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| Author | Longbin Qiu, Sisi He, Yan Jiang, Dae-Yong Son, |
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| | Luis K. Ono, Zonghao Liu, Taehoon Kim, |
| | Theodoros Bouloumis, Said Kazaoui, Yabing Qi |
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Hybrid Chemical Vapor Deposition Enables Scalable and Stable Cs-FA Mixed Cation Perovskite Solar Modules with a Designated Area of 91.8 cm² Approaching 10% Efficiency

Longbin Qiu, †a Sisi He, †a Yan Jiang, a Dae-Yong Son, a Luis K. Ono, a Zonghao Liu, a Taehoon Kim, a Theodoros Bouloumis, a Said Kazaoui, b Yabing Qi*a

The development of scalable deposition methods for stable perovskite layer is a prerequisite for the development and the future commercialization of perovskite solar modules. However, scalability and stability are the two main challenges to overcome towards the realization of perovskite solar modules. In sharp contrast to previous report, here we develop a fully vapor based scalable hybrid chemical vapor deposition (HCVD) process for depositing Cs-formamidinium (FA) mixed cation perovskite films, which alleviate the problem encountered using conventional solution coating of mainly methylammonium lead iodide (MAPbl₃). Using our HCVD technique, we fabricate perovskite films of Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} with enhanced thermal and phase stabilities, by the intimate incorporation of Cs into FA based perovskite films. In addition, the sputtered SnO₂ electron transport layer (ETL) is found to be damaged during the HCVD process. In combination with precise interface engineering of SnO₂ ETL, we demonstrate relatively large area solar modules with efficiency approaching 10% with a designated area of 91.8 cm² fabricated on 10 cm × 10 cm substrates (14 cells in series). On the basis of our preliminary operational stability tests on encapsulated perovskite solar modules, we extrapolated that the Tgo lifetime is approximately 500 h (at 1 sun, 25 °C).

1 Introduction

stability, towards their 18 with the area.3

21 the efficiency would be related to perovskite deposition 22 methods. By far the most efficient perovskite solar cells are 3 The development of perovskite solar cells/modules remains 23 processed by the spin-coating method and reported on small 4 unprecedented with the certified stabilized output efficiency of 24 active areas, e.g., 0.1 cm² to 1 cm². A high quality perovskite film 5 20.9% for small cells (0.991 cm²), 17.25% for mini-modules 25 is easily obtained on this area.⁴ However, film quality 6 (17.277 cm²) and 11.7% for submodules (703 cm²).¹ However, 26 deteriorates as the film size increases to a typical module size, 7 there are still two main challenges to be addressed, namely 27 i.e., 100 cm^{2,2,3} It is thus necessary to develop perovskite potential 28 deposition methods capable of depositing large area films with 9 commercialization.² As learned from other photovoltaic ²⁹ high quality.⁵ Currently, scalable solution coating methods 10 technologies (Si, CdTe), the absolute efficiency decay with 30 including spray coating, slot-die coating and doctor blading are 11 upscaling the area is estimated to be approximately 31 in rapid development. Large area methylammonium lead iodide 12 0.8%/decade.³ Hence, when a 1 cm² solar cell with 20% 32 (MAPbl₃) modules up to 45 cm × 65 cm have been 13 efficiency is upscaled to a solar module of 100 cm² designated 33 demonstrated in a previous study.⁶ However, this coating 14 area, the efficiency is expected to be 18.4%, according to the 34 method is less efficient than that of the spin coating in small 15 estimated decay rate. However, at the current stage the 35 cells. Furthermore, the volatile solvents are not 16 published results in the literature indicate that for perovskite 36 environmentally friendly. Beside solution coating methods, 17 solar cell devices the efficiency decays faster upon upscaling 37 efforts have been made in utilizing vapor deposition for 38 perovskite solar cells/modules, as it benefits from the 39 conformal coating over large area, elimination of toxic solvents, 40 precise film thickness control, and outstanding compatibility 41 with tandem solar cell fabrication. For example, the hybrid 42 chemical vapor deposition (HCVD) process was developed to 43 fabricate MAPbI₃ and formamidinium lead iodide (FAPbI₃) 44 based solar modules.8 In this process, perovskite films are 45 formed in two steps. In the first step, a lead halide film such as 46 Pbl₂, PbCl₂ or PbBr₂ is deposited by thermal evaporation. In the 47 second step, the as prepared lead halide film is reacted with

20 This can be explained mainly by two causes. Firstly, the decay in

^{a.} Energy Materials and Surface Sciences Unit (EMSSU), Okinawa Institute of Science and Technology Graduate University (OIST), 1919-1 Tancha, Kunigami-gun, Onnason, Okinawa 904-0495, Japan.

^{*}Corresponding author: Yabing Qi, E-mail: Yabing.Qi@OIST.jp.

^{b.} Research Center for Photovoltaics (RCPV), National Institute of Advanced Industrial Science and Technology (AIST), Tsukuba, Ibaraki 305-8565, Japan.

[†] L. Qiu and S. He contributed equally to this work. Electronic Supplementary Information (ESI) available.

