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Analytical methods

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Quantification of N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL) and Total Lysine through Stable Isotope Dilution Assay and Tandem Mass Spectrometry

Antonio Dario Troise^{1,2*}, Alberto Fiore³, Markus Wiltafsky⁴, Vincenzo Fogliano¹

¹ Food Quality Design Group, Wageningen University, PO Box 8129, 6700 EV, Wageningen, The Netherlands

² Department of Agricultural and Food Science, University of Napoli "Federico II", Parco Gussone, 80055 Portici, Napoli, Italy

³ School of Science, Engineering & Technology Division of Food Science, Abertay University Dundee DD1 1HG, UK

⁴ Evonik Industries AG, Rodenbacher Chaussee 4, 63457 Hanau, Germany

***Corresponding author**

Antonio Dario Troise

antonio.troise@wur.nl

Phone number: +39 081 2539360

Abstract

The control of Maillard reaction (MR) is a key point to ensure processed foods quality. Due to the presence of a primary amino group on its side chain, lysine is particularly prone to chemical modifications with the formation of Amadori products (AP), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL). A new analytical strategy was proposed which allowed to simultaneously quantify lysine, CML, CEL and the N ϵ -(2-Furoylmethyl)-L-lysine (furosine), the indirect marker of AP. The procedure is based on stable isotope dilution assay followed by, liquid chromatography tandem mass spectrometry. It showed high sensitivity and good reproducibility and repeatability in different foods. The limit of detection and the RSD% were lower than 5 ppb and below 8%, respectively. Results obtained with the new procedure not only improved the knowledge about the reliability of thermal treatment markers, but also defined new insights in the relationship between Maillard reaction products and their precursors.

Keywords: Maillard reaction, LC-MS/MS, CML, CEL, lysine, furosine

Abbreviations: Maillard reaction (MR), Maillard reaction end products (MRPs) N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL),

List of compounds: N ϵ -(2-Furoylmethyl)-L-lysine (furosine, PubChem CID: no items) , N ϵ -(Carboxymethyl)-L-lysine (CML, PubChem CID: 123800), N ϵ -(Carboxyethyl)-L-lysine (CEL, PubChem CID: no items), lysine (PubChem CID: 5962).

1 1. Introduction

2 The final quality of many industrial food products depends on food formulation and processing
3 design resulting in the formation of a huge variety of molecules as a consequence of thermal
4 treatments and chemical changes (van Boekel, Fogliano, Pellegrini, Stanton, Scholz, Lalljie, et al.,
5 2010). Along with lipid oxidation, the Maillard reaction (MR) occupies a prominent place in the
6 final quality of food being responsible not only for the desired color and aroma compounds but also
7 for the formation of potentially toxic Maillard reaction end products (MRPs). The reaction between
8 reducing sugars and amino groups is the first step in the Maillard cascade: the formation of the
9 stable 1-amino-1-deoxy-2-ketose the Amadori product (AP) and 2-amino-2-deoxyaldose Heyns
10 products represents the starting point of the many chemical pathways of this reaction (Hodge,
11 1953). The presence of an amino group on the side chain of lysine makes this amino acid
12 particularly sensitive to the carbonyls attachments. The modifications arising from the lysine
13 blockage resulted in the formation of a bewildering array of molecules: N ϵ -(1-Deoxy-D-fructos-1-
14 yl)-L-lysine (fructosyl-lysine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine
15 (CEL), pentosidine, pyrrolidine, lysino-alanine, 5-hydroxymethylfurfural (HMF), α -dicarbonyls and
16 aroma key odorants (Yaylayan & Huyghuesdespointes, 1994). Fructosyl-lysine, CML and CEL
17 represent the most widely studied MRPs, and they are often used as biomarker of food quality
18 (Erbersdobler & Somoza, 2007; Nguyen, van der Fels-Klerx, & van Boekel, 2013). As highlighted
19 in **Figure 1**, the acid hydrolysis adopted to release free amino acids from the polypeptide chain
20 promote the conversion of the 1-deoxy-fructosyl-L-Lysine (AP) through a cyclized Schiff base, into
21 the of N ϵ -(2-furoylmethyl)-L-lysine (furosine) which is a compound that can be quantified after
22 protein hydrolysis and it has been widely used as marker of thermal treatment particularly in the
23 dairy products (Krause, Knoll, & Henle, 2003).

24 The formation of CML and CEL from the oxidation of ARP and HRP has been well characterized
25 (Nguyen, van der Fels-Klerx, & van Boekel, 2013). Carbohydrate fragmentation allows the

26 formation of glyoxal and methylglyoxal that readily react with lysine residues yielding the
27 glycoxidation products CML and CEL, respectively (Ahmed, Thorpe, & Baynes, 1986). Moreover,
28 CML and CEL can be formed via the Namiki-pathway through three subsequent steps: Schiff base
29 production, glycolaldehyde alkylimine synthesis, oxidation and formation of glyoxal or
30 methylglyoxal which react with lysine to yield CML and CEL. Another route of CML and CEL
31 formation is linked to lipid peroxidation as glyoxal and methylglyoxal can derive from
32 polyunsaturated fatty acids (Hidalgo & Zamora, 2005). Moreover, the two markers can be also
33 formed from fragmentation and subsequent glycation of ascorbic acid and dehydroascorbic acid
34 (Leclere, Birlouez-Aragon, & Meli, 2002).

35 From the analytical point of view the identification of these markers of heat treatment can be
36 approached in several ways (Tessier & Birlouez-Aragon, 2012). Furosine is used as indirect marker
37 of quality control of moderately heat-treated dairy samples. The golden standards for furosine
38 detection are ion-exchange chromatography, reverse phase high performance liquid
39 chromatography (RP-HPLC) with UV detection, (Henle, Zehetner, & Klostermeyer, 1995) capillary
40 electrophoresis and ion-pairing HPLC by using sodium-heptanosulphonate (Vallejo-Cordoba,
41 Mazorra-Manzano, & Gonzalez-Cordova, 2004). These procedures had several drawbacks mainly
42 related to the modifications occurring during sample preparation: the acidic hydrolysis does not
43 allow the differentiation between AP and glycosyl-amine; overestimation or underestimation linked
44 to the acidic hydrolysis might occur due to the formation of further intermediates and end-products
45 (Pischetsrieder & Henle, 2012).

