

Highway Materials Research Laboratory
132 Graham Avenue, Lexington 29, Ky.
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Memo. to Dean Terrell,
Director of Research

At the meeting of the Research Board on March 24, we presented our first Progress Report on Project C-22, entitled "A Study of the Properties of Coarse Aggregates", and at that time an oral discussion was given by Mr. James L. Young, Jr., Assistant Research Engineer, in the capacity of geologist for our laboratory. Work on this project since March 24 has made possible a second Report of Progress on this study, a copy of which is attached.

You will recall that the original plan for this investigation included very extensive tests for chemical and mineralogical properties for limestone from four widely separated and different quarries. These were sampled at one foot intervals and during inspection of the laboratory last November 25, Mr. Young gave an account of his sampling, the laboratory techniques developed to that date, and the objectives of the research so far as they could be formulated at that time. It is gratifying to note that at the present stage a very large majority of chemical and mineralogical properties have been determined for all four of the quarries, and many samples have been prepared for the purpose of correlating these properties of the limestone with performance in concrete and bituminous mixes.

In line with our policy heretofore Mr. Young and Mr. James H. Havens, Jr. Materials Engineer and Chemist in our laboratory, will give an oral presentation of this voluminous report in order to condense and interpret the data.

Respectfully submitted


L. E. Gregg
Associate Director of Research

LG:K

cc: Members of the
Research Board

Commonwealth of Kentucky
Department of Highways

Progress Report No. 2
on
Research Project C-22

A STUDY OF THE PROPERTIES OF COARSE AGGREGATES

by

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Assistant Research Engineer

and

James H. Havens
Junior Materials Engineer

Highway Materials Research Laboratory
Lexington, Kentucky

October, 1947

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INTRODUCTION

Research Project C-22 is an investigation of the geologic aspects of limestone aggregate that might cause failure if used in highway construction, as reported earlier. As the study was to be approached from a new standpoint, it was allowed to mature along significant lines. At present a working plan has been adopted and is shown here in some detail.

Four quarries were selected on the basis of performance of the aggregates in pavements. The four selected, together with their rating, were:

- (Fair) Quarry No. 1 - Composed of 31 feet of Ordovician limestone. Location - Lexington.
- (Good) Quarry No. 2 - Composed of 134 feet of Mississippian limestone. Location - Mt. Vernon.
- (Poor) Quarry No. 3 - Composed of 65 feet of Mississippian limestone. Location - Olive Hill.
- (Excellent) Quarry No. 4 - Composed of 35 feet of Silurian dolomite. Location - Avoca.

These quarries have been collected and tested in detail until a very sound idea of all their physical properties, their variations, structure, mineralogy, etc. is known. By results of present tests it is possible to locate a random sample very closely in the quarry and in most cases orient it.

As the evaluations placed on these quarries can at best be only general, physical tests were selected with which to evaluate the aggregate closer and thereby to correlate an aggregate's performance accurately with the results of significant tests performed on it. First, it is necessary to prove that one property or a series of properties combined cause the failure.

The quarries are collected a second time with a new purpose. Four to eight levels have been selected at which point one or more properties is outstanding: A large sample is selected 2' x 3½' x 1'. From it, six beams are sawed on a carborundum masonry saw, and the remaining rock is crushed to furnish the aggregate for other physical tests and approximately six concrete beams. These beams are to be subjected to the following physical and chemical tests to determine performance and to correlate it with physical properties measured by old tests as well as by those that are new.

1. Insoluble Residue
2. Permeability
3. Porosity
4. Bulk Specific Gravity
5. Absolute Specific Gravity
6. Thin Section
7. Size, Shape and Frequency of Pores
8. Clay Minerals
9. Los Angeles Abrasion
10. Sodium Sulfate Soundness
11. Sonic Modulus of Elasticity (rock beam)
12. Modulus of Rupture (rock beam)
13. Durability Index (rock beam)
14. Sonic Modulus of Elasticity (concrete beam)
15. Modulus of Rupture (concrete beam)
16. Durability Index (concrete beam)
17. Coefficient of Thermal Expansion.
18. Toughness
19. Hardness
20. Ductility by Compression Tests

A sample form is included on the next page to show what control may be expected in dealing with this size beam.

Commonwealth of Kentucky
Department of Highways

Highway Materials Research Laboratory

REPORT OF IMPROMPTU TESTS

File: R-2

Date: 8-11-47

Subject: Variations in Insoluble Residue over a Beam
2 inches x 3 inches x 12 inches

Location Sampled: Central Rock Surface

Date Sampled: July 26 Sampled By: T. Wolfe

Reason for Sampling (related work, etc.): To determine what
control may be expected in testing beams in Research Project
C-22

Results: Samples A - 1.16% (22A)

B - 1.20% (22B)

C - 1.20% (22C)

D - .90% (22D)

(All from one beam at 22 ft. level.)

Remarks: Samples taken from four areas in longest dimension
of beam after it was broken. Maximum deviation is 0.3%.

Reported By: J. Young & J. Havens

Copies: Gregg
Havens
Young
File

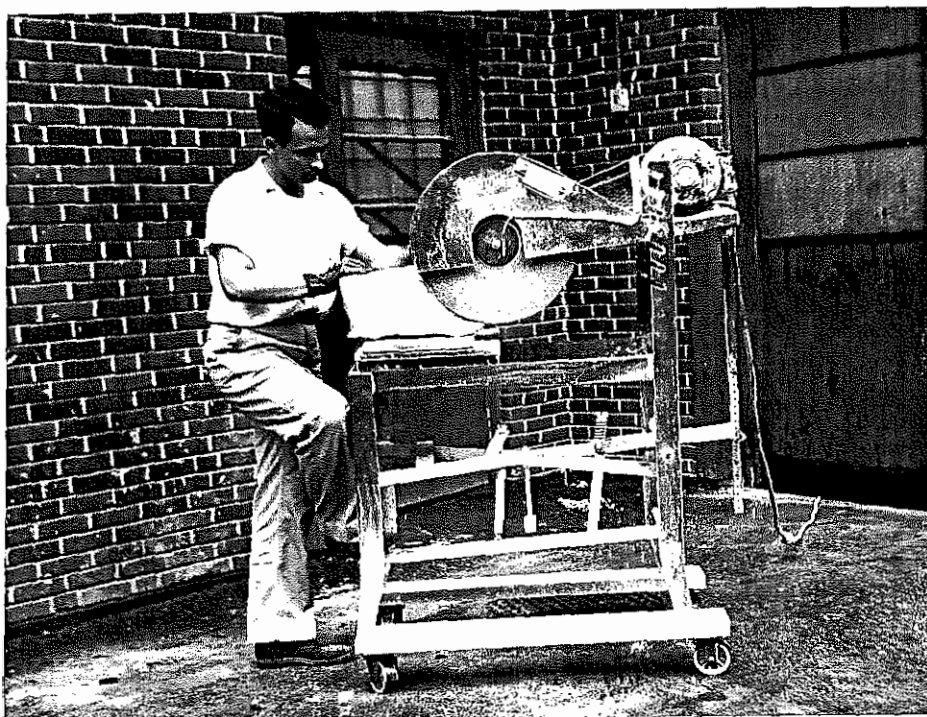


Figure 1. Beams of Limestone Being Sawed
on a Carborundum Masonry Saw

RESULTS

At the time of the last Research Board Meeting, tests and results were partially completed, and reported as such. At this time tests are more fully completed and will be reported with as much detail toward results as the progress allows.

Insoluble Residue

That portion of a limestone which refuses to go into solution upon digestion in a fairly active acid is termed the insoluble residue. It has been postulated that the insoluble residue should be a proper criterion in diagnosing the behavior of an aggregate. This seems to be true, and is the only test found to date that is discriminating.

Digestion is carried on at room temperature with one exception. Quarry No. 4 is dolomitic and the magnesium carbonate is more stable than calcium carbonate. In order to get the magnesium carbonate into complete solution, heating to just below boiling is necessary.

Formation of insoluble chlorides is a detrimental phenomena but is considered to be a negligible error. Silicates go into solution to such a small extent in hydrochloric acid that they are not considered.

Quarry Averages are the only criterion available for evaluation of a quarry as a whole and the averages place the quarries nearly in line with reference to the initial general performance ratings given to these quarries.

INSOLUBLE RESIDUE

QUARRY NO. 1, LEXINGTON, KY.

<u>Level</u>	<u>Insol. Res.</u>
0	10.5%
1	8.5%
2	6.5%
3	10.0%
4	10.0%
5	8.0%
6	3.0%
7	1.0%
8	1.0%
9	3.0%
10	4.0%
11	5.0%
12	2.5%
13	2.0%
14	4.5%
15	3.5%
16	4.0%
17	6.5%
18	6.0%
19	5.0%
20	2.5%
21	2.5%
22	6.5%
23	4.5%
24	2.5%
25	8.5%
26	5.5%
27	6.5%
28	6.5%
29	3.5%
30	6.0%
Quarry Average -	5.145%

INSOLUBLE RESIDUE

QUARRY NO. 2, MOUNT VERNON, KY.

<u>Level</u>	<u>Insol. Res.</u>	<u>Level</u>	<u>Insol. Res.</u>
1	1.5%	51	2.0%
2	4.0%	52	2.5%
3	5.0%	53	2.5%
4	7.5%	54	2.5%
5	4.5%	55	4.5%
6	3.5%	56	5.0%
7	2.5%	57	2.5%
8	0.5%	58	6.5%
9	2.0%	59	2.5%
10	1.0%	60	5.0%
11	1.0%	61	1.0%
12	0.5%	62	8.5%
13	1.0%	63	4.0%
14	1.0%	64	2.0%
15	1.0%	65	1.0%
16	3.0%	66	1.0%
17	6.0%	67	1.0%
18	8.0%	68	1.5%
19	7.0%	69	2.0%
20	6.5%	70	2.0%
21	4.0%	71	2.0%
22	8.0%	72	4.0%
23	3.0%	73	Covered
24	3.0%	74	2.5%
25	1.5%	75	1.0%
26	3.0%	76	2.0%
27	4.0%	77	1.0%
28	2.5%	78	0.5%
29	6.5%	79	1.0%
30	3.5%	80	3.0%
31	2.5%	81	2.5%
32	3.5%	82	3.0%
33	17.5%	83	6.0%
34	Shale	84	10.5%
35	6.5%	85	12.0%
36	5.5%	86	10.0%
37	4.5%	87	5.5%
38	3.0%	88	3.0%
39	3.5%	89	5.5%
40	2.5%	90	7.0%
41	3.0%	91	4.5%
42	3.0%	92	5.0%
43	5.0%	93	3.5%
44	3.5%	94	4.5%
45	5.5%	95	7.0%
46	6.5%	96	8.0%
47	10.0%	97	3.0%
48	5.5%	98	3.5%
49	7.5%	99	3.5%
50	8.0%	100	3.0%

QUARRY NO. 2 (Continued)

<u>Level</u>	<u>Insol. Res.</u>
101	5.0%
102	3.0%
103	2.5%
104	2.0%
105	3.5%
106	3.5%
107	1.5%
108	2.0%
109	1.0%
110	3.0%
111	Covered
112	5.5%
113	2.5%
114	1.5%
115	2.5%
116	1.5%
117	2.0%
118	10.0%
119	3.0%
120	3.5%
121	2.5%
122	4.0%
123	3.0%
124	2.5%
125	2.5%
126	5.0%
127	3.0%
128	4.0%
129	2.0%
130	5.5%
131	6.0%
132	Covered
133	7.0%
134	4.0%
135	5.5%
Quarry Average	3.88%

INSOLUBLE RESIDUE

QUARRY NO. 3, OLIVE HILL, KY.

<u>Level</u>	<u>Insol. Res.</u>	<u>Level</u>	<u>Insol. Res.</u>
1	12.5%	34	26.5%
2	9.5%	35	25.5%
3	10.5%	36	16.0%
4	12.5%	37	11.5%
5	14.5%	38	8.5%
6	9.0%	39	8.0%
7	4.5%	40	7.0%
8	20.0%	41	10.0%
9	15.0%	42	4.5%
10	11.0%	43	17.5%
11	13.0%	44	19.5%
12	19.5%	45	20.0%
13	12.0%	46	16.0%
14	8.5%	47	17.0%
15	9.0%	48	20.5%
16	14.5%	49	11.5%
17	7.5%	50	14.0%
18	11.5%	51	9.5%
19	59.5%	52	11.0%
20	28.0%	53	4.0%
21	Shale	54	10.5%
22	14.0%	55	6.5%
23	20.0%	56	4.0%
24	13.0%	57	8.0%
25	4.5%	58	4.5%
26	6.0%	59	4.0%
27	11.5%	60	5.0%
28	14.0%	61	3.0%
29	13.5%	62	1.5%
30	11.0%	63	2.0%
31	6.0%	64	2.5%
32	7.5%	65	1.5%
33	8.5%	Quarry Average	-11.9%

INSOLUBLE RESIDUE

QUARRY NO. 4, AVOCA, KY.

<u>Level</u>	<u>Insol. Res.</u>
1	8.26%
2	4.50%
3	3.73%
4	10.38%
5	3.25%
6	1.91%
7	3.15%
8	1.37%
9	1.05%
10	1.13%
11	1.31%
12	1.35%
13	3.04%
14	1.43%
15	1.97%
16	2.75%
17	2.59%
18	2.39%
19	1.32%
20	1.71%
21	--
22	5.32%
23	5.54%
24	3.49%
25	7.05%
26	7.66%
27	7.52%
28	6.12%
29	6.51%
30	7.00%
31	5.78%
32	5.77%
33	6.97%
Quarry Average	4.166%

Permeability

The permeability of a solid may be defined as the rate of flow of a given fluid through a unit cross section of the solid under a unit pressure gradient and conditions of viscous flow; permeability is therefore a measure of fluid conductivity of a solid. The standard unit of permeability as defined by the American Petroleum Institute (tentatively) is the "darcy", which is the rate of flow in milliliters per second, of a fluid of one centipoise viscosity through a cross section of one square centimeter of a porous medium, under a pressure gradient of one atmosphere (76.0 cm. Hg.) per centimeter and conditions of viscous flow.

Permeable mediums are necessarily porous but porous need not be permeable. The connections of the pore spaces included in a medium determine its permeability. Limestone has a very low permeability, as a rule, when compared to a reservoir rock such as an oil sand. However, small as the openings are, a limestone is usually quarried out fully saturated with water. This means then in the role of an aggregate in concrete pavements, the voids may be utilized in any exchange of water taking place. The total amount of hygroscopic water present in most cases will not exceed one per cent, but the influence of this small per cent cannot be ignored. If the size openings fall within the capillary limits, an unestimable force is brought into action. Such might be the case with Quarry No. 3 which had very low permeability as measured by the method adopted.

Specimens were cored out with a diamond studded core drill.

core used was 1.7 centimeters in diameter by 2.7 centimeters length, presenting a 2.27 square centimeter face to the testing ium.

