

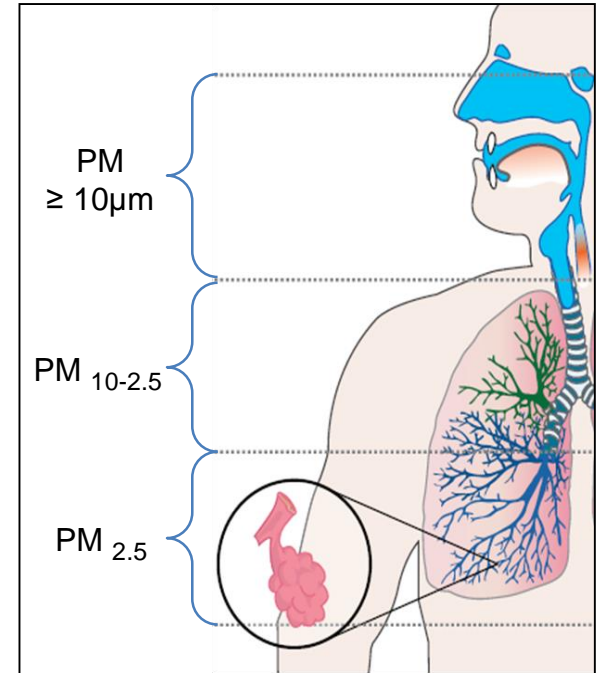
Chemical characterization and *in vitro* toxicity on human bronchial epithelial cells BEAS-2B of PM_{2.5} from an urban site under industrial emission influence

Yann LANDKOCZ, Sylvain BILLET, Frédéric LEDOUX, Anthony VERDIN,
Perrine MARTIN, Fabrice CAZIER, Dominique COURCOT

*Université du Littoral Côte d'Opale
Unité de Chimie Environnementale et Interactions sur le Vivant, UCEiV, EA 4492
SFR Condorcet, FR CNRS 3417
F-59140, Dunkerque, France*

Small size PM is attracting a particular attention:
the smaller the particle size,

- the higher time it remains suspended in air
- the higher probability of it being inhaled
- the higher ability to reach the deep lung



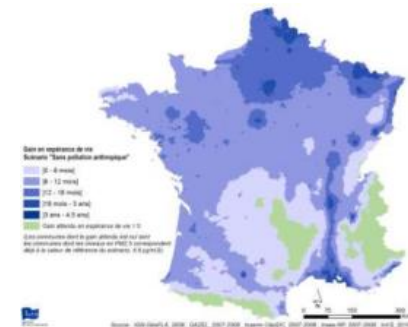
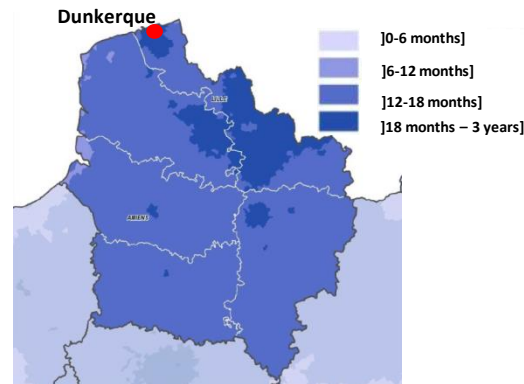
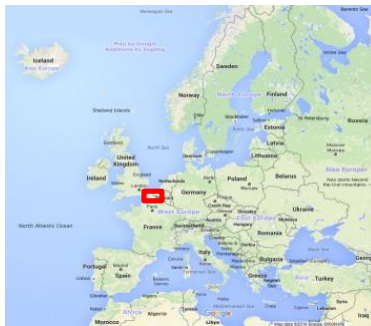
Epidemiological studies

↗ $10 \mu\text{g}/\text{m}^3$ of $\text{PM}_{2.5}$ concentration:

- 26 % mortality risk by cardio-vascular disease
(*Lepeule et al., 2012*)
- ↗ 8 % incidence rate of lung cancer (*ERS, 2010*)
- Respiratory insufficiencies : asthma, chronic obstructive pulmonary diseases (COPD)
(*Silverman et Ito, 2009 ; Tsai et al., 2013*)

