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## PROCESSING AND PROPERTIES OF BORON CARBIDE WITH HAFNIUM DIBORIDE ADDITION

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*This article presents the results of investigations on densification, mechanical and electrical properties of boron carbide ( $B_4C$ ) with the addition of  $HfB_2$ . High dense  $B_4C$ - $HfB_2$  (2.5 - 30 wt. %) composites were prepared by hot pressing at a temperature of 2173 K with 40 MPa mechanical pressure. The  $B_4C$ - $HfB_2$  composite mixture exhibited a better sintering aptitude compared with monolithic  $B_4C$ . Hardness and elastic modulus of  $B_4C$ - $HfB_2$  composites were measured to be in the range 36 - 28 GPa and 465 - 525 GPa respectively. Indentation fracture toughness of  $B_4C$  increased with  $HfB_2$  content and obtained a maximum of  $7 \text{ MPa m}^{1/2}$  at 30 wt. %  $HfB_2$ , which is  $\sim 3$  times higher than the monolithic  $B_4C$ . Crack deflection was identified to be the major toughening mechanism in the developed composite.  $B_4C$  - 10 wt. %  $HfB_2$  composite exhibited a maximum electrical conductivity of  $7144 \Omega^{-1} \text{ m}^{-1}$  which is 26 % higher than the conductivity of monolithic  $B_4C$  ( $5639 \Omega^{-1} \text{ m}^{-1}$ ) at 1373 K.*

### INTRODUCTION

Boron carbide is a promising candidate for many high performance applications in nuclear and defence sectors because of its unique characteristics. A combination of low density ( $2.51 \text{ g}\cdot\text{cm}^{-3}$ ), high elastic modulus (460 GPa) and high hardness (38 GPa) enables  $B_4C$  to find application in defence sector as an armour material.  $^{10}\text{B}$  isotope of  $B_4C$  offers significant neutron absorption cross-section for both thermal and fast neutrons and thus plays a major role in nuclear industries as neutron detectors, control rods and shielding materials [1, 2].

Refractory nature of  $B_4C$  necessitates a high sintering temperature close to 2273 K which makes densification difficult and often deteriorates mechanical properties of the material due to the grain coarsening effects. Also, the extreme brittleness/poor fracture toughness of  $B_4C$  limits the wide spread application of the material [1, 3-6].

Different sintering methods that assists the densification of high temperature ceramics are the use of submicron particles, binder additions and adoption of advanced sintering process like spark plasma sintering (SPS) [1, 2, 7-18]. Most of the research work has been devoted towards binder based additions as it is reported to enhance both densification and mechanical properties of  $B_4C$  [7-17]. Sintering study by Skorokhod et al. has brought into focus the approach to explore different oxide based binder additions to lower the sintering temperatures and increasing the strength of  $B_4C$  [13]. It is

reported that the chemical instability nature of  $B_4C$  with respect to oxide binders was favouring the formation of high dense  $B_4C$ -boride composite at comparatively lower sintering temperatures than that of monolithic  $B_4C$  [10, 19-22]. Till date, the ease of in-situ processing methodology overshadowed and limited the research on the densification of  $B_4C$  using pre-synthesized boride powders. In this study, hafnium diboride ( $HfB_2$ ) is chosen as a ceramic binder, as it possesses attractive properties like high melting point, high hardness, high elastic modulus and neutron absorption cross-section [23, 24]. Limited reports were available on the densification of  $B_4C$  with pre-synthesized diboride powders ( $TiB_2$ ,  $ZrB_2$  and  $CrB_2$ ) as ceramic additives [17, 25-27].

The present paper gives the results of investigations carried out on the effect of  $HfB_2$  addition on the densification and properties of  $B_4C$  by hot-pressing method.

### EXPERIMENTAL

Boron carbide ( $B_4C$ ; 78.5 % B, 19.5 % C, < 1 % O, 0.02 % Fe, 0.02 % Si; 5.3  $\mu\text{m}$  mean particle diameter; Boron Carbide India Pvt Ltd) and in-house synthesized [28] Hafnium diboride ( $HfB_2$ ; purity: 99 %, 0.5 %  $O_2$ , 0.4 % C; 3.1  $\mu\text{m}$  mean particle diameter) were used as the starting materials. 0 to 30 wt. % of  $HfB_2$  was added to  $B_4C$  and the powder mixtures were allowed to mix homogeneously in a motorized mortar and pestle

(Pulverisette 2, Fritsch, Germany) for 2 h. The homogeneously mixed  $B_4C$ - $HfB_2$  composite powder was loaded into the 17 mm dia. graphite die and further the graphite die assembly was kept inside the hot-pressing chamber. The chamber was evacuated to  $10^{-5}$  mbar vacuum level. Hot-pressing was carried out under dynamic vacuum condition at temperature of 2173 K for 1 h duration with 40 MPa mechanical pressure. After sintering, the samples were allowed to cool down to room temperature and measured for density by liquid displacement method. The dense compacts were polished to mirror finish using a series of diamond suspensions ranging from 15 - 0.5  $\mu m$  grades (Struers, Denmark). X-ray diffraction (XRD; XRG 3000, Inel, France) was performed with  $Cu-K\alpha$  radiation to identify the crystalline phases. Selected samples were characterized for microstructures using scanning electron microscopy (SEM; MB 2300 CT/100, CAMSCAN, UK) with simultaneous elemental analysis by energy dispersive spectrometer (EDS; X-MAX 80, OXFORD, UK). Vickers hardness and fracture toughness were measured based on indentation technique with a load of 1.961 N and 9.807 N respectively and for dwell time of 15 s. Anstis methodology [29] was adopted for indentation fracture toughness calculation. Elastic modulus was determined using ultrasonic wave velocity technique (UT 340 pulser receiver system, UTEX Scientific Instrument Inc., Canada). Electrical resistivity of the developed composites was measured by the four probe method at temperatures between 298 K and 1373 K.

