Probing quantum confinement of single-walled carbon nanotubes by resonant soft-x-ray emission spectroscopy

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(Received 13 March 2008; accepted 25 June 2008; published online 15 July 2008)

We report the band-structure changes near Fermi level for single-walled carbon nanotubes (SWNTs) with diameters down to 1 nm from the study of soft-x-ray absorption and resonant emission spectroscopy. The observed quantum confinement of SWNTs affects both π and σ bands and bandgap through the rehybridization of π and σ orbitals. The significant changes of electronic structure are proved to be a measure for the mean diameter of the macroscopic amounts of SWNTs. © 2008 American Institute of Physics. [DOI: 10.1063/1.2959058]

Carbon nanotubes (CNTs) have attracted much attention due to their variety of potential applications. Extensive research efforts have focused on the electronic structure.² The common model is graphene zone folding, which imposes periodical boundary conditions on the two-dimensional energy dispersion of graphene.³ However, such a model neglects the finite curvature of tube wall. Carbon atom has six electrons that occupy 1s, 2s, and 2p atomic orbitals. The 2s and 2patomic orbitals can readily mix to form sp^2 and sp^3 hybrids. In both graphite and CNTs, the hybridization is sp^2 type. In graphite, the π orbitals are perpendicular to the graphene plane, while in CNTs, the π orbitals are oriented along the normal to nanotube surface and thus are not parallel to each other due to the surface bending. They mix with σ orbitals of the neighboring sites, and such mixing would affect the band structure of CNTs. Optical absorption spectroscopy and scanning tunnel microscopy^{5–7} have shown the electronic structure changes near Fermi level along with the curvature for single nanotube. The curvature effect opens a small bandgap for metallic nanotubes.⁵ A semiconductor tube has a bandgap inversely proportional to its diameter d while the chiral metallic and zigzag metallic tubes are predicted to have narrow-gap scaling by $1/d^{2.5-7}$ The chemical reactivity showed dependence on the tube diameter or curvature. Although the electronic structure near Fermi level has been widely investigated, the curvature effect on the whole band structure and especially the σ bands is still largely unknown. In this letter we show the band structure changes in both π and σ bands using soft-x-ray absorption (XAS) and emission spectroscopy (XES), and the results indicate that the rehybridization of π and σ orbitals affects both π and σ bands in single-walled CNTs (SWNTs).

XAS probes the unoccupied density of states (DOS) and XES probes the occupied DOS. In addition to the inherent elemental selectivity of x-ray spectra, resonant excitation al-

lows separation of features that pertain to different elements or same element in different chemical environments. ¹⁰ Today the method is frequently applied in research fields ranging from atomic and molecular physics to materials science. The study of the quantum confinement of SWNTs demonstrates the important potential for investigating the electronic structure of SWNTs.

The XAS and XES experiments were performed at BL7 of ALS (Ref. 12) and the XES spectra were collected using a Nordgren-type spectrometer. ¹³ The resolution was set to 0.2 and 0.3 eV for XAS and XES measurements, respectively. The resolution of the spectrometer was set to 0.3 eV. The incident angle was 20° to sample surface in order to minimize the self-absorption effect with the spectrometer orthogonal to incoming photon beam. The XAS spectra were recorded using total electron yield mode with the energy scale calibrated to the π^* peak at 285.5 eV in XAS of highly oriented pyrolitic graphite. ¹⁴

Three SWNT samples were used in the experiments labeled as SWNT1, SWNT2, and SWNT3. SWNT1 was synthesized by a hydrogen arc discharge method¹⁵ and purified with 90% SWNTs in volume. SWNT2 was prepared by optimal floating chemical vapor deposition method¹⁶ and the purified sample contained over 70 wt % CNTs measured by thermogravimetric analysis. SWNT3 was purified HiPco SWNTs purchased from Carbon Nanotechnologies Incorporated¹⁷ (CNI) with purity better than 85 wt %. SWNT1 has a diameter distribution of 1.78-2.06 nm from Raman spectra with the mean diameter of 1.85 nm. SWNT2 has a diameter distribution of 1.08-1.25 nm from Raman spectra with the mean diameter of 1.15 nm. SWNT3 has a diameter distribution of 0.8–1.2 nm measured by CNI from transmission electron microscopy (TEM) micrographs with the mean diameter of 1.0 nm. The mean length is about several microns for SWNT1 and SWNT2 and 100 to 1000 nm for SWNT3 based on TEM images.

