

*Article*

In-situ Observation of Martensite Decomposition in HAZ of Cr-Mo Steel Weldment

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Abstract. In-situ observation of martensite decomposition at Heat Affected Zone (HAZ) was investigated on a dissimilar joining between 2.25Cr-0.5Mo grade T22 as base material and ER90S-B9 as filler metal using GTAW process using LEEM at a synchrotron facility. A post weld heat treatment (PWHT) cycle was simulated on a welded specimen in high vacuum chamber by heating cartridge and electron bombardment. Both effects PWHT duration and weld areas were studied for comparisons. At the simulated PWHT between 690°C -700°C in CGHAZ, martensite started to decompose by the dissolution of carbide flakes. The prior-austenite grain boundaries were also shown during the process. The same phenomena were also observed in FGHAZ with different extent. In un-affected base material, ferrite and new pearlite grains presented and grew at the expense of old pearlite. Longer PWHT duration resulted in more ferrite formed in all weld areas. Raising PWHT temperature to 730°C could push the reaction above Eutectoid temperature as the new austenite formed at grain boundaries. The proposed mechanism for martensite decomposition would be in steps as dissolution of carbide followed by formation of ferrite and growth as PWHT proceeded.

Keywords: PWHT, heat-affected zone (HAZ), martensite decomposition, LEEM, in-situ observation.

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1. Introduction

In steam turbine power generation, Cr-Mo steels have been selected owing to their elevated temperate strength and creep resistance [1-5]. Several grades have been used such as 2.25 Cr-1Mo (grade T22) for general application, and 9Cr-1Mo (grade T91) for higher operating temperature. In steam piping assemblies or equipment, there are process steps working at different temperatures such as economizer, heater, and super heater which require dissimilar joint among materials both during fabrication and repair. This could cause a concern during welding and post weld heat treatment (PWHT) in which soft zone and hard zone could occur at the weld interface in Heat Affected Zone (HAZ) and weld metal, respectively. Many researchers studied microstructures of the soft zone [4-12] as well as performed mechanical characterization in dissimilar metal weldments [4, 8-16]. In as-welded condition, martensite structure was found in heat-affected zone (HAZ). However, the martensite was changed to ferrite after PWHT, forming a soft zone. It has been proposed that this behavior is due to the carbon migration during PWHT [6-8, 13, 17-20] as the result of the affinity of Cr.

It has been believed that the formation of soft zone in the HAZ could cause the failure during in service of the material especially in creep mode. This was shown in the study of creep life assessment of Cr-Mo material by Petchsang et. al. [21] as the failure occurred close to the HAZ. Another investigation related the changes in local structures showed the decomposition of martensite after extend period in PWHT [22].

With many investigations stated above, those were performed in ex-situ investigation. There was still no in-situ investigation which could help provide information, in term of mechanism, for decomposition of martensite into ferrite in the HAZ of dissimilar welding between high and low Cr contents. This study shows sequences on microscopic transformations of the martensite to ferrite in real-time using Low Energy Electron Microscope (LEEM). A specimen from T22 welded with ER90S-B9 was heated under the simulated PWHT condition and the microstructures were observed at various times and locations.

2. Methods

2.1. Materials and Welding Process

The material used as base metal was 2.25Cr-1Mo steel tube (T22) with dimensions 65 mm (O.D.) \times 6.5 mm (thickness). The nominal and measured compositions of the base metal and the deposited Weld Metal (WM) are listed as shown in Table 1 and Fig. 1 by using the Scanning Electron Microscopy (SEM) and an Energy-Dispersive Spectroscopy (EDS) was performed in order to determine the alloying composition. The specimen was prepared using Gas Tungsten Arc Welding (GTAW) process with the ER90S-B9 filler metal. An overlay joint was selected to deposit the weld metal on the wall thickness of the tube as shown in Fig.1. Welding parameters implemented were 60 A with 11-12V using DCEN polarity.

Table 1. Chemical composition of the material and welding wire used

Material	Composition	Element (wt-%)			
		Cr	Mo	Mn	Fe
2.25Cr-1Mo	nominal ⁽¹⁾	1.90-2.60	0.87-1.13	0.30-0.60	95.1-96.8
base metal	measured ⁽³⁾	2.4	1	0.6	96.0
ER90S-B9	nominal ⁽²⁾	8.00-9.50	0.80-1.10	1.25	87.3-90.3
weld metal	measured ⁽³⁾	4.1	1.1	0.4	94.3

Note: (1) referring to [23]; (2) referring to [24]; (3) measured in by SEM/EDS analysis.

