

Supplementary material for the article:

Čobeljić, B.; Milenković, M.; Pevec, A.; Turel, I.; Vujčić, M.; Janović, B.; Gligorijević, N.; Sladić, D.; Radulović, S.; Jovanović, K.; et al. Investigation of Antitumor Potential of Ni(II) Complexes with Tridentate PNO Acylhydrazones of 2-(Diphenylphosphino)Benzaldehyde and Monodentate Pseudohalides. *Journal of Biological Inorganic Chemistry* **2016**, *21* (2), 145–162. <https://doi.org/10.1007/s00775-015-1315-x>

Investigation of antitumor potential of Ni(II) complexes with tridentate PNO acylhydrazones of 2-(diphenylphosphino)benzaldehyde and monodentate pseudohalides

Božidar Čobeljić¹, Milica Milenković¹, Andrej Pevec², Iztok Turel², Miroslava Vujčić³, Barbara Janović³, Nevenka Gligorijević⁴, Dušan Sladić¹, Siniša Radulović⁴, Katarina Jovanović⁴, Katarina Anđelković¹

¹ Faculty of Chemistry, University of Belgrade, Studentski trg 12–16, 11000 Belgrade, Serbia

² Faculty of Chemistry and Chemical Technology, University of Ljubljana, Večna pot 113, 1000 Ljubljana, Slovenia

³ Institute of Chemistry, Technology and Metallurgy, University of Belgrade, Njegoševa 12, P.O. Box 815, 11000 Belgrade, Serbia

⁴ Institute for Oncology and Radiology of Serbia, Department of Experimental Oncology, Laboratory for Experimental Pharmacology, Pasterova 14, Belgrade, Serbia

Keywords: Nickel(II) complexes, Girard's T reagent, Phosphine ligands, Cytotoxicity, DNA interactions

