



UNIVERSITI PUTRA MALAYSIA

***PERFORMANCE OF CHITOSAN FROM MUSHROOM AS
BIOCOAGULANT AGENT FOR KAOLIN AND PALM OIL
MILL EFFLUENT WASTEWATER***

ODAY ADNAN IRHAYYIM

FK 2017 8



**PERFORMANCE OF CHITOSAN FROM MUSHROOM AS
BIOCOAGULANT AGENT FOR KAOLIN AND PALM OIL
MILL EFFLUENT WASTEWATER**

By

ODAY ADNAN IRHAYYIM

**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,
in Fulfillment of the Requirements for the Degree of Master of Science**

January 2017

COPYRIGHT

All material contained within the thesis, including without limitation text, logos, icons, photographs, and all other artwork, is copyright material of Universiti Putra Malaysia unless otherwise stated. Use may be made of any material contained within the thesis for non-commercial purposes from the copyright holder. Commercial use of material may only be made with the express, prior, written permission of Universiti Putra Malaysia.

Copyright© Universiti Putra Malaysia



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the Degree of Master of Science

**PERFORMANCE OF CHITOSAN FROM MUSHROOM AS
BIOCOAGULANT AGENT FOR KAOLIN AND PALM OIL
MILL EFFLUENT WASTEWATER**

By

ODAY ADNAN IRHAYYIM

January 2017

Chairman : Associate Professor Zurina binti Zainal Abidin, PhD
Faculty : Engineering

This project aims to investigate the potential use of commercial chitosan produced from mushroom (CMs) as bio-coagulants which are water-soluble chitosan (WSC) and acid-soluble chitosan (ASC) in wastewater treatment. Palm Oil Mill Effluent (POME) is wastewater generated from the palm oil milling industry must be treated properly before being discharged into the environment. The use of inorganic coagulants like alum can potentially lead to the spread of chronic diseases due to the residual content of the coagulants in the treated wastewater. Thus, this study searched for an alternative coagulant using (CMs) for coagulation process. In this study, the optimum condition for synthetic water kaolin treatment when WSC was at pH 8, with a dosage of 10 mg/L, and with almost 100% turbidity removal. While ASC optimum conditions were at pH 11, dosage 10 mg/L and almost 100% turbidity removal. On the other hands, the performances of CMs coagulants as bio-coagulants were compared to aluminum sulphate (alum) coagulant. The results showed that alum gave 93% turbidity removal at optimum pH 4 and with an optimum dosage of 1200 mg/L.

The treatment of POME wastewater by using CMs as a bio-coagulant was successfully able to reduce COD with up to 73% in value, 70% of BOD reduction and 99% of TSS removal, whereas alum coagulant resulted in 71% of COD, 65% of BOD reduction and 86% of TSS removal. The second objective of this study was to examine the characterization of the WSC and ASC by using Fourier Transform Infrared (FTIR) analysis, Zeta Potential, and antimicrobial activity. The results of FTIR confirmed the existence of amino groups in the backbone of CMs. The results of Zeta Potential analysis showed that the surface charges for WSC and ASC were $40.46 \text{ mV} \pm 1.01$ and $+70.24 \text{ mV} \pm 3.69$, respectively. The antimicrobial activity analysis showed that ASC is more active than WSC where the concentration used of CMs was 10%. The ASC was active to inhibit microbes within around 7 mm zone which is considered clear zone of inhibition. The estimated cost for treatment using WSC, ASC and alum was RM 5.5, 4.52 and 84 per cubic meter respectively of POME treated wastewater.

In conclusion, this study has proven that chitosan mushroom possesses a significant potential to be used as a bio-coagulant for wastewater treatment and also economical.



Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia sebagai memenuhi keperluan untuk Ijazah Master Sains

KECEKAPAN CHITOSAN MUSHROOM SEBAGAI AGEN BIO KOAGULAN ON KEKERUHAN DIANGKAT DARJAT SISA

Oleh

ODAY ADNAN IRHAYYIM

Januari 2017

Pengerusi : Profesor Madya Zurina binti Zainal Abidin, PhD
Fakulti : Kejuruteraan

Projek ini bertujuan untuk mengkaji potensi chitosan yang dihasilkan oleh cendawan (CMs) sebagai agen bio-pengental iaitu chitosan larut-air (WSC) dan chitosan larut-asid (ASC) dalam rawatan air sisa. POME merupakan air sisa yang terhasil daripada industri minyak sawit dan mesti dirawat dengan sebaiknya sebelum dilepaskan ke persekitaran. Penggunaan pengental bukan organik seperti tawas telah membawa kepada penyebaran penyakit kronik akibat kesan kandungan sisa bahan pengental dalam air sisa yang dirawat. Oleh itu, kajian ini adalah bertujuan untuk mencari alternatif menggunakan CMs dalam proses pengentalan. Dalam kajian ini, keadaan optimum untuk rawatan air kaolin sintetik menggunakan WSC pada pH 8, dengan dos sejumlah 10 mg/L, dan dengan hampir 100% penyingkiran kekeruhan. Manakala keadaan optimum menggunakan ASC adalah pada pH 11, dengan dos sejumlah 10 mg/L, dengan hampir 100% penyingkiran kekeruhan. Pada masa yang sama, kebolehan pengental CMs sebagai bio-pengental telah dibandingkan dengan pengental aluminium sulfat (tawas). Keputusan menunjukkan bahawa tawas hanya memberikan 93% penyingkiran kekeruhan pada keadaan optimum pH 4 dengan dos optimum 1200 mg/L. Rawatan POME dengan menggunakan CMs sebagai bio-pengental telah berjaya mengurangkan COD sehingga 73%, 70% pengurangan BOD dan 99% pengurangan TSS manakala bio-pengental tawas menunjukkan hanya 71% COD, 65% pengurangan BOD dan 86% pengurangan TSS.

Objektif kedua kajian ini pula adalah untuk memeriksa ciri-ciri WSC dan ASC dengan menggunakan analisis Spektroskopi Fourier Menukar Sinaran Merah (FTIR), Potensi Zeta dan aktiviti antimikrob. Keputusan FTIR telah mengesahkan kewujudan kumpulan-kumpulan amino dalam tulang belakang CMs. Keputusan analisis Potensi Zeta menunjukkan bahawa cas permukaan untuk WSC dan ASC masing-masing ialah $40.46 \text{ mV} \pm 1.01$ dan $+70.24 \text{ mV} \pm 3.69$. Analisis aktiviti antimikrob pula menunjukkan bahawa ASC lebih aktif daripada WSC dimana kepekatan CMs yang digunakan adalah 10%. ASC aktif untuk merencat mikrob dalam zon lingkungan 7 mm yang dikira sebagai zon bebas perencatan. Anggaran kos untuk rawatan

menggunakan WSC, ASC dan tawas adalah RM 5.5, 4.52 dan 84 bagi setiap satu meter padu air sisa POME yang dirawat. Kajian ini telah membuktikan bahawa cendawan chitosan mempunyai potensi yang signifikan dalam penggunaannya sebagai bio-pengental serta kos yang efektif untuk merawat air sisa.



ACKNOWLEDGEMENTS

I would like to express my deep gratitude and thanks to my supervisor Prof. Madya Dr. Zurina bt. Zainal Abidin for her guidance, encouragement, advice and financial support throughout my studies, her valuable suggestion for the experimental arrangements, and her patience during the correction of the manuscript.

Next, I would like to express my deep gratitude to the Ministry of Science and Technology of Iraq - Oil Research Center for the Scholarship which given to me for pursuing this study.

I would also like to thank my friend Moothanna Al-qubaisi for his help and assistance in various phases during my studies. Thanks to all staffs and students of the Department of Chemical and Environmental Engineering.

Finally, I would like to express my thanks to my mother, my wife and my sisters in Iraq for their patience, ever constant encouragement and love during my studies.

I certify that a Thesis Examination Committee has met on 31 January 2017 to conduct the final examination of Oday Adnan Irhayyim on his thesis entitled "Performance of Chitosan from Mushroom as Biocoagulant Agent for Kaolin and Palm Oil Mill Effluent Wastewater" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Master of Science.

