Characterization of seed oils from different varieties of watermelon [*Citrullus lanatus* (Thunb.)] from Pakistan

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RESUMEN

Caracterización de aceites de semillas de diferentes variedades de sandía [*Citrullus lanatus* (Thunb.)] procedentes del Pakistán.

Se presentan las características físico-químicas de aceites de diferentes variedades de semillas de sandías (Citru-Ilus lanatus) cultivadas en Pakistán: Sugar Baby, QF-12, DWH-21 y Círculo rojo-1885. El aceite y el contenido de proteína cruda de las semillas de sandía están dentro de los rangos: 28,25-35,65% y 20,50-35,00%, respectivamente y varian significativamente (p < 0,05) entre las variedades ensayadas. Los niveles de humedad, fibra cruda y cenizas en las semillas se encontró entre 2.16-3.24%, 1.95-3.42% y 4.29-6.60%, respectivamente. Las características fisico-guímicas estudiadas de los aceites extraídos fueron: contenido de ácidos grasos libres (1.17-2.10% de ácido oleico), índice de yodo (97,10-116,32 g de I_2 100 g⁻¹ de aceite), índice de saponificación (190,20-205,57 mg de KOH g⁻¹ de aceite), insaponificable (0,54-0.82%) y color (1.12-4.30 de I + 12.20-33.40 y). Los aceites presentaron unos rangos de los parámetros de oxidación razonables, como se muestra en las determinaciones de la extinción específica a 232 y 270 nm (2.90-4.40 y 2.05-3.09 respectivamente), valores de *p*-anisi-dina (5.60-7.70) e índice de peróxidos (2,90-5,06 meq O_2 kg⁻¹ de aceite). El ácido linoleico es el principal ácido graso que se encuentra en todos los aceites de las semillas, con una contribución del 45.30-51.80% del total de ácidos grasos. Otros ácidos grasos determinados fueron oleico, 20,2-23,5%, palmítico, 15,1-16,9%) y esteárico, 11,5-14,4%. El contenido de α- y δ-tocoferol en los aceites fué de 120,6-195,6 y 9,1-58,3 mg kg⁻¹, respectivamente. Los atributos físico-químicos de los aceites de semillas de sandía variaron significativamente entre las variedades ensayadas. Los resultados del presente estudio indican que las semillas de las variedades de sandía ensayadas procedentes de Pakistán son una fuente potencial aceites con alto contenido en ácido linoleico y, por lo tanto, se puede explorar para usos comerciales y productos con valor añadido.

PALABRAS CLAVE: Aceite de semillas – Ácidos grasos – Atributos físico-químicos – Estado de oxidación – HPLC – Sandía – Tocoferoles.

SUMMARY

Characterization of seed oils from different varieties of watermelon [*Citrullus lanatus* (Thunb.)] from Pakistan.

This paper reports the physicochemical characteristics of the seed oils from different varieties of watermelon (*Citrullus*

lanatus) cultivated in Pakistan, namely Sugar Baby, Q-F-12, D-W-H-21 and Red Circle-1885. The oil and crude protein contents from watermelon seeds, within the range of 28.25 to 35.65% and 20.50 to 35.00%, respectively, varied significantly (p < 0.05) among the varieties tested. The levels of moisture, ash, and crude fiber in the seeds tested were found to be 2.16-3.24%, 1.95-3.42% and 4.29-6.60%, respectively. The physicohemical characteristics of the extracted oils were: free fatty acid contents (1.17-2.10% as extracted one were. The faity and containing $(100g^{-1} \text{ of oil})$, oleic acid), iodine value (97.10-116.32 g of l_2 100g⁻¹ of oil), saponification index (190.20-205.57 mg KOH g of oil). unsaponifiable matter (0.54-0.82%) and color (1.12-4.30 R + 12.20-33.40 Y). The oils revealed a reasonable oxidative parameter range as depicted by the determinations of specific extinction at 232 and 270 nm (2.90-4.40 and 2.05-3.09, respectively), p-anisidine value (5.60-7.70) and peroxide value (2.90-5.06 meqO₂ kg^{-1} of oil). Linoleic acid was the major fatty acid found in all the seed oils with contributions of 45.30-51.80% of the total fatty acids (FA). Other fatty acids detected were known to be oleic acid (20.2-23.5%), palmitic acid (15.1-16.9%) and stearic acid (11.5-14.4%). The contents of α - and δ -tocopherol in the oils accounted for 120.6-195.6 and 9.1-58.3 mg kg⁻¹, respectively. The physicochemical attributes of the watermelon seed oils showed a wider variation among the varieties tested. The results of the present study indicate that the seeds of the tested watermelon varieties from Pakistan are a potential source of high-linoleic oil and thus can be explored for commercial use and value addition.

