Influence of crystal structure on charge carrier effective masses in BiFeO₃

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Ferroelectric-based photovoltaics have shown great promise as a source of renewable energy, thanks to their in-built charge separation capability, yet their efficiency is often limited by low charge-carrier mobilities. In this paper, we compare the photovoltaic prospects of various phases of the multiferroic material BiFeO₃ by evaluating their charge-carrier effective masses from first-principles simulations. We identify a tetragonal phase with the promising combination of a large spontaneous polarization and relatively light charge carriers. From a systematic investigation of the octahedral distortions present in BiFeO₃, we clarify the relationship between structure and effective masses. This relationship is explained in terms of changes to the orbital character and overlap at the band edges that result from changes in the geometry. Our findings suggest some design principles for how to tune effective masses in BiFeO₃ and similar materials through the manipulation of their crystal structures in experimentally accessible ways.

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I. INTRODUCTION

The field of ferroelectric (FE) photovoltaics, dating back to 1956 [1], experienced something of a renaissance in 2009 with the discovery of the switchable diode and bulk photovoltaic effects in the multiferroic material, bismuth ferrite (BiFeO₃, BFO) [2]. The bulk photovoltaic effect, observed in BFO, arises from the absence of inversion symmetry in the crystal structure of the room-temperature R3c phase [3]. This asymmetry results in a giant spontaneous polarization (~100 μ C/cm²) [4,5] that aids in charge separation in a photovoltaic device.

Despite the intrinsic charge separation ability, FE materials such as BFO generally suffer from low charge-carrier mobilities, leading to high recombination losses and limited device efficiency [6–8]. Increasing the mobilities in such FE materials would therefore lead to enhanced photovoltaic device efficiency.

Although mobility is a macroscopic quantity, it is the mobility associated with the electronic band edges that is of particular relevance to recombination rates. Since mobility is directly related to the dispersion of these bands, a material's charge-carrier mobilities can be altered by modifying the curvature of the bands near the Fermi level. Manipulation of the crystal structure, for example, by strain engineering, is one such route to tuning the curvature of electronic bands. Strain engineering, together with chemical doping, have been widely exploited in the semiconductor industry to control mobility in silicon-based devices [9]. In BFO, a wide range of crystal structures can be stabilized through strain [10,11] and interface [12,13] engineering. Of these structures, those that lack a center of inversion symmetry are of significant interest for photovoltaic applications. There has been a great deal of interest in the tetragonal (space group *P4mm*) and tetragonal-like phases of BFO thin films, in particular, due to their giant spontaneous polarization of $\approx 150 \ \mu C/cm^2$ [10,11].

How do the mobilities of the charge carriers compare across the BFO phases? In this paper, we present an investigation within density functional theory (DFT) of the electronic properties of several FE and non-FE phases of BFO. We begin by considering the experimentally accessible groundstate R3c phase and the higher-temperature orthorhombic *Pnma* and cubic $Pm\bar{3}m$ phases. We also consider the prototypical tetragonal *P4mm* phase, accessible via epitaxial strain [10,11]. We further consider the theoretical $R\bar{3}c$ phase, which is similar to the R3c phase but without the polar distortion that gives the latter its spontaneous polarization. We compare the charge-carrier effective masses (m^*), which are inversely proportional to the mobilities, across these phases. We will

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show that the carriers in the room-temperature FE (R3c) phase have considerably larger m^* compared to some non-FE phases. Nonetheless we find that the tetragonal phase of BFO, which has a large spontaneous polarization, has relatively low electron and hole effective masses.

In order to explain the differences in m^* across the BFO phases, we systematically study the geometric transformations that map the BFO phase with the lowest m^* , to the phase with the highest m^* . The effects of these transformations on the m^* are explained in terms of changes to the orbital character and overlap at the band edges. Previous works [11] have indicated that, with a judicious choice of substrate material, a BFO lattice with the desired spontaneous polarization and m^* can be fabricated. Our results therefore provide insight into the rational design of materials with optimum properties, particularly for applications in light harvesting.

II. COMPUTATIONAL DETAILS

We consider the R3c, Pnma, P4mm, Pm $\bar{3}m$, and R $\bar{3}c$ phases of BFO. Simulations of these phases were performed using DFT as implemented in the Vienna ab initio Simulation Package (VASP version 5.4.1) [14–17]. We employ the version of the generalized gradient approximation parametrized by Perdew, Burke, and Ernzerhof (PBE) [18] as the exchangecorrelation functional, with an effective Hubbard correction, $U_{\rm eff}$, applied to the Fe d orbitals using the method of Dudarev et al. [19]. This PBE + U method has been employed in this paper to account for the known failures of standard local density and generalized gradient approximations to accurately describe the strong correlations in transition-metal oxides [19]. We use a $U_{\rm eff}$ of 4 eV for all calculations as this value has been found to best capture the electronic structure of BFO, particularly the orbital characters at the valence-band maximum (VB_{max}) and conduction-band minimum (CB_{min}) [20].