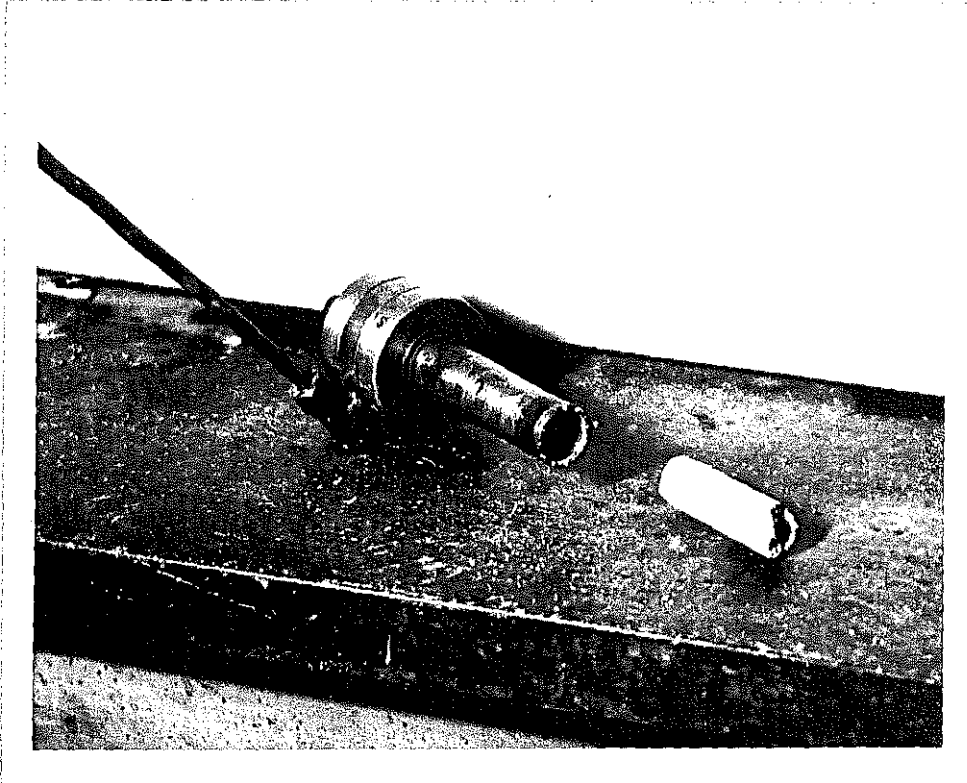


Fig. 2. Diamond Core Drill and Limestone Core

The sample was mounted in a simple adaptation of equipment described by Wycoff, Botset, Muscat and Reed in the February, 1934 bulletin of the American Association of Petroleum Geologists. The medium used was compressed, dried air. The sample was mounted in a rubber gasket, subjected to a known pressure and temperature and the air passing through the specimen was trapped in a calibrated glass tube. The computations are in "millidarcies" or one one thousandth of a "darcy". Two directions were tested from each level. The formula is shown:

$$= \frac{2000 \times l. \times \text{viscosity of air} \times \text{vol. of air}}{\text{area in sq. cm.} \times \text{difference in pressure}} = \text{Millidarcies}$$

RESULTS OF PERMEABILITY IN MILLIDARCYS
 QUARRY NO. 1, CENTRAL ROCK COMPANY, LEXINGTON, KY.

<u>Level</u>	<u>Position of Sample</u>	
	<u>Vert.</u>	<u>Horiz.</u>
0	.741	0.903
1	.463	1.160
2	2.000	.879
3	1.493	1.864
4	.625	.786
5	.417	2.512
6	1.094	1.967
7	3.422	1.059
8	1.700	.862
9	1.303	.827
10	.897	.485
11	.489	1.094
12	.256	.583
13	.384	.886
14	.373	.233
15	.419	.360
16	.129	.210
17	.484	.922
18	.265	.231
19	.553	.311
20	.507	.138
21	.196	.276
22	.553	.576
23	.369	.829
24	.150	.300
25	.289	.553
26	.174	.510
27	.417	.753
28	--	.382
29	.278	.406
30	.510	.463
Quarry Average -	.6983	.7399

RESULTS OF PERMEABILITY IN MILLIDARCYS

QUARRY NO. 2, MOUNT VERNON, KY.

Level	Position of Sample		Level	Position of Sample	
	Vert.	Hor.		Vert.	Hor.
1	0.000	0.000	51	0.000	0.000
2	0.000	0.000	52	0.000	0.000
3	0.000	0.000	53	0.000	0.000
4	0.000	0.000	54	--	--
5	0.000	0.000	55	0.000	0.000
6	0.000	0.000	56	0.000	0.000
7	0.000	--	57	0.000	--
8	0.000	0.000	58	0.000	0.000
9	0.593	0.000	59	0.000	0.000
10	66.480	21.928	60	0.000	0.000
11	1.941	9.079	61	--	--
12	293.300	188.550	62	--	--
13	0.265	0.000	63	0.139	0.007
14	0.000	0.000	64	--	--
15	0.000	0.000	65	0.056	0.111
16	0.000	0.000	66	0.042	0.011
17	0.000	0.000	67	0.000	0.000
18	0.000	0.000	68	0.000	0.000
19	--	--	69	0.000	0.005
20	0.000	0.000	70	0.000	0.000
21	0.000	0.000	71	0.000	0.000
22	0.000	0.000	72	0.000	0.000
23	0.000	0.000	73	0.000	0.000
24	0.000	0.000	74	0.000	0.000
25	0.000	0.000	75	0.000	0.000
26	0.000	0.000	76	0.000	0.000
27	0.000	0.000	77	0.000	0.000
28	0.000	0.000	78	0.000	0.000
29	0.000	0.000	79	0.000	0.000
30	0.000	0.000	80	0.000	0.000
31	0.000	0.000	81	0.000	0.000
32	0.000	0.000	82	--	--
33	0.000	0.000	83	--	--
34	0.000	0.000	84	0.000	0.000
35	0.000	0.000	85	0.111	0.028
36	0.000	0.000	86	0.000	0.000
37	0.000	0.000	87	--	--
38	0.000	0.000	88	0.000	0.000
39	0.000	0.000	89	0.000	0.000
40	0.000	0.000	90	0.000	0.000
41	0.000	0.000	91	0.000	0.000
42	0.000	0.000	92	0.000	0.000
43	0.000	0.000	93	0.000	0.000
44	0.000	0.000	94	0.000	0.000
45	0.000	0.000	95	0.154	0.028
46	0.000	0.000	96	--	--
47	0.000	0.000	97	0.000	0.000
48	0.000	0.000	98	0.000	0.000
49	0.000	0.000	99	0.000	0.000
50	0.000	0.000	100	0.098	0.000

QUARRY NO. 2 (Continued)

<u>Level</u>	<u>Position of Sample</u>	
	<u>Vert.</u>	<u>Horiz.</u>
101	0.000	0.000
102	0.000	0.000
103	0.000	0.000
104	0.000	0.000
105	--	--
106	0.391	0.000
107	0.000	0.000
108	0.000	0.000
109	0.000	0.000
110	0.000	0.000
111	0.000	0.000
112	0.000	0.000
113	0.139	0.000
114	0.000	0.000
115	0.000	0.000
116	0.000	0.000
117	0.000	0.000
118	0.000	0.000
119	0.000	0.000
120	0.000	0.000
121	0.000	0.000
122	0.000	0.000
123	0.000	0.000
124	0.000	0.000
125	0.000	0.000
126	0.000	0.000
127	0.000	0.000
128	0.000	0.000
129	0.000	0.000
130	0.000	0.000
131	0.000	0.000
132	--	--
133	0.000	0.000
134	0.000	0.000
135	0.000	0.000
Quarry Average -	5.272	1.773

RESULTS OF PERMEABILITY IN MILLIDARCYS

QUARRY NO. 3, OLIVE HILL, KY.

Level	Position of Sample		Level	Position of Sample	
	Vert.	Hor.		Vert.	Hor.
1	0.000	0.000	34	0.000	0.000
2	0.000	0.000	35	0.000	0.000
3	0.000	0.000	36	0.000	0.000
4	0.000	0.000	37	0.000	0.000
5	0.000	0.000	38	0.000	0.000
6	0.000	0.000	39	0.000	0.000
7	0.000	0.000	40	0.000	0.000
8	0.000	0.000	41	0.000	0.000
9	0.000	0.000	42	0.000	0.000
10	0.000	0.000	43	0.000	0.000
11	0.000	0.000	44	0.000	0.000
12	0.000	0.000	45	0.000	0.000
13	0.000	0.000	46	0.000	0.000
14	0.000	0.000	47	0.000	0.000
15	0.000	0.000	48	0.000	0.047
16	0.000	0.000	49	0.000	0.000
17	0.140	0.000	50	0.000	0.047
18	0.000	0.114	51	0.000	0.000
19	0.081	0.000	52	0.000	0.000
20	0.000	0.000	53	0.000	0.000
21	SEAM	--	54	0.000	0.000
22	0.292	0.760	55	0.000	0.000
23	0.421	0.632	56	0.000	0.000
24	1.100	1.100	57	0.000	0.000
25	0.000	--	58	0.081	0.000
26	0.000	0.000	59	0.000	0.000
27	0.529	0.682	60	0.000	0.000
28	1.165	1.176	61	0.046	0.000
29	1.117	1.741	62	0.000	0.000
30	0.988	1.024	63	0.000	0.000
31	0.400	0.882	64	0.000	0.000
32	0.717	1.176	65	0.046	0.000
33	0.894	1.211	Quarry Average	0.1233	0.1630

RESULTS OF PERMEABILITY IN MILLIDARCYS

QUARRY NO. 4, AVOCA, KY.

Level	Position of Sample	
	Vert.	Hor.
0	0.386	0.327
1	0.397	0.410
2	0.222	0.152
3	0.281	0.468
4	0.937	.117
5	.235	.046
6	11.940	47.750
7	.316	.246
8	.117	.000
9	.000	.000
10	.000	.000
11	.282	.671
12	.047	.318
13	.483	.165
14	1.810	.282
15	.000	.000
16	.118	.130
17	.000	.095
18	.163	.046
19	.200	.163
20	.262	14.921
21	.338	5.670
22	3.043	28.297
23	.081	.081
24	.211	.163
25	.000	.046
26	.000	.000
27	.000	.000
28	.164	.000
29	.046	.000
30	.095	.000
31	.000	.000
32	.000	.000
33	2.824	1.024
Quarry Average	0.735	3.017

Hygroscopic Water and Carbon Dioxide

An experiment was run to determine if water content or CO₂ content is indicative enough to be characteristic of different phases of sedimentation.

If true, this would be a good criterion for establishing zones differing in sedimentation in any geologic column of calcareous rocks.

Procedure- (H₂O content) Six capsules are heated to 725° C. in a muffle furnace or to constant weight. They are placed in an oven and heated @ 100 - 105° C. for four hours or more. They are placed in a dessicator and allowed to cool to room temperature at which point they are weighed. About 2 grams of powdered limestone (passing 150 mesh) are placed in the capsules. They are weighed, put in an oven @ 100 - 105° C. and left for 12-14 hours. They are placed in a dessicator and allowed to cool to room temperature at which point they are weighed. Loss of weight is a measure of the water content of the rock and will be expressed as a percentage of the total weight of the sample. (CO₂ and organic matter) The samples are now placed in the muffle furnace, taken to 750° C. and left for four hours. At the conclusion of this time they are placed in a dessicator and brought to room temperature. The loss of weight is a direct measure of the CO₂ and organic matter combined.

For this experiment arbitrary zones assigned in Quarry No. 1 were utilized and samples taken from the material used in insoluble residue determination. Samples 1-5 are from the corresponding zone number. Sample 6 is taken from the 18 foot level. Quarry No. 3.

Weights are indicated as follows:

A. Wt. of Crucibles at Constant Weight

No. 1 - 12.0314 gms.
 No. 2 - 12.2219 "
 No. 3 - 12.6611 "
 No. 4 - 12.1040 "
 No. 5 - 11.9056 "
 No. 6 - 12.1969 "

B. Wt. of Crucibles at Constant Weight + Wt. of Crushed Ls. at Room Temp. & Humidity

2.1540 No. 1 - 14.1854 (Zone 1-Q-1)
 1.7429 No. 2 - 13.9648 (Zone 2-Q-1)
 1.8991 No. 3 - 14.5602 (Zone 3-Q-1)
 1.7297 No. 4 - 13.8337 (Zone 4-Q-1)
 1.9136 No. 5 - 13.8192 (Zone 5-Q-1)
 2.5462 No. 6 - 14.7431 (18 ft. Olive Hill)

C. Wt. of Crucible + Sample after 14 hrs. at 100°C.

No. 1 - 14.1830 - .1114%
 No. 2 - 13.9627 - .1205%
 No. 3 - 14.5587 - .079%
 No. 4 - 13.8341 - ?
 No. 5 - 13.8199 - ?
 No. 6 - 14.7412 - .0746%

D. Wt. of Crucible + Sample after 750°C. decomp.

No. 1 - 13.3222 gms.
 No. 2 - 13.2877 "
 No. 3 - 13.7866 "
 No. 4 - 13.1514 "
 No. 5 - 13.0431 "
 No. 6 - 13.7487 "

This indicates -

No. 1 - 39.96%	CO ₂	+	Org.	Mat.
No. 2 - 38.72%	"	"	"	"
No. 3 - 40.65%	"	"	"	"
No. 4 - 39.65%	"	"	"	"
No. 5 - 40.59%	"	"	"	"
No. 6 - 38.97%	"	"	"	"

With intent to check results on the first determination the same samples were run again under similar conditions. Results were as follows:

- A. No. 1 - 13.5961 gms.
 No. 2 - 19.3577 "
 No. 3 - 17.3164 "
 No. 4 - 15.5076 "
 No. 5 - 16.2475 "
 No. 6 - 19.0676 "

		<u>Wt. of Ls. Sample</u>	
B.	No. 1 - 16.4499 gms.	-	2.8538 gms.
	No. 2 - 21.8909 "	-	2.5332 "
	No. 3 - 20.1121 "	-	2.7957 "
	No. 4 - 18.7636 "	-	3.2560 "
	No. 5 - 19.4851 "	-	3.2376 "
	No. 6 - 22.3581 "	-	3.2905 "

		<u>Wt. of Water</u>	
C.	No. 1 - 16.4447 gms.	-	.0052 gms.
	No. 2 - 21.8838 "	-	.0071 "
	No. 3 - 20.1040 "	-	.0081 "
	No. 4 - 18.7596 "	-	.0040 "
	No. 5 - 19.4825 "	-	.0026 "
	No. 6 - 22.3536 "	-	.0045 "

In terms of percentage this indicates -

- No. 1 - .182% of water
 No. 2 - .280% "
 No. 3 - .289% "
 No. 4 - .122% "
 No. 5 - .080% "
 No. 6 - .136% "

- D. No. 1 - 15.3030 gms.
 No. 2 - 20.8976 "
 No. 3 - 18.9708 "
 No. 4 - 17.4890 "
 No. 5 - 18.1571 "
 No. 6 - 21.0636 "

This indicates -

- No. 1 - 40.00% of CO₂ and Organic Matter
 No. 2 - 39.45% " " " " "
 No. 3 - 40.55% " " " " "
 No. 4 - 39.05% " " " " "
 No. 5 - 40.96% " " " " "
 No. 6 - 39.20% " " " " "

Explaining any difference between the first results and the second, the second set were "burned" at 750°C. for at least two hours longer than Group I, therefore the reaction did not go to completion in the first set. A minimum of 6 hours must now be recommended in the muffle furnace. While not conclusive, the results show no positive value for correctly delineating one zone of sedimentation from another by the amount of CO₂ and organic matter. The water content of the samples seems to be a good criterion but must be checked over the vertical extent of a quarry in order to determine its accuracy.

