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A STUDY OF POROUS MEMBRANE EVAPORATION

FOR

DESALINATION IN A FLOW SYSTEM

ΒY

JING MING LEE, 1934

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THESIS

submitted to the faculty of the

UNIVERSITY OF MISSOURI - ROLLA

in partial fulfillment of the requirements for the

Degree of

MASTER OF SCIENCE IN CHEMICAL ENGINEERING

Rolla, Missouri

1968

Approved by

M.E. Findley (advisor) Frank 1. Conrod Spagan

ABSTRACT

The purpose of this investigation was to study the simultaneous mass and heat transfer mechanism in evaporation through a porous membrane with a non-wettable surface. Such water repellent membranes permit the passage of water vapor, but not liquid water. The investigation concerned the mass transfer rate through the membrane pores with flow on one or both sides of the membrane.

The water-repellent membrane separated a hot salt solution from the fresh water, and a copper sheet separated the fresh water from a cold salt solution. A three-channel evaporator-condenser was used, and the membrane consisted of glass fiber paper treated with a teflon dispersion. The temperature range studied was from 93 to 190°F.

A temperature difference and a corresponding vapor pressure difference maintained across the membrane provided the driving force both for mass and heat transfer through the membrane and heat recovered through the copper sheet to cold salt solution. Theoretical and empirical correlations were employed to fit the experimental data. It was observed that heat transfer resistance and diffusion in the membrane pores were the major resistances to total mass transfer. The correlation predicted rates of mass transfer resistance close to the experimental values. The heat transfer coefficient was affected by the mass diffusion. The ratio of heat transfer coefficient with diffusion to that without diffusion was 1.5, and was slightly dependent on flow.

The mass transfer coefficient varied from 0.22 to 0.516 $lb/(hr)(ft^2)(in-Hg)$. The overall heat transfer coefficient for

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the membrane varied from 48 to 104 BTU/($hr(ft^2)(\circ F)$, and the overall heat transfer coefficient for the copper sheet varied from 54 to 84 BTU/(hr)(ft^2)($\circ F$).

ACKNOWLEDGEMENT

The author wishes to express his sincere thanks to Dr. Marshall E. Findley for suggesting this investigation and for his encouragement, guidance and help through the course of this study. The author also would like to express his appreciation to the University of Missouri - Rolla for the financial aid which enabled him to continue his work without interruption.

The author also is deeply indebted to his family and friends for their help and encouragement.

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NOMENCLATURE

Ar	= Transfer area of the membrane and copper sheet, ft ² .
Ъ	= Thickness of the membrane, ft.
С	= Total molar density, lb-mole/ft ³ .
C _{AF}	= Pure water concentration, (lb-mole)/ft ³ .
C _{AS}	= Molar concentration of water in the hot salt solution, lb-mole/ft ³ .
D _{AB}	= Binary gas diffusion constant, diffusivity, ft ² /hr.
DW	= Amount of water transferred to the fresh water side from
	the hot salt solution, lb/hr.
DPM	= Log mean vapor pressure, in-Hg.
E	= Boiling point elevation of the salt solution, °F.
F ₁ , F ₂ F ₃ , F ₄	= Correlation equation functions.
G	= Mass flow rate, lb/(hr)(ft ²).
h	= Film heat transfer coefficient, BTU/(hr)(ft ²)(°F).
h _F	= Film heat transfer coefficient of fresh water, $BTU/(hr)(ft^2)(\circ F)$.
h _R	= Film heat transfer coefficient of cold salt solution
	(recovery side), BTU/(hr)(ft ²)(°F).
h _S	= Film heat transfer coefficient of hot salt solution,
	$BTU/(hr)(ft^2)(°F).$
k	= Thermal conductivity, BTU/(hr)(ft)(°F).
kc	= Thermal conductivity of copper sheet, BTU/(hr)(ft)(°F).
k e	= Thermal conductivity of the membrane, BTU/(hr)(ft)(°F).
к _м	= Overall mass transfer coefficient, lb/(hr)(ft ²)(in-Hg).

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- M_A = Molecular weight of component A.
- M_B = Molecular weight of component B.
- N_A = Mass flux of component A, (lb-mole)/(hr)(ft²) or lb/(hr)(ft²).
- p = Total pressure of the system, (in-Hg).
- P_{A1} = Vapor pressure at the membrane surface on hot salt solution side, (in-Hg).

- = Vapor pressure of the hot salt solution at the bulk temperature (in-Hg).
- Δp_m = Log mean partial pressure difference of the water vapor, (in-Hg).
 Q = Heat flux.
- q = Conduction heat through the membrane.
- q = Total heat transferred from the hot salt solution to
 fresh water.
- q = Total heat transferred from fresh water to cold salt solution as recovery heat.
- q_{1.1} = Heat loss from the hot salt solution channel.
- $q_{1,2}$ = Heat loss from the fresh water channel.
- q_{1.3} = Heat loss from the cold salt solution channel.
- q_R = Total heat recovered by cold salt solution.
- q₊ = Total heat input to the system.
- q_v = Heat content in the vapor.
- R = Gas constant
- T = Absolute temperature, °R.

- \overline{T} = Average absolute temperature in the membrane, R.
- ΔT_m = Log mean temperature difference.
- TFM = Average temperature of fresh water channel, °F.
- TMl = Average temperature at the membrane surface on hot salt solution side, °F.
- TM2 = Average temperature at the membrane surface on fresh water side, °F.
- TRI = Cold salt solution inlet temperature in the tube, °F.
- TRM = Average temperature of the cold salt solution channel, °F.
- TRO = Cold salt solution channel outlet temperature in the tube, °F.
- TSI = Hot salt solution inlet temperature in the tube, °F.
- TSM = Average temperature of the hot salt solution channel, °F.
- TSO = Hot salt solution channel outlet temperature in the tube, °F.

 U_m = Overall heat transfer coefficient for the membrane.

^U3

= Heat transfer coefficient by conduction through the membrane, BTU/(hr)(ft²)(°F)

- W_c = Circulation flow rate, lb/hr.
- X = Mole fraction.
- Z = Diffusion direction and path, ft.

Subscripts:

- A, B, referring to water vapor and air, respectively.
- 1, 2, referring to positions of salt water and fresh water membrane surface, respectively.

S, F, referring to salt water and fresh water, respectively.

I. INTRODUCTION

Membrane transfer phenomena and theory has been studied for over a hundred years in biological systems. However, the potential of the application of membranes in engineering has been developed only in recent years (10). One recent method of evaporation through porous membranes has been proposed by Findley (4). This method, applied to the conversion of sea water to fresh water, is the subject of this investigation.

Membrane desalination processes have the potential advantages of economical operating costs, and simple equipment. The method studied in this investigation requires only a suitable porous membrane to separate the hot salt solution and the coolant fresh water and heat exchange equipment to supply and recover heat.

The salt solution at a higher temperature has a higher water vapor pressure than the vapor pressure of fresh water at a little lower temperature. The higher vapor pressure will produce diffusion through the membrane pores to the coolant fresh water which serves as a vapor condenser. The surface tension (5) keeps the fresh water from entering the pores. With counter-current flow, each pore is at a different temperature and functions as a single stage of flash evaporation. Thus, a porous membrane can provide an infinite number of stages of flash evaporation. Flash evaporation refers to evaporation from sensible heat in this paper.

Before this process can be applied to industrial production, it is necessary to thoroughly understand the relationship of the heat and mass transfer. Previous studies (7, 13, 14, 19) provided

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some useful information on this method, but further study of this process is still required, primarily to establish the effects of flow on the transfer relationship.

The purpose of this work was to study evaporation through the porous membranes in a flow system in order to determine the relationships of flow to heat and mass transfer.

II. THEORY AND LITERATURE REVIEW

This chapter is to discuss the theory and mechanism associated with the simultaneous heat and mass transfer through a water repellent porous membrane with a temperature gradient to provide the driving force for both heat and mass transfer.

A. Theory of Heat Transfer

1. Conduction

Fourier's law applied to one-dimensional heat transfer by conduction is,

$$Q = -k \frac{dT}{dZ}$$
(2-1)

Q = heat flux, BTU/(hr)(ft²)(°F)

k = conductivity, BTU/(hr)(ft)(°F)

 $T = temperature, \circ F$

Z = thickness of the transfer path and direction, ft.

2. Convection

There are two types of convection, one is natural (or free) convection due to density differences without external force, another is forced convection due to a pressure drop which causes turbulence in the fluid flowing. Convection is accompanied by a transfer of heat by conduction. Most liquid, and nearly all gases conduct heat so poorly that the heat transferred by conduction is, in general, negligible compared to that heat transferred by. convection (15). In this study with laminar flow, both natural and forced convection and possibly liquid conduction could be important. The conventional concept of heat transfer film coefficient is given by the equation,

$$Q = h(T_w - T_b) = h\Delta T$$
 (2-2)

h = film heat transfer coefficient, BTU/(hr)(ft²)(°F)
ΔT = the temperature difference between the surface of the
wall and the bulk temperature of the fluid, °F.

3. Radiation

Since temperature differences are small, radiation was assumed negligible in this study.

B. Mass Transfer

1. Diffusion in Binary System

Fick's law of diffusion with concentration gradients is similar to Fourier's law of heat conduction applied to diffusion in a binary system in one-dimensional diffusion. Fick's law is (1),

$$N_{A} = -CD_{AB} \frac{\partial XA}{\partial Z} + X_{A} (N_{A} + N_{B})$$
 (2-3)

where

 N_A , N_B = molar flux of A and B respectively X_A = mole fraction of A C = total molar density D_{AB} = diffusivity, ft²/hr Z = diffusion path and direction, ft. This equation shows that molar flux N_A is the result of two vector quantities, one of which is the bulk flow term $X_A(N_A + N_B)$, and the other is the diffusion term $CD_{AB} \frac{\partial XA}{\partial Z}$.

If a system is at steady state with constant molar flux equation 2-3 can be applied directly, but we must know the relation of N_A and N_B , C and D_{AB} in order to solve the equation. Fuller, Schettler and Giddings (6), gave an equation for a diffusion coefficient in gases as follows,

$$D_{AB} = K \frac{\overline{T}^{1.75}}{p(V_A^{1/3} - V_B^{1/3})} \left(\frac{1}{M_A} + \frac{1}{M_B}\right)^{1/2}$$
(2-4)

 V_A , V_B = molecular volume of A and B respectively. M_A , M_B = molecular weight of A and B respectively K = a constant for a pair of gases \overline{T} = average absolute temperature of the system, °R p = total pressure of the system, in-Hg

The equation shows that D_{AB} is independent of concentration and dependent on total pressure and temperature. It is reasonable to assume that D_{AB} is a constant in a system with small changes of temperature and constant pressure.

C. Simultaneous Heat and Mass Transfer in Porous Membrane Desalination

1. Mass Transfer Through the Membrane

In this desalination process the temperature gradient produce partial pressure gradients across the pores of the membrane (see Fig. 2-1), and at any given temperature there is a corresponding partial pressure of the water vapor whether vapor is present or not.



Figure 2-1.

Temperature, partial pressure and concentration profile in the membrane condenser system.

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TSM, TFM, TRM = bulk mean temperatures, refer to hot salt,

fresh and cold salt water respectively.

- TC1, TC2 = surface temperatures of copper sheet, refer to fresh and cold salt water.
- C_{AS}, C_{AF} = concentration of water in the hot salt water and pure water.

$$P_{AS}$$
, P_{AF} = equilibrium vapor pressures at TSM and TFM
 P_{A1} , P_{A2} = equilibrium vapor pressure of water at the interface
of the membrane at TMl and TM2.

The higher partial pressure of water vapor on the hot side provides a driving force for diffusion through a stagnant air film in the membrane pores with condensation occurring at the other cooler side of the membrane. Therefore, for the flux of air, $N_B^{=0}$, and by applying equation 2-3 (2),

$$N_{A} = -CD_{AB} \left(\frac{1}{1-X_{A}}\right) \frac{\partial XA}{\partial Z}$$
 (2-5)

For a steady state using equation 2-4 and combining constants,

$$D_{AB} = k' \overline{T}^{1.75}$$

For moderate temperature and atmospheric pressure, the ideal gas law can be applied.

$$C = p/RT$$

so,

$$CD_{AB} = \frac{P}{R\overline{T}} k' \overline{T}^{1.75} = k_D \overline{T}^{0.75}$$
(2-6)

For moderate temperature change we can assume CD_{AB} is a constant, and since the flux is constant through its transfer path, N_A is a constant and we get from equation 2-5,

$$\frac{\mathrm{d}}{\mathrm{d}Z} \left[\left(\frac{1}{1 - \mathrm{XA}} \right) \left(\frac{\partial \mathrm{XA}}{\partial Z} \right) \right] = 0$$

Integration with boundary conditions as follows,

 $Z = ZI = 0 \qquad X = X_{AI}$ $Z = Z2 = b \qquad X = X_{A2}$

We get (2),

$$\left(\frac{1-X_{A}}{1-X_{A\perp}}\right) = \left(\frac{1-X_{A2}}{1-X_{A\perp}}\right)^{Z/b}$$
 (2-7)

 X_A , X_B = mole fraction of A and B respectively. X_{A1} , X_{A2} = mole fraction of A at each interface of the membrane respectively.

b = thickness of the membrane, ft.

