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Lake Whatcom Water Quality - Presentation to Bellingham City Council

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Lake Whatcom Water Quality

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July 6, 2015



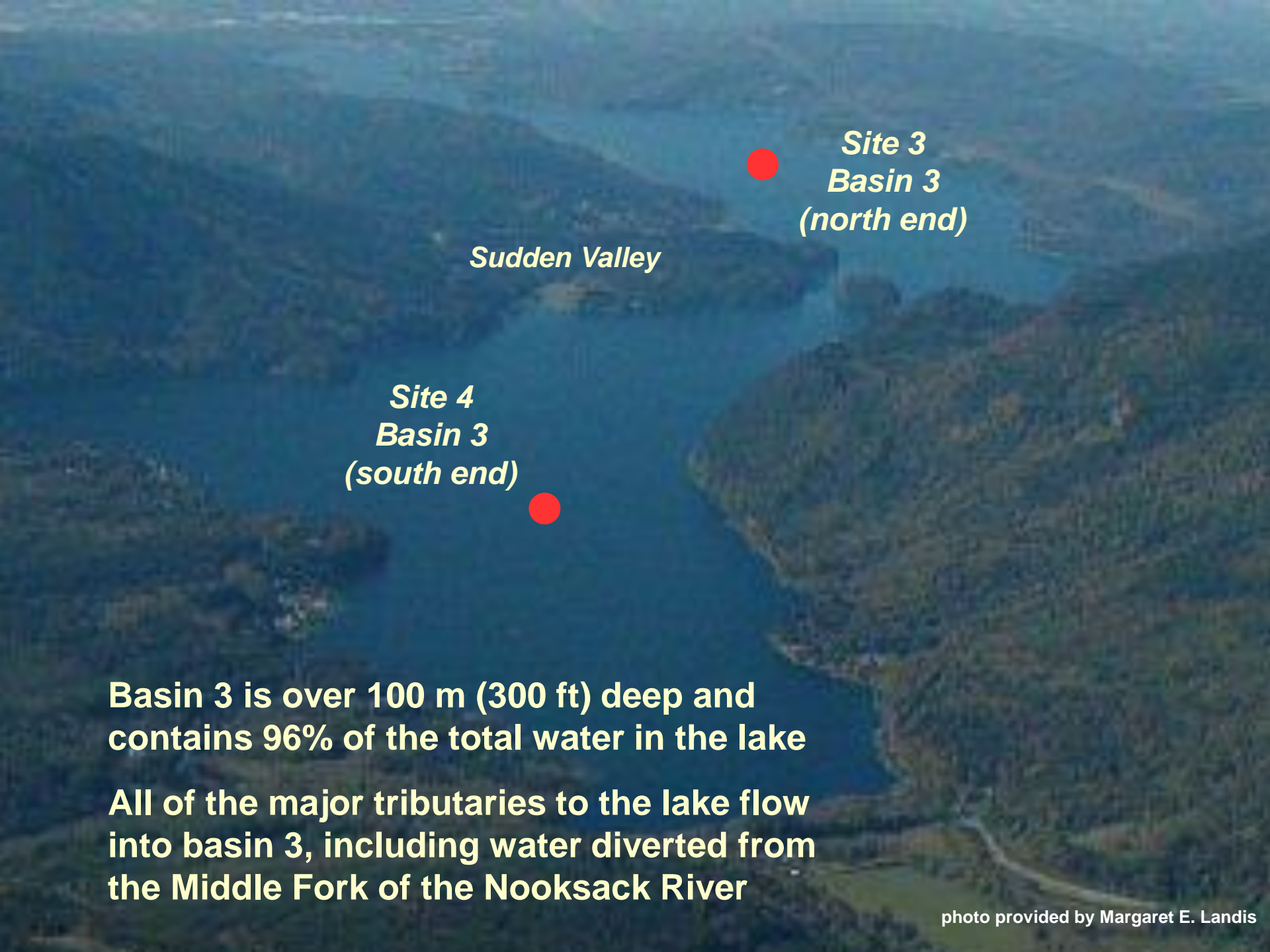
**Site 2
Basin 2**

**Site 1
Basin 1**

**Bloedel/
Donovan
Park**

**Lake Whatcom is
comprised of 2 small,
shallow basins and one
large, deep basin**

**Each shallow basin is only
~20 m (60 ft) deep and
contains about 2% of the
total water in the lake**



**Site 3
Basin 3
(north end)**

Sudden Valley

**Site 4
Basin 3
(south end)**

**Basin 3 is over 100 m (300 ft) deep and
contains 96% of the total water in the lake**

**All of the major tributaries to the lake flow
into basin 3, including water diverted from
the Middle Fork of the Nooksack River**

Lake Whatcom Monitoring Objectives

❖ Conduct long-term lake and stream monitoring

- 💧 Emphasis on lake and **storm event** monitoring
 - Silver Beach Creek 2009-2012
 - Anderson, Austin, and Brannian Creeks 2013-2015
- 💧 Tributaries in alternate years (annual 2016-2018)

❖ Collect stream hydrologic data

- 💧 Annual hydrographs for Austin and Smith Creeks
- 💧 Other tributaries monitored by USGS

Sampling Parameters



Lake, Tributaries, and Storm Event Sampling

Alkalinity
pH

Conductivity
Temperature

Dissolved oxygen
Nutrients (N/P)

Lake Only

Chlorophyll
Secchi depth

Turbidity
T. organic carbon*

Plankton
Hydrogen sulfide*

Tributaries and Storm Events Only

T. suspended solids

*infrequent sampling

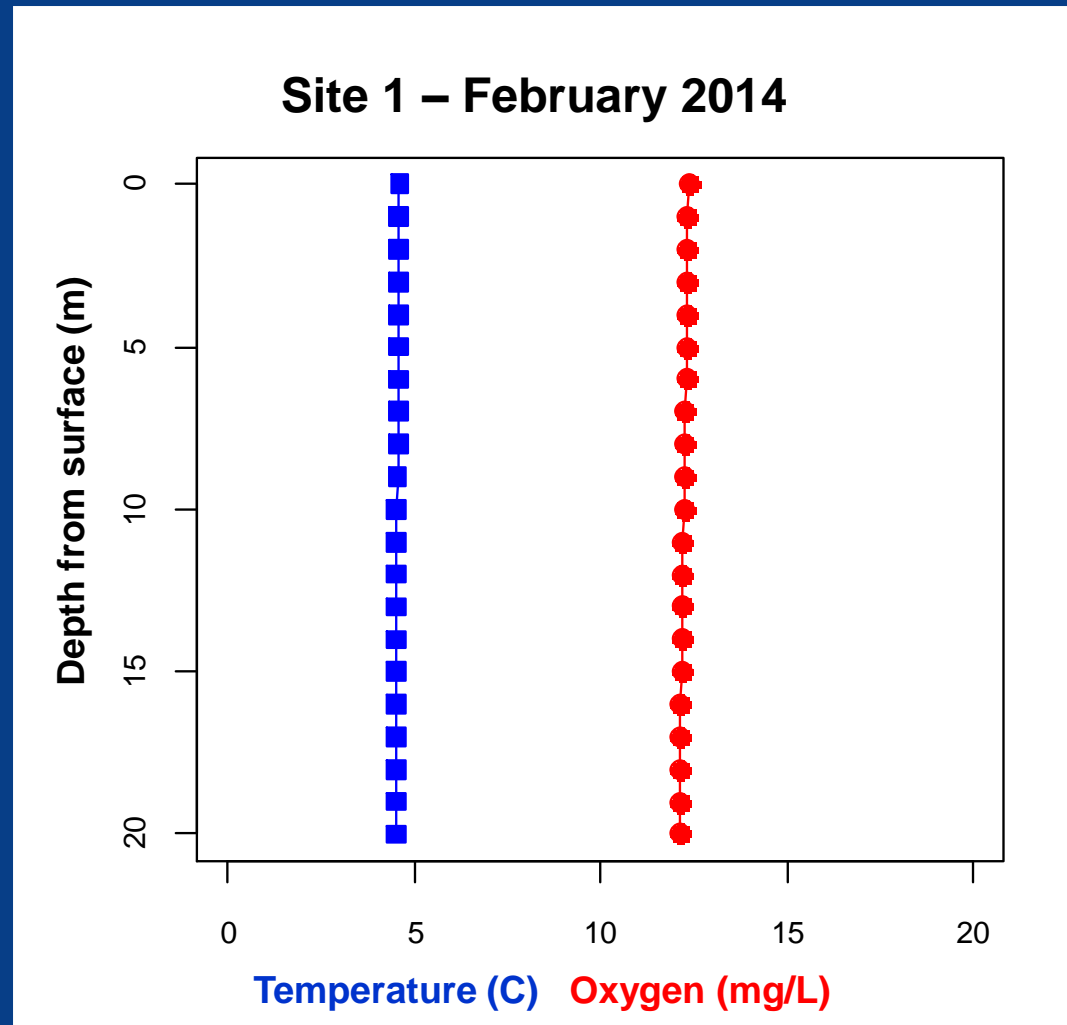


Winter Water Quality in Lake Whatcom

Lake is cold and *unstratified*;
water column mixes from
surface to bottom ... even
basin 3 (100 m)

Temperature is nearly
uniform from surface to
bottom

Dissolved oxygen and most
other compounds are nearly
uniform from surface to
bottom

