From passive degassing to violent Strombolian eruption: the case of the 2008 eruption of Llaima volcano, Chile.

Dawn C.S. Ruth[1,2*‡], Elizabeth Cottrell[2], Joaquín A. Cortés[3], Katherine A. Kelley[4], Eliza S. Calder[3]

* Corresponding author: sdawn@ntu.edu.sg

† Now at Earth Observatory of Singapore, Nanyang Technological University, Singapore, 639798

1. Department of Geology, State University of New York at Buffalo, Buffalo, NY 14260, USA
2. Smithsonian Institution, National Museum of Natural History, Washington, DC, 20560, USA
3. School of Geosciences, University of Edinburgh, Scotland, EH9 3JW
4. Graduate School of Oceanography, University of Rhode Island, Narragansett, RI, 02882, USA

ABSTRACT

On 1 January 2008 Llaima volcano, a basaltic andesite stratocone in southern Chile, entered a phase of violent Strombolian eruption. Llaima, like many passively degassing systems, has experienced prolonged (decades-long) periods of persistent summit degassing from its open vent. The rapid transition from long-lived passive degassing to violent explosive eruption occurred with limited precursory monitoring signals. This paper is motivated by the desire to understand what occurs in these systems when that switch takes place. To this end, we study the products of the 2008 violent Strombolian eruption of Llaima volcano, Chile. We present new textural analyses of scoria and geochemical data from five whole rock samples, troctolite glomerocrysts with and without Cr-spinel, and 182 olivine-hosted melt inclusions from tephra samples. Two populations of scoria ("brown" and "black") are distinguished by variable crystallinity and vesicularity, but are geochemically indistinguishable. Black scoria contain abundant microlites with tabular to acicular morphologies and convolute vesicles up to 1.75 mm in effective diameter. The brown scoria tend to have fewer, acicular microlites, abundant matrix glass, and round vesicles with a narrower size distribution constrained to <0.4 mm in diameter. Overall, textures from the black and brown scoria evidence a textural maturation process where shallow system magma becomes more crystal rich and likely rheologically stiffer as a result of prolonged passive degassing. The Cr-spinel and Cr-spinel free troctolite glomerocrysts have plagioclase and olivine compositions of An_{65-92} and Fo_{81}, respectively. Cr# in the Cr-spinel range from 26-37; these are consistent with magma originating from the deeper plumbing system. Whole rock compositions for the tephra average 51 wt% SiO_2, 18 wt% Al_2O_3, and ~6 wt% MgO. The major elements of olivine-hosted melt inclusions range from 49 to 56 wt% SiO_2, and 3.72 to 7.76 wt% MgO; there was no distinct compositional difference between olivine-hosted melt

41 inclusions sourced from the different scoria. Melt inclusion volatile contents range from below

42 detection up to 2.95 wt% H_2O, and 1973 ppm CO_2 (though not in the same melt inclusion). H_2O

43 and CO_2 concentrations are consistent with open system degassing and, when compared with

44 fractionation indices (e.g. K_2O), indicate coupled degassing and crystallization throughout the

45 system. The majority of melt inclusions denote a single liquid line of descent indicative of

46 plagioclase and olivine fractionation. Entrapment pressures range from 8-342 MPa and fall into

47 two groups: 8-100 MPa (300 m to ~4 km depth) and > 100 MPa (4-14 km), revealing that this

48 eruption tapped a deep plumbing system.

49   We propose here that passive degassing at Llaima is maintained by periodic small-batch

50 magma injections. Consequently, due to extensive degassing the upper plumbing system magma

51 crystallized and increased in viscosity. Before the 2008 eruption, some volatiles sourced from the

52 repeatedly-injected magmas, exsolved from the inferred crystal mush, and ascending from

53 deeply-sourced degassing magmas gradually accumulated within the crystal mush and beneath

54 the stiffened conduit magma. Our results support a model where eruption triggering occurred

55 when magma injection remobilized the mush and, importantly, unlocked the accumulated gasses

56 that ascended rapidly, and generated the observed violent Strombolian explosive activity. Our

57 proposed model contrasts with those models for explosive mafic volcanism which require rapid

58 magma ascent under closed system degassing conditions. Importantly, our proposed mechanism

59 provides a means for systems with dominantly open system degassing behavior to transition

60 from passive degassing to explosive eruptions.

61

INTRODUCTION

On the 1st of January 2008, Llaima volcano in the Southern Andean Volcanic Zone, Chile, entered a new phase of intense explosive activity. The eruption, described as a violent Strombolian type eruption, was characterized by highly pulsatory fire fountaining and the development of a 10 km high ash plume (Ruth and Calder, 2014). Prior to the eruption the volcano exhibited persistent passive degassing at least since its previous eruption in 1994 (OVDAS, 2008). Passive degassing (sometimes called persistent degassing, or quiescent degassing) is characterized by volatile outgassing, usually from a summit vent, but is commonly associated with limited or negligible magma net flux out of the system. In many cases volcanoes exhibiting this behavior are typified by mafic magmas and open vents, and the persistent degassing can continue for months to years (Rosi et al., 2000; Rose et al., 2013). In contrast, violent Strombolian eruptions (often referred to as paroxysms, or cineritic eruptions) are characterized by moderately-sized, sustained ash plumes (> 10 km in altitude) and dispersed tephra deposits (Francis et al 1990; Valentine and Gregg, 2008; Houghton and Gonnermann, 2008). Violent Strombolian eruptions are known to occur periodically at volcanoes that commonly present passive degassing, including, Stromboli, Italy (Barberi et al., 1993; Rosi et al., 2000; Rosi et al., 2013); Llaima and Villarrica, in Chile (Petit-Breuihl, 2006; OVDAS, 2008), Masaya, Nicaragua (Williams, 1983); and Fuego, Guatemala (Rose et al., 2008). Where these violent explosive eruptions occur from open, already passively degassing systems, their onset can be sudden, sometimes with little to no advance warning or signals in monitoring data. The hazards associated with such events (ballistics, tephra fall, lava flows, and sometimes lahars) can be challenging to ameliorate due to the sudden onset nature, especially if the volcano is located in a region of high population (e.g. Rose et al., 2008; Rose et al., 2013; Rosi et al., 2013).

87   Understanding the mechanisms that initiate this rapid change in behavior, from passive

88   degassing to violent explosive activity, is therefore a key challenge to improving our capacity to

89   deal with explosive basaltic volcanism.

90   Investigations on this style of explosive basaltic volcanism have been undertaken at

91   Stromboli, Mt. Etna, Parícutin, and Cerro Negro (Allard, 1997; Roggensack et al., 1997; Métrich

92   et al., 2001; Lautze and Houghton, 2007; Pioli et al., 2008; Johnson et al., 2010; Métrich et al.,

93   2010). Violent Strombolian eruptions necessitate not only high mass fluxes of magma (e.g., mass

94   eruption rates 10^3-10^5 kg s^{-1}), but also, critically, high instantaneous gas fluxes (Pioli et al., 2009;

95   2009). In several eruption cases, the triggering of these events have been attributed to bouts of

96   recharge of mafic magma that ascend rapidly through the system, e.g. Stromboli (Métrich et al.,

97   2001; Bertagnini et al, 2003; Métrich et al., 2010), Parícutin (Lurh, 2001; Pioli et al., 2008), and

98   Cerro Negro (Roggensack et al., 1997). Deeply-sourced magma injection into a relatively

99   shallow reservoir is inferred from the volatile contents of mineral-hosted melt inclusions. The

100   inferred rapid closed system ascent and subsequent degassing of H_2O in these magmas then drive

101   the sporadic explosive eruptions, often producing texturally bimodal scoria populations with

102   chemically distinct olivine hosted melt inclusions (e.g., Métrich et al., 2010). However this

103   model does not work for systems where erupted products record dominantly open system

104   degassing behavior (e.g. Anak Krakatau, Mandeville et al., 1998), or systems that are not well

105   characterized by either closed or open system degassing (e.g. Johnson et al., 2010; Oppenheimer

106   et al., 2011). Alternative mechanisms that have been proposed to explain the transition from

107   passive degassing or low intensity eruptions to violent explosive eruptions include changes in

108   magma system geometry (Palma et al., 2011; Genereau et al., 2010) and changes in the fluid

109   dynamics during magma ascent (Jaupart, 1998; Houghton and Gonnermann, 2008).

110   Studies of the products of the 2008 eruption of Llaima have been undertaken by Bouvet

111   de Maisonneuve et al. (2012a, 2012b) and Ruth and Calder (2014). Ruth and Calder (2014)

112   report on morphological, textural and dispersal characteristics of the diverse tephra types,

113   including distinctive plate tephra. Bouvet de Maisonneuve et al. (2012a) present olivine hosted

114   melt inclusion data for the 2008 scoria, as well as for three historical eruptions in 1957, and

115   ~1850 AD (separated into upper and lower tephras). Their melt inclusion dataset evidences a

116   persistent shallow plumbing system (< ~130 MPa) through this time, but provides limited

117   evidence for consistent degassing and fractionation trends. For the 2008 eruption, mineral zoning

118   and chemistry data suggest repeated pre-eruption magma injection (Bouvet de Maisonneuve et

119   al., 2012b). Based on these findings, they propose that the 2008 eruption was triggered by

120   remobilization of a crystal mush (i.e. magmatic regions with >50% crystals) by magma recharge.

121   The Bouvet de Maisonneuve et al. (2012a, 2012b) work provides an excellent framework for

122   understanding reservoir processes associated with several of the recent major eruptive episodes

123   at Llaima, but key aspects of the 2008 eruption remain unaccounted for.

124   In this work we focus specifically on the products of the 2008 eruption. We extend the

125   work undertaken by both Bouvet de Maisonneuve et al., (2012a, 2012b) and Ruth and Calder

126   (2014) and use a combined textural and geochemical analysis that both complements previous

127   work, and provides new information about the shallow and deeper system. Specifically, we build

128   on the previous work in the follow ways: 1) Ruth and Calder (2014) identified texturally

129   heterogeneous tephra produced during the violent explosive phase of the 2008 eruption. Here we

130   use textural data, including vesicle and crystal size distributions, from these different tephra to

131   investigate crystallization and degassing processes in the shallow system and discuss these in the

132   context of the explosive eruptive activity. This information is important because shallow

133   degassing and crystallization are considered critical processes that can lead to increases in the

134   intensity of mafic eruptions (e.g. Sable et al., 2006; Houghton and Gonnermann, 2008). 2) From

135   the different tephra types we provide major, trace, and volatile elements from 182 olivine-hosted

136   melt inclusions, thereby extending the work of Bouvet de Maisonneuve et al. (2012a, 2012b).

137   Our comprehensive geochemical data show evidence of a deeper, more CO_2-rich plumbing

138   system than previously reported. 3) We consider it important that any conceptual model for the

139   Llaima plumbing system places the 2008 eruptive episode in the context of the persistently active

140   passive degassing system, which typified the activity both before that eruption occurred and

141   afterwards. Although Bouvet de Maisonneuve et al. (2012a, 2012b) focused dominantly on the

142   nature of the magmatic plumbing and crystal mush processes, the role of gas accumulation was

143   not investigated. Here, bearing in mind that high gas fluxes resulting in highly pulsatory lava

144   fountaining is a key driver of violent Strombolian eruptions (e.g. Pioli et al., 2009), we discuss

145   potential scenarios for gas accumulation within the Llaima plumbing system. Finally we propose

146   a model for the transition in activity style from passive degassing to violent explosive eruptions.

147

148   **LLAIMA VOLCANO AND THE 1 JANUARY 2008 ERUPTION**

149   Llaima volcano (38°41′45″S 71°43′54″W, 3179 m a.s.l) is a basaltic andesite stratocone

150   located in the Southern Volcanic Zone (SVZ) of the Chilean Andes about ~600 km south of

151   Santiago (Fig. 1). Llaima is one of the most historically active volcanoes in the SVZ with 54

152   eruptions since 1640 (Naranjo and Moreno, 2005; Petit-Breuihl, 2006). Statistical analysis of the

153   historic record for eruptions of VEI ≥ 2 indicates an eruption recurrence interval of ~6 years

154   (Dzierma and Wehrmann, 2010). Major explosive activity producing significant eruption plumes

155   (>10 km) and associated tephra deposits occurred most recently in 1957 (Hantke, 1962; Naranjo

156 and Moreno, 2005), with a similar, but lower intensity eruption occurring in May 1994 (Moreno

157 and Fuentealba, 1994). The 1994 eruption lasted 8 hours, produced lava fountains, a 7 km

158 altitude plume and associated tephra deposit, and lahars down the western flanks; the eruption

159 was characterized at VEI 2 (Moreno and Fuentealba, 1994).

160   On 1 January 2008, between 1730 and 1820 local time (GMT - 04:00), a summit eruption

161 began at Llaima. The eruption was characterized by violent pulsating Strombolian and fire

162 fountaining activity which produced a plume reaching ~11 km high. Lava flows were generated

163 on the west flank, and lahars were generated in the Rio Calbuco and Rio Captrén valleys,

164 respectively (Global Volcanism Program, 2008) and resulted in the evacuation of the Conguillío

165 National Park. The main explosive phase lasted ~13 hours, but eruptive activity continued with

166 reduced but intermittent Strombolian activity continuing through July 2008. A second explosive

167 phase occurred on 3 April 2009; and the extended eruptive phase ended by July 2009. In all, the

168 2008-2009 eruptive phase produced lava flows extending 3 km from the vent onto the western

169 and southern flanks, a tephra deposit with a minimum eruption volume of 1.3 x 10^6 kg m^{-3} (Ruth

170 and Calder, 2014), pyroclastic flow deposits over snow (i.e. mixed avalanches) associated with

171 an observed vent on the (north) eastern flank, and lahars formed from the interaction of lava and

172 summit glacier. Retrospectively, minor increases in seismic activity and minor ash explosions

173 were recognized in May and June 2007 but otherwise there was little/no precursory activity

174 (OVDAS, pers. comm.). There is also no evidence that significant edifice deformation occurred

175 prior to January 2008 (Fournier et al., 2010; Remy et al., 2015; Delgado et al., submitted).

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SAMPLE COLLECTION AND PREPARATION

179

180   The tephra samples studied here are from the initial phase of the 2008 eruption, collected

181   in January 2009 and January 2011 from a total of 94 different sites in the fall deposit on the

182   eastern flanks of the volcano (Ruth and Calder, 2014). The tephra samples were characterized in

183   the field based on hand sample color, texture, and morphology as either black scoria, brown

184   scoria, or plate tephra. The black scoria and plate tephra only differ in shape, and can otherwise

185   be considered similar (Ruth and Calder, 2014). From these, twenty thin sections of representative

186   tephra types were made for textural analysis using conventional petrography and back-scattered

187   electron (BSE) imaging. BSE images were collected using the Hitachi SU-70 scanning electron

188   microscope at the University at Buffalo South Campus Instrument Center (SCIC) and the FEI

189   Nova NanoSEM 600 in the Department of Mineral Sciences at the National Museum of Natural

190   History (NMNH), Smithsonian Institution.

191   Tephra were separated by type (black and brown scoria, and plate tephra) and were

192   lightly crushed using a jaw crusher and shaker sieved with screen sizes of 25, 40, and 80 mesh

193   (e.g., openings of 710, 425, and 180 μm, respectively). Whole rock analyses were conducted on

194   the > 25 mesh separate, as well as gravel-sized hand-picked samples where homogeneity could

195   be ensured. Olivine phenocrysts were hand-picked from the 25 and 40 mesh separates of the

196   crushed tephra.

197   Clear, naturally quenched, glassy melt inclusions hosted in olivine crystals were selected

198   for analysis (see supplemental material for database of melt inclusions) (Fig. 2). Melt inclusions

199   containing daughter crystals were excluded. Some melt inclusions showed visible fractures but

200   the fractures did not extend to the crystal surface. Melt inclusions contained zero to one vapor

201   bubble. We include in the supplemental data the approximate location in the host olivine of the

202 melt inclusions. Acceptable samples were mounted in epoxy in single-grain mounts to create

203 wafered samples with the inclusions double-side exposed and polished. A total of 182 melt

204 inclusions were analyzed.

205

ANALYTICAL METHODS

X-ray Fluorescence

208   Whole rock major elements and some trace elements (Rb, Sr, Y, Zr, V, Ni, Cr, Nb, Ga,

209 Cu, Zn, Co, Ba, La, Ce, U, Th, Sc, Pb) for five tephra samples were analyzed with a

210 PANanalytical 2404 X-ray fluorescence spectrometer (XRF) hosted at Franklin and Marshall

211 College, Pennsylvania. Major element analyses were conducted on dried powder mixed with

212 lithium tetraborate. The mixture was melted in a platinum crucible and then quenched into a flat,

213 glass disk. Total iron as Fe_2O_3 was reported from the XRF analysis and the speciation of iron

214 was measured using titration on a separate aliquot from the crushed samples (after Reichen and

215 Fahey, 1962). Loss on ignition was determined after samples were heated for one hour at 950 °C.

216 The whole-rock powder was mixed with a high purity Copolywax powder and then pressed into

217 a briquette. Trace elements analyses were conducted on this briquette. Calibration curves were

218 constructed with standards listed in Abbey (1983) and Govindaraju (1994). One sample was run

219 in triplicate to determine reproducibility. For SiO_2, Al_2O_3, CaO, Na_2O the replicate analyses

220 showed a < 1% difference; Fe_2O_3, TiO_2, MgO, and K_2O had a 1-2% difference. Replicate P_2O_5

221 and MnO analyses were identical.

222

Electron Microprobe

224   Major element analyses of carbon-coated melt inclusions, matrix glass, plagioclase, and

225   olivine were collected using the five-spectrometer JEOL 8900 electron microprobe (EMPA)

226   located at The National Museum of Natural History, Smithsonian Institution. Select plagioclase

227   and matrix glass analyses were obtained with the JEOL 8900 electron microprobe at Cornell

228   University. Spinel inclusions in the glomerocrysts were analyzed using the five-spectrometer

229   JEOL JXA-8430 field emission electron microprobe located at the Facilities for Analysis,

230   Classification and Testing (FACTS) at Nanyang Technological University, Singapore. Glass

231   analyses were collected on at least five spots per melt inclusion with a 10 μm beam diameter, 15

232   keV accelerating voltage, and 10 nA beam current (see supplemental materials for details). To

233   minimize alkali loss, we analyze Na and K first, and for shorter on peak durations (10 sec). Our

234   chosen wide beam diameter also minimizes alkali loss (e.g. Hayward, 2011). These techniques

235   are reasonable since analysis of standards as unknowns (e.g. VG-2) returns Na_2O totals within

236   3% of reported values. Sulfur and chlorine were run after the major element analysis on a

237   minimum of three new spots with 15 keV, 80 nA beam current, and 10 μm beam diameter.

238   Sulfur peaks were initially obtained via a peak search routine on individual melt inclusions.

239   Since the peak position did not shift more than 2%, the peak position was determined using the

240   search routine on scapolite during standardization. The closest (~5-10 μm from the melt

241   inclusion) olivine probe point to the melt inclusion was considered the host olivine composition.

242   Olivine analyses were collected with 15 keV, 40 nA beam current, and 1 μm beam diameter.

243   Plagioclase (in thin section and the glomerocrysts) analyses were obtained with an accelerating

244   voltage of 15 keV, 15 nA beam current, and 5 μm beam diameter. Beam conditions for the spinel

245   analyses were: 15 keV, 20 nA, and 1 μm beam diameter. For all analyses, backgrounds on each

246   side of the peak were counted each for half the peak count times. Beam drift was assessed by

247 running standards (glass: VG-2; plagioclase: Bytownite; olivine: San Carlos olivine) as an

248 unknown throughout the analytical run and calculating the percent deviation from reported

249 values.