46 As for furosine, CML and CEL analysis implies acidic hydrolysis to hydrolyze peptide bonds
47 followed by their quantification that could be performed by different instrumental methods
48 (Nguyen, van der Fels-Klerx, & van Boekel, 2013). In some papers a pre-column derivatization
49 with o-phthalaldehyde was used to allow the detection by fluorescence detector (Hartkopf, Pahlke,
50 Ludemann, & Erbersdobler, 1994), while a widely used approach for CML and CEL detection is

51 gas or liquid chromatography coupled with tandem mass spectrometry. Specifically, multiple
52 reaction monitoring (MRM) mode improves the sensitivity, reduces the coefficient of variability
53 and ruled out the problems of derivatization (Delatour, Hegele, Parisod, Richoz, Maurer, Steven, et
54 al., 2009). A double derivatization is required for GC separation and this bottleneck highlights the
55 advantages of LC-MS/MS detection: no derivatization, highest sensitivity and good reproducibility
56 (Charissou, Ait-Ameur, & Birlouez-Aragon, 2007; Fenaille, Parisod, Visani, Populaire, Tabet, &
57 Guy, 2006). Moreover CML, CEL and lysine detection is possible also by matrix-assisted laser
58 desorption/ionization time-of-flight mass spectrometry (MALDI-TOF-MS) that allows relative
59 quantification of protein lactosylation and it is a reliable method to monitor the early Maillard
60 reaction as well as MRPs during milk processing (Meltretter, Becker, & Pischetsrieder, 2008).

61 The aim of the present paper, was to further improve the existing methodologies for the detection of
62 lysine and MRPs. A new method was designed which included direct hydrolysis along with stable
63 isotope dilution assay coupled with solid phase extraction and ion pairing liquid chromatography
64 tandem mass spectrometry (LC-MS/MS). The developed procedure allowed the simultaneous
65 detection of total lysine, furosine, CML and CEL. The method was tested on several foods: milk,
66 infant formulas, cookies, bread slices. The robustness after several injections and the reliability of
67 the results obtained were evaluated in soybean-based feed products obtained under severe thermal
68 treatment conditions. Data demonstrated satisfactory analytical performances on all tested samples
69 and results were perfectly in line with those previously obtained.

70 **2. Material and methods**

71 *2.1 Chemicals and reagents*

72 Acetonitrile, methanol and water for solid phase extraction (SPE) and LC-MS/MS determination
73 were obtained from Merck (Darmstadt, Germany). The ion pairing agent perfluoropentanoic acid,
74 trichloroacetic acid, hydrochloric acid (37%) and the analytical standards L-lysine hydrochloride

75 and [4,4,5,5- d_4]-L-lysine hydrochloride (d_4 -Lys) were purchased from Sigma-Aldrich (St. Louis,
76 MO). Analytical standards N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine
77 (CML) and its respective deuterated standard N ϵ -(Carboxy[$^2\text{H}_2$]methyl)-L-Lysine (d_2 -CML) were
78 obtained from Polypeptide laboratories (Strasbourg, France), N ϵ -(Carboxyethyl)-L-lysine and its
79 internal standard N ϵ -(Carboxy[$^2\text{H}_4$]ethyl)-L-lysine (d_4 -CEL) were purchased from TRC-Chemicals
80 (North York, Canada).

81 *2.2 Foods samples*

82 Powdered infant formula and milk samples were purchased in a local market, biscuits samples and
83 bread slices were prepared according previous papers published by our group (Fiore, Troise, Mogol,
84 Roullier, Gourdon, Jian, et al., 2012; Vitaglione, Lumaga, Stanzione, Scalfi, & Fogliano, 2009).
85 UHT milk was prepared according to the procedure previously described (Troise, Fiore,
86 Colantuono, Kokkinidou, Peterson, & Fogliano, 2014). Raw milk (protein, 3.5%; fat, 1%) was
87 purchased in a local market.

88 *2.2.1 Soybean samples*

89 One batch of quartered raw soybeans was purchased from Rieder Asamhof GmbH & Co. KG
90 (Kissing, Germany). The raw soybeans were further processed at the hydrothermal cooking plant of
91 Amandus Kahl GmbH & Co. KG (Reinbeck, Germany). First, the beans were short-term
92 conditioned to reach a temperature of 80 °C after 45 seconds. Afterwards, the beans entered a
93 hydrothermic belt cooker at 72 °C and left inside for 3 min at a temperature of 70 °C. Then they
94 were expanded at 117 °C using an annular gap expander (Typ OEE 8, Amandus Kahl GmbH & Co.
95 KG, Reinbeck, Germany). The expanded soybeans were collected in a drying wagon for 10 min.
96 Then they were dried with air at 65 °C for 10 min and cooled for another 10 min to reach a final
97 moisture content of 12%. Afterwards, the expanded soybeans were autoclaved for 0, 5, 10, 15, 20,

98 25, 30, 35, 40, 45, 50 and 60 min at 110 °C and 1470 mbar using a fully controlled autoclave (Typ
99 HST 6x9x12, Zirbus Technology GmbH, Bad Grund, Germany).

100 2.3 Samples preparation

101 Lysine and its derivatives N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine
102 (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL) were analyzed considering previous papers (Delatour, et
103 al., 2009; Fenaille, Parisod, Visani, Populaire, Tabet, & Guy, 2006; Troise, Dathan, Fiore, Roviello,
104 Di Fiore, Caira, et al., 2014) and introducing several modifications. Briefly, 100 mg of each sample
105 was accurately weighed in a screw capped flask with PTFE septa and 4 mL of hydrochloric acid (6
106 N) was added. The mixture was saturated by nitrogen (15 min at 2 bar) and hydrolyzed in an air
107 forced circulating oven (Mettler, Schwabach, Germany) for 20 h at 110° C. The mixture was
108 filtrated by polyvinylidene fluoride filters (PVDF, 0.22 Millipore, Billerica, MA) and 400 μ l was
109 dried under nitrogen flow in order to prevent the oxidation of the constituents. The samples were
110 reconstituted in 370 μ l of water and 10 μ L of each internal standard d_4 -Lys, d_2 -CML and d_4 -CEL
111 was added in order to obtain a final concentration of 200 ng/mg of samples for both standards.
112 Samples were loaded onto equilibrated Oasis HLB 1 cc cartridges (Waters, Wexford, Ireland) and
113 eluted according to the method previously described, then 5 μ l was injected onto the LC/MS/MS
114 system.

