

Evaluation of Fate of Added Copper and Zinc in Prairie Soils Using Chemical Speciation and XANES Spectroscopy

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Justification

Micronutrient deficiency issues in prairies:

❖ *Site specific:*

a patchy distribution related to spatial variability of soil properties



❖ *Crop and element specific:*

- Cu deficiency for cereal crops
- Zn for pulse crops like pea

Justification

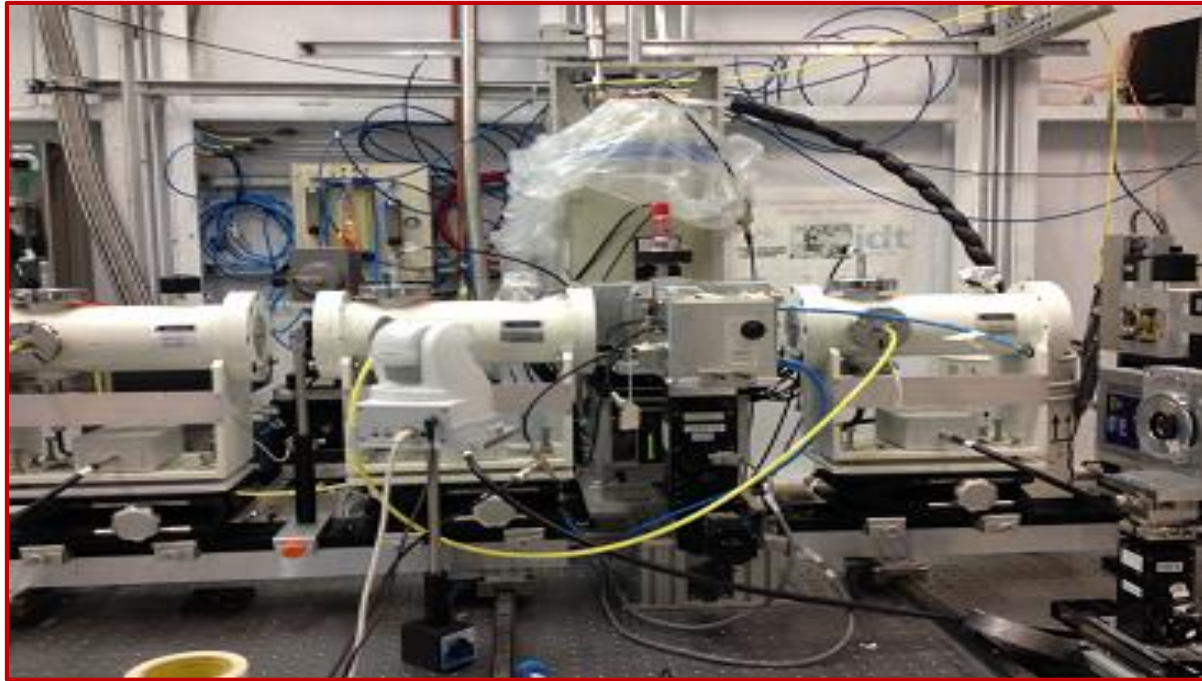
- Inconsistent yield response to fertilization reported
 - *Uncertainty about soil test critical levels*

- Soil properties can control fate and efficacy
 - *Transformation and complexation*
 - *Redistribution to unavailable pools*

Liang et al., 1990,91; Karamanos and Goh, 2001; Karamanos et al. 2003

Objective

To understand fate of different micronutrient fertilizer formulations added to contrasting soils



Methods and Materials

Whitewood Dark Gray Chernozem (Danbury)

Echo Brown Solodized Solonetz (Central Butte)

Whitefox Dark Gray Chernozem (Nipawin)

Sceptre Orthic Vertisol (Sceptre)

Ukalta Black Chernozem (Alix, AB)



