

# An efficient and accurate approach to modelling the microstructure and defect properties of $\text{LaCoO}_3$

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Complex perovskite oxides are promising materials for cathode layers in solid oxide fuel cells. Such materials have intricate electronic, magnetic and crystalline structures that prove challenging to model accurately. We analyse a wide range of standard density functional theory approaches to modelling a highly promising system, the perovskite  $\text{LaCoO}_3$ , focussing on optimising the Hubbard  $U$  parameter to treat the self-interaction of the B-site cation's  $d$ -states, in order to determine the most appropriate method to study defect formation and the effect of spin on local structure. By calculating structural and electronic properties for different magnetic states we determine that  $U = 4$  eV for Co in  $\text{LaCoO}_3$  agrees best with available experiment. We demonstrate that the generalised gradient approximation (PBEsol+ $U$ ) is most appropriate for studying structure versus spin state, while the local density approximation (LDA+ $U$ ) is most appropriate to determine accurate energetics for defect properties.

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## I. INTRODUCTION

Solid oxide fuel cells (SOFCs) work by using catalytic processes to oxidise a variety of fuels at the anode while reducing oxygen at the cathode side, balanced by ion transport through the cell, thus generating electrical power with water as the waste product when  $\text{H}_2$  is used as fuel.<sup>1</sup> They are a promising clean energy resource, but due to the chemical processes involved high temperatures are required for efficient operation.<sup>2-6</sup> For next-generation SOFCs, cathode layers that can conduct both ions and electrons at intermediate temperatures ( $\sim 500-750^\circ\text{C}$ ), while remaining stable and compatible with the other layers in the cell, are required.<sup>7-14</sup> One of the most promising materials for such cathode layers is the  $\text{LaCoO}_3$ -based system  $\text{La}_{1-x}\text{Sr}_x\text{Fe}_{1-y}\text{Co}_y\text{O}_3$  (LSCF).<sup>15-21</sup> Finding the optimum doping concentrations for efficient fuel cell operation is, however, challenging, and input from computational modelling of the material properties in order to help formulate design improvements is crucial.<sup>22</sup>

Many computational techniques have been employed to study different aspects of SOFCs, from mesoscopic models<sup>23-25</sup> to interatomic potential-based methods<sup>26-31</sup> and *ab initio* calculations.<sup>32-37</sup> To understand the key properties of LSCF, such as defect formation, ionic conductivity, the mechanism of electronic conductivity, magnetic and electronic structure, and surface catalysis, requires an accurate but computationally tractable approach.<sup>22,38-43</sup> A fundamental requirement of such an approach is a sufficiently accurate description of the parent compound  $\text{LaCoO}_3$ .

At low temperatures, the perovskite  $\text{LaCoO}_3$  stabilises in the rhombohedral phase ( $R\bar{3}c$ , no. 167).<sup>44-46</sup> As temperature  $T$  is varied, an interesting magnetic effect is observed. At low  $T$  (below  $\sim 50$  K)  $\text{LaCoO}_3$  is a diamagnetic insulator.<sup>47</sup> As  $T$  is increased above 50 K, a

pronounced spin-state transition occurs, where the system becomes a paramagnetic semiconductor, with another transition possibly occurring at  $T > 500$  K, where the system becomes metallic.<sup>47-49</sup> Associated with the spin transitions are variations in the local structure and splittings in optical phonon modes, possibly indicating a Jahn-Teller distortion.<sup>50-54</sup> The nature of the spin transition has been extensively studied experimentally and computationally,<sup>55-58</sup> focussing on the  $d$  orbitals of the octahedrally coordinated Co ions, which are split by the crystal field. Initial proposals of a low spin (LS) to intermediate spin (IS) transition, possibly followed by a transition to high spin (HS),<sup>48,59-63</sup> have been superseded by more complicated scenarios involving different HS-LS orderings and possible defect-related effects to explain the experimental results.<sup>64-70</sup> Theoretical approaches applied include density functional theory (DFT),<sup>71</sup> using the local density approximation including a Hubbard  $U$  parameter (LDA+ $U$ ),<sup>60,72-75</sup> dynamical mean-field theory,<sup>69,76</sup> and higher-level quantum chemical approaches.<sup>61,77-80</sup> Despite the sophistication of the methods applied, which become more computationally intense as complexity is increased, the nature of the spin state transition remains a topic of debate.<sup>49,70,76,81-85</sup>

To study defect properties, surface structure and catalysis, the most common approaches have been DFT (including Hubbard  $U$ )<sup>86-90</sup> and interatomic force field methods.<sup>26,27,91-93</sup> Such methods have been successful in modelling oxygen vacancy formation, ion migration, surface defect formation and oxygen reduction, but their application has not concerned the effect of local structure variations on macroscopic magnetic properties.