1 organic halide species with controlled vapor atmosphere and 58 cation (especially Cs cation) perovskites to improve stability. 19 2 pressure to form perovskite films inside a tube furnace. The 59 In FA based mixed cation perovskite, a small amount of Cs could 3 composition of perovskite can be readily tuned by changing the 60 stabilize the perovskite phase for high performance and 4 organic vapor atmosphere, from MAI to FAI or MABr, to obtain 61 stability.²⁰ In addition, with the Cs concentration in the range of 5 MAPbI₃, FAPbI₃ or MAPbBr₃, respectively.⁹ With the 62 10-30%, charge-carrier mobility, charge-carrier lifetime and 6 controllable deposition of perovskite, HCVD is promising for 63 crystal quality are substantially improved. 21 A few works 7 large scale fabrication of high performance/stable perovskite 64 reported the solution coating of Cs containing FA based 8 solar cells and light emitting diodes. 9 Secondly, the rapid decay 65 perovskite solar cells, however, very few studies reported solar 9 in the efficiency can be also originated from the nonoptimal 66 modules.^{2, 22} Our group previously reported the successful 10 module design (non-ohmic interconnection contact between 67 fabrication of mini-modules with Cs_{0.07}FA_{0.93}PbI₃ based 11 FTO/Au leading to high series resistance, large dead area due to 68 perovskite utilizing cation exchange process.²³ These mini-12 P1, P2, P3 pattering). A likely cause responsible for the faster 69 modules showed high stability under continues light 13 efficiency decay upon upscaling area comes from the 70 illumination. A recent report during the preparation of this work 14 transparent conducting electrode sheet resistance, which often 71 is related to CsxFA1-xPbI3-yBry based perovskite solar modules 15 leads to reduction in the fill factor. 10 The module design of 72 fabricated using vapor-solid reaction method that could be 16 stripes in series connection is a solution that can significantly 73 considered as a combination of physical vapor deposition and 17 increase the aspect ratio of each sub-cell and reduce the 74 low pressure chemical vapor deposition.²⁴ We would like to 18 resistance effect. The maximum stripe width is limited by the 75 stress that these reported modules are characterized by a 19 transparent conducting electrode sheet resistance. 11 In this 76 relatively modest GFF, and relative small active areas (12 cm² 20 design, the relation between the stripe width and the 77 on a 5 cm × 5 cm substrate, with 6 cells in series²³ and 41.25 cm² 21 transparent conducting electrode sheet resistance should be 78 on an 8 cm \times 8 cm substrate, with 9 cells in series²⁴). To further 22 considered, as for each stripe there will be an area for making 79 scale up FA based mixed cation perovskites with large area and 23 interconnection, which is a dead area. To get lower resistance a 80 also high GFF is important for the development of perovskite 24 narrower stripe is required, and the active area to the total area 81 solar modules. 25 ratio (geometric fill factor, GFF) is also lower. In addition, using 82 26 mask-patterning instead of other fine controlled patterning 83 Here we show the scalability of the HCVD process in deposition 27 techniques such as laser scribing or mechanical scribing results 84 of FA-Cs based mixed perovskites for large area stable solar 28 in the GFF as low as 50%, at the early stage of the perovskite 85 modules on 10 cm × 10 cm substrates with a high GFF (90%). This 29 solar module design.^{8, 12} On the basis of the calculation result, 86 fully vapor based process eliminates the use of toxic solvents 30 for highly conductive FTO substrates (10 Ω/\Box), the stripe width 87 such as dimethylformamide and dimethyl sulfoxide, which are 31 could be up to 7 mm with a small dead area to reach the 88 not environmentally friendly. The composition of the perovskite 32 maximum power output.¹0 GFF as high as 95% has been 89 layer could be readily changed from MAPbl₃ to FAPbl₃ and FA-33 demonstrated by using the laser patterning technique and this 90 Cs mixed cation perovskite, by controlling the precursor 34 offers a better design for active area approaching the total area 91 composition during HCVD process. With co-evaporation of CsBr 35 of the substrate. 13 92 and PbI₂ in the first step, followed by exposure to FAI vapor in 93 the second step, we successfully fabricated Cs-FA perovskite 37 Another aspect to be considered is the operational lifetime. 94 such as Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} through a fully vapor based process. 38 Although number of reports on the perovskite solar modules is 95 We also find that one critical issue for HCVD perovskite solar 39 increased gradually, they are mainly based on solution coated 96 cells/modules is the vacuum annealing process which might 40 MAPbl₃.² With modified precursor solution properties by adding 97 damage the SnO₂ layer. A C₆₀ layer with a thickness of 5 nm was 41 surfactant, large area 57.2 cm² solar modules with doctor 98 inserted between SnO₂ and perovskite, which passivate and 42 blading MAPbl₃ have been fabricated with an efficiency up to 99 modulate the energy levels at these interfaces. Benefiting from 43 14.6%.14 In another work, by adding excess MACI into the 100 the fully vapor deposition HCVD process, the film and interface 44 MAPbl₃ precursor ink, high quality film has been deposited by101 showed little difference between small area cells and large area 45 blade coating and module with 12.6 cm² showed efficiency of 102 modules. With the area increasing by 1000-fold from 0.09 cm² 46 13.3%.15 The efficiency was further improved to 15.6% with103 to 91.8 cm², we demonstrate that the device designated area 47 MA_{0.7}FA_{0.3}Pbl₃ and interconnection optimization. From an 04 PCE decreases slightly (from 13.3% to 9.34%, which corresponds 48 intermediate design of Pbl₂·nMA and MAI·mMA precursor,105 to a PCE drop of 1.3%/decade), suggesting that this fully vacuum 49 solvent- and vacuum-free tape assisted coating process has106 deposition process is a promising route towards scalability. In 50 been developed for MAPbI₃ based modules, with a certified₁₀₇ addition, the HCVD method is promising because it allows the 51 efficiency of 12.1% on an area of 36.1 cm^{2,5} The slot-die coated₁₀₈ synthesis of relatively stable perovskite composition such as 52 perovskite solar module with an area up to 168.75 cm 2 was also 109 Cs $_{0.1}$ FA $_{0.9}$ PbI $_{2.9}$ Br $_{0.1}$. The operational lifetime of these 53 based on MAPbI₃, with the efficiency reaching 10%.¹⁷ Although₁₁₀ encapsulated Glass/FTO/SnO₂/C₆₀/perovskite/spiro-MeOTAD/ 54 the area and efficiency are increasing with research efforts,111 Au/parylene/Glass solar modules has been studied, and showed 55 MAPbl₃ is not suitable for commercialization, due to its 56 instability under heat, light, humidity and oxygen. 18 A new trend 57 in the field is to employ thermally more stable FA based mixed

