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1 **Evolution of coal-bed methane in Southeast Qinshui Basin, China: Insights**
2 **from stable and noble gas isotopes**

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8 **Abstract**

9 The late Carboniferous-early Permian coal seams of the Qinshui Basin in Shanxi Province are
10 the most prolific producer of coalbed methane (CBM) in China. Methane formed in the late
11 Triassic during deep burial and reheating in late Jurassic-early Cretaceous driven by
12 magmatic underplating. Basin inversion brought the coal seams to 400-700 m from the
13 surface in the mid-late Cenozoic. Here we present results of a study aimed at understanding
14 the origin of the methane, and how it was affected by Cenozoic exhumation of the basin.
15 Methane from a 12 km traverse perpendicular to the basin margin in the southeast part of the
16 basin have stable isotope compositions ($\delta^{13}\text{C} = -30.2$ to -35.2% , and $\delta\text{D} = -155$ to -194%)
17 indicating a thermogenic origin with limited biogenic input. They are, however, lighter than
18 expected based on coal maturity, and $\text{C}_1/(\text{C}_2+\text{C}_3) (>1000)$ are significantly higher than typical
19 thermogenic methane (<50). This is due to diffusive fractionation during commercial gas
20 extraction. He-Ne-Ar isotopes are a mixture of crustal-radiogenic gas with air-derived noble
21 gases. ^4He concentrations (0.52 to 33.25 ppmv) and $^4\text{He}/^{40}\text{Ar}^*$ ratios (0.06-1.74) are unusually
22 low. He-Ne-Ar concentrations are consistent with the open system Rayleigh fractionation of
23 noble gases derived from air-saturated water with $^4\text{He}/^{40}\text{Ar}^* = 1$ during gas extraction. The
24 low $^4\text{He}/^{40}\text{Ar}^*$, compared with average crust (5) or local production (13) values, implies that
25 more than 90% of the radiogenic ^4He produced in the coals has been lost prior to equilibrium
26 between gas and water phase in the reservoir. This likely occurred in response to gas loss
27 process during rapid exhumation in Cenozoic, showing that the He and Ar content of natural
28 gases is a sensitive indicator of gas loss event caused by recent basin inversion. The event
29 may have led to the loss of up to 44% of the methane from the coal seams. This study
30 demonstrates the importance of basin inversion on gas preservation in shallow CBM, and
31 shows that, in contrast to $\delta^{13}\text{C}_{\text{CH}_4}$, the light noble gases are essential for tracing such a
32 process.

33

34 **Keywords**

35 Coalbed methane (CBM), Southeast Qinshui basin, noble gas isotopes, gas loss, high coal
36 rank

37 **Highlights**

- 38 • Gas compositions and stable isotopes indicate diffusive fractionation during
39 commercial gas extraction.
- 40 • There is no indication of mantle volatiles in gas samples, no evidence of mantle
41 heating during the Yanshanian Orogeny.
- 42 • The low radiogenic He and He/Ar ratio indicate loss of free gas during basin
43 exhumation.

44 **1. Introduction**

45 Coalbed methane (CBM) was first commercially exploited in USA in the late 1970s (Flores,
46 1998; Golding et al., 2013; Moore, 2012) and accounted for 8-10% of the national natural gas
47 production (Al-Jubori et al., 2009; USEIA, 2013). Australia and China commenced the
48 exploitation of CBM in 1990s, followed by Canada and India (Golding et al., 2013; Moore,
49 2012). The presence of shallow coalbeds in many parts of the world and the recent
50 developments on horizontal drilling techniques means that CBM can be cheaper to exploit
51 than other conventional onshore and offshore natural gas reservoirs (Golding et al., 2013;
52 Moore, 2012).

53 Coalbed methane is adsorbed on the coal matrix or absorbed in the micropores (Al-Jubori et
54 al., 2009; Hildenbrand et al., 2012; Rice, 1993). It can be generated by biogenic processes, at
55 the early stage of coalification by methanogens through acetate fermentation or CO₂
56 reduction (up to 70°C) (Whiticar, 1996), or it can be thermogenic in origin, produced by
57 thermal breakdown of kerogen with increasing temperature and further coalification (>150°C,
58 vitrinite reflectance (R_o) >0.5%) (Clayton, 1991; Strapoć et al., 2011). The gas yield is
59 suggested to be positively correlated with the coal rank (Moore, 2012; Rice, 1993; Zhang et
60 al., 2008). Generally, the adsorption capability of coal to gas also increases with higher coal
61 rank if all other variables (e.g. coal ash, moistures) are equal (Laxminarayana and Crosdale,
62 1999; Moore, 2012). The inversion of coal-bearing basins results in a decrease of the
63 reservoir pressure that can induce gas loss by desorption, diffusion and free gas flow (Rice,
64 1993; Xia and Tang, 2012). Interaction with shallow groundwaters can result in the
65 generation of secondary biogenic methane, the consumption of methane and wet
66 hydrocarbons (C₂₊) and the dissolution and removal of methane (Qin et al., 2006; Rice, 1993;
67 Strapoć et al., 2007; Tao et al., 2007).

68 The stable isotope and major gas compositions of coalbed gases have been used to constrain
69 their origin and accumulation history (Aravena et al., 2003; Flores et al., 2008; Hoşgörmez et
70 al., 2002; Kinnon et al., 2010; Rice, 1993; Strapoć et al., 2007; Zazzeri et al., 2016; Zhang et
71 al., 2018). The methane is generally accompanied by minor amounts of wet gases, carbon
72 dioxide and nitrogen. CBM is generally deficient in wet gases compared to conventional
73 natural gases (Song et al., 2012). The stable isotope composition of CBM worldwide is
74 highly variable ($\delta^{13}\text{C}_{\text{CH}_4} = -83$ to -17‰ , $\delta\text{D}_{\text{CH}_4} = -415$ to -117‰) reflecting the formation
75 mechanism and temperature, and the basin history (Aravena et al., 2003; Flores et al., 2008;
76 Rice, 1993; Song et al., 2012).

77 The concentration and isotopic composition of the noble gases (He, Ne, Ar, Kr and Xe) in
78 natural gases provide complementary information on gas source and the physical processes
79 that have acted upon them (Ballentine and O'Nions, 1992; Ballentine et al., 1991; Barry et al.,
80 2016; Battani et al., 2000; Byrne et al., 2018; Sherwood Lollar et al., 1997; Torgersen and
81 Kennedy, 1999; Xu et al., 1995; Zartman et al., 1961). They have, however, only been used
82 sparingly in studies of CBM. Zhou et al. (2005) demonstrated that the isotopic and elemental
83 ratios of noble gases fractionate during the gas extraction process in San Juan Basin, and they
84 developed a model on gas-water interaction in the reservoir and its influence on CBM
85 production. The ^4He content in CBM gas samples has been used to determine the age of
86 formation waters in reservoirs to constrain the timescale of hydrodynamic processes which
87 have affected hydrocarbon preservation (Zhou and Ballentine, 2006). Györe et al. (2018)
88 argued the influence of external heating on gas generation based on the presence of mantle-
89 derived noble gases in CBM from central Scotland, and the possibility of using high helium
90 concentrations in CBM (100s-1000s ppmv) to monitor gas emission during CBM extraction
91 activities. Noble gases in CBM from Illinois Basin also has been used to identify and quantify
92 the exogenous thermogenic gas in the reservoir, and improve the understanding of the origin
93 of methane in the reservoir (Moore et al., 2018).

94 The Qinshui Basin is the largest coal bed methane basin in China. It has proven reserve of
95 $4.35 \times 10^{11} \text{ m}^3$ methane and by the end of 2017 it was responsible for more than 70% of
96 Chinese national CBM production (Song et al., 2018). The main commercial CBM extraction
97 focuses on the coal seams in southeast part of the basin (SQB) (Su et al., 2005) where the
98 pore pressure is approximately equal to the hydrostatic pressure (Meng et al., 2011). The coal
99 of the SQB is high rank (vitrinite reflectance, $R_o=2.4-4.5\%$) which is believed to have been
100 the result of a thermal pulse caused by the intrusion of mantle-derived melts in the late
101 Mesozoic (Song et al., 2018; Su et al., 2005). The SQB gases are typically methane-rich (>
102 95%) (Li et al., 2014; Zhang et al., 2018; Song et al., 2018) with C isotopic compositions that
103 imply a thermogenic origin with only minor contributions of biogenic methane (Li et al.,
104 2014; Su et al., 2005; Zhang et al., 2018; Zhang et al., 2019). The basin exhumed to its
105 current state during the Cenozoic (Cao et al., 2015; Ren et al., 2005; Zeng et al., 1999). The
106 extent to which this affected CBM preservation is poorly understood. Thus, the basin is an
107 ideal natural laboratory to detect the role of recent exhumation to gas migration from the
108 reservoir rocks.

109 In this study we use the C, H isotopes of methane and noble gas isotope composition of well
110 gases from the SQB to determine the role the late Mesozoic mantle-derived melts may have
111 played in methane generation, and to understanding the extent to which the high coal rank
112 and Cenozoic exhumation have played in accumulation of the CBM.

113 **2. Geological setting**

114 The Qinshui Basin in Shanxi Province, northeast China, is a syncline of Palaeozoic-Mesozoic
115 sediments in the central belt of the North China craton (Figure 1) (Cai et al., 2011; Song et
116 al., 2018; Su et al., 2005; Zhang et al., 2015). Coal deposition started in the Late
117 Carboniferous in response to marine transgression and ended in the Early Permian with
118 fluvial-deltaic sedimentation (Ma et al., 2016; Su et al., 2005). Subsidence continued until the
119 Late Triassic by which time the coal had reached $\sim 135^{\circ}\text{C}$ (corresponding to more than 4 km
120 depth), inducing the first period of methane generation and maturation of coal to medium
121 volatile bituminous ($R_o \approx 1.2\%$) (Cai et al., 2011; Zeng et al., 1999). The Indosinian orogeny
122 resulted in uplift and basin inversion in the early Jurassic, followed by modest sedimentation
123 until Late Jurassic (Ma et al., 2016; Zeng et al., 1999). Magmatic activity during the
124 Yanshanian orogeny (Late Jurassic to Early Cretaceous) is evident from diorite porphyry
125 bodies that are exposed at the margin of the basin. These are widely accepted to have caused
126 a second peak of gas generation (Ren et al., 2005; Xu et al., 2004; Zeng et al., 1999). Zircon
127 fission track data suggest that temperatures reached over 250°C , which resulted in the
128 conversion of the bituminous coals to semi-anthracite and anthracite with R_o values reaching
129 2.2-4.5% (Ren et al., 2005; Su et al., 2005). The basin underwent a major phase of
130 exhumation during the Cenozoic that is ultimately related to the extensional tectonic regime
131 induced by the subduction of the Indian Plate in the southwest, or roll back of the subducting
132 Pacific Plate in the east (Cai et al., 2011; Cao et al., 2015). Fission tracks in apatites from
133 sediments overlying coal seams in the SQB reveal that the basin cooled from over 250°C to
134 $\sim 100^{\circ}\text{C}$ in the last 50 million years then experienced a pulse of rapid cooling to less than
135 60°C in the last 11 million years (Cao et al., 2015; Ren et al., 2005). Denudation is continued
136 until at least 5 million years (Cao et al., 2015).

