



Missouri University of Science and Technology Scholars' Mine

Materials Science and Engineering Faculty Research & Creative Works

Materials Science and Engineering

01 Apr 2015

Impact of Symmetry on the Ferroelectric Properties of CaTiO3 Thin Films

Michael D. Biegalski

Liang Qiao

Yijia Gu

Missouri University of Science and Technology, guyij@mst.edu

Apurva Mehta

et. al. For a complete list of authors, see https://scholarsmine.mst.edu/matsci_eng_facwork/2479

Follow this and additional works at: https://scholarsmine.mst.edu/matsci_eng_facwork



Part of the Materials Science and Engineering Commons

Recommended Citation

M. D. Biegalski et al., "Impact of Symmetry on the Ferroelectric Properties of CaTiO3 Thin Films," Applied Physics Letters, vol. 106, no. 16, American Institute of Physics (AIP), Apr 2015. The definitive version is available at https://doi.org/10.1063/1.4918805

This Article - Journal is brought to you for free and open access by Scholars' Mine. It has been accepted for inclusion in Materials Science and Engineering Faculty Research & Creative Works by an authorized administrator of Scholars' Mine. This work is protected by U. S. Copyright Law. Unauthorized use including reproduction for redistribution requires the permission of the copyright holder. For more information, please contact scholarsmine@mst.edu.



Impact of symmetry on the ferroelectric properties of CaTiO₃ thin films

Michael D. Biegalski, ¹ Liang Qiao, ^{1,a)} Yijia Gu, ² Apurva Mehta, ³ Qian He, ⁴ Yayoi Takamura, ^{5,b)} Albina Borisevich, ⁴ and Long-Qing Chen²

¹Center for Nanophase Materials Sciences, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA

²Department of Materials Science and Engineering, Pennsylvania State University, University Park, Pennsylvania 16801, USA

³Stanford Synchrotron Lightsource SLAC National Accelerator Laboratory, Menlo Park, California 94025, USA

⁴Materials Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA

⁵Department of Chemical Engineering and Materials Science, University of California Davis, Davis, California 95616, USA

(Received 17 February 2015; accepted 10 April 2015; published online 22 April 2015)

Epitaxial strain is a powerful tool to induce functional properties such as ferroelectricity in thin films of materials that do not possess ferroelectricity in bulk form. In this work, a ferroelectric state was stabilized in thin films of the incipient ferroelectric, CaTiO₃, through the careful control of the biaxial strain state and TiO₆ octahedral rotations. Detailed structural characterization was carried out by synchrotron x-ray diffraction and scanning transmission electron microscopy. CaTiO₃ films grown on La_{0.18}Sr_{0.82}Al_{0.59}Ta_{0.41}O₃ (LSAT) and NdGaO₃ (NGO) substrates experienced a 1.1% biaxial strain state but differed in their octahedral tilt structures. A suppression of the out-of-plane rotations of the TiO₆ octahedral in films grown on LSAT substrates resulted in a robust ferroelectric *I4 mm* phase with remnant polarization \sim 5 μ C/cm² at 10 K and T_c near 140 K. In contrast, films grown on NGO substrates with significant octahedral tilting showed reduced polarization and T_c . These results highlight the key role played by symmetry in controlling the ferroelectric properties of perovskite oxide thin films. © 2015 AIP Publishing LLC. [http://dx.doi.org/10.1063/1.4918805]

Calcium titanate (CaTiO₃) is the prototype for the perovskite structure. However, at room temperature, CaTiO₃ has the orthorhombic GdFeO₃ crystal structure (a = 5.38 Å, $b = 5.44 \,\text{Å}, c = 7.65 \,\text{Å}$) and only exhibits the cubic $(Pm\bar{3}m)$ perovskite structure above \sim 1575 K. Below this temperature, it takes the tetragonal I4/mcm structure and then transitions to the orthorhombic *Pnmb* structure below $\sim 1525 \, \mathrm{K.}^{1,2}$ It has been shown that the orthorhombic Pnmb tilt structure can be altered using epitaxial strain in thin films, 3,4 where the symmetry changes at the phase transitions are due to tilting of the TiO₆ oxygen octahedral away from cubic symmetry. The $Pm\bar{3}m$ to 14/mcm transition results from out-of-phase rotations of the octahedra around the c-axis, from $a^0a^0a^0$ to $a^0a^0c^-$ tilts according to Glazer notation.⁵ The *I4/mcm* to *Pnmb* phase transition involves a change from out-of-phase to in-phase tilts along the c-axis and additional out-of-phase tilts along the a- and b-axes, which can be represented by the $a^-a^-c^+$ tilt pattern.

