

University of Nebraska - Lincoln

DigitalCommons@University of Nebraska - Lincoln

Chemical and Biomolecular Engineering -- All
Faculty Papers

Chemical and Biomolecular Engineering,
Department of

2019

Comparative kinetic modeling of growth and molecular hydrogen overproduction by engineered strains of *Thermotoga maritima*

Raghuveer Singh

University of Nebraska-Lincoln, raghuveer@huskers.unl.edu

Rahul Tevatia

University of Nebraska-Lincoln, rahultevatia_83@yahoo.co.in

Derrick White

University of Nebraska-Lincoln


Yasar Demirel

University of Nebraska-Lincoln, ydemirel2@unl.edu

Paul H. Blum

University of Nebraska - Lincoln, pblum1@unl.edu

Follow this and additional works at: <https://digitalcommons.unl.edu/chemengall>

 Part of the [Biomedical Engineering and Bioengineering Commons](#), and the [Catalysis and Reaction Engineering Commons](#)

Singh, Raghuveer; Tevatia, Rahul; White, Derrick; Demirel, Yasar; and Blum, Paul H., "Comparative kinetic modeling of growth and molecular hydrogen overproduction by engineered strains of *Thermotoga maritima*" (2019). *Chemical and Biomolecular Engineering -- All Faculty Papers*. 88.
<https://digitalcommons.unl.edu/chemengall/88>

This Article is brought to you for free and open access by the Chemical and Biomolecular Engineering, Department of at DigitalCommons@University of Nebraska - Lincoln. It has been accepted for inclusion in Chemical and Biomolecular Engineering -- All Faculty Papers by an authorized administrator of DigitalCommons@University of Nebraska - Lincoln.

Comparative kinetic modeling of growth and molecular hydrogen overproduction by engineered strains of *Thermotoga maritima*

Raghuveer Singh,¹ Rahul Tevatia,² Derrick White,¹
Yaşar Demirel,² Paul Blum¹

1 Beadle Center for Genetics, University of Nebraska-Lincoln, 68588, USA

2 Department of Chemical and Biomolecular Engineering,
University of Nebraska-Lincoln, 68588, USA

Corresponding authors — R. Singh & P. Blum; E234 Beadle Center for Genetics,
School of Biological Sciences, University of Nebraska-Lincoln, Lincoln NE 68588- 0666, USA;
email raghuveer@huskers.unl.edu ; pblum1@unl.edu

Abstract

Thermotoga maritima is an anaerobic hyperthermophilic bacterium known for its high amounts of hydrogen (H₂) production. In the current study, the kinetic modeling was applied on the engineered strains of *T. maritima* that surpassed the natural H₂ production limit. The study generated a kinetic model explaining H₂ overproduction and predicted a continuous fermentation system. A Leudking-Piret equation-based model predicted that H₂ production by Tma200 (0.217 mol-H₂ g⁻¹-biomass) and Tma100 (0.147 mol-H₂ g⁻¹-biomass) were higher than wild type (0.096 mol-H₂ g⁻¹ -biomass) with reduced rates of maltose utilization. Sensitivity analysis confirmed satisfactory fitting of the experimental data. The slow growth rates of Tma200 (0.550 h⁻¹) and Tma100 (0.495 h⁻¹) are compared with the wild type (0.663 h⁻¹). A higher

Published in *International Journal of Hydrogen Energy* 44 (2019), pp 7125–7136

doi 10.1016/j.ijhydene.2019.01.124

Copyright © 2019 Hydrogen Energy Publications LLC; published by Elsevier Ltd. Used by permission.

Submitted 5 September 2018; revised 7 January 2019; accepted 10 January 2019; published 11 February 2019.

maintenance energy along with growth and non-growth H_2 coefficients corroborate the higher H_2 productivity of the engineered strains. The modeled data established a continuous fermentation system for the sustainable H_2 production.

Keywords: Bio hydrogen beyond Thauer limit, Biofuel, Dark fermentation, Kinetic modeling, Continuous stirred tank reactor, Natural gas phase out

Introduction

The need for energy is increasing across the world and this need is anticipated to increase for the next twenty years. The U.S. Energy Information Administration (EIA) based on the "International Energy Outlook 2013 and 2018 project" indicated that the world's energy consumption will grow by 56% between 2010 and 2040, from 524 quadrillion Btu to 820 quadrillion Btu accompanied by a sudden surge in electricity use after 2020. Among various sources of energy, molecular hydrogen (H_2) offers a promising clean fuel because of its higher energy content per unit weight (142 kJ g^{-1} or $61,000 \text{ Btu lb}^{-1}$) relative to alternatives [1]. Hydrogen is an energy carrier and energy from hydrogen can be released either by direct combustion which generates water or by conversion to electricity via a fuel cell to power electric vehicles [2–4]. However, the majority of H_2 is derived from nonrenewable sources such as natural gas and coal that threaten environmental quality because of the associated production of CO_2 , a green-house gas [5]. In response to this issue, the German *Energiewende* projects to supplant fossil fuel with renewable energy by 2050 [6]. Additionally, several countries have proposed to abandon the use of vehicles powered by fossil fuels [7] and the US Department of Energy (DOE) plans to phase out the use of natural gas by 2050 [8]. Renewable and affordable H_2 production may help mitigate the adverse effects of fossil fuels. When fossil fuels are phased out, continuation of the established hydrogen market would demand a technology for renewable hydrogen production technology.

Biological methods for H_2 production could provide a renewable energy source though low productivity has remained an obstacle for sustainable production. The stoichiometric conversion of organic acids to H_2 during photofermentation by photoautotrophic microbes employ light as an energy source for H_2 production, however the key enzyme, nitrogenase, produces H_2 only in response to nitrogen deprivation, an extreme physiologic stress. Therefore, this bioprocess has

limitations preventing its application at the industrial scale [9]. A sustainable and biologically stress-free method for renewable H₂ production uses microbial fermentation. This is an anaerobic process operating without the need for oxygen or light sometimes called dark fermentation that yields H₂ as a co-product from the partial oxidation of sugar [10]. Microbial H₂ production through dark fermentation is energetically favored as it does not require external energy [11,12] and is compatible with the growing availability of renewable sugar sources obtained from lignocellulosic biomass feedstocks. Among all H₂ producing bacteria, hyperthermophiles are the most promising biological choice [1,13,14] because high cultivation temperature favors the energetics of H₂ production, promotes solubilization and hydrolysis of complex substrates to fermentable sugars, and inhibits growth of H₂ consuming contaminating microorganisms naturally resident in feedstocks. However, in closed batch culture systems accumulation of H₂ inhibits cell growth [15]. This can be alleviated by continuous stripping of H₂ using molecular nitrogen (N₂) influx or the use of an in-line vacuum of the fermentation head-space resulting in improved H₂ production and better growth [16,17]. Therefore, bench-scale bioreactor studies have become crucial since they can be used to address end-product growth inhibition and to enable prediction of process parameters required for scale up [18].

Thermotoga maritima is a hyperthermophilic anaerobic bacterium that grows optimally at 80 °C and excretes H₂ to levels that approach the biological/physiological limit of hydrogen production [14,19,20] using simple and complex carbohydrates [21]. Natural strains of *T. maritima* employs the pathways of Embden–Meyerhof–Parnas (85% relative contribution) and the Entner–Doudorof (15% relative contribution) [13,14,22] to produce H₂, CO₂ and organic acids (lactate, acetate) from sugar oxidation (**Fig. S2**). Four moles of H₂ are produced per mole of glucose by a bifurcating hydrogenase that accepts electrons and protons simultaneously from reduced NADH and ferredoxin [23]. Organic acids including acetate and lactate are also produced in proportional amounts. Lactic acid production is dependent on culturing conditions as batch cultivation has been reported to shift the metabolism towards lactic acid production under a high partial pressure of H₂ [24] but not in its absence [25].

Kinetic modeling using various approaches [10] has been applied to biological H₂ formation to optimize H₂ production and to achieve

process scale up [26]. However, the Thauer limit (Thauer et al., 1977), the thermodynamic limit of 4mol of H_2 per mole of hexose imposed by cell-based energetic constraints, is the major obstacle limiting economic industrial production of H_2 through dark fermentation [27]. Overcoming this limit by improving the conversion of sugar to H_2 could lead to a superior H_2 production system and a new technology that may supersede fossil fuel-based H_2 production. Previously, genetically engineered cell lines of *T. maritima* were developed using a method called transient gene inactivation that were called Tma100 and Tma200 [25,28]. These cell lines were found to over-produce H_2 relative to the natural (wild type) lineage and thereby surpassed the H_2 productivity previously proposed for a fermentative bacterium. While applications of these strains are of some interest, kinetic parameters must be obtained first by utilizing modeling and bioreactor-based studies.

Here kinetic modeling was performed to predict the maximum H_2 molar yields and growth of the evolved *T. maritima* cell lines using maltose as a fermentable substrate. The kinetic modeling can be employed for modeling hydrogen production from engineered strains overproducing hydrogen and may provide a basis for large scale and continuous fermentation processes.