KEY-WORDS: Fatty acids – HPLC – Oil seeds – Oxidation state – Physicochemical attributes – Tocopherols – Watermelon.

1. INTRODUCTION

Several *Cucurbitaceae* species have been studied as potential sources of nutrients such as protein, minerals and lipids as well as ingredients for native medicine. Especially, the seed kernels from cucurbit fruits and vegetables such as melon, watermelon, cucumber and gourd, traditionally called "Charmagaz" are well-known in the subcontinental region for their therapeutic properties (Achu *et al.*, 2005; Mabaleha *et al.*, 2007).

Watermelon [(*Citrullus lanatus* (Thunb.)], belonging to the *Cucurbitaceae* family, is widespread in tropical and subtropical regions of the world (Achu *et al.*, 2005). Watermelon is

primarily consumed as a fruit because of its sweet flavor and high nutritional value. It has been widely used in several regions of the world, especially in Africa and Asia, not only as a source of nutrition but also as a folk remedy for the treatment of strangury (Artwatermelon, 2007; Ziyada and Elhussien, 2008).

Watermelon seeds have high nutritional value and are utilized for human consumption as snacks after salting and roasting in Arabian and Asian regions. The seeds are a rich source of protein 25.2-37% and oil 37.8-45.4% (Ziyada and Elhussien, 2008). El-Adawy and Taha (2001) investigated the chemical composition and characteristics of watermelon, pumpkin, and paprika seed oils and flours from Egypt. Watermelon seed oil is light in appearance, and is valued as a potential source of essential fatty acid (linoleic acid) with health benefits (Tiawo et al., 2008). Watermelon seed oils, often characterized by high-linoleic acid content, have been studied in different parts of the world such as West Africa and the Middle East (Taiwo et al., 2008; Ziyada and Elhussien 2008). Mabaleha et al. (2007) investigated the properties of selected melon seed oils to determine their potential uses for edible purposes.

All over the world, due to growing vegetable oil demand, much interest is currently being focused on the possibilities of exploring some newer and underutilized plant resources for the production of oils (Anhwange, 2004). For this reason, many fruit seed oils, as newer sources, have been evaluated for specific industrial or food uses (Parry *et al.*, 2005; Mabaleha *et al.*, 2007; Ramadan *et al.*, 2006; Adhikari *et al.*, 2008; Anwar *et al.*, 2008; Celik and Ercisli, 2009; Matthaus and Ozcan 2009; Nehdi, 2011).

According to Pakistan Oilseeds and Products Annual Report (POPA, 2011) vegetable oils imported to Pakistan have been estimated to be 2.16 million metric tons (MMT) for the marketing year 2011-2012. Nearly 75 percent of the total domestic consumption (3.5 MMT) of vegetable oil in Pakistan is met through imports (POPA, 2011) while only 25 percent is derived from the local vegetable oilseed resources. Therefore, it is worthwhile to investigate some non-conventional oil seed resources of Pakistan, to bridge the gap between supply and demand. Although considerable work has been carried out on the nutritional analysis of seed and seed oils from several cucurbit fruits, no data are known on the detailed physicochemical characteristics of fruit seed oils from important cultivars of watermelon (Citrullus lanatus) grown in Pakistan. Watermelon fruit is widely consumed in Pakistan yielding a considerable amount of seeds, which are often discarded as waste. In line with the present needs, these seeds can be used for oil extraction. The present work was therefore designed with the main objective of evaluating the physicochemical characteristics including the fatty acid and tocopherol profiles of seed oils produced from different cultivars of locally grown watermelon.

2. MATERIALS AND METHODS

2.1. Materials

The seeds of four watermelon (*Citrullus lanatus*) varieties (Sugar Baby, Q-F-12, D-W-H-21 and Red Circle 1885) were obtained from the Ayub Agricultural Research Institute, Jhang Road, Faisalabad, Pakistan. Three different seed samples for each of the varieties were harvested. The chemicals and reagents (analytical and HPLC grade) used throughout the experiment were from Merck (Darmstadt, Germany) or Sigma Aldrich (Buchs, Switzerland). Pure standards of tocopherols [DL- α -tocopherol, (+)- γ -tocopherol, (+)- δ -tocopherol], and fatty acid methyl esters (FAMEs) were purchased from Sigma Chemical Co. (St Louis, MO, USA).

