gel [27,28], thermal evaporation [29], electron-beam evaporation [30], RF magnetron sputtering [31] and pulsed laser deposition (PLD) techniques [11-14]. PLD is highly unique to produce high quality

1 films in broad range of deposition conditions, such as substrate temperature, ambient gas,
2 pressure of the gases, laser energy and repetition rate [32]. All these parameters can be
3 effectively monitored to obtain the desired properties from thin films. In this research, we
4 have used PLD technique to deposit CeO₂ films on the Pt coated substrate Pt (111)/Ti/SiO₂/Si
5 (100), which is crucial for the electrical and dielectric property analysis. We have deposited
6 CeO₂ films at different temperatures to investigate the microstructural, optical and electrical
7 quality of the films. We have used to take advantage of the deposited thin film to measure
8 electrical and dielectric properties through metal-insulator–metal structure.
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19 **2. Materials and Methods**

21 CeO₂ powder with 99.99 % purity was compacted using uni-axial press by applying
22 120 bar pressure and the pellet size was maintained at a specification of 25 mm dia x 4 mm
23 thick. The compact green pellet was sintered at 1473 K for 6 hours to improve the packing
24 density ~90%. Prior to deposition, the substrates were subjected to ultrasonic cleaning with
25 acetone and dried under N₂ gas flow. The experiments were carried out using KrF excimer
26 laser ($\lambda \sim 248$ nm) and 300 mJ energy with 10 Hz repetition rate. The spacing between the
27 substrate and target was kept at 4.5 cm with the optimized deposition time of 1 hour [11-14].
28 The films were prepared under 2×10^{-2} mbar oxygen partial pressure and 300 K, 573 K and
29 873 K temperatures. Thickness of the films were measured from the height difference
30 between coated and uncoated region. Dektak profilometer (DEKTAK 6M-stylus profiler
31 Veeco, USA) was used for the thickness measurement. The crystallinity of the films were as
32 analyzed using X-ray diffractometer (X'pert PW 3040 D-8, PANalytical) attached with point
33 detector using CuK α_1 radiation. XRD pattern was recorded between 20-90⁰ and a step size of
34 0.2^o with a continuous scanning mode. The Scherrer formula is used to evaluate the
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$$58 \quad D = \frac{K\lambda}{59 \quad \beta \cos \theta}$$

1 where $\beta = \sqrt{(B^2 - b^2)}$ K is Scherrer constant, λ is the wavelength of incident radiation, ,
2 D-Crystallite size, β is Full width at half maximum (FWHM), θ -Angle of diffraction, B-
3 Observed FWHM of the film and b-Instrumental broadening. The photoluminescence spectra
4 of the films were analysed using (Shimadzu, RF-5301PC) spectrofluorophotometer. The
5 surface morphology and surface roughness of the films were analyzed using Atomic force
6 microscope (AFM) (XE-100 Park systems) in contact mode. Dielectric properties of all
7 films were measured using low frequency impedance analyser (4192A; HP, USA).
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10 **3 Results and Discussion**

