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# Modelling of General Biotechnological Processes* 

A. ILCHMANN ${ }^{\dagger}$ and M.-F. WEIRIG ${ }^{\ddagger}$


#### Abstract

A general biotechnological process is modelled by a finite dimensional ordinary differential equation. The stoichiometry is only modelled qualitatively.It is shown that the usual biochemically motivated assumptions are not sufficient to guarantee boundedness of the solution. To overcome this, the concept of non-cyclic biotechnological processesis introduced. Loosely speaking it means that the process does not contain any "reaction loop". The assumption of non-cyclicity replaces the common assumption of Conservation of Mass. An algorithm is presented so that after finitely many steps it is decidedwhether a process is non-cyclic or cyclic. Non-cyclicity is also characterisedin terms of an echelon matrix derived from the stoichiometric matrix viapermutations of columns and rows.


Keywords: biochemical processes, bioreactors, non-cyclic processes, reaction models.

## NOMENCLATURE

$\mathbb{R}_{+}$the set of non-negative real numbers
$R_{+}^{*}$ the set of positive real numbers

## 1 INTRODUCTION

In this paper we model certain biotechnological processes, such as chemical, biochemical and microbiological processes in batch, fed-batch or continuous stirred tank reactors. If the process is completely mixed and the medium is

[^0]homogeneous in the reactor, then it can be modelled by a set of ordinary differential equations
\[

$$
\begin{equation*}
\dot{\xi}_{i}(t)=\sum_{j=1}^{M} k_{i j} \varphi_{j}(\xi(t))-D(t) \xi_{i}(t)-q_{i}(t)+F_{i}(t), \quad i=1, \ldots, N \tag{1}
\end{equation*}
$$

\]

Here $M$ is the number of reactions of the process and
$\xi_{i}(\cdot): \mathbb{R}_{+} \rightarrow \mathbb{R}_{+} \quad$ are the concentrations of the different components,
$\varphi_{j}(\cdot): \mathbb{R}_{+}^{N} \rightarrow \mathbb{R}_{+} \quad$ is the reaction rate of the $j$-th reaction,
$D(\cdot): \mathbb{R}_{+} \rightarrow \mathbb{R}_{+} \quad$ is the dilution rate,
$F_{i}(\cdot): \mathbb{R}_{+} \rightarrow \mathbb{R}_{+} \quad$ are the feed rates of the different components,
$q_{i}(\cdot): \mathbb{R}_{+} \rightarrow \mathbb{R}_{+} \quad$ are the gaseous outflow rates of the different components, $k_{i j} \in \mathbb{R}$ are the so-called stoichiometric (or yield) coefficients. This model is derived from the mass balance dynamics of each component in the reactor where the term $\sum k_{i j} \varphi_{j}(\xi)$ models the reaction kinetics, and the term $-D \xi_{i}-q_{i}+F_{i}$ models the exchange with the environment.

Notice that if $D(\cdot) \equiv 0$ and $F(\cdot) \equiv 0$, then we have a batch reactor, i.e. no inflow and no outflow; if $D(\cdot) \equiv 0$, then (1) describes a fed-batch reactor, i.e. no outflow; and generally (1) models a continuous stirred tank reactor.

To simplify the notation we may rewrite (1) as

$$
\begin{equation*}
\dot{\xi}=K \dot{\varphi}(\xi)-D \xi-Q \xi+F \tag{2}
\end{equation*}
$$

where $\varphi(\xi):=\left(\varphi_{1}(\xi), \ldots, \varphi_{M}(\xi)\right)^{T}, Q:=\operatorname{diag}\left\{\mathrm{q}_{1}, \ldots, \mathrm{q}_{\mathrm{N}}\right\}, F:=\left(F_{1}, \ldots, F_{N}\right)^{T}$ and $k=\left(k_{i j}\right)=\left[K_{1}, \ldots, K_{M}\right]$ denotes the stoichiometric matrix.

We stress that, although the matrix $K$ is called stoichiometric matrix, we do not require that it represents an exact stoichiometric relationship betweenthe components, it represents a qualitative relationship. Components that do not play an important role in the process, such as by-products ofa reaction or substrates which are not limiting, have been omitted. The advantage of this description is that the model might be of much smaller dimension. This approach is quite common; see for instance Bastin and Dochain [2]. However, the reaction scheme may not follow the law of conservation of mass.This is different to many traditional approaches, such as for example the well-known contribution on chemical reacting systems by Gavalas [6]. See in particular Section 1.1, where he introduces systems which can be described by an ordinary differential equation of the form (2). Although Gavalas does not explicitly say so (see Section 1.1 and also the sentence after equation (1.8.11)), the Principle of Mass Conservation implies the existence of a positive vector

$$
\begin{equation*}
\gamma \in\left(\mathbb{R}_{+}^{*}\right)^{N} \text { so that } \gamma^{T} K_{j}=0 \text { for all } j=1, \ldots, M \tag{3}
\end{equation*}
$$

Intuitively, this property means that the concentration remains bounded and is in fact crucial in proving it.
Instead of conservation of mass, we introduce the new notion of non-cyclic processes. It therefore has to be proved that (2) still exhibits practical relevant features,such as existence and uniqueness, positivity and boundedness of the solution. Roughly speaking, Cyclic means that the process contains a reaction loop, i.e. there exists a subset of reactions $S$, so that everysubstrate $\xi_{i}$ involved in a reaction $j \in S$ is also an autocatalyst or aproduct of one of the reactions $j \in S$. One of the crucial consequences of non-cyclic processes, which is proved at the end of the present paper, is the existence of a positive vector

$$
\begin{equation*}
\gamma \in\left(\mathbb{R}_{+}^{*}\right)^{N} \quad \text { so that } \quad \gamma^{T} K_{j} \leq 0 \text { for all } j=1, \ldots, M . \tag{4}
\end{equation*}
$$

Note that the condition (4) is only slightly less restrictive than (4). However it is sufficient to guarantee boundedness of the trajectories.
Bastin and Dochain [2] refer to (1) as the "general dynamical model" of biological reactors. Modifications and generalisations of (1) have been considered in Gavalas [6], Moser [10] and various authors in Rehm and Reed (eds.) [11]. For control theoretic applications, models of this type were used by $[7,3,12,1]$ to name but a few.
The paper is organised as follows. In Section 2 we describe the model and its assumptions in detail. Existence and uniqueness, invariance of the positive orthant and boundedness of the solution is proved in Section 3. Finally, in Section 4 the concept of non-cyclic processes is introduced. We present an algorithm which checks in finitely many steps whether a biotechnological process iscyclic or non-cyclic. Furthermore, non-cyclicity is characterised in terms of an echelon form of the stoichiometric matrix and a corollary of that is that non-cyclic process do have bounded trajectories.

## 2 THE GENERAL DYNAMICAL MODEL

Loosely speaking, we study biotechnological processes where $N$ concentrations $\xi_{1}, \ldots, \xi_{N}$ in the liquid phase of the reactor are related via $M$ reactions $\varphi_{1}, \ldots, \varphi_{M}$. More precisely such a process is commonly specified by the reaction scheme for each reaction

$$
\begin{equation*}
\sum_{i \in L_{j}} c_{i j} \cdot \xi_{i} \xrightarrow{\varphi_{j}} \sum_{i \in R_{j}} c_{i j} \cdot \xi_{i} \quad \text { for } \quad j=1, \ldots, M \tag{5}
\end{equation*}
$$

Here
$L_{j} \subseteq\{1, \ldots, N\}, L_{j} \neq \emptyset$, is the set of indices of the components $\xi_{i}$ which are the reactants of the $j$ th reaction,
$R_{j} \subseteq\{1, \ldots, N\}, R_{j} \neq \emptyset$, is the set of indices of the components $\xi_{i}$ which are the reaction products of the $j$ th reaction.
The quantities of each component involved in the reaction are specified by the nonnegative stoichiometric coefficients $c_{i j}$, sometimes also called yield coefficients. The rate of consumption of the reactants, which is equal to the rate of formation of the reaction products, iscalled the reaction rate and denoted by $\varphi_{\text {; }}$