## RESULTS AND DISCUSSION

### Densification and microstructures

Table 1 shows the sinter density and mechanical property results of  $B_4C$  and  $B_4C$ - $HfB_2$  composites processed at 2173 K. Sinter density of monolithic  $B_4C$  was measured as 96 %  $\rho_{th}$ . On addition of 2.5 wt. % of  $HfB_2$  to  $B_4C$ , the sinter density of the compact increases and reaches a value of 99 %  $\rho_{th}$ . Sinter densities of composites with higher content of  $HfB_2$  remained same as 99 % (Figure 1). It has to be noted that around 3 % improvement in sinter density was realized on mere

addition of 2.5 wt. %  $HfB_2$  to  $B_4C$  and it remained substantially unaltered with respect to increased weight fraction of  $HfB_2$  content. Zrozi et al. have reported that with the addition of 4 wt. %  $TiB_2$  to  $B_4C$ , no increase in density was observed in  $B_4C$ - $TiB_2$  composite under similar processing conditions [30]. Wenbo et al. observed that the sinter density of  $B_4C$  increased on addition of  $ZrB_2$  up to 30 vol. %, further increase of  $ZrB_2$  addition resulted in slight reduction in the sinter density (97.8 %) of the compacts [26]. On the other hand, Baharvandi et al. reported about the increase in densification of  $B_4C$  on  $TiB_2$  addition and mentioned  $TiB_2$ 's role as a grain growth inhibitor [31]. Fully dense  $B_4C$ - $HfB_2$  composite was fabricated by in-situ processing of  $B_4C$  with  $HfO_2$  as additive at 2173 K under 40 MPa mechanical pressure [21]; whereas in the present study, with  $HfB_2$  addition, the highest density achieved was  $\sim 99$  %  $\rho_{th}$ . Goldstein et al. has made similar observations while processing  $B_4C$ - $ZrB_2$  composite by an in-situ method [17].

X-ray diffraction pattern of sintered composites indicates the presence of  $B_4C$  and  $HfB_2$  phases only (Figure 2), which emphasize the high temperature phase stability of  $HfB_2$  with respect to  $B_4C$ . Figure 3 shows the scanning electron micrograph of 99 % dense  $B_4C$ - $HfB_2$  composite that indicates random distribution of bright phase particles in the dark matrix. Bright and dark phases were analysed to contain (a) hafnium (Hf) and

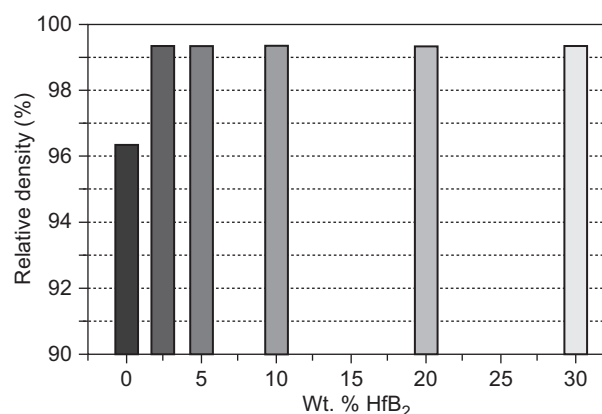


Figure 1. Relative densities of  $B_4C$  and  $B_4C$ - $HfB_2$  ceramics that was hot pressed at 2173 K under 40 MPa mechanical pressure.

Table 1. The sinter density and mechanical property results of  $B_4C$  and  $B_4C$ - $HfB_2$  hot pressed at 2173 K under 40 MPa mechanical pressure.

Sample composition	Theoretical density ( $g\cdot cm^{-3}$ )	Relative density (%)	Hardness (GPa)	Elastic modulus (GPa)	Fracture toughness ( $MPa\cdot m^{1/2}$ )
$B_4C$	2.52	96	$36 \pm 0.6$	$465 \pm 7$	$2.3 \pm 1.2$
$B_4C$ -2.5 wt. % $HfB_2$	2.56	99	$35 \pm 0.3$	$504 \pm 5$	$3.6 \pm 1$
$B_4C$ -5.0 wt. % $HfB_2$	2.60	99	$32.6 \pm 0.5$	$514 \pm 15$	$4.4 \pm 1.2$
$B_4C$ -10 wt. % $HfB_2$	2.69	99	$30.9 \pm 0.5$	$525 \pm 20$	$5.5 \pm 0.8$
$B_4C$ -20 wt. % $HfB_2$	2.89	99	$29.5 \pm 0.3$	$502 \pm 10$	$6.6 \pm 0.8$
$B_4C$ -30 wt. % $HfB_2$	3.12	99	$28 \pm 0.5$	$484 \pm 5$	$7.1 \pm 0.8$

boron (B), and (b) boron (B) and carbon (C), respectively (Figure 4a-b). HfB<sub>2</sub> particles appear as a bright phase in the microstructure due to the larger atomic weight of Hf compared to B and C elements.