Equation 2-7 gives the concentration profile of the diffusion path in the membrane pores. A combination of equation 2-5 and 2-7 gives,

$$N_{A} = \frac{CD_{AB}}{D} \ln \frac{X_{B2}}{X_{B1}} = \frac{CD_{AB}}{D} \frac{X_{B2} - X_{B1}}{(X_{B})_{Lm}}$$
(2-8)

where

$$(X_{B})_{Lm} = (X_{B2} - X_{B1})/(\ln \frac{X_{B2}}{X_{B1}})$$

 $X_{B2} = 1 - X_{A2}$
 $X_{B1} = 1 - X_{A1}$

If the concentration is represented by partial pressure of the water vapor, equation 2-8 becomes,

$$N_{A} = \frac{P}{RT} \cdot \frac{D_{AB}}{D} \ln \frac{P_{B2}}{P_{B1}} = \frac{PD_{AB}}{DRT} \frac{P_{A1} - P_{A2}}{(p_{B})_{Lm}}$$
(2-9)

since $C = p/R\overline{T}$

where

$$(p_B)_{Lm} = \frac{p_{B2} - p_{B1}}{\ln(p_{B2}/p_{B1})}$$

Equation 2-9 shows that the rate of diffusion of gas A is directly proportional to the pressure difference between the interfaces, and inversely proportional to the length of the diffusion path and to the logarithmic mean partial pressure of the stagnant gas B in the path.

Since we do not know the temperature at each interface of the membrane, we do not know the corresponding vapor pressure p_{A1} and p_{A2} . For application of the equation, a mass transfer coefficient was defined for convenience (16).

$$N_{A} = K_{M} (p_{AS} - p_{AF})$$
 (2-10)

 K_{M} = overall mass transfer coefficient, lb/(hr)(ft²)(in-Hg). P_{AS} , P_{AF} = vapor pressure of the hot salt solution and fresh

water at bulk temperatures respectively.

The two films concept was first suggested by W. G. Whiteman in 1923 (17), and has proved to be a great aid in understanding the process of diffusion between two fluids. We assume the interface of two phases on the surface of contact are in equilibrium and there is no appreciable diffusion resistance at the actual interface (18). Then in this case all the resistance to the mass diffusion is present in the membrane pores and in the salt water film. For the rates encountered in this investigation, it was also assumed that the salt water film resistance was negligible compared to the resistance of the vapor phase in the pores.

2. Heat Transfer Coefficient

The main heat transfer resistance for a fluid, cooling or heating, depends on the fluid layer in contact with the heat transfer surface. The thickness of this boundary layer depends on internal motion of the fluid. A number of heat transfer coefficient correlations appear in the literature, Sieder and Tate derived the equation for laminar stream flow as follows (9),

$$\frac{hD}{k} \left(\frac{\mu_{s}}{\mu_{b}}\right)^{0.14} = 1.86 \left[\left(\frac{DG}{\mu_{b}}\right)\left(\frac{Cp\mu}{k}\right)_{b} (D/L)\right]^{1/3}$$

For small temperature ranges $\mu / \mu = 1$

Then,

h =
$$a \frac{k^{0.67}}{D^{0.33}} \left(\frac{CpG}{L}\right)^{0.33}$$
 = af(G.T) (2-11)

where

a = constant. a = 1.86 for cylindrical pipes L = length of heat transfer area, ft. K = thermal conductivity of the fluid. C_p = heat capacity, BTU/(lb)(°F) G = mass flow, lb/(hr)(ft²) D = diameter, ft.

$$f(G.T) = (k^{0.67}/D^{0.33})(C_{PG}/L)^{0.33}$$

h = heat transfer coefficient, BTU/(hr)(ft²)(°F)

Perry's handbook gives the following equation for natural convection (12),

h = 43
$$\left(\frac{\Delta T}{D}\right)^{0.25}$$
 = df($\Delta T/D$) (2-12)

where

43 = constant for horizontal plate

d = constant

 ΔT = temperature drop between the wall and the bulk temperature. f($\Delta T/D$) = ($\Delta T/D$)^{0.25}

3. Theoretical Correlation of Heat and Mass Transfer

First, heat must be transferred from the bulk of the stream of hot salt solution to the membrane interface, then the same amount of heat should also be transferred through the membrane by evaporation and conduction and again through the fresh water film to the bulk of the fresh water. The same mechanisms transferred heat to the recovery cold salt solution through two films and through a copper sheet by conduction (see Fig. 2-2).

Heat transfer through the copper sheet may be expressed as,

$$q_{O2} = U_{O}(TFM-TRM)$$
(2-13)

The resistances to heat transfer are:

$$1/U_c = 1/h_F + 1/h_R + x/k_c$$

U_c = overall heat transfer coefficient for the copper sheet, BTU/(hr)(ft²)(°F)



Figure 2-2. Heat transfer in the process.

q_v = heat carried by vapor
q_c = heat transferred through the membrane by conduction
q_{cl} = q_v+q_c = total heat transferred from hot salt solution
 to the fresh water.

q_c2 = total heat transferred from fresh water to cold salt
 solution (recovery heat)

TFM, TRM = bulk mean temperature of the fresh water and cold salt solution respectively

X = thickness of the copper sheet, ft.

Since X/k_c is very small and can be neglected, then

$$1/U_{c} = 1/h_{F} + 1/h_{R}$$
 (2-14)

Heat balances over the membrane are as follows:

$$q_c = U_3(TSM-TFM) = \frac{k_e}{b}(TM1-TM2)$$

$$q_{cl} = h_{S}(TSM-TML) = h_{F}(TM2-TFM)$$

Then

$$(TSM-TFM) - (TM1-TM2) = q_{c1}(\frac{1}{h_S} + \frac{1}{h_F})$$
 (2-15)

and

$$q_{cl} = q_{v} + q_{c} = N_{A} H + \frac{k_{e}}{b}$$
 (TM1-TM2) (2-16)

Combining equations 2-15 and 2-16,

$$(TSM-TFM) = [N_A^H + \frac{k_e}{b} (TM1-TM2)](\frac{1}{h_S} + \frac{1}{h_F}) + (TM1-TM2)$$
 (2-17)

where

TMl, TM2 = temperatures on the membrane surfaces of hot salt and fresh water side respectively h_S = heat transfer coefficient of hot salt solution k_p = effective conductivity of the membrane U₃ = heat transfer coefficient across the membrane by conduction

TSM = bulk mean temperature of the hot salt solution, °F For moderate ΔT across the membrane, we assume the vapor pressure is linear relation with temperature at atmospheric pressure (see Fig. 2-3), and assumed $P_{AS}^{+P}_{AF} \stackrel{=}{=} P_{A1}^{+P}_{A2}$.

Therefore,

$$TM1-TM2 = \frac{TSM-TFM}{P_{AS}-P_{AF}} (P_{A1}-P_{A2})$$
(2-18)

By expansion of equation 2-8 (21),

$$N_{\mathbf{A}} = \frac{CD_{AB}}{b} (\ln X_{B2} - \ln X_{B1})$$

$$X_{B2} = 1 - X_{A2} = 1 - (p_{A2}/p)$$

$$X_{B1} = 1 - X_{A1} = 1 - (p_{A1}/p)$$

$$N_{\mathbf{A}} = \frac{CD_{AB}}{b} [\frac{1}{p} (p_{A1} - p_{A2}) + \frac{1}{2p^{2}} (p_{A1}^{2} - p_{A2}^{2}) + \frac{1}{3p^{3}} (p_{A1}^{3} - p_{A2}^{3}) + \dots]$$

Taking the first two terms and combining with equation 2-6, we get

$$\dot{p}_{A1} - p_{A2} = \frac{b}{k_{D}} \frac{N_{A}}{\overline{T}^{0.75}(1 + \frac{p_{AS} + p_{AF}}{2p})}$$
 (2-19)

A combination of equations 2-18 and 2-19 gives,

$$(TM1-TM2) = \frac{TSM-TFM}{P_{AS}-P_{AF}} \frac{b}{k_{D}} \frac{2p^{2}}{\overline{T}^{0.75}(2p+P_{AS}+P_{AF})}$$

We should consider the tortuous diffusion path (3) and the effective diffusion area of the membrane. Therefore, the thickness b should





be multiplied by 1.414 to correct for tortuousity (3). The diffusion constant k_D , should be multiplied by void fraction of the membrane to correct for the effective area of diffusion. Then,

$$(TM1-TM2) = \frac{(TSM-TFM)}{(P_{AS}-P_{AF})} \frac{b \cdot 1.414}{k_{D} po} \frac{2p^2 N_A}{\overline{T}^{0.75}(2p+P_{AS}+P_{AF})}$$
 (2-20)

where

po = porosity of the membrane, or void fraction Substitute equation 2-20 into equation 2-17 and combine with equation 2-10, then

$$\frac{1}{K_{M}} = \frac{(P_{AS} - P_{AF})H}{(TSM - TFM)} \left(\frac{1}{h_{S}} + \frac{1}{h_{F}}\right) + \frac{k_{e}}{k_{D}} \frac{2p^{2} l \cdot 4l4}{p_{o} \overline{T}^{0} \cdot 75} \frac{2p^{2} l \cdot 4l4}{(2p + P_{AS} + P_{AF})}$$

$$\left(\frac{1}{h_{S}} + \frac{1}{h_{F}}\right) + \frac{b}{k_{D}} \frac{2p^{2} l \cdot 4l4}{p_{o} \overline{T}^{0} \cdot 75} \frac{2p^{2} l \cdot 4l4}{(2p + P_{AS} + P_{AF})}$$
(2-21)

The first term is equivalent to the film resistance associated with the heat of vaporation and condensation. The third term is the resistance to diffusion in the membrane and the second term is an effect of membrane heat conduction parallel to mass transfer of the liquid film. For flow systems mean ΔP and ΔT are used.

Let
$$(p_{AS} - p_{AF}) = (\Delta p)_{Lm}$$
 (2-22)

$$(TSM - TFM) = (\Delta T)_{Lm}$$
(2-23)

$$\frac{2p^2 \cdot 1.414}{p_0 \overline{T}^{0.75}(2p+p_{AS}+p_{AF})} = p_p \qquad (2-24)$$

Then equation 2-21 becomes

$$\frac{1}{K_{M}} = \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} \left(\frac{1}{h_{S}} + \frac{1}{h_{F}}\right) + \frac{k_{e}}{k_{D}} p_{p} \left(\frac{1}{h_{S}} + \frac{1}{h_{F}}\right) + \frac{b}{k_{D}} p_{p}$$
(2-25)

Where

$$(\Delta p)_{Lm} = \frac{(P_{SI} - P_{F0}) - (P_{S0} - P_{FI})}{\ln (\frac{P_{SI} - P_{F0}}{P_{S0} - P_{FI}})}$$
$$(\Delta T)_{Lm} = \frac{(T_{SI} - T_{F0}) - (T_{S0} - T_{FI})}{\ln (\frac{T_{SI} - T_{F0}}{T_{S0} - T_{FI}})}$$

 T_{FI} , T_{F0} = the fresh water temperature at the inlet and outlet ends of the channel respectively.

A. Apparatus

A rectangular membrane Evaporator-Condenser was used in this investigation. The apparatus was made of acrylic plastic plate and consisted of three channels. The channel dimensions were $16.25" \times 1.5" \times 0.5"$ at both sides and $16.25" \times 1.5" \times 0.6"$ at the middle (see Fig. 3-1). Two zenith metering pumps running at the same speed maintained the circulation. The mass transfer area was 0.1695 ft^2 . Ten copper-constantan thermocouples were used to measure the desired temperature (see Appendix for details).



Figure 3-1. The Evaporator-Condenser.

Where

- HS = hot salt solution channel
- FW = fresh water channel

RS = heat recovery section, cold salt solution channel

1, 2, 3, 4, 5, 6 = thermocouples measuring the channel temperatures

7, 8, 9, 10 = thermocouples measuring the solution inlet and outlet temperatures (EMF) in the inlet and outlet tubes

B. Experimental

The procedure in brief was to assemble and connect the equipment of the system, to switch on motor-pumps, and to heat up the salt solution with a heated copper coil connecting the cold salt solution outlet to the hot salt solution inlet. After temperatures and level were steady, data were taken of time, temperature, amount of water transferred, flow rate and room temperature every 10 to 30 minutes (for details see Appendix, page 40).

C. Method of Calculation

Experimental data were used to calculate the following: 1) The overall mass transfer coefficient, K_M ; and 2) The overall heat transfer coefficients, U_M , U_C and U_3 .

 U_{M} is the overall heat transfer coefficient for the membrane. U_{c} is the overall heat transfer coefficient for the copper sheet.

 U_3 is the heat transfer coefficient for the membrane by conduction.

Figure 3-2 shows the heat and mass balance in the evaporatorcondenser.



Figure 3-2. Heat and mass flow in the evaporator-condenser.