Figure 1(a) shows the C K-edge XAS spectra of SWNTs of different diameters. All spectra have been normalized to

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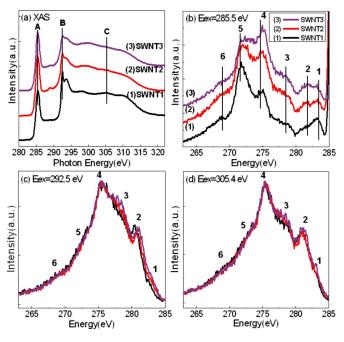


FIG. 1. (Color online) C *K*-edge absorption spectra and emission spectra of SWNT1, SWNT2, and SWNT3. (a) C *K*-edge absorption spectra (XAS) and C *K*-edge emission spectra at different excitation energies of (b) 285.5 eV,(c) 292.5 eV and (d) 305.4 eV.

the pre-edges and postedges. The absorption features labeled to A and B at 285.5 and 292.5 eV are attributed to the transitions from C 1s level to C–C π^* and σ^* states, respectively. Along with the increasing diameter of SWNTs, the XAS spectra show no significant angular dependence on the intensity ratio of π^*/σ^* . However, the σ^* feature of exciton nature becomes less prominent as the diameter of SWNTs reduces.

Figures 1(b)–1(d) shows XES spectra of SWNTs at excitation energies of (b) 285.5 eV, (c) 292.5 eV, and (d) 305.4 eV, respectively. Six main features (labeled with 1–6) are readily seen. The XES spectra at normal excitation energy have been shown in Fig. 1(d). No significant change is observed from the SWNTs of different diameters. The spectra at normal excitation energy are the sum of different orbitals and symmetries, leading to less symmetry dependence and momentum selectivity. The XES spectra excited on the σ^* state shown in Fig. 1(c) show no significant difference.

The big changes appear in Fig. 1(b), where the resonant XES spectra of SWNTs of different diameters at π^* excitation energy of 285.5 eV are displayed. All emission peaks show strong intensity variations upon SWNT's diameter. In addition, a shift to lower energy for peak1 and peak2 near Fermi level is clearly seen when the diameter decreases from SWNT1 (1.85 nm) to SWNT3 (1.0 nm).

To clarify the changes in band structure, Fig. 2 shows the curve fitting results in (a) for the energy positions of peak1 and peak2 and in (b) for the intensity ratio of peak5/peak4. The pictures on top of both Figs. 2(a) and 2(b) show the peak fitting curves for the corresponding peaks (SWNT2 as an example). The inset on top of Fig. 2(a) also shows the whole XES spectrum with elastic peak. In Fig. 2(a), the peak positions of peak1 and peak2 shift to lower energy when the SWNT's diameter decreases. Resonant inelastic x-ray scattering (RIXS) theory predicts that the control over excitation energy allows one to select the crystal momentum of the

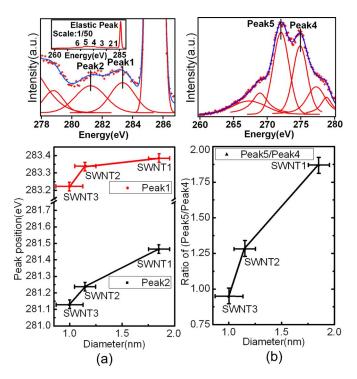


FIG. 2. (Color online) (a) The energy positions of peak1 and peak2, and (b) the intensity ratio of peak5/peak4 at the excitation energy of 285.5 eV vs the diameter of SWNT1, SWNT2, and SWNT3. The panels on top of both (a) and (b) show the peak fitting curves for the corresponding peaks (SWNT2 as an example).

photoelectron's final state in the conduction band. 19 The excitation energy of 285.5 eV is just above Fermi level and the corresponding emission spectra show the band structure near K point. According to previous reports, core hole effects will contribute to the π^* feature in XAS of graphite. However, The final state in RIXS consists of a hole in the valence band and an electron in the conduction band, so energetic considerations for allowed emission lines imply that transition process is largely an interband transition, unaffected by core-hole final-state effects. ^{21,22} The excitation energy of π^* peak can be used to excite the unoccupied electronic structure as shown in theory²² and experiments.^{9,19} The down shift of peak1 and peak2 near Fermi level indicates the lowering of π band near K point due to the curvature. For SWNTs, the π orbitals are not parallel to each other due to the curved surface and will hybridize with the σ orbitals. The band structure thus can be modified due to the rehybridization of π and σ orbitals. The lowering of π valence band with the increased curvature (reduced diameter) in the resonant XES spectra indicates the bandgap opening for metallic or semiconductor nanotubes.