All of the calculations were carried out using the projectoraugmented plane-wave method [21,22] and a plane-wave cutoff energy of 520 eV, treating explicitly 15 electrons for Bi $(5d^{10}6s^26p^3)$, 14 for Fe $(3p^63d^64s^2)$, and 6 for O $(2s^22p^4)$ [23]. The 40-atom pseudocubic (pc) unit-cell setting was consistently adopted for all but the effective mass calculations in Sec. III B. The pc unit-cell setting allows us to capture the C- or G-type antiferromagnetism (AFM) across all of the BFO phases and allows for a more straightforward comparison of the structures than the various primitive cells. For reasons of efficiency, the effective mass calculations in Sec. III B were performed using a ten-atom rhombohedral cell setting. Brillouin-zone integrations for the relaxations and static calculations were performed on Γ-centred Monkhorst-Pack (MP) [24] grids: $9 \times 9 \times 9$ for the pc unit cells and $11 \times$ 11×11 for the rhombohedral unit cells. Density-of-states (DOS) calculations, requiring finer sampling of the Brillouin zone, were performed using a Γ -centred 11 × 11 × 11 MP grid in the pc unit-cell setting. We relaxed the low-energy phases of BFO such that all force components were less than 5 meV/Å. The unit-cell shape and sizes of these phases were optimized such that all stress components were smaller than 0.06 GPa. All of the structure files and raw data are freely available in the Supplemental Material (SM) [25].

TABLE I. Computed effective masses in units of electron rest mass m_0 , spontaneous polarization values, and relative energies of five PBE + U relaxed phases of BFO. Note that energies, E, are taken relative to the most stable phase: R3c.

Space group	$ m_{h}^{*} $ (m_{0})	m_e^* (m_0)	Polarization $(\mu C/cm^2)$	$\frac{E - E_{R3c}}{(\text{meV per f.u.})}$
Pm3m	0.34	0.24	0.0	971
R3c	0.63	0.37	0.0	268
P4mm	0.54	0.33	185.3	84
Pnma	0.95	0.99	0.0	57
R3c	0.67	3.06	90.0	0

We use the modern (Berry's phase) theory of polarization [26,27] to calculate the spontaneous polarization of each structure. As noted by Neaton *et al.* [28], much care is needed when performing these calculations, especially in systems such as BFO for which the spontaneous polarization, P_s , is of the same order of magnitude as the quantum of polarization, Q. We follow the procedure outlined in Ref. [28].

We calculate the hole and electron effective masses for each of the considered BFO phases as follows. The curvature [29] effective mass of a given band, n, at a particular location in reciprocal space, k, is a 3×3 tensor quantity the magnitude of which in a given direction is inversely proportional to the band curvature in that direction. m^* can therefore be defined as

$$\left(\frac{1}{m^*}\right)_{ij} = \frac{1}{\hbar^2} \frac{\partial^2 E_n(\boldsymbol{k})}{\partial k_i k_j}, \quad i, j = x, y, z,$$
(1)

where $E_n(\mathbf{k})$ is the energy dispersion relation for the *n*th band, and *i* and *j* represent reciprocal space components. To obtain m^* , we first compute the full band structure along a path of high symmetry. We then identify the location of the VB_{max} and the CB_{min} in reciprocal space. The band curvatures at these points correspond to the hole effective mass, m_h^* , and the electron effective mass, m_e^* , respectively. Having identified these k points of interest, we employ the method and code outlined in Ref. [31] to obtain the full m^* tensors. In brief, the method involves generating a fine mesh around the kpoint of interest, calculating the energy eigenvalues, and using a finite difference method to build up the tensor of second derivatives. The dependence of m^* on the spacing of this mesh was investigated, and spacings of less than 0.05 bohr⁻¹ were found to give consistent results. We calculate the eigenvalues of the effective mass tensor, which correspond to the effective masses along the principle directions. For the purposes of this paper, in which we consider the photovoltaic prospects of different phases, we envisage a device oriented in such a way as to take advantage of any anisotropy in the m^* tensor. With that in mind, we compare the smallest effective masses of each structure, with the full m^* tensors presented in the SM [25].

To make the calculations tractable, we limit ourselves to a collinear treatment of spin, thus neglecting the long-wavelength (~ 620 Å) spiral spin structure found experimentally [32]. All of the low-energy phases in the present paper, with the exception of the *P4mm* phase, are found to adopt a G-type AFM ordering, consistent with previous work [33]. The C-type AFM ordering was found to be slightly lower (6 meV/f.u.) in energy than the G-type ordering for the