250

251 **Fourier Transform Infrared Spectroscopy**

252 H_2O and CO_2 concentrations were determined by transmission Fourier Transform

253 Infrared spectroscopy (FTIR) on 49 of the 182 melt inclusions using the Bio-Rad Excalibur

254 spectrometer or the Nicolet 6700 Analytical FTIR hosted at The National Museum of Natural

255 History, Smithsonian Institution, following the methods of Luhr (2001). To minimize

256 atmospheric interference on the spectra, the system (bench, microscope, and samples) was

257 continually purged with dry, CO_2-free air. The spectral region of interest, 1000-6000 cm^{-1}, was

258 obtained with a liquid nitrogen-cooled MCT detector, KBr beam splitter, and tungsten halogen

259 source. Aperture size was selected to ideally obtain three independent spectra per melt inclusion;

260 they ranged from 11x22 μm to 44x44 μm. For smaller melt inclusions, the spectra measurements

261 overlapped. Thickness data were collected either with a piezometric micrometer, and/or

262 indirectly with the wavelength of fringes from 2000-2700 cm^{-1} (Nichols and Wysoczanski,

263 2007).

264 Water concentrations in the melt inclusions were calculated using the total H_2O peak at

265 3535 cm^{-1} (Stolper, 1982a, 1982b). Dissolved CO_3^{2-} concentrations were quantified using the

266 asymmetric stretch doublet between 1515 cm^{-1} and 1435 cm^{-1} (Dixon et al., 1995). The

267 background-corrected peak intensities were obtained through an Excel fitting routine which

268 builds an optimized, synthetic signal that accounts for the vertical offset, the 1630 cm^{-1} total

269 water peak, the CO_3^{2-} doublet, and a straight line (Newman et al., 2000).

270   When fringe amplitude prevented a straightforward assessment of the intensity of the

271   carbonate doublet, we calculated the CO_2 content of the fringe-filtered signal. The amplitude and

272   wavelength of the fringes were modeled for each spectrum between 1800 and 2000 cm^{-1} using

273   the curve fitting module in MATLAB (see supplemental material for the MATLAB script). The

274   user imports the FTIR absorbance and wavenumber data via a Microsoft Excel file and selects a

275   region where the script will determine the fringes signal. The fringes signal between 1800 and

276   2000 cm^{-1} can be modelled simply as a linear function overprinted by a sine function. Note that

277   this region may be overprinted by the Si-O signal from analyzed olivine (Nichols and

278   Wysoczanski, 2007); if overprinting is present then this technique is not suitable to correct for

279   fringes. The script fits a sine curve to the fringes, subtracts this from the original spectrum, and

280   outputs the results into a comma-separated text file. CO_2 values were then re-calculated for the

281   fringe-subtracted signal using the Excel fitting routine described above. The difference between

282   the fringe-filtered and unfiltered calculation is less than 5% (Fig. 3), indicating that the presence

283   of fringes does not significantly affect the quantification of carbonate in these samples. In some

284   cases, the CO_3^{2-} doublet was completely overwhelmed by the fringes and CO_2 values could not

285   be reliably retrieved. The CO_2 detection limit for these melt inclusions (up to 80 μm thick) was ~

286   400 ppm. For these samples, we only report H_2O values. In total, we performed this correction

287   on spectra from 12 melt inclusions (see supplemental material for raw and fringe-corrected data).

288   Water was quantified using the spline function built-in in the Thermo Fischer OMNIC

289   spectral analysis software package. Peak heights were collected on the background corrected

290   spectra. Molar absorptivities at 1515 and 1435 cm^{-1} for C and 1635, 4500, and 5200 cm^{-1} for H

291   species were corrected for major element composition according to Dixon et al. (1995). A

292 constant value for molar absorptivity of 63 mol^{-1} cm^{-1} was applied to all 3535 cm^{-1} peaks as it is

293 not sensitive to composition (Dixon et al., 1995; Kelley and Cottrell, 2012).

294

Secondary Ion Mass Spectrometry

296   Secondary ion mass spectrometry (SIMS) was performed on 39 additional melt

297 inclusions because the detection limit for CO_2 by SIMS is lower (~20 ppm) relative to FTIR (~50

298 ppm) (Hervig et al., 2003). Additional volatile analyses (H_2O, CO_2, S, Cl, F) were collected on a

299 subset of olivine-hosted melt inclusions at the Northeast National Ion Microprobe Facility,

300 Woods Hole Oceanographic Institution, using the Cameca IMS 1280 secondary ion mass

301 spectrometer following the analytical procedure of Helo et al. (2011). These samples were

302 polished to 0.5 micron grit (single-side only), mounted in indium and gold coated samples were

303 then placed under vacuum (<5x10^{-9} torr) and bombarded with a 15 μm, 1.2-1.5 nA, 10kV Cs^{+} ion

304 beam to produce the secondary ions (^{12}C, ^{16}O^{1}H, ^{19}F, ^{30}Si, ^{32}S, and ^{35}Cl) with ^{30}Si as the

305 reference mass. The samples were pre-sputtered for 4 minutes and the primary beam rastered

306 over an area of 30 x 30 μm. Analysis continued after pre-sputtering with an aperture reduced to

307 15 x 15 μm. Count times for ^{16}O^{1}H and ^{12}C were 10 s for 10 cycles for a total of 100 s. ^{19}F, ^{30}Si,

308 ^{32}S, and ^{35}Cl count times were 5 s for 10 cycles for a total of 50 s. Calibration curves for

309 measured ^{12}C/^{30}Si, ^{16}O^{1}H/^{30}Si, ^{19}F/^{30}Si, ^{32}S/^{30}Si, ^{35}Cl/^{30}Si versus the respective volatile

310 concentration were built from the analysis of 11 standard basaltic to andesite glasses (49-57 wt%

311 SiO_2) (Helo et al., 2011) using a York regression (see supplement for calibration curves). R^2

312 values were 0.99 for all volatiles measured; the reported error includes the measurement error

313 and the error in the linear fit (see supplemental material).

314

Laser Ablation Inductively Coupled Plasma Mass Spectroscopy

315

316   129 melt inclusions were analyzed for thirty-nine trace elements using laser ablation

317   inductively coupled plasma mass spectroscopy (LA-ICPMS). These analyses were done using

318   the Thermo X-Series II quadrupole ICP-MS coupled with a New Wave UP 213 Nd-YAG laser

319   ablation system located at the Graduate School of Oceanography, University of Rhode Island.

320   Spot sizes ranged from 20 – 80 μm depending on melt inclusion size. Ablation time was

321   maximized for the wafered samples with an energy output of 60% and 5 Hz repeat rate.

322   Background and washout were collected for 30 seconds, and the maximum analytical time was

323   60 seconds. Our methodology was modified from Kelley et al. (2003) and Lytle et al. (2012). We

324   used ^{43}Ca as an internal standard and calibrated on eight natural glasses from the USGS (BIR-1g,

325   BHVO-2g, BCR-2g) and Max Planck Institute (GOR132-G, StHls-G, T1-G, ML3B-G, KL2-G);

326   reference values are as reported in Kelley et al. (2003), Jochum et al. (2006), and Lytle et al.

327   (2012). Acceptable linear calibration curves had $R^2 > 0.995$ (see supplemental material for more

328   information). For Ag, Pt, and W, a lower R^2 value is reported due to limited studies on the

329   standards used. Lab reproducibility is ≤ 5% for most trace elements.

330

X-ray Absorption Near Edge Structure Spectroscopy - (micro-XANES)

331

332   Two melt inclusions (blk3-2011 and blk1-2010) were analyzed *in situ* for $Fe^{3+}/\sum Fe$ ratios

333   using micro X-ray Absorption Near Edge Structure (micro-XANES) spectroscopy with precision

334   of ±0.005 (Cottrell et al., 2009). Following the methods of Cottrell et al. (2009), the two melt

335   inclusions were analyzed on beamline X26a located at the National Synchrotron Light Source,

336   Brookhaven National Lab. Spectra that included olivine interference were excluded (e.g., Kelley

337   and Cottrell, 2009; Brounce et al., 2014).

TEXTURAL ANALYSIS METHODS AND DEFINITIONS

Definitions

Textural analysis of crystal assemblages provides information about the kinetics of magmatic processes, especially crystallization and degassing (Hammer, 2008). We define phenocrysts here as crystals that are > 100 μm and microphenocrysts as crystals that range from 30 to 100 μm. Microlites are crystals under 30 μm. Vesicles are described as rounded or convolute (i.e. irregular perimeters). Qualitative crystal shape terms used in sample characterization include acicular, tabular, swallow-tail and hopper morphologies (e.g. Lofgren, 1974; Couch et al., 2003; Crabtree and Lange, 2011) (Fig. 4). Acicular crystals are characterized by needle shape habits whereas tabular crystals are rectangular. Swallow-tail plagioclase crystals have forked terminations, often with acicular habits. Hopper (or skeletal) morphology plagioclase crystals are those with voids often in the center (Fig. 4).

Vesicle Size Distributions

We measured the vesicle size distributions (VSDs) to quantify the textural differences between the scoria (e.g. Polacci et al., 2006; Moitra et al., 2013). BSE images at x70 and x90 magnification were made binary for vesicles or crystals. Vesicle walls were manually drawn in the images in the case of breakage during sample preparation. First, we determined the phenocryst areas by manually tracing each crystal in an image; these values were subtracted from the total area measured. Then, where necessary, we reconstructed all bubble walls if portions of the septum were visible to quantify the maximum number of vesicles per image. More than 1000 vesicles were measured per thin section for two black and brown scoria samples, respectively. The vesicle data obtained include: area, major axis, minor axis, and perimeter. Equivalent radius

361 and diameter were calculated from the area values. Since we are interested in an internal

362 comparison, here we report 2D values to avoid complications due to stereological conversion

363 (Polacci et al., 2006).

364

365 **Crystal Size Distributions**

366 We used BSE images to collect crystal size data of plagioclase crystals from the brown

367 and black scoria. Long and short axes were measured using ImageJ freeware

368 (http://rsb.info.nih.gov/ij/). We estimated crystal aspect ratios using the CSDSlice technique

369 (Morgan and Jerram, 2006) and considered the textural fabric as massive; we did not correct for

370 crystal roundness. Using petrography and built-in features of ImageJ, we determined the total

371 slide area, volumetric phase abundance, and vesicularity. With these data we calculated the

372 crystal size distributions (CSDs) of the plagioclase using the CSD Corrections 1.51

373 (http://www.uqac.ca/mhiggins/csdcorrections.html) after Higgins (2006). Higgins (2000) reports

374 maximum errors of ±0.5 ln units in the population density.

375

376 **PETROGRAPHIC RESULTS**

377 **Scoria Textures**

378 The black scoria usually occurs as sub-angular lapilli, although flatter examples (i.e.

379 "plate tephra" - Ruth and Calder, 2014) are common in the proximal (< 9 km) deposit. The

380 brown scoria only occurs as sub-angular, highly vesicular lapilli. General crystallinity

381 (groundmass + phenocrysts) for the black scoria is between 50 to 60 vol%, whereas the brown

382 scoria crystallinity ranges from 10 to 15 vol% (Ruth and Calder, 2014). Average vesicularity for

383 both scoria types was derived from density measurements (after Houghton and Wilson, 1989).

384 Black scoria vesicles tend to be large, convoluted, with an average vesicularity of 79%.

385 Conversely, vesicles in the brown scoria tend to be smaller and rounder, with an average

386 vesicularity of 88% (Ruth and Calder, 2014).

387   Phenocryst phase assemblage is plagioclase (~1-2 vol% in brown scoria, 4-7 vol% in

388 black scoria), olivine (<1 vol% in both scoria), rare titanomagnetite and orthopyroxene (<< 1

389 vol%). Plagioclase is the dominant mineral in the groundmass and is found in all scoria types.

390 Titanomagnetite and orthopyroxene are present as groundmass phases in the black scoria, but are

391 rarely observed in the brown scoria.

392   Plagioclase phenocrysts show significant signs of disequilibrium including sieve textures

393 and compositional zoning (also observed by Bouvet de Maisonneuve et al., 2012b). Some

394 plagioclase phenocrysts show additional features such as dendrite-like or swallow-tail

395 terminations (Fig. 5a). Plagioclase microphenocrysts have an euhedral, tabular crystal habit and

396 instances of hollow cores, hopper shapes, and minor swallow-tail terminations. They show little

397 to no sieve textures and are sometimes normally zoned; these are found only in the black scoria.

398   A wide range of microlite morphologies are observed in the black scoria, where habits

399 range from tabular to acicular, often in an interlocking-like configuration (Fig. 5b). Within the

400 brown scoria, microlite habits are dominantly acicular, though tabular microlite are sometimes

401 observed. Crystal free matrix glass is limited within the black scoria, whereas it is more abundant

402 in the brown scoria (Fig. 5c). Additionally, microlite-rich textural domains, similar to the black

403 scoria, are present in the brown scoria (Fig. 5d). Late-stage oxide microlites exhibiting cruciform

404 habits are often found along the boundary separating the two textural domains (Ruth and Calder,

405 2014). Additional images are provided in the supplemental material. A summary of the textural

406 features of the brown and black scoria is given in Table 1.

407   Plagioclase and olivine (i.e. troctolite) glomerocryst clusters are also present in all tephra.

408   They are subdivided into two groups based on textural features and mineralogy (Fig. 6). One

409   suite of glomerocrysts shows anhedral, equigranular, plagioclase and olivine with limited

410   amounts of matrix glass containing round vesicles (Fig. 6a). The plagioclase phenocrysts are

411   minimally sieve-textured. Subhedral to euhedral spinel are observed as inclusions in pristine

412   plagioclase and olivine crystals; these are hereafter referred to as Cr-spinel troctolite

413   glomerocrysts. In the other suite of glomerocrysts, the plagioclase phenocrysts show significant

414   sieve textures, the olivine phenocrysts are much smaller and less abundant, matrix glass is

415   generally more abundant with convolute and round vesicles, and the spinel phase is rare (Fig.

416   6b). These are referred to as Cr-spinel-free troctolite glomerocrysts. Cr-spinel troctolite

417   glomerocrysts are generally found in the brown scoria, whereas the Cr-spinel-free troctolite

418   glomerocrysts are found in the brown and black scoria. The glomerocrysts were noted but not

419   studied by Bouvet de Maisonneuve et al. (2012a), and were thought to originate from a crystal

420   mush. This inference is reasonable given that the glomerocrysts mineral chemistry is similar to

421   that of the scoria. Additionally, it is unlikely that they are assimilated country rock given that the

422   underlying basement comprises evolved Tertiary intrusives (Lucassen et al., 2004).

423

Crystal and Vesicle Size Distributions

425   Results from the vesicle textural analysis are shown in Table 2. Vesicle number densities

426   are similar for the brown scoria samples (e.g. 124 and 142), whereas the black scoria samples

427   exhibit higher and lower values (e.g. 106 and 166). The larger vesicle number density value for

428   1Black is puzzling given that the crystallinity and vesicularity are distinctly different than the

429   brown scoria (see below). One explanation is that the textures could reflect mingling of magma

430   at different textural stages (e.g. Lautze and Houghton, 2007). The black scoria have a larger

431   average vesicle diameter (0.074 mm) and a wider range of vesicle sizes (<0.01 to 1.74 mm),

432   whereas vesicles in the brown scoria are smaller (0.044-0.055 mm) and are characterized by

433   narrower range of sizes (<0.01 to 0.4 mm) (Fig. 7). The black and brown scoria have broadly

434   similar vesicle size distributions (VSDs) with linear segments (Fig. 7), indicative of steady state

435   nucleation and growth (Mangan and Cashman, 1996). However, the VSD of the black scoria

436   diverges from the brown scoria around 0.1 mm (see Fig. 7 inset), and is characterized by a lower

437   slope indicative of longer growth rates. The brown scoria shows a small plateau in the smallest

438   vesicle diameters, which may reflect the initial stages of vesicle ripening (Mangan and Cashman,

439   1996).

440   The statistical summary for the plagioclase crystal size data is presented in Table 3.

441   Crystal number density (CND), the number of crystals per unit area, generally decreases from the

442   black scoria (including plate tephra for comparison) to brown scoria. Sample BrownLL has an

443   exceptionally large CND and the average crystal size for this sample (~4 μm) is almost half the

444   average for the other brown scoria (7 and 9 μm). The black and brown scoria have generally

445   concave-up crystal size distributions (CSDs), whereas the plate tephra have a kinked profile (Fig.

446   8). The plate tephra have a significant component of larger crystal sizes, which manifests as

447   another kink in the CSD; it is unclear to what extent this might be a sampling artifact. However,

448   the plate tephra formed from ruptured bubble films, probably from magma that resided near the

449   conduit margins (Ruth and Calder, 2014). Magma near the conduit margins likely had the longest

450   residence times, thus producing the highest crystallinities and largest crystals. One sample of

451   brown tephra, brn1, exhibits textures similar to that of the black scoria, which could suggest it is

452   a transitional texture. The concave-up distribution has been attributed to super exponential

453 nucleation events (i.e. growth rate is a function of crystal size; Marsh, 1998) while a kinked

454 profile is commonly observed in igneous systems and interpreted as a signature of magma

455 mixing (Marsh, 1998; Higgins, 2006; Vinet and Higgins, 2010). Overall, the slope decreases

456 from the brown to black scoria and plate tephra. Similarly, plot intercepts decrease slightly from

457 the brown to black scoria and plate tephra (see Fig. 8 insets). Changes in slope can be attributed

458 to increased residence time and/or coarsening (Higgins, 2006). Plot intercepts are not expected to

459 change with increased residence time, however they do decrease with coarsening (Vinet and

460 Higgins, 2010). BrownLL has the highest intercept of all the tephra, which combined with the

461 high crystal number density, suggests a rapid nucleation event. Since the plot intercepts decrease

462 in these datasets, the change in slope might result from coarsening via Ostwald ripening. Both

463 plate tephra and one black scoria (1Black) samples show a kink in their CSD at the smallest

464 crystal sizes, which is not observed in the brown or black scoria. This extra kink could suggest

465 that these samples may have experienced coarsening with a nucleation event right before

466 eruption. Similar distributions and interpretations were presented theoretically (Marsh, 1998) and

467 in samples from Soufrière Hills volcano (Higgins and Roberge, 2003).

468

GEOCHEMICAL RESULTS

Whole Rocks - Major Elements

471 The whole rock composition of the tephra produced during the 2008 eruption is basaltic

472 with ~51 wt% of SiO_2 (n=5 whole rock samples), consistent with previously reported whole rock

473 chemistry from recent Llaima products (Naranjo and Moreno 2005; Reubi et al., 2011; Bouvet de

474 Maisonneuve et al., 2012a). These and additional geochemical data are provided in the

475 supplemental material. The black scoria samples have SiO_2 values of 51.2 to 51.59 wt%; brown

476   scoria samples have similar SiO_2 values, at 51.20 to 51.22 wt%. These data suggest, unlike the

477   textural data, that there is no discernible geochemical difference in whole rock chemistry

478   between the black and brown scoria. This finding is consistent with texturally bimodal tephra

479   reported elsewhere (e.g. Métrich et al., 2010; Cimarelli et al., 2010).

480

481   **Phenocrysts - Major Elements**

482   Mineral chemistry was determined on the phenocryst phases, the plagioclase

483   microphenocrysts, and all phases in the glomerocrysts (Table 4). The composition of the isolated

484   plagioclase phenocrysts and microphenocrysts ranges from An_{85-65} with an average of An_{69} (Fig.

