



Missouri University of Science and Technology
Scholars' Mine

Chemistry Faculty Research & Creative Works

Chemistry

01 Jul 2016

The First Potential Energy Surfaces for the C_6H^- -H₂ and C_6H^- -He Collisional Systems and their Corresponding Inelastic Cross Sections

Kyle M. Walker

Fabien Dumouchel

François Lique

Richard Dawes

Missouri University of Science and Technology, dawesr@mst.edu

Follow this and additional works at: https://scholarsmine.mst.edu/chem_facwork

 Part of the [Chemistry Commons](#), and the [Numerical Analysis and Scientific Computing Commons](#)

Recommended Citation

K. M. Walker et al., "The First Potential Energy Surfaces for the C_6H^- -H₂ and C_6H^- -He Collisional Systems and their Corresponding Inelastic Cross Sections," *Journal of Chemical Physics*, vol. 145, no. 2, American Institute of Physics (AIP), Jul 2016.

The definitive version is available at <https://doi.org/10.1063/1.4955200>

This Article - Journal is brought to you for free and open access by Scholars' Mine. It has been accepted for inclusion in Chemistry Faculty Research & Creative Works by an authorized administrator of Scholars' Mine. This work is protected by U. S. Copyright Law. Unauthorized use including reproduction for redistribution requires the permission of the copyright holder. For more information, please contact scholarsmine@mst.edu.

The first potential energy surfaces for the $C_6H^- - H_2$ and $C_6H^- - He$ collisional systems and their corresponding inelastic cross sections

Kyle M. Walker,^{1,a)} Fabien Dumouchel,^{1,b)} François Lique,^{1,c)} and Richard Dawes^{2,d)}

¹LOMC-UMR 6294, CNRS-Université du Havre, 25 rue Philippe Lebon, BP 1123, 76 063 Le Havre Cedex, France

²Department of Chemistry, Missouri University of Science and Technology, Rolla, Missouri 65409, USA

(Received 29 April 2016; accepted 21 June 2016; published online 14 July 2016)

Molecular anions have recently been detected in the interstellar and circumstellar media. Accurate modeling of their abundance requires calculations of collisional data with the most abundant species that are usually He atoms and H_2 molecules. In this paper, we focus on the collisional excitation of the first observed molecular anion, C_6H^- , by He and H_2 . Theoretical calculations of collisional cross sections rely generally on *ab initio* interaction potential energy surfaces (PESs). Hence, we present here the first PESs for the $C_6H^- - H_2$ and $C_6H^- - He$ van der Waals systems. The *ab initio* energy data for the surfaces were computed at the explicitly correlated coupled cluster with single, double, and scaled perturbative triple excitations level of theory. The method of interpolating moving least squares was used to construct 4D and 2D analytical PESs from these data. Both surfaces are characterized by deep wells and large anisotropies. Analytical models of the PESs were used in scattering calculations to obtain cross sections for low-lying rotational transitions. As could have been anticipated, important differences exist between the He and H_2 cross sections. Conversely, no significant differences exist between the collisions of C_6H^- with the two species of H_2 (*para*- and *ortho*- H_2). We expect that these new data will help in accurately determining the abundance of the C_6H^- anions in space. *Published by AIP Publishing.* [<http://dx.doi.org/10.1063/1.4955200>]

I. INTRODUCTION

The interstellar medium (ISM) is host to a number of molecules including both neutral and ionic species. Although linear carbon-chain anions have been included in interstellar chemistry models for some time, their detection has only recently occurred due to the previous lack of fundamental laboratory data.

The first observed molecular anion was C_6H^- , detected in the circumstellar envelope of the evolved carbon-rich star IRC +10216 and within the dense cold molecular cloud TMC-1.¹ Since then, the identification of five other molecular anions, C_4H^- , C_8H^- , CN^- , C_3N^- , and C_5N^- , has been made possible mainly through laboratory spectroscopic data.^{2–6} The detection of these anions and subsequent abundance analysis is not only important in constraining the chemical network of the ISM, the most modern of which includes over 4000 chemical reactions and 400 species,⁷ but their presence also directly impacts the free electron density and therefore affects the rates of cloud collapse and star formation.

However, to model the physical and chemical conditions in astrophysical environments containing anions, collisional rate coefficients with the dominant colliders are needed. Indeed, emission spectra are interpreted through detailed radiative transfer calculations, which requires the knowledge of both collisional and radiative rates. If theoretically calculated

or experimentally generated collisional rate coefficients are unavailable, the quantum level populations are usually estimated by assuming local thermodynamic equilibrium (LTE), the approximation that the level populations follow a Boltzmann distribution. Generally, though, this is a poor approximation in cool, low-density regions where anions are detected.

Rate coefficients for the rotational excitation of C_2H^- and OH^- molecules due to collision with He have been computed recently by Dumouchel *et al.*⁸ and Hauser *et al.*⁹ respectively. However, these molecules, even if suspected to be in the ISM, have not been detected and the only fully relevant astrophysical data are those for the $CN^- - H_2$ collisional system computed by Klos and Lique.¹⁰ The $CN^- - H_2$ rate coefficients were obtained from a new reliable *ab initio* potential energy surface of the $CN^- - H_2$ complex and the quantum scattering calculations were performed to investigate rotational energy transfer in collisions of the CN^- molecule with both *para*- H_2 ($j = 0$) and *ortho*- H_2 ($j = 1$) molecules. However, no experimental scattering data for anionic systems of interstellar interest have yet been obtained.

Hence, there is a real lack of collisional data for (interstellar) molecular anions that prevent accurate analysis of the anionic emission spectra that can be recorded with highly resolved telescopes like the Atacama Large Millimeter/submillimeter Array (ALMA) interferometer. Among the interstellar molecular anions, C_6H^- is of particular interest. C_6H^- has been found in several star-forming regions^{1,11–13} and is one of the most abundant anionic species. He and H_2 are generally the dominant colliders in interstellar and circumstellar media, the predominant one being H_2 .