To specify the process in more detail, we introduce the following notation for each reaction $j=1, \ldots, M$ :

$$
\text { Cat }_{j}:=L_{j} \cap R_{j}
$$

i.e. the set of the indices of those components, called catalysts, which are involved in the $j$ th reaction, but maintained by the reaction;

$$
\operatorname{Sub}_{j}:=L_{j} \backslash\left(L_{j} \cap R_{j}\right) \neq \emptyset,
$$

i.e. the set of those components, called substrates, that are consumed by the $j$ th reaction;

$$
\operatorname{Prod}_{j} \subseteq R_{j} \backslash\left(L_{j} \cap R_{j}\right) \neq \emptyset
$$

is the set of the indices of those components, called products, that are produced by the $j$ th reaction;

$$
A^{\prime u t_{j}}:=R_{j} \backslash\left(\left(L_{j} \cap R_{j}\right) \cup \text { Prod }_{j}\right),
$$

i.e. the set of the indices of those components, called autocatalysts, that are accumulated by the $j$ th reaction;

$$
\text { Rest }_{j}:=\{1, \ldots, N\} \backslash\left(L_{j} \cup R_{j}\right),
$$

i.e. the set of the process components that do not take part in the $j$ th reaction. Notice that, for all $j=1, \ldots, M$,

$$
{A u t_{j} \cup \operatorname{Prod}_{j}=R_{j} \backslash\left(R_{j} \cap L_{j}\right), ~}_{\text {l }}
$$

and $\{1, \ldots, N\}$ can be represented as the disjoint union:

$$
\{1, \ldots, N\}=\operatorname{Cat}_{j} \cup \operatorname{Sub}_{j} \cup \mathcal{A u t}_{j} \cup \operatorname{Prod}_{j} \cup \operatorname{Rest}_{j} .
$$

Another helpful specification is the distinction between an external substrate of the process, i.e. a component that is added from the outside to the reactor and only consumed in the reactor, and an internal substrate, i.e., a component that
is produced by some reaction of the process and consumed by another. Formally we define
$\substack{\xi_{i} \text { is called an } \\ \text { external substrate }}$
if, and only if, $i \in\left(\bigcup_{j=1}^{N} S u b_{j}\right) \backslash\left(\bigcup_{j=1}^{N}\left(A u t_{j} \cup\right.\right.$ Prod $\left.\left._{j}\right)\right)$,


We assume that the dynamics of the process (5) can be described by the ordinary differential equation:

$$
\begin{equation*}
\dot{\xi}(t)=K \cdot \varphi(\xi(t))-D(t) \cdot \xi(t)-Q \xi(t)+F(t), \quad \xi(0) \in\left(\mathbb{R}_{+}^{*}\right)^{N} \tag{6}
\end{equation*}
$$

where

$$
\begin{aligned}
& \xi(t)=\left(\xi_{1}(t), \ldots, \xi_{N}(t)\right)^{T} \quad \in \mathbb{R}_{+}^{N} \quad \text { are the concentrations, } \\
& \varphi(\xi)=\left(\varphi_{1}(\xi), \ldots, \varphi_{M}(\xi)\right)^{T} \in \mathbb{R}_{+}^{M}, \quad \varphi_{j}: \mathbb{R}_{+}^{N} \rightarrow \mathbb{R}_{+} \text {is the reaction rate } \\
& \text { of the } j \text { th reaction, } \\
& F(t)=\left(F_{1}(t), \ldots, F_{N}(t)\right)^{T} \quad \in \mathbb{R}_{+}^{N}, \quad F_{i}(\cdot): \mathbb{R}_{+} \rightarrow\left[0, \bar{F}_{i}\right] \text { is the feed rate } \\
& \text { of the } i \text { th component, and } \overline{\mathrm{F}}_{\mathrm{i}} \geq 0 \text {, } \\
& Q=\operatorname{diag}\left\{\mathrm{q}_{1}, \ldots, \mathrm{q}_{\mathrm{N}}\right\} \quad \in \mathbb{R}^{\mathrm{N} \times \mathrm{N}}, \mathrm{q}_{1}, \ldots, \mathrm{q}_{\mathrm{N}} \geq 0 \text {, is the matrix of } \\
& \text { gaseous outflow rates, } \\
& D(\cdot): \mathbb{R}_{+} \longrightarrow[\underline{D}, \bar{D}] \quad \text { is the dilution rate, } 0 \leq \underline{\mathrm{D}} \leq \overline{\mathrm{D}}, \\
& K=\left(k_{i j}\right)=\left[K_{1}, \ldots, K_{M}\right] \in \mathbb{R}^{N \times M} \text { is the matrix of the stoichiometric } \\
& \text { coefficients. }
\end{aligned}
$$

The function $\varphi$ is supposed to be locally Lipschitz continuous and $F$ and $D$ are assumed to be locally integrable. The entries of $K$ relaterhe reactions as follows:

$$
\begin{array}{rl}
k_{i j}=c_{i j}>0 & \text { if, and only if, } \\
k_{i j}=0 & i \in R_{j} \backslash\left(L_{j} \cap R_{j}\right),  \tag{7}\\
k_{i j}=-c_{i j}<0 & \text { if, and only if, only if, } \quad i \in R_{j} \cap L_{j} \text { or } i \notin L_{j} \backslash\left(L_{j} \cap R_{j}\right) .
\end{array}
$$

Notice that all of the entries of $K$ are determined by (7) since $\{1, \ldots, N\}$ is the disjoint union of the sets $R_{j} \backslash\left(L_{j} \cap R_{j}\right), L_{j} \backslash\left(L_{j} \cap R_{j}\right), R_{j} \cap L_{j}$ and $\{1, \ldots, N\} \backslash\left(R_{j} \cup L_{j}\right)$.

We already stressed and explained in the Introduction that, although the matrix $K$ is called stoichiometric matrix, we do not require that it represents an exact stoichiometric relationship between the components, it represents a qualitative relationship.

Remark 1 It is easy to see, and useful in the following, that the catalysts, substrates and products can be characterized as follows:

$$
\begin{aligned}
\text { Cat }_{j} & =\left\{i \in L_{j} \cup R_{j}: k_{i j}=0\right\} \\
\text { Sub }_{j} & =\left\{i \in\{1, \ldots, N\}: k_{i j}<0\right\}, \\
\text { Aut }_{j} \cup \text { Prod }_{j} & =\left\{i \in\{1, \ldots, N\} \mid k_{i j}>0\right\} .
\end{aligned}
$$

For notational convenience we also define the following sets:

$$
\begin{aligned}
& R_{\text {Cat }_{i}}:=\left\{j \in\{1, \ldots, M\} \mid k_{i j}=0, i \in L_{j} \cup R_{j}\right\} \text {, i.e. the set of numbers of } \\
& \text { the reactions that involve } \xi_{i} \text { as a catalyst, } \\
& R_{S_{u b_{i}}}:=\quad\left\{j \in\{1, \ldots, M\} \mid k_{i j}<0\right\} \text {, i.e. the set of the numbers of the reac- } \\
& \text { tions that involve } \xi_{i} \text { as a substrate, } \\
& R_{A u t_{i}}:=\left\{j \in\{1, \ldots, M\} \mid k_{i j}>0, i \in A u t_{j}\right\} \text {, i.e. the set of numbers of } \\
& \text { the reactions that involve } \xi_{i} \text { as an autocatalyst, } \\
& R_{\text {Prod }_{i}}:=\left\{j \in\{1, \ldots, M\} \mid k_{i j}>0, i \in \text { Prod }_{j}\right\} \text {, i.e. the set of numbers of } \\
& \text { the reactions that involve } \xi_{i} \text { as a product. }
\end{aligned}
$$

Notice that

$$
R_{\text {Auti }_{i}} \cup R_{\text {Prod }_{i}}=\left\{j \in\{1, \ldots, M\} \mid k_{i j}>0\right\} .
$$

Note that if $k_{i j_{0}}=0$ for some $i_{0} \in\{1, \ldots, N\}$ and some $j_{0} \in\{1, \ldots, M\}$, then without further knowledge of the process details we cannot decide whether $i_{0} \in$ Cat $_{j_{0}}$ or $i_{0} \in$ Rest $_{j_{0}}$.
Similarly, if $k_{i j_{0}}>0$ for some $i_{0} \in\{1, \ldots, N\}$ and some $j_{0} \in\{1, \ldots, M\}$, we do not know whether $i_{0} \in A u t_{j_{0}}$ or $i_{0} \in \operatorname{Prod}_{j_{0}}$.

Remark 2 In order to distinguish between a reaction which involves an autocatalyst and others which do not, we use in (5) either - respectively - and write

$$
\sum_{i \in S u b_{j} \cup C a t_{j}} c_{i j} \cdot \xi_{i} \xrightarrow{\varphi_{j}} \sum_{k \in P \operatorname{rod}_{j} \cup C a t_{j}} c_{k j} \cdot \xi_{k}: \quad \text { if } \quad A u t_{j}=\emptyset
$$

and

$$
\sum_{i \in S u b_{j} \cup C a t_{j}} c_{i j} \cdot \xi_{i} \stackrel{\varphi_{j}}{-} \sum_{k \in P_{r o d}, \cup C a l, \cup A u t_{i}} c_{k j} \cdot \xi_{k}, \quad \text { if } \quad A u t_{j} \neq \emptyset .
$$

The following example illustrates how a biochemical process presented by (5) is associated with the differential equation (6).