485   9). Isolated olivine phenocryst compositions range between Fo_{83-66}. Plagioclase within the Cr-

486   spinel-free troctolite glomerocrysts ranges from An_{92-65}, whereas the plagioclase in the Cr-spinel

487   troctolite glomerocrysts is slightly more calcic at An_{92-75}. Olivine composition within both types

488   of glomercrysts clusters around Fo_{81}. Zoning information is limited to a small subset (n=14) of

489   olivine crystals; these are dominantly reversed zoned, which is consistent with previously

490   reported observations (Bouvet de Maisonneuve et al., 2012a). The spinel phase is Cr and Al-rich

491   with Cr# (molar $Cr/(Cr+Al+Fe^{3+})$) ranging from 26-37. Cr-spinel in conjunction with high An

492   plagioclase is consistent with origins in the deeper part of the magmatic plumbing system (e.g.

493   Streck et al., 2002). Conversely, the lack of spinel and slightly lower An plagioclase in the Cr-

494   spinel-free troctolite glomerocrysts is consistent with origins in the shallower parts of the

495   plumbing system.

496

497

498

Post Entrapment Modification of Melt Inclusions and Fe Oxidation State

499

500   Melt inclusions may experience compositional modification after trapping. Processes

501   may include crystallization, melting, re-equilibration with the host mineral, and elemental

502   diffusion through the host mineral (for a review, see Kent, 2008). In order to undertake

503   meaningful analysis of the data, estimating the extent of these modifications is important.

504   Post entrapment modifications through Fe loss or crystallization were determined by

505   comparing the Fe-Mg exchange coefficients (K_D) of the melt inclusion and host olivine (Putirka,

506   2008) (Fig. 10). Melt inclusion compositions were corrected for post entrapment crystallization

507   by adding host olivine in 0.01% increments until $K_D(Fe^{2+}-Mg)^{ol-liq} = 0.3$ (Roeder and Emslie,

508   1970) using the Ford et al. (1983) olivine-melt model in the PETROLOG 3 software package

509   (Danyushevsky and Plechov, 2011). $Fe^{3+}/\sum Fe$ ratios were measured via XANES on two melt

510   inclusions; blk3-2011 yielded 0.194 ± 0.002 whereas blk1-2010 was 0.163. These values were

511   used for post entrapment corrections for these samples. For samples that were not analyzed with

512   XANES, our post entrapment crystallization (PEC) corrections assumed an oxygen fugacity

513   equivalent to the nickel-nickel oxide (NNO) oxygen buffer typical of arc systems (Toplis and

514   Carrol, 1995) and a $Fe^{3+}/\sum Fe$ ratio of 0.2, based on literature values for other arc systems (e.g.,

515   Cervantes and Wallace, 2003; Lloyd et al., 2013). We tested this assumption using equation 7

516   from Kress and Carmichael (1991) to estimate the oxygen fugacity buffer with a range of

517    $Fe^{3+}/\sum Fe$ values (Table 5). When 20% Fe is Fe_2O_3, the buffer is ΔNNO +0.06 to +0.14 for the

518   samples shown. At $Fe^{3+}/\sum Fe = 15\%$, the buffer is much closer to ΔNNO (-0.72 to -0.63), or

519   ΔQFM. The buffer did not change considerably with changing pressures in the model. Under this

520   assumption, most samples experienced < 10% post entrapment crystallization.

521   We assessed Fe loss to the host crystal (e.g. Danyushevksy et al., 2000) by comparing the

522   FeO* and Mg# of the melt inclusions (anhydrous, normalized data) to the FeO* and Mg# trend

523   of whole rock samples (Fig. 11). Melt inclusion compositions are consistent with the whole rock

524   array, but extend to significantly more mafic compositions. Four samples showed evidence of

525   post-entrapment Fe-loss. These inclusions were all less than 40 microns, considerably smaller

526   than other melt inclusions in the suite, which ranged from <20 to 425 microns in diameter

527   (average 68 μm). As such, they could re-equilibrate faster than larger melt inclusions and

528   therefore, their measured compositions may not reflect actual magmatic compositions (Cottrell et

529   al., 2002). Due to well documented post entrapment complications from Fe-loss (e.g.

530   Danyushevsky et al., 2000), these melt inclusions were filtered from the suite.

531   Water concentrations in olivine-hosted melt inclusions may be affected by post

532   entrapment diffusion of H, and possibly molecular water, through the olivine crystal lattice

533   (Métrich and Wallace, 2008; Johnson et al., 2008; Portnyagin et al., 2008). Three lines of

534   evidence argue against significant diffusive water loss from the majority of our inclusions. First,

535   significant water loss may cause extensive (>10%) post entrapment olivine crystallization by

536   increasing the inclusion liquidus temperature (Lloyd et al., 2013; Bucholz et al., 2013). We

537   therefore do not include any inclusions with > 10 % post-entrapment crystallization in this study.

538   Second, H loss may initiate precipitation of fine magnetite within the melt inclusion

539   (Danyushevsky et al., 2002; Métrich and Wallace, 2008), though this has been contested in other

540   studies that propose that proton diffusion into the inclusion is rapid and would mitigate any

541   changes in oxidation state (Gaetani et al., 2012; Bucholz et al., 2013). We do not include any

542   inclusions with daughter crystals in this study. Third, comparisons of olivine-hosted melt

543   inclusions sourced from ash, lapilli, and bombs show that samples sourced from lapilli, such as

544   those we report on here, may experience limited amounts of water diffusion (Lloyd et al., 2013).

545   We consider diffusive water loss minimal, though it cannot be entirely excluded. In summary,

546   going forth we consider only glassy melt inclusions that are free of daughter crystals although

547   vapor bubbles may be present, are without significant Fe loss, and have less than 10% post

548   entrapment crystallization.

549   Recent studies have shown that CO_2 concentrations in melt inclusions are also altered by

550   post-entrapment processes (e.g. Neave et al., 2014; Wallace et al., 2015; Aster et al., in

551   preparation). In particular, post-entrapment modifications of the melt inclusion via cooling and

552   crystallization causes a decrease in pressure. CO_2 partitions into the resultant vapor bubble

553   because of the pressure dependence of its solubility. This can be augmented if there is H^+ loss via

554   diffusion into the host mineral. Ultimately, these processes lower the CO_2 content of the melt

555   inclusion by as much as 40 to 90% (Wallace et al., 2015; Aster et al., 2016). Recovering the

556   original CO_2 contents of the melt inclusion requires knowing the volume of the vapor bubble and

557   the CO_2 density within the bubble; however, we do not have these data for our suite of

558   inclusions. Alternatively, original melt inclusion CO_2 contents may be estimated assuming that a

559   bubble forms during thermal contraction associated with crystallization and cooling (Aster et al.,

560   2016); the bubble volume can be estimated with crystallization and thermal contraction models.

561   However, this correction assumes that the olivine have higher entrapment temperatures than

562   eruption temperatures. Of the 182 melt inclusions analyzed in this study, entrapment

563   temperatures were lower than eruption temperatures for 171 melt inclusions (calculated using the

564   technique outlined in Aster et al. (2016)). These melt inclusions are less likely to have

565   experienced thermal contraction. This suggests that the Aster et al. (2016) protocol may not be

566   appropriate for these samples. We report uncorrected CO_2 values and consider these as minima.

567

Major Elements in Melt Inclusions

Our melt inclusion dataset show a wide compositional range, with the most mafic and most evolved compositions between 49 and 56 wt% SiO_2 respectively, expanding on the previously reported compositional range (Bouvet de Maisonneuve et al., 2012a) (Table 6). The matrix glasses show similar trends to the melt inclusions, but with a smaller range of values for all elements (e.g., 52 - 55 wt% of SiO_2). The majority of the samples (152 of 182 melt inclusions) are consistent with a single liquid line of descent (LLD) fractionating of olivine and plagioclase (Fig. 12). CaO and Al_2O_3 trends do not change slope with respect to SiO_2, suggesting that clinopyroxene did not crystallize during the evolution of the melt. These observations are consistent with sample petrography, as the dominant phenocryst assemblage is plagioclase + olivine. MnO and FeO^T (i.e. $FeO + 0.8998Fe_2O_3$) do not exhibit obvious trends with respect to SiO_2. Na_2O and K_2O behave as incompatible elements for the range of SiO_2 presented, all increasing with increasing SiO_2 content.

We observe a tight LLD for TiO_2 and P_2O_5 vs $SiO_2 < 52$ wt%. After 52 wt% SiO_2, we observe a factor of five range in TiO_2 and P_2O_5 values, falling into three groups (see Fig. 12). One group follows the main LLD. The second group shows decreasing TiO_2 and P_2O_5 (23 of 182 melt inclusions). For this subset, Al_2O_3 and Na_2O show slightly elevated concentrations with respect to the main LLD. These findings are consistent with localized assimilation of a lithology more evolved than the main LLD, possibly Tertiary intrusives (e.g. Lucassen et al., 2004) or evolved ancestral magmas potentially with Ti-magnetite (Reubi et al., 2011). A third group shows higher values of TiO_2 and P_2O_5 (7 of 182 melt inclusions). These samples also have some of the lowest CaO, Al_2O_3, and Na_2O concentrations in the dataset. These data are consistent with

590 formation in a residual, interstitial melt in the crystal mush (e.g. Bouvet de Maisonneuve et al.,

591 2012a).

592   The P_2O_5 values reported here are significantly higher than those reported in Bouvet de

593 Maisonneuve et al. (2012a) (see Fig. 12). Our melt inclusion data are consistent with observed

594 matrix glass and whole rock trends. Additionally, the whole rock values for P_2O_5 reported by

595 Bouvet de Maisonneuve et al. (2012a) overlap with the whole rock P_2O_5 values reported in our

596 study. These differences could be attributed to differences in analytical methodology such as

597 count times, or primary standards.

598   The chemical compositions of melt inclusions sourced from the black and brown scoria

599 overlap (Fig. 13); therefore, in this study we treat the melt inclusions as a single set. This

600 observation contrasts with those of Métrich et al. (2010), who propose, for the Stromboli system,

601 that olivine sourced from Stromboli brown scoria (i.e., "golden pumice" in their paper) originate

602 from a different magma than their black scoria.

603

604 **Olivine-Hosted Melt Inclusion Trace Elements**

605   The multi-element spider plot (Fig. 14) shows the range and average of key elements

606 including select rare earth elements (REE) and large-ion lithophile elements, which are used to

607 infer information about petrogenetic processes. All data collected with LA-ICPMS are plotted

608 and overlain by the calculated average value; we also included results from previously reported

609 whole rock trace element data from Llaima Holocene tephras for comparison (dark gray region;

610 Jacques et al., 2014). Our trace element data show a signature generally consistent with

611 subduction zone magmas as evidenced by enrichments of fluid-mobile elements (e.g., Rb, Ba, K,

612 U, Pb), a negative Nb-Ta anomaly, and moderately enriched light REE with respect to heavy

613 REE, as previously reported elsewhere (e.g. Hickey-Vargas et al., 1986; Reubi et al., 2011;

614 Jacques et al., 2014). However, several elemental signatures do exhibit some notable variation

615 from those data in the literature. First, we observe positive Eu anomalies in some of our melt

616 inclusions, and negative Eu anomalies in others. Second, Ba/La in our dataset ranges from 17-46,

617 whereas literature values for Llaima range from 23-27 (e.g., Reubi et al., 2011; Jacques et al.,

618 2014). Similarly, the range of observed Nb-Ta anomalies is wider than previously reported (see

619 Fig. 14). Conversely, Ce/Pb values in our samples (2-4) overlap well with literature values (2-3)

620 (Reubi et al., 2011; Jacques et al., 2014).

621

622 **Volatiles in Melt Inclusions and Matrix Glass**

623   Volatile contents within the melt inclusions were measured with the methodologies

624 discussed earlier; the results are shown in Table 6. H_2O and CO_2 were not measured in the matrix

625 glass because it was not possible to prepare FTIR wafers without crystals that interfered with the

626 path of the IR beam. S and Cl in the matrix glass were measured with EMPA and range from 6-

627 600 ppm and 330 – 500 ppm, respectively.

628   FTIR results for H_2O in the melt inclusions range from 0.36 to 2.97 wt%. Similarly, the

629 SIMS analytical results have H_2O contents ranging from 0.57 to 2.37 wt% H_2O. H_2O values

630 correspond with those reported in previous work (Bouvet de Maisonneuve et al., 2012a) (Fig.

631 15). CO_2 values from the FTIR analysis range from 397 to 1973 ppm, contrasting with the lower

632 CO_2 from SIMS analysis (22 to 274 ppm). We would expect the FTIR analysis to record the

633 lower CO_2 contents. However, many of our samples exhibited fringes (e.g. Nichols and

634 Wysoczanski, 2007), the presence of which could have obscured the small CO_3^{2-} doublet that

635 would result from low CO_2 in the melt inclusions. The CO_2 values presented here are the highest

636 measured in recent Llaima samples to date.

637 Sulfur contents from the EPMA analyses range from 23 to 1534 ppm. SIMS analyses

638 report sulfur abundances from 92 to 1277 ppm. Similarly, chlorine results from EPMA show

639 concentrations from 47 to 1342 ppm, whereas SIMS results have abundances from 236 to 641

640 ppm.

641 We were unable to measure volatiles using all techniques on each melt inclusion, thus

642 direct comparisons between the techniques were not possible with this dataset. However, the

643 measured values from the SIMS for H_2O, S, and Cl overlap well with those reported from EPMA

644 and FTIR (see Table 6). For S and Cl, the EPMA results show a wider range of values compared

645 with the SIMS results. Similarly, the range of FTIR H_2O concentrations is wider than those

646 obtained via SIMS. CO_2 concentrations measured using SIMS were lower, on average, than

647 those measured using FTIR; however, we were only able to retrieve CO_2 values using FTIR from

648 12 samples.

649

650 **Entrapment and Magma Temperature Calculations**

651 Entrapment temperature for each melt inclusion-olivine pair was estimated using olivine-

652 liquid equilibria, which is based dominantly on the Mg partitioning behavior between olivine and

653 the melt and has been calibrated for hydrous melts (Putirka et al., 2007; Putirka et al., 2008). For

654 this study, we used equation 22 from Putirka (2008), which includes the effect of H_2O on the

655 temperature calculation. H_2O contents measured with FTIR or SIMS were included for the

656 respective melt inclusions. We assumed an anhydrous melt for those melt inclusions without

657 H_2O content data. Using Equation 22 from Putirka (2008), the melt inclusion entrapment

658   temperatures ranged from 1094 °C to 1233 °C, with an average of 1175 °C. This range of

659   entrapment temperatures strongly suggests that olivine grew in a wide variety of magmatic

660   thermal conditions, and that cooling-induced crystallization was likely an important process.

661   To further constrain magmatic temperatures in the shallow plumbing system, plagioclase-

662   liquid equilibria, based on the alkali/alkali earth partitioning between the plagioclase and the

663   melt (Putirka, 2008) was also considered. Assuming equilibrium and using the composition of

664   the microphenocryst plagioclase and the anhydrous matrix glass, we estimated the average

665   temperature just prior to eruption at 1146 °C (low 1109 °C, high 1167 °C). We compared these

666   magmatic temperatures with those calculated from the plagioclase hygrometer proposed by

667   Lange et al., (2009). Using the matrix glass H_2O values (0.88 wt%, rounded to 0.9 wt% - Bouvet

668   de Maisonneuve et al, 2012a), the average matrix glass composition (SiO_2: 53.36 wt%, TiO_2:

669   1.34 wt%, Al_2O_3: 15.56 wt%, FeO^T:10.69 wt%, MgO: 5.69 wt%, CaO: 8.91 wt%, Na_2O: 3.18

670   wt%, K_2O: 0.76 wt%), the average plagioclase composition (An_{69}), and pressure at 50 MPa, we

671   calculated the minimum shallow magmatic temperatures as 1115 °C. This is lower than the

672   results from the Purtirka model, but within the range of expected uncertainty (±25-30 °C) for

673   geothermometers in general (Putirka, 2008). Based on this, we assume the calculated 1146 °C

674   temperature as a maximum magmatic temperature.

675

Melt Inclusion Entrapment Pressures

677   The pressure-dependent solubility behavior of CO_2 and H_2O can be used to estimate the

678   entrapment pressures of the individual melt inclusions (Dixon et al., 1995; Métrich and Wallace,

679   2008). That H_2O and CO_2 vary widely in our melt inclusions indicates melt inclusion formation

680   (and associated olivine crystallization) occurred over a wide range of pressures. Two solubility

681 models have been commonly used to estimate entrapment pressures: VolatileCalc (Newman and

682 Lowenstern, 2002) and the Papale et al. (2006) model. Moore (2008) established that the Papale

683 model was the best choice for calc-alkaline magmas because VolatileCalc neglects the effect of

684 Ca on CO_2 solubility. Note that we did not correct for the possible CO_2 migration into vapor

685 bubbles in the melt inclusions and, therefore, consider these values as minimum entrapment

686 pressures. The entrapment pressures for our suite of melt inclusions range from 8 to 342 MPa,

687 based on the Papale model (Fig. 15). Assuming a crustal density of 2500 kg m^{-3}, depths range

688 between ~340 m and 13.9 km. We observe a strong peak between 50-100 MPa (1-4 km),

689 suggesting a crystal rich region (i.e. the crystal mush). These data increase the depth of the

690 crystallizing portion of the plumbing system as reported by Bouvet de Maisonneuve et al.

691 (2012a) by a factor of three.

692

693 **DISCUSSION**

694 **System Wide Processes**

695 *Magmatic Evolution*

696 The majority of our dataset shows evidence for a connected magmatic system that

697 fractionates olivine + plagioclase. The most primitive melt inclusions follow the main liquid line

698 of descent (LLD) (see Fig. 12). More compositional variability is observed in melt inclusions

699 with > 52 wt% SiO_2. Low TiO_2 melt inclusion compositions are most consistent with

700 assimilation of more evolved lithologies, probably with SiO_2 values between 56-58 wt% and

701 very low TiO_2 and P_2O_5. These could represent the basement Teritary intrusives (e.g. Lucassen et

702 al., 2004) or evolved ancestral magmas (Reubi et al., 2011). Positive Eu anomaly values (1.10

703 ±0.24) are indicative of plagioclase dissolution which supports assimilation. This assimilation

704 occurred within the inferred crystal mush zone as evidenced by entrapment pressures ranging

705 from 50 to 150 MPa. Our interpretation is consistent with previous work that suggests limited

706 assimilation occurs potentially as a late-stage shallow process (e.g. Reubi et al., 2011).

707   The high TiO_2 melt inclusions have elevated P_2O_5 and K_2O, and low values of Al_2O_3 and

708 CaO. Eu anomaly values average 0.77 ± 0.05, indicative of plagioclase fractionation. These data

709 are consistent with melt inclusion formation in melts that fractionated a higher modal abundance

710 of plagioclase than the main LLD inclusions. We propose that these melt inclusions formed from

711 interstitial melts in crystal rich regions in the plumbing system (e.g. Bouvet de Maisonneuve et

712 al., 2012a). Only two melt inclusions from the high TiO_2 group have pressure estimates and these

713 are 73 and 309 MPa, respectively. The latter entrapment pressure possibly suggests that crystal

714 rich zones with evolved interstitial melts may be present in the deep system, as well as the crystal

715 mush. Although we cannot quantitatively link the number of melt inclusions to magma volumes,

716 the large number of melt inclusions analyzed (n=182) suggests that melt inclusions from the high

717 TiO_2 (n=7, 3.8% of the sample suite) and low TiO_2 (n=23, 13% of the sample suite) groups both

718 represent volumetrically minor amounts of magma compared to the main LLD. Our dataset

719 contrasts with previously reported interpretations that suggest melt inclusions dominantly formed

720 in isolated melts with different evolution histories (Bouvet de Maisonneuve et al., 2012a).