Members of the Thesis Examination Committee were as follows:

Rozita binti Omar, PhD

Senior Lecturer
Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Shafreeza binti Sobri, PhD

Senior Lecturer
Faculty of Engineering
Universiti Putra Malaysia
(Internal Examiner)

Rubiyah Baini, PhD

Associate Professor
Universiti Malaysia Sarawak
Malaysia
(External Examiner)



NOR AINI AB. SHUKOR, PhD
Professor and Deputy Dean
School of Graduate Studies
Universiti Putra Malaysia

Date: 22 March 2017

This thesis was submitted to the Senate of the Universiti Putra Malaysia and has been accepted as fulfillment of the requirement for the degree of Master of Science. The members of the Supervisory Committee were as follows:

Zurina binti Zainal Abidin, PhD

Associate Professor
Faculty of Engineering
Universiti Putra Malaysia
(Chairman)

Azni bin Hj Idris, PhD

Professor
Faculty of Engineering
Universiti Putra Malaysia
(Member)

Suryani binti Kamarudin, PhD

Senior Lecturer
Faculty of Engineering
Universiti Putra Malaysia
(Member)

ROBIAH BINTI YUNUS, PhD

Professor and Dean
School of Graduate Studies
Universiti Putra Malaysia

Date:

Declaration by graduate student

I hereby confirm that:

- this thesis is my original work;
- quotations, illustrations and citations have been duly referenced;
- this thesis has not been submitted previously or concurrently for any other degree at any institutions;
- intellectual property from the thesis and copyright of thesis are fully-owned by Universiti Putra Malaysia, as according to the Universiti Putra Malaysia (Research) Rules 2012;
- written permission must be obtained from supervisor and the office of Deputy Vice-Chancellor (Research and innovation) before thesis is published (in the form of written, printed or in electronic form) including books, journals, modules, proceedings, popular writings, seminar papers, manuscripts, posters, reports, lecture notes, learning modules or any other materials as stated in the Universiti Putra Malaysia (Research) Rules 2012;
- there is no plagiarism or data falsification/fabrication in the thesis, and scholarly integrity is upheld as according to the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) and the Universiti Putra Malaysia (Research) Rules 2012. The thesis has undergone plagiarism detection software

Signature: _____ Date: _____

Name and Matric No: _____

Declaration by Members of Supervisory Committee

This is to confirm that:

- the research conducted and the writing of this thesis was under our supervision;
- supervision responsibilities as stated in the Universiti Putra Malaysia (Graduate Studies) Rules 2003 (Revision 2012-2013) were adhered to.

Signature: _____
Name of Chairman
of Supervisory
Committee: Associate Professor Dr. Zurina binti Zainal Abidin

Signature: _____
Name of Member
of Supervisory
Committee: Professor Dr. Azni bin Hj Idris

Signature: _____
Name of Member
of Supervisory
Committee: Dr. Suryani binti Kamarudin

TABLE OF CONTENTS

	Page
ABSTRACT	i
ABSTRAK	iii
ACKNOWLEDGEMENTS	v
APPROVAL	vi
DECLARATION	viii
LIST OF TABLES	xiii
LIST OF FIGURES	xiv
LIST OF ABBREVIATIONS	xvii
CHAPTER	
1 INTRODUCTION	1
1.1 General Overview	1
1.2 Problem Statement	2
1.3 Research Objectives	3
1.4 Scope of Research	4
1.5 Thesis Organization	4
2 LITERATURE REVIEW	5
2.1 Wastewater	5
2.2 Water Quality	5
2.2.1 Chemical Properties	7
2.2.2 Physical Properties	9
2.2.3 Biological Characterization	10
2.3 Coagulation	11
2.3.1 Coagulation Theory Mechanism	13
2.4 Factors Affecting Coagulation	17
2.4.1 pH	18
2.4.2 Turbidity	18
2.4.3 Coagulating Materials	18
2.4.3.1 Metal Coagulants	18
2.4.3.2 Inorganic Coagulants	19
2.4.3.3 Organic Coagulants	20
2.4.4 Coagulant Dosage and Rapid Mixing Degree	22
2.4.5 Sedimentation Time	23
2.5 Mushroom	23
2.6 Chitosan	24
2.6.1 Chitosan Preparation and Production	26
2.6.2 Chitosan Applications	27
2.6.3 Chitosan as a Coagulant in Wastewater Treatment	28
2.6.4 Chitosan Characteristics	30
2.7 Palm Oil Mill Effluent Wastewater (POME)	31
2.8 Summary	32

3	METHODOLOGY	33
3.1	Materials	33
3.1.1	Chemicals	33
3.2	Wastewater Sample	34
3.2.1	Preparation of Synthetic Kaolin Wastewater	34
3.2.2	Palm Oil Mill Effluent (POME) Wastewater	35
3.3	Initial Characterization of Palm Oil Mill Effluent (POME) Wastewater	35
3.4	Preparation of Coagulants	35
3.4.1	Preparation of Chitosan Coagulant	35
3.4.2	Preparation of Alum Coagulant	36
3.5	Coagulation Experiment of Water-Soluble Chitosan (WSC), Acid-Soluble Chitosan (ASC) and Alum	36
3.5.1	Synthetic Kaolin Wastewater	36
3.5.2	Palm Oil Mill Effluent Wastewater (POME)	38
3.5.2.1	Main Coagulation Experiment	38
3.6	Effect of Storage Time	39
3.7	Effect of Storage Condition	40
3.8	Analytical Methods of Treated Water	40
3.8.1	Turbidity	40
3.8.2	pH	40
3.8.3	COD	40
3.8.4	BOD	41
3.8.5	TSS	41
3.8.6	Sludge Volume	42
3.9	Chitosan Characterization	42
3.10	Summary	43
4	RESULTS AND DISCUSSION	44
4.1	Chitosan Characterization	44
4.1.1	FTIR Analysis	44
4.1.2	Zeta Potential	46
4.1.3	Antimicrobial Activity	47
4.2	Coagulation Experiment by Using Synthetic Kaolin Wastewater	49
4.2.1	Effect of pH	49
4.2.2	Effect of Dose	53
4.2.3	Comparison WSC and ASC with Alum	56
4.2.4	Comparison WSC and ASC with Chitosan Originated from Invertebrate	59
4.3	Initial Wastewater POME Characterization	60
4.4	Coagulation Experiment with POME Wastewater	60
4.4.1	Coagulation by Chitosan Mushroom	60
4.4.2	Alum	66
4.4.3	Final Characterization of Treated POME	68
4.5	Efficiency Degradation of Chitosan	73
4.5.1	Effect of Storage Time	73
4.5.2	Effect of Storage Condition	74
4.6	Cost of treatment by CM and alum	75

5	CONCLUSIONS AND RECOMMENDATIONS	76
5.1	Conclusions	76
5.2	Recommendations	77
	REFERENCES	78
	APPENDICES	91
	BIODATA OF STUDENT	104



LIST OF TABLES

Table		Page
2.1	Malaysia water quality standard a and b	6
2.2	Sedimentation of small particles	12
2.3	Wide applications of chitosan	28
2.4	The previous work of using chitosan in water and wastewater treatment.	29
4.1	Peaks with different intensity were detected by FTIR analysis for ASC and WSC	46
4.2	Antimicrobial activity of the chitosan obtained from mushroom WSC and ASC	48
4.3	Initial characteristics of the wastewater palm oil mill effluent (POME)	60

