




# Assessment of metal and metalloid contamination in soils through compositional data: the old Mortórios uranium mine area, central Portugal

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**Abstract** Soils from the old Mortórios uranium mine area were studied to look for contamination, as they are close to two villages, up to 3 km away, and used for agriculture. They are mainly contaminated in U and As and constitute an ecological threat. This study attempts to outline the degree to which soils have been affected by the old mining activities through the computation of significant hot clusters. Traditional geostatistical approaches commonly use raw data (concentrations) accepting that the analyzed elements represent the soil's entirety. However, in geochemical studies these elements are just a fraction of the total soil composition. Thus, considering compositional

data is pivotal. The spatial characterization, considering raw and compositional data together, allowed a broad discussion about not only the concentrations' spatial distribution, but also a better understanding on the possibility of trends of “relative enrichment” and, furthermore an insight in U and As fate. The highest proportions (compositional data) on U (up to 33%), As (up to 35%) and Th (up to 13%) are reached in the south-southeast segment. However, the highest concentrations (raw data) occur in north and northwest of the studied area, pointing out to a “relative enrichment” toward the south-southeast zone. The Mondego Sul area is mainly contaminated in U and As, but also in Co, Cu, Pb and Sb. The Mortórios area is less contaminated than the Mondego Sul area.

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A. M. R. Neiva: This paper is dedicated to the memory of our friend, Prof. Ana Neiva.

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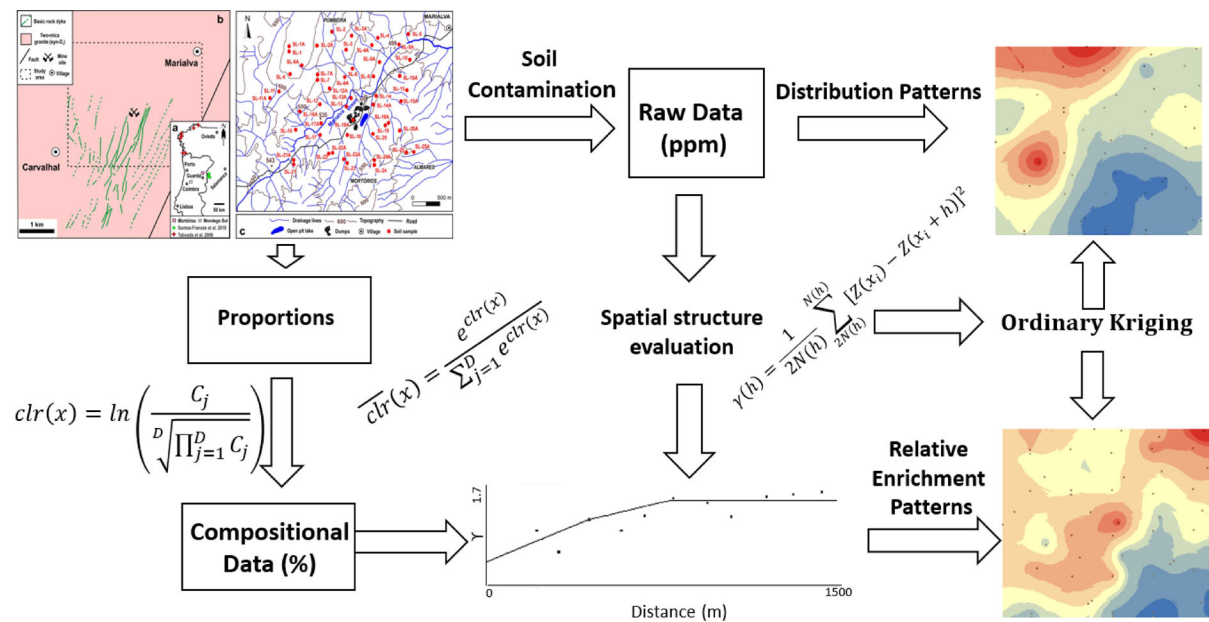
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## Graphic abstract



**Keywords** Raw data · Compositional data · Spatial analysis · Health risks · Open pit lake · Dumps

## Introduction

The assessment of soil contamination using trace elements is a key process in environmental assessment and subsequent management. The type of soil and geological setting determine the mobility of uranium and other metals in soil. The concentration of pollutant species in the soil varies according to the source of contamination—natural and anthropogenic—such as mining activities (e.g., Neiva et al. 2016; Albuquerque et al. 2017; Antunes et al. 2018; Boente et al. 2018a; Fu et al. 2018; González-Fernández et al. 2018; Selvakumar et al. 2018).

Contaminants leaching from open pits and dumps going toward streams, sediments and soils are causing a significant contamination, which affects water, plants, animals and people. Contaminant dispersion and enrichment areas will allow a contamination degree assessment that have been affected by previous or existing mining activities. At contaminated sites, uranium can enter in the food chain through soil–

plant–animal interactions, surface water and groundwater (Neiva et al. 2016). The contaminant interaction with soil type will change with alteration, adsorption and dispersion processes (Selvakumar et al. 2018).

Risk maps are a very powerful tool to support policy-making in environmental risk assessment framework allowing a spatial pattern visualization of contaminant distribution, exposure and its effects, and vulnerability assessment (Lahr and Kooistra 2010; Li et al. 2015). It is possible to determine contaminant distribution and identify the more contaminated or enriched areas (hotspots). Geochemical data are typically reported as compositions (raw data), in the form of proportions subject to a constant sum (e.g., 100%, 1 ppm), which implies that such data are “closed” and for an individual component composition without all environmental contributions (Bucianti and Grunsky 2014). Raw data treatment methodologies must be able to capture our perception of the natural scale of the data and a better understanding of environmental processes. However, a spatial distribution of proportions defined to a specific batch of elements (compositional data) will also allow the definition of contamination trends in an area.

In the present research, the metals and metalloids total concentration in the studied soils of the

abandoned U-mine area from Mortórios (Central Portugal) were used to interpret the contamination outbreaks and therefore locate the main sources of in situ enrichment (hotspots) and the distribution of contamination patterns. This study will promote the identification of contaminated areas and consequent availability assessment and site specific-risk environmental assessment.

**Analytical methods and data modeling**

**Study area**

The old Mortórios uranium mine area is located between the Marialva and Carvalhal villages, in the Guarda county, central Portugal (Fig. 1a, b) and in the Central Iberian Zone of the Iberian Massif (e.g., Farias et al. 1987). It is one of the many mines from the uranium-bearing Beiras area. This mine consists of basic rock dykes of 0.1 to 14 m thick containing torbernite and autunite which cut the syn-D3 coarse-grained porphyritic granite. These dykes are associated with older small quartz veins containing wolframite, pyrrhotite, arsenopyrite, pyrite and chalcopyrite (JEN 1959). This uranium-bearing area

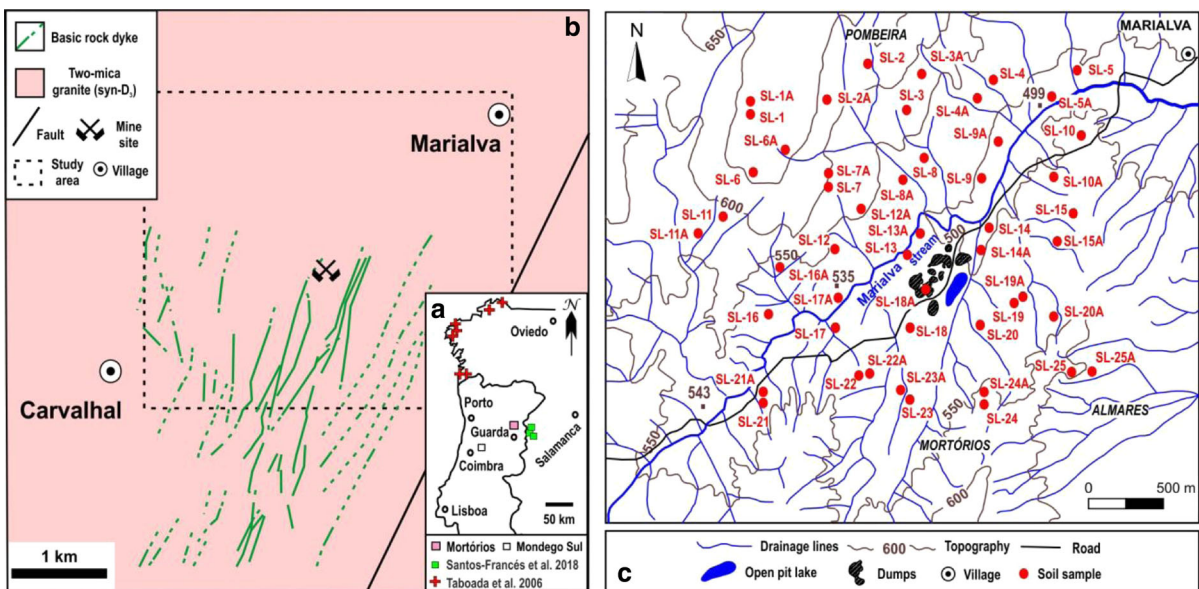
was 470 m long. It produced about 4.5 tonnes of uranium every year from 1982 to 1988 (Fortes et al. 2001).