721

722 *Degassing and Crystallization*

723   The melt inclusions display a wide range of CO_2 contents, from below detection to nearly

724 2000 ppm, but tightly cluster near 2 wt% water down to CO_2 contents of ~100 ppm. The

725 population of inclusions with CO_2 below 100 ppm and with 2 wt% water argues for open system

726 degassing (Fig. 15). In this work, we understand open system degassing as the process in which

727   volatiles exsolve, decouple from the parent magma and then escape through bubble ascent (i.e.

728   fractional degassing) (Dixon and Stolper, 1995; Cashman, 2004). Our findings contrast with

729   previous studies that suggest that explosive eruptions in mafic systems result from rapid

730   magmatic ascent under closed system degassing conditions (e.g. Coltelli et al., 2005; Métrich

731   and Wallace, 2008). For example, paroxysmal eruptions (i.e. violent Strombolian) at Stromboli

732   show distinctly different degassing trends (Fig. 15a), which have been interpreted to record

733   closed system degassing and ascent (Métrich et al., 2010). In closed system degassing, volatiles

734   exsolve to form a vapor phase, and then remain with the parent melt (i.e. equilibrium degassing).

735   When ascent is rapid, the vapor phase increases in volume fraction, and as it is coupled to the

736   magma, it leads to explosive fragmentation. Although, passive degassing processes at the summit

737   have been linked elsewhere to both open and/or closed system degassing processes in deeper

738   plumbing systems (e.g. Burton et al., 2007), the melt inclusion evidence for open system

739   degassing at Llaima indicates that volatiles, even at depth, are decoupled from the magma, and

740   allowed to ascend. Therefore, to produce the 2008 violent Strombolian eruption, a mechanism

741   must be invoked which can explain the observed transition from passive degassing to explosive

742   activity, but is independent of the established deeply-rooted passive degassing system.

743   Water loss through degassing promotes crystallization (e.g. Crabtree and Lange, 2011;

744   Applegarth et al., 2013). Although crystallization does occur during syn-eruptive cooling and

745   degassing, melt inclusion entrapment in olivine is not likely to occur syn-eruption. We explored

746   the relationship between degassing and crystallization by comparing the respective proxy ratios

747   (e.g. Blake et al., 2010; Johnson et al., 2010). Figure 16 displays the volatile/K_2O ratios for H_2O,

748   S, Cl, and CO_2 versus MgO/K_2O. If crystallization occurred without degassing, the data would

749   follow a horizontal trend. If degassing was dominant, the data would follow vertical trend. Our

750   results show strong correlations between MgO/K_2O and H_2O/K_2O, Cl/K_2O, and S/K_2O,

751   indicating that degassing and crystallization are coupled processes. CO_2/K_2O correlated with

752   MgO/K_2O at lower CO_2 values (i.e. lower pressures), but do not correlate at higher CO_2 values.

753   Indeed, the majority of the sample suite falls on the coupled crystallization and degassing trend,

754   including samples with high volatile contents (e.g. high volatile/K_2O ratios). Similarly, scoria

755   matrix glass values fall on the trend with some of the lowest S/K_2O and Cl/K_2O values in the

756   dataset. This indicates that the matrix glass was almost completely degassed due to

757   crystallization. Some melt inclusions have similarly low S/K_2O values, suggesting that the melt

758   inclusions trapped largely degassed magma. These inferences suggest that crystallization and,

759   importantly, degassing were coupled in the deep plumbing system (at least to 14 km depth), as

760   well as the shallow system. These data combined with the wide range of olivine entrapment

761   temperatures suggest that cooling-induced and degassing-induced crystallization processes may

762   be equally important.

763

Shallow System Processes – Passive Degassing

Textural Maturation and the Formation of Black and Brown Scoria

766   The brown scoria exhibit textural data indicative of rapid nucleation at relatively high

767   undercooling, i.e. higher intercept on the CSD plot and high vesicle number density. The inferred

768   initial stages of vesicle ripening in the brown scoria likely occurred during ascent in the conduit

769   just before quenching. Similar interpretations and short timescales were presented for vesicle

770   ripening in reticulite from Hawaii (e.g. Mangan and Cashman, 1996). Conversely, textural data

771   from the black scoria suggest a rapid nucleation event but also more prolonged crystal growth,

772   while vesicle textures suggesting steady-state nucleation with possibly some coalescence. These

773   data are indicative of longer residence times before eruption. Together, the observed textures are

774   indicative of a continuum of textural maturation from the brown to black scoria that we propose

775   results from background passive degassing. The presence of normally-zoned microphenocryst

776   plagioclase predominantly in the black scoria supports this, as it suggests growth-dominated

777   conditions associated with lower, possibly steady-state, rates of degassing (Cashman, 2004;

778   Castro et al., 2005). In summary, the textural features of the black scoria reflect passive

779   degassing conditions in the conduit and likely represent magma slowly circulating within the

780   shallow plumbing system.

781   Texturally bimodal scoria were first observed and described at Stromboli (Francalanci et

782   al., 2004) and have since been observed in other systems (e.g. Croscat complex cone system –

783   Cimarelli et al., 2010). The formation of texturally bimodal scoria in explosive systems has been

784   attributed to two processes, depending on whether they are geochemically distinct, or not. Where

785   scoria originated from geochemically-similar magma sourced from similar locations within the

786   shallow plumbing system, textures are interpreted to record end members on a textural maturity

787   continuum (Lautze and Houghton, 2007; Erlund et al., 2010; Cimarelli et al., 2010). In these

788   cases, differences among scoria types are a function of the physical processes experienced.

789   Alternatively, where distinctions can be made in the geochemistry, such as melt inclusion

790   chemistry, the source magmas for the different scoria are chemically distinct and originate from

791   different locations in the system (Métrich et al., 2001, 2010). At Llaima, the geochemical

792   similarity of the scoria whole rock compositions (Fig. 12) and the fact that melt inclusions from

793   the different scoria overlap (Fig. 13) indicate that differences are textural alone, and likely

794   formed as a result of differing degassing histories, rather than from geochemically different

795   magmas (at least with respect to major element chemistry).

796

Plumbing System Conceptual Model

Summarizing the scoria textures, olivine hosted melt inclusions, as well as findings from previous work (e.g. Bouvet de Maisonneuve et al., 2012a, 2012b, Ruth and Calder 2014) we present here a conceptual model of the plumbing system at Llaima from the vent to ~14 km depth (Fig. 17). We infer two magma accumulation regions based on the clustering of melt inclusion entrapment pressures (histogram in Fig. 15). The main magmatic region is the shallowest, extending from the conduit (300 m) to ~4 km. Magma that produced the black 'resident' scoria is likely sourced from this region. Most melt inclusions formed (or re-equilibrated) at pressures from between 50-100 MPa. This strongly supports the crystal mush zone proposed in previous work (e.g. Bouvet de Maisonneuve et al., 2012a, 2012b). Crystallization of olivine and plagioclase occurred throughout the plumbing system as evidenced by the main LLD. Minor amounts of localized assimilation of evolved lithologies occurred in the crystal mush zone (e.g. low TiO_2 melt inclusions, 50-150 MPa entrapment pressures). High TiO_2 and P_2O_5 melt inclusions with similar crystal mush pressures also evidence entrapment of interstitial melts. Reverse zoning in olivine crystals suggests this as the main region of magma mixing (Bouvet de Maisonneuve et al., 2012b; Ruth, 2014). This magma mixing probably produced the multiply zoned plagioclase and the sieve textured plagioclase phenocrysts, textures which form under dynamic conditions, especially with respect to magmatic H_2O and temperature (Landi et al., 2004; Streck, 2008; Crabtree and Lange, 2011; Bouvet de Maisonneuve et al., 2012b). Since plagioclase crystals within the Cr-spinel-free troctolite glomerocrysts have similar sieve textures, we infer that they are also sourced from the crystal mush zone.

818   Olivine hosted melt inclusions show that the deeper plumbing system extends from 4 to

819   14 km depth. Although we do not have many samples from the deep system, we infer

820   information about the processes occurring within this region from melt inclusion chemistry and

821   the Cr-spinel troctolite glomerocrysts. Degassing occurs at up to 14 km depth, and is coupled

822   with crystallization. Extensive crystallization is suggested by the one melt inclusion with high

823   TiO_2 and P_2O_5 with high entrapment pressure, which may have formed in the interstitial melts

824   within this region. Moreover, the presence of Cr-spinel in some troctolite glomerocrysts is

825   consistent with deeper origins that have experienced limited fractionation (e.g. Streck et al.,

826   2002); these are most likely to be found at depths lower than the main crystal mush zone. This

827   potentially suggests a crystal rich zone in the deeper system (e.g. Annen et al., 2006). Similar

828   crystal rich zones have been proposed for the deeper system beneath Mt. Etna (e.g. Laiolo and

829   Cigolini, 2006).

830

831   **Changes in Magmatic Physical Properties Due to Passive and Syn-eruption Degassing**

832   Magma viscosity is affected by crystal abundance and morphology, which can be altered

833   due to degassing processes. Analog experiments show that increasing microlite abundance with

834   respect to phenocryst abundance can potentially increase effective viscosity by up to three orders

835   of magnitude (Cimarelli et al., 2011). The effective viscosity may increase further as a result of

836   syn-eruption (or passive) degassing and crystallization. Heating experiments on volatile-poor

837   porphyritic basalts reveal that rapid, significant degassing and crystallization (up to 35%) can

838   occur in degassed conduit melts, resulting in additional dendritic to spherulitic crystal nucleation

839   on existing crystals (i.e. overgrowths) (Applegarth et al., 2013, see their Figure 12 and 13). We

840   estimated the viscosity indirectly using the MAGMA software package (version 2.50.0133

841   http://www.lanl.gov/orgs/ees/geodynamics/Wohletz/KWare/Index.htm), the average whole rock

842   composition, and the average total crystallinity of the brown and black scoria (Table 8). The

843   majority of the crystallinity in both cases is microlites since the phenocryst content for both

844   scoria types is less than 10%. The brown scoria magma has a vesicle-free effective viscosity

845   ranging from $5.4{\times}10^1$ to $1.33{\times}10^2$ Pa s. The black scoria magma has a vesicle-free effective

846   viscosity that is generally two to three orders of magnitude higher ($2.29{\times}10^4$ to $5.64{\times}10^4$ Pa s).

847   Our first order approximation of viscosity differences is consistent with findings reported in

848   Cimarelli et al. (2011). Additionally, in the black and brown scoria we observe dendritic

849   overgrowth textures similar to those found in Applegarth et al. (2013), especially around

850   boundaries between textural domains (Fig. 5d), and near vesicles and fractures (see supplemental

851   material Fig. S3). The proximity to vesicles, in particular, suggests formation resulting from late

852   stage degassing. Based on the experimental results from the literature and our textural data, we

853   propose that significant viscosity changes resulted from both passive degassing prior to, and syn-

854   eruptive degassing during, the 2008 eruption at Llaima.

855

Transitions in Eruptive Style at Llaima

857   Maintenance of an open conduit commonly associated with passive degassing requires

858   that heat and magma be continually replenished (Kazahaya et al., 1994; Stevenson and Blake,

859   1998). Heat and magma supply to the upper plumbing system can occur through density-driven

860   convective mixing between the degassed, and cooler shallow magmas and relatively volatile-rich

861   and hotter, deeper magmas (Kazahaya et al., 1994; Allard, 1997; Stevenson and Blake, 1998;

862   Beckett et al., 2014). An additional mechanism for inducing axial mixing which is likely to be

863 important, is that the ascending gas flux in a conduit will itself drive a return flow of magma

864 (Pansino, 2012).

865   We propose that the background activity at Llaima is maintained through periodic small-

866 batch mafic injections, as evidenced by crystal zoning, into the base of a crystal mush (1-4 km

867 depth) (Fig. 17). Resultant hybrid magmas with initially moderate H_2O content (~ 2 wt%) are

868 mobilized and entrained into the axial mixing processes. The ascent and mixing results in

869 degassing and crystallization, and we speculate that this magma would ultimately result in the

870 black scoria type products with the observed mature crystal and vesicle textures.

871   For the 2008 eruptive episode, magma injection and mush remobilization, evidenced by

872 reversely zoned olivine (e.g. Bouvet de Maisonneuve et al., 2012a,b; Ruth, 2014), have been

873 proposed as a triggering mechanism (Bouvet de Maisonneuve et al., 2012a). However, the lack

874 of significant ground deformation prior to the eruption suggests that any magma injection into

875 the system must have been volumetrically minor (Remy et al., 2015; Delgado et al., in

876 preparation). Further, violent Strombolian eruptions are highly pulsatory in nature and require

877 high mass and gas fluxes to drive them (Pioli et al., 2009). Therefore, any robust discussion of

878 the triggers of such an eruption must account for substantial gas accumulation and its rapid

879 ascent. Three mechanisms of potential gas accumulation are discussed below:

880   (i) Crystal mush zones may facilitate gas accumulation. Analog experiments suggest that

881 small vesicles can become trapped within particle-rich suspensions (i.e. gas holdup), which may

882 regulate gas flux through the plumbing system (Belien et al., 2010; Tran et al., 2015).

883 Populations of small vesicles can be added to the crystal mush through volatile saturation and

884 subsequent nucleation. A crystal mush zone at Llaima is inferred at depths between 1 and 4 km,

885 based on the melt inclusion entrapment pressures. With the average whole rock composition, and

886   2 wt% H_2O, saturation pressures (calculated with Papale et al. (2006)) are between 70 and 80

887   MPa (~ 3 km depth), thus it would be reasonable to expect vesicle nucleation within this region.

888   Cr-spinel-free troctolite glomerocrysts have matrix glass with vesicles, lending further support to

889   this hypothesis. These vesicles may have formed during syn-eruption ascent, but their convoluted

890   morphologies suggest some maturity. Further, ascending CO_2-rich vesicles from depth may also

891   be partially trapped in the crystal mush (Thomas et al., 1993; Costa et al., 2006). The passive

892   degassing behavior at the summit and the open system degassing signature of the melt inclusions

893   indicate that gas bubbles within the system are mobile. Some mobility may occur due to crystal

894   settling within the mush (e.g. Boudreau, 2016), with some bubble ascent feeding the passive

895   degassing. However, significant bubble ascent might be inhibited by the surrounding crystals.

896   We propose that gas accumulates within the mush until a critical gas volume is reached. Once

897   the bubbles are released from the crystal mush via remobilization (i.e. unlocked) the accumulated

898   gas would ascend rapidly to generate explosive activity (e.g. Vergniolle and Jaupart, 1990;

899   Parfitt and Wilson, 1995).

900   (ii) Extensive passive degassing could result in increased viscosity for magma residing

901   within the upper conduit. Over time, this magma would stagnate and potentially form a

902   rheologically stiffer (but still permeable) cap, above which magma circulation does not take

903   place. Experimental work suggests that rheological barriers lead to elevated gas overpressures,

904   ultimately increasing the intensity of Strombolian eruptions (Del Bello et al., 2015). Although we

905   do not have direct evidence for such rheological barrier, the black scoria and plate tephra from

906   the 2008 eruption have textures consistent with prolonged residence and slightly higher

907   viscosities (relative to crystal-free magmas of the same composition), as implied by their higher

908   crystallinities (Table 8). Moreover, with increased viscosity, vesicle coalescence and growth

909 could be retarded (Cashman, 2004; Tran et al., 2015) further augmenting gas accumulation

910 beneath the barrier.

911 (iii) Complex plumbing system geometry involving sills could provide additional traps

912 for gas accumulation. Experimental results indicate that sills in passively degassing systems may

913 serve as effective traps by segregating gas via buoyant exchange flow between the sill and

914 conduit (Menand and Phillips, 2007). The plumbing system geometry of Llaima is not well

915 constrained. Regional tectonics and the alignment of vents may indicate that the plumbing

916 system follows a NE-SW orientation (Cembrano and Lara, 2008; Schonwalder-Angel, 2015).

917 However, there is limited discussion of sills and/or other sub-horizontal bodies. Recent studies

918 using InSAR inferred the presence and inflation (6-15 cm) of a sill to the west of the summit at ~

919 5 km depth prior to a subsequent explosive phase on 3 April 2009 (Delgado et al, in preparation);

920 there is assumed post-eruptive subsidence after this event. Although inflation was not observed

921 before the 2008 eruption, the findings in Delgado et al. (in preparation) do suggest that a

922 potential short-lived trap for magma and gas accumulation could have occurred.

923 We propose the following scenario to explain the transition to the violent Strombolian

924 eruption at Llaima on 1 January 2008 (Fig. 17). The eruption occurred in a system that was

925 already passively degassing which was likely sustained by repeated, minor, magma injection

926 events. Upon injection these magma batches degassed volatiles that partially filtered through the

927 crystal mush. Simultaneously, some of these volatiles, as well as volatiles ascending from depth,

928 accumulated in the mush and under the stiffened conduit magma until a critical gas volume was

929 reached. Disruption and remobilization of the mush by injecting magmas unlocked the

930 accumulated gas phase, which ascended rapidly to generate the explosive activity at the vent.

931 Clearing the upper portions of the shallow magma reservoir allowed rapid, and likely chaotic,

932   ascent of the deeper, relatively volatile-rich, magma from depths up to 14 km. Partial mixing

933   between the influx, crystal mush and the resident magma produced the glomerocrysts and the

934   crystal-rich domains represented by the clots of black scoria in the brown scoria (Fig 5b). Thus

935   magma injection and gas accumulation processes equally contributed to eruption triggering.

936

937   **Explosive Eruptions in Mafic Systems**

938   Explosive mafic eruptions are commonly associated with closed system degassing and

939   ascent of magmas from depth (e.g. Roggensack et al., 1997; Métrich et al., 2010). Bouvet de

940   Maisonneuve et al. (2012a, 2012b) suggest that the dominant cause of the 2008 explosive

941   eruption of Llaima was injection-related, heat-induced rheological changes in a crystal mush. We

942   agree with Bouvet de Maisonneuve et al. (2012a, 2012b) that magma injection is an important

943   contribution in triggering the eruption. However, the injection and remobilization model alone

944   does not fully explain transitions from passive summit degassing that continuously leaks gas to

945   an explosive eruption characterized by pulsatory behavior that must be related to gas

946   accumulation (Pioli et al., 2009). Previous studies of mafic volcanoes elsewhere highlight the

947   importance of accumulated gas in producing explosive and pulsatory behavior (e.g. Andronico et

948   al., 2009; Pioli et al., 2008; Belien, 2011; Preece et al., 2016). Gas accumulation within crystal

949   mushes has been invoked to explain eruptive behavior in intermediate to silicic systems (Sisson

950   and Bacon, 1999; Pistone et al., 2015; Edmonds et al., 2016). We would have more insight into

951   this process if comparable studies of gas accumulation within mafic systems were available.

952   Additionally, magma reservoirs with accumulated gas are more compressible, and thus able to

953   accommodate more influx with limited ground deformation (Voight et al., 2010); this could

954   explain the lack of observed ground deformation prior to the 2008 activity. Therefore, we posit

955 that gas accumulation within the crystal mush, as well as other traps within the upper plumbing

956 system, and its subsequent sudden disruption and ascent is an equally important contribution to

957 eruption triggering. The proposed mechanism provides an alternative means for mafic systems to

958 generate explosive eruptions that does not rely on rapid ascent of magmas under closed system

959 degassing conditions. This is especially pertinent for understanding major explosive eruptions in

960 mafic systems like Llaima that exhibit a dominantly, open system degassing signature.

961

962 **CONCLUSIONS**

963 We conducted an extensive geochemical and textural study of the products from the 2008

964 eruption of Llaima. Our aim was to establish how the processes associated with the explosive

965 eruption initiate and occur, specifically in the context of the established background passive

966 degassing of the system. Here we present the three main conclusions from this work.

