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Abstract: The interest of Galician oil producers (NW Spain) in recovering the ancient autochthonous olive varieties Brava and Mansa has increased substantially in recent years. Virgin olive oils produced by co-crushing both varieties in two different proportions, reflecting the usual and most common practice adopted in this region, have gradually emerged for the production of virgin olive oils. Herein, the sensory and chemical characteristics of such oils were characterized by quality and genuineness-related parameters. In particular, minor components such as phenolic and volatile compounds were investigated. The results of chemical analysis are also discussed in terms of their effective contribution to the sensory profile, which suggests useful recommendations for olive oil producers to improve the quality of oils. A major content of phenolic compounds as well as a predominance of trans-2-hexen-1-al and cis-3-hexen-1-ol, responsible for bitter and green-leaf sensory notes, were seen when co-crushing with a higher proportion of Brava olives.

1 **Characterization of virgin olive oils**

2 **produced with autochthonous Galician varieties**

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18 **Abstract**

19 The interest of Galician oil producers (NW Spain) in recovering the ancient
20 autochthonous olive varieties *Brava* and *Mansa* has increased substantially in recent
21 years. Virgin olive oils produced by co-crushing both varieties in two different
22 proportions, reflecting the usual and most common practice adopted in this region, have
23 gradually emerged for the production of virgin olive oils. Herein, the sensory and
24 chemical characteristics of such oils were characterized by quality and genuineness-
25 related parameters. The results of chemical analysis are discussed in terms of their
26 effective contribution to the sensory profile, which suggests useful recommendations for
27 olive oil producers to improve the quality of oils. Antioxidant compounds, together with
28 aromas and coloured pigments were determined, and their contribution in determining
29 the functional value and the sensory properties of oils was investigated. In general,
30 given the high levels of phenolic compounds (ranging between 254 and 375 mg/kg oil),
31 tocopherols (about 165 mg/kg oil) and carotenoids (10-12 mg/kg oil); these are oils with
32 long stability, especially under dark storage conditions, because stability is reinforced
33 with the contribution of chlorophylls (15-22 mg/kg oil). A major content of phenolic
34 compounds, as well as a predominance of *trans*-2-hexen-1-al within odour-active
35 compounds (from 897 to 1645 µg/kg oil), responsible for bitter sensory notes. This
36 characterization allows to developing new antioxidant-rich and flavour-rich VOOs,
37 when co-crushing with a higher proportion of *Brava* olives, satisfying the consumers'
38 demand in having access to more healthy dishes and new sensorial sensations.

39

40 **Keywords:** *characterization, Galician virgin olive oils, phenolic compounds, volatile*
41 *compounds, sensory analysis.*

42

43 1. Introduction

44 Olive oil is a valuable product that is traditionally produced in Mediterranean countries.
45 Olive groves are present in 34 of the 50 Spanish provinces and occupy an area of
46 2,584,564 ha (AICA, 2015). Spain is the largest olive oil producer worldwide. In the
47 last few years, half of Spanish olive oil production is consumed domestically and the
48 other half is exported (Morales, Aparicio-Ruíz and Aparicio, 2013). Within the Spanish
49 territory, Galicia (NW Spain) has gradually emerged as a new olive-growing zone
50 producing virgin olive oils (VOOs) with autochthonous cultivars growing in particular
51 environmental and pedoclimatic conditions that characterize this area. Although
52 traditional Spanish varieties, such as Arbequina and Picual cv., are predominant in the
53 new Galician plantations, the interest of oil producers in ancient autochthonous varieties
54 (known by producers as *Brava* and *Mansa*) has increased substantially in recent years
55 due to their suitable edafo-climatic adaptation.

56 Two previous studies have evaluated the potential of these ancient cultivars (Reboredo-
57 Rodríguez *et al*, 2015a, 2015b). In the first, with the aim of providing extra value to the
58 final VOOs, Galician Arbequina or Picual fruits, separately, were co-crushed with low
59 proportions of a mixture of *Brava* and *Mansa* varieties (such a mixture is known by
60 producers as *Local*). The experimental results showed that the effect of co-crushing on
61 minor compounds, phenolics, and C₆ volatiles, both responsible for the sensory profile,
62 cannot be easily modulated because of a complex, non-progressive, and non-predictable
63 change in their composition, in contrast to most quality indices (*viz.* free acidity,
64 peroxides, and UV extinction coefficients) and fatty acid composition, which change
65 linearly in strict correlation with the fruit mass ratio. On the other hand, blending *Local*
66 VOOs (also in low percentages) with Arbequina or Picual monovarietal VOOs might be
67 another strategy to produce high quality VOOs with pre-established characteristics. In

68 this case, previous knowledge of the quality-related indices and fatty acid composition
69 as well as the concentrations of minor compounds of monovarietal VOOs make it
70 possible to obtain oils “à la carte”.

71 Nowadays, the current trend of the VOO market is production of high quality products
72 from traditional minor olive varieties with a specific designation of origin and
73 characteristic, well-defined sensory, nutritional, and health promoting properties
74 (especially with respect to the aromatic and phenolic composition) (Del Monaco *et al*,
75 2015; Bajoub *et al*, 2015). Up to now, no investigations have been carried out on the
76 chemical and sensory characterization of VOOs produced exclusively from *Brava* and
77 *Mansa* autochthonous Galician cultivars. Nevertheless, since the local Galician
78 producers traditionally co-crush different proportions of fruits from *Brava* and *Mansa*
79 cultivars, the aim of this work was to characterize VOOs produced by mixing fruits of
80 these varieties in different proportions, similar to the ones adopted by producers.
81 Towards this aim, chemical parameters and sensory analysis were first investigated to
82 classify olive oils according to EU Regulation 2568/91 and subsequent amendments.
83 Moreover, antioxidant compounds, together with aromas and coloured pigments were
84 determined, and their contribution in determining the functional value and the sensory
85 properties of oils was investigated. This characterization allows to developing new
86 VOOs by mixing these new varieties with other high-yield varieties with two main
87 purposes: the search for new antioxidant-rich and flavour-rich oils for dressings. These
88 are the two major trends driving the market sells today, since the consumers demand
89 more healthy dishes and new sensorial sensations. Food chemists have the goal to
90 satisfy consumers’ demands and help food companies to increase the level of sales with
91 this kind of innovations regarding olive oil-derived products and dishes.

92

93 2. Materials and Methods

94 2.1. Olive oil samples

95 Olives were harvested in the 2013/2014 crop season (specifically, between November
96 2013 and January 2014) in a cultivation area under organic agricultural practices located
97 between two municipalities, Ribas do Sil (42° 27'59.8'' N 7° 17'15.8''W) and Quiroga
98 (42° 29'04.8'' N 7° 12'33.4''W) in the Lugo province (NW Spain). **Table 1** shows the
99 climatic conditions for the study area over the crop year 2013.

100 Four VOOs (coded as VOO1, VOO2, VOO3, VOO4) were produced by co-crushing of
101 different proportions of two varieties known by the local producers as *Brava* and
102 *Mansa*. Neither of the autochthonous varieties are included in the database of the World
103 Olive Germplasm Bank of Córdoba, Spain (WOGBC), which is one of the world's
104 largest collections of olive germplasm (Trujillo *et al*, 2014).

105 It should be noted that obtaining monovarietal oils of *Brava* and *Mansa* at a semi-
106 industrial scale in this area is economically unprofitable due to the low production. In
107 particular, VOO1 and VOO2, crushed on 27 November and 9 December, respectively,
108 were the result of mixing 70% *Brava* and 30% *Mansa*. On the other hand, VOO3 and
109 VOO4 were both obtained by co-crushing 90-100% *Mansa* and 0-10% *Brava* olives, but
110 on 9 and 17 January, respectively. To summarize data from oils obtained by the same
111 percentages of the two varieties, despite their different maturation indexes, all analyses
112 individually performed on VOO1 and VOO2 were averaged, resulting in VOOa (n=2).
113 The same was performed with the other oils (VOO3 and VOO4), resulting in VOOb
114 (n=2). These percentages are among the most common that it is possible to find in the
115 cultivation area under study.

116 To suppress variability due to the extraction procedure, oils were obtained under
117 identical conditions at a semi-industrial scale in a local two phase mill. Oils were
118 allowed to settle and racked several times for about 4 months before sampling, since
119 this is the procedure typically used by local producers before marketing their oil. Three
120 replicates of each of the four oils were sampled and analyzed. Once in the laboratory,
121 samples were kept at a constant temperature of $10\pm 2^{\circ}\text{C}$ in amber bottles without
122 headspace until analysis.

123

124 **2.2. Analytical plan**

125 *2.2.1. Chemical and sensory parameters for classification of olive oils in different* 126 *commercial classes*

127 In conjunction with sensory analysis, the following chemical parameters, valid for
128 establishing quality and authenticity criteria of VOOs, were determined by using the
129 analytical methods proposed in the different Annexes of [EU Regulation 2568/91 and](#)
130 [subsequent amendments](#). The acceptable values for these parameters in olive oil are
131 regulated by the European Union ([EU Regulation 1348/2013](#)).

132 **Free acidity (FA).** FA, expressed as a percentage of oleic acid, was determined by a
133 simple acid-base titration with 0.1 M KOH of free fatty acids in an oil sample
134 previously dissolved in ethanol/ether 2:1 (v/v) ([EU Regulation 702/2007](#)).

135 **Peroxide value (PV).** PV, expressed as milliequivalents of active oxygen per kg of oil
136 (meq O_2/kg) was determined by titration with 0.01 N $\text{Na}_2\text{S}_2\text{O}_3$ of an oil sample
137 previously dissolved in acetic acid/chloroform 3:2 (v/v) in the presence of KI ([EU](#)
138 [Regulation 2568/91](#)).