The welded tube was cut in longitudinal direction to reveal the cross section of deposited weld metal in the size about 5 mm long and 3 mm wide (Fig. 2). Conventional preparation for microstructural examination was then performed by sandpaper No. 100, 320, 400, 600, 800, 1000 and 1200 successively. In this case, the specimen must be grinded to the thickness less than 1 mm (Fig. 3(a)) to reduce the heat mass in order to minimize the chance of outgassing and spontaneous arc during LEEM operation. To reveal its microstructure in as-welded condition and locate the weld interface prior observation, the specimen was etched using 10 ml HNO₃ + 20ml HCl + 30ml water for 60 seconds. The weld interface was then marked

by a micro-vicker hardness tester using 1 kgf with a spacing of 300 μm (Fig. 3(b)). Then, the surface was polished again with 1000 and 1200 sandpaper following by 1 μm diamond.

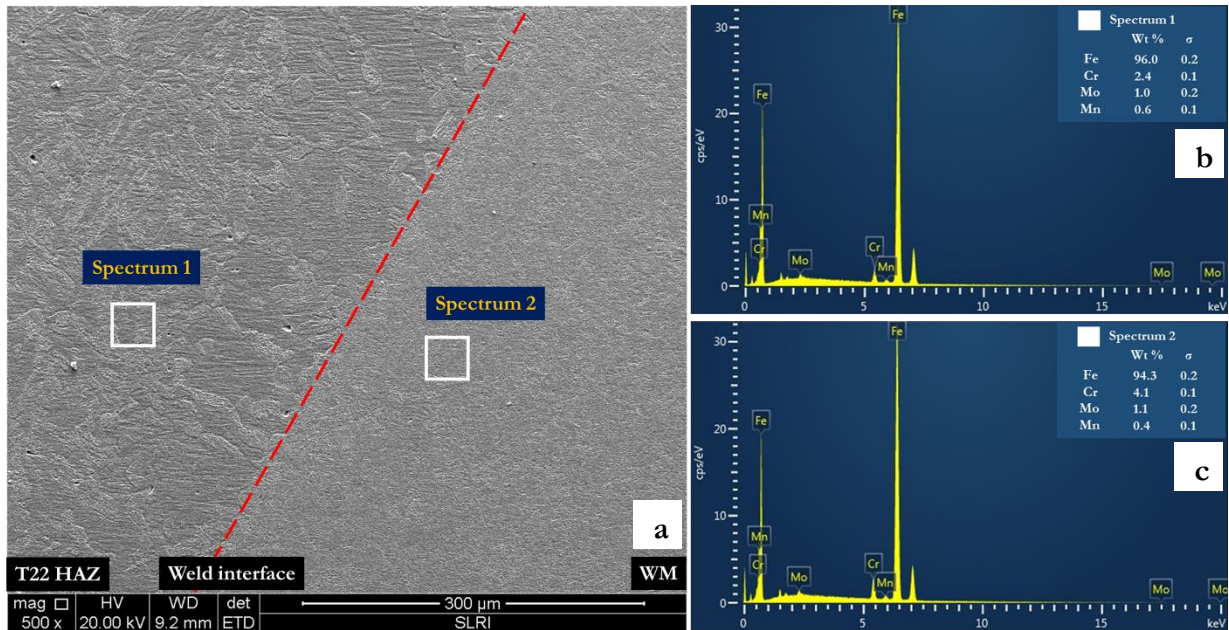


Fig. 1. (a) SEM results of microstructure in weld interface; (b) to (c) the EDS results of alloy composition in T22 HAZ and weld metal side adjacent to the fusion line, respectively.