CaTiO₃ is similar to ferroelectric titanates such as BaTiO₃ and PbTiO₃ in bonding and composition; however, it does not undergo any ferroelectric transitions. Rather, CaTiO₃ is an incipient ferroelectric, akin to KTaO₃ and SrTiO₃, in which it has a polar soft mode but does not exhibit a ferroelectric phase transition due to quantum fluctuations.^{6,7} This instability can be overcome and a robust ferroelectric state can be created in bulk materials with the substitution of Ca with larger ions.⁸ For SrTiO₃ and KTaO₃, the ferroelectric

state can also be stabilized through epitaxial strain. ^{9,10} The instability of the ferroelectric state in CaTiO₃ has been shown to be correlated to the competition between octahedral rotations and displacement of the Ca and Ti ions. It was predicted that suppression of these octahedral tilts would stabilize the $P4\,mm$ structure and thus allow for a large polarization. ¹¹ Recent calculations have shown that an epitaxial strain of $\sim 1.5\%$ (tensile) can enhance the ferroelectric instability and lead to a ferroelectric state. ^{12,13} Experimental evidence of ferroelectricity in CaTiO₃ has been limited to extrinsic switching due to oxygen vacancies. ¹⁴

In order to better understand the effects of contrasting strain states and in-plane symmetries of CaTiO₃ on the stabilization of ferroelectric properties, we examine the structural and ferroelectric properties of CaTiO₃ thin films grown on orthorhombic (101)-oriented NdGaO₃ (NGO) and cubic (001)-oriented La_{0.18}Sr_{0.82}Al_{0.65}Ta_{0.35}O₃ (LSAT) substrates. These substrates have nearly identical pseudo-cubic lattice parameters, but differing octahedral tilt structures. The octahedral tilt structures imprinted into the CaTiO₃ films were determined by mapping intensities of half order Bragg peaks and scanning transmission electron microscopy (STEM). Both experiment and phase-field modeling indicate that the combined effects of the suppression of octahedral tilts and epitaxial strain lead to robust ferroelectric properties in thin films grown on LSAT substrates.

CaTiO₃ films were grown by pulsed laser deposition on LSAT and NGO substrates with lattice parameters $a_{LSAT} = 3.87 \text{ Å}$, $a_{NGO} = 5.43$, $b_{NGO} = 5.50 \text{ Å}$, and $c_{NGO} = 7.71 \text{ Å}$. The films were grown in 100 mTorr pressure at 650 °C with

a)Present address: School of Materials, the University of Manchester, Manchester M13 9PL, United Kingdom.

b)ytakamura@ucdavis.edu

a laser fluence of 1.5 J/cm² from a CaTiO₃ ceramic target. These conditions were optimized to ensure the structural properties and cation stoichiometry of the films. The films have roughness of approximately 1 unit cell and grew in a layer-by-layer growth mode as evidenced from reflection high energy electron diffraction oscillations. The film thickness was limited to $\sim 10 \,\mathrm{nm}$ to prevent relaxation of the epitaxial strain which was found to occur for thicknesses exceeding ~20 nm. X-ray diffraction indicated that both films were fully commensurate with the substrate with cubeon-cube-type epitaxy. In Fig. 1(a), the out-of-plane lattice spacing for the films grown on LSAT substrates was found to be slightly smaller with a pseudo-cubic lattice parameter, $c_{PC} = 3.801 \pm 0.002 \,\text{Å}$, as compared to films on NGO substrates with $c_{PC} = 3.816 \pm 0.002 \,\text{Å}$, yet both films exhibit thickness fringes indicative of a smooth surface. From reciprocal space maps (not shown), the in-plane lattice constants for films on LSAT substrates were determined to be $a_{PC} = b_{PC} = 3.868 \pm 0.002 \,\text{Å}$, yielding an in-plane strain of 1.2%, while CaTiO₃ films grown on NGO substrates have $a_{PC} = 3.855 \pm 0.003 \,\text{Å}$ and $b_{PC} = 3.865 \pm 0.003 \,\text{Å}$, giving an average in-plane strain of $\sim 1.1\%$. Though both films have similar strain states with nearly identical unit cell volumes, their crystal structures differ markedly, as seen by scans around the 202_{PC}-type peaks (Figs. 1(b) and 1(c)). CaTiO₃ films on LSAT substrates show nearly overlapping scans along the orthogonal 202_{PC} and 022_{PC} directions indicating a tetragonal-like structure. In contrast, similar scans for CaTiO₃ films grown on NGO substrates show three distinct lattice spacings along different 202_{PC}-type directions (orthorhombic 004, 242, and 400 reflections) indicating a monoclinic structure.

