Supporting Information

Living β -Selective Cyclopolymerization Using Ru Dithiolate Catalysts

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1. General experimental

Materials

All reagents which are commercially available from Sigma-Aldrich[®], Tokyo Chemical Industry Co. Ltd., Acros Organics, Alfa Aesar[®], without additional notes, were used without further purification. Dichloromethane for the polymerization were purified by Glass Contour Organic Solvent Purification System, and degassed further by Ar bubbling for 10 minutes before performing reactions. Thin-layer chromatography (TLC) was carried out on MERCK TLC silica gel 60 F254 and flash column chromatography was performed using MERCK silica gel 60 (0.040~0.063 mm).

Characterization

¹H-NMR and ¹³C-NMR were recorded by Varian/Oxford As-500 (500 MHz for ¹H and 125 MHz for ¹³C) and Agilent 400-MR (400 MHz for ¹H and 100 MHz for ¹³C) spectrometers. ¹³C NMR for the polymers were mainly recorded by Bruker (600 MHz for ¹H and 150 MHz for ¹³C) spectrometers in the National Instrumentation Center for Environmental Management (NICEM) at SNU. High resolution mass spectroscopy (HRMS) analyses were performed by the ultrahigh resolution ESI Q-TOF mass spectrometer (Bruker, Germany) in the Sogang Centre for Research Facilities. Size exclusion chromatography (SEC) analyses were carried out with Waters system (1515 pump, 2414 refractive index detector) and Shodex GPC LF-804 column eluted with THF (GPC grade, Honeywell Burdick & Jackson®) and filtered with a 0.2 μm PTFE filter (Whatman®). Flow rate was 1.0 mL/min and temperature of column was maintained at 35 °C.

2. Experimental procedures for the preparation of the monomers

Ru1,¹ **Ru2**,² **M1**, **M3-M6**, **M8-M11**³ were prepared by literature methods.

M2 (tert-butyl 2-(prop-2-yn-1-yl)-2-(2-((triisopropylsilyl)oxy)propan-2-yl)pent-4-ynoate)



To a solution of **M3** (880 mg, 3.0 mmol) in THF (9 ml), methylmagnesium bromide (3 M in ether, 12 mmol, 4 ml) was slowly added at 0 °C. and The mixture was stirred for 2 hr at room temperature then quenched by saturated NH₄Cl aqueous solution at 0 °C. The organic layer was washed with NaCl aqueous solution, extracted by EtOAc, dried with MgSO₄, and concentrated. The product was purified by flash column chromatography on silica gel (EtOAc:Hexane=1:20) to afford **M2-1** as a colorless liquid (510 mg, 68% yield). ¹H-NMR (500 MHz, CDCl₃): δ 3.28 (s, 1H), 2.83 (q, J = 61.0, 17.2 Hz, 4H), 2.05 (s, 2H), 1.49 (s, 9H), 1.31 (s, 6H).; ¹³C-NMR (125MHz, CDCl₃): δ 172.7, 82.9, 81.9, 74.1, 71.3, 56.2, 28.1, 26.6, 21.8.; HR-MS (ESI) m/z for C₁₅H₂₂NaO₃ [M+Na]⁺, calcd. 273.1461, found: 273.1460.

To a solution of **M2-1** (750mg, 3.0 mmol) in DCM (9 mL), 2,6-lutidine (1.4 mL, 12 mmol) was added and the mixture was cooled down to 0 °C, followed by the addition of triisopropylsilyl trifluoromethanesulfonate (1.1 mL, 6 mmol). After stirring overnight at room temperature, the mixture was quenched by saturated NH₄Cl aqueous solution. The organic layer was washed with NaCl aqueous solution, extracted by EtOAc, dried with MgSO₄, and concentrated. The product was purified by flash column chromatography on silica gel (hexane only) to afford **M2** as a colorless liquid (650 mg, 53% yield). ¹H-NMR (400 MHz, CDCl₃): δ 2.83 (q, J = 16.8 Hz, 4H), 1.98 (s, 2H), 1.47 (s, 9H), 1.44 (s, 6H), 1.09 (s, 21H).; ¹³C-NMR (150MHz, CDCl₃): δ 171.8, 82. 7, 81.7, 77.0, 70.4, 57.6, 28.1, 22.4, 18.6, 13.8.; HR-MS (ESI): m/z for C₂₄H₄₂NaO₃Si, [M+Na]⁺, calcd. 429.2795, found: 429.2797.