Materials and methods

Strains and bioreactor set up

Thermotoga maritima MSB8 (wild type, type, strain) and excess H_2 producing strains (Tma100 and Tma200) [25], used in this study, were cultivated in 3L double-jacketed glass bioreactors (Applikon, MA) at 80 °C containing 1.5 L complex medium [29]. Bioreactor-based cultivation was employed to overcome growth inhibition caused by H_2 accumulation through headspace exchange. Prior to inoculation, the cultivation medium was reduced by addition of 0.1% (wv^{-1}) Na_2S followed by supplementation with maltose at a final concentration of 15 mM (mM). As shown in **Fig. 1**, anaerobic conditions in the bioreactor were maintained by continuous sparging of N_2 at 15 mL min^{-1} . The medium was stirred at 200 rpm using two axial impellers. Temperature, pH and dissolved oxygen were monitored using immersed sensors and a pH of 7.0 was maintained by metered addition of HCl

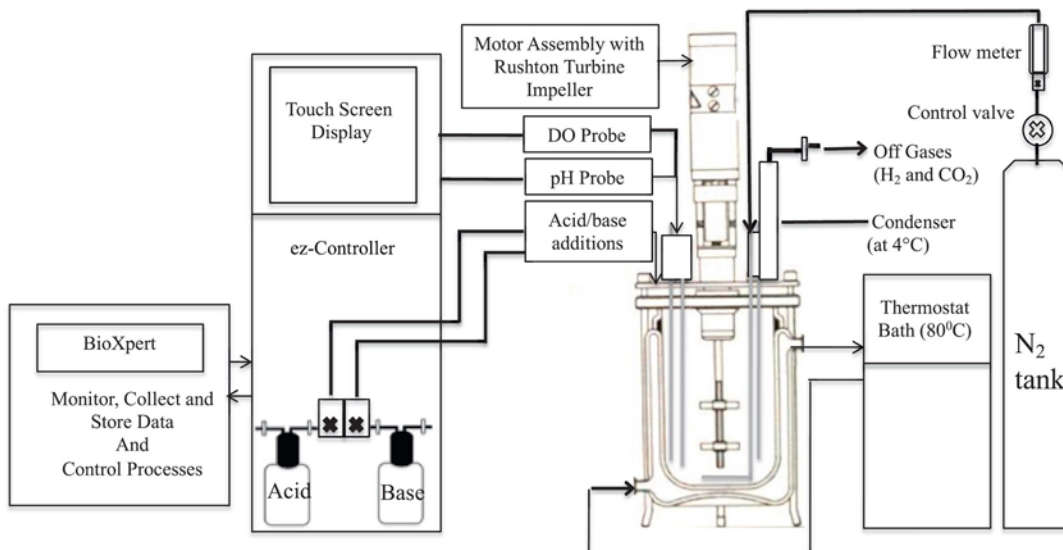


Fig. 1. Schematic representation of a 3L anaerobic fermenter set-up for wild type, Tma100 and Tma200 with associated control units and data acquisition system.

or NaOH as needed using peristaltic pumps. To minimize water loss, the water vapor present in the outgassing headspace was returned to the vessel by condensation using a chilled water supply. Samples were removed periodically using an external syringe to determine culture cell density determined as the optical density (OD_{600}), and subsequent organic acid and residual maltose analysis. For H_2 sampling, a gas tight syringe (Hamilton) was used to withdraw samples from a rubber septum located on the head plate of the bioreactor.

Analysis of metabolites

Analysis of headspace gas was performed by injecting 500 mL volumes into a gas chromatograph (GC 400 Series, GOWMAC, PA) fitted with a thermal conductivity detector. N_2 gas was used as a carrier and separation of the sample gas was carried out at 70 °C using a molecular sieve column (GOWMAC). For headspace CO_2 analysis, a Varian (430) GC equipped with an Alltech Porapak C-5000 column was used. Helium was used as the carrier gas, and column temperature was maintained at 65 °C to separate CO_2 . Calibration curves were obtained by injecting various volumes of pure H_2 and CO_2 , and the amount of H_2 and CO_2 in the headspace was estimated by comparison to these

values. The ideal gas law was used to calculate the amount of H₂ and CO₂ that was produced at standard temperature and pressure (STP). Organic acids and maltose concentrations were determined in culture supernatants by HPLC with comparison to standards. Prior to injection, samples were clarified by centrifugation at 10,000×g rpm for 10 min and then filtered (AcroDisc, 0.45 μM). Samples (1 μL) were analyzed using an Agilent 1200 HPLC system and an automated sampler equipped with a Refractive Index Detector and a Hi-Plex H column (ChromTech) operated at 65 °C. Isocratic separations used 4 mM sulfuric acid at a rate of 0.4 mL per minute. Aqueous metabolite concentrations were calculated by regression analysis relative to standards. Cell dry weights (cdw) were determined using cell samples from mid-exponential phase cultures.

Kinetic modeling

Mathematica 10.0 package (Wolfram Research Inc., Champaign, IL, USA) was used to solve all ordinary differential equations, data fitting, calculating parameters with standard errors, and performing ANOVA sensitivity analysis. In Mathematica, the Fitmodel equation was used to perform 1000 iterations. The entire data were statistically analyzed by ANOVA to estimate standard error, t-statistic, and *P*-values. The *P*-value is the most critical parameter necessary to assess the quality of data fitting and must be below 0.05 to be statistically significant.

T. maritima growth

The growth of *T. maritima* was modeled assuming cell growth was dependent on first order kinetics:

$$dX/dt = \mu X \quad (1)$$

where *X* was cellular biomass (g L⁻¹), *t* was time (h), and μ (h⁻¹) was the proportionality constant generally known as the specific growth rate.

The doubling time of the individual cell lines was estimated using the condition: $X = 2X$ at $t = t_d$, where t_d is the doubling time (h in Eq (1):

$$t_d = 0.695/\mu \quad (2)$$

Monod's equation or the logistic approach [30] has been widely used for modeling growth. Here the logistic approach was used for modeling the growth of *T. maritima* due to its simplicity for calculation of batch fermentation data and the utilization of significant biological and bioreactor geometric parameters.

The logistic model can be presented as:

$$dX/dt = \mu_{\max} (1 - X/X_{\max}) X \quad (3)$$

where μ_{\max} is the maximum specific growth rate (h^{-1}) and X_{\max} is the maximum attainable biomass (g L^{-1}). Applying initial condition: $X = X_0$ at $t = t_0$, Eq (3) can be simplified to the biomass equation:

$$X = \frac{X_0 X_{\max} e^{\mu_{\max} t}}{X_{\max} - X_0 + X_0 e^{\mu_{\max} t}} \quad (4)$$

Product formation

The Leudking-Piret equation [30] was used to model the productions of H_2 and acetate, where the rate of product formation was dependent on both growth and non-growth associated production, as shown in the following equation:

$$dP/dt = \alpha dX/dt + \beta X \quad (5)$$

where P is the concentration of product i.e. H_2 or acetate (mmol L^{-1}), α is the growth associated coefficient (mmol g^{-1}), and β is the non-growth associated coefficient ($\text{mmol g}^{-1} \text{h}^{-1}$).

The formation of products can be divided into three different classes [31]: (i) Class I that represents products connected to biomass formation exclusively ($\alpha \neq 0$; $\beta = 0$), (ii) Class II that concerns products moderately connected with biomass formation ($\alpha \neq 0$; $\beta \neq 0$), and (iii) Class III represents products that are unrelated to biomass formation ($\alpha = 0$; $\beta \neq 0$). The experimental data and modeling show that H_2 and acetate formation by *T. maritima* fall into the Class II category.

Maltose consumption

An equation that describes maltose consumption can be represented by the utilization of the sugar in biomass formation, maintenance, and metabolite (product) formation:

$$-\frac{dS}{dt} = \frac{1}{Y_{X/S}} \frac{dX}{dt} + mX + \frac{1}{Y_{P/S}} \frac{dP}{dt} \quad (6)$$

where S is the substrate concentration (mmol L^{-1}), $Y_{X/S}$ is the biomass yield coefficient ($\text{g-biomass mmol-maltose}^{-1}$), m is the maintenance coefficient ($\text{mmol g}^{-1} \text{h}^{-1}$), and $Y_{P/S}$ is the product yield coefficient ($\text{g-biomass mmol-maltose}^{-1}$). H_2 and acetate productions in *T. maritima* can be related to biomass using the expression: $Y_{P/X} = -dP(t)/dX(t)$, where $Y_{P/X}$ is the yield of biomass-based product.

Sensitivity analysis

The parameters obtained from the modeling of biomass ($t_d, \mu_{\max}, X_{\max}$), H_2 production ($\alpha_{\text{H}_2}, Y_{\text{H}_2/S}$), acetate production ($\alpha_A, \beta_A, Y_{A/S}$), and substrate (maltose) consumption ($m, Y_{X/S}$) were calibrated and analyzed using ANOVA sensitivity analysis. Apart from the values of t-statistic, P-value, and R^2 , the data were subjected to analysis with fitted residuals and estimated variance.

Simulation of continuous H_2 production

Continuous H_2 production was simulated with a continuous stirred tank reactor (CSTR) consistent with the experimental conditions. The assumptions for the continuous culture simulation were (i) the inflow stream to CSTR had a maltose concentration of 15 mM with no biomass, (ii) the inflow and outflow were set to the same flow rate, and (iii) the respective calculated batch parameters for the three cell lines were assumed to be the same in the experimental conditions. The following equations were used to represent the growth, maltose consumption, and product formation (H_2 and acetate):

$$dX/dt = -DX + \mu X \quad (7)$$

$$\frac{dS}{dt} = D(S_0 - S) - \left(\frac{1}{Y_{X/S}} \frac{dX}{dt} + m_s X + \frac{1}{Y_{P/S}} \frac{dP}{dt} \right) \quad (8)$$

$$P = Y_{P/X} X \quad (9)$$

where D is the dilution rate (h^{-1}), and S_0 is the initial substrate amount (g L^{-1}).