The peak intensity evolution in σ band due to the curvature difference is presented in Fig. 2(b). The intensity of peak4 increases while peak5 decreases when the tube curvature increases. Peak4 is even stronger than peak5 when the tube diameter is down to 1.0 nm. From SWNT1 to SWNT3, it shows a dramatic diameter dependence for the ratio of peak5/peak4. Peak4 and peak5 are σ band features. Prominent intensity changes of peak4/peak5 evident the σ band evolution due to different diameters. Previously, less study has focused on the electronic structure of σ bond because of the constant bond length of σ bond in SWNTs of different diameters. Most of the calculations discussed only the bandgap changes. This result indicates that the σ bond can also be

beingy allows one to select the crystal momentum of the gap changes. This result indicates that the σ bond can also be Downloaded 10 Dec 2009 to 163.13.32.114. Redistribution subject to AIP license or copyright; see http://apl.aip.org/apl/copyright.jsp

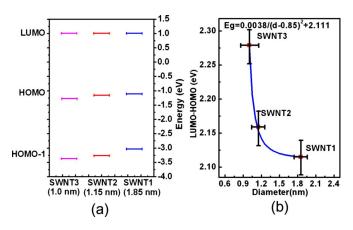


FIG. 3. (Color online) (a) The energy levels of LUMO, HOMO, and HOMO-1 for SWNT1, SWNT2, and SWNT3. (b) The energy difference between LUMO and HOMO for three samples and the fitting curve.

strongly affected by the curvature bending. The possible origins are the bond angle difference or other structural distortion and so further calculation insight would be interesting. It is an intrinsic property of the tubes and the spectral behavior can be used for the characterization of tube diameter. Interestingly, the ratio of peak5/peak4 does not change at different experimental geometry because of σ band properties. Thus it could be an effective way for determining the mean diameters of large amounts of SWNTs.

In regards to the changes near Fermi level, we show in Fig. 3(a) the energy levels of lowest unoccupied molecular orbital (LUMO), highest occupied molecular orbital (HOMO), and HOMO-1 for SWNT1, SWNT2, and SWNT3 derived from the XAS and XES spectra. To simplify the analysis process, we choose the lowest peak in XAS (π^*) as LUMO, the highest peak in XES (peak1) as HOMO and the next peak for HOMO-1. The LUMO shows nearly no energy shift for different SWNTs while the HOMO and HOMO-1 show a shift to lower energy with decreased diameter. The energy difference between LUMO and HOMO reflects the bandgap (E_{o}) of SWNTs and thus Eg (LUMO-HOMO) versus d (diameter) can reflect the change for SWNTs with different diameters. Figure 3(b) shows the energy difference for the three SWNTs samples and the corresponding curve fitting. The fitting result shows a relation between bandgap and SWNT's diameter with $E_g = 0.0038/(d - 0.85)^2 + 2.111$. The increased curvature alters the π -electron wave function and changes the bandgap or spike like density of states due to the van Hove singularity.

In conclusion we have investigated the quantum confinement of SWNTs on bonding and rehydridization of π and σ orbitals and bandgap change by using resonant x-ray emission spectroscopy. At the excitation energy of 285.5 eV (π^*), the emission spectral features change drastically due to the surface bending of SWNTs. The changes can be related to

the π and σ band structure evolution near K point. The results indicate that the rehybridization of π and σ orbitals has changed both π and σ bonds in SWNTs, and elude a possible way of determining the mean tube diameter for macroscopic amounts of SWNTs.

We thank Yi Luo for the fruitful discussion. Z.Y.W. acknowledges the financial support of the Outstanding Youth Fund (10125523), the Key Important Nano-Research Project (90206032) of the National Natural Science Foundation of China, and the Knowledge Innovation Program of the Chinese Academy of Sciences (KJCX2-SW-N11 and KJCX2-SW-H12-02). The Advanced Light Source is supported by the U.S. Department of Energy under Contract No. DE-AC02-05CH11231.

