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## Research papers

# The impact of supraglacial debris on proglacial runoff and water chemistry

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#### ABSTRACT

Debris is known to influence the ablation, topography and hydrological systems of glaciers. This paper determines for the first time how these influences impact on bulk water routing and the proglacial runoff signal, using analyses of supraglacial and proglacial water chemistry and proglacial discharge at Miage Glacier, Italian Alps. Debris does influence the supraglacial water chemistry, but the inefficient subglacial system beneath the debris-covered zone also plays a role in increasing the ion contribution to the proglacial stream. Daily hydrographs had a lower amplitude and later discharge peak compared to clean glaciers and fewer diurnal hydrographs were found compared to similar analysis for Haut Glacier d'Arolla. We attribute these observations to the attenuating effect of the debris on ablation, smaller input streams on the debris-covered area, a less efficient subglacial system, and possible leakage into a raised sediment bed beneath the glacier. Strongly diurnal hydrographs are constrained to periods with warmer than average conditions. 'Average' weather conditions result in a hydrograph with reverse asymmetry. Conductivity and discharge commonly show anti-clockwise hysteresis, suggesting the more dilute, rapidly-routed melt component from the mid-glacier peaks before the discharge peak, with components from higher up-glacier and the debris-covered areas arriving later at the proglacial stream. The results of this study could lead to a greater understanding of the hydrological structure of other debris-covered glaciers, with findings highlighting the need to include the influence of the debris cover within future models of debris-covered glacier runoff.

# 1. Introduction

Glacial melt is an important component of downstream runoff, especially during dry seasons (Maurya et al., 2011; Xu et al., 2009). Understanding the runoff hydrograph is important for water resources management, for example for irrigation (e.g. in the Tien Shan region of China (Wang et al., 2014)); and the development of hydropower projects (e.g. in the Indian Himalaya (Srivastava et al., 2014)). There is a need to provide a more complete assessment of how glacier retreat may impact on hydropower revenues (Milner et al., 2017). The diurnal amplitude of runoff from clean glaciers is expected to increase and the lag time to peak runoff to decrease in a warmer climate as the snowpack has a progressively smaller hydrological influence (Escher-Vetter and Reinwarth, 1994).

However, given the likely increase in debris cover on alpine glaciers (Stokes et al., 2007), and the widespread occurrence of debris-covered glaciers in high-mountain regions (Scherler et al., 2018), projections of

runoff hydrographs need to include debris cover effects. Debris-covered glaciers are especially common in the Pamirs and Himalaya (Bolch et al., 2012; Scherler et al., 2011), Caucasus Mountains, Russia (Stokes et al., 2007) and the Western Alps (Deline et al., 2012) with the extent and thickness of debris cover increasing in many areas (Bhambri et al., 2011; Bolch et al., 2008; Kirkbride and Deline, 2013; Lambrecht et al., 2011). Supraglacial debris cover reduces melt compared to bare ice, except where the cover is thin or discontinuous (Kirkbride and Dugmore, 2003; Mattson et al., 1993; Østrem, 1959). Debris also reduces the amplitude of the melt signal and causes an increase in the lag from peak air temperature to peak melt, with the effect greater for thicker covers (Fyffe et al., 2014).

Recent dye tracing work (Fyffe et al., 2019) found that a supraglacial debris cover reduced meltwater inputs and resulted in less efficient flow beneath the debris-covered area. In contrast, up-glacier, the higher ablation rates of the clean and dirty ice and the focussing of supraglacial streams between moraine ridges produced an efficient conduit

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drainage system. Despite this new understanding, dye tracing can only ever elucidate the system efficiency for specific drainage routes during the time of the test, and cannot identify the proportion of water in each part of the system and how this varies diurnally and seasonally.

However, hydrochemistry and proglacial hydrograph investigations can reveal variations in bulk water routing, providing a complementary perspective on this rarely studied system. Glacier hydrochemistry can elucidate the conditions through which meltwater travels before emerging at the snout (see Brown (2002) for a review), with the glacial runoff pattern a function of the recharge (input) hydrograph and how it has been influenced by the glacier hydrological system (Covington et al., 2012). Essentially, the proglacial discharge and its chemistry provides a cumulative insight into all of the meltwater sources and their transport paths, so that if the chemical signature of a flow component (usually defined by a meltwater source or transport path) is known, then its lag time and proportion of discharge can be determined (Brown, 2002). Understanding bulk water routing also allows a clearer understanding of the processes leading to sediment transport and solute acquisition. Variations in suspended sediment have implications for nutrient cycling (Hawkings et al., 2016; Hodson et al., 2004), and also influence turbidity, affecting light penetration and primary productivity (Milner et al., 2017).