11 **3.1 Microstructural analysis**

12 The crystalline structure of the CeO₂ films is investigated by XRD studies. Figure 1
13 shows the XRD pattern of the CeO₂ thinfilms prepared at different temperatures. The CeO₂
14 films exhibit cubic structure without any impurity phases. CeO₂ film prepared at 300 K
15 shows low intense peaks at angles of 28° and 32° corresponding to (111) and (200)
16 reflections from the cubic CeO₂ (JCPDS No. 34-0394). The films deposited at 573 and 873
17 K show the high intense peaks corresponding to (111) reflection, which indicated the
18 improved crystallinity at higher temperature. Furthermore, the (111) reflection of the films
19 deposited at 573 and 873 K temperatures is quite sharp with decreased FWHM as
20 compared to CeO₂ film deposited at 300K. The decrease of FWHM and increase of intensity
21 revealed increase in crystallite size caused due to the ad-atom mobility, which in turn leads to
22 the densification of CeO₂ films. The calculated crystallite size is 19-22 nm range [11-13]. The
23 high intense peaks at 40°, 69° and 86° denotes the Pt peaks corresponding to (111), (220) and
24 (222) respectively. Many researchers worked on CeO₂ thinfilms for different applications.
25 CeO₂ films were deposited on Si (100) and glass substrates under various oxygen partial
26 pressures and substrate temperatures using PLD [11,12] for optical applications. The results
27 indicated the polycrystalline cubic nature of the films deposited at various oxygen partial
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1 pressures. CeO₂ films deposited at various substrate temperatures (RT -973 K) and the XRD
2 analysis indicated the polycrystalline cubic structure with (111) orientation at room
3 temperature, while the orientation changed to (200) at higher temperatures [11]. Kanakaraju
4 *et al.* [26] prepared the CeO₂ films using ion beam sputtering and structural studies indicated
5 (220) orientation up to 473 K and it changed to (200) orientation at 673 K. Murugan *et al*
6 deposited CeO₂ films at 3x10⁻⁴ mbar pressure using RF magnetron sputtering. XRD studies
7 indicated cubic structure of the films [33].
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17 AFM studies showed the images of the CeO₂ films deposited at different temperature
18 on Pt coated Si substrates. Fig. 2a shows the surface morphology of CeO₂ film deposited at
19 300 K shows uniform distribution of vertically grown crystallites having dense
20 microstructure over the entire substrate with good crystallinity. The CeO₂ film prepared at
21 573 K (Fig. 2b) also shows similar surface morphology as observed in the film deposited at
22 300 K. However, the crystallite size significantly increased for the CeO₂ film prepared at 573
23 K. On further increasing the deposition temperature to 873 K coalescence of crystallites is
24 observed as shown Fig. 2c. The root mean square (RMS) surface roughness values were
25 quantitatively measured using the AFM images and shown in Table 1. RMS surface
26 roughness increased as 2.09 nm, 2.68 nm and 3.62 nm at 300, 573 and 873 K temperatures
27 respectively. The surface roughness increased due to increased crystallite size with the
28 increasing substrate temperature. The increased crystallite size with increasing temperature is
29 well correlated with the decrease of FWHM as observed in XRD study [11].
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49 **3.2 Optical characterization**

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51 Figure 3 shows the photoluminescence (PL) spectra of CeO₂ thinfilms deposited at
52 various temperatures. All thin films are excited at 325 nm wavelength using He-Cd laser.
53 PL spectra indicated a strong peak ~366 nm. The peak position is ~366-368 nm wavelength,
54 corresponding to the bandgap of ~3.38- 3.37 eV. The measured bandgap values well agreed
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1 with the bandgap of CeO₂ reported in literature [12]. CeO₂ film prepared at room
2 temperature (300 K) shows higher intensity and PL intensity is decreased at higher
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4 temperatures (573 K and 873 K) [13,30]. The PL studies revealed the energy gap of 3.05-
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6 3.10 eV range[13].
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10 **3.3 Electrical and Dielectric properties**