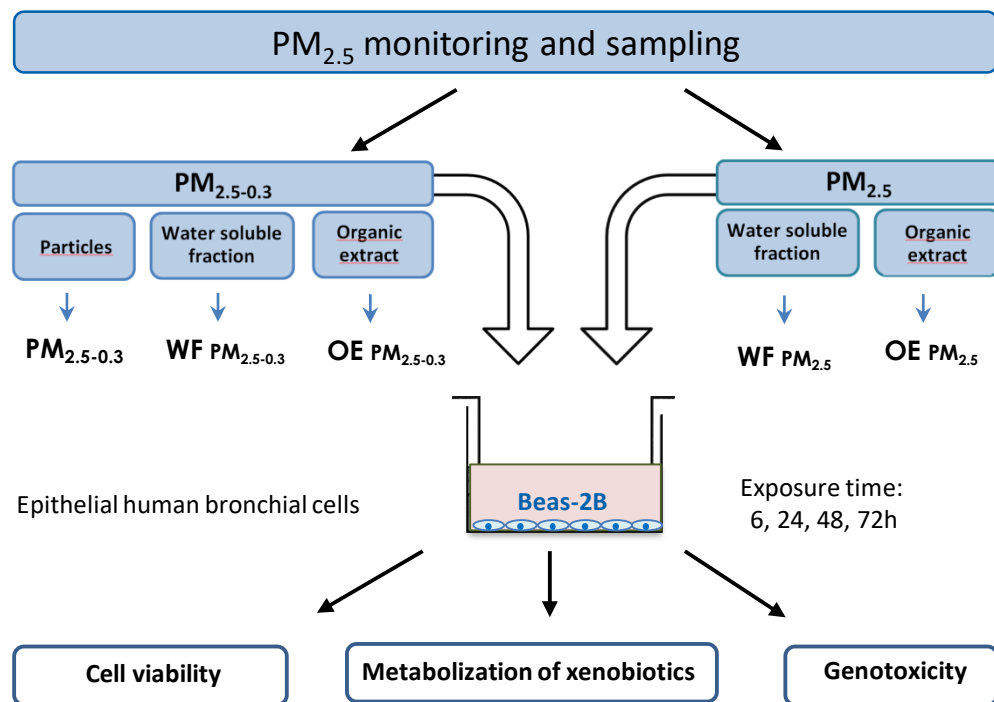
- Exposure to PM: 6 millions people in the Region
- Impact of PM on Life expectancy : average loss of 16 months (*InVS, 2015*)
- High road traffic density
- Local industrial emissions of particles in the atmosphere (>3000 tons/y) (*European-PRTR data*)

Gain in life expectancy in a scenario "without air pollution"



Objectives of the study

- Determine metals and PAHs concentrations in PM_{2.5} and identify their sources in Dunkerque, a coastal urban site
- Compare toxic effects of PM_{2.5} from *in vitro* experiments on epithelial lung cells, depending on their exposure to solid particles, water soluble or organic extracts



Physico-chemical characterization :

- ⇒ PM composition
- ⇒ Source apportionment

In-vitro toxicity assessment :

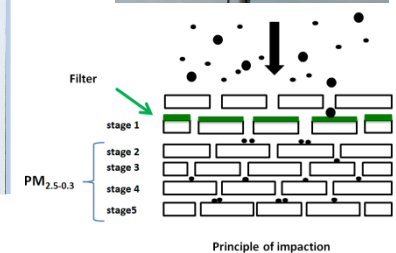
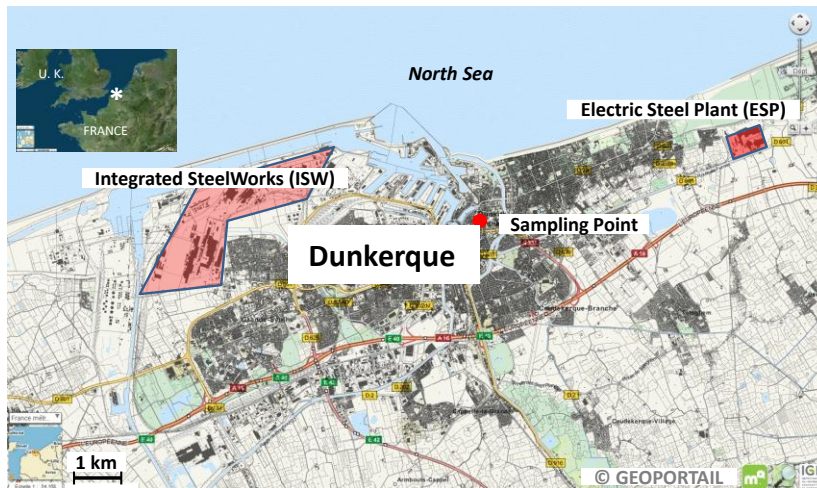
- ⇒ Effect of the different fractions
- ⇒ Effect of PM components in the solid state vs. in extracts

■ PM_{2.5} monitoring and sampling

Location : Dunkerque, 200 000 inhabitants

Period : March to July, 2011

Devices : - MP101 RST, Environnement SA[®] for PM_{2.5} concentration
 - Sampler Digitel[®] DA 80 (30 m³/h): 24h PM_{2.5} samples (on filter)
 - Cascade impactor STAPLEX[®] (68 m³/h): 5 days PM_{2.5-0.3} samples (on plate)