In this paper, we analyse simple DFT approaches to modelling the structural, magnetic, and electronic properties of  $\text{LaCoO}_3$ , in order to determine the optimum method to be used in studying the defect and catalytic properties of the material, as well as the magnetic effects on local structure. Our concern is to find the best method

81 that is both accurate and computationally tractable. We  
 82 find that, for defect studies, LDA+ $U$  is most advanta-  
 83 geous, while for local structure and magnetism the gener-  
 84 alised gradient approximation with a Hubbard  $U$  param-  
 85 eter (GGA+ $U$ ) is best from those surveyed. We find that  
 86 a value of  $U = 4$  eV is suitable in both cases. We demon-  
 87 strate the applicability of our approaches by studying  
 88 oxygen vacancy formation in the case of LDA+ $U$ , and  
 89 studying phonon mode splitting and local structure mod-  
 90 ification as the spin state varies in the case of GGA+ $U$ .  
 91 Our results serve as a guide to future computational stud-  
 92 ies of the spin and defect properties of LaCoO<sub>3</sub>.

93 The paper is now structured as follows. In Sec. II we  
 94 describe the DFT approaches used; in Sec. III we present  
 95 our results, and in Sec. IV we summarise the main points  
 96 of our study.

## 97 II. CALCULATIONS

98 We have used DFT to calculate the structural, elec-  
 99 tronic and magnetic properties of LaCoO<sub>3</sub> using a range  
 100 of density functionals. All our DFT calculations were  
 101 carried out using the VASP code,<sup>94–97</sup> utilizing the projec-  
 102 tor augmented wave (PAW) method<sup>98</sup> to model core and  
 103 valence electron interactions (using the ‘regular’ PAW  
 104 pseudopotential for O). The valence configurations used  
 105 were: La ( $5s^2 5p^6 6s^2 5d^1$ ), Co ( $4s^2 3d^7$ ), O ( $2s^2 2p^4$ ). To  
 106 account for exchange and correlation, we have compared  
 107 the LDA functional, the Perdew-Burke-Ernzerhof (PBE)  
 108 GGA functional,<sup>99</sup> and the PBE functional corrected for  
 109 solids (PBEsol),<sup>100</sup>. Moreover, the effect of adding a  
 110 Hubbard  $U$  parameter (LDA+ $U$ , PBE+ $U$ , PBEsol+ $U$ )  
 111 has been investigated, using the rotationally invariant  
 112 approach of Dudarev *et al.*<sup>101</sup> As PBEsol was devel-  
 113 oped in order to reproduce lattice parameters more ac-  
 114 curately than PBE,<sup>100</sup> one would expect improved struc-  
 115 tural properties over those found using other GGA ap-  
 116 proaches<sup>86–88,102,103</sup> (but less accurate cohesive energies).  
 117 Furthermore, PBEsol is known to model well interatomic  
 118 forces, resulting in accurate phonon frequencies.<sup>36,104,105</sup>  
 119 Hybrid functionals, where a fraction of Hartree-Fock ex-  
 120 act exchange is included,<sup>106–108</sup> were tested but we do  
 121 not report any results here as we found that, as well as  
 122 being prohibitively intensive computationally for larger  
 123 systems, they represented the Co  $d$  states and material  
 124 band gap in an erroneous manner, a result known from  
 125 previous studies<sup>109–111</sup> on similar systems (we have in-  
 126 cluded a comparison of our calculated structural proper-  
 127 ties using hybrid DFT with those of Gryaznov *et al.*,<sup>112</sup>  
 128 see the Supplemental Material at [URL inserted by pub-  
 129 lisher]).

130 To avoid the problem of Pulay stress, the ion coordi-  
 131 nates in the primitive rhombohedral cell (10 atoms) and  
 132 the cell shape were optimised at constant volume for a  
 133 series of different volumes, without enforcing symmetry  
 134 constraints, and the resulting data fitted to the Mur-  
 135 naghan equation of state to determine the lowest energy

