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Article

Screening and Selection of Ferric Reducing Bacteria by Electrical Current for Microbial Fuel Cell

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Abstract. The efficiency of microbial fuel cell (MFC) performance depends on the competence of microorganisms on being an essential role in primarily converting organic compound into electricity. In this study, the possibility of using electrical current to select electrochemical active bacteria from sub-sediment for use in MFC was conducted. By using an alternating current (AC) of 0.6-12 mA and selective media, 16 Gram-positive ferric reducing bacteria (GP-FRB), 15 Gram-negative ferric reducing bacteria (GN-FRB) and 9 Gram-negative non FRB (GN-nonFRB) were characterized. GN-FRB and GP-FRB were obtained from the current of 0.6-6 and 9-12 mA, respectively. After tested in MFC, GN-FRB had a greater current density and power density than those from GN-nonFRB and GP-FRB. However, the greatest voltage was obtained from GP-FRB, followed by those of GN-nonFRB and GN-FRB, respectively. The highest current and power density of 13.33 mA/m² and 0.32 mW/m², respectively were from GN-FRB namely KL14 which was identified later as *Proteus* sp. This research could contribute a promising method for screening and selection of ferric reducing bacteria using electrical current. MFC inoculated with our selected bacteria could be a model for next study in wastewater treatment.

Keywords: Ferric reducing bacteria, microbial fuel cell, electrical current selection.

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1. Introduction

Microbial fuel cells (MFC) are the future trend of wastewater treatment technology which not only reproduce water but also generate electricity simultaneously. MFC can directly convert organic substances in wastewater into electricity by catalytic activity of microorganisms [1]. They oxidize organic substances and extracellular electron transfer (EET) to anode of MFC; consequently, electricity is generated by the flow of electron from anode across circuit to cathode. Electrical current of MFC depends on the efficiency of microorganisms to extracellularly transfer electron to anode [2]. In general metabolic pathway of microorganisms, electron transferred to oxygen as a main electron acceptor to generate ATP by oxidative phosphorylation, but there are groups of bacteria that could transfer electron to electron acceptor outside the cell. According to the metabolism of ferric reducing bacteria (FRB) in anoxic environment they will seek for another electron acceptor other than oxygen to complete its electron transport chain. Their electron will extracellularly transfer to soluble Fe(III) and insoluble Fe(III) oxide in anoxic sediment [3-4]. Whereas, in anoxic anodic compartment, FRB such as *Shewanella putrefaciens* can oxidize organic substances and efficiently transfer electron to anode [5]. Additionally, electrochemical activity of bacterial cell is an important concern for EET and determine also the ability of cell to transfer to anode. Only *S. putrefaciens* in anoxic but not in aerobic culture has an electrochemical activity and efficiently transfer electron to anode [6]. More extensive studies found in *Geobacter* spp. and *Shewanella* spp. that their electron can export out of the cell via cytochrome and pili to anode in anoxic condition [7]. The other mechanism of EET is self-mediated electron transfer which transfer electron to anode by redox compounds as electron shuttle between cell and anode. This occurred by self-secreted mediator bacteria such as *Ps. aeruginosa* which can transfer electron to anode by its pyocyanin compound [8-9]. Selection of electrochemical active bacteria (EAB) are capable of performing wastewater treatment and have effective EET, therefore should be selected and their potential will be suitable for use in MFC.

EAB could be isolated from anoxic environment or from anodic biofilm of wastewater feeding MFC. In anoxic sediment, *Geobacter* spp. and *Shewanella* spp. which are dominant genera of FRB were isolated by using Fe(III) oxide as electron acceptor and using acetate and lactate as electron donor, respectively [5, 10-11]. In addition, *Rhodospirillum rubrum*, a psychrotolerant FRB isolated from sea-sediments could directly utilize glucose to electricity in MFC [12-13]. EAB such as *Clostridium butyricum* [14], *Aeromonas hydrophila* [15], *Rhodospseudomonas palustris* [16], and *Orchrobactrum anthropi* [17] were isolated from anodic biofilm of MFC. Dominant bacterial species in biofilm depend on source of wastewater feeding into anodic compartment of MFC [18-19]. After MFC fed with wastewater for long periods, mix bacterial communities will form biofilm on anode and these biofilms are responsible for electron transfer to electrode while treating wastewater [20-22]. In addition, EAB selection by electric potential was demonstrated to be possible. Poising electrical potential at anode was applied in benthic microbial fuel cell (BMFC) for facilitating EET of bacteria in sediment attaching on electrode thoroughly [23]. Poised potential can reduce lag time of the pure culture and mix consortium in producing current; however, there is no difference in the maximum current output [24-25]. Furthermore, negative poised potential (V vs a standard hydrogen electrode, SHE) are useful to enrich *Geobacter sulfurreducens* on anode [26]. Moreover, direct current (DC) was also claimed to be used to select EAB for use as inoculums in anodic compartment of MFC [27].

Based on natural selection, the bacteria survive in selective pressure will have higher performance. In other words, they have to survive during current flow in and out of the cells. Hence, we expect that the bacteria survive under current biasing should have higher (electrochemical) performance than the others. Therefore, in this research, a constant alternating current (AC) poising on stainless-steel electrode will be used to select the electrochemically active microorganisms from sea sediment. To characterize the electrochemical performance, ferric reduction activity will be studied. Our selected microorganism should facilitate electron flow towards the electrode and will be used in MFC practicably.