137 The main coal-bearing strata are the 50-135 m thick Pennsylvanian Taiyuan Formation and
138 the 20-40 m thick Early Permian Shanxi Formation (Figure 2) (Su et al., 2005). Coal seam 3
139 (2-7 m thick) in the Shanxi Formation ($R_o=2.5-4.5\%$) and coal seam 15 (1-6 m thick) in the
140 Taiyuan Formation ($R_o=2.7-4.5\%$) are present across the whole basin and are the most
141 economically important coals (Song et al., 2018; Su et al., 2005). The Taiyuan Formation

142 coal seams are mostly located at less than 700 m depth, approximately 80 m deeper than the
143 main Shanxi Formation coals. In the study region, a third major coal seam (no. 9) in the
144 Taiyuan Formation, which is about 40 m deeper than coal seam 3, is co-exploited with seams
145 3 and 15 (Wang et al., 2013).

146 The main hydrological influence on coal seams is sourced from the confined C_{3t} aquifer and
147 P_{1s}-P_{1x} aquifer (Figure 1D) (Zhang et al., 2015). The no. 9 and 15 coal seams in the Taiyuan
148 Formation are isolated from the underlying aquifer by bauxite and shale in Benxi Formation
149 (Figure 2) (Zhang et al., 2015). The roof of the no. 15 coal seam is a 2-30 km thick limestone,
150 belonging to the C_{3t} aquifer. The roof and floor of mudstones isolated the no.9 coal seam
151 (Wang et al., 2013). The no. 3 coal seam belongs to P_{1s}-P_{1x} aquifer (Zhang et al., 2015). The
152 roof and bottom of the no. 3 coal seam are fine sandstone and mudstone, respectively, which
153 seals the layer as an individual aquifer (Su et al., 2005). The outcrops of the east and south
154 coals in the SQB are the main source of recharging water (Su et al., 2005; Zhang et al., 2015).
155 The groundwater flows northwest and is prevented by a gas and water-sealed fault, Sitou
156 Fault (Su et al., 2005).

157 **3. Sampling and analytical techniques**

158 Nineteen gas samples were taken in April 2017 from wellheads along an E-W transect
159 perpendicular to the boundary fault in the Panzhuang block (Figure 1). This transect is
160 approximately along the groundwater recharge pathway. Seventeen gas samples are from
161 vertical wells, all but one combines the three main coal seams. Well gas Qs19 is from coal
162 seam 15. only. Two gas samples (Qs15 and Qs17) are from horizontal wells within coal seam
163 15. Well depths vary from 300 to 650 m (Table 1). The wells have been producing methane
164 for between 3 months and 15 years (Table 1).

165 Samples were collected and stored in Cu tubes at 1.5-2 bar using the method described in
166 Györe et al. (2015). Approximately half of the gas in each tube was used for the
167 determination of the major gas composition and stable isotopes. Gases were expanded into a
168 glass gas-purification system ($p < 0.01$ mbar) and an aliquot was extracted by a 100 μ L
169 Hamilton syringe. Major gas composition was determined in nine gas samples by a newly set
170 up Hewlett Packard 5890 gas chromatograph (GC) in SUERC. It is equipped with a single
171 filament thermal conductivity detector, a 2.13 m long, 1.0 mm internal diameter Restek
172 100/120 packed column operated with helium carrier gas. The temperature of the column
173 during analysis was kept at 30°C for 3 minutes and increased to 150°C at a rate of

174 15°C/minute and then held at the maximum temperature for 5 minutes. The reproducibility of
175 $\text{CH}_4/(\text{C}_2\text{H}_6 + \text{C}_3\text{H}_8)$ and CH_4/CO_2 ratios is $\pm 1\%$.

176 The carbon isotopic composition was determined using a VG SIRA II dual inlet isotope ratio
177 mass spectrometer (Dunbar et al., 2016) using established procedures in Györe et al. (2018).
178 $\delta\text{C}_{\text{CH}_4}$ values were determined relative to PDB (Craig, 1957). The hydrogen isotope
179 composition of the water vapour was analysed by a VG Optima dual inlet isotope ratio mass
180 spectrometer (Donnelly et al., 2001). $\delta\text{D}_{\text{CH}_4}$ values were determined relative to V-SMOW
181 (Gonfiantini, 1978). Experimental uncertainties (1σ) of $\delta^{13}\text{C}_{\text{CH}_4}$ and $\delta\text{D}_{\text{CH}_4}$ values are $\pm 0.3\%$
182 and $\pm 3\%$, respectively.

183 The remaining gas was used for noble gas analysis using a MAP 215–50 mass spectrometer
184 following procedures described elsewhere (Györe et al., 2015). Blank levels for all isotopes
185 are negligible compared with the signal of samples. Mass spectrometer sensitivity and mass
186 fractionation were determined using the HESJ standard for He (Matsuda et al., 2002), and air
187 (Eberhardt et al., 1965; Györe et al. 2019; Lee et al., 2006; Mark et al., 2011; Ozima and
188 Podosek, 2002) for Ne and Ar. The reproducibility of the absolute concentrations is $< 4\%$, and
189 isotopic ratios are less than 1%.

190 **4. Results**

191 Methane is the dominant gas ($> 95\%$) in all samples, with minor N_2 and trace CO_2 . The wet
192 gases (C_2H_6 , C_3H_8) were below detection limit (0.1%) of the GC, which indicates the
193 $\text{CH}_4/(\text{C}_2\text{H}_6 + \text{C}_3\text{H}_8)$ ratio is higher than 1,000. The CH_4/CO_2 ratio varies from 152 to 806
194 (Table 1). The carbon and hydrogen isotopic compositions of methane range from -30.2 to -
195 35.2‰ and -155 to -194‰, respectively (Table 1).

196 ^4He concentrations range from 0.52 to 33.3 ppmv. $^3\text{He}/^4\text{He}$ ratios vary from 0.009 ± 0.002 to
197 $0.482 \pm 0.007 R_A$, where R_A is the atmospheric ratio of 1.34×10^{-6} (Mishima et al., 2018).

198 ^{20}Ne concentrations range from 0.001 to 5.61 ppmv. Neon isotopes appear to be largely air-
199 derived without evident mantle or crustal contribution; $^{20}\text{Ne}/^{22}\text{Ne}$ and $^{21}\text{Ne}/^{22}\text{Ne}$ ratios vary
200 from 9.69 ± 0.09 to 10.18 ± 0.03 and from 0.0285 ± 0.0005 to 0.0301 ± 0.0007 , respectively.

201 Radiogenic ^{21}Ne is unequivocally present in Qs6, Qs21 and Qs22. ^{40}Ar concentrations range
202 from 23 to 2290 ppmv. $^{40}\text{Ar}/^{36}\text{Ar}$ range from 291 ± 1 to 497 ± 1 , with samples Qs7, Qs 9 and
203 Qs14 having ratios that are lower than the air value (298.6 ± 0.3 ; Lee et al. (2006); Mark et
204 al. (2011)). $^{38}\text{Ar}/^{36}\text{Ar}$ ratios are indistinguishable from air (0.1885 ± 0.0003 ; Lee et al. (2006);
205 Mark et al. (2011)) (Table 2). The gas from coal seam 15 (Qs15, Qs17 and Qs19) is not

206 evidently distinct from the combined-seam gases in gas compositions, stable isotopes and
207 noble gases.

208 **5. Discussion**

209 **5.1. The post-generation fractionation of molecular and stable isotopic compositions**

210 The C, H isotopic compositions ($\delta C_{CH_4} = -30.2$ to -35.2‰ and $\delta D_{CH_4} = -155$ to -194‰) of the
211 methane is indicative of a thermogenic origin (Figure 3). However, based on the coal
212 maturity in SQB, the lowest $\delta^{13}C_{CH_4}$ value is predicted to be -27‰ (Hu et al., 2001; Li et al.,
213 2014). This is up to 8‰ higher than measured values and hints at modification by other
214 process(es). Stable isotopes of the SQB methane are apparently lighter than reported for
215 methane from other CBM reservoirs of similar maturity (Kotarba and Rice, 2001; Schoell,
216 1980). Further, C_1/C_{2+} ratios are significantly higher than typical thermogenic gases (<50 ;
217 Bernard et al. (1976)). The abnormally lighter stable isotopes and high C_1/C_{2+} in SQB have
218 been commonly reported (Chen et al., 2007; Hu et al., 2001; Li et al., 2014; Qin et al., 2006;
219 Zhang et al., 2018).

220 Similar major gas and methane C isotope values for SQB gases have previously been
221 explained as due to the preferential dissolution of $^{13}CH_4$ and heavy hydrocarbons in
222 groundwater (Li et al., 2014; Qin et al., 2006). However, this is difficult to reconcile.
223 Experimental work has demonstrated that dissolution of methane should result in only small
224 depletion of $\delta^{13}C$ ($< -0.5\text{‰}$) in the free methane phase, and a δD enrichment ($< -16\text{‰}$) that is
225 in the opposite sense to that recorded by the methane (Bacsik et al., 2002).

