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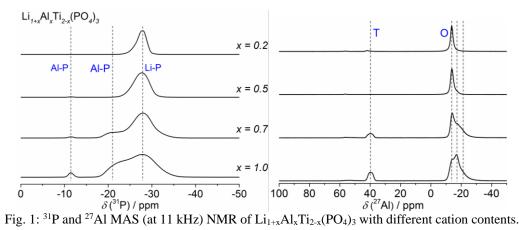
Solid-State NMR Studies on Li⁺ Ion Dynamics and Structure of Li_{1+x}Al_xTi_{2-x}(PO₄)₃

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Solid-state nuclear magnetic resonance (NMR) studies of Li⁺ ion dynamics based on spinlattice relaxation (SLR) experiments on lithium aluminum titanium phosphates (LATP) are reported. The samples were synthesized by conventional solid-state reaction using stoichiometric amounts of Li₂CO₃, TiO₂, (NH₄)H₂PO₄ and Al₂O₃ at high pressure and heated in a Pt crucible up to 1000 °C for 12 hours [1-4]. LATP with different lithium contents is studied with ⁷Li SLR measurements (in the laboratory frame (*T*₁) and rotating frame (*T*₁_ρ)) as well as multinuclear magic-angle spinning (MAS) NMR experiments. The solid-state SLR experiments were done with a Bruker 600 AV-III spectrometer at a ⁷Li Larmor frequency of 233 MHz over a wide range of temperature. The high-temperature range was achieved using LASER-heating in a 7mm MAS probe at 3 kHz spinning. ⁷Li, ²⁷Al and ³¹P MAS NMR measurements were done using a 4mm MAS probe at 7-11 kHz spinning frequencies.



From the diffusion induced ⁷Li NMR SLR rates Li jump rates and activation energies were deduced and the LATPs were found to be fairly fast ion conductors. The different compositions provided signature NMR signals for different cation stoichiometries (Fig. 1). From the various NMR methods, a detailed characterization of LATPs is obtained, with information on local ion dynamics and many structural aspects.

References

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