

Self-labeled magneto nanoprobes using tri-aminated polysorbate 80 for detection of human mesenchymal stem cells†

Eun-Kyung Lim,^a Jaemoon Yang,^b Jin-Suck Suh,^b Yong-Min Huh^{*b} and Seungjoo Haam^{*a}

Received 22nd June 2009, Accepted 23rd September 2009

First published as an Advance Article on the web 22nd October 2009

DOI: 10.1039/b912149h

Self-labeled magneto nanoprobes (SLMNs) were synthesized for one-step labeling of human mesenchymal stem cells (hMSCs) *via* non-invasive magnetic resonance (MR) image tracking. The surface of superparamagnetic nanocrystals (SMNCs) were enveloped by tri-aminated polysorbate 80 (TP80) not only for enhancement of the affinity for hMSCs, but also to achieve increased colloidal stability in the aqueous phase by the nanoemulsion method. The synthesized SLMNs exhibited excellent labeling efficiency of hMSCs and produced highly sensitive MR signals as well as safe cell cytotoxicity.

1. Introduction

Stem cell therapy has come of age for the treatment of various diseases through cell transplantation at defunct cell populations.^{1–7} The trafficking and monitoring of cell migration during stem cell therapy is an important issue for understanding the temporal and spatial fate and distribution of transplanted stem cells.^{1,3,4,6,8–13}

Because non-invasive magnetic resonance (MR) imaging can rapidly acquire multi-informational high resolution images, magnetic labeling of stem cells with super-paramagnetic iron oxide particles can provide critical information about the cellular migration of the stem cells.^{2,6,7,14,15}

For intracellular labeling agents, the FDA-approved ferumoxides have been conventionally available and can be complexed with polycationic transfection agents (*e.g.* poly-L-lysine (PLL), protamine sulfate (PS)). The positively charged surface properties of ferumoxides enable highly efficient cellular uptake through electrostatic interactions with the negatively charged cellular membranes.^{16–21}

However, we have noticed that the use of ferumoxide–PLL complexes results in oxidative stress and injury of the cells due to formation of reactive oxygen species (ROS) and hydroxyl-free radicals, which, may also alter cell metabolism or increase the rate of apoptosis or cell death.^{10,22,23} The intracellular labeling efficiency and long period tracking capability is dependent on the concentration of tracking agents, even after cell differentiation. However, the use of large amounts of labeling agents results in increased cellular toxicity.^{1,8,13}

The necessary and sufficient conditions for appropriate cellular labeling with MR contrast agents are (i) a high intracellular labeling efficiency, (ii) ultrasensitivity of magnetic

properties and (iii) maintenance of cellular-proliferation and differentiation capacity (biocompatibility).

Herein, we synthesized self-labeled magneto nanoprobes (SLMNs) by the nanoemulsion method using polycationic tri-aminated polysorbate 80 (TP80) which stabilizes and envelopes magnetic nanocrystals (MNCs) (Fig. 1). For enhancing the cationic property and stabilizing the capability in the aqueous phase, tri-aminated polysorbate 80 (TP80) was synthesized so that the terminal groups of polysorbate 80 would substitute the hydroxy groups with amine groups. We evaluated the physico-chemical properties of nanoparticles to confirm their nano-structures and characteristic features. Furthermore, we carefully investigated the cellular labeling and imaging efficiency of SLMNs using MR imaging and an *in vitro* MTT assay.

2. Experimental

2.1. Materials

Polysorbate 80 (polyoxyethylene sorbitan monooleate), ethylenediamine (EDA), 1,1'-carbonyldiimidazole (CDI), 1,4-dioxane

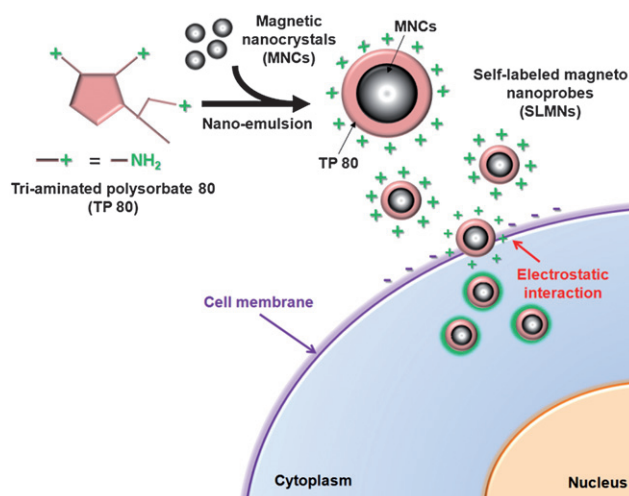


Fig. 1 Schematic of preparation of self-labeled magneto nanoprobes (SLMNs) for MR imaging.