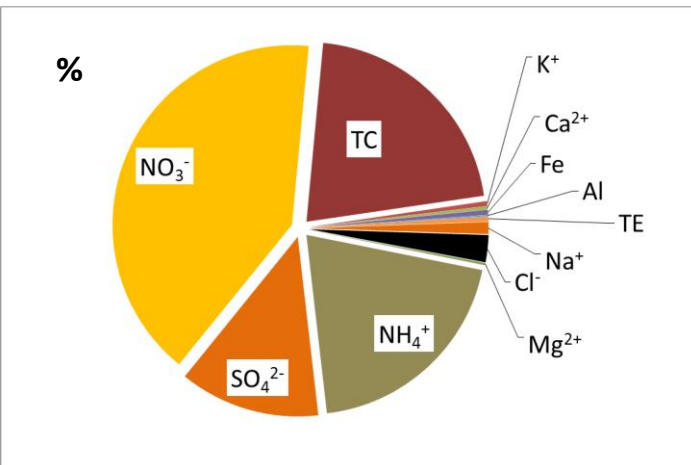
■ PM_{2.5} chemical characterization

Major and trace elements (ICP-AES and ICP-MS), water soluble ions (IC)

Total carbon (CHON analyzer) and Polycyclic Aromatic Hydrocarbons PAHs (GC-MS)

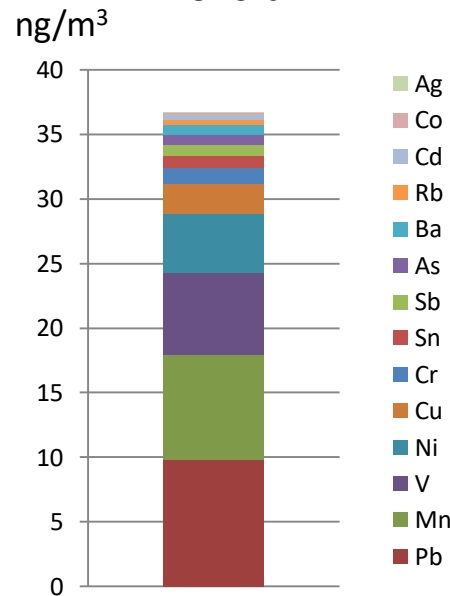
- Average PM_{2.5} concentration (march-july 2011) : 14 µg/m³

Major Components



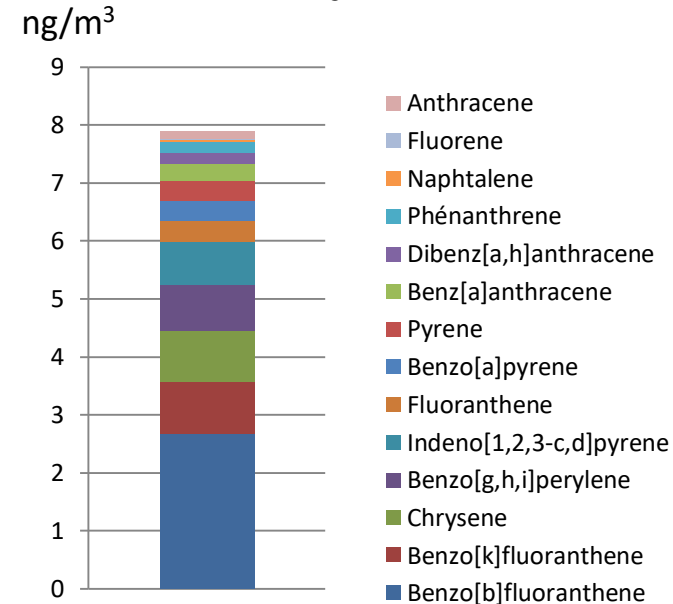
- NO₃⁻, SO₄²⁻, NH₄⁺ and carbon :
> 90% of PM_{2.5} mass