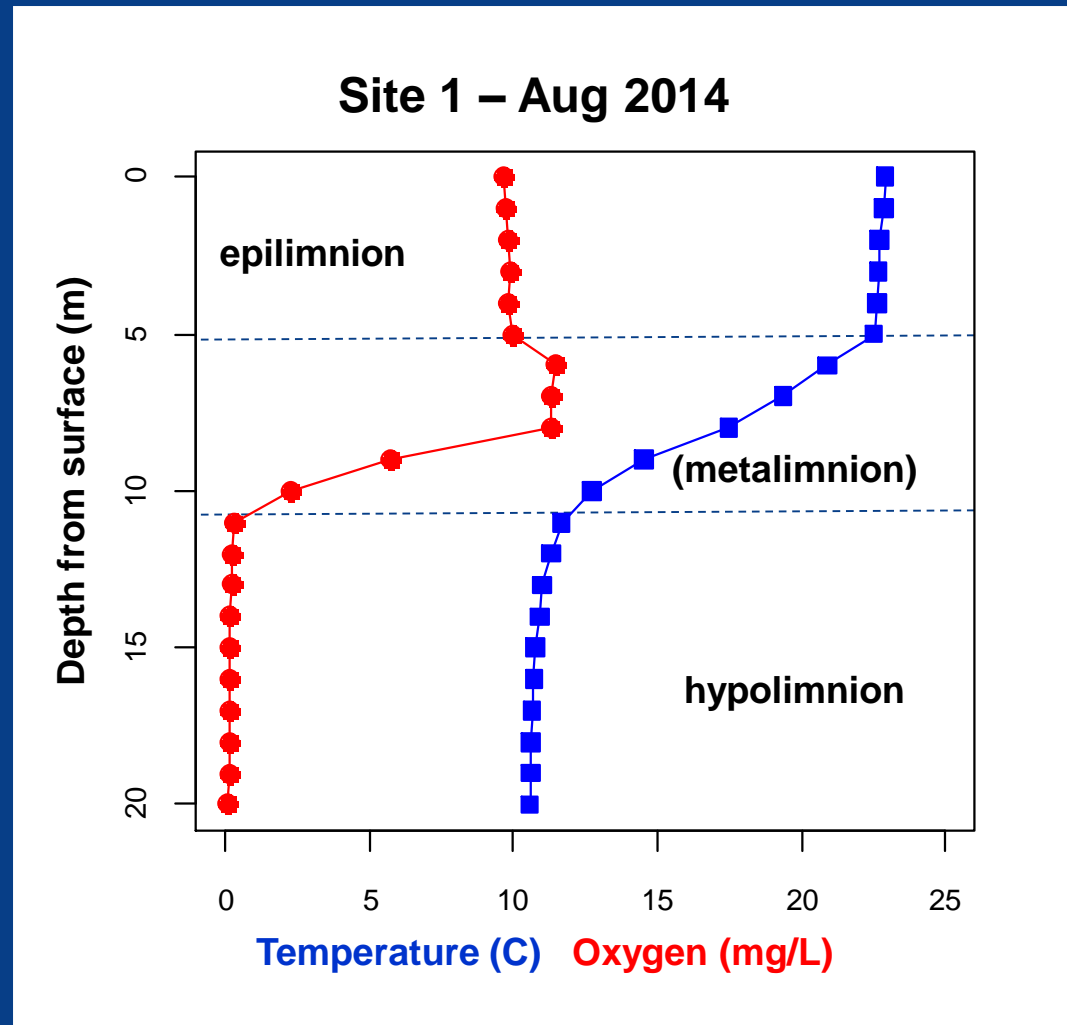


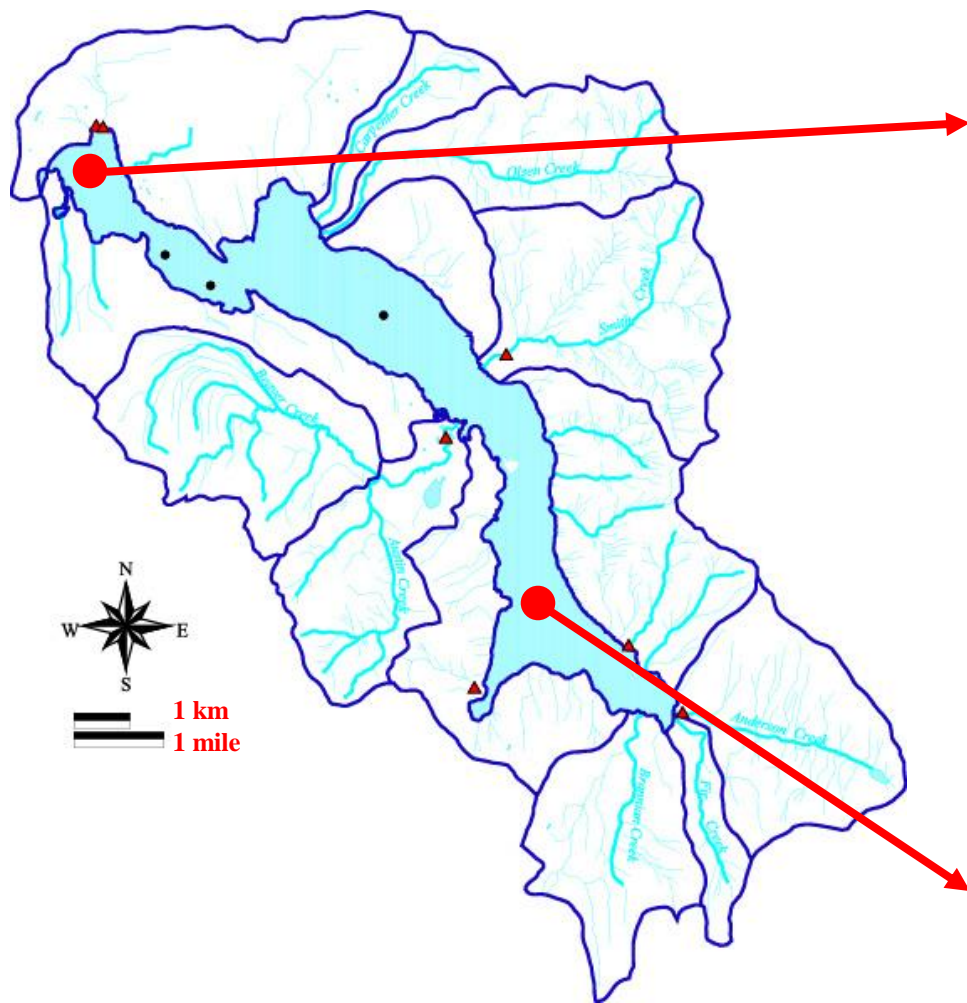
Summer Water Quality in Lake Whatcom

Lake becomes *stratified* into a warm surface layer (*epilimnion*) and cold bottom layer (*hypolimnion*)

Once stratified, wind can't mix the entire water column

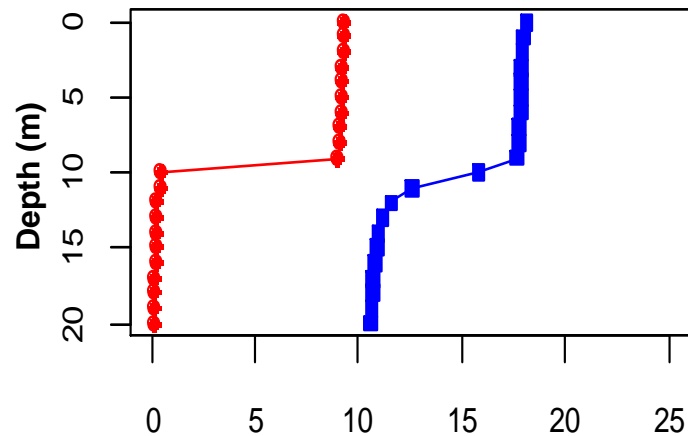
In parts of the lake (Sites 1-2), oxygen is depleted in the hypolimnion as bacteria decompose organic matter (dead algae, leaf fragments, etc.)





Basins 1-2 have different oxygen profiles than basin 3

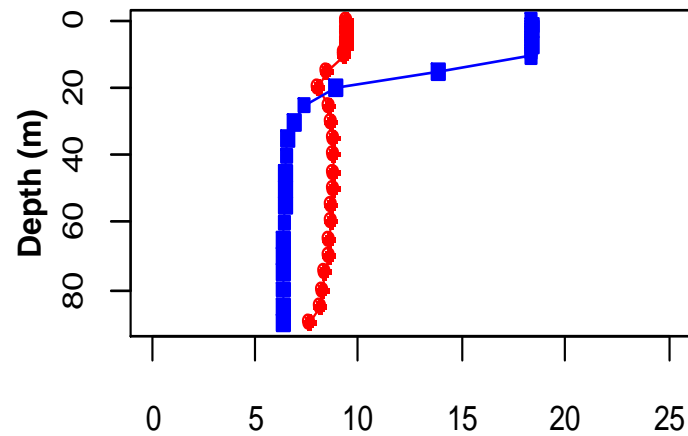
Site 1 - Oct 2014



Oxygen

Temperature

Site 4 - Oct 2014

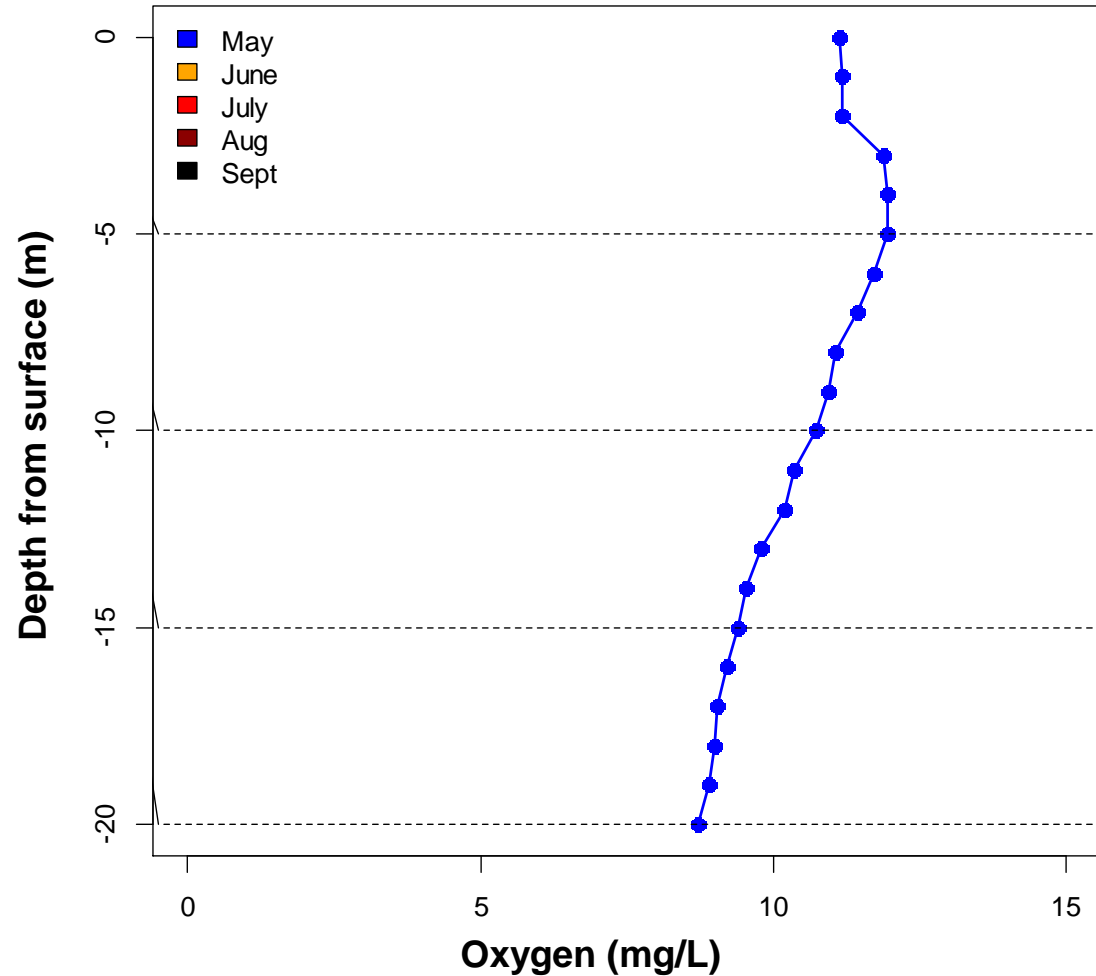


At Sites 1-2
hypolimnetic oxygen depletion begins after the lake stratifies

This may start as early as April* but usually begins in May or early June, depending on weather conditions

