Differential Gene Regulation in the Ag Nanoparticle and Ag<sup>+</sup>-induced
 Silver Stress Response in *Escherichia coli*: a Full Transcriptomic Profile

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## JONATHAN S. MCQUILLAN and ANDREW M. SHAW

Biosciences, College of Life and Environmental Sciences, University of Exeter, EX4
4QD, UK

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8 Correspondence: Dr Andrew M. Shaw, Biosciences, College of Life and Environmental
9 Sciences, University of Exeter, Exeter, EX4 4QD, UK. Email: <u>andrew.m.shaw@exeter.ac.uk</u>
10 Phone: +441392723495. Fax: +44139272263434.

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### 12 Key words

13 Microarray, Silver, Nanoparticle, Toxicity, Mechanism

14

### 15 Abstract

We report the whole-transcriptome response of Escherichia coli bacteria to acute 16 treatment with silver nanoparticles (AgNPs) or silver ions (Ag<sup>+</sup>) as silver nitrate using 17 gene expression microarrays. In total, 188 genes were regulated by both silver 18 treatments, 161 were up-regulated and 27 were down-regulated. Significant regulation 19 was observed for heat shock response genes in line with protein denaturation associated 20 with protein structure vulnerability indicating Ag<sup>+</sup>-labile –SH bonds. Disruption to 21 iron-sulfur clusters led to the positive regulation of iron-sulfur assembly systems and 22 the expression of genes for iron and sulphate homeostasis. Further, Ag ions induced a 23 24 redox stress response associated with large (>600-fold) up-regulation of the E. coli soxS transcriptional regulator gene. Ag<sup>+</sup> is isoelectronic with Cu<sup>+</sup>, and genes associated with 25 26 copper homeostasis were positively regulated indicating Ag<sup>+</sup>-activation of copper signalling. Differential gene expression was observed for the silver nitrate and AgNP 27 silver delivery. Nanoparticle delivery of Ag<sup>+</sup> induced the differential regulation of 379 28 genes; 309 genes were uniquely regulated by silver nanoparticles and 70 genes were 29 uniquely regulated by silver nitrate. The differential silver nanoparticle-silver nitrate 30 response indicates that the toxic effect of labile Ag<sup>+</sup> in the system depends upon the 31 mechanism of delivery to the target cell. 32

#### 1 Introduction

Anti-microbial silver (Ag) is increasingly prevalent in the clinic and in general 2 healthcare (Lansdown, 2006). Specifically, novel silver nanoparticles (AgNPs) are effective 3 broad-spectrum agents that are added to wound dressings, and hygiene products. Their 4 5 antimicrobial effects are enhanced by a large surface area favouring a high rate of dissolution 6 and release of Ag ions (oxidation of Ag(0) and release of Ag(I)). Dissolved Ag(I) can interact 7 with sulphur- and nitrogen-containing compounds which include protein amino acid side chains (Bauman and Wang, 1964, Vickery and Leavenworth, 1930, Clement and Jarrett, 8 9 1994, Bell and Kramer, 1999) and metabolically essential iron-sulfur clusters (Fe-S). Thus, 10 the protein targets are potentially pan-metabolic.

Bacteria respond to the dissolved Ag(I) by producing small metal-binding proteins 11 12 that sequester the silver and membrane transporters that remove it from the cytosol. This was first reported for a silver resistant strain of Salmonella typhimurium which contained a cluster 13 14 of plasmid-borne genes encoding dual silver ion exporters and a small soluble silver ion binding protein under the control of a 2-component (Ag(I) sensor- transcriptional responder) 15 16 signalling system (Gupta, 1999). Orthologues in other species including the enterohaemorragic pathogen Escherichia coli (Franke, 2001) perform similar roles. The E. 17 18 coli cus (Cu sensitivity) regulon encodes an RND (Resistance-Nodulation-cell Divison) 19 family Ag(I)/Cu(I) exporter (CusCBA) and a small Ag(I)/Cu(I) binding protein (CusF) 20 (Kittleson et al., 2006, Franke et al., 2003). The genes are over-expressed in silver resistant strains (Lok et al., 2008a) and inactivation in the wild-type is consistent with increased 21 sensitivity (Franke, 2001). The association with copper is logical as the Ag(I) and Cu(I) ions 22 have the same  $d^{10}$  electron configuration, similar charge and ionic radii. However, there is no 23 evidence to suggest that a second copper export system in E. coli, CopA, has any effect on 24 silver tolerance. Silver resistant strains of E. coli raised in the laboratory lack a sub-set of 25 constitutive outer membrane Porin proteins, OmpC and OmpF, indicating a chemiosmotic 26 27 defence, but gene knockout mutants had no detectable sensitivity compared to the wild type 28 (Li et al., 1997).

Previous studies have addressed the role of a limited sub-set of *E. coli* genes in response to Ag(I) in solution but the potentially pan-metabolism action of Ag(I) on proteins alludes to large-scale genetic regulation. For AgNPs, the toxic mechanism may be enhanced by association of the nanoparticle and bacterial surface and the subsequent localised dissolution and ion release directly against the cell wall. In our earlier study, we reported that the AgNP toxicity mechanism induces a quantitatively greater transcriptional response to

1 silver stress than Ag(I) added as silver nitrate, even though the measured bulk solution phase Ag(I) concentration was the same. This study was restricted to a sub-set of E. coli Ag-2 responding genes but differences in the global genetic response were not investigated 3 (McQuillan et al., 2012). In eukaryotes, including Saccharomyces cerevisae, a differential 4 5 dissolved Ag(I)-AgNP response has been measured using microarrays (Niazi et al., 2011, Kawata et al., 2009, Roh et al., 2009, Lim et al., 2012) but to our knowledge, these 6 7 experiments have never been performed in prokaryotes, which are clearly an important target group. In this study, we report the findings from whole transcriptome gene expression 8 9 microarray experiments to capture the overall genetic response to (a) 142 nm AgNPs and (b) AgNO<sub>3</sub> in the Gram negative bacterium, E. coli K12. The response was measured at the early 10 stage 10 minutes following silver shock at a sub-inhibitory dose to reduce background gene 11 12 regulation from secondary effects including a change in growth phase. Genes that responded to both treatments are described in terms of the response to dissolved Ag(I), the common 13 14 toxicant, and we report on genes that responded differentially in the two treatments reflecting differences in the mechanism of action for the two physical forms of Ag. 15