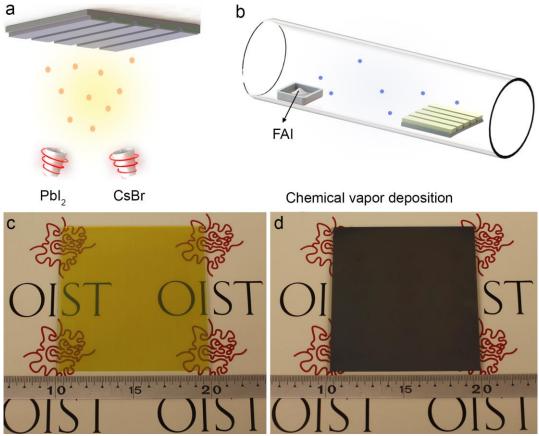


Figure 1. Schematic drawing showing the HCVD process. a) Thermal evaporation of inorganic precursors including Pbl₂ and CsBr. b) HCVD of FAI organic precursor for the formation of perovskite. Optical image of c) co-evaporated PbI₂/CsBr film and d) HCVD Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite film (96 cm² on the 100 cm² patterned FTO substrate).

1 an extrapolated T₈₀ lifetime of approximately 500 h under one 24 respectively. The deposition rate and thickness of the co-2 sun continuous illumination and steady state power output.

3 Results and discussion

4 Hybrid Chemical Vapor Deposition

9 modified, with the vacuum co-evaporation of inorganic 38 film. 10 precursors including PbI₂ and CsBr. The intended composition in 11 this work is 10% Cs and 90% FA cation, considering a suitable

25 evaporated film are monitored by quartz crystal microbalance 26 (QCM). The thickness of the co-evaporated PbI₂/CsBr film is 27 approximately 185 nm. In the second step (Figure 1b), the 28 Pbl₂/CsBr coated substrate is transferred into a multi-zone tube 29 furnace with the FAI organic precursor in the upstream side and 30 the substrates are placed in the downstream side. Upon 31 exposure to FAI vapor and at 150 °C, PbI₂/CsBr film readily 32 converts into Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite films. As shown in 33 Figure 1c and d, the yellow co-evaporated Pbl₂/CsBr film and a 5 Figure 1 displays the fully vapor based HCVD process for the 34 dark brown Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite film are uniformly 6 fabrication of perovskite films with controlled compositions. 35 coated on the stripe-patterned FTO substrate, with a size of 10 7 The perovskite formation proceeds into two steps as we 36 cm × 10 cm. These results demonstrate the excellent apparent 8 previously developed.⁹ The first step (Figure 1a) is slightly 37 uniformity of the as-prepared Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite

12 band-gap for single junction solar cells, phase stability, and high 39 The microstructure of the co-evaporated Pbl₂/CsBr film and 13 quality.²¹ A small amount of Br is incorporated to further ⁴⁰ Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite film after HCVD is further studied 14 improve the perovskite stability.²⁵ Although a layer-by-layer ⁴¹ (Figure 2). The cross-section image of co-evaporated Pbl₂/CsBr 15 deposition of CsBr and Pbl₂ might help precisely control the 42 film on top of FTO confirms that the Pbl₂/CsBr film consists of 16 thickness of each layer, [26] after the HCVD process, the resulted 43 nano-plates with large roughness (Figure 2a and c). The 17 film will have a gradient composition structure because the 44 thickness of this PbI₂/CsBr layer is expected to be 185 nm, 18 interdiffusion of CsBr and Pbl₂ (both are much less volatile than 45 monitored by QCM during the thermal evaporation. However, 19 FAI) during the HCVD process is expected to be much slower 46 due to the rough surface as shown in Figure 2a, it is difficult to 20 than FAI. Based on such consideration, we employed co- 47 determine the thickness accurately. After the HCVD process 21 evaporation on this work to ensure uniform distribution of the 48 with the FAI vapor, the perovskite film thickness is 22 CsBr - PbI₂ mixed film. To achieve 10% Cs, the evaporation rate 49 approximately 400 nm (Figure 2b), and is characterized by large 23 of PbI₂ and CsBr are controlled to be 0.10 nm/s and 0.01 nm/s, 50 grains and ultra-smooth surface (Figure 2d). The film thickness