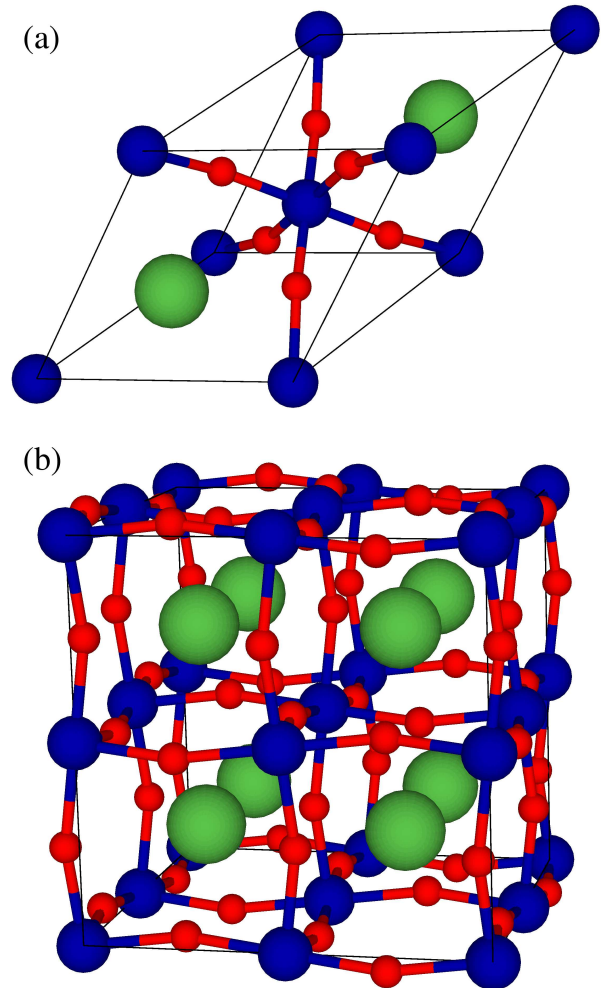


FIG. 1. (Color online) The unit cells of LaCoO<sub>3</sub> used in this work. (a) the 10 atom primitive rhombohedral cell. (b) The 40 atom pseudocubic expansion of the primitive cell. Where necessary, periodically repeated atoms are shown for clarity. La ions are represented by large light grey/green spheres, Co ions by intermediate-sized blue/darker grey spheres, and O ions by smaller red/dark grey spheres.

136 structure. For IS and HS configurations relaxations were  
 137 performed using the pseudocubic expansion of the prim-  
 138 itive cell (40 atoms), which allows symmetry-breaking  
 139 Jahn-Teller distortions to occur if favourable. The two  
 140 cells are shown in Fig. 1. The plane-wave cut-off en-  
 141 ergy used was 650 eV and Brillouin zone sampling was  
 142 performed, employing Gaussian smearing with a smear-  
 143 ing width of 0.05 eV, on a  $8 \times 8 \times 8$  Monkhorst-Pack<sup>113</sup>  
 144  $k$ -point mesh for the primitive cell, and a  $4 \times 4 \times 4$   $k$ -point  
 145 mesh for the pseudocubic cell, which provided conver-  
 146 gence in the total energy of up to  $10^{-4}$  eV. Geometry  
 147 optimisation was deemed to be converged when the in-  
 148 teratomic forces were less than  $10^{-2}$  eV/Å. For defect

149 calculations, a  $2 \times 2 \times 2$  expansion of the pseudocubic cell,  
 150 i.e. a 320 atom supercell, was used, with  $k$ -point sam-  
 151 pling performed at the  $\Gamma$  point only. With this supercell  
 152 the minimum distance between periodic images of point  
 153 defects is 14.95 Å.

154 Phonon frequencies at the  $\Gamma$  point were determined  
 155 using the frozen phonon approach, where the dynamical  
 156 matrix is derived by displacing atoms from their equilib-  
 157 rium positions and calculating the resulting forces, thus  
 158 giving the force constants. Atomic displacements of 0.01  
 159 Å were used and the convergence criterion for the self-  
 160 consistent field iterative procedure was  $10^{-7}$  eV. These  
 161 force calculations were performed using the pseudocubic  
 162 cell, the geometry of which had been relaxed so that the  
 163 interatomic forces were less than  $10^{-4}$  eV/Å, in order to  
 164 determine accurate phonon frequencies. The dynamical  
 165 matrix was diagonalised and the eigenvectors analysed  
 166 using the post-processing program PHONOPY.<sup>114</sup>

167 The formation energy of a neutral oxygen vacancy,  
 168  $E_f[V_O^\times]$  (where we use the standard Kröger-Vink<sup>115</sup> no-  
 169 tation), assuming thermodynamic equilibrium, was de-  
 170 termined from the equation:

$$E_f[V_O^\times] = E_{tot}[V_O^\times] - E_{tot}[\text{bulk}] + \frac{1}{2}\mu_{O_2}, \quad (1)$$