139 **Specific UV extinction coefficients (K_{232} and K_{270}).** K_{232} and K_{270} extinction
140 coefficients were calculated from absorption values at 232 and 270 nm, respectively,
141 and measured with an UV-Vis 1800 spectrophotometer (Shimadzu Co., Kyoto, Japan),
142 previously calibrated with an optical glass filter containing holmium oxide (Type 666-
143 F1, Hellma GmbH & Co., Müllheim, Germany) by analyzing a 1% solution of the oil in
144 cyclohexane and quartz cuvette with a path length of 1 cm ([EU Regulation 299/2013](#)).

145 **Sensory analysis.** Sensory evaluation of olive oil samples was carried out by nine fully
146 trained judges of the panel of the Department of Food Science (University of Bologna,
147 Italy) recognized in 2006 by the Italian Ministry (MIPAAF, Ministry of Agricultural
148 Policies, Food and Forestry), and in accordance with the official method of International
149 Olive Oil Council ([IOC/T.20/Doc.N°15/Rev.7/2015](#)) within the framework of [EU](#)
150 [Regulations 1348/2013](#).

151 The trained tasters evaluated positive gustatory (bitter), olfactory-gustatory (fruity), and
152 tactile/kinaesthetic attributes (pungent), as well as negative attributes, namely
153 fusty/muddy sediment, musty-humid-earthly, winey-vinegary-acid-sour, frostbitten
154 olives, rancid, and others (heated, burnt, hay-wood, rough, greasy, vegetable water,
155 brine, metallic, esparto, grubby, cucumber). In addition, the tasters had the possibility to
156 describe each oil with positive olfactory descriptors (i.e. greenly fruity, ripely fruity, red
157 fruits, exotic fruits, apple, almond, grass, green-leaf, floral, aromatic herbs, thistle,
158 tomato), according to the list of descriptors reported in [IOC/T.20/Doc. N° 22 November](#)
159 [2005](#). All the attributes were assessed on an oriented 10 cm linear scale and their
160 intensities were quantified by measuring the location (in cm) of the mark from the
161 origin of the scale. The data obtained for the five descriptors that finally characterized
162 the selected oils (*viz.* fruity, bitter, pungent, green-leaf, ripely fruity) were used to define
163 the sensory profile of each sample.

164 **Total sterols, erythrodiol, and uvaol.** The sterol fraction was separated by preparative
165 chromatography. Next, a silanization reaction was performed to analyze the sterol
166 composition by gas chromatography-flame ionization detection (GC-FID) (EU
167 Regulation 1348/2013).

168 **Δ ECN42.** The percentage difference between the theoretical and actual content of
169 triglycerides with ECN42 (ECN42 is corresponded with trilinolein) provides the
170 Δ ECN42 parameter (EU Regulation 2568/91). Equivalent Chain Number ECN= CN-2n,
171 where CN is the actual acylcarbon number and *n* the number of double bonds of fatty
172 acids constituting the triacylglycerols.

173 **Waxes.** Waxes were determined by GC-FID after passing the oil sample through a
174 preconditioned packed silica gel column with *n*-hexane and then eluted with *n*-
175 hexane/ethyl ether 99:1 (v/v) (EU Regulation 2568/91).

176 **Tocopherols.** Their content was determined according to the IUPAC 2.432 method. A
177 sample of 1.5 g of VOO was dissolved in 10 mL hexane and injected into the HPLC
178 system. Tocopherols were quantified by external α -, β -, γ - and δ - tocopherol standards
179 from Sigma Chemical Co.

180 **Squalene.** Squalene was obtained by adsorption chromatography following the method
181 described by Lanzón, Albi, Cert and Gracián (1994). GC was performed by using an
182 Agilent 6890A chromatograph equipped with a cold on-column injector with oven-track
183 system and a flame-ionization detector. Concentration of squalene was obtained
184 comparing the total area and the squalene internal standard area.

185 **Pigment content.** Carotenoids and chlorophylls were determined as described by
186 Minguéz-Mosquera, Rejano-Navarro, Gandul-Rojas, Sanchez-Gomez, and Garrido-
187 Fernandez (1991). 3 g of VOO was accurately weighed and dissolved in cyclohexane up

188 to a final volume of 10 mL; the carotenoid and chlorophyll content were determined by
189 measuring the absorbance at 470 and 670 nm, respectively. The results were expressed
190 as mg kg^{-1} and calculated using the following equations:

$$191 \quad [\text{Chlorophylls}] \text{ mg kg}^{-1} = (A_{670} \times 10^6) / (E_1 \times 100 \times d)$$

$$192 \quad [\text{Carotenoids}] \text{ mg kg}^{-1} = (A_{470} \times 10^6) / (E_2 \times 100 \times d)$$

193 where A is the absorbance, d is the spectrophotometer cell thickness (1 cm) and E_1 and
194 E_2 are, respectively, the values of the specific extinction coefficients: $E_1 = 613$ for
195 pheophytin (as major component in the olive oil chlorophyll fraction), and $E_2 = 2000$ for
196 lutein (as major component in the olive oil carotenoid fraction).

197

198 **2.2.2. Phenolic compounds**

199 **Determination of phenolic compounds by HPLC.** Phenolic compounds in the VOOs
200 samples were also determined following the procedure developed by Mateos *et al*
201 (2001). Briefly, a solution of the internal standard (250 mL of 15 mg/kg of syringic acid
202 in methanol) was added to a sample of VOO (2.5 g) and the solvent was evaporated
203 with a rotary evaporator at 35 °C under vacuum. The oil was then dissolved in 6 mL of
204 hexane and a diol-bonded phase cartridge (Supelco Co., Bellefonte, USA) was used to
205 extract the phenolic fraction. Phenols were eluted with methanol (15 mL) and the
206 solvent was removed with a rotary evaporator at 35 °C under vacuum until dryness. The
207 phenolic residue was dissolved in methanol/water (1:1 v/v; 250 μL) and analysed by
208 HPLC using an Agilent Technologies 1100 series system equipped with an automatic
209 injector, a column oven and a diode array UV detector. A ZORBAX SB-C18 column
210 (250 x 4.6 i.d. mm, 5 μm particle size) (Agilent Technologies, USA) was used,
211 maintained at 30 °C, with an injection volume of 20 μL and a flow rate of 1.0 mL/min.

212 Mobile phase was a mixture of water/acetic acid (95:5 v/v) (solvent A), methanol (B)
213 and acetonitrile (C): from 95% (A) – 2.5% (B) – 2.5% (C) to 34% (A) – 33% (B) – 33%
214 (C) in 50 min. Phenolic compounds were quantified at 280 nm using syringic acid as
215 internal standard.

216 ***Determination of phenolic compounds by spectrophotometric assays***

217 ***Extraction of phenolic compounds***

218 The phenolic fraction was extracted from olive oils following the International Olive
219 Council method (IOC/T.20/Doc N° 29), with some modifications. Briefly, 2.0 g of olive
220 oil were weighed in a 15 mL screw-cap tube, 6 mL of MeOH/H₂O mixture (80/20, v/v)
221 was added and shaken for 1min. The extraction was performed using an ultrasonic bath
222 for 15 min at room temperature. The tube was then centrifuged at 4000 rpm for 25min.
223 The MeOH/H₂O phase was collected and was filtered through a 0.2 µm filter with a
224 nylon membrane.

225 ***Determination of the total phenolic content using the Folin-Ciocalteu (FC) assay.*** FC
226 was applied according to Mateos, Espartero, Trujillo and Rios (2001). A volume of 0.1
227 mL of the phenolic extract was added to 0.5 mL FC reagent and 2 mL of Na₂CO₃ (15%
228 w/v) in a 10 mL volumetric flask reaching the final volume with purified water. Each
229 sample was stored for at least 2h at room temperature before reading the absorbance;
230 total phenolic compounds were detected at 750 nm and quantified using both gallic acid
231 (GA) ($r^2=0.9970$) and hydroxytyrosol (HTyr) ($r^2=0.999$) calibration curves. Data were
232 expressed as mg GA/g of oil and mg HTyr/g of oil, respectively.

233 ***Determination of total o-diphenolic content.*** An aliquot of 4 mL of a solution prepared
234 by mixing 0.5 mL of phenolic extract and 5 mL of MeOH/H₂O (1:1 v/v) were added to
235 1 mL of a 5% solution of sodium molybdate dihydrate in EtOH/H₂O (1:1 v/v) and

236 vortexed for 1 min. After 10 min at room temperature, the mixture was centrifuged for 5
237 min at 3000 rpm; *o*-diphenolic compounds were detected at 370 nm and quantified
238 using GA calibration curves ($r^2=0.9985$). Data were expressed as mg GA/g of oil (Cert,
239 [Romero and Cert, 2007](#)).

240 **2.2.3. Volatile compounds**

241 The analysis of volatile compounds was performed by automated SPME extraction
242 coupled with GC/qMS. SPME fibers were coated with a
243 divinylbenzene/carboxen/polydimethylsiloxane (DVB/CAR/PDMS) phase (50/30 μm , 2
244 cm long from Supelco Ltd., Bellefonte, PA, USA). In particular, a GCMS-QP2010
245 Ultra (Shimadzu) provided with an auto sampler AOC-5000 plus (Shimadzu) was used.
246 For preparation of the sample, 4-methyl-2-pentanone (Fluka, New York, USA)
247 dissolved in refined peanut oil (0.1 g at a concentration of 5 ppm) was added as an
248 internal standard to 1 g of oil. After pre-incubation (2 min), volatile compounds were
249 extracted for 30 min at 40 °C, and then desorbed in the injector for 5 min at 240 °C.