Example 3 The mechanism of yeast growth on glucose with ethanol production is describedby the following set of microbiological reactions, see Sonnleitner and Kappeli [13].

$$
\begin{aligned}
& c_{11} \xi_{1}+c_{21} \xi_{2} \xrightarrow{\varphi_{1}} c_{31} \xi_{3}+c_{41} \xi_{4} \\
& c_{52} \xi_{5}+c_{22} \xi_{2} \xrightarrow{\varphi_{2}} c_{32} \xi_{3}+c_{42} \xi_{4} \\
& c_{13} \xi_{1} \stackrel{\varphi_{3}}{\hookrightarrow} c_{33} \xi_{3}+c_{53} \xi_{5}+c_{43} \xi_{4}
\end{aligned}
$$

where $\xi_{1}, \ldots, \xi_{5}$ denote the glucose, dissolved oxygen, yeast, dissolved carbon dioxide and the ethanol concentration, respectively.
It is immediate that

$$
\begin{array}{ll}
L_{1}=\{1,2\}, & R_{1}=\{3,4\} \\
L_{2}=\{2,5\}, & R_{2}=\{3,4\} \\
L_{3}=\{1\}, & R_{3}=\{3,4,5\},
\end{array}
$$

and hence

$$
\begin{array}{rlrl}
\text { Cat }_{j} & =L_{j} \cap R_{j}=\emptyset \quad \text { for } j=1,2,3, \\
\text { Sub }_{1} & =\{1,2\}, & k_{11}=-c_{11}, k_{21}=-c_{21} . \\
\text { Prod }_{1} \cup \text { Aut }_{1}=\{3,4\}, & k_{31}=c_{31}, k_{41}=c_{41} . \\
\text { Sub }_{2}=\{2,5\}, & k_{22}=-c_{22}, k_{52}=-c_{52} . \\
\text { Prod }_{2} \cup \text { Aut }_{2}=\{3,4\}, & & k_{32}=c_{32}, k_{42}=c_{42} . \\
\text { Sub }_{3}=\{1\}, & k_{13}=-c_{13} . \\
\text { Prod }_{3} \cup \text { Aut }_{3}=\{3,4,5\}, & & k_{33}=c_{33}, k_{43}=c_{43}, k_{53}=c_{53} .
\end{array}
$$

Therefore, the mathematical model (6) of the yeast growth process is

$$
\left(\begin{array}{l}
\dot{\xi}_{1}(t) \\
\dot{\xi}_{2}(t) \\
\dot{\xi}_{3}(t) \\
\dot{\xi}_{4}(t) \\
\dot{\xi}_{5}(t)
\end{array}\right)=K\left(\begin{array}{c}
\varphi_{1}(\xi(t)) \\
\varphi_{2}(\xi(t)) \\
\varphi_{3}(\xi(t))
\end{array}\right)-D(t)\left(\begin{array}{l}
\xi_{1}(t) \\
\xi_{2}(t) \\
\xi_{3}(t) \\
\xi_{4}(t) \\
\xi_{5}(t)
\end{array}\right)-\left(\begin{array}{c}
0 \\
0 \\
0 \\
q_{4} \xi_{4}(t) \\
0
\end{array}\right)+\left(\begin{array}{c}
F_{1}(t) \\
F_{2}(t) \\
0 \\
0 \\
0
\end{array}\right) .
$$

where

$$
K=\left(k_{i j}\right)=\left[\begin{array}{ccc}
-c_{11} & 0 & -c_{13} \\
-c_{21} & -c_{22} & 0 \\
c_{31} & c_{32} & c_{43} \\
c_{41} & c_{42} & c_{43} \\
0 & -c_{52} & c_{53}
\end{array}\right]
$$

To be able to prove that the concentrations in Example 3 remain positive and bounded, we need further assumptions on the growth rates. More generally, we assume that the process (6) satisfies the following.
(A1) For $j=1, \ldots, M$ the reaction rates are of the form

$$
\varphi_{j}(\cdot): \mathbb{R}_{+}^{N} \longrightarrow \mathbb{R}_{+}, \quad \xi \mapsto \varphi_{j}(\xi)=\alpha_{j}(\xi)\left(\prod_{k \in A u t, \cup L} \xi_{k}\right)
$$

and the growth rates

$$
\alpha_{j}(\cdot): \mathbb{R}_{+}^{N} \longrightarrow\left(\underline{\alpha}_{j}, \bar{\alpha}_{j}\right], \quad 0 \leq \underline{\alpha}_{j} \leq \bar{\alpha}_{j}
$$

are continuous functions.
(A2) If the lower bound of the dilution rate $D(\cdot)$ is zero, i.e. $\underline{D}=0$, then the feed rate is proportional to the dilution rate

$$
F(\cdot) \equiv D(\cdot) \xi^{i n}(\cdot)
$$

where, for all $j=1, \ldots, M$, the components of the inflow rate satisfy for some $\bar{\xi}_{1}^{i n}, \ldots, \bar{\xi}_{N}^{i n} \geq 0$,

$$
\xi_{i}^{i n}(\cdot): \mathbb{R}_{+} \longrightarrow\left[0, \vec{\xi}_{i}^{i n}\right\}
$$

Remark 4 Assumption (A1) corresponds to the physical fact that a reaction rate is always nonnegative and that a reaction can only take place if all its reactants are present in the reactor, which means that as soon as the concentration of one reactant becomes zero the reaction rate is zero.

If the concentrations of the reactants are also bounded from above - which will be shown in Section 3 under additional assumptions - then the chosen reaction rates $\varphi_{j}$ agree with the conventional reaction rates. For a comprehensive list see e.g. the Appendix in [2]. The reaction rate $\varphi$ is often assumed to be proportional to the microbial growth rate $\mu$. Consider for instance the models of Monod or Haldane, where the function $\mu$ is used to explain the dependency of
$\varphi$ on the substrate concentrations. The relationship between $\alpha(\xi)$ and $\mu(\xi)$ is then

$$
\begin{array}{ll}
\text { Monod: } & \alpha_{j}(\xi)=\prod_{i \in S u b_{j}}\left(\frac{\hat{\mu}_{i}}{K_{i}+\xi_{i}}\right) \quad \text { and } \mu_{j}(\xi)=\alpha_{j}(\xi) \cdot \prod_{i \in S b_{j}} \xi_{i}, \\
\text { Haldane: } & \alpha_{j}(\xi)=\prod_{i \in S u b_{j}}\left(\frac{\hat{\mu}_{i}}{K_{i}+\xi_{i}+\frac{\xi_{i}^{2}}{K_{i}}}\right) \text { and } \mu_{j}(\xi)=\alpha_{j}(\xi) \cdot \prod_{i \in S u b_{j}} \xi_{i} .
\end{array}
$$

Since $A u t_{j} \cup L_{j}=S u b_{j} \cup C a t_{j} \cup A u t_{j}$, the reaction rate $\varphi$ in (A1) can be written as

$$
\varphi_{j}(\xi)=\mu_{j}(\xi)\left(\prod_{k \in A u_{j} \cup C a t_{j}} \xi_{k}\right) .
$$

Assumption (A2) means that if $D(\cdot)$ is not bounded away from zero, then one requires instead that $F(\cdot) \equiv D(\cdot) \xi^{i n}(\cdot)$. This means that all the substances that are added from the exterior to the reactor are fed with the influent water stream and $\xi_{i}^{i n}$ is the concentration of the substance $\xi_{i}$ in the influent water stream.
There might be substances $\xi_{i}$ which are soluble in the liquid phase and gasifiable at atmospheric pressure. In model (6), where the gaseous outflow is given by the proportion $Q \xi$, the saturation concentration between the liquid and the gas phase is assumed to be "very high".

Remark 5 It is a common assumption of various authors, see e.g. [6], that $N>M$ and rk $\mathrm{K}=\mathrm{M}$. This means that the number of components involved in the process exceeds the number of reactions, which usually is the case since one reaction involves more than one component. Furthermore it is often desirabie to describe composition changes by the smallest possible number of reactions. If rk $K=M$, then the reactions are called independent, see e.g. [6], Definition 1.1.1. If $K$ does not have full column rank, then there are certain strategies to pick a set of "key reactions", see e.g. Section I. 2 in [10]. However, in our setup there is no need to assume any of these assumptions.

## 3 INVARIANCE OF THE POSITIVE ORTHANT AND BOUNDNESS OF THE SOLUTION

In this section we show existence and uniqueness of the solution of (6) on a maximal interval of existence $[0, \omega)$, where $\omega$ might be finite. (A1) does not guar-
antee that finite escape time does not occur, this will be ensured by an extra assumption later on.

Proposition 6 If the general dynamical model ( 6 ) satisfies (A1), then there exists a unique solution $\xi(\cdot):(0, \omega) \longrightarrow \mathbb{R}^{N}$ on its maximal interval of existence $[0, \omega)$, for some $\omega \in(0, \infty)$. Furthermore, the positive orthant is invariant under the flow, i.e.

$$
\begin{equation*}
\text { if } \quad \xi(0) \in\left(\mathbb{R}_{+}^{*}\right)^{N}, \quad \text { then } \quad \xi(t) \in\left(\mathbb{R}_{+}^{*}\right)^{N} \quad \text { for all } \quad t \in[0, \omega) \tag{8}
\end{equation*}
$$

Proof: Existence and uniqueness of a solution $\xi(\cdot):[0, \omega) \rightarrow \mathbb{R}^{N}$ of (6) on a maximal interval of existence follow from the theory of ordinary differential equations, see e.g. Coddington and Levinson [4].