^{a)}Electronic mail: kyle.walker@univ-lehavre.fr

^{b)}Electronic mail: fabien.dumouchel@univ-lehavre.fr

^{c)}Electronic mail: francois.lique@univ-lehavre.fr

^{d)}Electronic mail: dawesr@mst.edu

Our aim is then to compute collisional rate coefficients for the C_6H^- -He and C_6H^- - H_2 systems in order to provide the astrophysical community with data for a second detected anion besides CN^- . Within the Born-Oppenheimer approximation, the study of inelastic collisions requires two steps: (i) the calculation of an *ab initio* PES describing the interactions between the particles in collision (ii) the study of the dynamics of nuclei on this surface. Accurate rate coefficient calculations require a high-quality PES. Interaction PESs between long carbon chains and He or H_2 are complex and difficult to represent because of the size of the target.¹⁴ Indeed, for short intermolecular distances, the interaction is typically moderate to possibly weakly attractive for a T-shape approach and is often extremely repulsive upon linear approach. This may lead to singularities in the angular expansion and severe oscillations in the numerical fit of the PES over the usual Legendre expansions that are used in quantum scattering calculations.

In this paper, we present the first PES for the C_6H^- - H_2 and C_6H^- -He collisional systems. Special care has been put to their computation in order to have analytical PESs that can be used with confidence in quantum scattering calculations. Details of the electronic structure calculations are given in Section II A, while Section II B describes the surface fitting procedure. As an application, the first dynamics calculations are presented in Section III and the conclusions and prospects for the future are given in Section IV.

II. COMPUTATIONAL DETAILS

A. Electronic structure calculations

For the C_6H^- - H_2 system, both monomers were held rigid in collinear arrangements with the internuclear bond distance for H_2 fixed at $r_{HH} = 0.76665 \text{ \AA}$ (the vibrationally averaged bond distance for *para*-hydrogen, $j = 0$). The geometry of C_6H^- was optimized at the explicitly correlated coupled cluster level of theory using an explicit correlation consistent valence triple zeta basis set [CCSD(T*)-F12b/VTZ-F12].^{15,16} The alternating pattern of bond distance parameters and Jacobi coordinates is shown in Figure 1.

Since the PES represents the van der Waals (vdW) interaction of two closed shell species, single reference coupled-cluster based methods were chosen. The Molpro electronic structure package was used for all of the calculations reported here.¹⁷ In previous studies we have compared the performance of standard and explicitly correlated (F12) coupled cluster for a range of basis sets.¹⁸⁻²⁰ Table I compares the interaction energy at the global minimum (collinear structure with intermonomer separation of 6.426 \AA) obtained by several high-level methods. The higher symmetry for the collinear structure enables benchmarking with large basis sets and testing the effect of core-correlation. Explicitly correlated CCSD(T)-F12b calculations were performed using the VTZ-F12 and VQZ-F12 bases, correlating the valence electrons and the CVTZ-F12 and CVQZ-F12 bases, correlating all electrons. Standard CCSD(T) results for valence correlation only (since the effect of core-correlation was found to be negligible) were produced using the AVTZ and AVQZ bases

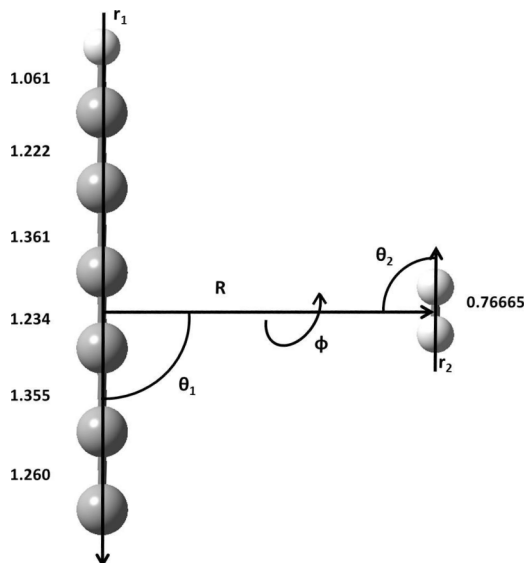


FIG. 1. The four intermolecular coordinates are illustrated. The optimized bond distance parameters for C_6H^- are listed at left. The r_{HH} distance at right is the empirical vibrationally averaged distance for *para*-hydrogen.

for comparison. The explicitly correlated results are already remarkably well converged at the triple zeta level and indicate negligible contribution from core-correlation. Well depths of -710.5 cm^{-1} and -710.7 cm^{-1} were obtained at the VTZ-F12 and CVQZ-F12 levels, respectively. These are both close to the value of -712.1 cm^{-1} on the fitted surface. In contrast, the standard CCSD(T) interaction energy (without counterpoise correction or mid-bond functions) is -765.6 cm^{-1} for AVTZ and -707.0 cm^{-1} for AVQZ. Given the large number of electrons in this system (40), an explicitly correlated method was chosen in order to obtain well converged energies with a moderate basis set at relatively low cost. We and others²¹ have noted some slight numerical issues with F12 methods when considering the finest details of the long range part of the potential for neutral systems. No discrepancies in our data were noted for this ion-neutral system ranging out to 22 \AA separation. The perturbative triples (T) contribution is not directly included in the F12b explicit correlation formalism and so in some cases we have used basis extrapolation in order to better converge that contribution. Here we scaled the (T) contribution based on the MP2-F12/MP2 correlation-energy ratio from second-order Møller-Plesset perturbation theory, implemented in Molpro as (T*). A lower-level guide surface using an explicit correlation consistent valence double

TABLE I. Collinear arrangement of rigid fragments with center-of-mass separation of 6.426 \AA .

Method	Well depth (cm^{-1})
CCSD(T)-F12b/VTZ-F12	-710.5
CCSD(T)-F12b/VQZ-F12	-710.7
(AE)-CCSD(T)-F12b/CVTZ-F12	-709.4
(AE)-CCSD(T)-F12b/CVQZ-F12	-711.9
CCSD(T)/AVTZ	-765.6
CCSD(T)/AVQZ	-707.0

zeta basis set [CCSD(T*)-F12a/VDZ-F12] was used to avoid computing expensive high-quality data in high-energy regions. The T_1 -diagnostic was monitored and found to be roughly 0.017 for all geometries in the high-level data set. For the C_6H^- -He system (with the same number of electrons, 40) the same level of theory was used, but without a low-level guide surface. Since default auxiliary basis sets necessary for F12 calculations are not defined for helium with the VTZ-F12 basis in Molpro, the augmented correlation consistent valence quintuple zeta basis from second-order Møller-Plesset perturbation theory (av5z/mp2fit)²² and segmented contracted highly polarized quadruple zeta valence quality (def2-qzvpp/jkfit)²³ basis sets were specified for the density fitting (DF) and resolution of the identity (RI) bases, respectively. In contrast to the good numerical behavior of the CCSD(T*)-F12b calculations for the C_6H^- - H_2 system out to 22 Å separation, some small numerical discrepancies were noted beyond 17 Å for C_6H^- -He. Tests with various other auxiliary bases did not improve the behavior. Thus, to determine an analytic long-range for the 2D C_6H^- -He system, standard CCSD(T)/AVTZ and CCSD(T)/AVQZ energies were extrapolated to the complete basis set (CBS) limit. The CCSD(T)/CBS data matched the CCSD(T*)-F12b/VTZ-F12 interaction energies very closely in the range of 9–15 Å, allowing a smooth switch to be applied (see below).