The old Mortórios uranium mine area has a pit lake containing 126,800 m<sup>3</sup> of water and eight dumps (Fig. 1c). The total volume of dump materials is 135,732 m<sup>3</sup> (Quintela 2015). The dumps contain tailings, waste rocks and rejected materials. Some dumps are located near the road and were used to get sand and materials for buildings and roads (Madruga et al. 2010). The area has mainly herbaceous species, but there are also Pinus. Small agricultural areas occur around the old uranium mine. They contain potatoes crops, vineyards and pastures.

**Analytical methods**

Overlaying the survey area of the old Mortórios uranium mine area, a regular grid of 500 × 500 m was created for point sampling. Within each square, two twin samples were generally picked up at a length of about 50 m. The total of 50 samples were collected in an area about 4 km<sup>2</sup> on the 30th of June 2014 (Fig. 1c).

The samples were collected at a depth of 20–30 cm and were transported in polyethylene bags to the



**Fig. 1** Mortórios uranium mine area and collected soil samples. **a** Location of the study area and the Mondego Sul area in the map of Portugal. Locations of three areas for the baseline of U and of seven profile areas for the baseline of Th in

the partial Spanish map; **b** Geological map and delimitation of the Mortórios area (Adapted from Oliveira et al. 1992; JEN 1959); **c** Location of the open pit lake, dumps and soil samples in the topographic map (adapted from IGE 1994)

laboratory. They were dried at 40 °C, disaggregated with a rubber hammer and sieved through a 2-mm nylon sieve. The pH was measured in a solid-water suspension with a liquid/solid ratio of 2.5 (British standard 1995a). The electrical conductivity (EC) was measured in a solid-water suspension with a liquid/solid ratio of 2.5 (British standard 1995b).

Soil samples < 250 µm were digested with *aqua regia* (3:1 HCl–HNO<sub>3</sub>), filtered through a 2 µm filter and analyzed by ICP-OES using a Horiba Jobin–Yvon JV20002 spectrometer with a monochromator to determine several metals and metalloids total contents. Duplicate blanks and a laboratory standard were included in each batch of twenty-six samples. An internal reference soil, prepared with *aqua regia* and validated using the certified sewage sludge amended soil BCR-143R, was used (Table 1). Detection limits were obtained using a blank signal and dispersion estimated by sampling signal in the 0.5 s interval following the ICP-OES (User manual 2001). The conversion of the detection limit from mg/L to mg/kg was estimated using the following equation (DL mg/kg) = (X \* 0.1)/m, where X is the DL at mg/L given by the ICP-OES and m is the average of the heavy masses of the samples in kg. The results were obtained in the Department of Earth Sciences of the University of Coimbra and were accepted if they were lower than those shown in Table 1.

The textural characteristics were obtained in four soil samples from north and eight samples from south of the open pit lake and dumps. A Coulter laser granulometer (2 mm–0.04 µm; precision up to 5%) was used to get the grain distribution of the fraction < 2 mm of the samples by laser diffraction analysis. A Philips PW3710 X-ray diffractometer, with a Cu tube at 40 kV and 20 nA, was used for X-ray diffraction to enable the mineralogical identifications

of clay minerals in the < 2 µm fraction in oriented samples before and after treatments with ethylene glycol and heating up to 550 °C. All determinations were carried out in the Department of Earth Sciences, University of Coimbra, Portugal.

Organic matter (OM) and cation exchange capacity (CEC) were determined in twelve soil samples. The organic carbon (OC) was determined by dry combustion with an elemental analyzer (SC 144DR, LECO). The samples were heated at 590 °C, oxidized to CO<sub>2</sub> and quantified in the infra-red gas cell. To quantify the organic matter, the van Bemmelen factor of 1.724 (Tabatabai 1996) was used (% OM = % OC × 1.724), with an accuracy of 1%. The exchange of cations (cmol/kg) was carried out with 1 M ammonium acetate pH 7 (1/10) extraction (Thomas 1982) and determined by atomic absorption spectrometry (PinAAcle 900T, PerkinElmer). The CEC was calculated at pH 7.0 from the individual cation values (CEC = Sum of exchangeable bases + exchange acidity) with quantification of the exchange acidity by extraction with 1 N potassium chloride (1/20) and titration with NaOH of a known title until pH 7.0 (Chernov 1947; Nye et al. 1961), with an accuracy of 0.38%. These determinations were carried out at the Agrarian School of Coimbra, in the Soil and Fertility Laboratory.

## Data modeling

### Data transformation—geochemical data and the closure problem

In geochemistry, compositional data respect to the transformation of each original raw concentration (i.e., mg/kg) into proportions of a considered whole (in the case at hands: As; Co; Cr; Mn; Th; U; Cu; Ni; Pb; Sr; W;

**Table 1** Allowance of accuracy and precision for routine analysis

Concentration range	Accuracy $\Delta \lg \bar{C} =  \lg \bar{C}_i - \lg C_s $	Precision $\lambda = \sqrt{\frac{\sum_{i=1}^n (\lg C_i - \lg C_s)^2}{n-1}}$
Soil		
< 3 detection limit	≤ 0.024	0.031
> 3 detection limit	≤ 0.015	0.018

$\bar{C}_i$ —the average determined value of the standards BCR<sub>i</sub> for soil;  $C_i$ —the determined value of the standards BCR for soil;  $C_s$ —the recommended value of the standards BCR for soil;  $n$ —number of standard subsamples located within each batch of samples

Zn), whose elements must sum one—100% (Pawłowsky-Glahn and Egozcue 2006). This topic has been much debated and referred in the literature as the closure problem (Filzmoser et al. 2009). In environmental science studies, it is in general accepted that the chosen analyzed elements correspond to the entirety of the soil, if a suitable number of those attributes is included in the survey (Campbell et al. 2009; Reimann et al. 2012). However, other authors work with what is called subcompositions which define a subset of parts of a composition (Pawłowsky-Glahn and Buccianti 2011). Subcompositions are accurate when the principles of compositional data are respected (Greenacre and Lewi 2009), including the subcompositional coherence principle (Aitchison 1986).

The log-ratio transform functions most used (alr; clr and ilr) have advantages but also disadvantages (Pawłowsky-Glahn and Buccianti 2011; Boente et al. 2018b). The clr transformation is the most used in geochemical studies since the geometric mean is used as a normalizer parameter, and it was the chosen transformation for the purposes of the herein survey.

The equation of the centered log-ratio transformation (clr) was adjusted from (Aitchison 1986):

$$\text{clr}(x) = \ln \left( \frac{C_j}{\sqrt[D]{\prod_{j=1}^D C_j}} \right) \quad (1)$$

where  $C_j$  corresponds to the concentration of the contaminant  $j$  and  $D$  corresponds to the number of sections into which the composition is segmented, namely the number of considered contaminants.

The back-transformation equation is as follows:

$$\overline{\text{clr}}(x) = \frac{e^{\text{clr}(x)}}{\sum_{j=1}^D e^{\text{clr}(x)}} \quad (2)$$

After this transformation, the represented data (clr-transformed data) may be considered as compositional data (proportions) and the sum of all the considered elements after back-transformation must be equal to 1. The clr transformation and its back-transformation were performed using CoDaPack v2.02.21 software (<http://www.compositionaldata.com/codapack.php>).

### Spatial modeling

The spatial characterization of the chosen elements' distribution—U, As, Co and Th—was performed

using ordinary kriging (OK) aiming the definition of spatial distribution patterns and hotspots of U, As, Co and Th location, in the study area. For this purpose, the concentrations of the elements at study (raw data) were used, allowing to interpret the contamination outbreaks and therefore locate the main sources of natural in situ contributions. However, more than simply looking at the elements, concentrations distribution, it is keen to understand their fate. Thus, the development of a new approach aiming to examine the changes in their proportions along the survey area (compositional data) allowed to define trends of dissemination. The compositional dataset went after through a stochastic spatial modeling through “OK,” and the computed spatial patterns used to tackle this issue (Boente et al. 2018b).

In synthesis, a two-step methodology was adopted as follows: (1) The selected attributes went through structural analysis, and experimental variograms were computed for both raw and compositional data and the best fitted theoretical models adjusted. The variogram is a vector function used to calculate the spatial variation structure of regionalized variables (Matheron 1963; Journel and Huijbregts 1978; Gringarten and Deutsch 2001); (2) spatial estimation through geostatistical interpolation (OK) allowed to predict spatial distribution patterns of the variables at hand for any arbitrary location (e.g., Albuquerque et al. 2017). The raw dataset was used to define concentrations spatial patterns as the compositional dataset was used for the definition of trends of dissemination and relative enrichment in U, As, Co and Th.