226 The addition of secondary biogenic methane has been proposed in several studies to explain
227 the high C_1/C_{2+} and light $\delta^{13}C$ of SQB methane (Chen et al., 2007; Li et al., 2014; Rice, 1993;
228 Zhang et al., 2018). The light $\delta^{13}C_{CH_4}$ (up to -61.7‰) of methane desorbed from SQB coal
229 cores supports the addition of biogenic methane. However, fewer than 10% of all SQB
230 samples have $\delta^{13}C_{CH_4}$ lighter than -40‰ . Using an upper limit of $\delta^{13}C = -50\text{‰}$ for biogenic
231 methane, and $\delta^{13}C = -27\text{‰}$ for thermogenic methane (Li et al., 2014), the methane sampled in
232 this study should be up to 50% biogenic in origin. The factor of two difference in C_1/C_{2+} is
233 not high enough to explain the complete absence of heavy hydrocarbons in most of the SQB
234 gases in this study. Thus, while there may be modest contributions of biogenic methane, it is
235 unlikely to be the key to the ubiquitous high C_1/C_{2+} and light $\delta^{13}C_{CH_4}$.

236 Desorption and diffusion of deep methane to shallow depths during basin exhumation has
237 also been proposed to explain the high C_1/C_{2+} and light $\delta^{13}C_{CH_4}$ (Chen et al., 2007; Zhang et

238 al., 2018; Zhang and Tao, 2000; Zhang et al., 2019). The $\delta^{13}\text{C}$ of methane gas desorbed for 4
239 hours from SQB anthracite is 8‰ lighter than the methane collected after 96 hours of
240 desorption, supporting the hypothesis that light methane readily desorbs at early stages
241 (Zhang and Tao, 2000). The negative trends in $\delta^{13}\text{C}_{\text{CH}_4}$ vs. C_1/C_{2+} and $\delta^{13}\text{C}_{\text{CH}_4}$ vs. C_1/CO_2
242 (Figure 4A, B) reveal that desorption and diffusion have also caused molecular fractionation
243 as C_{2+} and CO_2 are harder to desorb and diffuse in coalbeds than CH_4 (Bae and Bhatia, 2006;
244 Rice, 1993). However, if isotopically light methane migrates up-seam (Zhang et al., 2019),
245 there should be a positive relationship between coal seam depth and $\delta^{13}\text{C}_{\text{CH}_4}$, C_1/C_{2+} ratio and
246 C_1/CO_2 . This is not apparent in this study (Table 1). A positive relationship between $\delta^{13}\text{C}_{\text{CH}_4}$
247 and gas production rate is evident from the dataset (Figure 4C). This implies that the change
248 of $\delta^{13}\text{C}_{\text{CH}_4}$ has occurred during natural gas extraction. At the early stage of gas extraction, the
249 most easily desorbed gases are produced, and as production continues and the rate of gas
250 production rises, the isotopically heavier CH_4 and the more adsorbent gases (C_{2+} , CO_2) are
251 released (Figure 4).

252 It is important to note that even when $\delta^{13}\text{C}_{\text{CH}_4}$ approaches the predicted value, the C_1/C_{2+} ratio
253 (> 400) is evidently higher than typical value (< 50) (Figure 4A) (Zhang et al., 2018). Even
254 considering the addition of biogenic methane, the maximum C_1/C_{2+} ratio can only be two
255 times higher (< 100). Ethane and propane are cracked to methane by pyrolysis or the
256 existence of redox couples at temperature in excess of 250°C (Burruss and Laughrey, 2010).
257 The absence of heavy hydrocarbon gases in SQB is thus likely to be the result of cracking of
258 wet gases into methane at high temperature. This is consistent with the detrital zircon fission
259 track thermochronology that indicates such temperatures were attained during the early
260 Cretaceous (Ren et al., 2005). This gives further credence to the idea that the methane
261 generation peak occurred during the Yanshanian Orogeny.

262 **5.2. Identifying the noble gas components**

263 The He and Ne in the SQB gases appear to plot on mixing lines between a deep gas with high
264 He/Ne (> 700)-low $^3\text{He}/^4\text{He}$ ($< 0.03 R_A$) and noble gases with composition that can be
265 derived from mass fractionated air (Figure 5A). The five gases with the lowest $^4\text{He}/^{20}\text{Ne}$
266 (grey circles in Figure 5A) have air-like $^{40}\text{Ar}/^{36}\text{Ar}$ ratios and high concentrations of ^{20}Ne and
267 ^{36}Ar (Table 2) that are consistent with being dominated by air-derived noble gases. Both the
268 high and low He/Ne samples have a similar range of ^4He concentrations but the air-rich (low
269 He/Ne) gases have higher ^{20}Ne concentrations (Figure 5B), suggesting that air-derived noble
270 gases have been added to deep gas. As well as fractionated He/Ne (Figure 5A), the five air-

271 rich samples have Ne isotopes that plot along the mass fractionation line in the $^{20}\text{Ne}/^{22}\text{Ne}$ vs.
272 $^{21}\text{Ne}/^{22}\text{Ne}$ space (Figure 6), and three samples have $^{40}\text{Ar}/^{36}\text{Ar}$ that are lower than air value,
273 also suggestive of fractionation (Table 2). The fractionated air-derived light noble gases may
274 originate in liquids injected during hydraulic fracturing before pumping (Barry et al., 2018)
275 or acquired during the sampling process. It dominates the noble gases of the five low He/Ne
276 samples and these samples are not discussed further.

277 The high $^4\text{He}/^{20}\text{Ne}$ ratio (37-691) of the remaining samples rules out a significant
278 contribution of air. After correcting the measured $^3\text{He}/^4\text{He}$ for minor air-derived He on the
279 basis of the atmospheric He/Ne (0.32), these gases yield values of between 0.008 and 0.035
280 R_A . This range is typical of crust-derived gases where radiogenic production of ^4He occurs by
281 U and Th decay and nucleogenic production of ^3He by the reaction $^6\text{Li}(n,\alpha)^3\text{H}(\beta^-)^3\text{He}$
282 (Morrison and Pine, 1955). Using the range of Li measured in coal seam 15 (12-78 ppm, $n=12$),
283 and the concentration of the large neutron cross-section elements (B, Be, Nd, Gd, etc.)
284 in the Jincheng coals which is the closest coal mine to the study site (Zhang et al., 2004;
285 Zhao, 1997; Zhao et al., 2018), we calculate a $^3\text{He}/^4\text{He}$ production ratio of 0.015 to 0.085 R_A
286 (Figure 5A). This covers much of the range measured in the high He/Ne gases. The extremely
287 low $^3\text{He}/^4\text{He}$ of samples Qs18 and Qs22 implies that, locally, Li concentrations may be lower,
288 or the concentration of large neutron cross-section elements may be higher.

289 There is a possibility that the range of $^3\text{He}/^4\text{He}$ is due to the presence of a small contribution
290 of mantle-derived He ($^3\text{He}/^4\text{He}_{\text{magma}} = 6-7 R_A$; Gautheron and Moreira (2002)) from the early
291 Cretaceous Yanshanian magmatism (Ren et al., 2005; Xu et al., 2004). Using the lowest
292 measured $^3\text{He}/^4\text{He}$ (0.008 R_A) as representative of the crustal radiogenic He composition, and
293 6.1 R_A to represent the magmatic end-member, the highest measured $^3\text{He}/^4\text{He}$ requires a
294 maximum mantle He contribution of 0.5%. This is trivial and provides no support for the
295 contention that the main methane generation occurred in response to magmatic heating.
296 Mantle-derived Ne is present in CBM deposits from sedimentary basins that have been
297 intruded by magmatic bodies (e.g. Györe et al. 2018). The absence of mantle-derived Ne in
298 the SQB methane is consistent with the He isotope record.

299

300 **5.3. Helium loss during Cenozoic exhumation**

301 Compared with other CBM reservoirs of a similar age (e.g., Illinois coals, Moore et al.
302 (2018); Airth coals, Györe et al. (2018)), the ^4He concentrations in the high $^4\text{He}/^{20}\text{Ne}$ SQB
303 gases are extremely low (Figure 7A). However, the concentration of radiogenic ^{40}Ar (denoted

304 as $^{40}\text{Ar}^*$ and calculated by removing atmospheric ^{40}Ar and assuming negligible magmatic
 305 $^{40}\text{Ar}^*$) in the SQB natural gases overlaps the range recorded by CBM from similar age coals
 306 above (Figure. 7B). Consequently, $^4\text{He}/^{40}\text{Ar}^*$ ratios (0.06 to 2) in the SQB gases are lower
 307 than the average production ratio of the upper continental crust (4.9, Ballentine and Burnard
 308 (2002)) or the local coals (> 13), calculated using the lowest U (0.5 ppm) and Th (0.6
 309 ppm) (Zhang et al., 2004) and highest K concentration (0.16%, Zhao (1997)).

310 $^4\text{He}/^{40}\text{Ar}^*$ in natural gases are commonly higher than produced in the shallow crust
 311 (Ballentine et al., 1994), a function of the low temperature mobility of He relative to Ar.
 312 Considering the evident fractionation of C isotope composition and the C_1/C_{2+} ratio of the
 313 SQB gases during pumping process, it is reasonable to assume that $^4\text{He}/^{40}\text{Ar}^*$ ratio also
 314 experienced fractionation during this process. We use the Rayleigh fractionation law (Eq. 1)
 315 to test it:

$$316 \left(\frac{A}{B}\right)_w = \left(\frac{A}{B}\right)_o f^{(\alpha-1)} \quad (1)$$

317 where $\left(\frac{A}{B}\right)_w$ is the current elemental ratio in the water phase, $\left(\frac{A}{B}\right)_o$ is the initial elemental ratio
 318 in the water phase, f is the proportion of residue B in the water, and α is the fractionation
 319 factor. Assuming the ratio of water to gas volume (V_w/V_g) at each degassing stage is close to
 320 zero, the partition of noble gases reaches equilibrium. The elemental ratios of noble gas in
 321 gas bubbles can be estimated by Eq.2 :

$$322 \left(\frac{A}{B}\right)_g = \left(\frac{A}{B}\right)_w \alpha \quad (2)$$

323 where $\left(\frac{A}{B}\right)_g$ is the elemental ratio in the gas. Initial $^{20}\text{Ne}/^{36}\text{Ar} = 0.154$ in formation water is
 324 calculated from ASW equilibrated 10°C at an altitude of 740 m, with 10% Ne excess air
 325 (Kipfer et al., 2002; Peeters et al., 2003). The fractionation factor is calculated for the
 326 Henry's constants of noble gas and corrected from liquid phase activity coefficients and gas
 327 phase fugacity coefficients (Ballentine et al., 2002; Crovetto et al., 1982; Smith and Kennedy,
 328 1983) in groundwater at 27°C and 4 MPa, and salinity of 0.04 mol/L (Wang et al., 2015). If
 329 the kinetic diffusion under non-equilibrium condition controls the noble gases partition, the
 330 Rayleigh equation is still applicable, only with the fractionation factor proportional to the
 331 inverse of the square root of the reduced mass which could be assumed to be equal to the
 332 mass of noble gas isotope (Ballentine et al., 2002; Zhou et al., 2005). However, the extent of
 333 $^{20}\text{Ne}/^{36}\text{Ar}$ change for mass-controlled fractionation cannot explain the range observed in the
 334 samples (Figure 8). The least air-contaminated SQB gases (i.e. $^4\text{He}/^{20}\text{Ne} > 400$) plot on a

