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ABSTRACT

SIMULATION OF ELECTRONIC AND OPTICAL PROPERTIES OF GRAPHENE

by Biao Leng

Graphene is a recently discovered two-dimensional crystal. Due to its excellent electronic properties, transport properties, optical properties, and many other features, it has tremendous potential for applications in many areas. This thesis discusses the structure and properties of graphene using several different models of graphene and carries out detailed theoretical studies and calculations of its electronic and optical properties. Using two modules of Materials Studio, CASTAP and Doml³, four graphene models have been constructed. Their electronic and optical properties have also been calculated via these two modules. By comparing the results of calculations with experimental results and the literature, the influence of different structures of these models has been discussed.

SIMULATION OF ELECTRONIC AND OPTICAL PROPERTIES OF GRAPHENE

by Biao Leng

A Thesis Submitted to the Faculty of New Jersey Institute of Technology in Partial Fulfillment of the Requirements for the Degree of Master of Science in Materials Science and Engineering

Interdisciplinary Program in Materials Science and Engineering



APPROVAL PAGE

SIMULATION OF ELECTRONIC AND OPTICAL PROPERTIES OF GRAPHENE

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This thesis	is dedicated to	it was their su	pport that gave	e me the abili	ty to

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CHAPTER 1

INTRODUCTION

1.1 Objective

Since the discovery of graphene, it has been intensely researched in recent decades due to its unique structure and properties. Enormous amounts of theoretical and experimental work has been done. However, in order to study its properties, we have to synthesize graphene first, and it is expensive. Even after more than 10 years of developing different synthesis methods, the cost of making 1g of graphene can still be as expensive as \$150 or more. In order to not to waste this expensive material, a necessary and low-cost method to study the target material is needed, which is simulation and modeling. By simulating the target material utilizing appropriate software, we can build its model first and study its properties before we synthesize it so that we can test the structure and reduce costs in processing and subsequently manufacturing. We can also compare the outcomes of simulations and the results of experiments to see the similarities and differences between theoretical and experimental work. Materials Studio 7.0 has been used as simulation software in this work to study the electronic and optical properties of graphene with different structures.

1.2 Structure of Graphene

Graphene has a two-dimensional crystalline structure. It is an allotrope of carbon. Carbon atoms are densely packed in a regular sp2-bonded atomic-scale hexagonal pattern in graphene. Graphene can be described as a one-atom thick layer of graphite. [1]. It is the basic structural element of other allotropes, including graphite, fullerenes and carbon nanotubes. Perfect pure graphene is strong, light, nearly transparent and an excellent conductor of heat and electricity. Graphene has significant stability due to its tightly packed carbon atoms and a sp2 orbital hybridization – a combination of orbitals s, p_x and p_y that constitute the σ -bond. The final p_z electron makes up the π -bond. The excellent electrical properties of graphene are due to the half-filled band that permits free-moving electrons. Each atom has four bonds, one σ bond with each of its three neighbors and one π -bond that is oriented out of plane. The atoms are about 1.42 Å apart [1].

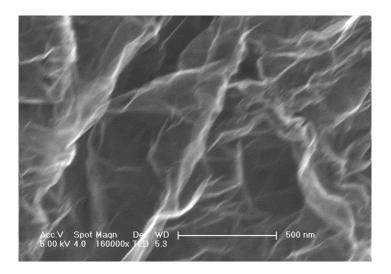


Figure 1.1 SEM micrograph of monolayer graphene.

Source: Factory, A.M.-G. Single Layer Graphene (Graphene Factory). 2015 [cited 2015 April]; Available from: http://www.acsmaterial.com/product.asp?cid=25&id=137 [2].

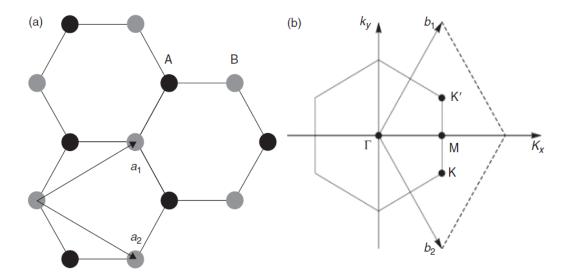


Figure 1.2 A honeycomb lattice of graphene, sub-lattices A and B are shown as black and grey. (b) Reciprocal lattice vectors and some special points in the Brillouin zone. Source: Katsnelson, M. I., & Katsnel'son, M. I. (2012). Graphene: carbon in two dimensions. Cambridge University Press [3].

Graphene has a honeycomb crystal lattice as shown in Figure 1.2 (a); it has two atoms in each elementary cell which belong to two sublattices A and B.

$$\overrightarrow{a1} = \frac{a}{2}(3,\sqrt{3}), \qquad \overrightarrow{a2} = \frac{a}{2}(3,-\sqrt{3})$$
 (1.1)

where, $a \approx 1.42 \text{Å}$ is the nearest distance between carbon atoms in the honeycomb lattice. It corresponds to a conjugated carbon-carbon bond.

As for the reciprocal lattice in Brillouin zone,

$$\overrightarrow{b1} = \frac{2\pi}{3a}(1,\sqrt{3}), \qquad \overrightarrow{b2} = \frac{2\pi}{3a}(1,-\sqrt{3})$$
 (1.2)

where, $a \approx 1.42 \mbox{\normalfont\AA}$ is the nearest distance between carbon atoms in the honeycomb lattice.

1.3 Electronic Properties of Graphene

The band gap for perfect graphene is zero, because its conduction and valence bands meet at the Dirac points at the Fermi level. The Dirac points are six locations in momentum space, on the edge of the Brillouin zone; they can be divided into two non-equivalent sets of three points- K and K'.

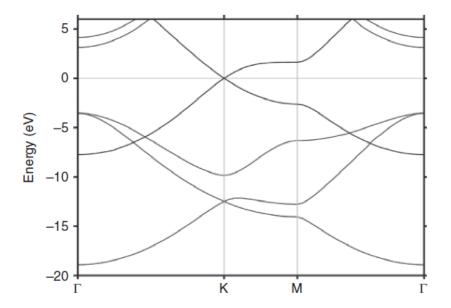


Figure 1.3 Band structure of graphene.