¹R. H. Baughman, A. A. Zakhidov, and W. A. De Heer, Science **297**, 787 (2002).

²M. Burghard, Surf. Sci. Rep. **58**, 1 (2005).

³R. Saito, M. Fujita, G. Dresselhaus, and M. S. Dresselhaus, Appl. Phys. Lett. **60**, 2204 (1992).

⁴P. Lambin, C. R. Phys. **4**, 1009 (2003).

⁵C. L. Kane and E. J. Mele, Phys. Rev. Lett. **78**, 1932 (1997).

⁶Y. J. W. Ding, X. H. Yan, J. X. Cao, D. L. Wang, Y. Tang, and Q. B. Yang, J. Phys.: Condens. Matter **15**, L439 (2003).

⁷T. W. Odom, J. L. Huang, P. Kim, and C. M. Lieber, Nature (London) **391**, 62 (1998).

⁸S. Park, D. Srivastava, and K. Cho, Nano Lett. 3, 1273 (2003).

⁹P. Skytt, P. Glans, D. C. Mancini, J.-H. Guo, N. Wassdahl, J. Nordgren, and Y. Ma, Phys. Rev. B **50**, 10457 (1994).

¹⁰J.-H. Guo, Y. Luo, A. Augustsson, S. Kashtanov, J.-E. Rubensson, D. K. Shuh, H. Agren, and J. Nordgren, Phys. Rev. Lett. 91, 157401 (2003).

¹¹J.-H. Guo, Int. J. Nanotechnol. **1-2**, 193 (2004).
 ¹²T. Warwick, P. Heimann, D. Mossessian, W. McKinney, and H. Padmore,

Rev. Sci. Instrum. 66, 2037 (1995). 13 J. Nordgren, G. Bray, S. Cramm, R. Nyholm, J.-E. Rubensson, and N.

Wassdahl, Rev. Sci. Instrum. **60**, 1690 (1989). ¹⁴L. J. Terminello, D. K. Shuh, F. J. Himpsel, D. A. Lapiano-Smith, J. Stöhr,

D. S. Bethune, and G. Meijer, Chem. Phys. Lett. 182, 491 (1991).
 ¹⁵C. Liu, H. T. Cong, F. Li, P. H. Tan, H. M. Cheng, K. Lu, and B. L. Zhou,

Carbon 37, 1865 (1999).

16L. Song, L. J. Ci, L. Lv, Z. P. Zhou, X. Q. Yan, D. F. Liu, H. J. Yuan, Y.

Gao, J. X. Wang, L. F. Liu, X. W. Zhao, Z. X. Zhang, X. Y. Dou, W. Y. Zhou, G. Wang, C. Y. Wang, and S. S. Xie, Adv. Mater. (Weinheim, Ger.) 16, 1529 (2004).

¹⁷P. Nikolaev, M. J. Bronikowski, R. K. Bradley, F. Rohmund, D. T. Colbert, K. A. Smith, and R. E. Smalley, Chem. Phys. Lett. 313, 91 (1999).

¹⁸A. Kuznetsova, I. Popova, J. T. Yates, M. J. Bronikowski, C. B. Huffman, J. Liu, R. E. Smalley, H. H. Hwu, and J. G. G. Chen, J. Am. Chem. Soc. 123, 10699 (2001).

¹⁹J. A. Carlisle, E. L. Shirley, E. A. Hudson, L. J. Terminello, T. A. Callcott, J. J. Jia, D. L. Ederer, R. C. C. Perera, and F. J. Himpsel, Phys. Rev. Lett. 74, 1234 (1995).

²⁰P. A. Brühwiler, A. J. Maxwell, C. Puglia, A. Nilsson, S. Andersson, and N. Mårtensson, Phys. Rev. Lett. **74**, 614 (1995).

²¹J. A. Carlisle, E. L. Shirley, E. A. Hudson, L. J. Terminello, T. A. Callcott, J. J. Jia, D. L. Ederer, R. C. C. Perera, and F. J. Himpsel, Phys. Rev. Lett. 76, 1762 (1996).

²²Y. Ma, Phys. Rev. B **49**, 5799 (1994).