

# Indonesian Journal of Fundamental and Applied Chemistry

Article http://ijfac.unsri.ac.id

Determination of Magnetic Mineral Crystal Structure Using X-Ray Diffraction (XRD) on Igneous Rock from Ogolowe and Bajugan Village, Toli toli District Central Sulawesi

Musfirah Cahya Fajrah<sup>1,\*</sup>, Alia Mapesso <sup>2</sup>, and Sandra Kasim <sup>3</sup>

### Abstract

Study of igneous rocks derived from Ogowele Village and Village Toli-Toli Bajugan has been conducted to determine the magnetic minerals contained in the igneous rocks, as well as determine the crystal system, lattice constants and field diffraction by comparison sieve 60 mesh and 80 mesh. Process characterization and analysis using X-Ray Diffraction (XRD). Characterization and analysis of the results showed that the compound was detected both rock samples Fe<sub>3</sub>O<sub>4</sub>, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and MnO<sub>2</sub>. Igneous rocks derived from Ogowele village is dominated by compounds SiO<sub>2</sub> (quartz) by 52% and rocks from Bajugan village dominated by the compound Fe<sub>3</sub>O<sub>4</sub> (magnetite) by 40%. The crystal structure is cubic and hexagonal compound to compound Fe<sub>3</sub>O<sub>4</sub> SiO<sub>2</sub>. Fe<sub>3</sub>O<sub>4</sub>compound cubic lattice constants are a = b = c is 8.375 Å. To compound the hexagonal SiO<sub>2</sub> ie a = b of 4.898 Å and c of 5.385 Å. Field diffraction on Fe<sub>3</sub>O<sub>4</sub>compound is (3 1 1) and the SiO<sub>2</sub> compound that is (1 0 1).

Keywords: X-Ray Diffraction, Fe<sub>3</sub>O<sub>4</sub> Compound, SiO<sub>2</sub> Compound, cubic, hexagonal

# Abstrak (Indonesian)

Telah dilakukan penelitian terhadap batuan beku yang berasal dari Desa Ogowele dan Desa Bajugan Kabupaten Toli-Toli untuk mengetahui mineral-mineral magnetik yang terkandung pada batuan beku tersebut, serta menentukan sistem kristal, konstanta kisi dan bidang difraksi dengan perbandingan ayakan 60 mesh dan 80 mesh. Proses karakterisasi dan analisis menggunakan *X-Ray Diffraction* (XRD). Hasil karakterisasi dan analisis menunjukkan bahwa pada kedua sampel batuan terdeteksi senyawa Fe<sub>3</sub>O<sub>4</sub>, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> dan MnO<sub>2</sub>. Batuan beku yang berasal dari Desa Ogowele didominasi oleh kandungan senyawa SiO<sub>2</sub> (kuarsa) sebesar 52 % dan batuan yang berasal dari Desa Bajugan didominasi oleh senyawa Fe<sub>3</sub>O<sub>4</sub> (*magnetite*) sebesar 40 %. Struktur kristal yang terbentuk yaitu kubik pada senyawa Fe<sub>3</sub>O<sub>4</sub> dan heksagonal pada senyawa SiO<sub>2</sub>. Konstanta kisi senyawa Fe<sub>3</sub>O<sub>4</sub> kubik yaitu a = b = c adalah 8,375 Å. Untuk senyawa SiO<sub>2</sub> heksagonal yaitu a = b sebesar 4,898 Å dan c sebesar 5,385 Å. Bidang difraksi pada senyawa Fe<sub>3</sub>O<sub>4</sub> yaitu (3 1 1) dan pada senyawa SiO<sub>2</sub> yaitu [1 0 1]

Article Info

Received 6 June 2017
Received in revised 5 Juy 2017
Accepted 10 July 2017
Available online 17 July 2017

Kata kunci: X-Ray Diffraction, Senyawa Fe<sub>3</sub>O<sub>4</sub>, Senyawa SiO<sub>2</sub>, kubik, heksagonal

## INTRODUCTION

The mining sector in Indonesia is one of the sectors which are the mainstay of government in generating foreign exchange. This is because Indonesia is one country that has many mineral resources. One area in Indonesia, Central Sulawesi, especially Toli-Toli is an area of huge potential for mining materials such as igneous rock which contains magnetic minerals. The spread of igneous rocks almost in 10 subdistricts in Toli Toli, including in the village and village Ogowele Bajugan (Department of Industry and Energy and Mineral Resources Kab.Toli-Toli,

2006).