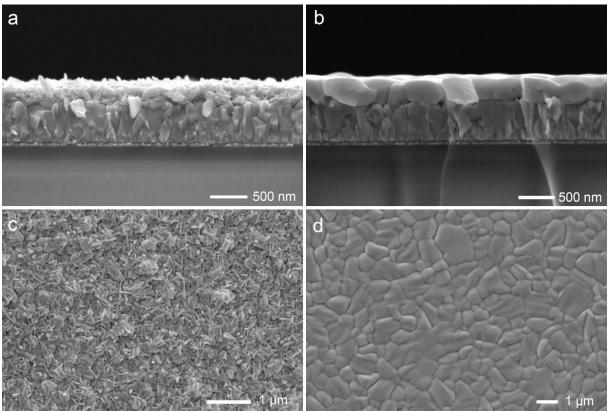


Figure 2. Microstructure of Pbl₂/CsBr before and after HCVD. Cross-section SEM image of a) Pbl₂/CsBr and b) Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite films. Surface SEM image of c) Pbl₂/CsBr and d) Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite films.

1 after the HCVD is found to increase by a factor of 2.16. The 31 confirms the uniform distribution of Cs and Br in the perovskite 9 Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} perovskite film.

11 Cs_{0.1}FA_{0.9}PbI_{2.9}Br_{0.1} perovskite film have been further 43 (Figure 3f). 12 characterized. From Figure 3a the (110) perovskite peak at 13 13.99° is clearly observed without any residual PbI₂ signal at 14 12.6° after HCVD (inset in Figure 3a), which confirms that FAI 44 Small area solar cells 15 could diffuse into the bottom and completely react with 16 Pbl₂/CsBr film. Furthermore, no diffraction peaks appear at 17 11.7° corresponding to the non-perovskite phase of FAPbI₃, 45 The carrier lifetime of Cs_{0.1}FA_{0.9}PbI_{2.9}Br_{0.1} on bare glass made by

2 thickness increase of the fully converted perovskite film could 32 layer (Figure S2). To confirm the composition, 3 be estimated from the density of PbI₂ (6.16 g/cm³) and FAPbI₃ 33 Cs_{0.1}FA_{0.9}PbI_{2.9}Br_{0.1} film made by solution coating process has $4 (4.10 \text{ g/cm}^3).^{9b, 27}$ With the assumption of full conversion and 34 been prepared and compared with $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ made by 5 exclusion of the small amount of Cs, the expected thickness 35 HCVD.²⁵ As shown in Figure S3-4, the XRD and UV-vis 6 would increase by a factor of 2.03, which is in good agreement 36 absorbance of Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} synthesized by solution and 7 with the cross-section SEM images (Figure 2d). These results 37 HCVD methods are very similar and thus we infere that their 8 confirm the excellent uniformity of the as-prepared 38 composition are also similar. The reaction of PbI₂/CsBr with FAI 39 and annealing at ~150 °C seems to relatively decrease the 40 concentration of Br on the top surface of perovskite (Figure S5). 41 The core level of N 1s is ascribed to FA cation, which also 10 The crystal structure and composition of the HCVD deposited 42 indicates the incorporation of FAI for the perovskite formation

18 suggesting the good phase purity of the $Cs_{0.1}FA_{0.9}PbI_{2.9}Br_{0.1}$ 46 HCVD was investigated by time-resolved photoluminescence 19 perovskite films prepared by HCVD. The optical absorbance 47 (TRPL), as presented in Figure S6. The upper limit of the open-20 edge of the perovskite film is 798 nm (Figure 3b), which 48 circuit voltage (Voc) due to the non-radiative recombination in 21 corresponds to an optical band of 1.56 eV from Tauc-plot 49 the perovskite is calculated to be approximately 1.04 V at room 22 (Figure S1). To study the composition of the perovskite film, X-50 temperature (Supporting Note 1). 11, 29 Further study are 23 ray photoelectron spectroscopy (XPS) is performed (Table S1). 51 necessary to improve the film quality and to reduce the defects 24 Before and after HCVD the ratio between Cs and Pb is 52 density to achieve long carrier lifetime. 30 To check the 25 approximately 0.1, which is consistent with deposition rate 53 photovoltaic performance of this HCVD deposited 26 monitored by QCM (Figure 3c and d). After HCVD, the core level 54 Cs_{0.1}FA_{0.9}PbI_{2.9}Br_{0.1} film, a planar junction structured device has 27 of Br 3d is weaker (Figure 3e). [28] Regarding the decreased Br 3d 55 been applied with sputtered SnO₂ as electron transport layer 28 signal, whether this is because there is phase separation on the 56 (ETL) (Figure 4a). The sputtered SnO $_2$ film has been optimized 29 top surface or the Br actually evaporates after HCVD process is 57 according to our recent work and achieves an efficiency over 30 currently under further investigation. The XPS mapping further 58 20% based on spin coated perovskite films. 31 However, the