2.2. Extraction of oil

Samples of watermelon seeds (dried at 40 °C in an oven for 24 h) were crushed using a commercial blender (TSK-949, Westpoint, France). One hundred grams of well-crushed seeds of each variety were placed in a paper thimble and fed into a Soxhlet extractor. The extractor was fitted with a 1 L round bottomed flask and a water condenser. The oil extraction was carried out with *n*-hexane on a water bath for 6 h. The extra hexane, after oil extraction, was removed by distilling off under low pressure through a rotary evaporator (EYELA, N-N Series; Rikakikai Co Ltd., Tokyo, Japan).

2.3. Analysis of oilseed cake

The oilseed cake, obtained after the oil extraction from the seeds, was analyzed for protein, fiber, and ash contents. Protein content (percent N \times 6.25) was estimated according to the Association of Official Analytical Chemists AOAC standard method 976.06 (AOAC, 1990) using a Kjeldahl apparatus. The fiber and ash contents were determined according to the ISO method 5983 (ISO, 1981) and ISO method 749 (ISO, 1977), respectively.

2.4. Physical and chemical parameters of oils

The extracted oils were analyzed for density, refractive index, iodine value (IV), peroxide value (PV), acidity, saponification index and unsaponifiable matter following AOCS official methods Cc 10a–25, Cc 7–25, Cd 1–25, Cd 8–53, F 9a–44, Cd 3-25, and Ca 61–40, respectively (AOCS, 1997). The oil color, in a 1-in. cell, was checked using a Lovibond Tintometer (Tintometer Ltd., Salisbury, Wiltshire, United Kingdom). Specific extinctions at 232 and 270 nm, representing the magnitude of primary and secondary oxidation products of oils, were monitored using a spectrophotometer (U-2001; Hitachi, Instruments Inc., Tokyo, Japan). The oil samples were mixed with *iso*-octane to take the absorbance

readings and $[(\epsilon^{1\%}_{1cm}(\lambda)]$ were calculated according to the IUPAC method II D.23 (IUPAC, 1987).The *p*-anisidine value was determined according to IUPAC method II. D. 26 (IUPAC, 1987). The oil samples, dissolved in *iso*-octane, were allowed to react with *para*-anisidine for 10 min. The absorbance of the colored complex formed was recorded at 350 nm, using a spectrophotometer.

2.5. Fatty acid composition

The oil samples were transformed into fatty acid methyl esters (FAMEs) through transmethylation according to the IUPAC standard method 2.301 (IUPAC, 1987) and analyzed by a SHIMADZU gas chromatograph model 17-A, fitted with an FID. The separation of FAMEs was done on an SP-2330 (SUPELCO, INC., Supelco Park, Bellefonte, PA, 16823-0048 USA) methyl-lignocerate-coated (film thickness 0.20 μ m), polar capillary column (30 m \times 0.32 mm). Extra pure nitrogen gas at a flow rate of 3.5 mL min-1 was used as the mobile phase. The column oven temperature started at 180°C and was raised to 220°C at the linear ramp rate of 5°C min⁻¹. The initial and final column temperature was held for 2 and 10 minutes, respectively. The injector and detector were adjusted to 230 °C and 250 °C, respectively. The identification of the unknown FAMEs was based on matching their retention times with those of pure standards and quantified on the basis of peak areas by a Chromatography Station for Windows (CSW32) data-handling program (Data APEX Ltd., Pague 5, The Czech Republic). The FA composition was expressed as a relative percentage of the total peak area.