Magnetic minerals in fact there is always naturally in rocks, soil, or sediments, despite the abundance quantitatively quite small at around 0.1% of the total mass of rock or sediment (Bijaksana, 2002). Magnetic minerals are minerals that have high magnetic properties. The magnetic mineral properties, the type and morphology varied depending on source (Hunt and Moskowitz, 1995).

Igneous rocks (from Latin: ignis, "fire") is a rock that resulted from the freezing process of magma (Sipatriot, 2013). Basically, the majority (99%) igneous rock

<sup>&</sup>lt;sup>1</sup>Physics Departement, Faculty of Mathematics and Natural Science, Institut Sains Dan Teknologi Nasional, Jakarta, Indonesia <sup>2,3</sup>Physics Departement, Faculty of Mathematics and Natural Science, Universitas Tadulako, Central Sulawesi, Indonesia

<sup>\*</sup>Corresponding Author: fira.toana@gmail.com

composed of main elements, namely: Oxygen, Silicon, Aluminum, Iron, Calcium, Sodium, Potassium and Magnesium. These elements form the main silicate minerals: Feldspar, Olivine, Piroksen, amphibole, quartz and mica. These minerals occupy more than 95% by volume of igneous rocks, and became the basis for the classification and explain the origin of magma (Graha, 1987).

Due to the lack of studies of magnetic minerals in igneous rocks in Toli-Toli it is necessary to identify more of the magnetic minerals contained in igneous rocks. Igneous rocks derived from Ogowele Village and Village Toli-Toli Bajugan can draw objects that contain magnets then chances are igneous rocks that have a fairly high Fe content. For it will be the characterization of magnetic minerals contained in the igneous rocks using X-Ray Diffraction (XRD) which includes the formation of compounds, crystal system, lattice constants and fields of diffraction of the magnetic minerals of igneous rocks.

The objectives of this study is to determine the magnetic minerals contained in igneous rocks of the Ogowele and Bajugan village Toli-Toli District and determine the crystal system, lattice constants, and the diffraction plane of the magnetic minerals of igneous rocks.

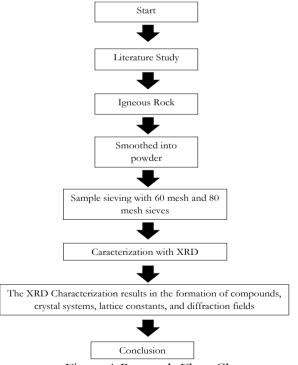


Figure 1 Research Flow Chart

### **EXPERIMENTAL SECTION**

Research for the characterization of magnetic minerals of igneous rocks using XRD performed on two laboratories, namely: Laboratory of Soil Science, Faculty of Agriculture UNTAD and Microstructure Laboratory of Physics, State UNM, Makassar. The timing of the study is from March to August 2013.

In the preparation phase materials used in the form of powder igneous rocks. Making the test material using the method of variation of grain size of 60 mesh and 80 mesh. At this stage of characterization using X-Ray tool Difraction (XRD) to determine its content and its compounds and crystal system of the sample. Research flow chart showed;

#### RESULT AND DISCUSSION

Characterization using XRD is characterization process that aims to determine the crystal structure of igneous rock samples that include several parameters, i.e: the crystal structure, lattice constants, and the field diffraction (Sartono, 2006) The results of XRD characterization of igneous rock samples can be seen in Table 1 to 4 and Figure 2 to 5.