Results and discussion

Kinetic modeling

The kinetic modeling of growth (Equation (4)), H₂ production (Equation (5)), acetate production (Equation (5)), and maltose utilization (Equation (6)) resulted in best-fit plots as shown in **Fig. 2**. The related kinetic parameters are listed in **Table 1**.

Growth kinetics

To maintain anaerobic conditions and to avoid H₂ associated growth inhibition of the cell lines, the headspace of the bioreactor was continuously replaced by nitrogen (N₂) sparging at a constant rate of 15 mL min⁻¹ throughout the duration of the fermentation. Fig. 2a shows

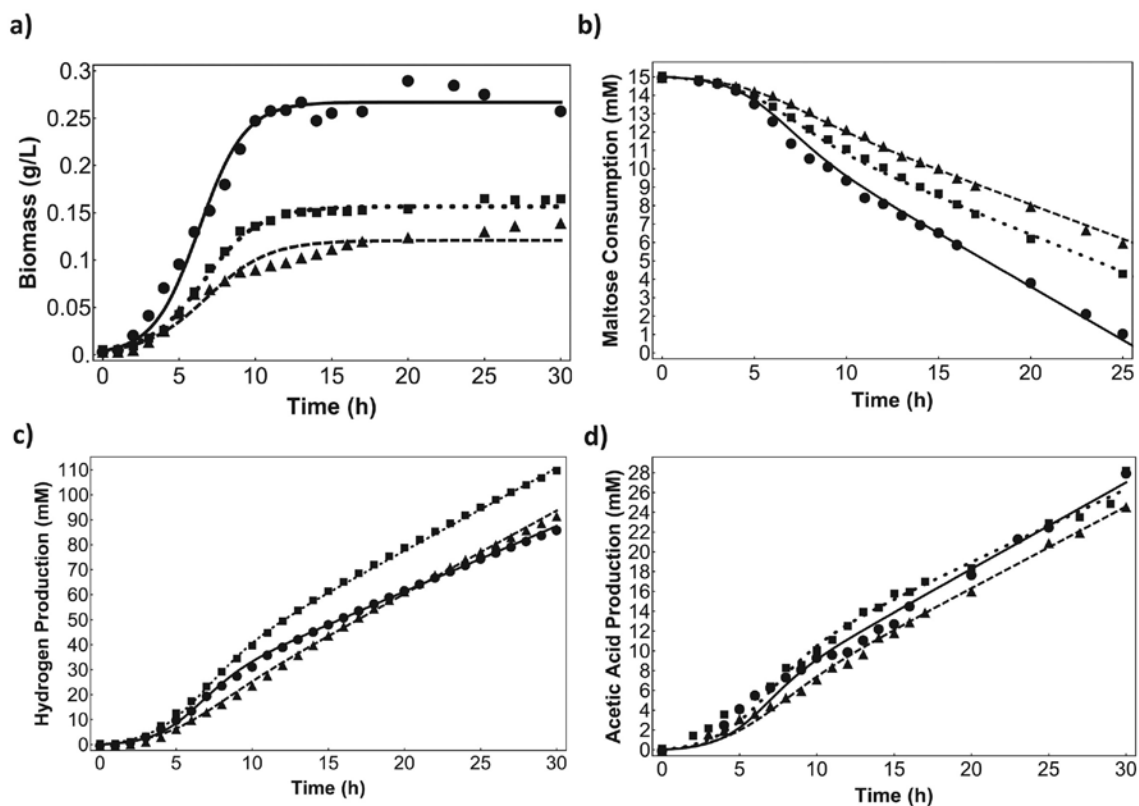


Fig. 2. Fitting of experimental data of growth, maltose consumption, H₂ production and acetate production in wild type, Tma100 and Tma200 to the kinetic models. Experimental: wild type (●), Tma100 (▲) and Tma200 (■). Kinetic model fitted curves for wild type (—), Tma100 (- -) and Tma200 (···).

Table 1. Fitted values of estimated parameters.

Parameters	<i>Thermotoga maritima</i> strains		
	Wild type	Tma100	Tma200
(i) Biomass			
Doubling time, t_d (h)	1.05	1.40	1.26
Maximum specific growth, μ_{max} (1/h)	0.663 ± 0.017	0.495 ± 0.022	0.550 ± 0.008
Maximum biomass, X_{max} (g/L)	0.267 ± 0.004	0.121 ± 0.003	0.157 ± 0.001
R^2	0.9961	0.9893	0.9988
(ii) H₂ Production			
H ₂ formation coefficient, α (mol-H ₂ /g-biomass)	0.096 ± 0.001	0.142 ± 0.004	0.218 ± 0.002
H ₂ non- growth associated coefficient, α (mol-H ₂ /g-biomass)	0.009 ± 0.001	0.028 ± 0.001	0.021 ± 0.002
Maximum H ₂ production (% cdw)	4.30	7.96	10.20
Biomass based hydrogen yield, $Y_{H_2/X}$ (mol-H ₂ /g-biomass)	0.204 ± 0.012	0.435 ± 0.023	0.471 ± 0.024
R^2	0.9998	0.9997	0.9999
(iii) Acetate (AA) Production			
AA formation coefficient, α (mol-AA/g- biomass)	0.024 ± 0.021	0.047 ± 0.025	0.060 ± 0.02
AA non- growth associated coefficient, β (mol-AA/(g-biomass * h))	0.0033 ± 0.0002	0.0068 ± 0.0002	0.0047 ± 0.0002
Biomass based AA yield, $Y_{AA/X}$ (mol-AA/g-biomass)	0.059 ± 0.015	0.112 ± 0.026	0.117 ± 0.031
R^2	0.9944	0.9981	0.9971
(iv) Maltose Consumption			
Maximum biomass yield, $Y_{X/S}$ (g-biomass/mol-maltose)	30.4 ± 1.05	22.3 ± 0.87	24.5 ± 0.93
Maintenance coefficient, m (g-biomass/(mol-maltose * h))	0.370 ± 0.063	0.509 ± 0.084	0.940 ± 0.071
Substrate based H ₂ yield, $Y_{H_2/S}$ (mol-H ₂ /mol-maltose)	6.22 ± 0.13	9.69 ± 0.20	11.54 ± 0.22
Substrate based AA yield, $Y_{AA/S}$ (mol-A/mol-maltose)	1.82 ± 0.02	2.66 ± 0.02	2.87 ± 0.02
R^2	0.9998	0.9999	0.9999
Ratio (H ₂ /AA)	3.41	3.64	4.02

AA — Acetic Acid; H₂ — Hydrogen gas.

the fitting of the experimental growth data using Equation (4). The ANOVA analysis (p -values < 0.05) and optimal fitting (R^2 value of 0.99) of the data confirm the reliability of the model. As evident by the initial slopes of growth curves shown in **Fig. 2a**, the model fitting demonstrated higher parametric determination of specific growth rates in the wild type followed by Tma200 (approximately 75% lower than wild type) and finally by Tma100 (approximately 83% lower than wild type). These variable growth rates resulted in approximately 45% and 59% lower overall biomass for Tma100 and Tma200 than that of the wild type, respectively. The differences in values of the biomass production rates observed between the cell lines were due to different doubling times and specific growth rates as listed in **Table 1** and were independent of H₂-associated growth inhibition [20] since a continuous flow of N₂ avoided H₂ accumulation [16]. Biomass based productivity of H₂ (96 mmol H₂ g⁻¹ cdw) in the wild type was comparable with prior fermentation studies (94 mmol g⁻¹ cdw) carried out with *T. maritima* [32].

Relationships between maltose consumption, growth and product formation

The maltose consumption rates, growth and product yields were estimated on the basis of limiting substrate (maltose) utilization during fermentation. The experimental values of maltose consumption in the cell lines were fitted ($p < 0.05$ and $R^2 = 0.99$) for their respective experimental data points using Equation (6) (**Fig. 2b**). The wild type grew faster (specific growth rate = 0.663 h^{-1}) and therefore utilized more than 95% of the available maltose in 30 h of fermentation. The reduction in the growth rates of Tma100 (0.495 h^{-1}) and Tma200 (0.550 h^{-1}) was consistent with a slower rate of maltose uptake and consumption. The poor growth of Tma100 resulted in a residual amount of unutilized maltose (6.75 mM) in the bioreactor after 30 h of fermentation. Tma200 showed an intermediate growth pattern again resulting in a residual amount of unutilized maltose (4.70 mM). The reason for the residual maltose was due to a slower maltose uptake by the mutated maltose transporter of Tma100 and Tma200 [25].

It has been reported that non-growing or slowly growing microbial cultures can recycle much of the metabolic energy derived from the initial fermentation of sugar into driving additional H_2 production [33]. To determine if this might be occurring in this study, we determined the amount of substrate utilized for non-growing biomass and the maintenance coefficient for all three strains (**Table 1**). The estimated values of the maintenance coefficients showed that Tma100 and Tma200 strains utilized ~ 1.37 and ~ 2.54 -fold more maltose respectively than the wild type for their non-growing components that contributed towards the formation of comparatively higher fermentation products than that of wild type. Importantly, fermentations in fermentative organisms have been optimized by evolution to produce cell biomass and not H_2 [34]. This has been a major obstacle for the production of biological H_2 in an economically viable manner but could be overcome if H_2 formation could be uncoupled from biomass formation and cell growth. From the data presented here it was evident that a greater proportion of maltose contributed to formation of fermentative products (H_2 and acetate) rather than biomass in strains Tma100 and Tma200 and was accompanied by slower growth (**Table 1**). The higher g cdw mol^{-1} maltose ratio in the wild type indicated that the maltose consumed by the wild type strain resulted in

an excess of biomass formation in deference to formation of fermentative products including H_2 (**Fig. 5**).