#### 1 Methods

#### 2 Materials

3 All reagents were purchased from Sigma-Aldrich unless otherwise stated. The Silver Nanoparticles (AgNPs) were synthesised in the vapour phase (QinetiQ Nanomaterials Ltd, 4 5 Farnborough, UK) and supplied as a dry powder. The mean equivalent spherical diameter 6 was  $142 \pm 20$  nm (mean  $\pm$  standard error of the mean), determined in transmission electron 7 images after dispersion in the experimental medium (low-salt Luria broth as defined below) using a JEOL 1400 TEM. The specific surface area was determined by BET adsorption 8 9 isotherm and was 4  $m^2/g$ . Scanning Electron Microscopy with Energy dispersive X-ray (EDX) analysis was carried out to confirm that the nanoparticles were silver with no other 10 elements detected (HITACHI S3200N SEM fitted with EDAX detection; INCA, Oxford 11 12 Instruments). The characterisation data including an assessment of the antibacterial activity of this specific material batch has been determined previously (McQuillan et al., 2012). 13

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### 15 Bacterial Culture and Ag Treatment

16 E. coli K12 (MG1655) was received from the Coli Genetic Stock Centre and maintained on Luria agar at 37°C. All cultures were carried out in a low-salt Luria broth (LB) 17 which was 10 g Tryptone and 5 g yeast extract in 1 L of water and pH 7.5. The salt was 18 omitted as this improved the colloidal stability of the AgNPs and reduced precipitation of 19 20 AgCl, but the medium still supported rapid growth and replication of the E. coli. Nanoparticle dispersion in the LB was achieved by sonicating the mixture for 2 minutes using a Soniprep 21 150 (MSE Instruments, London, UK). The bacterial growth curve was determined for a 100 22 mL culture, under aerobic conditions at 37°C with constant agitation. Viable cell numbers 23 were measured at 30 minute intervals using the plate counting method. For AgNP treatment 24 the dry nanopowder was dispersed in 10 mL of the LB by sonication at  $10 \times$  the required 25 concentration, then diluted to 100 mL with a log-phase culture of the E. coli containing 10<sup>7</sup> 26 27 cfu/mL. For silver nitrate we used an identical procedure; log-phase cultures were diluted 28 with fresh medium containing a  $10\times$  concentrated solution of AgNO<sub>3</sub>. The exposure 29 concentration was 40 µg/mL for AgNPs or 0.4 µg.mL for Ag(I) as AgNO<sub>3</sub>. Control cultures 30 were similarly diluted at the time of exposure and each Ag treatment was performed using 31 quadruplicate treated/untreated control pairs.

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33 Microarray Experiments

1 RNA was stabilised and isolated from each culture using the RNAprotect Reagent with the RNeasy Mini Kit (Qiagen, Crawley, UK). Residual DNA was digested with RQ1 2 RNase free DNase (Promega, Southampton, UK). The complete removal of the DNA was 3 confirmed by a null result in a Taq polymerase-based PCR from the samples, wherein Taq is 4 5 a DNA-specific polymerase and cannot amplify from an RNA template.. The RNA was 6 purified using the RNeasy clean-up protocol and analysed by agarose electrophoresis and spectrophotometry. High quality RNA was amplified, reverse transcribed and labelled (Cy3 7 for treated and Cy5 for untreated, including at least one dye swap) using the MessageAmp-II 8 9 Bacteria Kit (Applied Biosystems, Warrington, UK). Hybridisation was carried out according to the instructions of the microarray manufacturer (Agilent Technologies, USA). Briefly, the 10 labelled probes were mixed with fragmentation and blocking buffer at 60°C for 30 minutes. 11 12 The fragmentation reaction was terminated by mixing (1:1) with a hybridisation buffer containing 25% formamide, 5× Saline Sodium Citrate, 0.1% Sodium Dodecyl Sulfate and 1 13 14 % salmon sperm DNA. Then, 40 µL of hybridization sample was loaded onto each array using the SureHyb assembly apparatus. For each Ag treatment, the quadruplicate 15 16 treated/untreated control pairs were hybridised with quadruplicate gene expression microarrays (Product G4813A-020097) which were printed on glass in an 8 by 15,000 feature 17 18 format. The hybridization reaction was carried out in a rotisserie oven at 65°C for 17 hours. 19 The array was washed with Agilent gene expression wash buffers in a 1 L staining dish that 20 had been cleaned with acetonitrile and ultrapure water. All steps were carried out in an ozone 21 controlled environment.

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### 23 Data Analysis

The microarray slide was scanned on a GenePix 4000B array scanner (Molecular 24 Devices, USA) and feature extraction was carried out with Agilent Feature Extraction 25 software (Agilent Technologies, USA). Defective spots were excluded and the dye intensity 26 27 for each spot was normalised using local background subtraction. Overall normalisation of 28 dye intensity bias was performed using the global 'within array' LOWESS method. Gene expression filtering and statistical analysis was carried out using GeneSpring (Agilent 29 Technologies, USA). Genes were filtered by excluding those whose expression failed to 30 change by more than 2-fold. The remaining genes were subject to confidence testing using 31 the t-test with correction for multiple testing using the Benjimani-Hochberg False Discovery 32 33 Rate (Benjamini and Hochberg, 1995). Gene Ontology clustering and enrichment analysis 34 was performed using GeneCoDis (Carmona-Saez et al., 2007, Nogales-Cadenas et al., 2009).