In order to accurately determine the intensities of the weak half integer peaks that correspond to the oxygen octahedral tilts, these films were examined using high brightness synchrotron based diffraction. Since NGO and bulk CaTiO₃ share the same *Pnmb* space group, both substrate and thin film half order peaks are present as shown in Figures 2(a)–2(c). The observed peaks corresponded to reflections with *hkl* indices of the form (odd-even-odd)/2 and (odd-odd-odd)/2. The $\frac{3}{2}$ 2 $\frac{5}{2}$ peak (Fig. 2(a)) corresponds to in-phase tilts along the *b*-axis (i.e., b^+ tilts). In contrast, the $\frac{3}{2}$ $\frac{3}{2}$ $\frac{5}{2}$ and $\frac{1}{2}$ $\frac{5}{2}$ $\frac{5}{2}$ peaks correspond to tilts along the *a*- and *c*-axes, respectively, that are antiphase, i.e., neighboring octahedra along these axes tilt in opposite directions. The $\frac{1}{2}$ $\frac{5}{2}$ $\frac{5}{2}$ peak is much

weaker than the $\frac{3}{2}\frac{3}{2}\frac{5}{2}$ peak, indicating that the c^- tilts are greatly reduced. These results show that the film possesses the $a^-b^+c^-$ tilt structure with out-of-plane tilts greatly reduced from bulk symmetry. This finding is in agreement with the previous work asserting that the octahedral tilt structure for tensile-strained Pnmb SrRuO3 films on NGO substrates yielded an $a^-b^+c^0$ or $a^-b^+c^-$ tilt structure.⁴ Thus, the 1% tensile strain state suppresses the out-of-plane tilts but CaTiO₃ films grown on NGO substrates still remain members of the *Pnmb* space group. 15 In contrast, CaTiO₃ films grown on LSAT substrates possess a strikingly different octahedral tilt structure. Only half order peaks of the (odd-even-odd)/2 or (even-odd-odd)/2 type are present, indicating that the octahedral tilt structure has in-phase tilts around the in-plane directions and suppressed tilting around the out-of-plane direction. Moreover, these films appear to be twinned since all half order peaks are doublets as seen in Figures 2(d) and 2(e). This twinning is consistent with the fact that the LSAT substrate has equal in-plane lattice constants and therefore lends no preferred in-plane orientation allowing the CaTiO₃ film to twin around the c-axis. These twins are consistent with 110-type twin boundaries which are prevalent in bulk CaTiO₃. ¹⁷ The pairs of half order peaks have nearly equal intensities, indicating approximately equal proportions of both domains. This result strongly suggests that CaTiO₃ films on LSAT substrates assume the P42/nmc or I4/mmm space group, depending on whether the tilting is maintained along the c-axis.

The octahedral tilts in the CaTiO₃ films were further investigated by STEM high angle annular dark field (HAADF) and annular bright field imaging (ABF) along the pseudo-cubic [010]_{pc} direction. HAADF and ABF STEM images were taken using an aberration-corrected Nion UltraSTEM operating at 200 kV, equipped with a cold fieldemission electron gun and a corrector of third- and fifthorder aberrations. The convergence semi-angle for the electron probe was ~ 30 mrad. ABF signals for the samples were collected from a detector angle range of ~15-30 mrad, and the inner collection angle of HAADF images was 63 mrad. Figure 3 shows that although the interfaces of the CaTiO₃ films on both the LSAT and NGO substrates show a similar level of crystalline quality, the BO₆ octahedral connectivity at these two interfaces differs dramatically. The CaTiO₃/ NGO interface (Figs. 3(c) and 3(d)) maintains a nearly constant tilt angle (β) across the interface due to the common

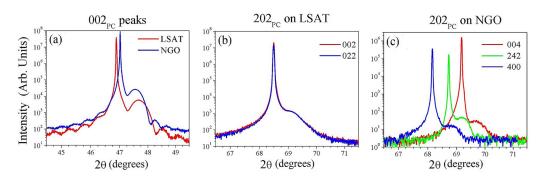


FIG. 1. (a) Out-of-plane diffraction around the 002_{PC} peak showing the different lattice spacing for the films grown on LSAT and NGO substrates. The diffraction from orthogonal 202_{PC} type peaks for the film grown on (b) LSAT and (c) NGO substrates indicating a tetragonal structure of CaTiO₃ grown on LSAT and a monoclinic structure for the films grown on NGO.

