### Field-aged phosphate impacts on glyphosate and phosphorus sorption patterns in two Prairie soils



#### University of Manitoba

### Sirajum Munira<sup>\*</sup>, Annemieke Farenhorst, Wole Akinremi and Mulikat Bammeke

Department of Soil Science University of Manitoba Winnipeg, MB, Canada



Introduction Objective Material and Methods

### Glyphosate is the world's best-selling chemical herbicide.





**Results** Conclusion

http://en.wikipedia.org/wiki/Glyphosate

Introduction Objective Material and Methods Results Conclusion

Potential concerns about glyphosate use:

Surface waters
Human health
Plant diseases
Nutrients interaction



Introduction Objective Material and Methods **Results** Conclusion **Phosphate structure Glyphosate structure** 1111



### **General concept of sorption mechanism**

Ligand Exchange Inner sphere complexes Metal-Glyphosate-Complex IntroductionObjectiveMaterial and MethodsResultsConclusion□ Phosphorus (P), an essential nutrient for<br/>crop nutrition

- Plants typically recover less than 20% of the applied phosphorus
- □ The release of P from soil to water can accelerate freshwater eutrophication





## To examine the impact of field-aged phosphate on glyphosate and phosphorus sorption patterns in two prairie soils.



# Fields sites were near Carman (49° 29.7' N, 98° 2.4' W), and Forrest, (50° 1.2' N, 99° 53.3' W) Manitoba.





Properties	Carman	Forrest
Texture	Sandy - clay - loam	Clay - Ioam
OC (%)	2.81 ± 0.04	3.2 ± .07
pH (soil/solution 1:2)	4.7 ± 0.02	7.3 ± 0.02
Fe <sub>2</sub> O <sub>3</sub> (mg/kg)	200 ± 8.85	12.52 ± .22
Al <sub>2</sub> O <sub>3</sub> (mg/kg)	6.41 ± .64	1.07 ± .47
Ca (mg/kg)	2252 ± 35	4791 ± 158
Olsen-P (mg/kg)	43 ± 7.34	23 ± 2.88

Introduction Objective Material and Methods Results Conclusion

### Introduction Objective Material and Methods



#### **Research Field**



#### **Soil + Solution**



Results Conclusion

Incubation 🔶



Samples into LSC Adding scintillation cocktail Centrifuge

### Introduction Objective Material and Methods Results Conclusion







#### Soil + Solution End to end shaker

Centrifuge 棏



Spectrophotometer Color development

Extraction<sub>2</sub>

### Two solutions (0.01M CaCl<sub>2</sub> and 0.01M KCl) were used as background liquid



### Background liquid effect on glyphosate sorption



Introduction Objective Materials and Methods Results

### Background liquid effect on phosphorus sorption



### **Carman Soil**

**Forrest Soil** 

### Introduction Objective Materials and Methods Results Conclusion Aged-P effect on glyphosate sorption



#### Results Conclusion Introduction Objective **Materials and Methods** Aged-P effect on Phosphorus sorption y = -0.5915x + 64.1980.01M CaCl2 80 0.01M CaCl2 y = -0.9773x + 77.171 $R^2 = 0.77$ K (L<sup>N</sup> mg<sup>1-N</sup> kg<sup>-1</sup>) 05 05 05 05 $R^2 = 0.75$ 0.01M KCl y = -0.4827x + 47.45680 0.01M KCl y = -0.806x + 55.085K (L<sup>N</sup> mg<sup>1-N</sup> kg<sup>-1</sup>) 0 05 09 09 $R^2 = 0.74$ $R^2 = 0.85$ 78% 86% **37% 52%** 0 0 0 15 30 45 60 20 80 0 60 100 40 Olsen-P (mg/kg) Olsen-P (mg/kg) **Carman Soil Forrest Soil**

#### Introduction Objective Materials and Methods

### Fresh-P effect on glyphosate sorption



### **Carman Soil**

**Forrest Soil** 

Results

## Roundup Ultra 2 effect on Phosphorus sorption (mg/kg)

Results

Phosphorus Level	Carman		Forrest		
	Roundup Ultra 2 level (mg/L)				
	0	100	0	100	
0P plots	244 ± 20	228 ± 6	459 ± 21	427 ± 22	
80P plots	229 ± 25	228 ± 12	450 ± 21	410 ± 30	

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□ Glyphosate and phosphorus sorption was higher with background liquid CaCl2. Long term application or co-application of fresh of phosphate fertilizer reduced the retention capacity of glyphosate and phosphorus.

- Phosphorus sorption was not significantly impacted when Roundup Ultra 2 was added to batch slurries in the laboratory.
- Differences in glyphosate and phosphorus K
   values between the soils resulted from
   differences in soil pH, calcium content, iron
   and aluminum oxide concentrations in soil.







### **Acknowledgements**

- □ I would like to thank Dr. Don Flaten, Dr. Cynthia Grant, Dr. Francis Zvomuya.
- □ I would like to acknowledge the University of Manitoba Graduate Fellowship Award, the Manitoba Graduate Scholarship Award, and NSERC. And Md. Mofizul Islam, Mulikat Bammeke, Rob Ellis, Summer students for their help.