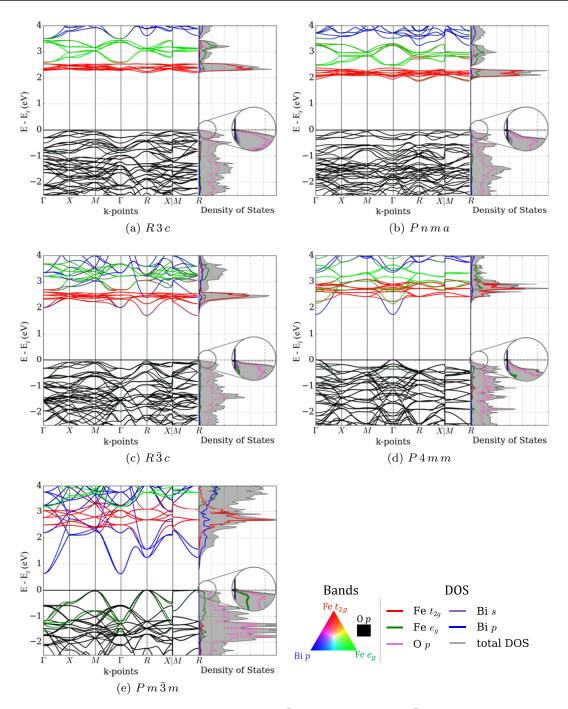


FIG. 1. Projected bands and DOS for the (a) R3c, (b) Pnma, (c) R3c, (d) P4mm, and (e) Pm3m phases of BFO, in their (pseudo-) cubic settings. The bands are coloured, at each k point, based on wave-function projections onto chosen orbitals. As indicated in the legend, red, green, blue, and black represent projections onto Fe t_{2g} , Fe e_g , Bi p, and O p states, respectively.

P4mm phase, again in agreement with previous work [33,34]. A change from G-type to C-type ordering in the *P4mm* phase had little effect on the character or curvature of the band edges.

III. RESULTS

A. Effective masses of the phases of BFO

Table I shows the calculated m^* , polarization, and relative stability of the considered BFO phases. The results for the

relative stability agree well with the literature, differing by at most 5% with respect to the PBE + U ($U_{eff} = 4 \text{ eV}$) work of Diéguez *et al.* [33]. For the FE *R*3*c* phase, the calculated P_s agrees well with both experiment and theory, which report values of between 90 and 100 μ C/cm² [4,28,35,36]. For the tetragonal *P*4*mm* phase, we find a P_s of 185 μ C/cm², larger than those reported in previous theoretical (151–152 μ C/cm² [33,37]) and experimental (\approx 130 μ C/cm² [38]) works. Our larger value of P_s may be due to the known tendency of the PBE functional to overestimate *c*/*a* ratios in complex oxides. Previous theoretical works employed the PBEsol [33] and

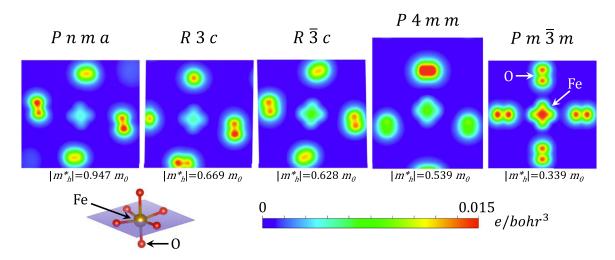


FIG. 2. Kohn-Sham orbitals corresponding to top VB, summed over all k points, for the five phases of BFO investigated. These are representative cross sections through the FeO₄ plane in the case of the perfect cubic $Pm\bar{3}m$ structure, and at least two O atoms and one Fe atom in the other phases, as illustrated in the bottom left diagram. The locations of the Fe and O atoms are labeled in the $Pm\bar{3}m$ section, and the hole effective masses m_h^* are shown below each panel for ease of reference. The full 3D data files are available in the SM [25].

LDA [37] functionals with Hubbard corrections and found c/a ratios of 1.28 and 1.27, respectively. Experimentally, a c/a ratio of about 1.24 was estimated [38]. The c/a ratio found here was 1.30, consistent with the larger predicted P_s . Indeed, when we constrained the *P4mm* lattice constants to those estimated in the experimental study [38], we obtained a reduced value of $P_s \approx 150.0 \ \mu\text{C/cm}^2$.

Comparing the charge-carrier effective masses, reported in Table I, we find the room-temperature FE phase of BFO (*R*3*c*) to have the largest m_e^* and the second largest m_h^* . The tetragonal *P*4*mm* phase, which is also ferroelectric, has an m_e^* an order of magnitude smaller and an m_h^* 20% smaller than those of *R*3*c*. The paraelectric $R\bar{3}c$ phase has an m_e^* comparable to that of *P*4*mm* and an m_h^* comparable to that of *R*3*c*. Both m^* are smallest in the paraelectric $Pm\bar{3}m$ phase. From these results we therefore observe that the ferroelectric phases do not necessarily have lower m^* than the paraelectric phases.