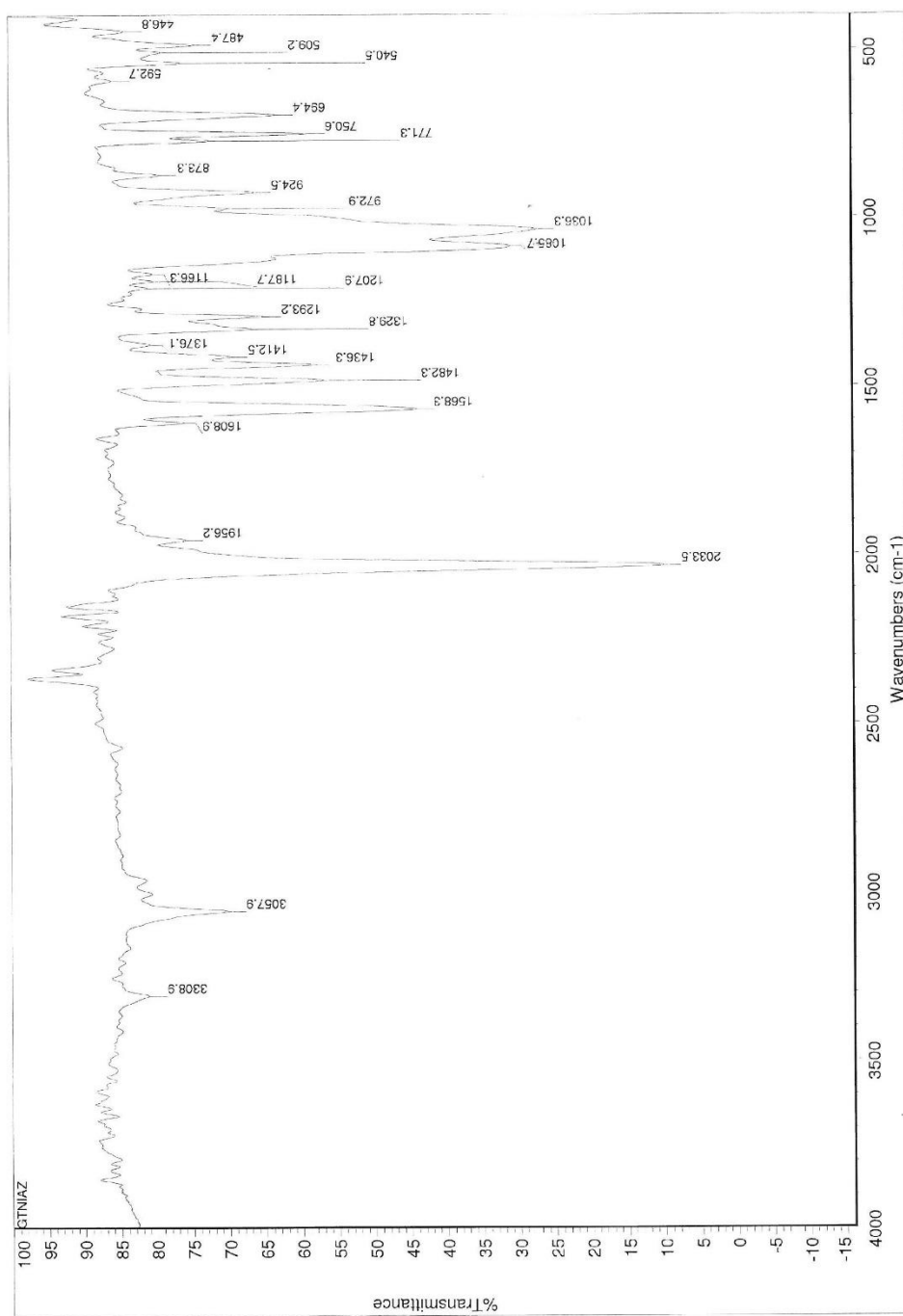


Fig. S1. IR spectrum of [NiLN₃]BF₄.

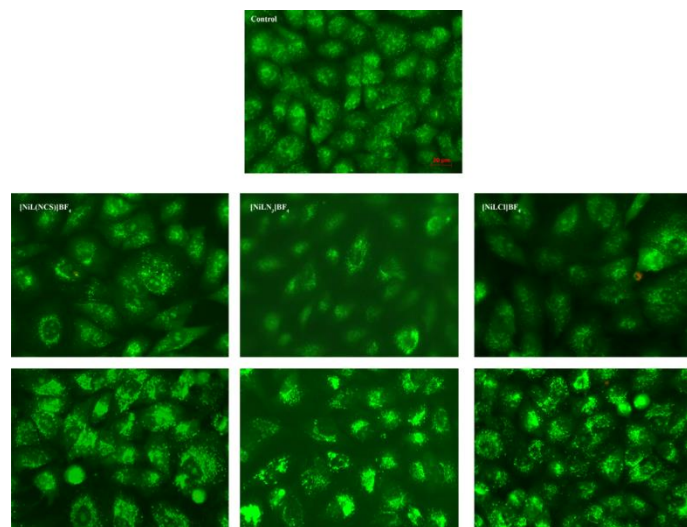


Fig. S2. Photomicrographs of acridine orange/ethidium bromide-stained control A549 cells and A549 cells exposed for 24h (upper images) and 48 h (lower images) to complexes $[\text{NiL}(\text{NCS})]\text{BF}_4$ and $[\text{NiLCl}]\text{BF}_4$. Applied concentrations of complexes corresponded to $0.5 \times \text{IC}_{50}$ values determined for 48 h.