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15 MIM capacitor structure is formed and its electrical properties are studied. Platinum
16 electrodes with $1.54 \times 10^{-4} \text{ cm}^2$ area was deposited on the surfaces of the CeO₂ films to form a
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18 capacitor structure. Figure 4 shows the schematic of the MIM structure. The variation of
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20 leakage current densities (J) of CeO₂ films with respect to electric fields (E) are shown in Fig.
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22 5 as a function of temperature. The measured leakage current densities were $2.40 \times 10^{-5} \text{ A/cm}^2$
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25 (300 K), $1.8 \times 10^{-6} \text{ A/cm}^2$ (573 K) and $2.0 \times 10^{-7} \text{ A/cm}^2$ (873 K), respectively at 100 KV/cm.
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31 The leakage current density of the CeO₂ thinfilm deposited at 873 K, indicates the
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33 minimum leakage current density and was two orders magnitude lower than the CeO₂ film
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35 deposited at 300 K. The high electrical leakage of the CeO₂ thin film deposited at room
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37 temperature might be attributed to the poor crystallinity. The improved crystallinity with
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39 better microstructure caused low leakage current density of the CeO₂ films prepared at 573
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41 and 873 K, respectively [1,2,5,16]. In dielectric thin films, the oxygen vacancies are act as
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43 mobile carriers for the large conduction. The dielectric thin films with large concentration of
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45 oxygen vacancies lead high electrical leakage current. The low leakage current density at 873
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48 K, indicating the reduced oxygen vacancies in the film.
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53 Frequency dependent dielectric constant was measured by varying frequencies from
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55 10^2 to 10^6 Hz (Fig.6.). The dielectric constant values (ϵ) for the CeO₂ films were 6 (300 K), 8
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57 (573 K) and 10 (873 K), respectively at 1 kHz frequency. The observed dielectric loss values
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59 were 1.13 (300 K), 1.07 (573 K), and 1.03 (873 K), respectively (Fig.6) at 1 kHz. The
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1 discrepancies in the dielectric constants and dielectric loss in these films might be due to the
2 ionic vacancies and microstructural features, such as crystallite size and porosity [7].
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5 Jung-Ho Yoo et al. [4] prepared CeO₂ films by reactive DC magnetron sputtering and
6 found the polycrystalline cubic structure of the films and these studies indicted the maximum
7 capacitance (C_{max}) and reduced leakage current with increasing interfacial SiO layer
8 thickness. CeO₂ films were prepared on Si (100) substrates by PLD method and formed a
9 MOS capacitor (Pt/Si/CeO₂/Pt) and were found to have a dielectric constant of 23 [16]. CeO₂
10 films were prepared on silicon at various temperatures using atomic layer deposition
11 technique [20]. The as-deposited films were cubic structure for 423 K-623 K temperatures
12 range. The permittivities were found to be 25-42 range at 1 MHz [20]. The CeO₂ films
13 prepared on Si (100) using electron beam evaporation at 473 K to form a MIS configuration
14 and the $C-V$ plots showed dielectric constant value of 3.4 [23]. Anil G Khairnar et al [33]
15 prepared CeO₂ films of thickness 5-56 nm using sol-gel spin coating technique and the C-V
16 measurement indicated the dielectric constant of 18.92 at 1 MHz.
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34 El-Nahass et al investigated the structural, electrical and dielectric behavior of bulk
35 CeO₂. XRD results revealed the cubic structure of CeO₂ powder. The conductivity of the
36 charge carriers was analyzed using electrical modulus formulism [36]. Takenori Yamamoto
37 et al. [37] theoretically studied the dielectric properties of CeO₂ using first-principles. The
38 dielectric constant values of the films were found to increase with lattice expansion and
39 decrease of oxygen vacancy. The dielectric constants values of CeO₂ calculated from the
40 first-principles are in agreement with our experimental results. These results indicated that
41 the dielectric properties of the films with frequency are strongly dependent on preparation
42 method, microstructure and crystallite size, controlled by the growth temperature.
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4 Conclusions

CeO₂ thinfilms were prepared on Pt (111)/Ti/SiO₂/Si (100) substrates at different temperatures using PLD. XRD studies revealed the formation of polycrystalline ceria of cubic structure with (111), (200) and (222) orientation. The AFM analysis confirmed the nanocrystalline nature of the films and improved crystallinity with increasing temperatures. RMS surface roughness of the films is ~ 2.1-3.6 nm. The PL studies indicated the bandgap of ~ 3.38 eV. The dielectric constant of CeO₂ films was found to increase with increasing substrate temperature, while the dielectric loss was found decreasing with increase of temperature. The low leakage current density of 2.0×10^{-7} A/cm² was obtained at 100 KV at the 873K temperature. This is due to the lattice expansion, improved crystalline nature and decrease of oxygen vacancy formation as the substrate temperature increased.