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## 2 *Real Time PCR*

3 The microarray data were validated by real-time PCR. First strand cDNA was synthesised using the same RNA samples and the Thermoscript RT System (Invitrogen, 4 5 Paisley, UK) Real time PCR was performed on a select gene set using the Stratagene MxPro 6 system and the SYBR green DNA detection chemistry (Biorad, Hemel Hempstead, UK). All 7 RT-PCR experiments were repeated in triplicate. Data were analysed according to the method of Pfaffl (Pfaffl, 2001), using a dilution series based on pooled cDNA samples to determine 8 9 the primer efficiency. The internal reference gene was rrsB, encoding the 16s rRNA subunit, which is not regulated by Ag. The primer pairs were as follows. For cueO: Forward, 10 TACCGATCCCTGATTTGCTC, Reverse, GACTTCACCCGGTACTTCCA; cusA: Forward, 11 12 TGGATGGGCTTTCATCTTTC, Reverse, TTCTGCTCGCTGAATGTTTG; ompF: TGCGCAACTAACAGAACGTC, Reverse, AGGCTTTGGTATCGTTGGTG; 13 Forward, 14 soxS: Forward, GTAATCGCCAAGCGTCTGAT, Reverse, 15 CCCATCAGAAAATTATTCAGGATCT. Primers were designed to amplify a 200-300 bp 16 region of the target gene.

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#### 1 Results

The microarrays measured the global changes in gene expression in exponentially 2 replicating (log-phase) E. coli after 10 minutes exposure to 142 nm AgNPs or AgNO<sub>3</sub>. The 3 nanoparticles have been characterised in our earlier study (McQuillan et al., 2012) and are 4 5 composed of silver, with no surface ligand. They dissolve in the experimental medium at a rate that is linearly related to the surface area, associate directly with the cell surface and are 6 acutely toxic to the *E. coli*. The bacteria were treated with 40 µg/mL of AgNPs or 0.4 µg.mL 7 of Ag<sup>+</sup> for precisely 10 minutes in a rich medium (low salt Luria Broth). The dose did not 8 9 inhibit bacterial replication in order to avoid background gene regulation associated with a 10 change in growth phase, Figure 1.

After the 10 minute exposure the mRNA pool was stabilised and used to synthesise 11 12 cDNA labelled with Cy3 (control samples) or Cy5 (Ag treated samples) including at least one dye swap per experiment. Treated/untreated control pairs were hybridised with Agilent gene 13 14 expression microarrays. Gene regulation was subject to confidence testing and filtered using a >2-fold change cut-off to generate lists of significantly up-regulated and down-regulated 15 16 genes, summarised in Table 1. In total, 188 genes were regulated in both Ag treatments, 161 were up-regulated and 27 were down-regulated. However, the response to each treatment was 17 18 also clearly different; 309 genes were regulated exclusively by AgNPs whereas only 70 genes were regulated exclusively by silver nitrate and overall the response to AgNPs was almost 2-19 20 fold greater in magnitude.

Biological interpretation of the microarray data was carried out using GeneCoDis to 21 find significantly enriched (hypergeometric test,  $p \le 0.05$ ) Gene Ontology (GO) terms against 22 a whole-genome reference set containing 4,619 E. coli genes. GO term enrichment analysis 23 was carried out for the lists of up- and down-regulated genes, regulated by both treatments or 24 regulated independently. The results of the enrichment analysis are summarised in Figure 2. 25 The full gene lists are given in the supplementary information, Table S1 (AgNPs) and Table 26 S2 (silver nitrate). Genes which we refer to in our discussion of the results are presented in 27 28 Tables 2-6.

The microarray data were validated by comparing the expression ratio of 4 genes (*ompF*, *cueO*, *cusF* and *soxS*) with the results of expression analysis using real-time PCR for the same RNA samples. We found that the results from microarray analysis and real-time PCR had a strong correlation, Figure 3.

#### 1 Discussion

2 The overall genetic response to the two physical forms of silver was quite different. In total 379 genes were differentially regulated; 309 genes were only regulated by AgNPs and 3 70 genes were only regulated by silver nitrate. For both forms of Ag, the primary toxicant is 4 5 Ag(I). For Ag added as AgNO<sub>3</sub> the labile Ag(I) can form Ag-complexes with components in 6 the medium and labile Ag ions enter the *E. coli* through the cation selective outer membrane 7 porin proteins. Dissolved Ag(I) is also supplied to the medium from disperse AgNPs which dissolve following oxidation of the silver surfaces. Therefore, we expect that the differential 8 9 response is a result of additive toxic effects from the Ag(I) delivery mechanism. In Eukaryotes, an AgNP toxicity process independent of ion release is described for 10 imperfections in the crystal lattice structure and highly reactive electron configurations at the 11 12 NP surface (George et al., 2012, Nel et al., 2006). However, there is no evidence to support 13 this in bacteria, which have structurally and chemically distinct membranes, in line with 14 recent evidence that AgNPs have no anti-bacterial activity if ion release is blocked under anaerobic conditions (Xiu et al., 2012). In our earlier study, we proposed a hypothesis based 15 16 on the observation that the nanoparticles associate directly with the cell surface, and dissolve on the outer membrane to create a high interfacial Ag(I) concentration, which enhances the 17 18 anti-bacterial effects as a function of the labile Ag(I) concentration in the bulk solution of the 19 medium (McQuillan et al., 2012). Accordingly in our Ag exposures, the AgNP-treated 20 bacteria may experience membrane proximity damage and a gradient of Ag(I) from the dissolving nanoparticle acting as a point source. In contrast, exposure to silver nitrate may 21 generate entirely different concentration gradients of  $Ag(I)^+$  within the cell. AgNPs with a 22 diameter of 12 nm, far smaller than those used in this study, have been shown to penetrate 23 into the cell wall and enter the cytoplasm of E. coli, and interact directly with nucleic acids 24 (Sondi and Salopek-Sondi, 2004, Jose Ruben and et al., 2005) which could further 25 differentiate the AgNP-AgNO<sub>3</sub> toxicity. However, we consider this process unlikely to occur 26 27 under these conditions because (1) the nanoparticles are relatively large (142 nm) and (2) 28 sufficient membrane damage to allow entry of a particle this size would be lethal to the cell 29 whereas the dose used in our experiments was sub-inhibitory. There are no known 30 nanoparticle transport processes in E. coli or other prokaryotes that could facilitate uptake in 31 this size range.