M7 (di(adamantan-1-yl) 2,2-di(prop-2-yn-1-yl)malonate)



Malonic acid (310 mg, 3.0 mmol) and 1-adamantanol (1.0 g, 6.6 mmol) were solvated in THF (15 ml). A mixture of *N*,*N*'-dicyclohexylcarbodiimide (1.4 g, 6.6 mmol) and 4-dimethylaminopyridine (36 mg, 0.30 mmol) in THF (15 mL) was slowly added at 0 °C. The mixture was stirred overnight at room temperature then quenched by acetic acid. After partially removing dicyclohexylurea (generated as a byproduct) by filtering, the organic layer was washed with water and extracted by DCM, dried with MgSO₄, and concentrated. The product was purified by flash column chromatography on silica gel (EtOAc:Hexane = 1:10) to afford **M7-1** as white solid (740 mg, 66% yield). ¹H-NMR (500 MHz, CDCl₃): δ 3.18 (s, 2H), 2.17 (s, 6H), 2.13 (s, 12H), 1.66 (s, 12H). ¹³C-NMR (150 MHz, CDCl₃): δ 166.09, 81.83, 44.78, 41.33, 36.28, 30.97. HR-MS (ESI) m/z for C₂₃H₃₂NaO₄ [M+Na]⁺, calcd. 395.2193, found: 395.2196.

Sodium hydride (60%, dispersion in mineral oil) (88 mg, 2.2 mmol) in THF (2 mL) was prepared at 0 $^{\circ}$ C in RBF purged with argon and a solution of **M7-1** (370 g, 1.0 mmol) in THF (1 mL) was added drop-wisely. After 10 minutes of stirring, propargyl bromide (80 wt%, in toluene) (0.50 mL, 2.5 mmol) was added and stirred for 2 hr. The reaction was quenched by adding NH₄Cl aqueous solution and the organic layer was extracted with EtOAc, dried with MgSO₄, and concentrated. The product was purified by flash column chromatography on silica gel (EtOAc:Hexane = 1:30) to afford **M7** as a white solid (410 mg, 91% yield). ¹H-NMR (500 MHz, CDCl₃): δ 2.88 (d, J = 2.6 Hz, 4H), 2.18 (s, 6H), 2.11 (d, J = 2.9 Hz, 12H), 2.02 (t, 2H), 1.66 (s, 12H). ¹³C-NMR (150 MHz, CDCl₃): δ 167.7, 82.3, 79.2, 71.5, 57.2, 41.2, 36.3, 31.0, 22.5.; HR-MS (ESI) m/z for C₂₉H₃₆NaO₄ [M+Na]⁺, calcd. 471.2506, found: 471.2509.

3. General procedure for the cyclopolymerization

A 5-mL sized sealed vial with septum was flame dried and charged with monomer and a magnetic bar. The vial was purged with argon four times, and degassed anhydrous DCM was added. After the Ar-purged catalyst (**Ru1** and **Ru2**) and pyridine additive in another 5-mL vial were dissolved in DCM, the solution was rapidly injected to the monomer solution at experimental temperature under vigorous stirring. The reaction was quenched by excess ethyl vinyl ether after desired reaction time, and partially precipitated in hexane or methanol at -78 °C, remaining small amount of crude mixture solution (~10%). Obtained solid was filtered and dried in vacuo. Monomer conversion was calculated from the ¹H NMR spectrum of the remaining crude mixture.

4. Calculation of the regioselectivity for P5 using ¹H and ¹³C NMR



<Figure S2. ¹³C NMR spectrum of **P5** from entry 1 in Table 3>

Composition of five-membered ring $=\frac{C_{a,5}}{C_{a,5}+C_{a,6}}$ or $\frac{C_{b,5}}{C_{b,5}+C_{b,6}}$ e.g. (entry 1 in Table 3) Composition of five-membered ring $=\frac{1}{16.35} = 0.061$ ($\therefore \beta$ -selectivity=94%)

5. In situ NMR experiment: procedure and data

Initiation experiment of Ru1

To an NMR tube was added a solution of **Ru1** (2.3 mg, 0.003 mmol) in 0.6 mL DCM-*d2*. The tube was then sealed with a rubber septum, taken out of the glovebox, and placed in a dry ice/acetone bath. Butyl vinyl ether (12 μ L, 0.090 mmol) was injected into the tube, and the reaction was monitored by observing the disappearance of the benzylidene signal by ¹H NMR using an array at the appropriate temperature.