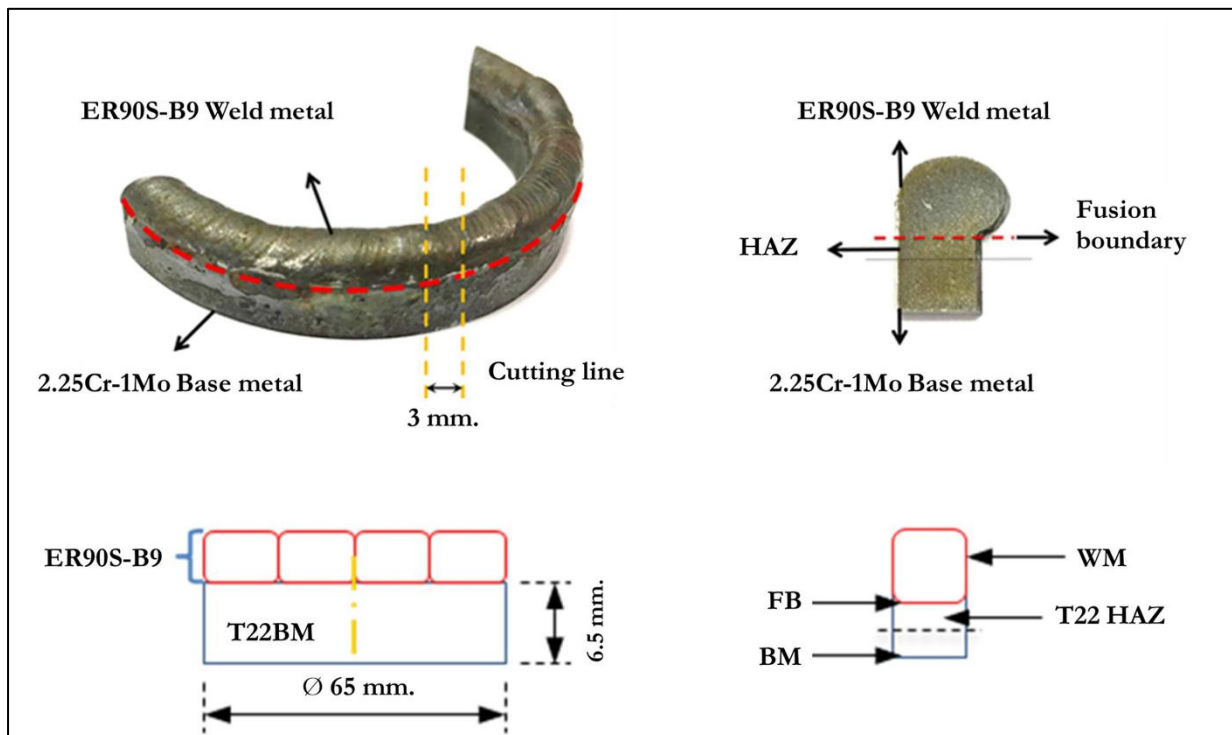


Fig. 2. 2.25Cr-1Mo steel welds were performed using GTAW process by ER90S-B9.

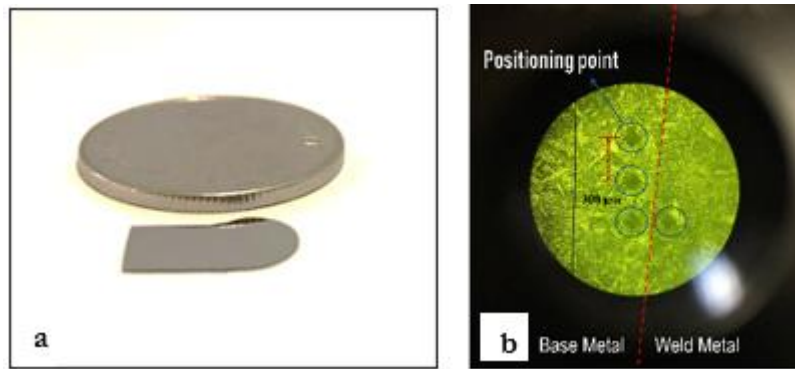


Fig. 3. (a) The specimens passed through polish; (b) Positioning by micro-vicker hardness tester.

2.2. 2.2 Observation

In this research, Low Energy Electron Microscope (LEEM) was utilized at the experimental station BL 3.2Ub at the Synchrotron Light Research Institute (Public Organization), Nakhon Ratchasima, Thailand as shown in Fig. 4. To study the martensite decomposition at HAZ of T22 material, the polished specimen was cleaned using ethyl acetone and followed by alcohol in an ultrasonic cleaner for 15 minutes. Then it was mounted in the sample holder (Fig. 5) and sputter-cleaned by Ar^+ ion under vacuum before being loaded into the analysis chamber. This operation was performed in high vacuum state at the level of 10^{-9} torr or below. Heating was introduced by both radiative from W filament and electron bombardment on the back of the specimen inside the holder. The temperature was measured by a pyrometer with the emissivity of Cr-Mo steel set at 0.28. As shown in Fig. 6, four positions of weld areas were observed during the experiment.

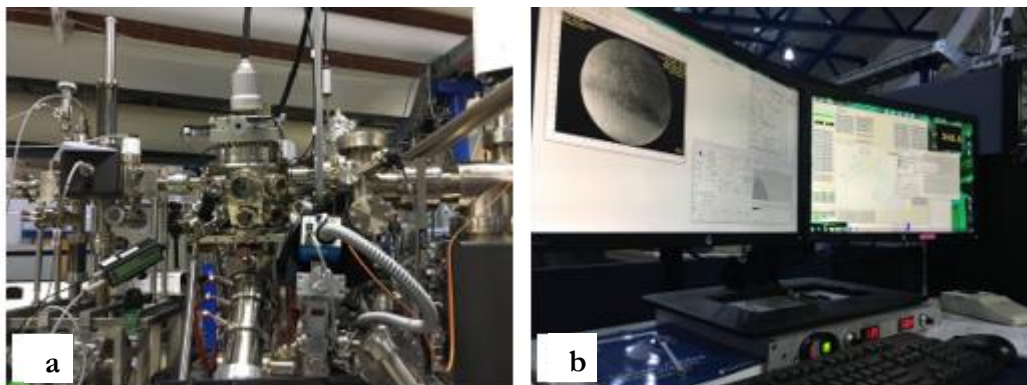


Fig. 4. (a) LEEM system operation; (b) Display during LEEM experiment.

