## Comparison of the amorphization ability of two polymorphic forms of olanzapine

## Catarina Chendo<sup>1</sup>, Nuno F. da Costa<sup>1</sup>, João F. Pinto<sup>1</sup> and Ana I. Fernandes<sup>2</sup>

<sup>1.</sup> iMed.ULisboa, Faculdade de Farmácia, Universidade de Lisboa, Lisboa, Portugal; <sup>2.</sup> CiiEM – Centro de Investigação Interdisciplinar Egas Moniz, Instituto Universitário Egas Moniz, Monte de Caparica, Portugal.

## aifernandes@egasmoniz.edu.pt

**Introduction:** The production of amorphous and co-amorphous (CAM) materials has been used as a procedure to overcome the poor water solubility shown by most of the drugs currently under development (1). Ball milling has been considered to convert the crystalline state of a substance into its amorphous counterpart (2,3). At present, the impact of the initial polymorphic form of a drug substance on its final amorphous state upon milling, has not been clearly studied. This work aims to compare the co-amorphization efficiency of two different polymorphic forms of olanzapine (OLZ; a BCS class II drug) using saccharin (SAC) as a co-former.

**Materials and Methods:** OLZ (forms I and II) were the starting polymorphic forms to be milled with SAC (2:1 molar ratio). OLZ form I was used as received while OLZ form II was obtained from crystallization of OLZ in dichloromethane. Ball milling was performed using 2.5g of 3.0mm Ø balls, for 2 h. The conversion of the crystalline states into the amorphous counterpart was monitored by calorimetry (DSC) and diffractometry (XRPD).

**Results:** The thermograms and the diffractograms obtained (Fig. 1) have shown a clear difference between the final products of the two polymorphic forms of OLZ. XRPD peaks were less intense for OLZ form II and most representative of SAC, indicating that OLZ was indeed mostly converted into the amorphous state. On the other hand, when OLZ form I was the starting material, a richer diffractogram resulted, suggesting that a crystalline fraction of OLZ remained in the particles' network of the final product, a feature also confirmed by DSC.

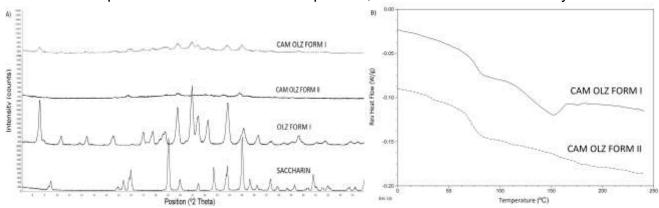


Figure 1- X-ray diffractograms (A) and DSC thermograms (B) of the different molecular entities.

**Discussion and Conclusions:** Since OLZ form II has a higher thermodynamic energy than form I, it is not surprising that the former exhibits better amorphization ability. Thus, it is expected that either an increase on milling time, or milling speed, would enhance the co-amorphization of OLZ form I with SAC.

## References:

- 1. Skieneh J, Rohani S. Screening new solid forms of pharmaceuticals to enhance solubility and dissolution rate. Austin Pharmacol Pharm. 2017; 2(1):1007.
- Chieng N, Aaltonen J, Saville D, Rades T. Physical characterization and stability of amorphous indomethacin and ranitidine hydrochloride binary systems prepared by mechanical activation. Eur J Pharm Biopharm. 2009; 71(1):47–54.
- 3. Lim AW, Löbmann K, Grohganz H, Rades T, Chieng N. Investigation of physical properties and stability of indomethacin-cimetidine and naproxen-cimetidine co-amorphous systems prepared by quench cooling, coprecipitation and ball milling. J Pharm Pharmacol. 2016; 68(1):36–45.

Acknowledgements: Fundação para a Ciência e a Tecnologia (PTDC/CTM-BIO/3946/2014) for funding.