Metals



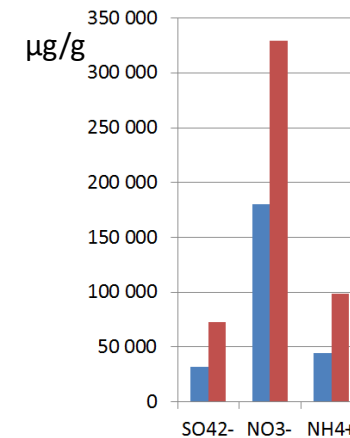
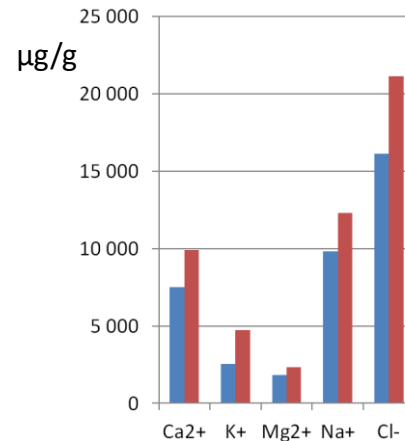
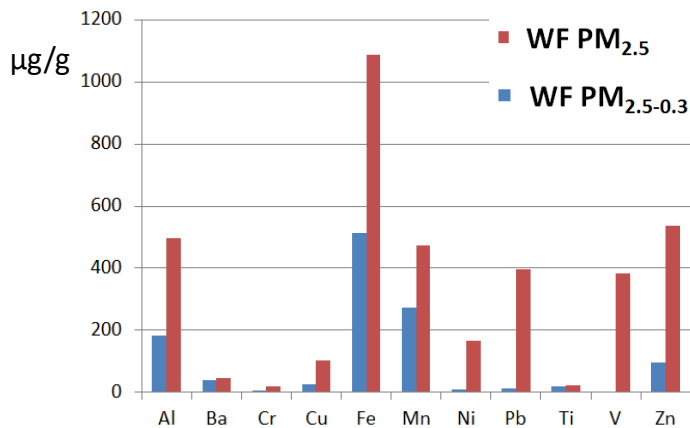
- Metals explained mainly by emissions from steelmaking industry, traffic non-exhaust and heavy fuel oil combustion

PAHs



- PAHs ratios (Fla/Pyr, InPy/BghiP) convenient with diesel exhaust, heavy fuel oil combustion and cokemaking industry emissions

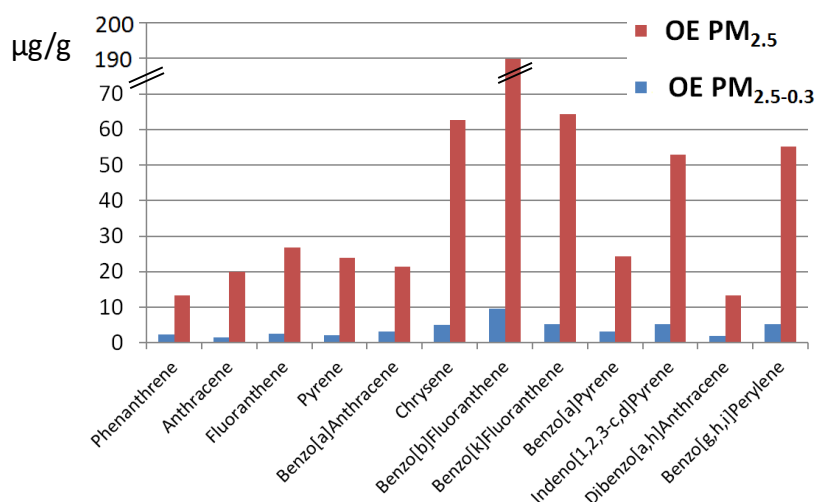
- Water soluble fraction (WF) : particles placed in pure water



- Proportion of elements in PM_{2.5-0.3} / PM_{2.5} :
 - Ba, Ti : > 75%
 - Al, Fe, Mn : between 75 and 25 %
 - Cr, Cu, Ni, Pb, V, Zn: < 25%

- Water soluble ions (Ca²⁺, K⁺) and sea salts (Na⁺, Mg²⁺, Cl⁻) predominantly found in the PM_{2.5-0.3}, contrary to secondary inorganic ions

- Organic extracts (OE): Soxhlet extraction using DCM
Organic compounds concentrated in DMSO