*2015 – may have very early stratification

Site 1 - May 2013



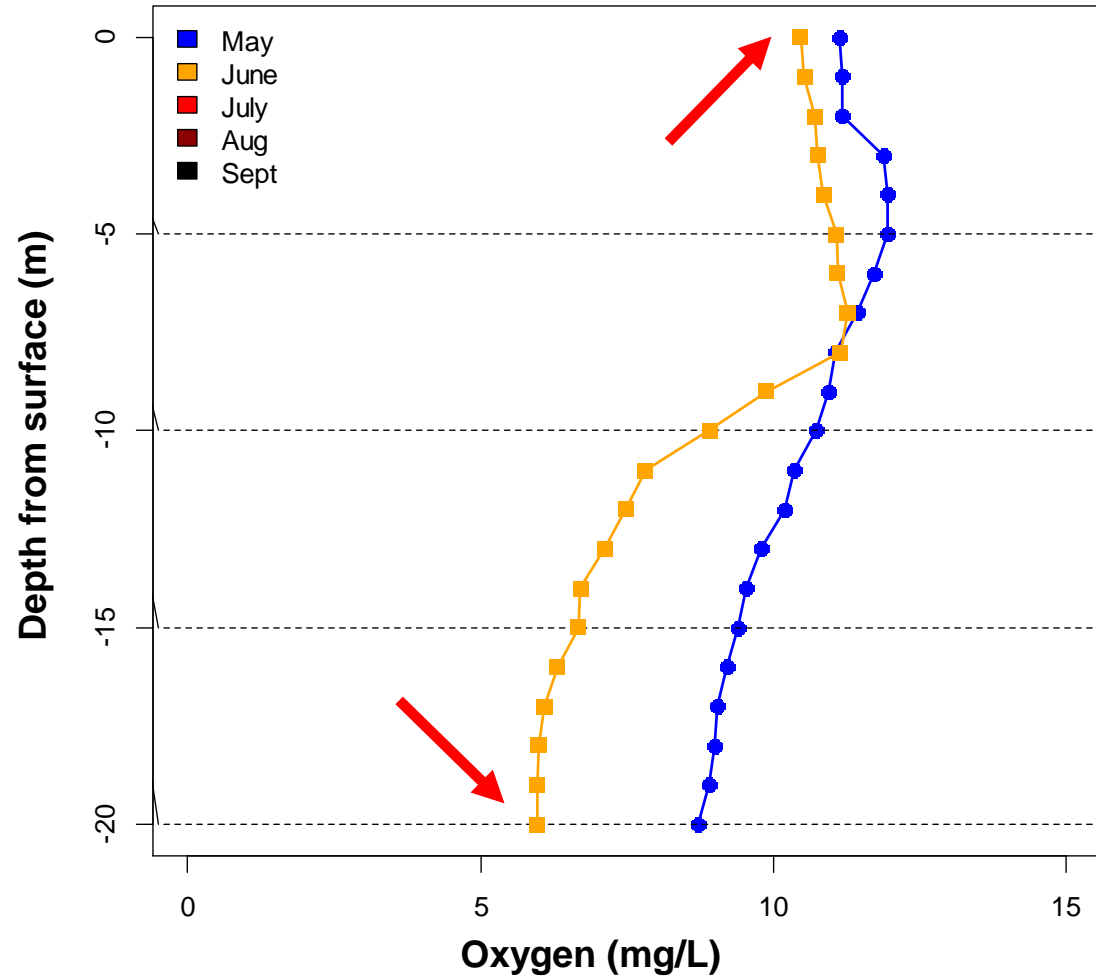
An oxygen sag typically develops by June

By June 2013, oxygen levels were 4.5 mg/L lower at the bottom compared to the surface

Surface oxygen levels fall slightly at the surface because the water is warmer

→ warm water holds less oxygen than cold water

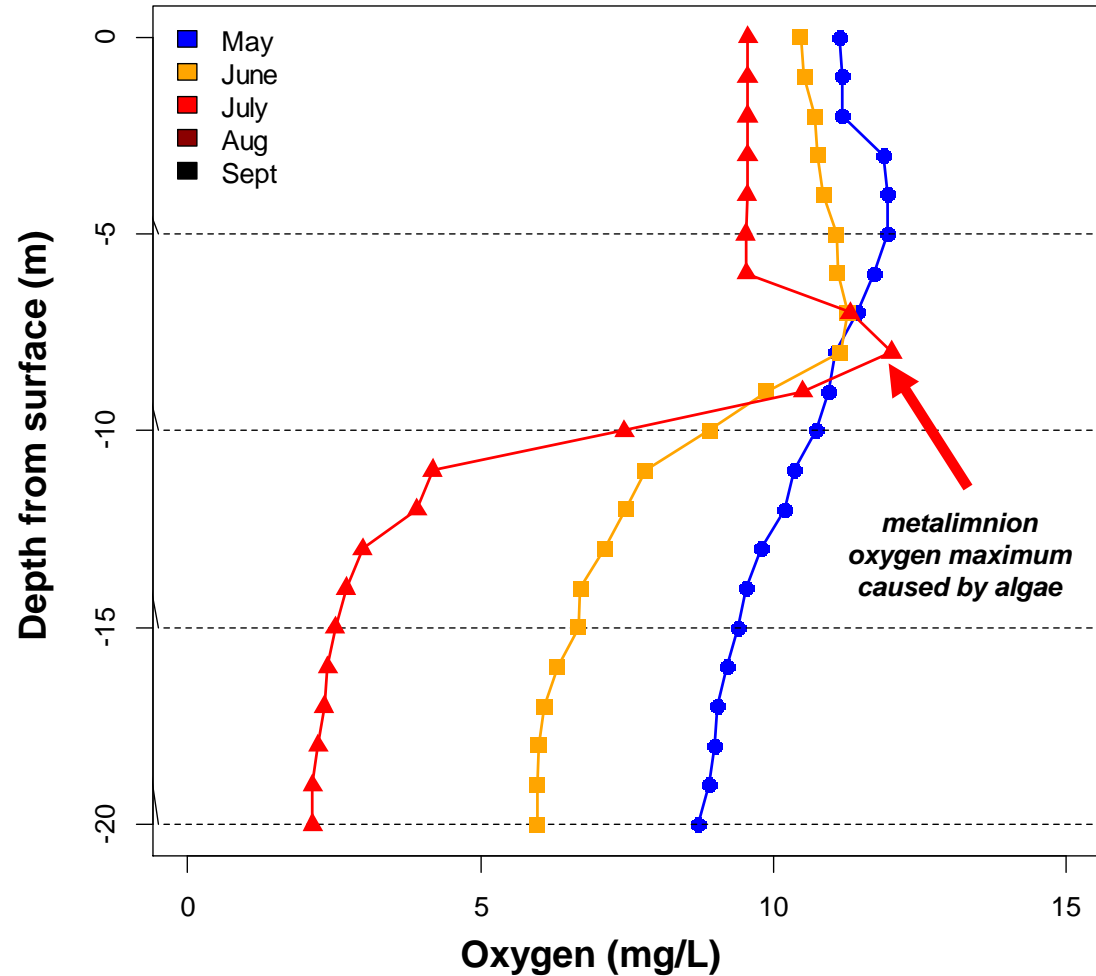
Site 1 - May and June 2013



As the summer progresses, the oxygen depletion in the hypolimnion becomes increasingly evident

The bulge between 5-10 meters is a *metalimnetic oxygen maximum* caused by bands of algae