The biological interpretation of the differentially regulated genes gave no clear indication as to why the responses were different. Genes for lipid and fatty acid biosynthesis were down-regulated only after exposure to AgNPs but we would predict that localised 1 membrane damage would lead to an increase in the expression of these genes. If the cell 2 cycle is temporarily arrested upon sudden addition of Ag, the lipid biosynthetic processes 3 may be reduced and captured in the early phase 10 minute response. Although the overall 4 growth profiles after exposure to both forms of Ag are equivalent, Ag(I) addeds as AgNO<sub>3</sub> 5 has the greater lability in the medium and the initial metabolism response to AgNPs may 6 have been delayed.

7 In total 188 genes were regulated after exposure to both forms of Ag; 161 genes were up-regulated and 27 genes were down-regulated. This response follows a logical pattern for 8 9 the indiscriminate action of Ag(I) on proteins, leading to potentially pan-metabolism toxic effects which require a substantial regulation of the E. coli genome. This was up to 11.1 % of 10 the identified genes after applying a 2-fold change cut-off and confidence filters. Studies on 11 12 silver sulfadiazine, a topical agent for anti-sepsis of superficial burns, demonstrate that Ag(I) can also bind to nucleic acids (Rosenkranz and Carr, 1972, Rosenkranz and Rosenkranz, 13 14 1972) but DNA is typically localised to the core of the cell and surrounded by high concentrations of proteins which will be attacked first. Accordingly, at the sub-inhibitory 15 16 dose used in our microarray experiments there was no evidence for a genotoxic response.

The action of Ag(I) on protein structure led to the induction of the *E. coli* heat-shock 17 18 response (HSR) and the positive regulation of genes encoding protein chaperones and proteolytic enzymes for the stabilisation and re-folding or proteolysis of denatured 19 polypeptides. Protein molecular chaperones DnaK-DnaJ-GrpE (DJE) and GroEL-GroES 20 were induced by up to 19-fold, and genes encoding the small heat shock proteins IbpA and 21 IbpB by up to 180-fold, Table 2. At the same 10 minute time point there was down-regulation 22 of genes associated with translation, consistent with the requirement to mount an adaptive 23 response before consuming cellular resources to generate more proteins (Lindquist, 1981). 24 25 The HSR is positively regulated by a sudden increase in the cytosolic concentration of the sigma 32 ( $\sigma^{32}$ ) subunit of RNA polymerase (RNAP). Regulation of the response by Ag could 26 be based on the competitive binding of Ag-denatured protein substrates and  $\sigma^{32}$  with the 27 protein chaperone DnaK (Arsene et al., 2000, Bukau, 1993), which allows for a temperature 28 independent activation of the HSR genes. 29

Ag also induced genes belonging to the operons *isc* and *suf*, encoding iron-sulfur cluster assembly proteins (Py and Barras, 2010), Table 3. This demonstrates that Ag<sup>+</sup> can perturb Fe-S metabolism in line with evidence that Ag causes uncoupling of the respiratory chain and respiratory arrest (Holt and Bard, 2005). Predictably, the biosynthesis of new Fe-S

1 clusters increases the cellular demand for iron and the positive regulation of genes under the control of the Ferric Uptake Regulator (Fur) regulon which increase the supply of ferric iron 2 (Fe<sup>3+</sup>) from the medium, Table 4. The low iron response is enhanced by the expression of 3 *cueO*, which was up-regulated by up to 320-fold. CueO is a Cu(I)/Ag(I)-inducible cuprous 4 5 oxidase which can oxidise and inactivate the enterobactin siderophore (Grass et al., 2004), 6 reducing the rate of iron acquisition. Accordingly, the same low iron response is stimulated 7 by excess copper (Kershaw et al., 2005). Ag also induced the genes cysA and cysW, which 8 encode subunits of the ABC family sulphate/thiosulfate transporter, and a complement of 9 genes required for intracellular sulphate reduction and assimilation during the biosynthesis of cysteine, Table 5. Activation of cysteine biosynthesis is logical as the functional thiol side 10 chain is a specific molecular target for Ag(I), and the supply of sulfur would further support 11 12 the assembly of novel Fe-S.

As the Fur regulator protein responds directly to cytosolic Fe(II) concentration the 13 14 activation of this regulon indicates that the cytosolic pool of iron is depleted quickly, within the 10 minutes following Ag exposure. For pathogenic *E. coli* the availability of iron is a key 15 16 factor in virulence associated with successful colonisation of the urinary tract and proliferation in the small intestine (Litwin and Calderwood, 1993). Destruction of essential 17 18 iron-sulfur proteins and an increase in the iron requirement could represent a fundamental anti-bacterial mechanism for Ag in vivo where iron availability is necessarily kept minimal as 19 20 part of the innate host defences.