967 1. Texturally bimodal scoria were observed and analyzed. Crystal and vesicle size

968 distributions reveal quantitative texturally differences between the "brown" and "black"

969 scoria from the 2008 eruption of Llaima Volcano. The black scoria samples generally

970 have higher crystal number densities and larger vesicles, whereas the brown scoria have

971 lower crystal number densities and smaller vesicles. Overall, this suggests increasing

972 textural maturation from the brown to the black scoria. Black scoria likely formed from

973 magma that experienced prolonged passive degassing and crystallization in the shallow

974 conduit whereas the brown scoria is characterized by features resulting from rapid syn-

975 eruptive ascent, degassing and crystallization.

976 2. Our dataset provides the first evidence for an extended deep, mafic plumbing system with

977 olivine-hosted melt inclusions that sampled magmas that range from 49 to 56 wt% SiO_2

at pressures from 8-342 MPa (300 m to ~ 14 km). The majority of samples formed at pressures between 50 and 100 MPa, consistent with a shallow crystal mush. Major elements record a main LLD consistent with plagioclase and olivine fractionation in a well-connected system. Two smaller groups of melt inclusions show higher and lower TiO_2 and P_2O_5 with respect to the main LLD. Low TiO_2 and P_2O_5 samples suggest localized assimilation of more evolved lithologies. The high TiO_2 samples may have formed in isolated pockets of evolved melt; they form at high and low pressures suggestive of extensive crystallization throughout the system. H_2O and CO_2 in the majority of the melt inclusions show evidence for open system degassing, which is coupled with crystallization throughout the system.

3. Based on the combined textural and geochemical data, we propose that the main magmatic locus is a crystal mush, located between 1 and 4 km depth. Small volumes of magma from the deeper system are periodically injected, quasi steady state, into the shallow system. This contributes to maintaining passive degassing at the summit. We propose that concurrent with the passive degassing, gradual gas accumulation in the crystal mush and below the rheologically stiffer upper conduit magma occurred over time such that a critical gas volume in the system was reached, and the system overpressure was surpassed by magma injection prior to the 2008 eruption. Magma remobilization unlocked the accumulated gas, which was able to ascend rapidly and generate the observed explosive and pulsatory activity at the summit. Our proposed scenario provides an alternative model to those that posit that explosive mafic volcanism results from rapid ascent of magma due under closed system degassing conditions. This mechanism

1000   provides a means for systems with dominantly open system degassing behavior to

1001   transition from passive degassing to violent explosive eruptions.