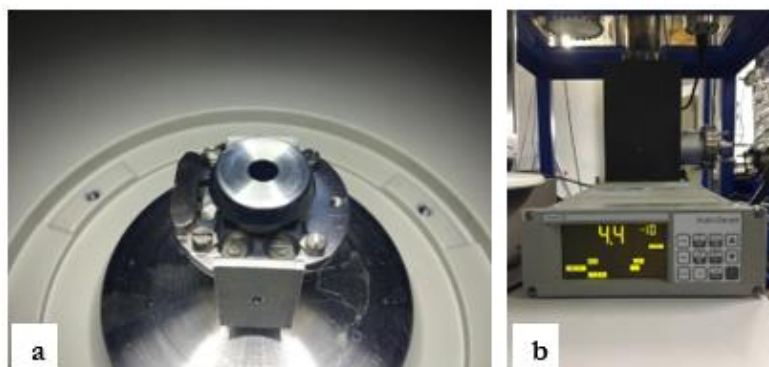


Fig. 5. (a) The specimen in the sample holder LEEM system; (b) pressure display on the internal pressure monitor.

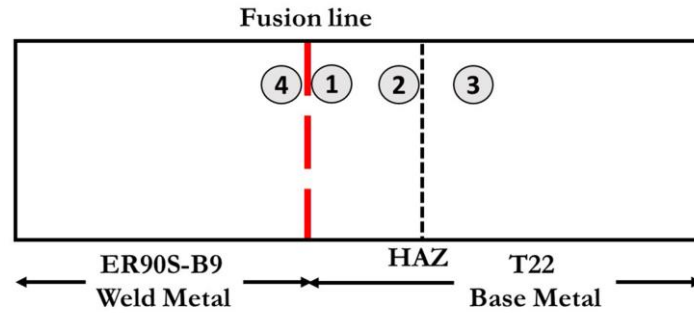


Fig. 6. Position of the weldment in each area:

1. Position 1: Coarse-grained heat affected zone (CGHAZ).
2. Position 2: Fine-grained heat affected zone (FGHAZ) 250 micron away from the fusion line.
3. Position 3: T22 Base metal (BM) 500 micron away from the fusion line.
4. Position 4: Heat affect zone in weld metal side adjacent to the fusion line

Figure 7 showed the heating history during the experiment. In order to simulate the condition of PWHT, the specimen was heated from room temperature to below the Eutectoid temperature (A1) at 690°C as recommended by ASME BPVC as minimum of 675°C for T22 (P-No. 5A Group No.1) and 705°C for ER90-B9 which could be comparable to T91 (9Cr 1Mo). However, since the PWHT temperature was determined by base material which T91 was not used in this case, therefore, the only material concerned would be T22 which required 675°C. In addition, using PWHT temperature of ER90S-B9 would be too high for T22 as it might be above the eutectoid point which material would become austenite and subsequent martensite.

The temperature was held between 690°C - 700°C for 14 hrs. under LEEM to observe effect of PWHT duration on microstructure. All 4 areas were observed from time to time for comparisons on PWHT effects on each weld area. Finally, the temperature was raised to 730°C which was above the Eutectoid. This was to determine the change of microstructure into the 2 phase-region or the starting of the decomposition of ferrite to form austenite.