B. Analytic representation

The four dimensional (4D) intermolecular C_6H^- - H_2 potential includes 4691 symmetry-unique high-level *ab initio* energy data and is represented analytically by the interpolating moving least squares (IMLS) method^{19,24,25} using a weight function to interpolate between local fitting basis expansions. The C_6H^- - H_2 system is similar to the previously studied CO_2 - CS_2 ,²⁶ $(NNO)_2$,^{19,27} $(OCS)_2$,²⁸ and $(CO)_2$ ^{20,29} systems from a fitting standpoint (weakly interacting rigid linear monomers) but has some differences with respect to symmetry. With two different monomers, the system lacks monomer exchange symmetry, but the H_2 monomer is symmetric with respect to exchange of the two end-atom nuclei (180° rotation). The fitting basis (mentioned below) can be adapted to treat this symmetry by placing a simple constraint on the basis indices. As discussed previously,^{26,28} this is more complicated when the basis is used interpolatively (with changing weights) so here we employ the simple procedure of adding the symmetry partner for each symmetry-unique *ab initio* data point to the fitting set (flipping the H_2 fragment). There is no additional cost in terms of electronic structure calculations and for cases of relatively low permutation symmetry (a factor of only two here) the fitting set does not become too unwieldy. For systems with very high permutation symmetry, the development of a permutation invariant basis would be preferred.³⁰ The automated procedure that was developed to construct 4D PESs for CO_2 - CS_2 , $(NNO)_2$, $(OCS)_2$, and $(CO)_2$, has been described in detail previously,^{19,20,25,26} and was employed here. The same inter-monomer coordinates and a fitting basis of 301 functions composed of products of radial functions with associated Legendre bend functions were used. The same distance metric, interpolative weight

function, and SVD-based dynamic conditioning procedure were also used. For C_6H^- - H_2 the range of inter-monomer center-of-mass distances was $R = [2.3, 22.0]$ Å, while the fitted energy range included all stable geometries (the global minimum has a well depth of -712.1 cm^{-1}), but was restricted to about 2100 cm^{-1} above the separated monomers asymptote. As was done previously, to avoid computing and discarding costly high-level *ab initio* data in highly repulsive regions, an initial lower-level guide surface was constructed (at the CCSD(T*)-F12a/VDZ-F12 level). For the low-level surface, a set of 3000 symmetry unique points was distributed according to a Sobol sequence³¹ subject to an exponential R-dependent bias that favors points at $R = 2.3$ Å over points at $R = 22.0$ Å by a factor of about 20 (making the short-range repulsive region much more densely sampled). As before, the guide surface was fit using the same IMLS scheme as the final high-level PES, but with a smaller fitting basis of only 40 functions per local expansion. For the high-level PES (at the CCSD(T*)-F12b/VTZ-F12 level), 2000 initial seed points were distributed the same way according to the exponentially radially biased Sobol sequence, but with high-energy regions excluded by the lower level guide surface. Starting from the 2000 seed points, sets of 48 automatically determined points were added in each of a series of iterations until the estimated RMS fitting error was reduced to below 1.0 cm^{-1} . The accuracy of the final PES was tested using a random set of 288 points, confirming the estimated sub-wavenumber accuracy. The PES generation algorithm and fitting error estimate method have been described previously and were applied here with the entire coordinate and energy range fit without bias, in an automated fashion. The PES generation algorithm was terminated and finalized for use in this case with a total of 4691 symmetry unique points. This is considerably more than the roughly 2000 points used to fit other 4D systems.^{19,20,26} The necessity for more points is likely due to the extreme anisotropy in the PES plotted in Figures 2 and 3.

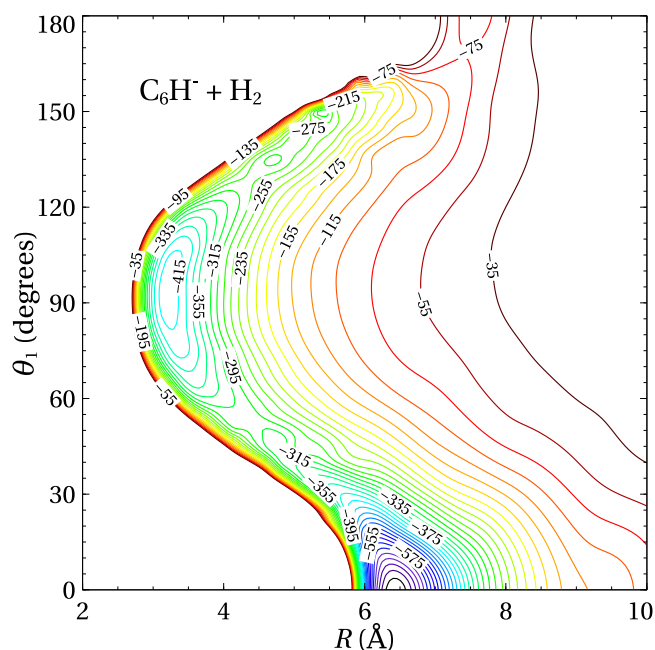


FIG. 2. Plot of C_6H^- - H_2 PES with $\theta_2 = 0$ (end-on approach).