To observe the amount of fermentative products (H_2 , CO_2 and acetate) synthesized by the strains (Tma100 and Tma200) [25], the substrate-based product yields were calculated. Tma100 and Tma200 were found to be superior in H_2 and acetate yields as compared to the wild type; the yields of H_2 production were ~ 1.56 and ~ 1.86 -fold higher, respectively, whereas acetate yields were ~ 1.46 and ~ 1.58 -fold higher, respectively compared with the wild type. Tma100 and Tma200 were also found to produce 1.40-fold and 1.65 fold higher molar yield of CO_2 than the wild type suggesting increased oxidation of the sugar.

Relationship of H_2 production to kinetic parameters determined from biomass formation and maltose consumption

To obtain H_2 -related kinetic parameters such as the H_2 formation coefficient, maximum production, and maximum yield, the experimental data for H_2 production for each cell line were fitted using the Leudking-Piret Equation (5) as shown (**Fig. 2c**). The fitted values for these parameters are listed in **Table 1**. The statistical analysis ($p < 0.05$) and model fitting ($R^2 = 0.99$) verified a higher confidence in the predictions. It was observed that the H_2 formation coefficient (a), that represented growth-associated H_2 production, remained ~ 1.53 and ~ 2.26 -fold higher for Tma100 and Tma200 relative to the wild type, respectively. This highlighted that H_2 formation was growth dependent, which was greater for Tma100 and Tma200 even though their growth rates were lower than that of the wild type. This could mean that production of H_2 in Tma100 and Tma200 was higher on a per cell basis. The production of H_2 in the stationary phase is represented by b , which is the non-growth H_2 coefficient. A comparatively low value of b for the wild type than that of Tma100 and Tma200 suggested that H_2 production remained lower in the wild type under non-growth conditions, while Tma100 and Tma200 produced H_2 constantly even under non-growth conditions. This underscores a unique capacity of Tma100 and Tma200 to produce H_2 continuously and concomitantly with slow biomass production. To determine the value for H_2 production per unit biomass, the respective exponential growth phase values of H_2 were linearly regressed (**Fig. 3**). The slopes represented the

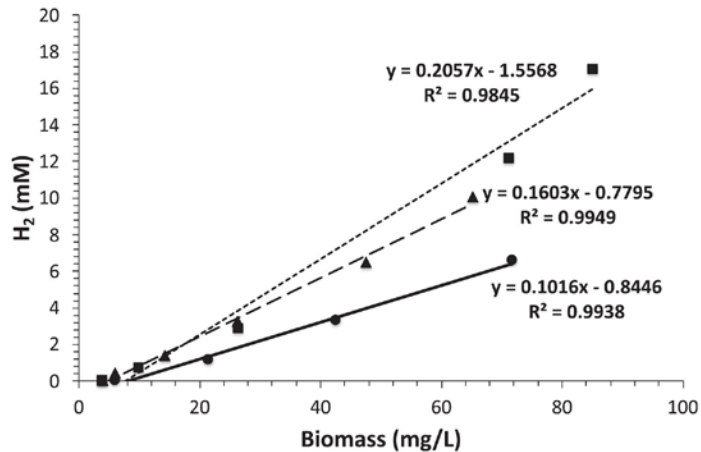


Fig. 3. Relationship between biomass and H₂ production for wild type, Tma100 and Tma200. The slope represents the H₂ production rate per mg cdw L⁻¹ biomass.

H₂ production per mg-cdw. The higher slopes of H₂ production for Tma200 (2 fold) and Tma100 (1.6 fold) relative to the wild type were also consistent with higher H₂ productivity per unit biomass. The intercept, representing the minimum concentration of biomass (on a cdw basis) that evolved H₂ and was found to be twice as high for Tma200 (1.56 mg L⁻¹) relative to those of both the wild type (0.84 mg L⁻¹) and Tma100 (0.78 mg L⁻¹). Therefore, the results indicated that a lower quantity of biomass and corresponding growth resulted in H₂ over-production by Tma100 and Tma200 (**Fig. 5**). This is consistent with prior studies in other organisms where reducing the substrate feeding to decrease biomass resulted in higher H₂ production [35]. As reported previously, the differential growth patterns observed for wild type, Tma100 and Tma200 were a result of their respective ability to transport and ferment maltose [25]. Based on the modeling reported here, H₂ production was a function of both growth and maltose utilization, and a reduction in both yielded more fermentative products.

A kinetic and redirected metabolism pathway-based explanation for the higher molar yield of H₂

The most interesting outcome from studies using Tma100 and Tma200 was that H₂ production per g cdw increased relative to the wild type and the molar yield of H₂ surpassed the previously

proposed limit of 4 mol of H₂ per hexose [19]. For Tma100, the ratio of moles of H₂ per mole of maltose was 9.69 ± 0.20 and for Tma200 the ratio was 11.54 ± 0.22 . This confirms the hydrogen yield to go beyond the predicted Thauer limit. (**Table 2**). It also emphasizes the unique nature of these strains.

Higher values of H₂ formation coefficient in combination with higher non-growth H₂ coefficients for Tma100 and Tma200 suggested that H₂ production in these strains was a continuous process. This could be one of the kinetic modeling-based explanations of the higher molar yield of H₂. Furthermore, the maintenance energy coefficient, (a physiological parameter) that specifies the amount of energy cells require for maintaining homeostasis in the absence of growth [36], was found to be twice as high for Tma100 and three times higher for Tma200 relative to that of the wild type. This suggests that energy from maltose catabolism is used to maintain the cellular integrity of Tma100 and Tma200 instead of being used for cellular multiplication. A relationship of higher maintenance coefficients to the highest yields of H₂ is consistent with prior studies in other hyperthermophiles [37].

The direction of the carbon flux through the Entner-Doudoroff (ED) or Embden-Meyerhof-Parnas (EMP) or the Pentose Phosphate Pathway (PPP) can determine the amount of H₂ and other metabolites. Normally, *T. maritima* utilizes the EMP (85%) and the ED (15%) pathways without using the PPP ultimately leading to a maximum production of 4 mol of H₂ and 2 mol of acetate per mol of glucose consumed [14]. If the carbon flux is rerouted through the PPP, the additional reductants can be produced leading to either 8 mol H₂ per mole of glucose (maximum) along with 1 mol acetate per mole of glucose⁻¹ or 5.33 mol H₂ mol⁻¹ glucose (minimum) along with 1.67 mol acetate per mole of glucose depending on reutilization of intermediates (**Fig. S1**). The carbon flux is rerouted through the PPP in the engineered strains and the glycolytic pattern resembles the letter "p". This can be defined as the p-type metabolism that results in maximum hydrogen productivity. In contrast, carbon flows through the EMP and ED pathway in the wild type and the overall carbon flow pattern resembles the letter "q". This can be defined as the q-type metabolism that is less productive (**Fig. 5**). The production of ATP is an important consideration because it is required to sustain bacterial growth. Generation of 8 mol H₂ mol⁻¹ glucose leads to 1 mol ATP formation whereas 5.33 mol H₂ mol⁻¹ glucose leads to 1.63 mol ATP (**Fig. S1**). The lower

Table 2. Hydrogen productivity and molar yield of H₂ by various hyperthermophiles.

Organism	Growth temperature (T_{opt}) (°C)	Cultivation Mode	(working volume/fermenter vol)	Substrate	Molar yield (mol H ₂ mol glu ⁻¹)	H ₂ production rate (mmol g _{cdw} ⁻¹ h ⁻¹)	Reference
<i>Thermotoga maritima</i> DSM 3109	80	Batch	100mL/1.25L serum bottle	Glu	4.0	75-80 (cal)	[14]
<i>Thermotoga maritima</i>	80	Batch	1.5L/2.3L	Glu	2.12 (growth phase)	10.65 (cal)	[32]
<i>Thermotoga neopolitana</i>	80	Batch	600mL/2.4L glass reactor	Glu	3.46	NA	[49]
<i>Thermotoga neopolitana</i> DSM 435	85	Batch	50mL/160 mL serum bottle	Glu	3.8	NA	[50]
<i>Thermotoga maritima</i>	80	Batch	60mL/120 mL bottle	Ara	3.2	NA	[38]
<i>Thermotoga elfii</i> DSM 9442	65	Batch	1L/3L	Glu	3.3	5.1	[51]
<i>Caloranaerobacter azorensis</i> H53214	60	Batch	20mL/50 mL bottle	Glu	1.46	25.7	[52]
<i>Thermococcus kodakaraensis</i> KOD1	85	Continuous D = 0.2 h ⁻¹	3L/10L	Star	3.3	14	[53]
<i>Thermococcus kodakaraensis</i> KOD1	85	Continuous	3L/10L	Pyr	1.1	24.9–59.6	[53]
<i>Thermobrachium celere</i>	67	D = 0.2–0.8 h ⁻¹	50mL/120 mL serum bottle	Glu	3.36	30 (cal)	[54]
<i>Caloramator celer</i> formerly known as <i>Thermobrachium celere</i>	67	Batch (pH = 8)	1L/3L	Glu	1.90	26.52 (cal)	[55]
<i>Caldicellulosiruptor saccharolyticus</i>	70	Continuous D = 0.05–0.15	1L/3L	Glu	3.48 (D = 0.05) and 2.90 (D = 0.15)	19.80 (D = 0.05) and 29.50 (D = 0.15) (cal)	[24]
<i>Caldicellulosiruptor saccharolyticus</i>	72.5	Continuous D = 0.09–0.30	0.75L/2L	Glu (1.9 g L ⁻¹)	3.0–3.60	13.33–25.39 (cal)	[37]
<i>Caldicellulosiruptor saccharolyticus</i>	70	Batch	1L/3L	Suc	3.3	11.7	[51]
<i>Caldicellulosiruptor owensensis</i> OL ₁	70	Batch	1L/3L STR	Xyl	3.5	21.59 (cal)	[56]
<i>Caldicellulosiruptor owensensis</i> OL ₁	70	Batch	1L/3L STR	Glu	4	19.73 (cal)	[56]
<i>Thermoanaerobacter tengcongensis</i>	75	Continuous	10L/12L	Glu	4.0 (predicted)	NA	[57]
<i>Thermoanaerobacter mathranii</i> A3N	70	Batch	10mL/125 mL serum bottle Glu/Suc/Xyl	@ 10 g	2.64/2.68/2.07	6.23 (glu)/6.13 (suc)/4.90 (xyl)	[58]
<i>Pyrococcus furiosus</i>	90	Continuous D = 0.1–0.75 h ⁻¹	1L/2L	Mal	3.17	25–100 (cal)	[42]