Seeking a contradiction to the invariance of the positive orthant under the flow, suppose there exist some $i \in\{1, \ldots, N\}$ and $t_{0} \in(0, \omega)$ such that

$$
\begin{equation*}
\xi_{i}\left(t_{0}\right)=0 \text { and } \xi_{k}(t)>0 \text { for all } t<t_{0}, k \in\{1, \ldots, N\} \tag{9}
\end{equation*}
$$

The $i$ th coordinate of (6) satisfies, for all $t \in\left[0, t_{0}\right]$,

$$
\dot{\xi}_{i}(t)=\sum_{j=1}^{M} k_{i j} \varphi_{j}(\xi(t))-D(t) \cdot \xi_{i}(t)-\beta_{i} \cdot \xi_{i}(t)+F_{i}(t)
$$

and since by Remark $1 k_{i j} \geq 0$ for $j \notin R_{S u b_{i}}$ and $F_{i}(t) \geq 0$ for all $t \geq 0$, it follows that

$$
\dot{\xi}_{i}(t) \geq-\sum_{j \in R_{S_{u b_{i}}}}\left|k_{i j}\right| \varphi_{j}(\xi(t))-D(t) \cdot \xi_{i}(t)-\beta_{i} \cdot \xi_{i}(t)
$$

Now Assumption (A1) yields

$$
\dot{\xi}_{i}(t) \geq-\sum_{j \in R_{S u b_{i}}}\left|k_{i j}\right| \alpha_{j}(\xi(t))\left(\prod_{k \in\left(A u t_{j} \cup L_{j}\right) \backslash\{i\}} \xi_{k}(t)\right) \xi_{i}(t)-\left[D(t)+\beta_{i}\right] \xi_{i}(t)
$$

and defining

$$
\psi_{i}(t):=-\left[\sum_{j \in R_{\text {sub }}^{i}}\left|k_{i j}\right| \alpha_{j}(\xi(t))\left(\prod_{k \in\left(A u t_{j} \cup L_{j}\right) \backslash\{i\}} \xi_{k}(t)\right)+D(t)+\beta_{i}\right]
$$

leads to

$$
\dot{\xi}_{i}(t) \geq \psi_{i}(t) \xi_{i}(t)
$$

Now integration yields, see Lemma 19,

$$
\xi_{i}(t) \geq e^{\int_{0}^{t} \psi_{i}(\tau) d \tau} \xi_{i}(0) \text { as long as } t \in\left[0, t_{0}\right],
$$

and hence

$$
\xi_{i}\left(t_{0}\right) \geq e^{\int_{0}^{i_{0}} w_{i}(\tau) d \tau} \xi_{i}(0) .
$$

Since $\xi_{i}(0)>0$, this contradicts the assumption $\xi_{i}\left(t_{0}\right)=0$ in (9). Therefore a component cannot leave the positive orthant and the state variables stay positive on their maximal interval of existence.

The assumptions (A1)-(A2) are however not sufficient to guarantee existence of the solutionon the whole of $[0, \infty)$, finite escape time might occur. This is demonstrated in the following example.

Example 7 Consider the following reaction scheme ${ }^{1}$

$$
\begin{array}{ll}
c_{11} S_{0} \stackrel{\varphi_{1}}{\leftrightarrows} X+c_{21} S_{1}+c_{41} P_{1}, & \varphi_{1}\left(S_{0}, X, S_{1}, P_{1}, P_{2}\right):=S_{0} \cdot X, \\
c_{22} S_{1} \stackrel{\varphi_{2}}{\leftrightarrows} X+c_{12} S_{0}+c_{52} P_{2}, & \varphi_{2}\left(S_{0}, X, S_{1}, P_{1}, P_{2}\right):=S_{1} \cdot X .
\end{array}
$$

The corresponding process (6) of the reaction scheme is

$$
\begin{aligned}
\dot{S}_{0}(t) & =k_{11} S_{0}(t) X(t)+k_{12} S_{1}(t) X(t)-D(t) S_{0}(t)+D(t) S^{i n} \\
\dot{S}_{1}(t) & =k_{21} S_{0}(t) X(t)+k_{22} S_{1}(t) X(t)-D(t) S_{1}(t) \\
\dot{X}(t) & =\left[S_{0}(t)+S_{1}(t)\right] X(t)-D(t) X(t) \\
\dot{P}_{1}(t) & =k_{41} X(t) S_{0}(t)-D(t) P_{1}(t) \\
\dot{P}_{2}(t) & =k_{52} X(t) S_{1}(t)-D(t) P_{2}(t)
\end{aligned}
$$

Setting $\quad k_{11}=-c_{11}=0.5, k_{21}=c_{21}=1.5, k_{12}=c_{12}=1.5, k_{22}=-c_{22}=-0.5$ yields
${ }^{1}$ This example may be chemically absurd, but it is a general biotechnological process satisfying (A1)-(A2).

$$
K=\left(\begin{array}{cc}
-0.5 & 1.5 \\
1.5 & -0.5 \\
1 & 1 \\
k_{41} & 0 \\
0 & k_{52}
\end{array}\right)
$$

And since $c_{41}$ and $c_{52}$ are associated with products, we conclude $k_{41}>0$ and $k_{52}>0$. Furthermore assume that the initial values of the different state variables are all positive and the dilution rate is a positive constant $D(\cdot) \equiv D>0$.

Differentiation of the new variable

$$
Z(t)=S_{0}(t)+S_{1}(t)-X(t)
$$

yields

$$
\dot{Z}(t)=-D Z(t)+D S^{i n}
$$

and hence

$$
Z(t)=S^{i n}+e^{-D t}\left(Z_{0}-S^{i n}\right)
$$

Now $X(\cdot)$ satisfies

$$
\begin{aligned}
\dot{X}(t) & =X(t)(Z(t)+X(t))-D X(t) \\
& =X^{2}(t)+(Z(t)-D) X(t) \\
& \geq(Z(t)-D) X(t)
\end{aligned}
$$

and hence, by Lemma 19 and substitution of $Z$, we obtain

$$
X(t) \geq e^{\int_{0}^{t} Z(s)-D d s} X_{0} \geq e^{\int_{0} S^{i n}+\left(Z_{0}-S^{n}\right) \cdot e^{-D s}-D d s} X_{0}
$$

If we choose $D<S^{i n}<Z_{0}$, then

$$
X(t) \geq e^{\left[S^{i n}-D \mid t\right.} X_{0}
$$

and hence $\lim _{t \rightarrow \infty} X(t)=\infty$. Therefore the concentration of the bacteria is unbounded.

If $D(t)=1-\cos t$, so that the alternative in (A2) is satisfied, then a similar calculation proves unboundedness of $X(\cdot)$ in case of $Z_{0}>S^{i n}>1$.

Example 7 shows that the assumptions (A1)-(A2) do not guarantee a bounded solution. Since $k_{41}$ and $k_{52}$ are positive, it is easy to see that there does not exist
a vector $\gamma \in \mathbb{R}^{j}$ with positive components such that $\left\langle\gamma, K_{j}\right\rangle \leq 0$ for $j=1,2$. Therefore, condition (4) is not satisfied for this example. This motivates a further assumption which replaces the more restrictive assumption (3) on exact stoichiometriccoefficients.
(A3) There exists some $\gamma \in\left(\mathbb{R}_{+}^{*}\right)^{N}$ such that

$$
\left\langle\gamma, K_{j}\right\rangle \leq 0 \quad \text { for all } j=1, \ldots, M
$$

We are now in a position to prove boundedness of the solution. The main result of Section 4 will be the definition and characterisation of the concept of noncyclic processes which will be shown to be sufficient for (A3).