Source: Katsnelson, M. I., & Katsnel'son, M. I. (2012). Graphene: Carbon in two dimensions. Cambridge University Press [3].

In momentum space, the positon of K and K' are given by:

$$K = (\frac{2\pi}{3a}, \frac{2\pi}{3\sqrt{3}a}), \qquad K' = (\frac{2\pi}{3a}, -\frac{2\pi}{3\sqrt{3}a})$$
 (1.3)

In real space, the 3 nearest vectors are:

$$\delta_1 = \frac{a}{2}(1,\sqrt{3}), \qquad \delta_2 = \frac{a}{2}(1,-\sqrt{3}), \qquad \delta_3 = -a(1,0)$$
 (1.4)

where, $a \approx 1.42 \text{Å}$ is the nearest distance between carbon atoms in the honeycomb lattice.

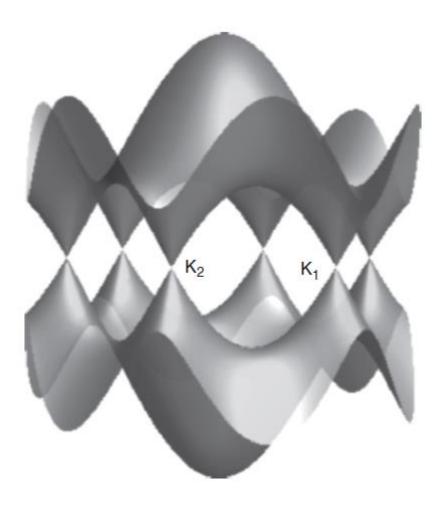


Figure 1.4 Electron energy spectrum of graphene in the nearest-neighbor approximation.

Source: Katsnelson, M. I., & Katsnel'son, M. I. (2012). Graphene: carbon in two dimensions. Cambridge University Press [3].

At the near-neighbor of K and K', the electrons' linear dispersion relation is

$$E = \hbar v_F \sqrt{k_x^2 + k_y^2} \tag{1.5}$$

where the wavevector k is measured from the Dirac points[4].

Electron transfer measurement results show that, at room temperature, graphene has surprisingly high electron mobility. Its value exceeds

15,000cm² V⁻¹S⁻¹. Symmetry from the measured conductivity data obtained show that the mobility of holes and electrons should be equal. Between 10 K and 100 K [5], the mobility is almost independent of temperature, and may be subject to scattering defects within a graphene lattice. At room temperature and carrier density of 10¹²/cm², the scattering in graphene is due to phonon scattering. The mobility of charge carriers in graphene is 200,000cm² V⁻¹S⁻¹. This value corresponds to a resistivity of $10^{-6} \Omega \cdot \text{cm}$, slightly less than the resistivity of silver ~ $1.59 \times 10^{-6} \Omega \cdot$ cm. At room temperature, the lowest resistivity material is silver [6]. Therefore, graphene is an excellent conductor. For graphene sheet on SiO₂, the scattering of phonons in graphene is relatively larger than in SiO₂, the upper limit of which leads to a mobility of 40,000cm² V⁻¹S⁻¹ [7]. Due to the chemical dopant in graphene, the carrier mobility might be affected; so the experiment can detect the degree of influence. Experimentalists, with the option of using a variety of gas molecules (some donor; some acceptor) incorporated in graphene, have found that even when the chemical dopant concentration exceeds $10^{12}/\text{cm}^2$, the carrier mobility changes slightly. Due to the two-dimensional nature of graphene, scientists believe that the charge fraction of the apparent charge (apparent charge in a low-dimensional material is less than the quasi-particle quantum units) will occur in graphene. Therefore, graphene may be a suitable material for any desired quantum computer subcomponents [8].

1.4 Optical Properties of Graphene

Graphene sheet has a unique ability to absorb a rather large 2.3% of white light, especially considering that it is only 1 atom thick. This is due to its aforementioned electronic properties; the electrons act like massless charge carriers with very high mobility. Not long ago, it has been proved that the amount of white light absorbed is based on the Fine Structure Constant, rather than being dictated by material specifics. Adding another layer of graphene increases the amount of white light absorbed by approximately the same value (2.3%). Graphene's opacity of $\pi\alpha \approx 2.3\%$ equates to a universal dynamic conductivity value of $G = 2e^2/4\hbar$ ($\pm 2-3\%$) over the visible frequency range.

Due to these impressive characteristics, it has been observed that once optical intensity reaches a certain threshold (known as the saturation fluence), saturable absorption takes place (very high intensity light causes a reduction in absorption) [9].

CHAPTER 2

THEORY AND BACKGROUND

2.1 Density functional theory

Density functional theory (DFT) is a computational quantum mechanical modeling method used in physics, chemistry and materials science to investigate molecules and electronic structure of many-body systems. Originally, density functional theory of quantum systems was proposed in 1927 by Thomas and Fermi. It was not accurate enough at that time. However, their approach illustrates the way how density functional theory works. Although density functional theory has its conceptual roots in the Thomas–Fermi model, DFT was not considered as a firm theory until the two Hohenberg–Kohn theorems (H–K) [10].

Hohenberg–Kohn Theorems

1. If there are two systems of electrons, one trapped in a potential $v_1(\vec{r})$ and the other in $v_2(\vec{r})$, with both of them having the same ground-state density $n(\vec{r})$, then necessarily,

$$v_1(\vec{r}) - v_2(\vec{r}) = const.$$
 (2.1)

In other words, the potential and all the properties of the system are mainly determined by the ground state density uniquely including the many-body wave function.

In particular, the "HK" functional, defined as F[n] = T[n] + U[n] is a universal functional of the density (not depending explicitly on the external potential) [11].