^aDepartment of Chemical and Biomolecular Engineering, Yonsei University, Seoul 120-749, South Korea. E-mail: haam@yonsei.ac.kr; Fax: +82-2-312-6401; Tel: +82-2-2123-3554

^bDepartment of Radiology, College of Medicine, Yonsei University, Seoul 120-752, South Korea

† Electronic supplementary information (ESI) available: Supporting spectra. See DOI: 10.1039/b912149h

(99.8%), and poly-L-lysine were purchased from Sigma Aldrich Chemical Co. Ferumoxides (Feridex I.V.[®]) as commercial magnetic resonance (MR) imaging contrast agents were provided by the manufacturer (11.2 mg Fe/mL, Bayer HealthCare Pharmaceuticals Inc., Wayne, NJ, USA). Phosphate buffered saline (PBS; 10 mM, pH 7.4), DMEM and fetal bovine serum were purchased from Gibco. Vybrant[®] Apoptosis Assay Kit #3 was purchased from Molecular probes (Eugene, USA). 2,4,6-Trinitrobenzene sulfonic acid (TNBSA) was purchased from Pierce (Thermo scientific, USA) and dialysis membrane (MWCO: 1000) was obtained from Spectrum laboratory. Human mesenchymal stem cells (hMSCs) were purchased from Young@science. All other chemicals and reagents were of analytical grade.

2.2. Synthesis of MnFe₂O₄ nanocrystals (MNCs)

We synthesized monodispersed MnFe₂O₄ nanocrystals (MNCs), which are soluble in non-polar organic solvents.^{24,25} Briefly, 2 mmol of iron(III) acetylacetonate, 1 mmol of manganese(II) acetylacetonate, 10 mmol of 1,2-hexadecanediol, 6 mmol of dodecanoic acid, and 6 mmol of dodecylamine were dissolved in 20 mL of benzyl ether under an ambient nitrogen atmosphere. The mixture was then pre-heated to 200 °C for 2 h and refluxed at 300 °C for 30 min. After the reactants were cooled at room temperature, the products were purified with an excess of pure ethanol. Approximately 12 nm of MNCs were synthesized using the seed-mediated growth method.

2.3. Synthesis of tri-aminated polysorbate 80 (TP80)

To modify the surface of MNCs with amine groups, we first synthesized tri-aminated polysorbate 80 (TP80) using CDI as

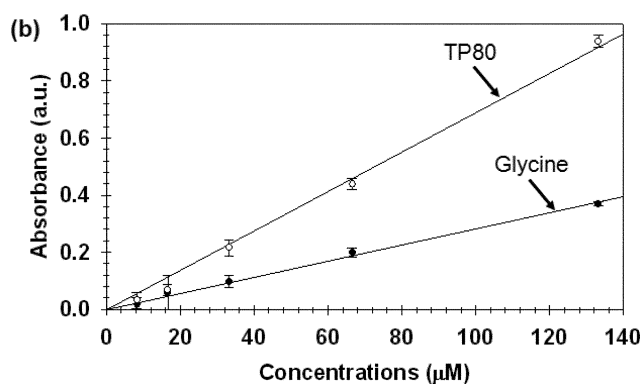
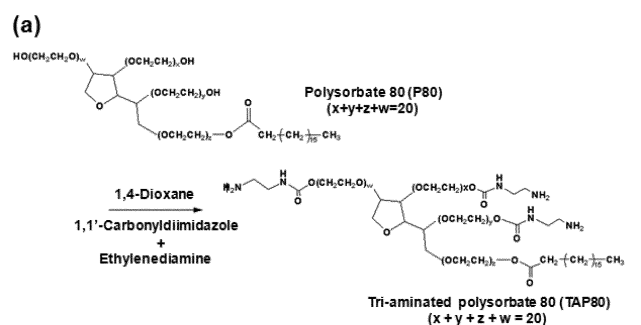


Fig. 2 (a) Synthetic diagram of tri-aminated polysorbate 80 (TP80) and (b) the determination of amine groups in tri-aminated polysorbate 80 (TP80) and glycine using TNBS assay.

a zero-length crosslinker (Fig. 2 (a)).^{26,27} 5 mmol of P80 and 15 mmol of CDI were dissolved in a flask containing 100 mL of 1,4-dioxane and the reactant solution was stirred for 1 h. Subsequently, 15 mmol of EDA was added into the reactor at room temperature. After further reaction for 24 h, the un-wanted solvent was rapidly eliminated using a rotary evaporator. The obtained transparent gel type product was then dissolved in 10 mL of deionized water and dialyzed against excess aqueous phase for over seven days. The purified product was freeze-dried and stored under vacuum for later use. The chemical structure of TP80 was characterized by ¹H-NMR (400 MHz, Varian INOVA400 NMR spectrometer) using CDCl₃ as a solvent. The characteristic bands of synthesized TP80 and P80 were confirmed by Fourier-transform infrared spectroscopy (FT-IR, Varian, Excalibur[™] series).