Furthermore, there is currently little knowledge of how the unusual hydrological system identified from dye tracing affects the glacier hydrochemistry and proglacial runoff. Some work has dealt with solute, SSC and discharge characteristics of debris-covered glaciers (see Table 1), but these studies do not relate their findings to the existence of a supraglacial debris cover. We hypothesis that proglacial solute will be enhanced due to prolonged contact with debris both supraglacially and subglacially, especially for the delayed flow component composed of sub-debris melt. It is expected that the proglacial hydrograph will be relatively subdued, with a long time to peak discharge and a less strongly peaked diurnal hydrograph, given the known lag time to peak melt beneath debris (Fyffe et al., 2014), and the less efficient hydrological system beneath the lower glacier (Fyffe et al., 2019).

Using data collected from the debris-covered Miage Glacier, Italy, this paper therefore has the following aims:

- Determine the influence of supraglacial debris on supraglacial and proglacial water chemistry.
- Determine how the debris' impact on the supraglacial hydrograph and englacial and subglacial drainage system structure determine the form and magnitude of the proglacial hydrograph.
- 3. Establish an understanding of the main flow components in debriscovered glacier runoff.

#### 2. Study site

Miage Glacier is a 10.5 km<sup>2</sup> glacier with an elevation range of 1740-4640 m a.s.l., situated in the western Italian Alps (Fig. 1). It sits within a catchment of approximately 22.7 km<sup>2</sup>, bounded by the ridge crest from Pointe Baretti around to Petit Mont Blanc (Deline, 2005), and by large lateral moraines surrounding the lower glacier. Miage Glacier has four steep icefall tributaries: the Mont Blanc, Dome, Bionassay and Tête Carrée Glaciers. The main tongue occupies a deep trough from which it flows into Val Veny, bending eastwards and dividing into the large northern and southern lobes and a smaller central lobe. The glacier has a variety of surface types: at high altitudes clean ice and snow dominate; in the mid-part of the glacier (c. 2400-2600 m a.s.l.) relief is controlled by medial and lateral moraines, with a patchy covering of debris (hereafter 'dirty ice') in the intervening troughs; whereas the lower 5 km of the glacier is almost continuously covered by debris with a mean thickness of 0.25 m (Foster et al., 2012). The described extent of the continuous debris cover (relevant to the 2010/2011 field seasons) has existed since the 1930s (Deline, 2005), though there has been an observed increase in debris cover at the expense of the moraine-trough

dirty ice since at least 2012. The debris originates predominantly from rockfalls and mixed snow and rock avalanches from the steep valley sides (Deline, 2009).

#### 3. Methodology

Meteorological, hydrological and water chemistry measurements were conducted on Miage Glacier over the 2010 and 2011 summer seasons (Table 2).

#### 3.1. Meteorology and melt modelling

Two energy balance weather stations were installed on continuous debris on the upper (UWS) and lower (LWS) glacier in 2010 and 2011, with a further station (IWS) measuring only air temperature situated over an area of thin and patchy debris in 2011 (Fig. 1). For full details of the instrumentation and variables measured see Brock et al. (2010) and Fyffe et al. (2014).

A distributed energy balance melt model was constructed for Miage Glacier (see Fyffe et al. (2014) for detailed model methods and results). The model runs at an hourly timestep over a  $30 \times 30$  m grid over the Miage Glacier catchment, with exactly the same inputs and parameters as outlined in Fyffe et al. (2014). The difference between simulated and measured melt at long term stakes was  $-0.001 \,\mathrm{m}\,\mathrm{d}^{-1}$  (5%), with the RMSE between simulated and measured debris surface temperature at LWS  $1.8 \,^{\circ}$ C ( $R^2 = 0.95$ ). Comparison between simulated melt plus effective precipitation (defined as rainfall minus evaporation, with rainfall distributed using a rainfall lapse rate derived from measured rainfall at tipping bucket rain gauges situated at LWS and UWS) and measured discharge resulted in a RMSE of 3.26 m<sup>3</sup> s<sup>-1</sup> in 2010 and  $3.77 \,\mathrm{m}^3 \,\mathrm{s}^{-1}$  in 2011, with a model bias of -0.94 in 2010 and -2.60 in 2011. This underestimation of discharge is likely due to raingauge undercatch since values corresponded well during dry periods. Note that no routing routines were applied to account for water transport through the hydrological system.