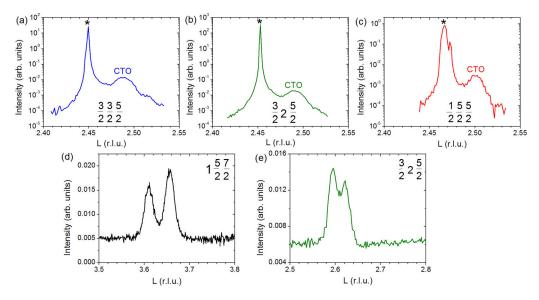


FIG. 2. L-scans around the (a) $\frac{3}{2}$ $\frac{3}{2}$ $\frac{5}{2}$, (b) $\frac{3}{2}$ $\frac{2}{2}$ $\frac{5}{2}$, and (c) $\frac{1}{2}$ $\frac{5}{2}$ $\frac{5}{2}$ half order peaks for CaTiO₃ films grown on NGO substrates. Substrate peaks are denoted by (*) and film peaks by (CTO). L-scans around the (d) $1\frac{5}{2}\frac{7}{2}$ and (e) $\frac{3}{2}2\frac{5}{2}$ half order peaks for CaTiO₃ films grown on LSAT substrates. The doublet features on the peaks suggest that the film has a twinned structure.

symmetry (Pnmb for both CaTiO₃ and NGO), consistent with synchrotron diffraction data. However, the CaTiO₃/ LSAT interface (Figs. 3(a) and 3(b)) clearly exhibits a transition region across which β varies from zero in the cubic LSAT substrate $(Pm\bar{3}m)$ to a bulk CaTiO₃ value (Pnmb)over a distance of \sim 7 atomic rows.

Given that the tilts in these materials are strongly coupled to the suppression of the ferroelectric transition, the differing octahedral tilt structures of CaTiO₃ films grown on LSAT and NGO substrates is expected to lead to disparate

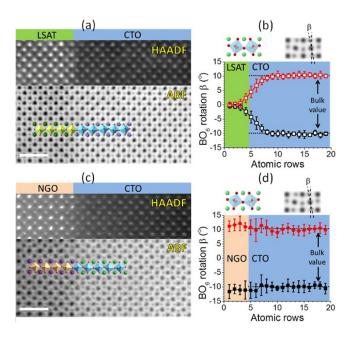


FIG. 3. STEM-HAADF and ABF images of (a) CaTiO₃/LSAT and (c) CaTiO₃/NGO interfaces, viewed from the [010]_{PC} zone axis. Polyhedral models showing the octahedral tilts are overlaid on the images. The size bar is 1 nm. BO₆ rotation angle, β for (b) CaTiO₃/LSAT and (d) CaTiO₃/NGO interfaces. The angles are projected onto the [010]_{PC} direction and displayed as a plane averaged value. Adjacent BO₆ octahedra, which rotate in opposite directions, are denoted separately as red and black symbols. The dashed lines represent the CaTiO3 bulk value along this axis.

ferroelectric properties. To compare the dielectric and ferroelectric properties of these films, they were measured with interdigitated electrodes to obtain the polarization in the plane of the film. 18 Despite theoretical predictions that a strain of at least 1.5% is needed to induce a ferroelectric state, ^{12,13} both of these films with lower strain exhibit ferroelectricity at low temperatures. Figure 4 shows peaks in the dielectric properties for films on both LSAT and NGO substrates; however, the temperatures of the maximum (T_{max}) differ. For films on LSAT substrates, T_{max} is \sim 140 K; for films grown on NGO, T_{max} is $\sim 70 \,\mathrm{K}$. The ferroelectric behavior was also confirmed with polarization-electric field hysteresis loops at 10 K showing remnant polarizations of 5 and $1.5 \,\mu\text{C/cm}^2$ for films on LSAT and NGO substrates, respectively. This difference in ferroelectric transition temperature and polarization is much larger than is expected for a $\sim 0.1\%$ change in strain state. This discrepancy along with the measured differences in symmetry strongly suggest that, unlike what has been reported for theoretical calculations of epitaxial strain effects, 12,13 the suppression/alteration of the octahedral tilts plays a crucial role in inducing ferroelectricity in CaTiO₃. Underestimation of transition temperatures and overestimation of strains required to achieve the ferroelectric state in CaTiO₃ in the current theoretical framework can perhaps be attributed to the assumption that the room temperature paraelectric state is (001)-oriented orthorhombic *Pnmb* structure for the calculations, ¹² whereas this work clearly shows a (101)-orientation for orthorhombic Pnmb CaTiO₃ on NGO substrates and a tetragonal symmetry for CaTiO₃ on LSAT substrates.

