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Enhanced Metallic Properties of SrRuO₃ Thin Films via Kinetically Controlled Pulsed Laser Epitaxy

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5 Electronic Measurement Pitfalls to Avoid

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Enhanced metallic properties of SrRuO₃ thin films via kinetically controlled pulsed laser epitaxy

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Metal electrodes are a universal element of all electronic devices. Conducting SrRuO₃ (SRO) epitaxial thin films have been extensively used as electrodes in complex-oxide heterostructures due to good lattice mismatches with perovskite substrates. However, when compared to SRO single crystals, SRO thin films have shown reduced conductivity and Curie temperatures (T_C), which can lead to higher Joule heating and energy loss in the devices. Here, we report that high-quality SRO thin films can be synthesized by controlling the plume dynamics and growth rate of pulsed laser epitaxy (PLE) with real-time optical spectroscopic monitoring. The SRO thin films grown under the kinetically controlled conditions, down to ca. 16 nm in thickness, exhibit both enhanced conductivity and T_C as compared to bulk values, due to their improved stoichiometry and a strain-mediated increase of the bandwidth of Ru 4d electrons. This result provides a direction for enhancing the physical properties of PLE-grown thin films and paves a way to improved device applications. *Published by AIP Publishing.* [<http://dx.doi.org/10.1063/1.4964882>]

Complex-oxide materials with intriguing properties have dawned a new era of so-called “all epitaxial functional devices” such as non-volatile memories,^{1–4} high speed switching devices,^{5–7} piezoelectric nano-generators,⁸ and ultraviolet lasers.⁹ The recent advances in epitaxial growth techniques for complex oxides enabled the investigations of high quality ultrathin films with thickness as thin as a few nanometers, yielding novel physical properties.¹⁰ Exploration of device applicability requires suitable metal electrodes that maintain such emergent physical properties. Normal metals and alloys such as Pt, Au, Ag, and Cu with high electrical conductivity lack interface adhesion and structural compatibility with complex oxides, which is essential for the fabrication of high performance devices. SrRuO₃ (SRO) is one of the most extensively studied and widely used metallic oxides.^{11,12} The perovskite structure yields SRO to have excellent chemical stability, which makes it an ideal electrode for oxide heterostructures.¹³ However, so far, SRO thin films have shown inferior metallic properties as compared to their bulk counterparts. SRO single crystals typically have a Curie temperature (T_C) around 160–165 K and a room temperature resistivity of $\sim 150 \mu\Omega\text{cm}$ (Refs. 14–17), yet no films have been synthesized that maintain these properties (see Table I). For example, the SRO thin films grown on GdScO₃ (GSO) substrates have a significantly higher resistivity of $\sim 650 \mu\Omega\text{cm}$ at room temperature and low T_C (100–130 K) (Refs. 18 and 19). The SRO thin films grown on SrTiO₃ (STO) substrates also exhibit a high room-temperature resistivity of $\sim 225 \mu\Omega\text{cm}$ and low T_C , which approaches 150 K

for thickness above ~ 25 nm (Refs. 20–23). Hence, it is essential to ask, “Are the transport properties of SRO thin films inherently inferior to SRO single crystals, or is there a way to enhance the metallic properties in the thin film limit?”

To answer these questions, we have investigated epitaxial thin films of SRO with various thicknesses (1–65 nm) grown on atomically flat GSO (110)_o substrates (see Fig. S1 of [supplementary material](#)), using pulsed laser epitaxy (PLE). The pseudo-cubic lattice parameters of SRO and GSO are 3.93 Å and 3.96 Å, respectively. Therefore, the SRO thin films grown on GSO substrates will experience in-plane tensile strain of +0.76% (see Fig. S2 of [supplementary material](#)). According to our first-principles calculations^{24,25} (see Fig. S3 of [supplementary material](#)), SRO thin films under tensile strain are expected to exhibit enhanced metallic properties as a result of the increase in the Ru 4d electron bandwidth and the average exchange energy (J_{avg}). Hence, we have chosen GSO substrates to study the effect of kinetically controlled growth on the physical properties of SRO thin films. We have prepared atomically flat GSO (110)_o substrates (from CrysTec GmbH) by annealing at 1000 °C in air for 1 hr.²⁶ The SRO thin films are deposited at 600 °C in an oxygen partial pressure of 100 mTorr, with a KrF excimer laser ($\lambda = 248$ nm) with a fluence of 1.6 J/cm² at 10 Hz using a ruthenium rich polycrystalline target.