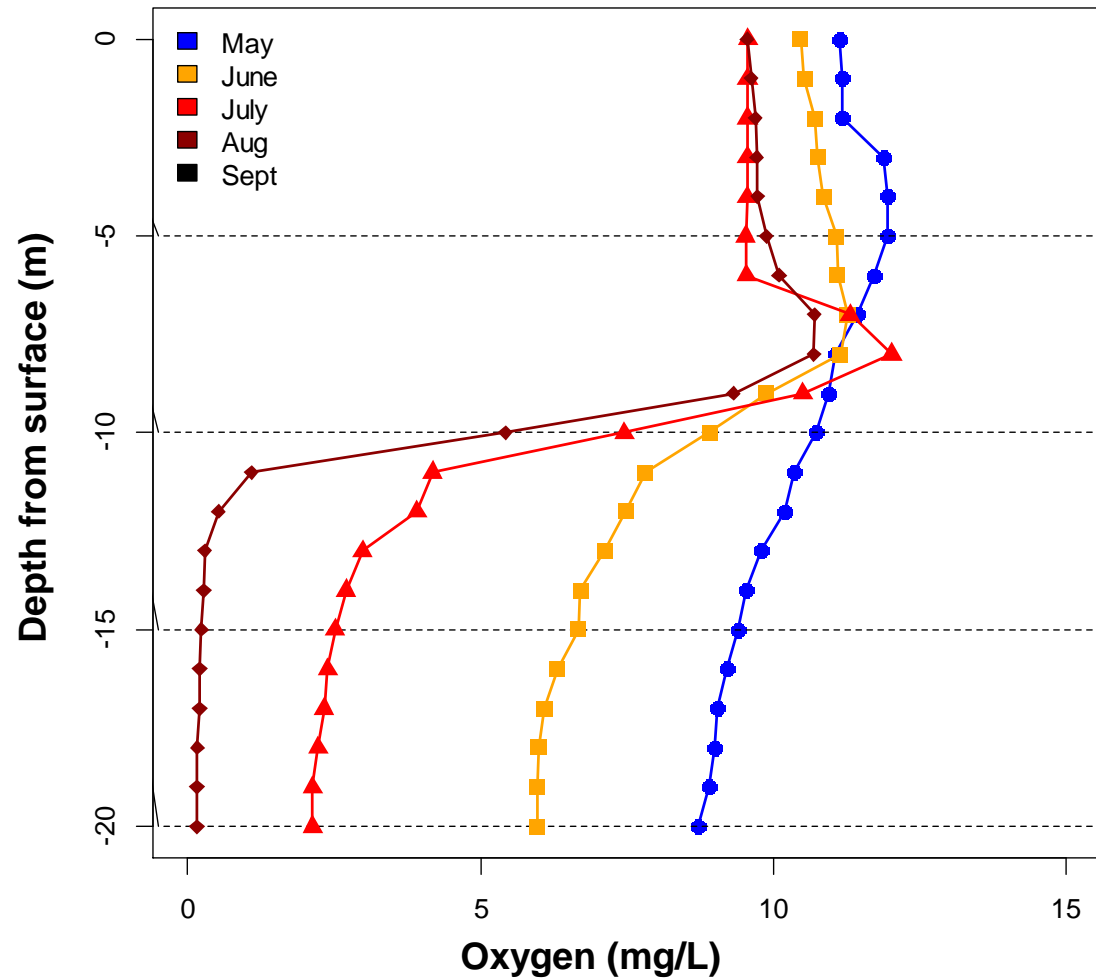
Site 1 – May to July 2013



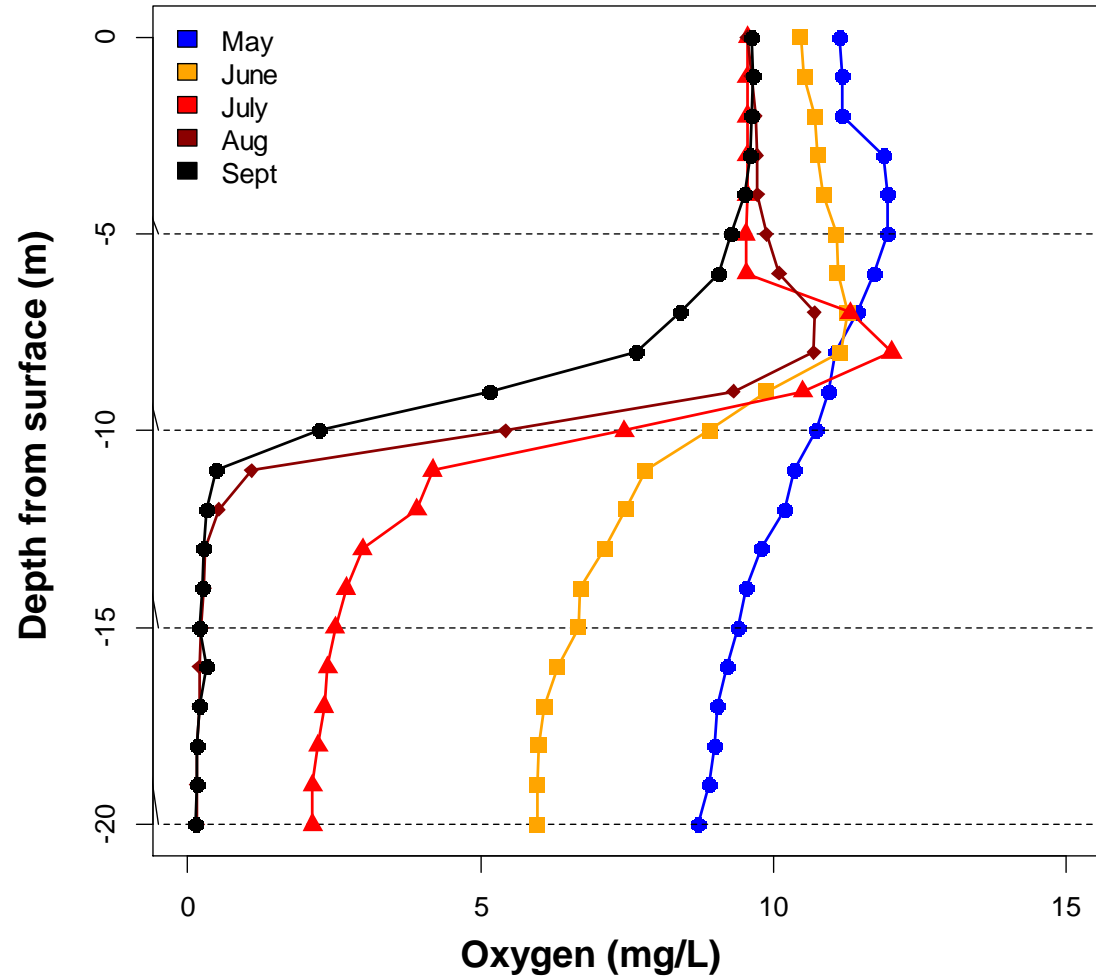
By August, there is almost no oxygen in the hypolimnion

Once oxygen levels fall below ~2 mg/L, the only aquatic organisms that thrive are *anaerobic bacteria*

Site 1 – May to August 2013

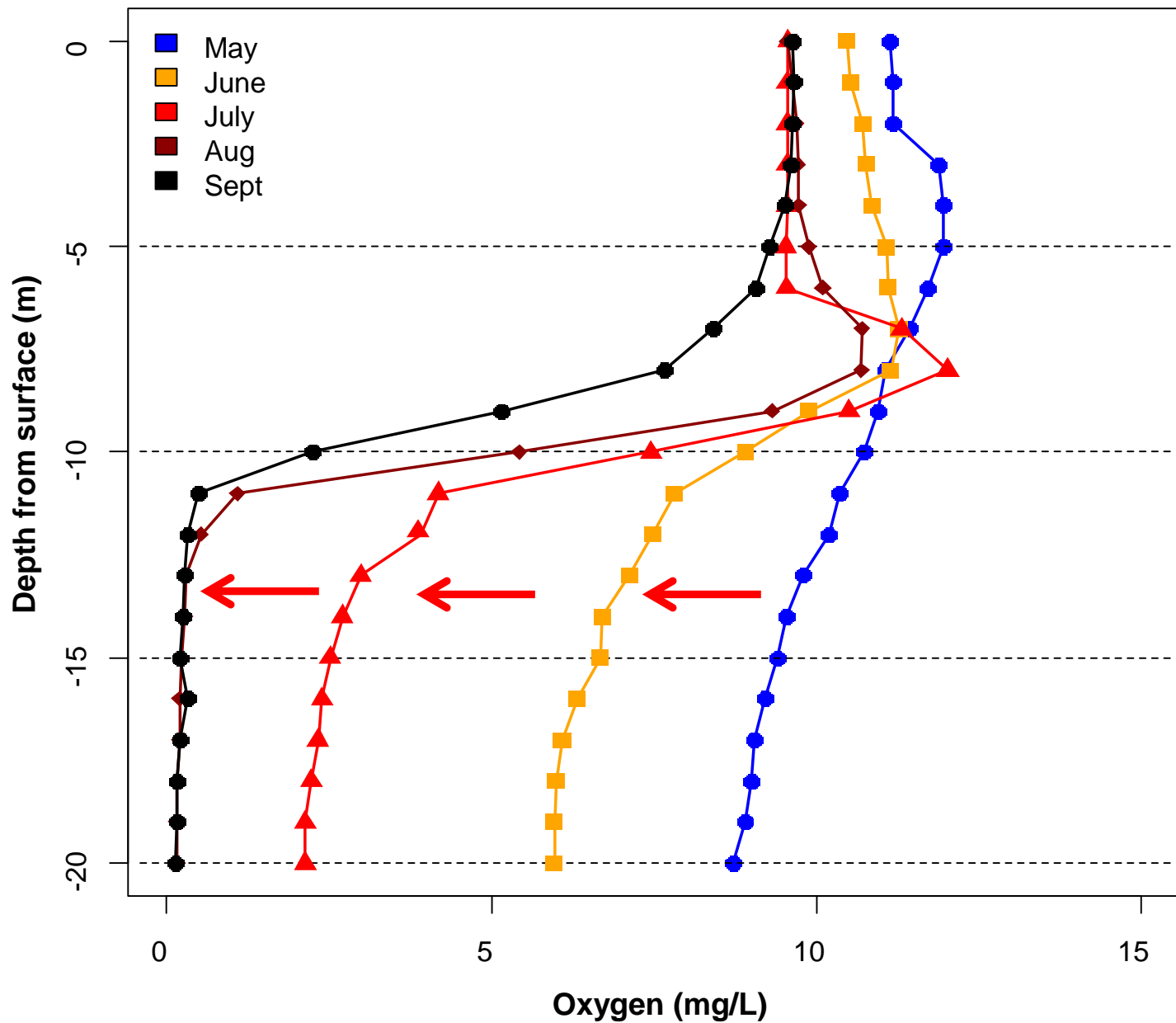


Site 1 - May- to September 2013



The September hypolimnion oxygen concentrations resemble August because additional oxygen won't be introduced until destratification (Oct/Nov at Sites 1-2; Dec/Jan at Sites 3-4)

The rate of hypolimnetic oxygen consumption is increasing



Water Quality Problems Associated With Low Dissolved Oxygen

- ❖ Loss of aquatic habitat

 - 💧 fish need at least 4-6 mg/L dissolved oxygen

- ❖ Release of nutrients and other compounds from the sediments

 - 💧 Dissolved metals, methylated mercury, hydrogen sulfide

 - 💧 **Phosphorus**

More Phosphorus = More Algae

More Phosphorus = More Algae



Internal sources

**Phosphorus released from
sediments (low oxygen)**

More Phosphorus = More Algae



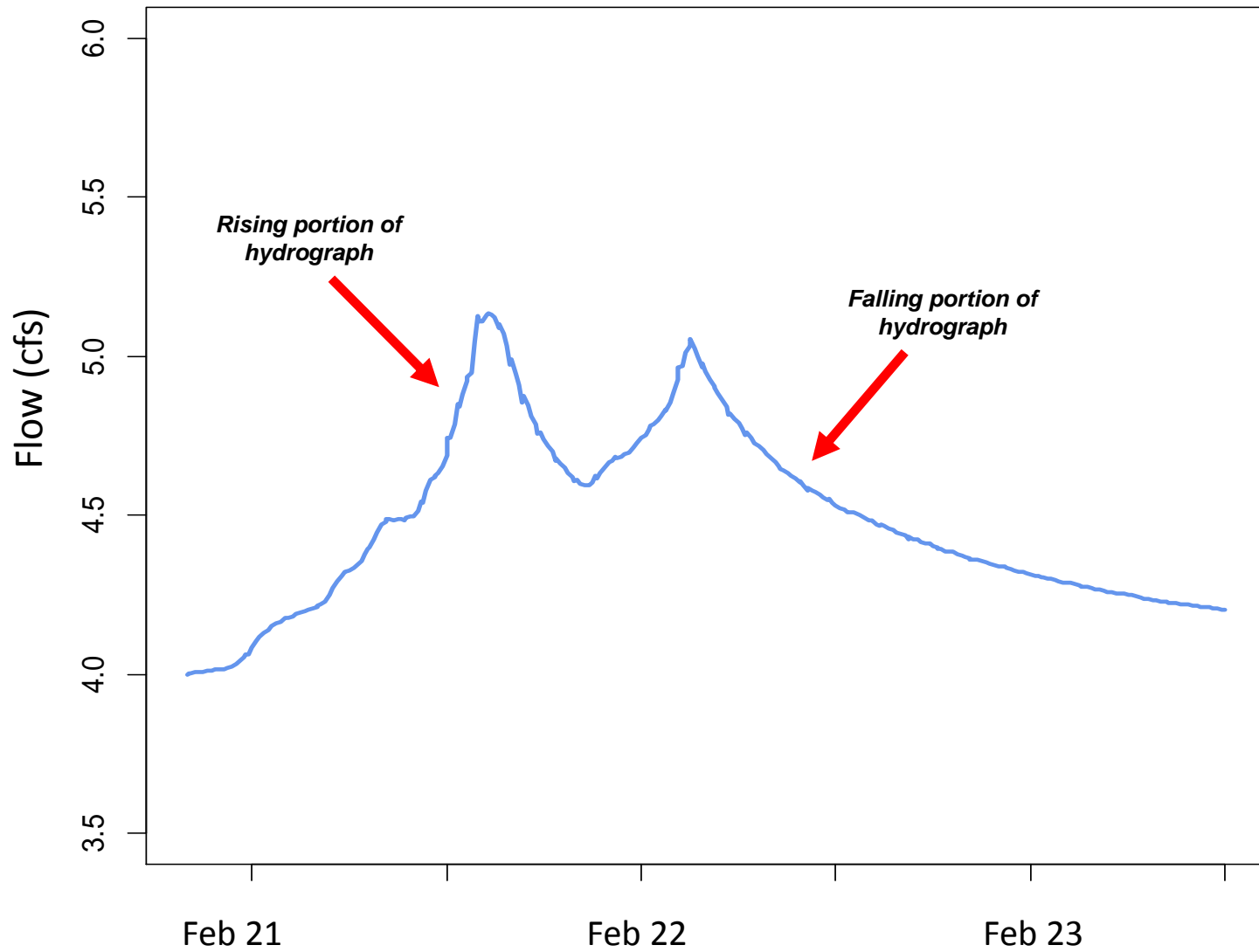
Internal sources
Phosphorus released from
sediments (low oxygen)