LIST OF FIGURES

Figure		Page
1.1	The distribution of total world water	1
2.1	Coagulation of fine particles	11
2.2	Double layer compression	14
2.3	Charge Neutralization	15
2.4	Chain of polymer connects to several colloid particles	16
2.5	Colloids become enmeshed in the growing precipitate	17
2.6	Coagulants categories	19
2.7	Fungi structure	24
2.8	Chemical structure of chitosan	25
2.9	Invertebrates are a significant source of chitosan	25
2.10	Fungi (e.g. mushroom) is another source of chitosan	26
2.11	Deacetylation of chitin to obtain chitosan	27
3.1	Summary of procedure for conducting the study	33
3.2	WSC and ASC specifications	34
3.3	Diagram of jar testing device	38
4.1	Infrared spectroscopy for WSC	45
4.2	Infrared spectroscopy for ASC	45
4.3	Photograph of antimicrobial test results of ASC (labeled as no. 2) and WSC (labeled as no. 3) against <i>P. aeruginosa</i> . No. 1 was standard	48
4.4	Percentage of turbidity removal against pH of synthetic kaolin wastewater using WSC	50
4.5	(a) Floccs formation during coagulation process at pH 5 (b) Floccs formation during coagulation process at optimum pH8 10 mg/l and 30 minutes of sedimentation time	51

4.6	Percentage of turbidity removal against pH of synthetic kaolin wastewater using ASC	52
4.7	(a): Synthetic wastewater kaolin before treatment; and (b): Synthetic wastewater kaolin after treatment 10 mg/l, 30 minutes' sedimentation time	54
4.8	Turbidity removal against coagulant dosage of WSC (mg/L) of synthetic kaolin wastewater	55
4.9	Turbidity removal vs. dose of ASC for kaolin wastewater	56
4.10	Turbidity removal % using alum vs initial pH for kaolin solution	57
4.11	Turbidity removal % using alum vs. dose for kaolin solution	57
4.12	Initial pH vs. final pH of WSC, ASC and alum for kaolin wastewater	58
4.13	Effect of pH on the removal of turbidity in POME treatment for WSC and ASC	61
4.14	(a) Wastewater POME turns into darker color at higher pH value with poor turbidity removal. (b) POME sample at lower pH value with significant removal	62
4.15	Effect of WSC and ASC dosage on the removal of suspended solids in POME pretreatment	64
4.16	(a): Real wastewater POME before treatment; and (b): Real wastewater POME after treatment, pH 3, dose 20 mg/l, sedimentation time 30 min	65
4.17	Effect of pH for alum coagulant in POME pretreatment	66
4.18	Effect of alum dosage on turbidity reduction	67
4.19	COD reduction from POME using WSC and ASC	68
4.20	TSS reduction from POME using WSC and ASC	69
4.21	Effect of alum coagulant dosage on COD and TSS removal of POME	69
4.22	Effect of the coagulants used on the reduction of COD, BOD and TSS in POME pretreatment	70
4.23	Effect of the final pH of the treated POME by using different coagulant	71

4.24	Sedimentation time of WAS, ASC, and alum coagulant	72
4.25	Sludge volume for WSC, ASC and alum coagulant during POME wastewater treatment	73
4.26	Effect of storage time for the WSC and ASC performances in POME treatment	74
4.27	Percentage of turbidity removal by different type of coagulants used due to the effect of the storage condition	75



LIST OF ABBREVIATIONS

CMs	Water-soluble chitosan and acid-soluble chitosan
WSC	Water-soluble chitosan
ASC	Acid-soluble chitosan
POME	Palm Oil Mill Effluent
Alum	Aluminum Sulphate
COD	Chemical Oxygen Demand
BOD	Biological Oxygen Demand
TSS	Total Suspended Solids
FTIR	Fourier Transform Infrared
TS	Total Solids
N	Total Nitrogen
P	Phosphorus
VOC	Volatile Organic Compounds
SS	Suspended Solids
DO	Dissolved Oxygen
K _w	Dissociation Constant
NOM	Natural Organic Matter
DADMAC	Diallyl Dimethyl Ammonium Chloride
PAC	Polyaluminium Chloride
MW	Molecular Weight
DD	Degree of Deacetylation
Ctnp	Mushroom Chitosan Nanoparticle

CHAPTER 1

INTRODUCTION

1.1 General Overview

Water is definitely the main secret of life. Global water consists of approximately 95.6% oceans, 2.5% fresh water, and the rest is saline water; in addition, the surface water is only about 1.2% of the total freshwater of the earth. (figure 1.1) (Gleick, 1993). The water crisis has become one of the current issues nowadays. The population growing led to a reduction in clean water sources, thereby threatens a numerous people; for instance, the report of (Unicef, 2008) estimated that one billion people or more are unable to obtain safe water.

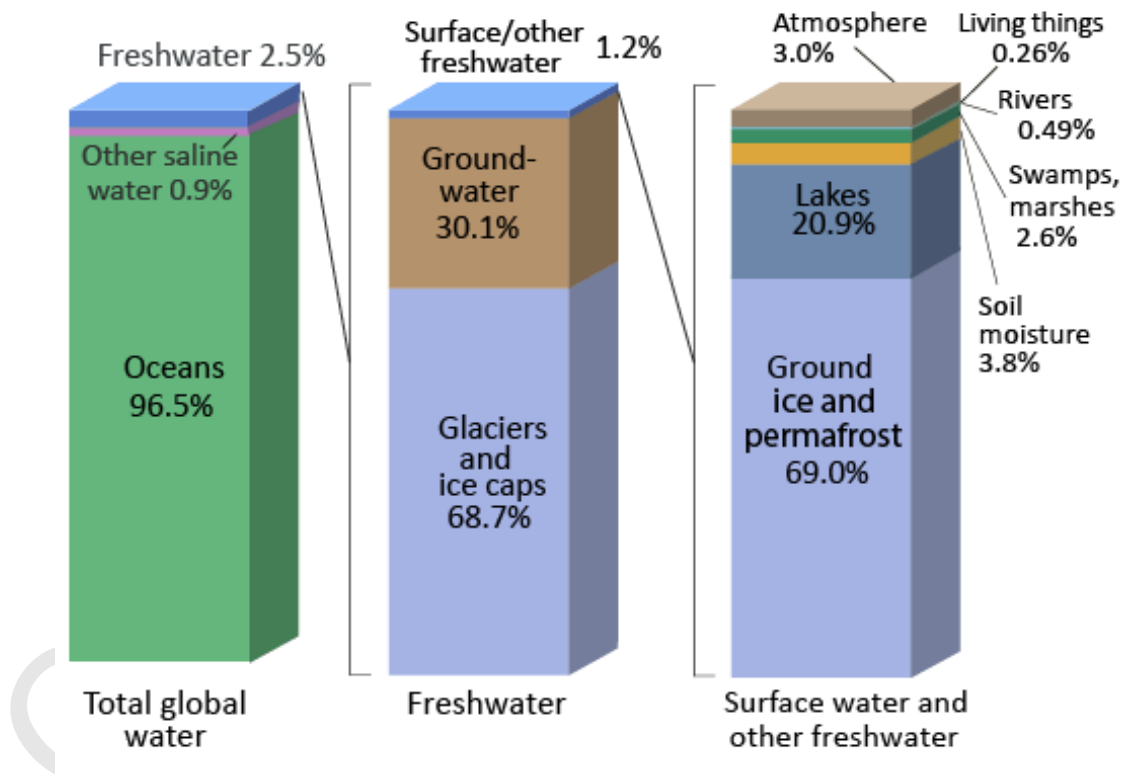


Figure 1.1 : The distribution of total world water
(<http://water.usgs.gov/edu/earthwherewater.html>)

Discharging the wastewater without remedy into the environment is no longer accepted choice because it carries a high load of organic and inorganic pollutants, pathogenic, and serious microorganisms. When the wastewater released without treatment into water bodies, that generates accumulated contamination in the aquatic environment (Riffat, 2012), thus, it may cause many diseases. The report issued by

(WATER, 2014) had stated that approximately 502,000 diarrheal deaths occurred in low and middle-income countries due to the inadequate and unsafe drinking-water.

