

Determination of Phosphorus in Low-Alloy Steels by Charged-Particle Activation Analysis

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Determination of small amounts of phosphorus in low alloy steels has always presented a rather difficult problem. Since this element in the low-alloy steels is usually determined by spectrophotometry, it is unavoidable that determination procedures applied in the analysis are relatively complicated. In analyzing small amounts of phosphorus in unknown samples, as a matter of course, it is best to use a nondestructive technique because of the difficulty of dissolving the material without losing the trace element or introducing the contaminant. It can be expected that charged-particle activation analysis has an interesting capability for nondestructive analysis.

In a previous paper 1 , we have studied fundamentally the charged-particle activation analysis of phosphorus in some biological materials using the 31 P(α ,n) 34m Cl reaction. The knowledge of relevant thick-target yield curves obtained in there made it possible to choice the bombarding energy in order to produce the highest yield of the required activity and at the same time to suppress interfering activities as much as possible.

In the present work, nondestructive determination of phosphorus in some low-alloy steels were examined by charged-particle activation analysis using the same reaction as above.

Low-alloy steels were purchased from the National Bureau of Standard and the Iron and Steel Institute of Japan. All irradiation samples were rolled into metal plates having thick enough to stop the incident energy of alpha external beam. The low-alloy steel of NBS-461 was used as a most suitable calibration standard of phosphorus in a series of the present experiment. In order to reduce errors due to the nonhomogeneity and drift of beam, 12 samples were irradiated simultaneously by using a special designed rotating-target assembly. For measurements of the relative thick-target yields of interference elements in the low-alloy steel such as nickel, copper and iron, the target holder inserted 12 different aluminum absorbers was used to degrade the energy of incident alpha particles. The degraded energy of each absorber was calculated on the basis of the range-energy relationships given in literature. The rotating targets to measure the relative thick-target yields were bombarded with 1.5 μ A beam of 22 MeV alphas for 32 min, whereas the sample to determine the phosphorus concentration were irradiated with 4 μ A beam of 17 MeV for 32 min.

Radioactive nuclides produced were identified by gamma-ray spectrometry using a high-resolution Ge(Li) detector connected to a multichannel pulse height analyser. In the cases of relative thick-target yield measurements, the number of counts in the relevant area below photopeak were corrected by means of the

counting efficiency of detector and the gamma-ray branching ratio.

A typical gamma-ray spectrum recorded from the NBS-461 low-alloy steel A is given in Fig. 1. As seen in Fig. 1, it is obvious that the Compton background of positron annihilation due to 61 Cu, 63 Zn, 66 Ga and 57 Ni, which are produced mainly through the 58 Ni(α ,p) 61 Cu plus $(\alpha,n)^{61}$ Zn \rightarrow 61 Cu, 60 Ni(α ,n) 63 Zn, 63 Cu(α ,n) 66 Ga and 54 Fe(α ,n) 57 Ni reactions, in particular prevents the determination of phosphorus at low concentration levels. On the assumption that the low-alloy steel contains one percent of each element as a constituent, the relative thick-target yields of radioactive nuclides produced from nickel and copper were measured, and the results obtained as a function of alpha energy are shown in Fig. 2, together with those of the 31 P(α ,n) 34 mCl reaction. In Fig. 2, those of 57 Ni produced from the sample in which no other elements contain are also given. From the analytical point of view, it is suggested that the determination of phosphorus seriously depends on the relative abundance of nickel in the sample. Since the presence of large amounts of nickel is undesirable as described above, it may be concluded that the present method is unsuitable for high-alloy steel samples.

On the other hand, the accuracy of the present method was examined by using a number of low-alloy steel samples with widely varing concentration of phosphorus. The results of analyses as given in Table 1 were excellent agreement with the certified values given by NBS and ISIJ. From these experimental data, it was confirmed that the phosphorus concentration in the low-alloy steel can be determined nondestructively and accurately by the present method. The detection limit for a 5 $\mu\text{A}-\text{one}$ half-life irradiation was found to be 0.3 μg of phosphorus.

On the basis of the above results, it is concluded that the present method is applicable to a wide variety of low-alloy steels as a useful complement to other analytical methods.

Reference

 Masumoto K. and Yagi M., J. Radioanal. Chem., (in press), CYRIC Ann. Rept. (1981) 106.

Table 1. Concentration of phosphorus in low-alloy steels.

Sample	Concentration of P (%)				
	Found		Average	Certified value	
	0.0166,	0.0160			
	0.0156,	0.0166	0.0161±0.0004	0.017	
NBS-466	0.0109,	0.0117	0.0116±0.0007	0.012	
	0.0125,	0.0111			
JSS-150-6	0.0448,	0.0430	0.0439±0.0009	0.044	
JSS-151-6	0.0282,	0.0286	0.0284±0.0002	0.028	
JSS-152-6	0.0286,	0.0242	0.0264 ± 0.0022	0.026	
JSS-153-6	0.0072,	0.0100	0.0086±0.0014	0.012	
JSS-154-6	0.0062,	0.0064	0.0063±0.0001	0.007	
JSS-155-6	0.0073,	0.0057	0.0065±0.0008	0.006	
JSS-163-3	0.0228,	0.0218	0.0223±0.0005	0.020	
JSS-164-3	0.0581,	0.0646	0.0614±0.0035	0.058	

JSS: Japanese standard of iron and steel

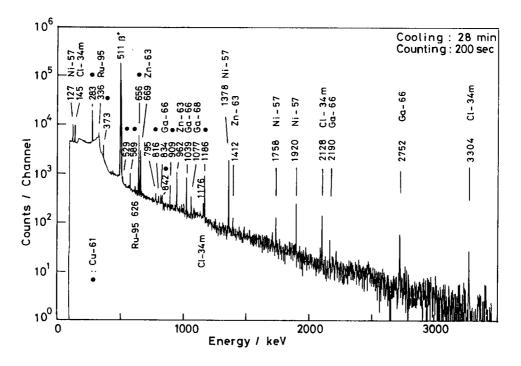


Fig. 1. Gamma-ray spectrum of irradiated NBS-461 low-alloy steel.

