

RSM approach for stochastic sensitivity analysis of the economic sustainability of a methanol production plant using renewable energy sources

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ABSTRACT

This study aims at investigating the economic viability, at the pre-feasibility level, of a 5MW electrolyser base-methanol production plant, coupled with a PV power plant. The Authors investigated the impact of different parameters, such as the PV plant size, the electrical energy cost and the components capital costs on the methanol production cost and on the system economic viability. It was also analyzed the minimum recommended sale price of the methanol in order to assure an adequate time frame for the return of the investment, considering a different combination of the investigated parameters.

An economic sensitivity analysis, based on the RSM approach, was performed in order to define the most promising economic conditions under which the plant can be considered a profitable investment in terms of ARR. A guide for an economically viable plant design, allowing for the identification of the most suitable combination of the economic parameters, was proposed as a kind of “maps of existence”. For the reference case, the Methanol Production Cost (MPC) resulted around 324 €/ton and the minimum methanol sale price to achieve a PBP of 10 years. The sensitivity analysis identified the cost of electricity and the capital cost of the electrolyser as the most affecting parameters for the system economic viability. In terms of ARR, the methanol price represents the most significant factor. Considering a methanol sale price ranging between 400 and 1200 €/ton, the ARR varied from 5% (20 year of PBP) to 20% (5years of PBP). From the environmental point of view, it is worth underling that the methanol production plant here proposed allows to recycle about 5800 tons of CO₂ per year and to avoid the consumption of about 5.2 MNm³ of NG per year (compared to the traditional production).

Keywords: CO₂ utilization, renewable energy, power to methanol, economic sensitivity analysis, Response Surface

Methodology

NOMENCLATURE

Abbreviation

ANOVA	<i>Analysis of Variance</i>
ARR	<i>Average Rate of Return</i>
CC	<i>Capital Cost</i>
CCD	Central Composite Design
CCS	<i>Carbon Capture Sequestration</i>
DoE	<i>Design of Experiment</i>
FCC	Face-Centered Central Composite
GHG	<i>Greenhouse gas</i>
ICE	<i>Internal Combustion Engine</i>
MPC	<i>Methanol Production Cost</i>
NOx	<i>Nitrogen Oxide</i>

<i>NPV</i>	<i>Net Present Value</i>
<i>PBP</i>	<i>Pay Back Period</i>
<i>PEC</i>	<i>Purchased Equipment Cost</i>
<i>PEM</i>	<i>Polymer electrolyte membrane</i>
<i>PM</i>	<i>Particulate Matter</i>
<i>PV</i>	<i>Photovoltaic</i>
<i>RES</i>	<i>Renewable Energy Sources</i>
<i>RSM</i>	<i>Response Surface Methodology</i>
<i>TCI</i>	<i>Total Capital Investment</i>
<i>TPG</i>	<i>Thermochemical Power Group</i>

27 **1. Introduction**

28 Until today, the scientific community agreed with the fact that the increase of Earth's temperature during the last century is
 29 due to an increase in the GHG emissions as consequence of human activities (Intergovernmental Panel On Climate Change
 30 IPCC, 2014). In particular, the CO₂ emissions represent about 65% of the total GHG emission. Some important actions need to
 31 be taken in order to face the problem of global climate change and, indeed, new technologies need to be developed in order
 32 to reduce the global GHG emission.

33 Beyond hypotheses as evocative as unrealistic about the total replacement of the fossil fuels with RES, still, for several decades
 34 to come, coal, oil, and natural gas will continue to play a primary role in both industrial and civil uses (U.S. Energy Information
 35 Administration, 2017). Therefore, the possibility of producing an innovative fuel with a low environmental impact, as the
 36 methanol, is an option that deserves to be investigated.

37 The methanol production by the CO₂ captured from the flue gases of the fossil-fueled power plant and by the H₂ produced
 38 through the water electrolysis employing the renewable energy is expected to be one of the most promising technologies for
 39 the emission reduction in the next future (Blumberg et al., 2019) (Bozzano and Manenti, 2016).

40 In the last years, the methanol market is significantly grown considering that it is becoming more and more interesting as
 41 electrical energy storage medium, as hydrogen carrier or directly as fuel for transport and power production.

42 The methanol showed great potential as a substitute of the diesel and gasoline for automotive transportation and offers
 43 significant benefits from the environmental impact point of view thanks its "soot-free" combustion and the lower CO₂
 44 emissions compared to the fossil fuels (Zhen and Wang, 2015)

45 The maritime sector has shown in recent years an increasing interest in methanol in place of the traditional fuel to face the
 46 main issue related to the more and more strictly emission regulation. (Ellis et al., 2018; "Methanol Institute," 2018)

47 The methanol synthesis process through CO₂ hydrogenation is rather well known and studies on the reaction mechanism and
 48 catalyst have been carried out in order to investigate the possibility to improve the system conversion and efficiency (Leonzio,
 49 2018). Up to now, the main challenge to the diffusion of this kind of technology is mainly related to its economic feasibility.

50 Several thermo-economic analyses have been proposed in literature considering different potential applications of the
 51 electrolyzer-based methanol synthesis process depending on the electrical energy and CO₂ sources.

52 The integration of the power to methanol plant with a fossil-fueled power plant for the valorization of the CO₂ captured from
 53 the flue gases and the improvement the system flexibility was investigated by Atsonios et al. (2016), and by Bellotti et al. (2019).

54 Szima and Cormos (2018) analyzed the methanol production from CO₂ provided by an industrial plant and H₂ produced
55 employing renewable energy, including a gas turbine and the integration of an ORC cycle to improve the system efficiency.
56 Instead, the potentialities of the methanol as renewable energy storage were analyzed by Matzen et al. (2015) where the
57 methanol was synthesis utilizing hydrogen produced by water electrolyzer power by wind energy and CO₂ supplied by a bio-
58 ethanol plant. The results were compared to the traditional fossil-based process by a multi-decision matrix on the base of
59 economic and sustainability indicators; the renewable-integrated concept gained the highest overall weighted score.

60 Hank et al. (2018) evaluated the production cost of sustainable methanol employing wind energy compared to a grid connect
61 option. All the cited works performed an economic analysis of the system comparing and evaluating the impact of different
62 parameters (energy cost, hydrogen production cost, methanol price, etc..) on the methanol production cost and on the system
63 economic viability. All of them agree that the most critical component is the electrolyzer due to its high capital cost and the
64 significant energy consumption required. Some works report also a sensitivity analysis to evaluate the percentage variation of
65 different economic parameters on the system profitability. Nevertheless, in all the cases, the analysis of effect was limited to
66 qualitative analysis and to a superposition principle.

67 The present work aims to innovate by proposing a different methodology for the sensitivity analysis based on the use of the
68 Response Surface Methodology-RSM approach that allows, for the quantification of the effects of the different parameters on
69 the outputs and their interactions that, in some cases, can result even more effective than the single parameters.

70 The main advantage of such approach is that helps to achieve a more comprehensive description of the problem and a more
71 effective analysis, as already put in evidence by other recent studies conducted by the authors (Bendato et al., 2016) (Bendato
72 et al., 2015).

73 In this study the Authors intend to analyze the economic viability, at the pre-feasibility level, of a 5MW electrolyser based-
74 methanol plant coupled with a PV power plant , by varying some parameters (such as the PV plant size, the electrical energy
75 cost, and components capital costs). The scope is to evaluate their impact on the system feasibility and to identify the most
76 promising conditions. Moreover, the minimum price at which the methanol should be sold in order to assure an adequate time
77 of the return of the investment is investigated considering a different combination of the parameters above mentioned.

78 At first, the analysis has been carried out on a 10MW PV plant as reference case and considering the actual Italian economic
79 scenario, hence the current capital cost of the components and the current market values for the electrical energy purchase
80 and the methanol sale. Then, an economic sensitivity analysis has been performed in order to define the most promising
81 economic conditions under which the plant can represent a profitable investment. Moreover, it was possible to sketch a kind
82 of “maps of existence” that can represent a guide to the economically viable plant design allowing for the identification of the
83 best combination of the economic parameters.

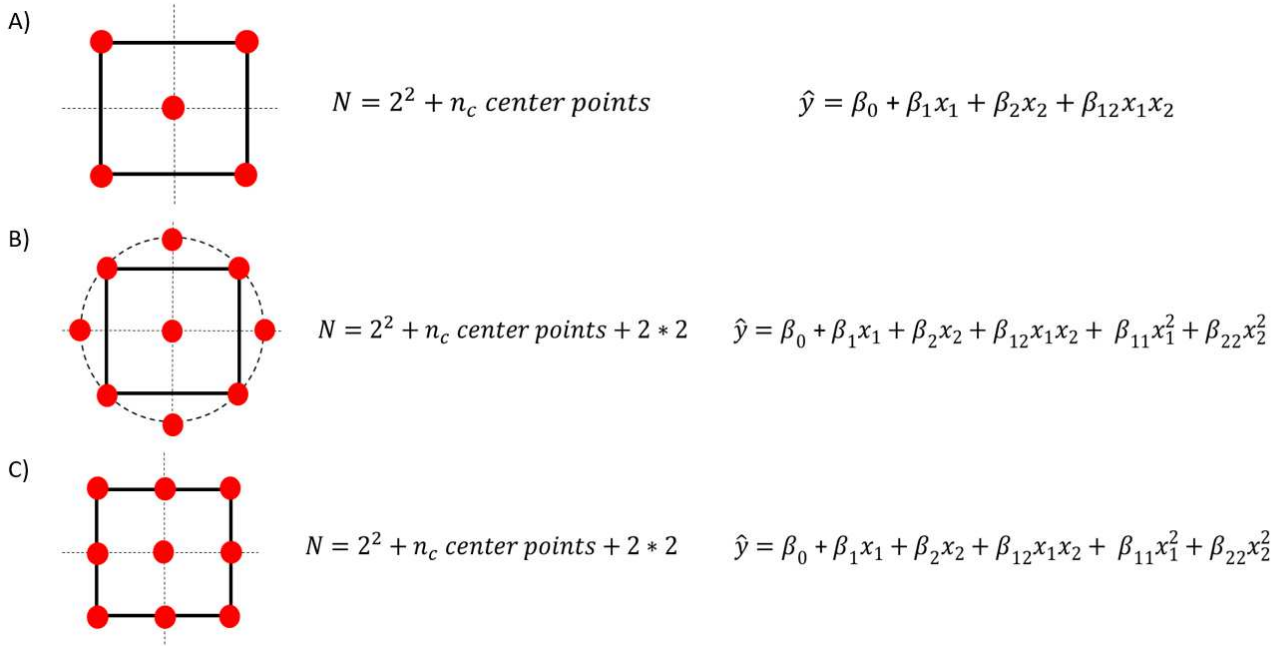
84 The main goal is to provide a comprehensive overview of the problem from the economic standpoint according to an exhaustive
85 sensitivity analysis performed by using the RSM approach, that at the best knowledge of the authors was not already proposed.

