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1	In vivo predictive dissolution and simulation workshop report: Facilitating the
2	development of oral drug formulation and the prediction of oral bioperformance
3	
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- 34 _____
- 35

36 Abstract

37 This summary report for the "in vivo predictive dissolution and simulation workshop" 38 highlights presentations from a two-day workshop held on September 11-12, 2017. This 39 workshop was aimed to present scientists at FDA, EMA, industry and academia the most 40 recent advances in dissolution methodologies and scientific knowledge for oral drug 41 products, which could be useful for guiding early phase development, bioavailability (BA) and bioequivalence (BE) studies and Scale-Up and Post-Approval Changes (SUPAC) of 42 43 oral products. Presentations and discussions focused on appropriate in vitro and in silico 44 applications and tool selections to predict *in vivo* bioperformance of oral formulations. 45 Product developability and Quality by Design (QbD) would be determined by the 46 physicochemical characteristics of active pharmaceutical ingredients (API), in vitro 47 dissolution and in silico models/computer simulation. Many methodologies and 48 applications are available to predict *in vivo* bioperformance of oral products/formulations. 49 It is crucial that the selections of appropriate tools based on API and formulations to 50 maximize in vivo prediction by in vitro/in silico results. This workshop presented cutting-51 edge tools/methodologies and how to select the right tools from a methodology toolbox and 52 testing parameters to predict best *in vivo* bioperformance of test products. The 53 combinations of *in vivo* minded *in vitro* dissolution methodologies and computational 54 approaches become mainstream to predict oral absorption/plasma profiles of oral products. 55 This workshop provides the degree of advancement within state-of-the-art scientific 56 knowledge, validation, and development and the extent to which the regulatory community

has absorbed and accepted these advancements in science-based mechanistic approaches tooral drug product development.

59 Introduction

A two-day workshop entitled "In Vivo Predictive Dissolution and Simulation" was held 60 61 September 11-12, 2017 in Washington DC focused on the selection of applications, 62 methodologies, and scientific advancements to predict in vivo bioperformance of oral drug 63 products/oral drug formulations based on the active pharmaceutical ingredient (API) and 64 drg product formulation. This workshop was fully sponsored by the AAPS and featured 65 speakers from industry, academia, and regulatory agencies to introduce the state-of-the-art 66 in cutting-edge applications, methodologies and latest initiatives in in vivo prediction of 67 oral drug product performance to attendees worldwide. A broad range of dissolution 68 methodologies and simulations together with the determination of developability based on 69 physicochemical characteristics were discussed specific considerations for in vivo 70 prediction implementing bioavailability (BA), bioequivalence (BE), and quality by design 71 (QbD), in this two-day workshop. 72 The objectives of this workshop were to: 73 Present scientists at regulatory agencies, industry and academia the most recent •

resent scientists at regardory agencies, industry and academia the most recent
 advances in dissolution methodologies, computational applications and science for
 oral drug products to predict *in vivo* behavior of oral drug products, which could be
 useful for guiding early phase development, bioavailability (BA) and
 bioequivalence (BE) studies and Scale-Up and Post-Approval Changes (SUPAC)
 of oral products.

79	• Present state-of-the-art <i>in vivo</i> predictive dissolution methodologies for drug
80	products, including determination in vitro testing parameters to achieve in vivo
81	predictive and desired outcomes, and how to interpret in vitro results and
82	translating them into potential IVIVCs.
83	• Present state-of-the-art scientific analysis and knowledge using the latest
84	mechanistic BCS-subclass-based in vivo and in silico predictive dissolution
85	methodologies.
86	• Present a mechanistic basis for more efficiently reviewing pharmaceutical product
87	change applications and new generic product applications, including BE studies,
88	assuring therapeutic benefits and safety of oral drug products for public health.
89	• Provide a forum to discuss <i>in vitro</i> dissolution and <i>in silico</i> simulation through
90	case studies.
91	Workshop participants learned the newest mechanistic, BCS Subclass based, in vivo
92	predictive dissolution methodologies and physiologically-based computer simulation and
93	science, and were presented with discussion on state-of-the-art dissolution methodologies
94	based on physicochemical characteristics of API. Case studies were presented where
95	current quality control (QC) dissolution methodologies have been inadequate predicting in
96	vivo performance, and bioequivalence failure. An in vivo predictive dissolution could
97	provide mechanistic explanation of in vivo results which could help guide an early
98	formulation development effort, bridge scale up work and understand reference product
99	profiles for generic formulation development.