2. For any positive integer N and potential $v(\vec{r})$, a density functional F[n] exists such that,

$$E_{(v,N)}[n] = F[n] + \int v(\vec{r})n(\vec{r})d^3r$$
(2.2)

obtains its minimal value at the ground-state density of N electrons in the potential $v(\vec{r})$. The minimal value of $E_{(v,N)}[n]$ is then the ground state energy of this system [11].

The first H–K theorem demonstrates that the ground state properties of a many-electron system are uniquely determined by an electron density that depends on only three spatial coordinates. It lays the groundwork for reducing the many-body problem of N electrons with 3N spatial coordinates to three spatial coordinates, through the use of functionals of the electron density. This theorem can be extended to the time-dependent domain to develop time-dependent density functional theory (TDDFT), which can be used to describe excited states [10].

The second H–K theorem defines an energy functional for the system and proves that the correct ground state electron density minimizes this energy functional.

Within the framework of Kohn-Sham DFT (KS DFT), the intractable many-body problem of interacting electrons in a static external potential is reduced to a tractable problem of non-interacting electrons moving in an effective potential. The effective potential includes the external potential and the

effects interactions between of the Coulomb the electrons, e.g., the exchange and correlation interactions. Modeling the latter two interactions becomes the difficulty within the KS DFT. The simplest approximation is the local-density approximation (LDA), which is based on exact exchange energy for a uniform electron gas, which can be obtained from the Thomas-Fermi model, and from fits to the correlation energy for a uniform electron gas. Non-interacting systems are relatively easy to solve as the wave function can be represented as a Slater determinant of orbitals. Further, the kinetic energy functional of such a system is known exactly. The exchange-correlation part of the total-energy functional remains unknown and must be approximated.

Another approach, less popular than the KS DFT but arguably more closely related to the spirit of the original H-K theorems, is the orbital-free density functional theory (OFDFT), in which approximate functionals are also used for the kinetic energy of the non-interacting system [12].

2.2 Kohn-Sham Equations

In physics and quantum chemistry, specifically density functional theory, the Kohn–Sham equation is the Schrödinger equation of a fictitious system (the "Kohn–Sham system") of non-interacting particles (typically electrons) that generate the same density as any given system of interacting particles. The Kohn–Sham equation is defined by a local effective (fictitious) external potential in which the non-interacting particles move, typically denoted as $v_s(r)$ or $v_{eff}(r)$, called the Kohn–Sham potential. As the particles in the Kohn–Sham system are non-interacting Fermions, the Kohn–Sham wave function is a single Slater determinant constructed from a set of orbitals that are the lowest energy solutions to:

$$\left(-\frac{\hbar^2}{2m}\nabla^2 + v_{\text{eff}}(\mathbf{r})\right)\phi_i(\mathbf{r}) = \varepsilon_i\phi_i(\mathbf{r})$$
(2.3)

This eigenvalue equation is the typical representation of the Kohn–Sham equations. Here, ϵ_i is the orbital energy of the corresponding Kohn–Sham orbital, ϕ_i , and the density for an N-particle system is:

$$\rho(\mathbf{r}) = \sum_{i}^{N} |\phi_i(\mathbf{r})|^2. \tag{2.4}$$

The Kohn-Sham equations are named after Walter Kohn and Lu Jeu Sham, who introduced the concept at the University of California, San Diego in 1965 [13, 14].

2.3 Local-Density Approximations (LDA)

Local-density approximations (LDA) are a class of approximations to the exchange-correlation (XC) energy functional in density functional theory (DFT) that depend solely on the value of the electronic density at each point in space (and not, for example, derivatives of the density or the Kohn–Sham orbitals). Many approaches can yield local approximations to the XC energy. However, overwhelmingly successful local approximations are those that have been derived from the homogeneous electron gas (HEG) model. In this regard, LDA is generally synonymous with functionals based on the HEG approximation, which are then applied to realistic systems (molecules and solids).

In general, for a spin-unpolarized system, a local-density approximation for the exchange-correlation energy is written as:

$$E_{xc}^{\text{LDA}}[\rho] = \int \rho(\mathbf{r}) \epsilon_{xc}(\rho) \, d\mathbf{r} , \qquad (2.5)$$

where, ρ is the electronic density and ϵ_{xc} is the exchange-correlation energy per particle of a homogeneous electron gas of charge density ρ . The exchange-correlation energy is decomposed into exchange and correlation terms linearly,

$$E_{xc} = E_x + E_c {,} {(2.6)}$$

so that separate expressions for E_x and E_c are sought. The exchange term takes on a simple analytic form for the HEG [15]. Only limiting expressions for the correlation density are known exactly, leading to numerous different approximations for ϵ_{xc} .

Local-density approximations are important in the construction of more sophisticated approximations to the exchange-correlation energy, such as generalized gradient approximations or hybrid functionals. A desirable property of any approximate exchange-correlation functional is that it reproduce the exact results of the HEG for non-varying densities. As such, LDA's are often an explicit component of such functionals.

2.4 Generalized Gradient Approximations (GGA)

Generalized gradient approximations (GGA) are still local but also take into account the gradient of the density at the same coordinate:

$$E_{XC}^{GGA}[n_{\uparrow}, n_{\downarrow}] = \int \epsilon_{XC}(n_{\uparrow}, n_{\downarrow}, \vec{\nabla}n_{\uparrow}, \vec{\nabla}n_{\downarrow}) n(\vec{r}) d^{3}r.$$
(2.7)

Using the latter (GGA), very good results for molecular geometries and ground-state energies have been achieved [16, 17].

Potentially more accurate than the GGA functionals (Perdew, Burke et al. 1996) are the meta-GGA functionals, a natural development after the GGA (generalized gradient approximation). Meta-GGA DFT functional, in its original form, includes the second derivative of the electron density (the Laplacian) whereas GGA includes only the density and its first derivative in the exchange-correlation potential [18].

Functionals of this type are, for example, TPSS and the Minnesota Functionals. These functionals include a further term in the expansion, depending on the density, the gradient of the density and the Laplacian (second derivative) of the density.