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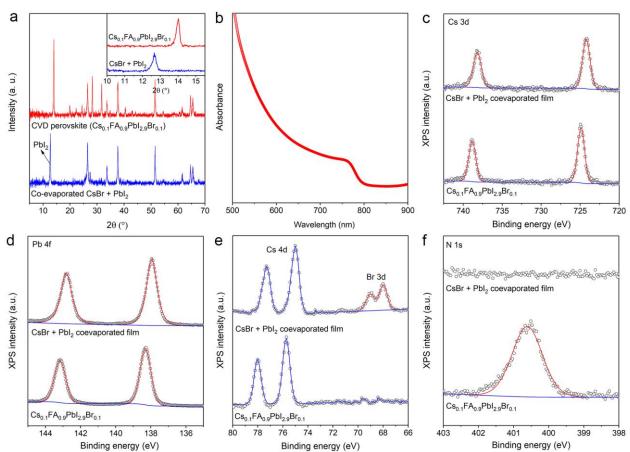


Figure 3. Structure and composition of HCVD perovskite layer. a) XRD before and after HCVD, which indicate high quality perovskite structure. b) Optical absorbance of Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}. Core level of c) Cs 3d, d) Pb 4f, e) Br 3d and f) N 1s before and after HCVD, which indicate the formation of Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}.

1 HCVD device performance is lower than expected. As shown in 30 glove box) where oxygen defect deteriorate the perovskite solar 2 Figure S7, from the typical J-V curves of the solar cells, the V_{OC} 31 cells performance.³⁴

3 is average only 0.81 V (Table S2), which is much lower than the

4 calculated upper limit due to the non-radiative recombination

 $_{10}$ believed to be mainly caused by oxygen deficiency after vacuum $_{37}$ S11, which confirms the smooth transfer of charges crossing the 11 annealing (Figure S9). These gap states after the vacuum 38 interface. 35 Regarding the influence on the SnO₂ films induced 12 annealing process lower the hole-blocking properties of the 39 by vacuum annealing, we further performed a control 13 SnO₂ ETL.^{31, 32} To confirm the effect of the vacuum annealing ⁴⁰ experiment on a sputtered SnO₂/C₆₀ electron transport layer 14 process, a control experiment has been performed. Solution 41 before and after vacuum annealing to study the effect of C_{60} 15 processed devices were fabricated on sputtered SnO₂ with and 42 passivation (Table S3). Note that in the case of "SnO₂/C₆₀", the 16 without vacuum annealing process. As shown in Table S3 and 43 efficiency with vacuum annealing is only slightly lower than that 17 Figure S10, the photovoltaic performance of solution coated 44 without vacuum annealing, suggesting that the negative effect 18 perovskite solar cells on sputtered SnO₂ with and without 45 of vacuum annealing is largely mitigated by adding a thin film of 19 vacuum annealing process differ significantly. After vacuum 46 C₆₀ with a thickness of 5 nm. However, when comparing the 20 annealing the SnO₂ performance degrades significantly, and all ⁴⁷ case of SnO₂/C₆₀ with the case of SnO₂ before vacuum 21 the device photovoltaic parameters decrease. Especially the V_{OC} 48 annealing, the device efficiency is about 10-20% lower in the 22 decreases from 1.03 V to 0.73 V. As a control experiment the 49 case of SnO_2/C_{60} . The factor that might influence the device $_{23}$ vacuum annealing effect has also been studied based on mostly 50 performance in the case of SnO_2/C_{60} is air exposure during the 24 used SnO $_2$ film prepared from Alfa Aesar solution. $^{[33]}$ This 51 sample transfer. It has been shown that C_{60} has poor air 25 crystalline SnO₂ film also shows reduced performance after 52 stability, which may induce oxygen-trap states and decrease 26 vacuum annealing process (Table S3). It shows that the negative 53 charge carrier lifetime. [36] Air exposure after C₆₀ deposition is 27 effect induced by vacuum annealing is not limited to sputtered 54 proposed to be the main reason for the lower performance.

5 (1.04 V). We find that the vacuum annealing process during 32 In this case, SnO2 surface modification is necessary to achieve 6 HCVD damages the SnO₂ layer. Ultraviolet photoemission ³³ high quality for electron extraction and hole blocking. A thin 7 spectroscopy (UPS) and XPS measurements of the SnO₂ layer ³⁴ layer of C₆₀ has been evaporated on top of SnO₂ before co-8 before and after vacuum annealing process have been 35 evaporation of PbI₂/CsBr as a buffer and passivation layer. The 9 performed to study the gap states (Figure S8). These states are 36 energy level diagram from UPS measurement is shown in Figure $28\,SnO_2$ films. Note that this is not unique to SnO_2 , because it also $55\,$ With this vacuum deposited SnO_2/C_{60} bilayer as ETL, solar cells $_{29}$ occurs for TiO_2 annealed in inert atmosphere (such as in N_2 $_{56}$ based on perovskite $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ made by HCVD have