86

87 **2. Methodology**

88 The aim of the Design of Experiments (DoE) techniques, is to determine, in stochastic systems, the influence on a selected
89 objective function for one or more independent variables (named factors), varying among different levels or treatments. The
90 significance of such factors is determined through a statistical comparison of the average of the observations under each
91 treatment (Box and Draper, 1987),(Montgomery, 2013). An important evolution of DoE is the so-called Response Surface

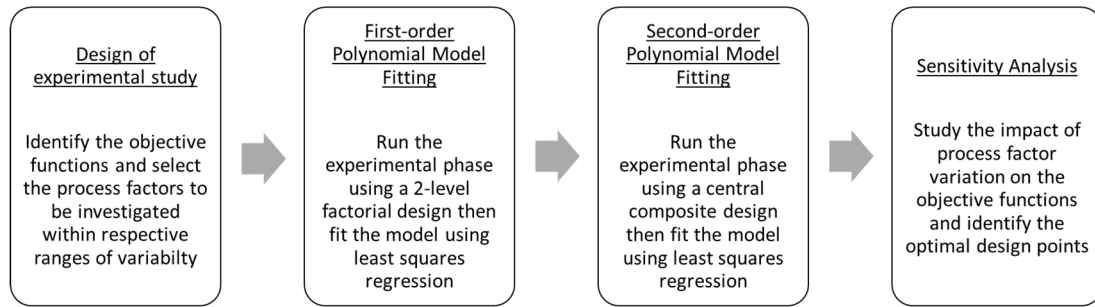
92 Methodology (RSM) that aims to define the optimal design (the grid of candidate points in the experimental region) in order
 93 to build regression models for the objective function.
 94 To fit a first-order regression model, the RSM identifies as best experimental design the Two-Level Factorial Design. to fit
 95 second-order regression models, the Central Composite Design (CCD) or the Face-Centered Central Composite (FCC) design are
 96 adopted. D-optimal, I-Optimal or user-defined designs are suitable to fit higher order regression models. Figure 1 shows the
 97 grid of candidate points, the total numbers of candidate points and the fitted regression model for a two-level factorial design,
 98 a CCD and a FCC in a 2-dimensional experimental region (two factors) are respectively represented.



99
 100 **Figure 1 A) Two-level Factorial Design: grid of candidate points and first-order regression model for a 2-dimensional space; B) Central**
 101 **Composite Design: grid of candidate points and second-order regression model for a 2-dimensional space; C) Face-Centered Composite**
 102 **Design: grid of candidate points and second-order regression model for a 2-dimensional space**

103
 104 Response surface methodology proved to be an adequate modeling tool for the mathematical representation of several
 105 systems and also a useful tool for optimizing process conditions in the industrial behavior. Brown and Brown (2012) used the
 106 RSM approach to optimise the process parameter of an auger reactor for the bio-oil production. Grahovac et al. (2012)
 107 performed the optimization of multiple responses in the context of the ethanol production from thick juice. Applications of the
 108 RSM approach correlated to an economic analysis of an industrial process are reported by Rodrigues et al. (2019), in which a
 109 statistical optimization of the supercritical CO₂ extraction of Eucalyptus bark at industrial scale was performed; the RSM
 110 optimization performed in this work intended to maximize the Total Yield and Productivity, and to minimize the cost of
 111 Manufacturing (COM) and Process Energy of the supercritical fluid extraction process. The analysis was carried out considering
 112 different process factors and three (of the four) responses modeled by RSM (i.e. COM, Productivity, and Process Energy), it
 113 required the knowledge of economic parameters such as capital investment, process costs, and human labor expenses.
 114 Ascough et al. (2013) used RSM to develop an integrated farm-level economic/environmental risk framework for trade-off
 115 analysis between farm profitability and environmental externalities (impacts). The RSM approach in this study uses a surface
 116 regression least squares method to fit linear, quadratic and cross product response combined surfaces. Ekren and Ekren (2008)
 117 used response surface methodology (RSM) in size optimization of an autonomous PV/wind integrated hybrid energy system

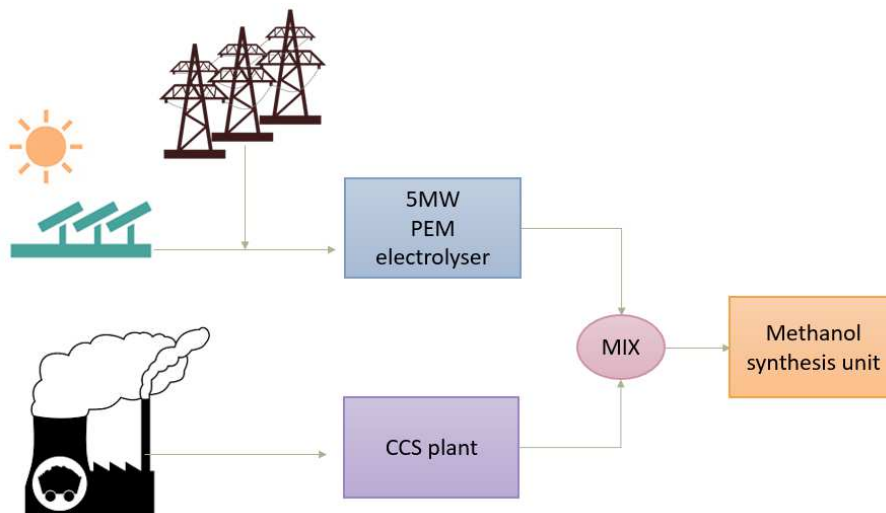
118 with battery storage. In this study the response surface output performance measure is the hybrid system cost and the design
 119 parameters are the PV size, wind turbine rotor swept area and the battery capacity. The optimum result obtained by RSM was
 120 confirmed by using the loss of load probability (LLP) and autonomy analysis.
 121 The main steps followed to build the regression model are outlined in the following diagram.



122
123
124 **Figure 2 Steps for building response surface metamodelling**

125 **3. Plant layout description**

126 The conceptual block diagram of the system under investigation is reported in Figure 3.
 127 The methanol is synthesized from the carbon dioxide captured from the exhaust gas of a coal-fired power plant and the
 128 hydrogen produced by a 5MW PEM water electrolyzer. A PV plant is installed for the methanol plant electrical energy supply.
 129 During the period in which the solar energy is not available, it is assumed to purchase the required electrical energy from the
 130 grid.



131
132 **Figure 3 Simplified plant layout**

133 Below, the main components of the system under investigation are described:

134 Photovoltaic power plant

135 The PV plant is installed for the methanol plant energy supply. The PV panels' average efficiency is assumed to be equal to 18%
 136 with a specific power of about 200 W/m². The PV panels production is calculated on the basis of the average monthly solar
 137 radiation related to the Northern Italy (ENA, 2013). Moreover, the plant equivalent operating hours are set equal to the Italian
 138 average for 2015 (GSE, 2016) of about 1200 hours. Figure 4 shows the average monthly solar radiation for Northern Italy.

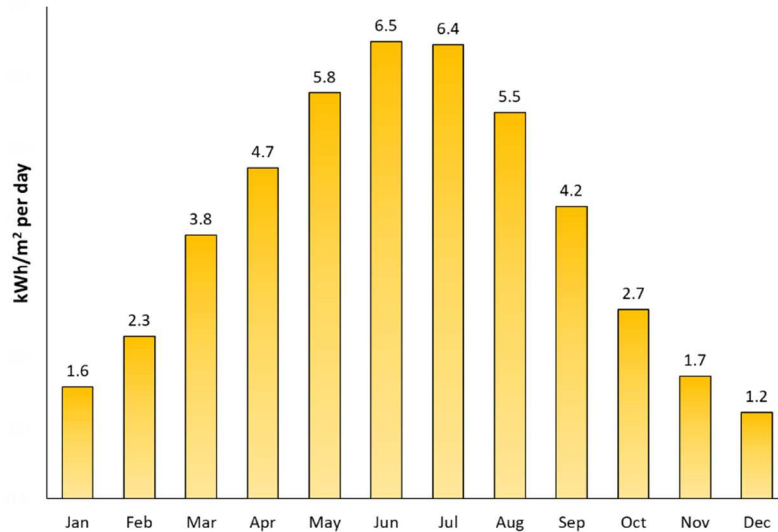


Figure 4 Average monthly insolation curve (ENA, 2013)(GSE, 2016)

According to the previous works from the authors and taking into account the low solar energy availability (about 1200 h_{eq}), the PV plant size needs to be, at least, twice the PEM size, so that the energy produced can be a relevant part of the plant energy balance (at least 25%). Therefore, assuming to install a 5MW PEM electrolyzer, a 10MW PV plant is the minimum size to consider (Rivarolo et al., 2014)(Bellotti et al., 2015)

PEM water electrolyzer

The PEM electrolyzer is a device that produces hydrogen and oxygen throughout the water electrolysis process. The energy consumption is assumed to be equal to about 4.7 kWh/Nm³ of H₂, meaning that for each MWh consumed, about 19 kg/h of H₂ and 152kg/h of oxygen are produced.