Another mechanism in Ag toxicity might be to displace metabolically important 21 metal ion cofactors from their native coordination sites on proteins. Specifically the parallels 22 between silver and copper chemistry in E. coli are well established (Franke et al., 2001, 23 Franke et al., 2003, Loftin et al., 2007). The Ag(I) and Cu(I) ions have the same  $d^{10}$  electronic 24 configuration, charge and similar ionic radii, and have been shown to have a similar protein 25 coordination chemistry (Loftin et al., 2007). If Ag(I) displaces Cu(I) from its native 26 27 coordination sites on proteins then the labile Cu(I) released into the cell may lead to the 28 generation of hydroxyl radicals (Simpson et al., 1988). Our microarray results indicate that 29 Ag(I) may interact with Cu(I) sensor proteins, CusS and CueR, which activate genes encoding copper ion homeostasis systems; CusCFBA, CopA and CueO (Franke et al., 2003, 30 31 Lok et al., 2008b, Munson et al., 2000). Both forms of Ag induced a complement of CusS 32 and CueR regulated genes, Table 6, but interestingly, the metal ion binding domain of CusS 33 is located on the periplasmic face of the plasma membrane whereas CueR is a soluble

cytosolic protein, so activation by Ag treatment indicates that dissolved Ag(I) could have
 been present in multiple cellular compartments.

3 Labile copper displaced from cupro-protein complexes could undergo redox cycling to generate highly reactive oxygen radicals, in line with evidence that AgNPs induce 4 5 oxidative stress responses in human hepatoma cells (Kim et al., 2009), zebra fish hepatocytes 6 (Choi et al.), fruit fly larvae (Ahamed et al., 2009) and in the bacterium Staphylococcus 7 aureus (Dagmar Chudobova, 2013). In the E. coli model, a small complement of antioxidant systems belonging to the soxRS regulon were expressed at high levels but there was no 8 9 significant enrichment of redox stress-associated GO annotations in the gene lists. The SoxR 10 protein is a constitutive cytosolic transcription factor which is activated following oxidation of iron-sulfur clusters [2Fe-2S], and could be directly compromised by Ag(I). Active SoxR 11 positively regulates the expression of soxS, expressed by up to 600-fold following Ag 12 treatment, encoding a second transcription factor that acts sequentially to initiate a cascade of 13 14 anti-oxidant responses. The high level of gene induction for soxS, which was up-regulated more than any other gene, is evidence that redox stress is an important determinant of Ag 15 16 toxicity.

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#### 18 Conclusions

In conclusion, our data for the differential AgNP-AgNO<sub>3</sub> response support a growing 19 20 body of evidence for a nanoparticle-specific silver ion dependenttoxicity mechanism. We propose, based upon our earlier observations of AgNPs dissolving in the medium and 21 attaching to the cell surface (McQuillan et al., 2012), that dissolution at the cell wall produces 22 an enhanced interfacial concentration that enters the cell. We have previously shown that 23 nanoparticle-delivery can enhance the anti-bacterial activity of Ag(I), but this is only 24 applicable if AgNPs are free to interact with the cell surface. Products which contain AgNPs 25 fixed in a gel matrix (Jain et al., 2009) that cannot interact directly with bacteria may not 26 27 benefit from this enhanced effect. For the overall genetic response to both physical forms of 28 Ag the comprehensive induction of genes for the heat shock response is evidence that Ag(I)29 acts on protein structure, and consistent with genetic responses to silver in eukaryotic models 30 Drosophila melanogaster (Ahamed et al., 2009), and Caenorhabditis elegans (Roh et al., 31 2009). The unfolded protein response has been linked with serious disease in humans and 32 prolonged non-essential silver use should be monitored. Additional disruption occurs at iron-33 sulfur components leading to disruption of metabolically essential processes and could 34 represent the critical Ag target, leading to respiratory arrest and a demand for iron, which is

typically a limiting nutrient in various *E. coli* infection scenarios. There was clear evidence that Ag causes redox stress but the greatest response in terms of the number of genes regulated was the response to unfolded proteins, reflecting the pan-metabolism action of Ag(I) on protein structure and function. Accordingly, we consider that this process is the primary mechanism in Ag toxicity against *E. coli*.

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# 1 Acknowledgements

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#### 1 Figure Legends

**Figure 1.** Growth Plots for the *E. coli* in low salt Luria Broth. The solid line shows the growth plot for the *E. coli* in the low salt Luria Both. The bacteria were treated with Ag when the cultures reached a density of approximately  $10^7$  CFU/mL (2 hours). The dashed lines show the sub-inhibitory effect of the exposure concentration for AgNPs ( $\Box$ ) and Ag(I) as AgNO<sub>3</sub> ( $\Delta$ ) on the post-exposure growth plot. The error bars represent the standard error of the mean for triplicate cultures.

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9 Figure 2. GO Term Enrichment Clustering Analysis using GeneCoDis. Bar charts showing 10 the number of genes sharing specific Gene Ontology (GO) terms that were significantly 11 enriched in the lists of up-regulated (top) or down regulated (bottom) genes including those 12 that were independently regulated by AgNPs or Ag(I) as AgNO<sub>3</sub>. The bar labels include the 13 term accession number.

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Figure 3. Validation of microarray data by real-time PCR. The microarray gene expression data were validated by comparing the results with a set of genes (*ompF*, *cusA*, *cueO* and *soxS*) which were measured by real-time PCR using the same RNA samples. In both experiments, either (A) Ag(I) as AgNO<sub>3</sub> or (B) AgNPs linear regression analysis shows that the expression values had a strong correlation with  $R^2$  values of 0.987 and 0.983 respectively. The units are Log 2 gene expression ratio between Ag treated and untreated control samples.

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# 1 Tables

	Total genes regulated >2-fold			Genes un	iquely regu	lated >	>2-fold	
	Up Regulated	Down Regulated	Total	% of Genome	Up regulated	Down Regulated	Total	% of Genome
Silver Nanoparticles	273	224	497	11.1 %	112	197	309	6.9 %
Silver Nitrate	220	38	258	5.8 %	59	11	70	1.7 %

Table 1. Gene Regulation	in E. coli Exposed to	AgNO <sub>3</sub> or AgNPs at	10 minutes.