Difficulties in expressing the exchange part of the energy can be relieved by including a component of the exact exchange energy calculated from Hartree–Fock theory. Functionals of this type are known as hybrid functionals.

CHAPTER 3

MATERIALS STUDIO

3.1 Introduction

Materials Studio is a software platform for simulating and modeling materials. It is developed and distributed by Accelrys. This software is used in advanced research of various materials, such as polymers, nanotubes, catalysts, metals, ceramics, and so on, by universities, research centers, and high-technology companies.

Materials Studio is a client–server software package with Microsoft Windows-based PC clients and Windows and Linux-based servers running on PCs, Linux IA-64workstations (including Silicon Graphics (SGI) Altix) and HP XC clusters. There are many modules in Materials Studio. We will now introduce two modules used in this research.

3.2 Materials Studio CASTEP

CASTEP (Cambridge Sequential Total Energy Package) is an ab initio quantum mechanical program employing density functional theory (DFT) to simulate the properties of solids, interfaces, and surfaces for a wide range of materials classes such as ceramics, semiconductors, and metals. First principle calculations allow researchers to investigate the nature and origin of the electronic, optical, and structural properties of a system without the need for any experimental input. In that case, Materials Studio CASTEP is the perfect simulation method to research problems in solid state physics, materials science, chemistry, and chemical engineering where empirical models and experimental data are lacking. With the help of CASTEP, researchers can save tremendous of time and costly experiments. CASTEP is capable of computing many electronic, optical, physical properties. In particular, it can predict the electronic properties such as band gaps, density of states and Schottky barriers; optical properties such as reflectivity, absorption, IR spectra, and dielectric functions; or physical properties such as elastic constants and so on.

A total energy plane wave pseudopotential method has been used by Materials Studio CASTEP. In order to reduce the complexity of calculation, core electrons was replaced by effective potentials which acting only on the valence electrons in the system. Electronic wave functions were expanded through a plane-wave basis set, and the local density (LDA) or generalized gradient (GGA) approximations were used to calculate the interaction, exchange and correlation

effects of electrons in the system. Due to the use of pseudopotentials and plane wave basis sets the geometry optimizations of molecules, solids, surfaces, and interfaces are efficient [19].

3.3 Materials Studio DMoL³

Materials Studio DMoL³ is one of fastest *ab initio* modeling program that uses density functional theory (DFT) to simulate chemical processes and predict properties of materials. It can predict processes in gas phase, solution, and solid environments, which lead to its great capability to research problems in chemistry, pharmaceuticals, materials science, and chemical engineering, as well as solid state physics.

Using numerical functions on an atom-centered grid as its atomic basis, Materials Studio DMoL³ achieves its speed and accuracy. By solving the DFT equations for individual atoms the atomic basis functions are obtained. The high quality of these basis sets minimizes basis set superposition effects and allows an improved description of molecular polarizabilities. The electron density in Materials Studio DMoL³ is expanded in terms of multipolar, atomic-centered partial densities. It provides a compact yet highly accurate representation of the density, and allows for a good scaling with growing system size. Thus, the evaluation of the Coulomb potential and Hamiltonian matrix elements scales linearly with the size of the system. Both all Electron and pseudo-potential calculations can be performed in Materials Studio DMoL³. Accurate DFT semi-local pseudo-potentials (DSPP) or the more conventional Effective Core Potentials (ECP) can be used. Geometry and transition state optimizations are performed using delocalized internal coordinates, both for molecular as well as for periodic calculations. This includes the ability to impose Cartesian geometry constraints while performing the optimization in internal

coordinates. A transition state search scheme has been applied which is based on a combination of traditional LST/ QST methods. This new robust and fast scheme allows rapid location of transition states. Solvent effects are included using the COSMO model to simulate a continuum [20].

CHAPTER 4

SIMULATION METHOD AND PROCESS

Initially, we created a new project and imported the graphite structure into it.

A graphite structure was constructed by going to File>Import, then selecting Structures>Ceramics>graphite.

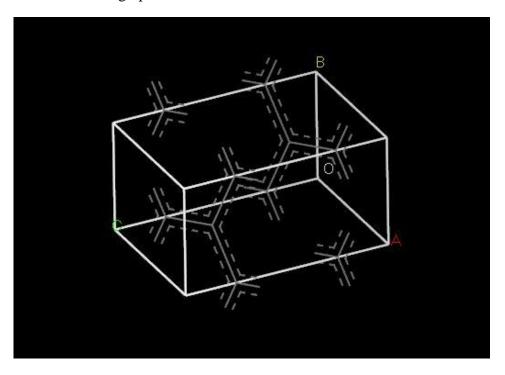


Figure 4.1 Graphite model.

In order to create the graphene sheet structure, some adjustment was needed. First, we utilized Build>Symmetry>Make P1 and then deleted one layer of graphite. This step was to break the symmetry of these two layers so that we could delete one layer without deleting the other layer; so we could build a single layer of graphite or graphene.

Afterwards, we added symmetry back for later calculations. Build>Symmetry>Find symmetry > impose symmetry.

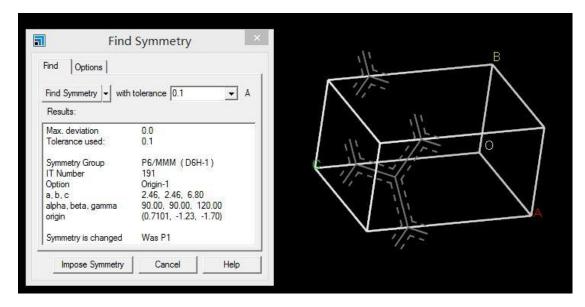


Figure 4.2 Adding symmetry back to the model.

Later, we accessed the honeycomb lattice primitive cell. In order to build the orthogonal graphene sheet, we made the primitive cell 2×2 via Build>Symmetry>supercell. We adjusted its lattice parameters as follows.