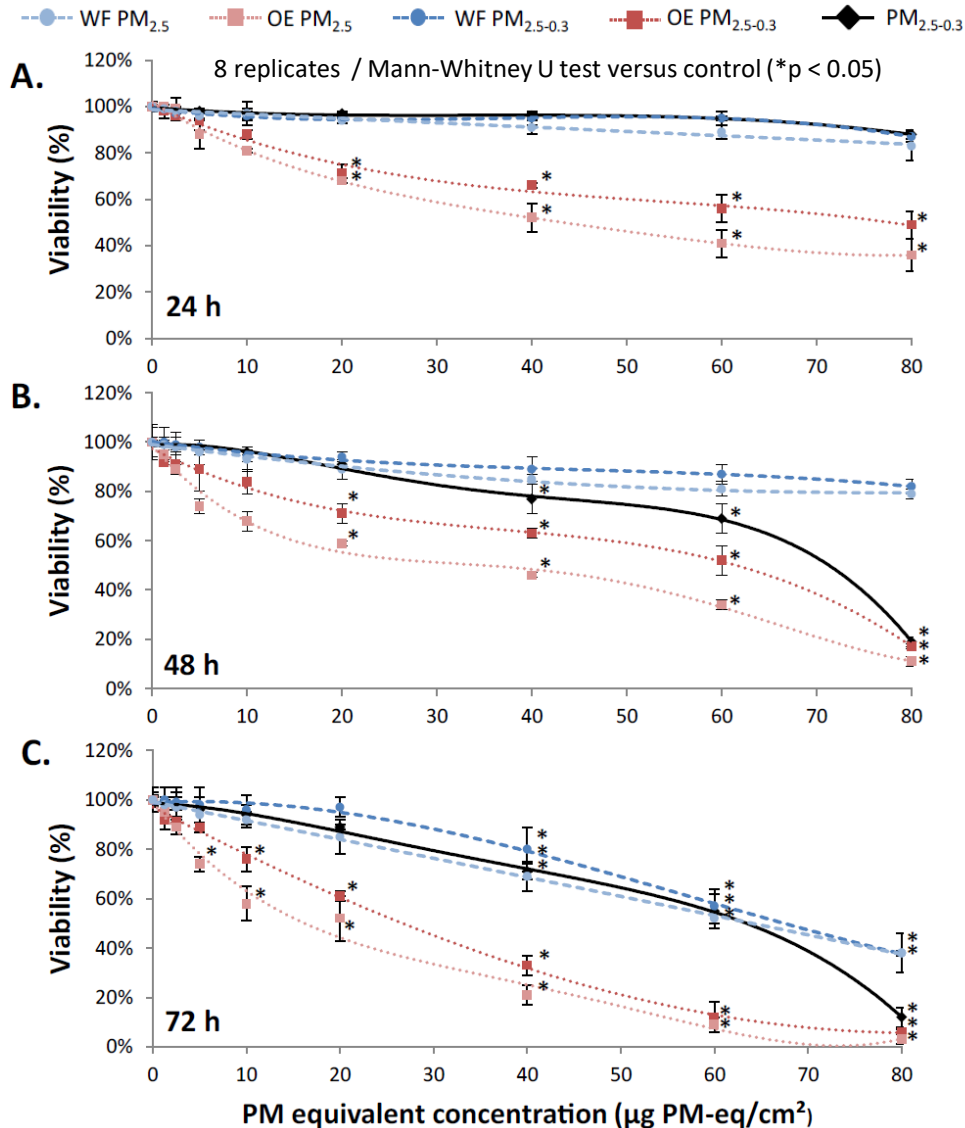


	PM _{2.5-0.3}		PM _{2.5}	
	(µg/g)	(pg/m ³)	(µg/g)	(pg/m ³)
Phenanthrene (Phe)	2.4	33	13.2	186
Anthracene (Ant)	1.5	21	–	–
Fluoranthene (Fla)	2.5	35	26.9	378
Pyrene (Pyr)	2.2	31	23.9	336
Benzo[a]anthracene (BaA)	3.2	46	21.3	330
Chrysene (Chr)	5.0	71	62.7	882
Benzo[b]fluoranthene (BbF)	9.5	134	189	2,670
Benzo[k]fluoranthene (BkF)	5.3	74	64.2	904
Benzo[a]pyrene (BaP)	3.2	44	24.4	343
Indeno[1,2,3-c,d]pyrene (InPy)	5.3	74	52.8	743
Dibenzo[a,h]anthracene (DahA)	2.0	28	13.4	189
Benzo[ghi]perylene (BghiP)	5.2	73	55.2	777
Total PAHs	48.5	669	548	7,708