Where

- ${\bf q}_{\rm L1}, \, {\bf q}_{\rm L2},$ and ${\bf q}_{\rm L3}$ are the heat losses in each channel based on the area exposed to the air, and temperature differences with the outside air.
- T_1, T_2 = hot salt solution temperatures at inlet and outlet tubes respectively, °F.
- T₃, T₄ = cold salt solution temperatures at inlet and outlet tubes respectively, °F.
- W_S , W_F = the mass flow rate in lb/hr of salt solution and fresh water respectively.

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TSM, TFM, TRM = the mean bulk temperatures of the fluids and refer to hot salt solution, fresh water and cold salt solution respectively.

The calculation of the mean bulk temperature of the fluids was as follows:

$$TSM = [T_{SI}(N) + T_{SI}(N+1) + T_{SO}(N) + T_{SO}(N+1)]/4$$

$$TFM = [T_{FI}(N) + T_{FI}(N+1) + T_{FO}(N) + T_{FO}(N+1)]/4$$

$$TRM = [T_{RI}(N) + T_{RI}(N+1) + T_{RO}(N) + T_{RO}(N+1)]/4$$

Where

N = the subscript referring to the measuring order according to time.

 T_{SI} , T_{SO} = the hot salt solution temperature at inlet and outlet ends of the channel.

T_{FI}, T_{FO} = the fresh water temperature at the inlet and outlet ends of the channel.

 T_{RI} , T_{RO} = the cold salt water temperature at the inlet and outlet ends of the channel.

For enthalpy loss on the hot salt solution side, \boldsymbol{q}_+

$$q_t = W_S(T_1 - 32) - (W_S - DW)(T_2 - 32)$$

Where 32°F is the enthalpy reference temperature and specific heat = 1.0.

$$q_{cl} = q_t - q_{Ll} = q_v + q_c = N_A H + \frac{k}{b}$$
 (TM1-TM2)

$$U_{\rm m} = \frac{q_{\rm cl}}{Ar(TSM-TFM)} = \frac{q_{\rm cl}}{Ar(\Delta T)_{\rm LM}}$$

where

Ar = the heat and mass transfer area, ft^2

H = the vapor enthalpy, BTU/lb

 $U_{\rm m}$ = overall heat transfer coefficient for the membrane The enthalpy increase on the recovery side, $q_{\rm R}$, may be calculated as

$$q_R = W_S(T_4 - T_3)$$

The total heat transferred through the copper sheet should include the heat loss, $q_{1,3}$, therefore

$$q_{c2} = q_R + q_{L3}$$

This was the amount of heat transferred from the fresh water to the cold salt solution. Then

$$U_{c} = \frac{q_{c2}}{Ar(TFM-TRM)}$$
(3-2)

where

 U_c = overall heat transfer coefficient for the copper sheet. For heat conduction through the membrane,

$$U_{3} = \frac{q_{c}}{Ar(TSM-TFM)} = \frac{q_{c}}{Ar(\Delta T)_{LM}}$$
(3-3)

where

$$(\Delta T)_{LM} = \frac{(T_{SI} - T_{FO}) - (T_{SO} - T_{FI})}{\frac{T_{SI} - T_{FO}}{T_{SO} - T_{FI}}}$$

(3-1)

 T_{SI} , T_{SO} = hot salt solution bulk temperatures at the inlet

and outlet channel ends.

For mass transfer as defined in equation 2-10, it was assumed that

$$\kappa_{\rm M} = \frac{N_{\rm A}}{(P_{\rm AS} - P_{\rm AF})} = \frac{DW}{Ar(\Delta p)_{\rm LM}}$$
(3-4)

where

$$(\Delta p)_{LM} = \frac{(P_{SI} - P_{FO}) - (P_{SO} - P_{FI})}{\ln \frac{(P_{SI} - P_{FO})}{(P_{SO} - P_{FI})}}$$

 P_{SI} , P_{SO} = the vapor pressure of the hot solution at the inlet and outlet channel ends.

p_{FI}, p_{F0} = the vapor pressure of the fresh water at the inlet and outlet channel ends.

Calculations of the salt solution vapor pressure and pure water vapor pressure are made as follows.

$$a = (f_s/f_f)_T = (p_s/p_F)_T$$

a. = activity of water in solution

f = fugacity of water in salt solution

f_f = fugacity of pure water

For small temperature changes and constant concentration in the system, the activity a. is approximately constant and equal to 0.96 (21).

Therefore,

$$P_S = 0.96 \times P_F$$

where,

 p_{S} = vapor pressure of salt solution

 $p_{_{\rm F}}$ = vapor pressure of pure water

The vapor pressure of pure water is obtained from the following equation (8):

$$\log_{10} \frac{P_{c}}{p} = \frac{x}{T} \left(\frac{a' + b'x + c'x^{3}}{1 + d'x} \right)$$

where

p = vapor pressure in atm $p_c = 218.167$ atm $T = t^{\circ}C + 273.16$, absolute temperature $x = T_c^{-T}$ $T_c = 647.27$ a' = 3.2437814 $b' = 5.86826 \times 10^{-3}$ $c' = 1.1702379 \times 10^{-8}$ $d' = 2.1878462 \times 10^{-3}$

D. Data and Results

All the data and results are tabulated in the Appendix from page 47 to 64.
IV. DISCUSSION

A. System with Both Salt and Fresh Water Flowing

The hot salt water, fresh water and cold salt water are in countercurrent flow with adjacent stream in the system.

1. Correlation

From equation 2-25 and 2-11,

$$\frac{1}{K_{M}} = \frac{1}{a} \frac{(\Delta p)_{LM}}{(\Delta T)_{LM}} \left(\frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}\right) + \frac{k_{e}}{k_{D}a}$$

$$p_{p} \left(\frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}\right) + \frac{b}{k_{D}} \cdot p_{p} \qquad (4-1)$$

a, k_e , k_D are assumed approximately constants. Yeh (22) found molecular diffusion to be the rate controlling factor for mass transfer in the membrane. The same molecular diffusion was assumed in this study and the other constants were evaluated by the least squares method. The least squares equation for the high flow rate (96 cc/min to 210 cc/min) data is as follows,

$$Fl = \frac{1}{K_{M}} = 0.0683 \frac{(\Delta p)_{LM}^{H}}{(\Delta T)_{LM}} (\frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}) + 13.4 p_{p}$$

$$(\frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}) + 2540 b p_{p} \qquad (4-2)$$

A plot was made of $1/K_{M}$ versus function Fl (see Fig. 4-1). In evaluating the constants, unreasonable results were obtained when using low flow rate (42 cc/min to 96 cc/min) data. For example, the mean of temperatures measured in the hot salt water channel were almost equal or slightly below measured outlet temperatures in the tube, which is impossible if the temperatures represent the





$$Fl = \frac{1}{K_{M}} = 0.0683 \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} (\frac{1}{f(G.T.)_{S}} + \frac{1}{f(G.T)_{F}})$$

+ 13.4 $p_{p} (\frac{1}{f(G.T.)_{S}} + \frac{1}{f(G.T)_{F}}) + 2540 b p_{p}$

true conditions. This indicates the flows are non-uniform, probably with higher temperature liquids channeling in the upper part of the chambers. For high flow rate data the temperatures were more reasonable and predicted relationships were consistent with the experimental results. For the above reasons the low flow rate data were not used in the correlations.

Comparing equation 4-1 with equation 4-2, where the constant $1/k_{\rm D}$ = 2540 is a theoretical value at 32°F and atmospheric pressure, a and $k_{\rm e}$ can be obtained from the least squares coefficients in equation 4-2. The constants determined in this way are a=14.65, $k_{\rm e}$ =0.0775.

The heat transfer coefficient is around 250 BTU/(hr)(ft²)(°F), which is reasonable in the heat transfer range for a laminar flow system. The effective k_e , is the conductivity of the solid phase and vapor phase of the membrane. Actually k_e is a variable with solid phase fraction and vapor phase composition, however, the changes are small. The calculated k_e , is 0.057 to 0.058, but this calculation did not consider the convection heat transfer of the vapor. However this convection in the membrane is very small. Other factors are thickness of the membrane which is not uniform, and any effects of moisture in the membrane. Therefore, it is reasonable for calculated k_e , 0.0775, from least squares coefficient to be larger than the theoretical value of k_e , 0.057 to 0.058.

The constant a=14.65, is considerably larger than the value found in the literature, 1.86. However, this constant depends on internal flow, and equipment geometry, and the value in the

literature is for round tubes whereas this study was made using rectangular channels.

Rao (13) pointed out that boiling point elevation has a strong effect on overall mass transfer resistance. A parameter was added to equation 4-1, to determine whether such a term would improve the correlation. The equation became,

$$F2 = \frac{1}{K_{M}} = 0.065 \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} (\frac{1}{f(G.T.)_{S}} + \frac{1}{f(G.T.)_{F}}) + 1.9 \text{ pp}$$

$$\cdot (\frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}) + 2540 \text{ x b } p_{p}$$

$$+ 7761 \frac{[(\Delta T)_{Lm}^{-E})b]}{\overline{T}} \qquad (4-3)$$

Where E is boiling point elevation for 7% salt water, 1.4°F. A plot was made of $1/K_{\rm M}$ versus function F2 given in equation 4-3. The predicted values of equation 4-3 were closer to the experimental values (see Fig. 4-2), but this did not appear to justify the strong effect on the second term of the equation 4-3 and 4-2. Further studies of the effect of ΔT seem to be desirable.

2. Heat Transfer

For heat conduction through the membrane, the equation for heat transfer is as follows.

$$q_{c} = U_{3}(TSM-TFM) = \frac{k_{e}}{b} (TM1-TM2)$$
$$k_{e} = \frac{U_{3}b (TSM-TFM)}{(TM1-TM2)} = \frac{U_{3}b (\Delta T)_{Lm}}{(TM1-TM2)}$$



Figure 4-2. Experimental $1/K_{M}$ versus F2.

$$F_{M}^{P} = \frac{1}{K_{M}} = 0.065 \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} (\frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}) + 1.9 p_{p}$$

$$\cdot (\frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}) + 2540 b p_{p} + 7761 \frac{[(\Delta T)_{Lm}^{-E})b]}{\overline{T}} .$$

Combining with equations 2-20 and 2-10,

$$k_{e} = \frac{U_{3}}{\left[\frac{1}{k_{D}} \frac{2p^{2} \cdot 1.414 K_{M}}{T^{0.75} (2p + p_{AS} + p_{AF})p_{0}}\right]}$$

$$= \frac{U_{3}}{(2540 p_{p} K_{M})}$$
(4-4)

The average k_{e} , of each set of data is as follows,

run 1 = 0.072
run 2 = 0.064
run 3 = 0.059
run 4 = 0.059
run 5 = 0.062
run 6 = 0.075
run 7 = 0.064
run 8 = 0.083
run 9 = 0.061

Since the latent heat of water is very high, a small error in the measurement of the transferred water would introduce a large error in U_3 . However, the value of k_e obtained from equation 4-4 is reasonably close to the value calculated by least squares, 0.0775.

For heat transfer coefficient for the copper sheet, we can obtain from equation 2-14 and 2-11,

$$\frac{1}{U_{c}} = \frac{1}{h_{F}} + \frac{1}{h_{R}} = \frac{1}{a'f(G.T)_{F}} + \frac{1}{a'f(G.T)_{R}}$$

a' = a constant

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By analysis of experimental results,

F3 =
$$U_c = \frac{9.5f(G.T)_F \times f(G.T)_R}{f(G.T)_F \times f(G.T)_R}$$
 (4-5)

Thus the constant a'=9.5. A plot of U versus F3 is shown in Figure 4-3.

Comparing the constants obtained for the membrane and the copper sheet, a/a'=1.5. This factor is believed to be due to the effect of the mass diffusion at the membrane surfaces on the film coefficient.

B. System with No Fresh Water Flowing

1. Empirical Correlation

In the case where there is no fresh water flow, the heat transfer coefficient of fresh water is by natural convection, and is proportional to a power of (TM2-TFM). For convenience, it was assumed that (TM2-TFM) was proportional to (TSM-TFM)/2.

Then from equation 3-25, combined with equations 2-11 and 2-12,

$$\frac{1}{K_{M}} = \frac{1}{a} \frac{(\Delta p)_{Lm}}{(\Delta T)_{Lm}} \frac{1}{f(G.T)_{S}} + \frac{1}{d} \frac{(\Delta p)_{Lm}}{(\Delta T)_{Lm}} \frac{1}{f(\Delta T/D)_{F}} + \frac{k_{e}}{k_{D}a} p_{p}$$

$$\cdot \frac{1}{f(G.T)_{S}} + \frac{k_{e}}{k_{D}d} \cdot p_{p} \cdot \frac{1}{f(\Delta T/D)_{F}} + \frac{b}{k_{D}} \cdot p_{p} \quad (4-6)$$

From the previous analysis the constants k_e , a, k_D were known and the constant d was evaluated and found to be d=52.6, when high flow rate data were used.