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8 **Figure captions**
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11 Fig.1 XRD patterns of the CeO₂ thin films deposited at 300, 573 and 873K on Pt
12 (111)/Ti/SiO₂/Si (100) substrates.
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15 Fig. 2 3D and 2D AFM images of CeO₂ films deposited at different substrate temperature (a)
16 300 K (b) 573 K (c) 873 K.
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20 Fig. 3 PL spectra of the CeO₂ films deposited at different substrate temperature.
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23 Fig. 4 Schematic diagram of the deposited MIM Capacitor structure
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26 Fig.5 Effect of applied electric field on Leakage current density of CeO₂ films
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29 Fig. 6 Frequency dependent dielectric properties of the CeO₂ thin films deposited at different
30 temperature.
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Table captions

Table 1 Crystallite size, RMS roughness and Dielectric constant of the CeO₂ films deposited at different substrate temperature.

S. No	Substrate temperature	Oxygen partial pressure (mbar)	Thickness (nm)	Crystallite size (nm)	RMS roughness (nm)	Bandgap (eV)	Dielectric constant
1	300 K	2×10^{-2}	300	-	2.1	3.37	6
2	573 K	2×10^{-2}	300	19	2.7	3.38	8
3	873 K	2×10^{-2}	300	22	3.6	3.37	10