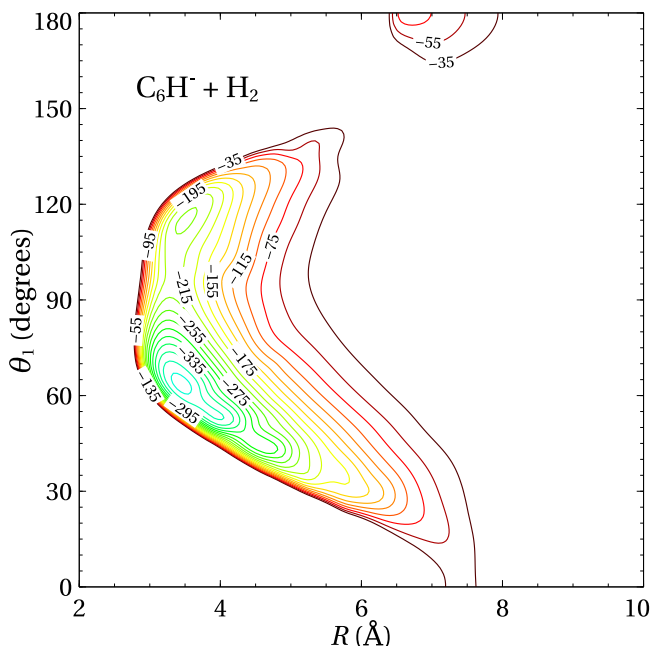


FIG. 3. Plot of $C_6H^-H_2$ PES with $\theta_2 = 90$, and $\phi = 0$ (side-on, planar approach).

The long length of the C_6H^- fragment introduces anisotropy with respect to approach of the H_2 fragment either towards the ends or side of C_6H^- . In addition, there is a remarkably strong anisotropy with respect to the orientation of the H_2 fragment (see Figures 2 and 3). End-on approach (θ_2 fixed at 0 or π) of H_2 includes geometries such as the completely collinear global minimum (-712.1 cm^{-1}) with H_2 interacting with the C-end of the C_6H^- fragment. The side-on approach of H_2 (θ_2 fixed at $\pi/2$), however, contrasts with

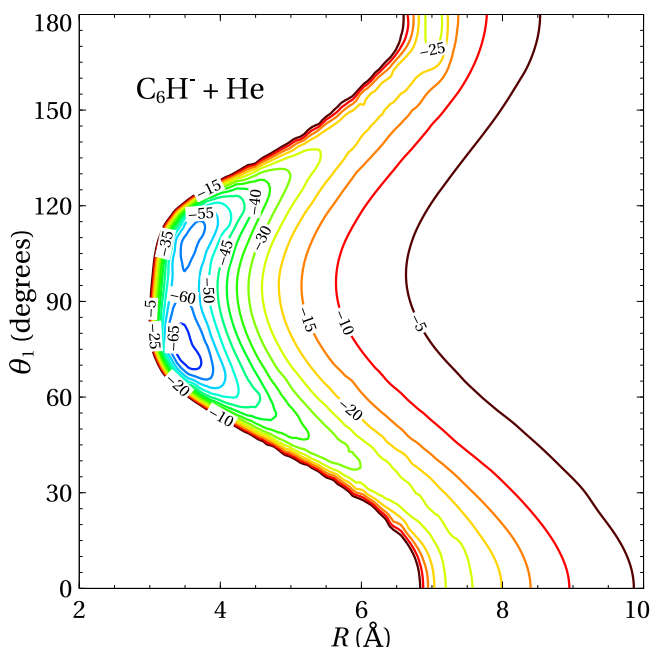


FIG. 4. Plot of 2D C_6H^-He PES. The global minimum energy (-68.76 cm^{-1}) is found at $R = 3.5055 \text{ \AA}$ and $\theta_1 = 74.23^\circ$.

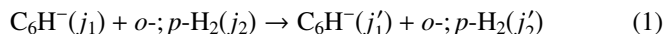
the end-on approach since the completely collinear minimum disappears.

To fit the 2D PES representing the interaction potential of the C_6H^-He system, a similar, but simplified procedure, was used. A pruned fitting basis of 39 functions (radial and Legendre, each with maximum order of 10) was used for the interpolation. Similar to the 4D PES described above, an initial radially biased set of points was determined, again using a Sobol sequence. Beginning with 400 points in the range $R = [2.3, 17] \text{ \AA}$, generations of automatically placed points were added, stopping at 953 points when the mean fitting error was below 0.3 cm^{-1} . As mentioned above, a long range analytic form was determined using CCSD(T)/CBS data. An R-dependent switch of the form $\tanh(3.5 * (R - 11))$ was used to smoothly switch between the IMLS and analytic forms. Thus, the PES is a 50:50 mixture of the two contributions at 11 \AA , and the switch is numerically complete before 15 \AA . The C_6H^-He PES is plotted in Figure 4. The global minimum energy found on the fitted PES is -68.76 cm^{-1} at $R = 3.5055 \text{ \AA}$ and $\theta_1 = 74.23^\circ$.

III. INELASTIC CROSS SECTIONS

A. Methods

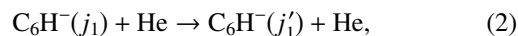
We consider collisions of C_6H^- with *para*- H_2 and *ortho*- H_2 such as



where j_1 and j_2 designate the rotational levels of C_6H^- and H_2 , respectively.

The calculations are restricted to low/moderate collisional energies ($E_c < 500 \text{ cm}^{-1}$). The rotational energy levels of the C_6H^- and H_2 molecules were computed using the experimental spectroscopic constants of $B_e^{C_6H^-} = 0.045927$, $\alpha^{C_6H^-} = 3.3356 \times 10^{-5}$, $D_e^{C_6H^-} = 1.079 \times 10^{-9}$, and $B_e^{H_2} = 60.853$, $\alpha^{H_2} = 3.062$, $D_e^{H_2} = 0.0471$.^{1,32,33}

We also consider collisions of the C_6H^- anion with helium,



where j is the orbital angular momentum quantum number.

Since the $C_6H^-H_2$ PES is extremely anisotropic, the surface was further adapted from the analytical fits. Chapman and Green³⁴ found that extremely anisotropic interactions, specifically HC_3N-He , lead to expansions that may be slowly convergent and subject to numerical instability. This was the case in our present study where the PES rapidly varies, so a regularized potential was introduced to eliminate the numerical issues. The procedure detailed below, adapted from Wernli *et al.*¹⁴ describes the modification to the $C_6H^-H_2$ PES. This modification was not necessary for the less expensive calculations on the 2D C_6H^-He PES.

For low energies the new regularized potential $v_{\text{reg}}(V(R, \theta_1, \theta_2, \phi))$ remained identical to the original potential, but at a threshold energy V_a it begins to smoothly saturate along the repulsive curve until reaching a limiting value based on the parameter V_b . A function S_f was used in the saturation area to smoothly switch the original PES to a constant value.