250 Desorption time was 5 min. Analytes were separated on a ZB WAX column 30 m, 0.25
251 mm i.d., 1.00 μm f.t. (Phenomenex, Torrence, CA, USA) coated with polyethylene
252 glycol phase. The carrier gas, helium, was circulated at 1 mL/min in constant flow
253 mode. A split/splitless injector in the split mode was used (split ratio, 1:10). Column
254 temperature was as follows: 40°C for 10 min; 3°C/min ramp to 200°C (held for 3 min)
255 and 10°C/min ramp to 240°C (held for 5 min). The ion source and the transfer line were
256 set at 200°C and 240°C, respectively. Electron impact mass spectra were recorded at 70
257 eV ionization energy in the 30-250 amu mass range.

258 Volatile identification was tentatively obtained by comparison of mass spectral data
259 with information from the NIST library (2005 version), MS literature data, and
260 available standards. A calibration curve was made using mixtures prepared by

261 dissolving the internal standard in refined peanut oil at different concentrations; volatile
262 compounds were quantified and expressed as μg of internal standard per kg of oil.

263

264 **2.3. Statistical analysis**

265 All chemical analyses were performed in triplicate. Data were subjected to Student's *t*-
266 test using the statistical software package Statgraphics Centurion XV for Windows v.
267 15.2.06 from Statistical Graphics Corp. (Herndon, VA, USA).

268

269 **3. Results and discussion**

270 **3.1. Classification of olive oils according to European Regulations**

271 [EU Regulation 2568/91](#) and subsequent amendments establishes limit values for several
272 chemical parameters, in conjunction with organoleptic analysis (Panel test) of olive oils,
273 with the purpose of assessing their quality, checking their genuineness, and classifying
274 them within the different categories established by the European Commission ([EU](#)
275 [Regulation 1348/2013](#)).

276

277 ***3.1.1. Quality-related parameters***

278 Olive oils are classified by the European law ([EU Regulation 1348/2013](#)) according to
279 quality, which is established on the basis of set parameters that depend on the quality of
280 olives, accuracy in milling technologies, and mode of preservation of the oils.

281 **FA:** FA percentage is a direct measure of the quality of the oil and represents the extent
282 of hydrolytic activities reflecting the care taken from blossoming and fruit set to
283 eventual sale and consumption of the oil. Both olive oil samples had FA values that
284 were much lower than 0.8% (**Table 2**), which is the maximum value established by EU
285 Regulation for extra virgin olive oils (EVOOs).

286 **PV:** Peroxides are the primary products of oxidation of olive oil. They may adversely
287 affect the nutritional value of the oil and originate secondary oxidation products that
288 also have an unpleasant flavour. A limit value of 20 meqO₂/kg has been established for
289 VOOs in EU Regulation. The PVs for all olive oils were under this limit (**Table 2**).

290 **Absorbance in ultraviolet region:** Specific absorbances are measured in the region at
291 the wavelengths corresponding to the maximum absorption of the conjugated dienes and
292 trienes, at 232 and 270 nm, respectively. They are formed during autoxidation from the

293 hydroperoxides of unsaturated fatty acids and their fragmentation products. For both
294 oils, K_{232} and K_{270} values were lower than the limits established for EVOOs (2.50 and
295 0.22, respectively) by [EU Regulation 299/2013](#) (**Table 2**).

296 **Sensory evaluation:** Sensory analysis is an important tool for classifying oils obtained
297 from olives in different commercial categories (extra virgin, virgin, lampante) and to
298 differentiate them within the same category of extra virgin. Sensory analysis of the two
299 *Local* olive oils was performed by defining the main attributes, both positive (such as
300 fruity, bitter, and pungent) and negative (fusty/muddysediment, musty-humid-earthly,
301 winey-vinegary-acid-sour, metallic, and rancid). The intensity of the attribute values
302 perceived by tasters, expressed as cm, were statistically assessed to calculate the median
303 of each positive and negative sensory characteristic, as seen in **Table 3**. According to
304 the results, VOOa can be classified as “extra virgin” due to the median of defects equal
305 to 0 and the median of the fruity attribute more than 0. On the contrary, VOOb must be
306 classified as “virgin” because the defect of fusty/muddy sediment (the most perceived
307 one) was found with a median above 0, but not higher than 3.5 ([EU Regulation](#)
308 [1348/2013](#)). The intensity of both defects found -fusty/muddy sediment and winey- was
309 low, namely 1.0 and 0.8, respectively. Fusty/muddy sediment is the characteristic flavor
310 of oils obtained from olives in an advanced stage of anaerobic fermentation or of oils
311 which have been left in contact with the sediment that settles in underground tanks and
312 vats, which has also undergone a process of anaerobic fermentation. On the other hand,
313 winey-vinegary is a sensory note due to high concentrations of acetic acid, ethyl acetate,
314 and ethanol ([EU Regulation 1348/2013](#)). It seems that olives of the cultivars *Mansa* (90-
315 100%) and *Brava* (0-10%) harvested in January had undergone some type of
316 preservation under inadequate conditions before olive oil processing, even if they were
317 processed on the same day of harvest.

318 According to the intensity of perceptions, as specified by EU legislation ([EU Regulation](#)
319 [1348/2013](#)), both oils were “well-balanced”. Nevertheless, there was a clear decline in
320 the positive attributes of VOOa with respect to VOOb, as confirmed by the lower values
321 in fruity, bitter, and pungent seen in VOOb compared with VOOa (**Table 3**). Actually,
322 VOOb was characterized by a low intensity for all their positive attributes (i.e. the
323 median of attributes is less than 3). Its score was close to the “mild” olive oils (for
324 which the median of the bitter and pungent attributes is 2 or less) ([EU](#)
325 [Regulation1348/2013](#)).

326

327 ***3.1.2. Olive Oil genuineness-related parameters***

328 The following parameters established by the [EU Regulation 2568/91](#) and subsequent
329 amendments help to guarantee the authenticity of the oil.

330 ***Fatty acid composition:*** Fatty acid composition of olive oils is affected by several
331 factors including production area, latitude, climate, and fruit ripeness, but is dependent
332 primarily on genetic factors. Fatty acid composition of the oils are shown in **Table 2**
333 and fell within the recommended ranges for EVOOs set by [EU Regulation 1348/2013](#).
334 As can be seen, there are no statistically significant differences between oils, except for
335 the minor *trans*-oleic isomers (C18:1 T), behenic (C22:0), and margaroleic (C17:1)
336 acids. As expected, the major fatty acid was oleic acid (C18:1; 70.77% on average),
337 followed by palmitic acid (C16:0, 13.49% on average), linoleic acid (C18:2; 10.34% on
338 average), stearic acid (C18:0; 2.03% on average), palmitoleic acid (C16:1, 1.19% on
339 average), and linolenic acid (C18:3; 1.00% on average). Finally, *trans*-oleic isomers
340 were lower than 0.05 and the sum of *trans*-linoleic and *trans*-linoleic isomers were not
341 detected. Saturated fatty acids comprised about 16% of the total fatty acids, whereas

342 monounsaturated and polyunsaturated fatty acids represented 72% and 11%,
343 respectively.

344 Several studies performed in Spain on cultivar collection have concluded that
345 Arbequina oils are characterized by a high C18:2 (12.57-13.06%) content and a low
346 C18:1 content (68.20-65.83%), whereas Picual oil shows an opposite trend (4.43-4.44%
347 and 78.28-78.34%, respectively) (Uceda, Beltrán and Jiménez, 2005). The main fatty
348 acid composition of Galician *Local* oils (i.e. C16:0, C18:0, C18:1 and C18:2) was
349 intermediate between the above mentioned varieties. It is also worth noting that there
350 was a high concentration of linolenic acid, near 1%. Some genuine Moroccan olive oils
351 can be also characterized by 1% of this acid (Angerosa, Campestre and Giasante, 2006).

352 ***Sterol and triterpene dialcohol composition:*** Sterols are present in the unsaponifiable
353 fraction of olive oil, reaching between 20-23% of its total amount in olive oil. The sterol
354 profile is highly species-specific (EU Regulation 1348/2013). **Table 2** shows the sterol
355 content of the samples analyzed. The predominant sterol in olive oils is β -sitosterol
356 (sum of β -sitosterol and some of its degradation products such as $\Delta^{5,23}$ - and $\Delta^{5,24}$ -
357 stigmastadienols, cholesterol, sitostanol and Δ^5 -avenasterol) followed by campesterol.
358 All levels complied with the legislation established for EVOOs (EU Regulation
359 1348/2013). Statistically significant differences were detected for all individual sterols
360 between the two samples, except for Δ^7 -stigmastenol. However, no difference was
361 found in total sterol content. The sum of the triterpene diols could not differentiate the
362 oils, although none had levels above the established limit (4.5%) for EVOOs.

363 ***Δ ECN42:*** Δ ECN42 is a very useful and effective tool in detecting adulteration of olive
364 oils (naturally lacking in linolenic acid) with other oils rich in linolenic acid (seed oils).
365 In both *Local* olive oils, Δ ECN42 was within the range considered appropriate for
366 EVOOs and listed in EU Regulation 796/2002.

367 **Wax content:** The determination of the sum of C₄₀₋₄₆ aliphatic waxes can be considered
368 to be a reliable parameter to detect olive-residue oil (e.g. pomace olive oil) in VOO; in
369 this case, their sum was lower than 150 mg/kg, the maximum limit established for
370 EVOOs (EU Regulation 183/1993).

371 **Tocopherols and tocotrienols:** Tocopherols are the main lipid-soluble antioxidants
372 present in olive oil. Four isomers can be found in olive oil (α , β , γ and δ). α -tocopherol
373 (vitamin E) is the most abundant (90-95%). The concentration of these compounds in
374 olive oil range from 150 to 250 mg/Kg (Amelio, 2003). **Table 2** shows the content of
375 the four tocopherol isomers determined in the present oils, as well as, the total
376 tocopherol and tocotrienol content. As can be seen in such table, no significant
377 differences are observed in both cultivars.