In order to control the deposition rate, we have used a variety of laser spot sizes (0.16–0.41 mm²) by changing the aperture size in our laser optics. Note that the shape of the laser spot (square) was the same for all of the spot sizes. In general, a larger (smaller) laser spot size produces a larger (smaller) PLE plume; therefore, by changing the size of the

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TABLE I. Transport properties of SrRuO₃ thin films and single crystals.

	ρ (2 K) ($\mu\Omega\text{cm}$)	ρ (300 K) ($\mu\Omega\text{cm}$)	RRR	T_C (K)
This report, SRO/GSO (tensile strain), $t > 15$ nm	25–30	125–140	5–6	160–163
SRO single crystals ^{14–17}	1–15	150–200	20–192	160–165
SRO/STO (comp. strain) ^{20,21} (1–50 u.c.)	25–80	225–300	2–14	130–150
SRO/GSO (tensile strain) ¹⁸ (27–64 u.c.)	350–375	650–700	2–4	100–130

laser spot, we can effectively control the deposition rate (for technical details, see Ref. 27). Note that the deposition rates we have used are between 150 and 800 pulses/u.c. (0.027–0.005 nm/s), which are significantly slower than the typical deposition rates (10–50 pulses/u.c.) of conventional pulsed laser deposition. Figure 1 shows that the deposition rate can be controlled by keeping our growth parameters fixed and only changing the laser spot size. Using an isotropic slab model²⁸ with the complex dielectric functions of an SRO thin

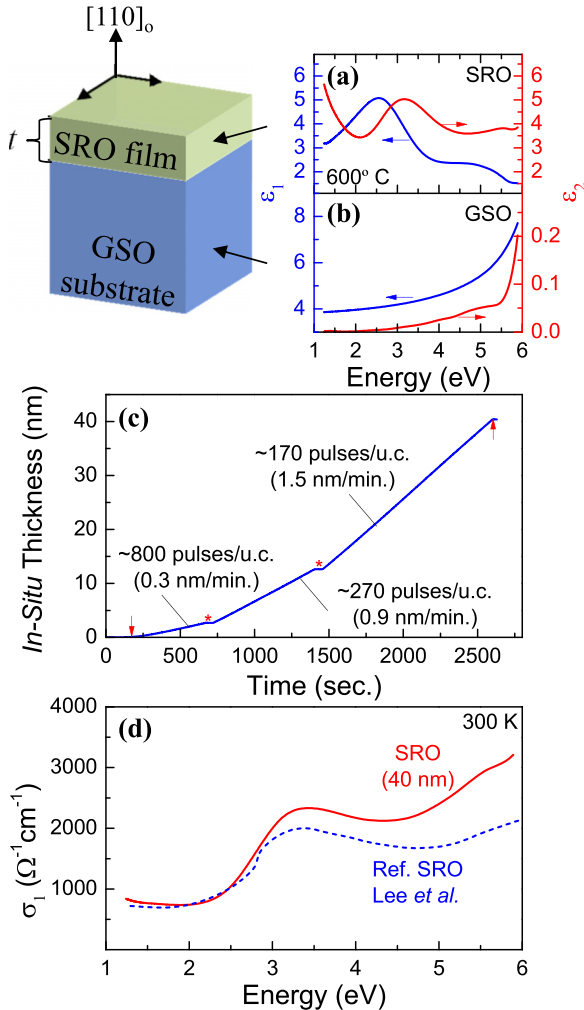


FIG. 1. Real-time monitoring of the thin film thickness via *in-situ* optical spectroscopic ellipsometry. Schematic of the sample geometry with the associated *in-situ* real (blue) and imaginary (red) dielectric functions as a function of photon energy for (a) an SRO thin film and (b) a GSO substrate, collected after and before deposition, respectively. (c) Real time thickness of an SRO thin film extracted from the *in-situ* spectroscopic ellipsometry data, using a single slab model during the growth. The red arrows indicate the start and stop points for deposition, and the red asterisks represent when the deposition is stopped to change the laser spot size. (d) Comparison of room-temperature optical conductivity spectrum of our SRO thin film to the data previously reported for SRO.²⁹

film and a GSO substrate, as shown in Figs. 1(a) and 1(b), respectively, the real-time thickness of SRO thin films was monitored using *in-situ* optical spectroscopic ellipsometry²⁸ and the deposition rate was determined for each laser spot size. Figure 1(c) shows the SRO *in-situ* film thickness as a function of time with the laser spot sizes of 0.16 mm², 0.30 mm², and 0.41 mm². The total thickness of the SRO thin films was also confirmed from the interference fringes in the X-ray diffraction (XRD) θ - 2θ scans. Figure 1(d) shows the room temperature optical conductivity spectrum of an SRO thin film, which is consistent with a reference spectrum.²⁹