115 2.4 Liquid chromatography tandem mass spectrometry (LC-MS/MS)

116 Separation of furosine, CML, CEL, lysine and their respective internal standards was achieved on a
117 reversed – phase core shell HPLC column (Kinetex C18 2.6 μ m, 2.1 mm x 100 mm, Phenomenex,
118 Torrance) using the following mobile phases: A, 5 mM perfluoropentanoic acid and B, acetonitrile
119 5 mM perfluoropentanoic acid. The compounds were eluted at 200 μ L/min through the following
120 gradient of solvent B (t in [min]/[%B]): (0/10), (2/10), (5/70), (7/70), (9/90), (10/90), (12/10),
121 (15/10). Positive electrospray ionization was used for detection and the source parameters were

122 selected as follows: spray voltage: 5.0 kV; capillary temperature: 350 °C, dwell time 100 ms, cad
123 gas and curtain gas were set to 45 and 5 (arbitrary units). The chromatographic profile was recorded
124 in MRM mode and the characteristic transitions were monitored in order to improve selectivity
125 using an API 3000 triple quadrupole (ABSciex, Carlsbad, CA). All relevant parameters are
126 summarized in **Table 1**.

127 *2.5 Analytical performances*

128 CML, CEL, furosine and total lysine were quantified using a linear calibration curve built with
129 specific solutions of CML spiked with d_2 -CML, lysine and furosine spiked with d_4 -lysine and CEL
130 spiked with d_4 -CEL (final concentration of internal standards: 200 ng/ml) dissolved in water. The
131 limit of detection (LOD) and the limit of quantitation (LOQ) were monitored according to the signal
132 to noise ratio (Armbruster, Tillman, & Hubbs, 1994). The coefficients of determination r^2 for the 4
133 analytes were tested plotting the ratio between the pure compounds and their respective internal and
134 the concentration of the pure compounds in the linearity range 5-1000 ng/mL. The internal standard
135 ratio was used for the quantification and the relative standard deviation of intraday and interday
136 assay was monitored three times each day and six times in different days. The recovery test was
137 monitored according to the concentration of the internal standards used and to the ratio between
138 labeled compounds and native compounds.

139 *2.6 Statistical analysis*

140 All of the analyses were performed in quadruplicate and the results expressed as mg/100 g of
141 protein. Statistical calculations were performed using Matlab R2009b (Natick, MA) while for mass
142 spectrometry data, Analyst version 1.4.2 (Applied Biosystems, Carlsbad, CA) was used.

143 **3. Results and discussion**

144 *3.1 Liquid chromatography set up*

145 Under the above described chromatographic conditions, typical retention time of CML and d_2 -CML
146 was 7.11 min, for d_4 -Lys and Lys it was 7.23 min, for furosine it was 7.91 min, while for CEL and
147 d_4 -CEL it was 7.36 min (**Figure 2**). Previous papers highlighted the problems due to the poor
148 retention of amino acids and their derived molecules on silica bonded and C-18 column (Frolov &
149 Hoffmann, 2008). Preliminary trials performed using C-18 column without the ion pairing agent
150 confirmed this feature: the retention was poor and the analytes co-eluted with the impurities on the
151 front of the chromatographic run with the consequent partial suppression of the signal associated to
152 the markers. Inadequate separation of the analytes was obtained also using polar end-capped
153 column; however a significant improvement was obtained using with this column
154 perfluoropentanoic acid as ion pairing agent. In these experimental conditions, the retention time
155 followed a typical reversed phase profile according to the polarity and to the steric hindrance of
156 each molecule, as previously observed by other papers published earlier (Fenaille, Parisod, Visani,
157 Populaire, Tabet, & Guy, 2006; Troise, Fiore, Roviello, Monti, & Fogliano, 2014). The presence of
158 the ion pairing agent charged the core shell residues increasing the retention and promoting the
159 selectivity of the positively charged CML, CEL, furosine, lysine and their respective internal
160 standards. The presence of a core shell phase increased of the resolution which directly reflects the
161 good performances of the reported method, the shape of the peak was maintained over each batch
162 and the retention time shift was lower than 0.5 min, highlighting the robustness of the analytical
163 performances.

164 *3.2 Mass spectrometry set up*

165 Mass spectrometry conditions were optimized by infusing singularly the seven standards directly in
166 the ion source. Collision energy, declustering potential, tube lens voltage along with spray voltage
167 and interface temperature were monitored in order to favor the formation of the typical
168 fragmentation pattern (Delatour, et al., 2009). The lysine derived compounds underwent the
169 formation of the fragment ion at 130 m/z which corresponds to the pipercolic acid generated by the

170 subsequent cyclization of the side chain of lysine and the loss of ϵ - amino group, similarly the mass
171 shift for deuterated standards *d4*-CEL and *d4*-Lys was +4 Da as consequence of the fragmentation
172 occurred on the side chain of lysine (**Figure S1** in supplementary material section) (Yalcin &
173 Harrison, 1996). The MRM revealed the loss of formic acid giving the typical fragment at m/z 84;
174 the mass shift for the deuterated molecules was +4. The seven standards were also infused inside
175 the chromatographic flow in order to evaluate the interferences due to the ion pairing agent or to the
176 solvent and the results revealed that no enhancement or suppression effect can be ascribed to the
177 parameters monitored.