This regularized potential allows the Legendre expansion to be performed with computational ease and numerical accuracy and is given by

$$v_{\text{reg}}(V) = \begin{cases} V, & V \leq V_a, \\ V_a + (V_b - V_a)S_f\left(\frac{V - V_a}{V_b - V_a}\right), & V < V_b, \\ V_a + (V_b - V_a)\left(\frac{2}{\pi}\right)^2, & V \geq V_b, \end{cases} \quad (3)$$

where the switching function is defined as

$$S_f(u) = \left(\frac{2}{\pi}\right)^2 \sin\left[\frac{\pi}{2} \sin\left(\frac{\pi u}{2}\right)\right]. \quad (4)$$

The threshold energy chosen for V_a was 300 cm^{-1} , while $V_b = 2000 \text{ cm}^{-1}$. These values yield a regularized potential that is extremely accurate up to 500 cm^{-1} , wherein it only diverges from the original PES by $\sim 4.0 \text{ cm}^{-1}$. The value for V_b prompts the PES to terminate at $\sim 990 \text{ cm}^{-1}$. A comparison between the original PES and the regularized PES is shown in Figures 5 and 6 for the θ_1 fixed values of 0° and 90° .

One can see in these figures that the regularized PES smoothly saturates therefore eliminating the cusp that is problematic for the lambda terms. The regularization procedure simply amounts to applying a slightly lower ceiling to the PES and with a smoother onset. The ceiling of the regularized potential is sufficiently high enough in energy for our scattering study here, but it should be noted that this modified PES should not be used for high collision energies.

Collisional cross sections were calculated using the quantum non-reactive molecular scattering code MOLSCAT.³⁵ The interaction potential was expanded over the complete set of functions of the angular coordinate. In the case of $\text{C}_6\text{H}^- - \text{H}_2$,

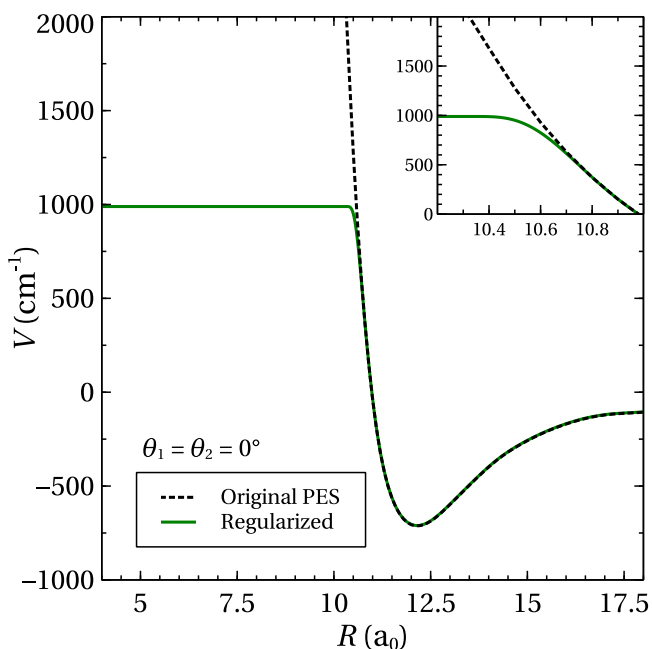


FIG. 5. The original $\text{C}_6\text{H}^- - \text{H}_2$ PES (dashed) and the regularized PES (solid) for $\theta_1 = \theta_2 = 0^\circ$.

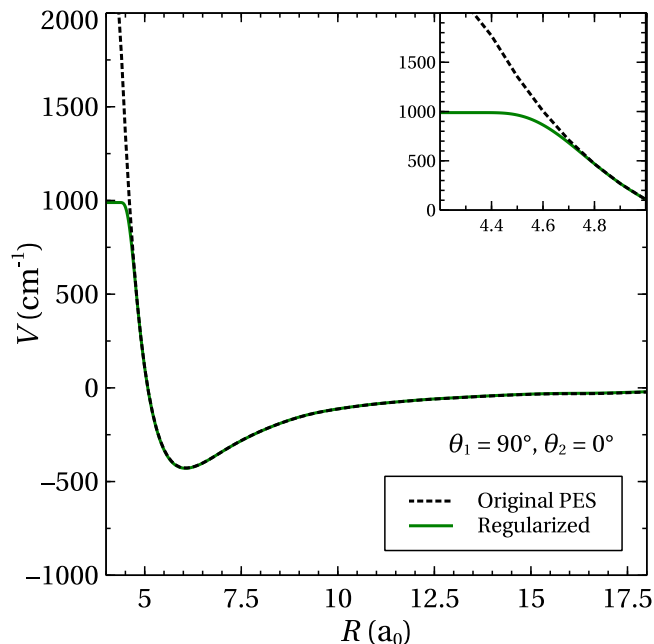


FIG. 6. The original $\text{C}_6\text{H}^- - \text{H}_2$ PES (dashed) and the regularized PES (solid) for $\theta_1 = 90^\circ, \theta_2 = 0^\circ$.

the PES was expanded as follows:

$$V(R, \theta_1, \theta_2, \phi) = \sum_{l_1, l_2, l} v_{l_1, l_2, l}(R) A_{l_1, l_2, l}(\theta_1, \theta_2, \phi), \quad (5)$$

where $A_{l_1, l_2, l}$ are contracted normalized spherical harmonics.³⁶ The angular dependence of C_6H^- was expanded up to order $l_{1, \text{max}} = 48$, while H_2 was expanded up to order $l_{2, \text{max}} = 4$.

The $\text{C}_6\text{H}^- - \text{He}$ PES was expanded according to

$$V(R, \theta) = \sum_{l_1=0}^{l_{1, \text{max}}} v_{l_1}(R) P_{l_1}(\cos \theta), \quad (6)$$

where P_{l_1} are the Legendre polynomials. Radial coefficients up to order $l_1 = 50$ were considered in the expansion.

For collisions between two linear rigid rotators, the fully quantal close-coupling (CC) approach of Green³⁷ was used to determine the integral cross sections. The log-derivative propagator of Manolopoulos³⁸ was used to solve the coupled-channel equations from 4 to $40 a_0$. The rotational basis included all open channels and several closed channels to secure convergence of the cross sections to within 10% compared to fully converged calculations for a given PES. For H_2 , only the $j_2 = 0$ and $j_2 = 1$ levels were included in the basis set; the additional inclusion of $j_2 = 2$ and $j_2 = 3$ for the *para*- and *ortho*- calculations, respectively, led to computed cross sections within 5%, of the previous more affordable result.