Quarry No. 1 was selected for study, due to its slight vertical extent and accessibility. It contains two different geologic formations and several horizons predominating in silicious replacement.

Samples were run in duplicate to check results. Some few did not check and were run again in duplicate until a check was assured. Results listed both in table form and graph form are submitted.

TABLE 3. PER CENT H₂O (HYGROSCOPIC) FOR QUARRY NO. 1

Level	Percentage		Mean	Level	Percentage		Mean
	H ₂ O				H ₂ O		
0a -	.0429)	.0426	16a -	.0919)	.0903
0b -	.0423)		16b -	.0887)	
1a -	.1432)	.1366	17a -	.0505)	.0610
1b -	.1301)		17b -	.0716)	
2a -	.0497)	.0492	18a -	.0397)	.0436
2b -	.0487)		18b -	.0475)	
3a -	.1325)	.1307	19a -	.1998)	.1978
3b -	.1290)		19b -	.1959)	
4a -	.0486)	.0489	20a -	.1018)	.0979
4b -	.0493)		20b -	.0941)	
5a -	.0755)	.0675	21a -	.1127)	.1127
5b -	.0596)		21b -	.1127)	
6a -	.0128)	.0162	22a -	.0767)	.0756
6b -	.0196)		22b -	.0745)	
7a -	.1695)	.1589	23a -	.1859)	.1763
7b -	.1484)		23b -	.1767)	
8a -	.0617)	.0601	24a -	.0122)	.0130
8b -	.0585)		24b -	.0139)	
9a -	.1441)	.1433	25a -	.0336)	.0364
9b -	.1425)		25b -	.0343)	
10a -	.1946)	.1933	26a -	.0698)	.0659
10b -	.1920)		26b -	.0621)	
11a -	.1115)	.1093	27a -	.0469)	.0452
11b -	.1072)		27b -	.0435)	
12a -	.1378)	.1324	28a -	.0985)	.0912
12b -	.1270)		28b -	.0839)	
13a -	.0096)	.0089	29a -	.0839)	.0872
13b -	.0083)		29b -	.0895)	
14a -	.0379)	.0382	30a -	.0505)	.0564
14b -	.0386)		30b -	.0663)	
15a -	missing						
15b -	"						

TABLE 4. PER CENT ORGANIC CONTENT (CO₂) FOR QUARRY NO. 1

<u>Level</u>	<u>Per Cent</u>	<u>Mean</u>	<u>Level</u>	<u>Per Cent</u>	<u>Mean</u>
0a - 41.55)			16a - 40.87)		
0b - 41.43)		41.49	16b - 40.92)		40.89
1a - 41.85)			17a - 42.05)		
1b - 41.81)		41.83	17b - 42.00)		42.02
2a - 41.80)			18a - 41.90)		
2b - 41.80)		41.80	18b - 42.28)		42.09
3a - -)			19a - 38.51)		
3b - -)		41.55	19b - 38.63)		38.57
4a - 41.70)			20a - 39.20)		
4b - 41.73)		41.72	20b - 39.30)		39.25
5a - 40.89)			21a - 40.14)		
5b - 40.89)		40.89	21b - 40.18)		40.16
6a - 41.67)			22a - 40.48)		
6b - 41.56)		41.62	22b - 40.49)		40.48
7a - 41.21)			23a - 41.13)		
7b - 41.16)		41.19	23b - 41.11)		41.14
8a - 42.97)			24a - 42.148)		
8b - 42.94)		42.96	24b - 41.995)		42.07
9a - 40.91)			25a - 41.7747)		
9b - 40.70)		40.80	25b - 39.6972)		40.73
10a - 40.34)			26a - 38.466)		
10b - 40.36)		40.35	26b - 38.606)		38.53
11a - 40.72)			27a - 45.3679)		
11b - 40.78)		40.75	27b - 40.9774)		43.17
12a - 39.58)			28a - 40.954)		
12b - 39.62)		39.60	28b - 41.038)		40.99
13a - 42.73)			29a - 37.952)		
13b - 42.70)		42.71	29b - 38.088)		38.02
14a - 41.14)			30a - 41.725)		
14b - 41.32)		41.23	30b - 41.078)		41.40
15a - 41.79)					
15b - 41.89)		41.84			

Bulk Specific Gravity

Bulk specific gravity was reported incompletely in Progress Report No. 1 and will be reported completely here. The method used was:

$$\text{Bulk S. G.} = \frac{\text{wt. of sample in air}}{\text{loss of weight in water}} \quad (\text{no drying precautions})$$

This was used to help evaluate the limestone and separate into zones differing in character. It is especially valuable in calculating the amount of pore space in a sample. A large enough sample was used in every case to make the error negligible when not observing drying precautions.

BULK SPECIFIC GRAVITY

QUARRY NO. 1, LEXINGTON, KY.

<u>Level</u>	<u>Bulk Sp. G.</u>
0	2.71
1	2.64
2	2.59
3	2.67
4	2.67
5	2.68
6	2.69
7	2.66
8	2.69
9	2.68
10	2.66
11	2.66
12	2.68
13	2.66
14	2.65
15	2.65
16	2.66
17	2.65
18	2.65
19	2.66
20	2.65
21	2.66
22	2.64
23	2.64
24	2.65
25	2.68
26	2.67
27	2.66
28	2.71
29	2.67
30	2.66
Quarry Average	2.666

BULK SPECIFIC GRAVITY

QUARRY NO. 2, MOUNT VERNON, KY.

<u>Level</u>	<u>Bulk Sp. G.</u>	<u>Level</u>	<u>Bulk Sp. G.</u>
1	2.69	51	2.67
2	2.69	52	2.67
3	2.69	53	2.68
4	2.70	54	2.67
5	2.70	55	2.72
6	2.70	56	2.70
7	2.70	57	2.72
8	2.70	58	2.51
9	2.70	59	2.69
10	2.70	60	2.67
11	2.62	61	2.67
12	2.52	62	2.62
13	2.43	63	2.68
14	2.43	64	2.66
15	2.43	65	2.58
16	2.54	66	2.59
17	2.69	67	2.59
18	2.68	68	2.58
19	2.68	69	2.58
20	2.69	70	2.57
21	2.69	71	2.58
22	2.69	72	2.59
23	2.65	73	Cover
24	2.63	74	2.45
25	2.61	75	2.42
26	2.61	76	2.36
27	2.62	77	2.49
28	2.68	78	2.56
29	2.70	79	2.31
30	2.68	80	2.61
31	2.68	81	2.62
32	2.70	82	2.67
33	2.71	83	2.66
34	Shale	84	2.64
35	2.69	85	2.65
36	2.70	86	2.67
37	2.69	87	2.66
38	2.70	88	2.55
39	2.72	89	2.61
40	2.70	90	2.65
41	2.65	91	2.64
42	2.65	92	2.70
43	2.65	93	2.65
44	2.66	94	2.63
45	2.70	95	2.67
46	2.66	96	2.64
47	2.66	97	2.67
48	2.66	98	2.64
49	2.70	99	2.66
50	2.66	100	2.69

BULK SPECIFIC GRAVITY
QUARRY NO. 2 (Continued)

<u>Level</u>	<u>Bulk Sp. G.</u>
101	2.69
102	2.69
103	2.69
104	2.65
105	2.53
106	2.56
107	2.62
108	2.56
109	2.63
110	2.69
111	Cover
112	2.68
113	2.55
114	2.55
115	2.68
116	2.57
117	2.61
118	2.64
119	2.69
120	2.65
121	2.65
122	2.48
123	2.57
124	2.62
125	2.69
126	2.69
127	2.69
128	2.66
129	2.69
130	2.68
131	2.65
132	Cover
133	2.62
134	2.69
135	2.66
Quarry Average	2.636

BULK SPECIFIC GRAVITY

QUARRY NO. 3, OLIVE HILL, KY.

<u>Level</u>	<u>Bulk Sp. G.</u>	<u>Level</u>	<u>Bulk Sp. G.</u>
1	2.68	34	2.30
2	2.69	35	2.30
3	2.68	36	2.40
4	2.68	37	2.57
5	2.67	38	2.56
6	2.69	39	2.62
7	2.67	40	2.55
8	2.67	41	2.56
9	2.65	42	2.62
10	2.68	43	2.59
11	2.67	44	2.60
12	2.67	45	2.61
13	2.68	46	2.63
14	2.68	47	2.56
15	2.69	48	2.59
16	2.67	49	2.51
17	2.66	50	2.51
18	2.67	51	2.69
19	2.64	52	2.54
20	2.64	53	2.68
21	Shale	54	2.53
22	2.65	55	2.66
23	2.65	56	2.64
24	2.67	57	2.63
25	2.67	58	2.56
26	2.67	59	2.57
27	2.60	60	2.63
28	2.61	61	2.48
29	2.58	62	2.52
30	2.50	63	2.63
31	2.44	64	2.64
32	2.66	65	2.64
33	2.57		
		Quarry Average - 2.606	

BULK SPECIFIC GRAVITY
 QUARRY NO. 4, AVOCA, KY.

<u>Level</u>	<u>Bulk Sp. G.</u>
0	2.69
1	2.65
2	2.66
3	2.70
4	2.74
5	2.71
6	2.62
7	2.78
8	2.73
9	2.71
10	2.77
11	2.72
12	2.79
13	2.69
14	2.59
15	2.79
16	2.72
17	2.73
18	2.70
19	2.66
20	2.56
21	2.54
22	2.47
23	2.66
24	2.67
25	2.73
26	2.71
27	2.66
28	2.64
29	2.68
30	2.63
31	2.68
32	2.68
33	2.56
Quarry Average	2.677

Absolute Specific Gravity

A liberal definition of absolute specific gravity as applied to limestone may be described as the greatest possible concentration of matter per unit of volume. For example, a piece of stone as it is taken from the quarry is greatly expanded by a great number of minute void spaces. Its specific gravity in this state is referred to as its bulk specific gravity. If such stone could be compressed and compacted until all voids were eliminated, there would be a greater concentration of matter per unit of volume, and in the case of limestone there would be corresponding increase in the specific gravity. Having provided a basis for an absolute determination, the absolute values must be arrived at indirectly.

To measure the absolute specific gravity of a consolidated stone it is necessary to crush it and to break up the structures enclosing the voids. More simply, the stone is pulverized to pass a 100 mesh screen and the volume of a known mass of the material is measured by immersion.

In more detail, the procedure is described as follows: Representative samples of the various levels were pulverized using a desk type unit.

250 ml. calibrated and tared volumetric flasks were used as pycnometers. Approximately 200 - 250 grams of the samples were weighed into the flasks, the samples were covered with distilled water and attached to a vacuum outlet, 74 - 75 mm., agitated periodically until complete saturation was effected, usually from 2 to 3 hours, and then made up to volume with distilled water. The flask and contents were quickly weighed and the

temperature taken. An example of the method of calculating is shown below:

$$(\text{Wt. of flask + sample}) - \text{wt. of flask} = \text{wt of sample}$$

$$(\text{Wt. of flask + sample + water}) - (\text{wt. of flask + sample}) = \text{wt. of water}$$

$$\text{Wt. of water} + (\text{wt. of water} \times .0010 \text{ correction for bouyancy of air}) = \text{actual wt. of water}$$

$$\text{Actual wt. of water/density of water at } T^{\circ}\text{C} = \text{actual vol. of water}$$

$$\text{Vol. of flask} - \text{vol. of water} = \text{actual vol. of sample}$$

$$\text{Wt. of sample} + (\text{vol. of sample} \times .0010 \text{ correction for bouyancy of air}) = \text{actual wt. of sample}$$

$$\text{Actual wt. of sample/actual vol. of sample} = \underline{\text{Absolute specific gravity}}$$

All weighings were made on a balance having a capacity of over 1 kg. and a sensitivity of .001 gm. Precautions were taken to assure complete saturation, cleanliness of the flasks and accuracy in making up to volume.

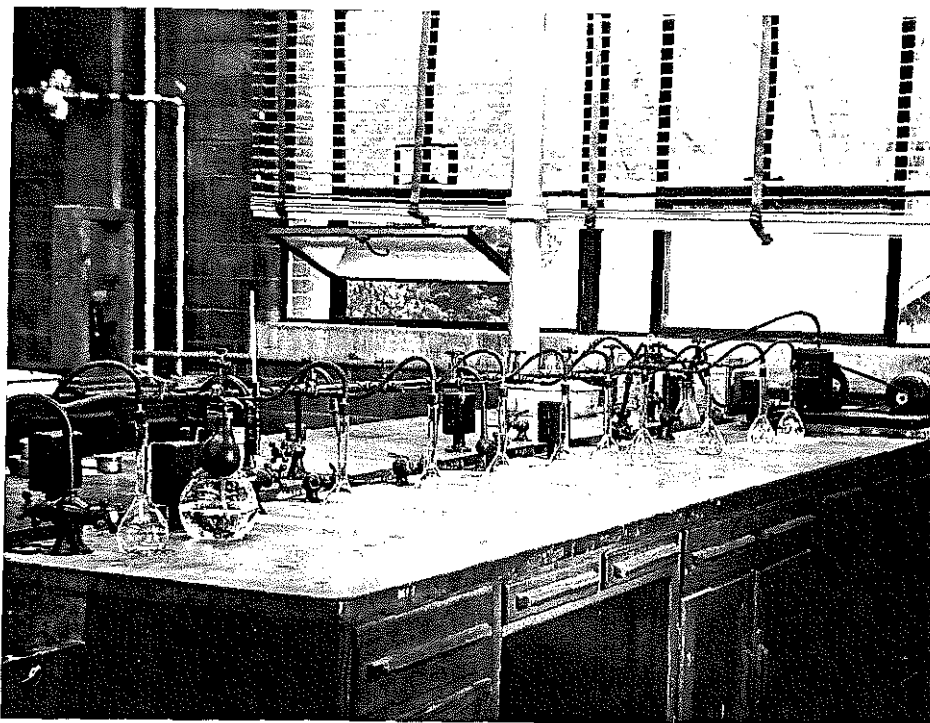


Fig. 3. Vacuum Line for Ten Simultaneous Evacuatiions

ABSOLUTE SPECIFIC GRAVITY

QUARRY NO. 1, LEXINGTON, KY.

<u>Level</u>	<u>Abs. Sp. G.</u>
0	2.7338
1	2.7138
2	2.7135
3	2.7249
4	2.7245
5	2.7297
6	2.7247
7	2.7241
8	2.7469
9	2.7320
10	2.7388
11	2.7238
12	2.7710
13	2.7454
14	2.7408
15	2.7278
16	2.7310
17	2.7262
18	2.7332
19	2.7379
20	2.7417
21	--
22	2.7503
23	2.7599
24	2.7480
25	2.7408
26	2.7387
27	2.7483
28	2.7292
29	2.7659
30	2.7069
Quarry Average -	2.7358

ABSOLUTE SPECIFIC GRAVITY

QUARRY NO. 2, MOUNT VERNON, KY.