178 *3.3 Analytical performances*

179 The analytical performances of the method were tested against reproducibility, repeatability, limit
180 of detection (LOD), limit of quantification (LOQ), linearity, precision, carry-over and coefficient of
181 correlation (r^2). Before and after each batch, three solutions of acetonitrile and water (90:10; 50:50
182 and 10:90) were injected in order to verify the absence of any contaminants with the same signal
183 and the same retention time of the analyzed molecules. The limit of detection and the limit of
184 quantitation were determined according to the procedure previously described. The concentration
185 0.1 ppb resulted in no signal, while the LOD was 0.5 ppb for CML and lysine while for CEL and
186 furosine it was 1 and 3 ppb, respectively. The slight differences among CML, CEL and furosine can
187 be related to the different stability in the injection conditions. By injecting these concentrations the
188 signal to noise ratio was always higher than 3. The LOQ were 5 ppb for CML, CEL and Lysine
189 while for furosine it was 9 ppb, as highlighted in **Table 2**. These values were perfectly in line with
190 those previously described for CML, CEL and lysine quantification by MS/MS (Delatour, et al.,
191 2009; Tareke, Forslund, Lindh, Fahlgren, & Ostman, 2013) while for furosine the performance of
192 LOD and LOQ were below the values previously reported in milk (Bignardi, Cavazza, & Corradini,
193 2012). According to the LOD and LOQ, linearity was achieved in the range 5-1000 ppb for CML,
194 CEL and lysine, while for furosine the linearity range was between 9 and 1000 ppb. The carryover

195 effect was tested injecting after each point of the calibration curves a solution consisting in
196 acetonitrile and water (50:50, v/v) and verifying the absence of the target compounds. The linearity
197 of the calibration curves was evaluated three times in the same day (intraday assay for the
198 reproducibility) and three times for three subsequent days (interday assay for the repeatability)
199 using the ratio between the target compounds and their respective internal standard. The RSD (%)
200 among the three curves was always lower than 8%, demonstrating that external factors had marginal
201 impact on the performance of the method. Each point of the calibration curves was monitored using
202 two specific transitions: the most intensive fragment was used as quantifier, the lowest as qualifier.
203 For CML, CEL, furosine and lysine, the respective transitions of m/z 205–84.1, m/z 219.1–84.1, m/z
204 255.1–130.2, and m/z 147.2–130.2 were used as quantifier, whereas m/z 205–130.2, m/z
205 219.1–84.1, m/z 255.1–84, and m/z 147.2–84.1 were used as qualifier. CML was quantified using
206 d_2 -CML as internal standard (m/z 207–144.1 and 207–84 for quantification and confirmation,
207 respectively), CEL was quantified using d_4 -CEL (m/z 223–134.1 and 223 – 84 for quantification
208 and confirmation, respectively) whereas for furosine and lysine, d_4 -lysine was used (m/z
209 151.2–134.1 and m/z 151.1– 88 for quantification and confirmation, respectively). The use of d_4 -
210 lysine as internal standard for the quantification and recovery of furosine was optimized by
211 monitoring the relative intensity of furosine standard towards d_4 -CEL, d_2 -CML and d_4 -lysine. A
212 mixture of the four standards (10 ppm) was directly infused in the ion source. Results revealed that
213 the intensity of the signal at m/z 151.2 and m/z 255 were similar and both were 15% higher than the
214 signal of d_2 -CML and d_4 -CEL.

215 The response of the method in food was tested during each batch evaluating the ratio between the
216 target compounds and the internal standard, these procedures confirmed and deepened the aspects
217 linked to the recovery assay: in each sample the ratio between the area of the analyte and the area of
218 the deuterated compounds was compared towards the calibration curve in order to obtain the final
219 concentration of the analytes in the matrix. The intensity of the internal standard in the samples and

220 in the standard was compared and the RSD (%) between the spiked samples and the spiked
221 standards was always lower than 10%. The recovery test was monitored in all the food matrix
222 according to the intensity of the internal standard, the results were 91.1 ± 8.4 , 84.2 ± 7.4 , 88.0 ± 6.9
223 for *d2*-CML, *d4*-CEL and *d4*-Lysine.

224 3.4 CML, CEL, furosine and total Lysine in food

225 Powdered samples were freeze dried prior analysis in order to remove the interferences due to the
226 humidity. The extraction procedure of MRPs is characterized by three key steps: the reduction with
227 sodium borohydride, the hydrolysis with hydrochloric acid and the stable isotope dilution assay
228 prior ion pairing solid phase extraction. According to the nature of protein and to their concentration
229 each of the above listed can influence the yield and the efficiency of the extraction. The reduction
230 with sodium borohydride promotes the conversion of free fructosyl-lysine into hexitol-lysine in
231 order to avoid the overestimation of CML, CEL (Niquet-Leridon & Tessier, 2011). Moreover, the
232 use of sodium borohydride is recommended when the concentration of free unstable Amadori
233 products is high. Unfortunately, the use of this reducing agent had several drawbacks: protein
234 degradation and free counterpart losses during the reduction, precipitation and purification
235 procedure; moreover, the use of sodium borohydride can interfere with the release of furosine with
236 the above mentioned reduction of fructosyl-lysine into hexitol-lysine. After several preliminary
237 measurements it was decided to avoid the reduction. A good compromise between the detection of
238 furosine and that of CML/CEL was achieved controlling the oxidation under nitrogen. In particular,
239 prior the acidic hydrolysis the screw capped flasks were saturated with nitrogen in order to reduce
240 the effect of autoxidation and control the reaction pathway (Yaylayan & Huyghuesdespointes,
241 1994).

242 The use of hydrochloric acid is a mandatory step for the hydrolysis of peptide bonds and for the
243 release of amino acids, MRPs and for the conversion of fructosyl-lysine into furosine. Different

244 concentrations of protein per mL of hydrochloric acid can lead to different efficiency of the
245 hydrolysis with the consequent underestimation of lysine content. In the present study, the
246 extraction procedure was optimized in order to promote the dehydration reaction that leads to the
247 formation of furosine and to the release of MRPs (Krause, Knoll, & Henle, 2003; Mossine &
248 Mawhinney, 2007). Further studies will be conducted in order to compare the effect of time and
249 concentration of hydrochloric acid on lysine release, mainly in protein rich samples.