To stress that the primary application of geostatistical methodologies is the estimation and mapping of environmental attributes. Kriging is the generalized name for an assemble of generalized least-squares regression algorithms. Ordinary kriging respects to local fluctuations of the mean, limiting the correspondent stationary to the local neighborhood (Goovaerts 1997). For the computation, the Space-Stat Software V. 4.0.18, BioMedware was used (Albuquerque et al. 2017).

## Results and discussion

### Geochemistry of soils

The textural characteristics, physical–chemical parameters and concentrations of metals and



metalloids of soils from the old Mortórios uranium mine area are presented in Table 2. The sample 18A was collected close to the open pit lake.

The textural characteristics were determined in a selected number of samples (Table 2). In general, they are dominated by sandy gravels. They consist of granite, quartz, feldspars, muscovite, biotite and rarely contain shale. The clay (< 4 µm) content is very low (0.06–0.35%). In the clay mineralogy, dominates the association vermiculite, smectite, and kaolinite. Smectite was found in five of the twelve studied samples. The fraction < 2 µm presents kaolinite (9–38%), smectite (0–53%), vermiculite (0.58%) and illite (21–66%). The pH values are acidic 4.00–6.34, mainly due to the host two-mica granite influence. The EC values are low ranging between 1 and 84 µS/cm, suggesting a low contamination. The highest EC value occurs at north of the open pit lake and dumps (Table 2).

The OM concentrations in soils range from 0.84 to 1.27% at south of the open pit lake and dumps and from 0.49 to 1.65% at north of the same sampling locations. Organic carbon is of great importance since it is a major uranium binding partner in soils (e.g., Gruan et al. 2000; Cumberland et al. 2016; Bone et al. 2017). The CEC presents 1.84–3.5 cmol<sub>c</sub> kg<sup>-1</sup> at south of the open pit lake and dumps and 2.2–6.1 cmol<sub>c</sub> kg<sup>-1</sup> at north, revealing low CEC values in both cases.

The highest concentrations of metals Fe, Co, Cr, Cu, Mn, Ni, Pb, Sr, Th, U, W and Zn and metalloids As and Sb occur in soils of north of the open pit lake and dumps (Table 2, Fig. 2), because they were released from them, leaching through the two-mica granite mine area and retained in soils. The weathering of the host granite gave some contribution for the U and Th concentrations in soils, because U and Th occur in the accessory minerals, zircon, monazite, xenotime and apatite. The highest median values of Fe, As, Co, Cr, Cu, Ni, Sr and Zn occur at north. But the highest median values of Mn and U occur at south of the open pit lake and dumps, which is probably due to contamination by small local basic rock dykes containing uranium minerals associated with older quartz veins containing sulfides. There are also two other uranium mine areas: A-do-Cavalo at 50 km and Quinta das Vermelhas at 2.8 km at south of the studied area of Mortórios, and both contribute to increasing some metals and As concentrations in soils from south of the open pit lake and dumps.

The metals and metalloids are more mobile and available in acid than neutral conditions (Alloway 1995). At acid conditions U is present as UO<sub>2</sub><sup>2+</sup>, but under neutral conditions occurs as hydroxide complexes and its adsorption increases with the increase in the CEC. The maximum adsorption of U is at pH 5 and then decreases up to 7.7 (Tserenpil et al. 2013). The acid pH of soils from the Mortórios area influences the capacity on the adsorption of metals and metalloids.

There were significant positive correlations ( $r > 0.7$  at 5% significance level) between Fe and Co, Cr, U and Ni, and between Co and Cr with U, Ni and Sr (Table 3). These correlations could be associated with the occurrence of these metals in the uranium mine areas and their geochemical behavior. Some of these correlations were also found in other Portuguese abandoned uranium mine area (Neiva et al. 2015). The adsorption of arsenic by Fe oxides decreases above pH 7 (Goldberg 2002). The pH values of soils from Mortórios area range from 4.00 to 6.34, supporting that As is adsorbed by Fe-oxyhydroxides. Some metals Co, Cr, U and Ni are also adsorbed by Fe-oxyhydroxides (Table 3). The Fe concentration in the studied area is up to 67.7 g/kg (Table 2). The U is incorporated in Fe oxides (Duff et al. 2002). The correlation analysis was applied to soils, from Minas Gerais (Brazil) with both low and high levels of Fe subjected to oxidizing conditions, and suggests that Fe concentrations are the main controlling agent for metals and the metalloid As (de Souza et al. 2015). The correlations between Co and Cr vs. U, Ni and Sr (Table 3) are due to weathering of basic rock dykes, which contain torbernite and autunite.

#### Assessment of soil contamination

The comparison (Fig. 3) of the histograms of the raw and compositional datasets shows that: (1) the raw dataset shows heavily skewed distributions for the four considered elements, biased mainly by the presence of outliers; (2) the compositional dataset allows to assume the assumption of normality for U, As and Th. The Co is the exception, even if with a considerable decrease on the variation coefficient (Fig. 3). Thus, it is possible to reason that compositional data have two principal advantages, as it allows to work with proportions and at the same time improves data normalization.

**Table 2** Textural characteristics, physical–chemical parameters and concentrations of metals and the metalloids of soils from the old Mortórios uranium mine area

Sample	Textural characteristics				Physical–chemical parameters (mg/kg), except Al and Fe (g/kg)																					
	% gravels	% sand	% silt	% clay	pH	EC ( $\mu\text{S}/\text{cm}$ )	T ( $^{\circ}\text{C}$ )	OM (%)	CEC ( $\text{cmol}/\text{Kg}$ )	Al	Fe	As	Co	Cr	Cu	Mn	Ni	Pb	Sb	Sr	Th	U	W	Zn		
<i>South of the open pit lake and mine dumps</i>																										
SL18	42.5	53.3	4.1	0.11	5.23	8	17.5	0.84	1.84	31.5	11.5	80.2	*	*	3.1	271	*	20.6	*	5.9	25.5	54.1	6.7	48.9		
SL18A	51.6	42.9	5.4	0.15	4.17	9	17.4	0.87	3.3	17.8	16.5	52.5	5.2	9.4	*	494	9.9	16.5	*	4.9	33.6	83.8	3.5	69.0		
SL19	57.6	37.9	4.3	0.14	4.00	6	17.5	1.27	3.5	23.8	19.9	66.0	2.6	3.6	5.2	856	*	30.3	*	2.3	27.4	64.1	5.4	86.6		
SL19A	n.d.	n.d.	n.d.	n.d.	4.51	9	17.5	n.d.	n.d.	20.5	16.9	56.4	2.6	2.2	*	622	*	17.6	*	1.75	31.4	52.1	3.4	69.4		
SL17	n.d.	n.d.	n.d.	n.d.	4.57	40	17.5	n.d.	n.d.	40.2	23.3	98.1	3.6	6.2	*	374	5.2	32.8	*	4.2	23.9	79.1	4.6	69.0		
SL17A	n.d.	n.d.	n.d.	n.d.	4.84	42	17.5	n.d.	n.d.	21.4	15.8	86.2	5.4	11.4	7.8	643	7.8	22.2	*	4.9	26.1	71.8	2.7	73.8		
SL20	n.d.	n.d.	n.d.	n.d.	4.47	41	18.0	n.d.	n.d.	23.4	23.9	69.6	10.9	26.1	6.5	808	25.3	23.6	*	7.8	39.9	80.9	3.2	77.5		
SL20A	n.d.	n.d.	n.d.	n.d.	4.44	12	17.8	n.d.	n.d.	23.6	25.5	75.7	10.6	29.0	5.4	684	26.1	22.4	*	6.1	39.8	91.7	4.4	70.0		
SL23	n.d.	n.d.	n.d.	n.d.	5.37	17	17.9	n.d.	n.d.	25.1	17.5	58.0	3.6	4.1	7.2	813	3.7	14.0	*	5.2	32.7	80.0	*	63.4		
SL23A	n.d.	n.d.	n.d.	n.d.	5.06	38	17.8	n.d.	n.d.	17.8	15.3	47.1	3.0	3.7	9.3	636	4.1	12.7	*	5.8	34.5	77.1	2.0	63.4		
SL16	n.d.	n.d.	n.d.	n.d.	4.57	7	17.4	n.d.	n.d.	17.7	17.2	108	3.7	14.4	7.5	411	11.6	22.2	*	5.2	20.1	71.6	*	73.0		
SL24	n.d.	n.d.	n.d.	n.d.	6.14	13	17.8	n.d.	n.d.	26.7	27.1	82.7	7.4	27.5	5.6	634	22.5	29.6	*	13.6	52.5	206	8.2	91.3		
SL24A	n.d.	n.d.	n.d.	n.d.	5.45	32	18.8	n.d.	n.d.	22.3	21.3	66.2	6.7	15.6	3.6	877	15.1	24.2	*	14.5	27.9	96.2	6.1	63.4		
SL22	n.d.	n.d.	n.d.	n.d.	4.47	2	17.9	n.d.	n.d.	20.1	16.1	65.0	*	2.3	3.2	611	*	17.2	*	1.14	22.6	66.6	*	63.8		
SL22A	78.1	20.3	1.53	0.06	4.97	12	18.1	n.d.	n.d.	24.3	17.3	63.7	3.7	8.2	*	258	6.9	19.2	*	5.8	23.3	61.9	*	54.5		
SL25A	n.d.	n.d.	n.d.	n.d.	6.05	6	18.0	n.d.	n.d.	25.8	25.6	82.1	10.3	36.8	5.7	543	31.9	16.7	*	14.1	53.1	217	5.2	79.6		
SL25	n.d.	n.d.	n.d.	n.d.	5.64	11	17.9	n.d.	n.d.	23.1	19.0	73.2	3.0	6.8	*	390	4.5	21.1	*	9.7	40.2	88.8	4.4	87.7		
SL 21	n.d.	n.d.	n.d.	n.d.	5.66	17	17.8	n.d.	n.d.	28.2	23.4	80.5	9.6	24.1	8.9	643	18.5	18.7	*	11.5	36.8	78.4	5.4	71.6		
SL21A	n.d.	n.d.	n.d.	n.d.	4.87	84	18.0	n.d.	n.d.	44.7	41.9	129	20.1	67.0	16.1	712	51.2	28.0	*	17.0	43.9	141	4.0	74.9		
Median										23.6	19.0	73.2	3.7	9.4	5.4	634	7.8	21.1	0.0	5.8	32.7	79.1	4.0	70.0		
<i>North of the open pit lake and mine dumps</i>																										
SL14	54.7	34.9	10.0	0.33	6.34	12	17.4	0.81	6.1	21.5	22.0	104	6.0	23.4	21.4	896	19.2	24.7	2.0	17.0	41.5	138	*	95.6		
SL14A	n.d.	n.d.	n.d.	n.d.	6.23	14	17.4	n.d.	n.d.	18.2	19.4	77.0	4.6	16.9	11.3	939	14.6	16.3	*	9.4	35.7	74.8	*	78.8		
SL13	n.d.	n.d.	n.d.	n.d.	4.67	13	17.7	n.d.	n.d.	15.8	12.1	79.8	*	6.4	16.5	386	*	18.2	2.3	5.9	24.1	60.4	3.4	61.2		
SL13A	n.d.	n.d.	n.d.	n.d.	5.25	2	17.6	n.d.	n.d.	24.0	11.3	101	*	6.3	3.3	341	3.2	25.3	*	7.8	21.8	47.2	10.0	45.2		
SL15	71.0	25.0	4.0	0.07	5.41	6	17.5	0.66	4.5	22.8	24.6	109	8.3	22.3	7.1	771	22.9	21.3	*	12.0	43.9	108	2.3	87.2		
SL15A	n.d.	n.d.	n.d.	n.d.	5.40	2	17.6	n.d.	n.d.	24.6	20.6	107	3.0	6.9	6.1	1065	4.8	22.9	*	8.8	38.0	116	2.4	98.9		
SL12	n.d.	n.d.	n.d.	n.d.	5.79	2	17.5	n.d.	n.d.	26.8	18.5	108	2.8	7.8	3.4	882	5.4	26.3	5.7	6.2	29.1	70.1	6.6	76.2		