171 where  $E_{tot}[\text{bulk}]$  is the total energy of the pure LaCoO<sub>3</sub>  
 172 bulk supercell,  $E_{tot}[V_O^\times]$  is the total energy of the super-  
 173 cell containing a  $V_O^\times$ , and  $\mu_{O_2}$  is the chemical potential  
 174 of molecular oxygen.  $\mu_{O_2}$  has been determined using the  
 175 standard approach in supercell DFT calculations.<sup>116–119</sup>  
 176 We assume thermodynamical equilibrium with a reser-  
 177 voir of oxygen gas under oxygen-rich conditions, so that  
 178  $\mu_{O_2}$  is the energy of an O<sub>2</sub> molecule in the ground state  
 179 (a triplet), i.e. excluding thermal contributions to the  
 180 chemical potential.

### 181 III. RESULTS

TABLE I. Calculated rhombohedral lattice parameter ( $a$ ) and  
 angle ( $\theta$ ), determined using LDA, PBE, and PBEsol and com-  
 pared with the low temperature neutron diffraction measure-  
 ments from Ref. 44.

	$a$ (Å)	$\theta$ (°)
Experiment	5.3416	60.99
LDA	5.2447	61.34
PBE	5.3613	61.20
PBEsol	5.2887	61.12

182 We first discuss our calculated lattice parameter ( $a$ )  
 183 and rhombohedral angle ( $\theta$ ) of the ground state system  
 184 using different density functionals, presented in Table I  
 185 and Figure 2. As our simulations are at the athermal  
 186 limit, we compare our results with the low temperature  
 187 (4 K) neutron diffraction measurements of Thornton *et*  
 188 *al.*<sup>44</sup> We find that the GGA and GGA+ $U$  functionals

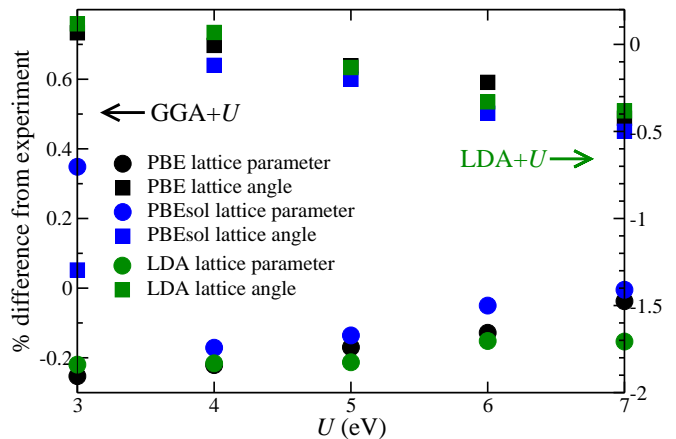


FIG. 2. (Color online) Percentage difference between the low  
 temperature experimental<sup>44</sup> and the calculated rhombohedral  
 lattice parameter (circles) and angle (squares), determined  
 using LDA+ $U$  (shown in green/light grey), PBE+ $U$  (black),  
 and PBEsol+ $U$  (blue/darker grey), shown as a function of  $U$ .  
 The scale on the left ordinate axis corresponds to GGA+ $U$ ,  
 while that on the right corresponds to LDA+ $U$ .

189 give values in good agreement with the experimental re-  
 190 sults (with differences of less than 1%), while LDA and  
 191 LDA+ $U$  underestimate the parameters by  $\sim 2\%$ . This  
 192 underestimation is a well-known feature of LDA.<sup>120</sup>

193 We have calculated the electronic density of states  
 194 (DOS) of LaCoO<sub>3</sub> (in the LS configuration) using differ-  
 195 ent density functionals and present our results in Fig. 4,  
 196 in comparison with the x-ray photoemission measure-  
 197 ments of the upper valence band from Ref. 61. We sum-  
 198 marise the calculated band gaps (LS state) in Table II, in-  
 199 cluding  $U = 4$  eV cases as representative examples of our  
 200 DFT+ $U$  results (see the Supplemental Material at [URL  
 201 inserted by publisher] for further data). The energy gap  
 202 has been experimentally determined using photoemission  
 203 techniques to be 0.6 eV<sup>121</sup> and 0.9 eV,<sup>122</sup> and using opti-  
 204 cal conductivity measurements to be 0.1 – 1.0 eV.<sup>123–125</sup>  
 205 We find that LDA and GGA result in a metallic system,  
 206 as expected due to the well-known self-interaction error  
 207 and resulting band-gap underestimation that is a feature  
 208 of these functionals. Adding a Hubbard  $U$  allows one to  
 209 open a gap, which may be tuned by varying  $U$  (although  
 210 one can derive a  $U$  parameter from first principles, as was  
 211 done in Refs. 75 and 86). From Fig. 4 it is evident that  
 212 varying  $U$  also varies the valence band width, indicating  
 213 that there is a trade-off between these two properties,  
 214 which must be balanced when choosing an appropriate  
 215  $U$  value.