Theorem 8 Suppose the general reactor model ( 6 ) satisfies (A1)-(A3), and let $\bar{F}:=\left(\bar{F}_{1}, \ldots, \bar{F}_{N}\right)^{T}$ denote the upper bound for the feed rates, $\bar{\xi}^{i n}:=\left(\bar{\xi}_{1}^{i n}, \ldots, \bar{\xi}_{N}^{i n}\right)^{T}$ the upper bound in (A2). Then the following holds.
(i) The initial value problem (6) has a unique solution $\xi(\cdot): \mathbb{R}_{+}^{\prime} \longrightarrow\left(\mathbb{R}_{+}^{*}\right)^{N}$ defined on the whole of the half axis $[0, \infty)$ and stays within the positive orthant.
(ii) If $\underline{D}>0$, then the solution is bounded from above as follows

$$
\langle\gamma, \xi(t)\rangle \leq \max \left\{\langle\gamma, \xi(0)\rangle, \frac{\langle\gamma, \bar{F}\rangle}{\underline{D}}\right\} \text { and } \quad \lim \sup _{t \rightarrow \infty}\langle\gamma, \xi(t)\rangle \leq \frac{\langle\gamma, \bar{F}\rangle}{\underline{D}} .
$$

(iii) If $\underline{D}=0$ (which by (A2) yields that $F(\cdot) \equiv D(\cdot) \xi^{i n}(\cdot)$ ), then the solution is bounded from above as follows

$$
\langle\gamma, \xi(t)\rangle \leq \max \left\{\langle\gamma, \xi(0)\rangle,\left\langle\gamma, \bar{\xi}^{i n}\right\rangle\right\} \text { and } \lim \sup _{t \rightarrow \infty}\langle\gamma, \xi(t)\rangle \leq\left\langle\gamma, \bar{\xi}^{i n}\right\rangle .
$$

Proof: By Proposition 6 we only have to show that the first inequalities in (ii) and (iii) hold true on $[0, \omega)$, the maximal interval of existence. Then boundedness of $\xi(\cdot)$ on $[0, \omega)$ yields $\omega=\infty$, and hence (i) follows.

Multiplying (6) from the left with $\gamma^{T}$ yields, for almost all $t \in[0, \omega)$,

$$
\gamma^{T} \dot{\xi}(t)=\gamma^{T} K \varphi(\xi(t))-\gamma^{T} D(t) \xi(t)-\gamma^{T}\left[\begin{array}{lll}
\beta_{1} & & 0 \\
& \ddots & \\
0 & & \beta_{N}
\end{array}\right] \xi(t)+\gamma^{T} F(t)
$$

and hence. for

$$
\Phi(t):=\langle\gamma, \xi(t)\rangle
$$

we obtain

$$
\dot{\Phi}(t)=\sum_{j=1}^{M}\left\langle\gamma, K_{j}\right\rangle \varphi_{j}(\xi(t))-D(t) \Phi(t)-\gamma^{T}\left[\begin{array}{lll}
\beta_{1} & & 0 \\
& \ddots & \\
0 & & \beta_{N}
\end{array}\right] \xi(t)+\langle\gamma, F(t)\rangle
$$

This yields, by assumption (A3) and positivity of $\xi(t)$, for all $t \in\{0, \omega)$

$$
\dot{\Phi}(t) \leq-D(t) \cdot \Phi(t)+\langle\gamma, F(t)\rangle
$$

If $\underline{D}>0$, then an application of Lemma $\frac{19}{a}$ and boundedness of $F(\cdot)$ yields, for all $t \in[0, \omega)$,

$$
\Phi(t) \leq \max \left\{\Phi(0), \frac{\langle\gamma, \bar{F}\rangle}{\underline{D}}\right\}
$$

Hence $\xi(\cdot)$ is bounded on $[0, \omega)$ and a repeated application of Lemma 19 proves the second inequality in (ii).

If $F(\cdot) \equiv D(\cdot) \xi^{i n}(\cdot)$, then we apply Lemma 19 to conclude

$$
\frac{d}{d t}\left[\Phi(t)-\gamma^{T} \bar{\xi}^{i n}\right]=\dot{\Phi}(t) \leq-D(t)\left[\Phi(t)-\gamma^{T} \bar{\xi}^{i n}\right]
$$

and the statements in (iii) follow. This completes the proof.
Remark 9
(a): Note that in Example 7 the matrix $K$ does not satisfy (A3).
(b): If assumption (A3) in Theorem 8 is weakened to hold only for some nonnegative (but not necessarily positive) $\gamma \in \mathbb{R}_{+}^{N}$, then we can only show boundedness for those concentrations $\xi_{i}$ for which $\gamma_{i}>0$.
(c): If the general dynamical model (6) consists of only one reaction, i.e. $M=1$, then (A3) is satisfied. To see this notice that $L_{j} \backslash\left(L_{j} \cap R_{j}\right)$ and $R_{j} \backslash\left(L_{j} \cap R_{j}\right)$ are both nonempty, and hence there exist some $i_{1} \neq i_{2}$ so that $k_{i_{1} j}>0$ and $k_{i j}<0$, whence every column of $K$ has at least one positive and one negative coefficient. Thus, if $K=K_{1}$ consists of one column only, we may define

$$
\gamma_{i}:=\left\{\begin{array}{lll}
1 & \text { if } & k_{i 1}<0 \\
\bar{\gamma} & \text { if } & k_{i 1}>0
\end{array} \quad \text { with } \quad \bar{\gamma}:=\frac{-\sum_{k_{11}<0} k_{i 1}}{\sum_{k_{i 1}>0} k_{i 1}}>0\right.
$$

and check that in this case $\left\langle\gamma, K_{1}\right\rangle=0$.

## 4 NON-CYCLIC PROCESSES

We now arrive at the main contribution of this paper, that is the concept of noncyclic processes, its characterization and its consequences such as it is a sufficient condition for the crucial assumption (A3). Cyclicity of (6) means that the process contains a reaction loop, i.e. there exists a subset of reactions $S$, so that every substrate $\xi_{i}$ involved in a reaction $j \in S$ is also an autocatalyst or a product of one of the reactions $j \in S$. Or in other words, (6) is non-cyclic respectively does not contain a reaction loop if, and only if, for each subset $S$ there exists a substrate $\xi_{i}$ of some reaction $j_{0} \in S$ such that $\xi_{i}$ is neither an autocatalyst nor a product of one of the reactions $j \in S$. This concept is more formally defined as follows.

Definition 10 The biotechnological process (6) is said to be cyclic if, and only if, there exists a nonempty subset $S \subseteq\{1, \ldots, M\}$, such that

$$
\begin{equation*}
\bigcup_{j \in S} S u b_{j} \subseteq \bigcup_{j \in S}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right) . \tag{10}
\end{equation*}
$$

For instance the biotechnological process in Example 7 is cyclic, since for $S=\{1,2\}$ all the substrates appear also as products.

Rephrasing this definition will be useful in the following.
Remark 11 Consider a biotechnological process (6). Then the following conditions are equivalent
(i) (6) is cyciic,
(ii) there exists some $S \subseteq\{1, \ldots, M\}, S \neq \emptyset$, such that

$$
\begin{equation*}
i \in \bigcup_{j \in S}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right) \quad \text { for all } \quad(i, j) \in S u b_{j} \times S \tag{11}
\end{equation*}
$$

and so are also the conditions
(i) (6) is non-cyclic,
(ii') for each $S \subseteq\{1, \ldots, M\}$,

$$
\begin{equation*}
\bigcup_{j \in S} \operatorname{Sub}_{j} \unrhd \subseteq \bigcup_{j \in S}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right), \tag{12}
\end{equation*}
$$

(iii) for each $S \subseteq\{1, \ldots, M\}$, there exists a pair

$$
\begin{equation*}
\left(i, j_{0}\right) \in S u b_{j_{0}} \times S \quad \text { such that } \quad i \notin \bigcup_{j \in S}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right) . \tag{13}
\end{equation*}
$$

The characterization of non-cyclicity is based on a constructive procedure which checks whether (10) is satisfied or not. This is comprised in the following pseudo-algorithm.

Algorithm 12 Consider the biotechnological process (6) with specified $M, N$, Sub $_{j}$, Aut $_{j}$, Prod $_{j}$ for $j=1, \ldots, M$.
SET $\quad S_{1}:=\{1, \ldots, M\} ;$
$\mathcal{N}:=\{1, \ldots, N\} ;$
FOR $\quad l=1, \ldots, M$
DO $\quad I F \quad l>1$ and there exists $j_{l} \in S_{l}$ such that
$i_{l-1} \in S u b_{j_{t}}$
SET $i_{l}:=i_{1-1} ;$
GOTO 10;
ELSE IF there exists $\left(i, j_{l}\right) \in \mathcal{N} \times S_{l}$ such that
$i_{l} \in \operatorname{Sub}_{j i}$ and $i_{l} \notin \bigcup_{j \in S_{l}}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right)$
GOTO 10;
ELSE PRINT "cyclic"; STOP;
10. $\quad S_{l+1}:=S_{l} \backslash\left\{j_{l}\right\} ;$
END
END

PRINT "non-cyclic"
END.

Remark 13 Instead of using subsets of substances, such as $S_{l}, S u b_{j}$, etc., it follows from Remark 1 that Algorithm 12 can equally be formulated for the stoichiometric matrix $K=\left(k_{i j}\right) \in \mathbb{R}^{N \times M}$ by replacing
$i_{l-1} \in S u b_{j_{l}}$ in condition IF by $k_{i_{1-}, j_{i}}<0$ and
$i_{l} \in S u b_{j}$ and $i_{l} \notin \cup\left(A u t_{j} \cup \operatorname{Prod}_{j}\right)$ in condition ELSE IF by $k_{i j_{l}}<0$ and $k_{i j} \leq 0$ for all $j \in S^{j} \in S_{i}$

Proposition 14 The Algorithm 12 terminates after finitely many steps with "noncyclic" if, and only if, the process (6) is non-cyclic; it terminates with "cyclic" if, and only if, the process ( 6 ) is cyclic.