# 3.2. Supraglacial and proglacial water chemistry

Water samples were taken in the morning and afternoon at the gauging station during field visits, almost daily in June 2010 and every second or third day during other field trips. The average time of morning measurements in 2010 was 10:44 and in 2011 was 10:06, with the average time of afternoon measurements 16:54 in 2010 and 16:43 in 2011. Conductivity and water temperature were measured continuously at the gauging station (see Table 3) at one minute intervals during field visits, and 30 minute intervals between visits. Unfortunately, clogging of the probe with sediment resulted in large periods of erroneous conductivity data. Conductivity data were collected from 23/06/2010 to 28/06/2010, from 28/07/2010 to 23/08/2010, from 06/09/2010 to 12/09/2010, from 05/06/2011 to 26/06/2011, from 26/07/2011 to 24/08/2011, and from 12/09/2011 to 15/09/ 2011, although there are smaller gaps within these periods. Water samples were also collected from supraglacial streams, ponds and lakes and three snow samples were melted and analysed, see Fig. 1 and Table 4. Sample locations were chosen to represent the range of supraglacial environments on the glacier and allow the conditions conducive to solute and sediment acquisition to be determined.

Water samples were filtered in the field using a Nalgene field filter holder and receiver and hand pump, and a pre-weighed Whatman cellulose nitrate membrane filter (0.45  $\mu m$  pore size and 47 mm diameter), following standard procedures (Hubbard and Glasser, 2005). The filter papers were oven dried and weighed in the laboratory to determine the suspended sediment concentration (SSC). Each sample was stored in an acid-rinsed bottle in a cool bag or fridge except during transit to the UK. Whenever a water sample was taken the conductivity, water temperature, turbidity and pH of the stream, pond or lake was

Table 1
Summary of relevant papers addressing the water chemistry and proglacial runoff characteristics of debris-covered glaciers. Key characteristics of the glaciers mentioned are given in Table 10. \*The Parlung No. 4 Glacier is not debris-covered but the paper has been included since it mentions findings related to debris-covered glaciers in the Tibetan region. The current paper is added for completeness.