The equation 4-6 became as follows:



Figure 4-3. Experimental U_c versus function F3.

$$F3 = U_{c} = \frac{9.5f(G.T)_{F} \cdot f(G.T)_{R}}{f(G.T)_{F} + f(G.T)_{R}}$$



Figure 4-4. Experimental $1/K_{M}$ versus function F4.

$$F4 = \frac{1}{K_{M}} = 0.0683 \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} \frac{1}{f(G.T)_{S}} + 0.019 \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} \cdot \frac{1}{f(\Delta T/D)_{F}}$$
$$+ 13.4 p_{p} \frac{1}{f(G.T)_{S}} + 3.74 p_{p} \frac{1}{f(G.T)_{F}} + 2540 b p_{p}$$

$$F4 = \frac{1}{K_{M}} = 0.0683 \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} \frac{1}{f(G.T)_{S}} + 0.019 \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} \frac{1}{f(\Delta T/D)_{F}}$$
$$+ 13.4 p_{p} \frac{1}{f(G.T)_{S}} + 3.74 p_{p} \frac{1}{f(\Delta T/D)_{F}} + 2540 b p_{p} \quad (4.7)$$

Comparing d of equation 4-6, with the literature value of 43, the ratio is about 1.2. Since (TSM-TFM)/2 is a little larger than the value of (TM2-TFM), this ratio would also be close to 1.5 if true film ΔT values were used.

In Figure 4-4, all low flow rate (below 131 cc/min flow) data will not fit the equation 4-7, and were all lower than the predicted value of F4. This could be explained, for low flow, if the detected temperatures were lower than actual temperatures in the channel as discussed previously. However, the temperature of fresh water was probably more uniform because of no flow. Therefore, the measured TSM-TFM would be smaller than the actual value and TFM-TRM would be greater than the actual value. Possibly for this reason a higher U_m , k_m , U_3 and a lower U_c were obtained (As shown in Results in the Appendix, from page 56 to 64). For both fluids flowing, the deviations of temperatures occur on both sides in the same direction. Therefore, the differences of temperatures are probably more reliable with both fluids flowing. However, there would be some effect on vapor pressure, P_{AS} and P_{AF} , but this effect should not be large.

2. Heat Transfer

The heat transfer with no fresh water flow was not correlated because of suspected incorrect temperatures.

V. CONCLUSION

The following conclusions have been obtained from the data and results during the investigation.

- 1. Mass transfer through the membrane is by diffusion.
- 2. The overall mass transfer resistance is the sum of three terms in theoretical equation 2-25. The first term is proportional to heat transfer resistance associated with evaporation and condensation at both sides of the membrane. The second term is due to the parallel heat conduction through the membrane and is small compared to other two. The third term is diffusion resistance in the membrane.
- 3. The heat transfer coefficient in the liquid films is affected by flow rate in the expected manner, and also appears to be affected by the mass diffusion.

VI. RECOMMENDATIONS

If this membrane process is to be applied in industrial desalination further studies of this process are recommended. The following studies are believed to be the most important.

- A high flow rate study of this process at atmospheric pressure should be made.
- The process should be studied as near the boiling point as possible in order to decrease the existance of non-condensable gases in the membrane pores.
- 3. Better membrane qualities should be investigated.
- 4. Further studies appear to be desirable on the effect of ΔT across the membrane.
- The diffusion resistance in liquid films and the effect of liquid diffusion on heat transfer coefficient should also be studied.

VII. APPENDICES

A. Apparatus

The apparatus was made of Acrylic plastic plate. It consisted of three parallel channels. The two outside channels were 1.5 inches in height and 16.25 inches long. Both exterior channels were 0.5 inch thick. The center channel was 0.6 inch in thickness. A membrane separated hot salt solution which flowed in one exterior channel from fresh water flowing in the middle channel. A copper sheet in turn separated the fresh water from cold salt solution flowing in the other exterior channel.

The circulation of the salt solution and fresh water was provided by two Zenith metering pumps running at the same speed. A speed controller on a Zero-Max drive power block, could be adjusted to the desired flow rate from 0 to 210 cc/min. The power was supplied by a 1/6 horse-power AC motor.

The energy required for evaporation was provided by a gas burner which heated the cold salt solution, which flowed through a water bath coil, to a desired temperature before it entered the hot salt solution chamber of the evaporator-condenser. Ten thermocouples were used to measure the temperatures. One was placed at each end of each of the three channels, another four were placed in the inlet and outlet tubes of the salt solution channels. The apparatus used 1/4 inch copper and plastic tubing and suitable fittings, so that a closed liquid circulation system could be maintained.

B. Equipment

1. Evaporator-Condenser Gaskets

The channels of the evaporator-condenser were sealed by . rubber gaskets.

2. Liquid Reservoirs

Two one gallon containers (plastic) were used as reservoirs for salt and fresh water. Each was connected to the pump inlet line with a 1/4 inch plastic tube and valve.

3. Level Guage Tubes

Two 5/8 inch glass tubes 15 cm long were used as level indicators. They were connected to the pump suction line at the bottom (see Fig. App. 1).

4. Heater

A copper coil in a stainless steel container 10 inches in depth served as a water bath and a gas burner supplied the heat.

5. Thermocouples

Ten copper-constantan thermocouples made from size 20 A.W.G. wire, were used to measure the inlet and outlet temperatures of the channel fluid. One was fixed in the middle of each channel entrance and exit. The other four were fixed in the inlet and outlet tubes of hot and cold salt solutions.

6. Potentiometer

A potentiometer (Indicator, portable, model No. 1324, by the Winslow Company, Inc.) was used to measure the EMF of each thermocouple, for conversion to temperature units. 7. Motor and Pumps

Two Zenith metering pumps were used to maintain the ' circulation of fluids. The pumps were driven by a Zero-Max drive power block, model EL. The energy was supplied by a Zero-Max motor model M2, 1/6 horse-power, 115 v, 725 RPM.

8. Graduated Cylinder

A 25 milliliter graduated cylinder was used to collect the amount of water transferred.

9. Miscellaneous

Copper and plastic tubing and fittings were used for connecting the system. A glass disk and electric oven were used in making the membrane.

C. Materials

The materials used in this investigation are listed below.

1. Membrane

A purchased glass fiber filter paper (Grade 934AH, H. Reeve Angel and Co.) 6"x18", was used for preparing membrane.

2. Teflon Solution

E. I. Dupont's Teflon 30-B dispersion was used for making the membrane. It is an aqueous dispersion containing 59 to 61 percent solids. It has a density of 1.5 g/cc, a pH of 10 and a viscosity of 15 centipoise at room temperature. The density of Teflon solids is 2.2 g/cc and thermal conductivity is 0.625 BTU/(hr)(ft)(°F). 3. Condensate Water

Steam condensate from a condensate line was used to prepare the salt solutions and to serve as fresh water.

4. Salt

Industrial grade salt (NaCl) was used for preparing 7% by weight salt solution.

5. Silver Nitrate

Reagent grade silver nitrate was used to detect any Cl ion present in the condensate.

D. Procedures

1. Membrane Preparation

The glass fiber paper (6"x10") was dipped in an aqueous Teflon dispersion (about 3 cc of Teflon 30-B mixed with 40 cc of water), then dried over night at room temperature. The membrane was then heated in an oven at 500°F for 30 minutes. After drying in the oven, drops of fresh water were placed on the membrane surfaces to check the water repellent character of the membrane. Any unsatisfactory membranes, those which absorbed water or became .wet on the surface, were discarded. The membrane in the apparatus had a transfer area of 0.1695 ft^2 with a void fraction of about 0.9 and a thickness of 0.019 inches.

2. Experimental Procedure (see Fig. App. 1)

The first step was to assemble the evaporator-condenser and tighten it with bolts. The circulation line in the system was

then connected and each reservoir was filled with salt and fresh water accordingly. The reservoir valves were opened to the pump suction line, and the motor was started while both discharge line drain valves were open (S2, F2). Any dirt in the pump was flushed out, then valve S2 was closed, and salt solution entered the recovery section. After salt solution over flowed from the exit, valve F2 was closed. The salt solution and fresh water entered each side of the membrane at the same volumetric rate to prevent rupturing the membrane due to unbalance of pressure. After circulation for a while, the glass tubes had the same level as in the reservoirs. Then the make up valves, S1, F1, to the pump suction were closed, and the gas burner was started for heating up the system. After heating to the desired temperature, the vapor from the salt water began to transfer and gradually lowered the level in LS and increased the level in LF. Valve F5 was opened to let any transferred water flow to a graduated cylinder. Valve F3 was opened to the salt solution system to automatically make up the water loss and maintain constant concentration. After about two hours, a steady state was reached, and data of temperatures, time, flow and amount of water transferred could be taken.

Usually data readings were made once every 10 to 30 minutes. At least 7 sets of data were taken in each run and the flow was adjusted as desired.

For runs with no fresh water flow, the procedures were the same, except that after steady state flows were reached, the fresh water pump was disengaged from the driving great and the valve F4 was closed.

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Figure App 1. Cycle flow diagram of experimental process.

E. Trouble Shooting

1. Temperatures

Before four extra thermocouples were added to the inlet and outlet tubes, a heat balance over the system was unreasonable, and negative heat transfer coefficients were obtained, similar to Hsu's work (6). It was found that the inlet and outlet tube temperatures were greatly different from the temperatures measured in each exit and inlet channel (see Fig. App 2).



Figure App 2. The position of thermocouples in the channels and tubes.

After four thermocouples were installed in the tubes, they provided a reasonably reliable heat balance.

2. Bubble Elimination

When a heated solution is near the boiling point, both air bubbles and vapor bubbles might be generated in the fluid. These bubbles caused the system to become unstable. A vent line was installed at the heater exit which vented any generated bubbles to the atmosphere before the liquid entered the evaporator-condenser.

F. Evaluation of Heat Loss

A special experiment was run with the purpose of determining the heat loss. By using the same apparatus without membrane and a copper sheet, so that there was only one channel, hot water was circulated through the channel for about two hours until the temperature in the channel reached a steady state. The pump was stopped and the temperature of the channel was measured every ten minutes. The drop of temperature was due to heat loss to the air. Then a heat transfer coefficient could be calculated based on the total area exposed and the temperature. The calculation was as follows,

Vp Cp $\Delta T = F(Al + A2 + A3)(T_W - T_A) \cdot t$ $F = [Vp Cp (\Delta T)]/[(Al + A2 + A3)(T_W - T_A)_m^t]$ V = volume of the channel p = density of water at the temperature in the channel ΔT = temperature change with time, °F Cp = heat capacity of water Al = A3, A2 = the areas exposed to the air and refers to channels 1, 2, 3, -ft²

F = heat transfer coefficient of heat loss

 T_W = the mean temperature of the channel water T_A = the mean temperature of the air (room temperature) t = time, hr

Three successive different measurements gave results as follows for heat loss coefficient F, BTU/(hr)(ft²)(°F)

1. F = 0.7930

2. F = 0.8269

3. F = 0.8172

The average F in 0.8124.

This value was used to estimate heat losses from each channel in the experimental runs.

G. Data and Results

All the data obtained during this investigation were tabulated in the following pages. The following nomenclatures were used in the tables.

Gl = flow rate in hot and cold salt water channels, lb/(hr)(ft²)
G2 = flow rate of fresh water channel, lb/(hr)(ft²)
TH = thickness of the membrane, inch

R = gram of Teflon contained per gram of glass fiber membrane

TSI, TSO = hot salt water temperature in inlet and outlet tubes,

respectively, °F

TRI, TRO = cold salt water temperatures in inlet and outlet tubes respectively, °F

TSM = bulk mean temperature of hot salt water in the channel, °F TFM = bulk mean temperature of fresh water in the channel, °F TRM = bulk mean temperature of cold salt water in the channel, °F DPM = logrithemic mean vapor pressure difference, (in-Hg) UM = overall heat transfer coefficient for the membrane.

- UC = overall heat transfer coefficient for the copper sheet, BTU/(hr)(ft²)(°F)
- U3 = overall heat transfer coefficient for the membrane by conduction, BTU/(hr)(ft²)(°F)
- DW = mass transfer rate, lb/hr
- KM = overall mass transfer coefficient, lb/(hr)(ft²)(in-Hg)

DATA AND RESULTS

TABLE 1.	Temperature,	Heat	and	Mass	Transfer	Coefficients	for	the	System	with	Fresh	Water
	and Salt Wat	er Flo	w.									