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1544
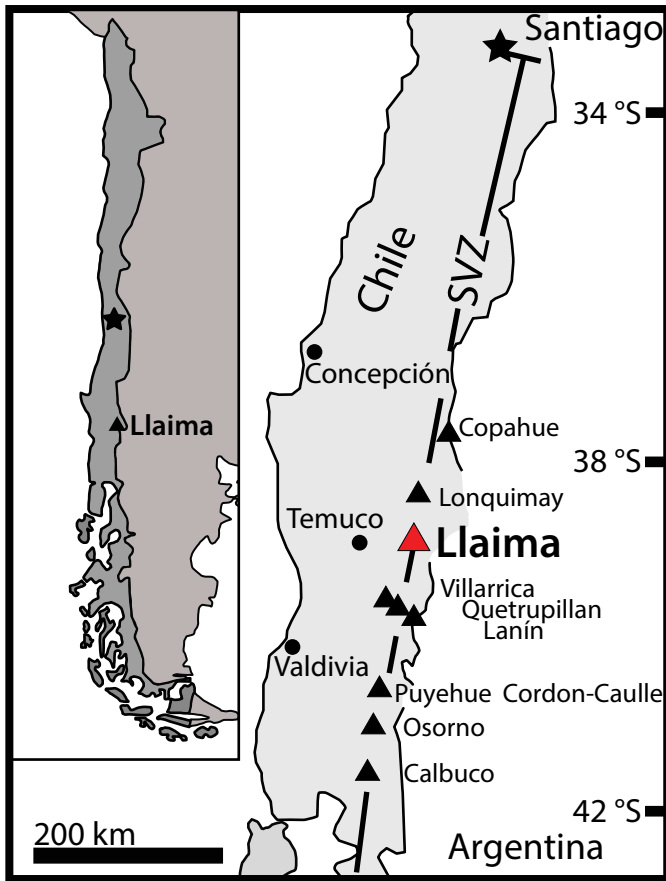
Figure 1. Location map of Llaima within the Southern Volcanic Zone of the Chilean Andes (33-46 °S). Shown for reference are the locations of select nearby volcanic systems, from north to south: Copahue, Lonquimay, Villarrica, Quetrupillan, Lanín, Puyehue Cordon-Caulle, Osorno, and Calbuco.
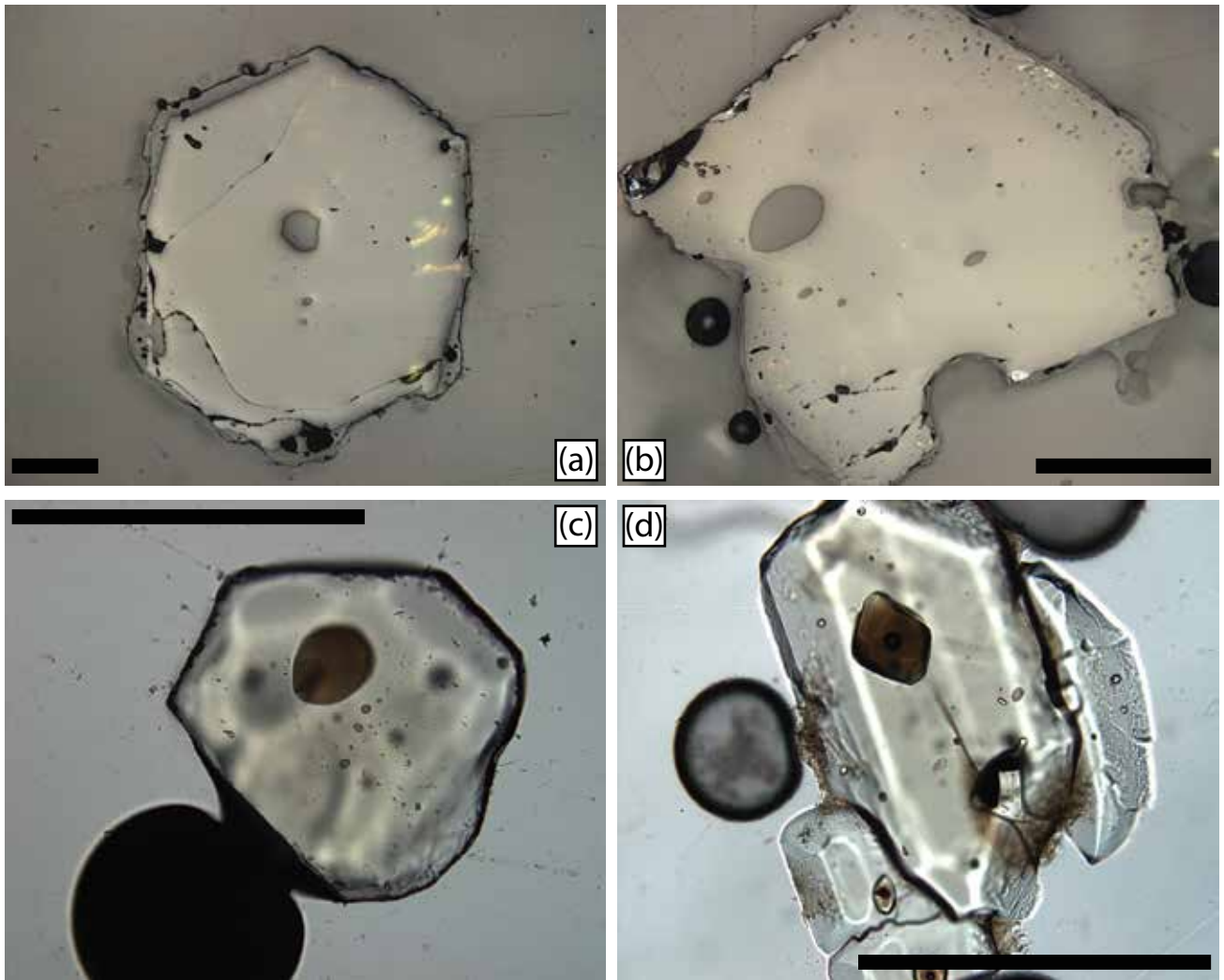
Figure 2. A range of different olivine and melt inclusion morphologies observed in the tephra samples. (A) and (b) are reflected light images whereas (c) and (d) ar e transmitted light images. All scale bars are 500 μm.
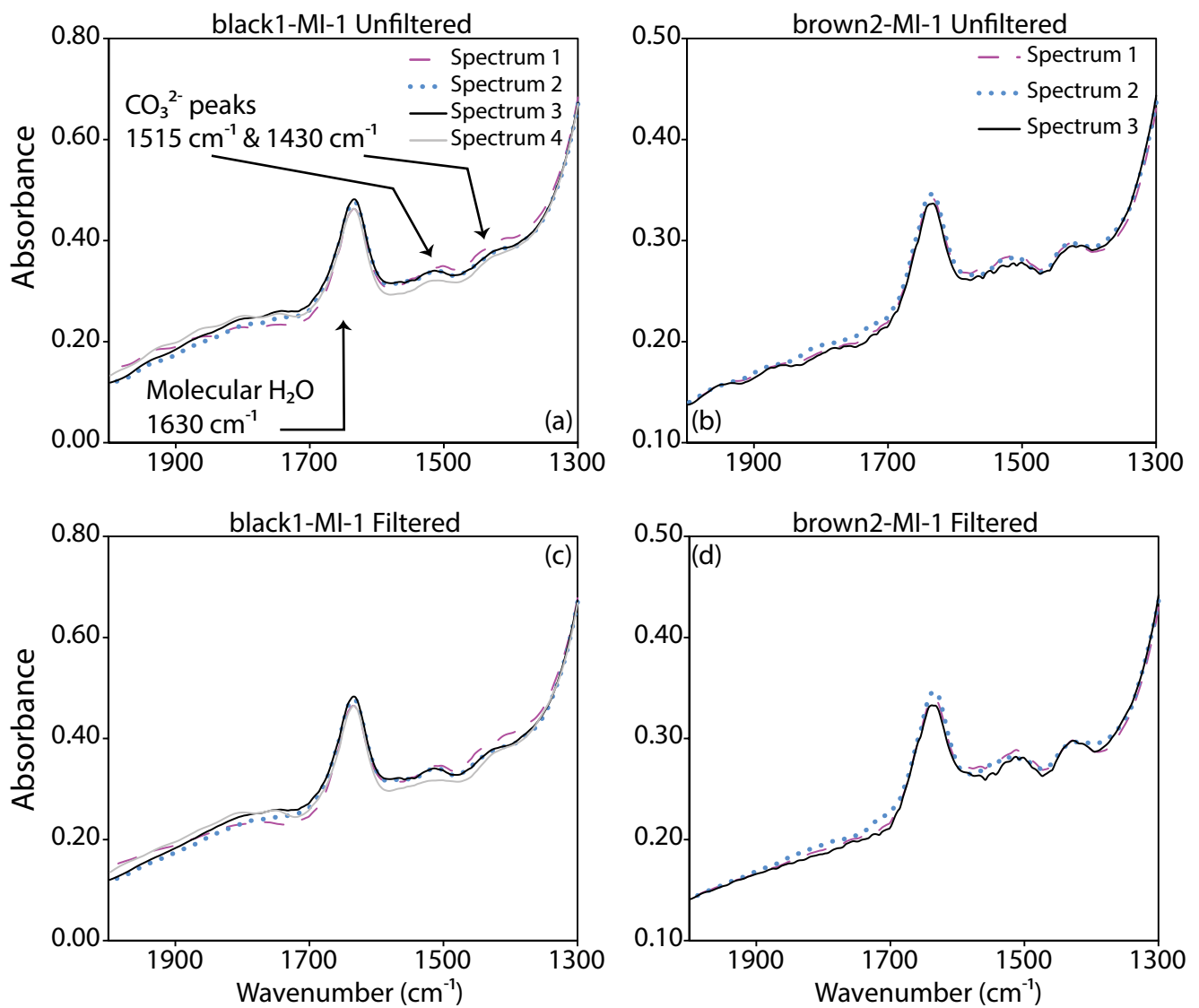
Figure 3. FTIR spectra showing the presence of fringes and the CO32- peaks. Panels (a) and (b) are unfiltered data whereas panels (c) and (d) show the fringes-filtered spectra. There was less than 5% difference between the raw and filtered spectra, indicating that the fringes did not affect the CO32- signal.
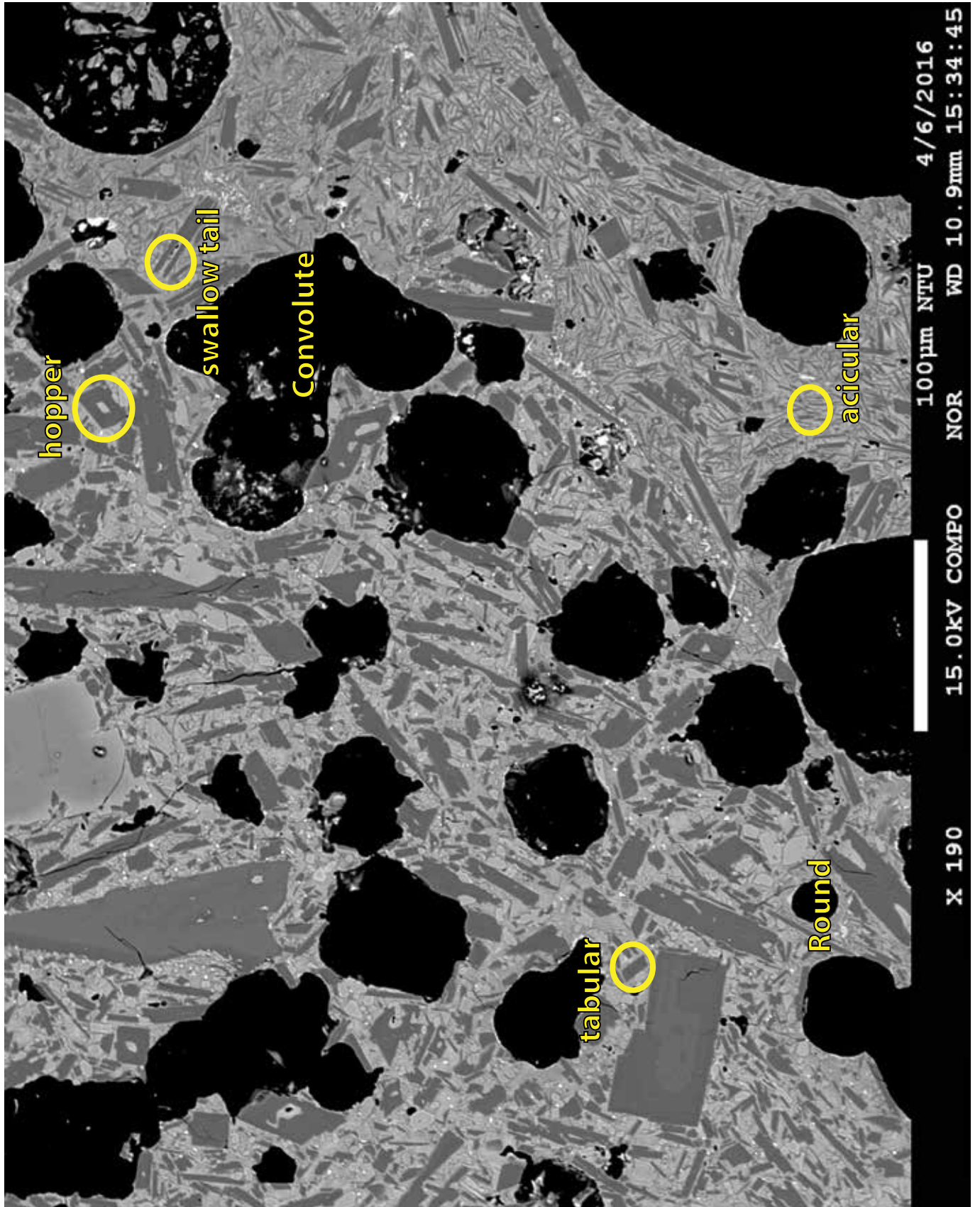
Figure 4. An example backscattered electron image of black scoria, highlighting specific plagioclase and vesicle textural features discussed in this study. With respect to plagioclase, we show hopper, swallow tail, tabular, and acicular morphologies. We highlight round and convolute vesicles.
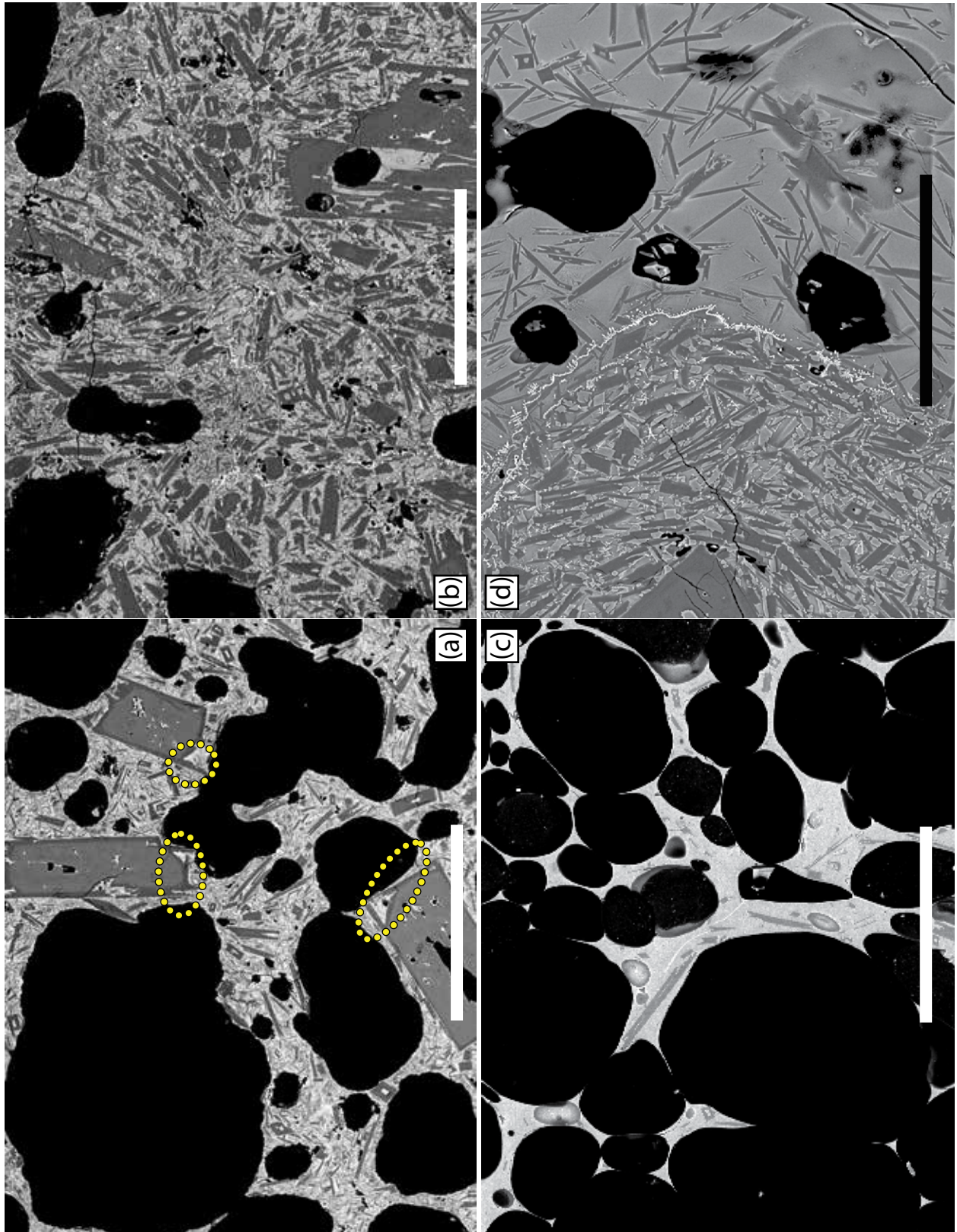
Figure 5. Backscattered electron images of the textural features of black and brown scoria. a) and b) The different microlite textures found in black scoria. In (a) there are abundant acicular plagioclase microlites (dark grey colour) alongside microphenocrysts and phenocrysts. Also note the circled plagioclase phenocrysts in showing the skeletal growths. In (b) the plagioclase microlites are dominantly tabular in morphology. c) The textural features of brown scoria, which show less abundant acicular microlites and abundant vesicles. d) The distinct boundary between two textural domains found in brown scoria. The boundary is highlighted by the spinel with cruciform morphology, suggesting rapid growth. The scale bar is 100 μm.
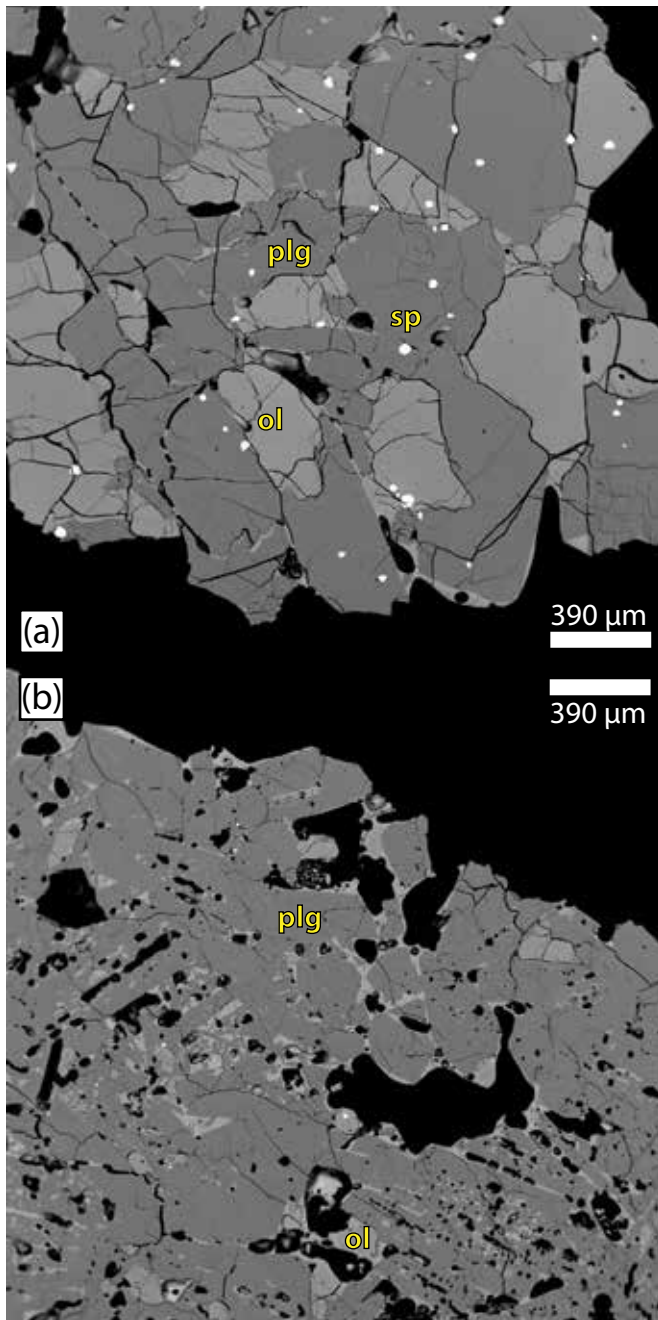
Figure 6. Backscattered electron images of plagioclase-olivine glomerocrysts. The bright phase is spinel, the medium gray phase is olivine, and the dark phase is plagioclase. a) Cr-spinel troctolite glomerocryst in equilibrium with olivine and spinel. Note there are a few vesicles in the surrounding matrix glass. b) Cr-spinel-free troctolite glomerocryst with significantly less olivine and no spinel present.
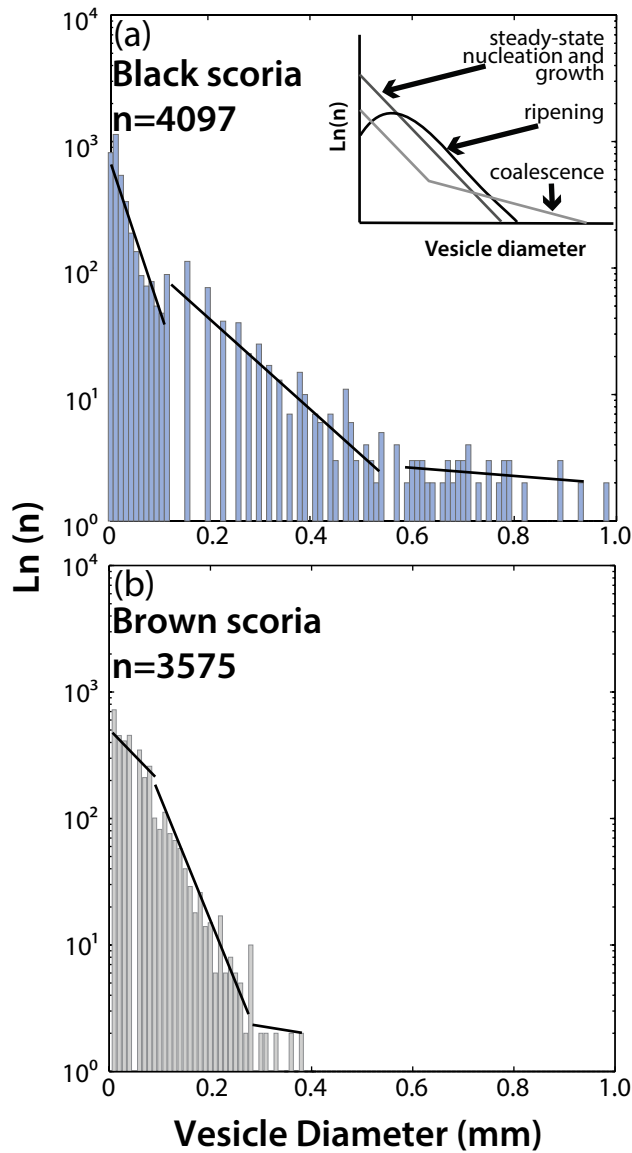
Figure 7. Vesicle size distribution histograms for a) black versus b) brown scoria. Inset figure shows interpretative models of different vesicle size histograms (after Mangan and Cashman, 1996). The lines on the histograms denote the approximate slope of each section. Note small plateau in the small vesicle sizes in the brown scoria, which are indicative of vesicle ripening (Mangan and Cashman, 1996). The black scoria exhibit two linear segments which suggests coalescence and residence.
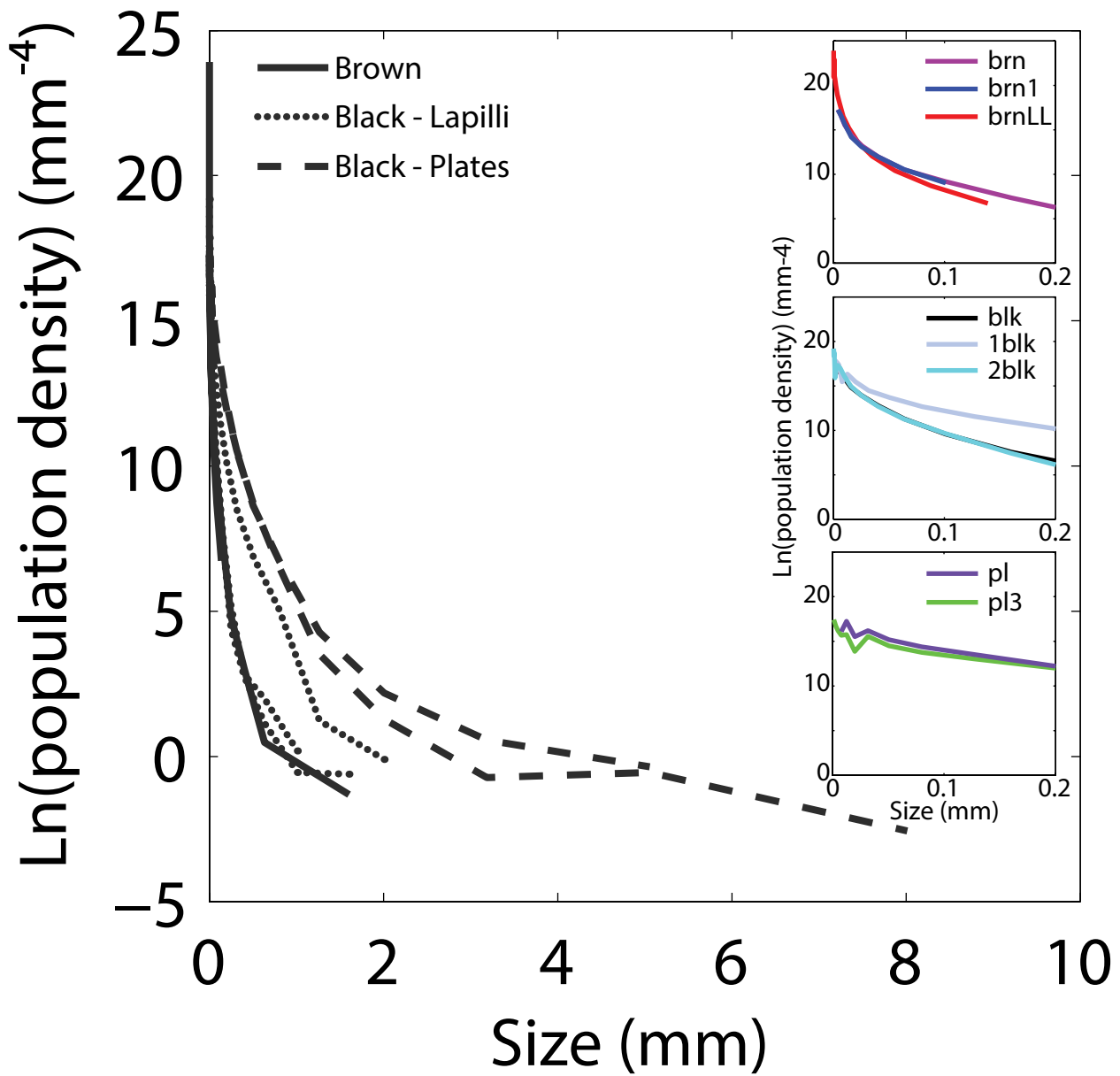
Figure 8. Crystal size distributions for the tephra erupted during the explosive phase. The insets show the breakdown by tephra type and sample. Note the decrease in slope from brown to black scoria to plates. Also note the decrease in y-intercept from brown to black scoria to plates.
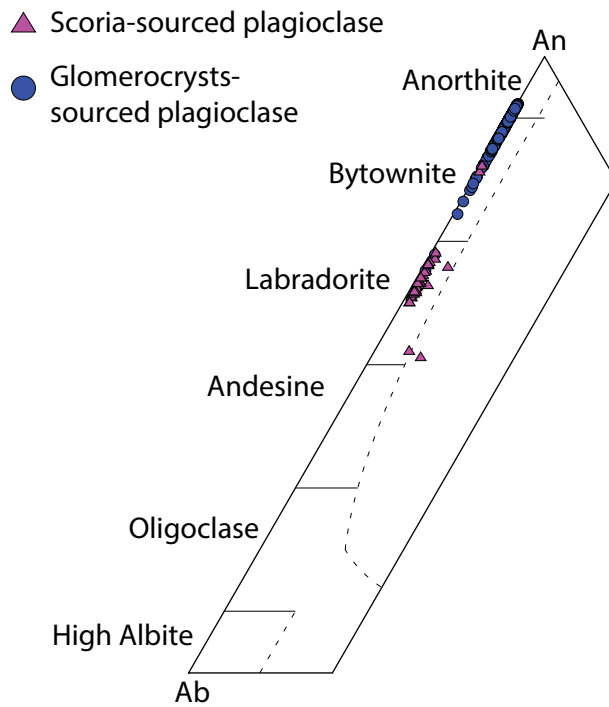
Figure 9. Modified plagioclase ternary diagram following disordered feldspar classification scheme of Deer et al. (1992). The triangles denote scoria-sourced microphenocryst plagioclase whereas the circles are plagioclases found in the glomerocrysts.
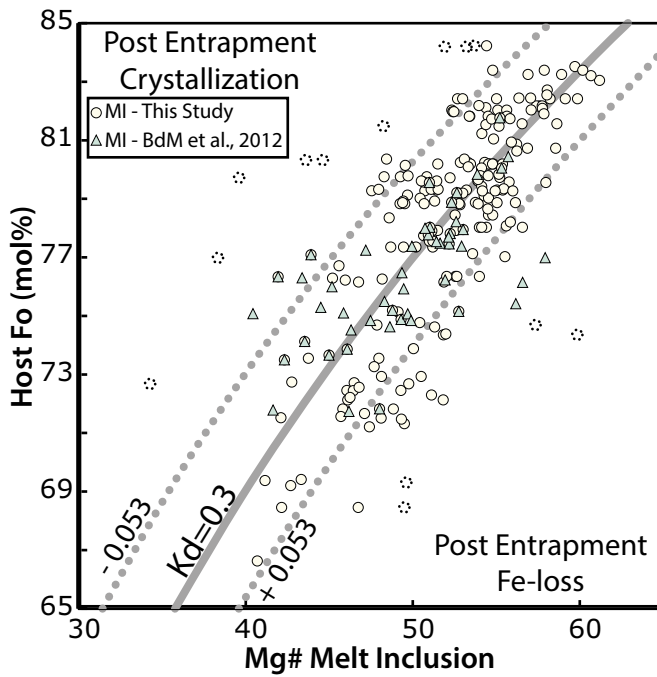
Figure 10. Rhodes diagram of Fo (mol%) of the host olivine and Mg# of the melt inclusion after Dungan et al., 1978; Rhodes et al., 1979). Samples are plotted assuming NNO, and Fe2O3 is 20% of the total Fe (Cervantes and Wallace, 2003). Those data in the top left experience post entrapment crystallization. Those data in the lower right experienced Fe-loss. The bold curve shows equilibrium which equates to Kd=0.3. The dotted lines are acceptable deviations from equilibrium (Putirka, 2008). The circles are melt inclusion analyses from this study whereas the triangles are analyses from the literature (Bouvet de Maisonneuve et al., 2012a). The dotted circles represent samples that were subsequently eliminated for further study due to likely post-entrapment modifications.
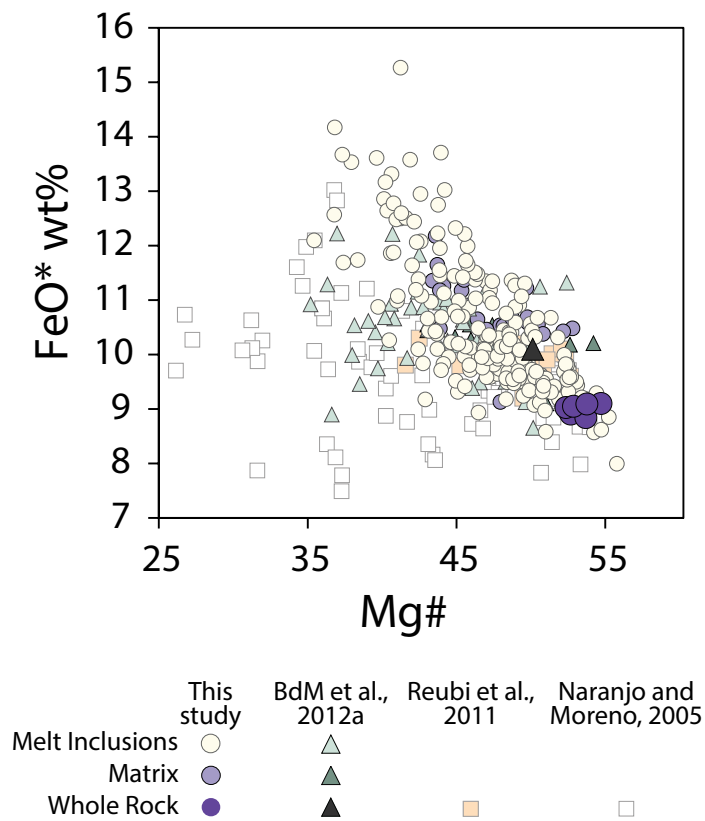
Figure 11. FeO* (total from EMPA) versus the Mg# of the melt inclusion. Also plotted are the matrix glass, and whole rock analyses from this study, Bouvet de Maisonneuve et al. (2012a) (abbreviated as BdM et al. 2012a), Reubi et al. (2011), and Naranjo and Moreno (2005). If the melt inclusions plot along the same trend for the whole rock values then Fe-loss can be considered minimal.
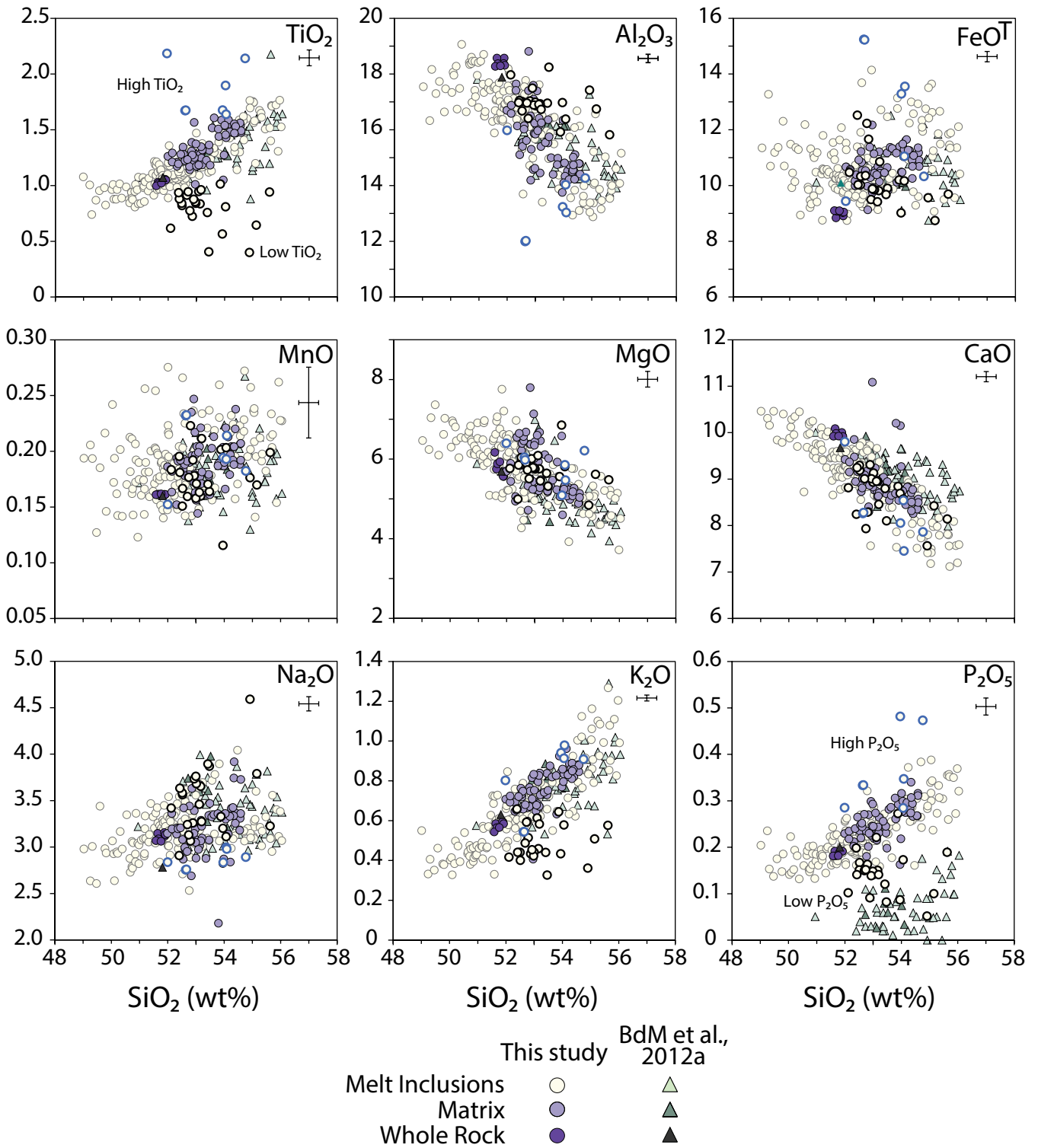
Figure 12. SiO2 diagrams showing the chemical range of the melt inclusions, matrix glass, and whole rock analyses. Samples from Bouvet de Maisonneuve et al. (2012a) are shown for comparison (triangles – same color scheme for whole rock, matrix glass, and melt inclusions). The crosses in the upper right corner represent the average standard deviation of the analyses. CaO, MgO, Al2O3 show negative correlations with SiO2, which is consistent with plagioclase and olivine crystallization. K2O, Na2O, P2O5, TiO2 show incompatible behavior. FeOT and MnO do not show any trends. All values are in wt%. Symbols with dark bold outlines highlight the low TiO2 and P2O5 samples whereas symbols with the lighter bold outlines denote samples with high TiO2 and P2O5.
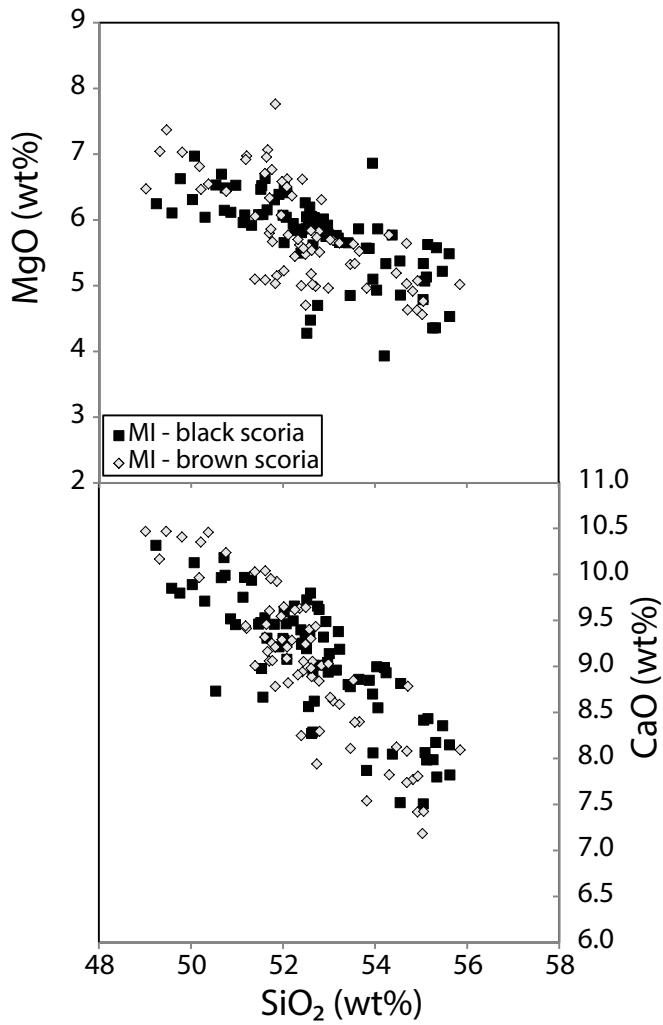
Figure 13. A comparison of melt inclusions sourced from the black and brown scoria to determine if there were distinct populations. The overlap of these data sets indicates that their respective melt inclusions are indistinguishable in terms of major elements (MgO and CaO).
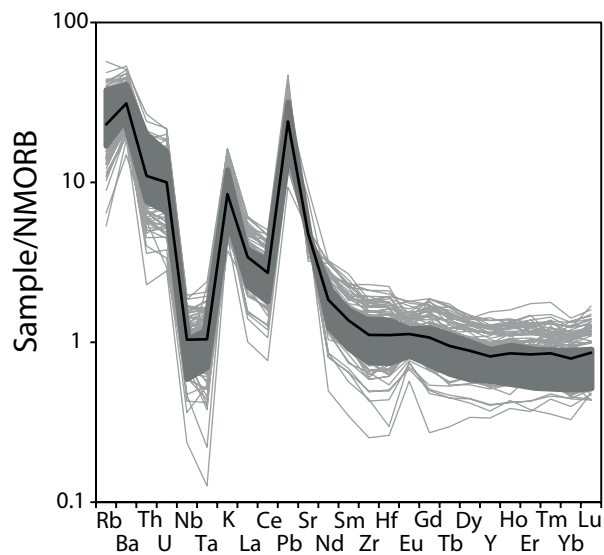
Figure 14. Multielement plot of trace elements in all analyzed melt inclusions with the average value shown in black. The dark gray field represents whole rock trace element data from Llaima Holocene tephra from Jacques et al. (2014). Note the subduction zone signature with the enriched large-ion lithophile elements, and light rare earth elements with respect to the heavy rare earth elements. Also note the range of Nb-Ta anomalies. Normalized values are from Sun and McDonough, 1989.
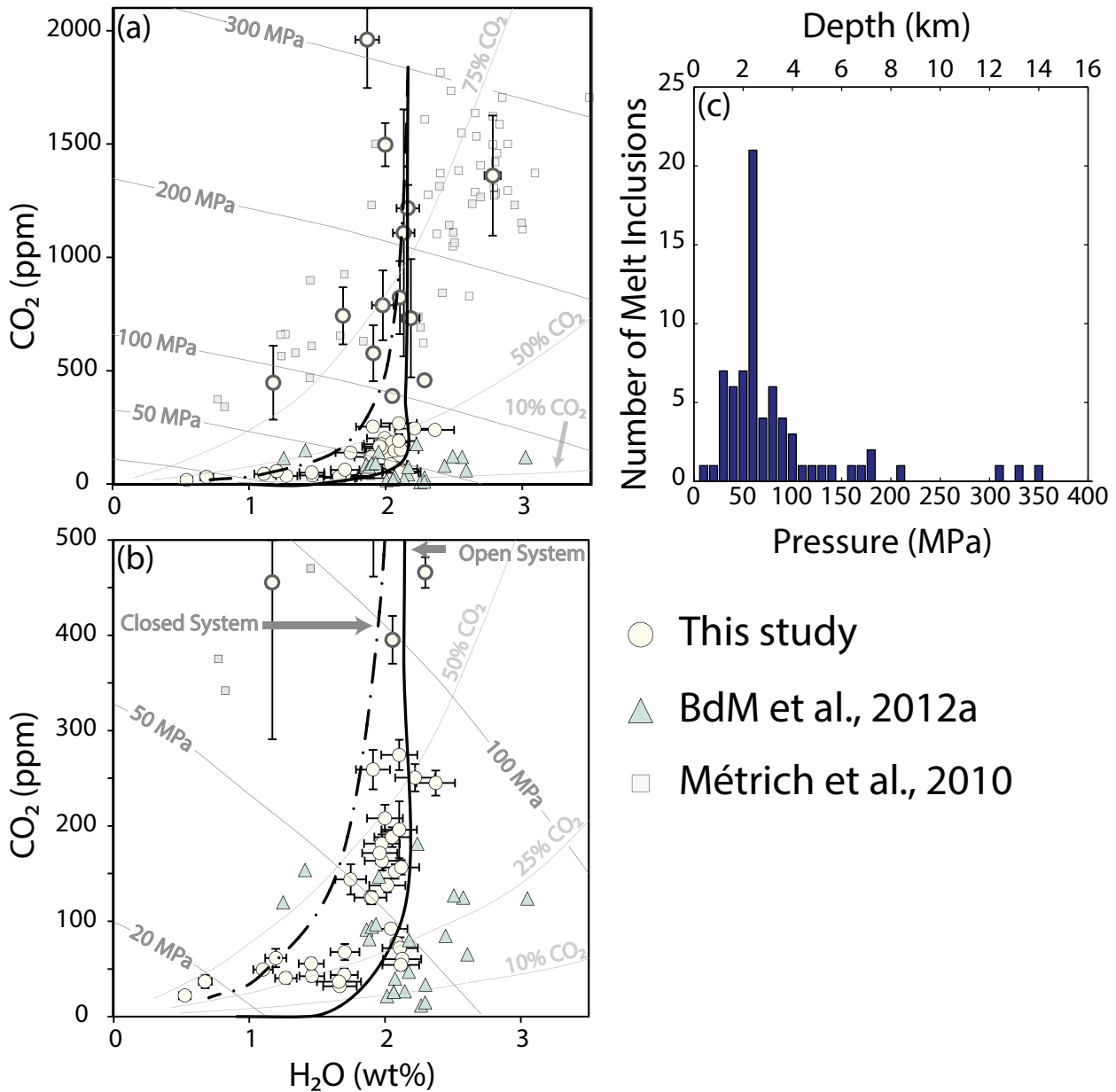
Figure 15. We used the model of Papale et al. (2006) to assess degassing style recorded by the melt inclusions. a) Plots of CO2 vs H2O from melt inclusions to show degassing behavior. Bold circles were analyzed with FTIR whereas the solid circles were analyzed with SIMS. The error bars are 2 sigma. Open (solid, black line) and closed (dash, black line) system degassing are modelled from 2000 ppm CO2, 2.15 wt% H2O. For the closed system, there is also an initial vapor phase of 1%. The dark gray lines are isobars and the light grey lines are isopleths showing the % CO2 in the vapor phase. All lines produced with Papale et al. (2006). b) Is a zoomed in version of a), note the vertical scale change. All symbols as in Figure 12. For comparison, melt inclusion volatile contents from paroxysmal eruptions at Stromboli reported in Métrich et al. (2010) are also plotted in a) and b). c) is a histogram showing the distribution of entrapment pressures for the plumbing system. Note the peak at 80 MPa.
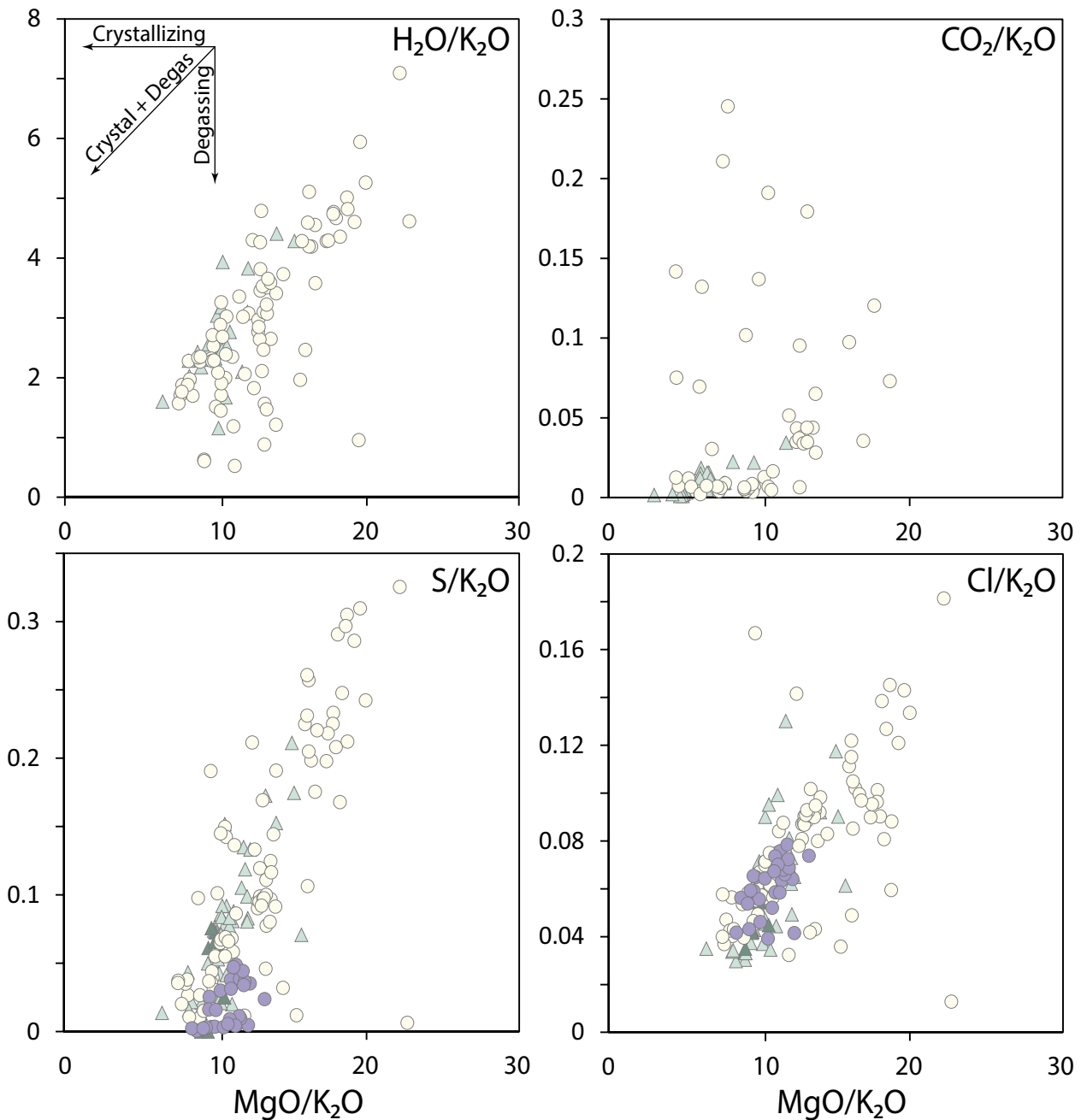
Figure 16. Volatiles ratios to K2O to show the relationship between degassing and crystallization. H2O, S, and Cl show good correlations with K2O, indicating that degassing and crystallization are coupled. CO2 shows more scatter, but at lower values (i.e. inferred shallower levels), degassing and crystallization are coupled. Matrix glass values for S and Cl provide information about degassing during the eruption. Both Cl and S in the matrix glass decrease, indicating syn-eruption degassing. Very low S values in the matrix glass suggest almost complete degassing whereas Cl did not completely degas.