Certain advantages can be observed in sintering of B<sub>4</sub>C with the addition of HfB<sub>2</sub> as compared to HfO<sub>2</sub>. Though the in-situ processing route has proven to fabricate high dense B<sub>4</sub>C composites, it often causes deviations from its stoichiometric composition [10, 19, 21]. This can be realized based on the reported changes in lattice parameters of B<sub>4</sub>C phase in the sintered product with respect to oxide additions [10, 21]. Since both boron and carbon atoms taking part in the reduction of oxides, B/C ratio of B<sub>4</sub>C will no longer remain the same as that of initial or starting material composition. In such cases, precise control over the material chemistry is often difficult. Further, the incomplete or partial completion of chemical reactions between B<sub>4</sub>C and oxides would

also cause compositional uncertainties in the material. B<sub>4</sub>C finds many uses in nuclear industry as a neutron control rods, detectors etc. [1, 2]. The discrepancies that arise from compositional uncertainties, particularly impurities and non-stoichiometry are of primary concern for nuclear applications. Hence applications of this type, demand materials whose compositions and impurity levels are precisely known [1]. Table 2 shows the measured lattice parameters of B<sub>4</sub>C phase that remain unaffected with respect to HfB<sub>2</sub> addition. Hence,

Table 2. Lattice parameters of sintered B<sub>4</sub>C compacts as a function of HfB<sub>2</sub> addition.

Sample Composition	Lattice parameters of B <sub>4</sub> C phase	
	a (Å)	c (Å)
B <sub>4</sub> C	5.6461	12.1116
B <sub>4</sub> C-2.5 wt. % HfB <sub>2</sub>		
B <sub>4</sub> C-5.0 wt. % HfB <sub>2</sub>		
B <sub>4</sub> C-10 wt. % HfB <sub>2</sub>		
B <sub>4</sub> C-30 wt. % HfB <sub>2</sub>		

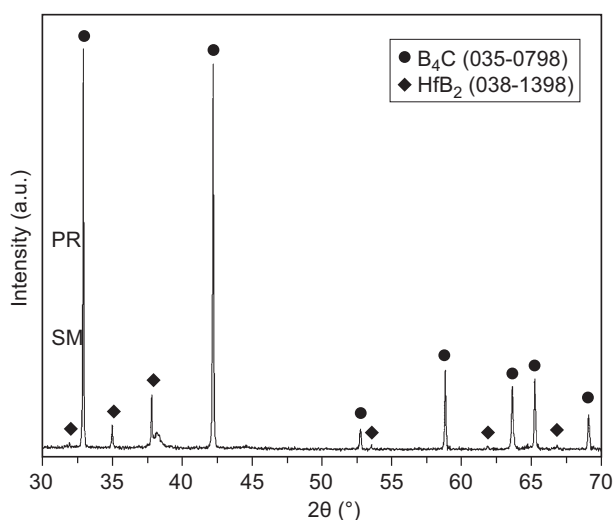


Figure 2. X-ray diffraction pattern of sintered B<sub>4</sub>C-30 wt. % HfB<sub>2</sub> composite indicating the presence of B<sub>4</sub>C and HfB<sub>2</sub> as the major phases.

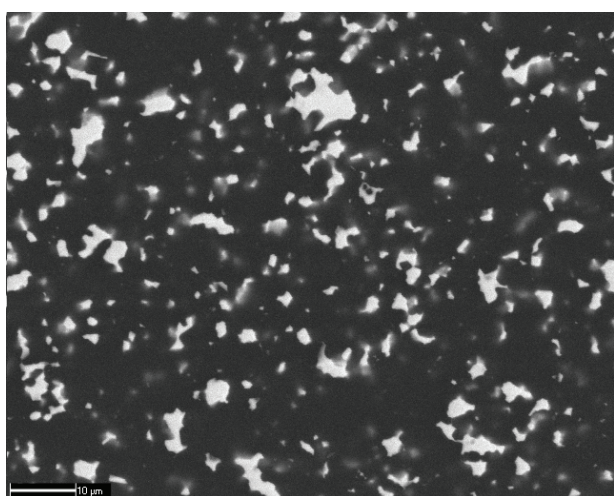


Figure 3. Electron back scattered image of 99 % dense B<sub>4</sub>C-30 wt. % HfB<sub>2</sub> composite showing the random distribution of bright phase (HfB<sub>2</sub>) in the dark matrix (B<sub>4</sub>C).