335 trend in $^4\text{He}/^{40}\text{Ar}^* - ^{20}\text{Ne}/^{36}\text{Ar}$ space that is consistent with open system Rayleigh fractionation
336 of gas with $^{20}\text{Ne}/^{36}\text{Ar}$ of air-saturated water (ASW) based on local conditions (Figure 8). This
337 demonstrates that the radiogenic ^4He and ^{40}Ar in the reservoir were dissolved in the formation
338 water and degassed during CBM extraction. Formation water with an initial $^4\text{He}/^{40}\text{Ar}^*$ ratio of
339 13.0 (local production in coals) and 4.9 (average continental crust) provide poor fits to the
340 data. The best fit to the data is derived for an initial $^4\text{He}/^{40}\text{Ar}^*$ that approaches a value of 1.0.
341 This implies that the formation water had an abnormally low $^4\text{He}/^{40}\text{Ar}^*$ prior to commercial
342 extraction of methane.

343 The main storage mechanism of gas in coal is by adsorption on the coal matrix or absorption
344 in the micro-pores (Al-Jubori et al., 2009; Hildenbrand et al., 2012; Rice, 1993). The rapid
345 uplift of sedimentary basins induces a drop of reservoir pressure, weakening the sorption
346 capability of coal (Rice, 1993; Xia and Tang, 2012). Adsorbed and absorbed gases are
347 released (desorbed) and accumulate in coal microstructures, generating a pressure gradient
348 that results in gas escape by diffusion or viscous flow (Hildenbrand et al., 2012; Rice, 1993).
349 Gas escape is mainly controlled by the permeability of the coals and pressure gradient
350 between gas phase and wetting phase (Hildenbrand et al., 2012). The rapid Cenozoic
351 inversion of the Qinshui Basin, from over 3 km to ~400 m deep, is likely to have resulted in
352 episodic gas expulsion from the coal seams. A series of detailed gas breakthrough
353 experiments on high rank coals ($R_o = 2.3\%$) from the Qinshui Basin mimicked the effect of
354 over-pressure (Han et al., 2010a). It revealed that at reservoir conditions, He and Ar could
355 escape from wet-coal matrix due to high pressure gradient by mainly diffusion and rarely
356 capillary-viscous flow if macro fractures (e.g. cleats) existed in coalbeds (Han et al., 2010a).
357 They also showed that the effective permeability of He for diffusion is significantly higher
358 than for Ar (Han et al., 2010a). Consequently, during continuous and intensive basin
359 inversion, Ar is likely to have remained trapped in the coalbeds while most of the He lost.
360 Further, the solubility of Ar in water is higher than He (Crovetto et al., 1982; Smith and
361 Kennedy, 1983). Thus the preservation of old formation waters in the SQB (Wang et al.,
362 2018) will tend to have further reduced the $^4\text{He}/^{40}\text{Ar}^*$. The loss of He after magmatic events
363 explains that absence of mantle He in the SQB methane.

364 ^{21}Ne excess (denoted as $^{21}\text{Ne}^*$) in samples is generated by nucleogenic reactions, $^{18}\text{O}(\alpha, n)^{21}\text{Ne}$
365 and $^{24}\text{Mg}(n, \alpha)^{21}\text{Ne}$, in the coals (Yatsevich and Honda, 1997). Only samples Qs6, Qs21 and
366 Qs22 have evident $^{21}\text{Ne}^*$ excess ($9.6 \pm 3.3 \times 10^{-14}$ to $3.1 \pm 1.6 \times 10^{-13}$ cm^3 STP/ cm^3) beyond
367 analytical uncertainty. It gives the $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ ratio of $1.1 \pm 0.6 \times 10^{-8}$ to $2.0 \pm 1.0 \times 10^{-8}$. The

368 minimum theoretical local production of $^{21}\text{Ne}^*/^{40}\text{Ar}^*$ in SQB is estimated to be 7.8×10^{-8} based
 369 on O (5.5%), Mg (0.4%) (Yin et al., 2016; Zhao, 1997) and lowest U, Th concentrations in
 370 coals. It is far higher than calculated value in the samples. The absence of crustal-radiogenic
 371 or mantle Ne in the samples, demonstrating that most of the non-atmospheric Ne was lost
 372 with ^4He .

373 As the He in the SQB well gases has mainly degassed from the coal formation water, the
 374 accumulation time of ^4He can be estimated (Craig and Lupton, 1976; Torgersen, 1980; Zhou
 375 and Ballentine, 2006) from Eq 3 & 4:

$$376 \quad [He]_{in\ situ} = \frac{\rho \Lambda J (1 - \varphi)}{\varphi} t \quad (3)$$

$$377 \quad J = 0.2355 \times 10^{-12} [U] \{1 + 0.123 ([Th]/[U] - 4)\} \quad (4)$$

378 where $[He]_{in\ situ}$ is the *in situ* production rate in $\text{cm}^3 \text{STP } ^4\text{He}/(\text{cm}^3_{\text{H}_2\text{O}} \text{ yr})$, ρ is the density of
 379 the coal in g/cm^3 , Λ is the fraction of He generated in the coal that enters the pore fluid and
 380 assumed to be 1 following Zhou and Ballentine (2006). φ is the porosity of the coal and J is
 381 the production rate of ^4He in the coal in $\text{cm}^3 \text{STP } ^4\text{He}/(\text{cm}^3_{\text{H}_2\text{O}} \text{ yr})$. [U], [Th] are the
 382 concentrations in the coal seams in ppm. Assuming that the coals have density of $1.6 \text{ g}/\text{cm}^3$,
 383 porosity of 5% (Cai et al., 2011), and all the ^4He has been generated in rock with average U
 384 and Th concentrations of 3.9 ppm and 8.5 ppm, respectively (Zhang et al., 2004), it gives
 385 the *in situ* production rate of $1.8 \times 10^{-11} \text{ cm}^3 \text{STP } ^4\text{He}/(\text{cm}^3_{\text{H}_2\text{O}} \text{ yr})$ in SQB coals. As He-Ne-Ar
 386 follows the open system Rayleigh fractionation, with assumption that the original $^{20}\text{Ne}/^{36}\text{Ar}$
 387 in formation water is ASW (0.154), the residual ^{36}Ar (f) in the formation water can be
 388 calculated based on equation 1 and 2, allowing the initial $^4\text{He}/^{36}\text{Ar}$ to be calculated. The
 389 initial ^{36}Ar in ASW under local conditions is about $1.3 \times 10^{-6} \text{ cm}^3 \text{STP}/\text{cm}^3_{\text{H}_2\text{O}}$. The initial
 390 concentration of ^4He in formation water before fractionation is therefore easily estimated and
 391 gives the He accumulation age of between 0.5 and 11 million years, which supports the
 392 contention that the majority of He has been lost.

393 This contrasts with the observed $^{40}\text{Ar}^*$ concentration. Using the maximum concentration of K
 394 (0.16%, Zhao (1997)) and assuming the transfer rate of Ar from rock to fluid of 1, it gives the
 395 maximum *in situ* production rate of $1.5 \times 10^{-13} \text{ cm}^3 \text{STP } ^{40}\text{Ar}^*/(\text{cm}^3_{\text{H}_2\text{O}} \text{ yr})$. As the solubility of
 396 $^{40}\text{Ar}^*$ and ^{36}Ar is indistinguishable at this scale, the calculated $^{40}\text{Ar}^*$ concentration in
 397 formation water requires 192-1,680 million years, which exceeds the deposition age of coal,
 398 and requires the incorporation of the crustal flux of ^{40}Ar flux and/or ^{40}Ar exsolved from

399 solidifying mantle melts (Castro et al., 1998; Györe et al., 2018; Torgersen et al., 1989). Both
400 of these processes will result in incorporation of ^4He in excess of *in situ* production.

401 Following the method of Zhou and Ballentine (2006), with total thickness of three coal seams
402 to be 8.6 m, we assume that the regional ^4He and $^{40}\text{Ar}^*$ fluxes are $4.9 \times 10^{-8} \text{ cm}^3 \text{ STP}$
403 $^4\text{He}/(\text{cm}^3_{\text{H}_2\text{O}} \text{ yr})$ and $8.6 \times 10^{-9} \text{ cm}^3 \text{ STP } ^{40}\text{Ar}^*/(\text{cm}^3_{\text{H}_2\text{O}} \text{ yr})$, respectively. With both *in situ* and
404 external production, the accumulated age for ^4He is $173 - 4 \times 10^3 \text{ yr}$ and $3.3 \times 10^3 - 2.9 \times 10^4 \text{ yr}$
405 for $^{40}\text{Ar}^*$. The older age defined by $^{40}\text{Ar}^*$ again supports the contention that He has been most
406 efficiently lost from SQB coals. The $^{40}\text{Ar}^*$ age overlaps the ^{14}C age of the production water
407 from other regions in SQB ($5 \times 10^3 - 4 \times 10^4 \text{ yr}$) (Wang et al., 2018).