Explosive Eruptive Activity

Transition Processes

Unlocking accumulated gas

Magma mixing & mush remobilization

Magma injection of deeply sourced volatile rich magma

Background Passive Degassing

Background Processes

Shallow degassing produces texturally mature, rheologically stiffer magma

Increasing gas accumulation also increases overpressure

Minor injection of deeply sourced volatile rich magma

Coupled degassing and crystalliza-tion at minimum depths of 14 km

Cr spinel free troctolite glom./crystal mush

Cr spinel troctolite glom.

Vescicles and Microlites

Relative magma and volatile flux

Depth Profile

8 MPa 300 m

25 MPa 1 km

Shallow zone

100 MPa 4 km

Deep zone

350 MPa 14 km

Figure 17. Schematic cartoon illustrating the plumbing system of Llaima and the key magmatic processes for the background passive degassing and transitions to the explosive eruption, respectively. Background passive degassing is characterized by continuous degassing from the summit, which results in the crystallization and rheological stiffening of the conduit magma. Minor volumes of magma are periodically injected into the crystal mush, which degasses and maintains the summit degassing. The deep plumbing system also experiences crystallization and degassing. Gas transport through the system is regulated mostly by the presence of the crystal mush, where vesicles get trapped within the interstitial spaces between crystals. Additional gas holdup may occur beneath the rheologically stiffer conduit magma. Gas may develop into larger bubbles, or more likely, foamy regions in the upper part of the crystal mush (the larger vesicles in the figure). The transition to the 2008 eruption at Llaima occurred after injection event that remobilized the crystal mush, and importantly, unlocked the accumulated gas that ascended rapidly to produce the observed explosive behavior at the summit.

Table 1. Summary table of crystalline and vesicle textures observed in the brown scoria and black lapilli scoria. For textural information on the black plate tephra, see Ruth and Calder (2014).

	Vesicles	Crystallinity (phenocrysts +groundmass)	Phenocrysts	Micro-phenocrysts	Microlites	Matrix glass	Other
Black scoria	79% convolute	50-60 vol%	4-7 vol% plg, ol	plg: normally zoned, pristine	plg: tabular and acicular, interlocking configuration	Limited	
Brown scoria	88%, round	10-15 vol%	1-2 vol% plg, ol	Rarely observed	plg: acicular, sometimes tabular	Abundant	Domains with black scoria textures

Table 2. Textural input and output for vesicle size distribution measurements. Average diameter = 2(area fraction/(π*vesicle number density))$^{1/2}$.

Sample	Total Area Measured (mm^2)	Phenocryst area (mm^2)	Total Area – Phenocryst Area	Number of vesicles measured	VND	Average diameter (mm)	SD (mm)
Brown	18.8	0.2	18.6	2315	124	0.055066	0.058757
brown	9.1	0.2	8.9	1260	142	0.044447	0.052181
1Black	13.7	0.6	13.0	2162	166	0.074175	0.155396
2Black	19.6	1.4	18.2	1935	106	0.074638	0.1865

Table 3. Textural inputs for crystal size distribution measurements. Crystal number density is the number of crystals per unit area.

Sample	Total Area (mm²)	Total Bubbles Area (mm²)	% Bubbles	Major axis	Intermediate axis	Minor axis	Number of crystals measured	CND
Brown	14.1374	10.41910152	73.7	2	1	1	5487	1476
Brown1	16.1369	11.8370804	73.4	2	1	1	5893	1371
BrownLL	1.5206	0.77426498	50.9	1.15	1.05	1	7485	10029
Black	6.9795	2.74364145	39.3	2	1	1	9298	2195
1Black	3.4565	1.72197592	49.8	10	1	1	8636	4979
2Black	7.511	3.7982712	50.6	2	1	1	9279	2499
Plate	2.2893	0.41159049	18	10	1	1	7917	4216
Plate3	3.7005	0.2743954	7.4	10	1	1	7332	2140

Table 4. Representative mineral analyses (olivine, plagioclase, spinel) obtained with the electron microprobe.