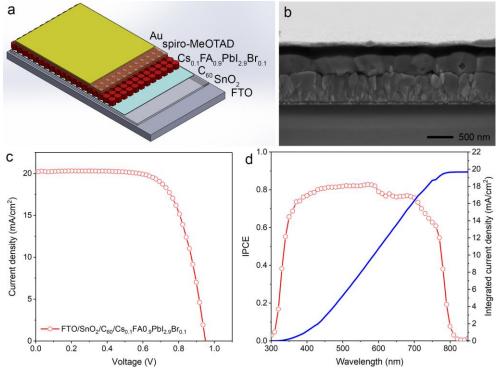


Figure 4. $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ based planar junction solar cell performance. a) Schematic drawing showing the planar junction device structure. b) Cross-section SEM image of the HCVD $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ based solar cell. Typical c) J-V curve and d) IPCE spectra of HCVD $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ based solar cells.

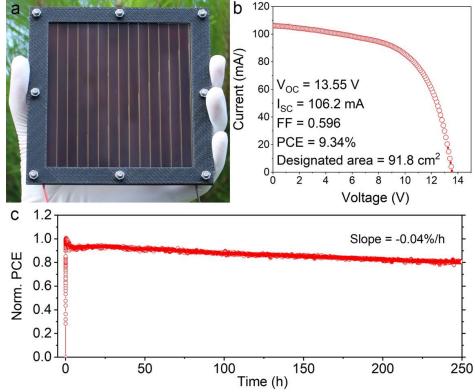


Figure 5. $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ based solar module performance. a) Optical image of a 10 cm \times 10 cm HCVD $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ based solar module. b) J-V curve of the champion 10 cm \times 10 cm HCVD $Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}$ based solar module (14 sub-cells in series). c) Operational stability of HCVD solar module under one-sun illumination and steady-state power output tracking.

¹ been fabricated with the structure 4 be observed in the cross-section SEM image (Figure 4b). The 2 Glass/FTO/SnO $_2$ /C $_{60}$ /Perovskite/spiro-MeOTAD/Au (Figure 4a). 5 active layer thickness is around 400 nm. With this planar 3 A clear layer by layer structure of the planar junction device can 6 junction structure the champion small cell efficiency is 13.3%

1 (0.09 cm²). Improving the perovskite optoelectronic quality and 58 PCE_{Forward})/PCE_{Reverse}) (Figure S16),³⁹ which need further 2 engineering the perovskite and ETL interfaces to reducing the 59 engineering of interface and perovskite composition.^{33, 39} As a 3 non-radiative carrier recombination rates at bulk and interfaces 60 demonstration of the broad composition adjustment of 4 are possible strategies to further push up the solar cell 61 perovskite by HCVD, the same 10 cm × 10 cm size modules but 5 performance. As shown in Figure 4c, from the typical J-V curve 62 with MAPbl₃ active layer have been fabricated. The 6 of solar cells with HCVD Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1}, the V_{OC} is on 63 performance is lower than that of the Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} based 7 average 0.90 V (Table S2). The short circuit current density (Jsc) 64 solar modules (Figure S17), due to the mismatch of the energy 8 is around 20.2 mA/cm², which is consistent with IPCE spectra 65 level alignment as shown in Figure S11. This demonstrates the 9 (Figure 4d). Currently, the average fill factor is around 0.67 66 scalability of our HCVD method for large area monolithic 10 leaving room for further improvement. 67 perovskite solar modules, with desired composition.

11 Large area solar modules

 $13 \ module \ fabrication. \ The \ scale-up \ of \ the \ Cs_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} \ solar_{72} \ series \ and \ exhibit \ designated \ area \ efficiency \ around \ 10\%, \ which \ area \ efficiency \ around \ 10\%, \ which \ area \ efficiency \ around \ 10\%, \ which \ efficiency \ effici$ $_{14}$ cells to 10 cm imes 10 cm solar modules through the HCVD method $_{73}$ is consistent with a decay rate of 1.3%/decade (Figure S15 and $_{15}$ is further demonstrated (Figure 5a). The module comprises $_{14}$ $_{74}$ $_{18}$). The operational lifetime was measured under a controlled 16 sub-cells in series connection. For each FTO stripe, the width is 75 temperature of 25 °C and continuous one-sun illumination with 17 6.6 mm, with 0.1 mm P1 pattering in between each FTO stripe. 76 steady state power output. From Figure 5c we can first observe 18 The length of the stripe is 98 mm results in the area of 6.468 77 a burn-in loss of power output within approximately the first ten 19 cm² for each stripe. The interconnection area is formed by CO_{2 78} hours with 90% of its initial performance remaining.⁴⁰ A slow $_{20}$ laser scribing, with a width around 0.15 mm (P2). Finally, $_{79}$ decay of the performance with a slope of -0.04%/h follows the 21 mechanical scribing is used to separate each sub-cell by forming 80 burn-in loss, which corresponds to a 780 lifetime of 500 h. 41 22 a P3 patterning (0.05 mm). The P1-P2-P3 patterning is shown in 81 These results demonstrate the good long-term stability of the 23 Figure S12. Considering the FTO stripe width plus P1 width (6.7 82 CS_{0.1}FA_{0.9}Pbl_{2.9}Br_{0.1} based solar module through our HCVD 24 mm) and the dead area for sub-cell series connection (0.68 $_{83}$ method. 25 mm), the geometric fill factor is up to 90%. 10 solar modules 26 have been fabricated, the photovoltaic parameters of which are 27 listed in Table S4. The champion solar module is plotted in 84 Conclusions 28 Figure 5b. The modules have been further encapsulated by 29 deposition of a Parylene film³⁷ on top of Au electrode side and 85 In summary, with the mix-cation Cs/FA based perovskite films 30 then a cover glass is sealed on top with UV-curable glue on the 86 deposited by HCVD, the SnO₂/C₆₀ ETL deposited by sputtering 31 edges to further prevent water and oxygen infiltration as well 87 and thermal evaporation, and the Parylene encapsulation layer 32 as to ensure a good thermal contact with cooling stage to 88 deposited by chemical vapor deposition, fully scalable 33 dissipate heat (Figure S13). This low temperature encapsulation 89 perovskite solar modules with 91.8 cm² designated area on 100 34 process does not affect the module performance (Figure S14). 90 cm² substrates are developed. Based on the fully vapor 37 (active area = 82.6 cm^2).