2.4. Determination of amine groups in TP80

To measure the primary amine groups in TP80, which were dissolved in 0.1 M sodium bicarbonate buffer (pH 8.5) at various concentrations (8.33–133.21 μM), 0.5 mL of TP80 solution was mixed with a 0.01% (w/v) TNBSA solution. Subsequently, the mixed solution was stored at 37 °C for 2 h. Then we added 0.25 mL of 1% sodium dodecyl sulfate (SDS) and 0.125 mL of 1 M hydrochloride (HCl) to the TP80 solution. As a control, glycine was prepared in the same manner. The absorbance of TP80 and glycine solution was analyzed using a UV-Vis spectrometer at 355 nm (Optizen 2120UV, MECASYS Co.).

2.5. Preparation of self-labeled magneto nanoprobes (SLMNs)

SLMNs were prepared by nanoemulsion method.²⁸ 50 mg of MNCs were dissolved in 4 mL of hexane (as the organic phase) and the organic phase was mixed with 20 mL of phosphate buffer saline (as the aqueous phase) containing 100 mg of TP80. After mutual saturation of organic and aqueous phases, the emulsion was ultra-sonicated in an ice-cooled bath for 10 min at 450 W. The organic solvent was evaporated overnight at room temperature and the products were purified by centrifugal filter (Centriprep YM-3, 3 kDa NMWL, Amicon and USA) with two cycles at 3000 g force for 1 h. The amount of amine groups in the SLMNs were measured by a TNBSA assay. The size distribution and zeta-potential of SLMNs were analyzed using laser scattering (ELS-Z, Otsuka electronics). The surface electrical charge of the prepared SLMNs was analyzed by gel electrophoresis in 1% agarose in buffer solution over a wide range of pH conditions (pH 3–11) at room temperature. The morphologies of SLMNs were confirmed using high resolution transmission electron microscope (HR-TEM, JEM-2100F, JEOL Ltd.). The magnetic properties of MNCs and SLMNs were evaluated using a vibrating-sample magnetometer (VSM; MODEL-7300, Lakeshore) at 298 K and the residual weight ratio (%) of the MNCs in the SLMNs was analyzed with a thermo-gravimetric analyzer (SDT-Q600, TA instrument).

2.6. Cell viability test of SLMNs against hMSCs

The cytotoxic effect of SLMNs against hMSCs was evaluated by measuring the inhibition of cell growth using MTT assay. Briefly, hMSCs were seeded at a density of 1×10^4 cells/well in a 96-well

plate and incubated overnight at 37 °C. hMSCs were then incubated with SLMNs at various concentrations for 24 h. The yellow tetrazolium salt of the MTT solution is reduced to purple formazan crystals in metabolically alive cells. The cell viabilities were obtained by calculating the ratio of the intensity of purple formazan in viable cells treated with SLMNs to the intensity in untreated control cells. Human mesenchymal stem cells (hMSCs) were seeded at a density of 1×10^6 cells/well in a culture flask for a week at 37 °C and incubated with $80 \mu\text{g mL}^{-1}$ of SLMNs with a medium containing serum for a further 24 h. Then hMSCs incubated with SLMNs were collected by centrifugation and washed twice with phosphate buffer saline (6.7 mM, pH 7.4).

Further to confirm whether labeled SLMNs caused cellular damage of hMSCs, *i.e.*, apoptosis and necrosis, we performed an annexin V-FITC/propidium iodide (PI) double staining test.^{29,30} After resuspending hMSCs in 100 μL of 1X annexin-binding buffer, 5 μL of annexin V-FITC solution and 1 μL of PI solutions were injected and incubated with hMSCs for 15 min at room temperature in a dark room. The fluorescence-stained hMSCs were analyzed by FACScalibur (Beckton-Dickinson, Mansfield, MA) by monitoring the fluorescence activated cell after addition of an additional 400 μL of 1X annexin-binding buffer.

2.7. Cellular labeling efficiency of SLMNs

Human mesenchymal stem cells (1×10^6 cells/well) were incubated for a week in a culture dish and were then further incubated with SLMNs ($80 \mu\text{g mL}^{-1}$) containing a medium with serum in 5% CO_2 for 24 h at 37 °C. Subsequently, hMSCs treated with SLMNs were washed twice with phosphate buffered solution (pH 7.4 and 10 mM), trypsinized (0.5 mM), and harvested. The collected hMSCs were washed in triplicate using 0.2% fetal bovine serum and 0.02% NaN_3 in phosphate buffered solution (pH 7.4 and 10 mM). The MR imaging capability of SLMNs was estimated by MR imaging, Prussian blue staining (BX51 upright microscope, Olympus) and transmission electron microscopy (TEM, JEOL-1100). Furthermore, the concentration of hMNCs (Fe + Mn) of labeled SLMNs into hMSCs was determined by inductively coupled plasma atomic emission spectroscopy (ICP-AES, Elan 6100, Perkin Elmer).