		`								
RUN	1 BOTH	FRESH AND SAL	T WATP	P FLOW	ING					
	C1	G2			тн		R			
	4186.781000	3488.9850	<u>)(</u>	0.0	19500	C	.52210)5		
TSI	TSC TRI	TPO TSM	TFM	TRM	ЛРМ	UM	UC	U3	NW	КM
159.6	147.4 115.5	118.5 148.5	123.0	116.5	3.22	62.15	70.57	14.02	1.08	0.334
159.9	147.5 115.6	118.9 148.8	123.6	116.8	3.21	63.94	74.67	15.15	1.08	0.336
160.4	147.6 115.6	113.2 149.0	124.1	117.1	3.20	66.26	77.45	17.74	1.06	0.332
160.5	147.8 115.6	115.4 149.1	124.5	117.3	3.16	67.26	80.16	19.70	1.03	Ü.324
162.3	148.3 115.5	119.7 149.2	124.8	117.5	3.15	74.08	83.57	25.02	1.05	0.333
162.9	148.9 115.6	120.2 149.8	125.1	117.7	3.22	73. 40	89.82	25.02	1.05	0.325

AVERAGE VALUE

160.9 147.9 115.6 119.3 149.1 124.2 117.1 3.19 67.35 79.37 19.44 1.06.0.331

TABLE	2.
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RUN	2	ROTH	FRESH	AND	SALT	WATER	FLOWING
-----	---	------	-------	-----	------	-------	---------

	(51		G2	·		ТН		R			
	3324.(153000	277	0.0440	00	0.0	19560	(.5221	5		
TSI	TSN	TRI	TEO	TSM	TEM	ткм	DPM	МU	UC	U3	DW	КМ
166.3	149.8	111.3	117.9	121.0	125.6	115.6	3.53	65.31	74.31	1.6.55	1.12	0.316
166.2	149.5	111.1	117.8	151.8	125.1	115.2	3.57	65.39	74.48	16.99	1.13	0.315
165.5	148.9	110.9	117.4	151.1	124.5	114.6	3.52	64.97	74.58	17.15	1.11	0.316
165.2	148.7	111.2	117.7	150.6	124.2	114.8	3.48	64.67	77.80	16.39	1.12	0.322
165.0	148.4	111.1	117.6	150.4	124.2	115.0	3.43	65.58	78.20	17.46	1.10	0.322
164.3	147.9	111.0	117.4	149.9	124.1	114.9	3.35	66.41	78.13	18.63	1.08	0.322

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AVERAGE VALUE

165.4 148.9 111.1 117.6 150.9 124.6 115.0 3.48 65.37 76.25 17.20 1.11 0.319

TABLE 3.

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	(31		6 <u>2</u>			TH		P			
	2435.9	247000	202	29,0560	90č	0.0	19500	(.52210)5		* <u>.</u> *
TSI	TSO	TRI	T P O	TSH	TEM	TRM	OPM	IJМ	UC	U3	DW	КМ
172.6	150.0	107.6	117.6	151.6	124.5	113.0	3.61	63.34	71.88	12.78	1.20	0.331
172.7	150.6	110.5	118.9	152.2	125.1	114.9	3.66	62.06	76.93	13,96	1.14	0.311
173.9	151.5	114.7	121.4	153.4	126.6	117.5	3.73	62.53	66.06	13.04	1.16	0.311
174.8	152.3	116.4	123.0	155.5	127.7	119.2	4.00	60.19	69.29	13.13	1.14	0.284
174.3	152.3	116.8	123.5	156.5	128.4	120.0	4.11	59.80	76.51	15.41	1.08	0.263
173.9	152.3	117.3	123.0	155.5	128.7	120.6	3.88	62.59	72.31	15.54	1.09	0.282

<u>173.7 151.5 113.9 121.4 154.1 126.9 117.5 3.83 61.75 7C.17 13.98 1.13 0.297</u>

TABLE 4.

	(G 1.		62	·		TH		R			
	1674.	713000] 30	15.594	100	0.0	19500	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	1.52210)5		
TSI	TSO	TRI	TEO	TSM	TFM	TRM	DPH	UΜ	UC	U3	DW	КM
178.1	149.9	108.0	119.5	149.5	125.0	115.9	3.17	60.06	69.85	15.17	0.96	0.303
176.1	148.7	107.3	118.0	148.5	124.1	114.1	3.08	58.17	62.55	14.05	0.94	0.304
174.9	147.6	168.9	118.3	147.7	123.5	114.3	3.01	50.31	58.08	15.97	U.91	0.303
173.0	147.2	110.1	118.6	146.8	123.2	114.6	2.89	59.06	59.97	14.14	0.92	0.320
174.0	147.1	110.6	115.0	146.5	122.9	115.2	2.88	58.83	64.02	15.52	0.89	0.310
174.0	146.5	110.7	112.2	146.7	122.7	115.3	2.94	58.92	64.49	15.68	0.91	0.309

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TABLE 5.

RUN	5	BOTH F	RESH A	ND SAL	T WATE	R FLOW	ING					
	(51		G 2			тн		R			
	4186.	781600	348	3.985(550	0.0	18000	(.65658	37		
TSI	ΤSO	TRI	TRO	T SM .	TEM	TRM	DPM	UM	UC	U3	DW	КM
154.1	142.0	105.3	111.6	143.6	110.0	108.0	2.92	62.23	82.46	20.96	0.94	0.320
151.4	140,0	104.0	169.8	141.5	116.0	106.5	2.78	58.87	81.78	17.95	0.93	0.333
149.3	138.4	102.9	168.3	139.6	114.5	105.4	2.62	57.10	80.39	17.21	0.89	Ú.338
147.9	137.0	102.2	107.4	138.2	113.5	104.7	2.51	57.10	81.19	17.02	0.88	0.350
146.9	136.1	101.7	106.5	137.1	112.6	104.1	2.44	57.24	76.25	18.87	0.84	0.343
146.1	135.1	101.3	105.9	135.9	111.7	103.5	2.34	59.01	76.47	21.70	0.80	0.342
				····	<u> </u>			· · · ·				

AVERAGE VALUE

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149.3 138.1 102.9 108.3 139.3 114.4 105.3 2.60 58.59 79.76 18.95 0.88 0.338

TABLE 6.

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RUN	1 6	BOTH F	RESHA	ND SAL	T WATE	R FLGW	ING					
	(51		<u>62</u>	•		тн		R			
	5328.0	532000	442	10.5270	<u> </u>	0.0	18000	(.65658	37	· · · · · · · · · · · · · · · · · · ·	
TSI	TSO	TRI	TRO	TSM	TEM	TRM	DPM	IJМ	UC	U3	DW	КM
143.4	.134.6	104.7	107.4	134.7	112.3	105.9	2.14	63.06	76.42	22.87	0.80	0.374
143.5	134.8	104.5	107.6	135.1	112.4	106.0	2.17	60,90	87.52	22.03	0.78	0.359
144.0	135.1	104.8	108.0	135.5	112.9	106.3	5.13	63.73	87.46	24.16	0.79	0.362
144.4	135.3	105.2	108.4	135.9	113.4	106.6	2.20	65.05	86.30	25.13	0.430	0.362
144.6	135.3	165.5	108.7	136.2	113.7	107.0	2.22	67.05	86.45	27.24	0.79	0.358
144.6	135.5	105.8	1(8.9	136.4	113.9	107.2	2.23	64.77	84.01	24.39	0.81	0.361
۸V	ERAGE	VALUE		· · · · · · · · · · · · · · · · · · ·								
144.1	135.1	105.1	108.2	135.6	113.1	106.5	2.19	64.10	84.69	24.30	0.79	0.363

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RUN	! 7	BOTH F	RESH	ND SAL	T WATE	R FLOW	ING					•
	٢	51		G2			T 14		R			
	1065.	726000	8.8	8.1057	700	0.0	19500	(.52210)5		
TSI	TSO	TRI	TRN	TSM	ТЕЧ	TRM ·	DP M	UM	UC	U3	DW	КМ
175.0	143.0	106.0	115.4	135.2	118.0	112.4	1.70	62.62	67.20	18.47	0.67	0.394
173.4	141.7	106.1	114.6	133.2	117.1	112.2	1.53	63.72	65.78	21.09	0.61	0.396
171.9	140.8	105.6	113.8	132.2	115.4	111.2	1.46	64.44	60.59	21.17	0.60	0.411
177.2	142.0	105.1	113.3	132.8	116.4	110.7	1.55	67.96	57.06	23.27	0.65	0.420
180.2	142.6	105.6	113.4	134.0	115.9	111.3	1.63	71.94	58.40	28.57	0.65	0.398
176.0	141.5	106.2	113.8	133.7	116.8	111.5	1.61	66.14	56.86	23.41	0.64	0.395

AVERAGE VALUE

175.6 142.0 105.8 114.0 133.5 116.9 111.6 1.58 66.14 60.98 22.66 0.64 0.402

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TABLE 8.

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PUP	18	вотн г	RESH A	ND SAL	T WATE	R FLOW	ING					
	(51		G 2			тн		R			
	2435.0	947666	202	9.956(100	0.6	19500	(.52210	15		
TSI	τsο	TRI	TRO	TSM	TEM	TRM .	0 P M	МU	UC	U3	DW	КM
175.8	160.9	127.7	125.8	159.9	144.1	132.2	2.60	70.30	56.82	21.26	0.55	0.254
175.3	160.2	125.7	134.4	158.9	143.9	131.1	2.43	74.29	58.74	20.20	0.70	0.286
174.4	159.6	125.9	134.5	158.4	144.0	131.5	2.32	76.24	59,50	17.52	0.73	0.314
176.2	160.0	125.5	134.9	159.2	144.4	131.5	2.42	73.71	61.41	15.78	0.74	0.304
177.5	152.2	125.9	135.4	160.9	145.1	131.8	2.63	71.65	61.90	13.74	0.77	0.294
176.7	161.4	126.2	135.5	161.6	145.4	132.2	2.69	76.74	59.84	18.79	0.70	0.261
	ERAGE	VALUE						· · · · · · · · · · · · · · · · · · ·				
176.0	160.9	126.2	135.1	159.8	144.5	131.7	2.52	72.82	59.71	17.88	0.72	0.286

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TABLE 9.

									· · · · ·			
G 1		G 2			ТН		R					
	2435.9	947600	202	29.0560	00	0.0	19500	(.52210	25		
ΤSΙ	TSO	TFI	TRO	TSM	TFM	TRM	DPM	ЦМ	UC	U3	DW	КM
84.2	164.6	112.9	131.7	167.1	142.9	125.6	4,40	60.69	86.41	9.48	1.04	0.23
84.2	164.9	112.6	131.5	167.6	142.9	125.4	4.53	63.01	84.97	12.96	1.04	0.22
82.0	163.0	113.5	130.8	166.7	142.6	125.2	4.35	59.92	78.82	8.43	1.04	0.23
81.3	163.5	114.0	130.4	166.6	142.9	125.4	4.24	59,88	75.57	7.49	1.03	0.24
82.1	164.4	113.6	130.3	164.9	142.6	125.2	3.90	62.70	76.11	5.55	1.06	0.27
182.9	163.7	113.0	129.9	164.3	142.4	124.7	3.80	68.33	75.88	12.36	1.03	0.27

AVERAGE VALUE

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182.8 164.0 113.3 130.8 166.2 142.7 125.2 4.20 62.42 79.63 9.38 1.04 0.249

		SALI M		- LUWING	AND N		H WAIE	R FLUW				
	C	-1		62			тн		R			
	2435.0	147000		0.0000	00	0.0	18000	· (.4126.	32		
TSI	TSO	TRI	TBU	TSM	TEM	TRM	D P M	UM	UC	113	РW	КМ
156.8	138.7	102.5	115.2	142.5	127.1	114.0	1.81	85.53	64.07	25.36	0.82	0.451
159.0	140.3	103.4	114.4	144.1	129.5	116.4	1.77	93.36	71.00	29.26	0.82	0.463
159.8	141.V	104.1	115.6	145.5	131.0	119.1	1.83	94.15	73.47	27.87	0.85	0.465
160.1	141.9	105.0	116.5	146.9	132.0	119.2	1.93	88,75	75.35	22.92	0.86	0.447
160.0	143.3	105.3	117.1	149.0	133.0	129.4	2.16	78.98	76.94	16.89	0.38	0.405
161.6	144.0	105.6	117.5	149.6	133.6	120.9	2.19	81.49	77.52	19.53	0.87	0.399

TABLE 10. Temperature, Heat and Mass Transfer Coefficients for the System with Salt Water

Flow but No Fresh Water Flow.

159.7 141.5 104.3 115.6 146.3 131.0 118.2 1.95 87.05 73.06 23.64 0.85 0.438

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TABLE 11.

RUN	2	SALT	WATER	FLOWING	AND N	D FRESH	WATER	FLOW
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	G	51		62			тн		8			
 	1065.7	726600		0.0000	00	0.0	20000	C	.51250	06		
TSI	T S O	TRI	TPO	TSM	ΤΓΜ	TRM	ПРМ	() M	UC	113	DW	КM
 171.6	135.1	104.2	115.9	142.2	128.3	116.2	1.56	80,18	42.57	31.80	0.57	0.365
175.2	138.8	104.9	119.8	147.3	133.2	120.5	1.78	82.26	47.01	24.33	0.70	0.391
 177.1	141.1	105.6	122.4	149.3	136.2	123.4	1.71	87.86	51.22	37.67	0.56	0.325
176.3	142.7	106.8	124.7	1.50.5	138.3	126.1	1.66	87.50	56.46	28.80	0.61	0.371
 176.9	143.9	108.4	126.3	150.9	139.5	127.5	1.59	90.79	56.55	29.17	0.61	0.386
175.6	143.5	108.2	125.7	149.4	139.1	126.8	1.36	100.80	52.19	45.40	0.49	0.362

AVERAGE VALUE

175.4 140.9 106.4 122.5 148.3 135.8 123.4 1.61 88.25 51.00 32.86 0.59 0.367

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TABLE 12.