SRO thin films display a change in crystal structure above 16 nm. Figure 2(a) shows the XRD θ - 2θ scans for our SRO thin films, which reveal the out-of-plane ($hh0$)_o reflections of the orthorhombic phase. The inset of Fig. 2(a) shows a rocking curve of a 16 nm-thick SRO thin film, which has a full width at half maximum (FWHM) value of 0.06°, indicating good crystallinity of the film. We performed the XRD reciprocal space mapping (RSM) around the GSO 620 and 260 reflections and defined Q_{\parallel} along the in-plane $[1-10]_o$ direction and along the out-of-plane $[110]_o$ direction (see Fig. S4 of [supplementary material](#)). For all of our films, the 620 and 260 film peaks are at the same Q_{\parallel} position as the

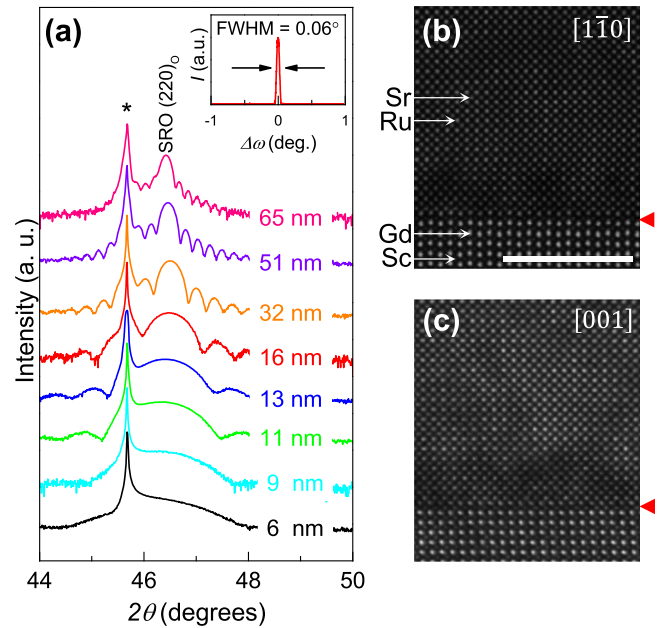


FIG. 2. X-ray diffraction patterns and cross-sectional high resolution scanning transmission electron microscopic images obtained for films of SrRuO₃ deposited on GdScO₃ (110) substrates. (a) Out-of-plane θ - 2θ XRD patterns for SRO films around the orthorhombic 220 peak of thickness ranging from 6 to 65 nm. The inset shows a typical rocking curve for all of the films in this thickness range. (b) and (c) High resolution scanning transmission electron microscopic images of cross-sections with the beam along the $[1-10]_o$ (b) and $[001]_o$ (c) directions. The size of the scale bar is 5 nm.

substrate, indicating they are coherently strained, whereas the position of the 260 peak moves to larger Q_{\perp} with a decreasing film thickness. Kan *et al.* (Ref. 18) have reported a similar result for their SRO films grown on GSO substrates. Our RSM data indicate a structural change from an orthorhombic (monoclinic) to a tetragonal phase, as the SRO thin film thickness is increased above 16 nm. Cross-sectional scanning transmission electron microscopic images of a 65 nm-thick SRO thin film are shown with the beam along the $[1-10]_o$ (Fig. 2(b)) and the $[001]_o$ (Fig. 2(c)) directions. A sharp interface (red arrows) is observed between the GSO substrate and SRO thin film, and there are no indications of misfit dislocations or defects.

The resistivity and T_C of thicker SRO thin films are similar to SRO bulk single crystals. The dc-transport behavior of SRO thin films is shown in Figure 3 as a function of temperature ($\rho(T)$). While a 1 nm thick film is insulating, the SRO thin films with increased thickness show a clear metallic behavior. Note that $\rho(T)$ is significantly reduced above ca. 16 nm and it is very similar to that of bulk crystals.¹⁷ Moreover, as summarized in Table I, the room temperature resistivity ($\rho(300\text{ K})$) of our SRO thin films is smaller than any previously reported values on SRO single crystals and thin films. The seemingly surprising improvement over the single crystal resistivity is in agreement with our first-principles calculations, which suggest an enhanced conductivity of SRO under tensile strain (see Fig. S3 of [supplementary material](#)). SRO is an itinerant ferromagnet described by the Ru 4d conduction bands using the Stoner model. The “kink” visible in the $\rho(T)$ data is due to the suppression of spin scattering as SRO transitions to a ferromagnetic state and represents the T_C of SRO thin films. The T_C is estimated