			<b>Expression Ratio</b>		
Gene	Function	AgNO <sub>3</sub>	AgNPs		
ipaA	Small heat shock protoins; hind and stabilise denstured polynantides	180.16	6.44		
ipaB	Shan heat shock proteins, bind and stabilise denatured polypeptides	4.76	17.55		
clpB	Disaggregation of insoluble protein aggregates	83.16	12.66		
clpP	Proteolysis	2.04	3.23		
dnaK		19.32	6.62		
dnaJ	DnaK-DnaJ-GrpE (DJE) complex; chaperone for protein folding; protein disaggregation; regulation of the heat shock response	3.46	2.06		
grpE		4.96	4.25		
groS	CroEl complex, chaperona for protein folding, protein ra folding	9.93	5.69		
groL	GOEL complex, chaperone for protein fording, protein re-roading	9.44	5.44		
htpG	Protein folding; homologue to mammalian HSP90	11.33	5.97		
htpX	Protease; degradation of denatured polypeptides	8.94	5.47		
hslJ		*	5.91		
hslO	Heat shock locus proteins	3.62	3.37		
hslR		2.90	2.62		
hslU		2.36	2.28		
hslV		2.33	2.72		
lon	DNA-binding ATP-dependent protease	2.85	3.11		
idhA	NAD-linked fermentative lactate dehydrogensase; associated with heat stress	3.01	2.48		
hflK		2.88	2.31		
hflX	Putative proteases; associated with heat stress	*	2.41		
rrmJ	Ribosome associated methyltransferase; associated with heat stress	2.76	3.12		
cpxR		*	2.80		
ppiA	CpxAR regulon; responds to protein unfolding in the periplasm; responds to $Cu^+$	*	2.67		
dsbA	Cu	*	2.57		

# Table 2. Heat Shock Response Genes Induced by Silver

1 \* failed expression cut-off or confidence filters

		Expressi	on Ratio
Gene	Function	AgNO <sub>3</sub>	AgNPs
iscA		4.11	2.09
iscU	Fe-S assembly complex	3.65	2.06
iscS		3.14	*
iscR	Regulatory protein for iscSUA	7.99	2.68
sufA		62.29	16.26
sufB		30.58	10.82
sufC	Fe-S assembly complex	20.58	11.97
sufD		18.06	7.83

# Table 3. Iron-Sulfur Cluster Assembly Genes Induced by Silver

Gene entA entB entC entD B entE entF	<b>Function</b> Biosynthesis of the enterobactin siderophore	AgNO <sub>3</sub> 146.78 116.39 161.96	AgNP: 9.19 10.65 4.91
entA entB entC entD B entE entF	Biosynthesis of the enterobactin siderophore	146.78 116.39 161.96	9.19 10.65 4.91
entB entC entD B entE entF	Biosynthesis of the enterobactin siderophore	116.39 161.96	10.65 4.91
entC entD E entE entF	Biosynthesis of the enterobactin siderophore	161.96	4.91
entD B entE entF	Biosynthesis of the enterobactin siderophore		
entE entF		6.82	4.69
entF		77.22	12.95
		53.56	15.07
ybdB		160.24	7.49
Cir		386.65	4.44
Fiu	Outer Membrane Fe <sup>3+</sup> /ligand receptors	41.44	3.32
fepA		47.48	12.21
fepB		55.40	2.87
fepC U	Jptake of Fe <sup>3+</sup> /enterobactin	24.10	*
fepD		20.76	*
fepG		27.12	*
fecR		29.91	2.05
fecI		35.64	2.27
fecA	Jptake of ferric citrate	5.21	*
fecB		2.99	*
fhuA		14.38	*
fhuC		5.62	*
fhuE U	Jptake of ferrichrome	20.93	16.0
fhuF		19.91	*
tonB		2.43	*
<i>exbB</i> II	nner membrane complex; energises the outer membrane Fe <sup>3+</sup> /ligand transporter	rs 4.19	*
exbD		4.03	*
* failed exp	pression cut-off or confidence filters		

## Table 4. Fur Regulon Genes Induced by Silver

			sion Ratio	
Gene	Function	AgNO <sub>3</sub>	AgNPs	
cysA		16.19	6.50	
cysC		4.29	5.94	
cysD		4.49	7.00	
cysH		13.66	6.97	
cysI	Sulfate assimilation and biosynthesis of cysteine	8.49	7.83	
cysK		3.70	2.49	
cysN		23.75	6.74	
cysP		2.02	2.72	
cysW		5.54	5.22	

# Table 5. Sulfate Transport and Assimilation Genes Induced by Silver

		Expressio	n Ratio
Gene	Function	AgNO <sub>3</sub>	AgNPs
copA	P-type ATPase; copper transporter	47.09	13.71
cueO	Periplasmic cuprous oxidase	320.45	36.72
cusC		72.48	40.54
cusF	RND-protein driven cation/proton exchanger; may transport Cu <sup>+</sup> and Ag <sup>+</sup> ; cusF	376.39	54.08
cusB	encodes a small Ag <sup>+</sup> /Cu <sup>+</sup> binding protein	99.98	26.90
cusA		50.32	20.72
cusR		2.53	5.11
cusS	Two-component regulatory system; regulates <i>cusCFBA</i>	*	2.83
* failed	expression cut-off or confidence filters		