To gain insight into the variation of m^* with the phases, we examine and compare the electronic structures of the BFO phases, particularly at the band edges-the locations at which the m^* are evaluated. In Fig. 1 the band structure and DOS, projected onto spherical harmonics, are shown for each phase of BFO studied here. Considering the lowest-lying conduction bands in order of decreasing m_e^* , i.e., from R3c to Pnma to $R\bar{3}c$ to P4mm to $Pm\bar{3}m$, we see that the major contribution to the states at CB_{min} gradually changes from Fe t_{2g} to Bi p. In Fig. S1 of the SM [25], we plot the corresponding Kohn-Sham (KS) orbitals. These orbital plots corroborate the spherical harmonic projections, indicating that Fe t_{2g} states are the primary contributors to the CB_{min} for the R3c, Pnma, and $R\bar{3}c$ phases, while the Bi p states make up the CB_{min} for the P4mm and $Pm\bar{3}m$ phases. Our results suggest that, at the CB_{min} , the presence of Bi p states leads to a lower m_{e}^{*} than that of Fe t_{2g} states.

Compared to the conduction bands, the topmost valence bands in Fig. 1 show a far less dramatic difference in character across the five phases. All of the phases considered here have an O p dominated VB_{max}. However, there are minor contributions from Fe e_g states to the VB_{max}, most notably in the $Pm\bar{3}m$ phase. To see more clearly the variation in the states at the VB_{max}, we show in Fig. 2 the cross sections through KS orbitals corresponding to the topmost valence band. The figures show that, as m_h^* decreases from *Pnma* to *R*3c to *R* $\bar{3}c$ to *P*4mm to *Pm* $\bar{3}m$, the contribution from the Fe e_g states increases. This observation indicates that the presence of Fe e_g states at the VB_{max} plays a role in decreasing m_h^* .

B. Effects of structural transformations on effective masses

The different crystal structures of the BFO phases considered have been found to have a wide variation in the electronic structures and, in particular, in their charge-carrier effective masses. As such, a substantial yet complex relationship exists between crystal structure and effective mass in BFO. In order to better understand this relationship, we now consider the geometric transformations that map the phase with the lowest effective masses, R3c. These transformations are (a) an antiphase rotation of the FeO₆ octahedra about the [111]_{pc} direction (i.e., $a^-a^-a^-$ in Glazer's notation [39]) and (b) a translation of the FeO₆ octahedra along the [111]_{pc} direction. These transformations are illustrated in Fig. 3. More generally, these and similar geometric transformations can describe the other BFO phases, portrayed in Fig. 4, as follows.

(1) *Pnma*: an octahedral rotation out of phase along the $[100]_{pc}$ and $[010]_{pc}$ directions but in phase along the $[001]_{pc}$ direction $(a^-a^-c^+)$, and alternating positive and negative translations in the $[010]_{pc}$ direction.

(2) $R\bar{3}c$: an out-of-phase rotation about the $[111]_{pc}$ axis $(a^-a^-a^-)$ with no translation.

(3) P 4 m m: no rotation but with a substantial translation in the $[001]_{pc}$ direction.

These transformations are accompanied by some strain to accommodate the change in the structure. We therefore consider the effects of strain, rotation, and translation

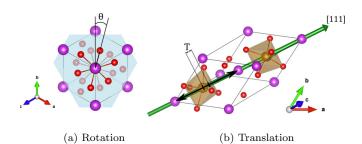


FIG. 3. Schematic of the FeO₆ octahedral (a) rotations about and (b) translations along the $[111]_{pc}$ direction present in the relaxed *R3c* structure. In (a) we separate neighboring octahedra along the $[111]_{pc}$ axis by a semitransparent plane in order to highlight the out-of-phase nature of these rotations. The green arrow in (b) shows the $[111]_{pc}$ direction, and *T* is the displacement of the Fe atom from the midpoint between successive Bi atoms along this direction. Purple, ochre, and red spheres represent Bi, Fe, and O, respectively.

separately, particularly in relation to the ground-state *R3c* structure, which has shown remarkable structural flexibility in the context of heteroepitaxially grown BFO films [10].

1. Strain

Various experimental works have found the R3c phase of BFO to be stable over a large range of epitaxial strain: from -2.6 to +1.2% [40–42]. In addition, spontaneous polarization in BFO has been found to be relatively insensitive to epitaxial strain values of up to $\pm 3\%$ [35,37]. Epitaxial strain can also be used to stabilize other phases of BFO, such as the tetragonal *P4mm* phase, as summarized by Sando *et al.* [10,11]. Since we are interested in the transformation from the *R3c* to the *Pm3m* phase, which involves an isotropic change in lattice constants, we consider the case of *uniform*, rather than epitaxial, strain. We apply uniform strains between -5 to +5% to a BFO unit

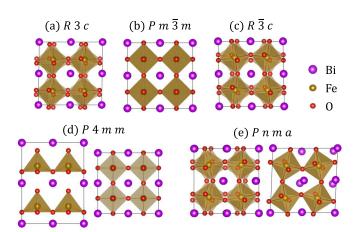


FIG. 4. Crystal structures of five phases of BFO. For ease of comparison, all structures are presented in their cubic or pseudocubic settings. For the (a) R3c, (b) $R\bar{3}c$, and (c) $Pm\bar{3}m$ structures, the three pseudocubic axes are equivalent. The (d) P4mm and (e) Pnma structures, however, have one nonequivalent axis. For these structures the left (right) figure has the nonequivalent axis parallel (perpendicular) to the page. Note that in the (d) P4mm case the elongated cell causes the FeO₆ octahedra to break into square pyramidal FeO₅ units.

cell constrained to the R3c symmetry. Figure 5(a) shows the effect of such strain on the charge-carrier effective masses.