2. Materials and Methods

2.1. Screening and Selection of Ferric Reducing Bacteria

Sea sediments nearby Ko Lan, Chonburi province, Thailand were collected. All samples were stored at 4°C before use. A random sample of 5 g was inoculated into 250 ml flask containing 150 ml of Nutrient broth (NB). To study the role of electrical current to the bacteria, we have developed the following protocol. A

couple of stainless steel (316L) electrode with the surface area of 7.1 cm² was put into the flask. It was used as stimulating electrode. Then paraffin oil was added as a top layer in order to generate an anaerobic condition. Subsequently, the 50 Hz AC generator from a laboratory-made current source was applied to the stainless steel electrode. To investigate the effect of the current, the amplitude of the current was varied from 0.6, 3, 6, 9 and 12 mA (or 0.084, 0.42, 0.84, 1.27 and 1.69 mA/cm²). After incubation for 5 days, electrodes were transferred into the new 250 ml flask contained 150 ml of NB covered with paraffin oil as indicated in Fig. 1. The biofilm coated electrode was incubated under the same electrical current selection for another 5 days. The same procedure was carried out every 5 days for 30 days. After incubation in alternating current for 5th transferring (30 days), the biofilm on the electrode was spread on NA and NA with 20 mmol/l of ferric citrate. It was incubated in anaerobic chamber, GENbox anaer (bioMérieux, France), at room temperature for 5-7 days. Mixed bacterial colonies were restreaked until pure culture was isolated. Pure cultures of selected bacteria were streaked on NA supplemented with 20 mmol/l ferric citrate, and incubated in anaerobic chamber, GENbox anaer, at room temperature for 5-7 days. The color change on NA due to ferric citrate from the reddish-brown color into the light-green color was used to evaluate the ferric reduction activity.

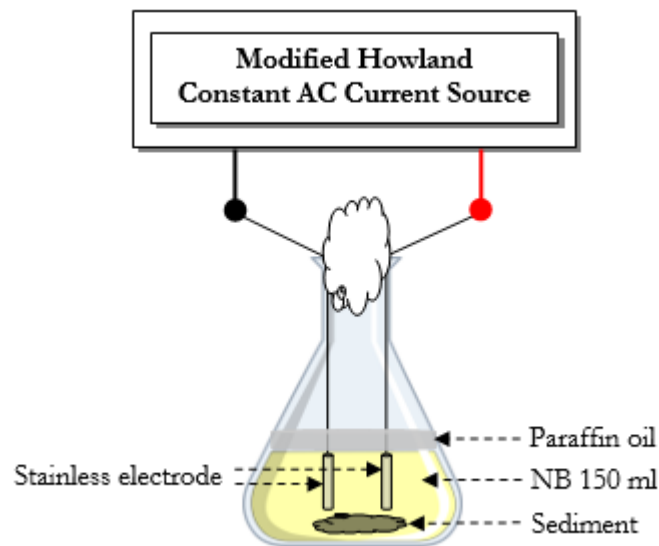


Fig. 1. Schematic diagram of experimental set up of AC electric current selection in flask.

2.2. Construction and Configuration of MFC

Dual glass chamber MFC was designed and used in this research. Neosepta® PEM (model CMS, ASTOM corporation, Japan), with 3 cm in diameter, functioning as cation exchange membrane, was installed between the anodic and cathodic compartment as illustrated in Fig. 2. It was sterilized by autoclaving at 110°C for 15 min before use. Carbon fiber cloth (ACELAN, Korea) with surface area of 18 cm², was used as anode and cathode. Electrode was installed inside the bottle and sterilized by autoclaving at 121°C for 15 min.

2.3. Microbial Fuel Cell Set-Up and Experiment

Phosphate buffer (K-buffer 100 mmol/l, pH 7.0) as electrically conductive medium was added into anodic and cathodic compartment, 100 mmol/l glucose as electron donor, and 1 mmol/l potassium ferric cyanide [K₃Fe(CN)₆] as electron acceptor, were added into anodic and cathodic compartment, respectively. Microorganism concentration of 10⁸-10⁹ CFU/ml was added into anodic compartment. All solutions were sterilized by autoclaving at 121°C for 15 min before use.

Forty isolates from sea sediments, each of them was further cultured in TSB added 0.02 mol/l of ferric citrate and incubated in GENbox anaer at room temperature for 5 days. Bacterial cells were collected by centrifugation at 8,000g, 25°C and washed twice with 100 mmol/l, phosphate buffer for removal of the

remaining medium broth. Cell pellet was resuspended into phosphate buffer for using as biological catalyst in the anodic compartment of MFC.

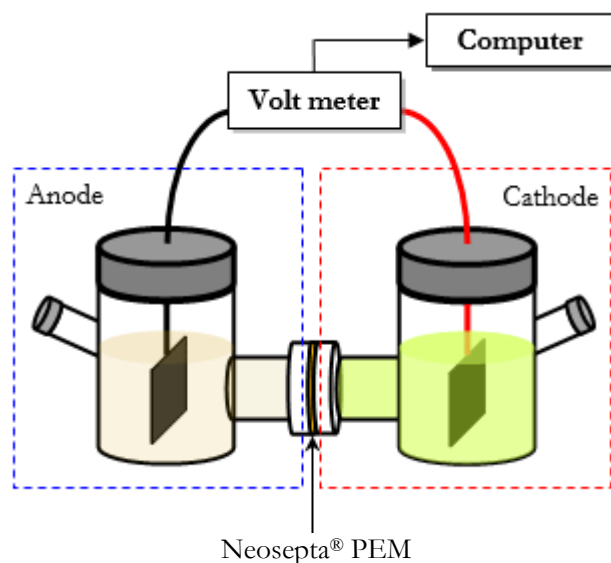


Fig. 2. Schematic diagram of experimental set up of dual chamber MFC.

