

## Study of the Recycle and Reuse of Domestic Wastes

A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of M.Sc. in Industrial Chemistry

by:

#### Aeisha Adam Osman Abdelkreem

B.Sc. (Industrial Chemistry) U.J

#### **Supervisor:**

Prof. Babiker Karama Abdalla

September 2016

## **Dedication**

This thesis is dedicated to my parentsat firstandmy family, whom there were motivated me all the time to see me successfuland prepotent, bysupporting and encouraged me invariably, Thanks a lot.

# Acknowledgement

First and foremost, I thank God for the numerous blessings he has bestowed upon me throughout. Thanks to my supervisor Prof/BabikerKramaAbdalla, for his advices and support. Thankful extended to Industrial researches and consultancy center (IRCC) / ahead of laboratories Mr. MdohamedHussein Ali andSalma Abdelkhalig Thanksalot for my friend MohammedGummaElshafie for helping and encourage me.

#### **Abstract**

Due to the modern trend in the field of bio-energy and replace it as an alternative to non-renewable energy, this study comes to illuminate light on biofuel production process from the waste cooking oil and reducing the harmful impact by protecting the environment from emission resulting comes from the production and use of diesel oil such as  $(CO_2, CO, H_2S)$ , in addition to the added value of the oil. Ease of production and cheap raw materials.

It produced by using trans-esterification process by using methanol mixed with Potassium hydroxide as catalyst, and then the mixture added to the oil in the magnetic stirring for two hour at 60 °C. It was separated in separator funnel by gravity. Lastly, bio-fuel was collected from the top layer and washed with hot water to give high purity. The glycerin from the bottom as by-product.

Laboratory analysis for biodiesel was obtained some results, the flash point is 115 C° and this useful for storage and transportation without cooling or holding in specific containers and this reversal for flash point of diesel oil which it has flash point 60C° lowers, it has viscosity 6 cp and density 0.88 g/cm<sup>3</sup>, in addition to the little amount of sulphated ash content which ensure the safety engine for a long time.

#### الخلاصة

نظرآ للتوجهات الحديثة في مجال الطاقة الحيوية المتجددة وإحلالها كبديل للطاقة غير المتجددة تأتي هده الدراسة لتنير الضوء على عملية إنتاج الوقود الحيوي من الزيت الراجع من الطعام لما له من أثر في حماية البيئة من إنبعاث الغازات الناتجة من عملية إنتاج وإستخدام الديزل البترولي مثل ( CO,CO<sub>2</sub>, H<sub>2</sub>S ) بالإضافة الي القيمة المضافة للزيت الراجع ، كما يفضل إستخدامه كبديل للديزل البترولي لسهولة إنتاجه ومن مواد خام مهملة وغير مكلفة لضررها البالغ بالبيئة.

تم إستخدام عملية الأسترة (Trans-esterification Process) في إنتاج الوقود الحيوي باستخدام الميثانول المخلوط مع هيدروكسيد البوتاسيوم المستخدم كعامل حفاز ثم يضاف الخليط إلى الزيت في الخلاط المغنطيسي عند درجة حرارة ٦٠ م لمدة ساعتين وبعد ذلك يتم الفصل في القمع الزجاجي لنتحصل على الوقود الحيوي في الجزء الاعلى والجلسرين في الاسفل كناتج ثانوي كما أن الفصل يعتمد على الجاذبية ، للحصول على نقاوة عالية من الديزل االحيوي يتم غسله بالماء الدافئ على عدة مراحل حتى نتحصل على طبقة نقية من الماء في الاسفل.

حللت عينات في المختبر و تم الحصول علي نتائج منها درجة اشتعال عالية تساوي (  $^{\circ}$  (  $^{\circ}$  )  $^{\circ}$  وهدايجعل ترحيله و تخزينه سهل من غير تبريد على عكس الديزل البترولي لان درجة حرارة اشتعاله قليلة ( $^{\circ}$   $^{\circ}$  ، كما وجدت درجة لزوجته ( $^{\circ}$ ) س ب ، وكثافته ( $^{\circ}$   $^{\circ}$   $^{\circ}$  م ، كما وجدت درجة لزوجته ( $^{\circ}$ ) س ب ، وكثافته ( $^{\circ}$   $^{\circ}$   $^{\circ}$  م ، كما وجدت درجة لزوجته ( $^{\circ}$ ) سيطة جدا مما يضمن سلامة المحرك لفترة اطول.

### **List of Contents**

Title	Page No
Dedication	I
Acknowledgment	II
English Abstract	III
Arabic Abstract	IV
List of Content	V
List of Table	VII
List of Figure	VIII
Chapter One: Introduction	
1.1 Introduction	1
1.2 Objective	5
Chapter One: Literature review	
2-1 Background	7
2-2 Classification of waste	7
2-3 Contamination from Agricultural Wastes	7
2-4Contamination from Industrial Wastes	9
2-5 Remediation and Rehabilitation of Soils Contaminated by	9
Heavy Metals	
2-6 Liquid Waste	11
2-6-1 Used Cooking Oil	11
2-6-2 Environmental problems for disposing used cooking oil	12
2-6-3The Different Techniques to Produce Biodiesel	13
2-6-4 Biodiesels	13
2-6-4-1Safety for Trans-esterification process	15
2-6-4-2 Physical Properties of Biodiesel	16
2-6-4-3 Six Factors Affecting Biodiesel Production	17
2-6-4-4 Storage Stability	17
2-6-4-5 Flash point	18
2-6-4-6Density	18
Chapter Three: Materials and Methods	
3-1 Collection of oil materials	21
3-2 Preparation of the raw material	21
3-3 Processing in Biodiesel	21
3-3-1 Esterification	21

3-3-2 Separation 1		
3-3-3 Mixing of Alcohol and Catalyst	22	
3-3-4Transesterification	22	
3-3-5 Separation 2	23	
3-3-6 Alcohol Removal	23	
3-3-7 Washing	24	
3.3.8 Total Biodiesel Produced Percentage	24	
3-3-9 Reagent used	25	
3-4 Chemical Properties Tests	25	
3-4-1 Acid value	25	
3-4-2 Peroxide value	26	
3-4-3Iodine value	26	
3-4-4 Saponification value	27	
3-4-5 Moisture content	28	
3-4-6 Sulfured ash	29	
3-5 Physical Properties Tests	30	
3-5-1Flash Point	30	
3-5-2 Distillation of Biodiesel Products at Reduced Pressure1	31	
3-5-3 Density	32	
3-5-4 API Gravity (American Petroleum Institute) Equation	33	
3-5-5 Calculated Cetane Index by Four Variable Equation	33	
3-5-6 Calculated Cetane Index by Iodine and Saponification	34	
Values		
Chapter Four: Results and Discussion		
4.1 Results	36	
4.2 Discussion:	39	
ChapterFive: Conclusion and Recommendations		
4.3 Conclusion	42	
4.4Recommendations	43	
References	44	

## **List of Tables**

Table No	Title	Page No
1	result analysis for Oil tests	36
2	result analysis for Biodiesel tests	37

## **List of Figures**

Figure No	Title	Page No
1	The Chemical Reaction of Biodiesel Production	14
	by Transesterification	
2	Stages of the Transesterification Reaction to	15
	Producing the Biodiesel	
3	Waste cooking oil	51
4	Biodiesel and Glycerin comes	51
	fromTransesterification Process in separation	
	form	
5	Biodiesel in first step of washing	51
6	Biodiesel in last step of washing	51
7	In the left Biodiesel after washing, and in the	52
	right Biodiesel and Glycerin before separation	
8	Biodiesel in pure form	52
9	IR OF Biodiesel	38