Methods and Materials

Treatments

<i>Wheat (Cu)</i>	<i>Pea (Zn)</i>
T ₁ : Control	T ₁ : Control
T ₂ : soil application @ 5 kg Cu ha ⁻¹ (CuSO ₄ ·5H ₂ O)	T ₂ : soil application @ 2 kg Zn ha ⁻¹ (ZnSO ₄ ·7H ₂ O)
T ₃ : soil application @ 2 kg Cu ha ⁻¹ (chelated product)	T ₃ : soil application @ 1 kg Zn ha ⁻¹ (chelated product)
T ₄ : foliar application @ 0.25 kg Cu ha ⁻¹ (chelated product)	T ₄ : foliar application @ 0.25 kg Zn ha ⁻¹ (chelated product)

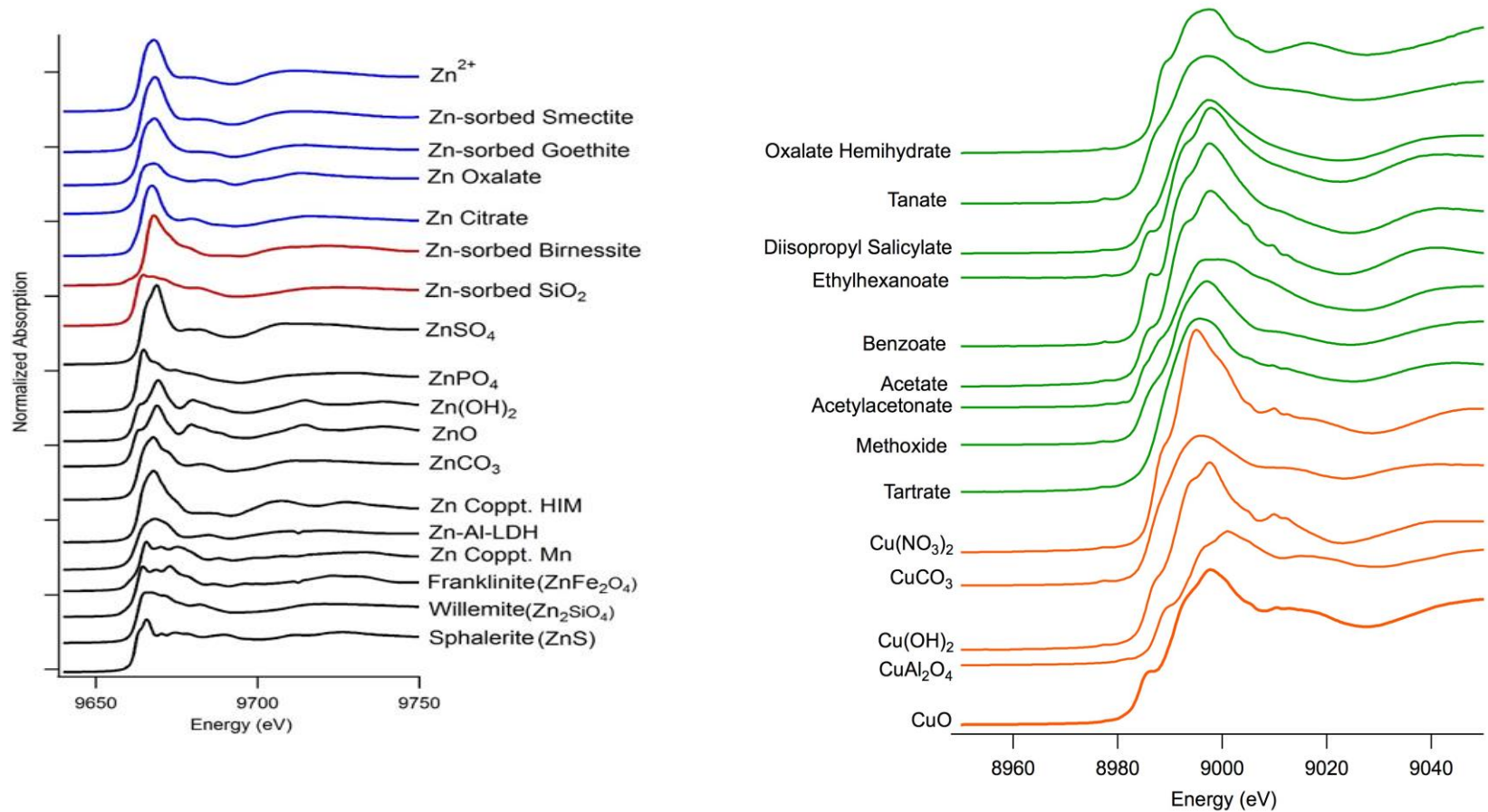
Why XANES

Traditional wet chemical analyses methods

- Unable to identify particular species of Cu and Zn in soil.