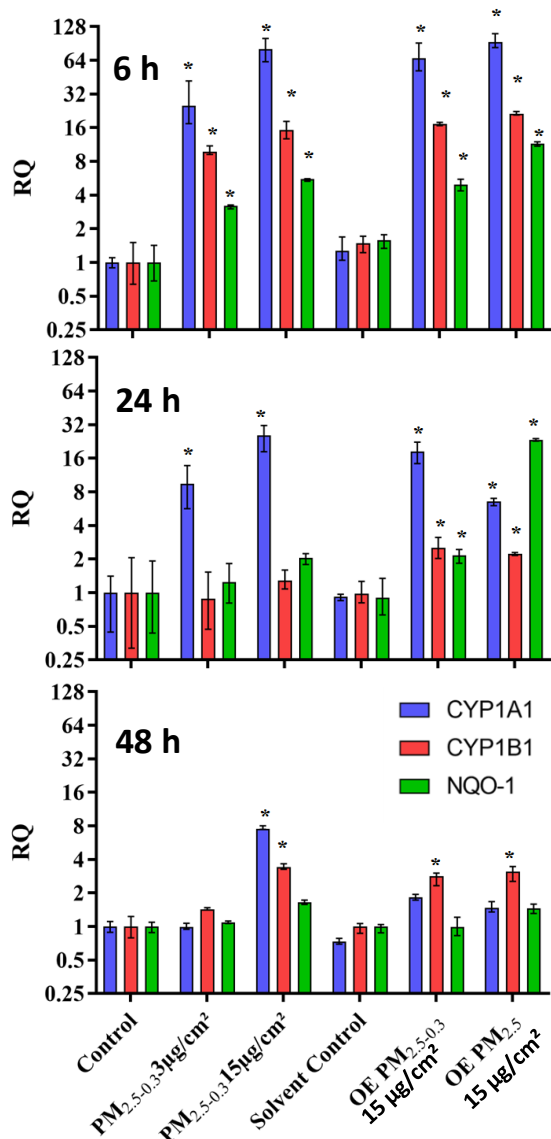
Landkocz et al. Env. Poll. 2017

- PAHs more concentrated in PM_{2.5} including the ultrafine fraction than in PM_{2.5-0.3}
 - Phe, BaA, BaP, DahA : 5-10 fold higher
 - Other compounds > 10- fold higher
- Contribution of combustion processes, known to form hydrocarbon- rich particles predominantly in the ultrafine mode (Kawanaka et al. 2009)

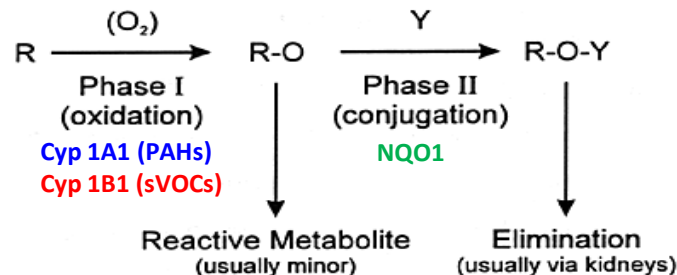
from Extracellular LDH release measurements



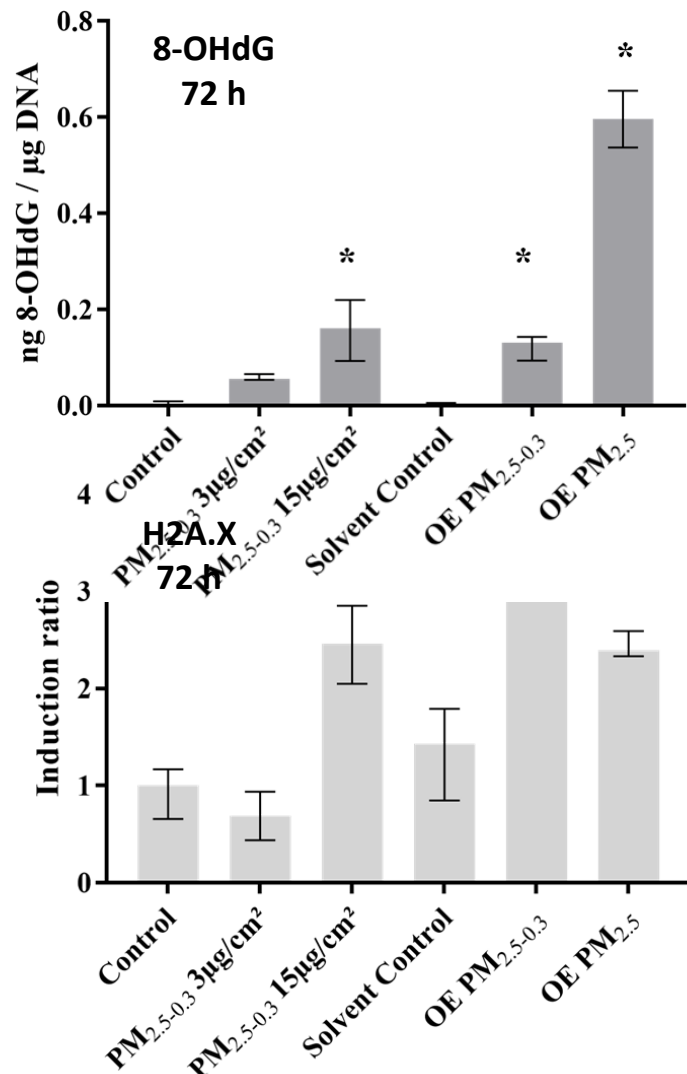
- $\text{PM}_{2.5-0.3}$, OE and WF samples : decrease of cell viability in a time- and dose-dependent manner
- Organic extracts (OE): more cytotoxic than $\text{PM}_{2.5-0.3}$ and the two WF extracts
- Cytotoxicity :
OE $\text{PM}_{2.5} > \text{OE PM}_{2.5-0.3}$
- ⇒ PAHs concentration : much higher in $\text{PM}_{2.5}$ than $\text{PM}_{2.5-0.3}$
- ⇒ Cytotoxicity mainly governed by organic compounds, and particularly PAHs
(Oh et al. 2011, Topinka et al. 2013)
- Tests using 2 doses : 3 and 15 $\mu\text{g}/\text{cm}^2$