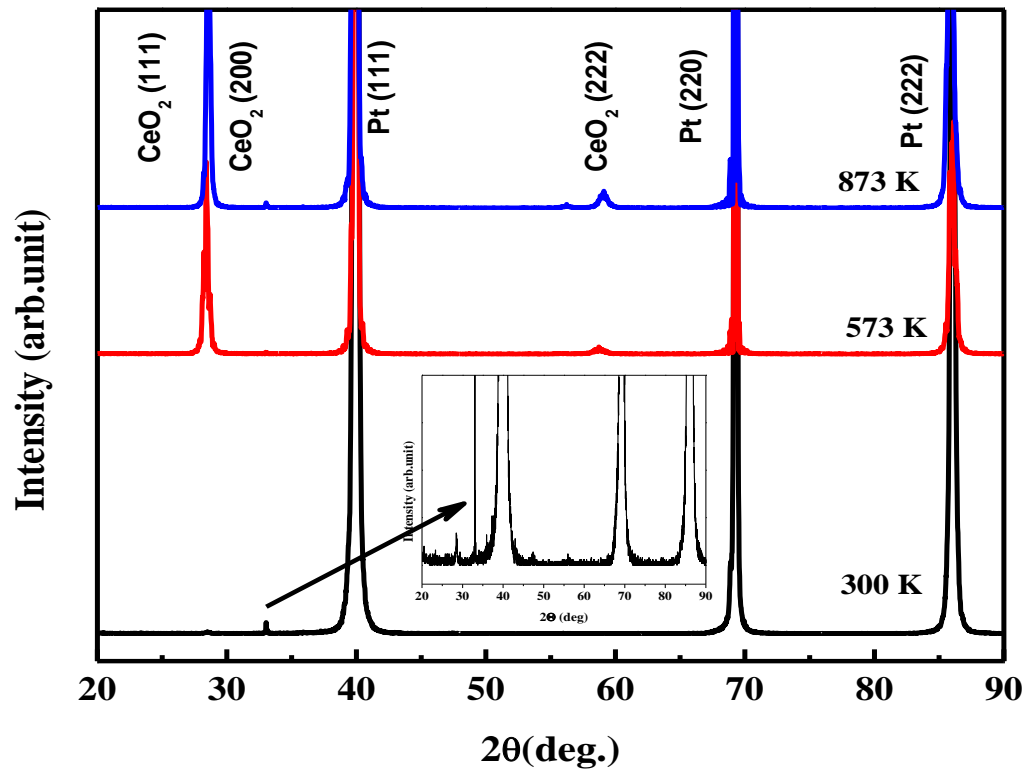
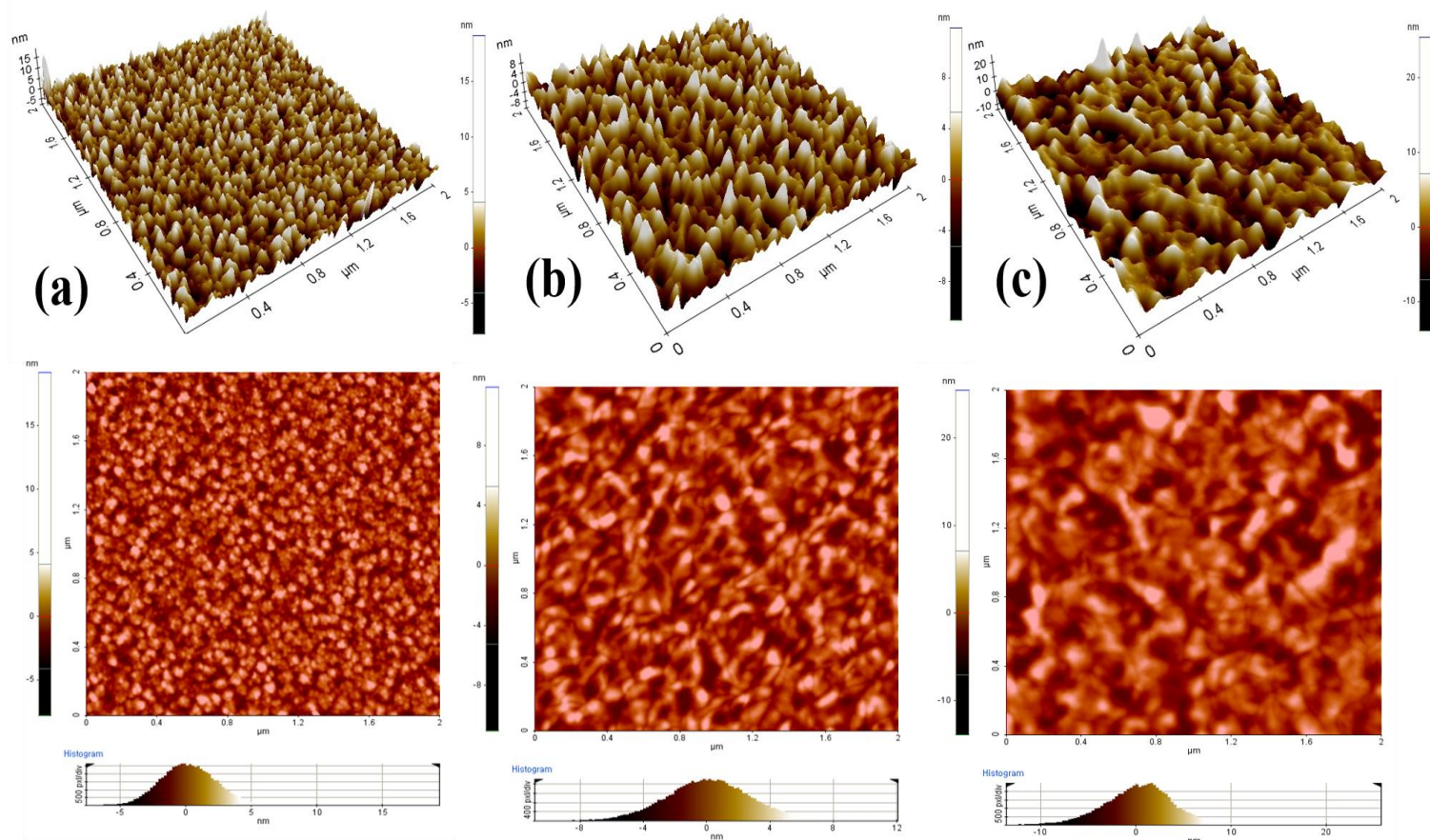