External sources
Phosphorus transported in
surface runoff

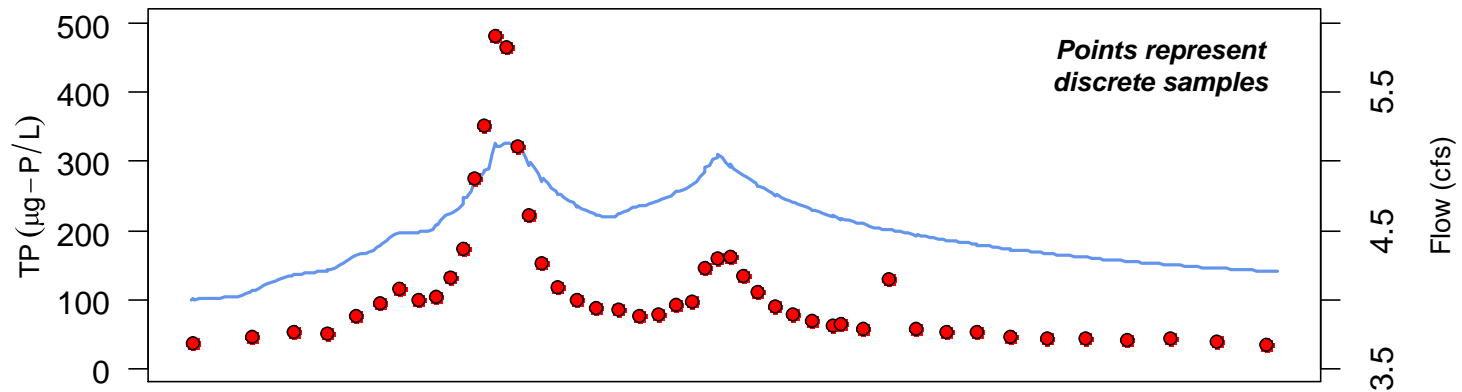
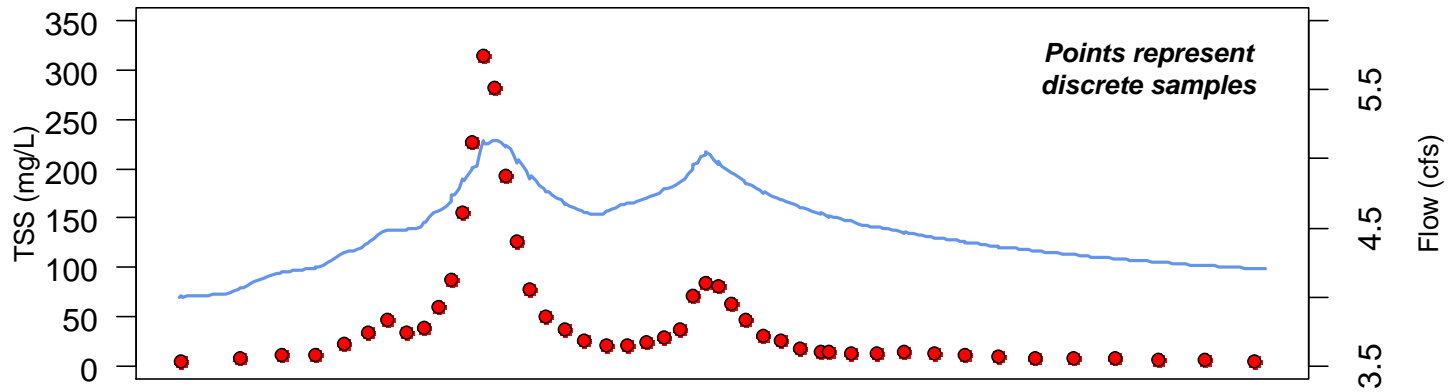
Storm Water Monitoring

- ❖ Samples collected during storms of ≥ 1 cm in 24-hr
 - 💧 At least 7 samples collected during storm event
 - 💧 Samples analyzed for total suspended solids and phosphorus
 - 💧 Data used to model phosphorus loading into the lake
- ❖ 2010-2012: Silver Beach Creek (24 events)
- ❖ 2013-2015: Austin, Anderson, Brannian Creeks (14 events)
- ❖ 2013-2014: Smith Creek (22 storm events; Beeler M.S. thesis)

Silver Beach Creek Storm Water Monitoring (Event #23: Feb 20-23, 2012)



Silver Beach Creek Storm Water Monitoring (Event #23: Feb 20-23, 2012)

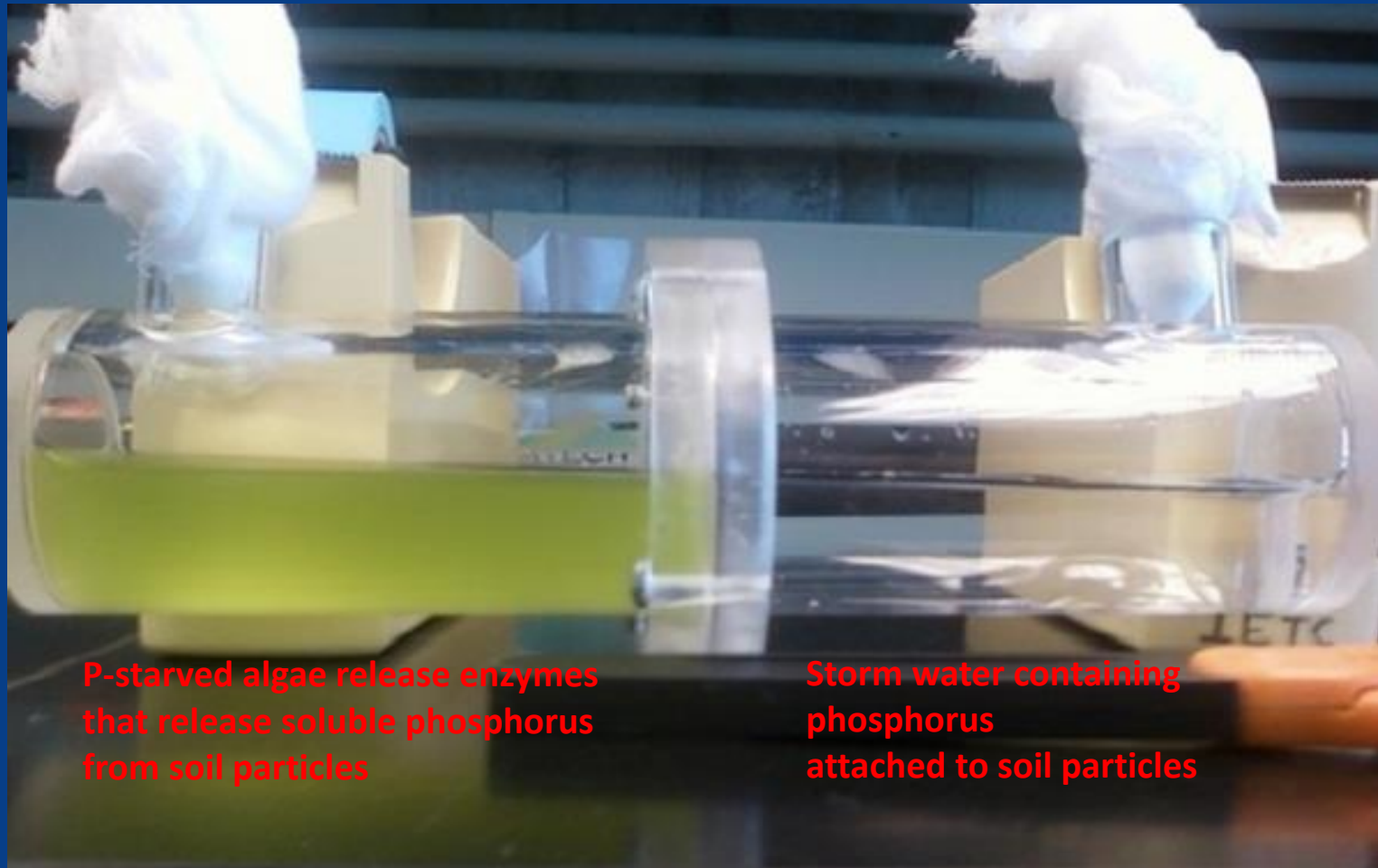


Relationship between Phosphorus in Storm Runoff and Algae

Lake Whatcom algae sample
Oct 2010 – magnified 200x



Although phosphorus enters Lake Whatcom attached to soil particles, it doesn't necessarily stay attached!

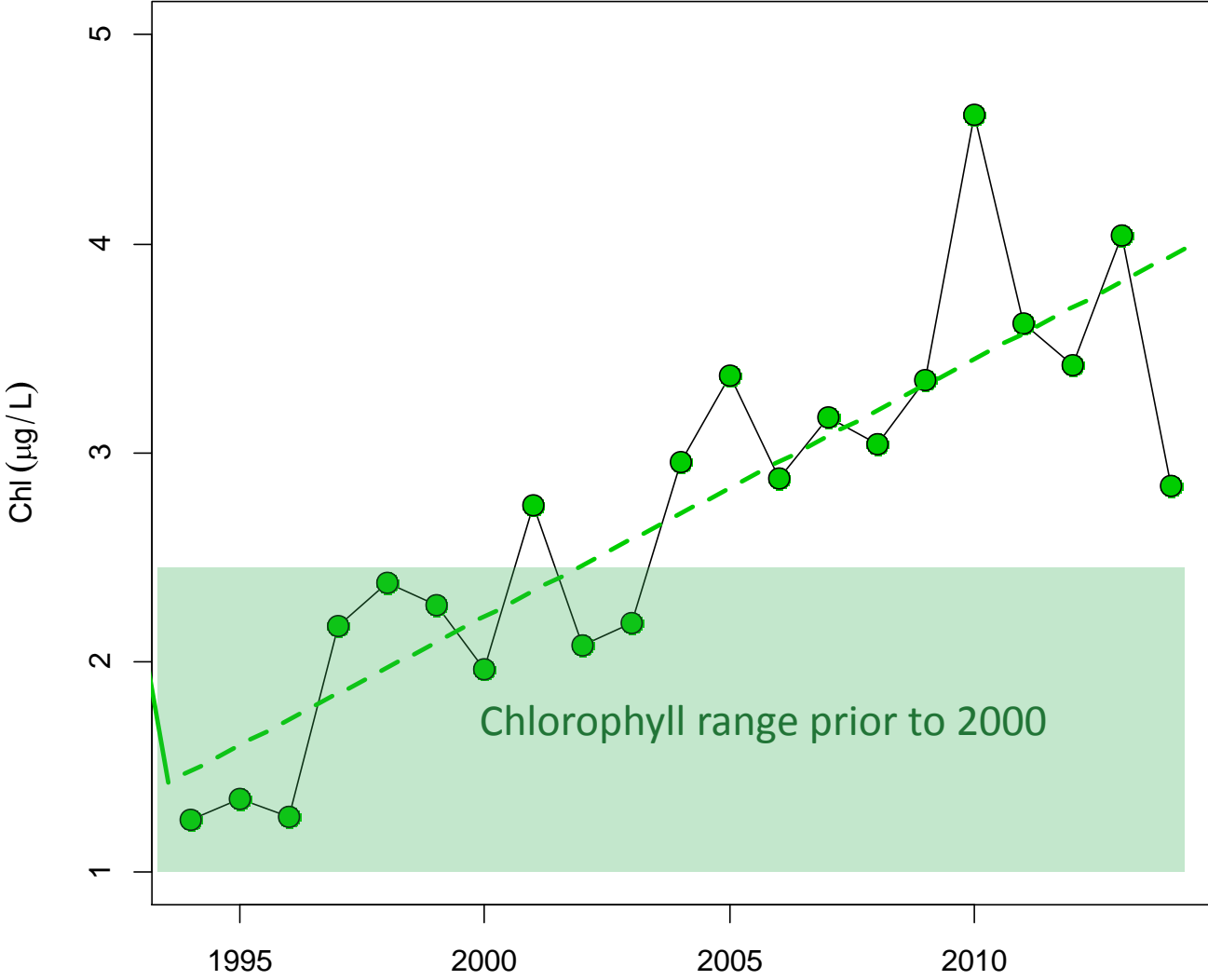


P-starved algae release enzymes that release soluble phosphorus from soil particles