2.6. Tocopherol content

Tocopherols (α and δ) were analyzed by HPLC according to the Current Protocols in Food Analytical Chemistry methods (Wrolstad, 2003). The oil (0.1g), mixed with 0.05 g ascorbic acid to protect from oxidation, was placed in an appropriate test tube. Five mL of 90% ethanol and 0.5 mL of 80% aqueous KOH solution were added to the test tube and the reaction mixture was mixed well by vortexing for 30 s. The test tube was flushed with nitrogen to provide an inert environment, capped and then incubated in a water bath (70°C) for 30 min with periodic vortexing. After that, the tube was removed and placed in an ice bath for 5 min, then 3 mL deionized water and 5 mL n-hexane were added and the mixture was vortexed for 30 s. The tube was subjected to centrifugation at $1,000 \times q$ for 10 min, at room temperature and the upper hexane layer was recovered in another test tube. The aqueous layer and the residue were reextracted using the same procedure. The upper hexane layers from both the extractions were combined and evaporated to dryness under nitrogen. The extract recovered was dissolved in one mL mobile phase, mixed well and then transferred to an HPLC sample vial. A $20-\mu$ L sample was injected into a Supelcosil LCSi column (250×4.6 mm, Supelco Inc.). Elution was made using a mobile phase of ethyl acetate/acetic acid/ hexane (1:1:198, v/v/v) at the flow rate of 1.5 mL min⁻¹. Detection was performed at 295 nm. Tocopherols were identified by matching their retention times with those of pure standards of α and δ -tocopherols and were quantified on the basis of peak areas based on an external standard method. A D-2500 Hitachi Chromatointegrator model with a built-in computer program for data handling was used for quantification.

2.7. Statistical Analysis

All the measurements were made in triplicate and the results statistically compared and analyzed by using one way analysis of variance (ANOVA). The values are reported as means with standard deviations for three replicates (Steel *et al.*, 1997). A probability value at $p \le 0.05$ was considered to denote statistically significant differences.

3. RESULTS AND DISCUSSION

3.1. Proximate composition of seeds

The results for the proximate composition of four different varieties of Pakistani watermelon seeds are depicted in Table 1. Hexane-extracted oil contents from watermelon seeds varied from 28.25-35.65% (g $100g^{-1}$ dried seed weight basis) among the varieties tested. The highest (35.65%) oil yield was exhibited by watermelon var. Q-F-12, whereas the lowest (28.25%) by var. Red circle-1885 showing significant variation (p < 0.05) among the varieties tested which might be linked to the their differing genetic makeup. The watermelon seed oil contents as determined in the present work was comparable to that reported (35.50%) for watermelon (Citrullus lanatus) seeds var. Colocynthoide investigated from Sudan (Ziyada and Elhussien, 2008). However, the present oil yield was lower than that reported by Ovedeji and Oderinde (2006) for Nigerian watermelon seeds (oil contents 45%). Another study conducted by Baboli and Kordi (2010) revealed the hexane-extracted oil content from watermelon seeds (50%) to be significantly higher than the present data. Such variation in oil yield from different regions might be attributed to the varied agroclimatic conditions employed for the growing of such fruits. The levels of seed oil contents (28.25%-35.65%) were found to exceed those of three other conventional oilseed crops including cotton (15.0-24.0%), soybean (17.0-21.0%) and olive (20.0-25.0%) (Pritchard, 1991).

The amount of seed protein, determined as 20.50-35.00%, varied significantly (P < 0.05) among the varieties tested. Watermelon var. D-W-H-21 exhibited the highest protein contents

Devemeters	Varieties				
Farameters	Sugar Baby	Q-F-12	D-W-H-21	Red Circle-1885	
Oil	30.08 ± 1.90^{ab}	35.65 ± 2.30^{bc}	32.22 ± 2.09^{b}	28.25 ± 1.50^{a}	
Ash	3.00 ± 0.12^{cd}	2.00 ± 0.10^{ab}	1.95 ± 0.90^{ab}	3.42 ± 0.15^{cd}	
Protein	20.5 ± 1.09^{a}	24.5 ± 1.50 ^b	35.00 ± 2.50^{d}	$29.00 \pm 1.42^{\circ}$	
Moisture	2.16 ± 0.04^{a}	3.14 ± 0.05^{cd}	$3.24 \pm 0.08^{\circ}$	$3.00 \pm 0.04^{\text{bc}}$	
Fiber	5.29 ± 0.50^{b}	$6.60 \pm 0.11^{\circ}$	4.29 ± 0.08^{a}	5.23 ± 0.09^{b}	

Table 1 Proximate composition (g 100 g⁻¹ dried seed weight basis) of different varieties of watermelon seeds

Different superscript letters within the same row indicate significant differences (p < 0.05) among

varieties tested

Values are mean ± SD of triplicate determinations

(35.00%) whereas the var. Sugar Baby had the lowest (20.50%). The protein contents of Pakistani watermelon seeds were in line with those reported (26.33%) for watermelon seeds from Nigeria (Taiwo *et al.* 2008). The ash and fiber contents of the tested watermelon seeds ranged from 1.95-3.42% and 4.29-6.60%, respectively; although they did not compare with watermelon seeds from other regions. However, they were in line with a previous investigation (3.95% and 3.44%, respectively) of another melon type, the *Cucurbita maxima* seed (Achu *et al.*, 2005).