<u>Level</u>	<u>Abs. Sp. G.</u>	<u>Level</u>	<u>Abs. Sp. G.</u>
1	2.7085	51	2.7083
2	2.7071	52	2.7096
3	2.7113	53	2.7049
4	2.7109	54	--
5	2.7108	55	2.7236
6	2.7060	56	2.7346
7	2.7134	57	2.7344
8	2.7160	58	2.7120
9	2.7135	59	2.7134
10	2.7085	60	2.7148
11	2.7119	61	2.7145
12	2.7019	62	--
13	2.7139	63	2.7204
14	2.7110	64	2.7174
15	2.7247	65	2.7127
16	2.7119	66	2.7133
17	2.7134	67	2.7677
18	2.7139	68	2.7455
19	2.7225	69	2.7595
20	2.7170	70	2.7122
21	2.7048	71	2.7365
22	2.7043	72	2.7186
23	2.7077	73	--
24	2.7079	74	2.7068
25	2.7065	75	2.7218
26	2.7065	76	2.7107
27	2.7044	77	2.7729
28	2.7093	78	2.6952
29	2.7073	79	2.7131
30	2.6973	80	2.8083
31	2.7077	81	2.7443
32	2.7090	82	--
33	2.7197	83	--
34	Shale	84	2.7145
35	2.7128	85	2.7100
36	2.7176	86	2.7134
37	2.7343	87	--
38	2.7430	88	2.7062
39	2.7234	89	2.7102
40	2.7117	90	2.7053
41	2.6967	91	2.7076
42	2.7016	92	2.7133
43	2.7015	93	2.7118
44	2.7117	94	2.6880
45	2.7262	95	2.7022
46	2.7102	96	--
47	2.7049	97	2.6909
48	2.7075	98	2.6961
49	2.7137	99	2.7084
50	2.7093	100	2.7089

QUARRY NO. 2 (Continued)

<u>Level</u>	<u>Abs. Sp. G.</u>
101	2.7047
102	2.7101
103	2.6925
104	2.7075
105	--
106	2.7170
107	2.7131
108	2.7108
109	2.7088
110	2.7319
111	2.7122
112	2.6973
113	2.7050
114	2.7125
115	2.7121
116	2.7137
117	2.7171
118	2.7171
119	2.7078
120	2.7115
121	2.7134
122	--
123	2.6629
124	2.6969
125	--
126	2.7013
127	2.7026
128	2.7104
129	2.7091
130	2.7282
131	2.7123
132	--
133	2.7046
134	2.7203
135	2.7097
Quarry Average -	2.7139

ABSOLUTE SPECIFIC GRAVITY
QUARRY NO. 3, OLIVE HILL, KY.

<u>Level</u>	<u>Abs. Sp. G.</u>	<u>Level</u>	<u>Abs. Sp. G.</u>
1	2.7042	34	2.7374
2	2.6938	35	2.7658
3	2.6978	36	2.8050
4	2.7002	37	2.7623
5	2.7098	38	2.7179
6	2.6899	39	2.7152
7	2.7005	40	2.7484
8	2.7037	41	2.7295
9	2.6961	42	2.7119
10	2.6983	43	2.7076
11	2.7070	44	2.7105
12	2.7288	45	2.7076
13	2.7504	46	2.7097
14	2.7201	47	2.7172
15	2.7302	48	2.7180
16	2.7396	49	2.7208
17	2.7679	50	2.7198
18	2.6992	51	2.7286
19	2.7113	52	2.7322
20	2.7108	53	2.7142
21	Shale	54	2.7117
22	2.7811	55	2.7079
23	2.7256	56	2.7097
24	2.7984	57	2.7094
25	2.8153	58	2.7204
26	2.8148	59	2.7071
27	2.7360	60	2.7241
28	2.8350	61	--
29	2.8132	62	2.7031
30	2.7924	63	2.8018
31	2.6686	64	2.7014
32	2.7236	65	2.7028
33	2.7112		
		Quarry Average - 2.7294	

ABSOLUTE SPECIFIC GRAVITY

QUARRY NO. 4, AVOCA, KY.

<u>Level</u>	<u>Abs. Sp. G.</u>
0	2.8816
1	2.8523
2	2.8493
3	2.8249
4	2.8432
5	2.8558
6	2.8527
7	2.8491
8	2.8362
9	2.8550
10	2.8542
11	2.9771
12	2.8599
13	2.8356
14	2.8461
15	2.8344
16	2.8413
17	2.8548
18	2.8421
19	2.8454
20	2.8510
21	--
22	2.8531
23	2.8470
24	2.8538
25	2.8316
26	2.8401
27	2.8313
28	2.8144
29	2.8295
30	2.8302
31	2.8395
32	2.8408
33	2.8421
Quarry Average -	2.8453

Porosity

From a quantitative standpoint, the porosity of limestone or other substance is the ratio of the volume of internal open spaces to the total volume of the specimen. The open spaces are variously referred to as "pores", "interstices", or "voids". The pores of a substance may be isolated from one another, or they may be intercommunicating, and it is in this latter sense that "available pore space", or "effective porosity" is spoken of in contradistinction to "total pore space" or "total porosity".

Porosity may be determined by direct or indirect method. The present method is an indirect method to be followed in the near future by a direct method. This portion of the report is concerned with "total porosity".

Total porosity by an indirect method is calculated from the previously determined physical properties, Bulk Specific Gravity and Absolute Specific Gravity. The Specific Gravity of a specimen is determined of the "bulk", including the limestone and its voids. This is the value referred to as the Bulk Specific Gravity. A large enough specimen was used in all cases to be able to ignore surface saturation phenomena as a negligible error.

The specimen was crushed to pass a 100 mesh screen and in the Absolute Specific Gravity determination all of the air was displaced by water, thus destroying all voids.

The difference in these two values represents the weight of material displaced by voids. This weight divided by the absolute specific gravity gives the volume of material displaced per unit volume and multiplying by 100 will place the figure as a percentage

per unit volume or, for example:

Absolute Sp. Gr. - 2.8816

Bulk Sp. Gr. - 2.69

.1916 = wt. of material displaced by voids

.1916/2.8816 = volume of material displaced per unit vol. =
.0665 ml.

.0665/1 ml. x 100 = 6.65% per unit volume

PERCENTAGE OF VOID SPACE PER UNIT VOLUME OF BULK MATERIAL

QUARRY NO. 1, LEXINGTON, KY.

<u>Level</u>	<u>Per Cent Voids</u>
0870%
1	2.719%
2866%
3	2.015%
4	2.000%
5	1.821%
6	1.273%
7	2.353%
8	2.071%
9	1.903%
10	2.877%
11	1.975%
12	3.284%
13	3.111%
14	3.313%
15	2.852%
16	2.600%
17	2.795%
18	3.044%
19	2.845%
20	3.345%
21	---
22	4.010%
23	4.334%
24	3.566%
25	2.218%
26	2.508%
27	3.213%
28703%
29	3.467%
30	1.733%
Quarry Average ...	2.523%

PERCENTAGE OF VOID SPACE PER UNIT VOLUME OF BULK MATERIAL

QUARRY NO. 2, MOUNT VERNON, KY.

<u>Level</u>	<u>Per Cent Voids</u>	<u>Level</u>	<u>Per Cent Voids</u>
1	0.683%	51	1.414%
2	0.632%	52	1.461%
3	0.786%	53	.920%
4	0.402%	54	--
5	0.398%	55	.315%
6	0.222%	56	1.265%
7	0.494%	57	.527%
8	0.589%	58	7.448%
9	0.497%	59	.862%
10	0.589%	60	1.650%
11	3.446%	61	1.693%
12	6.960%	62	--
13	10.395%	63	1.485%
14	10.365%	64	2.112%
15	10.449%	65	4.892%
16	6.339%	66	4.562%
17	0.862%	67	6.420%
18	1.249%	68	6.028%
19	1.561%	69	6.505%
20	0.994%	70	5.243%
21	0.547%	71	5.719%
22	0.529%	72	4.730%
23	2.131%	73	--
24	2.877%	74	9.487%
25	3.565%	75	11.088%
26	3.565%	76	12.938%
27	3.121%	77	10.202%
28	1.081%	78	5.016%
29	0.270%	79	14.857%
30	0.641%	80	7.061%
31	1.023%	81	4.529%
32	0.332%	82	--
33	0.357%	83	--
34	Shale	84	2.744%
35	0.840%	85	2.214%
36	0.648%	86	1.599%
37	1.620%	87	--
38	1.568%	88	5.772%
39	0.125%	89	3.697%
40	0.431%	90	2.044%
41	1.732%	91	2.497%
42	1.910%	92	.490%
43	1.906%	93	2.279%
44	1.906%	94	2.158%
45	0.961%	95	1.192%
46	1.852%	96	--
47	1.660%	97	.777%
48	1.754%	98	2.081%
49	0.505%	99	1.787%
50	1.820%	100	.698%

QUARRY NO. 2 (Continued)

<u>Level</u>	<u>Per Cent Voids</u>
101543%
102742%
103093%
104	2.124%
105	--
106	5.778%
107	3.431%
108	5.552%
109	2.909%
110	1.534%
111	--
112642%
113	5.730%
114	5.991%
115	1.183%
116	5.469%
117	3.942%
118	2.837%
119657%
120	2.268%
121	2.336%
122	--
123	3.488%
124	2.851%
125	--
126418%
127466%
128	1.859%
129705%
130	1.767%
131	2.297%
132	--
133	3.128%
134	1.114%
135	1.834%
Quarry Average	2.864%

PERCENTAGE OF VOID SPACE PER UNIT VOLUME OF BULK MATERIAL

QUARRY NO. 3, OLIVE HILL, KY.

<u>Level</u>	<u>Per Cent Voids</u>	<u>Level</u>	<u>Per Cent Voids</u>
1	.845%	34	15.979%
2	.141%	35	16.841%
3	.660%	36	14.438%
4	.748%	37	6.961%
5	1.469%	38	5.810%
6	.000%	39	3.506%
7	1.129%	40	7.219%
8	1.246%	41	6.210%
9	1.710%	42	3.389%
10	.678%	43	4.343%
11	1.367%	44	7.766%
12	2.155%	45	3.605%
13	2.560%	46	2.941%
14	1.474%	47	5.785%
15	1.472%	48	4.709%
16	2.540%	49	7.748%
17	3.898%	50	7.714%
18	1.082%	51	1.415%
19	2.630%	52	7.035%
20	2.612%	53	1.260%
21	shale	54	6.700%
22	4.714%	55	1.769%
23	2.774%	56	2.572%
24	4.588%	57	2.930%
25	5.161%	58	5.896%
26	5.144%	59	5.064%
27	4.971%	60	3.454%
28	7.936%	61	--
29	8.249%	62	6.774%
30	10.471%	63	6.132%
31	8.566%	64	2.265%
32	2.335%	65	2.323%
33	5.208%	Quarry Average	4.462%

PERCENTAGE OF VOID SPACE PER UNIT VOLUME OF BULK MATERIAL

QUARRY NO. 4, AVOCA, KY.

<u>Level</u>	<u>Per Cent Voids</u>
0	6.65%
1	7.10%
2	6.66%
3	4.42%
4	3.63%
5	5.11%
6	8.16%
7	2.43%
8	3.74%
9	5.08%
10	2.95%
11	8.64%
12	2.44%
13	5.13%
14	9.00%
15	1.57%
16	4.27%
17	4.37%
18	5.00%
19	6.52%
20	10.21%
21	--
22	13.43%
23	6.57%
24	6.44%
25	3.59%
26	4.58%
27	6.05%
28	6.20%
29	5.28%
30	7.07%
31	5.62%
32	5.66%
33	9.93%
Quarry Average ...	5.864%

Chemical Analysis

The analysis of Quarry No. 1 was reported in more detail (Appendix 2 of the previous report) than the later results. The later analyses do not involve the analysis of the insoluble residues. They are reported as the percentage of insoluble material present, and the composition of the residue is considered to be largely SiO_2 , in the form of Quartz and other silicates. This difference does not detract greatly from the significance of the results, if the per cent of insoluble residue is treated as being indicative of the per cent of SiO_2 .

The results for Quarry No. 1 and Quarry No. 4 are submitted here complete. Quarry No. 2 is reported partially complete. In addition to the incomplete data for Quarry No. 2, it is of interest to compare these results with an average for the quarry which was computed from data taken from the Annual Report of Quarries, prepared by the Testing Laboratory, January, 1947. These averages are listed below and properly labeled.

As a matter of further interest, four levels for Quarry No. 3 were listed in the Annual Report of Quarries. These values indicate approximate averages of 65% CaCO_3 , 20% MgCO_3 , 14% insoluble residue, and 1% R_2O_3 . Actual quarry average for insoluble residue is 11.9%. These approximations are mentioned merely to indicate the trend of the quarry.

Analysis of Quarry No. 1

<u>Level</u>	<u>Zone</u>	<u>CaCO_3</u>	<u>MgCO_3</u>	<u>Fe_2O_3</u>	<u>SiO_2</u>	<u>Al_2O_3</u>	<u>$\text{Ca}_3(\text{PO}_4)_2$</u>
0- 5	1	86.10%	6.94%	.75%	3.51%	1.76%	.93%
6-11	2	82.60%	8.51%	.84%	3.78%	1.69%	2.55%
12-18	3	85.80%	7.28%	.61%	2.15%	1.46%	2.61%
19-24	4	82.00%	8.00%	.83%	3.95%	2.32%	2.81%
25-30	5	87.40%	5.29%	.53%	2.26%	1.53%	2.93%
Averages -	-	84.78%	7.20%	.71%	3.13%	1.75%	2.37%

Analysis of Quarry No. 2

<u>Level</u>	<u>Zone</u>	<u>CaCO₃</u>	<u>MgCO₃</u>	<u>R₂O₃</u>	<u>Insol. Res.</u>
2-6	2	96.16%	trace	1.21%	4.70%
7-9	3	97.10%	"	1.02%	2.59%
10-16	4	98.22%	"	.71%	1.34%
17-22	5	94.10%	"	1.23%	4.56%
23-29	6	94.30	"	1.55%	3.98%
55-60	11	96.17%	3.24%	.71%	1.67%
61-63	12	93.71%	5.82%	.85%	1.98%
64-78	13	91.19%	4.59%	.79%	2.95%
79-82	14	73.72%	18.01%	3.47%	3.98%
83-86	15	94.15%	4.49%	.79%	3.15%
<hr/>					
Averages for quarry		as indicated by Annual Report of Quarries:			
		94.12%	1.68%	.51%	3.18%

Analysis of Quarry No. 4

<u>Level</u>	<u>Zone</u>	<u>CaCO₃</u>	<u>MgCO₃</u>	<u>R₂O₃</u>	<u>Insol. Res.</u>
0-3	1	55.89%	35.40%	2.76%	5.96%
4-9	2	59.10%	36.70%	1.90%	3.72%
10-11	3	61.37%	35.09%	2.22%	1.36%
12-18	4	59.57%	37.24%	1.78%	2.39%
19-23	5	57.37%	39.37%	.96%	3.96%
24-28	6	69.99%	23.01%	1.14%	6.34%
29-33	7	68.82%	23.78%	1.12%	6.19%
<hr/>					
Averages -		61.73%	32.94%	1.70%	4.27%

Thin Sections

Thin sections have been completed for all the quarries and a complete file of photomicrographs is on hand. The structure is nowhere exclusive as an identical structure may be found in an accepted and a rejected quarry.