100 This workshop was targeted to regulatory scientists, pre-formulation, formulation,

101 biopharmaceutics, and QC scientists in industry, and graduate students and scientists in the

102 academia. The workshop focused on presenting the most recent methods and scientific

103 understanding related to possible pharmacokinetic performance and bioequivalence (BE)

104 risk, in vivo dissolution/prediction for a test formulation/a test oral product to meet for

105 ensuring the therapeutic efficacy of modified/changed product. Formulation changes occur

106 frequently over the course of an innovator product's lifetime due to composition,

107 manufacturing, and site of manufacturing changes. BE provides an important standard for 108 the development and approval of multi-source and generic drug products, the most rapidly 109 expanding segment of the pharmaceutical industry worldwide. The workshop benefited the 110 audience by presenting the mechanistic basis for more efficiently designing pharmaceutical

111 product/formulation and for quality by design (QbD) studies.

112

113 Day 1

114 In vivo buffers and buffer properties for affecting solubility and dissolution rate.

115 Dr. Gregory E. Amidon (University of Michigan) led off the conference making the case

116 that the critical link between oral solid dosage form formulation, in vivo plasma levels, and

117 therapeutic effect is *in vivo* dissolution. He discussed several key aspects important to the

- 118 development of relevant *in vitro* methods focusing on our improved understanding of
- 119 bicarbonate as our primary lumenal buffer. Accurate prediction of dissolution rate requires

120 an understanding of the conditions at the dissolving drug surface (1, 2). For acidic or basic

121 drugs, an *in vitro* measurement of dissolution that reflects *in vivo* conditions requires

122 dissolution media that yields a surface pH (pH₀) representative of *in vivo* conditions (1-6).

123 The improved understanding of bicarbonate as a buffer is important and confirms that

124 lumenal bicarbonate buffer concentration and buffer capacity is very low and this is

125 critically important to developing methodologies that reflect *in vivo* pH₀(7). This more

126 comprehensive understanding of *in vivo* hydrodynamic and chemical conditions will allow

127 for physiologically and physicochemically relevant *in vitro* dissolution testing to be

128 performed on a sound, scientific basis.

129 In Vivo Gastrointestinal Fluid Composition and Effects of Drug Substance

130 Physiochemical Properties on Solubilization

131 Dr. Christel Bergström (Uppsala University) continued with a thorough presentation of 132 composition of human intestinal fluids. She emphasized that recent clinical studies pointed 133 at a higher pH in the stomach than that typically used in compendial media (median of 2.5 134 with a range of 1.7-3.3 in comparison to compendial pH of 1.0-1.2), a lower buffer capacity 135 in the upper gastrointestinal tract than previously thought, and a larger intra- and inter-136 individual variability in bile salts and phospholipids than previously has been reported (7-137 9). These factors may significantly affect both dissolution rate and solubilization in the 138 human intestinal tract. For this reason, there is not a single biorelevant medium that can be 139 used to provide insights into the expected variability of dissolution rate and solubilization; 140 rather a number of biorelevant media is likely to be needed to provide insights into the 141 expected variability in vivo. She then linked the performance of drugs to their 142 physicochemical properties and in particular pointed at the usefulness of understanding the 143 role of lipophilicity, solid state properties and extent of ionization on the dissolution in