Mal-maltose, Glu-glucose, Suc-sucrose, Xyl-xylose, Pyr-pyruvate, Cell-cellulose, Ara-arabinose, YE-yeast extract, COe carbon mono-oxide, STR- Stir Tank reactor. Cal-mmol H₂ g cdw⁻¹ h⁻¹ was calculated by diving the H₂ mmol h⁻¹ L⁻¹ to g cdw L⁻¹ using the original plots or values published in corresponding references.

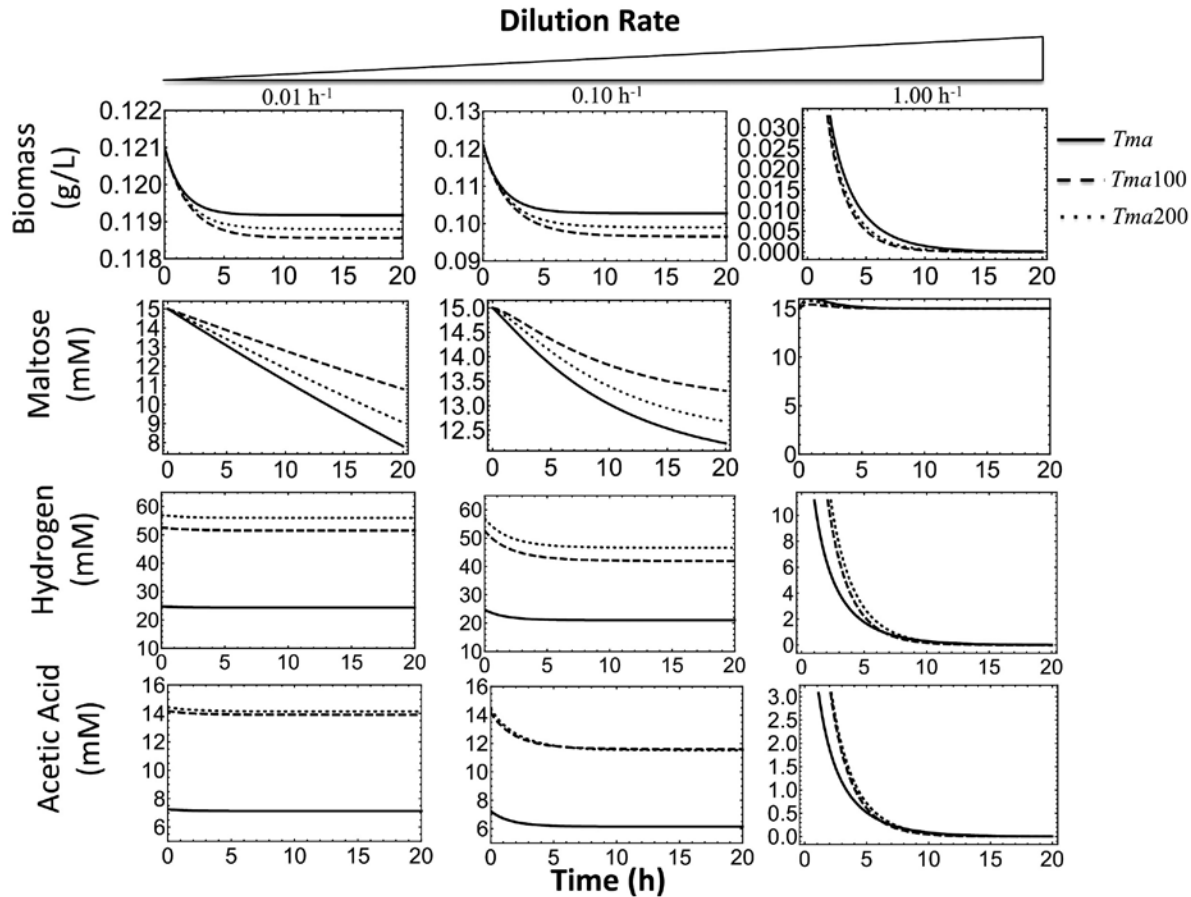


Fig. 4. Continuous flow simulation for biomass, maltose utilization, H₂ production and acetate production in wild type (—), Tma100 (- -) and Tma200 (···).

amount of ATP production does not offer any benefit to a fermentative bacterium instead it is likely to affect growth negatively. The organism would prefer to extract the maximum amount of ATP when carbon is redirected through the PPP. This was the case for Tma200 that generated 5.77 mol H₂ mol⁻¹ glucose and 1.87 mol acetate mol⁻¹ glucose representing the attainment of a stable metabolic condition that could generate enough ATP to sustain growth while synthesizing H₂ above the previously proposed hypothetical limit. In addition to the involvement of the PPP, the overall Gibbs free energy change for the metabolic reactions for Tma100 and Tma200 remained supportive of enough ATP formation and excess H₂ production [25].

Physiologically, a metabolic shift towards acetate offers another plausible reason of the increased H₂ molar yield by Tma100 and

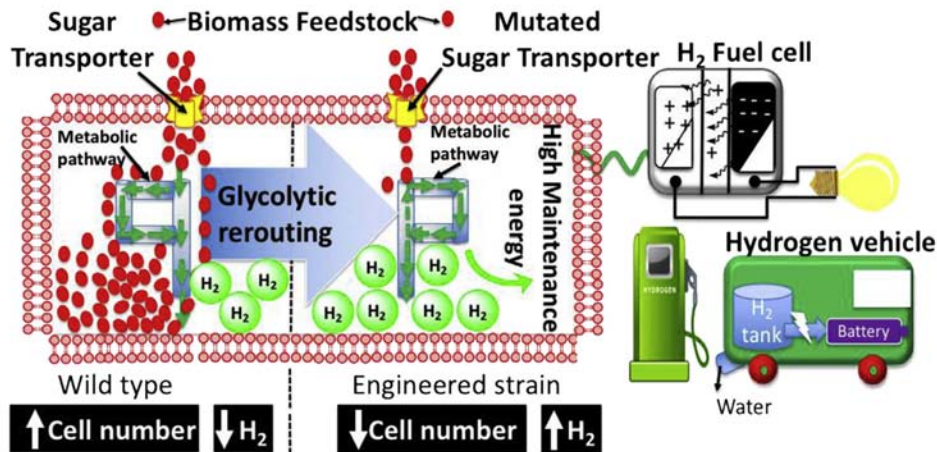


Fig. 5. A proposed model explaining feedstock uptake and the resultant metabolic transition from the letter “q” shaped (q-type metabolism) to “p” shaped (p-type metabolism). The wild type accumulates excessive carbon that forces q-type metabolism which is created by the use of the EMP pathway. In the engineered strain uptake of carbon is slow which activates p-type metabolism formed by the use of the PPP. Redirection of energy consumption from cell reproduction to maintenance energy enables more H₂ formation than the cell mass and the opposite is true in the wild type. Hydrogen can be used either to generate energy via a fuel cell or as a fuel for the hydrogen vehicles.

Tma200. In the absence of any detectable ethanol production [38] and a reduced quantity of lactate formation (none for Tma100 and a 75% reduction for Tma200 relative to the wild type) it appears that metabolism has been redirected in Tma100 and Tma200. A metabolic shift has been observed in different knockout mutants of another organism [39], however, in the present study this metabolic shift seems to occur as a result of variable maltose uptake and growth in Tma100 and Tma200. Prediction of the molar yield of acetate and a non-growth associated coefficient by Leudking-Piret equation verified the metabolic shift in Tma100 and Tma200 (**Table 1**).