2.8. *In vitro* MR imaging procedure

All MR imaging experiments were performed with a 1.5 T clinical MRI instrument with a micro-47 surface coil (Intera; Philips Medical Systems, Best, the Netherlands). The T2 weights of the SLMNs solution and ferumoxides and hMSC treated with SLMNs were measured by the Carr–Purcell–Meiboom–Gill (CPMG) sequence at room temperature: TR = 10 s, 32 echoes with 12 milliseconds even echo space, number of acquisition = 1, point resolution of $156 \times 156 \mu\text{m}$, section thickness of 0.6 mm. For T2-weighted MR imaging of the samples, the following parameters were adopted: resolution of $234 \times 234 \mu\text{m}$, section thickness of 2.0 mm, TE = 60 milliseconds, TR = 4000 milliseconds and the number of acquisitions = 1. The relaxivity coefficient ($\mu\text{M}^{-1} \text{s}^{-1}$) is equal to the ratio of R2 ($1/T2$, s^{-1}) to SLMNs and ferumoxides concentration.

2.9. Statistical analysis

All experiments were performed in triplicate and the values are expressed as the average \pm standard deviation. The statistical evaluation of the data was performed by analysis of variance (ANOVA), followed by Turkey's test for comparison of cell viability, and relative T2 signal in MR image from different groups. A $p < 0.01$ was considered to be statistically significant.

3. Results and discussion

For the synthesis of SLMNs, the tri-hydroxyl group of polysorbate 80 (P80) was first modified with amine groups using ethylenediamine (EDA) and 1,1'-carbonyldiimidazole (CDI) as a zero-length cross linker. The synthetic scheme of tri-amine terminated polysorbate 80 (TP80) is depicted in Fig. 2 (a). First, the tri-hydroxyl groups of P80 were activated with CDI, and the conjugated imidazole group of activated P80 was attacked by the amine group of EDA. Finally, a stable carbamate bond was formed by coupling the hydroxyl group from P80 and an amine group from EDA.^{26,27} After the conjugation process, the chemical structure of TP80 was characterized using $^1\text{H-NMR}$; δ values of 2.0 ($-\text{NH}_2$), 2.91, 3.22 ($-\text{CH}_2-$) and 7.11 ($-\text{NH}-$) (ESI \dagger). Furthermore, the characteristic bands of TP80 were verified by FT-IR spectra; 3550 cm^{-1} (N–H stretching of amine group), 3400 cm^{-1} (C–N stretching of an amide group), 2400 cm^{-1} (O–CO–CH₃ stretching) and 1600 cm^{-1} (N–H bending of and amine group) (ESI \dagger). Otherwise, the hydroxyl stretching at 3500 cm^{-1} from TP80 was no longer observed after modification of P80 terminal groups. The increment of molecular weights for TP80 was further confirmed using gel permeation chromatography (GPC). The retention time of tri-amine terminated polysorbate 80 was decreased compared to non-modified P80 due to conjugated EDA (ESI \dagger). The absorbance of the reactants was detected in proportion to the amount of amine groups ($-\text{NH}_2$) because TNBS (2,4,6-trinitrobenzene sulfonic acid) was reacted with an amine group and produced a chromogenic derivative. As shown in Fig. 2 (b), the absorbance of TP80 was three times higher in comparison with the one of glycine through the TNBS assay. These results confirmed that the tri-hydroxyl group of P80 were successfully modified with tri-amine groups.

To utilize the MR capability of SLMNs, the hydrophobic magnetic nanocrystals (MNCs) were encapsulated with amphiphilic TP80. The size and the zeta-potential value of SLMNs measured by laser scattering were $40.73 \pm 4.63 \text{ nm}$ and $25.36 \pm 1.87 \text{ mV}$, respectively. The size of the SLMNs was larger than the hydrophobic magnetic nanocrystals because the hydrophilic parts ($-\text{OCH}_2\text{CH}_2-$) of tri-aminated polysorbate 80 (TP80) on the surface of SLMNs were conjugated with water molecules *via* hydrogen bonding. Furthermore, the TEM image presented well dispersed SLMNs in the aqueous phase without any precipitation and aggregation (ESI \dagger).

The size of SLMNs was smaller compared to ferumoxides (commercial MR imaging agents, $62.30 \pm 2.94 \text{ nm}$) ($*p < 0.01$). Because of the large amounts ($2.78 \times 10^{20} \text{ ea/mL}$) of amine groups ($-\text{NH}_2$) on the surface of SLMNs, the zeta-potential value exhibited a strong positive charge compared with ferumoxide ($-24.79 \pm 0.37 \text{ mV}$) (Fig. 3 (a)).