39 be 119 mA considering the J_{SC} value obtained from the small cell 96 perovskite film post treatment, the device performance is 40 (20.2 mA/cm²) and the area of each cell stripe (5.909 cm²). 97 promising for improvement. We believe that HCVD will be a $^{\circ}$ Indeed, the measured I_{SC} from solar modules is mostly around $^{\circ}$ 98 strong 42 110 mA (Table S4), which clearly demonstrates the fine 99 commercialization. 43 pattering process and the scalability of our HCVD method. The 44 average V_{OC} for modules is 12.8 V and for each sub-cell is 0.91 45 V. The V_{OC} of sub-cells in the modules also matches well the 46 value obtained for the small area cells (Table S2 and S4). From $_{101}$ **Experimental** 47 the small area cells to modules, the area is enlarged by over 101 48 1000 time (from 0.09 cm² to 91.8 cm²). The champion efficiency₁₀₂ 49 decreased from 13.3% to 9.3% (active area efficiency 10.4%)₁₀₃ HCVD: The HCVD process is performed in a multi-zone tube 50 (Figure S15), which is only slightly lower than expectation 103 HCVD. The HCVD process is performed in a multi-zone tube 51 (10.9%) if assuming an efficiency-against-area decay rate of 104 furnace with solid FAI power as the precursor. The FAI power is 52 0.8%/decade. The decrease mainly comes from the lower fill 105 first converted to the gas phase in the upstream zone when 53 factor. Without removing bottom SnO₂ (17 nm), the contact¹⁰⁶ heated up to 190 °C. With the assistance of a low flux of carrier 54 resistance might also increase compared with Au/FTO interface 107 gas (air with a relative humidity of 45% in this case; flux = 6055 contact resistance, which might be the main cause for the lower 108 sccm), the gas phase FAI is driven uniformly towards the 56 fill factor.³⁸ This module shows hysteresis with a hysteresis109 downstream zone where the chemical reaction takes place

68 The HCVD method for mixed cation perovskite is promising for 69 stable solar modules. Here the operational lifetime is measured 70 with encapsulated 5 cm × 5 cm mini-modules with designated 12 The deposition process is very promising for large scale solar 71 area of 22.4 cm². The module consists of 7 cells connected in

35 The best performance is 9.34% with a designated area of 91.8 91 deposition process there is no limitation for the module area of 21.8 92 and the absolute efficiency decay is consistent with other 93 commercialized photovoltaic techniques when scaling-up. The 94 operational T_{80} lifetime under illumination is extrapolated 38 With a GFF of 90%, the short-circuit current (Isc) is expected to 95 approximately 500 h. With further interface engineering and tool for perovskite

 $_{57}$ factor around 0.17 (hysteresis factor = (PCE_{Reverse} $_{-110}$ between FAI and the pre-deposited CsBr/PbI $_{2}$ substrate. The