For the $\text{C}_6\text{H}^- - \text{He}$ system, the CC approach for the scattering of a rigid rotor and an atom of Arthurs and Dalgarno³⁹ was used to determine the cross sections. The same log-derivative propagator of Manolopoulos³⁸ was used to solve the coupled-channel equations from 4.8 to $50 a_0$. The resulting cross sections are converged to within 10%.

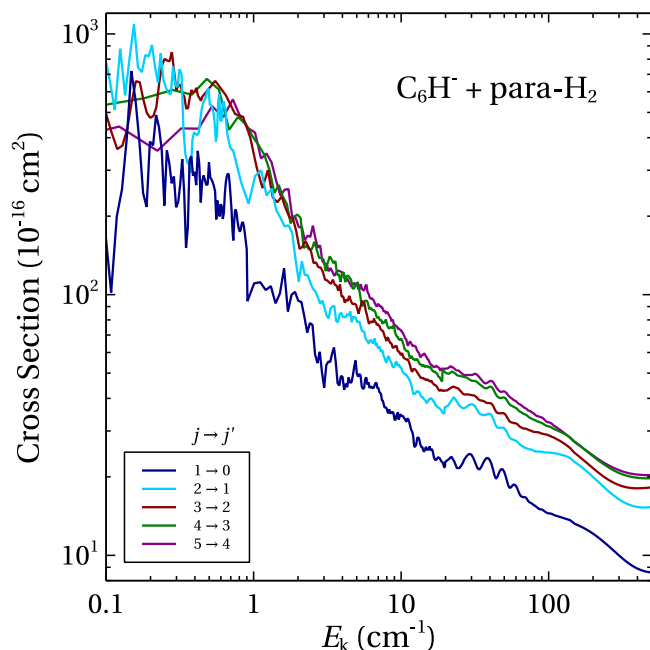


FIG. 7. The largest state-to-state de-excitation cross sections for C_6H^- in collisions with *para*- H_2 ($j_2=0$).

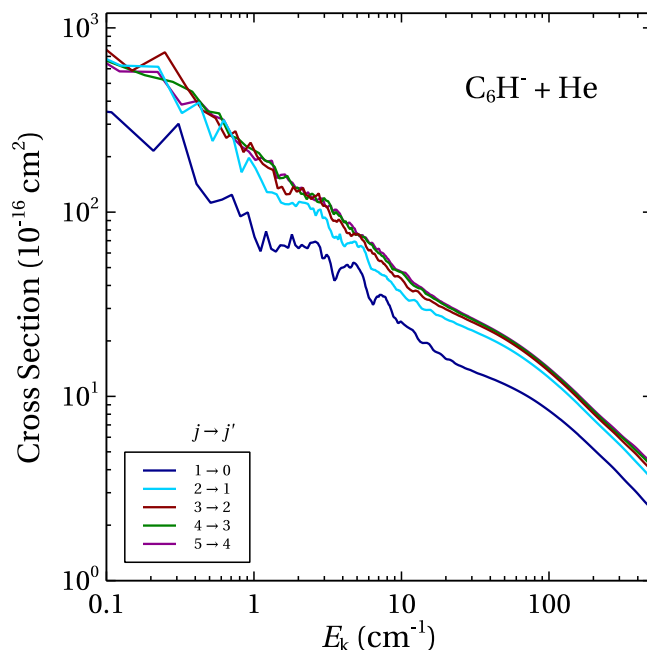


FIG. 9. The largest state-to-state de-excitation cross sections for C_6H^- in collisions with He.

B. Results

De-excitation cross sections for $\Delta j_1 = -1$ are presented in Figures 7 and 8 for collisions of C_6H^- with *para*- H_2 and *ortho*- H_2 , respectively, while de-excitation cross sections for collisions with He are shown in Figure 9.

The *ortho*- and *para*- H_2 cross sections are similar in magnitude and in fact there is no significant difference between the two moieties. The energy-dependent de-excitation cross sections for C_6H^- in collision with *ortho*- H_2 ($j_2 = 1$) appear to

have a smoother energy dependence than the cross section for collision with *para*- H_2 ($j_2 = 0$). This is a result of the fact that there are many more, and hence overlapping, resonances for *ortho*- H_2 than for *para*- H_2 . Hence, the resonance features are mostly washed out for *ortho*- H_2 . The $1/v$ dependence of the state-to-state cross sections can be clearly seen as the values decrease with increasing kinetic energy. Therefore, one can expect a slow temperature variation of the rate coefficients for these low temperature collisions in agreement with Langevin theory for ion-neutral interactions.

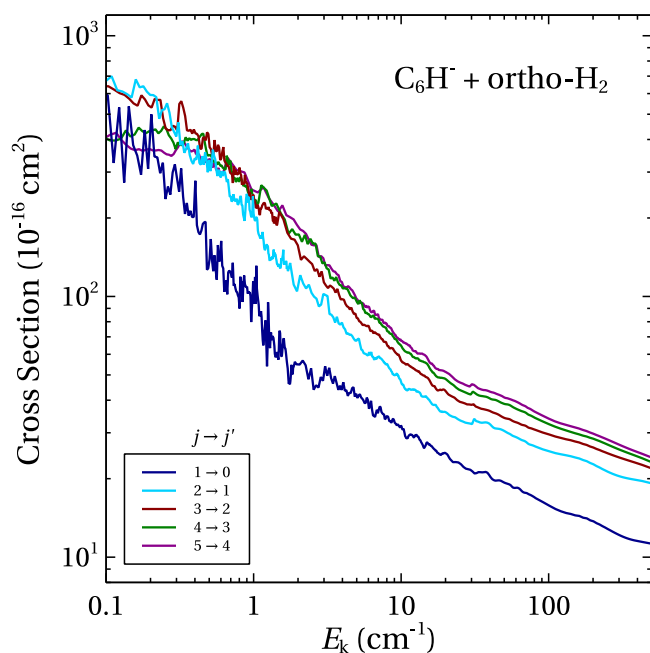


FIG. 8. The largest state-to-state de-excitation cross sections for C_6H^- in collisions with *ortho*- H_2 ($j_2=1$).