3.2. Physiochemical characteristics of oils.

The results of various physicochemical characteristics of the extracted oils from the different varieties of watermelon seeds are given in Table 2.

Among the varieties tested, the seeds oils investigated exhibited no significant (P > 0.05) variation with regard to the values of refractive index (40 °C) and density (24 °C), which ranged from 1.4665-1.4670 and 0.902-0.935 mg mL⁻¹, respectively. The present results were quite comparable to those reported for refractive index

(1.465-1.469) and density (0.874-0.954 mg/mL), in five different varieties of melon seeds including one var. of *Citrullus lanatus* from Ghana (Mabaleha *et al.*, 2007). Among the varieties tested, Red circle-1885 showed the lowest value for density (0.902) and the highest for refractive index (1.4670). While Sugar Baby had the highest density (0.935), it also revealed the lowest refractive index (1.4665). Pure oils have characteristic refractive index and density ranges thus the degree of variation of such data from the true value may indicate its relative purity.

The color values for the investigated watermelon seed oils (12.2-33.4Y/ 2.1-4.3R), although varying significantly (P < 0.05) within the varieties tested, were quite low, indicating that these oils could be employed for edible or other oleochemical uses after slight bleaching. The development of color in vegetable oils is mainly linked with the presence of some coloring matters/pigments, for example, chlorophyll and carotenoids which are extracted along with the oil during the process of oilseed extraction. Such coloring components have to be thoroughly removed through a step known as oil bleaching. Lighter colored oils, being valuable from

Table 2
Physicochemical characteristics of different varieties of watermelon seed oils

Deremetere	Varieties			
Farameters	Sugar Baby	Q-F-12	D-W-H-21	Red Circle-1885
Free fatty acid (% as oleic acid)	1.30 ± 0.30^{ab}	1.17 ± 0.35^{a}	$1.90 \pm 0.45^{\rm bc}$	$2.10 \pm 0.50^{\circ}$
lodine value (g of $I_2 100 \text{ g}^{-1}$ of oil)	97.10 ± 2.14^{a}	103.25 ± 2.50^{b}	116.32 ± 2.90^{cd}	$114.00 \pm 2.62^{\circ}$
Density (24 °C, mg mL ⁻¹)	0.93 ± 0.12^{b}	0.91 ± 0.10^{a}	0.91 ± 0.20^{a}	0.90 ± 0.15^{a}
Refractive index (40 °C)	1.4665 ± 0.002^{ab}	1.4668 ± 0.005^{cd}	1.4667 ± 0.001^{bc}	1.4670 ± 0.003^{cd}
Saponification index (mg of KOH g^{-1} oil)	199.81 ± 3.15 ^c	205.57 ± 3.20^{d}	196.84 ± 3.12^{bc}	190.20 ± 3.00^{a}
Unsaponifiable mater (%)	$0.82 \pm 0.10^{\circ}$	0.54 ± 0.16^{a}	0.56 ± 0.15^{ab}	0.63 ± 0.12^{b}
Color (yellow unit)	33.40 ± 1.60^{d}	33.30 ± 1.42^{d}	12.20 ± 1.60^{a}	16.85 ± 1.90^{b}
Color (red unit)	2.10 ± 0.21^{b}	4.30 ± 0.15^{d}	2.10 ± 0.08^{b}	1.12 ± 0.12^{a}

Different superscript letters within the same row indicate significant differences (p < 0.05) among varieties tested Values are mean \pm SD of triplicate determinations

a technical view-point, are thus more acceptable for edible or oleochemical applications.

The contents of free fatty acids (FFA), calculated as percent oleic acid, for the different varieties of watermelon seed oils were found to be 1.17-2.10%, with the highest value in watermelon var. Red circle-1885 while the lowest was found in var. Q-F-12. FFAs are the measure of extent to which hydrolysis has hydrolyzed and liberated the oil fatty acids from their parent molecule ester linkages. The contents of FFA as determined in the present analysis were found to be higher than those reported (1.20-4.00%) earlier by Mabaleha et al., (2007) for five different varieties of melon seed oils from Ghana. Generally, a higher FFA value of the oil correlates with a higher magnitude of hydrolytic deterioration, thus poor oil quality due to the generation of objectionable flavors and bad odors.