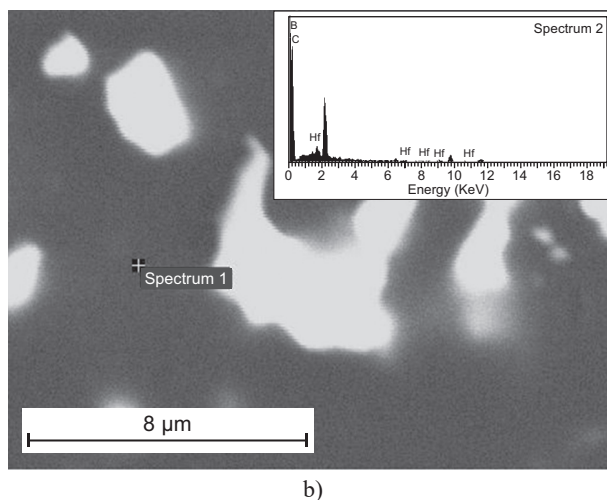
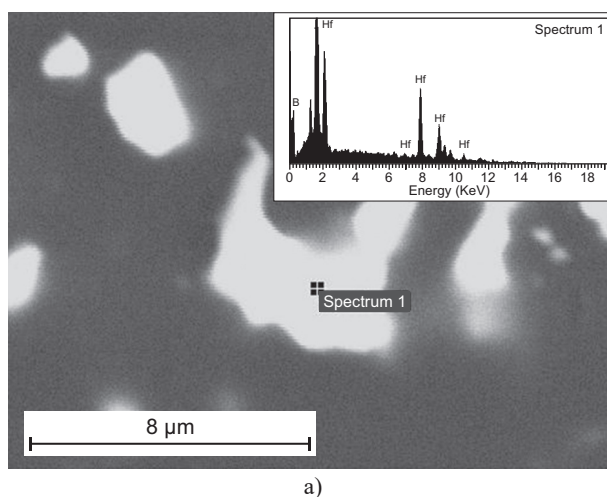


Figure 4. EDS spot analysis spectrum indicating: a) Hafnium and Boron and b) Boron and Carbon as the constituting elements in the bright and dark regions respectively.

the direct additions  $\text{HfB}_2$  could be preferred over the  $\text{HfO}_2$  addition while fabricating boron carbide-hafnium diboride composites for nuclear applications.

#### Mechanical properties

Hardness of 96 %  $\rho_{th}$   $\text{B}_4\text{C}$  was measured as 36 GPa. On addition of 2.5 wt. %  $\text{HfB}_2$  to  $\text{B}_4\text{C}$ , the hardness of the composite material reduced to 35 GPa. Further increase of  $\text{HfB}_2$  content to 30 wt. % has resulted in decreasing the hardness value to 28 GPa (Figure 5). The lower hardness of the developed  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composite was mainly due to addition of reinforcement phase ( $\text{HfB}_2$ ) which is relatively soft compared to monolithic  $\text{B}_4\text{C}$ . Elastic modulus of monolithic  $\text{B}_4\text{C}$  is measured as 465 GPa whereas the modulus of  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composites were measured to be in the range of 485 - 525 GPa. Elastic modulus increases with increasing  $\text{HfB}_2$  content up to 10 wt. %  $\text{HfB}_2$  and reaches a maximum of 525 GPa (Figure 6). Further increase of  $\text{HfB}_2$  up to 30 wt. % resulted in lowering the elastic modulus of  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composite.

Indentation fracture toughness of monolithic  $\text{B}_4\text{C}$  and  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composites were measured to be  $2.3 \text{ MPa m}^{1/2}$  and  $3 - 7 \text{ MPa m}^{1/2}$  respectively (Figure 7). Fracture toughness increases with increasing addition of  $\text{HfB}_2$  and reaches a maximum of  $\sim 7 \text{ MPa m}^{1/2}$  (for 30 wt. %  $\text{HfB}_2$ ) which is 3 times superior to that of monolithic  $\text{B}_4\text{C}$ . The observed increase in fracture toughness of the composite is mainly due to the crack tip deflections at the reinforced  $\text{HfB}_2$  particles as shown in Figure 8a-c. In addition to crack deflection, the stress intensity factor at the crack tip reduces as it experiences compressive stresses [21, 32] while traversing through  $\text{B}_4\text{C}$  matrix and thus contributes for enhancing the fracture toughness of the material. The developed  $\text{B}_4\text{C}$ -30 wt. %  $\text{HfB}_2$  composite exhibited a fracture toughness of  $7.1 \text{ MPa m}^{1/2}$  which is superior to some of the established boron carbide-boride composites such as  $\text{B}_4\text{C}$ -10 %  $\text{HfB}_2$ ,  $\text{B}_4\text{C}$ -15 vol. %  $\text{TiB}_2$ ,  $\text{B}_4\text{C}$ - $\text{TiB}_2$ -Mo and  $\text{B}_4\text{C}$ -MoSi<sub>2</sub> whose fracture toughness values were reported to be 5.24, 6.1, 4.3 and 4.8  $\text{MPa m}^{1/2}$  respectively [13, 33-35]. With improved fracture resist behaviour of  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composite as well as the inherent neutron absorption characteristics of Hf, particularly in the epithermal region, this composite can be a potential substitute for monolithic  $\text{B}_4\text{C}$  as a neutron absorbers in nuclear reactors.

#### Electrical properties

The voltage and current characteristics of  $\text{B}_4\text{C}$  and  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composite at temperatures 298 K, 773 K and 1273 K is shown in Figure 9. Figure 10 shows the electrical conductivity of  $\text{B}_4\text{C}$  and its dependence on  $\text{HfB}_2$  content at different temperatures. At 298 K, the electrical conductivity of  $\text{B}_4\text{C}$  was measured to be  $212 \Omega^{-1}\cdot\text{m}^{-1}$ .