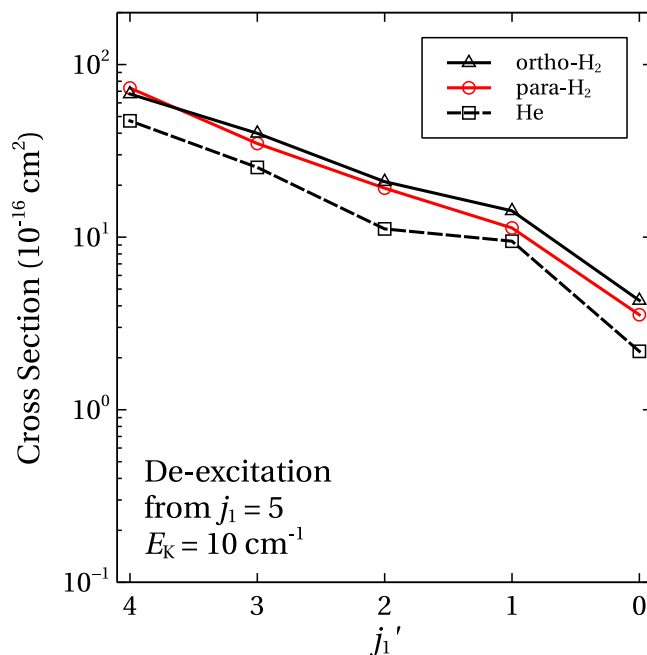


FIG. 10. Propensities for the de-excitation of C_6H^- from initial level $j_1=5$ in collisions with H_2 and He for $E_K=10\text{ cm}^{-1}$.

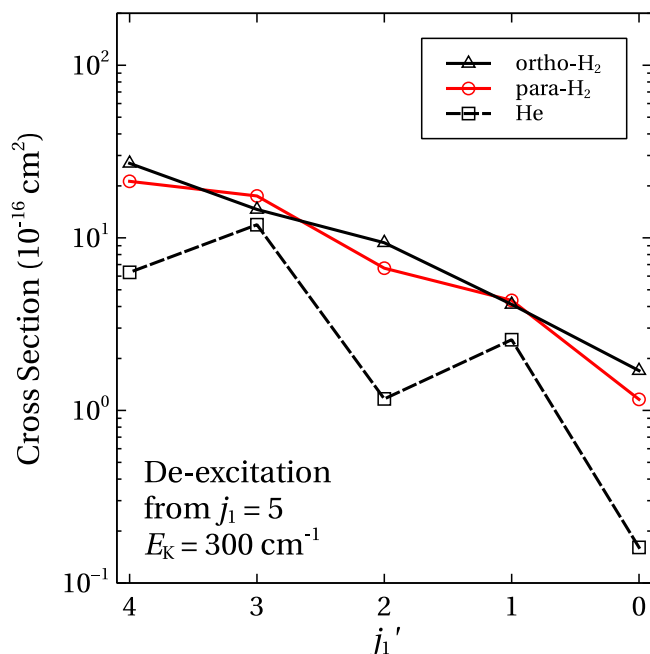


FIG. 11. Propensities for the de-excitation of C_6H^- from initial level $j_1 = 5$ in collisions with H_2 and He for $E_K = 300 \text{ cm}^{-1}$.

The C_6H^- -He cross sections have a similar magnitude to those of H_2 . The He resonances disappear after $20\text{--}30 \text{ cm}^{-1}$, however, while the H_2 resonances sustain until $\sim 80 \text{ cm}^{-1}$.

The propensities for the de-excitation of C_6H^- out of initial level $j_1 = 5$ in collisions with H_2 and He are shown for 10 and 300 cm^{-1} in Figures 10 and 11, respectively. These plots confirm that both the *ortho* and *para*- H_2 data exhibit similar values for the state-to-state cross sections. Similar trends are generally found for ionic systems (HCO^+ , N_2H^+ , CN^- , ...). This is because the inelastic process is mostly governed by long range interactions that are weakly anisotropic with respect to H_2 rotation. The plots also show that the magnitude of the cross sections decrease with increasing Δj_1 . Note that for the C_6H^- , the decrease is quite slow because of the small energy spacing between the rotational states and because of the large well depth of in the PESs that will significantly couple states with large Δj_1 . At low collision energies, the C_6H^- -He cross sections slowly decrease with increasing Δj_1 as well. However, at high collision energies, a propensity rules in favor of transitions with even Δj_1 due to the near-homonuclear symmetry of the potential energy surface.

Finally, it is interesting to compare the new C_6H^- - H_2 results with the CN^- - H_2 one of Kłos and Lique.¹⁰ At low collision energies, both C_6H^- and CN^- exhibit similar behavior with the largest cross section being the $\Delta j = -1$ transition.

IV. CONCLUSIONS

We present the first PESs for the C_6H^- - H_2 and C_6H^- -He systems. Fortran codes of the PESs are available upon request. Both *ab initio* data were computed at the CCSD(T*)-F12b level of theory. 4D and 2D PESs, respectively, were constructed using the IMLS method. Both surfaces similarly

exhibit large anisotropies for linear versus T-shape approaches and have deep van der Waals wells. We furthermore performed inelastic scattering calculations on these surfaces in order to describe the collisions of C_6H^- with *para*- H_2 , *ortho*- H_2 , and He for low internal excitations of C_6H^- . The H_2 and He state-to-state de-excitation cross sections are similar in magnitude, although the He cross sections are slightly smaller. At high collisional energies, the C_6H^- -He system exhibits propensity rules in favor of even Δj_1 transitions contrarily to the C_6H^- - H_2 system. For odd Δj_1 transitions, the difference between He and H_2 results can be up to an order of magnitude. Such behavior prevents from using He rate coefficients as a template for the H_2 ones. Collisional excitation data with H_2 are especially important for modeling astrophysical environments containing anions, such as TMC-1 or IRC +10216, and determining the physical and chemical conditions in the regions. In the future, this PES can be used to obtain a complete set of collisional rate coefficients, including highly excited rotational states, for the C_6H^- - H_2 and C_6H^- -He systems.

ACKNOWLEDGMENTS

This research was supported by the French National Research Agency (ANR) through a grant to the Anion Cos Chem Project (No. ANR-14-CE33-0013), the CNRS-INSU Programme National de Physique et Chimie du Milieu Interstellaire, and the National Science Foundation (Grant No. CHE-1300945 to R.D.). This work was granted access to the HPC resources of IDRIS under the Allocation No. 2016-047659 made by GENCI (Grand Equipement National de Calcul Intensif).