250 The above described analytical performances were tested in food and feed samples in order to
251 verify the robustness of the method. Several thermally treated foods were tested: powdered infant
252 formula, low lactose milk, lab scale UHT milk, biscuits samples, bread (all prepared according to
253 three different procedures previously described by our group) and powdered soybean-based feed
254 products (prepared at industry scale). All data are summarized in **Table 3**. The concentration of
255 CML in powdered infant formula analyzed ranged from 8.22 ± 0.31 mg/100 g of protein to $14.81 \pm$
256 0.92 mg/100 g of protein, while CEL and furosine ranged from 0.71 ± 0.02 mg/100 g of protein to
257 1.31 ± 0.11 mg/100 g of protein and 471.9 ± 22.3 mg/100 g of protein to 639.4 ± 21.1 mg/100 g of
258 protein, respectively. The concentration of total lysine varied from 9.89 ± 0.88 to 13.12 ± 0.78 % of
259 total protein. In low lactose milk the content of lysine was 5.21 ± 0.30 g/100 g of protein, while the
260 concentration of CEL and furosine was $0.28 \text{ mg} \pm 0.01$ mg/100 g of protein and 12.32 ± 0.31
261 mg/100 g of protein, respectively. CML was $1.28 \text{ mg} \pm 0.11$ mg/100 g of protein and this value was
262 perfectly in line with the one previously obtained. Lab scale UHT milk was prepared in order to
263 verify the effect on raw cow milk; while the lysine content was of the same order of magnitude of
264 the low lactose milk (4.71 ± 0.22 mg/100 g of protein), the concentration of the three markers of the
265 MR was 18.41 ± 0.93 , 1.12 ± 0.02 and 14.41 ± 1.02 mg/100 g of protein for CML, CEL and
266 furosine respectively. The results obtained were perfectly in line with those previously obtained for
267 the three categories of milk (Fenaille, Parisod, Visani, Populaire, Tabet, & Guy, 2006; Tareke,
268 Forslund, Lindh, Fahlgren, & Ostman, 2013), specifically the CML in low lactose milk was similar

269 to one previously obtained by our group for LC-MS/MS analysis (Troise, et al., 2014). The
270 concentration of CML and furosine was closed to the range previously obtained: 2.2 – 30.8 and 0.8
271 – 3.7 mg/100 g of protein for furosine and CML, respectively (de Sereys, Muller, Desic, Troise,
272 Fogliano, Acharid, et al., 2014).

273 In bakery products CML content was 43.75 ± 2.02 and 27.15 ± 0.61 mg/100 g of protein for biscuits
274 samples and bread slices, respectively, while CEL and furosine were 46.25 ± 3.01 and $10.01 \pm$
275 0.61 and 10.91 ± 0.01 and 98.55 ± 4.61 mg/100 g of protein for biscuits and bread, respectively.
276 The lysine content was almost similar in the two products: 5.01 ± 0.04 and 5.81 ± 0.04 g/100 g of
277 protein, even if the protein content was 6% and 8% for biscuits and bread. The results here reported
278 were of the same order of magnitude as the ones previously reported. Hull et al., analyzed several
279 kinds of bread and other bakery products and the concentration of CML ranged from 2.6 to 45.1
280 mg/100 g of protein for wheaten bread and potato bread, respectively (Hull, Woodside, Ames, &
281 Cuskelly, 2012). On the other hand He and coworker reported higher values for wholemeal bread:
282 CML ranged from 66.72 to 109.9 mg/100g of protein and CEL ranged from 53.30 to 82.04 mg/100
283 g protein for bread, while in biscuits samples the concentrations varied from 50.8 to 116.7 and
284 15.87 to 45.26 mg/100g protein for CML and CEL, respectively (He, Zeng, Zheng, He, & Chen,
285 2014). Interestingly, the concentration of furosine in bread (after 20 min at 200° C) is similar to the
286 one reported by Capuano and coworker: after 13 min the concentration of furosine increased up to
287 200 mg/100 g of protein and it quickly decreased up to 20 mg/100 g protein at the end of the
288 thermal treatment (Capuano, Ferrigno, Acampa, Ait-Ameur, & Fogliano, 2008). A similar kinetic
289 profile was observed also by Ramirez-Jimenez and coworker in sliced bread: the concentration of
290 furosine at the end of the process was 79.3 mg/100 g of protein while after 12 min it reached a
291 concentration higher than 200 mg/100 g protein (Ramirez-Jimenez, Garcia-Villanova, & Guerra-
292 Hernandez, 2001). In biscuit samples the kinetic profile revealed similar trends to the ones obtained
293 for bread; as a consequence at the end of the thermal process the concentration of furosine value of

294 10.01 ± 0.61 mg/100 g of protein was comparable to those of sucrose-containing cookies reported
295 by previous authors (Gökmen, Serpen, Açar, & Morales, 2008).

296 The above described analytical performances were evaluated in industrially prepared soybean feeds
297 in order to verify the main advantages of the method on industrial sampling. The simultaneous
298 quantification of the four analytes allowed a direct overview of the extent of the MR, where the
299 concentration of lysine and the formation of furosine, CEL and CML can be easily related to the
300 final quality of foods using a single extraction and a single injection. According to the procedure
301 described in material and methods section, soybeans were incubated at 110° C for one hour in an
302 autoclave and the kinetic profile was reported in **Figure 3**. The initial concentration of lysine was
303 3.45 ± 0.12 g/100 g of protein while CML, CEL and furosine were 9.94 ± 0.74, 0.98 ± 0.04 and
304 24.24 ± 1.74 mg/100 g of protein respectively. After 30 minutes the concentration of furosine
305 reached the highest values: 108.01 ± 8.97, then it rapidly decreased up to 60.58 ± 3.75 mg/100 g of
306 protein after 55 min. According to the reaction mechanism the degradation of the Amadori products
307 was followed by the increase of CML: at the end of the thermal treatment its concentration was
308 higher than 76 mg/100 g of protein. CEL reached the maximum concentration after 45 minutes
309 (2.41 ± 0.24 mg/100 g of protein), then it decreased probably due to degradation processes or to the
310 blockage of methylglyoxal by other compounds. The degradation of lysine was constant throughout
311 the thermal treatment, after 60 min lysine concentration was 2.60 ± 0.08 g/100 g of protein thus
312 around 23%. Several studies reported the effect of soy proteins in the development of the MR
313 focusing on soy health benefits and on the presence of functional molecules able to control the
314 extent of the MR (Palermo, Fiore, & Fogliano, 2012).