Proof: (a): We prove that printing "non-cyclic" yields that () is non-cyclic.
This is proved in two steps.
$\left(a_{1}\right)$ : It is proved that if Algorithm 12 prints "non-cyclic", then for each $l \in\{1 \ldots, M\}$ there exists a pair of indices $\left(i_{1}, j_{l}\right) \in\{1, \ldots, N\} \times S_{l}$ such that

$$
\begin{equation*}
i_{i} \in \operatorname{Sub}_{j_{t}} \quad \text { and } \quad i_{i} \notin \bigcup_{j \in S_{t}}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right) . \tag{14}
\end{equation*}
$$

Proceed by induction on $l$. By assumption either the condition IF or the condition ELSE IF is satisfied. For $l=1 \mathrm{IF}$ is excluded, and hence ELSE IF holds and there exists a pair $\left(u_{1}, j_{1}\right) \in\{1, \ldots, N\} \times S_{1}$ satisfying (14).

Now suppose that $\left(i_{1}, j_{1}\right), \ldots,\left(i_{l-1}, j_{l-1}\right) \in\{1, \ldots, N\} \times S_{l-1}$ satisfy (14).
If IF is true, then there exists $j_{l} \in S_{l}$ such that $i_{l-1} \in S u b_{j}$. Since $i_{l-1} \notin \bigcup_{j \in S_{l-1}}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right)$ and $S_{l} \subset S_{l-1}$ it follows that $i_{l-1} \notin \bigcup_{j \in S_{l}}\left(\right.$ Aut $\left._{j} \cup \operatorname{Prod}_{j}\right)$.
Setting $i_{l}:=i_{-1}$ yields the pair ( $\left.i_{1}, j_{l}\right)$ which satisfies (14).
If ELSE IF is satisfied, it is immediate that $\left(i_{l}, j_{l}\right) \in\{1, \ldots, N\} \times S_{l}$ satisfies (14).
$\left(a_{2}\right)$ : It is proved that if Algorithm prints "non-cyclic", then (iii') in Remark 11 is satisfied.
Let $S \subseteq\{1, \ldots, M\}$ be arbitrary but nonempty, set $r=$ cardS, and use the notation

$$
S=\left\{j_{l_{1}}, \ldots, j_{l}\right\}, \quad \text { where } \quad l_{1} \leq \ldots \leq l_{r} .
$$

Since ( $i_{1}, j_{h_{1}}$ ) satisfies (14) and $S_{j_{1}}=\left\{j_{l_{1}}, \ldots, j_{M}\right\}$, it follows that $S \subseteq S_{j_{1}}$. Therefore (iii') in Remark 11 holds. This proves ( $a_{2}$ ) and (a) follows from Remark 11. (b): We prove that non-cyclicity of (6) yields printing "non-cyclic".

If (6) is non-cyclic, then by (13) for each $l$ there exists a pair $\left(i, j_{l}\right)$ such that

$$
\begin{equation*}
\left(i_{l}, j_{l}\right) \in S u b_{j l} \times S_{l} \quad \text { and } \quad i_{l} \notin \bigcup_{j \in S_{l}}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right) \tag{15}
\end{equation*}
$$

This means ELSE IF is satisfied for all $l=1, \ldots, M$ and the algorithm finally terminates with printing "non-cyclic".
(c): We prove that if Algorithm prints "cyclic", then (6) is cyclic.

If Algorithm prints "cyclic" in the lth loop, then ELSE IF does not hold true and hence (11) is satisfied for $S=S_{l}$. Now the statement follows from Remark 11.
(d): We prove that cyclicity of (6) yields printing "cyclic".

Suppose Algorithm 12 pprints "non-cyclic". Then by (a) the process is non-cyclic which contradicts the assumption.
This completes the proof.

The following example illustrates a non-cyclic process.
Example 15 This production of strains for yogurt manufacturing described in [3] is represented by the following reaction scheme ${ }^{2}$

$$
\begin{aligned}
& \xi_{1}+\xi_{4} \xrightarrow{\varphi_{1}} \xi_{2}+\xi_{3}+\xi_{4} \\
& \xi_{2} \stackrel{\rho_{2}}{\longrightarrow} \xi_{5}+\xi_{4} \\
& \xi_{2}+\xi_{5} \xrightarrow{\varphi_{3}} \xi_{7}+\xi_{5} \\
& \xi_{2} \xrightarrow{\rho_{4}} \xi_{6}+\xi_{4}+\xi_{9} \\
& \xi_{2}+\xi_{6} \xrightarrow{\varphi_{5}} \xi_{8}+\xi_{9}+\xi_{6}
\end{aligned}
$$

with $\xi_{1}$ representing lactose, $\xi_{2}$ glucose, $\xi_{3}$ galactose, $\xi_{4}$ lactase, $\xi_{5}$ lactobacillus bulgarius, $\xi_{6}$ streptococcus thermophilus, $\xi_{7}$ D-lactate, $\xi_{8}$ L-lactate and $\xi_{9}$ carbon dioxide. Following Algorithm 12, we choose $i_{1}=1$ and $l_{1}=1$, because $1 \in S u b_{1}$ and $\xi_{1} \notin$ Aut $_{j} \cup \operatorname{Prod}_{j}$ for $j \in\{2,3,4,5\}$.

Since 1 does not appear in the reactions $2-5,2 \nexists A u t_{j} \cup \operatorname{Prod}_{j}$ for $j \in\{3,4,5\}$ and $2 \in S u b_{2}$, we choose $i_{2}=2$ and $l_{2}=2$.

For $i_{3}$ we choose again $i_{3}=i_{2}=2$ and $l_{3}=3$, because $2 \in S u b_{3}$ and. $2 \notin A u t_{j} \cup \operatorname{Prod}_{j}$ for $j \in\{4,5\}$.

Finally $i_{4}=i_{5}=2$ and $l_{4}=4, l_{5}=5$. Now the algorithm stops (after $M=5$ steps) and hence this process is non-cyclic.
We are now in a position to prove the main result of this paper. It is the characterization of the concept of non-cyclicity in terms of the stoichiometric matrix $K$ in (6).

Theorem 16 The biotechnological process ( 6 ) is non-cyclic if, and only if, there exist permutation matrices $L \in \mathbb{R}^{N \times N}$ and $R \in \mathbb{R}^{M \times M}$ and there exist, for some $r \in\{1, \ldots, M\}, 1=q_{1}<q_{2}<\ldots<q_{r-1}<q_{r} \leq M$, such that

$$
L K R=\tilde{K}=\left(\tilde{k}_{l m}\right)
$$

satisfies

$$
\begin{array}{rll}
\tilde{k}_{l m}<0 \text { for } & m=q_{l}, \ldots, q_{l+1}-1 & \text { and } l=1, \ldots, r \\
\tilde{k}_{l m}=0 \text { for } & m=q_{l+l}, \ldots, M & \text { and } l=1, \ldots, r-1, \tag{16}
\end{array}
$$

where $q_{r+1}-1:=M$.

[^1]This means that $\tilde{K}$ is of the form
where $\ominus$ denotes negative coefficients and $*$ some real coefficients. The form is not unique.

Proof: We show first that if the biotechnological process is non-cyclic, then the stoichiometric matrix can be transformed, by adequate permutations of its columns and its rows, into the form (17).
But before specifying the permutation matrices we need to use the construction of Algorithm 12 to specify certain elements of $K$. By Proposition 14, Remark 11, and Algorithm 12, for every $l=1, \ldots, M$ there exists a pair ( $i_{i}, j_{l}$ ) such that

$$
i_{l} \in S u b_{j t} \quad \text { and } \quad i_{l} \notin \bigcup_{j \in S_{t}}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right),
$$

and hence, by Remark 1, the stoichiometric coefficients of the process satisfy, for $l=1, \ldots, M$,

$$
k_{i j_{i}}<0
$$

and

$$
\begin{equation*}
k_{i j} \leq 0 \text { for all } j \in S_{l+1}:=S_{l} \backslash\left\{j_{l}\right\}=\left\{j_{l+1}, \ldots, j_{M}\right\}, \tag{18}
\end{equation*}
$$

where, for notational convenience, $\left\{j_{M+1}, j_{M}\right\}:=\emptyset$.
Notice that $\left(j_{1}, \ldots, j_{M}\right)$ is a permutation of $(1, \ldots, M) .\left\{i_{1}, \ldots, i_{r}\right\}$ is not a permutation; $i_{k}=i_{l}$ is possible for $k<l$ only if $i_{k}=i_{m}=i_{l}$ for all $m \in\{k+1, \ldots, l-1\}$. This follows directly from Algorithm 12, where the Ifcondition checks whether there exists another $j \neq j_{l-1}$ in the set $\left\{j_{l}, \ldots, j_{M}\right\}$ that fits to $i_{t-1}$ of the preceding loop.