216 In LaCoO<sub>3</sub>, the Co cations are octahedrally coordi-  
 217 nated with a formal oxidation state of 3+, meaning that  
 218 the six  $d$  electrons can occupy the  $e_g$  and  $t_{2g}$  orbitals in  
 219 the configurations shown in Fig. 3; that is in LS, IS or HS  
 220 states. Moreover, the spin states can have ferromagnetic  
 221 (FM) or antiferromagnetic (AFM) ordering amongst the

TABLE II. The energy band gap (in eV) and ground state spin configuration of  $\text{LaCoO}_3$  as determined using different density functionals and compared with experimental results. ‘Metal’ indicates zero gap. For brevity the  $U = 4$  eV cases are included as representative of the DFT+ $U$  functionals. In each case the band gap is calculated for the LS state (see text for the meanings of the acronyms used for spin states).

	Experiment			LDA	PBE	PBEsol	LDA+ $U$	PBE+ $U$	PBEsol+ $U$
Band gap (eV)	0.6, <sup>121</sup>	0.9, <sup>122</sup>	0.1–1.1 <sup>123–125</sup>	Metal	Metal	Metal	0.888	0.953	1.023
Spin state	LS			LS	LS	LS	LS	IS-HS FM	IS-HS FM

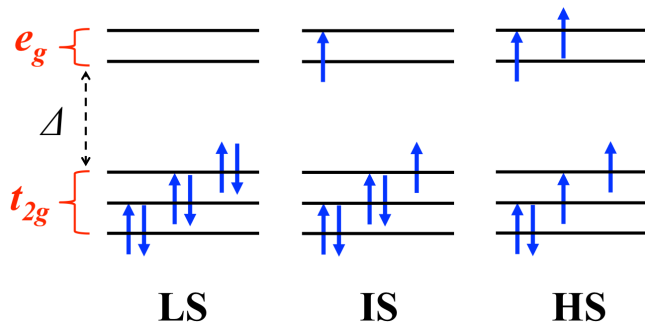


FIG. 3. (Color online) Schematic of the idealised spin states on the octahedrally coordinated Co cations in  $\text{LaCoO}_3$ : low spin (LS), intermediate spin (IS), and high spin (HS). The  $d$  states are split in energy ( $\Delta$ ) by the crystal field into  $e_g$  and  $t_{2g}$  orbitals. Upwards pointing arrows represent spin up electrons, downwards pointing arrows represent spin down.

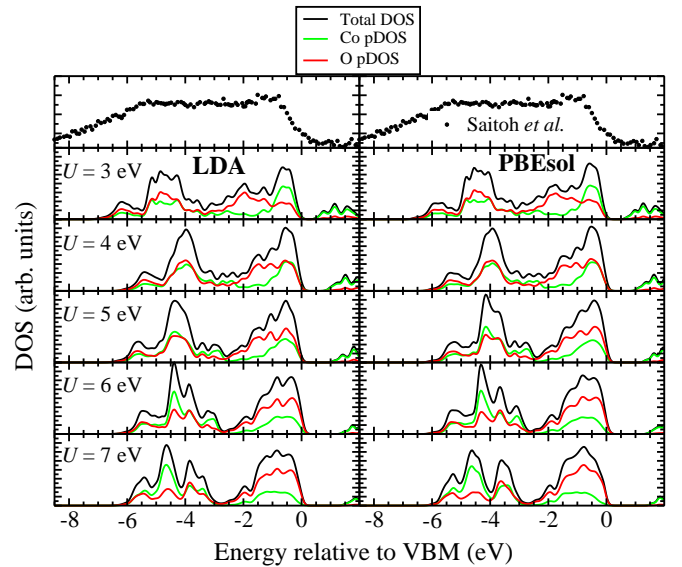


FIG. 4. (Color online) Calculated density of states (DOS) (black lines) and partial DOS (pDOS) (Co pDOS - light grey/green lines, O pDOS - dark grey/red lines) of  $\text{LaCoO}_3$  determined using LDA+ $U$  and PBEsol+ $U$ , for different values of  $U$ . The energy scale is with respect to the valence band maximum (VBM). For comparison the x-ray photoemission results of Saitoh *et al.*<sup>61</sup> are shown (black dots).