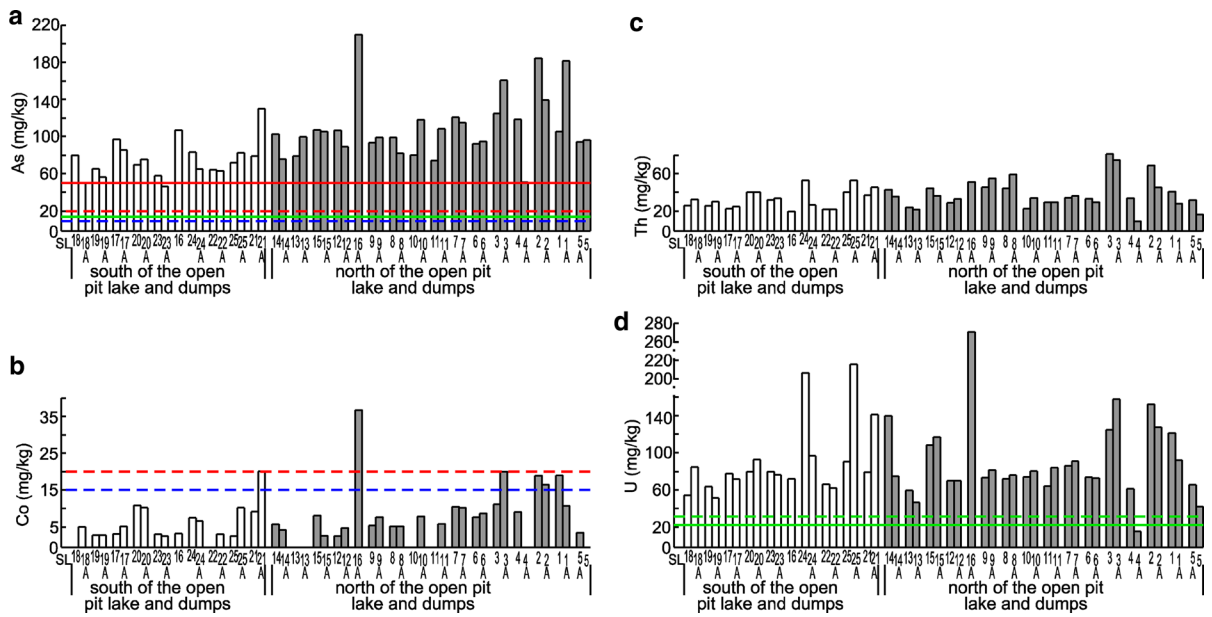
Table 2 continued

Sample	Textural characteristics			Physical-chemical parameters										(mg/kg, except Al and Fe (g/kg))									
	% gravels	% sand	% silt clay	pH	EC (µS/cm)	T (°C)	OM (%)	CEC (cmol/Kg)	Al	Fe	As	Co	Cr	Cu	Mn	Ni	Pb	Sb	Sr	Th	U	W	Zn
SL12A	n.d.	n.d.	n.d.	5.63	1	17.3	n.d.	n.d.	15.7	14.9	89.7	4.9	15.4	10.9	477	12.1	20.8	*	7.0	32.0	70.0	7.1	66.5
SL16A	n.d.	n.d.	n.d.	5.20	14	17.4	n.d.	n.d.	54.4	67.7	208	36.3	143	38.5	706	103	22.6	*	37.7	51.1	271	10.5	79.0
SL9	n.d.	n.d.	n.d.	4.86	6	17.5	n.d.	n.d.	19.2	18.1	94.5	5.7	19.1	11.5	532	10.8	20.4	*	10.8	45.2	73.6	3.8	88.5
SL9A	38.5	48.0	13.2	0.30	4.63	9	17.4	0.55	2.7	21.1	20.2	101	8.0	25.5	412	12.5	16.7	*	14.2	54.0	80.9	*	82.6
SL8	n.d.	n.d.	n.d.	5.41	7	17.2	n.d.	n.d.	18.2	17.4	102	5.3	17.9	10.1	398	10.4	25.9	*	10.4	43.6	72.4	*	81.6
SL8A	35.5	57.3	8.1	0.15	5.79	9	17.2	0.60	2.7	15.4	16.6	84.2	5.3	395	9.7	15.9	*	11.4	57.6	76.1	*	79.7	
SL10	n.d.	n.d.	n.d.	4.57	10	17.2	n.d.	n.d.	14.9	13.5	81.9	*	7.6	4.3	464	5.6	20.5	*	4.9	22.2	73.7	*	66.4
SL10A	59.2	34.4	6.1	0.24	5.32	8	17.4	0.49	2.9	26.7	22.6	120	8.3	600	19.4	20.2	*	8.6	32.8	79.5	2.3	80.0	
SL11	63.0	29.9	7.0	0.12	5.85	3	17.4	0.49	2.2	15.3	13.2	76.0	*	542	3.1	22.0	*	3.8	27.7	65.1	*	64.6	
SL11A	n.d.	n.d.	n.d.	5.45	4	17.3	n.d.	n.d.	22.8	23.4	110	8.7	30.6	11.9	760	23.0	25.6	2.3	7.7	28.5	82.9	6.5	75.2
SL7	n.d.	n.d.	n.d.	6.12	19	17.6	n.d.	n.d.	24.8	27.9	123	10.7	39.0	14.4	624	23.9	25.5	*	12.1	32.9	86.1	7.2	86.2
SL7A	n.d.	n.d.	n.d.	5.96	23	17.6	n.d.	n.d.	22.3	26.1	117	10.4	38.9	14.6	610	23.3	23.6	*	12.2	34.5	91.1	3.4	85.8
SL6	n.d.	n.d.	n.d.	5.67	10	17.6	n.d.	n.d.	18.7	22.5	94.0	8.1	31.0	10.3	821	17.4	17.5	*	7.9	31.5	73.8	7.7	72.1
SL6A	55.4	39.6	4.8	0.11	6.18	20	17.6	1.56	3.3	18.1	21.0	97.3	9.2	671	14.7	15.7	*	10.8	28.4	73.4	2.7	70.3	
SL3	n.d.	n.d.	n.d.	4.62	21	17.5	n.d.	n.d.	31.7	27.7	127	11.4	20.3	9.5	381	8.4	27.8	*	8.7	79.1	125	6.6	127
SL3A	n.d.	n.d.	n.d.	5.74	5	17.5	0.92	2.2	41.3	39.4	163	20.2	49.6	15.6	346	23.8	35.0	2.1	35.1	72.8	156	7.6	122
SL4	28.0	56.8	14.9	0.35	5.32	11	17.7	n.d.	2.2	26.7	22.5	121	9.3	249	14.9	28.3	3.7	11.3	32.6	62.3	9.7	100	
SL4A	n.d.	n.d.	n.d.	4.90	6	17.7	1.65	2.7	11.5	13.0	52.5	*	6.1	3.1	146	3.5	13.8	*	4.1	7.9	17.3	8.8	38.3
SL2	n.d.	n.d.	n.d.	5.55	20	17.5	n.d.	n.d.	45.0	40.0	186	19.1	51.6	13.7	383	24.3	36.7	*	33.5	65.9	151	7.8	117
SL2A	n.d.	n.d.	n.d.	5.75	51	17.6	n.d.	n.d.	27.2	29.8	142	16.4	44.8	10.6	361	26.0	25.7	1.53	27.3	44.4	127	9.1	86.3
SL1	n.d.	n.d.	n.d.	5.86	25	17.5	n.d.	n.d.	27.2	37.0	108	18.9	59.6	17.1	573	37.3	25.7	*	21.7	37.5	120	5.5	113
SL1A	n.d.	n.d.	n.d.	5.42	15	17.6	n.d.	n.d.	36.7	25.9	184	11.0	24.9	21.2	186	14.6	33.2	*	10.5	26.3	91.4	8.9	113
SL5A	n.d.	n.d.	n.d.	4.89	30	17.6	n.d.	n.d.	15.5	17.4	95.5	3.9	16.6	7.4	496	11.2	17.6	*	8.6	29.6	65.6	7.5	67.4
SL5	n.d.	n.d.	n.d.	4.62	15	17.5	n.d.	n.d.	19.7	17.3	99.1	*	10.8	3.4	227	7.2	20.3	*	4.5	14.6	43.0	*	48.0
Median									22.3	21.0	104	8.0	22.3	10.5	496	14.6	22.6	0.0	10.4	32.8	76.1	5.5	80.0