Study	Location (only debris- covered glaciers)	Focus	Key Results
This study.	Miage Glacier, Italian Alps	Proglacial runoff, solute and SSC	SSC and bicarbonate concentrations are enhanced by the supraglacial debris. The inefficient subglacial system (itself a result of the debris cover) increases the bicarbonate ad sulphate concentrations compared to clean glaciers. The baseflow component of discharge and the lag time between peak air temperature and discharge are increased due to the delaying effects of the debris. Clearly diurnal hydrographs are
Han et al. (2010)	Koxkar Glacier, China	Proglacial runoff	also confined to periods of warmer weather.  Seasonal runoff variability is greater than annual, with maximum discharge in August.  Maximum runoff is 1800–0200 with minimum 0700–1000. Lag time between peak melt and discharge varies between 4 and 10 h.
Hasnain and Thayyen (1994)	Dokriani Glacier, Garhwal Himalaya	Proglacial runoff and solute	Used conductivity measurements to separate the hydrograph into englacial and subglacial components. Conductivity varied diurnally and out of phase with discharge.
Hasnain and Thayyen (1999a)	Dokriani Glacier, Garhwal Himalaya	Proglacial runoff, solute and SSC	The chemical-denudation rate was particularly high, indicating intense chemical weathering. Chemical weathering likely dominated by coupled reactions of sulphide oxidation and carbonate dissolution. Monsoonal rainfall thought to enhance weathering of supraglacial debris and result in high proglacial concentrations of sulphate.
Hasnain and Thayyen (1999b)	Dokriani Glacier, Garhwal Himalaya	Proglacial runoff and SSC	The majority of runoff, suspended sediment transport and rainfall occurred during the monsoon months of July and August. The timing of discharge and SSC peak became progressively earlier in the day over the ablation season.
Hodson et al. (2002)	Batura Glacier, Pakistan	Solute dynamics	The majority of solute is derived from carbonate weathering. Chemical denudation is responsible for 75% of the solute yield with rates high and similar to the upper limits for temperate glaciers. The silicate-derived cation denudation rate is also high, likely due to
Irvine-Fynn et al. (2017)	Khumbu Glacier, Nepal	Supraglacial lake runoff	ablation and intense monsoon precipitation causing high flushing rates.  Supraglacial ponds act as storage reservoirs and modulate proglacial discharge, with ponds storing > 23% of mean daily discharge. As supraglacial pond and debris cover extent increases the authors expect the runoff signal to become further attenuated.
Kumar et al. (2009)	Gangotri Glacier, Garwhal Himalaya	Solute dynamics	The dominant cations were found to be magnesium and calcium with bicarbonate and sulphate the dominant anions. Carbonate weathering mainly controls the meltwater chemistry. Inverse relationships are found between discharge and some cations, but the relationships are weak, possibly due to high flow events corresponding with rising cation concentrations.
Li et al. (2016)	Parlung No. 4 Glacier <sup>*</sup>	Proglacial runoff	Air temperature, vapour pressure and incoming shortwave radiation were significantly positively correlated with discharge. Daily discharge could be simulated from a multiple regression model. Most discharge occurred in July and August with the daily discharge cycle altering over the season. Differences in the inter-annual proglacial discharge characteristics were found between glaciers in the Tibetan Plateau. These differences
Liu et al. (2010)	Rongbuk Glacier, Central Himalaya	Proglacial runoff	are explained by the existence of debris cover on some glaciers.  The high nighttime discharges suggested significant storage, explained by the large glacier area and existence of glacier lakes. The time lag between maximum daily air temperature and discharge ranged from 8 to 14 h, with this time reducing over the season. Air temperature correlated well with discharge. Most runoff occurred between April and October.
Singh et al. (1995)	Dokriani Glacier, Garhwal Himalaya	Proglacial runoff, SSC and water temperature	Analysis of the proglacial runoff included determination of mean amplitude, recession analysis and peak timing. Discharge and SSC did not have a significant relationship. Runoff and air temperature were significantly correlated suggesting that solely air temperature could be used for future melt modelling.
Singh et al. (2004)	Dokriai Glacier, Garwhal Himalaya	Proglacial runoff	Nighttime discharges are high and comparable to daytime discharges throughout the season, explained by storage which is more significant earlier in the season. Diurnal variations become clearer over the season, with the amplitude greatest in August. Lag times between peak air temperature and discharge ranged between 3 and 6 h, shortening over the ablation season.
Singh et al. (2006)	Gangotri Glacier, Garwhal Himalaya	Proglacial runoff	Proglacial runoff is highest in July, but with maximum diurnal variability in May and September. Daytime and nighttime runoff magnitudes are similar, demonstrating the strong storage characteristics of the glacier. The time lag from peak air temperature to peak discharge varied from 4 to 7.30 h. The relationship between air temperature and discharge was stronger on a monthly rather than daily timescale.

also measured using hand-held probes (Table 3).

Each sample was tested for the concentration of sulphate and bicarbonate ions. These ions were chosen due to their common use as indicators of the subglacial environment: sulphate ions are generated by sulphide oxidation, within water routed slowly within a distributed system (although snow and groundwater can also be a source of sulphate); with bicarbonate ions generated from acid hydrolysis, signifying contact of meltwater with sediment either at the bed or suspended in-stream (Brown, 2002; Brown et al., 1996; Raiswell, 1984; Tranter et al., 1996, 1993a; Tranter and Raiswell, 1991). Sulphate concentration was found by adding a powdered sulphate reagent, with the sample absorbance tested in a CECIL CE 4002 spectrophotometer

(accuracy of  $\pm$  0.005A or 1%, whichever is greater). The test has a stated measurement range of between 2 and 100 mg l $^{-1}$ . The measured absorbance was calibrated to sulphate concentration using a calibration curve. The results were not accepted if they differed by more than 100  $\mu eq\,l^{-1}$ . For the remaining samples the mean (median) absolute difference between samples was 28.7 (19.4)  $\mu eq\,l^{-1}$  or 18 (12)% in 2010 and 9.3 (5.3)  $\mu eq\,l^{-1}$  or 51 (8)% in 2011. The mean percentage difference is larger in 2011 due to the increased number of lower concentration supraglacial samples.