408 Based on the local $^4\text{He}/^{40}\text{Ar}^*$ production rate of 13, the initial $^4\text{He}/^{40}\text{Ar}^*$ ratio of 1 before
409 fractionation indicates that over 92% of He has been lost during basin inversion. However,
410 the SQB coals have, obviously, retained economic quantities of methane. This reflects
411 different physical behaviour of CH_4 and the light noble gases. Compared with He and Ar,
412 methane is strongly adsorbed onto the coal and only moves by diffusion in SQB coalbeds
413 (Han et al., 2010a; Han et al., 2010b; Hildenbrand et al., 2012). A model which considers
414 both adsorption/desorption and diffusion during methane loss has been introduced by Xia and
415 Tang (2012). For total organic carbon of 71% (Yin et al., 2016), assuming overpressure of 60
416 MPa with 250 °C before basin inversion and final conditions of 4 MPa and 27 °C after basin
417 inversion, the free methane concentration decreases from 692 mol/m^3 to 81 mol/m^3 , assuming
418 the Peng-Robinson equation of state. The adsorbed gas amount drops from 786 mol/m^3 to
419 749 mol/m^3 following the Langmuir equation (Xia and Tang, 2012). Considering both the
420 loss of free and adsorbed gas, approximately 44% of the methane has been lost from the SQB
421 during basin inversion. Based on the model in Xia and Tang (2012), the $\delta^{13}\text{C}$ of the residue
422 methane would be less than 2‰ heavier than the initial composition. This is contrast with
423 observed lighter $\delta^{13}\text{C}_{\text{CH}_4}$ compared with predicted value. Thus, although the carbon isotope of
424 methane might have been influenced by gas loss event, it has been obscured by later
425 fractionation caused by desorption during gas extraction. This suggests that basin inversion
426 and gas loss are not recorded by stable isotopes (or major gas composition). Although the
427 noble gases experienced open system Rayleigh distillation during gas extraction as well,
428 through modelling, the gas loss could be revealed by light noble gases (He-Ne-Ar). In
429 particular, the radiogenic ^4He and $^{40}\text{Ar}^*$ provide a keen tracer of the timing of gas loss during
430 basin inversion.

431 **6. Conclusion**

432 The stable isotope and major gas composition of CBM from the Qinshui Basin indicates a
433 thermogenic origin with limited biogenic methane. The characteristics have been over-printed
434 by fractionation during gas extraction. The noble gases reveal a mixture of air-derived gases
435 from groundwater (ASW) with deep gases in the reservoir. The mixing of radiogenic ^4He and
436 ^{40}Ar and ASW derived noble gases occurred in water phase. Then, they degassed from water
437 following open system Rayleigh fractionation during gas extraction. A gas loss process which
438 happened before the commercial extraction can explain the significantly lower $^4\text{He}/^{40}\text{Ar}^*$ in
439 all samples even considering the fractionation process. Meanwhile it also explains the
440 absence of mantle volatiles in the reservoir although the basin was thought to be heated
441 during a magmatic activity in Yanshanian Orogeny (Late Jurassic to Early Cretaceous). The
442 He loss is related to the exhumation of the basin mainly in Cenozoic. Compared with He, Ar,
443 with lower diffusive efficiency and higher solubility, was partly stored in the coal and
444 dissolved in water, which caused the low $^4\text{He}/^{40}\text{Ar}^*$ ratio in formation waters. Although about
445 half of the methane also lost during basin inversion, the diagnostic of stable isotopes is weak
446 and easily covered by post fractionation processes. This study demonstrates the possibility of
447 using He-Ne-Ar content in natural gases to track gas evolution caused by recent basin
448 activities.

449

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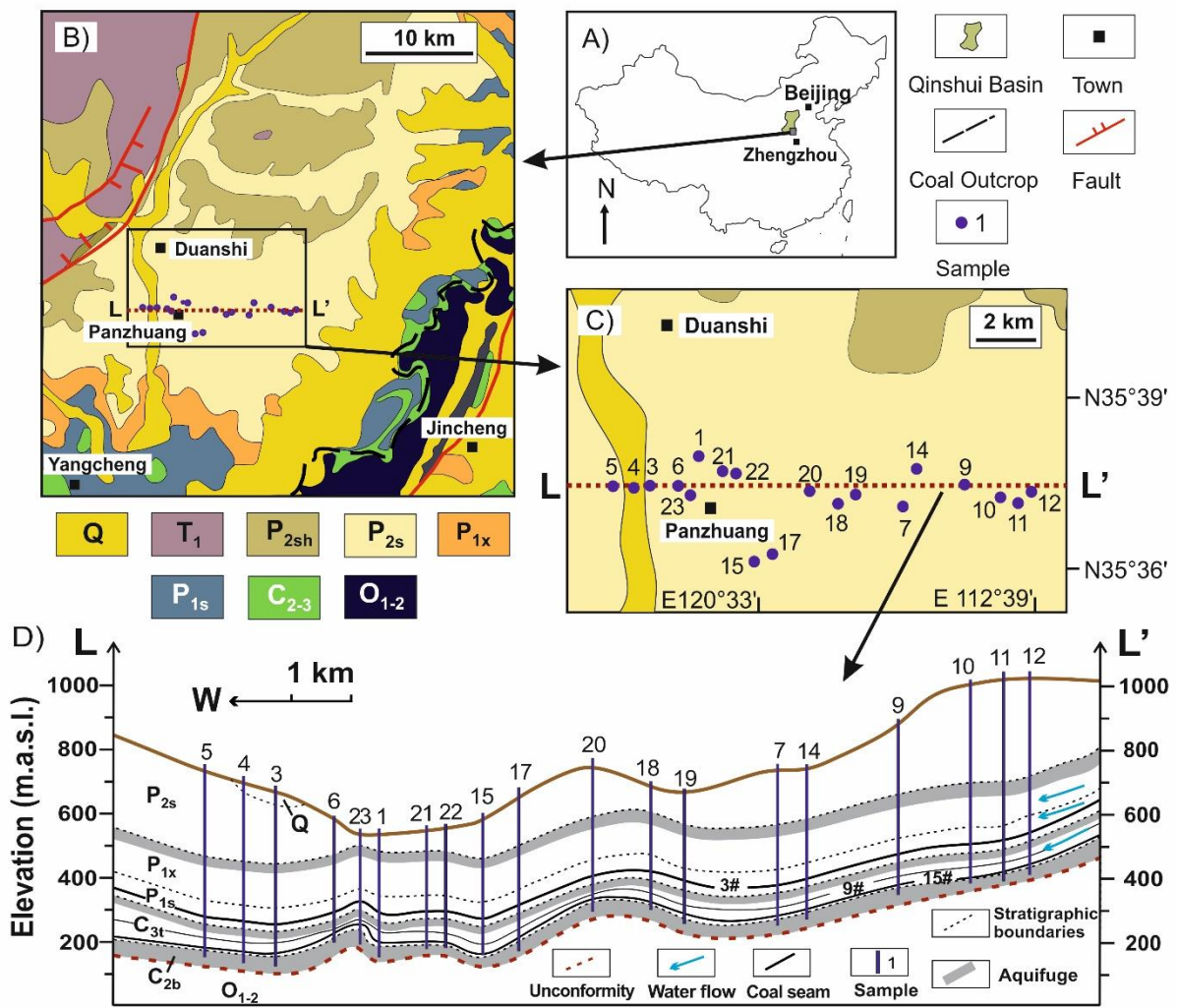
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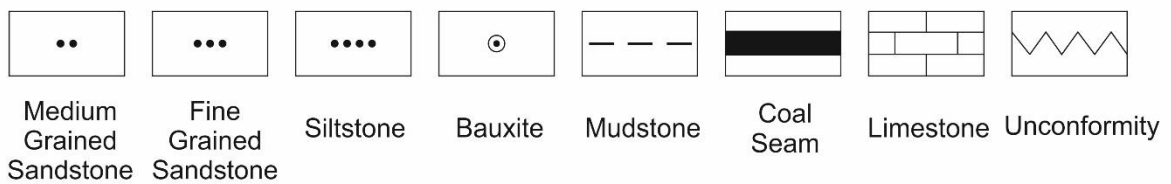
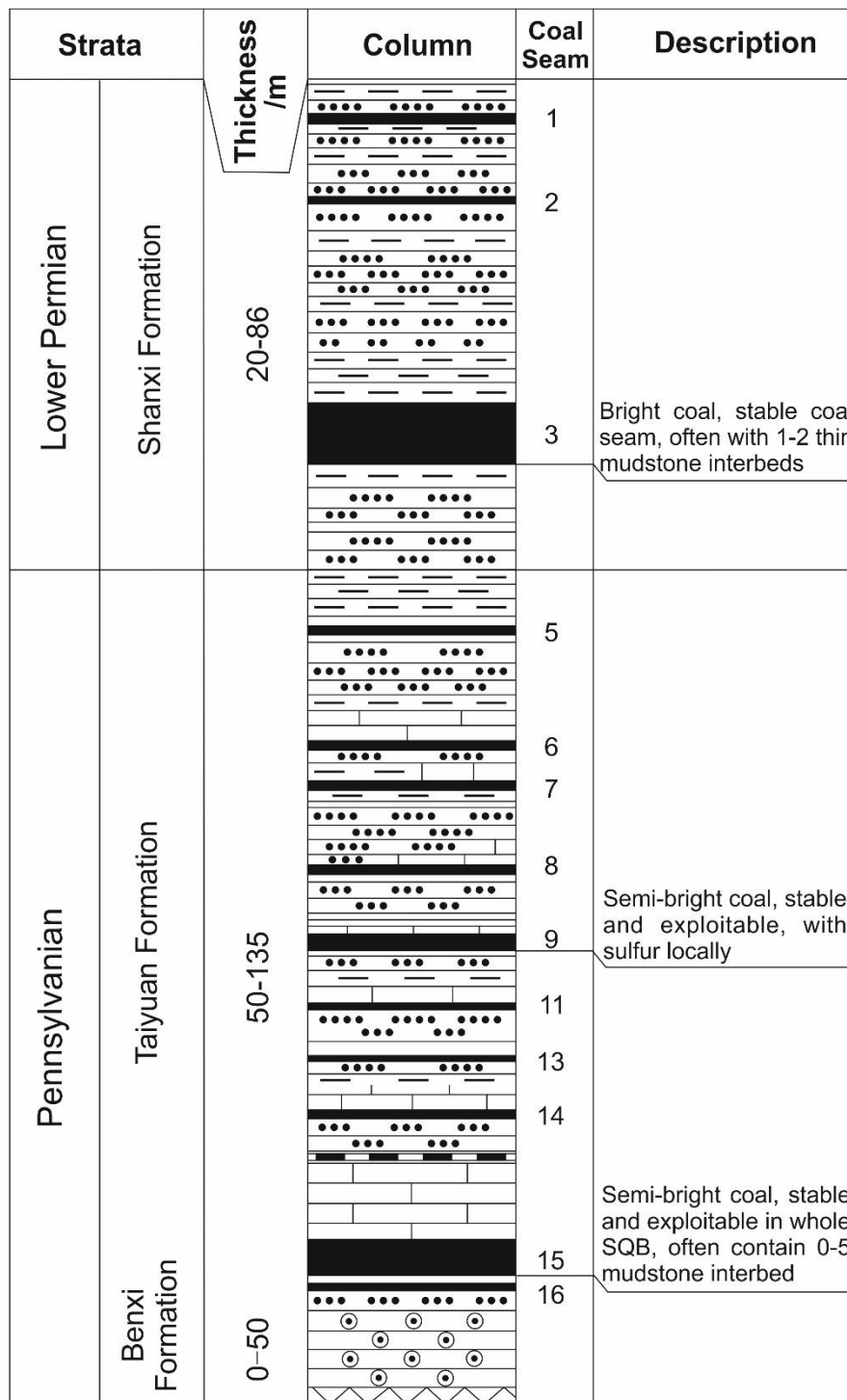
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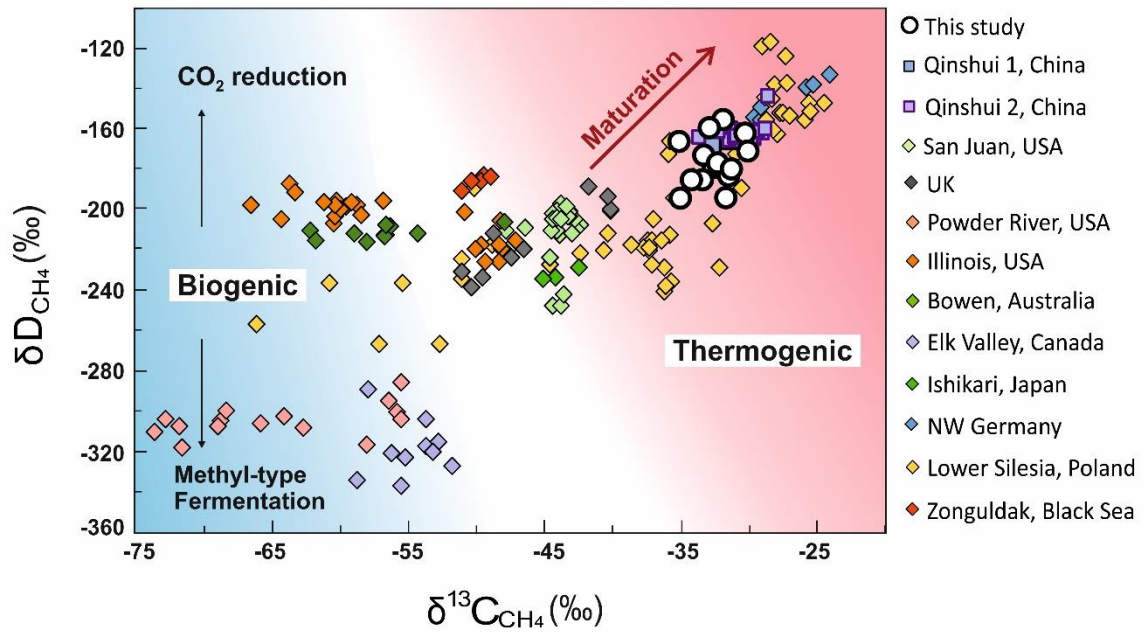
703 **Figure 1**