# Table 6. Copper Homoestasis Genes Induced by Silver

## 1 References

- 2
- AHAMED, M., POSGAI, R., GOREY, T. J., NIELSEN, M., HUSSAIN, S. M. & ROWE, J. J. 2009. Silver
   nanoparticles induced heat shock protein 70, oxidative stress and apoptosis in Drosophila
   melanogaster. *Toxicology and Applied Pharmacology*, 242, 263-269.
- ARSENE, F., TOMOYASU, T. & BUKAU, B. 2000. The heat shock response of Escherichia coli. Int J Food
   Microbiol, 55, 3-9.
- BAUMAN, J. E. & WANG, J. C. 1964. Imidazole Complexes of Nickel(II), Copper(II), Zinc(II), and
   Silver(I). *Inorganic Chemistry*, 3, 368-373.
- BELL, R. A. & KRAMER, J. R. 1999. Structural chemistry and geochemistry of silver-sulfur compounds:
   Critical review. *Environmental Toxicology and Chemistry*, 18, 9-22.
- BENJAMINI, Y. & HOCHBERG, Y. 1995. Controlling the False Discovery Rate: A Practical and Powerful
   Approach to Multiple Testing. *Journal of the Royal Statistical Society. Series B* (Methodological), 57, 289-300.
- BUKAU, B. 1993. Regulation of the Escherichia coli heat-shock response. *Molecular Microbiology*, 9,
   671-680.
- CARMONA-SAEZ, P., CHAGOYEN, M., TIRADO, F., CARAZO, J. M. & PASCUAL-MONTANO, A. 2007.
   GENECODIS: a web-based tool for finding significant concurrent annotations in gene lists.
   *Genome Biol*, 8, R3.
- CHOI, J. E., KIM, S., AHN, J. H., YOUN, P., KANG, J. S., PARK, K., YI, J. & RYU, D.-Y. Induction of
   oxidative stress and apoptosis by silver nanoparticles in the liver of adult zebrafish. *Aquatic Toxicology*, 100, 151-159.
- 23 CLEMENT, J. L. & JARRETT, P. S. 1994. Antibacterial silver. *Met Based Drugs*, 1, 467-82.
- DAGMAR CHUDOBOVA, J. D., LUKAS NEJDL, DARINA MASKOVA, MIGUEL ANGEL MERLOS RODRIGO,
   BRANISLAV-RUTTKAY NEDECKY, OLGA KRYSTOFOVA, JINDRICH KYNICKY, MARIE KONECNA,
   MIROSLAV POHANKA, JAROMIR HUBALEK, JOSEF ZEHNALEK, BORIVOJ KLEJDUS, RENE KIZEK,
   VOJTECH ADAM 2013. Oxidative Stress in Staphylococcus aureus Treated with Silver(I) Ions
   Revealed by Spectrometric and Voltammetric Assays. International Journal of
   Electrochemical Science, 8.
- FRANKE, S., GRASS, G. & NIES, D. H. 2001. The product of the ybdE gene of the Escherichia coli
   chromosome is involved in detoxification of silver ions. *Microbiology*, 147, 965-72.
- FRANKE, S., GRASS, G., RENSING, C. & NIES, D. H. 2003. Molecular analysis of the copper transporting efflux system CusCFBA of Escherichia coli. *J Bacteriol*, 185, 3804-12.
- FRANKE, S., GRASS, G. NIES, D. H. 2001. The Product of the ybdE Gene of the Escherichia coli
   Chromosome in Involved in the Detoxification of Silver Ions. *Microbiology*, 147, 965-972.
- GEORGE, S., LIN, S., JI, Z., THOMAS, C. R., LI, L., MECKLENBURG, M., MENG, H., WANG, X., ZHANG, H.,
   XIA, T., HOHMAN, J. N., LIN, S., ZINK, J. I., WEISS, P. S. & NEL, A. E. 2012. Surface defects on
   plate-shaped silver nanoparticles contribute to its hazard potential in a fish gill cell line and
   zebrafish embryos. *ACS Nano*, 6, 3745-59.
- 40 GRASS, G., THAKALI, K., KLEBBA, P. E., THIEME, D., MULLER, A., WILDNER, G. F. & RENSING, C. 2004.
   41 Linkage between catecholate siderophores and the multicopper oxidase CueO in Escherichia
   42 coli. J Bacteriol, 186, 5826-33.
- GUPTA, A., MATSUI, K., LO, J. F., SILVER, S. 1999. Molecular Basis for Resistance to Silver Cations in
   Salmonella. *Nature Medicine*, 5, 183-188.
- HOLT, K. B. & BARD, A. J. 2005. Interaction of silver(I) ions with the respiratory chain of Escherichia
   coli: an electrochemical and scanning electrochemical microscopy study of the antimicrobial
   mechanism of micromolar Ag+. *Biochemistry*, 44, 13214-23.
- JAIN, J., ARORA, S., RAJWADE, J. M., OMRAY, P., KHANDELWAL, S. & PAKNIKAR, K. M. 2009. Silver
   nanoparticles in therapeutics: development of an antimicrobial gel formulation for topical
   use. *Mol Pharm*, 6, 1388-401.