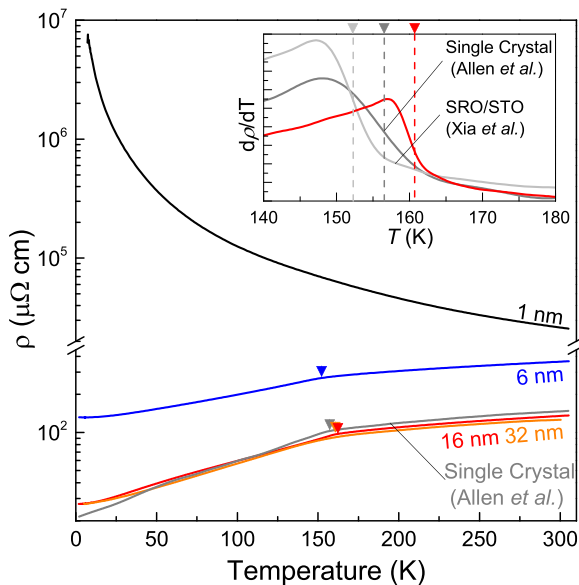


FIG. 3. *dc*-transport data for SrRuO₃ films of various thicknesses. Resistivity, $\rho(T)$, as a function of temperature for films of 1, 6, 16, and 32 nm in thickness. For comparison, we have also included the digitized data from our references for SRO single crystals (Ref. 17). The inset shows the derivative of the resistivity as a function of temperature for our 16 nm sample as well as the digitized data for a single crystal (Ref. 17) and an SRO/STO (compressive strain) film (Ref. 20). We can clearly see that the T_C and resistivity of our 16 nm sample are similar to the values reported for single crystals and significantly improved from SRO/STO.

by taking the first derivative of the resistivity ($d\rho/dT$), as shown in the inset of Figs. 3, S5, and S6 ([supplementary material](#)). The T_C values of our SRO thin films are close to (or higher than) the previously reported values of SRO single crystals (Refs. 14–16) and compressive-strained SRO thin films (Refs. 20–22).

The T_C values gradually increase as the thickness is increased, reaching a maximum ($\sim 163\text{ K}$) at 16 nm, and remains approximately constant above this thickness. Figure 4(a) shows the estimated T_C 's as a function of thickness. Note that these T_C values are significantly higher than previously reported T_C 's for SRO thin films under compressive (open triangles)²⁰ and tensile (open squares) strain,¹⁸ as shown in Fig. 4(a) for comparison. Although the room temperature resistivity and T_C of our films are enhanced from other SRO thin films, the residual resistivity ratio (RRR) remains low (see Table I and Fig. S7 ([supplementary material](#))). This is likely due to the appearance of a resistivity upturn at low temperatures ($<20\text{ K}$) in SRO thin films, which has been observed in previous studies, but its origin is not fully understood at this moment.

To investigate the discrepancy in the enhanced metallic properties observed in our SRO thin films compared to that of other thin film reports, we have performed a test growth of a 15 nm-thick SRO thin film with an increased laser fluence of $\sim 3.0\text{ J/cm}^2$. Note that the optimal SRO thin films are grown at $\sim 1.6\text{ J/cm}^2$. The XRD θ - 2θ scan of the test-grown SRO thin film shows a slight increase in the out-of-plane lattice parameter (decrease in 2θ values) compared to an optimal 16 nm-thick SRO thin film, as shown in Fig. 4(b), while the FWHM of the two thin films rocking curves are very similar, indicating that they both are homogeneous with comparable crystalline quality. Surprisingly, the $\rho(T)$ data show that the 15 nm-thick SRO thin film grown at the high fluence ($\sim 3.0\text{ J/cm}^2$) has roughly doubled the resistivity, as shown in Fig. 4(c), and a significantly lower T_C ($\sim 133\text{ K}$) than those from the 16 nm-thick SRO thin film.

It is well known that Ru vacancies can be formed in SRO thin films due to the volatile nature of the Ru atom. According to Dabrowski *et al.*,³⁰ the Ru vacancies have a profound impact on the metallic and magnetic properties of SRO thin films, i.e., decreasing T_C down to 86 K for only a 6% change in the Ru site occupancy. Therefore, we consider that the laser fluence in PLE may affect the stoichiometry of SRO thin films. Unfortunately, because the change in stoichiometry is so small, microscopic characterization measurements are not capable of resolving the differences in our films. However, the transport and magnetic properties of SRO thin films can be significantly affected as shown above. It is also known that structural distortions and symmetry mismatch across interface boundaries and other interfacial effects can deteriorate the electrical properties of complex-oxide thin films and heterostructures.^{31–33} However, in our SRO thin films grown on GSO substrates, the laser beam spot size and, hence, the film deposition rate are shown to have a more significant role in their transport properties than the interfacial contributions (see Fig. S7 of [supplementary material](#)).