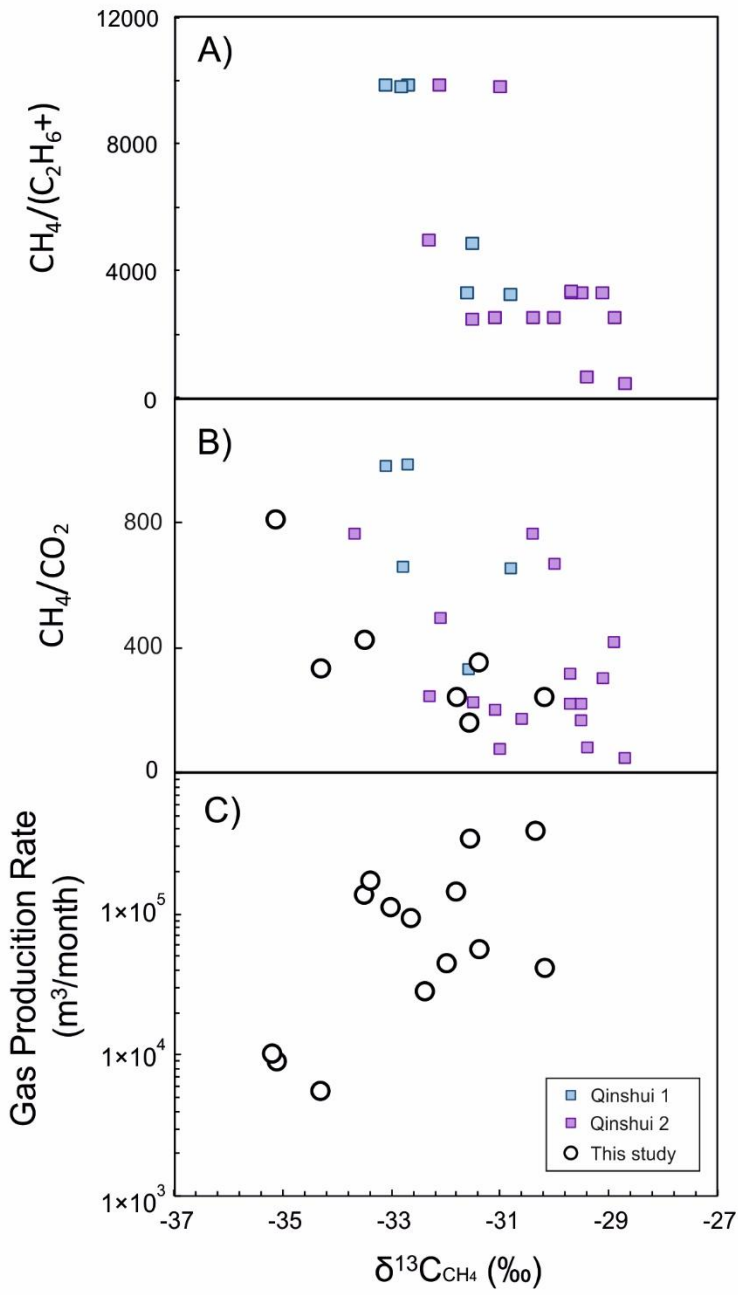
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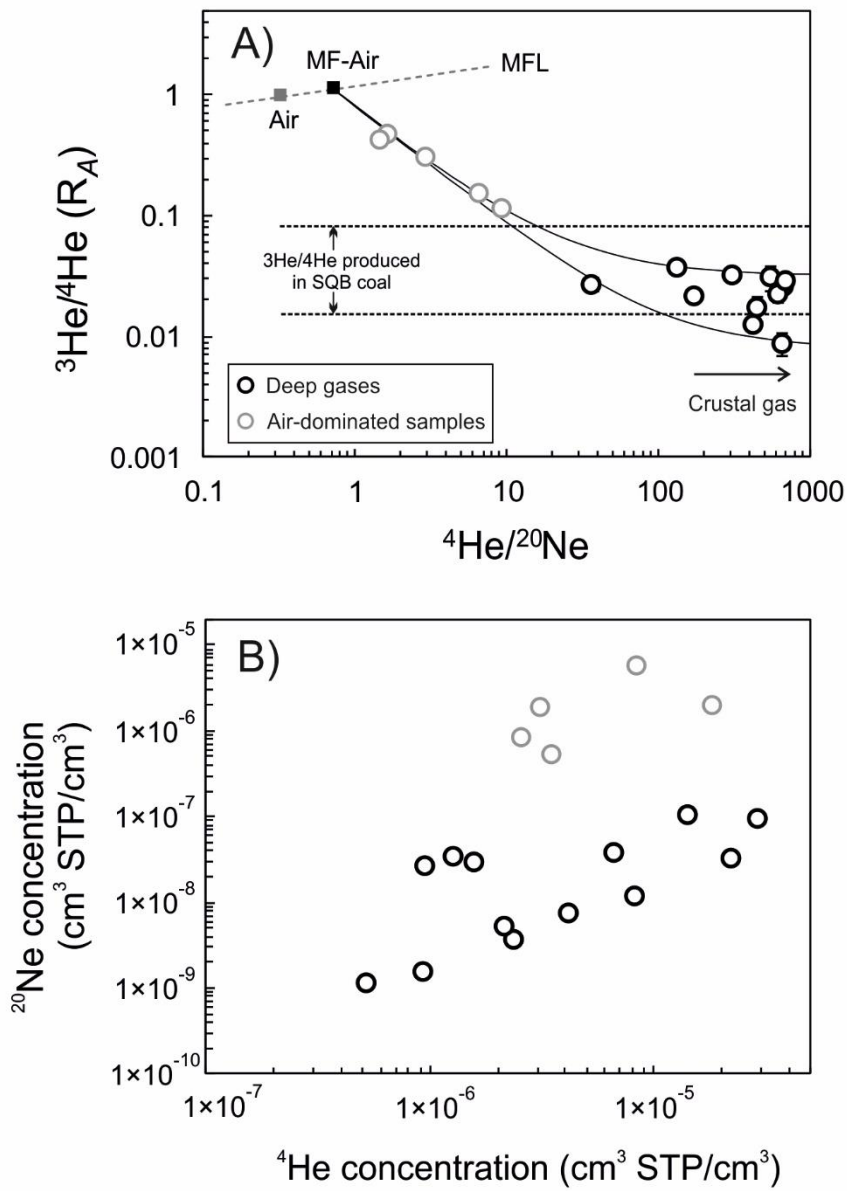