2.4. MFC Outputs Measurement and Calculation

MFC performances were evaluated in term of voltage, current density and power density that was supplied to external load or resistance. MFC was connected to pico ADC-11 data acquisition unit (Pico technology, UK). Voltage was recorded every 15 seconds and transferred to the personal computer via parallel port. The output voltage was displayed in pico recorder and pico player program (Pico technology, UK). The open circuit voltage (OCV) was obtained when the system was not connected to any load or resistance. The voltage output was measured at steady state after the external resistor (R) was connected. Resistor in the range of 1-100 k Ω , were used to investigate the performance. The current density (i) (mA/m²) and power density (P) (mW/m²) were calculated using Eq. (1) and Eq. (2), respectively, where V is the voltage (volts) and A is electrode surface area (m²).

$$i = (V/R)/A \quad (1)$$

$$P = (IV)/A \quad (2)$$

2.5. Identification of Bacterial Isolates

Bacterial cell morphology and Gram staining were observed using light microscopy at x1000 magnification. Biochemical tests were examined by using rapid identification kit API® 20E (bioMérieux, France). All the protocols are carried out followed the company direction. Results from API kit were interpreted by using program API®WEB. Bacterial DNA was extracted from bacterial grown in tryptic soy broth (TSB) at 37°C, 24 h by using Simax Genome DNA Extraction Kit (Beijing SBS Genetech Co., Ltd., China). DNA extraction was performed following the manufacturer's instructions manual. 16S rDNA was amplified by polymerase chain reaction (PCR) (94°C for 1 min, 55°C for 1 min, 72°C for 2 min, 35 cycles using primers 16F27 (5'-AGA GTT TGA TCC TGG CTC AG-3') and 16R1522 (5'-AAG GAG GTG ATC CAG CCG CA-3') as described in [28]. PCR products were sequenced at Macrogen Inc. co. Ltd. (Seoul, Korea). The 16S rDNA sequences were analysed using BIOEDIT and GenBank® nucleotide data base.

2.6. Statistical Analysis

The open circuit voltage, current density, and power density of GP-FRB, GN-FRB and GN-nonFRB was compared in term of the difference between those from GPB vs. GNB, FRB vs. non FRB, and GN-FRB

vs. GN-nonFRB, respectively. The differences of means between two-sample for variances were compared by F-test and T-test, using the accepting significance at the $p < 0.05$ level.

3. Results and Discussion

3.1. Bacterial Isolation from Sea-Sediments of Ko Lan

After sea-sediments from Ko Lan were selected by AC for 30 days, microbial biofilm was observed only at anode electrode stimulating at 6, 9 and 12 mA. Biofilm formation on electrode for bacteria selection using different AC was shown in Fig. 3. The pure culture of 40 isolates of biofilm from the sea-sediments were isolated and designated as KL1-40. All isolates were characterized based on Gram staining, cell morphology under microscopic examination. They were 16 Gram-positive bacteria (GPB) and 24 Gram-negative bacteria (GNB).

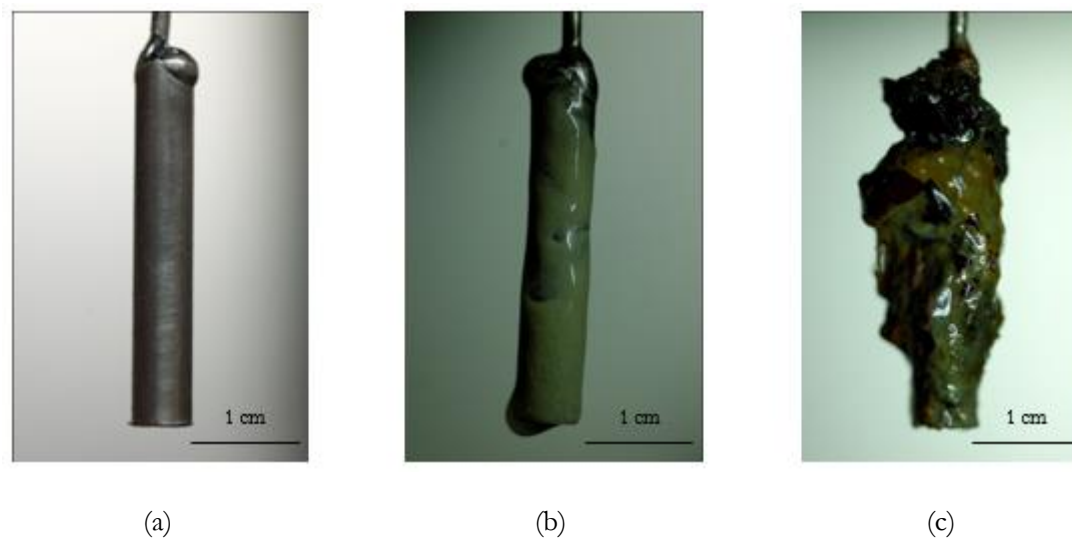


Fig. 3. Biofilm formation on electrode for bacteria selection using AC current (a) before supplied current (b) 6 mA and (c) 12 mA after 30 days.

3.2. Ferric Reduction of Isolates

Ferric reduction activity of all isolates was characterized under anaerobic condition by streaking pure cultures on NA plates containing ferric citrate. FRB could use Fe(III) as the electron acceptor under the anaerobic condition, thus Fe(III) were reduced to be Fe(II). FRB changed the reddish-brown color of NA with ferric citrate into the light green-color. A total of 40 isolates comprised of 16 Gram-positive FRB (GP-FRB), 15 Gram-negative FRB (GN-FRB), and 9 Gram-negative non FRB (GN-nonFRB) were identified and kept in anaerobic chamber for further study.