222 Co-centred octahedra in different combinations, while it  
 223 is also possible that there is ordered mixing of the LS, IS,  
 224 and HS states. Which configuration is most favourable  
 225 can be determined by calculating and comparing the total  
 226 energies of the different spin combinations. We have per-  
 227 formed such calculations to determine the ground state  
 228 configuration for each density functional considered in  
 229 this study. Our results are presented in Table II (see the  
 230 Supplemental Material at [URL inserted by publisher] for  
 231 further data). We find that LDA, GGA, and LDA+ $U$   
 232 result in a LS ground state configuration, which agrees  
 233 with experiment as our simulations are at the athermal  
 234 limit and  $\text{LaCoO}_3$  is a diamagnetic insulator at low  $T$ .  
 235 PBE+ $U$  and PBEsol+ $U$  result in an interesting ordered  
 236 HS-IS FM configuration as the ground state, a point to  
 237 which we return below.

238 To summarise the results presented so far, amongst  
 239 the density functionals studied: for structural proper-  
 240 ties, GGA functionals give the most accurate results;  
 241 for electronic properties LDA+ $U$  and GGA+ $U$  are most  
 242 accurate; and for magnetic properties LDA, GGA and  
 243 LDA+ $U$  are most accurate. Unsurprisingly, no simple  
 244 DFT approach can accurately reproduce all these prop-  
 245 erties of  $\text{LaCoO}_3$ . Nevertheless, progress can be made by  
 246 using LDA+ $U$  and PBEsol+ $U$ , as we demonstrate below.  
 247 With LDA+ $U$ , well reproduced electronic and magnetic  
 248 structure is gained at the expense of slightly underes-

249 timated structural parameters. For defect calculations  
 250 and studies of surface catalysis, the reproduction of ac-  
 251 curate energetics are required, while errors introduced  
 252 by underestimated structural parameters should largely  
 253 cancel, meaning that LDA+ $U$  will be a suitable func-  
 254 tional for such studies. We note that this approach has  
 255 been studied previously,<sup>86</sup> but was deemed inappro-  
 256 priate for oxygen vacancy formation calculations due to the  
 257 calculated energy being higher than that determined ex-  
 258 perimentally, a point to which we return below. We find  
 259 that PBEsol+ $U$  reproduces the structural parameters in  
 260 excellent agreement with experiment, while also provid-  
 261 ing accurate electronic energies (although, as shown in  
 262 Fig. 2, PBE+ $U$  also results in accurate structural prop-  
 263 erties, using PBEsol is known to model well interatomic  
 264 forces<sup>36,104,105</sup> which are key for phonon frequency cal-  
 265 culations). The calculated magnetic structures, however,  
 266 involve mixtures of LS, IS and HS all close in energy  
 267 (the ground state being ordered IS-HS FM). The differ-