- ❑ *XANES spectroscopy can distinguish different oxidation states of an element*
- ❑ *Offers unique ability to follow transformation of fertilizers through different forms.*

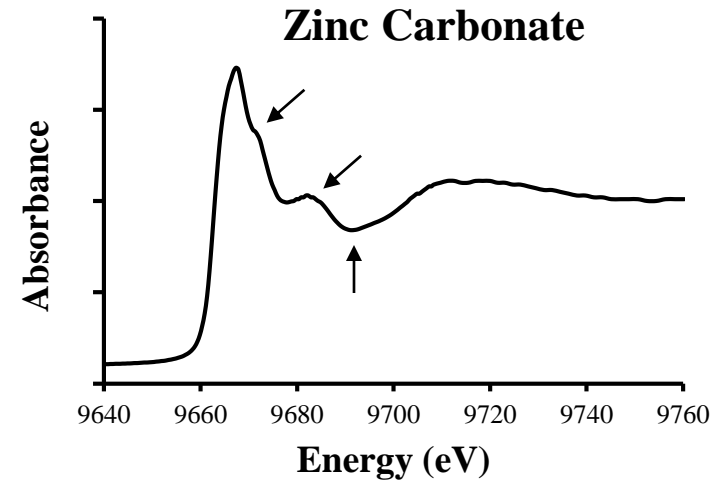
XANES spectra of Cu and Zn reference compounds



XANES Data Analysis Approaches

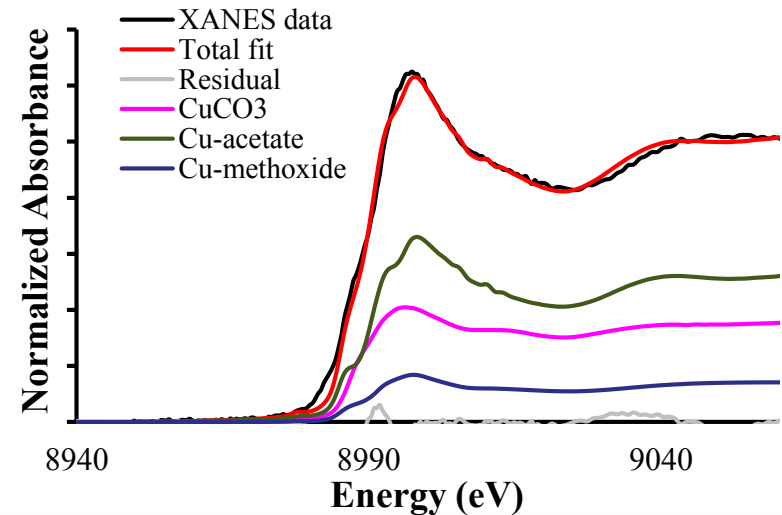
➤ Qualitative analysis

- Fingerprint technique



➤ Quantitative analysis

- LC fitting technique



Results

Observed increased plant available Cu and Zn in soil after harvest with soil applied fertilizer could have rotational benefits