Many of recent studies have focused on water control and wastewater treatment (Punmia et al., 1998). In fact, there are key methods for wastewater treatments; namely, physical, and chemical and biological treatment. Physical Treatment consists of preliminary, primary, and advanced primary process. Preliminary treatment aims to remove the gross solids such as large particles, and grit which destroy the apparatus. The primary treatment, which is a physical operation such as sedimentation, is typically used to harvest the floating particles of the wastewater. Subsequently, advanced primary treatment is used to boost the elimination process of suspended solids and to increase the precipitation rate. Secondly, chemical and biological treatment that includes secondary and advanced (or tertiary) treatment. Secondary treatment is applied to remedy the majority of the organic matters. Afterward, the advanced treatment uses supplementary unit operations and processes to deal with the residual suspended solids and other elements which were not removed considerably after the secondary treatment (Tchobanoglous, 1979).

Coagulation is a pivotal physicochemical process which has been broadly applied as a pre or post process step in wastewater treatment (Saini & Kumar, 2016). Coagulation is a principal process to reduce suspended solids, turbidity, viruses, and any particles within the size (0.1 to 10 μm) away from wastewater (Asano, 1998). This process involves basically, surface charge neutralization of particles and flocculation (i.e. floc creation) (Norulaini et al., 2001). When coagulation process starts, rapid mixing is applied in order to disperse the coagulant particles readily, thereby, destabilization of the colloidal particles is done. Slow mixing is applied to let the destabilized particles slowly aggregate and create settleable flocs. Undoubtedly, coagulation is applied to remedy diverse wastes, for instance, textile wastewater (Han et al., 2016), coal mining wastewater treatment (Galloux et al., 2015), biodiesel wastewater (Daud et al., 2015), and drinking water (Hu et al., 2015).

Coagulation efficacy depends on several factors such as coagulant category, coagulant concentration, sample PH, ionic strength and wastewater characteristics (Norulaini et al., 2001). Coagulants can be inorganic such as aluminum sulfate and magnesium chloride or polyelectrolyte (e.g., synthetic coagulants and natural coagulant).

1.2 Problem Statement

Alum coagulants are the most common agents used for water remedy (Okuda et al., 1999). However, recent studies approved that the use of aluminum salts showed extreme weakness in affecting the wastewater (Miller et al., 1984; Ndabigengesere et al., 1995). Abidin et al. (2011) mentioned that the use of alum salts may potentially caused many Alzheimer related disease cases. When these metals applied for water therapy purpose, a high aluminum concentration can be detected in the sludge residual of treated water as well as their efficacy is extreme pH- sensitive (Z. Yang et al., 2013). Subramonian et al. (2015) stated that the inorganic coagulants have serious side effects

on mankind; also their efficiency in cold water is low. Moreover, these types of coagulants are rarely applied in the developing countries due to their high prices. In addition, alum forms a large amount of residual sludge and gives dewatering problem. (Özacar, 2000). The use of the synthetic organic polymers such as acrylamide and poly aluminium chloride exhibited many adverse consequences (e.g. neurotoxicity and strong carcinogenic impacts) (Choy et al., 2014).

To manage above mentioned issues, many researchers have turned their faces toward safer coagulants that can be: 1) more effective, 2) human health friendly, 3) economic, and 4) biodegradable (Abidin et al., 2013; Choy et al., 2014; Shamsnejati et al., 2015). For this effect, natural coagulants have attracted considerable attention recently. Biocoagulants can be extracted from the sustainable sources (e.g. plants, animals, and microorganisms) (Abidin et al., 2011; Abidin et al., 2013; Zurina et al., 2014). In addition, polyelectrolytes are natural coagulants and they are able to condition the sludge and solve the dewatering problem (Özacar, 2000). Natural coagulants such as *Ocimum basilicum*, *Moringa Olivera*, chitosan, and other polymers are used in wastewater treatment as they are rich in polyelectrolytes in their contents (Bolto & Gregory, 2007; Oladoja, 2015; Shamsnejati et al., 2015).

Natural coagulant has proven to demonstrate a high efficacy in water treatment. For instance, (Chi & Cheng, 2006) studied the chitosan efficiency on an effluent that produced high fat and protein content in sludge where authors found out chitosan possess high potential. On the contrary of conventional coagulant, chitosan is more feasible, lower cost, low toxic and produce reasonable sludge volume. Generally, invertebrates (e.g. shrimp and crab shell) are the main source of the commercial chitosan. However, the raw materials required are seasonal and variable thereby leads to a difficult process. Hence, chitosan produced from this process is heterogeneous based on its physiochemical characteristics. Also, chitosan production rate is low due to the long time required to complete the reaction. In this scenario, exploiting the fungi is recommended as an alternative source for chitosan as it is rich in chitin and chitosan can be easily obtained from their wall cell (Pochanavanich & Suntornsuk, 2002).

The previous studies have dealt with the chitosan extracted from invertebrates; therefore, it is necessary to explore alternative source of chitosan of the new origin. In this study, chitosan of mushroom origin is studied to explore its ability as a bio-coagulant for wastewater treatment technology. From our knowledge, chitosan from mushroom can be produced by utilizing the waste accumulated due to mushroom production and harvest. This in turn will help to reduce waste accumulation and disposal to landfill which represents environmental issue for the producers (Tao et al., 2004).

1.3 Research Objectives

The current study presents an advanced water treatment through bio coagulant agent based on water-soluble chitosan mushroom (WSC) and acid-soluble chitosan mushroom (ASC). Therefore, there are three objectives of this study which are:

1. To characterize the ASC and WSC from mushroom as bio coagulant agent.
2. To investigate the coagulation performance of ASC and WSC for Kaolin and POME and compared with Alum.

1.4 Scope of Research

In this work the efficiency of chitosan mushroom (WSC and ASC) on kaolin wastewater and POME was investigated. This work included observing the turbidity reading, pH alteration, chemical oxygen demined (COD), biological oxygen demined (BOD), and total suspended solids (TSS). Furthermore, characterization of chitosan mushroom was limited to involve zeta potential, Fourier Transform Infrared (FTIR), and antimicrobial activity. Actually, the performance of chitosan was studied by test the alteration in the pH of treated water as well as by observing the effects of coagulant concentration on the coagulation process. The pH range in all pH optimization tests was from 2 to 12 while the dose of WSC or ASC ranged from 5 to 60 mg/L. This work involved a comparison between WSC and ASC as well as a comparison with alum. Moreover, this project included a comparison between the results obtained from the application of chitosan of mushroom origin with the results reported by other investigators in the literature which associated with chitosan produced from invertebrates (e.g. shrimp and crab shell).

1.5 Thesis Organization

The current manuscript is partitioned into five parts starting by the preface of this investigation which it affirms on the background of the study as well as the research problem statement with the research targets. Regarding chapter two, literature review discloses the important previous studies and their findings, potential, and the weakness, which mainly centering on the coagulant-flocculation process of wastewater therapy. On the other hand, chapter three indicates steps are taken throughout the study for data collections besides the provision steps which were deemed within the study. Chapter four confirms the data acquired and argues the outcomes with the previous studies. Chapter five includes the comprehensive conclusions and recommendations for next studies. Finally, this thesis is ended with the references and appendices associated with this thesis.