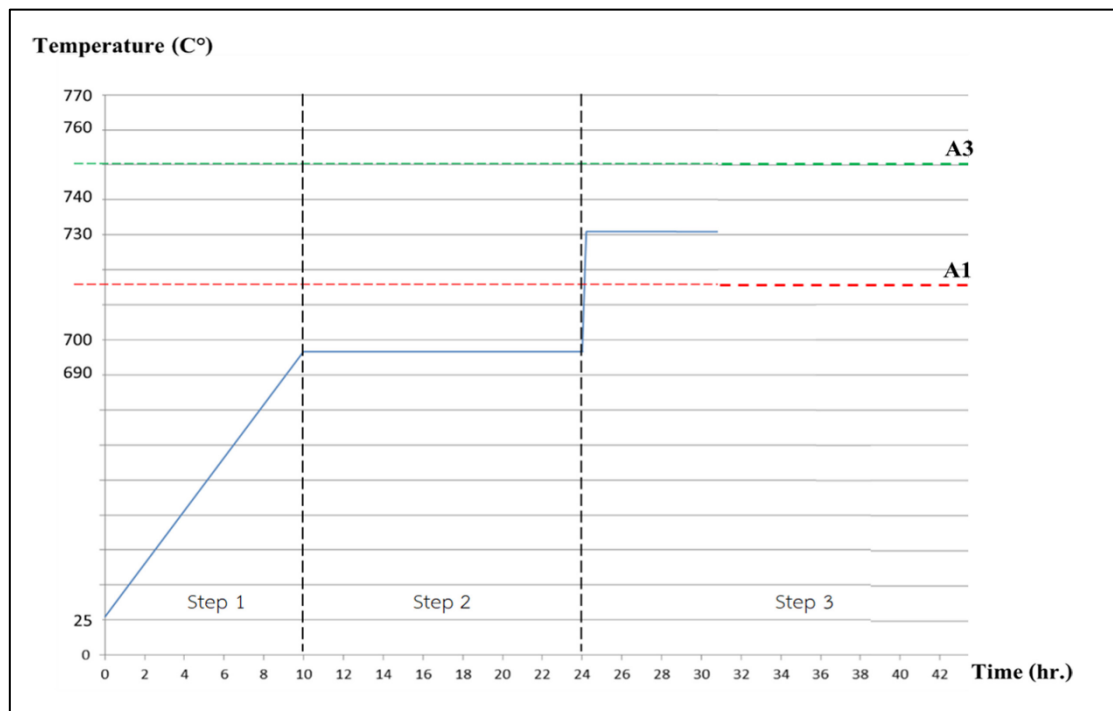


Fig. 7. Simulated PWHT history.

3. Results and Discussion

In this research, the effect of duration (time of PWHT) and distance from the fusion boundary (weld areas) were observed. Starting at room temperature, the microstructure of 2.25Cr-1Mo base material (position 3) consisted of ferrite, pearlite and carbide precipitate both within grain and along the grain boundaries [4]. Upon welding with GTAW using ER90S-B9 filler metal, the heating from welding on the specimen created different welding areas due to different in thermal history from various peak temperatures depending on distance from fusion boundary. This resulted the gradient of microstructures.

In CGHAZ (position 1), there was martensite structure resulted from fast cooling of prior-austenite in which the peak temperature was above the critical temperature. On the other hands, in FGHAZ (position 2), the peak temperature was between 2 phases among austenite and ferrite. This resulted in ferrite and martensite mix from transformed prior-austenite. The near fusion boundary weld metal (position 4) as a mixture between ER90S-B9 and melted base material was melted and re-solidified as martensite structure with columnar morphology. Since this area possibly was not mixed completely in macroscopic scale, the chemical composition especially for Cr would be high compared to CGHAZ.

The in-situ recorded micrographs by LEEM were shown in Table 2. For easy understanding changes shown the following abbreviations were used.

WM	: Weld Metal
Pr-A	: Prior Austenite
MxCy	: Carbides both from precipitations and formed inside martensite
F	: Ferrite
P	: Pearlite
A	: Austenite

As the simulated PWHT cycle began, prior-austenite grain boundaries started to present in CGHAZ at temperature about 650°C as shown in Table 2:1a-bottom. In the same way in Table 2:1b and 1c, both in FGHAZ and BM showed the prior-austenite grain boundaries. This areas martensite and its mixtures (ferrite + martensite) due to fast cooling from austenite phase upon welding. The base material was possibly martensite mixture as well due to its chemical composition. This also occurred in weld metal (Table 2:1a – top) with finer features and carbides.

When the temperature rose to between 690°C - 700°C, prior-austenite grain boundaries were even more pronounced as shown Table 2:2a-bottom. This phenomenon was similar to what it's called "Tinned Etching" in which the prior features could be shown upon heating. Carbide flakes in martensite started to dissolve showing clearer prior-austenite grain. The ferrite grain started to be more pronounce as the prior-austenite grain boundaries started to bend slightly. On the other hand, in FGHAZ, all features of microstructures were gone, while in BM there started to present the formation of ferrite and pearlite as more equilibrium phase condition achieved. The features in weld metal (Table 2:2a-top) was also more pronounced with a mix of small size phases and microstructures. The columnar morphology started to become rounded as inhering the effect of solidification direction from welding.

After holding at the temperature between 690°C - 700°C for 6 hours, there were little changes in CGHAZ as it reached the completion of martensite decomposition in which no more carbide flakes to be dissolved. The grain size started to grow bigger as the pinning effects had gone from decomposition of martensite (Table 2:3a). This was also due to the higher driving force to become larger prior-austenite grains together with residual stress from welding in this area. As a result, the soft zone with lower hardness was formed in this location corresponding to S. Petsang [8, 21, 22] and C. Sudha et al. [14]. In FGHAZ (Table 2:3b), ferrite grain boundary was somewhat shaper as well as feature of pearlite was more prominent. At this temperature, the equilibrium phase for the T22 material was more ferrite and a few pearlite. In the same way, in weld metal, microstructure features started to grow consuming each other. On the other hand, in base material the ferrite grains began to grow noticeably consuming some pearlite and grain boundaries became more prominent as shown in Table 2:3c.