Let $r \leq M$ be the number of different indices $i_{l}$ and define successively

$$
\begin{aligned}
& \tilde{i}_{1}:=i_{1} \\
& \tilde{i}_{l}:=i_{q_{l}}, \quad \text { where } q_{l}=\min \left\{k \mid i_{k} \notin\left\{\tilde{i}_{1}, \ldots, \tilde{i}_{i-1}\right\}\right\} \quad \text { for } l=2, \ldots, r .
\end{aligned}
$$

where, for notational convenience, $q_{1}:=1,\left(q_{r+1}-1\right):=M$ and $i_{q_{r+1}}>i_{M}$.
Since by construction we have, for $l=1, \ldots, r$, that $i_{q_{t}-1}<i_{l}=i_{q_{l}}=\ldots=$ $i_{q_{I+1}-1}<i_{I_{I+1}}$, the sets

$$
J_{l}:=\left\{j_{q_{l}}, \ldots, j_{q_{l+1}-1}\right\} \quad \text { for } l=1, \ldots, r
$$

are disjoint and $\{1, \ldots, M\}=\bigcup_{l=1}^{r} J_{l}$.
By (18) we have, for $l=1 \ldots, r$,

$$
\begin{align*}
k_{i j_{q_{I}}} & =k_{i q, q_{q,}} & <0 \\
& \vdots &  \tag{19}\\
k_{i j_{q_{l+1}-1}} & =k_{\left.i_{(q++1}-1\right) j_{\left(q q_{+1}-1\right)}} & <0
\end{align*}
$$

and, for $l=1, \ldots, r-1$,

$$
\begin{equation*}
k_{i, j}=k_{i_{\left(q_{1+1}-1\right.} j} \leq 0 \text { for all } j \in\left\{j_{q_{++1}}, \ldots, j_{M}\right\}=\bigcup_{r \geq m>l} J_{m} . \tag{20}
\end{equation*}
$$

Since the $i_{l} \mathrm{~S}$ are chosen according to Algorithm 12, we conclude that

$$
\tilde{i}_{i} \notin\left(\bigcup_{j \in\left\{j_{q_{l+1}}, \ldots, j_{M}\right\}} S u b_{j}\right)
$$

which is equivalent to $\tilde{i}_{l} \notin S u b_{j}$ for all $j \in J_{m}, r \geq m>l$, and a repeated application of Remark 1 yields, for ail $j \in J_{m}, r \geq m>l$, that $k_{i j} \geq 0$.

Therefore, by (19) and (20) we conclude that

$$
\begin{align*}
& k_{i, j}<0 \text { for } j \in J_{l}  \tag{21}\\
& k_{i, j}=0 \text { for } j \in J_{m}, \quad r \geq m>l .
\end{align*}
$$

We are now in a position to determine the permutation matrices.
Let $R \in \mathbb{R}^{M \times M}$ be the permutation matrix of

$$
\binom{1, \ldots, M}{j_{1}, \ldots, j_{M}} .
$$

Then $\bar{K}=\left(\bar{k}_{l m}\right):=K \cdot R$ satisfies, for all $l=1, \ldots, N$,

$$
\begin{equation*}
\vec{k}_{l m}=k_{l_{m},}, \quad m=1, \ldots, M \tag{22}
\end{equation*}
$$

Let $L \in \mathbb{R}^{N \times N}$ be the permutation matrix of

$$
\left(\begin{array}{cccccccccc}
1 & , & , & r & r+1 & , & \cdots & , & N \\
i_{1} & , & \cdots & , & i_{r} & & f_{r+1} & , & \cdots & , \\
f_{N}
\end{array}\right),
$$

where $\left(f_{r+1}, \ldots, f_{N}\right)$ is some permutation of $\{1, \ldots, N\} \backslash\left\{\tilde{i}_{1}, \ldots, \tilde{i}_{r}\right\}$.
Now $\tilde{K}=\left(\tilde{k}_{l m}\right):=L \cdot \bar{K}$ satisfies, for all $m=1, \ldots, M$,

$$
\tilde{k}_{l m}= \begin{cases}\bar{k}_{i, m}, & l=1, \ldots, r \\ \bar{k}_{f i m}, & l=r+1, \ldots, N\end{cases}
$$

which by (22) yields, for $m=1, \ldots, M$,

$$
\tilde{k}_{l m}= \begin{cases}\bar{k}_{i, m}=k_{i j m}, & l=1, \ldots, r  \tag{23}\\ \bar{k}_{f i m}=k_{f i l m}, & l=r+1, \ldots, N .\end{cases}
$$

Therefore (21) and (23) yield

$$
\begin{array}{ll}
\tilde{k}_{l m}<0 \text { for } m=q_{l}, \ldots,\left(q_{l+1}-1\right) & \text { and } l=1, \ldots, r \\
\tilde{k}_{i m}=0 \text { for } m=q_{l+1}, \ldots, M & \text { and } l=1, \ldots, r-1,
\end{array}
$$

which by (16) leads to the echelon form (17). This proves necessity.
To prove sufficiency assume that $\tilde{K}:=L K R$ has the form (17) for some permutation matrices $L \in \mathbb{R}^{N \times N}$ and $R \in \mathbb{R}^{M \times M}$.
Let $L \in \mathbb{R}^{N \times N}$ describe the permutation $\sigma$

$$
\left(\begin{array}{cccc}
1 & , & \cdots & N \\
\sigma(1) & , & \cdots & \\
\sigma(N)
\end{array}\right)
$$

and $R \in \mathbb{R}^{M \times M}$ the permutation $\pi$

$$
\left(\begin{array}{cccc}
1 & , & \cdots & , \\
\pi(1) & , & \cdots & M \\
& \pi(M)
\end{array}\right)
$$

Then the coefficients of $\tilde{K}=\left(\tilde{k}_{i j}\right)$ and of $K=\left(k_{i j}\right)$ are related as

$$
\begin{equation*}
\tilde{k}_{l m}=k_{\sigma(l) \pi(m)} \text { for } 1=1, \ldots, \mathrm{~N} \quad \text { and } \quad \mathrm{m}=1, \ldots, \mathrm{M} \tag{24}
\end{equation*}
$$

Since $\tilde{K}$ has the form (17), there exist $r$ numbers $q_{1}, \ldots, q_{r} \in\{1, \ldots, M\}$ such that (16) is satisfied. Setting $J_{l}:=\left\{q_{l}, \ldots, q_{l+1}-1\right\}^{3}$ for $l=1, \ldots, r$, yields the disjoint subsets $\bigcup_{j=1}^{r} J_{l}=\{1, \ldots, M\}$.

Now let $S \subseteq\{1, \ldots, M\}, S \neq \emptyset$, be arbitrary and choose $\bar{j} \in S$ such that $\pi^{-1}(\bar{j})=\min \left\{\pi^{-1}(\mathrm{j}) \mid \mathrm{j} \in \mathrm{S}\right\}$. Then there exists $\bar{l} \in\{1, \ldots, r\}$ such that $\pi^{-1}(\bar{j}) \in J_{\bar{l}}$.

In order to apply (13), we show that

$$
\begin{equation*}
\sigma(\bar{l}) \in S u b_{j} \quad \text { and } \quad \sigma(\bar{l}) \notin \bigcup_{j \in S}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right) \tag{25}
\end{equation*}
$$

From (16) it follows that $\tilde{k}_{l \pi^{-\mathrm{i}}(\bar{j})}<0$ and (24) yields $k_{\sigma(\bar{l}) \bar{j}}<0$, which by Remark 1 means that $\sigma(\bar{l}) \in S u b_{\bar{j}}$. This means the first condition in (25) is satisfied. Since $\pi^{-1}(\hat{j}) \leq \pi^{-1}(j)$, for all $j \in S$, we obtain

$$
\bar{k}_{l \pi^{-1}(j)} \leq 0 \quad \text { for all } \quad j \in S
$$

which by (24) yields $k_{\sigma(\bar{l}) j} \leq 0$, for all $j \in S$ and this means, again by Remark 1 , that $\sigma(\bar{l}) \notin \bigcup_{j \in S}\left(A u t_{j} \cup \operatorname{Prod}_{j}\right)$. This proves the second condition in (25) and hence (13) is satisfied, whence the process is non-cyclic.

In Example 17 we show that the form (17) is not unique. This completes the proof of the theorem.

Example 17 Consider the reaction scheme and the corresponding stoichiometric matrix

[^2]Algorithm 12 leaves some freedom how to choose the pairs $\left(i_{k}, j_{k}\right), k=1,2,3$. Three possibilities are as follows.