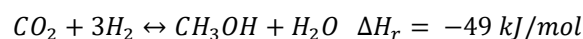
In the system under investigation, the considered PEM size is 5MW which enables a production of about 832 ton/yr of H₂, assuming system availability equal to 95%.

Carbon capture system

The amine-based CCS system is installed in order to sequester the necessary CO₂ for the methanol synthesis from the exhaust gas of a coal-fired power plant. The CO₂ content in the flue gas is assumed to be equal to 19% in mass. The capture efficiency is assumed to be equal to 90% and the thermal and electrical energy consumption of the CCS system is set equal to 3 GJ/tonCO₂ and 110kWh/tonCO₂, respectively. The CCS system is sized in order to be able to capture the required amount of CO₂, hence it is able to process about 3500kg/h of flue gases, sequestering about 700kg/h of CO₂. The CO₂ that exits the CCS is pre-compressed up to 30bar before being mixed with the hydrogen and sent to the methanol synthesis unit.

Methanol synthesis unit

The methanol is synthesized from hydrogen and carbon dioxide, according to the following reaction:



The catalytic reaction is exothermic and takes place in a range of temperature and pressure of 250 – 300 °C and 50 -100 bar on CuO/ZnO/Al₂O₃ as catalyst. In the present work, the H₂ and CO₂ flows are mixed in stoichiometric ratio (1:3) and sent to the reactor for the methanol synthesis. Then, the gaseous products enter the distillation section in order to separate the water and obtain the methanol in liquid form. The reactor conversion efficiency (defined as the ratio between the mass of methanol

167 actually produced and the mass of methanol that can be theoretically produced at the stoichiometric conditions) is assumed
 168 to be equal to 96%.

169 In Table 1, the main technical parameters of the plant component are reported.

170 **Table 1 Main technical parameters** (Rivera-Tinoco et al., 2016)(Van-Dal and Bouallou, 2013)(Jadhav et al., 2014)(Bellotti et al.,
 171 2017)(Mohammad R M Abu-Zahra et al., 2007)

Photovoltaic panels	
Panel average efficiency	18%
Panel specific power	196 W/m ²
Equivalent operating hours	1200 h
PEM Electrolyser	
Electrical consumption	4.7 kWh/Nm ³ di H ₂
Pressure	30 bar
Efficiency	75%
PEM availability	95%
Carbon Capture system	
Treatment kind	Amines (MEA) (30%)
Flue gases inlet T[°C] and p[bar]	40°C, 2bar
Thermal energy consumption per tonne of CO₂	3 GJ _{th} /kgCO ₂
CO₂ outlet temperature[°C] pressure[bar]	40°C, 2 bar
CO₂ capture rate	90%
Methanol Reactor	
Working Pressure	80 bar
Temperature	240 °C
Recirculation factor of unreacted syngas	0.85
Conversion efficiency	96%
Molar H₂ : CO₂ ratio	3:1

172

173 In Table 2, the energy and mass balance of the plant is reported, assuming a 10MW PV plant. The overall energy consumption
 174 includes both the PEM electrical energy demand and the auxiliaries (i.e. compressors, pumps).

175

Table 2 Main thermodynamic results

Annual 10MW PV plant production	12000 MWh
Annual electrical energy purchased from the grid	33771 MWh
Annual electrical energy consumption	45771 MWh/yr
Annual methanol production	4047 ton /yr
Annual Oxygen production	6324 ton/yr

176

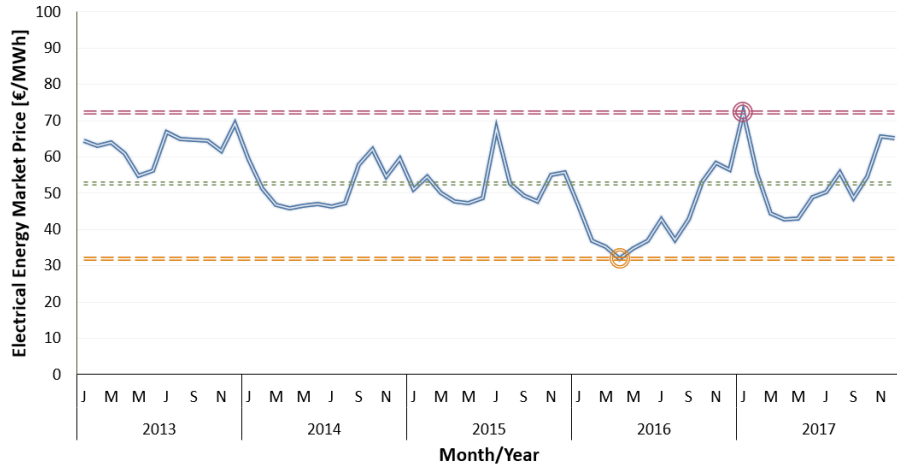
177 **4. Economic assumption**

178 The Italian economic scenario is taken as reference for the thermo-economic analysis. The main economic
179 assumptions referred to the base case are reported in the following.

180 Electrical energy cost

181 As demonstrated in previous works of the authors (Rivarolo et al., 2014), considering the low capacity factor of the PV plant,
182 the sole use of the renewable energy is not sufficient to assure an adequate exploitation rate of the plant; it is, therefore,
183 necessary to purchase energy from the grid when electricity from the PV plant is not available. Hence the cost of the electrical
184 energy represents a term of primary importance for the economic feasibility of the plant under investigation. In Figure 5, the
185 monthly average electrical energy market price between 2013 and 2017 is reported. The prices range between 30€/MWh and
186 70€/MWh and the average value is equal to about 50€/MWh (“GME - Gestore dei Mercati Energetici SpA,” 2018). The same
187 range of values is used in the sensitivity analysis in order to investigate the influence of the energy cost.

188 In the reference case analysis, the electrical energy for the electrolyzer supply is assumed to be purchased from the grid at 30
189 €/MWh.



190

191

192 **Figure 5 Italian monthly average electrical energy market price between 2013 and 2017** (“GME - Gestore dei Mercati Energetici SpA,”
193 2018)

194 Methanol selling price

195 The methanol selling price depends on the economic scenario where the plant is going to operate; in this analysis, the European
196 market is chosen as a target for the methanol sale. The average European Posted Contract Price of methanol between 2013
197 and 2018 is about 350 €/ton, fluctuating in the range of 225 and 450 €/ton and the mode is around the 370€/ton as reported
198 in (“Methanex Corporation,” 2018)(“Methanol Market - Global Industry Analysis, Size, Share, Growth, Trends and Forecast
199 2017 - 2026,” 2017). Moreover, the average Non-Discounted Reference price in the same period is about 415€/ton. For
200 simplicity and in consideration of the previous works from the authors (Bellotti et al., 2019, 2017), the methanol price for the
201 reference case is assumed equal to 400€/ton.

202 Oxygen selling price

203 As already discussed in previous works, the sale of oxygen co-produced by the electrolyzer is crucial for the methanol plant
204 economic feasibility. The oxygen selling price is assumed 150 €/ton, which represents the minimum selling price for the medical

205 use of oxygen (Intratec, 2018). It is worth noting that the oxygen purity produced by electrolyzer (>99.9%) is sufficient for
 206 medical and industrial applications, therefore no further purification treatments are needed. (EIGA and ASSOCIATION, 2015)

207 Purchased equipment cost estimation

208 The total Purchased Equipment Cost (PEC) is the sum of the capital cost of each plant component calculated in accordance to
 209 the cost functions reported in Table 3. The cost functions are extrapolated from literature data, applying the cost-capacity
 210 method or directly from private communication with the manufacturer (Mohammad R.M. Abu-Zahra et al., 2007; Asif et al.,
 211 2018; International Renewable Energy Agency (IRENA), 2018; Pérez-Fortes et al., 2016; “Private Communications by
 212 Hydrogenics,” 2018).

213

Table 3 Main components cost functions

Photovoltaic plant	$C_{PV} = 1822.6 \cdot (P_{inst[kW]})^{0.85}$
PEM Electrolyser	$C_{PEM} = 1.3 \cdot 10^6 (P_{inst[MW]})^{0.815}$
CCS system	$C_{CCS} = 4855.2 (M_{in [kg/h]})^{0.65}$
Methanol synthesis unit	$C_{MR} = 7106.85 (M_{in [kg/h]})^{0.7}$

214

215 Total Capital Investment cost estimation

216 The Total Capital Investment (TCI) cost is calculated starting from the PEC of the plant: it is assumed that the PEC is about the
 217 45% of the TCI (Mohammad R.M. Abu-Zahra et al., 2007). Moreover, it is assumed that the TCI corresponds to the Initial
 218 Investment.

219 Plant lifetime

220 The plant lifetime is assumed to be equal to 20 years, considering the lifetime of the electrolyzer (“Private Communications by
 221 Hydrogenics,” 2018) and PV plants, which represent the most expensive plant components.