After further holding at the PWHT temperature to 14 hours, ferrite grain boundaries in CGHAZ continued to grow as shown in Table 2:4a. This was also occurred in weld metal in which the microstructure features continued to grow. The observation might contradict to the previous literatures which reported that the carbides should form and become the hard zone. However, in this case, the temperature used for simulating the PWHT temperature was about 700°C which was higher than recommend for T22 material or P-No. 5A Group No. 1 (according to ASME BPVC) at 675°C. Since the temperature was well beyond

the recommendation which was close to the Eutectoid point, carbides were more prone to dissolution than precipitation. Therefore, its microstructures were complete matrix features with un-noticeable precipitates. However, another possibility, the carbides might form during cooling that could be observed in ex-situ observation which was not the case for this experiment. In FGHAZ, the pearlite was even more pronounced. As can be seen from Table 2:4b, the laminar pattern of pearlite showed with sharper grain boundaries. For BM, carbides in pearlite were also dissolved and new ferrite and pearlite formed consuming the prior features to reach the equilibrium stage between ferrite and pearlite conforming to the chemical composition of the material.

Next, the specimen was heated to 730°C in order to ensure that the simulated PWHT temperature was not beyond Eutectoid temperature, that could compromise the observation results, as well as to continue the observation of phase transformation at the 2-phase region of Fe-C phase diagram in which both ferrite and austenite should present together.

After holding the temperature at 730°C for 1 hours, the austenite phase started to form in both CGHAZ and weld metal splitting ferrite grains (at the triple point junctions) as shown in Table 2:5a-top and bottom. In the same way, both FGHAZ and BM experienced the formation of austenite phase starting at the triple point junctions of grain boundaries. Both of existing ferrite and pearlite, accordingly, were consumed by newly formed austenite/ferrite mixed microstructures. After further holding the temperature at 730°C to 7 hours, austenite grains were larger in all locations as shown in Table 2:6a-6c. This indicated that the temperature was well beyond the Eutectoid point at the 2-phase region. This information could also be applied to infer that the too high PWHT temperature could result in an adverse effect on the properties of the weldments in all weld areas, as the newly formed austenite could transform into martensite from being rapidly cooled to room temperature after completion of the PWHT procedure. It needs to be aware not to exceed the Eutectoid temperature as most codes and standards always indicate the minimum PWHT temperature but not recommend the maximum PWHT temperature boundary.

From the information above, the decomposition mechanism of martensite in weld areas especially in CGHAZ could be suggested. Upon, heating as PWHT progress, carbides in martensite started to dissolve into the matrix. This resulted in instability of martensite which then transformed to become ferrite. The dissolution of carbides could also free the grain boundaries to move as grain growth processed. Depending on the thermal history of each weld area, ferrite formed could grow and consume prior microstructure features both in the as-received in BM and as-welded in heat affected zone. In addition, the results were also consistent to other studies that the longer the PWHT duration and the higher the temperature of PWHT were, the larger the grain size was [14, 25]. Nevertheless, S. K. Albert et al. [6] also indicated that the width of microstructure features was affected by PWHT duration which corresponded to the results of observations.

4. Conclusion

The in-situ observation by LEEM of martensite decomposition in soft zone or CGHAZ of Cr-Mo steel grade T22 welded by ER90S-B9 could be concluded as follows:

1. Martensite decomposition started by dissolution of carbides in martensite flakes, followed by formation of ferrite phase in the CGHAZ.
2. After complete decomposition of the martensite, ferrite phase expanded and grew to form the soft zone.
3. FGHAZ also experienced the same situation as in CGHAZ but at different extent while the unaffected BM behaved differently as the ferrite grew, consuming the pearlite.
4. Longer PWHT duration resulted in larger grain size in all weld areas.
5. Too high PWHT temperature could result in formation of new austenite grains that could have adversary effects on weldments upon cooling.