6 replicates / * Significant (Relative Quantity, RQ < 0.5 or RQ > 2)



- CYP 1A1, CYP 1B1 and NQO-1 gene expression : induced in a dose-dependent manner (6h)
- CYP 1A1 gene expression : decrease over time, however, significant expression after 48h for PM_{2.5-0.3} (15 μg/cm²) contrary to OEs
- Interpretation :
 - ⇒ 1/ higher bioavailability of PAHs in OEs
 - 2/ in PM_{2.5-0.3}, PAHs strongly bounded on particle surface and inside pores
 - 3) in PM_{2.5-0.3}, a gene induction by metals can not be excluded (*Korashy et al. 2005*)



- **DNA oxidative alteration:** Dose-dependent increase of 8-hydroxydesoxyguanosine 8-OHdG level (*Billet et al. 2018; Dergham et al. 2015*)

- **Similar level :** PM_{2.5-0.3} ≈ OE PM_{2.5-0.3} ,
and OE PM_{2.5} > OE PM_{2.5-0.3}

⇒ oxidative DNA alteration mainly linked to the organic fraction (*Høgsberg et al. 2013*)

- **DNA Damage Response** (repair mechanism of double strands breaks) (*Foster et al. 2005*)

- **H2A.X :** significant increase of phosphorylation for PM_{2.5-0.3} (15 μg/cm²) and OEs :

⇒ 1/ organic compounds at high dose could limit the ability of cells to induce repair mechanism

2/ metals in PM_{2.5-0.3} known to cause double strands breaks but also to inhibit proteins involved in the DNA repair pathway