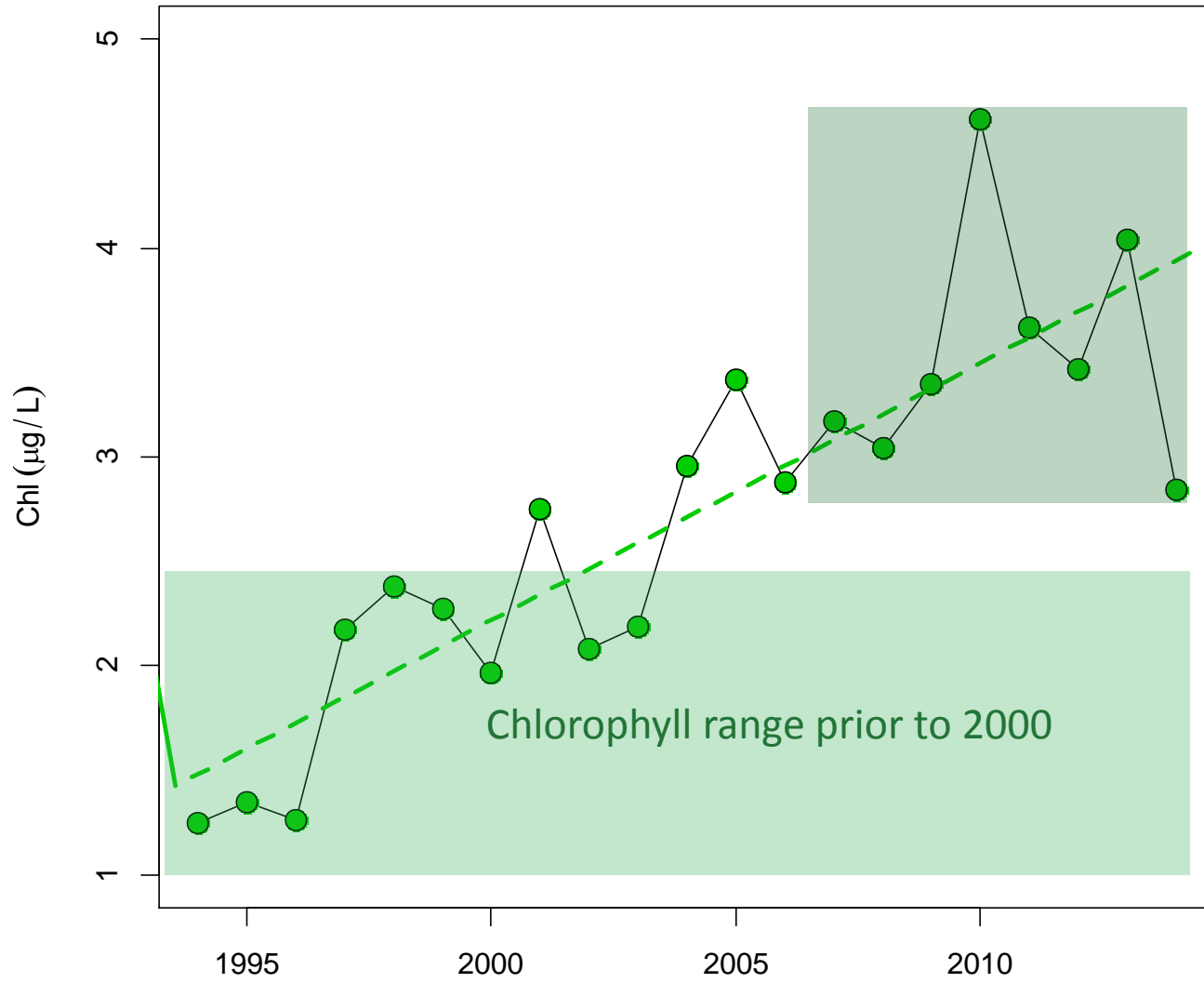
Storm water containing phosphorus attached to soil particles

Determining biological available phosphorus in storm water entering Lake Whatcom, WA using the dual culture diffusion apparatus, J. Deacon, 2015. M.S. thesis

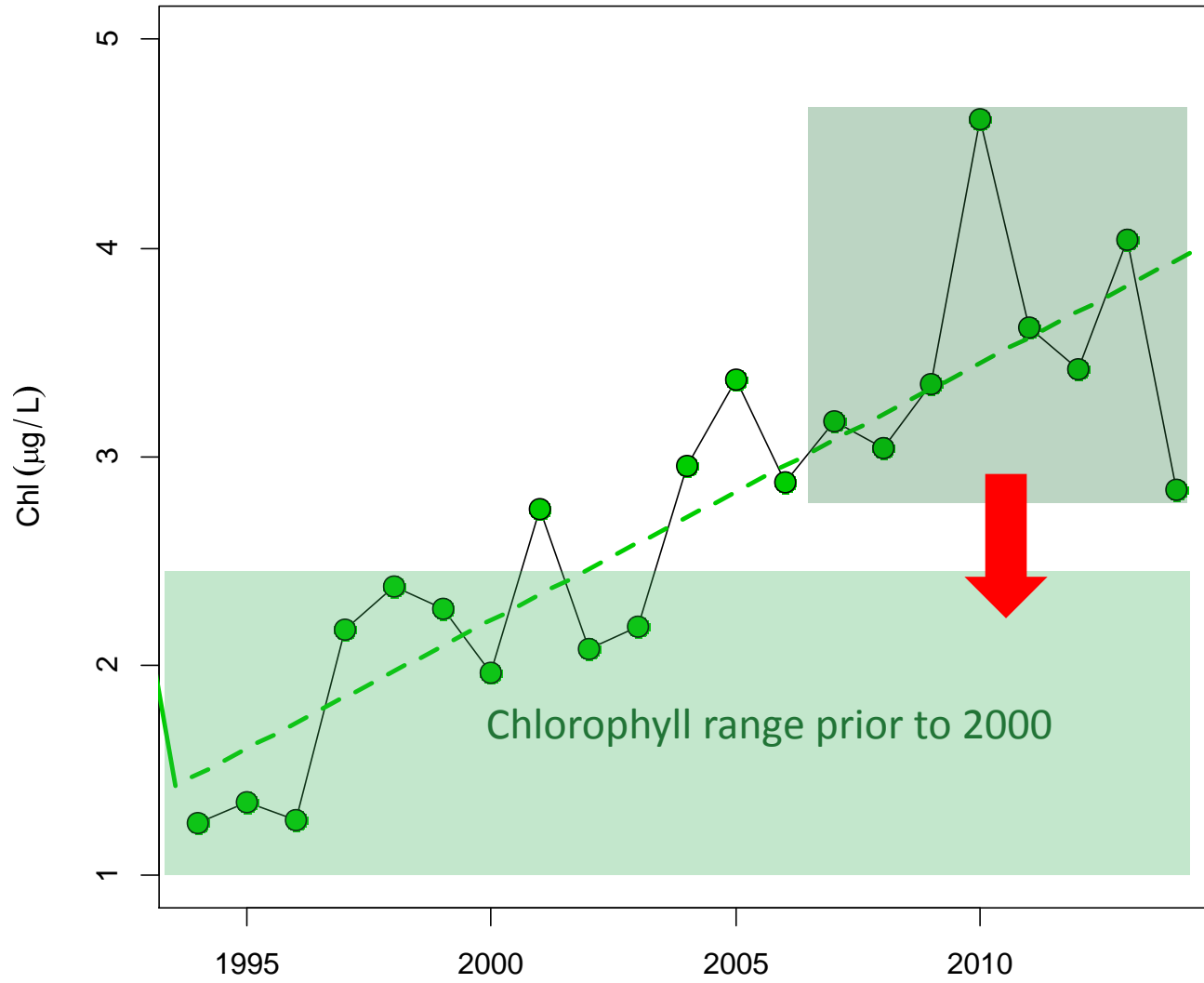
Increasing Chlorophyll at Site 4 in Lake Whatcom



Increasing Chlorophyll at Site 4 in Lake Whatcom



Increasing Chlorophyll at Site 4 in Lake Whatcom



Water Quality Problems Associated With High Concentrations of Algae

- ❖ Positive feedback loop between algae and phosphorus
 - 💧 Algae remove phosphorus from water and soil particles, causing algal growth
 - 💧 Decomposing algae release phosphorus, which causes more algal growth
- ❖ Some lakes have toxic algae blooms (currently not a problem in Lake Whatcom)
- ❖ Increased drinking water treatment costs
 - 💧 Disinfection byproducts (THMs)
 - 💧 Taste and odor problems
 - 💧 Decreased water filtration rates

Where are we now?

- ❖ Lake Whatcom Annual Report summarizes the current conditions
 - 💧 Hypolimnetic oxygen levels still low at Site 1
 - 💧 Storm runoff carries increased concentrations of sediment and phosphorus into the lake
 - 💧 Chlorophyll concentrations (and algal counts) still high throughout the lake

Where are we now?

- ❖ Lake Whatcom Annual Report summarizes the current conditions
 - 💧 Hypolimnetic oxygen levels still low at Site 1
 - We can't do much about this directly, but it should improve if the amount of algae in the lake can be reduced

Where are we now?