<Figure S3. Plots of (a) -ln([Ru]/[Ru]₀) and (b) -ln([M]/[M]₀) vs. time for the CP of **M2** for M/I=20, with and without 3-CIPy, and (c) SEC traces of the resulting polymers>

Kinetic experiments using Ru1 or Ru2

Ru1 or **Ru2** (0.003 mmol, 1 eq) and hexamethyldisilane (internal standard, 3 μ l) were dissolved in DCM-*d2* (400 μ L). Initial benzylidene was measured by integral ratio of **Ru1** or **Ru2** to hexamethyldisilane in ¹H NMR spectrum. (After the addition of 4-7 eq of the pyridine additive,) Monomer (0.06 mmol, 20 eq) solution in DCM-*d2* (200 μ l) was added to the catalyst solution and mixed by shaking the NMR tube for 5 seconds. The reaction was monitored by ¹H NMR over time. The k_i or k_p values were obtained from the slope of linear – ln [**Ru**]/[**Ru**]₀ or – ln [M]/[M]₀ vs. time graphs, respectively.



<Figure S4. Plot of the propagating carbene vs. reaction time for **M5** and **M2** using **Ru1** under 3-CIPy as an additive>



<Figure S5. Plot of the conversion and initial Ru catalysts vs. reaction time for **M5** using **Ru1** (left) and **Ru2** (right) under 3,5-Cl₂Py as an additive>

6. SEC traces of the polymers



<Figure S6. SEC traces of P5s in Table 2>



<Figure S7. SEC traces of **P5**s synthesized in different conditions>



<Figure S8. SEC traces of P5s in Table 3, entries 1 and 2>



<Figure S9. SEC traces of P6s in Table 3, entries 3-6>



<Figure S10. SEC traces of P7s in Table 3, entries 12-15>



<Figure S11. SEC traces of **P8**s in Table 4, entries 1 and 2>



<Figure S12. SEC traces of **P9**s in Table 4, entries 3 and 4>



<Figure S13. SEC traces of P4s in Table 4, entries 5 and 6>



<Figure S14. SEC traces of P1s in Table 4, entries 7 and 8>



<Figure S15. SEC traces of P10s in Table 4, entries 9 and 10>



<Figure S16. SEC traces of P11s in Table 4, entries 11 and 12>

7. ¹H and ¹³C NMR characterization of polymers

The ¹H NMR and ¹³C NMR of **P1**, **P3-6**, **P8-11** are described in the literature.³

P2

¹H (500 MHz, CDCl₃): δ 7.04 – 5.62 (br m, 2H), 3.68 – 2.40 (br m, 4H), 1.42 (br s, 6H), 1.29 (br s, 9H), 1.11 (br s, 21H); ¹³C (150 MHz, CDCl₃): δ 172.9, 140.3, 137.5, 136.9, 134.0, 131.4, 80.3, 75.9, 57.8, 33.2, 30.6, 27.8, 18.8, 13.8.

P2₁₅-*b*-P3₁₅

¹H (500 MHz, CDCl₃): δ 7.04 – 5.62 (br m, 4H), 3.68 – 2.40 (br m, 8H), 1.42 (br s, 24H), 1.29 (br s, 9H), 1.11 (br s, 21H); ¹³C (150 MHz, CDCl₃): δ 172.9, 170.1, 137.4, 134.6, 133.2, 131.7, 128.1, 81.5, 80.3, 75.92, 57.8, 55.6, 35.3, 32.5, 30.7, 28.0, 18.8, 13.8.

P2₁₅-*b*-P4₁₅

¹H (500 MHz, CDCl₃): δ 7.04 – 5.62 (br m, 4H), 3.88 – 2.07 (br m, 12H), 1.39 (br s, 6H), 1.29 (br s, 9H), 1.11 (br s, 21H), 1.04 (br s, 42H); ¹³C (150 MHz, CDCl₃): δ 172.9, 140.4, 137.4, 135.9, 133.4, 132.3, 80.3, 75.9, 66.0, 57.8, 41.3, 33.6, 27.8, 18.8, 18.3, 13.8, 12.2.