Acetic acid production and kinetic parameters

In a fermentative H₂ producing organism, organic acid excretion can determine the effectiveness of H₂ production. A metabolic shift towards lactate production decreased the H₂ production rate [40] whereas an increase in acetate improved the H₂ production rate [12,41]. As no lactate was produced by Tma100 whereas a reduction

of 75% of lactate was observed in Tma200, only the experimental data for acetate formation was modeled. Previously, in small batch culture studies, the molar yield of acetate has been reported to be 2 mol mol⁻¹ glucose in *T. maritima* [14]. However using a larger scale batch culture, a lower yield of 1 mol acetate mol⁻¹ glucose was observed [32]. Furthermore, in a chemostat study of *Pyrococcus furiosus*, a yield of 1.27 mol of acetate per mole of glucose was reported [42]. This suggested that the theoretical value of 2 mol of acetate mol⁻¹ of glucose may not be achieved using a larger scale of batch cultivation. In this study, however, the molar yield of acetate for wild type was estimated to be 0.90 mol mol⁻¹ of glucose (1.82 ± 0.02 mol mol⁻¹ maltose) which is comparable (1 mol acetate mol⁻¹ glucose) to prior larger scale batch culture studies [32]. Additionally, a lower theoretical molar yield of acetate could result from a loss of carbon as L-alanine excretion as has been reported previously in *T. maritima* [32,43]. Since H₂ production was tightly linked to acetate production, a proportional increase in the acetate molar yield was achieved by Tma100 and Tma200. The overall yields of acetate in both engineered strains were very close to as predicted by the involvement of the PPP (**Fig. S1**). The model fitting of the acetate produced by the strains (**Fig. 2d**) showed that acetate was both a growth and non-growth associated product. The growth associated coefficient (α) increased 1.96 and 2.50-times for Tma100 and Tma200, respectively relative to the wild type (**Table 1**). The values of non-growth associated growth (β) increased by 2.06 and 1.42-fold for Tma100 and Tma200, respectively relative to the wild type (**Table 1**). Biomass-based acetate yield by Tma100 and Tma200 was ~1.98-fold higher than that of the wild type (**Table 1**). Carbon recovery was over 90% indicating a balanced stoichiometry and indicated that the major carbon source was maltose while other carbon present in yeast extract added to the growth medium did not contribute significantly to fermentation product formation [25]. A separate validation of the fermentation was pursued by calculating the oxidation-reduction balance of the overall fermentation reaction for all strains. The oxidation-reduction balance of oxidized and reduced products was close to the theoretical value of 1.0. This showed that the products were in balance and were accurately determined without any indication of new products present in significant quantities [25].

Simulation of a continuous biohydrogen production system

Growth simulation in a bioreactor

Kinetic modeling can provide a comprehensive analysis of experimental data to predict operating conditions [44] that are required for continuous fermentation. Continuous culture studies are crucial to achieve a more stable and higher yield of productivity ([45,46]. Since, H₂ is mostly a growth dependent product, a continuous stirred tank reactor (CSTR) may be the best choice for continuous H₂ production. Here, kinetic modeling of the bioreactor-based experimental data was used to estimate various kinetic parameters to define H₂-specific attributes necessary for H₂ over-producing strains. These became the basis for simulating a continuous H₂ production system.

The dilution rate (D) is an important factor that maintains cultivated organisms in their most productive phase leading to stabilization of the continuous culture-based system [47]. To predict a stable system for H₂ production, the determined kinetic parameters were employed to simulate a continuous stirred tank reactor. For the simulation of continuous H₂ production, Equations (7)–(9) were solved using the parameters for the respective strains as listed in **Table 1**. The comparative simulation results for all the cell lines at three different dilution rates are presented in **Fig. 4**. A lower dilution rate (0.01 h⁻¹) maintained a majority of the biomass (~119.6, 118.8, and 118.6 mg L⁻¹ of wild type, Tma200 and Tma100, respectively) compared to the dilution rate of 0.1 h⁻¹ (~100.6, 100.0, and 109.6 mg L⁻¹ of wild type, Tma200 and Tma100, respectively). Both lower rates (0.01 and 0.1 h⁻¹) were likely to maintain a steady state after 5 hr until the end of the fermentation. A higher dilution rate (1 h⁻¹) would result in a loss of significant biomass. The loss of biomass at a higher dilution rate can be attributed to the higher dilution rate (D) that approached the value of μ_{\max} (**Table 1**). Consequently, at a higher dilution rate, the continuous culture system is predicted to become unstable due to wash-out of the biomass.

Simulation of maltose consumption during continuous cultivation

The effect of dilution rate on maltose consumption was estimated at three dilution rates. At a lower dilution rate, due to the presence of a

higher number of cells in the bioreactor, an excess of substrate utilization was likely. From a simulation plot of maltose consumption, it was evident that a lower dilution rate (0.01 h^{-1}) would result in fast utilization of maltose in the wild type with a slower rate of consumption by Tma100 and Tma200. Therefore, the inherent ability of Tma100 and Tma200 to utilize less maltose makes them more economic in regard to substrate utilization. Since, bioreactor washout is predicted to occur at 1.0 h^{-1} , no maltose utilization would take place even in the presence of 15 mM maltose.

H₂ production simulation in a bioreactor

At a lower dilution rate (0.01 h^{-1}) H₂ production would remain constant without any fluctuation and the H₂ production rate would be significantly higher in Tma100 and Tma200 relative to that of the wild type. At an intermediate flow rate (0.1 h^{-1}) H₂ production would fluctuate and would collapse at 1 h^{-1} due to washout of cells as has been seen in CSTR simulation studies on H₂ production [48]. This suggested that maintaining a flow rate of 0.01 h^{-1} would retain the optimum number of cells of Tma100 (118.6 mg L^{-1}) and Tma200 (118.8 mg L^{-1}) thereby, allowing them to produce more H₂ than that of the wild type at a reduced rate of maltose utilization. From this observation, it was reasonable to conclude that maintaining a lower biomass for the wild type (equivalent to Tma100 and Tma200) could improve H₂ production by the wild type organism. However, as the only way to maintain a lower biomass of the wild type would be to increase the flow rate, it is evident from the simulation that an increase in flow rate will decrease H₂ production even though the biomass of the wild type will become equivalent to strains Tma100 or Tma200 at the steady state. This supports the notion that the productions of H₂ from Tma100 and Tma200 are higher on a per cell basis and will remain higher than that of wild type under any kinetic condition.

Acetic acid production simulation in a bioreactor

Similar to H₂ production, a proportional amount of acetate would be produced under continuous cultivation conditions and the overall concentration would be higher than that of the wild type. A higher dilution rate (1 h^{-1}) would result in a decrease of acetate due to wash

out of acetate producers. This simulation-based prediction considered results obtained from an experimental batch study [48]. From the overall simulation, a dilution rate of (0.1 h^{-1}) would be suitable for a constant amount of H_2 production in the continuous culture. As excess H_2 production is an inherent property of Tma100 and Tma200, and the wild type cannot achieve this despite manipulation of cultivation conditions, the evolved *T. maritima* cell lines are ideal candidates for economic and large scale H_2 production.

Conclusions

The kinetic modeling of product formation by *Thermotoga maritima* cell lines (wild type, Tma100, and Tma200) presents the relationship between maltose utilization, biomass production, and the syntheses of H_2 and acetate. Fitting of the experimental data showed that the excess H_2 production by Tma100 and Tma200 resulted from the contributions of growth and non-growth coefficients. The slow growth of Tma100 and Tma200 resulted from their reduced maltose uptake rates that shifted the q-type metabolism to p-type metabolism in these strains. The kinetic parameters were used to simulate a continuous and stable culture-based H_2 production system. Besides, the study suggests an improved design for large-scale biological hydrogen system.

Acknowledgments — This research was supported by funds from the U.S. Department of Energy (DEPS02-08ER08-12) and the UNL Cell Development Facility to PB. We thank Robert M. Kelly and Christopher Straub, North Carolina University for providing us detailed information on the carbon flux redirected through the pentose phosphate pathway and resultant H_2 yield.

Appendix A. Supplementary data — Supplementary Figures 1 and 2 follow the References.

References

- [1] Das D. Advances in biohydrogen production processes: an approach towards commercialization. *Int J Hydrogen Energy* 2009; 34: 7349–57.