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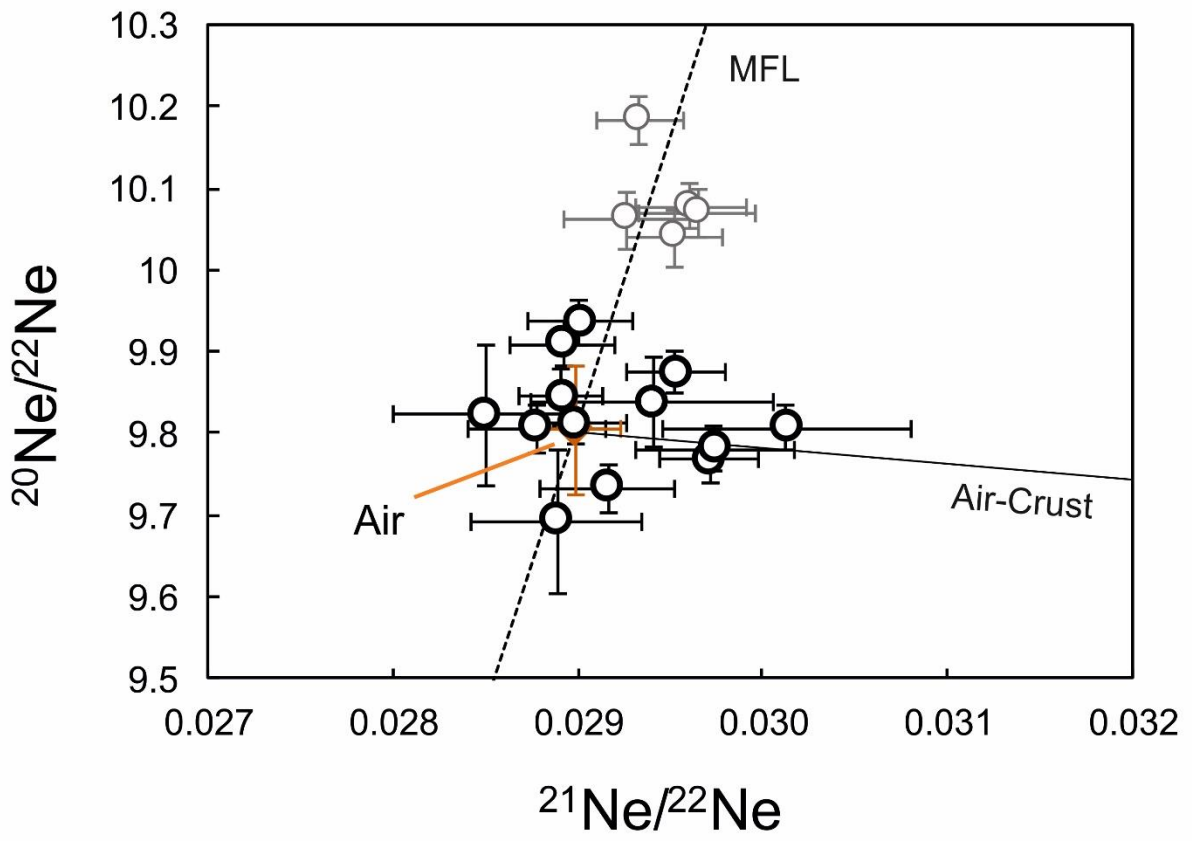


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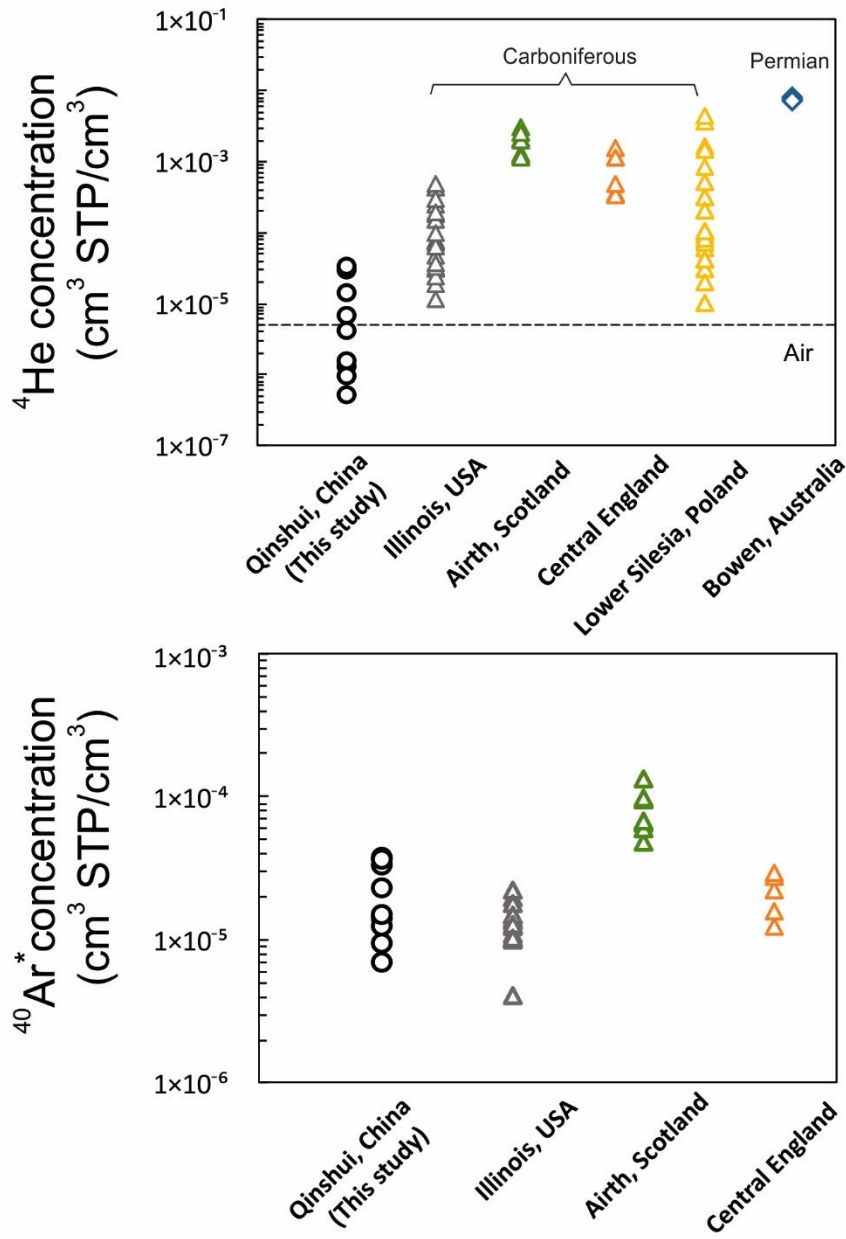




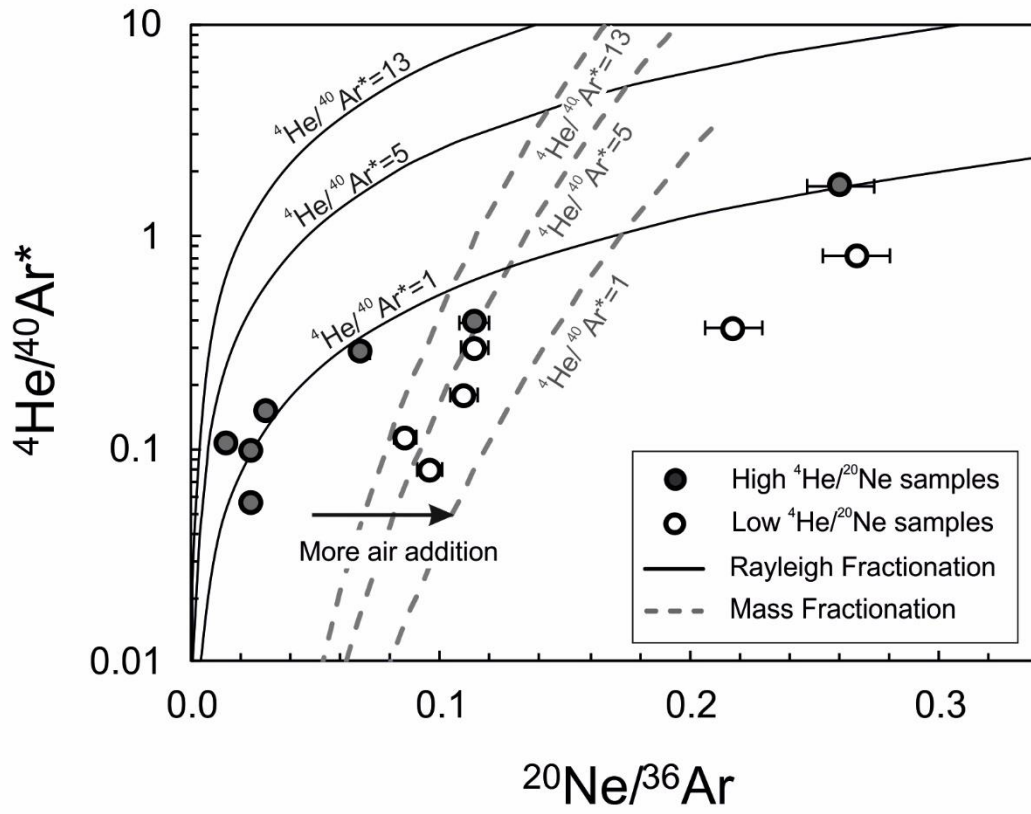
713 **Figure 6**



714



717 **Figure 8**



718

719 **Figure Captions**

720 **Figure 1.** Location and geologic map of the southeast Qinshui Basin (China). A) The location
721 of study area in China; B) Simplified geological map of southeast Qinshui Basin; C) Location
722 of wells sampled in this study; D) Simplified hydrological cross-section of the basin along the
723 line L-L' above. Figures are modified after Wang et al. (2014) and Zhang et al. (2015).

724 **Figure 2.** Stratigraphic column of the coal-bearing formations in southeast Qinshui Basin.
725 Modified after Su et al. (2005) and Cai et al. (2011).

726 **Figure 3.** Genetic characterization of methane from SQB. Data from this study show that
727 methane is thermogenic in origin. Data sources: San Juan (Zhou et al., 2005); UK (Györe et
728 al., 2018); Powder River (Flores et al., 2008); Illinois (Moore et al., 2018; Strapoć et al.,
729 2007); Bowen (Kinnon et al., 2010); Elk Valley (Aravena et al., 2003); Ishikari (Shimizu et
730 al., 2007); NW Germany (Schoell, 1980); Lower Silesia (Kotarba and Rice, 2001); Black Sea
731 (Hoşgörmez et al., 2002); Qinshui 1 (Li et al., 2014); Qinshui 2 (Zhang et al., 2018). 1 σ
732 errors are covered by printed symbols. Modified after Strapoć et al. (2011).