222 In this analysis the economic parameters such as inflation, interest rate and taxation are not considered, for simplicity, because
 223 the main purpose of the work is to evaluate the relative effect of some parameters over the economic feasibility of the system
 224 under investigation.

225 The economic indicators considered are the following:

- 226 • The methanol production cost: it is useful to define the minimum methanol sale price that needs to be applied in order to
 227 guarantee a positive cash flow and it is calculated in accordance with the following equation:

$$MeOH_{cost} = \frac{\text{annual fixed cost} + \text{annual net variable cost}}{\text{annual methanol production}} \quad [€/ton_{MeOH prod}] \quad (1)$$

228 where the annual fixed cost is the annual rate of the TCI that is calculated over the 20years of the plant lifetime and the
 229 annual net variable costs are the electrical energy purchase cost, net of the income, coming from the sale of the oxygen
 230 at 150€/ton.

- 231 • The PayBack Period calculated in accordance with the following equation:

$$PBP = \frac{\text{Initial Investment}}{\text{Annual Cash inflow}} \quad (2)$$

232 where the annual cash inflow is assumed to be constant over the plant lifetime and it is calculated as follow:

$$\text{Annual Cash inflow} = \text{Annual O}_2 \text{ income} + \text{Annual MeOH income} - \text{Annual variable cost} \quad (3)$$

Since the plant lifetime is assumed to be equal to 20yr, 10yr of PBP is chosen as the threshold for the plant economic viability.

- The *Average Rate of Return* (ARR) can be used as an alternative to the PBP parameter to evaluate the plant feasibility. The ARR divides the average profit by the initial investment, to get the expected ratio of return.

In this case the ARR is a percentage value calculated as the reciprocal of the PBP and the threshold value is set equal to 10%.

$$ARR = \frac{1}{PBP} \times 100 \quad [\%] \quad (4)$$

5. Reference Case results

At first, the analysis is carried out considering a reference case represented by a 10MW PV plant and a 5MW PEM electrolyzer based methanol plant. The resulting TCI is equal to about 25 M€. The PEC percentage distribution between the main plant components is reported in Figure 6.

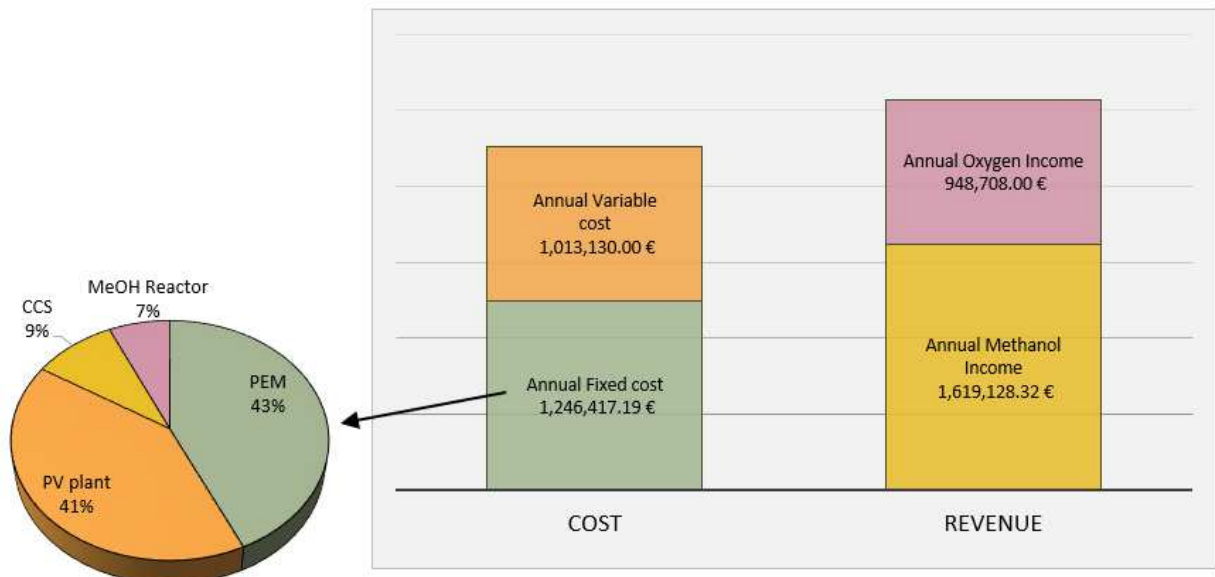


Figure 6 Reference case - Cost and revenues comparison and annual fixed cost percentage breakdown

The most expensive components are the PEM electrolyzer and the PV plant, that represent 43% and 41% of the PEC respectively. The CCS and Methanol unit costs are comparable and lower than 10% of the PEC.

Considering the strong impact of the PEM cost on the total capital cost and considering that the electrolyzer is the component with the lowest technology readiness level, it is reasonable to assume that its capital cost will decrease in the next future. For this reason, the percentage reduction in electrolyzer capital cost is taken into account as a parameter for the sensitivity analysis.

For the reference case, the MPC (Methanol Production Cost) results around 324 €/ton and the PBP and the ARR are equal to about 15.7% and 6.5% respectively. Therefore, on the base of the economic assumption presented above, the reference case presents a PBP higher than 10 years, meaning that the plant cannot be defined cost effective. However, considering the

254 environmental constraints, it is possible to perform a sensitivity analysis in order to identify the parameters that mostly affect
255 the plant economic feasibility and to define the minimum value of the methanol sale price, for achieving a PBP equal to 10
256 years at least.

257 6. Sensitivity Analysis Results And Discussion

258 In the present work, a sensitivity analysis is carried out in order to evaluate the influence of some parameters on the economic
259 feasibility of the methanol production plant. In particular, the methanol production cost and the ARR are investigated in
260 function of a number of parameters:

- 261 - PV plant size;
- 262 - Electrical energy purchasing cost;
- 263 - Percentage reduction of PEM electrolyzer capital cost;
- 264 - Methanol sale price.

265 The methanol sale price range is chosen taking into consideration the results coming from the methanol production cost
266 analysis.

267 The RSM approach and the ANOVA technique have been used to perform a sensitivity analysis, aimed at the evaluation of the
268 impact on the plant economic viability in consideration of different economic parameters. The statistical analysis and graphical
269 analysis of the data were performed by using Design Expert software (Version 10.0, Stat-Ease, USA). The analysis of variance
270 (ANOVA) was selected to assess the statistical significance of the effects, by using Fisher's test. The Lack of Fit F-test was used
271 to evaluate the goodness of the fit of the regression models.

272 In the following, the results related to the methanol production cost and the ARR analyses are reported.

273 Methanol production cost

274 In Table 4 the range of the parameters considered for the Methanol Production Cost (MPC) analysis are reported.

275 **Table 4 Methanol production cost analysis factors and respective ranges**

Factor	Name	Unit	Low level	Midrange level	High level
A	PV plant size	MW	10	15	20
B	Percentage reduction of PEM capital cost	%	0	25	50
C	Electrical energy purchasing cost	€/MWh	30	50	70

276
277 The resulting factorial design for the MPC sensitivity analysis is a 2^3 design with a center point, represented by a cube in the R^3
278 space. The vertices of the cube represent the experimental points that must be tested.

279 For each combination of the factors ($2^3 + 1$), three replications of the calculated MPC are taken into account, by considering
280 the minimum, average and maximum annual solar radiation.

281 The ANOVA analysis showed that the MPC passed the F test on Regression and, therefore, the first-order model can be
282 considered to be a satisfying approximation of the problem.

283 The test of the residual normality, throughout the "Residual vs predicted plot" (reported in **Error! Reference source not**
284 **found.**), showed that no transformation of the response is needed: the points on the plot appeared to be randomly scattered
285 around zero, so it was reasonable to assume that the error terms had a mean of zero. The vertical width of the scatter did not
286 appear to increase or decrease across the fitted values, so the Authors could assume that the variance in the error terms was
287 constant.

288

289 In Figure 7, the ANOVA results for the MPC are reported.

Analysis of variance table [Partial sum of squares - Type III]

Source	Sum of Squares	df	Mean Square	F Value	p-value	Prob > F
Regression model	1.026E+006	4	2.564E+005	539.66	< 0.0001	significant
A-PV size	27051.52	1	27051.52	56.94	< 0.0001	significant
B-Reduction %CC	52545.13	1	52545.13	110.60	< 0.0001	significant
C-Cost of energy	9.037E+005	1	9.037E+005	1902.31	< 0.0001	significant
AC	42185.30	1	42185.30	88.80	< 0.0001	significant
Curvature	4.67	1	4.67	9.820E-003	0.9215	
Residual	21378.28	45	475.07			
Lack of Fit	0.000	11	0.000	0.000	1.0000	not significant
Pure Error	21378.28	34	628.77			
Cor Total	1.047E+006	50				

290

291

Figure 7 MPC analysis – ANOVA results

292

The ANOVA table reports that the chosen regression model (the first-order model) is significant, meaning that the model is a good representation of the problem under investigation. Moreover, the ANOVA analysis shows that the most significant terms are A, B, C and the interaction term AC, while the other interactions can be neglected, as reported in Table 5.

293

294

Table 5 highlights that the factor that mainly affects the methanol production cost is the Cost of Energy (factor C) with a percentage contribution higher than the 86%.