EC electrical conductivity, OM organic matter, CEC cation exchange capacity, n.d. not determined

\*Below the detection limit. Detection limits in mg/kg for Al 4.0, Fe 2.7, As 1.08, Cd 1.44, Co 2.4, Cr 1.99, Cu 2.8, Mn 2.3, Ni 3.0, Pb 2.4, Sb 1.42, Sr 1.10, Th 1.39, U 2.5, W 1.73 and Zn 1.59; Cd is not detected. Analyst: A.C.T. Santos





**Fig. 2** Selected concentrations of metals and the metalloid As of soils from the abandoned Mortórios uranium mine area. **a** As, **b** Co, **c** Th, **d** U. Red dashed line: public, private and green areas and residential areas; Red solid line: commercial and industrial areas (Decreto Ministerial 1999). Blue dashed line: Swedish

guidelines for soils (2009). Blue solid line: As for all purposes; Green solid line: U for agricultural and residential parkland uses; Green dashed line: U for commercial use (Canadian soil quality guidelines 2017)

**Table 3** Correlation matrix (Spearman) in soils from the old Mortórios uranium mine area

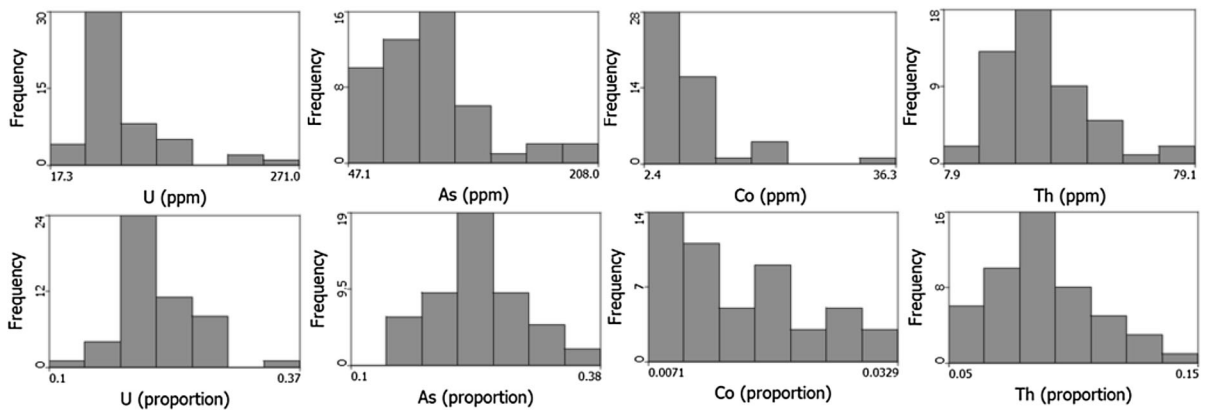
	Al	Fe	Mn	As	Co	Cr	Th	U	Cu	Ni	Pb	Sr	W	Zn	Sb
Al	1														
Fe	<b>0.7</b>	1													
Mn	0.0	0.2	1												
As	0.5	0.6	-0.1	1											
Co	0.6	<b>0.9</b>	0.1	0.6	1										
Cr	0.4	<b>0.8</b>	0.1	<b>0.7</b>	<b>0.9</b>	1									
Th	0.4	0.6	0.2	0.4	<b>0.7</b>	0.6	1								
U	0.5	<b>0.8</b>	0.3	0.5	<b>0.8</b>	<b>0.7</b>	<b>0.7</b>	1							
Cu	0.2	0.5	0.2	0.6	<b>0.7</b>	<b>0.7</b>	0.4	0.4	1						
Ni	0.4	<b>0.8</b>	0.2	0.6	<b>0.9</b>	<b>0.9</b>	0.6	<b>0.7</b>	0.6	1					
Pb	0.6	0.6	-0.1	0.6	0.4	0.4	0.2	0.4	0.2	0.3	1				
Sr	0.5	<b>0.7</b>	0.1	0.6	<b>0.8</b>	<b>0.8</b>	<b>0.7</b>	<b>0.7</b>	0.6	<b>0.8</b>	0.3	1			
W	0.5	0.4	-0.2	0.4	0.4	0.3	0.1	0.2	0.2	0.3	0.5	0.3	1		
Zn	0.5	0.7	0.1	0.6	0.6	0.6	<b>0.7</b>	0.6	0.5	0.5	0.6	0.6	0.2	1	
Sb	0.1	0.1	0.0	0.3	0.1	0.2	0.0	0.0	0.3	0.1	0.3	0.2	0.2	0.2	1

The highest Spearman correlation value with a significance level of 0.05 are given in bold

Spatial modeling—geostatistical approach

For spatial modeling, U, As, Co and Th were chosen as they are core in contamination forecasts, associated with uranium mining. The spatial stochastic patterns

of the four chosen contamination indicators were performed following a two-step geostatistical modeling approach: (1) structural analysis (variography) and (2) spatial estimation through ordinary kriging (OK).



**Fig. 3** Raw data histograms for U, As, Co and Th (ppm) and compositional data histograms for U, As, Co and Th (%)

The semi-variogram is a vectorial function applied to compute the spatial variability of a specific regionalized variable, defined as follows (Matheron 1963; Journel and Huijbregts 1978):

$$\gamma(h) = \frac{1}{2N(h)} \sum_{2N(h)}^{N(h)} [Z(x_i) - Z(x_i + h)]^2 \quad (3)$$

The distance is represented by  $h$ , where  $Z(x_i)$  and  $Z(x_i + h)$  are the values assumed by the observed variable at points  $x_i$ , and  $x_i + h$ . Indeed, it is the average value of the square differences, for an assemble of pairs of points, within a survey field at an  $h$  distance. The obtained variograms provide information about the spatial structure of the attribute at study. The nugget effect ( $C_0$ ), represents the behavior at short distances. The sill ( $C_1$ ) and the amplitude ( $a$ ) define, respectively, the used inertia in the inference process and the radius of influence for each of the considered variables.

Theoretical models  $\gamma^*(h)$  were fitted to the experimental functions for the raw and the compositional data (Isaaks and Srivastava 1989). The omnidirectional (isotropic structures) variograms obtained show, generally, a better fitting for the compositional dataset (Fig. 4). Therefore, the interpolation procedure's (OK) error is minimized when using the compositional dataset.