378 **Squalene:** Squalene exhibits antitumor activity against different cancer types (Rao,
379 Newqmark and Reddy, 1998; Smith, 2000; Murakoshi et al, 1992, Ohkuma, Otagiri,
380 Tanaka and Ikekawa, 1983) and may also be useful for the treatment of cardiovascular
381 diseases (Banks et al, 2004). In olive oil, squalene achieves a concentration of 700
382 mg/100 g (Newmark, 1999) with a range variation comprised between 90 and 870
383 mg/100g (De Leonardis, Maccionala and De Felice, 1998). Squalene levels varied from
384 7436 to 6518 mg/Kg for VOOa and VOOB, respectively (**Table 2**). VOOa was quite
385 rich in squalene content. Squalene content of *Local* olive oils was in the range as those
386 showed in Beltrán *et al* 2015. According to this paper, our oils can be included in the
387 category IV (high squalene content: 600-750 mg/100 g).

388 **Pigment content:** The evaluation of chlorophylls and carotenoids pigments profile in
389 olive oil is of great importance, given their antioxidant properties, as well as their
390 crucial role as colorful substances responsible of greenness (chlorophylls) and
391 yellowness (carotenoids) of olive oil (Giuliani, Cerretani, & Cichelli, 2011). As can be

392 seen in **Table 2**, significant differences in the levels of the analysed oils were observed
393 in the chlorophyll and in the β -carotene content. Chlorophyll levels varied from 22 to 15
394 mg/Kg for VOOa and VOOB, respectively. The content of β -carotene was quite similar
395 with levels of about 10 mg/Kg for both oils.

396

397 **3.2. Other quality-related parameters not included in current European** 398 **Regulations**

399 **3.2.1. Phenolic composition**

400 Phenols are considered natural antioxidants that are responsible for oil stability against
401 oxidation, and also contribute to the characteristic bitter, pungent, and astringent
402 sensory notes of olive oils. They have been shown to have a wide range of beneficial
403 effects from healing sunburn to lowering cholesterol, blood pressure, and risk of
404 coronary disease. VOOs contains five different classes of phenols clustered into
405 phenolic acids, simple phenols, complex oleuropein derivatives, flavonoids, lignans,
406 and hydroxy-isocromans (Servili, 2014).

407 **Table 4** shows the concentration of the individual phenolic compounds (mg/kg oil) of
408 the studied VOOa and VOOB. VOOa -obtained co-crushing 70% *Brava cv.* and 30%
409 *Mansa cv.*- shows 433 mg/kg oil and VOOB -obtained co-crushing 90-100% *Mansa cv.*
410 and 0-10% *Brava cv.*- shows 325 mg/kg oil. The content of hydroxytyrosol derivatives
411 for VOOa and VOOB oils (174 and 120 mg/kg, respectively) were lower than tyrosol
412 derivatives (257 and 201 mg/kg). The concentrations obtained were in accordance with
413 those previously determined (Reboredo-Rodríguez *et al*, 2015a, 2015b).

414 Moreover, **Table 4** shows the total *o*-diphenolic and phenolic content of the VOOs.
415 Total *o*-diphenolics are expressed as mg of GA/kg of oil, whereas total phenolics can be

416 expressed equally as mg of GA/kg of oil or mg of HTyr/kg of oil. VOO_b was
417 significantly less rich in phenols than VOO_a: they could be considered as low and
418 medium content oils, respectively, in accordance with the normal ranges established by
419 [Servili \(2014\)](#) for total phenols in VOOs (high content > 500 mg GA/kg oil; medium
420 content from 250 to 500 mg GA /kg oil; low content < 250 mg GA /kg oil).

421 It should be emphasized that the phenolic content was higher for VOO_a than for VOO_b,
422 which suggests that, in this investigation, the sample produced by co-crushing 70%
423 *Brava*/30% *Mansa* olives was an oil richer in these healthy compounds: this finding
424 could be due to the different varieties or to the earlier harvesting date of olives
425 processed to obtain the sample VOO_a.

426

427 ***3.2.3. Volatile composition***

428 C₅ and C₆ volatile compounds, responsible for the positive green sensory perceptions in
429 olive oil, were the main compounds identified in the Galician *Local* olive oils (**Table 5**).
430 C₆ compounds are mainly produced by endogenous olive enzymes through the
431 lipoxygenase (LOX) pathway, which consists of a cascade of oxidative reactions that
432 give rise to a variety of metabolites with different functions from polyunsaturated fatty
433 acids (either linoleic and linolenic acids as the initial substrates) ([Feussner and
434 Wasternack, 2002](#)). C₅ compounds are generated through an additional branch of the
435 LOX pathway that involves the production of a 13-alkoxyl radical by LOX. Many
436 factors influence the presence of these compounds: variety of olives, weather and
437 location of the orchard, care that went into growing, maturity of olives at harvest, oil
438 extraction, storage conditions, and age of the oil ([Kalua et al, 2007](#)).

439 ***C₆ compounds.*** The most abundant volatiles in all samples were those in the C₆ alcohol
440 fraction (primarily *cis*-3-hexen-1-ol), followed by aldehydes (*trans*-2-hexen-1-al was
441 the most abundant) and esters (*cis*-3-hexenyl acetate) (see **Table 5**). This is consistent
442 with the results of a previous study on oils obtained by olives of the same varieties,
443 where the process of sedimentation and further racking of oil samples for a minimum of
444 2 months was found to promote the formation of C₆ alcohols in most samples
445 ([Reboredo-Rodríguez, González-Barreiro, Cancho-Grande and Simal-Gándara,2013](#)).

446 VOOa, as a result of 70% *Brava* and 30% *Mansa* co-crushing, had a markedly higher
447 content in all relevant C₆ compounds than VOOB (0-10% *Brava* and 90-100% *Mansa*),
448 except for *trans*-3-hexen-1-ol and 1-hexanol.

449 ***C₅ compounds.*** C₅ compounds were detected at low levels compared with the C₆
450 volatile fraction (**Table 5**). Pentene dimers was the most abundant class, and 3-
451 pentanone (without significant differences between the two samples) followed by 1-
452 penten-3-ol (with statistically significant different concentrations for both oils) were the
453 major single components of the C₅ fraction. On the other hand, 1-penten-3-one and
454 *trans*-2-pentenal were minor compounds in both samples.

455

456 ***Minor volatile compounds***

457 Several terpenic hydrocarbons, such as 3-carene, limonene, and α -farnesene were found
458 (**Table 5**) in minor concentrations with respect C₆ and C₅ compounds. Carene was the
459 only compound present in both samples ranging from 11.0 to 20.5 $\mu\text{g IS/kg oil}$. Several
460 authors ([Vichi, Guadayol, Caixach, López-Tamames and Buxaderas, 2006](#)) consider
461 these compounds as genetic orgeographic markers of VOO.

462 Only ethylbenzene as a mono-ring aromatic hydrocarbon (MAH) was observed in
463 VOOa at low levels (9.6 µg IS/kg oil). Aromatic hydrocarbons in olive oils may arise
464 from both exogenous contamination and endogenous pathways ([Sabatini, Perriand](#)
465 [Marsilio, 2009](#)).

466 Other minor volatile compounds derived from autoxidation reactions and related to oil
467 rancidity, such as octane, nonanal, *trans*-2-heptenal, and *trans, trans*-2,4-hexadienal,
468 were also found in both samples.

469 Some products deriving from amino acid transformation were also detected. The highest
470 concentrations were 2-methyl-1-butanol and 3-methyl-1-butanol (221.8 and 319.3 µg
471 IS/kg oil in samples VOOa and VOOB, respectively), while 2-methyl-1-butanol, even if
472 present at a lower concentration (19.9 and 39.0 µg IS/kg oil in samples VOOa and
473 VOOB, respectively), exceeded its odor threshold (5.2 µg/kg oil). This latter compound
474 may contribute to the fusty defect ([Morales et al, 2005](#)) that was detected by tasters in
475 VOOB.

476 Ethyl acetate and acetic acid are metabolites normally associated with ethanol to
477 undesired fermentation occurring in the olive fruits before olive oil extraction. VOOB
478 showed higher levels of these compounds, and especially acetic acid (805.1 µg IS/kg
479 oil). Such a considerable amount of acetic acid, mainly originated by the fermentation
480 metabolism of *Acetobacter* in olives ([Angerosa, 2002](#)) and in amounts higher than its
481 sensory threshold, could explain the winey defect scored by judges. It is likely that
482 incorrect preservation of olives in terms of bad storage conditions was the cause of
483 alcoholic fermentation. Ethanol was also detected in olive oils: it should be noted the
484 low processing capacity of small mill plants located in this area may have produced a
485 conservation of the piling olives at room temperature for a few days before processing,
486 thus inducing yeast and bacteria growth that increased sugar fermentation and amino

487 acid degradation. Moreover, in the case of VOOB, 0-10% of olives of the cv. *Brava*
488 were probably over-ripe when collected (at a date after their optimal maturation stage),
489 thus promoting degradation.