144 human intestinal fluids (10, 11). These physicochemical properties will inform on which 145 types of biorelevant media to select for a particular compound. Further, computational 146 modeling was discussed and identified as a tool that merits to be used to predict e.g. 147 dissolution, solubility and biopharmaceutical performance (12). She identified that more 148 clinical data on the impact of the fed state on drug dissolution are warranted to better 149 understand inter-individual variability in the fed state. 150 Impacts of In Vivo Fluid Hydrodynamics on Dissolution and Absorption in the 151 **Human Intestines** 152 Dr. James G. Brasseur (University of Colorado) discussed the impacts of intestinal fluid 153 motions ("hydrodynamics") on the processes by which drug molecules are released from 154 clouds of small drug particles from a disintegrated tablet or capsule as particles and 155 molecular concentrations are transported within the intestinal lumen and drug molecules are 156 absorbed at the mucosal surface. Emphasis was placed on the varying impacts of different 157 classes of motility patterns (i.e., changes in luminal geometry along gut segments as a 158 function of time driven by contraction of the muscle fibers within the intestinal wall) 159 associated with the different migrating motor complex (MMC) phases of contraction when 160 the gut is in the fasting state vs. fed state motility. Whereas peristaltic motility in the fasting 161 state drives the transport of residual material from the gut, the dominant function in the fed 162 state is nutrient absorption, associated with segmental motions that locally mix intestinal 163 liquid content in addition to bulk transport by peristalsis. The rate of release of drug 164 molecules from drug particles (dissolution) is modulated by flow patterns that transport 165 thousands of drug particles preferentially within localized regions and by the hydrodynamic

166 enhancement in the rate of release of molecules from the surface of individual drug 167 particles from flow field characteristics local to the moving particle. Dr. Brasseur described 168 the mathematical framework for single particle dissolution rate and showed that the 169 hydrodynamic enhancement of particle dissolution rate was represented within a 170 normalized molecular flux, historically referred to as the "Sherwood number.". It was 171 shown that this normalized particle flux is at the core of mathematical model formulations for dissolution from clouds of drug particles. Dr. Brasseur then went into a detailed review 172 173 of recent research into two key hydrodynamic influences on particle dissolution rate (i.e., 174 normalized flux): (1) the convection effect which arises from "slip" velocity between the 175 moving particle and the surrounding fluid, and (2) a "shear-rate" effect that has been 176 recently discovered, quantified and experimentally validated that arises from drug particle spin induced by hydrodynamic shear-rate at the location of the particle. Using a 177 178 computational fluid dynamics in vivo simulation environment in which the particle 179 dissolution model was embedded, Dr. Brasseur showed that the hydrodynamic shear-rate 180 effect creates major enhancements in drug dissolution while the convection effect provides 181 only a minor influence due to the small size of the particles. Additional discussion was 182 presented of the physical processes underlying the balance between release and absorption 183 of ibuprofen *in vivo* in the presence of peristaltic motility and high permeability. This 184 balance involves the interplay between diffusion and hydrodynamic transport of drug from 185 the bulk to the mucosal surface and is strongly impacted by the size (or volume) of the 186 pocket of intestinal liquid in which drug molecules are released and transported.

187 Dissolution Methodologies and Selection of Study Conditions Based upon Drug 188 Physicochemical Characteristics (BCS subclass) & Dosage Forms

189 Dr. Deanna Mudie (Lonza Pharma & Biotech) presented a mechanistic approach for

190 selecting *in vitro* dissolution methodologies and testing parameters for designing oral drug

191 product formulations and differentiating them with respect to bioperformance. This

approach relies upon first predicting the rate determining steps to *in vivo* absorption based

193 upon the drug substance and product of interest, and an understanding of the complex and

194 heterogeneous gastrointestinal tract. For example, dimensionless numbers (e.g. Do, Dn &

195 Pn) can be used to predict whether a compound may be solubility-permeability,

196 permeability or dissolution rate limited *in vivo* (13, 14). BCS sub-classification can be used

197 together with knowledge of the drug product formulation as a basis for predicting relative

198 extent of gastric to intestinal dissolution (15). To demonstrate this methodology, Dr. Mudie

199 presented a case study of spray-dried amorphous solid dispersions of itraconazole with

200 hydroxypropyl methylcellulose acetate succinate dosed to rats (16). Using a material

201 sparing membrane flux apparatus (17), colleagues at Lonza Pharma & Biotech were able to

show that the maximum absorption rate for each formulation rank ordered with membrane

203 flux *in vitro* when the test was set up to be solubility-permeability limited and a biorelevant

204 fluid composition representative of fast rats was selected.