295

296

297

Table 5 MPC analysis – “Effect results”

Term	Stdized Effect	Sum of Squares	% Contribution
A-PV size	-47.48	27051.52	2.58
B-Reduction %CC	-66.17	52545.13	5.02
C-Cost of energy	274.43	9.04E+05	86.33
AB	-3.30E-12	0	0
AC	-59.29	42185.3	4.03
BC	1.70E-12	0	0
ABC	-1.70E-12	0	0
Pure Error		21378.28	2.04

298

299

The MPC Objective function in terms of coded factors is reported below:

$$MPC_{coded} = 433.64 - 23.7397 * A - 33.0861 * B + 137.215 * C - 29.6456 * AC \quad (5)$$

300

By default, the high levels of the factors are coded as +1 and the low levels of the factors are coded as -1. The coded equation is useful for identifying the relative impact of the factors by comparing the factor coefficients and their own sign.

301

302

The equation presents a constant term, three terms related to the single factors and an interaction factor. The constant term represents the MPC value, corresponding to the center of the design space (all factor equal to 0). Looking at the coefficients in

303

304 absolute value of the single factors, it is possible to note that the most influent factor is C (the electrical energy cost), with a
 305 coefficient equal to 137.215. In terms of percentage contribution on the response, the C-factor counts more than 86%. The
 306 positive sign indicates that increasing the C value the cost of methanol increases. The second factor in terms of magnitude is
 307 B, the percentage reduction of the PEM capital cost. In this case, the B-sign is negative, meaning that increasing the B value
 308 the MPC is reduced. Factor A has a negative coefficient (as C) but with a lower absolute value. Moreover, the influence of A is
 309 lower than the influence of its interaction with C. The coefficient of the AC term is 30.91 with negative sign, meaning that the
 310 higher is the product AC, the lower is the MPC value and vice versa. The interaction term makes the regression surface not flat
 311 but presents a slight twist, as shown in Figure 8.

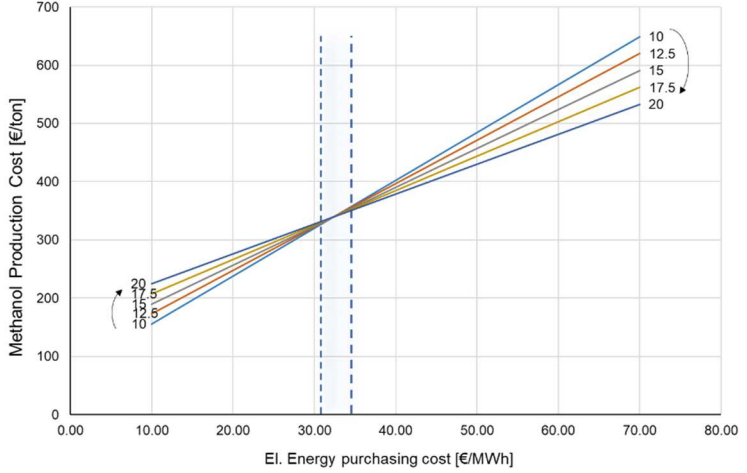
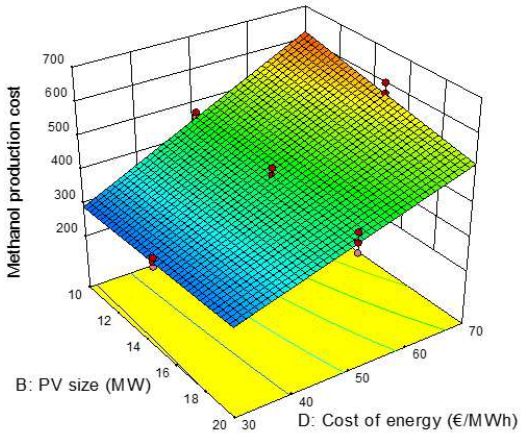


Figure 8 Methanol production cost as function of the factors B and D, for A and C equal to the midterm value **Figure 9 Methanol production cost as function of the electrical energy cost and for different value of the PV plant size [MW]**

312 The effect of the interaction is shown in Figure 9. For low values of the electrical energy cost, the MPC increases with the PV
 313 size, therefore, the impact on the cost of the PV plant PEC is higher than the cost related to the electrical energy purchase.
 314 Vice versa, for high electrical energy cost, the MPC decreases with the PV size. It is interesting to note the presence of a break-
 315 even area where the methanol production cost results almost constant for each PV plant size. The electrical energy purchase
 316 cost at the intersection corresponds to the value for which the total cost (made of the TCI and the electrical purchase) is the
 317 same in the intersecting cases. In other words, the lower is the PV size, the lower is the TCI, and the lower gets the annual
 318 amount of produced energy, hence, the higher is the amount of the purchased electrical energy and vice versa. For example,
 319 considering 10MW and 20MW PV plant installed, the value of electrical energy cost for which the total costs are equivalent
 320 (and hence the MPC results constant) is equal about to 32.38€/MWh.

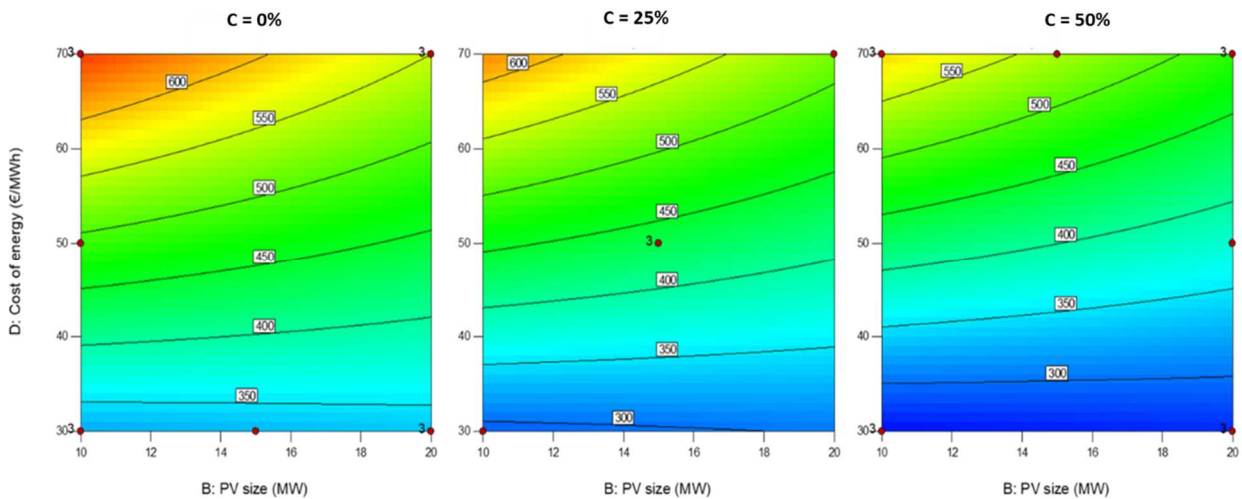


Figure 10 MPC as function of the PV size and the electrical energy cost for different values of the PEM capital cost reduction

The methanol production cost results in the range 200 ÷ 600 €/ton. Therefore, for the following analysis, the methanol sale price is assumed to be in the range 400 ÷ 1200 €/ton.

Pay Back Period

In order to evaluate the PBP, the methanol sale price shall be included as parameter in the analysis. Therefore, in this case, the sensitivity analysis was performed considering four variables. In Table 6, the variation range of the four considered parameters is reported.

Table 6 Methanol production cost analysis factors and respective ranges

Factor	Name	Unit	Low level	Midrange level	High level
A	Methanol selling price	€/ton	400	800	1200
B	PV plant size	MW	10	15	20
C	Percentage reduction of PEM capital cost	%	0	25	50
D	Electrical energy purchasing cost	€/MWh	30	50	70

The resulting factorial design for the PBP sensitivity analysis is a 2⁴ design with a center point, represented by a hypercube in the R⁴ space.

The PBP passes the F test on Regression and, therefore, the first-order model is capable of providing a satisfying approximation of the problem. However, the response-normality check and the constant error check show that a transformation of the response is needed: the “residual vs predict plot” shows the existence of a specific values-pattern (Figure 11).

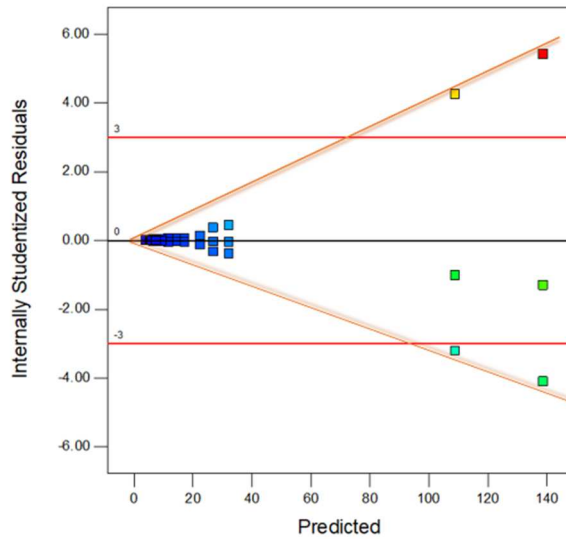


Figure 11 PBP analysis - Residual vs Predicted plot

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337

338 In particular, the mean residual does not change with the predicted values, but the spread of the residual increases
 339 proportionally to the predicted values. In accordance with the BOX-COX analysis, an inverse transformation of the response is
 340 applied as follows:

$$y' = \frac{1}{y} \quad (6)$$

341

Because the inverse of the PBP is indeed the ARR, in the following, the analysis will be performed in reference to the ARR.

342

Average Rate of Return

343

Similarly to the PBP, the ARR passes the F test on regression as shown in Figure 12.