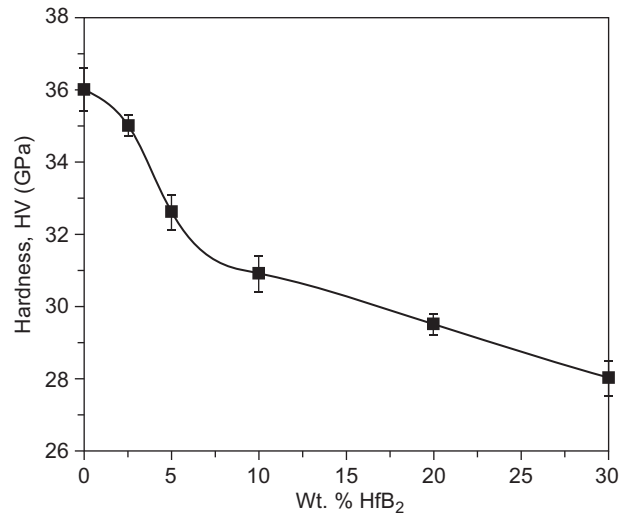


Figure 5. The effect of  $\text{HfB}_2$  addition on hardness of  $\text{B}_4\text{C}$ .

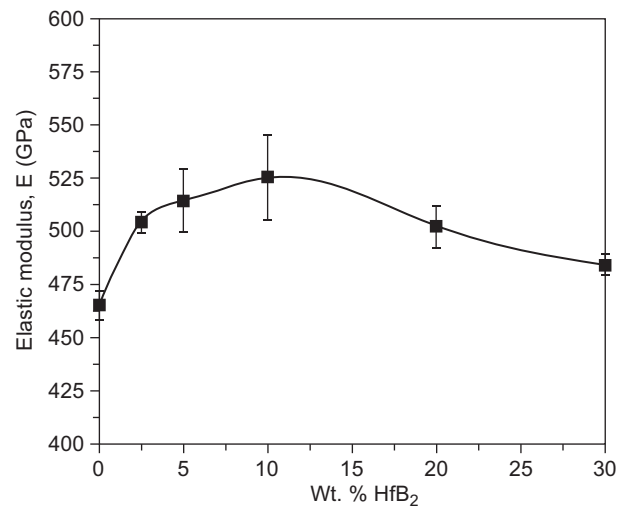


Figure 6. Variation of elastic modulus of  $\text{B}_4\text{C}$  with respect to  $\text{HfB}_2$  addition.

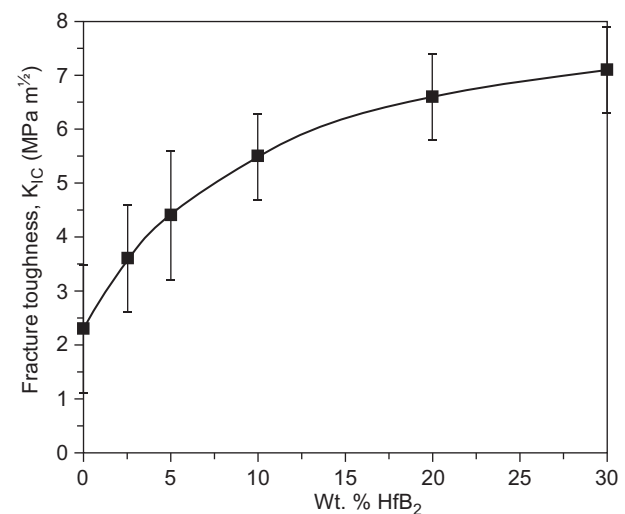


Figure 7. Effect of  $\text{HfB}_2$  addition on fracture toughness of  $\text{B}_4\text{C}$ .

With increasing temperature, the electrical conductivity was found to increase and attained  $2813 \Omega^{-1}\cdot\text{m}^{-1}$  at 773 K which is one order increase in conductivity compared to that of room temperature value. On further increasing the temperature to 1373 K, the conductivity of  $\text{B}_4\text{C}$  was