- [2] Gurz M, Baltacioglu E, Hames Y, Kaya K. The meeting of hydrogen and automotive: A review. *Int J Hydrogen Energy* 2017; 42: 23334–46.
- [3] Dutta S. A review on production, storage of hydrogen and its utilization as an energy resource. *J Ind Eng Chem* 2014; 20: 1148–56.
- [4] Hames Y, Kaya K, Baltacioglu E, Turksoy A. Analysis of the control strategies for fuel saving in the hydrogen fuel cell vehicles. *Int J Hydrogen Energy* 2018; 43: 10810–21.
- [5] Sinha P, Pandey A. An evaluative report and challenges for fermentative biohydrogen production. *Int J Hydrogen Energy* 2011; 36: 7460–78.
- [6] Ehret O, Bonhoff K. Hydrogen as a fuel and energy storage: Success factors for the German Energiewende. *Int J Hydrogen Energy* 2015; 40: 5526–33.
- [7] Coren MJ. Nine countries say they'll ban internal combustion engines. So far, it's just words. Quartz, August 7, 2018. [<https://qz.com/1341155/nine-countries-say-they-will-ban-internal-combustion-engines-none-have-a-law-to-do-so/> ; Accessed 29 December 2018].
- [8] Singh M, Moore J, Shadis W. Hydrogen demand, production, and cost by region to 2050. Argonne, IL USA: Argonne National Laboratory; 2005.
- [9] Hallenbeck P. Bioenergy from microorganisms: an overview. In: Zannoni D, De Philippis R, editors. *Microbial bioenergy: hydrogen production*. Netherlands: Springer; 2014. p. 3–21.
- [10] Hallenbeck PC, Ghosh D. Advances in fermentative biohydrogen production: the way forward? *Trends Biotechnol* 2009; 27: 287–97.
- [11] Chou CJ, Jenney Jr FE, Adams MW, Kelly RM. Hydrogenesis in hyperthermophilic microorganisms: implications for biofuels. *Metab Eng* 2008; 10: 394–404.
- [12] Nath K, Das D. Improvement of fermentative hydrogen production: various approaches. *Appl Microbiol Biotechnol* 2004; 65: 520–9.
- [13] Selig M, Xavier KB, Santos H, Schönheit P. Comparative analysis of Embden-Meyerhof and Entner-Doudoroff glycolytic pathways in hyperthermophilic archaea and the bacterium *Thermotoga*. *Arch Microbiol* 1997; 167: 217–32.
- [14] Schröder C, Selig M, Schönheit P. Glucose fermentation to acetate, CO₂ and H₂ in the anaerobic hyperthermophilic eubacterium *Thermotoga maritima*: involvement of the Embden-Meyerhof pathway. *Arch Microbiol* 1994; 161: 460–70.
- [15] van Niel EW, Claassen PA, Stams AJ. Substrate and product inhibition of hydrogen production by the extreme thermophile, *Caldicellulosiruptor saccharolyticus*. *Biotechnol Bioeng* 2003; 81: 255–62.
- [16] Nguyen TA, Han SJ, Kim JP, Kim MS, Sim SJ. Hydrogen production of the hyperthermophilic eubacterium, *Thermotoga neapolitana* under N₂ sparging condition. *Bioresour Technol* 2010; 101(Suppl 1): S38–41.
- [17] Mizuno O, Dinsdale R, Hawkes FR, Hawkes DL, Noike T. Enhancement of hydrogen production from glucose by nitrogen gas sparging. *Bioresour Technol* 2000; 73: 59–65.
- [18] Show K-Y, Lee D-J, Chang J-S. Bioreactor and process design for biohydrogen production. *Bioresour Technol* 2011; 102: 8524–33.
- [19] Thauer RK, Jungermann K, Decker K. Energy conservation in chemotrophic anaerobic bacteria. *Bacteriol Rev* 1977; 41: 100–80.

- [20] Huber R, Langworthy T, König H, Thomm M, Woese C, Sleytr U, et al. *Thermotoga maritima* sp. nov. represents a new genus of unique extremely thermophilic eubacteria growing up to 90°C. *Arch Microbiol* 1986; 144: 324–33.
- [21] Chhabra SR, Shockley KR, Conners SB, Scott KL, Wolfinger RD, Kelly RM. Carbohydrate-induced differential gene expression patterns in the hyperthermophilic bacterium *Thermotoga maritima*. *J Biol Chem* 2003; 278: 7540–52.
- [22] Singh R, White D, Blum P. Identification of the ATPase subunit of the primary maltose transporter in the hyperthermophilic anaerobe *Thermotoga maritima*. *Appl Environ Microbiol* 2017; 83.
- [23] Schut GJ, Adams MW. The iron-hydrogenase of *Thermotoga maritima* utilizes ferredoxin and NADH synergistically: a new perspective on anaerobic hydrogen production. *J Bacteriol* 2009; 191: 4451–7.
- [24] Willquist K, Pawar SS, Van Niel EW. Reassessment of hydrogen tolerance in *Caldicellulosiruptor saccharolyticus*. *Microb Cell Factories* 2011; 10: 111-2859-10-111.
- [25] Singh R, White D, Demirel Y, Kelly R, Noll K, Blum P. Uncoupling fermentative synthesis of molecular hydrogen from biomass formation in *Thermotoga maritima*. *Appl Environ Microbiol* 2018; 84.
- [26] Nath K, Das D. Modeling and optimization of fermentative hydrogen production. *Bioresour Technol* 2011; 102: 8569–81.
- [27] Ghimire A, Frunzo L, Pirozzi F, Trabaly E, Escudie R, Lens PNL, et al. A review on dark fermentative biohydrogen production from organic biomass: process parameters and use of byproducts. *Appl Energy* 2015; 144: 73–95.
- [28] Singh R, Gradnigo J, White D, Lipzen A, Martin J, Schackwitz W, et al. Complete genome sequence of an evolved *Thermotoga maritima* isolate. *Genome Announc* 2015; 3.
- [29] Chhabra SR, Shockley KR, Ward DE, Kelly RM. Regulation of endo-acting glycosyl hydrolases in the hyperthermophilic bacterium *Thermotoga maritima* grown on glucan- and mannan-based polysaccharides. *Appl Environ Microbiol* 2002; 68: 545–54.
- [30] Luedeking R, Piret EL. A kinetic study of the lactic acid fermentation. Batch process at controlled pH. *Biotechnol Bioeng* 2000; 67: 636–44.
- [31] Tevatia R, Demirel Y, Blum P. Kinetic modeling of photoautotrophic growth and neutral lipid accumulation in terms of ammonium concentration in *Chlamydomonas reinhardtii*. *Bioresour Technol* 2012; 119: 419–24.
- [32] Lakhal R, Auria R, Davidson S, Ollivier B, Dolla A, Hamdi M, et al. Effect of oxygen and redox potential on glucose fermentation in *Thermotoga maritima* under controlled physicochemical conditions. *Internet J Microbiol* 2010; 2010: 896510.
- [33] Keasling JD, Benemann J, Pramanik J, Carrier TA, Jones KL, Van Dien SJ. A toolkit for metabolic engineering of bacteria. *BioHydrogen*. Springer; 1998. p. 87–97.
- [34] Hallenbeck PC. Fundamentals of the fermentative production of hydrogen. *Water Sci Technol* 2005: 21–9.

- [35] Van Ginkel SW, Logan B. Increased biological hydrogen production with reduced organic loading. *Water Res* 2005; 39: 3819–26.
- [36] Pirt SJ. The maintenance energy of bacteria in growing cultures. *Proc R Soc Lond B Biol Sci* 1965; 163: 224–31.
- [37] de Vrije T, Mars AE, Budde MA, Lai MH, Dijkema C, de Waard P, et al. Glycolytic pathway and hydrogen yield studies of the extreme thermophile *Caldicellulosiruptor saccharolyticus*. *Appl Microbiol Biotechnol* 2007; 74: 1358–67.
- [38] Eriksen N, Riis M, Holm N, Iversen N. H₂ synthesis from pentoses and biomass in *Thermotoga* spp. *Biotechnol Lett* 2011; 33: 293–300.
- [39] Liu S, Nichols N, Dien B, Cotta M. Metabolic engineering of a *Lactobacillus plantarum* double *ldh* knockout strain for enhanced ethanol production. *J Ind Microbiol Biotechnol* 2006; 33: 1–7.
- [40] Levin DB, Pitt L, Love M. Biohydrogen production: prospects and limitations to practical application. *Int J Hydrogen Energy* 2004; 29: 173–85.
- [41] Cha M, Chung D, Elkins JG, Guss AM, Westpheling J. Metabolic engineering of *Caldicellulosiruptor bescii* yields increased hydrogen production from lignocellulosic biomass. *Biotechnol Biofuels* 2013; 6: 85.
- [42] Schicho RN, Ma K, Adams MW, Kelly RM. Bioenergetics of sulfur reduction in the hyperthermophilic archaeon *Pyrococcus furiosus*. *J Bacteriol* 1993; 175: 1823–30.
- [43] Ravot G, Ollivier B, Fardeau ML, Patel BK, Andrews KT, Magot M, et al. L-alanine production from glucose fermentation by hyperthermophilic members of the domains bacteria and Archaea: A remnant of an ancestral metabolism? *Appl Environ Microbiol* 1996; 62: 2657–9.
- [44] Mu Y, Wang G, Yu HQ. Kinetic modeling of batch hydrogen production process by mixed anaerobic cultures. *Bioresour Technol* 2006; 97: 1302–7.
- [45] Brethauer S, Wyman CE. Review: continuous hydrolysis and fermentation for cellulosic ethanol production. *Bioresour Technol* 2010; 101: 4862–74.
- [46] Ren N-Q, Tang J, Liu B-F, Guo W-Q. Biological hydrogen production in continuous stirred tank reactor systems with suspended and attached microbial growth. *Int J Hydrogen Energy* 2010; 35: 2807–13.
- [47] Chen CC, Lin CY, Chang JS. Kinetics of hydrogen production with continuous anaerobic cultures utilizing sucrose as the limiting substrate. *Appl Microbiol Biotechnol* 2001; 57: 56–64.
- [48] Frascari D, Cappelletti M, Mendes J de S, Alberini A, Scimonelli F, Manfreda C, et al. A kinetic study of biohydrogen production from glucose, molasses and cheese whey by suspended and attached cells of *Thermotoga neapolitana*. *Bioresour Technol* 2013; 147: 553–61.
- [49] d'Ippolito G, Dipasquale L, Vella FM, Romano I, Gambacorta A, Cutignano A, et al. Hydrogen metabolism in the extreme thermophile *Thermotoga neapolitana*. *Int J Hydrogen Energy* 2010; 35: 2290–5.
- [50] Munro SA, Zinder SH, Walker LP. The fermentation stoichiometry of *Thermotoga neapolitana* and influence of temperature, oxygen, and pH on hydrogen production. *Biotechnol Prog* 2009; 25: 1035–42.