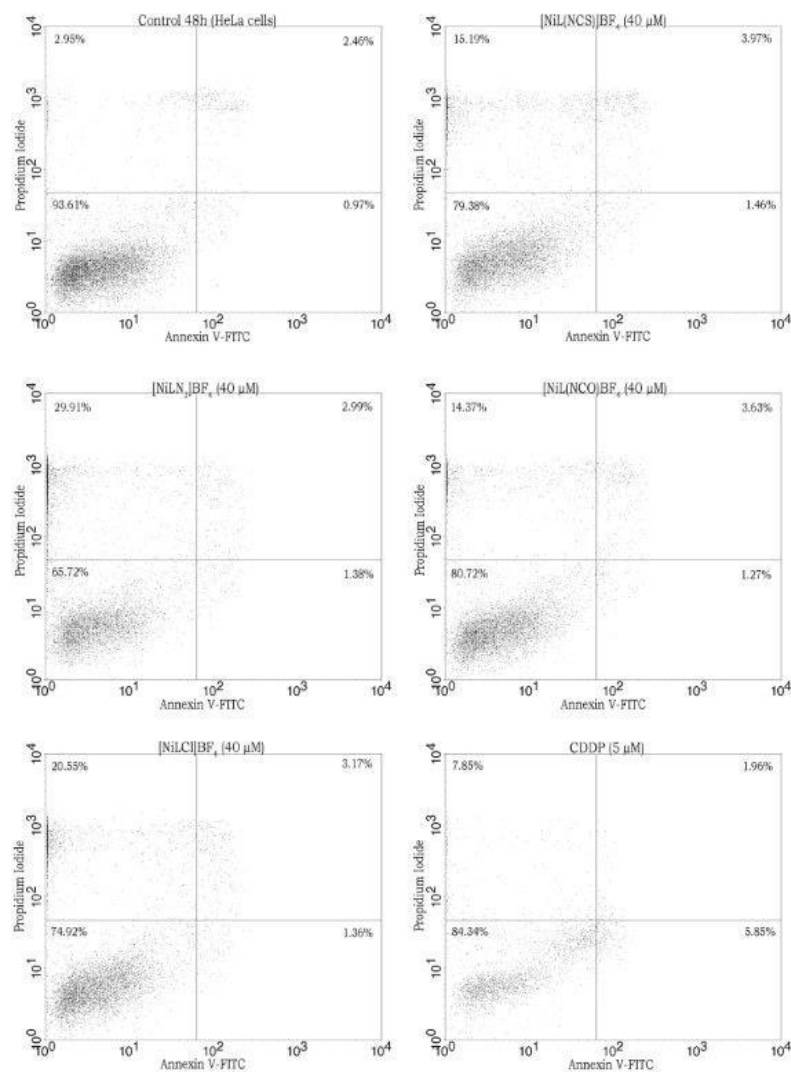


Fig. S3. Dot plot diagrams obtained by flow-cytometric analysis of treated HeLa cells after dual staining with Annexin V-FITC and PI. Annexin V-FITC/PI staining was performed after 48 h of HeLa cells exposure to nickel complex ($[\text{NiL}(\text{NCS})]\text{BF}_4$ $[\text{NiL}(\text{N}_3)]\text{BF}_4$ $[\text{NiL}(\text{NCO})]\text{BF}_4$, $[\text{NiLCI}]\text{BF}_4$) and CDDP with 40 μM concentration for nickel complexes and 5 μM concentration for cisplatin. Representative dot plots are given, presenting live cells at lower-left quadrant, FITC(-)/PI(-); early apoptotic cells at lower-right quadrant, FITC(+)/PI(-); late apoptotic or necrotic cells at upper-right quadrant, FITC(+)/PI(+); and dead cells at upper-left quadrant, FITC(-)/PI(+).