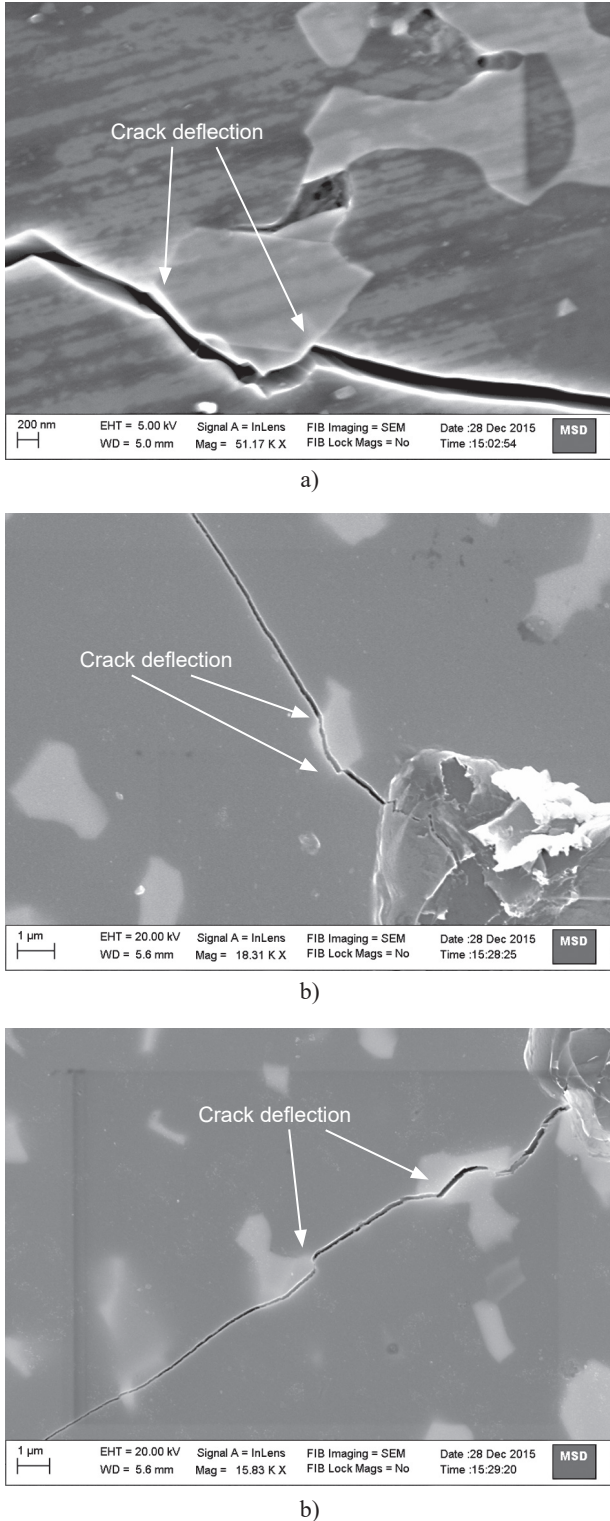


Figure 8. Microstructures of 99 % dense  $\text{B}_4\text{C}$ -30 wt. %  $\text{HfB}_2$  composite showing the deflection of crack propagation path due to the existence of second phase particle.

measured to be  $5639 \Omega^{-1}\cdot\text{m}^{-1}$  which is approximately twice the value of conductivity that was measured at 773 K. Overall, the monolithic  $\text{B}_4\text{C}$  has shown 25 times the initial conductivity value on increasing the temperature from 298 K to 1373 K. At 298 K, the electrical conductivity of  $\text{B}_4\text{C}$  sample containing 2.5 wt. %  $\text{HfB}_2$  was measured as  $340 \Omega^{-1}\cdot\text{m}^{-1}$  which is 60 % higher than the conductivity of monolithic  $\text{B}_4\text{C}$ . It was observed the reinforcement of  $\text{HfB}_2$  to  $\text{B}_4\text{C}$  matrix enhances

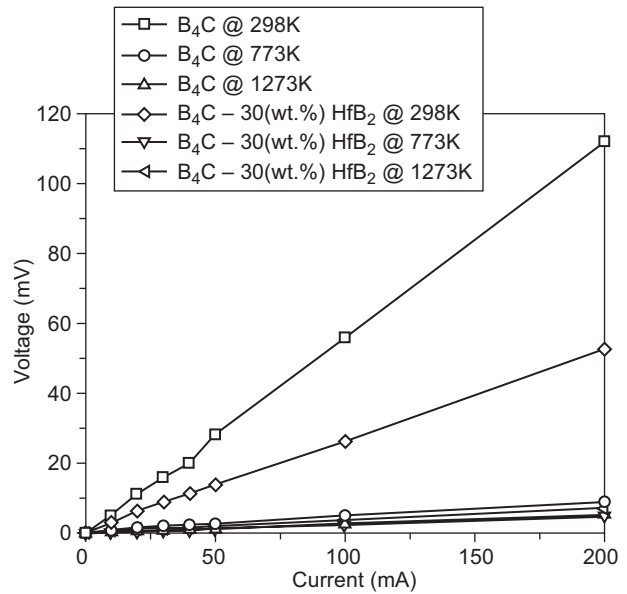


Figure 9. Voltage and current characteristics of  $\text{B}_4\text{C}$  and  $\text{B}_4\text{C}$ -30 wt. %  $\text{HfB}_2$  composite at temperatures 298 K, 773 K and 1273 K.

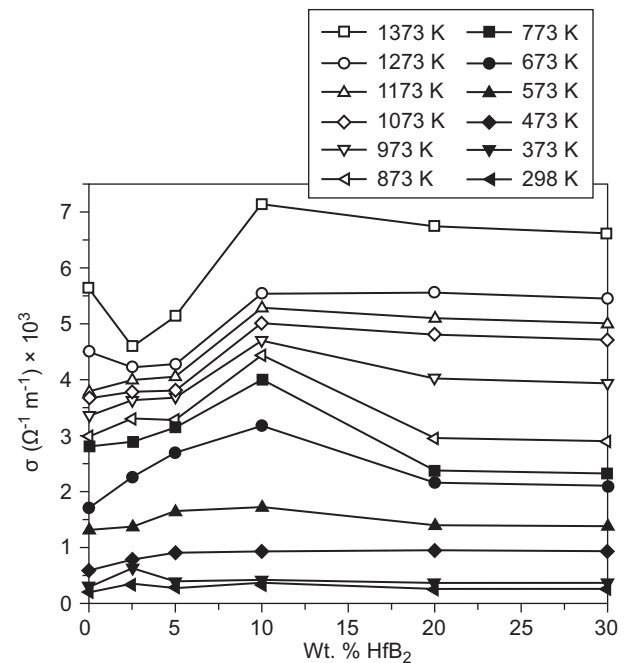


Figure 10. The electrical conductivity ( $\sigma$ ) of  $\text{B}_4\text{C}$  and its dependence on  $\text{HfB}_2$  content at different temperatures (298 K to 1373 K).