Quarry No. 4, the source rated as excellent, has so nearly a homogenous structure as to be phenomenal. Differences occur only in size of crystals and the impurities present, although to the observer, no measurable difference may be seen in the impurities. A photomicrograph is submitted for illustration:

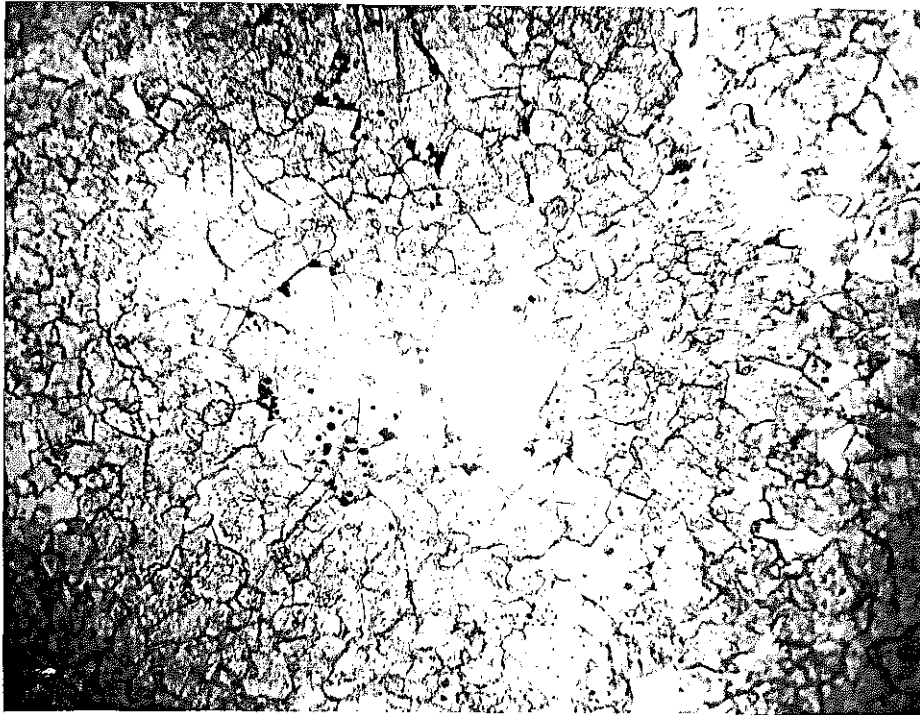


Fig. 4. Photomicrograph of Thin Section, Quarry No. 4

Until such time as the tests on individual zones show their relative durability, no special inference can be placed on structure.

CONCLUSIONS

Data presented in the foregoing section was subject to the usual experimental error and is used as such. It is the intent of the authors to discuss as specifically as possible with this information, the results and their significance in this problem.

The overwhelming fact this information presents is that due to the wide range in physical properties, it is fallacious to analyze a quarry as a whole. The act of breaking up a quarry into zones of sedimentation does not make these physical properties specific, so that generalities must always be used. The smaller divisions make these properties more nearly specific so that it is more useful.

In this discussion, there appears a set of graphs with figures presented for standard deviation, coefficient of variation, maximum deviation, and the usual quarry average. Standard deviation is calculated from the formula, $\sigma =$

$$\sqrt{\frac{x_1^2 + x_2^2 + \dots + x_n^2}{n} - \bar{x}^2}$$

where n = number of observations and \bar{x} = average figure.

Coefficient of variation is calculated from the formula,

$V = 100 \times \frac{\sigma}{\bar{x}}$ which is expressed as a percentage. Maximum deviation indicates the range over which the values occur.

Quarry average is merely the arithmetic mean.

Insoluble Residue is the most valuable criteria to date:

(Fair)	Quarry No. 1 -	5.145%
(Good)	Quarry No. 2 -	3.880%
(Poor)	Quarry No. 3 -	11.900%
(Excellent)	Quarry No. 4 -	4.166%

Quarries Nos. 2 and 4 are the lowest in insoluble residue and the best in performance. They do not fall exactly in line with our ratings, however the difference is slight. The only other obvious property which could serve to differentiate these is one of hardness or tenacity. This could involve a formula of multiplying the insoluble residue by the reciprocal of the hardness or breaking strength. Subsequent tests should show whether this is true or not.

Determinations of hygroscopic water and total organic content have been abandoned as being untenable. Not enough difference is shown over the vertical extent of a quarry to use this evaluation, particularly since a chemical analysis is available.

The percent porosity (absolute) has indicated nothing other than the fact that not much difference occurs in any of the quarries. Effective porosity should be the next test and an comparison to permeability may be a valuable test.

Permeability is interesting in that the horizontal direction generally shows more permeability than the vertical. This would possibly indicate that limestone has slight differences in strength with regard to orientation. No comparable differences may be noted over the total quarries. As mentioned earlier, a study of pore size and shape will be run and this should determine which fall into the capillary limits. In conjunction with permeability, this could be very pertinent.

Absolute Specific Gravity - It is only by correlation of the various properties of limestone that we are able to visualize the "over-all" picture of how these properties are related, and to understand how the limestone itself is put together. Absolute specific gravity is, as indicated, an absolute value and provides a firm basis upon which to base correlations. Absolute specific gravity has already been defined as being the greatest possible concentration of matter per unit of volume. In other words, it is the maximum density which could be obtained by "close packing" of the particles which make up the limestone. Under such ideal conditions, all voids are eliminated and the specific gravity measured is necessarily unit weight divided by the sum of the volumes of the components as determined from their specific gravity constants and percentage present.

For illustration, these relations are shown below:

<u>Component</u>	<u>Percent Present</u>	<u>Wt. per gram</u>	<u>Sp.Gr. Constant</u>	<u>Vol. per gram</u>
SiO ₂	3.13%	.0313 ÷	2.30 =	.01361
Fe ₂ O ₃	.71%	.0071 ÷	5.25 =	.00135
Al ₂ O ₃	1.75%	.0175 ÷	3.7 =	.00473
Ca ₃ (PO ₄) ₂	2.37%	.0237 ÷	3.14 =	.00755
CaCO ₃	84.78%	.8478 ÷	2.711 =	.31550
MgCO ₃	7.20%	.0720 ÷	3.037 =	.02371
Total Volume per gram				= .36645

$$1 \text{ gram} \div .36645 = 2.730$$

$$\text{Actual Quarry Average} = 2.7358$$

It is easy to see that if the composition is known, the absolute specific gravity can be calculated with fair accuracy. In addition, the absolute specific gravity itself is very indicative of the composition.

Bulk Specific Gravity - is related to the absolute value by virtue of the loose natural arrangement of the limestone as contrasted to the ideal arrangement, which eliminates the void spaces. The two differ only to the amount of void spaces.

It has been shown already that the two determinations are used to calculate the percentage of void space per unit volume of bulk material. This value leads up to questions as to how much of this value is effective and vulnerable to saturation with water, and how much of this void space is isolated or, in effect, provides "air cushions" for the expansions and contractions due to freezing and thawing. These questions provide incentive for future work. The possibility of establishing limits or a limiting ratio between the two values is worthy of further consideration.

Chemical Analysis. Chemical analyses add to the picture by establishing the percentages of materials or minerals which compose the rock. Limestone is, first, a carbonate rock. Other materials are superfluous and probably interfered with crystallization and caused imperfect cleavage in the three planes. These generalities add emphasis to the insoluble residue criteria for discriminating between limestones. High percentages of insoluble materials disrupt the homogeneity of the stone. Carbonates, CaCO_3 and MgCO_3 , are considered stable and occur with uniform crystallization and cleavage. Proof of this idea is seen in the performance record of Quarry No. 4.

There follow a tabulation of data which is intended to illustrate the absence of homogeneity between the various levels of the quarries as determined by their physical properties.

Standard Deviation

	Fair <u>Quarry No. 1</u>	Good <u>Quarry No. 2</u>	Poor <u>Quarry No. 3</u>	Excellent <u>Quarry No. 4</u>
Insol. Res.	2.588	3.039	8.63	2.542
Abs. Sp. G.0706	.063	.496	.143
% Voids893	2.874	3.424	2.447
Bulk Sp. G.0316	.0895	.1413	.105
Permeability117	21.717	.340	5.68

Coefficient of Variation

	Fair <u>Quarry No. 1</u>	Good <u>Quarry No. 2</u>	Poor <u>Quarry No. 3</u>	Excellent <u>Quarry No. 4</u>
Insol. Res.	50.3%	78.3%	72.5%	61.02%
Abs. Sp. G.	2.58%	2.32%	18.17%	5.03%
% Voids	35.39%	100.3%	76.92%	41.73%
Bulk Sp. G.	1.19%	3.39%	5.42%	3.92%
Permeability	68.1%	616.6%	237.6%	302.8%

Range - (Maximum - Minimum)

	Fair <u>Quarry No. 1</u>	Good <u>Quarry No. 2</u>	Poor <u>Quarry No. 3</u>	Excellent <u>Quarry No. 4</u>
Insol. Res.	9.5	17.0	58.0	9.25
Abs. Sp. G.641	.110	.166	.1627
% Voids	3.631	14.732	16.841	11.86
Bulk Sp. G.07	.41	.39	.32
Permeability	1.509	240.922	1.741	29.890

Quarry Averages

	Fair <u>Quarry No. 1</u>	Good <u>Quarry No. 2</u>	Poor <u>Quarry No. 3</u>	Excellent <u>Quarry No. 4</u>
Insol. Res.	5.145	3.88	11.90	4.166
Abs. Sp. G.	2.7358	2.7139	2.7294	2.8453
% Voids	2.523	2.864	4.462	5.864
Bulk Sp. G.	2.666	2.642	2.606	2.677
Permeability1719	3.522	.1431	1.876

10.5
10.0
9.5
9.0
8.5
8.0
7.5
7.0
6.5
6.0
5.5
5.0
4.5
4.0
3.5
3.0
2.5
2.0
1.5
1.0
0.5

Graphical Representation of
(1) Insoluble Residue and
(2) Porosity, Quarry No. 1,
Lexington, Ky.

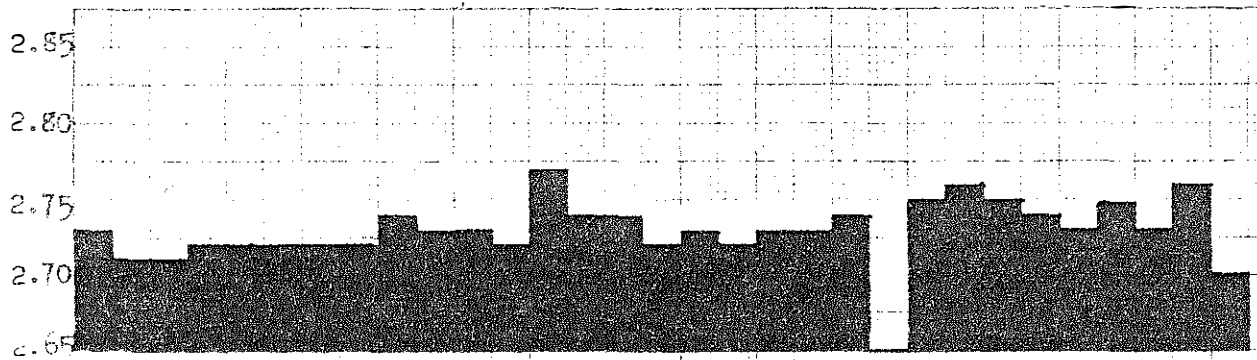
(1) Quarry Average - 5.145%
Standard Deviation - 2.588
Coefficient of Variation - 50.3%
Maximum Deviation - 9.5%

(2) Quarry Average - 2.523%
Standard Deviation - 0.893
Coefficient of Variation - 35.39%
Maximum Deviation - 3.631%

Insoluble Residue, Quarry No. 1 - Central Rock

Per Cent Porosity, Quarry No. 1 - Central Rock

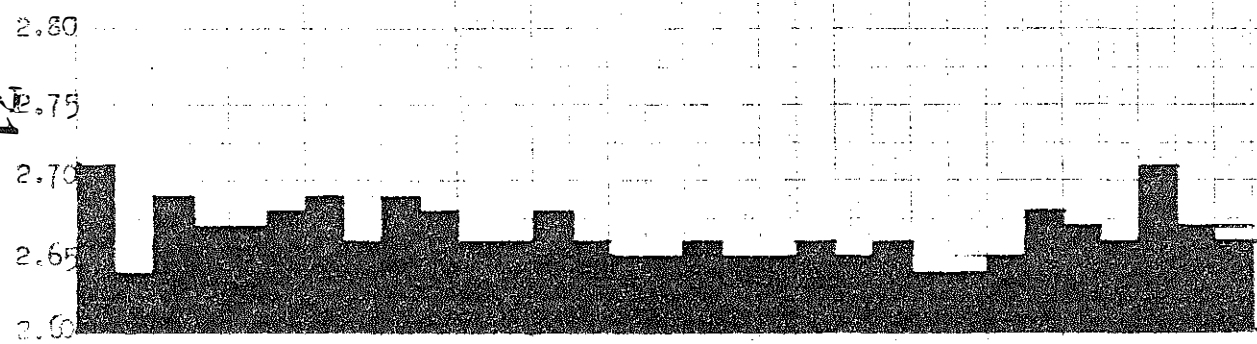
Level 0 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30



Graphical Representation of
 (1) Absolute Specific Gravity,
 (2) Bulk Specific Gravity, and
 (3) Permeability of Quarry No. 1,
 Lexington, Ky.

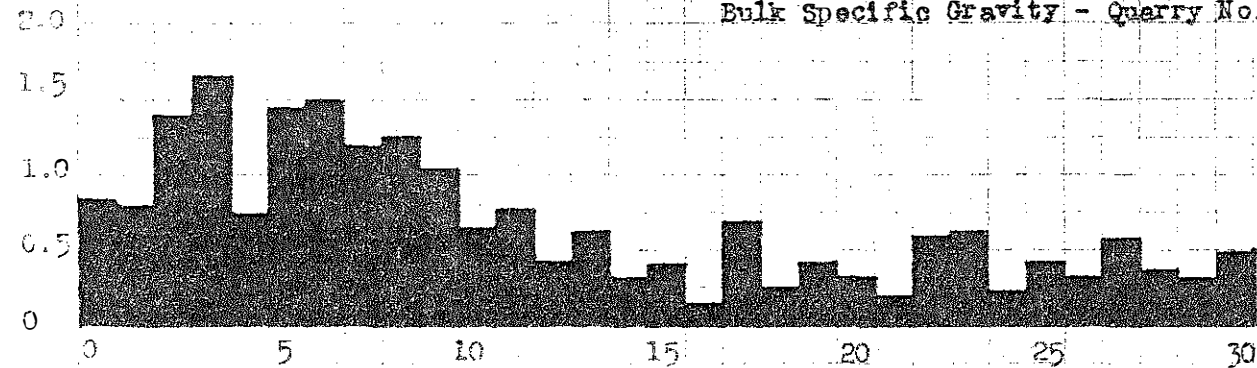
(1) Quarry Average - 2.7358
 Standard Deviation - 0.0706
 Coefficient of Variation - 2.58%
 Maximum Deviation - 0.0641

Absolute Specific Gravity - Quarry No. 1



(2) Quarry Average - 2.666
 Standard Deviation - .0316
 Coefficient of Variation - 1.19%
 Maximum Deviation - .07

Bulk Specific Gravity - Quarry No. 1



(3) Quarry Average - .1719
 Standard Deviation - .117
 Coefficient of Variation - 68.1%
 Maximum Deviation - 1.509%

Averaged Permeability - Quarry No. 1

Graphical Representation of
 (1) Organic Content (SO₂) and
 (2) Hygroscopic Water, Quarry
 No. 1, Lexington, Ky.