Based on the XRD results obtained for samples of igneous rocks from the village Ogowele with a size of 60 mesh looks compound Fe<sub>3</sub>O<sub>4</sub> (magnetite) has a 21.6% weight percent, SiO2 (quartz) of 52.2%, MnO<sub>2</sub> (pyrolusite) by 9, 22%, and 17% Al<sub>2</sub>.66O<sub>4</sub>. It is clear that the compound has the highest weight percent compound SiO2 (quartz). As for the compound Fe<sub>2</sub>O<sub>4</sub>very small. Fe<sub>2</sub>O<sub>4</sub> compound has a cubic crystal structure or isometric, based on the lattice constant of a = b = c = 8.375 Å with crystallographic angle  $\alpha = \beta = \gamma = 90$ °. SiO<sup>2</sup> and Al<sub>2</sub>O<sub>3</sub> compound has a hexagonal crystal structure with lattice constants  $a = b \neq c$  and crystallographic angle  $\alpha = \beta \neq \gamma =$ 120o. While the compound MnO2 seen that the lattice constant a = b = 4.492 Å Adan c = 2.924 and has a crystallographic angle  $\alpha = \beta = \gamma = 90$ ° so that the compound MnO2 has a tetragonal crystal structure.

Different results shown by the results of the characterization of the rock sample size of 80 mesh which  $Fe_3O_4$  by 22%, 53% compound  $SiO_2$ ,  $MnO_2$  by 8.6% and 17.1%  $Al_2O_3$ . The difference in results is due to the sample size. The finer the sample, the more accurate the phases were detected.

Based on the XRD results it can be said that the rock came from the ogowele village is classified as alkaline igneous rocks (andesite) because it is dominated by high SiO2 compounds.

In igneous rock samples from the village Bajugan 60 mesh size was detected only 3 compounds which amounted to 39.7% Fe<sub>3</sub>O<sub>4</sub>, 32.7% SiO<sub>2</sub>, and Al<sub>2</sub>O<sub>3</sub> compounds by 28%. While the sample size of 80 mesh detected four compounds, namely by 40% Fe<sub>3</sub>O<sub>4</sub>, SiO<sub>2</sub> compounds by 33%, Al<sub>2</sub>O<sub>3</sub> 24.9% and 2.1% Mn<sub>3</sub>O<sub>4</sub> compound. Igneous rock samples Bajugan village also has a crystal structure similar to igneous rock samples Ogowele village because the resulting compound is also the same. Fe<sub>3</sub>O<sub>4</sub> compound has a cubic crystal structure / isometric, based on the lattice constant of a = b = c = 8,385Å the angle is  $\alpha = \beta$  crystallography =  $\gamma = 90$ odan field

diffraction [311]. SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub>compound has a hexagonal crystal structure with lattice constants  $a = b \neq c$  and crystallographic angle  $\alpha = \beta \neq \gamma = 120^{\circ}$  having SiO<sub>2</sub> diffraction field is [101] and Al<sub>2</sub>O<sub>3</sub> is [100]. While the

compound MnO<sub>2</sub> seen that the lattice constant  $a=b=c=5,886\text{\AA}$  and 2,9,54Å and has a crystallographic angle  $\alpha=\beta=\gamma=90$ ° and the field diffraction [211] so that the compound MnO<sub>2</sub> has a tetragonal crystal structure.

Table 1. Compound with crystal system of XRD result of Ogowele rock 60 mesh size.

No	Compound	Content (%)	2θ (°)	d-Value (Å)	Int (cps)	Crystal Structure	crystallographic angle	Lattice Contant	Diffraction field
1	Fe <sub>3</sub> O <sub>4</sub>	21,6	35,530	2,5246	218	Cubik	$\alpha = \beta = \gamma = 90^{\circ}$	a = b = c	(3 1 1)
2			30,228	2,9542	47				
3			43,157	2,0944	74				
4			62,61	1,4825	150				
5		52,2	26,726	3,3328	1387	Hexagonal	$\alpha = \beta \neq \gamma = 120^{\circ}$	a = b ≠ c	(1 0 1)
6	SiO <sub>2</sub>		20,945	4,2378	319				
7			39,609	2,2735	51				
8			60,048	1,5394	167				
9		9,2	28,053	3,1781	498	Tetragonal	$\alpha = \beta = \gamma = 90^{\circ}$	$a = b \neq c$	(1 1 0)
10	$MnO_2$		36,627	2,4514	62				
11			57,04	1,6132	99				
12	Al <sub>2</sub> .66.O <sub>4</sub>	17	37,49	2,397	43	Cubik	$\alpha = \beta = \gamma = 90^{\circ}$	a = b = c	(4 0 0)
13			45,939	1,9739	130				
14			68,405	1,3703	125				