205 Direct Measurement of In Vivo Dissolution of IR and MR Drug Products in Human

206 GI Tract

207 Dr. Duxin Sun (University of Michigan) presented the *in vitro/in vivo* data analysis of a

208 human intubation study and the challenge of *in vivo-in vitro* correlation (IVIVC) for the

209 local acting drugs with the administration of modified release (MR) mesalamine oral 210 formulations, Pentasa, Apriso, and Lialda, along with oral mesalamine solution and an 211 immediate release (IR) ibuprofen formulation. The specialized catheter with 4 aspiration 212 channels allowed the measurement of luminal drug concentrations (18, 19). The idea is to 213 correlate the directly measured drug concentration in the human gastrointestinal (GI) 214 regions and the plasma drug concentration along with the drug dissolution in different GI 215 tract by computational modeling. Results indicated that in vivo dissolution of MR 216 mesalamine oral dosage forms were highly variable. Pentasa released mesalamine 217 throughout the GI tract including the stomach, while Apriso released mesalamine between 218 duodenum and jejunum regions. However, Lialda rarely released any mesalamine in first 7 219 hrs. Those MR formulations exhibited the different drug release profiles *in vivo* and *in* 220 *vitro*. However, the large amount of unmetabolized drugs was observed in feces, 221 suggesting unreleased and/or undissolved. In ibuprofen studies, high concentration of 222 ibuprofen was observed in the stomach and small intestine at 7 hrs after oral administration (18). With the elevation of gastric pH by the intake of liquid meal (Pulmocare[®]), higher 223 224 drug concentration of ibuprofen in the stomach was observed (19). However, the lower 225 C_{max} and delayed T_{max} in the plasma profiles in the fed state were observed compared to 226 ones in the fasted state suggesting the slower gastric emptying time in the fed state. 227 Overall, the challenges are the limited data of *in vivo* dissolution in the different GI sites to 228 validate the *in vitro* dissolution models and *in silico* simulation. It would be a mutually 229 beneficial if the industry, academia and the regulators to collaborate to produce and share 230 more in vivo dissolution data.

231 Interpreting Drug Concentration Profiles in Plasma and Relating Them to In Vitro

232 Dissolution Measurements/In Silico Predictions

233 Dr. Marival Bermejo (Universidad Miguel Hernández de Elche) presented the exploratory 234 data analysis of a human intubation study with the administration of an immediate release 235 (IR) ibuprofen (weak acid) oral formulation. The specialized manometric catheter with 4 236 sampling ports allowed the measurement of luminal drug concentrations, pH values as well as intestinal wall motility (19). Results indicated that ibuprofen in vivo dissolution depends 237 238 on luminal pH (7). Additionally, time to the next Phase III wave post dose (TMMC) 239 determined the arrival of most of the ibuprofen dose to the small intestine, consequently 240 longer TMMC is reflected in lower C_{max} and longer T_{max}. Absorption rates estimated from 241 plasma levels by deconvolution showed a good correlation with *in vivo* dissolution i.e. 242 maximal absorption rates corresponded with the maximal ibuprofen concentrations in 243 intestinal lumen. A compartmental (stomach-duodenum-jejunum-plasma) mass transport 244 analysis incorporating TMMC, and pH-dependent dissolution reproduced closely the 245 individual plasma levels and the inter-subject variability. These results confirmed the direct 246 link between intestinal dissolution, luminal solution concentration and systemic absorption 247 thus the impact of gastrointestinal variables as pH and motility in oral absorption. iPD 248 methodologies incorporating these variables in combination with mass transport 249 computational methods are necessary tools to optimize formulation development.