- Let $\left(i_{1}, j_{1}\right)=(1,1),\left(i_{2}, j_{2}\right)=(2,2)$ and $\left(i_{3}, j_{3}\right)=(3,3)$, and hence

$$
\tilde{K}=\left(\begin{array}{ccc}
-c_{11} & 0 & 0 \\
\hdashline-c_{21} & -c_{22} & 0 \\
c_{31} & -c_{32} & -c_{33} \\
c_{41} & 0 & 0 \\
0 & c_{52} & -c_{53} \\
0 & c_{62} & 0 \\
0 & 0 & c_{73} \\
0 & 0 & c_{83}
\end{array}\right)
$$

- Let e $\left(i_{1}, j_{1}\right)=(2,1),\left(i_{2}, j_{2}\right)=(2,2)$ and $\left(i_{3}, j_{3}\right)=(3,3)$, and hence

$$
\tilde{K}=\left(\begin{array}{ccc}
-c_{21} & -c_{22} & 0 \\
\hline c_{31} & -c_{32} & -c_{33} \\
-c_{11} & 0 & 0 \\
c_{41} & 0 & 0 \\
0 & c_{52} & -c_{53} \\
0 & c_{62} & 0 \\
0 & 0 & c_{73} \\
0 & 0 & c_{83}
\end{array}\right)
$$

Alternatively we may also choose $i_{3}=5$.

- Another choice is $\left(i_{1}, j_{1}\right)=(1,1),\left(i_{2}, j_{2}\right)=(3,2)$ and $\left(i_{3}, j_{3}\right)=(3,3)$, which yields

$$
\tilde{K}=\left(\begin{array}{ccc}
-c_{11} & 0 & 0 \\
c_{31} & -c_{32} & -c_{33} \\
-c_{21} & -c_{22} & 0 \\
c_{41} & 0 & 0 \\
0 & c_{52} & -c_{53} \\
0 & c_{62} & 0 \\
0 & 0 & c_{73} \\
0 & 0 & c_{83}
\end{array}\right)
$$

In every case we obtain a matrix $\bar{K}$ that has the form (17) and we conclude that the process is non-cyclic.

An immediate consequence of the form (17) is the existence of a positive vector $\gamma \in\left(\mathbb{R}_{+}^{*}\right)^{N}$ such that $\gamma^{T} \widetilde{K}_{j} \leq 0$ for each column $\widetilde{K}_{1}, \ldots, \widetilde{K}_{M}$ of $\widetilde{K}$. This
immediately yields the following corollary which is crucial to ensure boundedness of the trajectory of the biotechnological process.

Corollary 18 Theorem 8 remains valid if assumption ( $A 3$ ) is replaced by the assumption that (6) is non-cyclic.

## 1 CONCLUSIONS

General reactor models are frequently used in the engineering literature to study the dynamical behaviour and control of biotechnological processes. However, if the stoichiometric matrix is not exact, and this is always the case when components that do not play an important role in the process have been omitted to achieve a lower dimensional model, then these models are not described sufficiently accurate so that it is not clear whether they exhibit dynamical properties which are expected, for example boundedness of the trajectories. To overcome this problem, we introduced the concept of non-cyclic biotechnological processes. This is a restriction on the stoichiometric matrix, however, as far as we are aware, most papers on control of biotechnological processes satisfy this condition. We presented an algorithm which decides in finitely many steps whether a process is non-cyclic or cyclic. This algorithm is also essential when characterising non-cyclic processes in terms of permutations of columns and rows of the stoichiometric matrix $K$ into an echelon form. The main consequence from this form is that non-cyclic processes exhibit bounded trajectories. These dynamical properties can be used to achieve adaptive setpoint control in noncyclic processes. Preliminary results on this can be found in Ilchmann et al. [8].

## APPENDIX

In the following we present some modifications of the Bellman-Gronwall Lemma which are taylored for our needs.

Lemma 19 Let $\omega \in(0, \infty]$. Suppose $\alpha(\cdot), f(\cdot):[0, \omega) \rightarrow I R$ are locally integrable functions and $V(\cdot):[0, \omega) \rightarrow \mathbb{R}_{+}$is absolutely continuous. If

$$
\begin{equation*}
\dot{V}(t) \leq-\alpha(t) V(t)+f(t), \quad \text { for almost all } \quad t \in[0, \omega) \tag{26}
\end{equation*}
$$

then $V(t)$ satisfies the following:
(i) For all $t \in\left[t_{0}, \omega\right)$ and $t_{0} \geq 0$,

$$
\begin{equation*}
V(t) \leq e^{-\int_{t_{0}}^{t} \alpha(\tau) d \tau} V\left(t_{0}\right)+\int_{t_{0}}^{t} e^{-\int_{s}^{1} \alpha(\tau) d \tau} f(s) . \tag{27}
\end{equation*}
$$

(ii) If $\alpha(\cdot)$ and $f(\cdot)$ satisfy, for some numbers $\alpha \bar{f}$,
then, for all $t \in\left[t_{0}, \omega\right)$ and $t_{0} \geq 0$,

$$
\begin{equation*}
V(t) \leq\left[V\left(t_{0}\right)-\frac{\bar{f}}{\underline{\alpha}}\right] e^{-\underline{\underline{\alpha}}\left(t-t_{0}\right)}+\frac{\bar{f}}{\underline{\alpha}} \leq \max \left\{V\left(t_{0}\right), \frac{\bar{f}}{\underline{\alpha}}\right\} . \tag{28}
\end{equation*}
$$

If $\omega=\infty$, then

$$
\begin{equation*}
\limsup _{t \rightarrow \infty} V(t) \leq \frac{\bar{f}}{\underline{\alpha}} . \tag{29}
\end{equation*}
$$

(iii) Suppose instead of (26) we have, for almost all $t \in[0, \omega)$,

$$
\begin{equation*}
\dot{V}(t) \geq-\alpha(t) \cdot V(t)+f(t), \tag{30}
\end{equation*}
$$

then, for all $t \in\left[t_{0}, \omega\right)$ and $t_{0} \geq 0$,

$$
\begin{equation*}
V(t) \geq e^{-\int_{0} \alpha(\tau) d \tau} V\left(t_{0}\right)+\int_{t_{0}}^{t} e^{-\int_{s}^{i} \alpha(\tau) d \tau} f(s) d s, \tag{31}
\end{equation*}
$$

and if

$$
0<\bar{\alpha}=\operatorname{ess} \sup _{t \in[0, \omega)} \alpha(t) \quad \text { and } \quad f_{0} \leq e s \inf _{t \in[0, \omega)} f(t) \text {, }
$$

then, for all $t \in\left[t_{0}, \omega\right)$ and $t_{0} \geq 0$,

$$
\begin{equation*}
V(t) \geq\left[V\left(t_{0}\right)-\frac{f_{0}}{\bar{\alpha}}\right] e^{-\bar{\alpha}\left(t-t_{0}\right)}+\frac{f_{0}}{\bar{\alpha}} \geq \min \left\{V\left(t_{0}\right), \frac{f_{0}}{\bar{\alpha}}\right\} . \tag{32}
\end{equation*}
$$

$$
\text { If } \omega=\infty, \text { then }
$$

$$
\begin{equation*}
\lim \inf _{t \rightarrow \infty} V(t) \geq \frac{f_{0}}{\bar{\alpha}} \tag{33}
\end{equation*}
$$

Proof:
(i) This part of the proof is a straightforward generalization of the proof of Lemma 3.2.4 in Ioannou and Sun [9], where they consider constant $\alpha . f$ and $\omega=\infty$ instead. Let

$$
w(t):=\dot{V}(t)+\alpha(t) V(t)-f(t)
$$

and apply Variation-of-Parameters to

$$
\dot{V}(t)=-\alpha(t) V(t)+f(t)+w(t)
$$

so that, for all $t \in\left[t_{0}, \omega\right)$ and $t_{0} \geq 0$,

$$
V(t)=e^{-\int_{0}^{t} \alpha(\tau) d \tau} V\left(t_{0}\right)+\int_{t_{0}}^{t} e^{-\int_{s}^{1} \alpha(\tau) d \tau}[f(s)+w(s)] d s
$$

and since $w(t) \leq 0$, (27) follows.
(ii) Applying the bounds in (ii) to (27) yields

$$
V(t) \leq e^{-\underline{\alpha}\left(t-t_{0}\right\rangle} V\left(t_{0}\right)+\int_{t_{0}}^{i} e^{-\underline{\alpha}(t-s)} \bar{f} d s
$$

and evaluating the integral leads to the first equation in (28).
Now if $V\left(t_{0}\right)<\bar{f} / \underline{\alpha}$, then $V(t) \leq \bar{f} / \underline{\alpha}$; and if $V\left(t_{0}\right) \geq \bar{f} / \underline{\alpha}$, then $V(t) \leq V\left(t_{0}\right)$. This proves the second inequality in (28).
(29) is an immediate observation from (28).
(iii) The statements in (iii) are proved similarily and the proof is omitted for brevity.

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[^1]:    ${ }^{2}$ For simplicity the stoichiometric coefficients are neglected.

[^2]:    ${ }^{3}$ Put, for notational simplicity, $q_{r=1}-1:=M$.