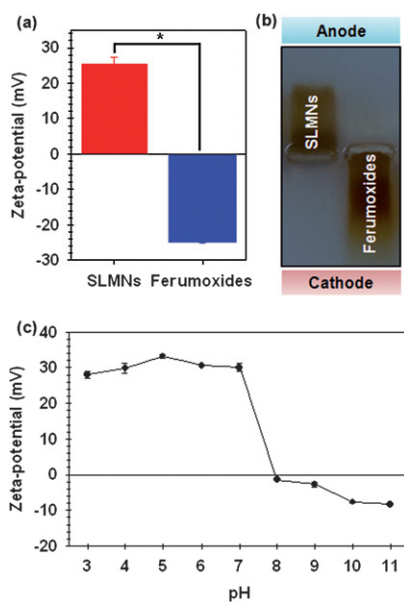


Fig. 3 (a) Size distribution and zeta-potential, (b) gel electrophoresis analysis of self-labeled magneto nanoprobes (SLMNs) and ferumoxides and (c) zeta-potential of SLMNs against pH condition (* $p < 0.01$).

Since cell membranes are mainly composed of phospholipids, which contain negatively charged phosphate groups, strong positively charged SLMNs can enhance cellular affinity and the labeling efficiency by electrostatic interaction. Hence, we envisage that SLMNs are suitable labeling agents even without the use of transfection agents. The nitrogen atom of the amine group featured a lone electron pair that can bind with hydrogen protons (H^+) of the water molecules to form an ammonium ion (NH_3^+).³¹ In Fig. 3 (b), gel electrophoresis analysis revealed that the band of SLMNs was shifted towards the anode (downward) due to the positive surface charge of SLMNs when 100 V of voltage was applied to the gel matrix. Furthermore, we confirmed the variation of zeta-potentials and stability of SLMNs according to various pH conditions (Fig. 3 (c) and ESI†). Because the nitrogen atoms of amine groups interacted with the hydrogen protons (H^+) in acidic condition, the surface of the SLMNs exhibited a positive charge below pH 8. In basic conditions, the nitrogen atoms were bereft of hydrogen ions (H^+) in the amine groups, and the surface charges were negative above pH 8. The results indicated that SLMNs exhibit positive charges under biological conditions.³²

While the amount of encapsulated MNCs in SLMNs was 70 wt% through thermogravimetric analysis, SLMNs exhibited superparamagnetic behavior at 298 K without remanence and coercivity and the saturation of magnetization at 0.8 T of SLMNs was 32.28 emu/g_{MNCs}, which was similar to that of naked MNCs (39.76 emu/g_{MNCs}) (Fig. 4).

To evaluate the utility of SLMNs as ultrasensitive labeling agents *via* MR imaging, we investigated MR images of the SLMN solutions and the signal intensities of SLMNs and then compared them with ferumoxide as a commercial MR imaging agent. The SLMN solution resulted in a stronger black color T2-weighted MR images and the corresponding relaxivity coefficient was 383.73 $\mu M^{-1} s^{-1}$ indicating that SLMNs were

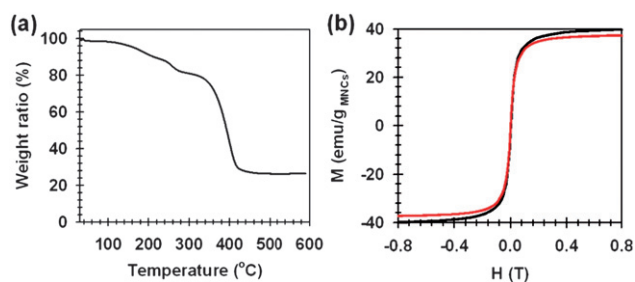


Fig. 4 (a) Thermogravimetric analysis of self-labeled magneto nanoprobes (SLMNs) and (b) magnetic hysteresis loops of MNCs (black solid line) and SLMNs (red solid line).

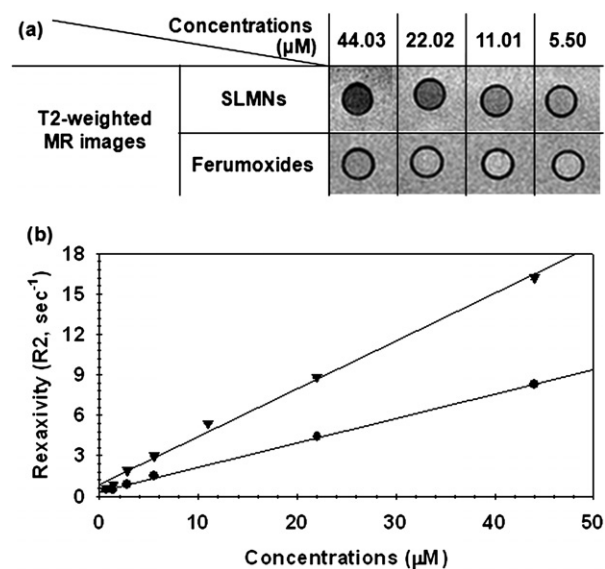


Fig. 5 (a) T2-weighted MR images and (b) relaxivity (R_2) graph *versus* various concentrations of self-labeled magneto nanoprobes (SLMNs) (\blacktriangledown) compared to ferumoxides (\bullet).

sufficiently sensitive labeling agents as compared with ferumoxide ($190.47 \mu M^{-1} s^{-1}$) (Fig. 5).