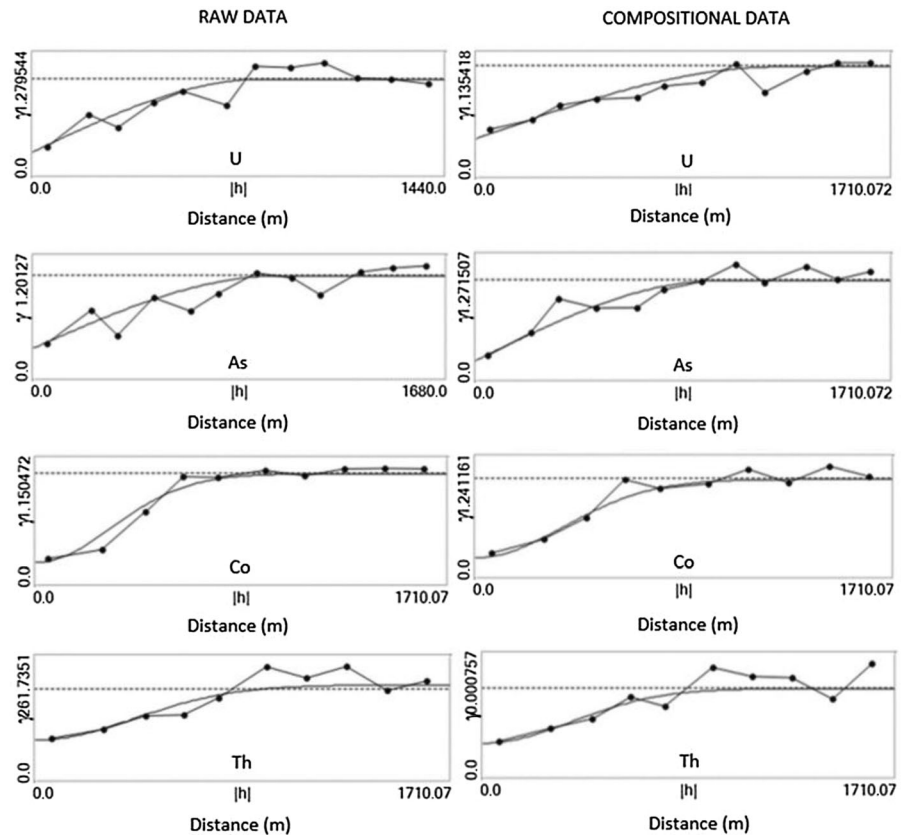
The distribution patterns were than computed using OK (Fig. 5). Evident contrasts are noteworthy when analyzing the maps obtained for the raw and the compositional datasets. Indeed, a careful interpretation of these representations must be done, as they support distinct data and, therefore, different

information. The RD's spatial patterns show the stochastic approach of the elements concentration distributions and targeting hotspots location for these contaminants. On the other hand, the CD's mapping allowed the visualization of proportions' spatial variability and, therefore, providing an important insight in contaminant fate, within the study area.

The contaminants higher concentrations (hotspots) are located north and northwest in the survey area (Fig. 5). However, it is in the south-southeast segment where the highest proportion on U—up to 33%; As—up to 35% and Th—up to 13% can be found. The Co relative enrichment is not relevant—up to 2.8%. The results lead to conclude that the old mining activity inflicted the depletion on U, As and Th concentrations. However, the high proportions on these elements, observed in the southeast segment, lead to consider that the fate of these contaminants shows a clear spatial trend and, therefore, a prevalent direction for contamination and local concentration (relative enrichment). Uranium is significantly more soluble than Th in moderately acidic soils, which justify the U prevalent direction of contamination to southeast part (Ahmed et al. 2012).

The studied soil samples from south and north of the open pit lake and dumps of the old Mortórios uranium mine area record As concentrations of 47.1–129 mg/kg and 52.5–208 mg/kg, respectively, which are higher than 1–20 mg/kg As for typical uncontaminated agricultural soils (Adriano 2001). They exceed the Italian guideline limits (Decreto Ministerial 1999) for public, private and green areas and residential areas (20 mg/kg) and generally for commercial and industrial areas (50 mg/kg) (Fig. 2a).

**Fig. 4** Omnidirectional experimental variograms and correspondent fitted models for raw and compositional data



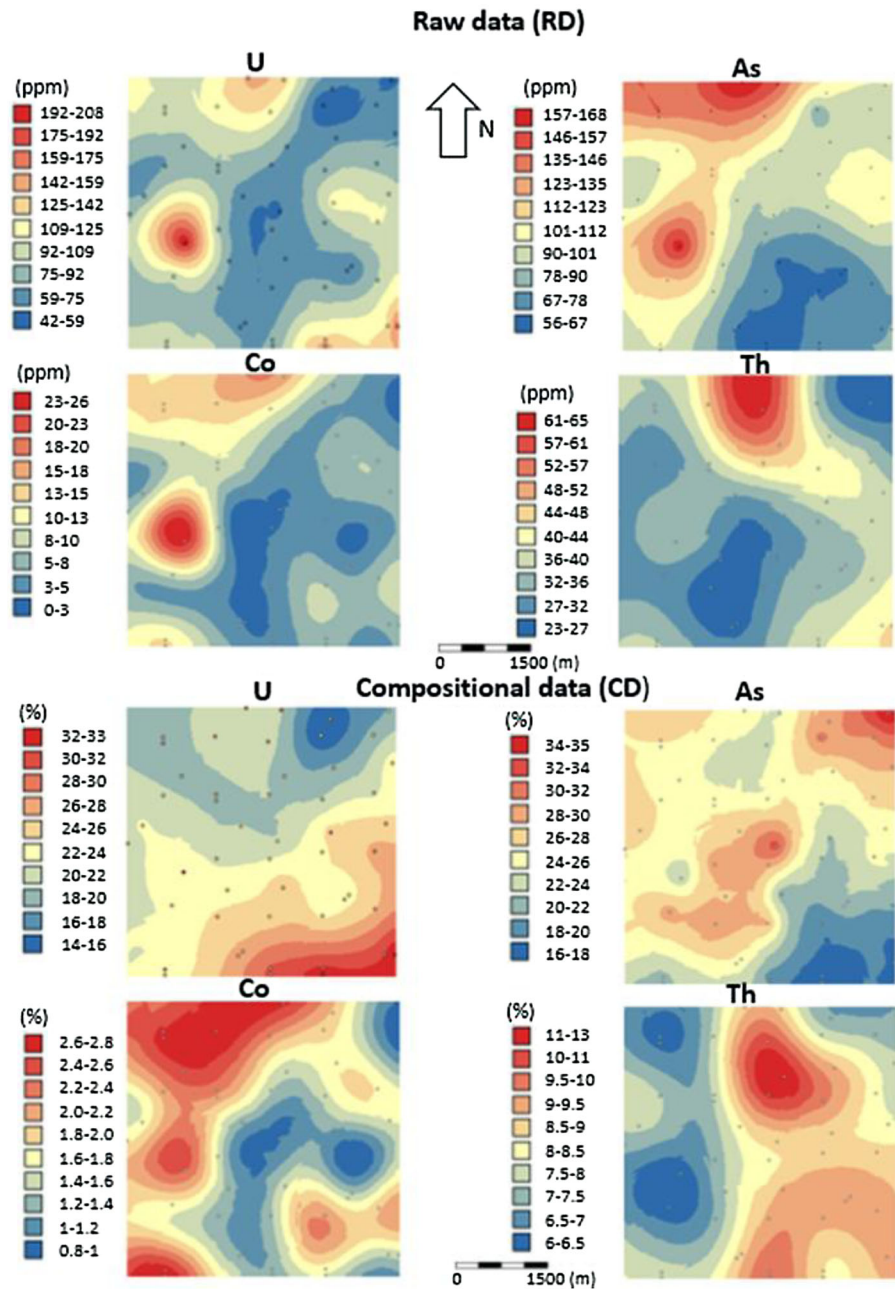
They also have higher As concentrations than 10 mg/kg (Fig. 2a) indicated for sensitive land use (Swedish guidelines for soils 2009) and 12 mg/kg for agricultural, residential parkland, commercial and industrial uses of the Canadian soil quality guidelines (2017). The different physical and chemical properties of soils control the fraction of As in soils available to plants.

The studied soil samples from south and north of the Mortórios area also have 52.1–217 mg/kg and 17.3–271 mg/kg concentrations of U, respectively, which are generally higher than 23 mg/kg (Fig. 2d) for agricultural and residential parkland uses and 33 mg/kg for commercial use, according to the Canadian guidelines (2017). One soil sample from south and four soil samples from north soil samples have Co concentrations of 16.4–36.3 mg/kg, which exceed that of 15 mg/kg (Fig. 2b) of the Swedish guidelines for soils (2009).