Analysis of variance table [Partial sum of squares - Type III]						
Source	Sum of Squares	df	Mean Square	F Value	p-value Prob > F	
Regression Model	2504.90	10	250.49	1881.36	< 0.0001	significant
A-MeOH price	2069.77	1	2069.77	15545.50	< 0.0001	
B-PV size	15.29	1	15.29	114.85	< 0.0001	
C-Reduction %CC	62.43	1	62.43	468.93	< 0.0001	
D-Cost of energy	246.06	1	246.06	1848.08	< 0.0001	
AB	49.03	1	49.03	368.29	< 0.0001	
AC	22.34	1	22.34	167.81	< 0.0001	
BC	2.64	1	2.64	19.79	< 0.0001	
BD	32.25	1	32.25	242.23	< 0.0001	
CD	2.99	1	2.99	22.44	< 0.0001	
ABC	2.09	1	2.09	15.70	0.0003	
Residual	5.33	40	0.13			
Lack of Fit	0.72	6	0.12	0.88	0.5195	not significant
Pure Error	4.61	34	0.14			
Cor Total	2510.23	50				

344

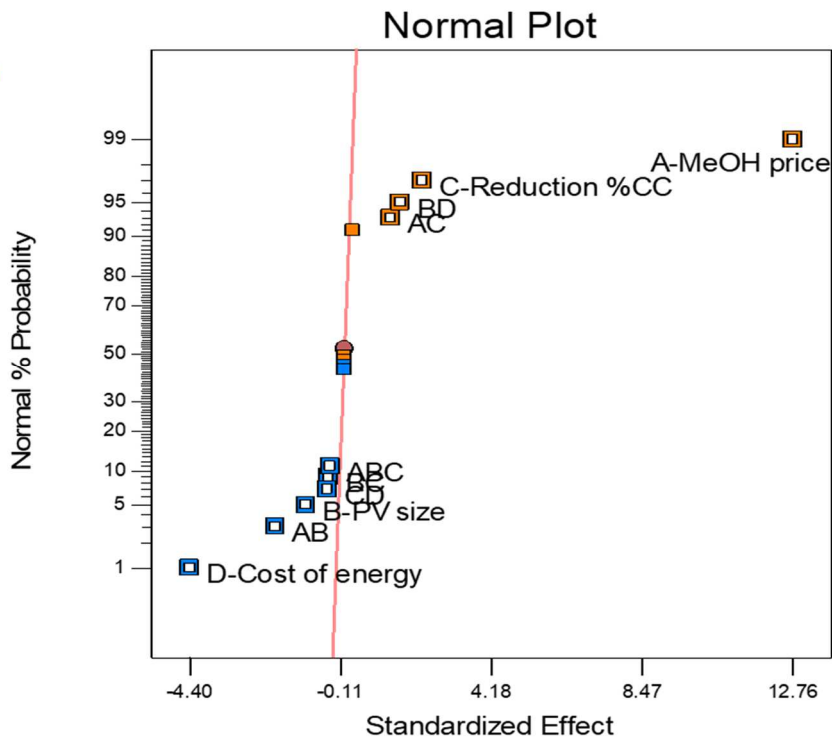
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Figure 12 ARR analysis – ANOVA results

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347 The “Lack of Fit F-value” of 0.88 implies that it is not significant, meaning that the regression model well-fits the problem. The
 348 model standard deviation and mean values are 0.36 and 10.47, respectively. The “residual vs predicted plot” shows that the
 349 constant error assumption can be confirmed.

350 The “effect analysis” based on the normal plot (Figure 13) shows that the significant model terms are A, B, C, D, AB, AC, BC, BD,
 351 CD, ABC. The only two-factor interaction resulting not significant is AD, which represents the interaction between the methanol
 352 sale price and the electrical energy cost. The most affecting term is A, with a contribution impact higher than 82%. The second
 353 most influent factor is D, with a percentage contribution of around 9.8%. Among the interaction terms, the most affecting are
 354 AB and BD with a percentage contribution of 1.95 % and 1.28 %, respectively. It is interesting to note that the factor B presents
 355 a percentage contribution value lower than both the interaction terms above mentioned.



356
 357 **Figure 13 ARR – Normal Plot Effect analysis**

358
 359 The regression model for the ARR, in terms of coded factors, is reported below.

$$ARR_{coded} = 10.77 + 6.34 * A - 0.55 * B + 1.13 * C - 2.24 * D - 1.00 * AB + 0.67 * AC - 0.23 * BC + 0.82 * BD - 0.23 * CD - 0.21 * ABC \quad (7)$$

360 The regression model for ARR, in terms of actual factors, is as follow:

$$ARR = 5.6664 + 0.0202635 * MeOH\ price - 0.155065 * PV\ size - 0.00598245 * Reduction\ \%CC - 0.223686 * Cost\ of\ energy - 0.000401009 * MeOH\ price * PV\ size + 0.000130836 * MeOH\ price * Reduction\ \%CC + 0.00146484 * PV\ size * Reduction\ \%CC + 0.00819699 * PV\ size * Cost\ of\ energy - 0.000499007 * Reduction\ \%CC * Cost\ of\ energy - 4.17407e - 006 * MeOH\ price * PV\ size * Reduction\ \%CC \quad (8)$$

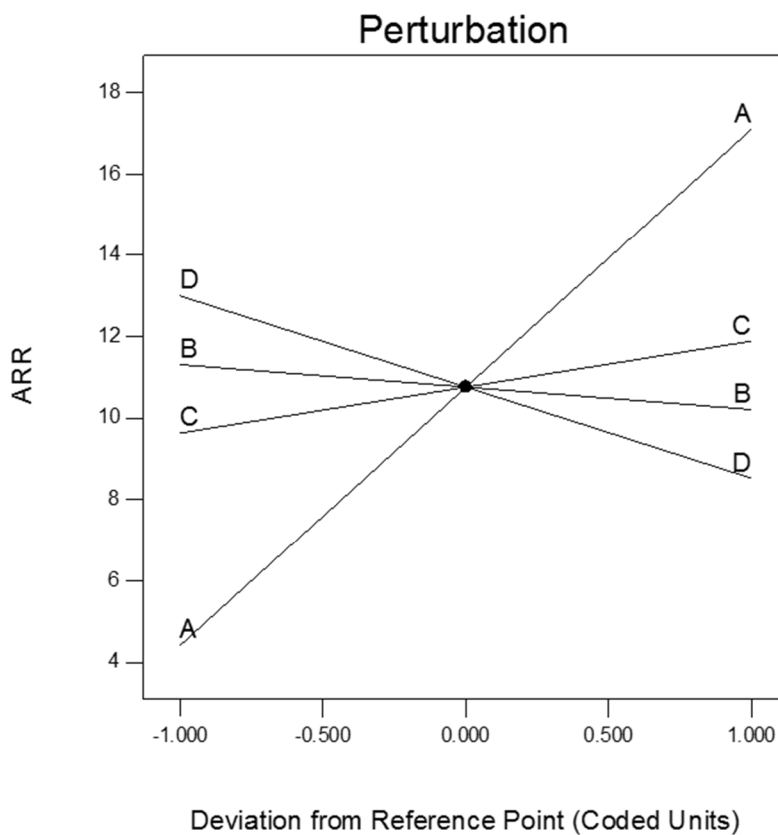
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362 The equation, in terms of actual factors, can be used to make predictions about the response, for given levels of each factor.

363 In Figure 14 the perturbation plot is reported. It shows the effects of all factors at the midpoint of the design. It is possible to
 364 note that while increasing the value of the factors A and D, the ARR consequently increases; vice versa, the increasing value of
 365 the factors B and C have a negative impact on the ARR.

Design-Expert® Software
 Factor Coding: Actual
 ARR

Actual Factors
 A: MeOH price = 800
 B: PV size = 15
 C: Reduction %CC = 25
 D: Cost of energy = 50



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Figure 14 ARR – perturbation factors plot

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369 In Figure 15, the response surfaces of the ARR, as function of factors A and B, A and C, B and C, C and D, are reported.