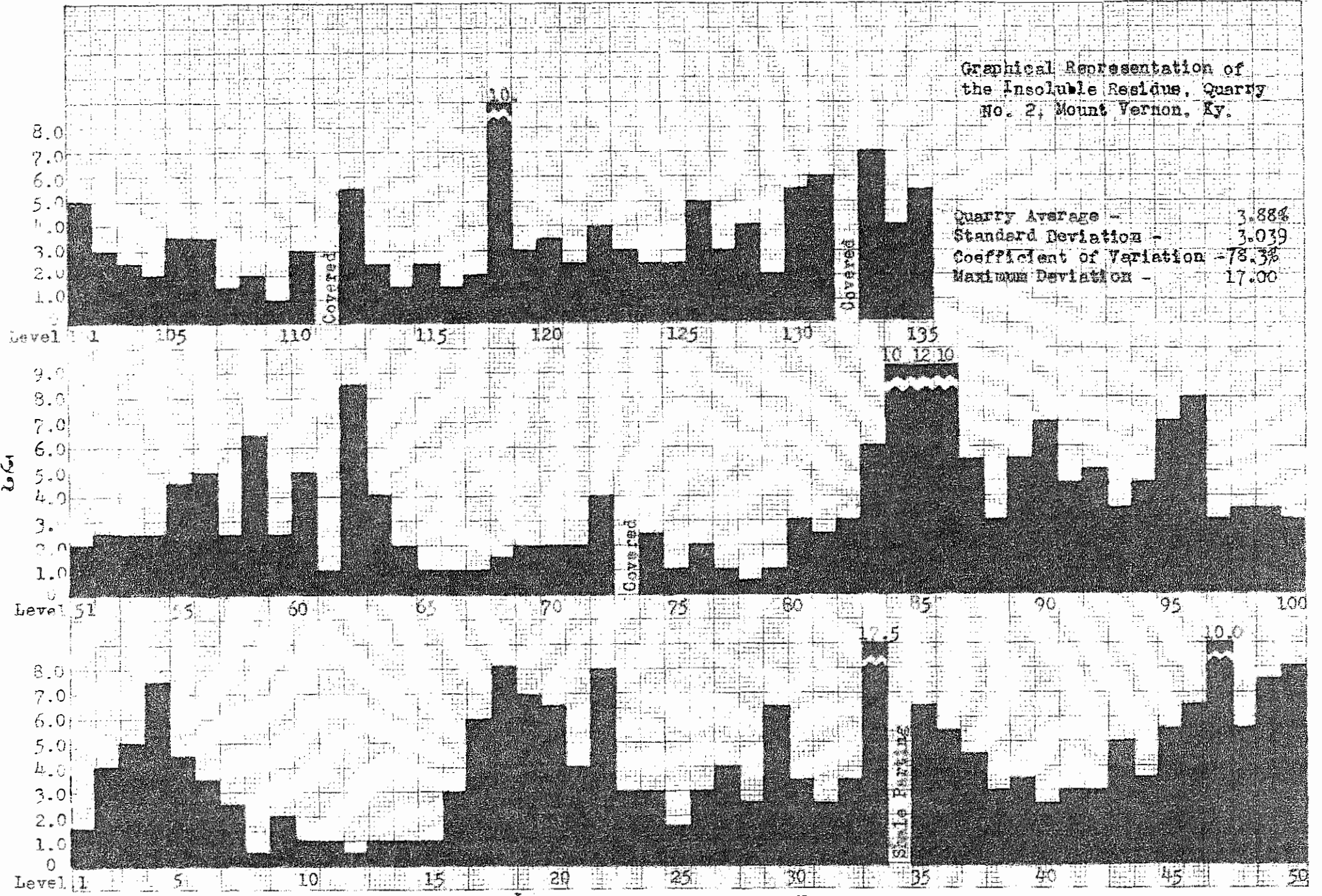
44.0
 43.5
 43.0
 42.5
 42.0
 41.5
 41.0
 40.5
 40.0
 39.5
 39.0
 38.5
 38.0
 37.5
 37.0
 .25
 .20
 .15
 .10

(1)

Per Cent Organic Content (SO₂)

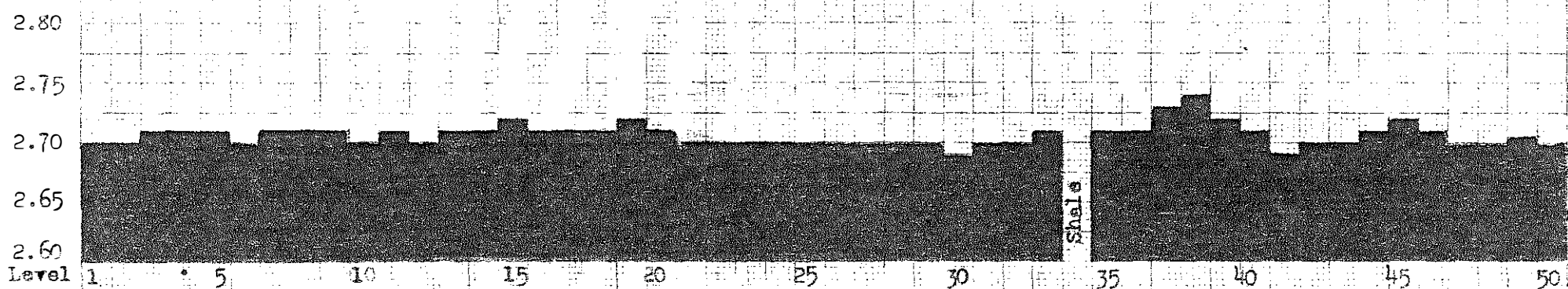
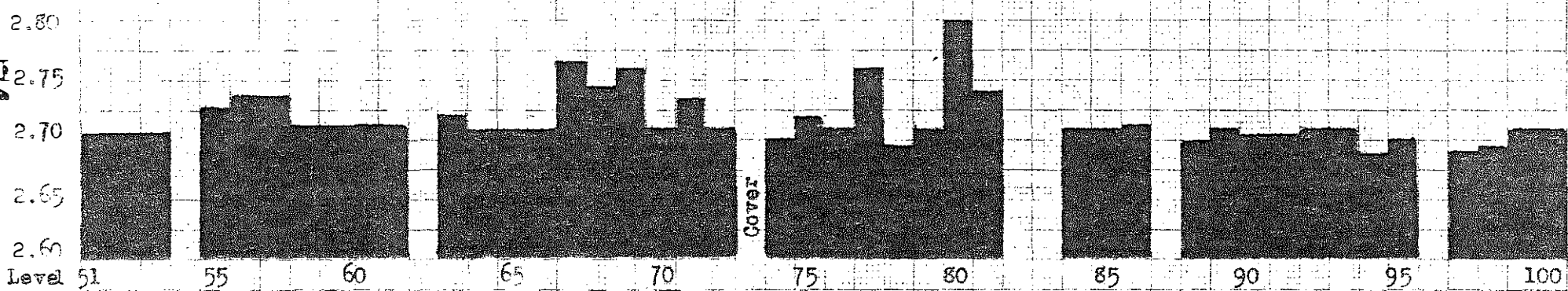
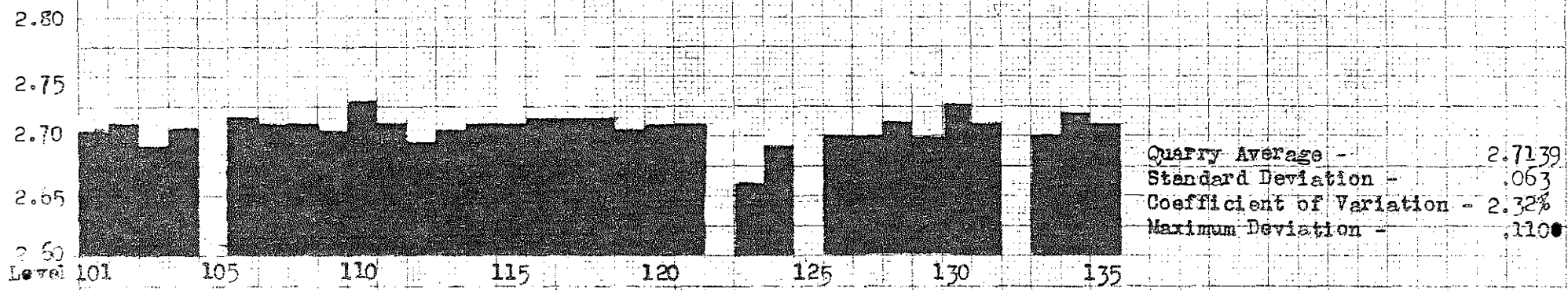
Level 0 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30
 Per Cent H₂O Content (Hygroscopic)

Graphical Representation of
the Insoluble Residue, Quarry
No. 2, Mount Vernon, Ky.



Insoluble Residue - Quarry No. 2

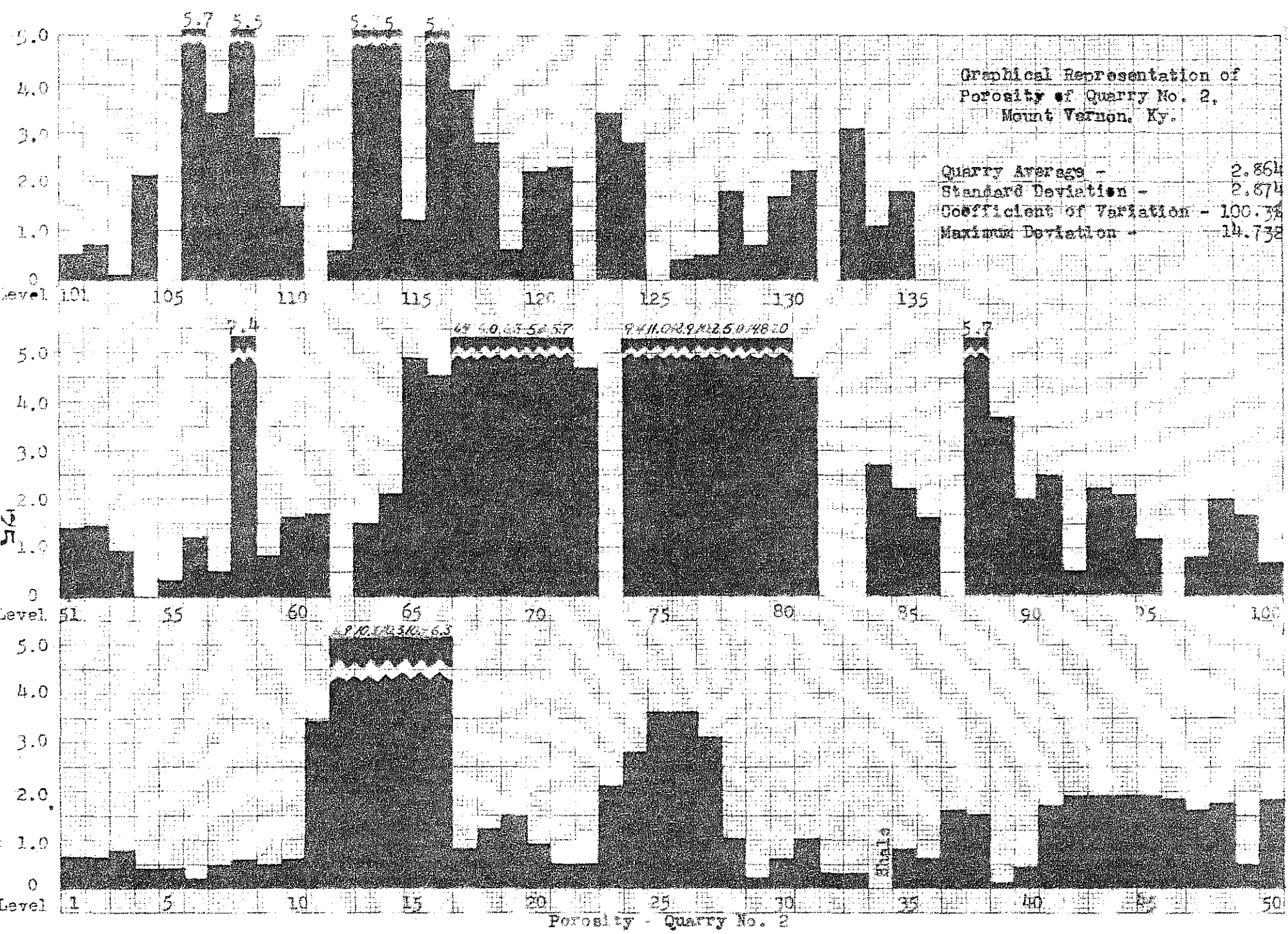
Graphical Representation of
 Absolute Specific Gravity
 Quarry No. 2, Mount Vernon, Ky.