The saponification index (SI) was determined to be 190.20-205.57 (mg of KOH g⁻¹ of oil) for the tested oils and differed significantly (P < 0.05) among the varieties analyzed. These SI are in line with those reported (200.00 mg of KOH g⁻¹ of oil) by Baboli and Kordi (2010) for watermelon seed oil, however higher than those (182.10-193.80 mg of KOH g⁻¹ of oil) investigated for different melon seed oils from Ghana (Mabaleha *et al.*, 2007). Such differences in SI might be attributed to the replacement of long chain (C₂₂) fatty acids by shorter chain C₁₈ fatty acids of the oils analyzed (Anwar *et al.*, 2006).

The amount unsaponifiable matter (USM) of the investigated varieties of watermelon oils ranged from 0.54-0.82%. These values were lower than those reported (0.68-4.18%) earlier for melon seeds oil by Anwer *et al.*, (2006). Higher oil USM reflects the presence of higher concentration of valuable minor components such as tocopherols, carotenoids, squalene and phytosteroals which not only impart oxidative stability to the oils but also have potential medicinal value. Mostly the concentration of USM constitutes up to 0.50 to 2.50%, however in exceptional cases 5 to 6% of vegetable oils (Bockisch, 1993; Gordon and Magos, 1983; Malecka, 1994).

The iodine value (IV) of the investigated oils, ranging from 97.10-116.32 (g of $I_2 100g^{-1}$ of oil), varied significantly (P < 0.05) among the varieties analyzed. These results are in close agreement to

those reported in for five varieties of melon oil $(95.80-124.00 \text{ g of } \text{I}_2 \ 100 \text{g}^{-1} \text{ of oil})$ investigated by Mabaleha et al., (2007) from Ghana. It is well recognized that IV is the measurement of the degree of unsaturation of the fatty acids present in an oil, principally the oleic and linoleic acids. A high iodine value means that the oil is more unsaturated and could be used for edible purposes. This value is helpful in determining the quality of the oil, whether as a drying, semi-drying or non-drying oil. The iodine value is also related to the melting point or hardness of an oil or fat. The iodine value of an oil is reported to decrease as a result of unfavorable storage conditions, especially temperature, due to the oxidation of unsaturated fatty acids (Rossell, 1991).

3.3. Oxidation parameters of seed oils

The oils extracted from the tested varieties of watermelon seeds showed good oxidation states as indicated by the data for the different parameters presented in Table 3.

The specific extinction at 232 nm and 270 nm, which revealed the magnitude of oxidative deterioration (Rossell, 1991; Frankel, 2005), of the watermelon seed oils ranged from 2.90-4.40 and 2.05-3.09, respectively, indicating a significant (p < 0.05) variation among the varieties tested. These values of specific extinctions at 232 nm and 270 nm which were found to be comparable with those investigated (4.80 and 3.52, respectively) in another cucurbit *i.e.* pumpkin (*Cucurbita pepo*) seed oil analyzed by Ardabili *et al.*, (2011) from Iran, could not be specifically compared with watermelon seed oils due to the lack of investigations in the literature.

The peroxide value (PV) is a measurement of those substances in a sample expressed in terms of mili-equivalents of active oxygen per kilograms of oil sample which oxidize potassium iodide under the conditions of the test. PV is usually used to measure the oxidative rancidity of an oil or fat. The PV of watermelon seed oils ranged from 2.90 to $5.06 \text{ meqO}_2 \text{ kg}^{-1}$ with the lowest value for the variety Sugar Baby while the highest value was found for Q-F-12. The present PV, noted to be considerably lower than that investigated in pumpkin seed oil

Table 3
Oxidation parameters of different varieties of watermelon seed oils

Devemetere	Varieties			
Parameters	Sugar Baby	Q-F-12	D-W-H-21	Red Circle-1885
Conjugated diene $\epsilon^{1\%}_{1cm}$ (λ_{232})	3.66 ± 0.02^{bc}	$3.94 \pm 0.11^{\circ}$	4.40 ± 0.25^{d}	2.90 ± 0.11^{a}
Conjugated triene $\epsilon^{^{1\%}}{}_{^{1cm}}$ ($\lambda_{^{270}})$	$3.09 \pm 0.15^{\circ}$	2.15 ± 0.13^{ab}	2.75 ± 0.12^{bc}	2.05 ± 0.09^{a}
Peroxide value (meqO ₂ kg ⁻¹ of oil)	2.90 ± 0.19^{a}	5.06 ± 0.25^{d}	3.3 ± 0.13^{b}	$4.62 \pm 0.08^{\circ}$
<i>P</i> -anisidine value	5.60 ± 0.09^{a}	6.23 ± 0.11^{ab}	7.70 ± 0.15^{d}	$6.35 \pm 0.12^{\circ}$