- JOSE RUBEN, M. & ET AL. 2005. The bactericidal effect of silver nanoparticles. *Nanotechnology*, 16,
   2346.
- KAWATA, K., OSAWA, M. & OKABE, S. 2009. In Vitro Toxicity of Silver Nanopartides at Noncytotoxic
   Doses to HepG2 Human Hepatoma Cells. *Environmental Science & Technology*, 43, 6046 6051.
- KERSHAW, C. J., BROWN, N. L., CONSTANTINIDOU, C., PATEL, M. D. & HOBMAN, J. L. 2005. The
   expression profile of Escherichia coli K-12 in response to minimal, optimal and excess copper
   concentrations. *Microbiology*, 151, 1187-1198.
- 9 KIM, S., CHOI, J. E., CHOI, J., CHUNG, K.-H., PARK, K., YI, J. & RYU, D.-Y. 2009. Oxidative stress 10 dependent toxicity of silver nanoparticles in human hepatoma cells. *Toxicology in Vitro*, 23,
   11 1076-1084.
- KITTLESON, J. T., LOFTIN, I. R., HAUSRATH, A. C., ENGELHARDT, K. P., RENSING, C. & MCEVOY, M. M.
   2006. Periplasmic metal-resistance protein CusF exhibits high affinity and specificity for both
   Cul and Agl. *Biochemistry*, 45, 11096-102.
- LANSDOWN, A. B. 2006. Silver in health care: antimicrobial effects and safety in use. *Curr Probl Dermatol*, 33, 17-34.
- LI, X. Z., NIKAIDO, H. & WILLIAMS, K. E. 1997. Silver-resistant mutants of Escherichia coli display
   active efflux of Ag+ and are deficient in porins. *J Bacteriol*, 179, 6127-32.
- LIM, D.-H., JANG, J., KIM, S., KANG, T., LEE, K. & CHOI, I.-H. 2012. The effects of sub-lethal
   concentrations of silver nanoparticles on inflammatory and stress genes in human
   macrophages using cDNA microarray analysis. *Biomaterials*, 33, 4690-4699.
- LINDQUIST, S. 1981. Regulation of protein synthesis during heat shock. *Nature*, 293, 311-314.
- LITWIN, C. M. & CALDERWOOD, S. B. 1993. Role of iron in regulation of virulence genes. *Clinical Microbiology Reviews*, 6, 137-149.
- LOFTIN, I. R., FRANKE, S., BLACKBURN, N. J. & MCEVOY, M. M. 2007. Unusual Cu(I)/Ag(I) coordination
   of Escherichia coli CusF as revealed by atomic resolution crystallography and X-ray
   absorption spectroscopy. *Protein Sci*, 16, 2287-93.
- LOK, C.-N., HO, C.-M., CHEN, R., TAM, P. K.-H., CHIU, J.-F. & CHE, C.-M. 2008a. Proteomic
   Identification of the Cus System as a Major Determinant of Constitutive Escherichia coli
   Silver Resistance of Chromosomal Origin. *Journal of Proteome Research*, 7, 2351-2356.
- LOK, C. N., HO, C. M., CHEN, R., TAM, P. K., CHIU, J. F. & CHE, C. M. 2008b. Proteomic identification
   of the Cus system as a major determinant of constitutive Escherichia coli silver resistance of
   chromosomal origin. *J Proteome Res*, 7, 2351-6.
- MCQUILLAN, J. S., INFANTE, H. G., STOKES, E. & SHAW, A. M. 2012. Silver nanoparticle enhanced silver ion stress response in Escherichia coli K12. *Nanotoxicology*, 6, 857-66.
- MUNSON, G. P., LAM, D. L., OUTTEN, F. W. & O'HALLORAN, T. V. 2000. Identification of a Copper Responsive Two-Component System on the Chromosome of Escherichia coli K-12. J.
   *Bacteriol.*, 182, 5864-5871.
- NEL, A., XIA, T., MÄDLER, L. & LI, N. 2006. Toxic Potential of Materials at the Nanolevel. Science, 311,
   622-627.
- NIAZI, J. H., SANG, B. I., KIM, Y. S. & GU, M. B. 2011. Global gene response in Saccharomyces
   cerevisiae exposed to silver nanoparticles. *Appl Biochem Biotechnol*, 164, 1278-91.
- NOGALES-CADENAS, R., CARMONA-SAEZ, P., VAZQUEZ, M., VICENTE, C., YANG, X., TIRADO, F.,
   CARAZO, J. M. & PASCUAL-MONTANO, A. 2009. GeneCodis: interpreting gene lists through
   enrichment analysis and integration of diverse biological information. *Nucleic Acids Res,* 37,
   W317-22.
- 47 PFAFFL, M. W. 2001. A new mathematical model for relative quantification in real-time RT-PCR.
   48 *Nucleic Acids Res*, 29, e45.
- 49 PY, B. & BARRAS, F. 2010. Building Fe-S proteins: bacterial strategies. *Nat Rev Microbiol*, 8, 436-46.

- ROH, J.-Y., SIM, S. J., YI, J., PARK, K., CHUNG, K. H., RYU, D.-Y. & CHOI, J. 2009. Ecotoxicity of Silver
   Nanoparticles on the Soil Nematode Caenorhabditis elegans Using Functional
   Ecotoxicogenomics. *Environmental Science & Technology*, 43, 3933-3940.
- ROSENKRANZ, H. S. & CARR, H. S. 1972. Silver sulfadiazine: effect on the growth and metabolism of
   bacteria. Antimicrob Agents Chemother, 2, 367-72.
- ROSENKRANZ, H. S. & ROSENKRANZ, S. 1972. Silver sulfadiazine: interaction with isolated
   deoxyribonucleicacid. Antimicrob Agents Chemother, 2, 373-83.
- SIMPSON, J. A., CHEESEMAN, K. H., SMITH, S. E. & DEAN, R. T. 1988. Free-radical generation by
   copperions and hydrogen peroxide. Stimulation by Hepes buffer. *Biochem J*, 254, 519-23.
- SONDI, I. & SALOPEK-SONDI, B. 2004. Silver nanoparticles as antimicrobial agent: a case study on E.
   coli as a model for Gram-negative bacteria. *J Colloid Interface Sci*, 275, 177-82.
- VICKERY, H. B. & LEAVENWORTH, C. S. 1930. THE BEHAVIOR OF CYSTINE WITH SILVER SALTS. Journal
   of Biological Chemistry, 86, 129-143.
- 14 XIU, Z. M., ZHANG, Q. B., PUPPALA, H. L., COLVIN, V. L. & ALVAREZ, P. J. 2012. Negligible particle 15 specific antibacterial activity of silver nanoparticles. *Nano Lett*, **12**, 4271-5.
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