Fig.1 XRD pattern of the films deposited at various substrate temperatures



**Fig.2 3D and 2D AFM images of CeO_2 films deposited at different substrate temperature
(a) 300 K (b) 573 K (c) 873 K**

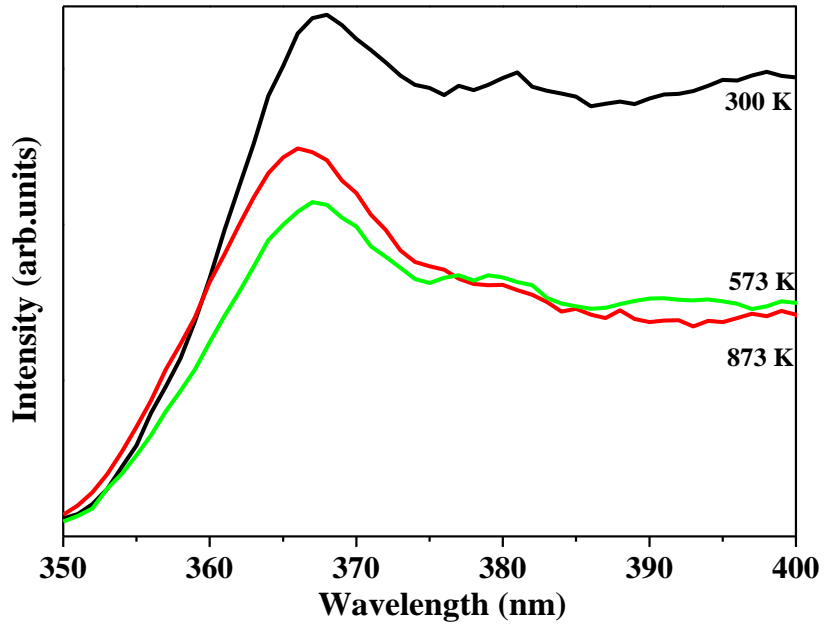


Fig.3 PL spectra of the CeO₂ films deposited at different substrate temperature