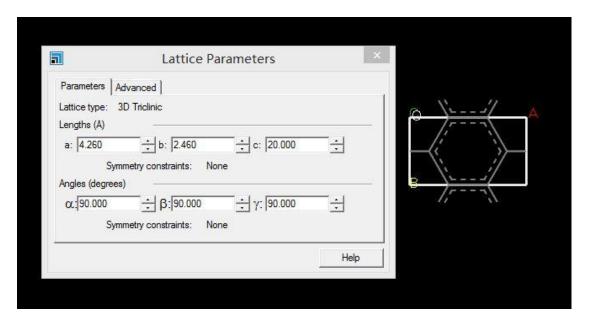


Figure 4.3 Adjustment of lattice parameters.

We made the primitive cell into a graphene sheet using the Supercell option and creating a 4×4 sturcture. Then we selected the whole structure and went to Modify>Modify bond type>partial double bond. Figure 4.4 presents our graphene sheet.

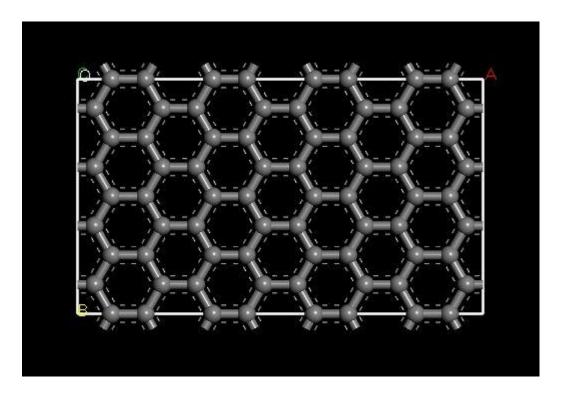


Figure 4.4 Graphene model armchair orientation.

In order to investigate the potential influence on graphene's electronic and optical properties introduced by the defects and number of layers, four models have been created using the same method each time: monolayer graphene, monolayer graphene with defect (containing a single missing carbon atom in the middle for reducing complexity), bilayer graphene and 3-layer-graphene. The calculation setup and results are presented in the next chapter.

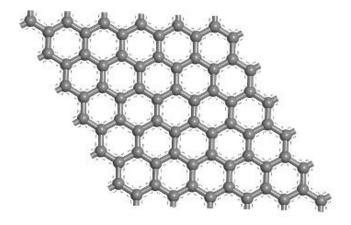


Figure 4.5 Monolayer graphene.

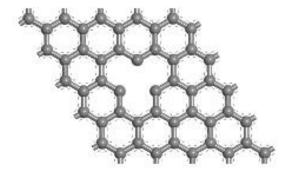


Figure 4.6 Monolayer graphene with defect.

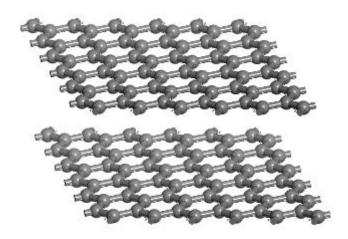


Figure 4.7 Bilayer graphene.

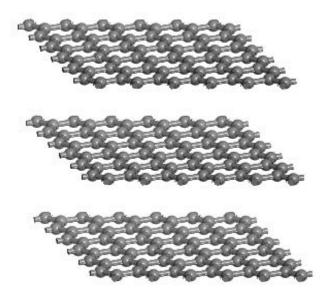


Figure 4.8 3-layer graphene.

CHAPTER 5

RESULTS AND DISCUSSION

The density functional theory (DFT) calculations of energy of these models were performed with Materials Studio CASTEP using norm conserving pseudopotential, plane-wave basis and periodic boundary conditions. The Generalized gradient approximations (GGA) with PW91 functional and a 750 eV cutoff energy for the plane-wave basis set were used in all relaxation processes. The k-point was set to $3 \times 3 \times 1$ for the Brillouin zone integration. The Brillouin zone path was set as G (0.000, 0.000, 0.000) \rightarrow K (-0.333, 0.667, 0.000) \rightarrow M (0.000, 0.500, 0.000) \rightarrow G (0.000, 0.000, 0.000).

The geometry optimization calculations for these models were also carried out via Materials Studio CASTEP modules using norm conserving pseudopotential, plane-wave basis and periodic boundary conditions. The Local density approximations (LDA) with CA-PZ functional and a 750 eV cutoff energy for the plane-wave basis set were used in all relaxation processes. The k-point and Brillouin zone path are the same as above. The reflectivity, absorption and Raman spectrum were calculated in this process.

The calculation results are shown as follows; the band structure of monolayer graphene is approximately linear at or near the Fermi level. The conduction band and valence band intersect with each other at the Fermi level (Figure 5.1) shown as zero band gap which agrees with the theoretical results.

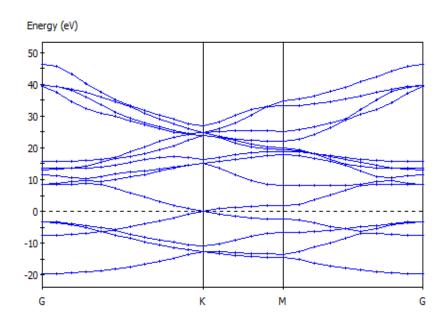


Figure 5.1 Band structure of monolayer graphene.

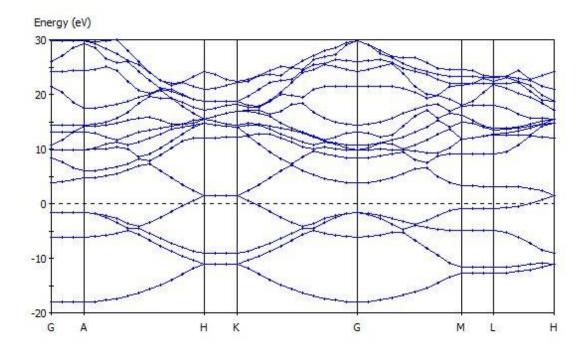


Figure 5.2 Band structure of bilayer and 3-layer graphene.