3.3. MFC Performance from Various Isolates

The current density and power density were determined by connecting the various external resistors, ranging from 1 to 100 k Ω , across the anode and cathode. However, to compare the current density and power density of the 40 isolates the same load of 1 k Ω was connected to the MFC. The results were shown in Fig. 4. After individual isolate of GP-FRB, GN-FRB and GN-nonFRB was performed in MFC. It was found that the maximum open circuit voltage (V_{max}) of the 40 isolates were in the range of ~ 300 -500 mV with no significant difference ($p > 0.05$) as shown in Fig. 4(a), whereas the highest $V_{max} \sim 500$ mV was produced by GP-FRB. Average current density from GP-FRB and GN-nonFRB (less than 6 mA/m 2) were different from the one of GN-FRB (~ 8 -12 mA/m 2) significantly ($P < 0.05$) (Fig. 4(b)). The power density of the GP-FRB and GN-nonFRB were less than 0.05 mW/m 2 while the GN-FRB gave the highest average power density of 0.236 mW/m 2 which was ~ 4 times greater than those from GP-FRB with the significant

difference at $P < 0.05$. This indicates that for the same ferric reducing capability, the GN-FRB offers the higher performance than the GP-FRB. Concurrently, the same higher performance can be obtained from GNB that having ferric reducing capability than those without ferric reducing capability significantly different at $P < 0.05$. The highest current and power density was 13.33 mA/m² and 0.32 mW/m², respectively obtained from GN-FRB, KL14 isolate.

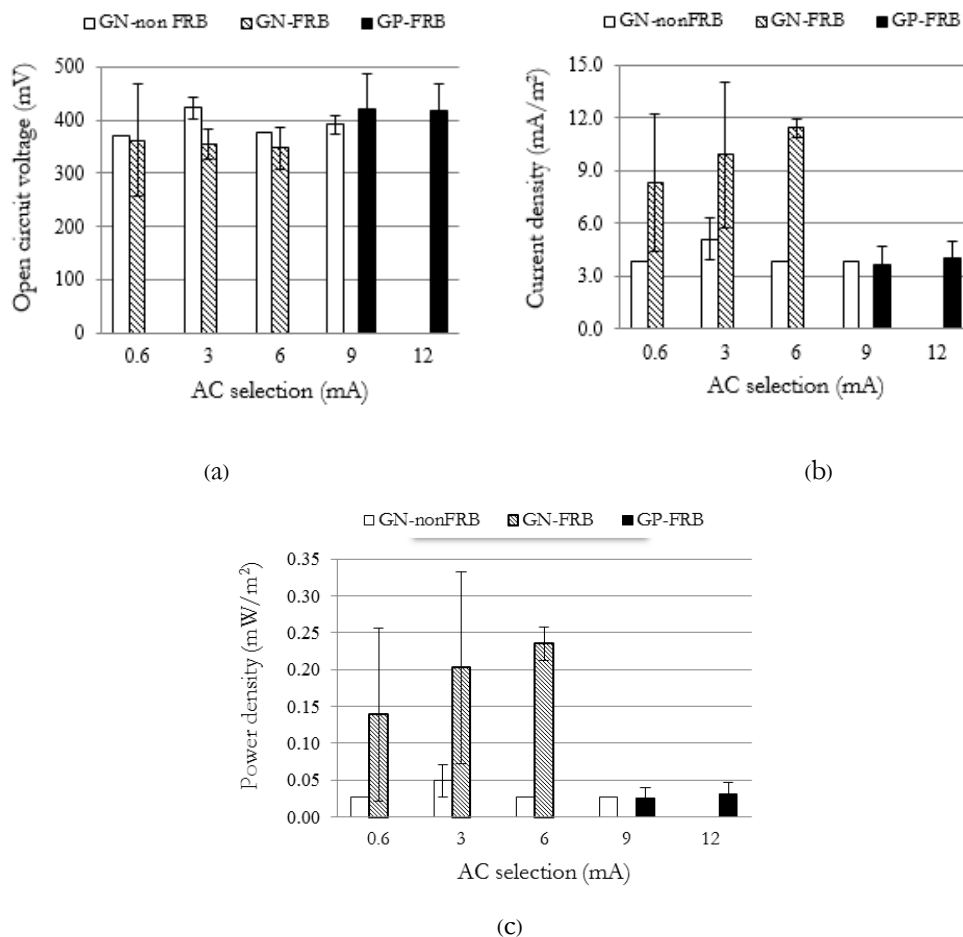


Fig. 4. AC selection (mA) and MFC performance of the 40 isolates in sea-sediments from Ko Lan: Gram negative–non ferric reducing bacteria (GN-non FRB) (n= 9), Gram negative-ferric reducing bacteria (GN-FRB) (n= 15) and Gram positive-ferric reducing bacteria (GP-FRB) (n=16). (a) open circuit voltage, (b) current density at 1 k Ω load, and (c) power density at 1 k Ω load.