	BRN3-1-2010	bk2-5-1-2012	br3-1-1-2012	br4-2-2-2012	3 blk_ol_2-52	15 brn_ol_3-199	7 blk_ol_6-128	6 blk_ol_5-95
Type	Host ol	Host ol	Host ol	Host ol	Glom. ol	Glom. ol	Glom. ol	Glom. ol
SiO_2	36.41	38.74	39.67	40.21	37.67	40.59	39.24	37.67
TiO_2	nd*	0.05	0.01	0.01	nd	nd	nd	nd
Al_2O_3	nd	0.02	0.01	0.02	nd	nd	nd	nd
Cr_2O_3	0.00	0.01	0.02	0.02	0.01	0.02	0.02	0.01
FeO	30.14	21.24	18.02	14.99	19.09	17.24	16.97	16.99
MnO	0.49	0.33	0.27	0.22	0.31	0.27	0.27	0.27
MgO	33.74	39.23	42.07	44.88	40.90	40.76	42.73	46.10
NiO	0.07	0.11	0.13	0.20	0.11	0.15	0.16	0.12
CaO	0.27	0.22	0.19	0.18	0.22	0.20	0.20	0.25
Total	101.11	99.94	100.38	100.72	98.32	99.22	99.60	101.40
Si	0.98	1.00	1.01	1.00	0.99	1.03	1.00	0.95
Ti	nd	0.00	0.00	0.00	nd	nd	nd	nd
Al	nd	0.00	0.00	0.00	nd	nd	nd	nd
Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Fe^{2+}	0.68	0.46	0.38	0.31	0.42	0.37	0.36	0.36
Mn	0.01	0.01	0.01	0.00	0.01	0.01	0.01	0.01
Mg	1.35	1.51	1.59	1.67	1.60	1.55	1.62	1.73
Ni	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ca	0.01	0.01	0.01	0.00	0.01	0.01	0.01	0.01
∑Cation	3.02	3.00	2.99	3.00	3.01	2.97	3.00	3.05
%Fo	66.25	76.42	80.39	84.03	78.98	80.58	81.54	82.64

Table 4. Continued.

	13 brn_plg_3-236	15 brn_plg_5-269	Black-655	WAQ-738	WAQ-737	Black-657	Black-656	ZRN -732
Type	Glom. plag	Glom. plag	Scoria plag	Scoria plag	Scoria plag	Scoria plag	Scoria plag	Scoria plag
SiO_2	44.46	44.07	53.75	52.52	52.57	52.18	49.38	47.70
TiO_2	0.03	0.00	0.036	0.175	0.068	0.036	0.007	0.018
Al_2O_3	33.95	34.39	28.75	29.11	29.89	30.00	32.56	32.81
FeO	0.57	0.54	1.22	1.08	0.92	0.84	0.95	0.58
MgO	0.07	0.07	0.31	0.33	0.26	0.26	0.30	0.12
CaO	18.08	18.39	11.80	12.24	13.21	13.54	13.08	16.85
Na_2O	1.09	0.87	4.24	4.14	3.87	3.74	3.32	2.01
K_2O	0.02	0.01	0.17	0.16	0.12	0.07	0.11	0.02
Total	98.27	98.34	100.28	99.75	100.90	100.67	99.70	100.11
Si	2.09	2.08	2.43	2.40	2.37	2.36	2.26	2.19
Ti	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00
Al	1.88	1.91	1.53	1.56	1.59	1.60	1.76	1.78
Fe^{2+}	0.02	0.02	0.05	0.04	0.03	0.03	0.04	0.02
Mg	0.00	0.01	0.02	0.02	0.02	0.02	0.02	0.01
Ca	0.91	0.93	0.57	0.60	0.64	0.66	0.64	0.83
Na	0.10	0.08	0.37	0.37	0.34	0.33	0.29	0.18
K	0.00	0.00	0.01	0.01	0.01	0.00	0.01	0.00
∑Cation	5.02	5.02	4.99	5.00	5.00	5.00	5.01	5.01
%An	90.07	92.05	59.97	61.41	64.90	66.42	68.05	82.18

Table 4. Continued

	17 Llaima5-287	18 Llaima6-294	18 Llaima7-295	19 Llaima7-296	20 Llaima8-301	21 Llaima9-304	24 Llaima12-311
Type	oxide	oxide	oxide	oxide	oxide	oxide	oxide
SiO_2	0.08	0.10	0.11	0.20	0.16	0.05	0.11
TiO_2	0.61	1.25	1.25	0.63	1.40	1.03	1.46
Al_2O_3	31.19	27.79	27.29	32.94	20.80	27.52	21.87
Cr_2O_3	23.06	22.50	22.84	22.99	27.28	23.41	25.57
FeO	29.80	36.23	36.89	29.83	39.07	35.92	39.70
MnO	0.19	0.23	0.19	0.16	0.23	0.19	0.24
MgO	14.51	12.74	12.68	14.59	10.18	12.94	10.99
CaO	0.24	0.21	0.16	0.20	0.37	0.12	0.26
Total	99.69	101.05	101.40	101.53	99.49	101.18	100.20
Si	0.00	0.00	0.00	0.00	0.00	0.01	0.00
Ti	0.03	0.03	0.03	0.03	0.07	0.01	0.03
Al	0.94	0.86	0.79	0.74	0.53	1.11	0.77
Cr	0.55	0.60	0.62	0.73	0.66	0.52	0.68
Fe^{3+**}	0.46	0.48	0.52	0.48	0.66	0.34	0.51
Fe^{2+**}	0.45	0.48	0.50	0.50	0.60	0.37	0.52
Mn	0.01	0.01	0.01	0.01	0.01	0.00	0.01
Mg	0.55	0.53	0.51	0.50	0.46	0.62	0.48
Ni	0.00	0.00	0.00	0.00	0.00	0.00	0.00
∑Cation	2.99	3.00	2.99	2.99	3.00	2.99	2.99
Cr#	28.31	30.87	32.10	37.27	35.86	26.38	34.68

[*]n.d. indicates not determined.
[**]Fe^{2+} and Fe^{3+} determined through charge balancing.

Table 5. Oxygen fugacity calculations using equation 7 from Kress and Carmichael (1991). All calculations at the given pressure and temperature.

Sample	BLK1-1-2010	blk3-MI-1-2011	blk8-MI-1-2011	brn8-MI-1-2011
SiO_2	52.04	53.09	51.68	50.93
TiO_2	1.01	1.55	0.92	1.06
Al_2O_3	15.74	12.89	16.95	16.96
FeO*	9.38	11.01	9.81	9.11
MnO	0.20	0.19	0.16	0.15
MgO	4.16	5.35	4.81	5.44
CaO	8.76	7.48	9.02	9.02
Na_2O	3.29	3.04	3.24	3.33
K_2O	0.64	1.04	0.49	0.60
P_2O_5	0.27	0.29	0.15	0.19
Sum	95.48	95.94	97.23	96.79
T (°C)	1147	1188	1131	1152
P (MPa)	50	50	50	50
$Fe^{3+}/\Sigma Fe$	0.15	0.15	0.15	0.15
ΔNNO	-0.72	-0.63	-0.66	-0.71
$Fe^{3+}/\Sigma Fe$	0.20	0.20	0.20	0.20
ΔNNO	+0.06	+0.14	+0.12	+0.07
$Fe^{3+}/\Sigma Fe$	0.25	0.25	0.25	0.25
ΔNNO	+0.693	+0.779	+0.75	+0.71

Table 6. Representative major element and volatile concentrations in melt inclusions. These measurements are after post entrapment correction. Uncorrected data can be found in the supplemental material. Melt inclusions labeled int were located in between the core and the rim.

Sample	br1-3-3-2012	blk11-MI-5-2011	br1-3-2-2012	BRN6-3-2010*	bk3-2-1-2012	bk1-2-1-2012	blk8-MI-4-2011	blk7-MI-2-2011
MI Location	int	rim	core	core	int	int	int	int
MI Size (um)	90	41	104	18	113	101	40	82
Bubble	no	no	yes	no	no	no	yes	no
% PEC	1	4	1	4	0	4	5	3
SiO_2	49.02	49.76	50.21	50.72	50.72	50.74	50.97	51.52
TiO_2	1.08	0.99	0.97	0.95	0.94	0.89	1.08	0.92
Al_2O_3	17.52	16.62	16.71	18.35	18.58	16.96	16.77	17.50
FeOt	11.34	12.29	11.34	9.19	9.45	11.30	11.17	10.05
Fe_2O_3	1.80	1.98	1.80	1.23	1.33	1.69	1.73	1.46
FeO	9.71	10.51	9.73	8.09	8.25	9.78	9.61	8.73
MnO	0.19	0.22	0.14	0.21	0.16	0.17	0.18	0.17
MgO	6.47	6.62	6.46	6.84	6.14	6.48	6.53	6.46
CaO	10.47	9.80	10.35	10.21	10.18	9.99	9.45	9.47
Na_2O	2.97	2.98	2.93	2.78	3.06	2.77	3.02	3.20
K_2O	0.55	0.38	0.50	0.42	0.48	0.38	0.48	0.40
P_2O_5	0.20	0.15	0.19	0.20	0.15	0.15	0.19	0.16
Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
mg#	0.50	0.49	0.50	0.57	0.54	0.51	0.51	0.53
Host Fo	80	79	80	83	82	80	80	81
S (ppm)	n.d.	n.d.	n.d.	n.d.	n.d.	927	1244	1153
Cl (ppm)	n.d.	n.d.	n.d.	n.d.	n.d.	511	549	488
F (ppm)	n.d.	n.d.	n.d.	n.d.	n.d.	305	0	0
H_2O (wt%)	n.d.	0.37	n.d.	n.d.	n.d.	2.01	n.d.	2.05
H_2O error	n.d.	0.14	n.d.	n.d.	n.d.	0.13	n.d.	0.02
CO_2 (ppm)	n.d.	n.d.	n.d.	n.d.	n.d.	138	n.d.	394
CO_2 error	n.d.	n.d.	n.d.	n.d.	n.d.	7	n.d.	25
Temp (°C)	1218	1180	1212	1179	1206	1173	1158	1185
Pressure (MPa)	n.d.	n.d.	n.d.	n.d.	n.d.	49	n.d.	81
Depth (km)	n.d.	n.d.	n.d.	n.d.	n.d.	2.0	n.d.	3.3

n.d. indicates not determined

Table 6. Continued.

Sample	blk9-MI-1-2011	BUL4-3-2010*	BUL4-1-2010*	br2-1-1-2012	brn5-MI-1-2011	bk2-8-1-2012	bk4-2-2-2012	bk4-1-1-2012
MI Location	rim	int	int	int	core	core	rim	core
MI Size (um)	34	16	20	30	50	158	65	184
Bubble	no	yes	yes	no	yes	no	no	no
% PEC	4	1	0	0	0	1	2	0
SiO_2	51.54	51.55	52.01	52.02	52.20	52.22	52.25	52.65
TiO_2	1.22	1.02	1.11	1.18	1.10	1.10	1.04	1.68
Al_2O_3	15.86	17.54	17.01	17.55	17.59	17.09	16.96	11.99
FeOt	11.53	10.34	10.66	9.92	10.14	9.71	9.94	15.22
Fe_2O_3	1.90	1.57	1.65	2.21	2.26	1.50	1.52	3.14
FeO	9.82	8.93	9.18	7.94	8.11	8.36	8.57	12.39
MnO	0.20	0.25	0.22	0.20	0.19	0.16	0.17	0.23
MgO	6.52	5.74	6.08	5.22	5.25	5.94	5.84	5.99
CaO	8.98	9.59	8.82	9.65	9.51	9.49	9.65	8.29
Na_2O	3.04	3.04	3.01	3.54	3.24	3.34	3.21	2.76
K_2O	0.73	0.55	0.66	0.56	0.59	0.61	0.61	0.55
P_2O_5	0.19	0.21	0.26	0.17	0.19	0.19	0.19	0.33
Total	100.00	100.00	100.00	100.22	100.23	100.00	100.00	100.00
mg#	0.50	0.50	0.50	0.48	0.48	0.52	0.51	0.41
Host Fo	80	79	79	76	77	81	80	68
S (ppm)	400	n.d.	n.d.	512	454	553	577	175
Cl (ppm)	545	n.d.	n.d.	515	597	495	532	454
F (ppm)	0	n.d.	n.d.	300	n.d.	311	324	289
H_2O (wt%)	n.d.	n.d.	2.00	0.68	n.d.	1.70	1.61	2.04
H_2O error	n.d.	n.d.	0.04	0.05	n.d.	0.10	0.10	0.12
CO_2 (ppm)	n.d.	n.d.	n.d.	37	n.d.	44	25	92
CO_2 error	n.d.	n.d.	n.d.	7	n.d.	7	2	5
Temp (°C)	1171	1193	1219	1197	1192	1196	1176	1190
Pressure (MPa)	n.d.	n.d.	n.d.	13	n.d.	31	26	73
Depth (km)	n.d.	n.d.	n.d.	0.5	n.d.	1.3	1.1	3.0

Table 6. Continued.

Sample	bk4-4-1-2012	brn6-MI-1-2011	bk3-4-3-2012	Plate1-2010*	blk9-MI-2-2011	brn4-MI-1-2011	blk5-MI-1-2011	bk2-6-1-2012
MI Location	int	core	int	core	int	int	core	int
MI Size (um)	90	93	105	33	38	106	157	102
Bubble	no	no	yes	no	yes	no	no	yes
% PEC	0	0	2	1	3	0	5	0
SiO_2	52.88	52.90	52.93	54.06	54.06	54.08	54.21	54.24
TiO_2	0.73	1.23	0.88	0.82	1.90	1.65	1.43	1.52
Al_2O_3	17.47	14.95	16.87	16.36	14.00	13.01	15.71	13.53
FeOt	9.48	14.13	9.86	10.15	11.04	13.54	10.92	12.23
Fe_2O_3	1.36	2.82	1.54	1.53	1.99	2.71	2.04	2.72
FeO	8.26	11.60	8.48	8.78	9.25	11.11	9.08	9.78
MnO	0.19	0.24	0.18	0.20	0.19	0.21	0.22	0.23
MgO	6.01	4.62	5.79	5.81	5.86	5.48	3.93	5.33
CaO	9.32	7.57	9.01	8.58	8.55	7.46	8.99	8.93
Na_2O	3.25	3.08	3.73	3.11	2.99	2.97	3.36	2.98
K_2O	0.44	0.73	0.46	0.58	0.92	0.98	0.72	0.64
P_2O_5	0.09	0.27	0.14	0.17	0.28	0.35	0.30	0.37
Total	100.00	100.00	100.00	100.00	100.00	100.00	99.99	100.28
mg#	0.53	0.37	0.51	0.51	0.49	0.42	0.39	0.44
Host Fo	81	68	80	80	79	71	72	72
S (ppm)	731	482	995	n.d.	895	104	618	n.d.
Cl (ppm)	352	412	435	n.d.	491	423	512	n.d.
F (ppm)	218	n.d.	257	n.d.	n.d.	n.d.	n.d.	n.d.
H_2O (wt%)	1.90	n.d.	1.96	1.98	n.d.	1.93	n.d.	n.d.
H_2O error	n.d.	n.d.	0.13	0.08	n.d.	0.10	n.d.	n.d.
CO_2 (ppm)	125	n.d.	171	796	n.d.	n.d.	n.d.	n.d.
CO_2 error	7	n.d.	11	155	n.d.	n.d.	n.d.	n.d.
Temp (°C)	1215	1186	1178	1197	1176	1209	1224	1213
Pressure (MPa)	45	n.d.	55	151	n.d.	n.d.	n.d.	n.d.
Depth (km)	1.8	n.d.	2.2	6.1	n.d.	n.d.	n.d.	n.d.

Table 6. Continued.

Sample	br1-2-1-2012	BLK8-1-2010*	br4-3-3-2012	br1-5-1-2012	brn14-MI-1-2011	brn12-MI-1-2011*	blk1-MI-1-2011	bk3-2-2-2012	BUL1-2010*
MI Location	core	int	rim	core	core	int	core	int	rim
MI Size (um)	131	60	111	42	92	123	236	57	60
Bubble	no	no	no	no	no	no	no	no	no
% PEC	0	5	1	0	0	0	0	2	0
SiO_2	54.31	54.38	54.47	54.92	54.93	55.03	55.05	55.15	56.04
TiO_2	1.55	1.35	1.15	1.77	1.70	1.76	1.75	0.65	1.53
Al_2O_3	12.91	16.05	16.65	13.02	13.36	12.85	13.34	16.72	15.13
FeOt	12.98	9.76	9.27	13.49	12.41	13.63	12.61	8.73	10.40
Fe_2O_3	2.50	1.54	1.53	2.85	2.42	2.93	2.56	1.30	1.81
FeO	10.73	8.38	7.90	10.93	10.23	10.99	10.31	7.55	8.78
MnO	0.22	0.16	0.17	0.20	0.26	0.24	0.24	0.17	0.23
MgO	5.77	5.77	5.19	4.63	5.07	4.56	4.79	5.62	4.51
CaO	7.82	8.05	8.12	7.42	7.81	7.18	7.50	8.43	7.60
Na_2O	3.08	3.12	4.04	3.14	3.09	3.11	3.09	3.78	3.06
K_2O	0.84	0.93	0.62	0.84	0.84	1.01	1.06	0.51	0.99
P_2O_5	0.26	0.29	0.16	0.28	0.29	0.34	0.32	0.10	0.32
Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
mg#	0.44	0.51	0.50	0.38	0.42	0.37	0.40	0.53	0.44
Host Fo	73	80	80	69	73	69	72	82	75
S (ppm)	299	n.d.	n.d.	330	258	268	215	n.d.	n.d.
Cl (ppm)	462	n.d.	n.d.	515	412	406	500	n.d.	n.d.
F (ppm)	478	n.d.	n.d.	606	n.d.	n.d.	n.d.	n.d.	n.d.
H_2O (wt%)	1.91	2.17	n.d.	2.12	1.91	2.30	2.00	n.d.	1.69
H_2O error	0.12	0.08	n.d.	0.14	0.04	0.04	0.04	n.d.	0.04
CO_2 (ppm)	259	1226	n.d.	61	584	n.d.	1508	n.d.	750
CO_2 error	21	103	n.d.	9	124	n.d.	96	n.d.	126
Temp (℃)	1225	1128	1182	1202	1201	1200	1196	1182	1175
Pressure (MPa)	108	205	n.d.	57	174	n.d.	342	n.d.	165
Depth (km)	4.4	8.4	n.d.	2.3	7.1	n.d.	13.9	n.d.	6.7

Table 7. Analytical results (range, average, and standard deviation) for melt inclusion volatiles measured with FTIR, SIMS, and EPMA.

	H_2O (wt%)	CO_2 (ppm)	S (ppm)	Cl (ppm)	F (ppm)
FTIR (H_2O, n=54; CO_2, n=13)	0.36-2.94 (1.8±0.61)	394-1973 (939±473)			
SIMS (n=32)	0.53-2.37 (1.76±0.44)	22-274 (110±77)	92-1277 (639±290)	236-641 (478±86)	155-606 (322±89)
EPMA (n=47)			23-1534 (699±415)	47-1342 (490±183)	

Standard deviations are 2σ. *N* equals number of analyses.

Table 8. Inputs and viscosity modeling results based on different crystallinity in the scoria. Model inputs include the average whole rock chemistry with all major elements in wt%, assumed water content of 1 wt%, and the temperature from the plagioclase melt thermometer. The reported values are for both the Shaw model and the Bottinga and Weill model.

Model Inputs												
SiO_2	TiO_2	Al_2O_3	Fe_2O_3	FeO	MnO	MgO	CaO	Na_2O	K_2O	P_2O_5	H_2O	T °C
51.35	1.02	18.25	2.76	8.94	0.16	5.75	9.89	3.09	0.57	0.19	1.0	1150

	8 MPa		15 MPa		30 MPa	
	Shaw	B&W	Shaw	B&W	Shaw	B&W
Black Scoria – 50% crystallinity	5.64×10^4	2.80×10^4	2.29×10^4	2.35×10^4	2.29×10^4	2.35×10^4
Brown Scoria – 10% crystallinity	1.33×10^2	1.59×10^2	4.45×10^1	5.40×10^1	4.45×10^1	5.40×10^1

Model inputs include the average whole rock chemistry with all major elements in wt%, assumed water content of 1 wt%, and the temperature from the plagioclase melt thermometer. The reported values are for both the Shaw (1972) model and the Bottinga and Weill (1972) model.