The band structure of bilayer and 3-layer graphene shows different results from monolayer graphene; it is approximately linear at near Fermi level but the Fermi

level itself has increased by about 1eV. The conduction band and valence band intersect with each other near the Fermi level shown as zero band gap (Figure 5.3). According to our analysis, it should be due to the influence of the other layer of graphene and the orientation.

The density of states describes the number of states per interval of energy at each energy level that are available to be occupied. It is consistent with the band structure. From Figure 5.3, the partial density of states of monolayer graphene, we can see that DOS of monolayer graphene at the Fermi level is very low, about 0, which is consistent with that obtained from the band structure at Fermi level (also 0).

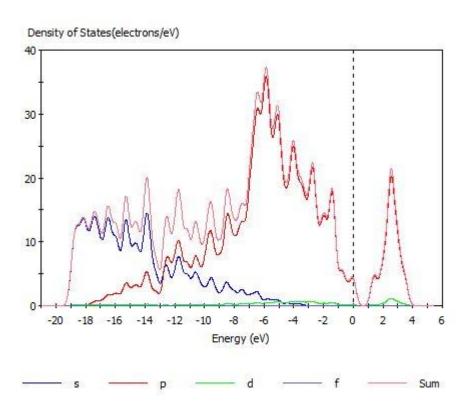


Figure 5.3 The partial density of states of monolayer graphene.

At or near the Fermi level (Figure 5.3), both the DOS and the 2p electrons show some significant peaks, but s electrons almost have no peak at all which means the DOS of graphene is mainly determined by 2p electrons.

For the density of states, as shown in Figures 5.4 and 5.5, bilayer and 3-layer graphene models shows similar results as the sum (of the density of states) in Figure 5.3.

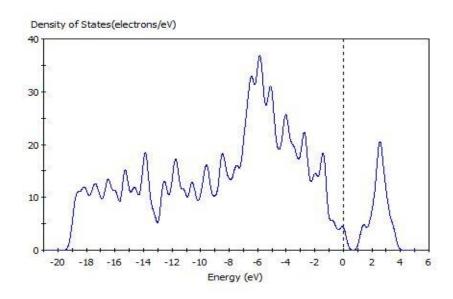


Figure 5.4 Density of states of bilayer graphene.

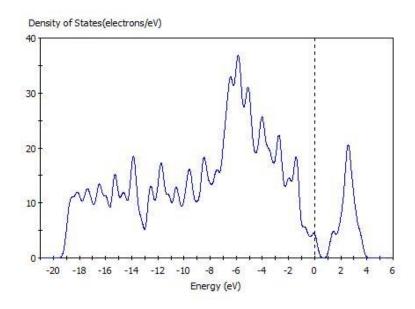


Figure 5.5 Density of state of 3-layer graphene.

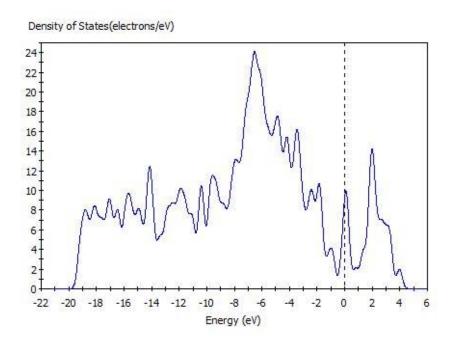


Figure 5.6 DOS of monolayer graphene with defect.

However, the results of the DOS of defect model are different from the above. There is a significant peak at the Fermi level as shown in Figure 5.6. This means the band gap, for the defect model, at the Fermi level is no longer 0, which means the absence of one atom in the graphene sheet has opened a band gap. This phenomenon has also been observed in earlier work [21].

The reflectivity and absorption showed no difference in monolayer, bilayer and 3-layer graphene models, which indicates that at the molecular level, the number of layers has little influence on the optical properties of graphene.

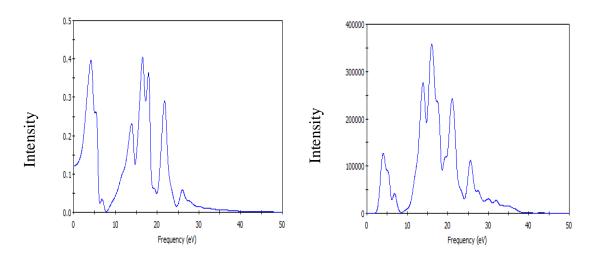


Figure 5.7 Reflectivity and absorption of monolayer graphene (from left to right).

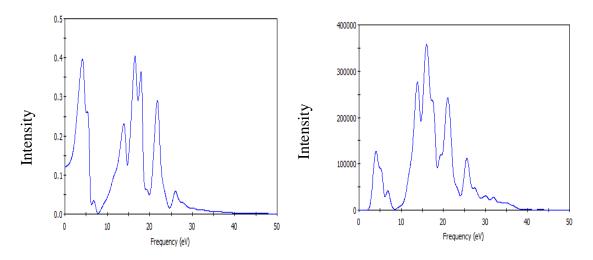


Figure 5.8 Reflectivity and absorption of bilayer graphene (from left to right).

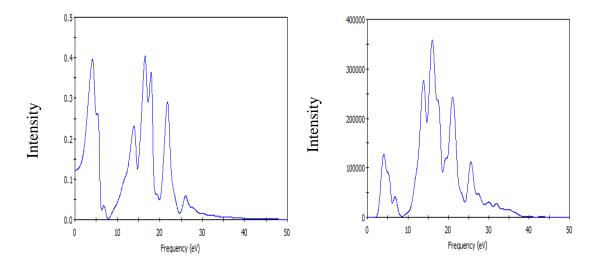
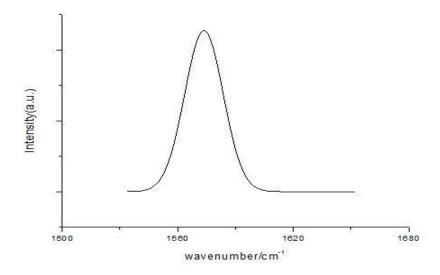


Figure 5.9 Reflectivity and absorption of 3-layer graphene (from left to right).