733 **Figure 4.** The relationship between $\delta^{13}\text{C}_{\text{CH}_4}$ with molecular composition of gases and gas
734 production rate. The evident negative trends between $\delta^{13}\text{C}_{\text{CH}_4}$ and $\text{CH}_4/(\text{C}_2\text{H}_6+)$ (A), CH_4/CO_2
735 (B) and positive trend between $\delta^{13}\text{C}$ and gas production rate (C) indicate that molecular and
736 isotopic compositions of produced gases have been affected by the diffusive fractionation
737 process during gas extraction. Data source: Qinshui 1 (Li et al., 2014); Qinshui 2 (Zhang et
738 al., 2018). 1 σ errors are covered by printed symbols.

739 **Figure 5.** The He and Ne systematics of CBM gases from SQB. A) $^3\text{He}/^4\text{He}$ ratio plotted
740 against $^4\text{He}/^{20}\text{Ne}$ of SQB methane. The solid black lines are mixing lines between deep gases,
741 with high $^4\text{He}/^{20}\text{Ne}$, and mass fractionated air. The range of $^3\text{He}/^4\text{He}$ of radiogenic He
742 produced within the coals (see text) is shown by black dashed lines. MFL: Mass fractionation
743 line after square root law (Kaneoka, 1980). The black circles represent deep gases; grey
744 circles represent air-dominated samples. The black and grey squares represent the mass
745 fractionated air and air, respectively. B) ^{20}Ne vs. ^4He concentrations. Air-dominated samples
746 have relatively higher ^{20}Ne concentration but similar ^4He concentration range with deep
747 gases. Uncertainties are 1 σ .

748 **Figure 6.** $^{20}\text{Ne}/^{22}\text{Ne}$ vs. $^{21}\text{Ne}/^{22}\text{Ne}$ for SQB methane-rich gases. Air is after Eberhardt et al.
749 (1965) and Györe et al. (2019). The air-dominated samples (grey circles) are following the
750 mass-controlled fractionation line (MFL) in consistent with He-Ne molecular trend. The deep
751 gases (black circles) are mainly air-derived without evident crustal addition of ^{21}Ne .
752 Uncertainties are 1σ .

753 **Figure 7.** He and Ar contents of Carboniferous and Permian coals worldwide. The
754 concentration of ^4He in the SQB is evidently lower than that in other basins with similar age
755 (A). $^{40}\text{Ar}^*$ concentrations in gases from SQB overlaps with that in other basins (B). Data
756 source: Illinois, USA (Moore et al., 2018); Airth, Scotland (Györe et al., 2018); Central
757 England (Györe et al., 2018); Lower Silesia, Poland (Kotarba and Rice, 2001), Bowen,
758 Australia (Kinnon et al., 2010). 1σ errors are covered by printed symbols.

759 **Figure 8.** $^4\text{He}/^{40}\text{Ar}^*$ vs. $^{20}\text{Ne}/^{36}\text{Ar}$ for Southeast Qinshui Basin CBM. The black lines reflect
760 solubility-controlled Rayleigh fractionation of open system with an initial $^{20}\text{Ne}/^{36}\text{Ar}$ of 0.154.
761 The grey dashed lines are mass-controlled kinetic fractionation (Ballentine et al., 2002; Zhou
762 et al., 2005). Open system Rayleigh fractionation of gas from formation water with initial
763 $^4\text{He}/^{40}\text{Ar}^* = 1$ provides the best fit to the least air-contaminated gas samples (black filled
764 circles, $^4\text{He}/^{20}\text{Ne} > 400$). This implies that radiogenic ^4He and ^{40}Ar in the reservoir were
765 dissolved in the formation water and degassed with air-derived ^{20}Ne and ^{36}Ar during pumping
766 process.

767 **Tables**

768 Table 1. Gas compositions and stable isotopes of coalbed methane from southeast Qinshui Basin.

Sample ID	Depth (m)	Distance to the basin margin (km)	Gas extraction duration (month)	Total gas production ($\times 10^5$)	CH ₄ /CO ₂	$\delta^{13}\text{C}$	δD
Qs1	406	20.2	132	125	na	-32.7	-178
Qs3	567	21.3	86	24.4	na	-32.4	-176
Qs4	585	21.8	25	41	na	na	na
Qs5	606	22.5	96	335	152	-31.6	-183
Qs6	390	20.4	96	377	na	-30.4	-162
Qs7	529	13.8	83	105	na	na	na
Qs9	552	12.0	69	39.1	349	-31.4	-180
Qs10	638	10.8	99	138	423	-33.5	-185
Qs11	673	10.2	125	217	na	-33.4	-173
Qs12	660	9.9	128	57.9	na	-32.0	-155
Qs14	487	13.6	71	6.51	806	-35.1	-194
Qs15	446	17.4	23	10.4	na	na	na
Qs17	519	16.8	15	0.85	329	-34.3	-185
Qs18	394	15.3	24	2.5	na	-35.2	-166
Qs19	420	14.9	3	0.29	na	na	na
Qs20	482	16.5	27	52.7	330	na	na
Qs21	375	19.3	25	36.7	237	-31.8	-194
Qs22	390	18.9	26	29.1	na	-33.0	-159
Qs23	361	20.0	96	40.5	235	-30.2	-171

769 1σ standard deviation for C₁/C₂₊, CH₄/CO₂, $\delta^{13}\text{C}$ and δD are 2%, 2%, 0.3‰ and 3‰, respectively.770 $\delta\text{C}_{\text{CH}_4}$ and $\delta\text{D}_{\text{CH}_4}$ are in permil and relative to PDB (Craig, 1957) and V-SMOW (Gonfiantini, 1978), respectively.771 Volume of produced gas is given in m³ STP, where standard conditions are: p=0.101 MPa and T=0 °C.

772 na: not analysed.

773

774 Table 2. Noble gas compositions of well gases from southeast Qinshui Basin.

Sample ID	⁴ He (×10 ⁻⁶)	²⁰ Ne (×10 ⁻⁸)	⁴⁰ Ar (×10 ⁻⁵)	³ He/ ⁴ He (R _A)	²⁰ Ne/ ²² Ne	²¹ Ne/ ²² Ne	⁴⁰ Ar/ ³⁶ Ar	³⁸ Ar/ ³⁶ Ar	⁴⁰ Ar* (×10 ⁻⁵)
Qs1	4.15(12)	0.75(3)	4.8(2)	0.031(8)	9.81(3)	0.0288(4)	431(2)	0.202(5)	1.46(5)
Qs3	29.23(86)	9.36(34)	13.8(5)	0.032(2)	9.94(3)	0.0290(3)	392(2)	0.188(3)	3.29(13)
Qs4	22.24(65)	3.25(12)	5.0(2)	0.026(2)	9.81(3)	0.0290(3)	401(2)	0.190(5)	1.28(5)
Qs5	8.27(23)	1.20(4)	5.2(2)	0.030(1)	9.73(3)	0.0292(4)	497(1)	0.193(5)	2.09(8)
Qs6	0.52(2)	0.115(4)	2.3(1)	0.017(4)	9.77(3)	0.0297(3)	493(2)	0.189(4)	0.92(3)
Qs7	8.47(25)	561(20)	229(8)	0.428(20)	10.06(4)	0.0293(3)	291(3)	0.187(4)	n/a
Qs9	3.10(9)	187(7)	72(3)	0.482(7)	10.07(3)	0.0296(3)	291(1)	0.187(6)	n/a
Qs10	1.58(5)	2.86(10)	11.3(4)	na	9.82(9)	0.0285(5)	341(1)	0.186(4)	1.39(6)
Qs11	0.96(3)	2.59(9)	9.2(3)	na	9.84(3)	0.0289(2)	343(2)	0.187(8)	1.20(6)
Qs12	1.27(4)	3.46(13)	10.1(4)	0.027(3)	9.84(6)	0.0294(7)	320(2)	0.188(5)	0.68(6)
Qs14	2.55(7)	84(3)	42(2)	0.307(10)	10.08(3)	0.0296(3)	298(1)	0.190(5)	n/a
Qs15	18.30(54)	195(7)	86(3)	0.115(4)	10.04(4)	0.0295(3)	302(2)	0.185(5)	0.91(43)
Qs17	33.25(96)	na	18.8(7)	0.026(1)	na	na	368(1)	0.188(10)	3.58(14)
Qs18	2.16(6)	0.51(2)	12.5(4)	0.013(1)	9.87(3)	0.0295(3)	358(2)	0.196(4)	2.06(9)
Qs19	3.50(10)	52(2)	31(1)	0.153(3)	10.18(3)	0.0293(2)	309(1)	0.186(4)	1.06(10)
Qs20	6.38(18)	3.83(14)	12.3(4)	0.022(1)	9.69(9)	0.0289(5)	365(1)	0.196(4)	2.23(8)
Qs21	0.93(3)	0.150(6)	2.8(1)	0.023(3)	9.81(3)	0.0301(7)	451(2)	0.187(1)	0.95(4)
Qs22	2.38(7)	0.37(1)	5.2(2)	0.009(2)	9.78(3)	0.0297(4)	428(3)	0.184(8)	1.56(6)
Qs23	14.14(42)	10.54(38)	19.1(7)	0.037(2)	9.91(3)	0.0289(3)	370(3)	0.193(10)	3.71(18)
Air	5.24(5)	16.45(4)	934(1)	1.000(9)	9.80(8)	0.0290(2)	298.6(3)	0.1885(3)	

775 1σ standard deviation is shown as the last significant figures in parentheses.

776 Noble gas concentrations are in unit of cm³ STP/cm³ with standard conditions after Ozima and Podosek, (2002) (p = 0.101 MPa, T = 0 °C).

777 Air composition is after Eberhardt et al. (1965); Györe et al. (2019); Lee et al. (2006); Mark et al. (2011); Ozima and Podosek (2002).

778 R_A is the atmospheric ³He/⁴He ratio of 1.34 × 10⁻⁶ (Mishima et al., 2018).

779 ⁴⁰Ar* is calculated air-free ⁴⁰Ar.

780 na: not analysed; n/a: not applicable.