250

251 *iPD Methodologies – Future*

252 Dr. Gordon L. Amidon (University of Michigan) presented his vision of in vivo predictive 253 dissolution (iPD) to the future of Biopharmaceutics and to the implications of oral product 254 development through the evolution of regulations on oral drug products, dissolution 255 methodologies, and technologies to advance the understanding of the human GI physiologies. 256 The improved understanding of complexed human GI physiology and the advancement of 257 technologies allows us to develop the in vitro dissolution apparatuses, which are 258 physiologically relevant to the human GI conditions, and the simulation and physiologically 259 based pharmacokinetics modeling for the prediction of in vivo dissolution and drug 260 absorption of oral dosage forms. Those movements have revolutionized and will keep 261 advancing the development of drug products, the design of oral drug products, and the 262 bioequivalent (BE) studies. However, the regulatory agencies, academia, and industries 263 should fully collaborate to facilitate this advancement and to validate *in vitro* models and to 264 share limited amount of human permeability and plasma data. The global harmonization will 265 be necessary to promote science based dissolution methodologies and BE standards.

266

267 Day 2

A Two-Phase Dissolution-Partition Test for Characterizing BCS II Drugs Products and Establishing IVIVR

270 Dr. Ping Gao (AbbVie) presented his work in developing a two-phase dissolution-partition

- 271 test for evaluation of BCS II drug formulations. This method, referred as to the biphasic
- test, permits dissolution in the aqueous media (with pH alteration) under a non-sink
- 273 condition and simultaneous partition of the dissolved drug into an organic phase that acts as

274 an "absorption compartment". The partition of the drug into the organic phase is driven by 275 the free drug concentration in the aqueous phase and this is to mimic absorption in 276 vivo. The theoretical model of the biphasic system was developed to reveal that the 277 physiological relevance of this test method is based on the *in vitro* partitioning rate 278 coefficient, kp, approximates the *in vivo* absorption rate coefficient, ka (20). Three case 279 studies of BCS II drug formulations including ABT-072 (weak acid) (21), ritonavir (weak 280 base) (22), and fenofibrate (23) were reviewed. Their in vitro profiles obtained in 281 biorelevant media under the optimal hydrodynamic condition by the biphasic test are 282 closely correlated with relative exposures of these drugs in human subjects. These cases 283 jointly reveal the significant impact of supersaturation upon oral exposure of BCS II drugs 284 and a complex interplay among the dissolution, precipitation, and partition processes that 285 dictates the oral exposure.

286

287 BCS IIb Drug Substances in the Gastro-Intestinal Simulator (GIS)

288 Dr. Yasuhiro Tsume (University of Michigan) presented his work in developing a multi-

289 compartment transfer system, gastrointestinal simulator (GIS), to evaluate the

bioperformance of weakly base drugs, ketoconazole and dasatinib as model drugs (24,

291 25). The GIS, which consists of three chambers, gastric, duodenal, and jejunal

292 compartments with secretion chambers to supply appropriate media back into the gastric

- and duodenal chambers (26). Using the GIS, Dr. Tsume demonstrated the occurrence of
- supersaturation and precipitation of BCS class IIb drugs and the enhanced absorption
- resulting from supersaturation effects by the combination study of infusion study and the

dissolution study and the potential to predict clinical outcome with *in vitro* dissolution
methods (24, 25). Dr. Tsume mentioned the importance of experimental conditions like
aqueous volume (volume to the dose), buffer species, buffer capacity, buffer pH and gastric
motility (gastric emptying rate and transit time) with experimental examples (27-30). He
also demonstrated the presence of absorption phase (biphasic setting) would be useful in
the dissolution methodologies for certain drugs for more accurate *in vivo* prediction (31).

302

303 Multicompartment Transfer Model to Predict Dissolution/Precipitation of Weakly Basic 304 Drug

305 Sanjaykumar Patel and Wei Zhu (Merck & Co., Inc., Kenilworth, NJ, USA) presented their 306 work in developing a multi-compartment transfer system for evaluation of dissolution and 307 precipitation of weakly basic drugs during the transfer out of the stomach into the 308 intestine. This transfer system includes a "gastric" compartment, an "intestinal" 309 compartment, a "sink" compartment for removal of the drugs from intestinal compartment, 310 and a "reservoir" compartment to re-supply FaSSIF media during the course of the 311 experiment. An *in silico* model was built to simulate the time-dependent dissolution and 312 precipitation processes when drugs/formulations were tested using the transfer system, and 313 the precipitation rate obtained from the model was used as the inputs for subsequent 314 absorption modeling. Two case studies, dypyridamole and ketoconazole, were reviewed, as 315 the in vitro dissolution and precipitation of these two drugs were analyzed using both 316 transfer system and traditional two-stage dissolution. Using the fitted precipitation rate from transfer system as the inputs for GastroplusTM modeling, the predicted 317