3.4. Stimulating Current and Bacterial Type

It should be noted that the GNB were obtained only from the low electrical current selection of 0.6–6 mA whereas GPB were obtained only when using electrical current selection higher than 9 mA. This implies the possibility in using electrical current for screening the types of bacteria. It was also observed that the GPB isolated were spore forming bacilli, facultative anaerobic bacteria and tentatively identified as *Bacillus* spp. This kind of bacteria was found to have a low electron transferring to electrode in anaerobic environment reflecting in low current density and power density output as reported previously [29]. From their study the facultative anaerobic GPB, *Brevibacillus* sp. PTH1 generated quite low electrical current in the MFC was demonstrated. In addition, the current densities from all the GPB from our isolates were low even they are ferric reducing bacteria. It is known that the cell wall composition of the GPB is different from GNB. That is the GPB have thicker peptidoglycan layer than the GNB. It could be understand that the higher ohmic loss or higher voltage drop will occur at the thicker peptidoglycan layer than the thinner one. This implies that in this case the cell wall composition might play more important role in decreasing the MFC performance than the reducing activities.

3.5. Open circuit Voltage, Current Density and Power Density

The open circuit voltage in our experiment is the voltage obtained at open circuit condition without any load connection. This value indicates the potential of the MFC that how much it can distribute power to the load. However the real power that the MFC could distribute depending on many conditions or parameters. The high OCV-MFCs may distribute power lower than the low OCV-MFCs if their internal impedances are higher than the low OCV-ones. Therefore, we had to measure the OCV before connecting our MFCs to the load. In our cases, we found that the open circuit voltage of all MFCs were quite similar; however, the current and power density were significantly varied when the resistor load was added. The lower current and power means there is more loss inside the MFCs due to the internal impedance. This implies that the internal resistance of the MFCs play a very important role in the MFCs performances. The average of open circuit voltage, current density and power density among GN-nonFRB (n=9), GN-FRB (n=15) and GP-FRB (n=16) are indicated in Fig. 5. It is obvious that the high open circuit voltage was obtained from the GP-FRB (Fig. 5(a)), while the highest of both current density (Fig. 5(b)) and power density (Fig. 5(c)) were obtained from the GN-FRB. This indicates that the ferric reducing capability plays important impact on the MFC performance.

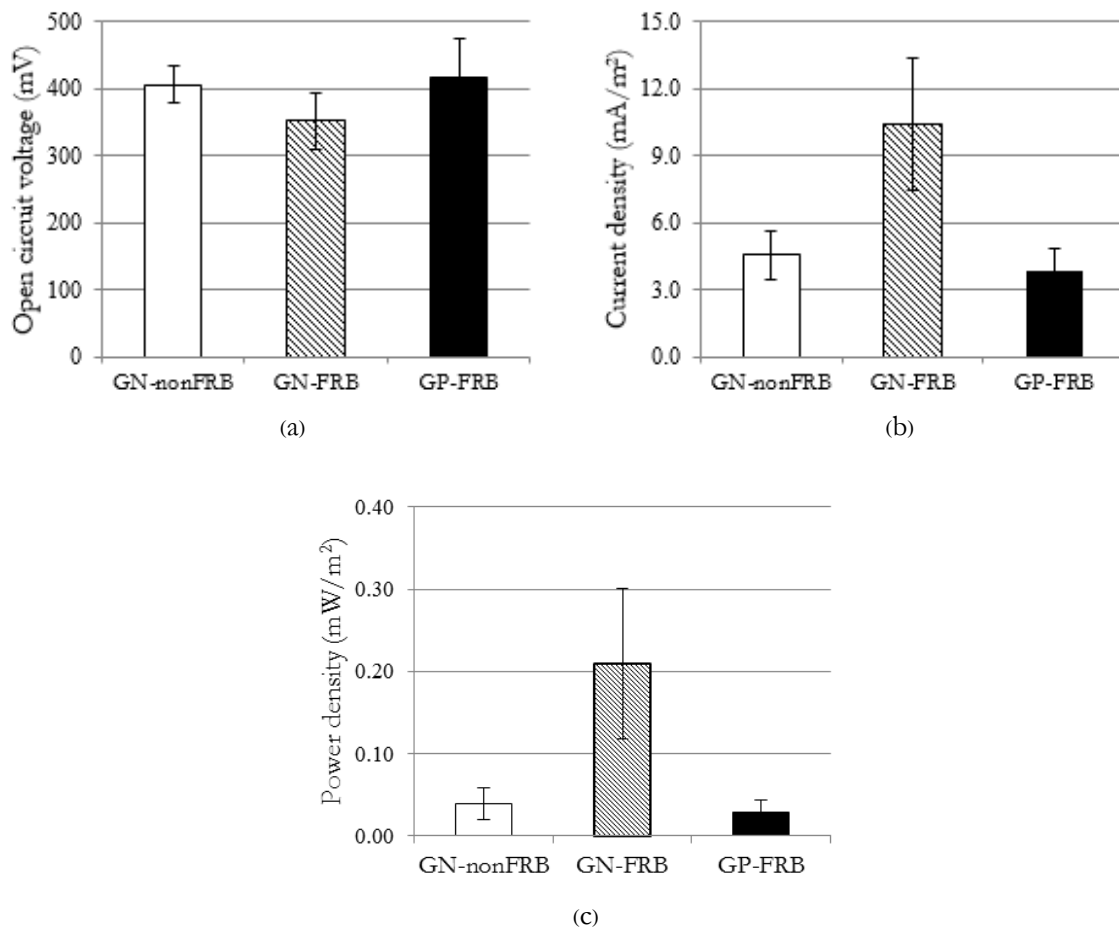


Fig. 5. Average and standard deviation of (a) open circuit voltage, (b) current density at 1 k Ω load and (c) power density at 1 k Ω load of each type of bacteria: GP-FRB (n=16), GN-non FRB (n=9) and GN-FRB (n=15).