REFERENCES

- Abidin, Z. Z., Ismail, N., Yunus, R., Ahamad, I. S., & Idris, A. (2011). A preliminary study on *Jatropha curcas* as coagulant in wastewater treatment. *Environmental technology*, 32(9), 971-977.
- Abidin, Z. Z., Shamsudin, N. S. M., Madehi, N., & Sobri, S. (2013). Optimisation of a method to extract the active coagulant agent from *Jatropha curcas* seeds for use in turbidity removal. *Industrial Crops and Products*, 41, 319-323.
- Ahmad, A., Sumathi, S., & Hameed, B. (2004). Chitosan: a natural biopolymer for the adsorption of residue oil from oily wastewater. *Adsorption science & technology*, 22(1), 75-88.
- Ahmad, A., Sumathi, S., & Hameed, B. (2005). Adsorption of residue oil from palm oil mill effluent using powder and flake chitosan: equilibrium and kinetic studies. *Water research*, 39(12), 2483-2494.
- Ahmad, A., Sumathi, S., & Hameed, B. (2006). Coagulation of residue oil and suspended solid in palm oil mill effluent by chitosan, alum and PAC. *Chemical Engineering Journal*, 118(1), 99-105.
- Ahmad, A. L., Ibrahim, N., Ismail, S., & Bhatia, S. (2002). Coagulation-sedimentation-extraction pretreatment methods for the removal of suspended solids and residual oil from palm oil mill effluent (POME). *IIUM Engineering Journal*, 3(1).
- Al-Hamadani, Y. A., Yusoff, M. S., Umar, M., Bashir, M. J., & Adlan, M. N. (2011). Application of psyllium husk as coagulant and coagulant aid in semi-aerobic landfill leachate treatment. *Journal of Hazardous Materials*, 190(1), 582-587.
- Ariffin, A., Shatat, R. S., Norulaini, A. N., & Omar, A. M. (2005). Synthetic polyelectrolytes of varying charge densities but similar molar mass based on acrylamide and their applications on palm oil mill effluent treatment. *Desalination*, 173(3), 201-208.
- Asano, T. (1998). *Wastewater Reclamation and Reuse: Water Quality Management Library*: Taylor & Francis.
- Ashmore, M., & Hearn, J. (2000). Flocculation of model latex particles by chitosans of varying degrees of acetylation. *Langmuir*, 16(11), 4906-4911.
- Aziz, H. A., & Sobri, N. I. M. (2015). Extraction and application of starch-based coagulants from sago trunk for semi-aerobic landfill leachate treatment. *Environmental Science and Pollution Research*, 22(21), 16943-16950. doi: 10.1007/s11356-015-4895-7

- Barikani, M., Oliaei, E., Seddiqi, H., & Honarkar, H. (2014). Preparation and application of chitin and its derivatives: a review. *Iranian Polymer Journal*, 23(4), 307-326.
- Benhamou, N., Lafontaine, P., & Nicole, M. (1994). Induction of systemic resistance to *Fusarium* crown and root rot in tomato plants by seed treatment with chitosan. *Phytopathology*, 84(12), 1432-1444.
- Benjamin, M. M., & Lawler, D. F. (2013). *Water Quality Engineering: Physical / Chemical Treatment Processes*: Wiley.
- Bhatia, S., Othman, Z., & Ahmad, A. L. (2007). Pretreatment of palm oil mill effluent (POME) using *Moringa oleifera* seeds as natural coagulant. *Journal of Hazardous Materials*, 145(1), 120-126.
- bin Idris, A., & Al-Mamun, A. (2003). Anaerobic Fluidised Bed Reactor For The Treatment Of High Solid Content Industrial Effluent. Paper presented at the Wastewater Treatment and Waste Management: Proceedings of the International Conference on Water and Environment (WE-2003), December 15-18, 2003, Bhopal, India.
- Binnie, C., Kimber, M., & Smethurst, G. (2002). *Basic Water Treatment*: Royal Society of Chemistry.
- Bolto, B., & Gregory, J. (2007). Organic polyelectrolytes in water treatment. *Water research*, 41(11), 2301-2324.
- Boyd, C. E. (2000). *Physical Properties of Water Water Quality: An Introduction* (pp. 5-20). Boston, MA: Springer US.
- Boyd, C. E. (2015). *Water quality: an introduction*: Springer.
- Bratby, J. (1980). *Coagulation and flocculation*. Uplands: Croydon, England.
- Burkatovskaya, M., Tegos, G. P., Swietlik, E., Demidova, T. N., Castano, A. P., & Hamblin, M. R. (2006). Use of chitosan bandage to prevent fatal infections developing from highly contaminated wounds in mice. *Biomaterials*, 27(22), 4157-4164.
- Cai, Z.-s., Song, Z.-q., Shang, S.-b., & Yang, C.-s. (2007). Study on the flocculating properties of quaternized carboxymethyl chitosan. *Polymer Bulletin*, 59(5), 655-665.
- Chen, C.-Y., & Chung, Y.-C. (2011). Comparison of Acid-Soluble and Water-Soluble Chitosan as Coagulants in Removing Bentonite -010-0613-8

- Cheng, W. Suspensions. *Water, Air, & Soil Pollution*, 217(1), 603-610. doi: 10.1007/s11270P., Chi, F. H., Yu, R. F., & Lee, Y. C. (2005). Using chitosan as a coagulant in recovery of organic matters from the mash and lauter wastewater of brewery. *Journal of Polymers and the Environment*, 13(4), 383-388.
- Chi, F. H., & Cheng, W. P. (2006). Use of chitosan as coagulant to treat wastewater from milk processing plant. *Journal of Polymers and the Environment*, 14(4), 411-417.
- Choy, S. Y., Prasad, K. M. N., Wu, T. Y., Raghunandan, M. E., & Ramanan, R. N. (2014). Utilization of plant-based natural coagulants as future alternatives towards sustainable water clarification. *Journal of environmental sciences*, 26(11), 2178-2189.
- Dan, D.-z., Yao, Y.-l., & Jiang, W.-j. (2008). Analysis of environmental samples. *Chinese Journal of Analysis Laboratory*, 27(4), 100.
- Daud, Z., Awang, H., Nasir, N., Ridzuan, M. B., & Ahmad, Z. (2015). Suspended Solid, Color, COD and Oil and Grease Removal from Biodiesel Wastewater by Coagulation and Flocculation Processes. *Procedia-Social and Behavioral Sciences*, 195, 2407-2411.
- Davoudi, M., Samieirad, S., Mottaghi, H., & Safadoost, A. (2014). The main sources of wastewater and sea contamination in the South Pars natural gas processing plants: Prevention and recovery. *Journal of Natural Gas Science and Engineering*, 19, 137-146.
- Day, R. P. (2005). Cosmetic compositions having improved tactile and wear properties: Google Patents.
- Desai, K., Kit, K., Li, J., Davidson, P. M., Zivanovic, S., & Meyer, H. (2009). Nanofibrous chitosan non-wovens for filtration applications. *Polymer*, 50(15), 3661-3669.
- Diaz, A., Rincon, N., Escorihuela, A., Fernandez, N., Chacin, E., & Forster, C. (1999). A preliminary evaluation of turbidity removal by natural coagulants indigenous to Venezuela. *Process Biochemistry*, 35(3), 391-395.
- Divakaran, R., & Pillai, V. S. (2002). Flocculation of algae using chitosan. *Journal of Applied Phycology*, 14(5), 419-422.
- Dorf, R. C. (2004). *The Engineering Handbook*, Second Edition: CRC Press.
- Dr. B. C. Punmia, A. K. J. A. K. J., Punmia, B. C., Jain, A. K., & Jain, A. K. (1998). *Waste Water Engineering: Laxmi Publications Pvt Limited*.
- Drinan, J. E., & Spellman, F. (2012). *Water and wastewater treatment: A guide for the nonengineering professional: Crc Press*.