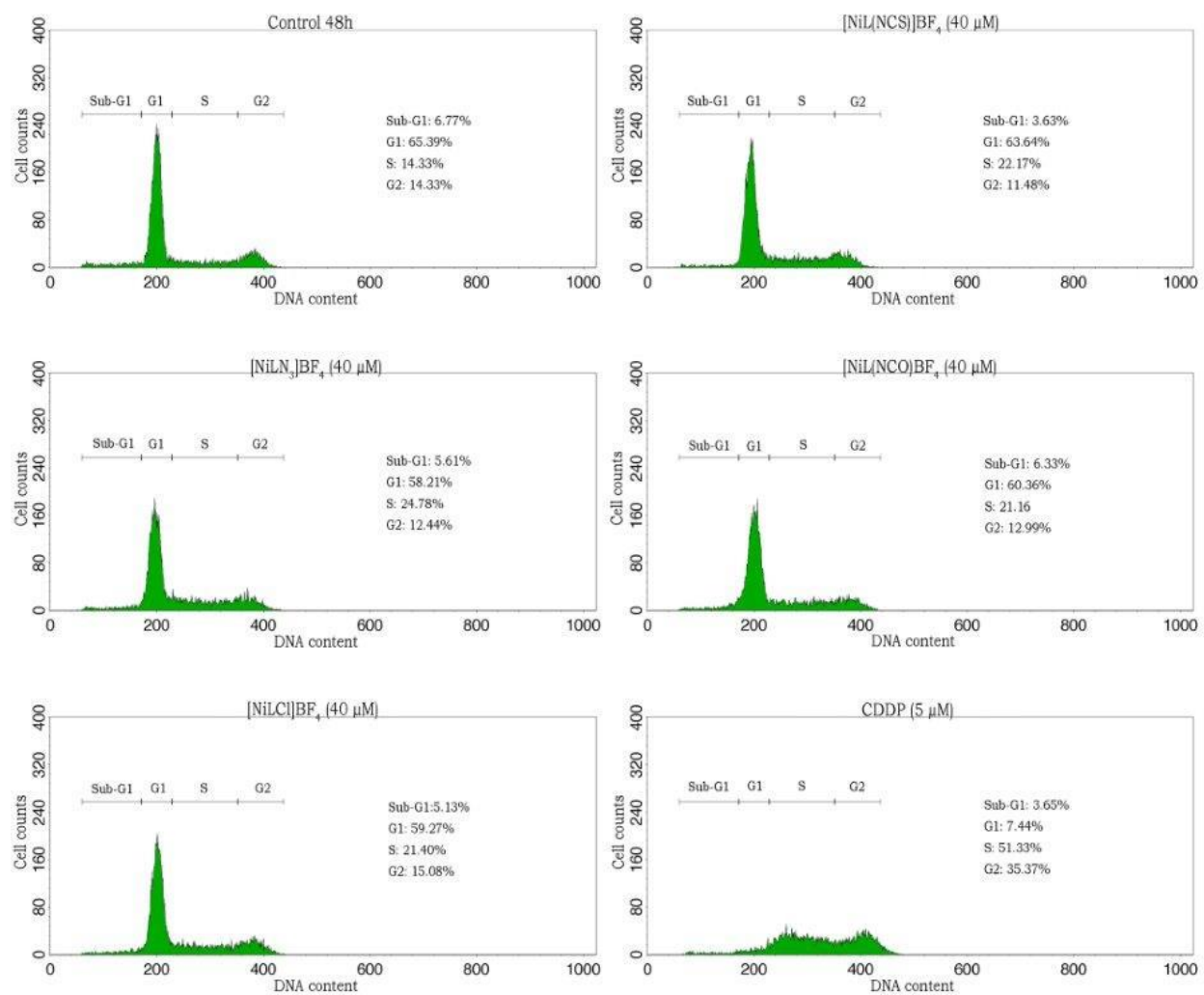


Fig. S4. Effect of the Ni(II) complexes ([NiL(NCS)]BF₄, [NiL(N₃)]BF₄, [NiL(NCO)]BF₄ and [NiLCl]BF₄) and cisplatin (CDDP) on cell cycle progression of HeLa cells following 48 h incubation with 40 μM concentration for nickel (II) complexes and 5 μM concentration for cisplatin. Histograms presented are representative of three independent experiments.

