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Laser performance of Cr⁴⁺:Y₂SiO₅

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Laser action in Cr^{4+} doped yttrium-orthosilicate Cr^{4+} : Y₂SiO₅ cooled to 77 K was demonstrated for the first time with 1064 and 532 nm pumping from a pulsed Nd: YAG laser and broad band (@ 840 nm) pumping from a *Q*-switched Cr:LiSAF laser.

In this letter we present, for the first time, the laser properties of a pure Cr^{4+} system, namely $Cr^{4+}:Y_2SiO_5$. We observed two interesting characteristics in this new laser: (1) the free running output spectrum depends on the pump wavelength, and (2) the output energy indicates some form of saturation at higher absorbed energies.

The discovery of laser action in the near-infrared (NIR) 1.2 μ m band in Cr:Mg₂SiO₄ (Forsterite)¹ and Ca,Cr:Y₃Al₅O₁₂,² pumped with 1064 nm radiation, indicated Cr⁴⁺ as the active ion^{2,3} responsible for lasing. However, existence of octahedral sites which accommodate Cr³⁺ ions in these crystals complicates a conclusive analysis of the role that Cr⁴⁺ plays in the laser action. In order to be able to address this problem, efforts have been underway for quite some time to search for new host materials for Cr⁴⁺ containing no octahedral site capable of accommodating Cr³⁺ ions.⁴

High quality Cr^{4+} : Y_2SiO_5 single crystals were grown by the Czochralski technique. The Cr-concentration in the melt was 2 at. % of the Si ions. The space group of Y_2SiO_5 is C_{2h}^6 (C2/c).⁵ The dimensions of the monoclinic cell are a=10.410 Å, b=6.721 Å, c=12.490 Å, $\beta=109^{\circ}39'$.⁶ Its melting point is 2070 °C. Cr^{4+} substitutes for Si⁴⁺ in a slightly distorted tetrahedral site. There is an octahedral symmetry at the Y³⁺ site, but size incompatibility of Cr^{3+} precludes it from entering that site.

The 77 K absorption and fluorescence spectra of the Cr^{4+} :Y₂SiO₅ sample used for the laser experiment are shown in Fig. 1. The characteristics of the fluorescence in the NIR depend on pump wavelength. The spectroscopy of Cr^{4+} :Y₂SiO₅ is discussed in more detail elsewhere.^{7,8}

The laser experiment with Cr^{4+} : Y_2SiO_5 is sketched in Fig. 2. It is a conventional longitudinal laser pumped laser cavity with a flat high reflector and a curved out coupler. The crystal used for the experiment was a parallelepiped with dimensions $5.4 \times 6.65 \times 12.5$ mm³. The end faces of the crystal as well as the fused silica windows of the liquid nitrogen dewar used in this experiment were antireflection coated between 1200 and 1300 nm.

The laser performance characteristics of the Cr^{4+} :Y₂SiO₅ crystal at 77 K are summarized in Table I.

However lasing at temperature up to 257 K was demonstrated.

Figure 3 is a plot of the single pulse output energy from the Cr⁴⁺:Y₂SiO₅ laser versus the absorbed energy for different pump wavelengths. With 1064 nm pumping (pulse duration 15 ns, beam diameter $\sim 400 \,\mu m$), the slope efficiency of the Cr⁴⁺:Y₂SiO₅ laser decreased drastically as the absorbed energy was increased above 1.0 mJ, showing a strong saturation behavior in the laser output. Similar saturation behavior is observed for 532 nm pumping (pulse duration 15 ns, beam diameter $\sim 400 \ \mu m$) when the absorbed energy is higher than 3.0 mJ. When pumped with 840 nm (pulse duration 35 ns, beam diameter \sim 400 μ m), on the other hand, the slope efficiency changes discontinuously to a smaller value at about 2.5 mJ of absorbed energy and the crystal continues to lase at this efficiency up to absorbed energy of about 7.5 mJ. The cause of this saturation behavior is yet to be determined.

The free running spectra due to 1064 nm pumping is distinct from that due to 532 nm pumping. The broadband pumping at 840 nm produces roughly the same output free-running laser spectrum (Fig. 4) as that due to 532 nm pumping. The pump wavelength dependence of the Cr^{4+} :Y₂SiO₅ laser indicates the existence of two types of emission in the NIR resulting from different excitations. The exact mechanisms involved in these emissions are not presently understood but are the subject of detailed spectroscopic studies in progress.

The near-infrared broadband luminescence and the laser action due to the 1064 nm excitation is attributed to the



FIG. 1. Absorption and fluorescence spectra of Cr⁴⁺:Y₂SiO₅ at 77 K.

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FIG. 2. Schematic diagram of the experimental arrangement for the laser experiment.

TABLE I. Laser performance characteristics of the free-running $Cr^{4+}{:}Y_2SiO_5$ laser at 77 K.

Dump			
wavelength (nm)	1064	532	840
Threshold		<u></u>	
energy (mJ)	0.32	1.5	0.93
Maximum slope			
efficiency (%)	6.3	12.4	8.2
Peak laser			
wavelength (nm)	1270	1225	1225
Spectral			
bandwidth (nm)	10	23	30



FIG. 3. Output-input curve of Cr^{4+} :Y₂SiO₅ laser at 77 K due to pumping with (a) 1064 nm and (b) 532 nm from a Nd:YAG laser, and (c) broadband @ 840 nm from a Cr:LiSAF laser.



FIG. 4. Free-running $\mathrm{Cr}^{4+}{:}\mathrm{Y}_2\mathrm{SiO}_5$ laser spectrum due to different pump wavelengths.

transition from the ${}^{3}T_{2}$ manifold of the Cr⁴⁺ ion to the ${}^{3}A_{2}$ ground state. It was observed that intensity of the nearinfrared fluorescence of the Cr^{4+} ion in Cr^{4+} :Y₂SiO₅ is much weaker than in forsterite $(Cr^{4+}:Mg_2SiO_4)$. This resulted in poorer laser performance of Cr4+:Y2SiO5 compared to forsterite and is attributed to the nearly ideal tetrahedral symmetry of the Cr^{4+} site in Cr^{4+} :Y₂SiO₅. The resulting symmetry selection rule forbids electric dipole radiative transitions between the ${}^{3}A_{2}$ ground state and the ${}^{3}T_{2}^{-}$ excited state in the ideal tetrahedral symmetry. The key to increasing the electric-dipole coupling between these two states is the degree of distortion of the $(CrO_4)^{-4}$ tetrahedron. As it becomes distorted the symmetry of the Cr^{4+} site is lowered, splitting the ${}^{3}T_{2}$ state into crystal field components and increasing the radiative transition probability between the ground state and one or more of the crystal field components of the ${}^{3}T_{2}$ state in the lower symmetry.⁷ Such effects are evident in forsterite in which the Cr^{4+} site has one of the lowest symmetry (C_s) in the tetrahedral group.⁹ This point is important in that it provides a guide to a systematic approach to the problem of developing more efficient Cr^{4+} doped crystals. It has been suggested that introduction of additional impurity elements having size and charge mismatch with the substituted ions in the host material may provide the necessary mechanism for distorting the $(CrO_4)^{-4}$ tetrahedron.⁷

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