Post-Harvest Soil Residual DTPA extractable Cu and Zn

Treatment	Soil type				
	Whitewood	Echo	Whitefox	Sceptre	Ukalta
	-----mg Cu kg ⁻¹ soil-----				
T ₁ (Con)	0.84a	0.70b	1.31c	1.83c	0.38b
T ₂ (Soil)	3.07a	1.79a	4.56a	3.80a	1.77a
T ₃ (Soil)	2.02a	1.35a	2.34b	3.47ab	0.60ab
T ₄ (Foliar)	1.11a	0.74b	1.45bc	1.95bc	0.30b
<i>p</i> -values	0.057	<.0001	<.0001	0.009	0.013
	-----mg Zn kg ⁻¹ soil-----				
T ₁ (Con)	1.44c	1.18b	1.73b	0.79b	2.43b
T ₂ (Soil)	2.43a	1.89a	2.66a	2.03a	3.58a
T ₃ (Soil)	1.86b	1.54a	2.01ab	1.20b	2.60b
T ₄ (Foliar)	1.42c	1.15b	1.67b	0.92b	2.29b
<i>p</i> -values	<.0001	<.0001	0.012	<.0001	<.0001

Chemical speciation results

Residence of added Cu in the soil solution-carbonate-exchangeable (F₁) and oxyhydroxide fractions (F₂)

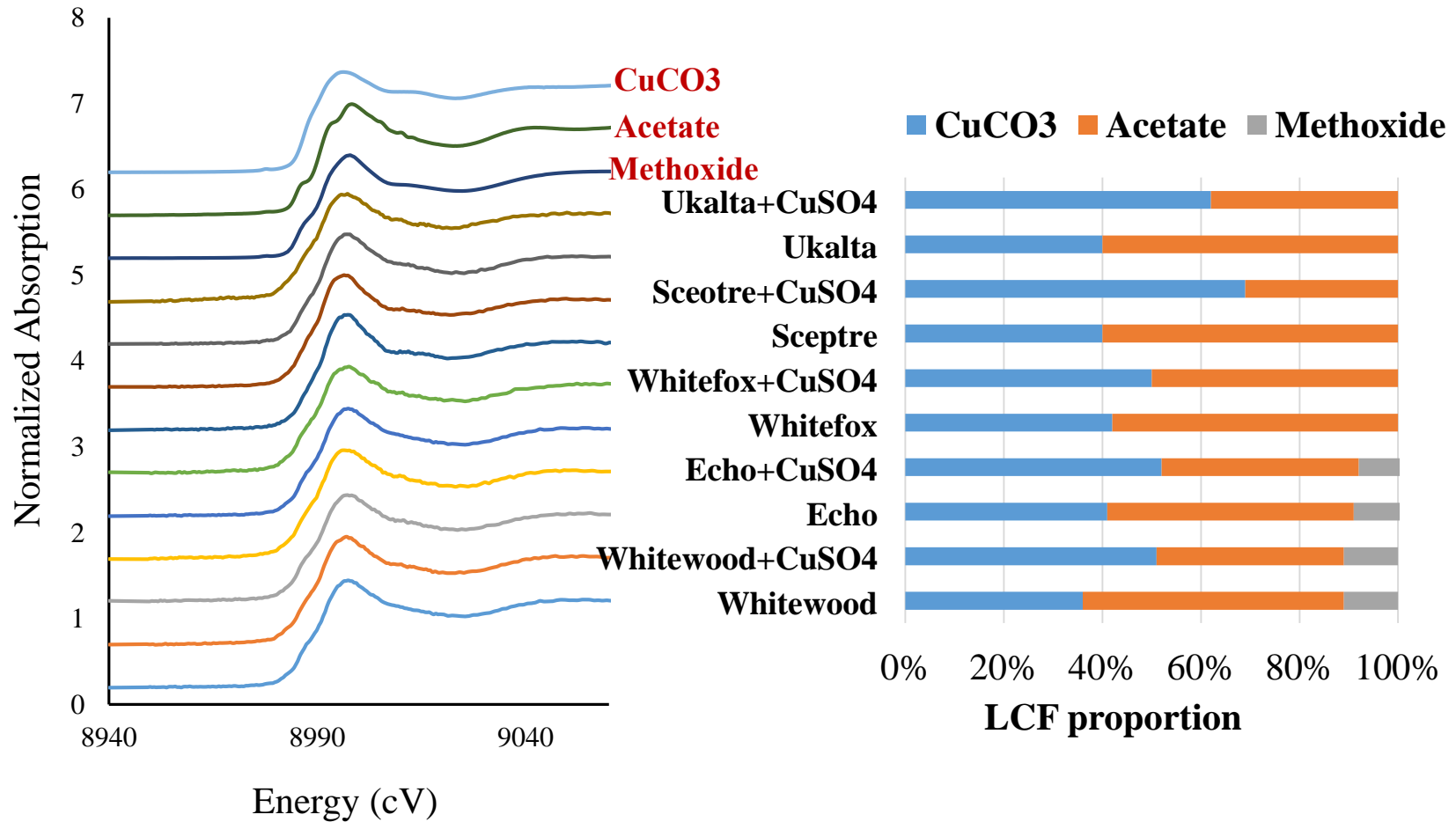
Soil	Treat.	mg Cu kg ⁻¹ soil				
		F ₁	F ₂	F ₃	F ₄	F ₅
Whitefox	T ₁	0.61c	0.58c	1.65b	4.65a	7.49c
	T ₂	2.30a	2.05a	3.30a	4.50a	12.2a
	T ₃	1.23b	1.13b	1.91b	5.27a	9.64b
	T ₄	0.60c	0.65c	1.72b	4.23a	7.19c
<i>p</i> -values		<.0001	<.0001	<.0001	0.064	<.0001
Ukalta	T ₁	0.31b	0.09c	1.57b	4.29a	6.26b
	T ₂	1.27a	0.69a	3.10a	2.80a	7.86a
	T ₃	0.72a	0.39b	2.34ab	4.00a	7.45a
	T ₄	0.28b	0.11c	1.68b	3.83a	5.91b
<i>p</i> -values		<.0001	<.0001	0.003	0.109	<.0001