Different superscript letters within the same row indicate significant differences (p < 0.05) among varieties tested Values are mean \pm SD of triplicate determinations

 $(10.85 \text{ megO}_2 \text{ kg}^{-1})$ (Ardabili *et al.*, 2011), varied significantly ($\overline{P} < 0.05$) among the varieties selected.

The *p*-anisidine values of watermelon seed oils, within the range of 6.35 to 7.70, indicated a reasonable resistance to secondary oxidation of these oils. Rancid and off-flavors are generally produced due to the presence of aldehydes of short or medium chain length. The present PV and p-anisidine values of Pakistani watermelon oils were found to be within the range of those reported (1.10-10.90 meqO₂ kg^{-1} and 0.20-9.00, respectively) earlier for five varieties of melon seed oils investigated from Ghana by Mabaleha et al., (2007).

3.4. Tocopherols of seeds oils

The data for the quantification of tocopherols (α and δ) of the seed oils of the four varieties of watermelon is presented in Table 4. The total amounts, ranging from 140.6 to 222.6 mg/kg, of these two tocopherol isomers varied significantly (p < 0.05) among the varieties tested. The contents (mg kg⁻¹) of α -tocopherol varied significantly (P < 0.05) from 120.62-195.60 mg/kg with the highest level for the cultivar Sugar Baby and lowest in D-W-H-21. Data are rarely available on the tocopherol contents of melon seed oil, yet the present α -tocopherol levels are considerably higher than those found (27.10-75.10 mg kg⁻¹) in other Cucurbita maxima melon seed oil (Stevenson et al., 2007). On the other hand, the present δ -tocopherol content, ranging from 9.08 to 58.29 mg kg⁻¹, was found to be lower than that reported (74.90-492.80 mg/kg) for Cucurbita maxima seed oil (Stevenson et al., 2007). Gamma-tocopherol (y-tocopherol) and β-tocopherol were not detected in the present analysis of watermelon oils.

A considerable amount of α -tocopherol (120.62-195.60 mg kg⁻¹) was detected in the present analysis of watermelon oils was found to be higher than those reported for palm oil (4-185 mg kg⁻¹) and coconut oil (not detected/ND-17 mg kg⁻¹), while it was within the limits for soybean (9-352 mg kg⁻¹), maize (23-573 mg kg⁻¹), ground nut (49-304 mg kg⁻¹) and high and low-erucic acid rapeseed (39-320 mg kg⁻¹) oils (Rossell,1991). The δ -tocopherol contents (9.08 and 58.29 mg kg^{-1}), which were in close agreement with that of maize (23-75 mg kg^{-1}), and lower than soybean (154-932 mg kg⁻¹), were established to be higher than those of coconut (ND-

2 mg kg⁻¹), cottonseed (ND-17 mg kg⁻¹), ground nut (3-22 mg kg⁻¹), palm (traces), sunflower (ND-7 mg kg⁻¹), low, and high-erucic acid rapeseed (4-22 mg kg⁻¹) oils (Rossell,1991).

Tocopherols are minor components that contribute significantly to the oxidative stability of vegetable oils during storage and processing. They are naturally present in vegetable oilseeds and extracted along with the oils, however, their concentration usually decreases as result of oil processing. In vegetable oils, their original concentration may vary from a few mg/kg to a thousand mg kg⁻¹ or even more depending on the oil type and oil fatty acid composition (Rossell, 1991). Of the different isomers of tocopherols reported so far, α isomer has the greatest vitamin E efficacy, while δ -exhibits the highest antioxidant activity (Timmermann, 1990; Sies and Murphy, 1991; Matthaus and Ozcan, 2009).