the room temperature electrical conductivity of the material. This is due to the fact that the  $\text{HfB}_2$  is a better conductor at room temperature than  $\text{B}_4\text{C}$  and hence, the overall conductivity of the composite was increased. Electrical conductivities of  $\text{B}_4\text{C}$  containing different proportion of  $\text{HfB}_2$  was estimated based on Maxwell's equation, which is in line with the present observations (Figure 11). Similar trend in conduction has been reported when  $\text{CrB}_2$  was added to  $\text{B}_4\text{C}$  [36]. Like  $\text{TiB}_2$  and  $\text{ZrB}_2$  ceramics,  $\text{HfB}_2$  is known to exhibit metallic like conduction behaviour whose conductivity decreases with increase of temperature due to the increased scattering of conducting electrons by lattice phonons [23, 37-38]. Hence, due to reinforcement of  $\text{HfB}_2$  to  $\text{B}_4\text{C}$ , it was expected that the conductivity of  $\text{B}_4\text{C}$  would decrease at high temperatures; but on the contrary, it increased with temperature. At 473 K, conductivity of the  $\text{B}_4\text{C}$  sample was measured to increase from  $797 \Omega^{-1}\cdot\text{m}^{-1}$  to  $936 \Omega^{-1}\cdot\text{m}^{-1}$  on addition of 2.5 and 10 wt. % of  $\text{HfB}_2$  respectively. On further increase of  $\text{HfB}_2$  content from 10 to 30 wt. %, the conductivity of  $\text{B}_4\text{C}$  remain unchanged at 473 K ( $\text{B}_4\text{C}$  - 30 wt. %  $\text{HfB}_2$ :  $938 \Omega^{-1}\cdot\text{m}^{-1}$ ); whereas conductivity measured at temperatures between 573 K and 1373 K showed a decreasing trend on increasing the  $\text{HfB}_2$  content beyond 10 wt. % (Figure 10).  $\text{B}_4\text{C}$  samples containing 2.5 and 5 wt. %  $\text{HfB}_2$  resulted in systematic rise in conductivities up to 1173 K, beyond this temperature, a slight dip in conductivity was noticed compared with monolithic  $\text{B}_4\text{C}$ . For all the tested temperatures between 298 K and 1373 K,  $\text{B}_4\text{C}$  with 10 wt. %  $\text{HfB}_2$  content exhibited a highest conductivity compared with that of all the other compositions. The conductivity of  $\text{B}_4\text{C}$ -10 wt. %  $\text{HfB}_2$  composite was measured as  $7144 \Omega^{-1}\cdot\text{m}^{-1}$  which is 26 % higher than monolithic  $\text{B}_4\text{C}$  ( $5639 \Omega^{-1}\cdot\text{m}^{-1}$ ) and 8 % higher than  $\text{B}_4\text{C}$ -30 wt. %  $\text{HfB}_2$  composite ( $6608 \Omega^{-1}\cdot\text{m}^{-1}$ ) at 1373 K. Dolor et al. made similar observations during the investigation

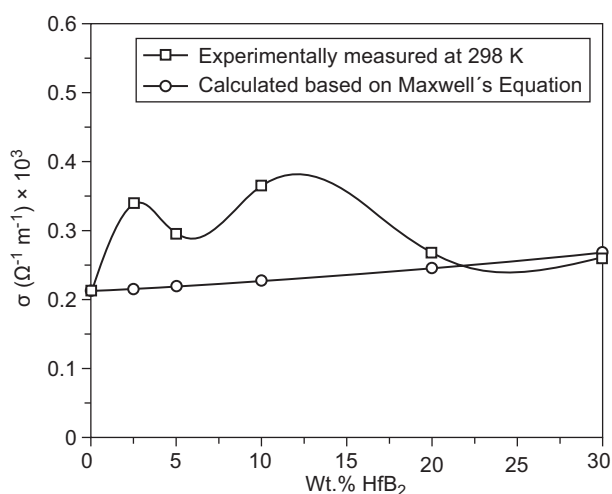


Figure 11. Comparison of experimentally measured and estimated room temperature electrical conductivity ( $\sigma$ ) of  $\text{B}_4\text{C}$  specimens as a function of  $\text{HfB}_2$  content.

of high temperature electrical properties of  $\text{B}_4\text{C}$ - $\text{HfB}_2$  system and reported the maximum in conductivity with 10 wt. %  $\text{HfB}_2$  at 1000 K [39]. Further he articulated this anomalous behaviour to the lower sinter density of sample that contained 10 wt. %  $\text{HfB}_2$  compared with other compositions. In the present study, all the tested samples had identical sinter densities ( $\sim 99\% \rho_{th}$ ) but still exhibited the similar behaviour. Hence the remark over density effects on this anomalous behaviour in conductivity could be considered invalid. The electrical conductivity of the composites is dependent on several factors including the intrinsic conductivity of matrix and particulates, its volume fraction, size, shape, distributions and porosity. These parameters were not fully investigated in this study.