RUN 3 SALT WATER FLOWING AND NO FRESH WATER FLOW

	(51		- 6 2			TH		R			
	1674.	713000		0.0000	000	0.0	20000	. (.51250	0		
TSI	TSO	TRI	TRO	TSM	TFM	TRM	прм	UM	UC	* U3	DW	КM
184.3	156.6	121.4	133.1	162.7	148.6	135.0	2.50	95.62	55.50	27.29	0.83	0.333
185.3	159.5	124.0	136.3	166.0	152.1	138.2	2.66	91.34	55.32	23.51	0.82	0.309
184.6	150.1	125.1	127.9	165+4	154.1	139.9	2.13	109.90	55.79	21.71	6.81	0.382
184.9	160.8	125.2	138.6	166.4	155.6	141.4	2.07	105.49	57.80	21.04	0.80	0.386
186.9	162.5	125.7	135.7	168.7	157.1	143.0	2.34	102.80	61.69	26.10	0.78	0.333
185.7	162.0	125.9	139.8	168.1	157.3	143.9	2.12	168.80	62.57	34.10	0.70	0.331

AVERAGE VALUE

185.3 160.3 124.5 137.6 166.2 154.1 140.2 2.30 102.33 58.11 26.62 0.79 0.346

TABLE 13.

RUN	4	SALT 6	ATER F	LOWING	S AND A	ID FRES	H MATE	R FLOW				
	~	-1		62			T'4		Ð			
	2435.9	947000		0.0000	юс. <u>Г</u>	0.0	20010	(.5125)(.		
TSI	TSN	T 61	100	TSM	TEM	ТЗМ	Орм	UM.	UC	U 3	D₩	ΚM
186.7	170.1	130.4	147.2	175.7	164.3	140.3	2,59	102.87	45.69	35.84	0.56	0.256
187.0	171.0	140.5	148.8	176.2	165.3	149.2	2.47	108.89	47.05	33.38	0.71	0.287
188.0	171.7	140.0	140.6	177.3	155.7	149.3	2.70	99.68	47.36	37.30	0.62	0.228
188.6	171.2	141.0	149.5	177.0	165.0	142.4	2.89	95.41	46.95	31.83	0.53	0.235
189.1	171.4	140.0	144.1	177.1	165.4	149.3	2.70	110.29	46.16	48.93	0.52	0.228
195.6	170.4	140.3	149.4	174.5	164.0	149.2	2.38	111.53	48.81	41.31	0.59	0.282

And a second consideration of the second sec

AVERAGE VALUE

<u>187.6 171.1 140.8 148.5 176.4 165.2 142.1 2.57 104.85 47.01 38.10 0.64 0.252</u>

TABLE 14.

RUN 5 5	ALI	WATER	FLUWING	AND NO	FRESH	WATER	FLOW
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	C	;1		G 2			тн		R			
	3324.0	53000	· · · · · · · · · · · · · · · · · · ·	0.0000	000	0.0	20000		51250	00		
TSI	TSO	TRI	TRO	TSM	T F M	TRM	DPM	UМ	UC	U3	DW	КM
186.7	175.9	149.9	154.0	179.2	167.8	153.8	2.73	88,90	48.49	26.94	0.61	0.222
188.4	177.5	151.4	156.3	178.8	169.4	155.4	1.78	128.55	45.99	37.92	0.65	0.364
189.4	178,4	151.2	156.5	179.6	171.0	155.8	1.73	122+28	45.27	42.91	C.54	0.312
189.2	178.5	151.2	156.8	181.9	171.1	156.2	2.77	90.32	48.26	28.21	0.59	0.211
189.1	178.4	151.3	156.7	182.2	170.9	156.3	2.92	87.83	47.25	28.24	0.59	0.201
188.3	177.0	151.1	156.4	182.0	170.6	155.2	2.93	85.36	47.64	27.77	0.57	0.195

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AVERAGE VALUE

188.5 177.8 151.0 156.2 190.6 170.1 155.6 2.48 100.54 47.15 32.00 0.59 0.251
TABLE 15.

	(31		G2			ТН		F			
	4212.	156000		0.0000		. 0 . 0	20000	(0.51250	00		
TSI	TSO	TRI	TRO	TSM	TFM	TRM	ŊРМ	Uм	UC	U3	DW	КM
189.8	179.9	156.0	160.5	181.9	172.0	159.1	2.53	119.03	48.98	55.97	0.55	0.216
189.3	179.6	157.0	160.5	181.3	172.Ŭ	159.1	2.31	124.82	46.35	6().47	0.52	0.226
188.3	179.9	157.0	160.2	181.7	172.3	159.2	2.36	101.84	44.15	34.42	0.55	0.234
188.2	180.6	157.3	160.6	182.4	172.7	159.5	2.50	90.08	44.78	26.81	0.54	6.216
187.1	179.9	157.0	160.4	181.8	172.2	159.7	2.46	84.69	48.13	22.86	0.52	0.213
187.9	179.5	156.7	160.0	181.9	171.9	159.7	2.56	96.87	47.53	33.92	0.55	0.216

AVERAGE VALUE

188.4 179.9 157.0 160.4 181.8 172.2 159.4 2.45 102.89 46.65 39.08 0.54 0.220

TABLE 16.

RUN	17	SALT W	ATER F	LOWING	S AND N	10 FRES	H WATE	R FLOW				
:	G	1		G 2	٠		TH		R			
	1065.7	26000	· · · ·	0.0000	100	C.Q	19000	(1.48750	00		
TSI	TSO	I A L	TRO	Т 5м	TFM	TRM	DPM	UM	UC	U3	DW	КМ
181.6	149.0	93.3	120.6	144.6	128.9	114.1	1.94	57.21	63.59	11.89	0.77	0.395
176.7	147.9	92.9	118.2	142.6	127.6	112.1	1.79	63:31	55.97	7.70	0.74	0.414
179.3	150.3	92.6	116.3	138.2	125.2	109.8	1.40	76.78	52.76	15.06	0.71	0.506
184.1	151.7	93.4	116.2	135.9	124.1	100.4	1.22	87.29	56.36	13.71	0.17	0.635
186.0	152.7	94.5	118.4	137.2	125.4	110.5	1.25	90.32	57.25	18.17	0.75	0.606
186.3	153.6	94.9	119.3	138.2	125.9	110.8	1.32	85.24	56.65	19.45	0.71	0.540

182.3 150.9 93.6 118.1 139.5 126.2 111.1 1.49 78.36 57.10 14.33 0.74 0.516

TUDDD TI.	ΤA	BL	E	l	7	•
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	(51		G2			тн		R			
	1065.7	726000		6.0000	200	0.0	19000	1	.4875(00		
TSI	TSO	TRI	TRO	ΤSΜ	TEM	TRM	DPM	Uм	UC	U3	DW	КM
184.5	153.8	105.4	124.0	140.0	120.1	116.2.	1.21	85.59	56,45	18.13	0.65	0.534
185.6	154.4	108.4	126.0	142.2	131.6	113.9	1.23	90.87	55.30	23.88	0.62	0.50
185.7	155.3	109.5	128.3	143.2	132.2	120.3	1.19	94.86	53.94	26.98	0.60	0.50
186.0	156.3	110.0	128.7	143.9	134.1	121.0	1.18	95.00	54.15	24,43	6.51	0.51
186.1	156.5	116.6	129.0	144.7	134.9	122.1	1.21	92.87	54.54	24.66	6.59	0.49
186.4	157.0	111.1	128.9	145.3	135.2	123.0	1.25	92.67	55.28	24.63	0.60	0.48

AVERAGE VALUE

185.6 155.5 109.2 127.6 143.2 133.0 120.3 1.21 91.98 54.94 23.78 0.61 0.506

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	(31		G2			TH		R			
	1674.	713600		0.000	500	0.0	19000		.4875	00		
TSI.	TSO	TRI	TRN	TSM	TFM	TRM	прм	UM	UC	U3	DW	КM
187.8	154.8	125.7	132.8	157.5	145.4	131.6	1.97	92.76	57.32	28.33	0.69	0.34
186.0	165.6	127.1	139.2	158.3	147.2	134.2	1.83	92:34	56.90	26.29	0.65	0.35
185.8	154.8	125.1	132.5	157.3	146.2	133.0	1.80	<u>-1.91</u>	58.57	26.10	0.65	0.35
187.4	165.1	123.9	138.0	157.9	145.9	132.3	1.97	90.15	61.06	25.02	0.59	0.35
188.1	166.1	123.3	137.0	157.8	145.7	131.8	1.99	90.69	60.30	28.44	0.66	0.33
188.4	166.0	123.3	138.0	158.0	146.1	131.8	1.96	90.20	61.92	17.68	0.76	0.38

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3. Data Used for Least Squares with Equation is Tabulated as Follows:

	Let $p_p = \frac{2p^2 \times 1.414}{\overline{T}^{0.75}(2p+p_{AS}+p_{AF})p_{AF}}$		
	$h_{h} = \frac{1}{f(G.T)_{S}} + \frac{1}{f(G.T)_{F}}$		
$\frac{1}{k_{\rm m}} - \frac{b}{k_{\rm D}}$	$P_{p} \qquad \frac{(\Delta p)_{Lm}^{H}}{(\Delta T)_{Lm}} h_{h}$	p _p h.	$\frac{b[(\Delta T)_{Lm} - Z]}{\overline{T}}$
1.64	16.59	3.78	
1.67	16.96	3.76	
1.78	18.57	4.05	
l.76	17.13	3.75	
l.85	19.13	4.00	
l.76	18.62	4.04	
1.76	18.44	4.05	
1.66	16.83	3.77	
1.64	16.72	3.78	
1.73	17.13	3.75	

The following data were used for least squares with equation 4-6

1.64	16.59	3.78	0.65
1.73.	17.13	3.75	0.62
1.66	16.83	3.77	0.63
1.83	19.13	4.00	0.66
1.76	18.44	4.05	0.65
1.76	18.62	4.04	0.67

H. Computer Programs

The computer programs were used for the computations described in this thesis and are given in this appendix. The programs were written in Fortran IV language and were run in an IBM 360 system.

. 19 . 1	/WAT4 CN120288,TIME=1,PAGES=10 D LEE,J.M. JOB C CNCS490F LEE,JING-MING RESEARCH ANALYSIS CACULATION C HEAT AND MASS TRANSFER ANALYSIS PROGRAM C WSI FLOW PER HOUR LETTR.
	C INC,HIC,HIRC ARE THE MASS AND HEAT TRANSFER COEFFICIENT. C TSIEM,TSOEM,TEIEM,TEOEM,TRIEM,TROEM ARE THE MEAN TEMP.DURING PERIOD C DW IS THE COLLECTED WATER AMOUNT
	C DPM,DTM ARE THE LOG MEAN VALUE CE PRESS AND TEMP, RESPECT TO S.AND F. WATER C V1 IS THE SALTY WATER CHAMBER VOLUME TIMES DENSITY C V2 IS THE FRESH WATER CHAMBER VOLUME TIMES DENSITY
2	DIMENSIONTSIF(10), TSOF(10), TFIF(10), TFOF(10), TRIF(10), TROF(10) DIMENSION
345	DIMENSION TF1(10), TF2(10), TF3(10), TF4(10) DIMENSION TFR(10)
6	$\frac{V_1 = (16.5 \times 1.5 \times 0.5 / 12 \times 2) \times 61.3}{V_1 = (16.5 \times 1.5 \times 0.5 / 12 \times 2) \times 61.3}$
8	V2=(16.5*1.5*0.6/12**3)*61.3 J=9
10	$\begin{array}{c} DD & 1C I=1, J \\ 44 F=0.81243 \end{array}$
11	222 READ(1,101) W,THICK,WT,TWT
13	
15	<u>APF=0.</u>
$16 \\ 17$	AH=0. AT=0.
$\frac{18}{19}$	AKM≠0. AUM=0.
20 21	AUC=0. AU=0.
22	ADW=0
24	ADPM=0
25	TT2=0.
27	TT3=0.
29	
31	TF7=0.
32	WRITE(3,100) 1 WRITE(3,105)
34	WST=W*60./454. 55 WFT=0.
36	$\frac{1}{1000} \frac{1}{1000} \frac{1}{1000$
38	CA1=1.5*0.5/12.**2