490 The aroma perceived by smelling can rarely be ascribed to a single, specific compound,
491 but rather to a mixture of several volatile molecules. However, not all compounds
492 contribute to oil aroma to the same extent; in fact, the contribution of a given compound
493 depends, among other factors, on its odor threshold (Delahunty, Eyres, and Dufour,
494 2006). For this reason, the odor activity value (OAV), which represents the
495 concentration of a volatile compound divided by its odor threshold value, allows an
496 estimation of the contribution of each individual compounds to the oil overall aroma
497 (Reboredo-Rodríguez *et al*, 2015a). However, considering each single OAV to
498 tentatively establish the aromatic profile of an oil can give rise to a model complex and
499 difficult to interpret. Thus, volatile compounds with similar odor descriptors were
500 grouped into five odorant series (**Table 5**; *viz.* sweet odor notes, pungent odor notes,
501 bitter odor notes, green-leaf, and ripely fruit). The total OAV of each odorant series was
502 calculated by summing the OAV calculated for each volatile compound belonging to a
503 particular series. This procedure, borrowed from the characterization of white wines
504 (Sánchez-Palomo, Gómez García-Carpintero, Alonso-Villegas and González-Viñas,
505 2010), can be useful in finding a relation between the quantitative information obtained
506 by the chemical analysis of the volatiles and the sensory perceptions in the form of a
507 tentative aroma profile. It should be noted that some volatile compounds, such as 1-
508 penten-3-one, could contribute to the bitter and pungent sensations perceived, even if it
509 is well known that these two attributes are basically due, respectively, to stimulation of
510 gustatory cells and the trigeminal nerve by phenolic soluble and not volatile compounds,
511 such as secoiridoids (Angerosa *et al*, 2000).

512 **Figure 1** shows the contribution of each odorant series to the aroma profile of VOOa (a)
513 and VOOB (b) according to instrumental data (OAV calculation) and Panel results,
514 respectively. Taking into account only sensory evaluation, pungent and bitter were
515 highlighted as the main attributes in VOOa compared to VOOB, while sweet, ripely
516 fruity, and green leaf descriptors showed a similar organoleptic score in both oils.

517 As reflected in **Figure 1**, considering the results of the two samples characterized by
518 different chemical composition, a good relation was found between sensory (mean of
519 the attributes) and instrumental (OAV values) data. Actually, the intensities of the
520 perceived sensory attributes (Panel test results) were substantially in accord with the
521 instrumental data (OAV) and the Panel test results, except for "pungent odor notes"
522 series in the case of VOOB.

523 The pungent sensation perceived by the tasters is not mainly produced by volatile
524 substances grouped in **Table 5** as "pungent", but is instead caused by trigeminal/tactile
525 stimulation of the mucosa by secoiridoids. Volatiles grouped in **Table 5** as "pungent"
526 can reinforce this action but can not alone be predictive. In addition, acetic acid present
527 in VOOa and VOOB could help to boost the perception of "pungent odor notes". This
528 compound was not considered because only C₅ and C₆ compounds were used to
529 construct the odorant series (Reiners and Grosch, 1998), but if the contribution of acetic
530 acid was included in that series, the OAV for this series would be 18.3 and 1.8 for VOOa
531 and VOOB, increasing by 0.6 and 1.6 units, respectively. Consequently both criteria,
532 sensory and instrumental data, are better suited. This synergistic effect of acetic acid in
533 the perception of "pungent odor notes" that here has been hypothesized requires,
534 however, additional tests to be confirmed.

535

536 4. Conclusions

537 The chemical and sensory characterization of VOOs obtained from different proportions
538 of autochthonous varieties from a new Spanish olive-growing zone has been
539 investigated herein. The fatty acid composition of the Galician *Local* oils (i.e. C16:0,
540 C18:0, C18:1 and C18:2) was intermediate between usual Arbequina and Picual oils.
541 The Galician *Local* oils showed a medium content in total phenolic compounds, but
542 samples produced with higher proportion of *Brava* variety had more abundant phenolic
543 content than those obtained using a high percentage of *Mansa*. In general, given the
544 high levels of phenolic compounds, tocopherols and carotenoids, these are oils with
545 long stability, especially under dark storage conditions, because stability is reinforced
546 with the contribution of chlorophylls.

547 Concerning volatile compounds, the most abundant volatiles in all samples were those
548 in the C₆ alcohol fraction, typical of good quality products. Oils obtained by co-crushing
549 70% *Brava* and 30% *Mansa* olives had a markedly higher content in all relevant C₆
550 compounds than oils with 90-100% of *Mansa* olives, except for *trans*-2-hexen-1-ol.

551 Pungent and bitter were highlighted as the main sensory attributes in oils with a higher
552 percentage of *Brava* cv., whereas oils with a higher percentage of *Mansa* cv. were
553 characterized by a low intensity for all positive attributes. Moreover, the sensory quality
554 of the oil obtained with the highest percentage of *Mansa* cv. was penalized by
555 decreasing its commercial category to VOO because of the presence of fusty/muddy
556 sediment and winey defects, as confirmed by the presence of some volatile compounds
557 deriving from amino acid transformation (2-methyl-1-butanol and 3-methyl-1-butanol)
558 and acetic acid. This is probably due to the fact that these oils are traditionally produced
559 in local, small size mills by non-expert producers who are not yet fully aware of the
560 optimal procedures to store olives before processing; of course, with the aim to improve

561 the quality of the product, better conditions in the preservation of olives, providing air
562 circulation and avoiding accumulation, should be pointed out to local producers in the
563 future. Moreover, it is also important to take into account the optimal degree of
564 maturation of olives when harvesting with the aim to reach the quality standard of
565 EVOOs.

566

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572

573 **Figure legends**

574 **Figure 1.** Sensory profile and OAV values calculated for the odorant series (see **Table**
575 **5**) of VOOs. For this calculation, the sums of the OAV evaluated for specific
576 compounds were made: **sweet odor notes series:** hexanal, hexyl acetate, *cis*-2-penten-1-
577 ol, 3-pentanone; **pungent odor notes series:** 1-penten-3-ol, 1-penten-3-one, 1-pentanol;
578 **bitter odor notes series:** *trans*-2-hexen-1-al, *trans*-3-hexen-1-ol, 1-penten-3-one,
579 pentanal, *trans*-2-pentenal; **green-leaf series:** *cis*-3-hexen-1-ol, *trans*-2-hexen-1-ol, 1-
580 penten-3-ol, 1-penten-3-one; **riply fruit series:** *cis*-3-hexenyl acetate, 1-hexanol, hexyl
581 acetate, 1-penten-3-ol, *cis*-2-penten-1-ol, 1-pentanol, 3-pentanone, as reported in **Table**
582 **5.**

583

584 **References**

- 585 AICA. Agencia de Información y Control de Alimentos. (2015). Spanish olive
586 cultivation.
587 [http://aplicaciones.magrama.es/pwAgenciaAO/OliverEspanol.aao?opcion_seleccionada](http://aplicaciones.magrama.es/pwAgenciaAO/OliverEspanol.aao?opcion_seleccionada=2100&control_acceso=SâPagina=2101&idioma=ING)
588 [=2100&control_acceso=SâPagina=2101&idioma=ING](http://aplicaciones.magrama.es/pwAgenciaAO/OliverEspanol.aao?opcion_seleccionada=2100&control_acceso=SâPagina=2101&idioma=ING)). Accessed 25.03.2015.
- 589 **Amelio, M. (2003). Chemical-physical characteristics of olive oils. Technical course for**
590 **olive oil testers. Organizzazione Nazionale Assaggiatori Olio di Oliva.**
- 591 Angerosa, F. (2002). Influence of volatile compounds on virgin olive oil quality
592 evaluated by analytical approaches and sensor panels. *European Journal of Lipid*
593 *Science and Technology*, 104, 639-660.
- 594 Angerosa, F., Campestre, C., & Giasante, L. (2006). Analysis and Authentication. In
595 *Olive oil: Chemistry and technology*. D. Boskou (Ed.), American Oil Chemists' Society,
596 pp. 113-172.
- 597 Angerosa, F., Mostallino, R., Basti, C., & Vito, R. (2000). Virgin olive oil odour notes:
598 Their relationships with volatile compounds from the lipoxygenase pathway and
599 secoiridoid compounds. *Food Chemistry*, 68(3), 283–287.
- 600 Aparicio, R., & Luna, G. (2002). Characterisation of monovarietal virgin olive oils.
601 *European Journal of Lipid Science and Technology*, 104, 614-627.
- 602 Aparicio, R., & Morales, M. T. (1998). Characterization of olive ripeness by green
603 aroma compounds of virgin olive oil. *Journal of Agricultural and Food Chemistry*, 46,
604 1116-1122.
- 605 Bajoub, A., Hurtado-Fernández, E., Ajal, E. A., Fernández-Gutiérrez, A., Carrasco-
606 Pancorbo, A., & Ouazzani, N. (2015). Quality and chemical profiles of monovarietal
607 north Moroccan olive oils from "picholine Marocaine" cultivar: Registration database
608 development and geographical discrimination. *Food Chemistry*, 179, 127-136
- 609 **Banks, W. A., Coon, A. B., Robinson, S. M., Moinuddin, A., et al. (2004) Tryglycerides**
610 **induce leptin resistance at the blood brain barrier. *Diabetes*, 53, 1253-1260.**
- 611 **Beltrán, G., Bucheli, M.E, Aguilera, M.P., Belaj, A., Jimenez, A. (2015). Squalene in**
612 **virgin olive oil: Screening of variability in olive cultivars. *European Journal of Lipid***
613 ***Science and Technology*, in press.**
- 614 Burdock, G. A. *Fenaroli's Handbook of Flavor Ingredients*, 4th ed.; CRC Press: Boca
615 Raton, FL, 2002.
- 616 Cayuela, J.A., Gómez-Coca, R.B., Moreda, W., & Pérez-Camino, M.C. (2015). Sensory
617 defects of virgin olive oil from a microbiological perspective. *Trends in Food Science*
618 *and Technology*, 43(2),227-235.
- 619 Cert., A., Romero, A., & Cert, R. (Revised Method: December 2007). Colorimetric
620 method for the determination of *o*-diphenolic compounds in olive oils. International