- Du, W.-L., Niu, S.-S., Xu, Y.-L., Xu, Z.-R., & Fan, C.-L. (2009). Antibacterial activity of chitosan tripolyphosphate nanoparticles loaded with various metal ions. *Carbohydrate Polymers*, 75(3), 385-389.
- Dutta, T., & Bhattacharjee, S. (2015). Arsenic Removal Through Combined Method Using Synthetic Versus Natural Coagulant. In *Advancements of Medical Electronics* (pp. 323-332). Springer India.
- El-Bestawy, E., Hussein, H., Baghdadi, H. H., & El-Saka, M. F. (2005). Comparison between biological and chemical treatment of wastewater containing nitrogen and phosphorus. *Journal of Industrial Microbiology and Biotechnology*, 32(5), 195-203.
- El-Nashar, A. M. (1980). Energy and water conservation through recycle of dyeing wastewater using dynamic Zr (IV)-PAA membranes. *Desalination*, 33(1), 21-47.
- Engineering, H. (2002). *Handbook of Public Water Systems*: Wiley.
- Gamage, A., & Shahidi, F. (2007). Use of chitosan for the removal of metal ion contaminants and proteins from water. *Food Chemistry*, 104(3), 989-996
- Galloux, J., Chekli, L., Phuntsho, S., Tijing, L., Jeong, S., Zhao, Y., . . . Shon, H. (2015). Coagulation performance and floc characteristics of polytitanium tetrachloride and titanium tetrachloride compared with ferric chloride for coal mining wastewater treatment. *Separation and Purification Technology*, 152, 94-100.
- Gamage, A., & Shahidi, F. (2007). Use of chitosan for the removal of metal ion contaminants and proteins from water. *Food Chemistry*, 104(3), 989-996.
- George, J. S., Ramos, A., & Shipley, H. J. (2015). Tanning facility wastewater treatment: Analysis of physical-chemical and reverse osmosis methods. *Journal of Environmental Chemical Engineering*, 3(2), 969-976.
- Ghernaout, D., Al-Ghonamy, A. I., Naceur, M. W., Boucherit, A., Messaoudene, N. A., Aichouni, M., . . . Elboughdiri, N. A. (2015). Controlling coagulation process: From zeta potential to streaming potential. *American Journal of Environmental Protection*, 4(5-1), 16-27.
- Gleick, P. H. (1993). *Water in crisis: a guide to the world's fresh water resources*: Oxford University Press, Inc.
- Goosen, M. F. A. (1996). *Applications of Chitan and Chitosan*: Taylor & Francis.
- Gregory, J., & Guibai, L. (1991). Effects of dosing and mixing conditions on polymer flocculation of concentrated suspensions. *Chemical Engineering Communications*, 108(1), 3-21.

- Guibal, E., & Roussy, J. (2007). Coagulation and flocculation of dye-containing solutions using a biopolymer (Chitosan). *Reactive and functional polymers*, 67(1), 33-42.
- Guibal, E., Van Vooren, M., Dempsey, B. A., & Roussy, J. (2006). A review of the use of chitosan for the removal of particulate and dissolved contaminants. *Separation science and technology*, 41(11), 2487-2514.
- Han, G., Liang, C.-Z., Chung, T.-S., Weber, M., Staudt, C., & Maletzko, C. (2016). Combination of Forward Osmosis Process (FO) with Coagulation/Flocculation (CF) for Potential Treatment of Textile Wastewater. *Water research*.
- Hang, Y. (1990). Chitosan production from *Rhizopus oryzae* mycelia. *Biotechnology letters*, 12(12), 911-912.
- Hare, W. (2003). Assessment of knowledge on impacts of climate change-contribution to the specification of art. 2 of the UNFCCC: Impacts on ecosystems, food production, water and socio-economic systems.
- Harper, T. R., & Kingham, N. W. (1992). Removal of arsenic from wastewater using chemical precipitation methods. *Water Environment Research*, 64(3), 200-203.
- Hassan, M. A. A., & Puteh, M. H. (2007). Pre-treatment of palm oil mill effluent (POME): A comparison study using chitosan and alum. *Malaysian Journal of Civil Engineering*, 19(2), 128-141.
- Heiderscheidt, E., Saukkoriipi, J., Ronkanen, A.-K., & Kløve, B. (2013). Optimisation of chemical purification conditions for direct application of solid metal salt coagulants: Treatment of peatland-derived diffuse runoff. *Journal of Environmental Sciences*, 25(4), 659-669.
- Hoven, V. P., Tangpasuthadol, V., Angkitpaiboon, Y., Vallapa, N., & Kiatkamjornwong, S. (2007). Surface-charged chitosan: Preparation and protein adsorption. *Carbohydrate Polymers*, 68(1), 44-53.
- Hu, C., Chen, Q., Chen, G., Liu, H., & Qu, J. (2015). Removal of Se (IV) and Se (VI) from drinking water by coagulation. *Separation and Purification Technology*, 142, 65-70.
- Huang, C., & Chen, Y. (1996). Coagulation of colloidal particles in water by chitosan. *Journal of Chemical Technology and Biotechnology*, 66(3), 227-232.
- Jadhav, M. V., & Mahajan, Y. S. (2013). Investigation of the performance of chitosan as a coagulant for flocculation of local clay suspensions of different turbidities. *KSCE Journal of Civil Engineering*, 17(2), 328-334.
- Jiang, J.-Q., & Graham, N. J. (1998). Pre-polymerised inorganic coagulants and phosphorus removal by coagulation- a review. *Water Sa*, 24(3), 237-244.

- Jun, H. K., Kim, J. S., No, H. K., & Meyers, S. P. (1994). Chitosan as a coagulant for recovery of proteinaceous solids from tofu wastewater. *Journal of agricultural and food chemistry*, 42(8), 1834-1838.
- Katayon, S., Noor, M. M. M., Asma, M., Ghani, L. A., Thamer, A., Azni, I., . . . Suleyman, A. (2006). Effects of storage conditions of *Moringa oleifera* seeds on its performance in coagulation. *Bioresource Technology*, 97(13), 1455-1460.
- Kathiravale, S., & Ripin, A. (1997). Palm oil mill effluent treatment towards zero discharge. Paper presented at the A paper presented at National Science and Technology Conference, held in July.
- Kemmer, F. N., & McCallion, J. (1988). *The NALCO water handbook* (Vol. 8): McGraw-Hill New York, NY.
- Kolhe, P., & Kannan, R. M. (2003). Improvement in ductility of chitosan through blending and copolymerization with PEG: FTIR investigation of molecular interactions. *Biomacromolecules*, 4(1), 173-180.
- Kumar, M. N. R. (2000). A review of chitin and chitosan applications. *Reactive and functional polymers*, 46(1), 1-27.
- Laboratory, I. E. R. (1983). *Design Manual: Neutralization of Acid Mine Drainage: The Laboratory*.
- Lang, G., Maresch, G., & Lenz, H.-R. (1990). O-benzyl-N-hydroxyalkyl derivatives of chitosan and nail polish containing the same: Google Patents.
- Lange, K. R., & Spencer, R. (1968). Mechanism of activated silica sol formation. *Environmental science & technology*, 2(3), 212-216.
- Lee, C. S., Robinson, J., & Chong, M. F. (2014). A review on application of flocculants in wastewater treatment. *Process Safety and Environmental Protection*, 92(6), 489-508.
- Lin, S. H., & Lin, C. M. (1993). Treatment of textile waste effluents by ozonation and chemical coagulation. *Water research*, 27(12), 1743-1748.
- Ma, C., Hu, W., Pei, H., Xu, H., & Pei, R. (2016). Enhancing integrated removal of *Microcystis aeruginosa* and adsorption of microcystins using chitosan-aluminum chloride combined coagulants: Effect of chemical dosing orders and coagulation mechanisms. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 490, 258-267.
- Mahmoodi, N. M., Salehi, R., Arami, M., & Bahrami, H. (2011). Dye removal from colored textile wastewater using chitosan in binary systems. *Desalination*, 267(1), 64-72.