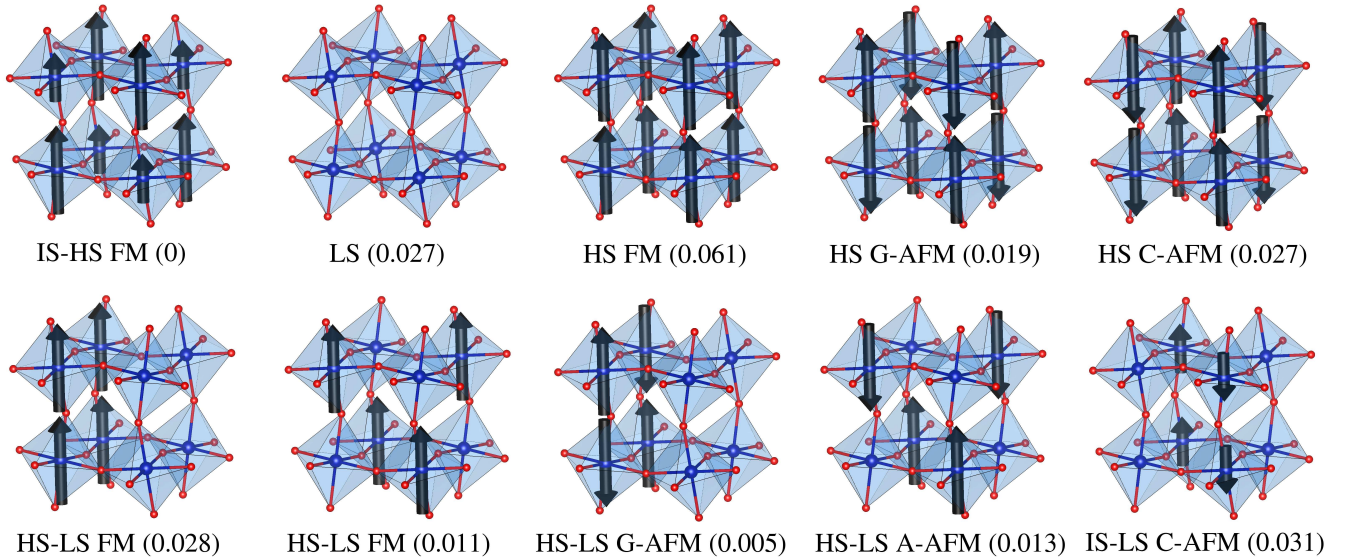


FIG. 5. (Color online) Schematic of the different spin configurations that are close in energy when using the PBEsol+ $U$  functional. The different states are combinations of low spin (LS), intermediate spin (IS), and high spin (HS) configurations with ferromagnetic (FM) or antiferromagnetic (AFM) ordering. AFM ordering can be of A-, C- or G-type. The numbers in parantheses are the energy differences per atom (in eV) between the spin configuration shown and the ground state (IS-HS FM). Spins are indicated by black arrows, the relative length of which distinguish between HS and IS. Co-centred polyhedra are shown, with blue spheres for Co and red for O. For clarity, La ions are not shown.

ent structures are shown in Fig. 5, where the standard  
notation to distinguish different types of AFM ordering  
(A-, C- and G-AFM) is used. For the HS-LS mixtures we  
find either layers that alternate along the [100] direction  
or channels of HS (with FM or AFM ordering) along  
[101], while for the IS-LS mixture we find alternating  
channels along [001]. The ground state IS-HS mix  
consists of alternating channels of each type along  $\bar{1}01$ . For  
pure HS A-AFM could not be stabilised. The accurate  
structural properties, coupled with the different magnetic  
structures lying close in energy, mean that this functional  
may be useful in studying local structural changes *vs* spin  
state. Considering the electronic DOS shown in Fig. 4,  
we see that in varying the  $U$  parameter there is a trade  
off between the energy band gap and the valence band  
width, as mentioned above. Setting  $U = 4$  eV offers  
a good compromise in this trade off for both LDA+ $U$   
and PBEsol+ $U$ . This value agrees well with that used  
in previous studies.<sup>72,86</sup> We also note from Fig. 2 that a  
higher value of  $U$  would result in slightly more accurate  
structural properties. The improvement in the percentage  
difference from experiment between  $U = 4$  and e.g.  
 $U = 7$ , however, is less than 0.4 %, which would not be a  
significant improvement given the drastically worse elec-  
tronic properties obtained with  $U = 7$ .  $U = 4$  offers the  
best compromise for electronic and structural properties  
(moreover, LS is no longer the ground state for LDA+ $U$   
with  $U = 7$  eV, see the Supplemental Material at [URL  
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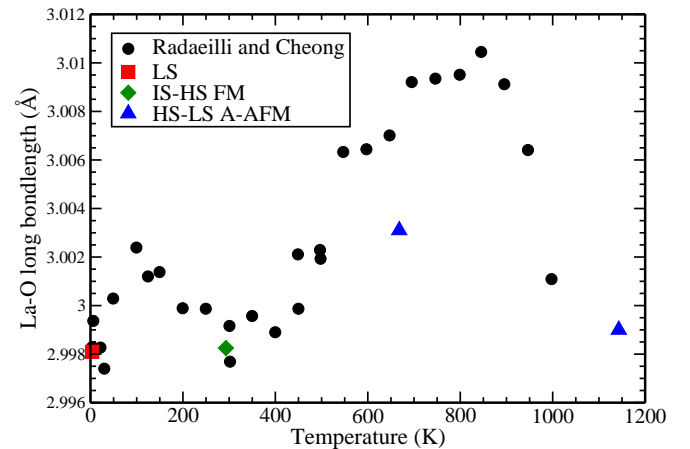


FIG. 6. (Color online) La-O long bond length calculated at different volumes, corresponding to different temperatures, for the spin transition LS to IS-HS FM to HS-LS A-AFM (red square, green diamond and blue triangles respectively), compared with neutron diffraction measurements from Ref. 53 (black circles).