- 621 Olive Council, URL ([http://www.internationaloliveoil.org/documents/viewfile/3857-](http://www.internationaloliveoil.org/documents/viewfile/3857-testing12eng)
622 [testing12eng](http://www.internationaloliveoil.org/documents/viewfile/3857-testing12eng)).
- 623 Del Monaco, G., Officioso, A., D'Angelo, S., La Cara, F., Ionata, E., Marcolongo, L.,
624 Squillaci, G., Maurelli, L., & Morana, A. (2015). Characterization of extra virgin olive
625 oils produced with typical Italian varieties by their phenolic profile. *Food Chemistry*,
626 *184(1)*, 220-228.
- 627 Delahunty, C. M., Eyres, G., & Dufour, J. P. (2006). Gas chromatography-olfactometry.
628 *Journal of Separation Science*, *29*, 2107-2125.
- 629 De Leonardis, A., Maccionala, V., De Felice, M. (1998). Rapid determination of
630 squalene in virgin olive oils using gas-liquid chromatography. *Italian Journal of Food*
631 *Science*, *10*, 75-80.
- 632 EU Regulation 1348/2013 of 16 December 2013 amending Commission Regulation
633 (EEC) No 2568/91 on the characteristics of olive oil and olive-residue oil and on the
634 relevant methods of analysis.
- 635 EU Regulation 183/1993 of 29 January 1993 amending Commission Regulation (EEC)
636 No 2568/91 on the characteristics of olive oil and olive-residue oil and on the relevant
637 methods of analysis.
- 638 EU Regulation 2568/1991 of 11 July 1991 on the characteristics of olive oil and olive-
639 residue oil and on the relevant methods of analysis.
- 640 EU Regulation 299/2013 of 26 March 2013 amending Commission Regulation (EEC)
641 No 2568/91 on the characteristics of olive oil and olive-residue oil and on the relevant
642 methods of analysis.
- 643 EU Regulation 432/2012 of 16 May 2012 establishing a list of 199 permitted health
644 claims made on foods, other than those referring to the 200 reduction of disease risk and
645 to children's development and health.
- 646 EU Regulation 702/2007 of 21 June 2007 amending Commission Regulation (EEC) No
647 2568/91 on the characteristics of olive oil and olive-residue oil and on the relevant
648 methods of analysis.
- 649 EU Regulation 796/2002 of 6 May 2002 amending Commission Regulation (EEC) No
650 2568/91 on the characteristics of olive oil and olive-pomace oil and on the relevant
651 methods of analysis and the additional notes in the Annex to Council Regulation (EEC)
652 No 2658/87 on the tariff and statistical nomenclature and on the Common Customs
653 Tariff.
- 654 Feussner, I., & Wasternack, C. (2002). The lipoxygenase pathway. *Annual Review of*
655 *Plant Biology*, *53*, 275-297.
- 656 Giuliani, A., Cerretani, L., & Cichelli, A. (2011). Chlorophylls in olive and in olive oil:
657 Chemistry and occurrences. *Critical Reviews in Food Science and Nutrition*, *51(7)*, 678-
658 690.

- 659 IOC/T.20/Doc. N° 15/Rev. 7. (2015). Sensory analysis of olive oil. Method for the
660 organoleptic assessment of virgin olive oil.
- 661 IOC/T.20/Doc. N° 22. (2005). Method for the organoleptic assessment of extra virgin
662 olive oil applying to use a designation of origin
- 663 IOC/T.20/Doc. N° 29. (2009). Determination of biophenols in olive oils by HPLC.
- 664 UPAC 1992. Determination of tocopherols and tocotrienols in vegetable fats by HPLC.
665 Method no. 2432. In *Standard Methods of Analyses of Oils, Fats and Derivatives*.
666 Blackwell, Oxford.
- 667 Kalua, C.M., Allen, M.S., Bedgood Jr., D.R., Bishop, A.G., Prenzler, P.D., & Robards,
668 K. (2007). Olive oil volatile compounds, flavour development and quality: A critical
669 review. *Food Chemistry*, *100*(1), 273-286.
- 670 Lanzón, A., Albi, T., Cert, A., Gracián, J. (1994). The hydrocarbon fraction of virgin
671 olive oil and changes resulting from refining. *Journal of the American Oil Chemists'*
672 *Society*, *71*, 285–292.
- 673 Mateos, R., Espartero, J. L., Trujillo, M., & Rios, J. (2001). Determination of phenols,
674 flavones, and lignans in virgin olive oils by solid-phase extraction and high-
675 performance liquid chromatography with diode array ultraviolet detection. *Journal of*
676 *Agricultural and Food Chemistry*, *49*, 2185-2192.
- 677 Minguéz-Mosquera, M. I., Rejano-Navarro, L., Gandul-Rojas, B., Sanchez-Gomez, A.
678 H., & Garrido-Fernandez, J. (1991). Color-pigment correlation in virgin olive oil.
679 *Journal of the American Oil Chemists Society*, *68*(5), 332–336.
- 680 Morales, M. T., Aparicio-Ruíz, R., & Aparicio, R. (2013).
681 Chromatographic methodologies: compounds for olive oil odor issues. In *Handbook of*
682 *Olive Oil: Analysis and Properties, Second Edition*; Aparicio, R., Harwood, J., Eds.
683 Springer Science + Business Media New York, pp 261-309.
- 684 Morales, M. T., Luna, G., & Aparicio, R. (2005). Comparative study of virgin olive oil
685 sensory defects. *Food Chemistry*, *91*, 293-301.
- 686 Murakoshi, M., Nishino, H., Tokuda, H., Iwashima, A., Okuzumi, J., Kitano, H.,
687 Iwasaki, R. (1992). Inhibition by squalene of the tumor-promoting activity of 12-O-
688 tetradecanoylphorbol-13-acetate in mouse skin carcinogenesis. *International Journal of*
689 *Cancer*, *52*, 950-952.
- 690 Newmark, H. I. (1999). Squalene, olive oil, and cancer risk. Review and hypothesis.
691 *Annals of the New York Academy of Sciences*, *889*, 193–203.
- 692 Ohkuma, T., Otagiri, K., Tanaka, S., Ikekawa, T. (1983). Intensification of host's
693 immunity by squalene in sarcoma 180 bearing icr mice. *Journal of pharmacobio-*
694 *dynamics*, *6*, 148-151.

695 Psomiadou, E.; Tsimidou, M. (1998). Simultaneous HPLC determination of
696 tocopherols, carotenoids, and chlorophylls for monitoring their effect on virgin olive oil
697 oxidation. *Journal of Agricultural and Food Chemistry*, 46, 5132-5138.

698 Rao, C. V., Newqmark, H. L., Reddy, B. S. (1998). Chemopreventive effect of squalene
699 on colon cancer. *Carcinogenesis*, 19, 287-290.

700 Reboredo-Rodríguez, P., González-Barreiro, C., Cancho-Grande, B., Fregapane, G.,
701 Salvador, M. D., & Simal-Gándara, J. (2015a). Characterization of extra virgin olive
702 oils from Galician autochthonous varieties and their co-crushings with Arbequina and
703 Picual cv. *Food Chemistry*, 176(1), 493-503.

704 Reboredo-Rodríguez, P., González-Barreiro, C., Cancho-Grande, B., Fregapane, G.,
705 Salvador, M. D., & Simal-Gándara, J. (2015b). Blending *Local* olive oils with
706 Arbequina or Picual oils produces high quality, distinctive EVOOs. *European Journal*
707 *of Lipid Science and Technology*, 117(8), 1238-1247.

708 Reboredo-Rodríguez, P., González-Barreiro, C., Cancho-Grande, B., & Simal-Gándara,
709 J. (2013). Effects of sedimentation plus racking process in the extra virgin olive oil
710 aroma fingerprint obtained by DHS-TD/GC-MS. *Food and Bioprocess Technology*,
711 6(5), 1290-1301.

712 Reiners, J., & Grosch, W. (1998). Odorants of virgin olive oils with different flavor
713 profiles. *Journal of Agricultural and Food Chemistry*, 46, 2754-2763.

714 Sabatini, N., Perri, E., & Marsilio, V. (2009). An investigation on molecular partition of
715 aroma compounds in fruit matrix and brine medium of fermented table olives.
716 *Innovative Food Science and Emerging Technologies*, 10, 621-626.

717 Sánchez-Palomo, E., Gómez García-Carpintero, E., Alonso-Villegas, R., & González-
718 Viñas, M. A. (2010). Characterization of aroma compounds of Verdejo white wines
719 from the La Mancha region by odour activity values. *Flavour Fragrance Journal*, 25(6),
720 456-462.

721 Servili, M. (2014). The phenolic compounds: A commercial argument in the economic
722 war to come on the quality of olive oil? *OCL - Oilseeds and fats, crops and lipids*,
723 21(5), 8.

724 Smith, T. J. (2000). Squalene: Potential chempreventive agent. *Expert Opinion on Drug*
725 *Discovery* 9, 1841-1848.

726 Trujillo, I., Ojeda, M. A., Urdiroz, N. M., Potter, D., Barranco, D., Rallo, L., & Diez, C.
727 M. (2014). Identification of the Worldwide Olive Germplasm Bank of Córdoba (Spain)
728 using SSR and morphological markers. *Tree Genetics & Genome*, 10(1), 141-155.

729 Uceda M., Beltrán, G. & Jiménez, A. (2005). Composición del aceite (Banco de
730 Germoplasma de Córdoba). In *Las Variedades de Olivo Cultivadas en España, Libro II:*
731 *Variabilidad y Selección*, ed. by Rallo L., Barranco D., Caballero J., Martín A., Del Río

- 732 C., Tous J., et al. Junta de Andalucía/MAPA/Ediciones Mundi-Prensa, Madrid, pp. 365-
733 372.
- 734 Vichi, S., Guadayol, J.M., Caixach, J., López-Tamames, E., Buxaderas, S. (2006).
735 Monoterpene and sesquiterpene hydrocarbons of virgin olive oil by headspace solid-
736 phase microextraction coupled to gas chromatography/mass spectrometry. *Journal of*
737 *Chromatography A*, 1125(1), 117-123.