- Maria, G., Maria, C., Salcedo, R., & de Azevedo, S. F. (2000). Databank transfer-of-information, shortcut and exact estimators used in the wastewater biological treatment process identification. *Computers & Chemical Engineering*, 24(2), 1713-1718.
- Matilainen, A., Vepsäläinen, M., & Sillanpää, M. (2010). Natural organic matter removal by coagulation during drinking water treatment: A review. *Advances in colloid and interface science*, 159(2), 189-197.
- Merrill, R. C. (1948). Activated silica sols in water treatment. *Industrial & Engineering Chemistry*, 40(8), 1355-1359.
- Michael-Kordatou, I., Michael, C., Duan, X., He, X., Dionysiou, D., Mills, M., & Fatta-Kassinos, D. (2015). Dissolved effluent organic matter: Characteristics and potential implications in wastewater treatment and reuse applications. *Water research*, 77, 213-248.
- Miller, R. G., Kopfler, F. C., Kelty, K. C., Stober, J. A., & Ulmer, N. S. (1984). The occurrence of aluminum in drinking water. *Journal (American Water Works Association)*, 84-91.
- Miyoshi, H., Shimura, K., Watanabe, K., & Onodera, K. (1992). Characterization of some fungal chitosans. *Bioscience, biotechnology, and biochemistry*, 56(12), 1901-1905.
- Mohire, N. C., & Yadav, A. V. (2010). Chitosan-based polyherbal toothpaste: As novel oral hygiene product. *Indian Journal of Dental Research*, 21(3), 380.
- Montgomery, J. M. Consulting Engineers, Inc.(1985) *Water Treatment Principles and Design*. John wiley & sons Inc. USA, 1(1), 6-1.
- Morales, J., De La Noüe, J., & Picard, G. (1985). Harvesting marine microalgae species by chitosan flocculation. *Aquacultural Engineering*, 4(4), 257-270.
- Moretto, L. M., & Kalcher, K. (2014). *Environmental analysis by electrochemical sensors and biosensors*: Springer.
- Mucha, M., & Miśkiewicz, D. (2000). Chitosan blends as fillers for paper. *Journal of applied polymer science*, 77(14), 3210-3215.
- Muttamara, S. (1996). Wastewater characteristics. *Resources, conservation and recycling*, 16(1), 145-159.
- Muzzarelli, R., Biagini, G., Bellardini, M., Simonelli, L., Castaldini, C., & Fratto, G. (1993). Osteoconduction exerted by methylpyrrolidinone chitosan used in dental surgery. *Biomaterials*, 14(1), 39-43.
- Ndabigengesere, A., & Narasiah, K. S. (1998). Quality of water treated by coagulation using *Moringa oleifera* seeds. *Water research*, 32(3), 781-791.

- Ndabigengesere, A., Narasiah, K. S., & Talbot, B. G. (1995). Active agents and mechanism of coagulation of turbid waters using *Moringa oleifera*. *Water research*, 29(2), 703-710.
- Norulaini, N., Norulaini, N., Zuhair, A., Zuhair, A., Hakimi, M., Hakimi, M., . . . Omar, M. (2001). Chemical Coagulation Of Settleable Solid-Free Palm Oil Mill Effluent (POME) For Organic Load Reduction. *Journal of Industrial Technology*, 10(1), 55-72.
- Okuda, T., Baes, A. U., Nishijima, W., & Okada, M. (1999). Improvement of extraction method of coagulation active components from *Moringa oleifera* seed. *Water research*, 33(15), 3373-3378.
- Oladoja, N. A. (2015). Headway on natural polymeric coagulants in water and wastewater treatment operations. *Journal of Water Process Engineering*, 6, 174-192.
- Oladoja, N. A. (2016). Advances in the quest for substitute for synthetic organic polyelectrolytes as coagulant aid in water and wastewater treatment operations. *Sustainable Chemistry and Pharmacy*, 3, 47-58.
- Osman, Z., & Arof, A. (2003). FTIR studies of chitosan acetate based polymer electrolytes. *Electrochimica Acta*, 48(8), 993-999.
- Özacar, M. (2000). Effectiveness of tannins obtained from *Valonia* as a coagulant aid for dewatering of sludge. *Water research*, 34(4), 1407-1412.
- Ozturk, E., Koseoglu, H., Karaboyaci, M., Yigit, N. O., Yetis, U., & Kitis, M. (2016). Minimization of water and chemical use in a cotton/polyester fabric dyeing textile mill. *Journal of Cleaner Production*, 130, 92-102.
- Panswad, T., & Wongchaisuwana, S. (1986). Mechanisms of dye wastewater colour removal by magnesium carbonate-hydrated basic. *Water Science and Technology*, 18(3), 139-144.
- Paulino, A. T., Simionato, J. I., Garcia, J. C., & Nozaki, J. (2006). Characterization of chitosan and chitin produced from silkworm crysalides. *Carbohydrate Polymers*, 64(1), 98-103.
- Pharino, C. (2007). *Sustainable water quality management policy: the role of trading: the US experience* (Vol. 10): Springer Science & Business Media.
- Pillai, J. (1997). *Flocculants and coagulants: The keys to water and waste management in aggregate production*. Naperville, IL: Nalco Company (Stone review). p, 1-6.
- Pochanavanich, P., & Suntornsuk, W. (2002). Fungal chitosan production and its characterization. *Letters in applied microbiology*, 35(1), 17-21.

- Poh, P., & Chong, M. (2009). Development of anaerobic digestion methods for palm oil mill effluent (POME) treatment. *Bioresource Technology*, 100(1), 1-9.
- Prashanth, K. H., & Tharanathan, R. (2005). Depolymerized products of chitosan as potent inhibitors of tumor-induced angiogenesis. *Biochimica et Biophysica Acta (BBA)-General Subjects*, 1722(1), 22-29.
- Qi, L., Xu, Z., Jiang, X., Hu, C., & Zou, X. (2004). Preparation and antibacterial activity of chitosan nanoparticles. *Carbohydrate research*, 339(16), 2693-2700.
- Ramawat, K. G., & Mérillon, J.-M. (2015). *Polysaccharides: Bioactivity and Biotechnology*: Springer.
- Ravina, L., & Moramarco, N. (1993). Everything you want to know about coagulation & flocculation. Zeta-Meter, Inc.
- Ravina, L., & Moramarco, N. (1993). Everything you want to know about coagulation & flocculation. Zeta-Meter Inc, 19–24.
- Ravindra, P. (2015). *Advances in Bioprocess Technology*: Springer International Publishing.
- Rhim, J.-W., Hong, S.-I., Park, H.-M., & Ng, P. K. (2006). Preparation and characterization of chitosan-based nanocomposite films with antimicrobial activity. *Journal of agricultural and food chemistry*, 54(16), 5814-5822.
- Ridwan, M., & Nobelia, J. (2009). Pengaruh Kekakuan, pH, Alkalinitas dan Zat Organik terhadap Dosis Koagulan pada Pengolahan Air Minum.(Studi Kasus: IPAM Ciparay PDAM Tirta Raharja Kabupaten Bandung): SW1-1. Penelitian, Teknik Lingkungan, Institut Teknologi Bandung, Bandung.
- Riffat, R. (2012). *Fundamentals of Wastewater Treatment and Engineering*: CRC Press.
- Riva, R., Ragelle, H., des Rieux, A., Duhem, N., Jérôme, C., & Préat, V. (2011). Chitosan and chitosan derivatives in drug delivery and tissue engineering Chitosan for biomaterials II (pp. 19-44): Springer.
- Robert, M. E., & Eric, J. B. (1992). *Chitin Chemistry*, The.
- Rodrigues, A. C., Boroski, M., Shimada, N. S., Garcia, J. C., Nozaki, J., & Hioka, N. (2008). Treatment of paper pulp and paper mill wastewater by coagulation–flocculation followed by heterogeneous photocatalysis. *Journal of Photochemistry and Photobiology A: Chemistry*, 194(1), 1-10.
- Roussy, J., Van Vooren, M., Dempsey, B. A., & Guibal, E. (2005). Influence of chitosan characteristics on the coagulation and the flocculation of bentonite suspensions. *Water research*, 39(14), 3247-3258.