To further explore the effects of different symmetries on the predicted ferroelectric transition in CaTiO₃, a thermodynamic analysis was performed to calculate the strain dependent ferroelectric phase diagram with different symmetries in the paraelectric state. The previously determined thermodynamic potential from Gu et al. 13 was utilized to describe the coupling between the antiferrodistortive and ferroelectric phases in CaTiO₃. The phase transitions are described with a

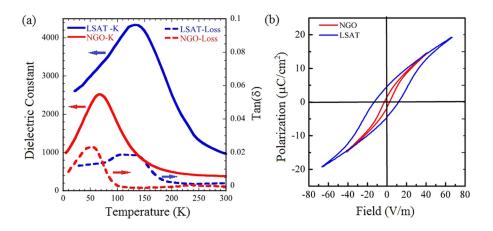


FIG. 4. (a) Dielectric properties vs. temperature for CaTiO₃ films measured at 1 kHz and (b) ferroelectric hysteresis loops at 10 K.

single Landau free energy expansion in terms of ε_i , P_i , θ_i , and φ_i , where ε_i (i=1–6) are the strain components using Voigt's convention; P_i (i=1,2,3) represent the three components of the spontaneous polarization in the Cartesian coordinate system; and θ_i and φ_i (i=1,2,3) represent the in-phase and out-of-phase oxygen octahedral tilt components, respectively. The free energy can be described as follows:

$$\begin{split} F\big(P_{i}, \varphi_{i}, \theta_{i}, \varepsilon_{ij}\big) \\ &= \alpha_{ij} P_{i} P_{j} + \alpha_{ijkl} P_{i} P_{j} P_{k} P_{l} + \alpha_{ijklmn} P_{i} P_{j} P_{k} P_{l} P_{m} P_{n} \\ &- \mu_{ijkl} \varphi_{i} \varphi_{j} \theta_{k} \theta_{l} + \beta_{ij} \varphi_{i} \varphi_{j} + \beta_{ijkl} \varphi_{i} \varphi_{j} \varphi_{k} \varphi_{l} \\ &+ \beta_{ijklmn} \varphi_{i} \varphi_{j} \varphi_{k} \varphi_{l} \varphi_{m} \varphi_{n} - \kappa_{ijkl} P_{i} P_{j} \theta_{k} \theta_{l} \\ &+ \gamma_{ij} \theta_{i} \theta_{j} + \gamma_{ijkl} \theta_{i} \theta_{j} \theta_{k} \theta_{l} + \gamma_{ijklmn} \theta_{i} \theta_{j} \theta_{k} \theta_{l} \theta_{m} \theta_{n} - t_{ijkl} P_{i} P_{j} \varphi_{k} \varphi_{l} \\ &+ \frac{1}{2} c_{ijkl} \varepsilon_{ij} \varepsilon_{kl} - g_{ijkl} \varepsilon_{ij} P_{k} P_{l} - \lambda_{ijkl} \varepsilon_{ij} \varphi_{k} \varphi_{l} - \varsigma_{ijkl} \varepsilon_{ij} \theta_{k} \theta_{l}, \end{split}$$

where α , β , and γ are Landau-Devonshire coefficients (only coefficients of the second order terms are temperature dependent) of polarization, out-of-phase tilt, and in-phase tilt, respectively; t_{ijkl} , κ_{ijkl} , g_{ijkl} , μ_{ijkl} , λ_{ijkl} , and ς_{ijkl} are coupling coefficients; c_{ijkl} is the elastic stiffness tensor. The indices i, j, k, and l run from 1 to 3. All the coefficients are defined as previously reported. Since the films are $(001)_{PC}$ -oriented, the coordinate system was rotated such that the x_l , x_2 , and x_3 directions correspond to the $[\bar{1}10]$, [001], and [110] orthorhombic directions, respectively. By applying the thin film boundary condition, i.e., $\varepsilon'_{11} = \varepsilon'_{22} = \varepsilon_s$, $\varepsilon'_{21} = \varepsilon'_{12} = 0$, $\sigma'_{13} = \sigma'_{23} = \sigma'_{31} = \sigma'_{31} = \sigma'_{32} = 0$, and minimizing the total free