- [51] van Niel EWJ, Budde MAW, de Haas GG, van der Wal FJ, Claassen PAM, Stams AJM. Distinctive properties of high hydrogen producing extreme thermophiles, *Caldicellulosiruptor saccharolyticus* and *Thermotoga elfii*. *Int J Hydrogen Energy* 2002; 27: 1391–8.
- [52] Jiang L, Long C, Wu X, Xu H, Shao Z, Long M. Optimization of thermophilic fermentative hydrogen production by the newly isolated *Caloranaerobacter azorensis* H53214 from deep-sea hydrothermal vent environment. *Int J Hydrogen Energy* 2014; 39: 14154–60.
- [53] Kanai T, Imanaka H, Nakajima A, Uwamori K, Omori Y, Fukui T, et al. Continuous hydrogen production by the hyperthermophilic archaeon, *Thermococcus kodakaraensis* KOD1. *J Biotechnol* 2005; 116: 271–82.
- [54] Ciranna A, Santala V, Karp M. Biohydrogen production in alkalithermophilic conditions: *Thermobrachium celere* as a case study. *Bioresour Technol* 2011; 102: 8714–22.
- [55] Ciranna A, Pawar SS, Santala V, Karp M, van Niel EW. Assessment of metabolic flux distribution in the thermophilic hydrogen producer *Caloramator celer* as affected by external pH and hydrogen partial pressure. *Microb Cell Factories* 2014; 13: 48.
- [56] Zeidan AA, van Niel EWJ. A quantitative analysis of hydrogen production efficiency of the extreme thermophile *Caldicellulosiruptor owensensis* OLT. *Int J Hydrogen Energy* 2010; 35: 1128–37.
- [57] Soboh B, Linder D, Hedderich R. A multisubunit membrane-bound [NiFe] hydrogenase and an NADH-dependent Fe-only hydrogenase in the fermenting bacterium *Thermoanaerobacter tengcongensis*. *Microbiology* 2004; 150: 2451–63.
- [58] Jayasinghearachchi HS, Sarma PM, Lal B. Biological hydrogen production by extremely thermophilic novel bacterium *Thermoanaerobacter mathranii* A3N isolated from oil producing well. *Int J Hydrogen Technol* 2012; 37: 5569–78.
-

Figs. S1 & S2 

Figure S1.

Singh et al

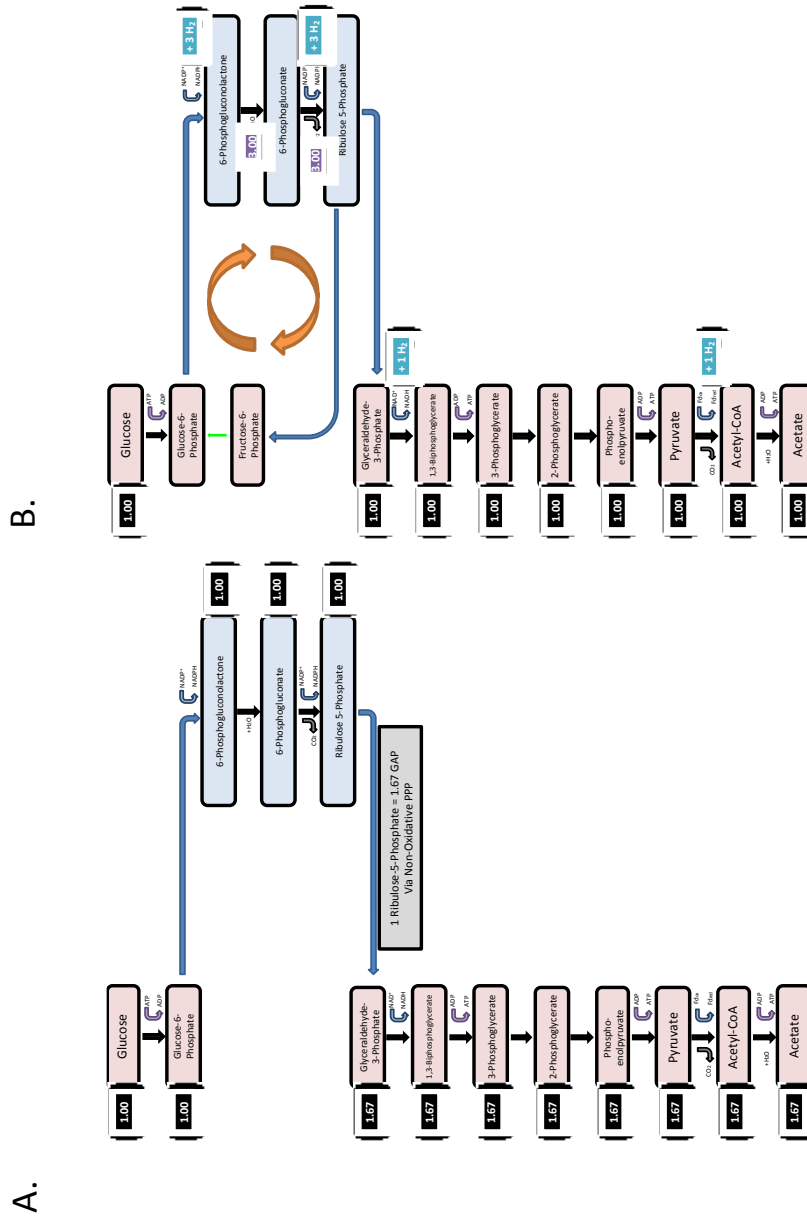


Figure S1. A proposed model of metabolites formation when carbon is redirected through the pentose phosphate pathway (PPP). The panel A shows the minimum amount of H₂ production and the panel B shows the maximum amount of H₂ production via the PPP.

Singh et al

Figure S2.

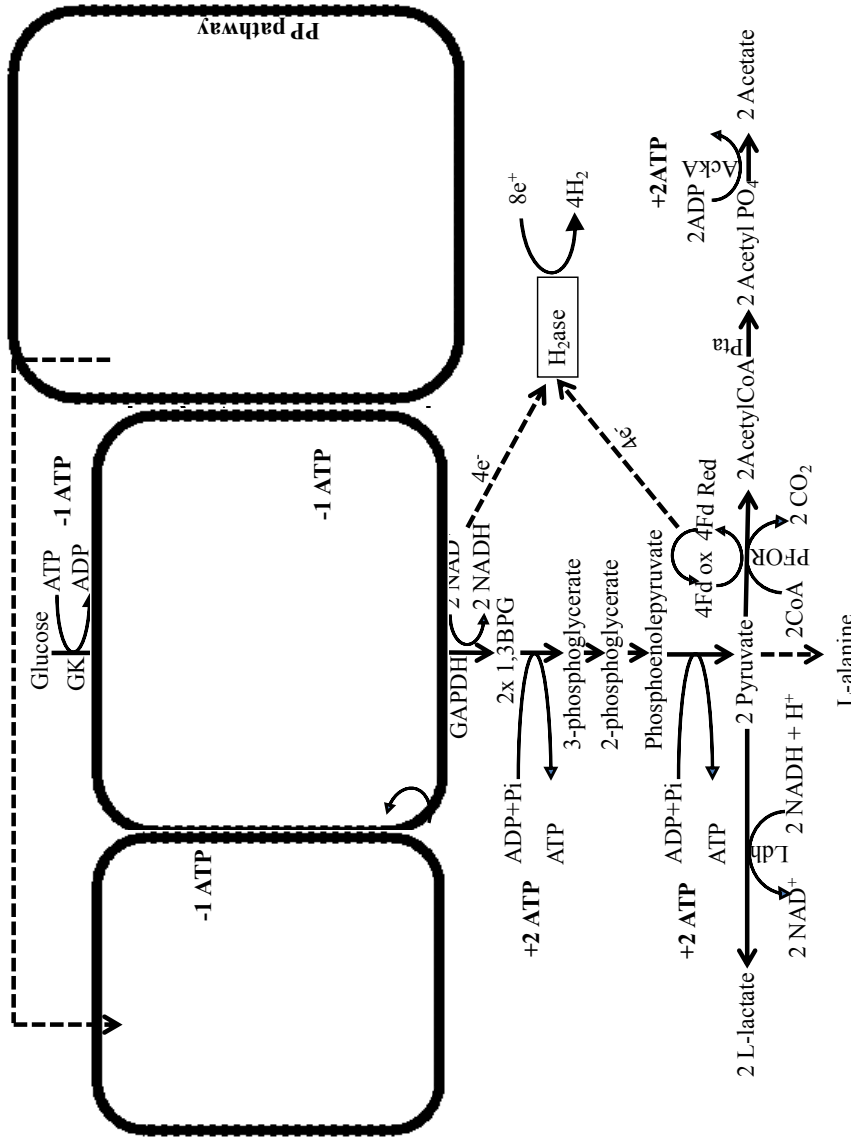


Figure S2. Proposed central metabolic pathways in *T. maritima*. **GK**- Glucose kinase, **PGI**- Phosphoglucose isomerase, **PFK**- Phosphofructokinase, **FBA**- Fructose-1,6-bisphosphate aldolase, **TIM**- Triose-phosphate isomerase, **G6PDH**- Glucose-6-phosphate dehydrogenase, **ilvD**- Phosphoglucuronate dehydratase, **gnd**- 6-phosphoglucuronate dehydrogenase, **KDG**- 2-keto-3-deoxygluconate, **KDPG**- 2-Keto-3-deoxy-6-phosphoglucuronate, **GAP**- Glyceraldehyde -3-phosphate, **1,3 BPG**- 1,3-bisphosphoglycerate, **GAPDH**-Glyceraldehyde-3phosphate dehydrogenase, **H₂ase**-Hydrogenase, **ldh**- lactate dehydrogenase, **PFOR**- Pyruvate Ferredoxin, **Pta**-Phosphate acetyltransferase, **ackA**- Acetate kinase.