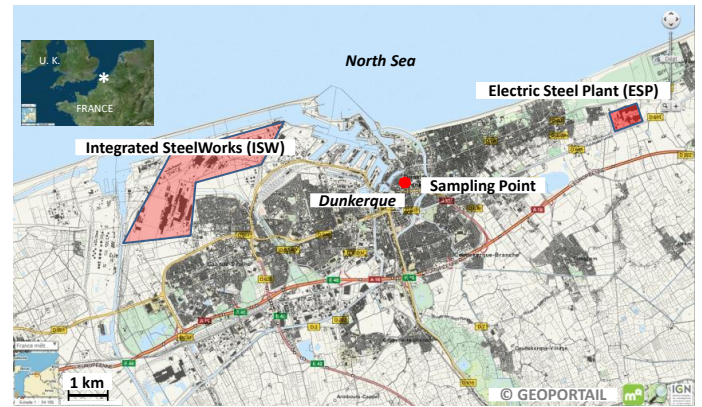
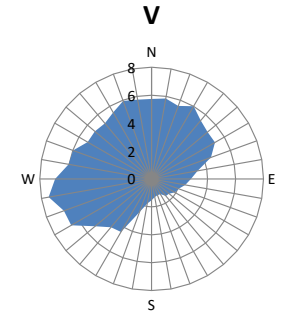
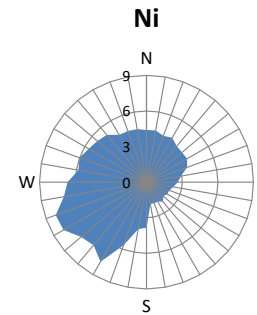
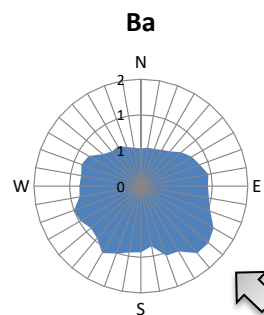
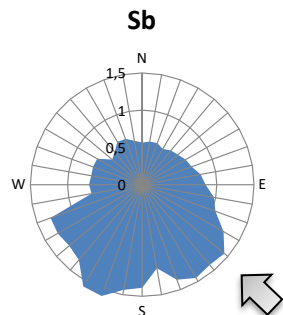
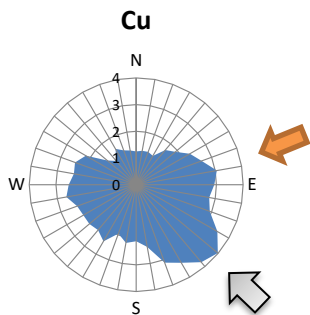
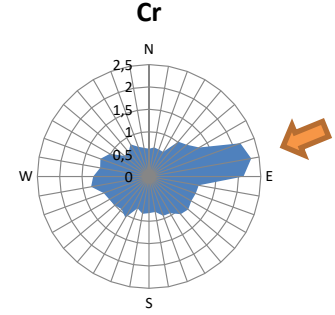
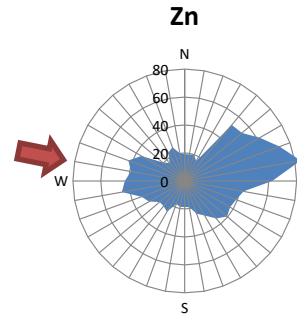
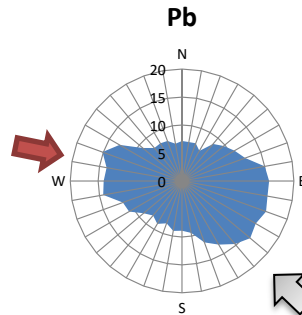
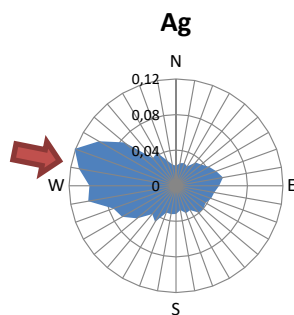
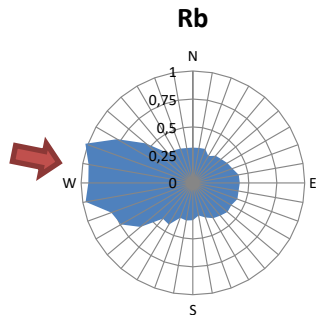
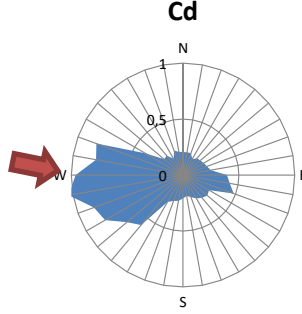
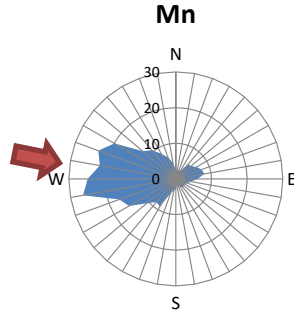
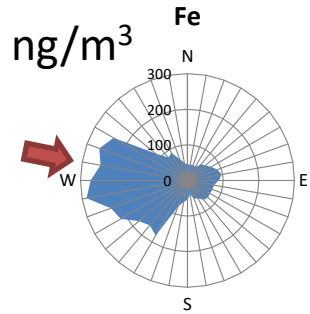
- Comparison of biological response of Beas 2B cells depending on the use of starting particles (as collected on plates) or water soluble and organic extracts
- Cytotoxicity and oxidative DNA alteration (8-OHdG) mainly governed by the organic fraction
- Considering Organic Extracts for *in vitro* toxicology tests does not reflect exactly the cell response (XME, DDR) in the presence of particles : role of the particle skeleton
- Further investigation on the signalisation pathways involving oxidative stress presented in the next talk....

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- the « Institut National du Cancer », France
- the European Union (EFRED)



Metal Concentration Roses



wind sector

elements

assignment

WNW

Fe, Mn, Cd, Rb, Ag, Pb, Zn

⇒ Integrated steelworks

ENE

Zn, Cr, Cu

⇒ Electric Steel plant

SE-SW

Cu, Sb, Ba, Pb

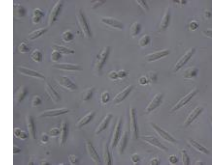
⇒ Traffic

SW-NE

Ni, V

⇒ Heavy fuel oil combustion

- Experiments using epithelial human bronchial cells (BEAS 2B) in culture : « *in vitro* » study
- Test considering PM_{2.5-0.3} and extracts :
Cytotoxicity, gene expression of XME, genotoxicity (8-OHdG, H2A.X)



	Preparation	PM _{2.5}	PM _{2.5-0.3}
Particles		DA80 filter	PM_{2.5-0.3} recovered on impaction plates
Organic Extract	Soxhlet extraction using DCM. Organic compounds concentrated in DMSO	OE PM_{2.5}	OE PM_{2.5-0.3}
Water soluble fraction	Solubilization in pure water	WF PM_{2.5}	WF PM_{2.5-0.3}

- Exposure time: 6, 24, 48 and 72 h