1 other end of tube furnace is connected to a pump, which results 57 of 2 mm. This Parylene film was reported to be resistant to 2 in an overall average vacuum level of approximately 1 torr in the 58 water. 37 3 furnace. The temperature of the downstream zone is 59 4 approximately 150 °C. The pressure at the end with the input 60 Characterization: The absorbance spectra of perovskite were 5 carrier gas is expected to be slightly higher and gradually 61 recorded with a UV-vis spectrometer (Jasco V-670). The 6 decrease towards the end connected to the pump. The gentle 62 ultraviolet photoemission spectroscopy (UPS) and X-ray 7 gradient of pressure along the tube serves as the driving force 63 photoelectron spectroscopy (XPS) spectra were recorded from 8 to move FAI vapor from one end to the other. During HCVD, the 64 XPS-AXIS Ultra HAS (Kratos) equipped with monochromatic Al-9 FAI powder source is abundant, therefore to the first order 65 Kα=1486.6 eV and non-monochromatic He-I=21.22 eV sources. 10 approximation we can consider the FAI concentration in the gas 66 Sample damage induced by UV and X-ray was monitored by 11 stream to be a constant along the tube, if we do not take into 67 taking five consecutive scans. Surface morphology and cross-12 account the slight decrease of the FAI concentration down the 68 section characterization were performed using a scanning 13 stream due to a small amount of FAI reacting with the CsBr/PbI₂ 69 electron microscope (FEI Quanta 250 FEG). Crystal structure of 14 substrates. According to the equation $D_g \propto T^{3/2}/P$ (where D_g is 70 perovskite was measured by an X-ray Diffractometer (XRD) 15 the diffusion constant, T is the temperature, and P is the 71 (Bruker D8 Discover). Time resolved photoluminescence (TRPL) 16 pressure), [9b] the diffusion rate of gas phase FAI to the substrate 72 of perovskite film on bare glass was acquired using the time-17 is approximately a constant. Also, the adsorption rate of FAI on 73 correlated, single-photon counting technique (Hamamatsu, 18 the substrate can be assumed to be a constant independent (or 74 C10627), and excitation was provided by a femtosecond mode-19 only has a weak dependence) of the deposited film thickness. 75 locked Ti:sapphire laser (SpectraPhysics, MAITAI XF-IMW) at 20 As a result, the film thickness of the deposited FAI is determined 76 450 nm. J-V curves of solar cells/modules were recorded by a 21 by the total deposition time and will be uniform across the 77 Keithley 2420 Source Meter. The simulated AM1.5 solar light 22 entire length of the tube furnace. 78 comes from a solar simulator (Oriel-Sol1A equipped with a 450 79 W Xe lamp and an AM1.5 filter), with an illumination intensity 24 Solar cell/module fabrication: Patterned FTO glass with sheet 80 of 100 mW/cm² calibrated by a reference Si solar cell. The 25 resistance around 7 Ω/□ (OPVT) was first ultrasonic washed 81 designated area of 0.09 cm² for small cells, 22.4 cm² for small 26 sequentially with 1 wt% sodium dodecyl sulfate aqueous, de-82 modules and 91.8 cm2 for large modules were defined by 27 ionized water, acetone and isopropanol for 15 min. The dry FTO 83 corresponding metal mask. For solar cells/modules 28 substrate was then sputtered coated with 17 nm SnO₂ and 84 characterization, the dwell time is 10 ms, with small cell in the 29 evaporated with 5 nm C_{60} , with deposition rate of 0.7 nm/min 85 range of -0.1 V to 1.2 V, 5 cm \times 5 cm modules in the range of -30 and 0.02 nm/s, respectively. The perovskite active layer was $86\,0.1\,V$ to $7.0\,V$ and $10\,cm\times10\,cm$ modules in the range of -1 V 31 deposited by HCVD developed in our early work,8-9 with 87 to 14.0 V. The EQE spectra of small cells were characterized by 32 different precursor composition. Pbl₂ and CsBr were thermally 88 Oriel IQE 200. 33 co-evaporated on C₆₀ coated FTO/SnO₂ substrate with a rate of 34 0.10 nm/s and 0.01 nm/s, respectively, for 185 nm. The 35 Pbl₂/CsBr coated substrate was then transferred to a multi-zone 89 Conflicts of interest 36 tube furnace. The Cs salt is sensitive to water. To avoid the 90 There are no conflicts to declare. 37 moisture effect, the tube furnace was quickly pumped down to 38 1 torr after loading the CsBr/PbI₂ substrate. The FAI vapor was 39 carried by air with a humidity of 45%. The mixed O₂/N₂ carrier 91 Acknowledgements 40 gas helps passivate shallow and deep traps in the converted
41 perovskite film. [42] On the other hand, with such a low pressure
42 This work was supported by funding from the Energy Materials
43 and Surface Sciences Unit of the Okinawa Institute of Science 41 perovskite film. 42 level of air (1 torr) and the substrate temperature at 43 approximately 150 °C, we expect the moisture in the carrier gas 44 to have a minimal negative impact on the final perovskite 45 film. 43 After reaction and cooling the perovskite film was 45 film. 46 JSPS KAKENHI Grant Number JP18K05266. We would like to

46 washed with saturated potassium iodide solution (isopropanol)

$_{49}$ Li-TFSI solution (520 mg/mL in acetonitrile) in 0.4 mL $_{99}$ Notes and references 50 chlorobenzene was spin coated on top of the perovskite layer. 51 For solar module the P2 patterning was performed by a ${\rm CO}_{101}^{100}$

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47 and annealed under 100 °C for 20 min. A hole transport material 48 solution containing 29 mg spiro-MeOTAD, 11.5 μL TBP, and 7 μL

52 laser with a power of 5.6 W. At the end, 120 nm Au was₁₀₂

53 thermally evaporated as back contact electrode and separated₁₀₃ 2

54 each sub-cell by mechanical scribing to form P3 patterning. The 104 3

 $_{55}$ solar modules were encapsulated by 2 μm thick chemical vapor 105 56 deposited Parylene and sealed by a cover glass with a thickness $^{106}_{107}$

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