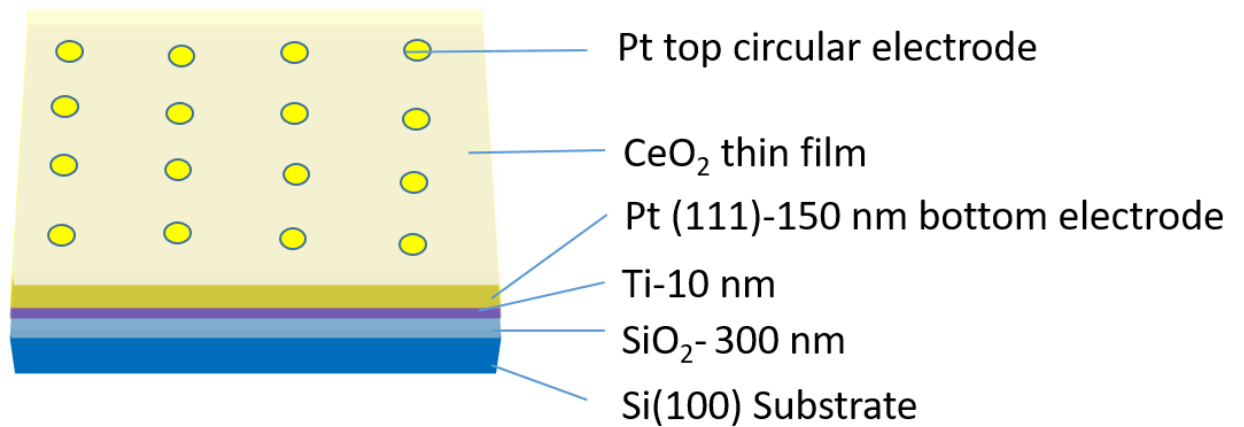


Fig.4 Schematic diagram of the device structure

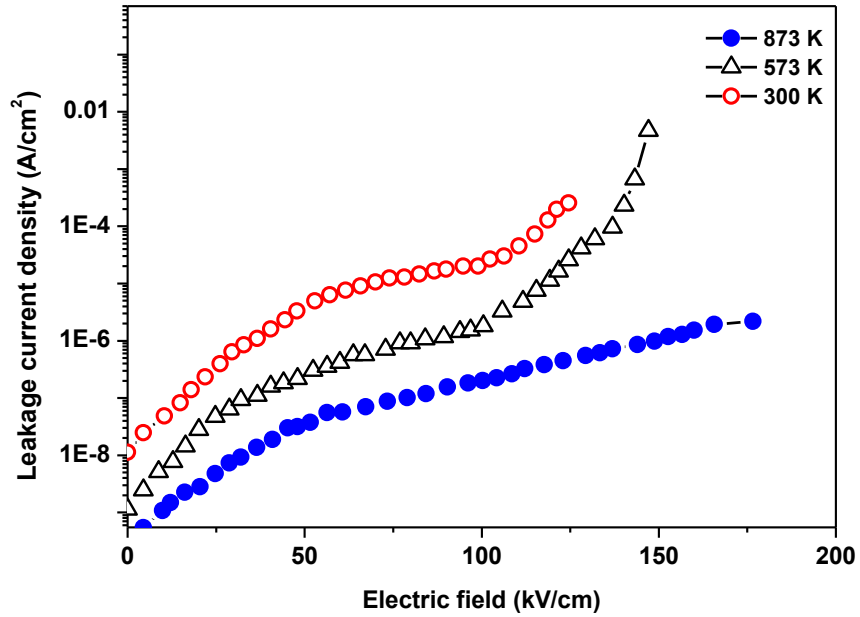


Fig.5 Effect of applied electric field on Leakage current density of CeO₂ films

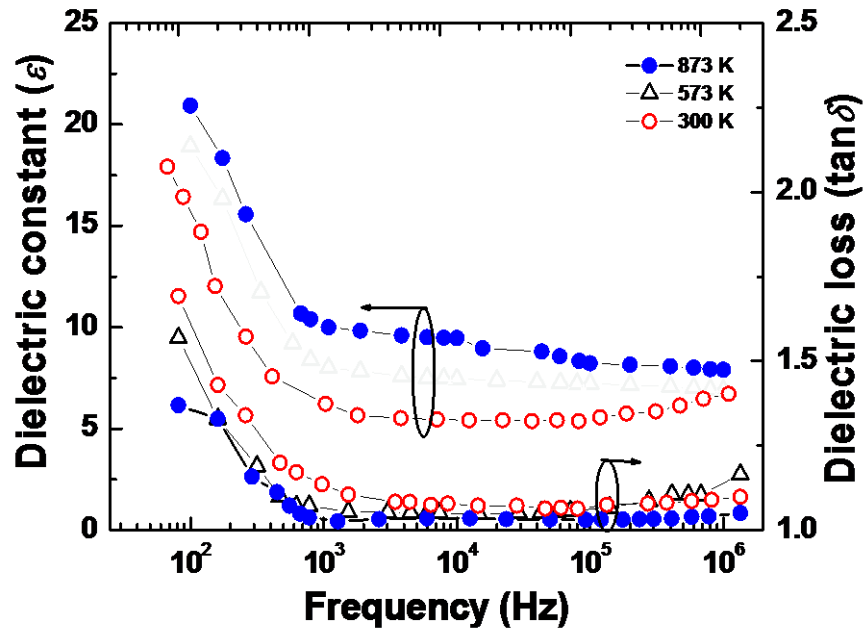


Fig.6 Frequency dependent dielectric properties of the CeO₂ thin films deposited at different temperature.