The m_e^* changes significantly, decreasing from 3.02 m_0 in the unstrained cell to 2.33 m_0 under 5% compressive strain. Under a tensile strain of between 1 and 3% we see an even larger decrease in m_e^* , from 3.19 m_0 to 0.60 m_0 , corresponding to a change in the location and character of the CB_{min}. Figures 5(b)–5(d) show the projected bands and DOS for the -5, 0, and +5% strain cases. Similar figures for the other strain values are available in Fig. S2 of the SM [25]. From these figures a shift from an Fe t_{2g} to an Fe e_g dominated CB_{min} can be observed from 1 to 3% strain, which explains the sudden drop in m_e^* for large tensile strain, since the e_g bands are more dispersive than the t_{2g} bands. We can understand the shift from a t_{2g} to an e_g dominated CB_{min} by considering the reduction in the splitting of the Fe d orbitals, due to the octahedral environment, as the Fe-O bond lengths increase.

The m_h^* , however, shows little dependence on strain. Over the range considered, $|m_h^*|$ is largest $(1.051 m_0)$ at -5% strain, and smallest $(0.753 m_0)$ at +5% strain. For all strain values considered, the VB_{max} remains strongly dominated by O p states as evident from Figs. 5(b)–5(d). Thus, the negligible changes in m_h^* as the cell is strained are reflected by the minor changes in the character of the VB_{max}.

2. Rotation of the FeO₆ octahedron

To examine the effect of octahedral rotation on the electronic properties of BFO, we rotate the FeO₆ octahedron from the perfect cubic perovskite geometry about the [111]_{pc} axis in the out-of-phase manner shown in Fig. 3(a). In Fig. 6 we plot the dependence of m^* on the rotation angle, θ . Both m_e^* and m_h^* increase with increasing θ . Around $\theta = 15^\circ$ we see a marked increase in the gradient of this relationship.

To understand the trends in m^* with θ , we examine the electronic structures at VB_{max} and CB_{min} as a function of θ . In Fig. 7 we plot the projected bands and DOS of BFO with the FeO₆ octahedron rotated at $\theta = 0$, 14, and 28°. Similar figures for the full range of θ considered can be found in Fig. S6 of the SM [25]. As θ increases up to $\theta = 14^\circ$, the Bi *p* bands move up in energy but we see little change in the character of the CB_{min}, and correspondingly little change in m_e^* . Around 14° we see a shift in the location of the CB_{min} from Γ to *R*. For $\theta > 14^\circ$, the character of the CB_{min} transitions from Bi *p* to Fe t_{2g} , and then to a possible [43] mix of Fe e_g and t_{2g} states for large θ . The Bi *p* to Fe t_{2g} transition is associated with the large increase in m_e^* with θ shown in Fig. 6.

Within the topmost valence bands, we find that O *p* states dominate for all values of θ considered. The states with minor contributions to the VB_{max} can be seen more clearly in the projected DOS. As θ increases, there is a decrease in the Fe e_g contribution, and an increase in the Bi *s* contribution to the top of the VB. We therefore associate the increase in m_h^* with a decrease in Fe e_g contribution and an increase in Bi *s* contribution to the top of the valence band. Real-space plots of the KS orbitals as a function of θ can be found in Fig. S7 of the SM [25]. We also observe that for $\theta = 0^\circ$ there is a band degeneracy at VB_{max}, and hence we present m_h^* values for both the heavy and light holes in Fig. 6. This degeneracy lifts as soon as we have any rotation.

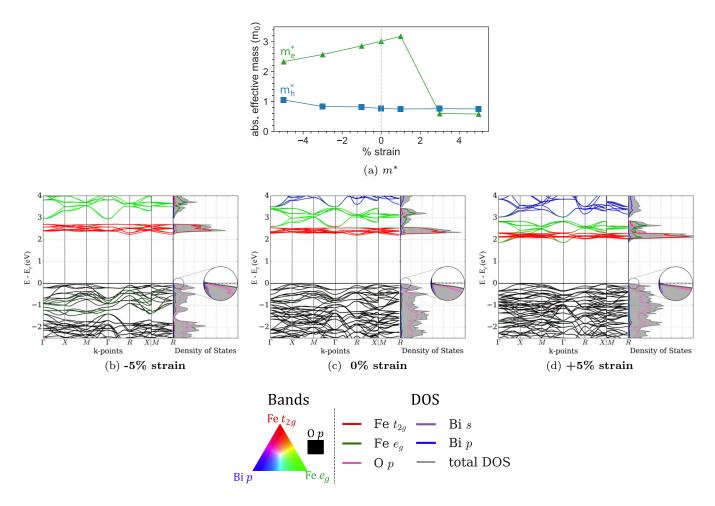


FIG. 5. Effects of uniform strain on the (a) m_e^* (green triangles) and m_h^* (blue squares) and (b–d) projected bands and density of states. Note that while the DOS axis has arbitrary units here the scale is the same in each subfigure. The bands are colored, at each k point, based on wave-function projections onto chosen orbitals. As indicated in the legend in the panel, red, green, blue, and black represent projections onto Fe t_{2g} , Fe e_g , Bi p, and O p states, respectively.