Table 1. Climatic conditions of the growing area in the crop year season 2013.

Climatic conditions					
Year	R (L/m ²)	T (°C)	TCT ₇ (days)	RH (%)	MBP (hPa)
2013	1044.5	13.3	6.8	76.9	987.9

R, total rainfall; **T**, mean air temperature; **TCT₇**, total cold time (T<7°C); **RH**, mean air relative humidity; **MBP**, mean barometric pressure.

Table 2. Quality and genuineness-related parameters of the studied VOOs.

Quality-related indices	VOOa	VOOb
<i>FA (% oleic acid)</i>	0.2±0.0 ^a	0.2±0.0 ^a
<i>PV (meq O₂/kg oil)</i>	9.2±0.7 ^a	10.4±0.6 ^b
<i>K₂₃₂</i>	2.01±0.08 ^b	1.89±0.09 ^a
<i>K₂₇₀</i>	0.15±0.01 ^a	0.14±0.04 ^a
<i>ΔK</i>	0.0020±0.0003 ^a	0.0013±0.0002 ^a

Genuineness-related indices	VOOa	VOOb
<u>Fatty acid composition (% m/m methyl esters)</u>		
Myristic C14:0	0.01±0.00 ^a	0.01±0.00 ^a
Palmitic C16:0	13.42±0.54 ^a	13.57±0.61 ^a
Palmitoleic C16:1	1.19±0.09 ^a	1.19±0.02 ^a
Margaric C17:0	0.11±0.02 ^a	0.13±0.01 ^a
Margaroleic C17:1	0.26±0.03 ^a	0.32±0.03 ^b
Stearic C18:0	2.07±0.18 ^a	1.99±0.13 ^a
Oleic C18:1	71.19±1.20 ^a	70.36±0.36 ^a
Linoleic C18:2	10.06±0.59 ^a	10.63±0.23 ^a
Linolenic C18:3	0.97±0.09 ^a	1.04±0.04 ^a
Arachidic C20:0	0.30±0.01 ^a	0.30±0.03 ^a
Eicosenoic C20:1	0.25±0.01 ^a	0.27±0.01 ^a
Behenic C22:0	0.08±0.01 ^a	0.12±0.02 ^b
Lignoceric C24:0	0.04±0.01 ^a	0.05±0.01 ^a
<i>trans</i> -Oleicisomers C18:1 T	0.02±0.00 ^a	0.04±0.00 ^b
<i>trans</i> -Linoleic + <i>trans</i> -Linolenic	n.d.	n.d.
<u>Sterol relative amounts (%)</u>		
Cholesterol	0.03±0.01 ^a	0.06±0.00 ^b
Brassicasterol	Tc	Tc
Campesterol	2.39±0.08 ^a	2.54±0.11 ^b
Stigmasterol	0.75±0.01 ^a	0.96±0.18 ^b
Apparent β-sitosterol	96.37±0.18 ^b	95.96±0.33 ^a
Δ ⁷ -Stigmasterol	0.08±0.01 ^a	0.11±0.02 ^a
Total sterols (μg/g)	1582.07±160.64 ^a	1686.08±190.51 ^a
Erythrodiol + Uvaol	1.12±0.12 ^a	1.22±0.19 ^a
<i>ΔECN42</i>	0.08±0.01 ^a	0.10±0.02 ^a
<i>Waxes</i> (μg/g)	32.32±4.35 ^a	39.77±6.14 ^a
<u>Tocopherols (mg/Kg)</u>		
<i>α-tocopherol</i>	160±8 ^a	157±8 ^a
<i>β-tocopherol</i>	2±0.1 ^a	2±0.1 ^a
<i>γ-tocopherol</i>	3±0.1 ^a	5±0.1 ^a
<i>δ-tocopherol</i>	<2±0.1 ^a	<2±0.1 ^a
<i>Total tocopherols</i>	165±8 ^a	164±8 ^a
<i>Squalene</i> (mg/Kg)	7436±372 ^b	6518±326 ^a
<i>Chlorophyll</i> (mg/Kg)	22±1.1 ^b	15±0.7 ^a
<i>β-carotene</i> (mg/Kg)	12±0.6 ^b	10±0.5 ^a

VOOa is the average of results obtained for VOO1 and VOO2, elaborated by co-crushing 70% *Brava* cv. and 30% *Mansa* cv. (n=2). VOOB is average of results obtained for VOO3 and VOO4, elaborated by co-crushing 90-100% *Mansa* cv. and 0-10% *Brava* cv. (n=2).

Values are mean \pm standard deviation. All analyses were performed in triplicate. Different letters within rows indicate significant differences by a Student's *t*-test ($p < 0.05$).

n.d.: not detected ($< \text{LOD}$).

Tc: Traces ($< \text{LOQ}$).

Table 3. Sensory analysis of VOOs.

Attributes	VOOa	VOOb
<i>Positive</i>		
Fruity	3.3 [3.6-3.0]	2.5 [2.9-2.1]
Bitter	4.7 [5.2-4.2]	2.5 [3.2-1.8]
Pungent	4.1 [4.6-3.7]	2.3 [2.8-1.8]
<i>Negative</i>		
Fusty/muddy sediment	n.p.	1.0 [1.2-0.8]
Winey-vinegary-acid-sour	n.p.	0.8 [1.3-0.4]
Panel test classification	EXTRA VIRGIN	VIRGIN

VOOa is the average of results obtained for VOO1 and VOO2, elaborated by co-crushing 70% *Brava cv.* and 30% *Mansa cv.* (n=2). **VOOb** is the average of results obtained for VOO3 and VOO4, elaborated by co-crushing 90-100% *Mansa cv.* and 0-10% *Brava cv.* (n=2).

Values are medians; confidence intervals of the median at 95% are reported in brackets.

s: significant.

n.p.: not perceived.

Table 4. Concentrations of the main phenolic compounds (mg/kg oil) of the studied VOOs and concentration of *o*-diphenolics and total phenolics of VOOs.

	Concentration \pm SD (mg/Kg)	
	VOOa	VOOb
Hydroxytyrosol (3,4-DHPEA)	11.23 \pm 0.21 ^b	9.18 \pm 0.37 ^a
Tyrosol (<i>p</i> -HPEA)	24.11 \pm 0.03 ^b	27.50 \pm 1.16 ^a
3,4-DHPEA-EDA	112.38 \pm 8.80 ^b	64.19 \pm 0.21 ^a
<i>p</i> -HPEA-EDA	195.27 \pm 20.67 ^b	142.56 \pm 2.26 ^a
3,4-DHPEA-EA	50.34 \pm 4.38 ^a	46.69 \pm 3.19 ^a
<i>p</i> -HPEA-EA	37.32 \pm 6.28 ^a	31.49 \pm 7.62 ^a
Σ Hydroxytyrosol derivatives	174	120
Σ Tyrosol derivatives	257	201
Vanillic acid	0.23 \pm 0.02 ^a	0.26 \pm 0.03 ^a
Vainillina	0.25 \pm 0.002 ^a	0.27 \pm 0.01 ^b
<i>p</i> -Coumaric acid	0.19 \pm 0.07 ^a	0.18 \pm 0.001 ^a
Pinoresinol	2.08 \pm 0.36 ^a	2.37 \pm 0.11 ^a
Total phenols	433.41\pm39.43^b	324.69\pm7.09^a
<i>o</i>-Diphenolics		
Expressed as GA	132.6 \pm 10.7 ^b	83.9 \pm 12.0 ^a
Total phenolics		
Expressed as GA	340.4 \pm 24.1 ^b	215.6 \pm 24.7 ^a
Expressed as HTyr	374.8 \pm 24.1 ^b	253.9 \pm 22.1 ^a

VOOa is the average of results obtained for VOO1 and VOO2, elaborated by co-crushing 70% *Brava* cv. and 30% *Mansa* cv. (n=2). **VOOb** is the average of results obtained for VOO3 and VOO4, elaborated by co-crushing 90-100% *Mansa* cv. and 0-10% *Brava* cv. (n=2).

Values are mean \pm standard deviation. All analyses were performed in triplicate. Different letters within rows indicate significant differences using a Student's *t*-test ($p < 0.05$).

Table 5. Concentrations of C₆ and C₅volatile compounds of VOOs.

C ₆ , C ₅ volatile compounds	Odor threshold* (µg/kg oil)	Odorant series	Concentration ± SD (µg IS/kg oil)	
			VOOa	VOOb
<i>trans</i> -2-Hexen-1-al	420	Bitter	1645.0±17.9 ^b	897.1±258.9 ^a
Σ C6/LnA-Aldehydes			1645±18	897±259
<i>cis</i> -3-Hexen-1-ol	1100	Green-leaf	1694.3±144.9 ^b	1196.9±47.5 ^a
<i>trans</i> -3-Hexen-1-ol	1500	Bitter	40.9±14.0 ^a	24.9±4.1 ^a
<i>trans</i> -2-Hexen-1-ol	5000	Green-leaf	482.2±47.3 ^a	887.4±162.0 ^b
Σ C6/LnA-Alcohols			2217±153	2109±169
<i>cis</i> -3-Hexenyl acetate	200	Ripely fruity	275.5±6.4 ^b	56.2±16.6 ^a
Σ C6/LnA-Esters			276±6	56±17
Hexanal	75	Sweet	173.8±1.6 ^b	85.9±4.3 ^a
Σ C6/LA-Aldehydes			174±2	86±4
1-Hexanol	400	Ripely fruity	561.1±32.8 ^a	582.5±84.4 ^a
Σ C6/LA-Alcohols			561±33	583±84
Hexyl acetate	1040	Ripely fruity, sweet	35.3±9.2 ^b	12.9±1.9 ^a
Σ C6/LA-Esters			35±9	13±2
1-Penten-3-ol	400	Ripely fruity, green-leaf, pungent	56.2±10.8 ^a	72.8±8.4 ^b
<i>cis</i> -2-Penten-1-ol	250	Ripely fruity, sweet	33.5±2.9 ^a	50.7±8.4 ^b
Σ C5/LnA-Alcohols			90±11	124±12
1-Penten-3-one	50; 0.73	Green-leaf; bitter, pungent	12.8±1.3	n.d.
Σ C5/LnA-Ketones			13±1	-
Pentanal	240	Bitter	44.8±12.3 ^a	26.7±0.4 ^a
<i>trans</i> -2-Pentenal	300	Bitter	4.7±2.0	n.d.
Σ C5/LnA-Aldehydes			49±13	27±0.4
1-Pentanol	470; 3000	Ripely fruity, pungent	25.6±1.3 ^a	31.4±1.7 ^b
Σ C5/LA-Alcohols			26±1	31±2
3-Pentanone	70000	Ripely fruity, sweet	96.2±12.0 ^a	112.1±14.4 ^a