Two contamination factors were calculated.  $Pi = Ci/Bi$  where  $Ci$  is the measured concentration of the pollutant and  $Bi$  is the natural background level, which gives for most metals and As the baseline values given

by median for values for < 2.00 mm Portuguese topsoil's of Al-1.73%, As-9 mg/kg, Co-8 mg/kg, Cr-21 mg/kg, Cu-16 mg/kg, Fe-2.22%, Mn-370 mg/kg, Ni-14 mg/kg, Pb-15 mg/kg, Sr-9 mg/kg and Zn-45 mg/kg (Inácio et al. 2008). However, the  $Ci$  value for U of Portuguese topsoil's is unavailable. Therefore, the  $Ci$  baseline of U for granitic subsoil, without the influence of the soil formation and anthropogenic processes, from three Spanish areas at West of Salamanca and not far from the Mortórios area in Portugal (Fig. 1a) with the value of 29.8 mg/kg (Santos-Francés et al. 2018) was used. The 40.8 mg/kg for Th of the C horizon PCG (porphyritic coarse-grained granite) developed on granites of the NW Spain (Taboada et al. 2006) (Fig. 1a) was used because it is from a granitic subsoil. The  $Pi$  pollution percentage increases for As, Co, Fe, Cr, Cu, Ni, Pb, Sr, Th, U and decreases for Al, Mn and Zn from south soils to north soils of the open pit lake and dumps from the Mortórios uranium mine area (Table 4). South soils present mainly low pollution percentage of Co, Cr, Cu, Fe, Ni, Sr and Th, but moderate pollution of Al,

**Fig. 5** Ordinary kriging results. Distribution patterns for raw data (RD) and compositional data (CD). Scale is expressed in ppm (RD) and in % (CD)



Mn, Pb, U and Zn and only very high pollution percentage of As. The north soils have mainly low pollution percentage of Cu and Th, moderate pollution percentage of Al, Co, Cr, Mn, Pb, Ni, U and Zn and very high pollution of Al, As and Fe.

The other contamination factor is the  $I_{geo} = \log_2(-Ci/1.50 Bi)$  with  $Ci$ —the measured concentration of metal or metalloid  $i$  in the soil and  $Bi$  is the geological

background level of the metal or metalloid  $i$ . The pollution percentage based on the  $I_{geo}$  increases for As, Co, Cr, Fe, Pb, Sr, Th and Zn from south soils to north soils of the open pit lake and dumps from the old Mortórios uranium mine area (Table 5). The increases in As, Pb, Fe and Zn are mainly due to the weathering of sulfides from the quartz veins. The increases in Co and Cr are due to the weathering of basic rock dykes.

**Table 4** The pollution percentage based on the factor Pi of soils from the old Mortórios uranium mine area. (Modified from Santos-Francés et al. 2018)

Pollution %	Al	As	Co	Cr	Cu	Fe	Mn	Ni	Pb	Sr	Th	U	Zn
<i>South of the open pit lake and mine dumps</i>													
Low	0	0	71	72	93	63	11	53	11	68	84	0	0
Moderate	100	0	29	28	7	37	89	40	89	32	16	74	100
High	0	11	0	0	0	0	0	7	0	0	0	16	0
Very high	0	89	0	0	0	0	0	0	0	0	0	10	0
<i>North of the open pit lake and mine dumps</i>													
Low	23	0	41	48	84	55	26	47	3	45	65	3	3
Moderate	77	0	55	49	16	42	74	50	97	42	35	62	97
High	0	0	4	0	0	3	0	0	0	13	0	32	0
Very high	0	100	0	3	0	0	0	3	0	0	0	3	0

Pi < 1—low pollution;  
 1 = Pi < 3—moderate pollution;  
 3 = Pi < 6—high pollution;  
 Pi ≥ 6—very high pollution

The small increase in Th is attributed to the weathering of basic rock dykes. South soils and north soils are mainly non-contaminated in Co, Cr, Cu, Sr and Th, mildly contaminated in U and Zn and strongly contaminated in As (Table 5).

The U is much more leached from the open pit lake and dumps and adsorbed by Fe-oxyhydroxides than Th (Table 2, Fig. 2c, d), because U is more abundant than Th. The Th is generally more resistant to weathering than U (e.g., Négrel et al. 2018). The acid pH (4.00–6.34) of the Mortórios soils increases the mobility of toxic elements (Loska et al. 2004).

Comparison of contaminations in soils from two Portuguese uranium mine areas exploited from open pits

The old Mondego Sul uranium mine area produced 75 tonnes of U<sub>3</sub>O<sub>8</sub> from 1987 to 1991 (Neiva et al. 2016). Therefore, it produced more 48 tonnes of U<sub>3</sub>O<sub>8</sub> for 4 years than the old Mortórios uranium mine area for 6 years. The Mondego Sul mine consists of quartz veins containing sulfides and autunite, torbernite, meta-uranocircite and meta-saleeite, whereas the Mortórios uranium mine consists of weathered basic rock dykes containing torbernite and autunite associated with older quartz veins containing sulfides. The soil data from the old Mondego Sul uranium mine area belong to 15 samples from outside and 36 samples from inside the mine influence area. In general, soils from south and north of the open pit lake and dumps from Mortórios present lower median values for metals and metalloids than those of soils from outside

and inside the mine influence area of Mondego Sul (Table 6), except for Mn.

Soils from the old Mondego Sul uranium mine area are contaminated in As, contain 70.4–147 mg/kg and 47.5–223 mg/kg from outside and inside the mine influence area, respectively (Table 6) and may not have any use according to the Decreto Ministerial (1999), Swedish guidelines for soils (2009) and Canadian soil quality guidelines (2017). They also contain 27.4–152 mg/kg of U from outside and 51.8–377 mg/kg of U from inside the mine influence area (Table 6) and may not also be used for agricultural, parkland and commercial uses according to the Canadian soil guidelines (2017). Some soil samples from inside the mine influence area also have a higher U concentration than the accepted value for industrial use (300 mg/kg), indicated by the Canadian guidelines (2017). Some soil samples from outside and inside also have higher Pb concentrations and a few soil samples from inside present higher Co rarely Sb concentrations than those of Pb (100 mg/kg), Co (20 mg/kg) and Sb (10 mg/kg) indicated by the Decreto Ministerial (1999) (Table 6). Some soil samples from outside and inside the mine influence area have higher Ni, Pb and Zn concentrations and some soil samples from inside also present higher Co concentrations than those of Co (15 mg/kg), Ni (40 mg/kg), Pb (50 mg/kg), Zn (250 mg/kg) given by the Swedish guidelines (2009) (Table 6). Some soil samples from outside and inside the mine influence area have higher Cr, Ni, Pb and Zn concentrations than those for agricultural and residential parkland uses (Cr 64 mg/kg; Ni 45 mg/kg; Pb 70 mg/kg, 140 mg/kg; Zn



**Table 5** The pollution percentage based on the geo-accumulation index Igeo of soils from the old Mortórios uranium mine area

Pollution %	Al	As	Co	Cr	Cu	Fe	Mn	Ni	Pb	Sr	Th	U	Zn
<i>South of the open pit lake and mine dumps</i>													
Class 0	74	0	94	89	100	95	37	53	68	79	100	0	32
Class 1	26	0	6	6	0	5	63	40	32	21	0	74	68
Class 2	0	5	0	5	0	0	0	7	0	0	0	16	0
Class 3	0	90	0	0	0	0	0	0	0	0	0	10	0
Class 4	0	5	0	0	0	0	0	0	0	0	0	0	0
Class 5	0	0	0	0	0	0	0	0	0	0	0	0	0
Class 6	0	0	0	0	0	0	0	0	0	0	0	0	0
<i>North of the open pit lake and mine dumps</i>													
Class 0	68	0	80	77	97	87	58	70	48	77	90	7	26
Class 1	29	0	16	20	3	13	42	27	52	10	10	58	74
Class 2	3	3	4	0	0	0	0	0	0	13	0	32	0
Class 3	0	55	0	3	0	0	0	3	0	0	0	3	0
Class 4	0	42	0	0	0	0	0	0	0	0	0	0	0
Class 5	0	0	0	0	0	0	0	0	0	0	0	0	0
Class 6	0	0	0	0	0	0	0	0	0	0	0	0	0

Class 0,  $I_{geo} < 0$ , non-contaminated; class 1,  $0 = I_{geo} < 1$ , from mildly contaminated; class 2,  $1 = I_{geo} < 2$  to moderately contaminated; class 3,  $2 = I_{geo} < 3$ , to strongly contaminated; class 4,  $3 = I_{geo} < 4$ , and strongly contaminated; class 5,  $I_{geo} < 5$ , from heavily contaminated; class 6,  $I_{geo} \geq 5$ , to extremely contaminated. (e.g., Santos-Francés et al. 2018)

200 mg/kg) given by the Canadian guidelines (2017) (Table 6). Therefore, soils from Mondego Sul area are more contaminated than those from the Mortórios area. However, in both areas the main contaminants are arsenic and uranium. At the former U site at Kadji Sai, Kyrgyzstan, the soils were also mainly enriched in U and As (Lind et al. 2013).