315 This paper represents the first example of a systematic study on the relationship between thermal
316 treatments, MR and soybean products in feeds and in pet food a topic recently attracting the
317 attention of the scientific community. In fact, it has been observed that the average daily intake
318 (mg/kg body weight^{0.75}) of HMF is 122 times higher for dogs and 38 times higher for cats than

319 average intake for adult humans. Possible health risks, such diabetes and renal failure, can be
320 associated to the intake of MRPs not only in human, but also in pets (van Rooijen, Bosch, van der
321 Poel, Wierenga, Alexander, & Hendriks, 2013).

322 4. Conclusion

323 The analytical method allowed a comprehensive approach in the analysis of MRPs, simultaneously
324 determining both lysine and its heat-induced derivatives. Up to now the golden standards for MRPs
325 detection were RP-HPLC with UVvis detection for furosine and LC-MS/MS for CML, CEL and
326 lysine, respectively. These results showed that the extraction procedure with nitrogen and
327 hydrochloric acid provided a good compromise for the simultaneous detection of the four analytes.
328 The analytical performances showed high sensitivity and good reproducibility and repeatability in
329 several foods. Quantitative data were fully in line with those previously obtained by other authors
330 on similar foods. The simultaneous detection of the four analytes offered a sensitive tool for the
331 kinetic modeling on neoformed contaminant reaction routes monitoring the precursor lysine, the
332 intermediate furosine via the indirect analysis of the Amadori products and the end-products CEL
333 and CML. The simultaneous monitoring of all compounds allowed to minimize the variability
334 among different samples and to combine the reaction steps starting from lysine blockage, Amadori
335 compounds formation and fragmentation, CML and CEL formation.

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338

339 **The authors declare no conflict of interests.**

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454 **Figure legend**

455 **Figure 1:** Effect of glucose and dicarbonyls on the formation of protein-bound MRPs. At the
456 bottom the structure of β -Lactoglobulin (Brownlow, Cabral, Cooper, Flower, Yewdall, Polikarpov,
457 et al., 1997).

458 **Figure 2:** Extracted ion chromatogram of the four target molecules and their respective internal
459 standards

460 **Figure 3:** Kinetic profile of the precursor lysine (green), intermediate, furosine (blue) and end-
461 products, CML and CEL (red).

462 **Figure S1:** Fragmentation pathway for lysine and its deuterated internal standard *d4*-lysine. The
463 structures of pipercolic acid and 1,2,3,4-tetrahydropyridin-1-ium ion was reported (Yalcin &
464 Harrison, 1996).

465

466 **Table legend**

467 **Table 1:** Mass spectrometry set up

468 **Table 2:** Analytical performances for the four analytes and their respective internal standards

469 **Table 3:** MRPs concentration after 8 replicates in different samples, the results for CML, CEL and
470 furosine were reported as mg/100 g of protein, except for lysine. The results were compared to the
471 AGE Database (Technische Universität Dresden, 2014).

Tables

Table 1

Compounds	[M+H] ⁺	Fragments	CE (V)	DP (V)
CML	205	84	29	30
		130.2	27	30
<i>d</i> ₂ -CML	207	84	30	20
		144	21	20
		130	17	20
Furosine	255.1	130	18	21
		84.4	28	21
Lys	147.2	130.2	16	30
		84.1	24	30
<i>d</i> ₄ -Lys	151.3	134.1	15	30
		88.2	26	30
CEL	219.2	130.3	20	30
		84.0	28	30
<i>d</i> ₄ -CEL	223	134.1	18	25
		88.0	30	25

Table 1: Mass spectrometry set up

Table 2

Compound	LOD	LOQ	RSD [%]	Linearity range	r ²	Recovery
CML	0.5 ppb	5 ppb	7	5-1000 ng/ml	> 0.99	91.1 ± 8.4
CEL	1 ppb	5 ppb	5	5-1000 ng/ml	> 0.99	84.2 ± 7.4
Lysine	0.5 ppb	5 ppb	5	5-1000 ng/ml	> 0.99	88.0 ± 6.9
Furosine	3 ppb	9 ppb	8	9-1000 ng/ml	> 0.99	88.0 ± 6.9

Table 2: Analytical performances for the four analytes and their respective internal standards

Table 3:

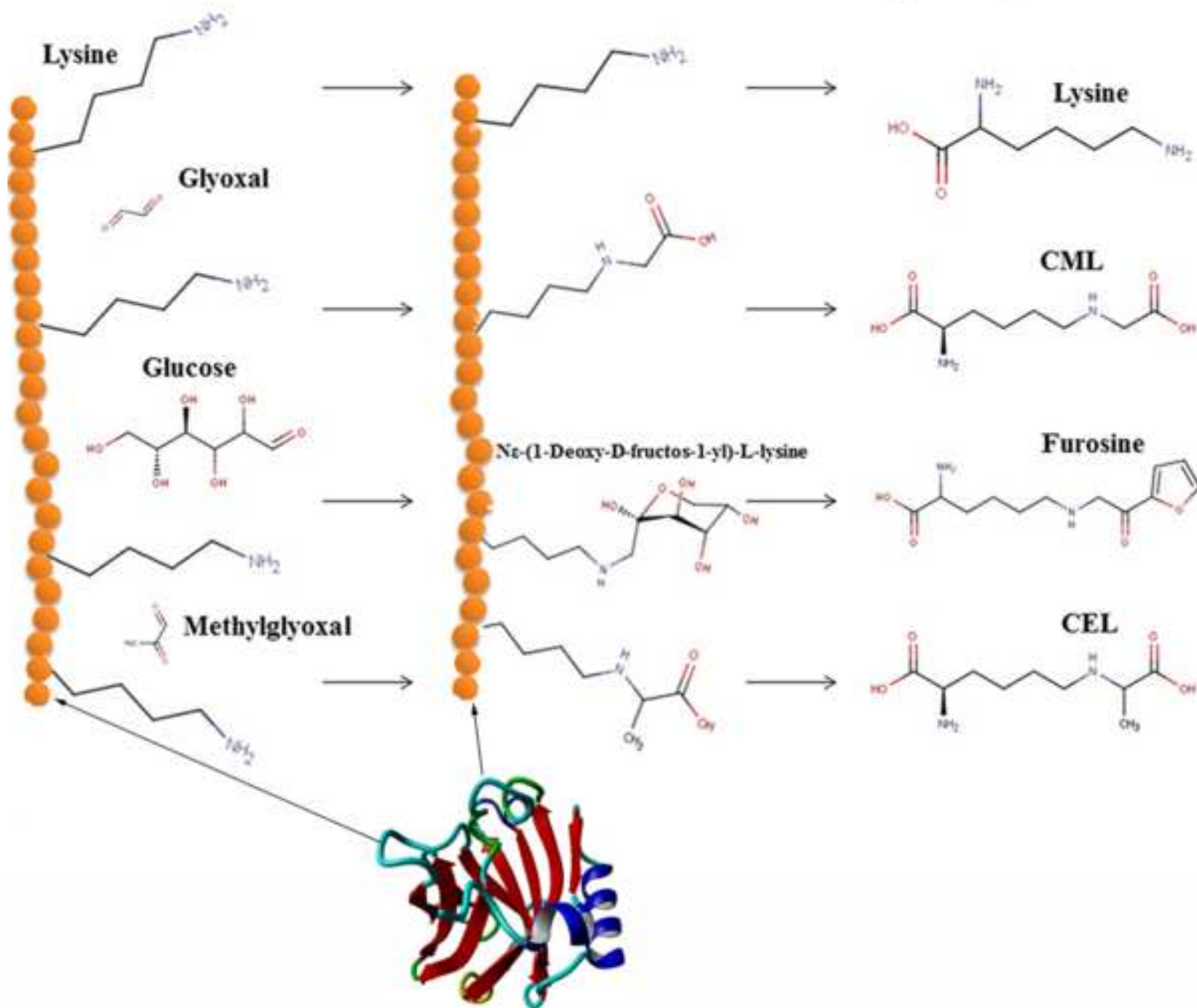
Food	CML	CEL	Furosine	Lysine (g/100 g protein)
Infant formula -1	8.22 ± 0.31	0.71 ± 0.02	471.91 ± 22.31	9.89 ± 0.88
Infant formula -2	10.4 ± 0.52	0.85 ± 0.06	542.53 ± 11.91	12.24 ± 0.91
Infant formula -3	10.9 ± 1.03	1.10 ± 0.05	574.5 ± 44.12	13.12 ± 0.78
Infant formula -4	14.81 ± 0.92	1.31 ± 0.11	639.4 ± 21.11	10.28 ± 1.01
<i>Age Database</i>	0.6 – 40.5	/	Up to 1819	/
Low lactose milk	1.28 ± 0.11	0.28 ± 0.01	12.32 ± 0.31	5.21 ± 0.30
<i>Age Database</i>	1.4	/	/	/
Lab scale UHT milk	18.41 ± 0.93	1.12 ± 0.02	14.41 ± 1.02	4.71 ± 0.22
<i>Age Database</i>	0.9-8.3	/	12.4 – 220.0	/
Biscuits	43.75 ± 2.02	46.25 ± 3.01	10.01 ± 0.61	5.01 ± 0.04
Bread slices	27.15 ± 0.61	10.91 ± 0.01	98.55 ± 4.61	5.81 ± 0.04
<i>Age Database</i>	2.6 – 45.1	/	/	/