Table 2. Compound with crystal system of XRD result of Ogowele rock 80 mesh size

No	Compound	Content (%)	2θ (°)	d- Value (Å)	Int (cps)	Crystal Structure	crystallographic angle	Lattice Contant	Diffraction field
1			30,142	2,9624	312			_	
2	Fe <sub>3</sub> O <sub>4</sub>	22	35,463	2,5292	389	Cubic	$\alpha = \beta = \gamma = 90^{\circ}$	a = b = c	(3 1 1)
3			62,54	1,4839	181				
4			20,890	4,2488	560				
5			26,674	3,3392	650		0. /	, ,	
6	SiO <sub>2</sub>	53	36,57	2,455	117	Hexagonal	$\alpha = \beta \neq \gamma = 120^{\circ}$	$a = b \neq c$	(1 0 1)
7			39,459	2,2818	66				
8			50,16	1,8173	104				
9	$MnO_2$	8,6	22,082	4,022	136	Orthorombik	$\alpha = \beta = \gamma = 90^{\circ}$	$a \neq b \neq c$	(110)
10			24,236	3,6692	81		- 0 - 4 -	, ,	(012)
11	$Al_2.O_3$	17,1	43,01	2,101	96	Hexagonal	$\alpha = \beta \neq \gamma = 120^{\circ}$	$a = b \neq c$	(1 1 3)
12			56,94	1,6158	99				(1 1 6)

DOI: 10.24845/ijfac.v2.i2.52

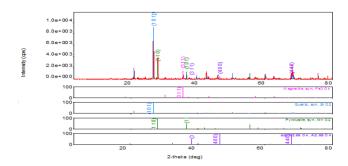


Figure 2. XRD result of igneous rock samples Ogowele village 60 Mesh size

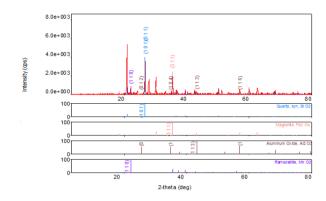


Figure 3. XRD result of igneous rock samples Ogowele village 80 Mesh size

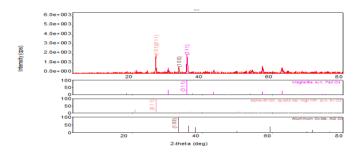


Figure 4. XRD result of igneous rock samples from Bajugan village 60 Mesh size

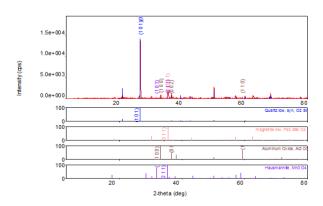


Figure 5. XRD result of igneous rock samples from Bajugan village  $80 \, \textit{Mesh}$  size

No	Compound	Content (%)	2θ (°)	d- Value (Å)	Int (cps)	Crystal Structure	crystallographic angle	Lattice Contant	Diffraction field
1			30,14	2,962	125				
2	Fe <sub>3</sub> O <sub>4</sub>	39,7	35,529	2,5247	550		$\alpha = \beta = \gamma = 90^{\circ}$	a = b = c	(3 1 1)
3	1.63.04	37,1	57,04	1,6132	216	Cubic	$\alpha - \beta - \gamma - 50$		(3 1 1)
4			62,61	1,4824	225				
4	SiO <sub>2</sub>	32,7	26,739	3,3313	285	Hexagonal	$\alpha = \beta \neq \gamma =$	$a = b \neq c$	(0 1 1)
5	5102	<i>52</i> , <sup>7</sup>	54,12	1,6931	216	Tiexagonai	120°		(0 1 1)
6	Al <sub>2</sub> .O <sub>3</sub>	28	33,300	2,6883	156	Hexagonal	$\alpha = \beta \neq \gamma = 120^{\circ}$	$a = b \neq c$	(012)