3.5. Fatty acid composition

The data related to fatty acid composition for the seed oils of different varieties of watermelon as analyzed by GLC-FID are shown in Table 5. The contents of palmitic acid and stearic acid varied from 14.3-16.2% and 11.2-13.8%, respectively. The total amounts of these two saturated fatty acids (TSFA) were the highest in the watermelon var. D-W-H-21 (30.0%) while the lowest were found in var. Sugar Baby (26.2%). The present TSFA levels were found to be higher than those reported (17.8%) earlier in Citrullus lanatus var. Colocynthoide (Ziyada and Elhaussien, 2007). These TSFA varied slightly from those (12.0-36.0%) investigated for different species of musk melon seed oils (Tilak et al., 2006).

The watermelon seed oils tested contained relatively higher levels (68.1-72.3%) of unsaturated fatty acids with linoleic acid (45.1-51.2%) as the major component. The oils also contained appreciable amounts of oleic acid with contributions of 20.2 to 23.0%. The contents of total unsaturated fatty acids (TUFA), comprising linoleic and oleic acids, was found to be the highest (72.3%) in the watermelon var. Sugar Baby while the lowest (68.1%) was found for D-W-H-21. The contents of TUFA in the present analysis were found to be lower than those reported by Baboli and Kordi (2010) (81.6%). However, these levels of TUFA were in line with those detected (67.93-82.36%) in five species of melon by Mabaleha et al., (2007).

Tocopherol contents of different varieties of watermelon seed oils				
To contain $(ma ka^{-1})$	Varieties			
	Sugar Baby	Q-F-12	D-W-H-21	Red Circle-1885
α-tocopherol	195.6 ± 12.20^{d}	$164.3 \pm 5.60^{\circ}$	122.0 ± 7.10^{ab}	120.6 ± 10.25^{a}
δ-tocopherol	12.3 ± 0.90^{a}	58.3 ± 1.20^{d}	9.1 ± 0.50^{a}	20.0 ± 1.10^{b}
Total tocopherol	207.9 ^c	222.6 ^{cd}	131.1 ^ª	140.6 ^{ab}

Table /

Different superscript letters within the same row indicate significant differences (p < 0.05) among varieties tested Values are mean ± SD of triplicate determinations

Faily acid composition (g roog of faily acids) of watermeton seed ons					
Fotty opida	Varieties				
Fally acids	Sugar Baby	Q-F-12	D-W-H-21	Red Circle 1885	
C _{16:0}	15.0 ± 0.05^{a}	15.1 ± 0.12^{ab}	16.2 ± 0.08^{b}	14.3 ± 0.99^{a}	
C _{18:0}	11.2 ± 0.52^{a}	$12.5\pm0.06^{\text{bc}}$	$13.8\pm0.06^{\text{cd}}$	$12.3 \pm 0.62^{\circ}$	
C _{18:1}	21.2 ± 0.45^{b}	20.2 ± 0.95^{a}	$23.0 \pm 0.09^{\circ}$	20.2 ± 0.65^{ab}	
C _{18:2}	51.1 ± 0.09^{d}	50.5 ± 0.09^{d}	45.1 ± 0.80^{a}	51.2 ± 0.05^{d}	
Unidentified	1.5	1.7	1.9	2.0	
TSFA	26.2	27.6	30.0	26.6	
TUFA	72.3	70.7	68.1	71.4	
TEFA	51.1	50.5	45.1	51.2	

	Table 5
Fatty acid composition (g 100g	⁻¹ of fatty acids) of watermelon seed oils

Different letters in superscript within the same row indicate significant differences (p < 0.05) among the varieties tested Values are mean \pm SD of triplicate determinations

TSFA: Total saturated fatty acids

TUSF: Total unsaturated fatty acids

TEFA: Total essential fatty acid

The amount of TUSFA in the present analysis are comparable with those reported (64.6-88.2%) for different species of muskmelon (Tilak *et al.*, 2006).

The contents of linoleic acid, an essential fatty acid (EFA), were 45.1-51.2%. EFA have potential health benefits related to cardiovascular disorders (Nicolosi *et al.*, 2004, Manzoor *et al.*, 2007). The fatty acid profile of watermelon seeds oil shows relatively low amounts (26.2-30.0%) of TSFA and high amounts of EFA, which is advantageous due to the fact that a diet low in saturates can benefit patients with cardiovascular diseases (Nicolosi *et al.*, 2004).

4. CONCLUSION

The results of the present study demonstrated that the seeds from the watermelon varieties tested are a good source of high-linoleic oil that might be utilized for human consumption as well as for other oleochemical purposes. In addition, it was evident from the present findings that the oil yield and most of the other physicochemical attributes of the oils extracted considerably varied among the varieties of watermelon tested.

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