Table 2. Shows in-situ recorded micrographs by LEEM

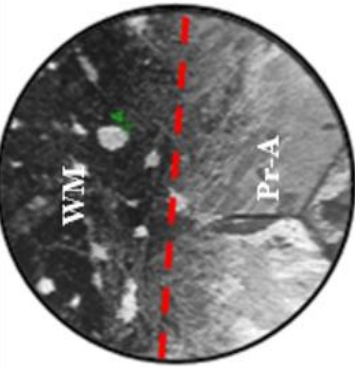
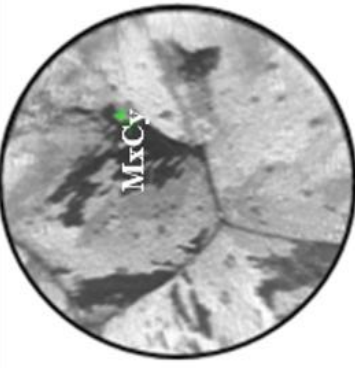
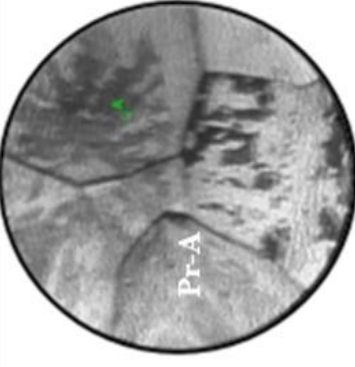
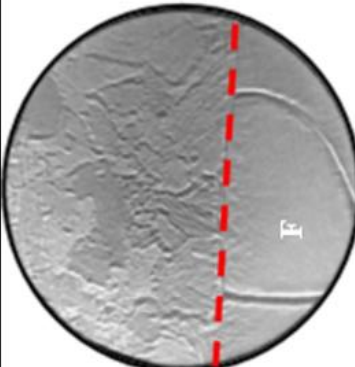
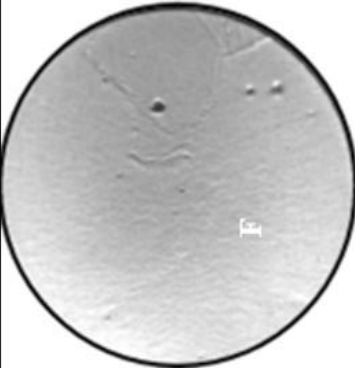
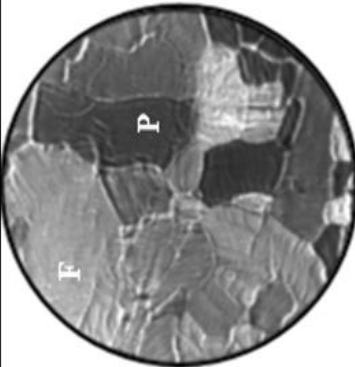
		LEEM images Zone Start voltage at ≈ 0.40 V, 75 μm field-of-view (FOV)		
Temp ($^{\circ}\text{C}$)	Position 1: CGHAZ (bottom) Position 4: Weld Metal at fusion boundary	Position 2: FGHAZ	Position 3: BM	
(1) Heating $T < T_{A1}$ 590 $^{\circ}\text{C}$	 <p>1a</p> <ul style="list-style-type: none"> - Martensite phase in CGHAZ. - Columnar morphology in WM. 	 <p>1b</p> <ul style="list-style-type: none"> - Prior-austenite grain growth at the expense of martensite and carbide dissolution. 	 <p>1c</p> <ul style="list-style-type: none"> - Prior-austenite grain growth at the expense of martensite and carbide dissolution. 	
(2) Heating $T < T_{A1}$ 690-700 $^{\circ}\text{C}$	 <p>2a</p> <ul style="list-style-type: none"> - Martensite in CGHAZ, prior-austenite grain completed decomposition showing ferrite grain. - In WM, as-welded features decomposed and a mix of small size phases presented. 	 <p>2b</p> <ul style="list-style-type: none"> - Complete dissolution of carbide and ferrite phase formed. 	 <p>2c</p> <ul style="list-style-type: none"> - Complete dissolution of prior features and finer phases formed, possibly ferrite and austenite. 	

Table 2. Shows in-situ recorded micrographs by LEEM (Continued)

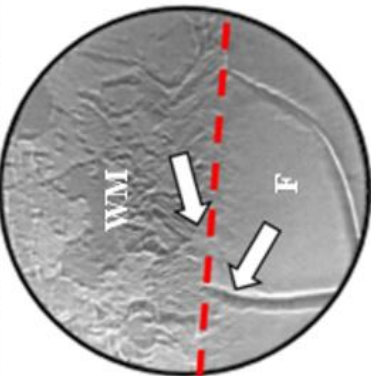
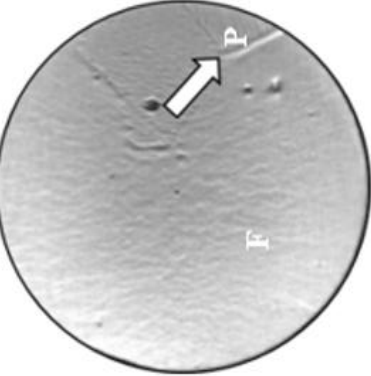
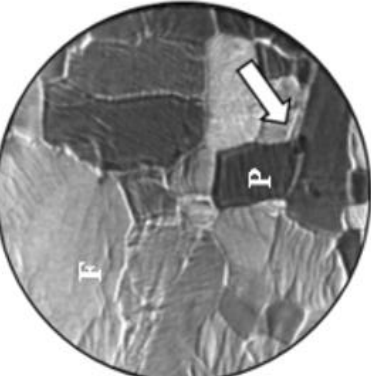
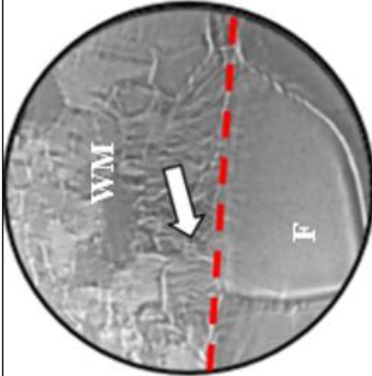
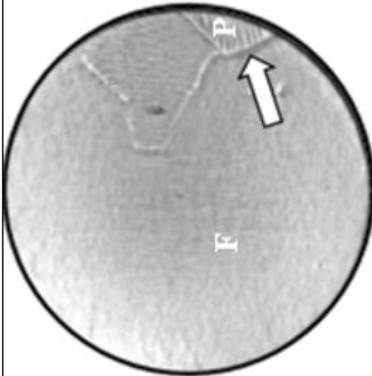
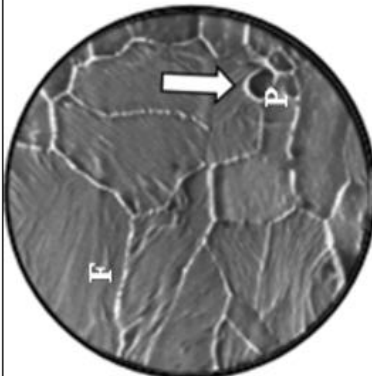
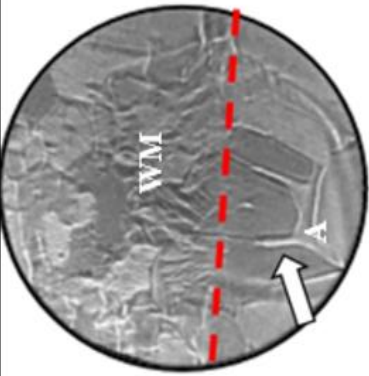