pharmacokinetic profiles of orally dosed IR formulations were generally in agreement with observed clinical data. A sensitivity analysis on *in vivo* precipitation in GastroplusTM suggested an optimal prediction accuracy when precipitation rates from the transfer system was utilized. These case examples showed promising results to support this integrated *in vitro/in silico* transfer system as an alternative approach to estimate in vivo precipitation in intestinal compartment, which is one of the critical attributes for prediction of clinical bioperformance for weak basic compounds.

325 BCS II/IV Drug Substances in the Artificial Stomach Duodenum (ASD) System

326 Dr. David C. Sperry (Eli Lilly and Company) presented his work in artificial stomach and 327 duodenum (ASD) as a tool to develop oral drug products. This dissolution apparatus, 328 which mimics the dynamic conditions of the human GI tract, helps predict the *in vivo* 329 impact of oral dosage forms properties such as salts, solid forms, formulation composition, 330 and particle size. The goal of this approach is to reduce the number of animal studies 331 required during formulation development while selecting the best possible oral dosage 332 forms for clinical studies. Certain drugs would supersaturate, precipitate, and/or dissolve in 333 the duodenal region, which have impact on their absorption. Those molecule/formulation 334 related phenomena can be captured by ASD, which mimics the dynamic GI conditions, to 335 support the *in vivo* prediction. The drug concentration in the duodenal chamber of ASD 336 can be predicted based on the drug concentration in the gastric chamber of ASD. The 337 difference between experimental results and calculated/expected results indicates additional 338 dissolution and/or precipitation, which will provide tremendous helps to understand the in 339 vivo dissolution and the potential problems of test drug/formulation. Dr. Sperry presented a

few case studies with the different API forms (free base form vs. salt form), the different dosage strengths (low vs. high), the different pH and buffer viscosity to demonstrate the impact of *in vivo* dissolution of test oral formulations. He demonstrated through those case studies that those *in vitro* dissolution profiles obtained with ASD combination of *in silico* absorption model, gCOAS, predict better *in vivo* performance and, hence, the usefulness and practicality of *in vivo* predictive dissolution methodology, ASD.

346

347 Implementing In Vitro Dissolution Data into PBPK Models for Evaluation of Absorption

348 from the Lower Intestine

349 Dr. Maria Vertzoni (National and Kapodistrian University of Athens) presented the impact 350 of absorption from the lower intestine on plasma pharmacokinetic profile. After oral 351 administration of a drug product, the drug absorption from the lower intestine was of 352 particular interest when considering the development of modified release products. It could 353 also be useful, for understanding the pharmacokinetic performance of poorly soluble active 354 pharmaceutical ingredients (APIs), BCS Class II and Class IV APIs, when those are 355 administered in immediate release products and their drug absorption is incomplete in the 356 upper intestine. For the evaluation of colonic absorption, knowledge of drug solubility and 357 dissolution rates in the region is required but relevant estimations remain problematic, due to 358 limited information on the conditions prevailing in the lower intestine. In recent years our 359 understanding on the environment in the lower intestine has been increased (32, 33).

360 Dr Vertzoni presented the usefulness of biorelevant in vitro data in PBPK models describing
361 oral absorption from upper / middle as well as from lower intestine with various case
362 examples.