As shown in Fig. 6 (a), *in vitro* cytotoxic activity measured by MTT assay showed that the cell viability of SLMNs retained over 80% of the concentration of 100 μg Fe + Mn/mL. In addition, we performed the biocompatibility test of SLMNs against hMSCs through the annexin V-FITC/propidium iodide (PI) double stain method to investigate whether SLMNs may induce apoptosis. In apoptotic cells, phosphatidylserine (PS) was exposed to the outer cell membrane and annexin V had a high affinity to PS as a phospholipid binding protein, since the integrity of the cell membrane was breaking. PI was also used for distinction of dead or damaged cells, because PI penetrates into the membrane of dead or damaged cells. In Fig. 6 (b), SLMN (80 μg Fe + Mn/mL)-treated hMSCs were detected in the lower left quadrant (99.51%) and did not stain with either annexin V-FITC and PI, which is comparable to results with non-treated control cells (97.21%). Also, the rate of apoptotic and necrotic cells did not differ between SLMN-treated and non-treated cells. The results demonstrate that SLMNs do not induce apoptosis and necrosis in hMSCs.

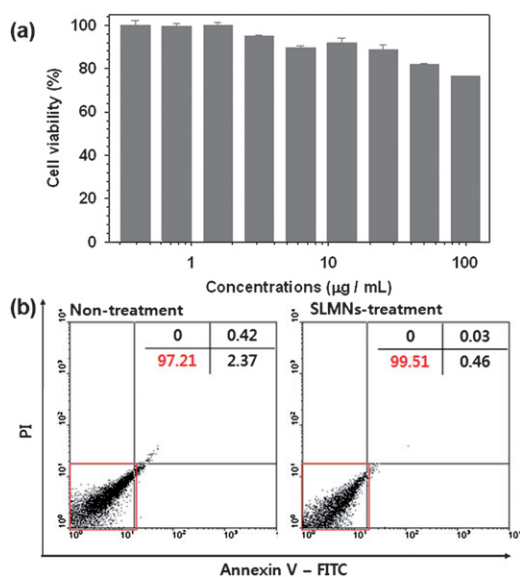


Fig. 6 (a) Cell viability of human mesenchymal stem cells (hMSCs) treated with various concentrations of self-labeled magneto nanoprobes (SLMNs) and (b) biocompatibility test of hMSCs treated with SLMNs using apoptosis detection assay (annexin V-FITC/PI double stain method); the numbers in all of the FACS data represent the percentages of each cell death mechanism: the percentages (%) of cells suffering necrotic death (annexin V⁻/PI⁺; upper left), apoptotic cells (annexin V⁺/PI⁻; lower right), late apoptotic or secondary necrotic death (annexin V⁺/PI⁺; upper right) and live cells (annexin V⁻/PI⁻; lower left).

We next evaluated the labeling efficiency of SLMNs against hMSCs through MR contrast effect at 1.5 T. In the T2-weighted MR images, a significant MR contrast effect was observed in SLMN-treated cells (black color) compared to non-treated control cells (bright gray color) (Fig. 7 (a)). The calculated R2 relaxivity value of hMSCs treated with SLMNs exhibited consistent results showing 253.81 s^{-1} (1408.12%, $\Delta R2/R2$ non-treatment), which is much higher than that for non-treated cells ($*p < 0.01$, Fig. 7 (a)). Furthermore, the results from Prussian blue staining and cellular TEM images showed similar results. Prussian blue and ferric ions of the MNCs rapidly exchange electrons thereby producing dark blue colors in the intra-cellular region. In Fig. 7 (b), the image of SLMN-treated hMSCs demonstrated significant MNC uptake manifested as numerous blue spots, indicating MNCs in the cytoplasm. In contrast, no dark blue spots were observed for the non-treated cells. In the cellular TEM image, the black clusters of hMSCs treated with SLMNs indicated the engulfment of SLMNs into the intra-cellular region (Fig. 7 (c)). The content of the MNCs was $0.19 \mu\text{g Fe + Mn}$ per 5.0×10^5 cells (measured by ICP-AES). Fig. 7 (d) shows the strongly gathered SLMN-treated hMSCs upon external magnetic field (Nd-B-Fe magnet, 0.3 T) for 1 min indicating the excellent magnetic properties of SLMNs.

Through the varieties of physical, chemical and biological experiments, the synthesized SLMNs demonstrated sufficient capability as ultrasensitive self-labeling and tracking agents *via* MR imaging.