In order to verify the effects of the kinetically controlled growth rate on the properties of SRO thin films, we have

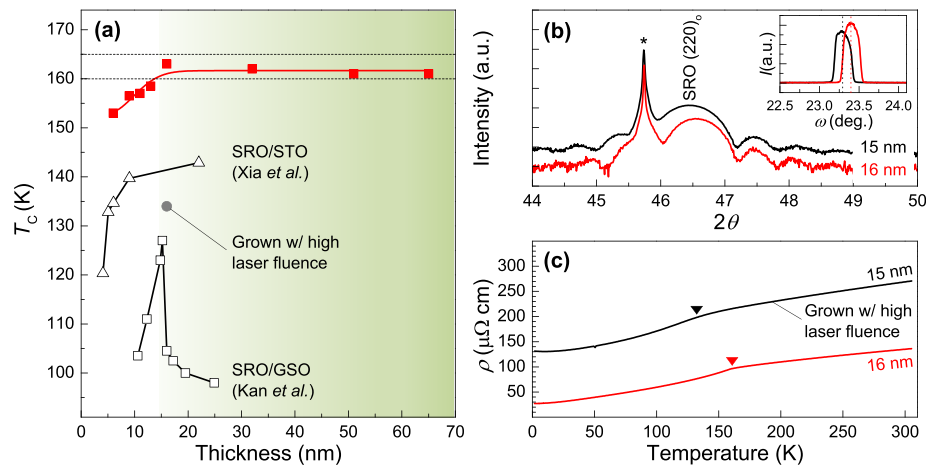


FIG. 4. Effects of moderated deposition rate on the Curie temperature (T_C) and physical properties. (a) The Curie temperature versus the film thickness is shown as red filled circles. The open triangles and squares represent previously reported values for SRO/STO (Ref. 20) and SRO/GSO (Ref. 18) films (compressive and tensile strain), respectively. The T_C reaches a maximum of ~ 163 K at 16 nm and remains essentially constant above this thickness. This value is significantly higher than the previous reports for SRO films under compressive or tensile strain and is even comparable to the values for single crystals. The green shaded area highlights the thickness region where the conductivity and T_C become maximum and comparable to the single crystals (black dashed lines). (b) Comparison of the XRD θ - 2θ data of two films grown at 350 pulses/u.c. but with different laser fluence. The red curve shows the 16 nm film (~ 1.6 J/cm²) and the black curve shows the 15 nm film (~ 3 J/cm²). (c) Comparison of the $\rho(T)$ data for the same two samples shown in (b).

grown two samples of similar thickness but with two different laser spot sizes. A 5 nm-thick SRO thin film was grown using the smallest laser spot size (0.16 mm²) with an extremely slow growth rate (~ 800 pulses/u.c.). This film is compared to a 6 nm-thick film that was grown using a laser spot size of 0.35 mm² resulting in ~ 270 pulses/u.c. Our dc-transport measurements show that the resistivity of the 5 nm film is reduced compared to the 6 nm-thick film (Fig. S8 of supplementary material). This is surprising because for our films grown with the same laser spot size (0.35 mm²), the resistivity increases as the film thickness decreases. Hence, the moderated PLE techniques, i.e., controlled growth rate and laser spot size, are the key for the enhanced transport properties of SRO thin films. Lee *et al.*²⁷ has shown that the laser spot size plays a significant role in the oxygen stoichiometry of STO thin films fabricated with PLE, and our result shows that the laser spot size can also have an impact on the cation stoichiometry (i.e., Ru ions). Although it is an arduous task to precisely measure the stoichiometric ratio of thin films directly, by comparing the data from the samples intentionally grown with the Ru or O vacancies,^{30,34} we have shown that the kinetically controlled (laser spot size) PLE deposition is a fundamental component of fabricating the highest quality complex oxide films and heterostructures.

This work dismisses the notion that SRO thin films are inherently inferior to single crystals, which has become a generally accepted problem in the solid state community. We have discovered that it is possible to achieve SRO thin films with metallic properties similar to SRO single crystals by adjusting the laser spot size and effectively reducing the laser plume kinetic energy, thereby improving the overall stoichiometry of the films. Our results show that by controlling the laser spot size (laser plume energy), it is possible to fabricate epitaxial thin film electrodes for functional oxide devices, which do not hinder the functionality of the device as a result of the degraded metallic properties.

See [supplementary material](#) for the results of atomic force microscopy, first-principles calculations, XRD RSM, and additional transport measurements.

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