363 She presented the media simulating the contents of lower intestine i.e. distal ileum and 364 proximal colon under conditions simulating the bioavailability and bioequivalence studies in 365 the fasted and in the fed states and a recently developed in vitro two-stage single-366 compartment models for evaluating dissolution characteristics in the lower intestine. This 367 approach evaluates the impact of dilution of ileal contents as they empty into the proximal 368 colon and the potential precipitation of weak acids, due to the decrease of the pH in the 369 proximal colon, particularly apparent in the fed state (34-36). To evaluate the importance of 370 specific luminal characteristics within a specific region of intestinal lumen two levels of 371 simulation of luminal composition were considered. Level I biorelevant media reflect luminal 372 pH and buffer capacity whereas Level II biorelevant media take additionally into account 373 luminal bile components and osmolality (35, 36). In addition, the importance of solid 374 particles [i.e. of Level III simulation] was evaluated (36). For the evaluation of the impact of 375 passive absorption from the lower intestine on the overall absorption process, in vitro 376 dissolution data collected under conditions simulating the environment in the upper 377 gastrointestinal lumen and under the conditions simulating the environment in the lower 378 intestinal lumen were coupled with physiologically based oral absorption modelling to 379 simulate the overall drug absorption process.

Based on data collected using high dose low solubility APIs and a colon targeting product,
dissolution characteristics in the lower intestine can be much different from that in upper
intestine with potential impact on PBPK modelling.

- 383 Dr Vertzoni concluded that in situations where stress effects are not expected to be of an
- issue (e.g. for immediate release products, pellets, products coated with pH sensitive
- 385 polymers) Level II or even Level I (if API is not very lipophilic) biorelevant media in
- 386 conjunction with the proposed two-stage in vitro methodology seem to be adequate for the
- 387 evaluation of dissolution in the lower intestine.

388 In Vivo Predictive Models for Oral Drug Absorption

- 389 Dr. Nikoletta Fotaki (University of Bath, UK) discussed the use of biorelevant in vitro data
- 390 within a physiologically-based pharmacokinetic (PBPK) model environment for the

391	prediction of <i>in vivo</i> performance with a focus on the points to be considered and the
392	challenges regarding the type of <i>in vivo</i> predictive data needed. Due to the pharmacokinetic
393	reasons for attrition in drug development the need for in vivo predictive in vitro tests and
394	the increased use of absorption modeling during drug development are evident (37). The
395	first aspects discussed related to the methodology of in vivo predictive solubility and
396	dissolution studies in terms of 1) the appropriate medium to be used (buffers,
397	pharmacopoeia media, biorelevant media), 2) the continuous update of the biorelevant
398	media based on physiological data (i.e. FaSSIF V1/ V2/ V3) and 3) the type of in vitro
399	dissolution apparatus to be used (USP dissolution apparatus I-IV and other approaches such
400	as Dissolution Stress Test Device, TNO Intestinal Models). A case study in which a
401	successful IVIVC for an immediate and a prolonged release formulation of a BCS Class II
402	compound was achieved based on appropriate selection of <i>in vitro</i> conditions (media,
403	apparatus) in combination with PBPK modeling was presented. The impact of in vitro
404	hydrodynamics on the development of in vitro-in vivo correlations for modified release
405	formulations of a BCS Class II compound, were discussed in the second case study (38). It
406	was shown that the hydrodynamics of USP apparatus II, III and IV may all be adequate as a
407	starting point for generating IVIVCs of up to 7 mm monolithic dosage forms with low drug
408	load, at least in the fasted state. The next point discussed related to the need of appropriate
409	in vivo predictive enzyme and transporter data apart from the solubility/ dissolution data in
410	the PBPK models. The third case study involved the development of a successful IVIVC
411	for an amorphous sustained release formulation of a BCS Class II compound based on
412	appropriate selection of <i>in vitro</i> conditions (media, apparatus) and enzyme/transporter data

413 in combination with PBPK modeling. In the cases that the compound undergoes in vivo 414 degradation, biorelevant *in vitro* degradation data has to be generated and used as an input 415 in the PBPK model. This was revealed through the fourth case study in which the 416 development of a successful IVIVC for an amorphous formulation of a BCS Class II 417 compound based on appropriate selection of *in vitro* conditions (media, apparatus) and *in* 418 vitro degradation data in combination with PBPK modeling was shown. In the last part of 419 her presentation she elaborated on the characterization of the dissolution of other 420 components of the formulation apart from the API, such as functional excipients or co-421 formers in co-crystals that can play a vital role in the assessment of bioavailability (39, 40)