39	CA2=1.5*C.6/12.**2
40	SF=WSI/CA1
41	FF=WFI/CA2
43	
44	VVF = WFI/(CA2*62, 4*3600,)
45	VVS = WSI/(CAI * 62.4 * 36CO.)
46	WRITE(3,400) SF, FE, THICK, TEW
47	WRITE(3,150)
48	400 FORMAT(/3X, 4(3X, F12.6))
49 50	100 FURMAILIHI,////4X,'RUN',14,/X,'FUR FRESH WALER NU FLUW') 105 FODMAT(///128.1011 128 1021 158 1741 128 1013
51	150 FORMAT(//2X, 1TS11, 3X, 1TS01, 3X, 1TR11, 3X, 1TR01, 3X, 1TRM1, 3X, 1TEM1, 3X,
7 1 7	1 TRM • 3X • DPM • 5X • UM • 4X • UC • 4X • U3 • 4X • DW • 4X • KM •)
52	DO 20 K=1,7
53	READ(1,102)ESI,ESO,EFI,EFO,ERI,ERO,TCR
54	READ(1,1,102)-ET1,ET2+ET3+ET4
55	ESI=ESI+16K/24.49 TE(ESI=2.45)1.1.2
57	1 + 1 + 2 + 4 + 2 + 1 + 2 + 2 + 2 + 2 + 2 + 2 + 2 + 2
59	$2 \text{ TSI} = 60.0 + (\text{ESI} - 2.45) \times 21.74$
60	3 ÉSO=ESO+TCR/24,49
61	$IF_{4} = SQ_{2} + 45) + 45$
62	$4-TSO_{=} FSO*24.49$
03	
65	6 FFT=FFT+TCR/24.49
66	IF(EFI-2.45)7.7.8
67	7 TFI= EFI*24.49
68	GO_{10} 9
69	$8 1 + 1 = 60 \cdot 0 + (1 + 1 - 2 \cdot 45) \times 21 \cdot 14$
	Y_ETU ETU TI UK/24 47
72	$11 T(1) = E(1)^{2} + 4 + 4 + 4$
73	
74	<u>12</u> TF0=60,0+(EF0=2,45)*21,74
75	13 ERI= ERI+TCR/24.49
76	IF(ERI-2.45)14,14,15
77	$14 \ 181 = E81 \times 24.49$
	15 TRI = 60.0 + (ERI-2.45) * 21.74
80	$16 FRD = FRD + TCR/24 \cdot 49$
81 -	IF (ERO-2.45) 17,17,18
Ř2	$17 \text{TRO} = \text{ERO} \times 24.49$
83	GO TO 21
84	18 IKU=6U+U+(EKU=2+47)*21+74
85	<u>とし、にししやてしませていたとなりが?</u>

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	86	6	
	87	7 22 T1=FT1*24.49	
	88		
	йŏ	0 - 23 I - 60 I - 2 (ET) - 2 (E) + 21 - 7(
	- Só	2 20 IL-000+T(L1122+40)*21+74 D 26 ET2-ET2+TC0 224 40	
	01		
	91		
	22		
	93	3 <u>GU 10 27</u>	
	-94-	+26-12=60++(F12-2-45)*21-74	
	95	27 ET3=ET3+TCR/24.49	
	96	5 IF (ET3-2,45)28,28,29	
	97	7 · · · · 28 T3=FT3*24,49	
	98		
	99	29 13-60 + (513-2) (5) + 21 7(
1	ιóó		•
1			
1		LEVEL472+421 32+32	
{	102-	32_14=+14*24.49	
1	103	3 · · · · · · · · · · · · · · · · · · ·	
3	04	22 T = 40 + 15T = 2 + 51 + 21 = 74	
1	24	$\frac{1}{2} = \frac{1}{2} $	
1	NS-	$34 (51F(K)=(151-2) \times (151-10K) / (100) \times (10K) \times (10K) \times (10K)$	
1	06	$1SUF(K) = (1SU - 2 \cdot *(1SU - 1CR)) (1CO - 1CR) 1 * 1 \cdot 8 + 32 \cdot 0$	
1	01	$IF IF (K) = (IF I - 2 * (TF I - ICR) / (100 - ICR)) * 1 * 8 + 32 \cdot 0$	
1	80.	3 TFOF(K)=(TFO-2.*(TFO-TCR)/(100TCR))*1.8+32.0	
1	09) TRIF(K)=(TRI-2.*(TRI-TCR)/(100TCR))*1.8+32.0	
1	10	TROF(K) = (TRO-2.*(TRO-TCR)/(10CTCR))*1.8+32.	
1	111	TE1(K) = (T1-2, *(T1-TCR)/(100, -TCR)) *1, 9+32,	
1	12	TE2(K) = (T2-2 - *(T2-TCR)/(100 - TCR)) * 1 - 8 + 32	
- î	112	TE3(k) = (T3-2) * (T3-TCR) / (100) = TCR) * (18432)	
		T = (1 + 1 + 1 + 2 + 1 + 1 + 2 + 1 + 1 + 1 +	
1		$\frac{1}{1} = \frac{1}{1} \left[\frac{1}{1} \left[$	
1	113		
1			
]	L18	1 + K(K) = 1 + K + 3 + 3 + 3 + 3 + 3 + 3 + 3 + 3 + 3	
	119_	920_CUNTINUE	
1	120	D = D = 0 = 30 = N = 1 + 6	
1	121	A=3.2438	
1	122	5 B=5.8683F-3	
-	123	$C = 1 \cdot 1702F - 8$	
	124	D=2.1878F-3	
÷	125		
2	122	$T_{5}T_{-}(T_{5}T_{-}(N_{1}+T_{5}T_{-}(N_{1}+1))/2 = -32 = 1/1 = 8$	i
2	122		
;	141-	$\frac{1}{1}$	
-	128		
	129		
	130	$0 \qquad 1R1 \pm ((1R1 \pm (N) \pm 1R1 \pm (N \pm 1))/2 = 32 \cdot 1/1 \cdot 3$	
	131	$1 \qquad x_{1} = 374 \cdot 11 - 151$	
	132	2 S1=2.303*X1/(TS1+273.16)*(A+B*X1+C*X1**3)/(1.+D*X1)	
	133	3 X2=374•11-TSO	