To demonstrate the effectiveness of the PBEsol+ $U$   
( $U = 4$  eV) density functional for studying the rela-  
tionship between local structure and spin state, we have  
calculated, using the pseudocubic cell, the average La-O  
long bond length for the range of spin states shown in

Fig. 5 at  $T=4$ , 273, 668 and 1143 K in order to compare with the neutron diffraction measurements of Radaelli and Cheong.<sup>53</sup> To simulate the different temperatures, we have fixed the lattice parameters to those determined experimentally by Thornton *et al.*<sup>44</sup> and allow the internal ionic coordinates to relax. If we fix the low  $T$  bond length to that of Radaelli and Cheong,<sup>53</sup> and analyse the differences in calculated bond length as the spin state is varied, we find that the transition from LS (at  $T = 4$  K) to the IS-HS FM state (occurring between  $T = 4$  and  $T = 273$  K), followed by a transition to the HS-LS AFM state (at  $T > 273$  K) reproduces the experimental trend well (see Fig. 6). Such a spin transition is consistent with experimental studies, where strong evidence is found of HS states after the initial transition at  $T > 50$  K, rather than just IS spin states.<sup>64–70</sup> We can conclude then that the PBEsol+ $U$  approach can indeed be used successfully for such structural *vs* spin state studies.

As a further example, we have calculated the zone-centre phonon modes of LaCoO<sub>3</sub> in different spin configurations, using PBEsol+ $U$ , in order to compare with the infrared (IR) measurements of Yamaguchi *et al.*<sup>52</sup> At low  $T$  (and hence the LS configuration), we calculate the IR stretching mode doublet to be 68 meV, in excellent agreement with experiment. Considering the transition to IS-HS FM ordering (see above), we find that the mode splits to 67 and 73 meV, again in excellent agreement with experiment.<sup>52</sup> (The associated phonon density of states are given in the Supplemental Material at [URL inserted by publisher]). This result further reinforces our conclusion that we can use this approach to study local structure *vs* spin state. Indeed, we find that, if we were to use LDA+ $U$  instead, the calculated low- $T$  IR stretching mode doublet is 73 meV, an overestimation of  $\sim 7\%$ . As using LDA+ $U$  results in underestimated lattice parameters (see Fig. 2), this increase in the calculated frequency is unsurprising.

Using LDA+ $U$ , we have calculated the formation energy of an oxygen vacancy to be 3.36 eV, with AFM ordering on the neighbouring reduced Co ions. The calculated magnetic moment  $\mu = 1.6\mu_B$ . This result is in good agreement with previous computational studies in the literature using a variety of theoretical approaches.<sup>33,86,87,92</sup> It is, however, significantly higher than the value of 2.2 eV determined experimentally,<sup>126</sup> but, given the low levels of non-stoichiometry observed in undoped LaCoO<sub>3- $\delta$</sub>  ( $\delta \leq 0.01$ ),<sup>126</sup> comparison with this value should take into account that vacancies on the surface may play a significant role in the reduction process. The surface vacancy formation energy has been determined to be lower than in the bulk by  $\sim 1 - 2$  eV<sup>87,93</sup> (this effect has also been determined in the related perovskite perovskite LaMnO<sub>3</sub><sup>40</sup>). Such a result demonstrates that the LDA+ $U$  approach can be used for studies of defect properties of this material. If instead we employ the PBEsol+ $U$  functional, we immediately have the problem that the ground state spin configura-

tion of the defect-free system is not LS. When forming a defect, many spin configurations can be converged, and choosing the most appropriate one is difficult given that the original configuration is incorrect. Using the lowest total energy results, we calculate a formation energy of 6.14 eV, a value that is substantially higher than that determined using LDA+ $U$  and inconsistent with experimental results. Similar problems are expected when using PBE+ $U$ , as the ground state spin configuration is also not LS in that case. These complications, which both lead to results that are most likely not comparable with experiment and increase the computational load (due to the necessity of checking the many different possible configurations), lead us to conclude that GGA+ $U$  is drastically less favourable than LDA+ $U$  when studying defects.

#### IV. SUMMARY/CONCLUSIONS

In summary, we have compared the results of calculated electronic, structural and magnetic properties of LaCoO<sub>3</sub> using a range of standard density functionals in order to determine the optimum DFT approach to study local distortions and defect formation. We found that no single DFT approach could model simultaneously all these aspects accurately, but that two clear approaches, LDA+ $U$  and PBEsol+ $U$ , offered the most advantages for defect properties and structural studies *vs* spin states, respectively. We found that in both cases  $U = 4$  eV gave results in good agreement with experiment. We demonstrated the applicability of these approaches by calculating the formation energy of an oxygen vacancy using LDA+ $U$ , finding excellent agreement with previous studies in the literature, and by determining the local structural variation and phonon mode splitting for different spin configurations, finding that the transition from LS to ordered HS-IS to HS resulted in good agreement with experiment. Our results demonstrate that simple DFT methods can be used to study complex features of LaCoO<sub>3</sub>.

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