- ¹M. C. McCarthy, C. A. Gottlieb, H. Gupta, and P. Thaddeus, *Astrophys. J., Lett.* **652**, L141 (2006).
- ²J. Cernicharo, M. Guélin, M. Agúndez, K. Kawaguchi, M. McCarthy, and P. Thaddeus, *Astron. Astrophys.* **467**, L37 (2007).
- ³S. Brünken, H. Gupta, C. A. Gottlieb, M. C. McCarthy, and P. Thaddeus, *Astrophys. J., Lett.* **664**, L43 (2007).
- ⁴M. Agúndez, J. Cernicharo, M. Guélin, C. Kahane, E. Roueff, J. Kłos, F. J. Aoiz, F. Lique, N. Marcelino, J. R. Goicoechea *et al.*, *Astron. Astrophys.* **517**, L2 (2010).
- ⁵P. Thaddeus, C. A. Gottlieb, H. Gupta, S. Brünken, M. C. McCarthy, M. Agúndez, M. Guélin, and J. Cernicharo, *Astrophys. J.* **677**, 1132–1139 (2008).
- ⁶J. Cernicharo, M. Guélin, M. Agúndez, M. C. McCarthy, and P. Thaddeus, *Astrophys. J.* **688**, L83 (2008).
- ⁷V. Wakelam, E. Herbst, J.-C. Loison, I. W. M. Smith, V. Chandrasekaran, B. Pavone, N. G. Adams, M.-C. Bacchus-Montabonel, A. Bergeat, K. Béroff *et al.*, *Astrophys. J., Suppl. Ser.* **199**, 21 (2012).
- ⁸F. Dumouchel, A. Spielfiedel, M. L. Senent, and N. Feautrier, *Chem. Phys. Lett.* **533**, 6 (2012).
- ⁹D. Hauser, S. Lee, F. Carelli, S. Spieler, O. Lakhmanskaya, E. S. Endres, S. S. Kumar, F. Gianturco, and R. Wester, *Nat. Phys.* **11**, 467 (2015).
- ¹⁰J. Kłos and F. Lique, *Mon. Not. R. Astron. Soc.* **418**, 271 (2011).
- ¹¹N. Sakai, T. Sakai, Y. Osamura, and S. Yamamoto, *Astrophys. J., Lett.* **667**, L65 (2007).
- ¹²H. Gupta, C. A. Gottlieb, M. C. McCarthy, and P. Thaddeus, *Astrophys. J.* **691**, 1494 (2009).
- ¹³M. A. Cordiner, S. B. Charnley, J. V. Buckle, C. Walsh, and T. J. Millar, *Astrophys. J., Lett.* **730**, L18 (2011).
- ¹⁴M. Wernli, L. Wiesenfeld, A. Faure, and P. Valiron, *Astron. Astrophys.* **464**, 1147 (2007); e-print [arXiv:physics/0611258](https://arxiv.org/abs/physics/0611258).
- ¹⁵T. B. Adler, G. Knizia, and H.-J. Werner, *J. Chem. Phys.* **127**, 221106 (2007).
- ¹⁶J. G. Hill, S. Mazumder, and K. A. Peterson, *J. Chem. Phys.* **132**, 054108 (2010).
- ¹⁷H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, and M. Schütz, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.* **2**, 242 (2012).

- ¹⁸F. Lique, J. Klos, and M. Hochlaf, *Phys. Chem. Chem. Phys.* **12**, 15672 (2010).
- ¹⁹R. Dawes, X.-G. Wang, A. W. Jasper, and T. Carrington, Jr., *J. Chem. Phys.* **133**, 134304 (2010).
- ²⁰R. Dawes, X.-G. Wang, and T. Carrington, Jr., *J. Phys. Chem. A* **117**, 7612 (2013).
- ²¹K. B. Gubbels, S. Y. van de Meerakker, G. C. Groenenboom, G. Meijer, and A. van der Avoird, *J. Chem. Phys.* **136**, 074301 (2012).
- ²²R. A. Kendall, T. H. Dunning, Jr., and R. J. Harrison, *J. Chem. Phys.* **96**, 6796 (1992).
- ²³F. Weigend and R. Ahlrichs, *Phys. Chem. Chem. Phys.* **7**, 3297 (2005).
- ²⁴R. Dawes, A. F. Wagner, and D. L. Thompson, *J. Phys. Chem. A* **113**, 4709 (2009).
- ²⁵M. Majumder, S. A. Ndengue, and R. Dawes, *Mol. Phys.* **114**, 1 (2015).
- ²⁶J. Brown, X.-G. Wang, T. Carrington, Jr., G. Grubbs II, and R. Dawes, *J. Chem. Phys.* **140**, 114303 (2014).
- ²⁷X.-G. Wang, T. Carrington, R. Dawes, and A. W. Jasper, *J. Mol. Spectrosc.* **268**, 53 (2011).
- ²⁸J. Brown, X.-G. Wang, R. Dawes, and T. Carrington, Jr., *J. Chem. Phys.* **136**, 134306 (2012).
- ²⁹S. A. Ndengué, R. Dawes, and F. Gatti, *J. Phys. Chem. A* **119**, 7712 (2015).
- ³⁰M. Majumder, S. E. Hegger, R. Dawes, S. Manzhos, X.-G. Wang, C. Tucker, Jr., J. Li, and H. Guo, *Mol. Phys.* **113**, 1823 (2015).
- ³¹I. M. Sobol, *USSR Comput. Math. Math. Phys.* **16**, 236 (1976).
- ³²U. Fink, T. A. Wiggins, and D. H. Rank, *J. Mol. Spectrosc.* **18**, 384 (1965).
- ³³J. V. Foltz, D. H. Rank, and T. A. Wiggins, *J. Mol. Spectrosc.* **21**, 203 (1966).
- ³⁴S. Chapman and S. Green, *J. Chem. Phys.* **67**, 2317 (1977).
- ³⁵J. M. Hutson and S. Green, Distributed by Collaborative Computational Project No. 6, Swindon, UK: Engineering and Physics Science Resource Council, 1994.
- ³⁶H. Rabitz, *J. Chem. Phys.* **57**, 1718 (1972).
- ³⁷S. Green, *J. Chem. Phys.* **62**, 2271 (1975).
- ³⁸D. E. Manolopoulos, *J. Chem. Phys.* **85**, 6425 (1986).
- ³⁹A. M. Arthurs and A. Dalgarno, *Proc. R. Soc. London A* **256**, 540 (1960).