3. Translation of the FeO₆ octahedron

We consider, independently, translations of distorted (rotated $\theta = 14^{\circ}$ about the $[111]_{pc}$ axis) and perfect ($\theta = 0^{\circ}$)

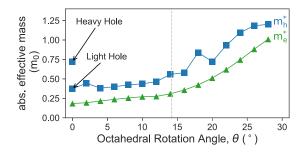


FIG. 6. Absolute charge-carrier effective mass vs octahedral rotation angle. The FeO₆ octahedra are rotated out of phase about the [111]_{pc} axis. For each angle, θ , the m_h^* (blue squares) and m_e^* (green triangles) were calculated. Notice that in the perfect cubic structure ($\theta = 0^\circ$) there is a band degeneracy at the top of the VB—hence the heavy and light holes. This degeneracy breaks as soon as we have any octahedral rotation. The rotation angle of the R3c PBE+ U relaxed structure ($\theta \approx 14.2^\circ$) is indicated with a vertical dashed line.

FeO₆ octahedra along the $[111]_{pc}$ direction. The former translation is as illustrated in Fig. 3(b). We find a marked difference between the two cases, as shown in Fig. 8 where we plot the dependence of m_e^* and m_h^* on the magnitude of the translation, *T*. Translating the perfect octahedron [Fig. 8(a)] has negligible effects on both m_e^* and m_h^* . Translating the distorted octahedron [Fig. 8(b)] also has little effect on m_h^* ; however, there is a large effect on m_e^* .

Figure 9 shows changes in the projected bands and DOS as the octahedra are translated. Once again we observe that changes in m^* are correlated with changes in the chemical character of the band extrema. Figures 9(a)-9(c) confirm that the character at the CB_{min} of the structure with the perfect octahedron does not change within the range of T considered. Figures 9(d)-9(f), however, show a shift from Bi p to Fe t_{2g} character in the case of the 14° rotated octahedron, corresponding to the substantial increase in m_e^* . Conversely, there is no change in character at the VB_{max}, and also little change in m_h^* with T, for both the perfect and the rotated octahedra. The VB_{max} degeneracy at $\theta = 0^\circ$, that was lifted with octahedral rotations in Sec. III B 2, does not vanish in the case of octahedral translations, as indicated in Figs. 9(a)-9(c).

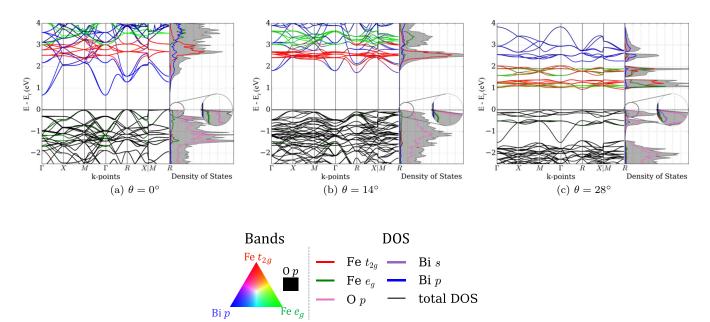


FIG. 7. Variation in projected bands and density of states as a function of out-of-phase FeO₆ octahedral rotation about the $[111]_{pc}$ axis. Note that while the DOS axis has arbitrary units here the scale is the same in each subfigure. The bands are colored, at each k point, based on wave-function projections onto chosen orbitals. As indicated in the legend in the panel, red, green, blue, and black represent projections onto Fe t_{2g} , Fe e_g , Bi p, and O p states, respectively.

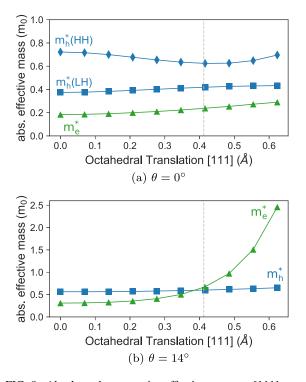


FIG. 8. Absolute charge-carrier effective mass vs $[111]_{pc}$ octahedral translation. In (a) we translate the perfect FeO₆ octahedron $(\theta = 0^{\circ})$, while in (b) we translate a distorted octahedron $(\theta = 14^{\circ})$. The angle, θ , is as defined in Sec. III B 2. The m_e^* and m_h^* are represented by green triangles and blue squares, respectively. In (a), the band degeneracy at VB_{max} gives rise to a heavy hole, HH (blue diamonds), in addition to the light hole, LH (blue squares). The vertical lines in each subfigure correspond to the PBE+ U relaxed R3c Fe translation along $[111]_{pc} (\approx 0.412 \text{ Å})$.