energy with respect to epitaxial strain, ε_s , a temperaturestrain phase diagram can be determined. The prime sign indicates rotated tensors, and σ_{ij} stand for the stress tensor. The calculations were performed as described in Refs. 13 and 19. Phase diagrams for the space groups were calculated as a function of strain for the *Pnmb* space group (i.e., $a^-b^+c^$ tilts) for CaTiO3 films grown on NGO substrates and both the $P4_2/nmc$ $(a^+a^+c^-)$ or I4/mmm $(a^+a^+c^0)$ space groups for CaTiO₃ films on LSAT substrates. The ferroelectric states in these materials could not be determined in a straightforward manner through simply minimizing the free energy due to the fact that several phases with and without polarization were close in energy. To overcome this complication, the most likely phases were determined from the symmetry consideration as shown by Stokes et al.²⁰ This procedure leads to the most likely ferroelectric phase for films on NGO substrates being Pmc2₁. For films on LSAT substrates, the most likely phases are more complex; P42mc and Aba2 being likely for the P4₁/nmc space group and I4 mm, Fmm2, and Cm phases being likely for the I4/mmm space group. These results are summarized in Figure 5 where ferroelectric transitions were determined as a function of strain for the three initial starting space groups. Figures 5(a) and 5(b) show that the ferroelectric transition is better described using the I4/mmm phase because the calculations over-predict the amount of strain required to induce the phase transition for the $P4_2/nmc$ phase. These results are consistent with the experimentally determined transition temperatures, though the results do not match precisely perhaps due to uncertainty in the coefficients used to describe CaTiO3, as seen previously for the

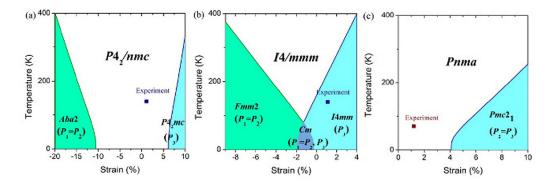


FIG. 5. The calculated temperature-strain phase diagram for CaTiO₃ films grown on (a) and (b) LSAT and (c) NGO substrates. The corresponding measured experimental transition temperatures are indicated on the phase diagrams.

calculation of transition temperatures in strained SrTiO₃. Additional experiments would be required to better refine the values of these coefficients.

In conclusion, epitaxial CaTiO3thin films grown on LSAT and NGO substrates display strong ferroelectric properties in agreement with theoretical predictions. These substrates induce different symmetries and octahedral tilt structures in the films. The tensile strain imposed by both substrates strongly suppresses the octahedral tilts around the c-axis; however, distinct tilt ordering was observed with a twinned $a^+a^+c^0$ tilt structure for films grown on LSAT substrates and the $a^-b^+c^-$ tilt structure on NGO substrates. Ferroelectric and dielectric measurements show that the larger suppression of tilts on LSAT substrates results in a more robust ferroelectric state with remnant polarization \sim 5 μ C/cm² at 10 K and T_c near 140 K. However, unlike previous calculations, the in-plane symmetry of the epitaxial films affected the ferroelectric phase transition. This work demonstrates an innovative mechanism for controlling ferroelectricity by altering octahedral tilts in perovskites, not via direct interfacial coupling as recently investigated by several groups for regions just a few unit cells from the interface, but by macroscopic strain and control of strain symmetry. This tool has the potential to be applied to a wide range of orthorhombic *Pnmb* perovskite materials.