<i>Σ C5/LA-Ketones</i>		96±12	112±14
3-Ethyl-1,5-octadiene (1)		50.6±7.7 ^b	31.5±2.5 ^a
3-Ethyl-1,5-octadiene (2)		50.7±3.6 ^b	28.0±2.5 ^a
3-Ethyl-1,5-octadiene (3)		27.4±3.1 ^a	23.3±3.2 ^a
3-Ethyl-1,5-octadiene (4)		34.5±1.3 ^b	20.4±1.3 ^a
<i>Σ Pentene dimers</i>		163±9	103±5
Total compounds		5345±160	4141±321
Minor volatile compounds	Odor threshold*	Concentration ± SD	
	(µg/kg oil)	(µg IS/kg oil)	
		VOOa	VOOb
Ethylbenzene	-	9.6±1.6	n.d.
Limonene	-	34.4±2.7	n.d.
3-Carene	-	20.5±5.1 ^b	11.0±3.9 ^a
α-Farnesene	-	8.2±0.9	n.d.
<i>Hydrocarbons</i>		73±6	11±4
Octane	940	21.9±1.8 ^a	59.0±8.6 ^b
Nonanal	150	9.1±2.3	n.d.
<i>trans</i> -2-Heptenal	5	n.d.	12.4±2.2
<i>trans, trans</i> -2,4-Hexadienal	2000	46.8±13.0	n.d.
<i>Autooxidation</i>		78±13	71±9
2-Methyl-1-butanal	5.2	19.9±6.0 ^a	39.0±5.4 ^b
2-Methyl-1-propanol	-	51.3±11.2 ^a	80.9±0.4 ^b
2-Methyl-1-butanol+3-methyl-1-butanol	480;100	221.8±66.4 ^a	319.3±35.7 ^b
<i>Aminoacids metabolites</i>		293±68	439±36
Methyl acetate	-	68.1±17.2 ^a	80.4±14.9 ^a
Ethyl acetate	940	146.7±13.0 ^a	205.3±9.2 ^b
Methanol	33000	378.1±20.7 ^a	402.8±93.7 ^a
Ethanol	30000	3594.0±841.0 ^b	2030.3±504.0 ^a

Acetic acid	500	279.1±17.3 ^a	805.1±96.8 ^b
Acetone	-	29.1±6.7 ^a	35.2±10.6 ^a
<i>Sugar fermentation</i>		4495±842	3559±522
Total compounds		4939±845	4080±523

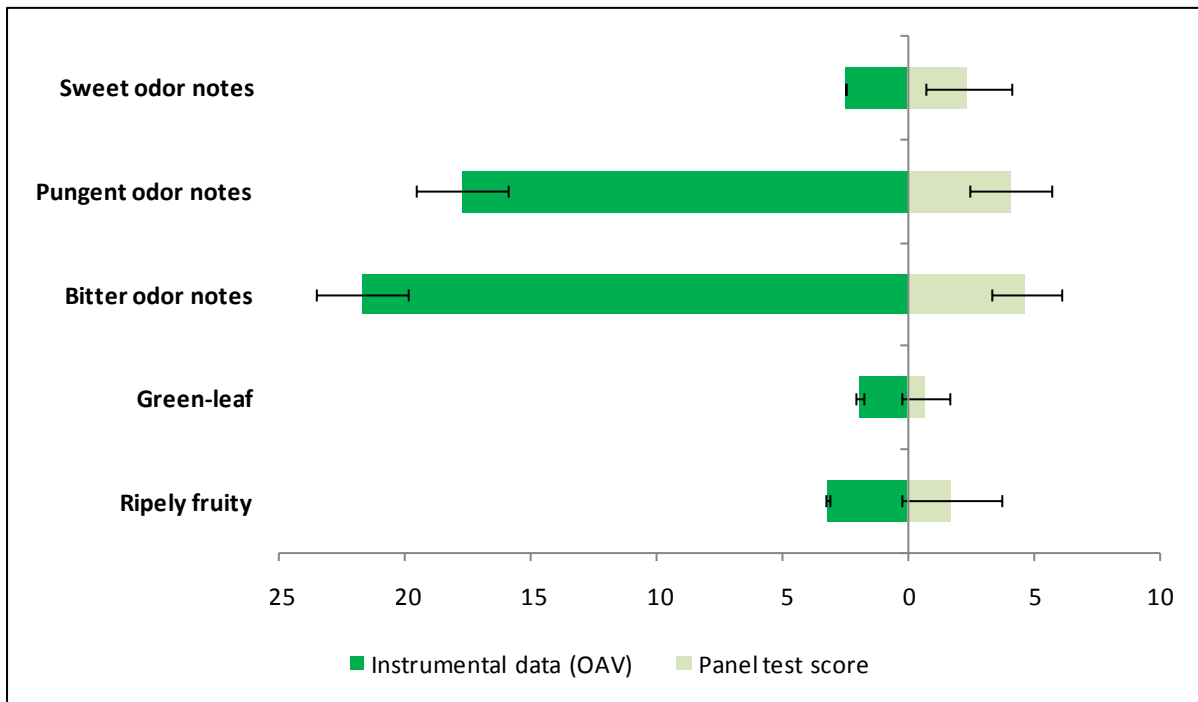
* As reported in [Aparicio and Morales, \(1998\)](#); [Reiners and Grosch, \(1998\)](#); [Aparicio and Luna, \(2002\)](#); [Burdock, \(2002\)](#); [Morales, Luna, and Aparicio \(2005\)](#); [Cayuela, Gómez-Coca, Moreda, & Pérez-Camino, \(2015\)](#).

VOOa is the average of results obtained for VOO1 and VOO2, elaborated by co-crushing 70% *Brava* cv. and 30% *Mansa* cv. (n=2). **VOOb** is the average of results obtained for VOO3 and VOO4, elaborated by co-crushing 90-100% *Mansa* cv. and 0-10% *Brava* cv. (n=2).

All analyses were performed in triplicate. Different letters within rows indicate significant differences using a Student's *t*-test ($p < 0.05$).

n.d.: not detected (< LOD).

VOOa



VOOb

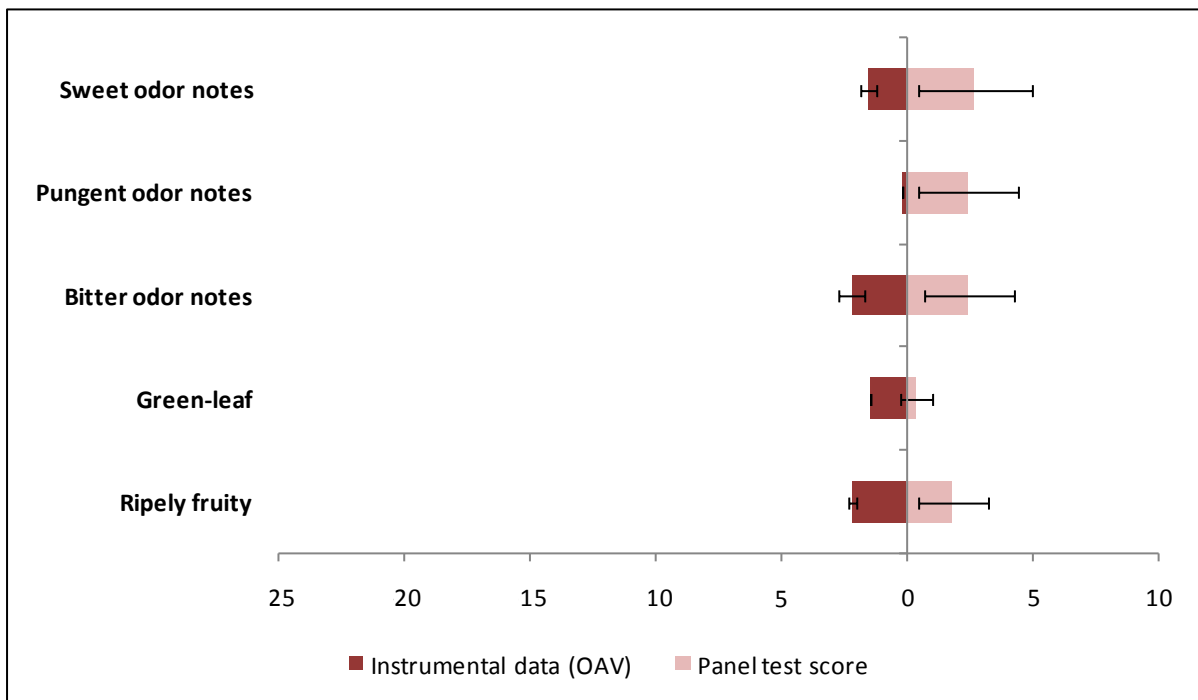


Figure 1.

Supplementary Material

[Click here to download Supplementary Material: Supplemental Figure S1.docx](#)