3.6. Voltage-Current Density Characteristics

Figure 6 depicted voltage vs current density of the representatives among three groups of our isolates. Although GP-FRB (KL37) offered the highest open circuit voltage, it gave the lowest current density. While the highest current density was obtained from the GN-FRB (KL21) even its open circuit voltage was

the lowest. The largest slope of voltage and current density observed from the GP-FRB (KL37) indicates that it possesses the largest ohmic loss than those from GN-FRB (KL21) and GN-nonFRB (KL10). Ouitrakul *et al.* have shown that the impedance of the electrode used in the MFC do effect the performance of MFC [30]. For our cases, all parameters used in forming MFC were the same, except for the microorganism used. Then it can be mentioned that the difference in impedance comes from the difference of the cell. The internal impedance of each MFC could be estimated from the slope (Fig. 6). It was found that the internal impedance of the GP-FRB, GN-nonFRB and GN-FRB were 70, 57 and 30 k Ω , respectively. It could be explained from the fact that most of the GPB has the different compositions of cell wall especially more peptidoglycan thicker than the one of GNB. This made the impedance of the GPB cell wall to be higher than the GNB one. It is believed that this impede the electron transfer from the inside of the cell to the electrode. That is why the voltage drop of the GP-FRB was higher than the GN-nonFRB and GN-FRB. It should be noted that the internal impedance of GP-FRB was more than 2 times of the one from GN-FRB.

Moreover it should also be mentioned that the internal impedance of the GN-nonFRB was two times higher than the one from GN-FRB. This implies that the ohmic loss of the GN-FRB was lesser than the GN-nonFRB. GN-FRB clearly generated higher current density and higher electron capability than those of GN-nonFRB. These results are consistent with previous report that GN-nonFRB such as *E. coli* can poorly transfer electron to electrode, and it required electron mediator for facilitating their electron transfer to the electrode [31]. It is possible to indicate that the ferric reducing capability also plays important role in the electron transfer from cell to electrode. Moreover, many researchers also succeeded in enhancing electron transport of GN-nonFRB by modifying anode electrode [32-33]. Hence, GN-FRB from this study shows the highest performance in current density and power density among all three bacterial groups.

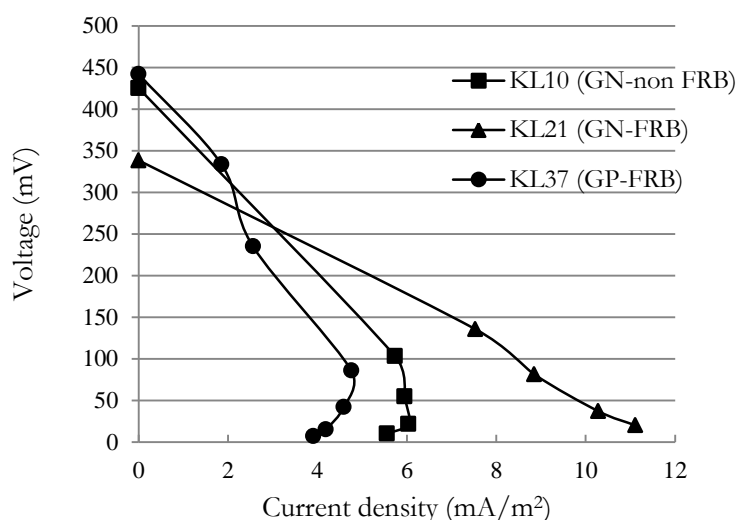


Fig. 6. Typical voltage and current density relation of GP-FRB (KL37), GN-non FRB (KL10), and GN-FRB (KL21).

3.7. Identification of Electrical Current Enriched Bacteria

Only KL11, KL14 and KL22 were identified. They are GNB with unique swarm colonies on agar plates. Using API 20E kit (BioMérieux, France) and program API®WEB (bioMérieux, France), KL11, KL14 and KL22 were primarily identified as *Proteus* sp. and *Proteus vulgaris*, respectively. The identity percentage of KL11, KL14 and KL22 were 96%, 97.7%, and 99.6%, respectively. Comply with BioEdit (Ibis Biosciences, California, USA) the 16S rDNA sequence of KL11, KL14 and KL22 was 1508, 1512, and 1513 bp, respectively. After compared with 16S rDNA sequence from database GenBank using the nucleotide-nucleotide Blast (BLASTN) program (www.ncbi.nlm.nih.gov/blast/), KL11, KL14 showed 99% similarity to *Proteus* sp. (accession number EF426445.1 and EF426446.1, consecutively) and 98% similarity to *Proteus vulgaris* for KL22 (accession number DQ499636.1). Three isolates belongs to the family *Enterobacteriaceae* which are fermentative and facultative anaerobic GNB. 16s rDNA sequence of KL11, KL14 and KL22 was

submitted to GenBank data base and acquired accession number KP313867, KP313868, and KP313869, respectively.

Proteus sp. had previously been used in MFC but the anode electrode of MFC was modified and mediator (thionin) was added in order to facilitate electron transfer of bacterial cell [34]. We hardly compared the outputs even with the same genus of bacteria was used. The mechanism of electron transfer of *Proteus* sp. to electrode was still unclear. However, further improvement by optimizing the physical and chemical parameters in MFC is required.