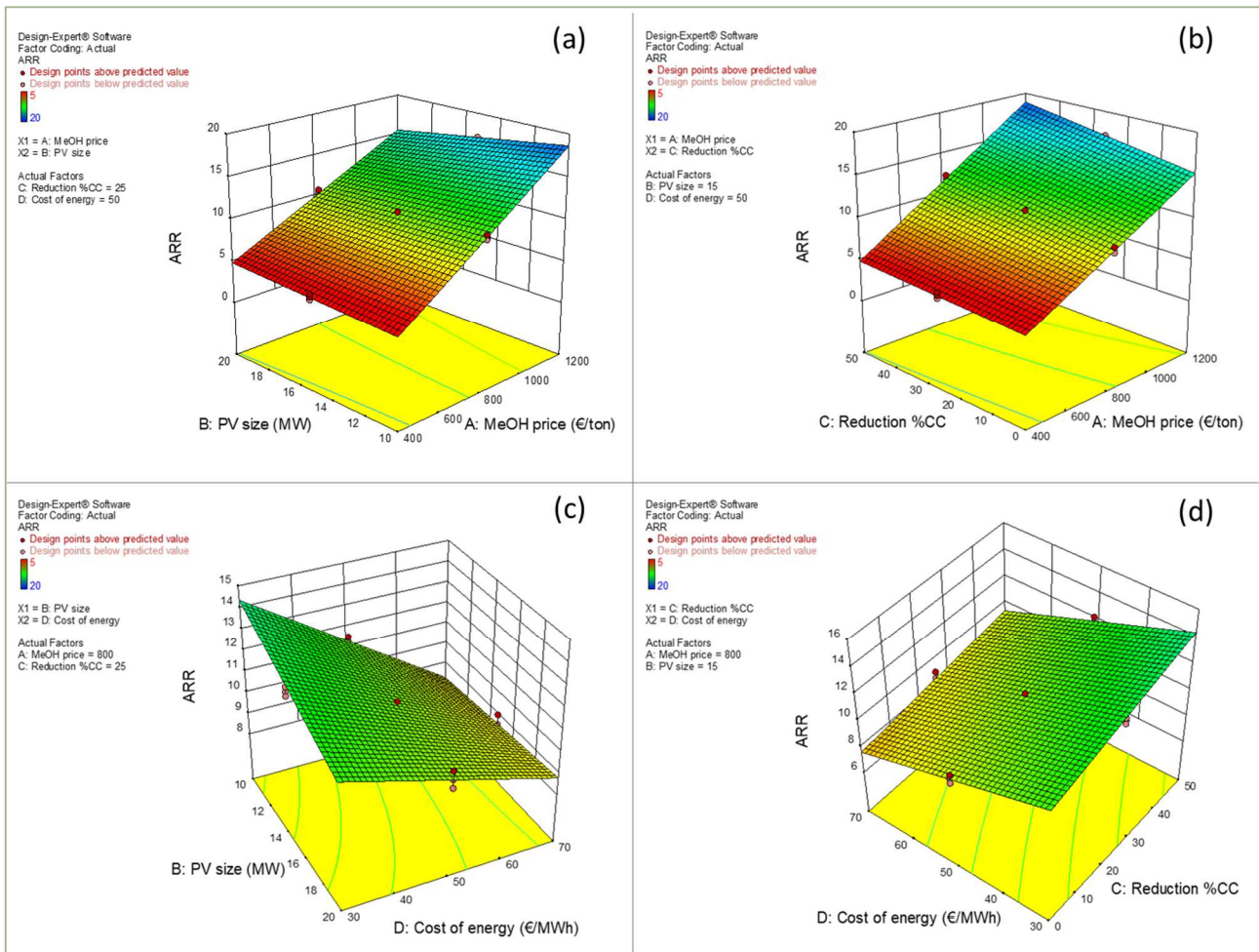


Figure 15 Response surfaces of ARR as function of the factors (a) A and B, (b) A and C, (c) B and D, (d) C and D.

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One of the most interesting outputs of the RSM, for the economic analysis, is the possibility to draw a map of the ARR values as function of the different factors.

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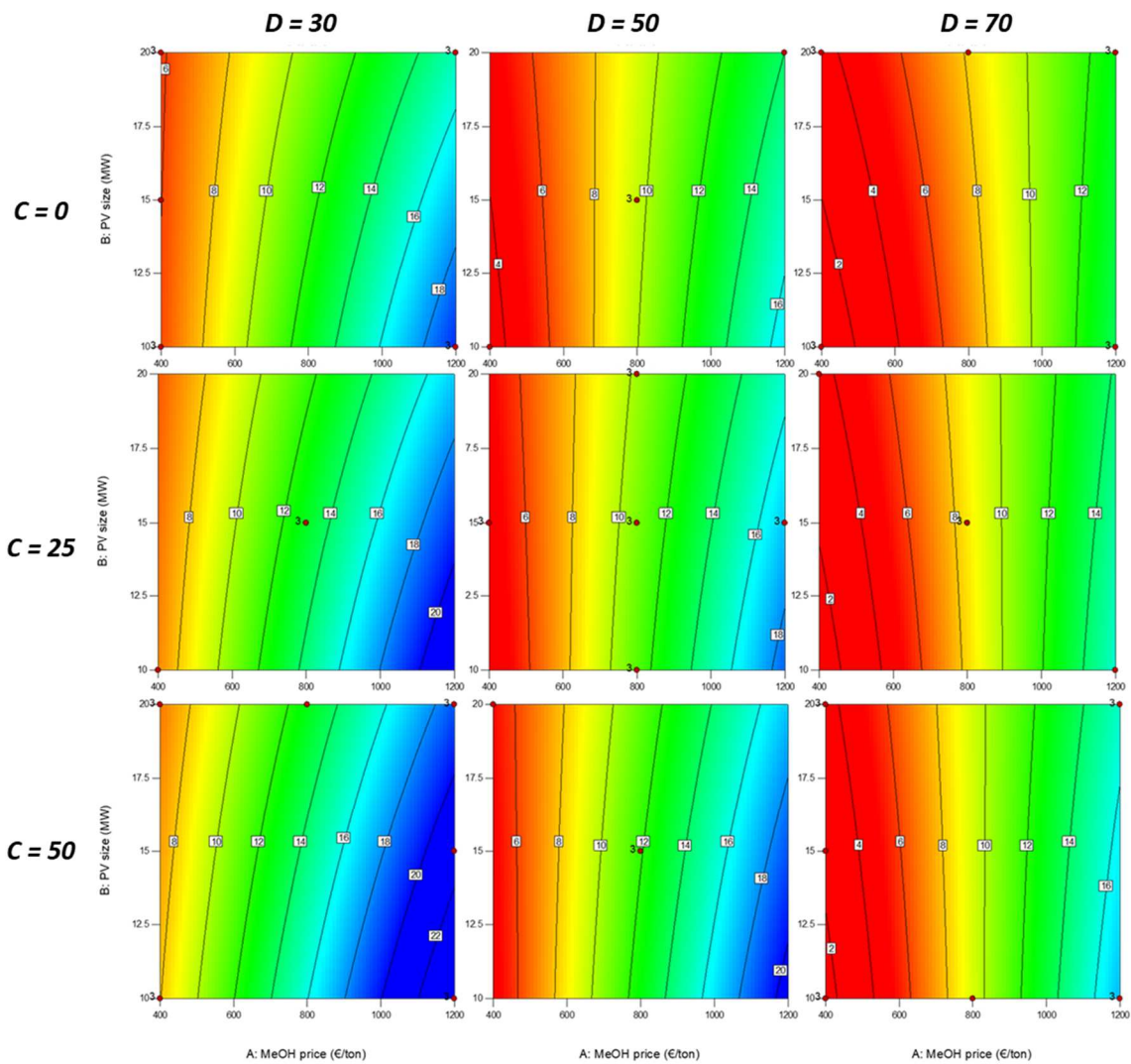
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Figure 16 reports a matrix of contour graphs of the ARR as function of the PV size (B) and the Methanol sale price (A) in dependence of the low, middle and high level of the Electrical energy purchasing cost (D) and the percentage reduction of the capital cost of the PEM (C).

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Figure 16 ARR contours matrix plots as function of the factors A (MeOH price) and B (PV size) for different values of factors C (PEM %CC reduction) and D (Energy cost)

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It is possible to define the minimum price for methanol sale, that allows for an ARR equal to 10% (i.e. 10 yrs. of PBP, that is set as the lowest acceptable value) taking into account the scenario constraints such as the electrical energy purchase cost. For example, having fixed the PV size equal to 20MW and not considering the reduction in PEM cost,, the ARR results equal to 10% for C= 30 €/MWh and A around 750 €/ton, or for C = 70 €/MWh and A is around 950 €/ton. Furthermore, it is possible to define the optimal PV size as function of the electrical energy price: it is worth noting that for a fixed A, at low value of D, the ARR decreases (i.e. PBP increases); for increasing PV size, instead, the trend diverts for high value of D (see also Figure 17). On the other hand, having fixed the PV size, the minimum value of A increases while increasing the electrical energy cost. This trend is more visible for C equal to zero and tends to fade reducing the capital cost of the PEM. Finally, the increase of C allows for a reduction of the methanol-sale price for the same values of D and B.