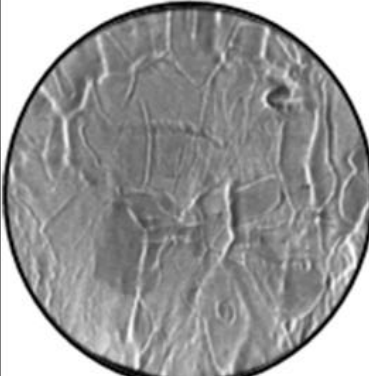
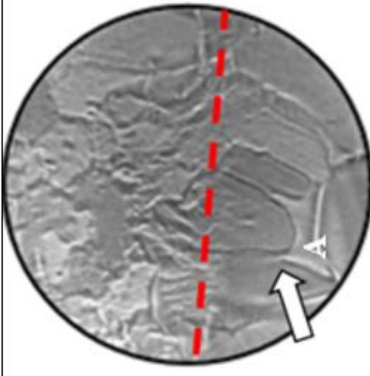

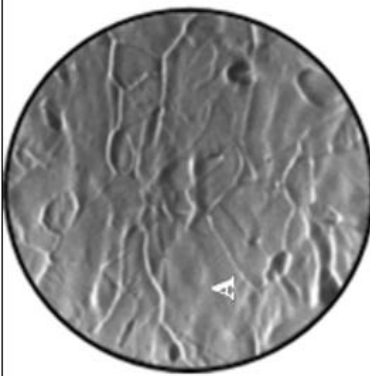
Temp (°C)		LEEM images Zone Start voltage at ≈ 0.40 V, 75 μm field-of-view (FOV)		
(3)	Heating $T < T_{A1}$ 690-700 °C @ 6 hrs.	Position 1: CGHAZ (bottom) Position 4: Weld Metal at fusion boundary	Position 2: FGHAZ	Position 3: BM
		 <p>3a - Grain growth both in CGHAZ (ferrite) and features changed in WM.</p>	 <p>3b - Slight grain growth and more prominent feature of pearlite (arrow).</p>	 <p>3c - Formation of ferrite and pearlite. - Ferrite growth consuming pearlite (arrow).</p>
		 <p>4a - Ferrite grain growth in CGHAZ. - Slightly larger features in WM.</p>	 <p>4b - More pronounced cementite features in pearlite.</p>	 <p>4c - Ferrite growth consuming old pearlite.</p>
	(4) Heating $T < T_{A1}$ 690-700 °C @ 14 hrs.			

Table 2. Shows in-situ recorded micrographs by LEEM (Continued)

LEEM images Zone Start voltage at ≈ 0.40 V, $75 \mu\text{m}$ field-of-view (FOV)	
Temp (°C)	Position 3: BM
(5) Heating $T_{A1} < T <$ T_{A3} $730 \text{ }^\circ\text{C}$ @ 1 hr.	Position 1: CGHAZ (bottom) Position 4: Weld Metal at fusion boundary  <p>5a - Austenite grain started to form both in WM and CGHAZ.</p>
	Position 2: FGHAZ  <p>5b - Austenite formed at the pearlite grain boundaries.</p>
(6) Heating $T_{A1} < T <$ T_{A3} $730 \text{ }^\circ\text{C}$ @ 7 hr.	 <p>5c - Austenite phase formed superimposed on ferrite/pearlite matrix.</p>
	 <p>6a - Austenite growth at the expense of ferrite.</p>
	 <p>6b - Extensive austenite growth.</p>
	 <p>6c - Growth of austenite phase.</p>

Acknowledgments and Legal Responsibility

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