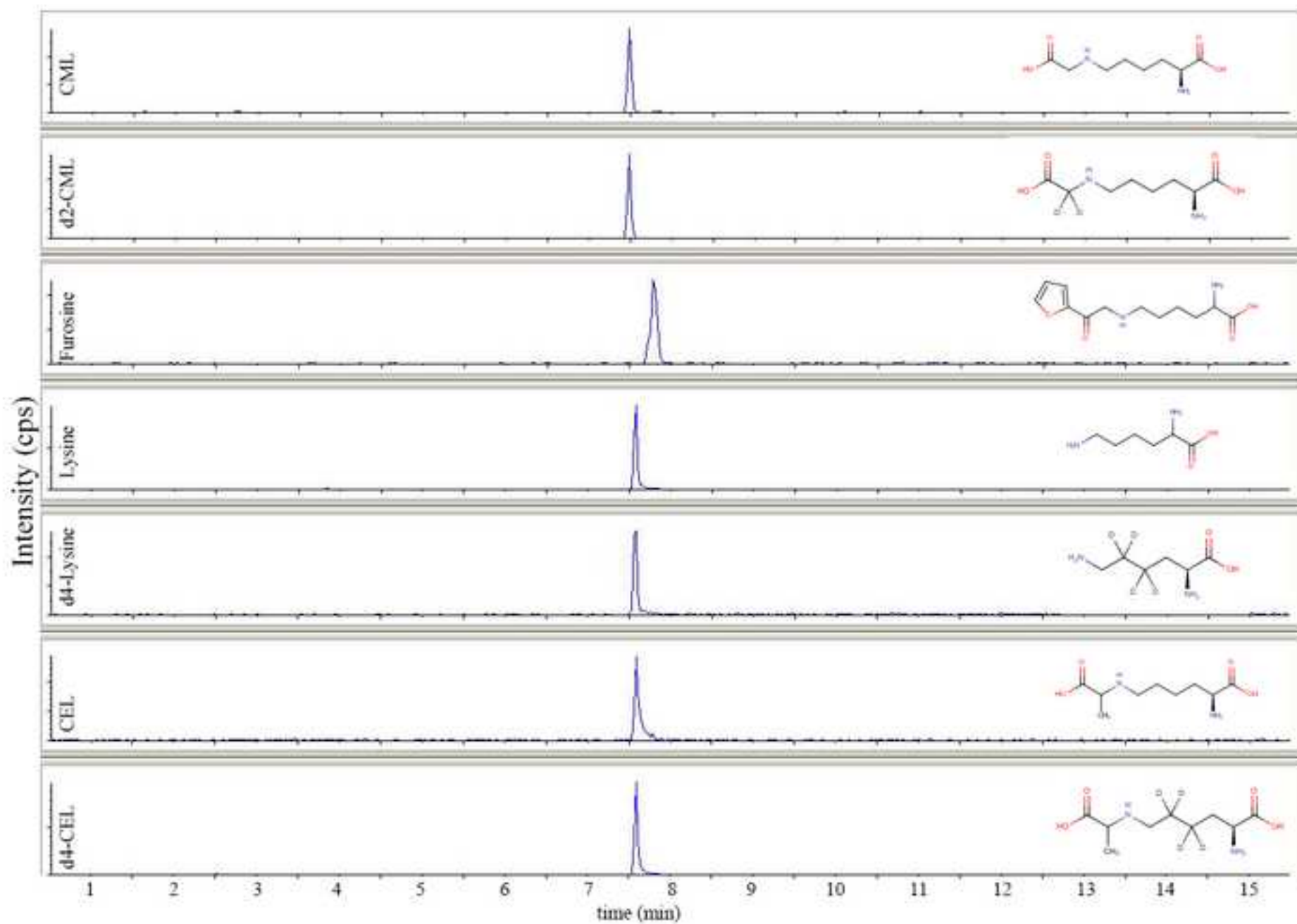
Table 3: MRPs concentration after 8 replicates in different samples, the results for CML, CEL and furosine were reported as mg/100 g of protein, except for lysine. The results were compared to the AGE Database (Technische Universität Dresden, 2014).

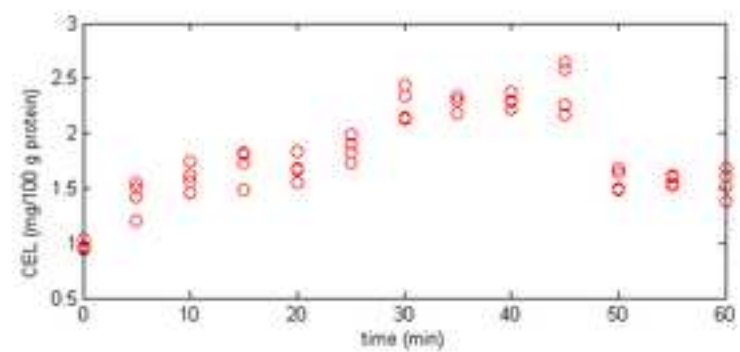
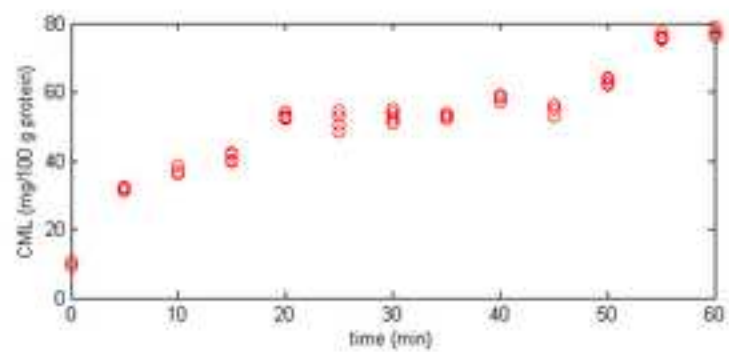
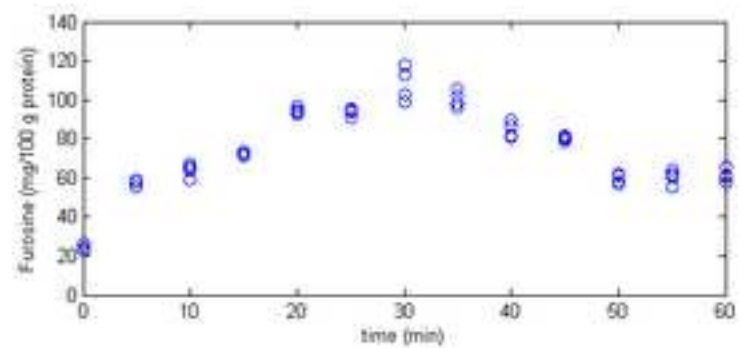
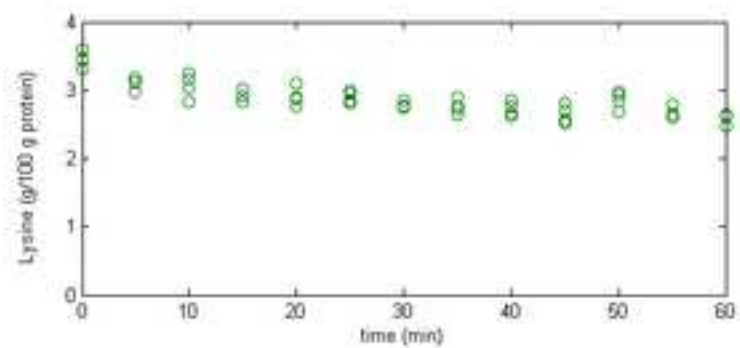
Figure-1

Maillard Reaction

Acidic Hydrolysis







- Tandem mass spectrometry and stable isotope dilution ensured reliable performances.
- The method achieved simultaneous detection of CML, CEL, Lysine and furosine.
- CML, CEL, Lysine and furosine were quantified in several foods.
- The analysis of the four markers paved the way for a better quality control.

ACCEPTED MANUSCRIPT