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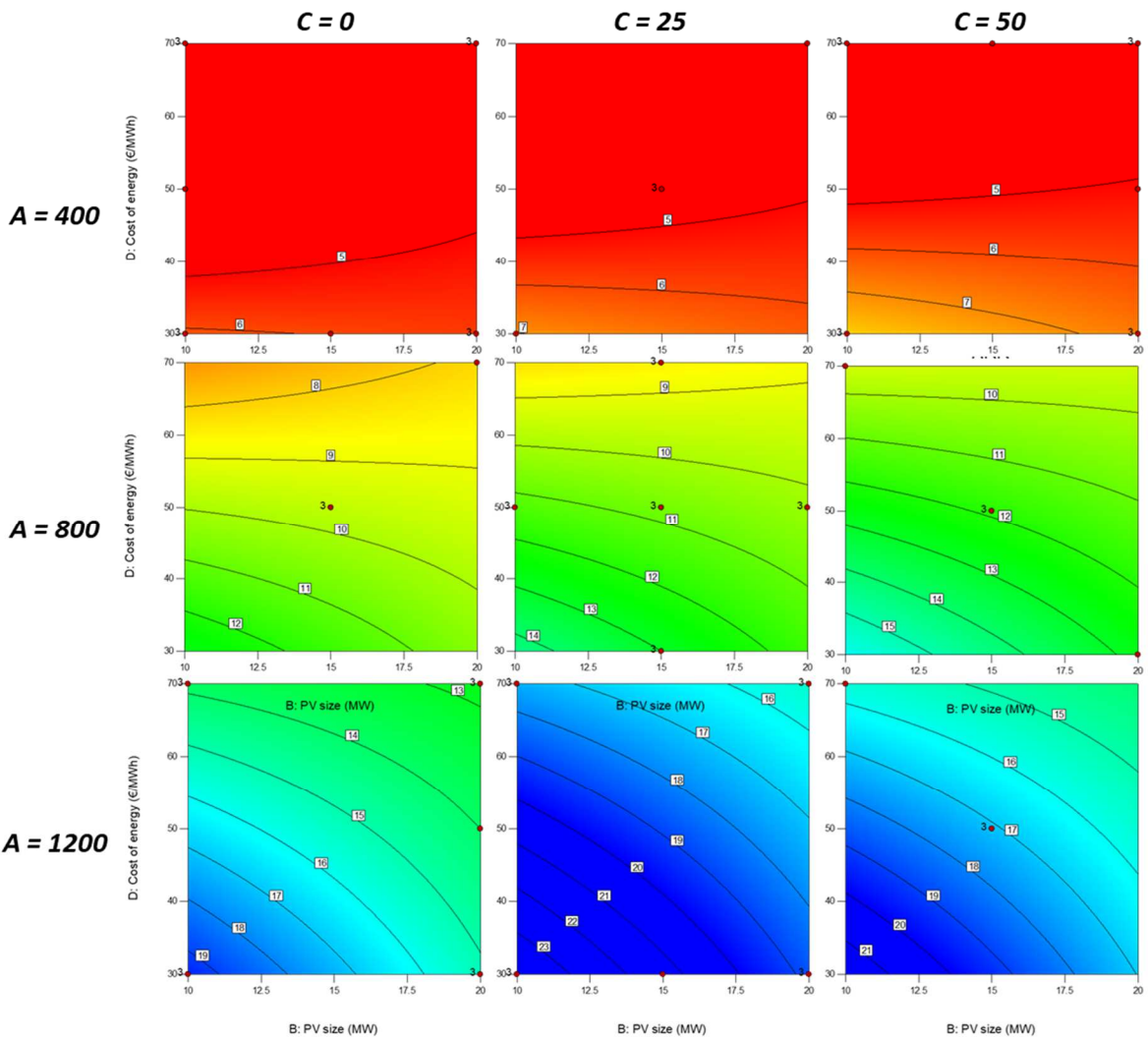
For example, having fixed the PV size equal to 10 MW and the electrical energy cost to 30€/MWh (as in the reference case), the values of A that make the ARR equal to 10% are 625€/ton for C=0%, 580€/ton for C=25%, and 500€/ton for C=50%. Hence, a reduction in the capital cost of PEM of 50% allows for a reduction in methanol sale price of the 20%. Or, in other words, for

394

395 a fixed value of the methanol sale price equal to 625€/ton, the reduction of the PEM capital cost of the 50% allows for an
 396 increase of 25% of the ARR, resulting equal to about 12.5% (i.e. 8 yrs. of PBP).

397 However, if the electrical energy cost increases up to 70 €/MWh, for the same PV size (10MW) and not considering the capital
 398 cost reduction, the minimum methanol sale price rises from 625 up to 975€/ton. Nevertheless, if the PV size increases up to
 399 20MW, the methanol sale price decreases to 950€/ton. Moreover, if the option to reduce the PEM capital cost to 50 % is
 400 considered, the lower value of A results equal to 827 €/ton, with 10MW of PV plant, against the 835€/ton with 20MW of PV
 401 plant. It is worth noting that the values of the methanol sale price above mentioned are rather high compared to the actual
 402 market value. Nevertheless, it cannot be forgotten that the methanol produced in this kind of plant has a low environmental
 403 impact, being synthesized by using wasted CO₂ and renewable energy.

404 In Figure 17, the ARR map as function of the PV plant size (factor B) and the electrical energy cost (factor D) is reported for
 405 different values of the methanol sale price (factor A) and of the percentage reduction of the PEM capital cost (factor C).



406
 407 **Figure 17 ARR contours matrix plots as function of the factors B (PV size) and D (energy cost)**
 408 **for different values of factors A (MeOH price) and C (PEM %CC reduction)**
 409

410 In Figure 18, the single plots of the ARR, as function of each factor used in the sensitivity analysis, are reported.

Design-Expert® Software
 Factor Coding: Actual
 ARR
 — 95% CI Bands
 - - 95% PI Bands
 - 95% TI Bands (p=99%)

Actual Factors
 A: MeOH price = 800
 B: PV size = 15
 C: Reduction %CC = 25
 D: Cost of energy = 50

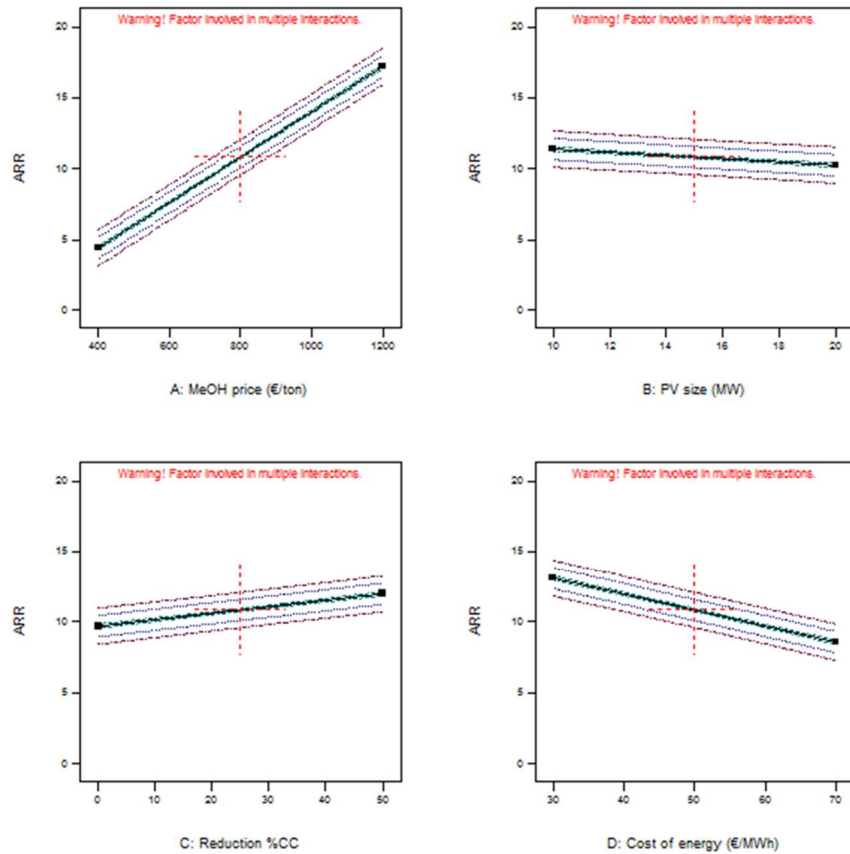


Figure 18 ARR as function of single factor

7. Conclusions

In the present work, the economic viability of a 5MW electrolyser base-methanol production plant, coupled with a PV power plant, was investigated by the use of the RSM approach. At first, a preliminary analysis on a reference case was performed, in order to identify the components' cost that most affect the economic viability of the plant under investigation. Afterwards, the RSM approach was used to perform a sensitivity analysis, that would lead to evaluate the capacity of the three main design variables (the PV plant size, the electrical energy purchasing cost, a percentage reduction in the PEM electrolyser capital cost) to affect the methanol production cost and, then, including the methanol sale price as variable, the ARR. The "maps of existence" created by using the RMS approach are one of the most important outcomes of the study as they may represent an useful baseline for an economically viable plant design.

The main results of the study are summarised below:

- From the capital cost point of view, the PEM electrolyser and the PV plant resulted to be the most influent elements (43% and 41% of the total investment cost, respectively); while the relevance of the CCS system and the methanol synthesis unit resulted marginal, representing an overall value lower than the 20% of the total cost;

- 429 • Considering the high electrical energy consumption of the electrolyser, the variable that most affects the
430 methanol production cost is the electrical energy purchasing cost, accounting the 86% of the total;
- 431 • In the case with an installed PV plant of 10MW, assuming 30€/MWh as cost of energy purchased from the
432 grid and considering the current cost of the electrolyser, the minimum methanol sale price able to ensure
433 a 10 years PBP, resulted 625 €/ton;
- 434 • In case that the cost of the electrolyser was reduced by 25% and 50%, the minimum methanol sale price
435 would have decreased to around 600 €/ton and 500€/ton, respectively.

436 The main implications of the study are the following:

- 437 • It is of utmost importance to continue the research for improving the electrolyser technology, in order to
438 achieve a significant reduction in the capital cost or a relevant increase in efficiency;
- 439 • The amount of energy produced by the PV plant is strictly dependent on the panel's efficiency. Therefore, future
440 studies on the materials designated to the solar radiation capture, can lead to an increase in energy efficiency
441 and, hence, in energy production. For example, for the same m² of installed panels, an increase of the 20% in
442 actual efficiency (18%) allows for an increase in energy production and the PBP can be reduced down to 25%;
- 443 • The same result can be obtained, for the same panel efficiency, in terms of an increase of the equivalent
444 operating hours, by choosing an installation site with a higher daily solar irradiation (as it changes significantly
445 with the latitude).

446 The methanol production cost resulted basically higher than the actual market value, but it could be justified
447 considering that the methanol has proved to be a valuable low carbon alternative to the diesel fuel in the
448 automotive transportation sector. Considering that, in the European scenario, the actual diesel fuel market average
449 value is around 1.5€/l (about 1.8€/kg) ("Global Petrol Prices," 2018), assuming the energy equivalence, the resulting
450 methanol sale price is about 860€/ton.

451 In the end, it is worth to underline that the methanol produced with this method allows to recycle more than 5800
452 ton/yr of CO₂ that can be recovered from the industrial plants. Nonetheless, compared to the traditional natural
453 gas-based production chain, this concept allows for saving about 5.2 MNm³ of NG and for avoiding the related
454 emission of about 10000 ton/yr of CO₂

455 **ACKNOWLEDGMENTS**

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