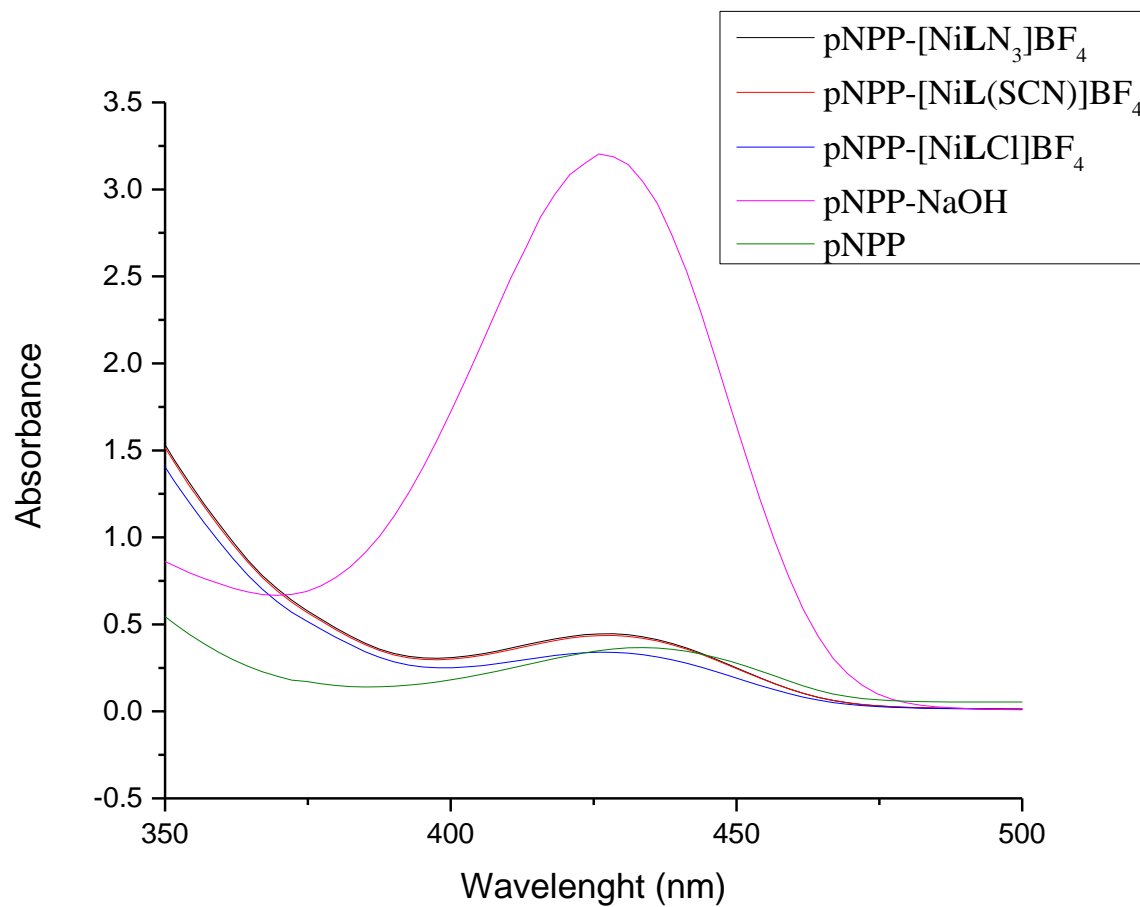


Fig. S5. Wavelength scan for hydrolysis of pNPP in the absence and presence of Ni(II) complexes and NaOH in 97.5 % DMSO. $[pNPP] = 1 \times 10^{-3}$ M; $[Ni(II) \text{ complexes}] = 5 \times 10^{-5}$ M (in the NaOH control $[NaOH] = 5 \times 10^{-5}$ M).

Table S1. Hydrogen bonding geometry for [NiLN₃]BF₄.

D – H ... A	$d(\text{D} - \text{H})/\text{Å}$	$d(\text{H} \cdots \text{A})/\text{Å}$	$d(\text{D} \cdots \text{A})/\text{Å}$	$\angle(\text{DHA})/\text{°}$	Symmetry transformation for acceptors
C1–H1A...F1	0.96	2.34	3.229(8)	154	x, 1/2-y, -1/2+z
C1–H1B...F3	0.96	2.48	3.358(9)	152	1/2-x, -y, -1/2+z
C2–H2A...F3	0.96	2.44	3.367(8)	163	x, 1/2-y, -1/2+z
C14–H14...N6	0.93	2.61	3.358(8)	138	1/2+x, y, 1/2-z