The Pi factor shows increases in the pollution percentages of Co, Cr, Fe, Cu, Mn, Ni, Pb, Sr and U increases from the soils outside to soils inside the mine influence area from the old Mondego Sul uranium mine area (Table 7). These behaviors are due to the weathering of quartz veins containing sulfides and uranium minerals. The outside soils and inside soils have mainly low pollution percentages of Mn and Sr, which also occur in Co from outside soils. They also have moderate pollution percentages in Al, Cr, Cu, Fe and Zn, but they also occur in Co and Ni in inside soils. The high pollution percentages are recorded for Pb, Th and U in soils from outside and Th, U and Zn in soils from inside. In general, the Pi factor percentage presents higher contamination mainly in soils from the

Mondego Sul area than in the soils from the Mortórios area (Tables 4, 7), which can be explained by the significant higher exploitation in the former area than in the latter one.

The Igeo pollution percentage shows increase in Co, Cu, Fe, Mn, Ni, Pb, Th and U and decrease in Cr from soils of outside to soils of inside of the mine influence area from the Mondego Sul uranium mine area (Table 8). Outside soils from the Mondego Sul area are mainly non-contaminated in Co, Cu, Fe, Mn, Ni, and Sr, mildly contaminated in Al, Cr, Th and Zn, to moderately contaminated in Pb and U and strongly contaminated in As. The inside soils are mainly non-contaminated in Fe, Co, Mn and Sr, mildly contaminated in Al, Cr, Cu, Ni and Zn, moderately contaminated in Pb, Th and U and strongly contaminated in As. The Igeo factor percentage shows higher contamination of Al, As, Cr, Cu, Pb, Th, U and Zn in outside and inside the mine influence area in soils from the Mondego Sul area than in the soils from south and north of the open pit lake and dumps of the Mortórios area (Tables 5 and 8).



**Table 6** Comparison of the values, of some metals and the metalloid As of the soil from the two abandoned uranium mines areas, before the remediation, with Portuguese background and FOREGS values, and with the limits of Canadian and Italian legislation

Parameter	Unit	Mortórios				Mondego Sul (Neiva et al. 2016)			
		South of the open pit lake and mine dumps		North of the open pit lake and mine dumps		Outside the mine influence area		Inside the mine influence area	
		Median	Range	Median	Range	Median	Range	Median	Range
As	(mg/kg)	73.2	47.1–129	104	52.5–208	112	70.4–147	111	47.5–223
Co	(mg/kg)	3.7	< 2.4–20.1	8.0	< 2.4–36.3	7.0	4.2–13.2	10.5	3.1–24.6
Cr	(mg/kg)	9.4	< 1.99–67.0	22.3	4.8–143	43.0	30.7–76.6	53.3	11.4–70.1
Cu	(mg/kg)	5.4	< 2.8–16.1	10.5	3.1–38.5	21.3	13.3–38.3	31.8	8.4–58.8
Ni	(mg/kg)	7.8	< 3.0–51.6	14.6	< 3.0–103	15.3	6.0–48.2	29.2	5.8–60.3
Pb	(mg/kg)	21.1	12.7–32.8	22.6	13.8–36.7	79.8	26.8–202	80.6	21.0–200
Th	(mg/kg)	32.7	20.1–53.1	32.8	7.9–79.1	161	72.2–315	149	81.2–236
U	(mg/kg)	79.1	52.1–217	76.1	17.3–271	90.7	27.4–152	124	51.8–377
Zn	(mg/kg)	70.0	48.9–91.3	80.0	38.3–127	119	60.9–281	133	37.4–364

Parameter	Unit	Portuguese background (Inácio et al. 2008)		Salminen et al. (2005) FOREGS		Canadian limit values (Canadian Soil Guidelines, 2017)		Italian limit values (Decreto Ministerial. 1999)	
		Median	Range	Median	Range	Agricultural	Residential parkland	Agricultural	Residential parkland
		As	(mg/kg)	9	< 2–266	6.0	< 5.0–220	12	12
Co	(mg/kg)	8	0.5–42	7.0	< 1.00–255	40	50	20	20
Cr	(mg/kg)	21	0.5–223	22.0	1.00–2340	64	64	150	150
Cu	(mg/kg)	16	0.5–111	12.0	1.00–239	63	63	120	120
Ni	(mg/kg)	14	< 1–880	14.0	< 2.0–2560	45	45	120	120
Pb	(mg/kg)	15	1–180	15.0	< 3.0–886	70	140	100	100
Th	(mg/kg)	5	1–49	7.2	0.3–75.9	n.d.	n.d.	n.d.	n.d.
U	(mg/kg)	n.d.	n.d.	2.0	0.21–53.2	23	23	n.d.	n.d.
Zn	(mg/kg)	45	0.5–589	48.0	4.0–2270	200	200	150	150

n.d. not defined

**Table 7** The pollution percentage based on the factor Pi of soils from the old Mondego Sul uranium mine area

Pollution %	Al	As	Co	Cr	Cu	Fe	Mn	Ni	Pb	Sr	Th	U	Zn
<i>Outside the mine influence area</i>													
Low	7	0	67	0	20	40	87	47	7	93	0	7	0
Moderate	93	0	33	93	80	60	13	47	0	7	20	33	60
High	0	0	0	7	0	0	0	6	60	0	53	60	33
Very high	0	100	0	0	0	0	0	0	33	0	27	0	7
<i>inside the mine influence area</i>													
Low	6	0	28	3	8	17	80	14	0	83	0	0	3
Moderate	94	0	69	78	81	80	17	72	22	17	19	17	50
High	0	3	3	19	11	0	3	14	36	0	81	58	42
Very high	0	97	0	0	0	0	0	0	42	0	0	25	5

Classes as in Table 4

**Table 8** The pollution percentage based on the geo-accumulation index Igeo of soils from the old Mondego Sul uranium mine area

Pollution %	Al	As	Co	Cr	Cu	Fe	Mn	Ni	Pb	Sr	Th	U	Zn
<i>Outside the mine influence area</i>													
Class 0	33	0	100	0	93	100	93	66	6	100	20	6	7
Class 1	67	0	0	60	7	0	7	27	0	0	53	27	53
Class 2	0	0	0	40	0	0	0	7	60	0	27	67	33
Class 3	0	7	0	0	0	0	0	0	27	0	0	0	7
Class 4	0	93	0	0	0	0	0	0	7	0	0	0	0
Class 5	0	0	0	0	0	0	0	0	0	0	0	0	0
Class 6	0	0	0	0	0	0	0	0	0	0	0	0	0
<i>Inside the mine influence area</i>													
Class 0	31	0	64	8	28	61	86	25	3	94	0	0	5
Class 1	69	0	33	72	67	39	11	61	19	6	19	17	47
Class 2	0	3	3	20	5	0	3	14	39	0	81	58	42
Class 3	0	39	0	0	0	0	0	0	31	0	0	22	6
Class 4	0	53	0	0	0	0	0	0	8	0	0	3	0
Class 5	0	5	0	0	0	0	0	0	0	0	0	0	0
Class 6	0	0	0	0	0	0	0	0	0	0	0	0	0

Classes as in Table 5

## Conclusions

At the old Mortórios uranium mine area, soils in the northern vicinity of the open pit lake and dumps have the highest concentrations of Fe, Co, Cr, Cu, Mn, Ni, Pb, Sr, Th, U, W and Zn and the highest median values of Fe, As, Co, Cr, Cu, Ni, Sr and Zn, because they were leached from weathered basic rock dikes containing torbernite and autunite and older quartz veins containing sulfides. However, the highest median values of U and Mn occur at south of the open pit lake and dumps.

The contaminants higher concentrations—hot-spots—are located north and northwest in the survey area. However, it is in the south-southeast segment where the highest proportion on U—up to 33%; As—up to 35% and Th—up to 13% are achieved. However, when analyzing the proportions, a clear concentration on U, As, and Th is settled in the southeast part. Therefore, the fate of these contaminants has a clear spatial southeast trend for contamination and relative enrichment on these elements.

The Mortórios area is less contaminated than the old Mondego Sul uranium mine area, before the remediation, where the main contamination is also due to the As and U concentration. The Mondego Sul area also presents contamination in other metals, such as Co, Cu, Pb and the metalloid Sb. The different contamination in the two areas are attributed to the

different production of U<sub>3</sub>O<sub>8</sub> tonnes, which are certainly due to the different origin of the two uranium mineralizations. At Mortórios, autunite and torbernite occur in basic rock dykes which are associated with older quartz veins containing sulfides. However, at Mondego Sul the same uranium minerals occur and meta-uranocircite and meta-saleeite were found and all belong to quartz veins containing sulfides.

The old Mondego Sul area is being remediated. The old Mortórios area must also be remediated.

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