Chemical speciation results

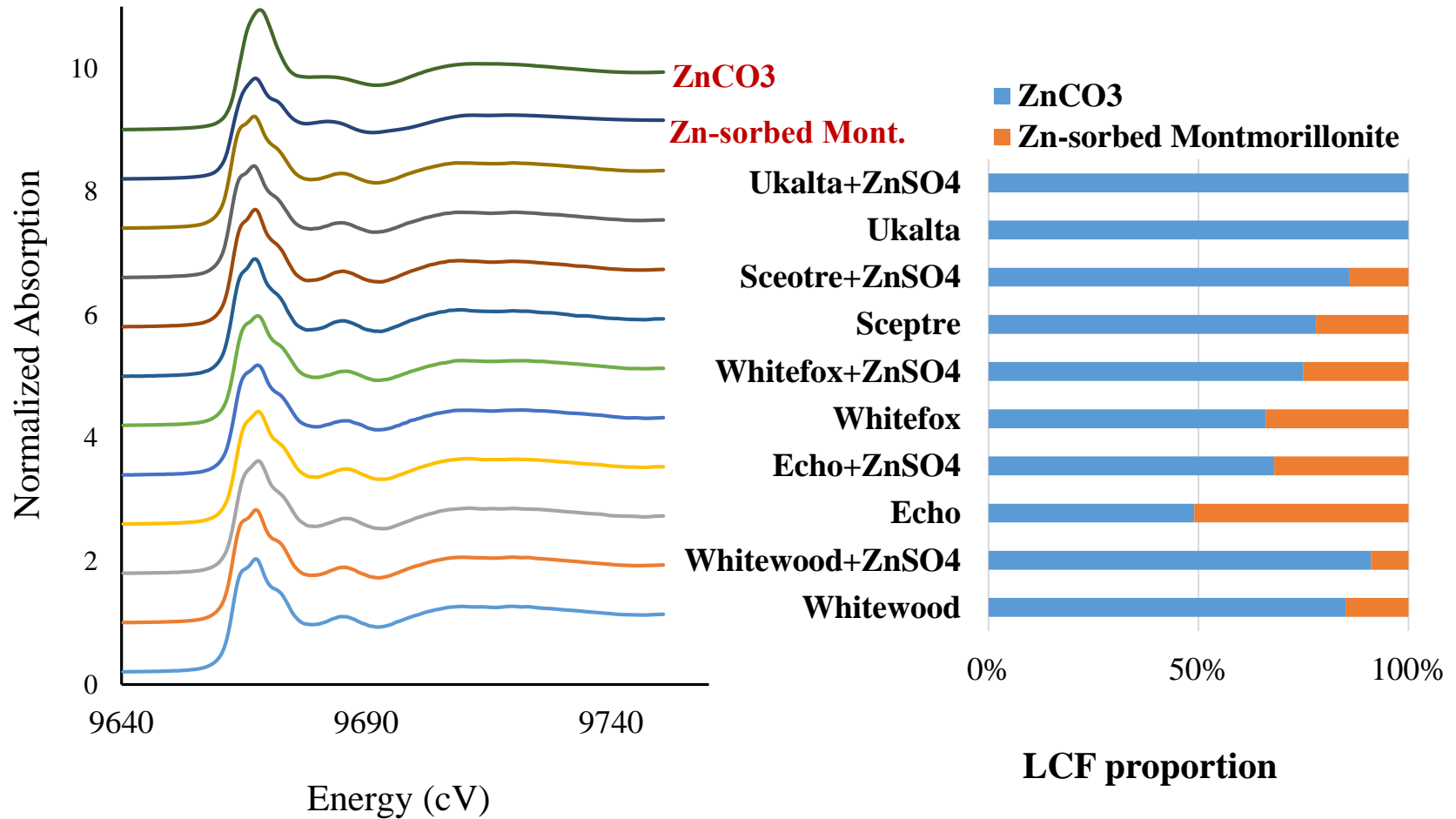
Residence of added Zn in the soil solution-carbonate-exchangeable (F_1) and oxyhydroxide fractions (F_2)

Soil	Treat	mg Zn kg ⁻¹ soil				
		F ₁	F ₂	F ₃	F ₄	F ₅
Echo	T ₁	0.77bc	4.24b	6.27a	33.9a	45.2a
	T ₂	1.11a	5.04a	6.75a	34.4a	47.3a
	T ₃	0.90b	4.33b	6.92a	35.8a	47.9a
	T ₄	0.71c	4.13b	6.46a	34.3a	45.6a
<i>p</i> -values		0.003	0.002	0.05	0.788	0.477
Sceptre	T ₁	0.41b	6.85c	12.3a	61.3a	80.9b
	T ₂	1.31a	8.68a	12.9a	61.6a	84.5a
	T ₃	1.04a	7.52b	12.8a	63.0a	84.3a
	T ₄	0.39b	7.19bc	12.2a	59.8a	79.6b
<i>p</i> -values		<.0001	<.0001	0.197	0.160	0.009

Cu XANES spectra and LC Fit Results



Zn XANES spectra and LCF Fit Results



Summary

- ❖ Carbonate is the dominant form of Cu and Zn, and adsorption to carbonates is likely the dominant process that determines fate of added fertilizer
- ❖ Organic forms of Cu (acetate) and Zn-sorbed montmorillonite can contribute to maintain bioavailability through chelation and cation exchange, respectively
- ❖ Applied CuSO_4 is mostly distributed to soil carbonates and oxyhydroxide fraction; but also to the organic fractions (e.g. Whitefox and Ukalta soils)
- ❖ Residual micronutrients from soil-applied fertilizer may favor performance of following crops grown in rotation

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Canadian
Light
Source

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synchrotron



AAFC AgriInnovation Program

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Thank you

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