Figure 12 shows the plot of  $\log \sigma T$  versus  $1/T$  that exhibit nearly a linear relationship for both  $\text{B}_4\text{C}$  and  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composites. The activation energy of  $\text{B}_4\text{C}$  was calculated to be 0.163 eV which indicates the domination of bipolaron hopping during electrical conduction; whereas activation energies of  $\text{B}_4\text{C}$  containing 2.5 to 30 wt. %  $\text{HfB}_2$  were ranged between 0.14 - 0.16 eV. In the present study, the activation energy reported for  $\text{B}_4\text{C}$  is consistent with the literature reported values (0.14 - 0.17 eV) [40-42]. These calculated activation energies for composites are more close to the value of monolithic  $\text{B}_4\text{C}$  which suggests that the electrical conduction in  $\text{B}_4\text{C}$  composites would primarily occur through  $\text{B}_4\text{C}$  grains with bipolaron hopping as the conduction mechanism. The temperature coefficient of electrical resistance (TCER) values were calculated to be  $1.03 \times 10^{-3} \text{ K}^{-1}$  for  $\text{B}_4\text{C}$  and  $1.54 \times 10^{-3} \text{ K}^{-1}$  for  $\text{B}_4\text{C}$ -30 wt. %  $\text{HfB}_2$ . The TCER values of the composites ranged between 1.2 to 1.64 except for 10 wt. %  $\text{HfB}_2$  composition. TCER of  $\text{B}_4\text{C}$ - $\text{HfB}_2$  composites is given in Table 3 which indicates a maximum value of  $2.23 \times 10^{-3} \text{ K}^{-1}$  corresponding to 10 wt. %  $\text{HfB}_2$  addition. The change in conductivity with respect to temperature

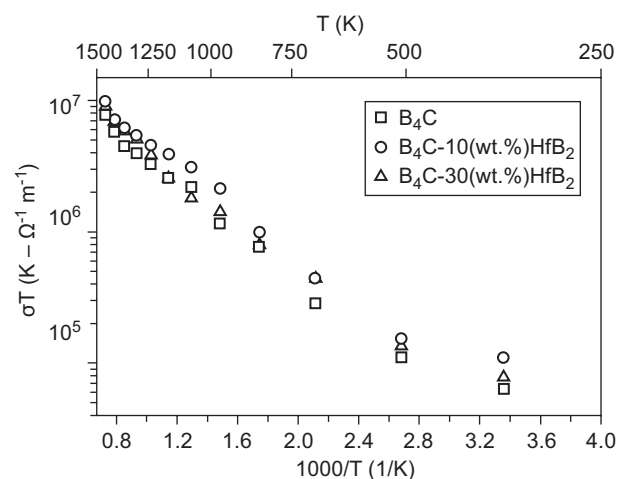


Figure 12. Plot of  $\log(\sigma T)$  versus reciprocal of temperature for  $\text{B}_4\text{C}$  with 0 wt. %, 10 wt.% and 30 wt.%  $\text{HfB}_2$ .

Table 3. Temperature coefficient of electrical resistance (TCER) of monolithic B<sub>4</sub>C and B<sub>4</sub>C–HfB<sub>2</sub> composites

Composition	TCER (10 <sup>-3</sup> K <sup>-1</sup> )
B <sub>4</sub> C	1.03
B <sub>4</sub> C-2.5 wt. % HfB <sub>2</sub>	1.42
B <sub>4</sub> C-5 wt. % HfB <sub>2</sub>	1.29
B <sub>4</sub> C-10 wt. % HfB <sub>2</sub>	2.23
B <sub>4</sub> C-20 wt. % HfB <sub>2</sub>	1.61
B <sub>4</sub> C-30 wt. % HfB <sub>2</sub>	1.54

is found to be predominant for B<sub>4</sub>C-10 wt. % HfB<sub>2</sub> composite compared to other proportions including monolithic B<sub>4</sub>C.

### CONCLUSION

- In the present study, 99 % dense B<sub>4</sub>C-HfB<sub>2</sub> was fabricated by hot-pressing route with the following processing conditions: Temperature of 2173 K under 40 MPa mechanical pressure for 1 h dwell time.
- Hardness, elastic modulus and fracture toughness of B<sub>4</sub>C-HfB<sub>2</sub> composites were measured to be in the range of 36 - 28 GPa, 465-525 GPa and 2.3 - 7.1 MPa·m<sup>1/2</sup> respectively. With increasing HfB<sub>2</sub> content, hardness decreases and fracture toughness increases. B<sub>4</sub>C-10 wt. % HfB<sub>2</sub> exhibited a maximum elastic modulus of 525 GPa.
- Crack deflection was observed and attributed to be the primary toughening mechanism for B<sub>4</sub>C-HfB<sub>2</sub> composites.
- Electrical conductivity of both B<sub>4</sub>C and B<sub>4</sub>C-HfB<sub>2</sub> composites increases with increase in temperature. B<sub>4</sub>C containing 10 wt. % HfB<sub>2</sub> exhibited a maximum conductivity of 7144 Ω<sup>-1</sup>·m<sup>-1</sup> at 1373 K. Activation energy of B<sub>4</sub>C-HfB<sub>2</sub> composites were measured to be in the range of 0.14 - 0.16 eV.

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