4. Conclusion

In this study, AC current was used to select bacteria from sub-sediment. Our 40 isolates were classified into three groups, GN-non FRB, GN-FRB and GP-FRB. The GNB were obtained at a low current in the range of 0.6-6 mA while the GPB did at the higher current of 9-12 mA. The results show feasibility in using current in selection or classification the type of bacteria. MFC outputs of GN-FRB generated greater current density and power density than GN-non FRB and GP-FRB, respectively. On the other hand, GP-FRB had greater open circuit voltage than GN-non FRB and GN-FRB, respectively. The results can be explained from the fact that GNB cell wall have thinner peptidoglycan layer than GPB or there are the difference in cell wall composition which lead to the lower voltage drop or lowering the ohmic loss at the cell, so GNB have electron transfer better than GPB. Moreover, it was also found that FRB have better electron transfer than non-FRB resulting in the higher current density and power density. It implies that the ferric reducing activities are also the one of key parameter in forming MFC. The results indicate that the GN-FRB should be used in construction of MFC due to the better electron transfer and lower the ohmic loss at the cell. It should be noted that our isolates are *Proteus* spp. It is interesting that these bacteria can be applied to form MFC without using any electron mediator. Moreover, the GN-non FRB can produce electricity higher than the GP-FRB, even it has no ferric reducing activities. Further study should be investigated to understand the detail mechanism of electrical current on the bacterial growth.

Acknowledgements

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References

- [1] M. C. Potter, "Electrical effects accompanying the decomposition of organic compounds," *Proc. R. Soc. Lond. B.*, vol. 84, no. 571, pp. 260-276, Sep. 1911.
- [2] U. Schröder, F. Harnisch, and L. T. Angenent, "Microbial electrochemistry and technology: terminology and classification," *Energy Environ. Sci.*, vol. 8, no. 2, pp. 513-519, Feb. 2015.
- [3] D. R. Lovley, and E. J. P. Phillips, "Organic matter mineralization with reduction with reduction of ferric iron in anaerobic sediments," *Appl. Environ. Microbiol.*, vol. 51, no. 4, pp. 683-689, Apr. 1986.
- [4] K. H. Nealson, and D. Saffarini, "Iron and manganese in anaerobic respiration: environmental significance, physiology, and regulation," *Annu. Rev. Microbiol.*, vol. 48, pp. 311-343, Oct. 1994.
- [5] B. H. Kim, H. J. Kim, M. S. Hyun, and D. H. Park, "Direct electrode reaction of Fe(III)-reducing bacterium, *Shewanella putrefaciens*," *J. Microbiol. Biotechnol.*, vol. 9, no. 2, pp. 127-131, Apr. 1999.
- [6] H. J. Kim, H. S. Park, M. S. Hyun, I. S. Chang, M. Kim, and B. H. Kim, "A mediator-less microbial fuel cell using a metal reducing bacterium, *Shewanella putrefaciens*," *Enzyme Microbiol. Technol.*, vol. 30, no. 2, pp. 145-152, Feb. 2002.
- [7] A. Sydow, T. Krieg, F. Mayer, J. Schrader, and D. Holtmann, "Electroactive bacteria—molecular mechanisms and genetic tools," *Appl. Microbiol. Biotechnol.*, vol. 98, no. 20, pp. 8481-8495, Oct. 2014.
- [8] K. Rabaey, N. Boon, S. D. Siciliano, M. Verhaege, and W. Verstraete, "Biofuel cells select for microbial consortia that self-mediate electron transfer," *Appl. Environ. Microbiol.*, vol. 70, no. 9, pp. 5373-5382, Sep. 2004.