- ❖ Lake Whatcom Annual Report summarizes the current conditions
 - 💧 Storm runoff carries increased concentrations of sediment and phosphorus into the lake
 - The TMDL is designed to reduce phosphorus loading from the watershed
 - This will ultimately reduce the amount of algae in the lake

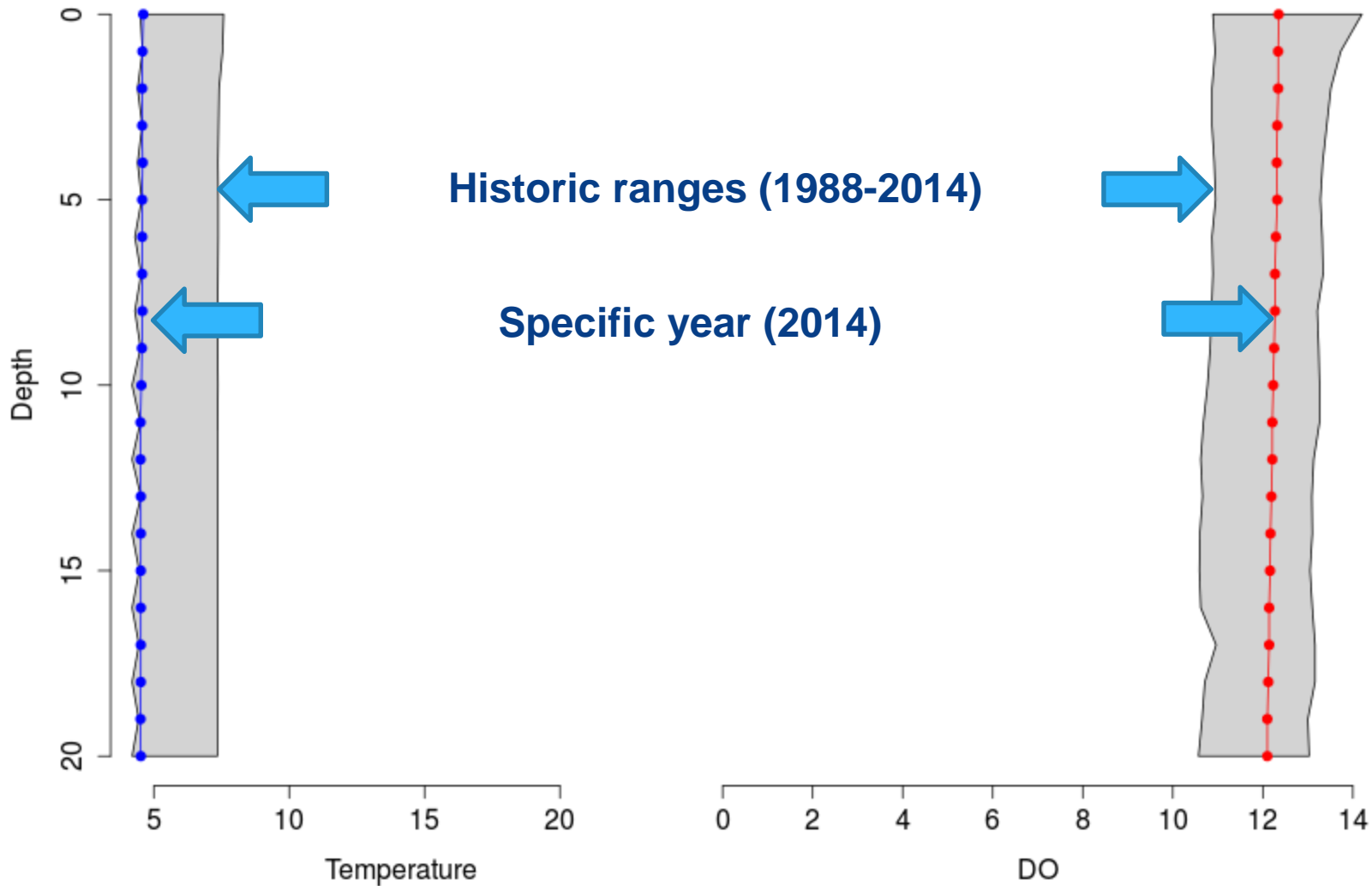
Where are we now?

- ❖ Lake Whatcom Annual Report summarizes the current conditions
 - 💧 Chlorophyll concentrations (and algal counts) still high throughout the lake
 - This will improve (slowly) if we reduce the amount of phosphorus entering the lake
 - To address current water quantity requirements, source water pretreatment may be needed

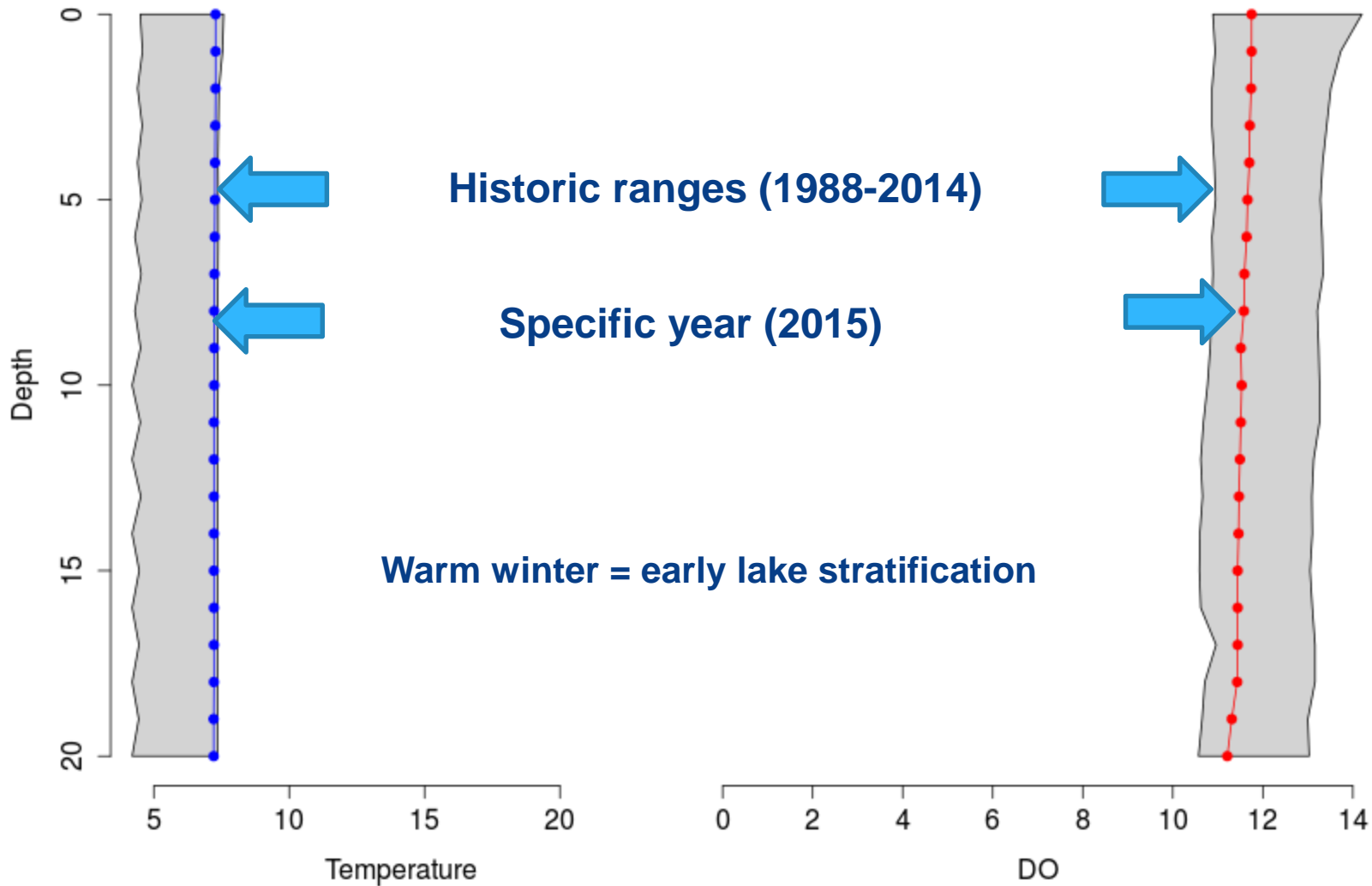
2015 – Not a Typical Year

The Feb-June 2015 weather has resulted in temperature and dissolved oxygen profiles that are unlike any patterns measured in the past 30 years