The sample growth and characterization in this research were conducted at the Center for Nanophase Materials Sciences, which is a Department of Energy (DOE) Office of Science User Facility. Use of the Stanford Synchrotron Radiation Lightsource, SLAC National Accelerator Laboratory, was supported by the U.S. DOE, Office of Science, Office of Basic Energy Sciences under Contract No. DE-AC02-76SF00515. Y.T. acknowledges the National Science Foundation (NSF, DMR 0747896); Y.G. and L.Q.C. acknowledge NSF (DMR 1410701, DMR 1420620). Electron microscopy research (Q.H. and A.B.) was supported by the U.S. DOE, Office of Science, Basic Energy Sciences, Materials Sciences and Engineering Division. The

authors L.Q., Y.G., A.M., Q.H., A.B., L.Q.C., and Y.T. submit this paper in honor and in memory of Michael Biegalski who was not only a well-respected research colleague but also a wonderful father and husband, a driven athlete, and a dear friend to all of us.

- ¹B. J. Kennedy, C. J. Howard, and B. C. Chakoumakos, J. Phys.-Condens. Matter 11, 1479 (1999).
- ²M. Yashima and R. Ali, Solid State Ionics **180**, 120–126 (2009).
- ³J. He, A. Borisevich, S. V. Kalinin, S. J. Pennycook, and S. T. Pantelides, Phys. Rev. Lett. **105**(22), 227203 (2010).
- ⁴A. Vailionis, H. Boschker, W. Siemons, E. P. Houwman, D. H. A. Blank, G. Rijnders, and G. Koster, Phys. Rev. B **83**(6), 064101 (2011).
- ⁵A. M. Glazer, Acta Crystallogr., Sect. B **28**, 3384 (1972).
- ⁶V. V. Lemanov, A. V. Sotnikov, E. P. Smirnova, M. Weihnacht, and R. Kunze, Solid State Commun. 110, 611 (1999).
- ⁷T. Nakamura, P. H. Sun, Y. J. Shan, Y. Inaguma, M. Itoh, I. N. Kim, J. H. Sohn, M. Ikeda, T. Kitamura, and H. Konagay, Ferroelectrics **196**, 205 (1997).
- ⁸V. V. Lemanov, A. V. Sotnikov, E. P. Smirnova, and M. Weihnacht, Appl. Phys. Lett. 81, 886 (2002).
- ⁹J. H. Haeni, P. Irvin, W. Chang, R. Uecker, P. Reiche, Y. L. Li, S. Choudhury, W. Tian, M. E. Hawley, B. Craigo, A. K. Taganstev, X. Q. Pan, S. K. Streiffer, L. Q. Chen, S. W. Kirchoefer, J. Levy, and D. G. Schlom, Nature **430**, 758 (2004).
- ¹⁰M. Tyunina, J. Narkilahti, M. Plekh, R. Oja, R. M. Nieminen, A. Dejneka, and V. Trepakov, Phys. Rev. Lett. **104**, 227601 (2010).
- ¹¹S. M. Nakhmanson, K. M. Rabe, and D. Vanderbilt, Phys. Rev. B 73, 060101R (2006).
- ¹²C. J. Eklund, C. J. Fennie, and K. M. Rabe, Phys. Rev. B **79**, 220101 (2009).
- ¹³Y. Gu, K. M. Rabe, E. Bousquet, V. Gopalan, and L. Q. Chen, Phys. Rev. B 85, 064117 (2012).
- ¹⁴S. M. Yang, S. J. Moon, T. H. Kim, and Y. S. Kim, Curr. Appl. Phys. 14, 757–760 (2014).
- ¹⁵A. M. Glazer, Acta Crystallogr., Sect. A **31**, 756–762 (1975).
- ¹⁶S. J. May, J.-W. Kim, J. M. Rondinelli, E. Karapetrova, N. A. Spaldin, A. Bhattacharya, and P. J. Ryan, Phys. Rev. B 82, 014110 (2010).
- ¹⁷H. F. Kay and P. C. Bailey, Acta Crystallogr. **10**, 219 (1957).
- ¹⁸M. D. Biegalski, Y. Jia, D. G. Schlom, S. Trolier-McKinstry, S. K. Streiffer, V. Sherman, R. Uecker, and P. Reiche, Appl. Phys. Lett. 88, 192907 (2006).
- ¹⁹Y. L. Li, S. Choudhury, J. H. Haeni, M. D. Biegalski, A. Vasudeyarao, A. Sharan, H. Z. Ma, J. Levy, V. Gopalan, S. Trolier-McKinstry, D. G. Schlom, Q. X. Jia, and L. Q. Chen, Phys. Rev. B 73, 184112 (2006).
- ²⁰H. T. Stokes, E. H. Kisi, D. M. Hatch, and C. J. Howard, Acta Crystallogr., Sect. B 58, 934 (2002).