- [9] K. Rabaey, N. Boon, M. Höfte, and W. Verstraete, "Microbial phenazine production enhances electron transfer in biofuel cells," *Environ. Sci. Technol.*, vol. 39, no. 9, pp. 3401-3408, May 2005.
- [10] D. R. Bond, and D. R. Lovley, "Electricity production by *Geobacter sulfurreducens* attached to electrodes," *Appl. Environ. Microbiol.*, vol. 69, no. 3, pp. 1548-1555, Mar. 2003.
- [11] D. R. Lovley, "Bug juice: harvesting electricity with microorganisms," *Nat. Rev. Microbiol.*, vol. 4, no. 7, pp. 497-508, Jul. 2006.
- [12] S. K. Chaudhuri, and D. R. Lovley, "Electricity generation by direct oxidation of glucose in mediatorless microbial fuel cells," *Nat. Biotechnol.*, vol. 21, no. 10, pp. 1229-1232, Oct. 2003.
- [13] K. T. Finneran, C. V. Johnsen, and D. R. Lovley, "*Rhodospirillum rubrum* sp. nov., a psychrotolerant, facultatively anaerobic bacterium that oxidizes acetate with the reduction of Fe(III)," *Int. J. Syst. Evol. Microbiol.*, vol. 53, no. 3, pp. 669-673, May 2003.
- [14] H. S. Park, B. H. Kim, H. S. Kim, H. J. Kim, G. T. Kim, M. Kim, I. S. Chang, Y. K. Park, and H. I. Chang, "A novel electrochemically active and Fe(III)-reducing bacterium phylogenetically related to *Clostridium butyricum* isolated from a microbial fuel cell," *Anaerobe*, vol. 7, no. 6, pp. 297-306, Dec. 2001.
- [15] C. A. Pham, S. J. Jung, N. T. Phung, J. Lee, I. S. Chang, B. H. Kim, H. Yi, and J. Chun, "A novel electrochemically active and Fe(III)-reducing bacterium phylogenetically related to *Aeromonas hydrophila* isolated from a microbial fuel cell," *FEMS Microbiol. Lett.*, vol. 223, no. 1, pp. 129-134, Jun. 2003.
- [16] D. Xing, Y. Zuo, S. Cheng, J. M. Regan, and B. E. Logan, "Electricity generation by *Rhodospseudomonas palustris* DX-1," *Environ. Sci. Technol.*, vol. 42, no. 11, pp. 4146-4151, Jun. 2008.
- [17] Y. Zuo, D. Xing, J. M. Regan, and B. E. Logan, "Isolation of the exoelectrogenic bacterium *Ochrobactrum anthropi* YZ-1 by using a U-tube microbial fuel cell," *Appl. Environ. Microbiol.*, vol. 74, no. 10, pp. 3130-3137, May 2008.
- [18] D. Pant, G. Van Bogaert, L. Diels, and K. Vanbroekhoven, "A review of the substrates used in microbial fuel cells (MFCs) for sustainable energy production," *Bioresour. Technol.*, vol. 101, no. 6, pp. 1533-1543, Mar., 2010.
- [19] P. D. Kiely, R. Cusick, D. F. Call, P. A. Selembo, J. M. Regan, and B. E. Logan, "Anode microbial communities produced by changing from microbial fuel cell to microbial electrolysis cell operation using two different wastewaters," *Bioresour. Technol.*, vol. 102, no. 1, pp. 388-394, Jan. 2011.
- [20] B. E. Logan, "Simultaneous wastewater treatment and biological electricity generation," *Water Sci. Technol.*, vol. 52, no. 1-2, pp. 31-37, Jan. 2005.
- [21] B. E. Logan, B. Hamelers, R. Rozendal, U. Schröder, J. Keller, S. Freguia, P. Aelterman, W. Verstraete, and K. Rabaey, "Microbial fuel cells: Methodology and technology," *Environ. Sci. Technol.*, vol. 40, no. 17, pp. 5181-5192, Jul. 2006.
- [22] J. C. Biffinger, J. Pietron, R. Ray, B. Little, and B. R. Ringeisen, "A biofilm enhanced miniature microbial fuel cell using *Shewanella oneidensis* DSP10 and oxygen reduction cathodes," *Biosens. Bioelectron.*, vol. 22, no. 8, pp. 1672-1679, Mar. 2007.
- [23] D. A. Finkelstein, L. M. Tender, and J. G. Zeikus, "Effect of electrode potential on electrode-reducing microbiota," *Environ. Sci. Technol.*, vol. 40, no. 22, pp. 6990-6995, Oct. 2006.
- [24] E. J. Cho, and A. D. Ellington, "Optimization of the biological component of a bioelectrochemical cell," *Bioelectrochem.*, vol. 70, no. 1, pp. 165-172, Jan. 2007.
- [25] X. Wang, Y. Feng, N. Ren, H. Wang, H. Lee, N. Li, and Q. Zhao, "Accelerated start-up of two-chambered microbial fuel cells: Effect of anodic positive poised potential," *Electrochim. Acta*, vol. 54, no. 3, pp. 1109-1114, Jan. 2009.
- [26] H. S. Lee, C. I. Torres, and B. E. Rittmann, "Effects of substrate diffusion and anode potential on kinetic parameters for anode-respiring bacteria," *Environ. Sci. Technol.*, vol. 43, no. 19, pp. 7571-7577, Oct. 2009.
- [27] H. J. Lin, X. Wu, C. Miller, and J. Zhu, "Improved performance of microbial fuel cells enriched with natural microbial inocula and treated by electric current," *Biomass Bioenerg.*, vol. 54, pp. 170-180, Jul. 2013.
- [28] A. Bayane, D. Roblain, R. D. Dauphin, J. Destain, B. Diawara, and P. Thonart, "Assessment of the physiological and biochemical characterization of a Lactic acid bacterium isolated from chicken faeces in sahelian region," *Afr. J. Biotechnol.*, vol. 5, no. 8, pp. 629-634, Apr. 2006.
- [29] T. H. Pham, N. Boon, P. Aelterman, P. Clauwaert, L. De Schamphelaire, L. Vanhaecke, K. De Maeyer, M. Höfte, W. Verstraete, and K. Rabaey, "Metabolites produced by *Pseudomonas* sp. enable a Gram-positive bacterium to achieve extracellular electron transfer," *Appl. Microbiol. Biotechnol.*, vol. 77, no. 5, pp. 1119-1129, Jan. 2008.

- [30] S. Ouitrakul, M. Sriyudthsak, S. Charojrochkul, and T. Kakizono, "Impedance analysis of bio-fuel cell electrodes," *Biosens. Bioelectron.*, vol. 23, no. 5, pp. 721-727, Dec. 2007.
- [31] D. H. Park and J. G. Zeikus, "Electricity generation in microbial fuel cells using neutral red as an electronophore," *Appl. Environ. Microbiol.*, vol. 66, no. 4, pp. 1292-1297, Apr. 2000.
- [32] U. Schröder, J. Nießen, and F. Scholz, "A generation of microbial fuel cells with current outputs boosted by more than one order of magnitude," *Angew. Chem. Int. Ed. Engl.*, vol. 42, no. 25, pp. 2880-2883, Jun. 2003.
- [33] T. Zhang, Y. Zeng, S. Chen, X. Ai, and H. Yang, "Improved performances of *E. coli*-catalyzed microbial fuel cells with composite graphite/PTFE anodes," *Electrochem. Commun.*, vol. 9, no. 3, pp. 349-353, Mar. 2007.
- [34] Y. Yuan and S. Kim, "Polypyrrole-coated reticulated vitreous carbon as anode in microbial fuel cell for higher energy output," *Bull. Korean Chem. Soc.*, vol. 29, no. 1, pp. 168-172, Jan. 2008.