- Ruiz-Herrera, J. (2016). *Fungal Cell Wall: Structure, Synthesis, and Assembly*, Second Edition: CRC Press.
- Rupani, P. F., Singh, R. P., Ibrahim, M. H., & Esa, N. (2010). Review of current palm oil mill effluent (POME) treatment methods: vermicomposting as a sustainable practice. *World Applied Sciences Journal*, 11(1), 70-81.
- Saifuddin, M., & Kumaran, P. (2005). Removal of heavy metal from industrial wastewater using chitosan coated oil palm shell charcoal. *Electronic journal of Biotechnology*, 8(1), 43-53.
- Saini, R., & Kumar, P. (2016). Simultaneous removal of methyl parathion and chlorpyrifos pesticides from model wastewater using coagulation/flocculation: Central composite design. *Journal of Environmental Chemical Engineering*, 4(1), 673-680.
- Sampa, M., Borrelly, S., Silva, B., Vieira, J., Rela, P., Calvo, W., . . . Somessari, E. (1995). The use of electron beam accelerator for the treatment of drinking water and wastewater in Brazil. *Radiation Physics and Chemistry*, 46(4), 1143-1146.
- Sanghi, R., Bhattacharya, B., Dixit, A., & Singh, V. (2006). Ipomoea dasysperma seed gum: An effective natural coagulant for the decolorization of textile dye solutions. *Journal of Environmental Management*, 81(1), 36-41.
- Sanghi, R., Bhattacharya, B., & Singh, V. (2006). Use of Cassia javahikai seed gum and gum-g-polyacrylamide as coagulant aid for the decolorization of textile dye solutions. *Bioresource Technology*, 97(10), 1259-1264.
- Šćiban, M., Klačnja, M., Antov, M., & Škrbić, B. (2009). Removal of water turbidity by natural coagulants obtained from chestnut and acorn. *Bioresource Technology*, 100(24), 6639-6643.
- Sethupathi, S. (2004). Removal Of Residue Oil From Palm Oil Mill Effluent (Pome) Using Chitosan [TD899. I27 S955 2004 f rb][Microfiche 7577]. Universiti Sains Malaysia.
- Shamsnejati, S., Chaibakhsh, N., Pendashteh, A. R., & Hayeripour, S. (2015). Mucilaginous seed of *Ocimum basilicum* as a natural coagulant for textile wastewater treatment. *Industrial Crops and Products*, 69, 40-47.
- Sharma, S. K., & Sanghi, R. (2012). *Advances in water treatment and pollution prevention*: Springer Science & Business Media.
- Shi, B., Li, G., Wang, D., Feng, C., & Tang, H. (2007). Removal of direct dyes by coagulation: The performance of preformed polymeric aluminum species. *Journal of Hazardous Materials*, 143(1), 567-574.

- Sila, A., Mlaik, N., Sayari, N., Balti, R., & Bougatef, A. (2014). Chitin and chitosan extracted from shrimp waste using fish proteases aided process: Efficiency of chitosan in the treatment of unhairing effluents. *Journal of Polymers and the Environment*, 22(1), 78-87.
- Singh, V., Malviya, T., & Sanghi, R. (2012). Polysaccharide-Based Macromolecular Materials for Decolorization of Textile Effluents. In *Advances in Water Treatment and Pollution Prevention* (pp. 377-403). Springer Netherlands.
- Smith, A. H., & Weber, N. S. (1980). *The Mushroom Hunter's Field Guide*: University of Michigan Press.
- Subramonian, W., Wu, T. Y., & Chai, S.-P. (2015). An application of response surface methodology for optimizing coagulation process of raw industrial effluent using *Cassia obtusifolia* seed gum together with alum. *Industrial Crops and Products*, 70, 107-115.
- Synowiecki, J., & Al-Khateeb, N. A. (2003). Production, properties, and some new applications of chitin and its derivatives.
- Tchobanoglous, M. E. I. (1979). *Wastewater engineering: treatment, disposal, re-use*. McGraw-Hill Book Company. New York, 2 nd Edition,(07 A MET), 938.
- Teh, C. Y., Wu, T. Y., & Juan, J. C. (2014a). Optimization of agro-industrial wastewater treatment using unmodified rice starch as a natural coagulant. *Industrial Crops and Products*, 56, 17-26.
- Teh, C. Y., Wu, T. Y., & Juan, J. C. (2014b). Potential use of rice starch in coagulation–flocculation process of agro-industrial wastewater: treatment performance and flocs characterization. *Ecological Engineering*, 71, 509-519.
- Theodoro, J. D. P., Lenz, G. F., Zara, R. F., & Bergamasco, R. (2013). Coagulants and natural polymers: perspectives for the treatment of water. *Plastic and Polymer Technology*, 2(3), 55-62.
- Tripathy, T., & De, B. R. (2006). *Flocculation: a new way to treat the waste water*.
- Unicef. (2008). *UNICEF handbook on water quality*. United Nations Childrens Fund, New York/USA.
- Verma, A. K., Dash, R. R., & Bhunia, P. (2012). A review on chemical coagulation/flocculation technologies for removal of colour from textile wastewaters. *Journal of Environmental Management*, 93(1), 154-168.
- Vijayaraghavan, G., Sivakumar, T., & Vimal Kumar, A. (2011). Application of plant based coagulants for waste water treatment. *International Journal of Advanced Engineering Research and Studies*, 1(1), 88-92.
- WATER, T. B. (2014). *PREVENTING DIARRHOEA THROUGH BETTER WATER, SANITATION AND HYGIENE*.

- Water Treatment. (2011). American Water Works Association.
- Wu, T., Mohammad, A. W., Jahim, J. M., & Anuar, N. (2007). Palm oil mill effluent (POME) treatment and bioresources recovery using ultrafiltration membrane: effect of pressure on membrane fouling. *Biochemical Engineering Journal*, 35(3), 309-317.
- Wu, T., Zivanovic, S., Draughon, F. A., & Sams, C. E. (2004). Chitin and chitosan value-added products from mushroom waste. *Journal of agricultural and food chemistry*, 52(26), 7905-7910.
- Xin-Yuan, S., & Tian-Wei, T. (2004). New contact lens based on chitosan/gelatin composites. *Journal of Bioactive and Compatible polymers*, 19(6), 467-479.
- Yang, R., Li, H., Huang, M., Yang, H., & Li, A. (2016). A review on chitosan-based flocculants and their applications in water treatment. *Water research*, 95, 59-89.
- Yang, Z., Li, H., Yan, H., Wu, H., Yang, H., Wu, Q., . . . Cheng, R. (2014). Evaluation of a novel chitosan-based flocculant with high flocculation performance, low toxicity and good floc properties. *Journal of Hazardous Materials*, 276, 480-488.
- Yang, Z., Liu, X., Gao, B., Zhao, S., Wang, Y., Yue, Q., & Li, Q. (2013). Flocculation kinetics and floc characteristics of dye wastewater by polyferric chloride–polyepichlorohydrin–dimethylamine composite flocculant. *Separation and Purification Technology*, 118, 583-590.
- Yang, Z., Yang, H., Jiang, Z., Cai, T., Li, H., Li, H., . . . Cheng, R. (2013). Flocculation of both anionic and cationic dyes in aqueous solutions by the amphoteric grafting flocculant carboxymethyl chitosan-graft-polyacrylamide. *Journal of hazardous materials*, 254, 36-45.
- Yang, Z., Yuan, B., Huang, X., Zhou, J., Cai, J., Yang, H., . . . Cheng, R. (2012). Evaluation of the flocculation performance of carboxymethyl chitosan-graft-polyacrylamide, a novel amphoteric chemically bonded composite flocculant. *Water research*, 46(1), 107-114.
- Yin, C.-Y. (2010). Emerging usage of plant-based coagulants for water and wastewater treatment. *Process Biochemistry*, 45(9), 1437-1444.
- Zhang, D., & Quantick, P. C. (1998). Antifungal effects of chitosan coating on fresh strawberries and raspberries during storage. *The Journal of Horticultural Science and Biotechnology*, 73(6), 763-767.
- Zhang, J., Zhang, F., Luo, Y., & Yang, H. (2006). A preliminary study on cactus as coagulant in water treatment. *Process Biochemistry*, 41(3), 730-733.

Zhang, L., Ao, Q., Wang, A., Lu, G., Kong, L., Gong, Y., . . . Zhang, X. (2006). A sandwich tubular scaffold derived from chitosan for blood vessel tissue engineering. *Journal of biomedical materials research Part A*, 77(2), 277-284.

Zheng, L.-Y., & Zhu, J.-F. (2003). Study on antimicrobial activity of chitosan with different molecular weights. *Carbohydrate Polymers*, 54(4), 527-530.

Zurina, A. Z., Mohd Fadzli, M., Ghani, A., & Abdullah, L. (2014). Preliminary Study of Rambutan (*Nephelium Lappaceum*) Seed as Potential Biocoagulant for Turbidity Removal. Paper presented at the Advanced Materials Research.