Absolute Specific Gravity - Quarry No. 2

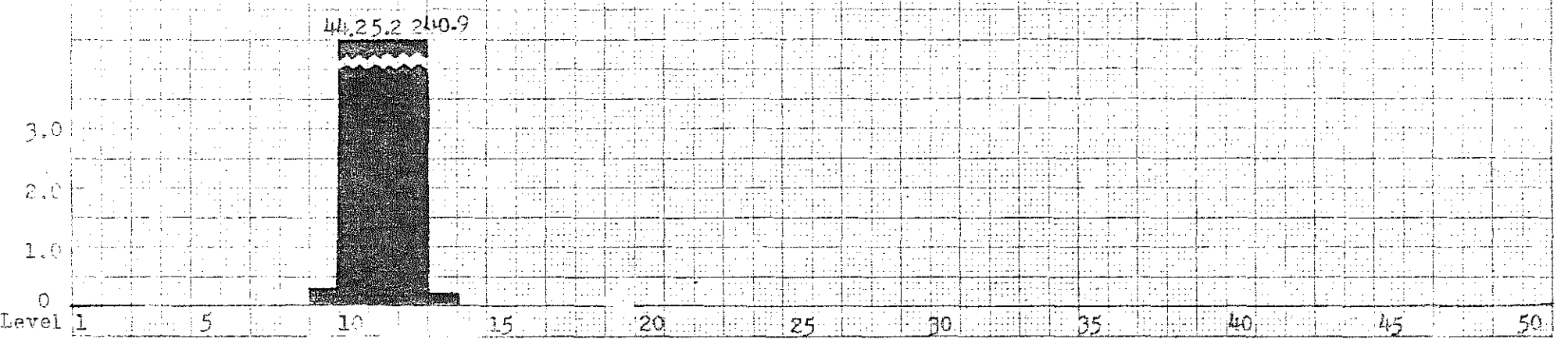
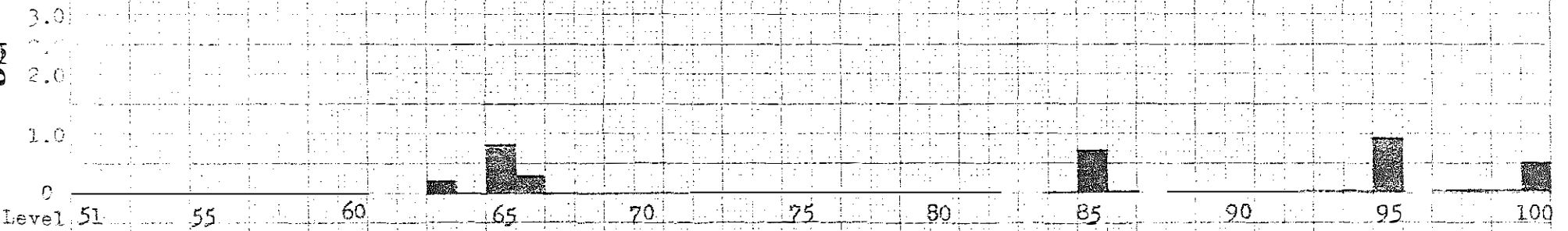
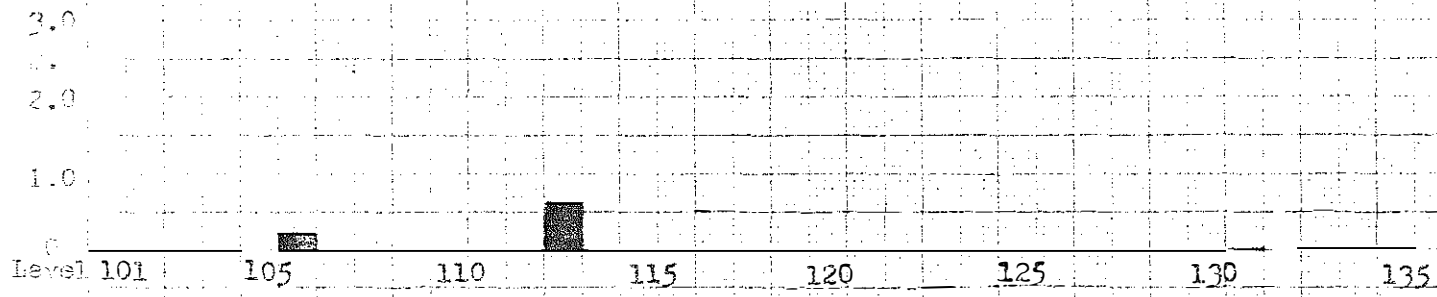
Graphical Representation of
Porosity of Quarry No. 2,
Mount Vernon, Ky.

Quarry Average - 2.864
Standard Deviation - 2.874
Coefficient of Variation - 100.38
Maximum Deviation - 14.738



Graphical Representation of
Permeability (in Millidarcys)
for Quarry No. 2

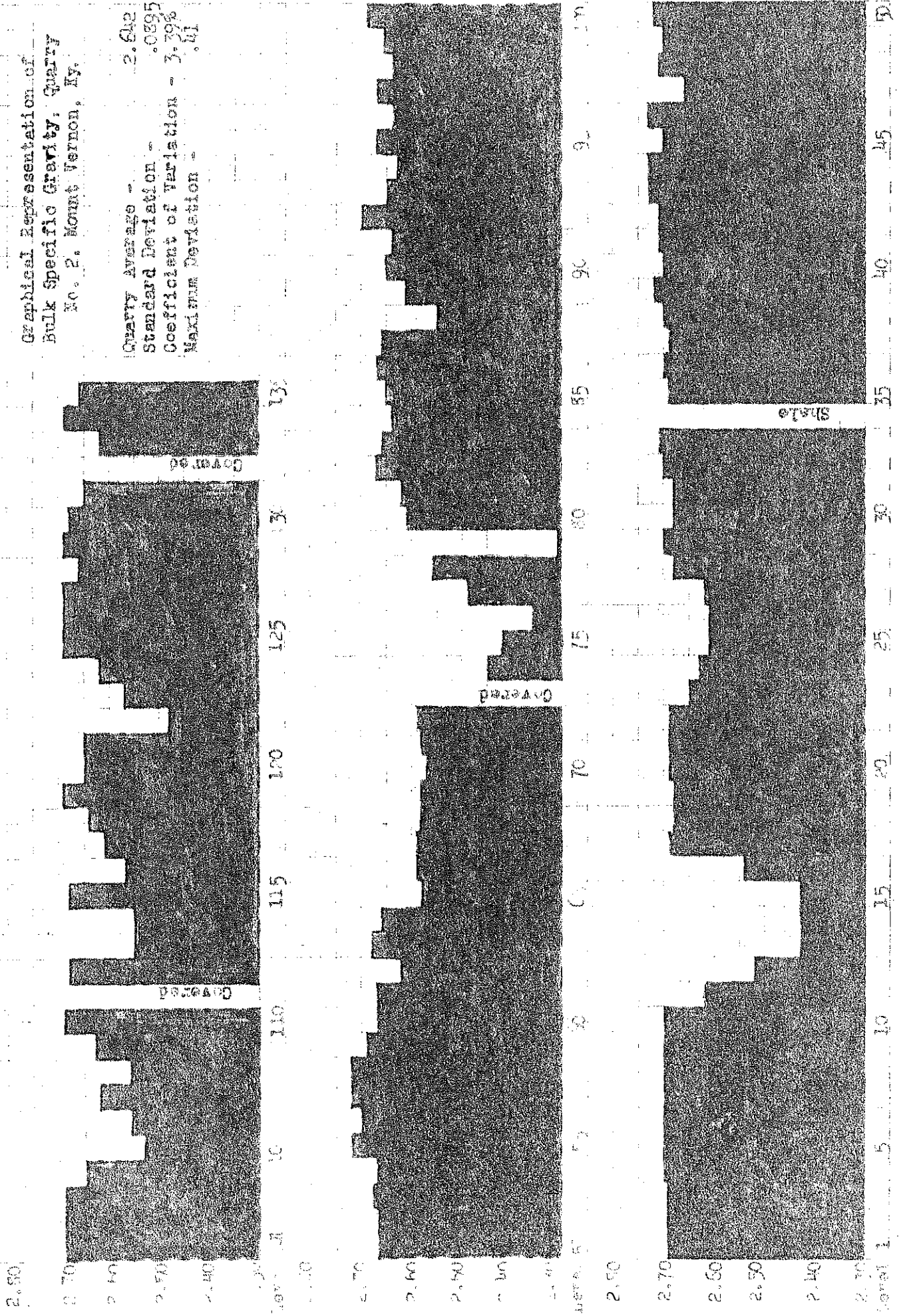
Quarry Average - 3.522
Standard Deviation - 21.717
Coefficient of Variation - 616.6%
Maximum Deviation - 240.922



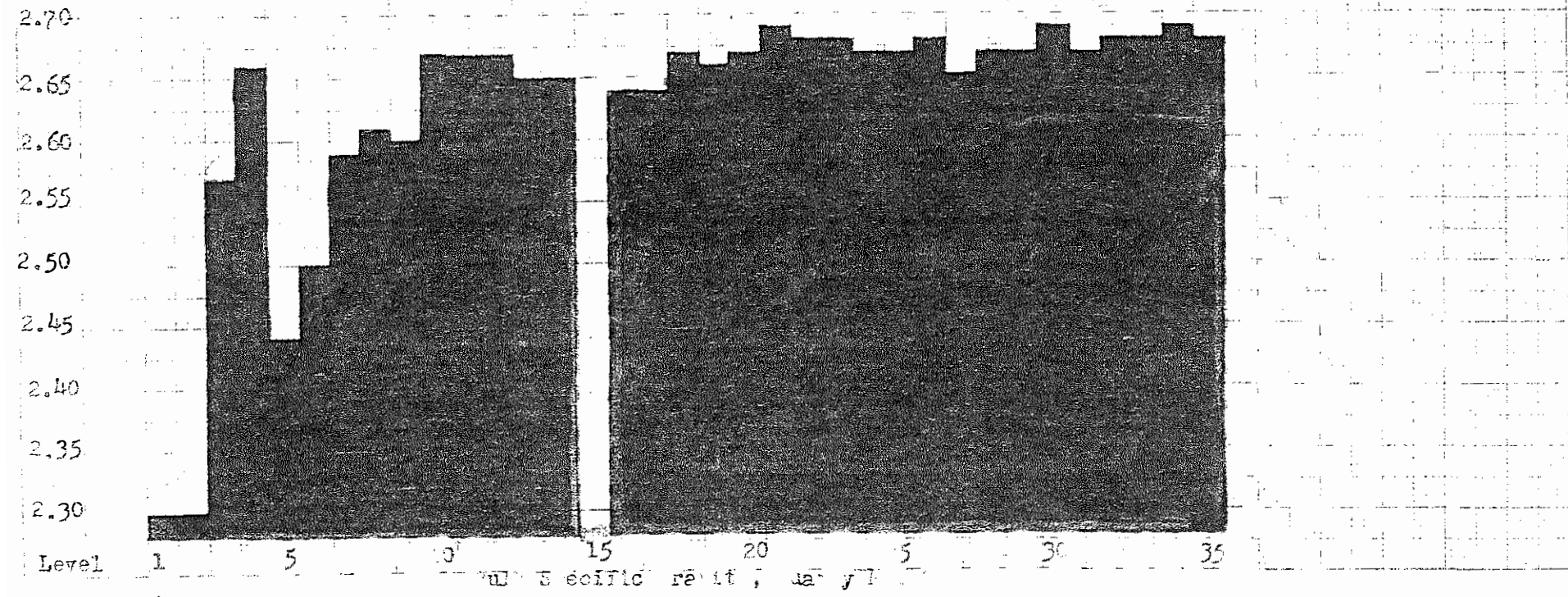
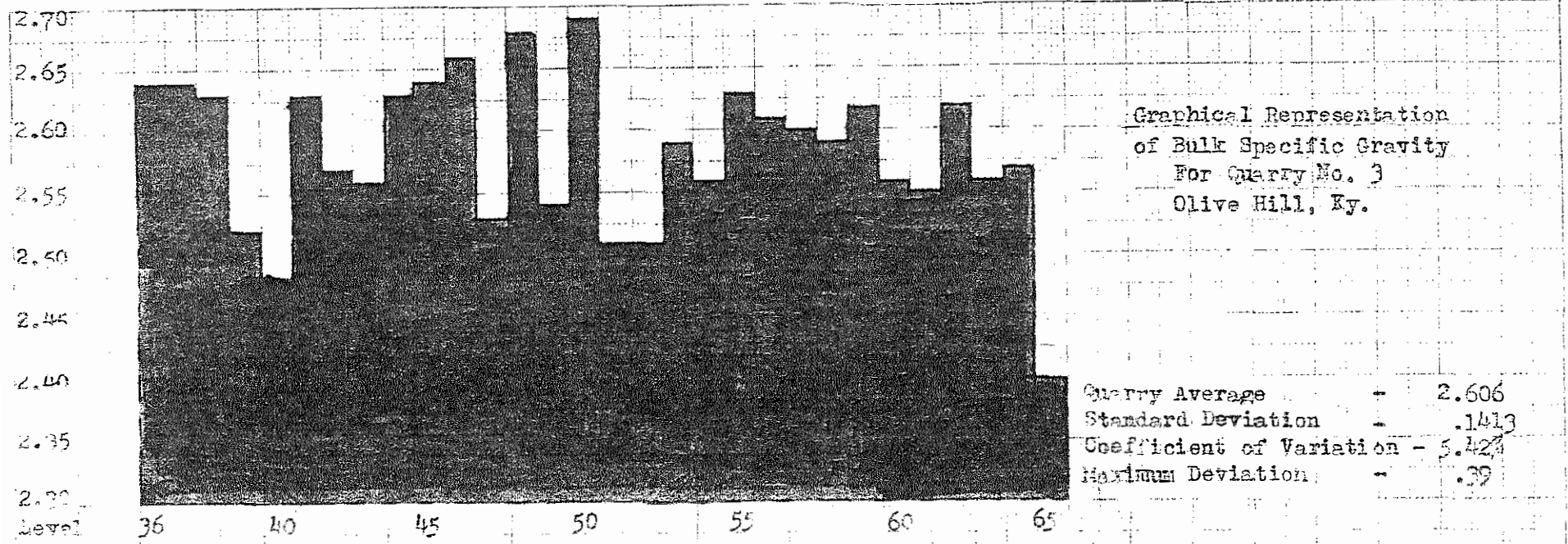
Permeability (in Millidarcys) - Quarry No. 2

Graphical Representation of
Bulk Specific Gravity, Quarry
No. 2, Mount Vernon, Ky.

Quarry Average - 2.642
Standard Deviation - .0395
Coefficient of Variation - 3.39%
Maximum Deviation - .41



Bulk Specific Gravity - Quarry No. 2



Bulk Specific Gravity, Quarry

Graphical Representation of
Permeability (in Millidarcys),
Quarry No. 3, Olive Hill, Ky.

Quarry Average - .1431
Standard Deviation - .340
Coefficient of Variation - 237.6%
Maximum Deviation - 1.741

1.5
1.0
0.5

36 40 45 50 55 60 65

1.5
1.0
0.5
0

1 5 10 15 20 25 30 35

Permeability (in millidarcys) - Quarry No. 3

Graphical Representation of
Absolute Specific Gravity
For Quarry No. 3,
Olive Hill, Ky.