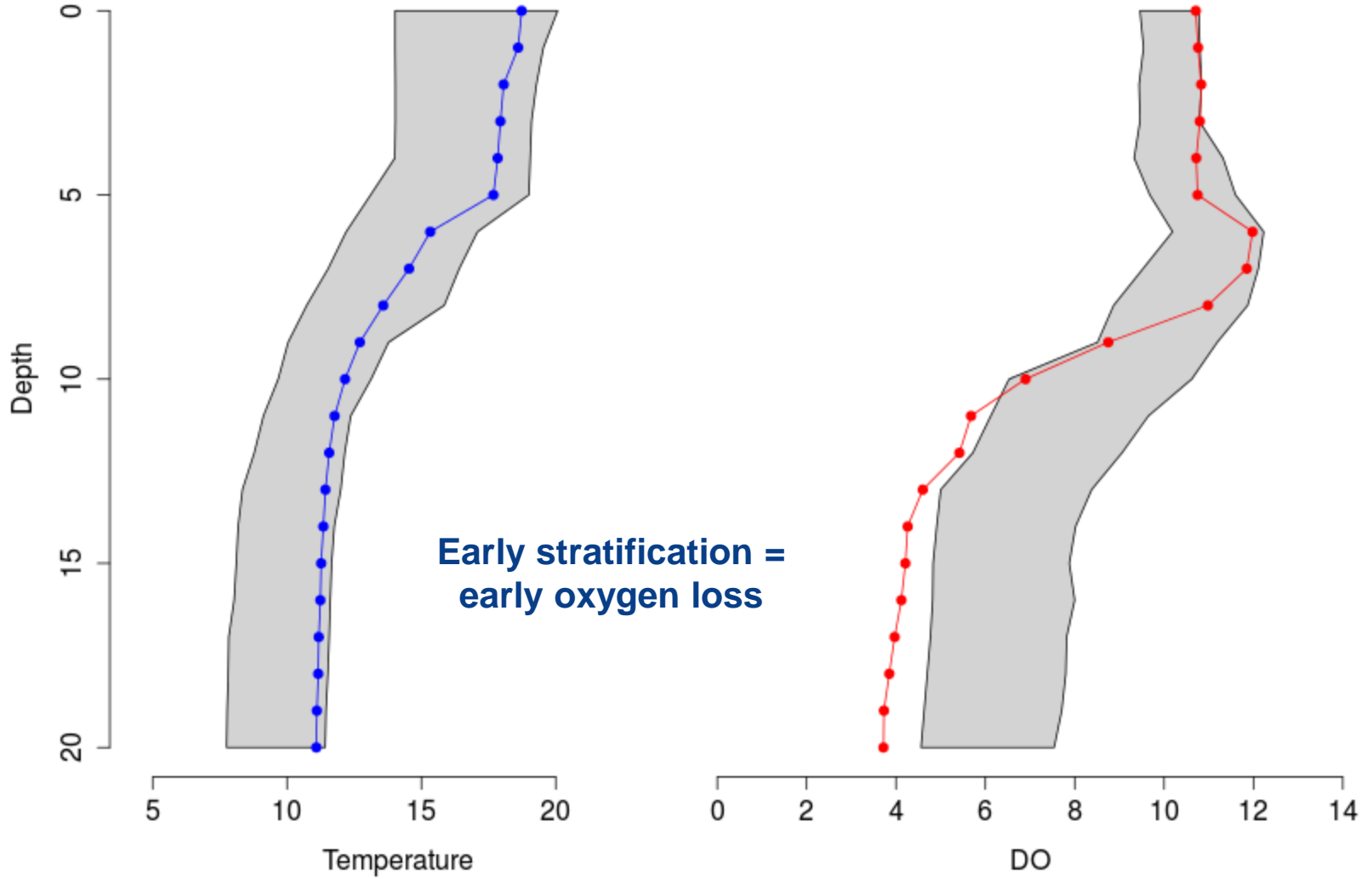
Site 1, February 2014



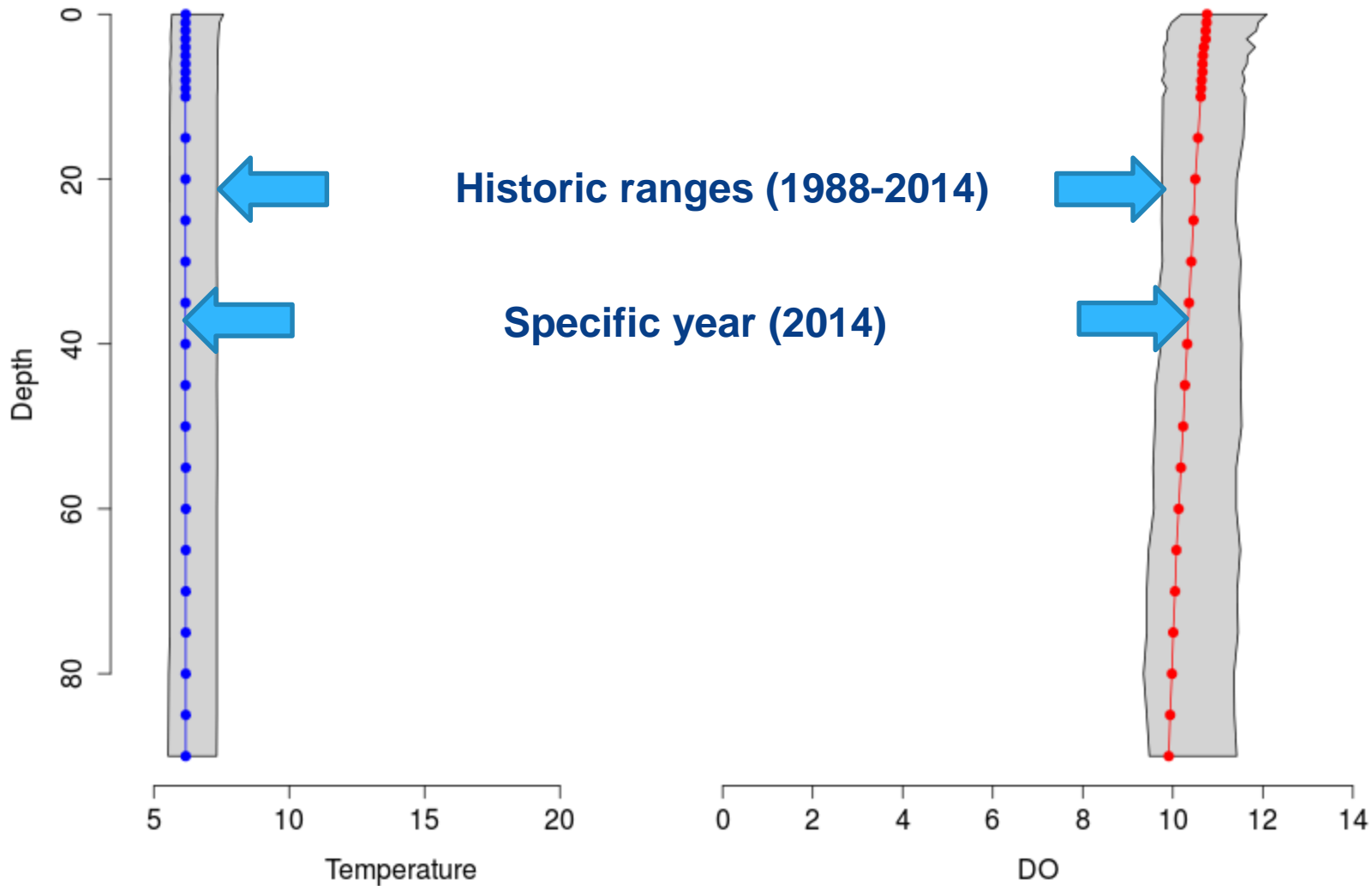
Site 1, February 2015



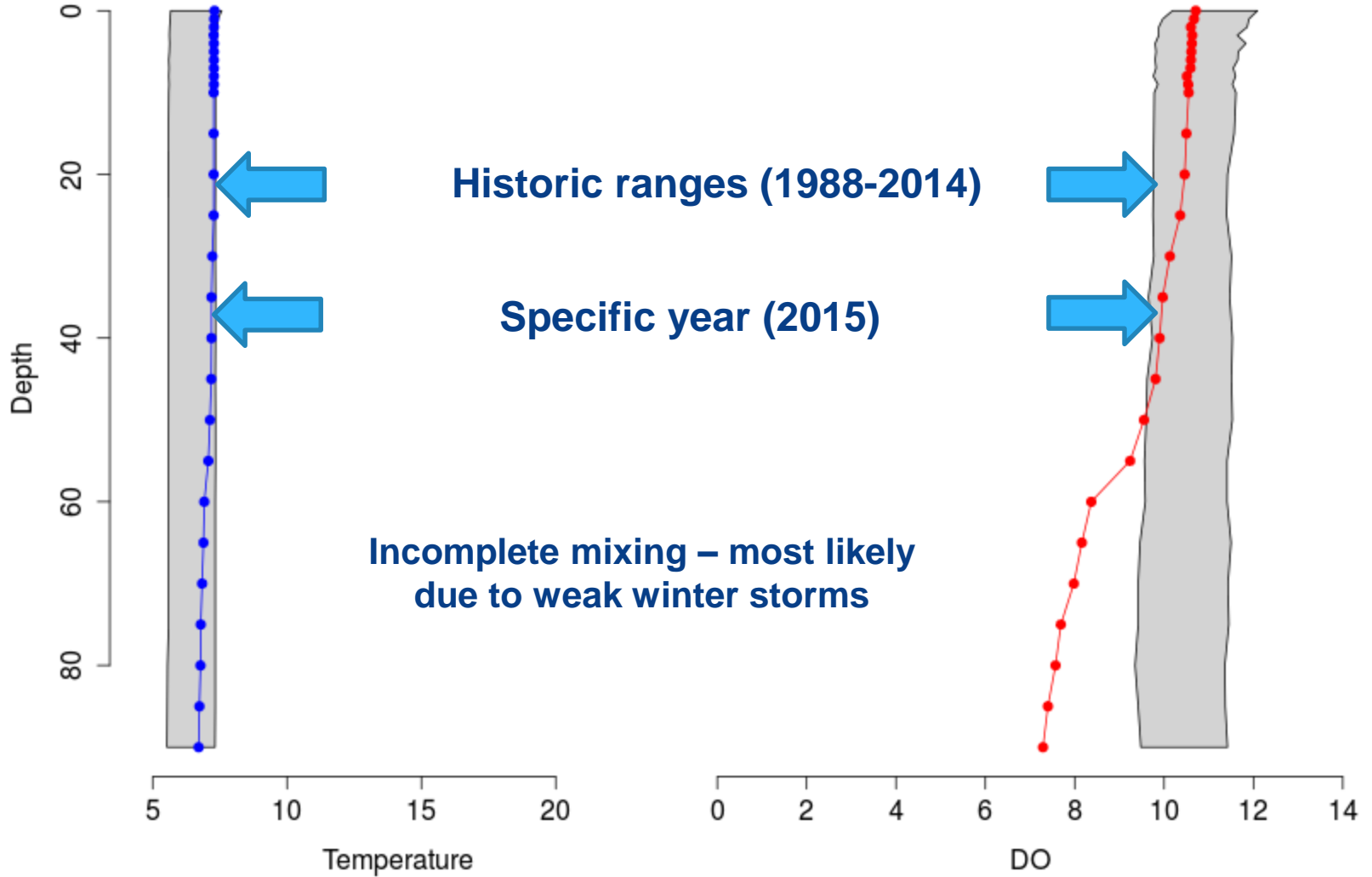
Site 1, June 2015



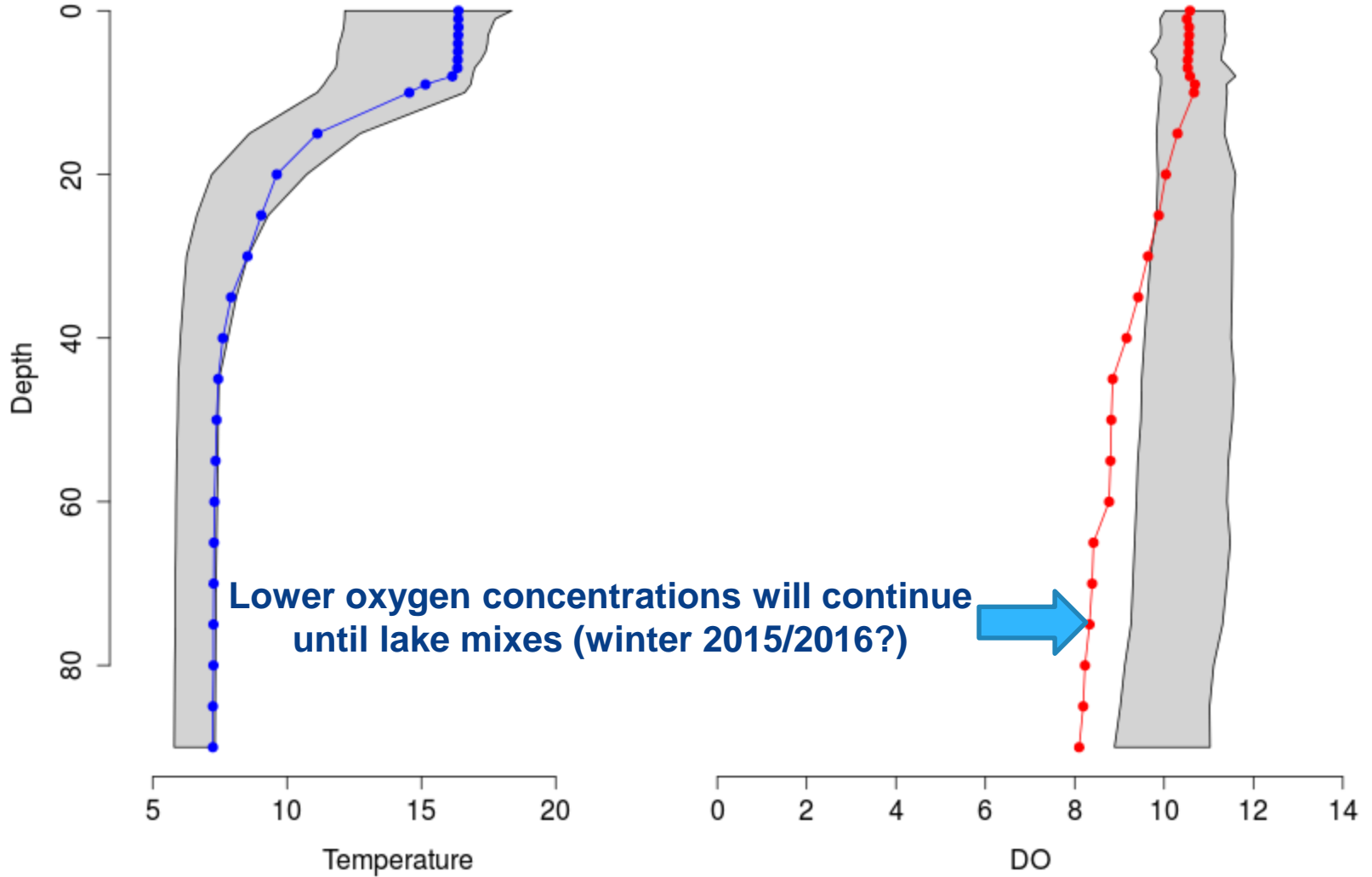
Site 4, February 2014



Site 4, February 2015



Site 4, June 2015



Thanks!

Mike Hilles

Joan Vandersypen

Marilyn Desmul

Dr. Robert Mitchell

Dr. Geoffrey Matthews

undergraduate and graduate students
working on the Lake Whatcom Project