 $\textbf{Fig 5.10} \ \textbf{Raman spectrum of monolayer graphene}.$

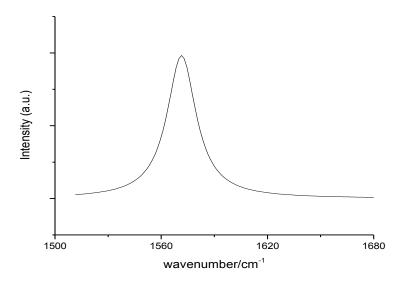


Fig 5.11 Raman spectrum of bilayer graphene.

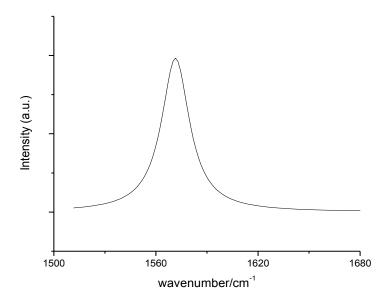


Fig 5.12 Raman spectrum of 3-layer graphene.

The results of the Raman spectrum of the three models is also similar; only one peak can be seen at around 1572.9 cm⁻¹ which is consistent with the experimental G peak wavenumber 1583cm⁻¹[22, 23].

CHAPTER 6

CONCLUSIONS

In summary, four different graphene models have been built in this work - monolayer graphene, monolayer graphene with defect, bilayer graphene and 3-layer-graphene. Calculations of the electronic properties and optical properties were performed with Materials Studio CASTEP and Doml³ modules. With different setting corresponding to energy and geometry optimization calculations, the band structure and density of states have been carried out to study the electronic properties of graphene and investigate of optical properties of different graphene models with a focus on reflectivity, absorption and Raman spectrum.

The band structure of monolayer graphene is approximately linear at or near Fermi level. The conduction band and valence band intersect with each other at the Fermi level and show zero band gap which is in agreement with the theoretical results. For bilayer and 3-layer graphene, the results are similar except the Fermi level has increased due to the influence of the electrons of other layer of graphene.

From density of states (DOS) of monolayer graphene, we can confirm that the major peak was mainly determined by 2p electrons. These peaks, at the DOS of bilayer and 3-layer graphene, show more significance and are distinguishable.

The reflectivity and absorption shown no difference in monolayer, bilayer and 3-layer graphene models. That indicates that, at the molecular level, the number of layers has little influence on the optical properties of graphene.

The Raman spectrum of 3 models has one peak at around 1572.9 cm⁻¹ which is about consistent with the experimental G peak wavenumber of 1583cm⁻¹.

Future work will focus on the influence of defects on the optical properties of graphene as well as the effects of electric and magnetic fields on the electronic and optical properties of graphene.

APPENDIX A

MONOLAYER GRAPHENE MODEL INPUT FILE

Task parameters

Calculate energy
Symmetry on
Max_memory 2048
File_usage smart

Electronic parameters

Spin_polarization restricted
Charge 0
Basis dnd
Pseudopotential none
Functional gga(p91)
Aux_density octupole
Integration_grid medium

Occupation thermal 0.0050
Cutoff_Global 3.3000 angstrom
Scf_density_convergence 1.0000e-005

Scf_charge_mixing 2.0000e-001

Scf_iterations 50

Scf_diis 6 pulay

Kpoint definition file (intervals/offset):

Kpoints file 1 1 2 0.0000 0.0000 0.0000

graphene.kpoints

Calculated properties

Band strucure input file

Calculate energy

Scf_iterations 0 use_old_density_and_keep on

Max_memory 2048
File_usage smart
empty_bands 12

Symmetry on

Electronic parameters

Spin_polarizationrestrictedCharge0BasisdndPseudopotentialnone

Functional gga(p91)
Aux_density octupole
Integration_grid medium

Occupation thermal 0.0050 Cutoff_Global 3.3000 angstrom

Kpoint definition file:

Kpoints file

graphene_BandStr.kpoints

plot_dos on

DOS input file

Calculate energy

Scf_iterations 0 use_old_density_and_keep on

Max_memory 2048
File_usage smart
empty_bands 12

Symmetry on

Electronic parameters

Spin_polarizationrestrictedCharge0BasisdndPseudopotentialnoneFunctionalgga(p91)

Aux_density octupole Integration_grid medium

Occupation thermal 0.0050 Cutoff_Global 3.3000 angstrom

Kpoint definition file:

Kpoints file 2 2 3 0.000000 0.000000 0.000000

graphene_DOS.kpoints

plot_dos on

Optics input file

comment: CASTEP calculation from Materials Studio

task: GeometryOptimization

xc_functional : PW91 spin_polarized : false opt_strategy : Default page_wvfns : 0

fine_grid_scale: 1.500000000000000

finite_basis_corr: 2 finite_basis_npoints: 3

elec_energy_tol: 2.0000000000000000e-006

max_scf_cycles: 100

fix_occupancy : true metals_method : dm mixing_scheme : Pulay

mix_history_length: 20

nextra_bands: 0

geom_max_iter: 100

geom_method : BFGS
fixed_npw : false

geom_modulus_est: 500.0000000000000 GPa

calculate_ELF : false
calculate_stress : true
popn_calculate : true
calculate_hirshfeld : true
calculate_densdiff : false

popn_bond_cutoff: 3.000000000000000

pdos_calculate_weights : false

num_dump_cycles: 0

task: Optics

continuation: default

spin_polarized : false

xc_functional: PW91

optics_nextra_bands: 12

calculate_stress : false
calculate_ELF : false
popn_calculate : false
calculate_hirshfeld : false
calculate_densdiff : false
pdos_calculate_weights : false

num_dump_cycles: 0

bs_write_eigenvalues: true

APPENDIX B

MONOLAYER GRAPHENE WITH DEFECT MODEL INPUT FILE

Task parameters

Calculate energy
Symmetry on
Max_memory 2048
File_usage smart

Electronic parameters

Spin_polarization restricted
Charge 0
Basis dnd
Pseudopotential none
Functional pwc
Harris off