IV. DISCUSSION

In our investigations of the electronic properties in BFO, a clear trend between the orbital characters at the band edges and the resulting m^* emerges. We find that presence of Fe e_g states at the VB_{max} and that of Bi p states at the CB_{min} lead to relatively low m^* . At the VB_{max}, hybridization between the Fe e_g and O 2p states leads to a more dispersive band and hence a lower m_h^* . In the conduction bands, the Bi p states are more dispersive than the Fe t_{2g} , and hence structures that have a CB_{min} that is dominated by Bi p states have a lower m_e^* than those the CB_{min} of which is dominated by Fe t_{2g} states.

We have demonstrated that the changes in the band-edge characters and hence m^* can be affected through geometric changes to the crystal structure. Extreme tensile strain, suppression of the octahedral rotation, and suppression of the octahedral translation yield a much reduced m^* in BFO. This trend explains the variation in m^* across the BFO phases. The high m^* values of the R3c and Pnma phases can be attributed to the presence of octahedral rotation and translation in these phases. When we compare the R3c and $R\bar{3}c$ structures, which contain similarly rotated octahedra, we find that the phase with the translated octahedron (R3c) has larger m^* values compared with the untranslated phase $(R\bar{3}c)$. Interestingly, we find that when the rotation is completely suppressed ($\theta = 0$) translation of the octahedron (within the range of study) has little effect on m^* . Hence the P4mm structure, which has translation without rotation, also has lower m^* values than those of R3c. The $Pm\bar{3}m$ structure has the lowest m^* values of all the structures considered since it has neither octahedral rotation nor translation.

For applications in light harvesting, materials with low m^* values are required to allow the charge carriers to be transported to the electrode before the carriers recombine.

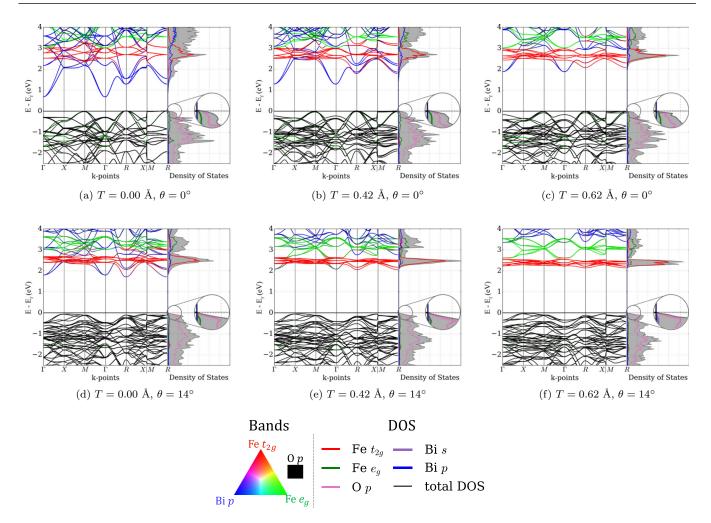


FIG. 9. Variation in projected bands and density of states with FeO₆ octahedral translation along the $[111]_{pc}$ direction. $\theta = 0$ and 14° indicates the translation of the perfect (a–c) and distorted (d–f) octahedra, respectively, and *T* gives the translation magnitude. The bands are colored based on projections onto chosen orbitals. As indicated in the legend, red, green, blue, and black represent projections onto Fe t_{2g} , Fe e_g , Bi p, and O p states, respectively.

Appreciable photovoltaic responses have already been recorded in BFO in the ground-state R3c phase [2,45–48]. Our paper indicates that its crystal structure may be modified to optimize m^* . Particularly in thin-film heteroepitaxial growth, some degree of control over the octahedral tilt angle and tilt pattern has been demonstrated experimentally in BFO and other oxide systems [13,49–52]. While the high-temperature $Pm\bar{3}m$ phase of BFO has the required geometric features to result in low m^* , it lacks the asymmetric potential required for the bulk photovoltaic effect. We find that the tetragonal P4mm phase, which is experimentally accessible under compressive epitaxial strain, has both a large spontaneous polarization and relatively low m^* . However, the P4mm phase has been reported to have a larger optical gap than the R3c phase, despite the smaller electronic gap, which would adversely affect the efficiency of the device [53]. Further work is required to find a balance between these competing effects. Besides geometry manipulation, the band-edge character can also be modified through chemical doping [54,55], which may be an interesting subject for future work.

Another avenue for future work would be to consider the effects of the crystal structure on charge-carrier relaxation

times, since the electron and hole *mobilities* of a given phase are, of course, not only determined by the effective mass of the charge carriers, but also by their relaxation times. These relaxation times depend on electron-phonon scattering [56] and are thus expected to also be sensitive to changes in crystal structure.