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	134	S2=2.303*X2/(TS0+273.16)*(A+B*X2+C*X2**3)/(1.+D*X2)
	136	$F_{1=2}^{+}$ 303*Y1/(TFI+273, 16)*(A+B*Y1+C*Y1**3)/(1, +D*Y1)
	137	Y2=374.11-TF0
	138	F2=2.303*Y2/(TF0+273.16)*(A+B*Y2+C*Y2**3)/(1.+D*Y2)
	39	<u>PS1=6528.5*EXP(-S1)</u>
i	40	PSZ=0028•0*EXP(-SZ) DE1-4529 5±EVP(-SZ)
. 1	42	
Í	43	PS1=PS1*0.96
]	44	P\$2=P\$2*0.96
.]	45	DP1=PS1-PF2
]	46	DP2=PS2-PF1
_	41	DPM = (DP1 - DP2)/(ALOG(DP1) - ALOG(DP2))
1	L48	
ļ	50	
1	51	$TSTEM=(TSTE(N)+TSTE(N+1))/2_{-}$
1	52	\dot{T} SOFM=(T SOF(N)+ T SOF(N + 1))/2.
j	53	TFIFM = (TFIF(N) + TFIF(N+1))/2.
	54	IF0EM=(IF0E(N)+IF0E(N+1))/2.
_]	55	$\frac{1}{1} R I F M = (TR I F (N) + TR I F (N+1)) / 2 $
		$\frac{1}{1} \left(1 \left(1 \left(1 \right) + 1 \left(1 \right) + 1 \right) \right) / 2 $
		1 F M L= (1 F 1 (N T 1 / T 1 F 1 (N) / / /) T F M D = (T F 2 (N 1)) T F D (N) / /)
4	159	$T = M_2 = (T = 2 (N + 1) + T = 3 (N))/2$
- :	60	TEM4 = (TE4(N+1) + TE4(N))/2
	161	DT1=TSIFM-TFOFM
:	162	DT2=TSQFM-TFIFM
	163	DTM = (DT1 - DT2) / (ALOG(DT1) - ALOG(DT2))
	164	
	166	TPM=(TPTFM=TPOFM)/2
-	167	$H_1 = 1.26 + 1.40 + 3.96 \times (1.50 + 1.50)$
	168	H2=TFM-32-
	169	DH=H1-H2
	170	DH1 = DW + H1
	171	
~~~	1-7-2	-{J}= {-1+M[ <del>~32,•()}×WS}= {++M2~32,•()}+×{WS}=0W}=VI×{{+S}{N+1}=+S{N+}}- &gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;&gt;</del>
	172	1*00•/1[NE - 00-/1TEM2_TEM4_1&UST_V1&(TD(N+1)_TD(N))*601TEME
	174	$\Delta \Delta C E = V2 \times (TE(N+1) - TE(N)) \times 60 \times (TEME+DW \times (TEME - 32))$
	175	_TCR=(TFR(N+1)+TFR(N))/2.
	176	Al=((21.*0.75)*2.+16.5*1.5)/144.
	177	$A2 = (21 \cdot *0 \cdot 6) * 2 \cdot /144$
	178	A3=A1

170	(9)T-M1*F*(TCM-TCP)	
180	$(1) 2 = \Lambda 2 \times F_{\pm} (T F M_{\pm} T \Gamma P)$	· · · · · · · · · · · · · · · · · · ·
100	$\Delta L 2 = A 2 + E + (T 0 M - T C 0)$	
101		
182		
-183	QC2=QR+QL3	
184	QC = QT - DH1 - QL1	
185	DT=TFM-TRM	
186	AU1=OC1/(DTM*AR)	
_187_	$\frac{112 = 0C2}{DT \times AR}$	
188	ΛU3-0C/(DTM±AP)	
100		
103	WKIIELS+20171FML+1FM2+1FM4+	(FM3, ISM, IFM, IRM, UPM, AU1, AU2, AU3,
	L DWL, IMC.	
_1.90	PS=(PS1+PS2)/2	
191	PE = (PE1 + PE2)/2	
192	$\Delta P S = A P S + P S$	
ີເດີລ		
101		
-1-9-4		
195	1=  1+ FM1	· · · · · · · · · · · · · · · · · · ·
196	TT2=TT2+TFM2	
197	TT3=TT3+TFM3	
104	TT4-TT4+TEM4	
100	ΤΕΚ-ΤΕΕ⊥ΤΟΜ	
133		
ZUU	1FO = 1FO + 1FM	· · · · · ·
201	1 + 1 = 1 + 1 + 1 RM	
_202	ADW=ADW+DW1	
203	$\Delta K M = \Delta K M + T M C$	·
204	$\Delta DTM = \Delta DTM + DTM$	
วักร่	$\Lambda D P M = \Lambda O P M + D P M$	
201	$A \square M = A \square M \pm A \square \square$	
200		
207	AUC=AUC+AUZ	
208	AU = AU + AU3	
209	AQC = AQC + QC	· · · · · · · · · · · · · · · · · · ·
-210-	30 CONTINUE	· · · · · · · · · · · · · · · · · · ·
211	WRITE(3.111)	
515		
212	TEM 2-TT2/4	
212		
-214	IFM 3=11370.	
215	TFM4=114/6.	
216	TSM=TE5/6	
517	TEM-TEA/A	•
210		
210		
_219_	$\underline{DPM=ADPM76}$	
220	IMC=AKM/6.	
221	DW1 = ADW/6.	
222	AU1 = AUM/6.	
222	AU2 = AUC/6.	
	$\Lambda U3 = \Lambda U/A$	
224		
227	UTH-AUTH/O.	

226	PS=APS/6.	
_227_	PE=APE/6.	
228	WRITE(3,201)TFM1, TFM2, TFM4, TFM3, TSM, TFM, TRM, DPM, AU1, AU2, AU3,	
	1DW1, TMC	
229	10 CONTINUE	
230	111_FORMAT(//3X, AVERAGE_VALUE')	
231	201 FORMAT(/7F6.1,F6.2,F7.2,3F6.2,F6.3)	
232	102 FORMAT(7 F10.5)	
233	STOP	
_234	END	

/DATA

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1	EV	FI	٠	1	ΜA	R	68	
1	EΥ	EL.		- 11	IT H	in.	υc	•

## IBM OS/360 BASIC FORTRAN IV (E) COMPILATION

	ç	GENERAL PROGRAM TO TEST THE NULL HYPOTHESIS
	6 C	-LINFAK-KEOKESSIUN
	č	NP=NO. OF PARAMETERS=IP+1
•	ç	REWRITE THE MODEL AS FOLLOWING:
	С	$WHERE \Lambda 1 IS \Lambda (IO*1) VECTOR$
•	č	A2 IS A ((IP-IQ)*1) VECTOR
C 0001	C	TESTING HO : A2=0
<u>-S-0001</u>		
Š.0003		DIMENSION $CR(20, 45) \cdot R(20, 45) \cdot S(20, 20) \cdot T(50, 20) \cdot YH(50) \cdot DY(50)$
S.0004		READ(1,500)MP
-5-0005	·····	
S.0007		$RFAD(1, 500) N \cdot IP \cdot IQ \to IP \cdot IQ$
Š.0008	-	NP=IP+1
<u> </u>	<u> </u>	- READ IN BY ROW
5.0010	1	DU = 1 = 1 + N READ(1 = 510) = Y(T) = (X(T = 1) = 1 + NP)
Š.0011		DO 455 MM=1,2
-S-0012		IF(MM-1)7,7,2
S-0013	4	
Š.0015		DO 3 K=1, IPQ
-5-0016		<u>- K1=K+1</u>
S-0017 S-0018	-	KZ=K1+1Q 3 T{T_K1}=X(T_K2)
S.0019		DO' 4 KK = 1, IO
-S.0020		-KK1 = KK + 1
S.0021	4	$(1 \cdot K \times 2) = X (1 \cdot K \times 1)$
S.0023	-	00 5 J=2,NP
<u>-S-0024</u>	i	
S.0026	Ĺ	WRITE(3,701)
S.0027	_	GO TO 9
-S-0028		/-WK11F(-3,-/(U))
S.0030	8	$\frac{1}{3}$ WRITE(3,710)(X(I,J),J=1,NP)
	C	TRANSFORM X TO SÍNGLE SÚBSCRIPT VECTOR
-5-00-3-1		-()) - () - () - () - () - () - () - ()
Š.0033		I = J + (K - 1) * N
S.0034	- 10	$\frac{1}{2} \times \frac{1}{2} \times \frac{1}$

Continued on next page.

5.0035			CALL GMTRA(XC,XTC,N,NP)
5.0036			$DD_{15}T=1$ · NP
5.0037			
<u><u><u></u><u>S</u><u>0038</u></u></u>			K=1+(1-1)*NP
5.0039		15	XT(T, J) = XTC(K)
	С		PRODUCT OF XT*X : XTX (NP BY NP)
S.0040	•		CALL GMPRD(XTC+XC+XTXC+NP+N+NP)
5.0041			DO 20 I=1.NP
5-0042			
S.0043			K = 1 + (1 - 1) + NP
<u>_S.0044</u>		20	XTX(1,J) = XTXC(K)
	C		PRODUCT OF XT*Y : XTY (NP BY 1)
S.0045			CALL GMPRD(XTC, Y, XTY, NP, N, I)
	С		IDENTITY MATRIX : AI (NP BY NP)
_ <u>S.0046</u>			DO 4C I=1,NP
5.0047			DO 35 J=1,NP
5.0048		<b>.</b>	1F(1-J)25,30,25
5.0049		25	
<u>S.0050</u>		20	
S-0052		35	
5.0053		40	CONTINUE
	С		AUGMENT XTX MATRIX TO INCLUDE XTY AND AI
S.0054			NP 1=NP+1
S.0055			NN=NP+1+NP
S+0056			
<u> </u>			
S.0059			
S.0060		45	XTX(I,JJ) = AI(I,J)
_S.0061		50	
5.0062	C		PRUCEDE DINCLITICE METHOD
5.0063			CR(1, I) = XTX(1, I)
_S.0064		55	$\underline{R(1, j)} = CR(1, j) ZCR(1, 1)$
5.0065			DO 68 I=2,NP
5.0000			
5.0068			
S.0069			DD 60 K=1,IM1
S.0070		60	SUM = SUM + P(K, I) * CP(K, J)
5.0072		65	$R(I_{\bullet}J) = CR(I_{\bullet}J) / CR(I_{\bullet}T)$
5.0073		68	CONTINUE
S.0074			IF(1P-4)69,69,72
S.0075		69	WRIIE(3,720)
5.0076			

5.0077	70	WRITE(3,730)(XTX(I,J),J=1,NN)
S-0078		$\begin{array}{c} UU & I \\ I = I \\ - \end{array}$
<u></u> <u>S_0080</u>		<u>GO TO (71,72,73,74,75),I</u>
S.0081	71	WRITE(3, 740)L, (CR(I, J), J=I, NN)
S-0082		$\frac{WRIIE(3, 750)L_{1}(R(1, J), J=1, NN)}{CO_{1}TO_{1}76}$
\$.0084	72	WRITE(3,741)L,(CR(I,J),J=I,NN)
S+0085		WRITE(3,751)L, (R(I,J), J=I, NN)
S-0085 S-0087	73	$WRITE(3, 742)I \cdot (CR(1, 1) \cdot I = I \cdot NN)$
S.0088	21	WRITE(3,752)L, (R(1,J), J=1,NN)
5.0089		GO TO 76
- S. 0090	74	$\frac{1}{10} \frac{1}{10} \frac$
5-0091		$\frac{1}{60} \frac{1}{76} \frac$
Š.0093	75	$WRITE(3, 744)L_{*}(CR(1, J), J=I.NN)$
S.0094		WRITE(3,754)L, (R(I,J), J=I,NN)
	76-	CONTINUE
5,0096	L	B(NP1)=0.
Š.0097		DO 80 I=1, NP
-\$-0098		L=NP1-I
5.0099		SUM=0.
5.0101		11 = 1 + 1
_Š_0102	78-	SUM=SUM+R(L+L)*B(LL)
S.0103	80	B(L)=R(L, NP1)-SUM
\$ 0104	L L	CALCULATE INVERSE UF XIX : S (NP BY NP)
-S.0105		
S.0106		00 90 J=I,NP
S.0107		JJ=NP1+J
5.0108		SUM=U.
5.0110	85	SUM=SUM+R(K,TI)*CR(K,JJ)
S.0111	20	S(I,J) = SUM
S•0112	95	
S.0114	**************************************	IM = [-]
Š.0115		DO 96 J=1, IM1
S.0116	96	S(1,J) = S(J,1)
		$n_{1} = 1.00$
š.0119	97	WRITE(3,731)(S(I,J),J=1,NP)
a	C	CALCULATE CROSS PRODUCT IN DOOLITTLE : CPID
-5-0120		$P(R_1) = 1 \cdot NP$

5.0122		III=I-1	· ·
5 0123		$\hat{\Gamma}\hat{P}\hat{\Gamma}\hat{\Gamma}\hat{I}\hat{I}=\Gamma \hat{R}(I,NP1) * \hat{R}(I,NP1)$	
	100	$\bigcup D T E (2, 52) ) T T (C P I O (T))$	
5.0125		$\frac{1}{10} \frac{1}{1} 1$	
5.0126		$\frac{1}{10}$ $\frac{1}{1}$ $\frac{1}{1}$ $\frac{1}{1}$ $\frac{1}{1}$ ND	
5.0127			
	101	$WRTTE(3, 530)TTT_B(T)$	
5.0129		TE(MM-1)102.102.105	
5.0130	102	WRITE(3, 770)	t
รักเริ่ม	102	DO = 1 O A = T = 1	
\$ 0135			
C /122			······
5.0136	100	UU 1US J=1,NP	
3.0134	103	$SUM=SUM+B(J) \times X(I,J)$	
5.0135		YH(I) = SUM	
-S.0136		DY(I) = Y(I) - YH(I)	
S.0137	104	WRITE(3,780)Y(I),YH(I),DY(I)	
	C	SST=TOTAL (UNCORRECTED) SSO	
	Ċ	SSRG=REGRESSION SSQ	
	Γ.	SSRS=RESIDUAL SSO	
5-0138	105	S1=A	· · · · · ·
C 0120	10.5		
S • 0159	104		
3.0140	100	21=21+1(1)*1(1)	۱
S.0141		SST=S1	
S.0142		\$2=0.	
S.0143		DO 107 I = 1.NP	
5.0144	107	S2=S2+CPID(I)	
5.0145		SSRG = S2	
\$ 0146		3922 - 722 = 2922	
\$ 0147		SSN = SSN	
5 • 01 <del>7</del> 1		550-5F10117 550-5F10117	
_ <u>_</u> .0140		33r0+33r0=330	
5.0149		NWZ = 1WZ	
S.0150		SUM=U.	
S.0151		DO 110 I = NQ2, NP	
_S.0152	110_	SUM=SUM+CPID(I)	
S.0153		SSPQ=SUM	
5.0154		NR=N-NP	
5.0155		ASSR0=SSR0/IP	
\$.0156		NPO = IP - IO	
-5-0157		A S SR S=SSR S/NR	
5 0158		$F_1 = \Delta S S F O / \Delta S S R S$	
C 0150		WRITE(3,540)	
5 0160		WRITE (3.550)N.SST	
		WRTFIZ,FGGISSG	······································
		WD TTEL3, 570 ND, SSPC	
2.0102		WARLAUJJIUNEJJJNU Udtelo sonije coda kooda ci	
2.0105		WRIELD, DOUIT, JONU, 400RU, EI TEIMM11115 115 120	
.5.0164		1F1MM=11112,112,140	

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S.0165	115	ASSPO=SSPO/NPO
5.0166		
5 01 (7		
3.0107		MK11E(3)3A01NKA122KA1422
<u></u>		<u>GO TO 125</u>
S.0169	120	ASSPQ=SSPQ/IQ
5.0170		
		1 CTAJJEV/AJJAJ UDITE/J EDINA CCDA ACCDA ED
5.01/1		MKTTE15+541118+22563+422563+65
_ <u>S.0172</u>	125	_WRITE(3,610)NR, SSRS, ASSRS
		MULTIPLE CORRELATION COFFEICIENT
5.0173		22 = 52 = 52 = 72
6 0174		
3.0114		WKIJE(D)02UJKZ
<u>S.0175</u>		WRITE(3,660)1P, NR
S-0176		IF(MM-1)130.130.135
č 0177	120	
3.0111	1 20	WKIJE(),000/NFW,NK
5.0178		60 10 140
5.0179	135	WRITE(3.660)IO.NR
5 0100	-126	
	140	
2.0181	4 7 7	CUNTINUE
S.0182	499	
S-0183	500	FORMAT(7110)
\$ 0184	505	EARMATILLI // 28 IDDODLEM 1 TIN
C 010E	- E 1 0	TONNAT(TINTT//TCATTENUOLUTETTITT
2.0102	510	+ORMAI(7+10.4)
S.0186	520	FORMAT(///,2X, PARAMETERS ARE')
5-0187	521	FORMAT(///.2X. ICROSS PRODUCT IN DOOLITTLE ARE!)
- c ^igo	7620	= CCMAT(7, 57, 10, 1, 10, 1) = 1 = 12 = 51
	250	
2.0189	231	
5.0190	540	FORMAT(//,8X,*SOURCE*,9X,*DF*,9X,*SS*,17X,*MS*,15X,*CAL_F*)
5.0191	550	FORMATU//.2X.ITOTAL (UNCORRECTED) 1.2X.I3.2X.E12.4)
5 0102	540	= COMAT(1/2) + D(20) + 177 + 11 + 27 = 12 + 1
3.0172	200	
2.0183	510	FURMAI(//,/X,'REGRESSIUN(B)',/X,13,2X,Fl2.4)
S.0194	580	FORMAT(//,2X,'R(B:B0)',13X,I3,2X,F12.4,5X,'MSR=',F12.4,F15.4)
C 0105	500	
2.0192	590	_EURMA!(//,2X,*R(A2;3(0),A1)*,/X,13,2X,E12.4,4X,*MS12=*,E12.4.
		11-15-4)
5.0196	591	FORMAT(//.2X.(A):B(A):A)(.A2)(.7X.(3.2X.F))(.4.4X.(MS2)) = F(2.4)
0002/0		1515 A)
5 0107	(10	
2.0191	610	<u>EURMAIL////28,*KES10UAL*+L28+L3+28+L2.4+48,*S**2=++E12.4</u>
S•0198	620	FORMAT(///,2X,'MULTIPLE CORRELATION COFFETCIENT IS!.//.8X.'R**2=!
		1F7.4)
\$ 0100	660	= EODMAT/// OV STABLEATED E/1 10 1 10 ()=()
-3.0177		= 0  Interval  + 1  ADUCATED = 0  Interval  + 1  Inte
2.0500	001	<u>たいというしょくメットアムないしん(ED) たく、エットッキススット)=+)</u>
S.0201	700	FORMAT(//+1X+*MODEL : Y=XO*B(0)+X1*A1+X2*A2*+//+2X+*X MATRIX IS*)
5.0202	701	= FORMAT(77.1X.*MODF[ : Y=XO*B(O)+XO*AO+XI*AI*77.5Y.*Y MATDIV +C*1
<u>s</u> _0202	715	= CDDMAT(A, RY, 1) = 10 - 2 + (AC, 0)
5.0204	120	FURMALLY/ 2X - UUULIILE PRINT OUT)
5.0205	/30	HURMAT(/,8X,11F10.2)
S.0206	731	FORMAT(/,8X,10F11.4)
-5.0207	740	FORMAT(2 2X TORITILITED 2)
	· · ·	I site it be week filler and the taken to the taken to the taken the taken to the t

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Continued on next page.

S.0208 S.0209 S.0210 S.0211 S.0212 S.0213 S.0214 S.0215 S.0216 S.0216 S.0217 S.0218 S.0218 S.0219	741 FORMAT(/,3 742 FORMAT(/,3 743 FORMAT(/,3 744 FORMAT(/,3 750 FORMAT(/,4 751 FORMAT(/,4 752 FORMAT(/,4 753 FORMAT(/,4 754 FORMAT(/,4 760 FORMAT(//, 770 FORMAT(//,	X, CR(',Il,') X, CR(',Il,') X, CR(',Il,') X, CR(',Il,') X, R(',Il,') X, R(',I1,') X, R(',I1,') X	<pre>,10X,10F1C.2) ,20X,9F10.2) ,30X,8F10.2) ,40X,7F10.2) 11F10.2) 10X,1CF10.2) 20X,9F10.2) 30X,8F10.2) 30X,8F10.2) 40X,7F10.2) XTX IS') YH',9X,'DY')</pre>		
S.0220 S.0221	END	·			

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