Aux_density octupole Integration_grid medium

Occupation thermal 0.0050
Cutoff_Global 3.3000 angstrom
Scf_density_convergence 1.0000e-005
Scf_charge_mixing 2.0000e-001

Scf_iterations 50

Scf_diis 6 pulay

Kpoint definition file (intervals/offset):

Kpoints file 1 1 2 0.0000 0.0000 0.0000

graphite.kpoints

Calculated properties

Band structure input file

Calculate energy
Scf_iterations 0
use_old_density_and_keep on

Max_memory 2048
File_usage smart
empty_bands 12

Symmetry on

Electronic parameters

Spin_polarization restricted
Charge 0
Basis dnd
Pseudopotential none
Functional pwc
Harris off

Aux_density octupole Integration_grid medium

Occupation thermal 0.0050 Cutoff_Global 3.3000 angstrom

Kpoint definition file:

Kpoints file

graphite_BandStr.kpoints

plot_dos on

DOS input file

Calculate energy

Scf_iterations 0 use_old_density_and_keep on

Max_memory 2048 File_usage smart empty_bands 12

Symmetry on

Electronic parameters

Spin_polarizationrestrictedCharge0BasisdndPseudopotentialnoneFunctionalpwcHarrisoff

Aux_density octupole Integration_grid medium

Occupation thermal 0.0050 Cutoff_Global 3.3000 angstrom

Kpoint definition file:

Kpoints file 2 2 3 0.000000 0.000000 0.000000

$graphite_DOS.kpoints$

plot_dos on plot_pdos on

APPENDIX C

BILAYER GRAPHENE MODEL INPUT FILE

comment: CASTEP calculation from Materials Studio

task: GeometryOptimization

xc_functional : PW91 spin_polarized : false opt_strategy : Default page_wvfns : 0

fine_grid_scale: 1.500000000000000

finite_basis_corr: 2 finite_basis_npoints:

elec_energy_tol: 2.0000000000000000e-006

3

max_scf_cycles: 100

fix_occupancy : true metals_method : dm mixing_scheme : Pulay

mix_history_length: 20

nextra_bands: 0

geom_max_iter: 100

geom_method : BFGS
fixed_npw : false

geom_modulus_est: 500.0000000000000 GPa

calculate_ELF : false
calculate_stress : true
popn_calculate : false
calculate_hirshfeld : false
calculate_densdiff : false
pdos_calculate_weights : false

num_dump_cycles: 0

Band structure input file task : BandStructure continuation : default

spin_polarized : false

elec_energy_tol: 2.000000000000000e-006 cut_off_energy: 750.00000000000000

xc_functional: PW91

bs_nextra_bands: 12

calculate_stress: false
calculate_ELF: false
popn_calculate: false
calculate_hirshfeld: false
calculate_densdiff: false
pdos_calculate_weights: false

num_dump_cycles : 0
bs_write_eigenvalues : true

DOS input file

task : BandStructure continuation : default spin_polarized : false

elec_energy_tol: 2.000000000000000e-006 cut_off_energy: 750.00000000000000

xc_functional: PW91

bs_nextra_bands: 12

calculate_stress : false
calculate_ELF : false
popn_calculate : false
calculate_hirshfeld : false
calculate_densdiff : false
pdos_calculate_weights : true

num_dump_cycles : 0
bs_write_eigenvalues : true

Optics input file task : Optics

continuation : default
spin_polarized : false

elec_energy_tol: 2.000000000000000e-006 cut_off_energy: 750.00000000000000

xc_functional : PW91

optics_nextra_bands: 12

calculate_stress : false
calculate_ELF : false
popn_calculate : false
calculate_hirshfeld : false
calculate_densdiff : false
pdos_calculate_weights : false

num_dump_cycles: 0

bs_write_eigenvalues : true

APPENDIX D

3-LAYER GRAPHENE MODEL INPUT FILE

comment: CASTEP calculation from Materials Studio

task: GeometryOptimization

xc_functional : LDA spin_polarized : false opt_strategy : Default page_wvfns : 0

finite_basis_corr: 2 finite_basis_npoints: 3

elec_energy_tol: 2.0000000000000000e-006

max_scf_cycles: 100

fix_occupancy : true metals_method : dm mixing_scheme : Pulay

mix_history_length: 20

nextra_bands: 0

geom_max_iter: 100

geom_method : BFGS
fixed_npw : false

geom_modulus_est: 500.0000000000000 GPa

calculate_ELF: false calculate_stress: true popn_calculate: true calculate_hirshfeld: true calculate_densdiff: false

popn_bond_cutoff: 3.000000000000000

pdos_calculate_weights : false

num_dump_cycles: 0

Band structure input file task: BandStructure

continuation : default
spin_polarized : false

elec_energy_tol: 2.000000000000000e-006 cut_off_energy: 750.00000000000000

xc_functional : LDA

bs_nextra_bands: 12

calculate_stress : false
calculate_ELF : false
popn_calculate : false
calculate_hirshfeld : false
calculate_densdiff : false
pdos_calculate_weights : false

num_dump_cycles : 0 bs_write_eigenvalues : true

DOS input file

task : BandStructure continuation : default spin_polarized : false

elec_energy_tol: 2.00000000000000e-006 cut_off_energy: 750.00000000000000

xc_functional : LDA

bs_nextra_bands: 12

calculate_stress : false
calculate_ELF : false
popn_calculate : false
calculate_hirshfeld : false
calculate_densdiff : false
pdos_calculate_weights : true

num_dump_cycles : 0
bs_write_eigenvalues : true

Optics input file task : Optics

continuation : default
spin_polarized : false

elec_energy_tol: 2.000000000000000e-006 cut_off_energy: 750.000000000000000

xc_functional : LDA

optics_nextra_bands: 12

calculate_stress : false
calculate_ELF : false
popn_calculate : false
calculate_hirshfeld : false
calculate_densdiff : false

pdos_calculate_weights : false

num_dump_cycles: 0

bs_write_eigenvalues : true

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