P7

¹H (500 MHz, CDCl₃): δ 6.90 – 5.70 (br m, 2H), 3.50 – 2.59 (br m, 4H), 2.14 (br s, 6H), 2.08 (br s, 12H), 1.64 (br s, 12H); ¹³C (150 MHz, CDCl₃): δ 169.9, 134.8, 134.3, 133.4, 131.9, 81.4, 55.7, 41.1, 36.3, 31.0.

P915-b-P315

¹H (500 MHz, CDCl₃): δ 6.92 – 5.66 (br m, 2H), 3.39 (br s, 4H), 3.16 – 2.06 (br m, 8H), 1.42 (br s, 18H), 0.88 (br s, 18H), 0.00 (br s, 12H); ¹³C (150 MHz, CDCl₃): δ 170.2, 135.6, 134.7, 134.4, 133.4, 132.0, 128.0, 81.4, 65.7, 65.3, 55.7, 47.7, 40.6, 35.3, 33.5, 32.4, 30.9, 28.0, 26.1, 18.4, -5.4.

P3₁₅-*b*-P7₁₅

¹H (500 MHz, CDCl₃): δ 6.90 – 5.70 (br m, 4H), 3.50 – 2.59 (br m, 8H), 2.14 (br s, 6H), 2.08 (br s, 12H), 1.64 (br s, 12H), 1.42 (br s, 18H); ¹³C (150 MHz, CDCl₃): δ 170.1, 169.9, 134.7, 134.4, 133.4, 132.0, 81.4, 55.7, 55.6, 41.2, 36.3, 35.3, 32.4, 31.0, 28.0.

P3₁₅-*b*-P7₁₅-*b*-P11₁₅

¹H (500 MHz, CDCl₃): δ 6.90 – 5.63 (br m, 6H), 3.50 – 2.46 (br m, 20H), 2.14 (br s, 6H), 2.08 (br s, 12H), 1.64 (br s, 12H), 1.42 (br s, 18H), 1.08 (br s, 12H); ¹³C (150 MHz, CDCl₃): δ 170.1, 169.9, 134.9, 134.3, 133.4, 132.0, 81.4, 55.7, 55.6, 53.6, 41.1, 36.3, 35.3, 32.4, 30.9, 27.9, 14.0, 12.9.

8. ¹H and ¹³C NMR spectra of the polymers

¹³C NMR spectra were used for the determination of the ratio between five- and six-ring on the polymer backbone.

<P2 from Table 1>

^tBuC OTIPS Ŧ'n

¹H NMR (500 MHz, CDCl₃) ~ 6.86 √6.14 √5.89 √5.75 -- 3.44 -- 3.19 39 -2.718.35 / 11.81-26.83 1.82 4.00 6.5 8.0 7.5 5.0 4.0 f1 (ppm) 2.5 0.0 7.0 5.5 4.5 3.0 2.0 1.5 1.0 0.5 6.0 3.5







































$^{\rm 13}{\rm C}$ NMR (150 MHz, CDCl₃)













^{13}C NMR (150 MHz, CDCl₃)



























<P9-b-P3 from Scheme 3>



¹H NMR (500 MHz, CDCl₃)



- 170.2	135.6 134.7 133.4 133.4 128.0		65.7 65.3	55.7 47.7	7 40.6 35.3 33.5 32.4	~ 30.9 28.0 18.4 18.4	ц
				ł	1		
	140 130 120 110 100 90	80 I (ppm)	70 6	0 50		hullandandan 30 20 1	0 0 -10





^{13}C NMR (150 MHz, CDCl₃)







^{13}C NMR (150 MHz, CDCl₃)



9. ¹H and ¹³C NMR spectra of the monomers



¹H NMR (500 MHz, CDCl₃)











M7-1









$^{\rm 13}{\rm C}$ NMR (150 MHz, CDCl₃)



10. MALDI-TOF Spectrum of P2

The molar masses of polymer were measured by Bruker UltrafleXtreme TOF/TOF using dithranol in THF as a matrix. Matrix Assisted Laser Desorption/Ionization Time-of-Flight Mass Spectrometry (MALDI-TOF) spectrum supports that isopropoxy-styrene from **Ru1** remains at one end of the polymer chain, and terminal olefin at the other end of the polymer chain (end-capping was conducted with ethyl vinyl ether).



11. References

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