2.85

2.80

2.75

2.70

2.65

2.60

35

40

45

50

55

60

65

Quarry Average	-	2.7294
Standard Deviation	-	.496
Coefficient of Variation	-	18.17%
Maximum Deviation	-	.1664

2.85

2.80

2.75

2.70

2.65

2.60

Level

1

5

10

15

20

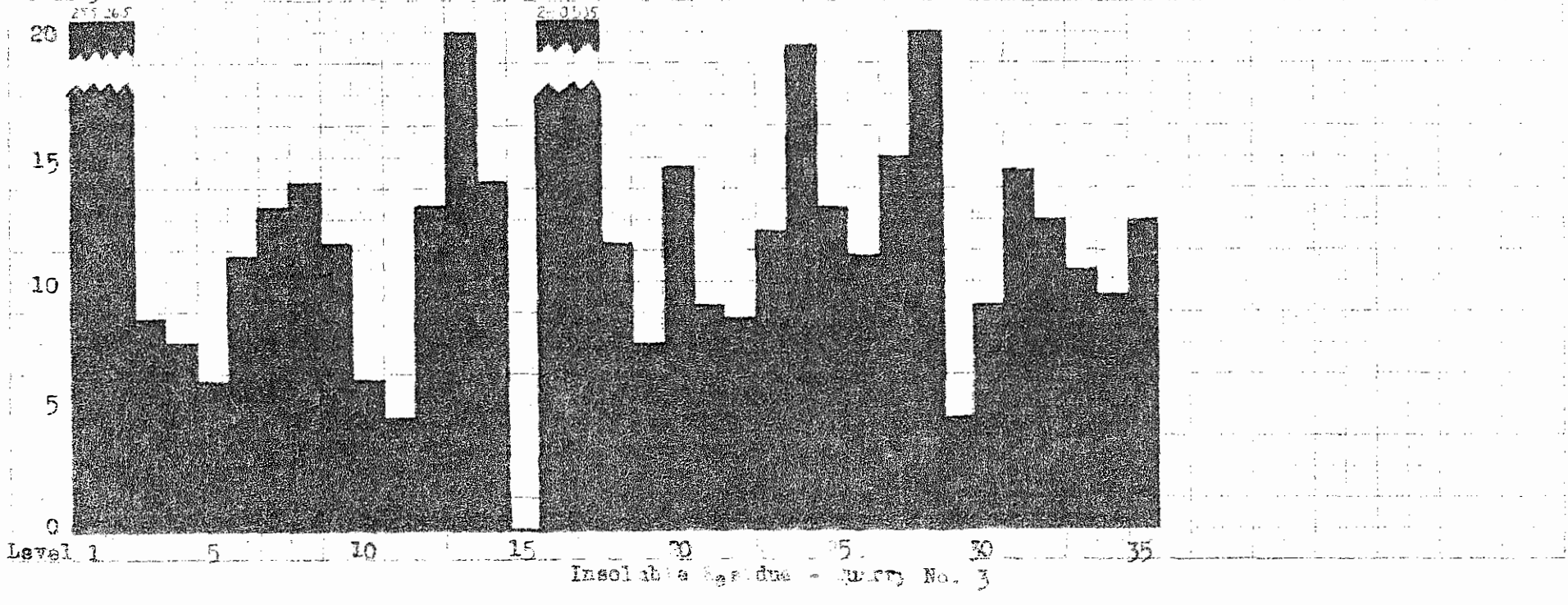
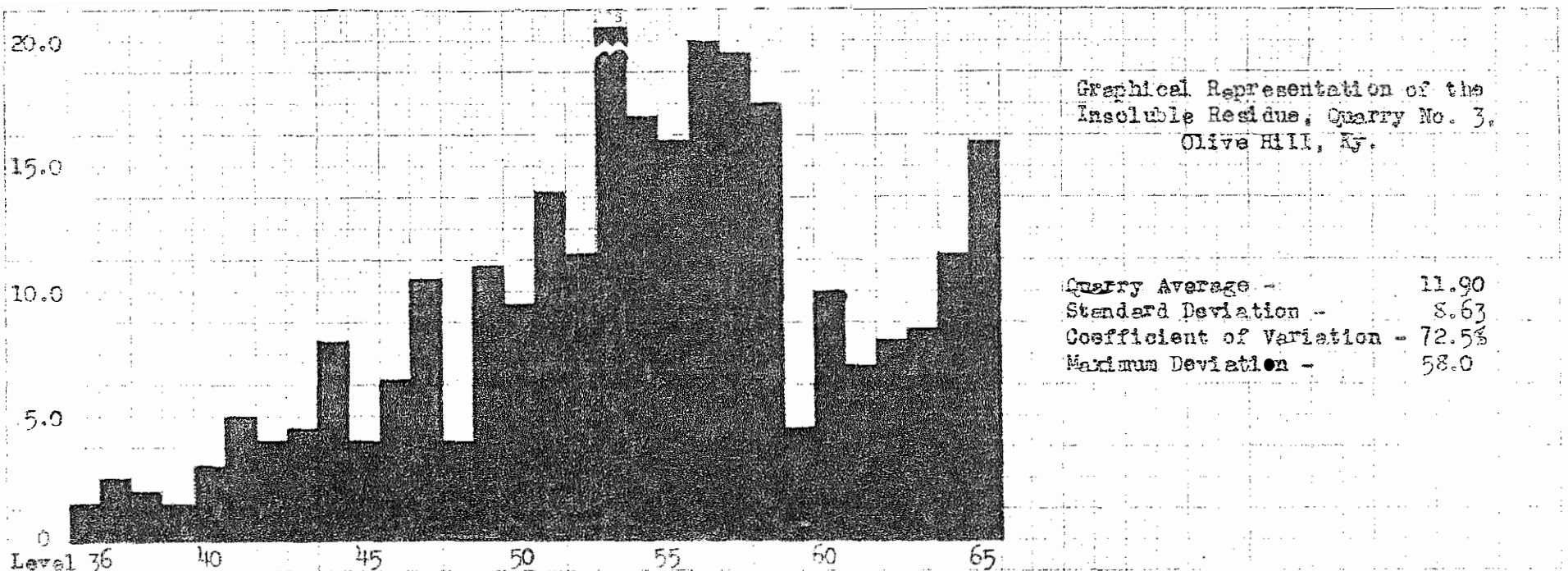
25

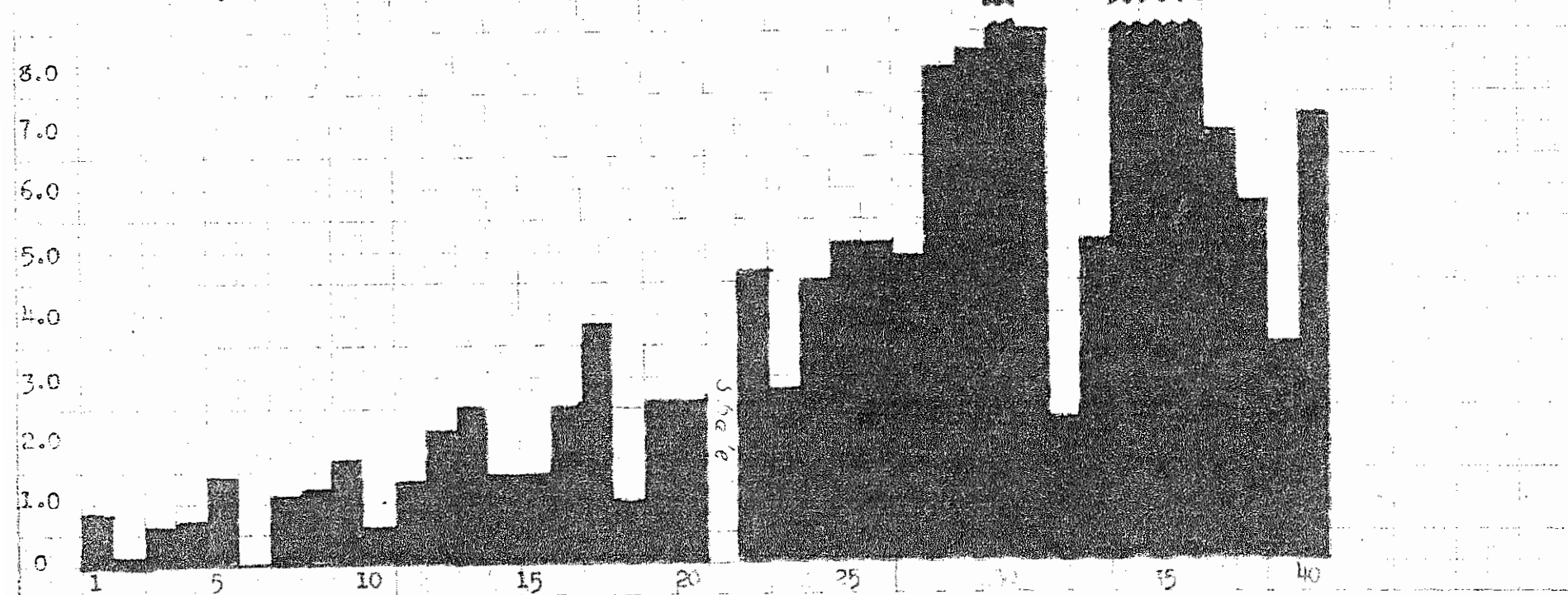
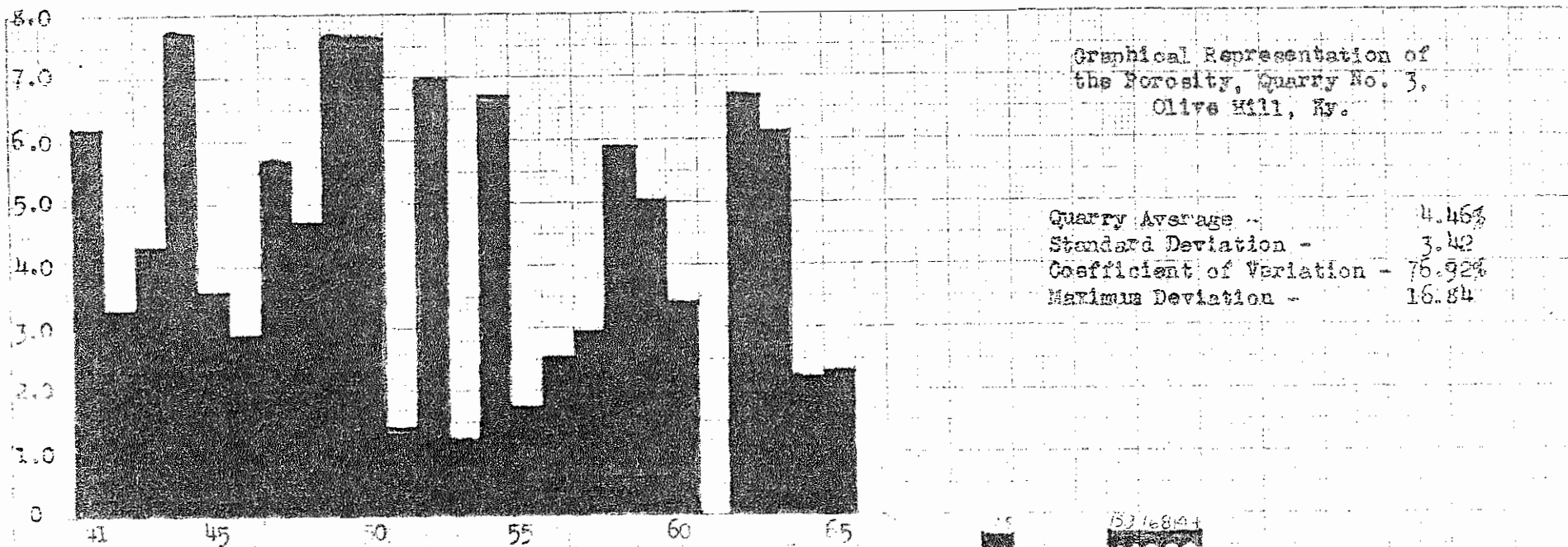
30

35

Absolute Specific Gravity, Quarry No. 3

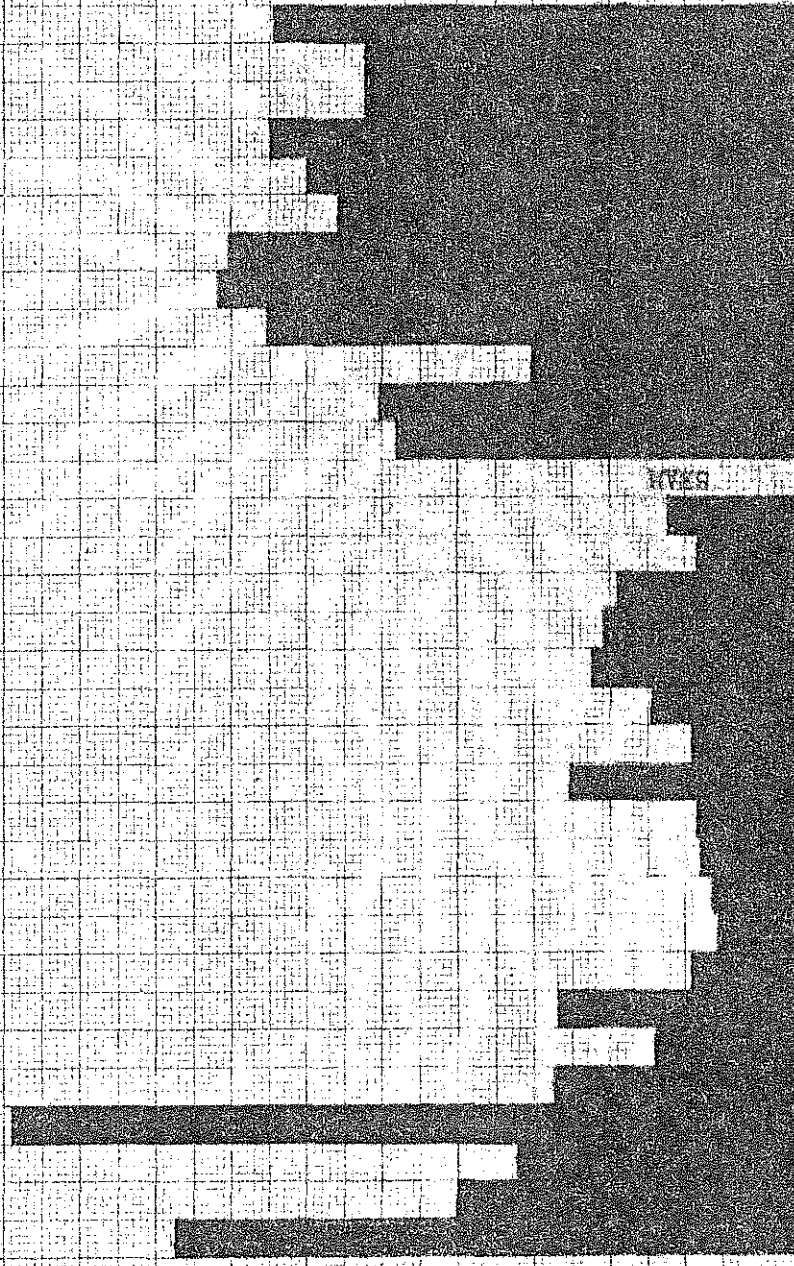
81





Graphical Representation
of the Insoluble Residue.
Quarry No. 4, Avoca, Ky.

11.0
10.0
9.0
8.0
7.0
6.0
5.0
4.0
3.0
2.0
1.0
0



Per Cent of Insoluble Residue

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33

Quarry Average - 4.1666
Standard Deviation - 2.542
Coefficient of Variation - 61.024
Maximum Deviation - 9.25

Graphical Representation of
 Porosity for Quarry No. 4,
 Avoca, Ky.

14.0
 13.0
 12.0
 11.0
 10.0
 9.0
 8.0
 7.0
 6.0
 5.0
 4.0
 3.0
 2.0
 1.0
 0

Quarry Average - 5.861
 Standard Deviation - 2.447
 Coefficient of Variation - 41.73%
 Maximum Deviation - 11.86%

0 5 10 15 20 25 30 33

SEAM

Porosity - Quarry No. 4

84