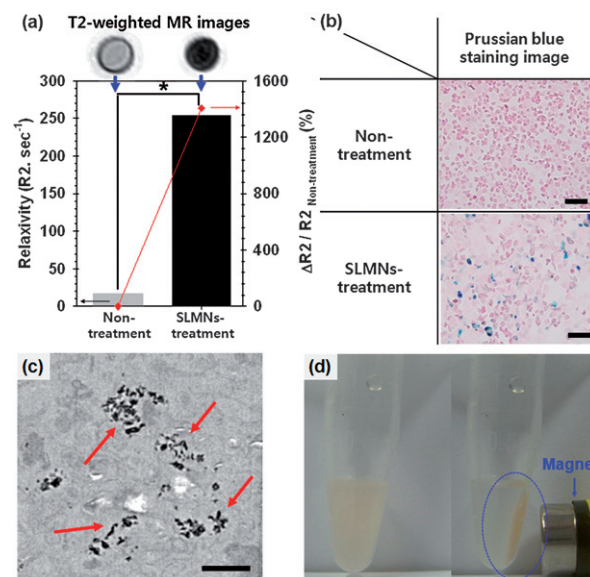


Fig. 7 (a) T2-weighted MR images and relaxivity (R2) graph (b) Prussian blue staining images, (c) TEM images of human mesenchymal stem cells (hMSCs) treated with self-labeled magneto nanoprobes (SLMNs) ($80 \mu\text{g Fe + Mn/mL}$). The red arrow of TEM images indicates the existence of SLMNs in hMSCs. And (d) dispersed in a microtube with a Nd-B-Fe magnetic field of hMSCs treated with SLMNs for 1 min (scale bar: 100 μm . $*p < 0.01$).

Conclusions

For cell tracking and monitoring *via* MR imaging, we successfully synthesized self-labeled magneto nanoprobes (SLMNs) for detection of human mesenchymal stem cells (hMSCs) without using additional transfection agents. Efficient and successful encapsulation of MNCs using tri-aminated polysorbate 80 (TP80) without loss of magnetic sensitivity produces highly water soluble nanoparticles as well as strong positively charged surface characteristics. These biologically safe and cationic SLMNs may be a potentially powerful and efficient cell tracking nanoprobe to use with MR imaging.

Acknowledgements

This study was supported by the Korea Science and Engineering Foundation (KOSEF) (No. M10641550001-08N4155-00110, M10755020001-08N5502-00110), a grant of the Korea Healthcare technology R&D Project, Ministry for Health, Welfare & Family Affairs, Republic of Korea. (A0851362), and grant from KRIBB Research Initiative Program.

Notes and references

- C.-W. Lu, Y. Hung, J.-K. Hsiao, M. Yao, T.-H. Chung, Y.-S. Lin, S. H. Wu, S.-C. Hsu, H.-M. Liu, C.-Y. Mou, C.-S. Yang, D.-M. Huang and Y.-C. Chen, *Nano Lett.*, 2007, 7, 149.
- H. Kalish, A. S. Arbab, B. R. Miller, B. K. Lewis, H. A. Zywicke, J. W. M. Bulte, L. H. Bryant Jr. and J. A. Frank, *Magn. Reson. Med.*, 2003, 50, 275.
- B. S. Shah, P. A. Clark, E. K. Moiola, M. A. Strosio and J. J. Mao, *Nano Lett.*, 2007, 7, 3071.