Table 3. Compound and crystal system of XRD result of Bajugan rock 60 mesh size

Table 4. Compound and crystal system of XRD result of Bajugan rock 80 mesh size

No	Compound	Content (%)	2θ (°)	d- Value (Å)	Int (cps)	Crystal Structure	crystallographic angle	Lattice Contant	Diffraction field
1		40	30,214	2,9555	110		$\alpha = \beta = \gamma = 90^{\circ}$	a = b = c	(3 1 1)
2	Fe <sub>3</sub> O <sub>4</sub>		35,472	2,5285	552	Cubic			
3	1 0304		43,12	2,0962	84				
4			62,60	1,4828	250				
5		33	20,981	4,2305	199	Hexagonal	$\alpha = \beta \neq \gamma = 120^{\circ}$	$a = b \neq c$	(1 0 1)
6			26,679	3,3386	1871				
7	SiO <sub>2</sub>		42,520	2,1243	57				
8			50,215	1,8153	368				
9			67,81	1,3809	36				
10	Mn <sub>3</sub> O <sub>4</sub>	2,1	53,98	1,697	182	Tetragonal	$\alpha = \beta = \gamma = 90^{\circ}$	$a = b \neq c$	(2 1 1)
11			56,96	1,6154	167				
12	Al <sub>2</sub> .O <sub>3</sub>	24,9	33,20	2,696	168	Hexagonal	$\alpha = \beta \neq \gamma = 120^{\circ}$	$a = b \neq c$	(1 0 0)

Based on the above results it is clear that the composition of rock samples from Bajugan village dominated by the  $Fe_3O_4$ compound. With a high content of the  $Fe_3O_4$ compound then rocks from the Bajugan village very useful as a main raw material of iron / steel.

#### **CONCLUSION**

Based on the analysis and XRD characterization in both the rock samples indicate that the detected magnetic minerals are compounds Fe<sub>3</sub>O<sub>4</sub> (Magnetite). Where the igneous rocks from the village Ogowele have Fe<sub>3</sub>O<sub>4</sub> content of 22%, while rocks from Bajugan village has a fairly high Fe<sub>3</sub>O<sub>4</sub> content of about 40%.

From the results of XRD and proof theory calculations for the two rocks is known that the compound Fe<sub>3</sub>O<sub>4</sub> from each sample has a cubic crystal structure / isometric with diffraction field or index field is [3 1 1] and has a lattice constant a=b=c. In a sample size of 60 mesh stone Ogowele Fe<sub>3</sub>O<sub>4</sub>obtained compound with the results of XRD lattice constant values are  $a=b=c=8,375\text{\AA}$  whereas the theoretical calculation that a=b=c=8.372 Å. While the rock sample size of 60 mesh Village Bajugan lattice constant values obtained XRD results are  $a=b=c=8,385\text{\AA}$  while the results of theoretical calculations are  $a=b=c=8,385\text{\AA}$  while the results of theoretical calculations are a=b=c=8,372 Å.

DOI: 10.24845/ijfac.v2.i2.52

Article http://ijfac.unsri.ac.id

### References

- [1] Bijaksana, S, F., 2012, Journal of Geophysics: *Analysis of magnetic minerals in ITB environmental issues*, ITB, Bandung.
- [2] Department of Industry, 2012, *Potential Mineral Resources in Toli-Toli*, Toli-Toli, Central Sulawesi.
- [3] Graha, Doddy, S., 1987, Rocks and Minerals, Nova, Bandung.
- [4] Hunt, Christopher P., Moskowitz, Bruce P., 1995, Magnetic properties of rocks and minerals, Rock Physics and Phase Relations: A Handbook of Physical Constants 3, Washington DC: American Geophysical Union, JLM 189-204.
- [5] Sartono, AA, 2012, X-ray diffraction (X-RD), Final project laboratory subject, Department of Physics, Faculty of Mathematics and Natural Science, University of Indonesia, Jakarta

DOI: 10.24845/ijfac.v2.i2.52