As discussed above, the "curvature" definition of m^* adopted here is one that emphasizes the importance of the band edges in conduction. However, in the case of materials that have fairly flat bands with multiple valleys of similar energy, the band curvature at these local minima (maxima) may also influence m^* . In Fig. 1 we see that, particularly in the R3c and Pnma phases, the valence bands of BFO do indeed exhibit several local maxima of similar energy. In order to explore the effect of such multiple valleys, we calculate the "conductivity mass" as defined in Eq. (12) of Ref. [57] and implemented in BOLTZTRAP2 [58] for the five phases of BFO considered above. The electron conductivity masses are found to be in good quantitative agreement with "curvature masses" listed in Table I. The hole conductivity masses, however, are generally larger than the hole curvature masses listed in Table I, consistent with the presence of multiple local maxima.

More importantly for the purposes of the present paper, the trends in both m_e^* and m_h^* generally agree between the two effective mass definitions. An exception to this is the hole effective mass of the *Pnma* phase. While *Pnma* has the largest hole curvature mass, its conductivity mass is slightly smaller than those of the $R\bar{3}c$ and R3c phases. Despite this, given the otherwise good agreement in the m^* trends across these BFO phases, our tests indicate that the central conclusions drawn in this paper are not dependent on the choice of effective mass definition. The conductivity mass tensors for the five BFO phases considered are also available in the SM [25].

In addition to the intrinsic factors considered above, a number of extrinsic factors may also play a role in determining the m^* in a given device. Factors such as temperature, chargecarrier doping concentrations, impurities, grain boundaries, and FE domain walls could all play important roles in determining m^* . However, since our aim is to compare m^* between the BFO phases, and to suggest approaches to improve m^* further, we only consider *differences* in m^* . That is to say, while factors such as temperature are likely to affect m^* , we have made the assumption in this paper that our 0-K results are a necessary precursor to including high-temperature effects.

The effect of spin-orbit coupling (SOC) has been found to be significant in describing certain properties of BFO, such as its weak (Dzyaloshinskii-Moriya) ferromagnetism [59]. We find a noticeable shift in the energy of the electronic bands and a small, Rashba-like shift in the location of the band edges. For example, the inclusion of SOC for the *R3c* structure shifts the location of the CB_{min} slightly off the high-symmetry *R* point. Such a shift might improve the photovoltaic prospects of BFO by inhibiting charge-carrier recombination. This Rashba effect, together with the effects of SOC on the *m** anisotropy, while beyond the scope of the present paper, would be an interesting avenue for further investigation.

Despite the changes to the positions of the bands, the band curvatures around the Fermi level (and hence m^*) are largely unaffected by SOC. For the R3c structure, we calculate the hole effective mass to be $-0.780 m_0$ with SOC and $-0.669 m_0$ without SOC. Similarly, the electron effective mass changed from 2.89 m_0 with SOC, to 3.06 m_0 without SOC. Note that in these comparisons we have kept constant the direction along which we evaluate m^* : [-0.28366, 1.00000, -0.17603] for the hole mass and [1,1,1] for the electron mass. These correspond to the principle directions that yield the lowest m^* in the non-SOC case, i.e., those that were used in Table I for R3c. Such differences are roughly an order of magnitude smaller than the differences in m^* that we found by comparing the various phases of BFO, and hence we neglect the effects of SOC throughout this paper. Further tests of the effect of SOC on the character of bands as a function of octahedral rotations are available in Fig. S6 (SM [25]).

V. CONCLUSIONS

An important and often limiting factor in perovskite-based photovoltaics is their low charge-carrier mobility. In this paper we have employed first-principles methods to compare the charge-carrier effective masses of both FE and non-FE phases of BFO and explore how the effective masses are influenced by changes in the crystal geometry. We find that the groundstate FE R3c phase has relatively large m^* values compared to those of some non-FE phases. However, these non-FE phases lack the mechanism required for charge separation via the bulk photovoltaic effect. We therefore investigate the m^* -determining factors in BFO, with the aim of uncovering a FE phase with reduced m^* values. We discover that the differences in the m^* values are related to the orbital character at the band edges. BFO structures with Bi p or Fe e_g states, instead of Fe t_{2g} states, at the CB_{min} have lower m_e^* due to the localized nature of the t_{2g} bands. Structures with hybridized Fe e_g -O p states at the VB_{max}, rather than pure O p states, have lower m_h^* since the hybridization leads to a more delocalized state. The change in the states at the CB_{min} leading to decreased m_a^* can be achieved with (i) high tensile uniform strain, (ii) reduction of $[111]_{pc}$ translations of distorted Fe-O octahedra, and (iii) a reduction in octahedral rotation about the $[111]_{pc}$ axis. The Fe e_g state emerges at the VB_{max} only with a reduction in octahedral rotations about $[111]_{pc}$. These findings explain the reasons why the $Pm\bar{3}m$ phase, which has unrotated and untranslated octahedra, has the lowest m^* values of the BFO phases we have considered in this paper. However, since we seek a FE phase and we find that m^* changes little with translation when the octahedral rotation about $[111]_{pc}$ is suppressed, then a structure resembling the tetragonal P4mm phase would provide both ferroelectricity and reduced m^* . Our results demonstrate that manipulation of crystal geometry, which is easily achieved with advanced growth techniques, is a viable avenue to tune the electronic properties of materials particularly at the band edges.

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