- 4 T. D. Jeff, W. M. Bulte, B. Witwer, S.-C. Zhang, E. Strable, B. K. Lewis, H. Zywicke, B. Miller, P. Gelderen, B. M. Moskowitz, I. D. Duncan and J. A. Frank, *Nat. Biotechnol.*, 2001, **19**, 1141.
- 5 K. S. Cahill, G. Gaidosh, J. Huard, X. Silver, B. J. Byrne and G. A. Walter, *Transplantation*, 2004, **78**, 1626.
- 6 L. Wang, J. Deng, J. Wang, B. Xiang, T. Yang, M. Gruwel, T. Kashour, B. Tomanek, R. Summer, D. Freed, D. S. Jassal, G. Dai, M. Glogowski, R. Deslauriers, R. C. Arora and G. Tian, *Magn. Reson. Imaging*, 2009, **27**, 108.
- 7 H. E. Rice, E. W. Hsu, H. Sheng, D. A. Evenson, A. J. Freerman, K. M. Safford, J. M. Provenzale, D. S. Warner and G. A. Johnson, *AJR, Am. J. Roentgenol.*, 2007, **188**, 1101.
- 8 J. A. Frank, B. R. Miller, A. S. Arbab, H. A. Zywicke, E. K. Jordan, B. K. Lewis, L. H. Bryant, Jr. and J. W. M. Bulte, *Radiology*, 2003, **228**, 480.
- 9 B.-H. Park, J.-C. Jung, G.-H. Lee, T.-J. Kim, Y.-J. Lee, J.-Y. Kim, Y. W. Kim, J.-H. Jeong and Y. Chang, *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 2008, **145**, 313–314.
- 10 A. S. Arbab, G. T. Yocum, L. B. Wilson, A. Parwana, E. K. Jordan, H. Kalish and J. A. Frank, *Mol. Imaging*, 2004, **3**, 24.
- 11 D. L. Kraitchman and J. W. M. Bulte, *Basic Res. Cardiol.*, 2008, **103**, 105.
- 12 S. Ju, G. Teng, Y. Zhang, M. Ma, F. Chen and Y. Ni, *Magn. Reson. Imaging*, 2006, **24**, 611.
- 13 H.-T. Song, J.-s. Choi, Y.-M. Huh, S. Kim, Y.-w. Jun, J.-S. Suh and J. Cheon, *J. Am. Chem. Soc.*, 2005, **127**, 9992.
- 14 J. Y. S.-B. Seo, E.-S. Lee, Y. Jung, K. Kim, S.-Y. Lee, D. Kim, J.-S. Suh, Y.-M. Huh and S. Haam, *J. Mater. Chem.*, 2008, **18**, 4402.
- 15 X. S. K. S. Cahill, G. Gaidosh, B. J. Byrne and G. A. Walter, *Proc. Intl. Soc. Mag. Reson. Med.*, 2003, **11**, 368.
- 16 A. S. Arbab, G. T. Yocum, H. Kalish, E. K. Jordan, S. A. Anderson, A. Y. Khakoo, E. J. Read and J. A. Frank, *Blood*, 2004, **104**, 1217.
- 17 E. Pawelczyk, A. S. Arbab, S. Pandit, E. Hu and J. A. Frank, *NMR Biomed.*, 2006, **19**, 581.
- 18 A. S. Arbab, G. T. Yocum, A. M. Rad, A. Y. Khakoo, V. Fellowes, E. J. Read and J. A. Frank, *NMR Biomed.*, 2005, **18**, 553.
- 19 S. A. Anderson, J. Glod, A. S. Arbab, M. Noel, P. Ashari, H. A. Fine and J. A. Frank, *Blood*, 2005, **105**, 420.
- 20 J. Yang, E.-K. Lim, E.-S. Lee, J.-S. Suh, S. Haam and Y.-M. Huh, *J. Biotechnol.*, 2008, **136**, S194.
- 21 H. Sadek, S. Latif, R. Collins, M. G. Garry and D. J. Garry, *Regener. Med.*, 2008, **3**, 807.
- 22 A. S. Arbab, L. A. Bashaw, B. R. Miller, E. K. Jordan, B. K. Lewis, H. Kalish and J. A. Frank, *Radiology*, 2003, **229**, 838.
- 23 M. Babic, D. Horak, P. Jendelova, K. i. Glogarova, V. t. Herynek, M. Trchova, K. n. Likavcanova, P. Lesny, E. Pollert, M. Hájek and E. Sykova, *Bioconjugate Chem.*, 2009, **20**, 283.
- 24 Y. M. Huh, Y. w. Jun, H. T. Song, S. Kim, J. s. Choi, J. H. Lee, S. Yoon, K. S. Kim, J. S. Shin, J. S. Suh and J. Cheon, *J. Am. Chem. Soc.*, 2005, **127**, 12387.
- 25 S. Sun, H. Zeng, D. B. Robinson, S. Raoux, P. M. Rice, S. X. Wang and G. Li, *J. Am. Chem. Soc.*, 2004, **126**, 273.
- 26 C. Yang, X. Wang, H. Li, S. H. Goh and J. Li, *Biomacromolecules*, 2007, **8**, 3365.
- 27 G. S. Bethell, J. S. Ayers, W. S. Hancock and M. T. W. Hearn, *The journal of biological chemistry*, 1979, **254**, 2572.
- 28 J. Yang, S. B. Park, H.-G. Yoon, Y. M. Huh and S. Haam, *Int. J. Pharm.*, 2006, **324**, 185.
- 29 S.-K. Yang, S. Attipoe, A. P. Klausner, R. Tian, D. Pan, T. A. Rich, T. T. Turner, W. D. Steers and J. J. Lysiak, *J. Urol.*, 2006, **176**, 830.
- 30 Y.-H. Chen, H.-Y. Chen, C.-L. Hsu and G.-C. Yen, *J. Agric. Food Chem.*, 2007, **55**, 1743.
- 31 D.-B. Shieh, F.-Y. Cheng, C.-H. Su, C.-S. Yeh, M.-T. Wu, Y.-N. Wu, C.-Y. Tsai, C.-L. Wu, D.-H. Chen and C.-H. Chou, *Biomaterials*, 2005, **26**, 7183.
- 32 N. C. B. M. Vamvakaki, S. P. Armes, J. F. Watts and S. J. Greaves, *J. Mater. Chem.*, 2001, **11**, 2437.