422

423 *Physiologically Based Pharmacokinetic Simulations Integrating In Vitro Dissolution*

424 Results for Preclinical and Clinical Formulation Development

425 Dr. Neil Parrott (F Hoffmann LaRoche) presented a pharmaceutical industry perspective on the utility of physiologically based absorption models integrating biorelevant in vitro 426 427 dissolution data to guide formulation development. Within Roche, absorption modeling 428 plays a key role in biopharmaceutics sub-teams which are formed to address formulation 429 challenges in a project. The sub-teams bring together expertise in drug metabolism and 430 pharmacokinetics, clinical pharmacology and formulation and the models provides an 431 invaluable platform for integration of data, hypothesis generation and extrapolation. This is 432 illustrated with 2 case studies. The first shows how an oral absorption model, developed in 433 GastroPlus[™], can be verified with Phase 1 data for immediate-release capsules and then 434 applied to understand drug release from Phase 2 tablets and granules and to develop an in

435 vitro-in vivo correlation (IVIVC) model with biorelevant USP2 dissolution data and the 436 mechanistic absorption model (41). The second example covers the application of 437 physiologically based absorption modeling during the late stage clinical development and 438 filing of Alectinib (42). The modelling helped to predict and understand the impact of food 439 and gastric pH changes on Alectinib absorption.

440 The Impact of In Vivo Predictive Dissolution on Generic Drug Development and Review

441 Dr. Robert Lionberger (Food and Drug Administration, USA) presented that in vivo

442 predictive dissolution (IVPD) could have its highest impact on generic drugs and be a path

to expand access to generic competition. Many generic products are in small markets where

444 the cost of an *in vivo* bioequivalence study could be a significant barrier to entry. This is an

445 opportunity for IVPD to make a positive public health impact by supporting *efficient in*

446 *vitro* bioequivalence standards. FDA has guidance that provides for BCS biowaivers for

447 class 1 and 3 drug products, but BCS class 2 and 4 are where IVPD is the critical step.

448 IVPD needs to be linked closely with modeling and simulation of drug absorption and

449 distribution to fully characterize risks of bioavailability or bioequivalence differences.

450 Between 2013 and 2017 under GDUFA I, FDA has support a wide variety of research to

451 close some of the scientific gaps related to bioequivalence. As we move in to GDUFA II, it

452 is time to move toward implementation of IVPD for generic drugs.

453

443

454 **Future improvement and direction**

455 In order to understand the bioperformance of the drug substance and product of interest,

456 great progresses have been made in recent years. Scientists have been developing and

457	conducting science-centric researches to advance the area of <i>in vivo</i> prediction. Many
458	scientists agreed that in vivo predictive dissolutions and computational approaches would
459	be the right direction and the future to improve the oral drug dosage forms and to predict in
460	vivo plasma profiles. The development of decision tree to select an appropriate dissolution
461	methodology and experimental conditions for the test API formulation was extensively
462	discussed to direct the formulation and analytical scientists and to harmonize the in vitro
463	dissolution methodologies based on BCS and physicochemical properties. However, there
464	is no clear answer for the selection of one methodology over the other methodologies.
465	Thus, the different dissolution and simulation methodologies can be offered to scientists
466	and regulatory agents as a toolbox and they can freely select the methodologies for their
467	purposes.
468	Two important questions are: 1) if the scientific community can cross-validate their own
469	experimental/computational methodologies and/or harmonize their experimental
470	methodologies so that the results and agreements/disagreements could be discussed on the
471	same ground, and 2) if the scientific community and the regulatory community can develop
472	the field of new in vitro dissolution methodologies for bioequivalence and in vivo
473	predictive dissolution and harmonize the common ground. Academia, industry and
474	regulators should collaborate to derive the maximum benefit from in vivo predictive
475	dissolution and computational applications. It would be mutual benefit to all to expand our
476	knowledge and advance this area of sciences.
477	

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