The concentration of bicarbonate was found by titrating 20 ml of sample with  $0.01\,\mathrm{mol}\,\mathrm{l}^{-1}$  hydrochloric acid using Kittiwake pH 4.5 indicator. Two titrations were carried out on each sample, and they

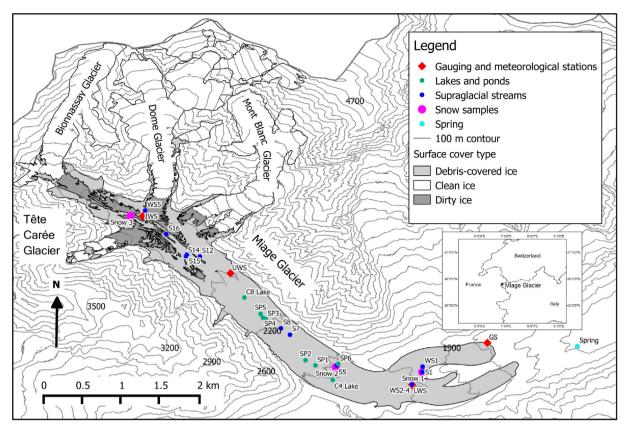


Fig. 1. Miage Glacier showing the location of water chemistry samples and gauging and meteorological stations. Note that 'S' supraglacial streams were also dye traced (see Fyffe et al. (2019)) whereas 'WS' streams were monitored only for water chemistry purposes. The prefix 'SP' is for small supraglacial ponds, in distinction to the large lakes (diameter of tens of metres) at C8 and C4 (GPS locations mentioned in (Fyffe, 2012)).

Table 2
Guide to fieldwork dates and data collected. 'Water Chemistry', 'Discharge' and 'Conductivity' measurements were of the proglacial stream and were not necessarily continuous throughout these periods (see individual sections for details).

Year	Months	Start day	End day	Measurements
2010	June July/Aug Sep June–Sep June–Sep	05/06/2010 28/07/2010 04/09/2010 05/06/2010 05/06/2010	24/06/2010 06/08/2010 10/09/2010 12/09/2010 13/09/2010	Water Chemistry Water Chemistry Water Chemistry Discharge, Conductivity Meteorology
2011	June July/Aug Sep June–Sep June–Oct	04/06/2011 26/07/2011 12/09/2011 04/06/2011 04/06/2011	15/06/2011 04/08/2011 15/09/2011 16/09/2011 11/10/2011	Water Chemistry Water Chemistry Water Chemistry Discharge, Conductivity Meteorology

were accepted if the volume of hydrochloric acid added agreed within 2 ml (59.8  $\mu eq\,l^{-1}$  of bicarbonate). If not then a third test was performed, and if two of the three tests agreed then their values were used instead. The average absolute difference between the tests was 24.7  $\mu eq\,l^{-1}$  of bicarbonate (5%). Once both the sulphate and bicarbonate concentrations had been found they were used to calculate the C-ratio [HCO $_3$ -/(HCO $_3$ -+SO $_4$ -)] (Brown et al., 1996). The partial pressure of CO $_2$  was also calculated from the pH and bicarbonate concentration (see Brown (2002), Eq. (9)).

#### 3.3. The impact of debris on the proglacial hydrograph

# 3.3.1. Field measurements

Proglacial runoff was derived from water level measurements and a rating curve collected at a gauging station located on the northern lobe stream, upstream of any confluences (GS in Fig. 1). It consisted of a pressure transducer hung in a stilling well, bolted onto a very large boulder (see Table 3). The Onset HOBO pressure data were compensated using air pressure data from Mont de la Saxe, 7.6 km from the gauging station. Dve tracing indicates that the small outflow from the southern lobe is derived locally from only the area of that lobe (Fyffe, 2012) so the runoff from the rest of the glacier exits via the northern lobe. A high flow event in June 2011 caused damage to the well, resulting in data loss between 18/06/2011 and 03/08/2011 and the repositioning of the well. Data was also lost between 27/08/2010 and 28/ 08/2010 and between 04/09/2010 and 08/09/2010. The measured stages from all time periods were adjusted to the datum of the June 2010 well so that one stage-discharge rating could be applied. A total of 16 rhodamine WT dye-dilution gaugings were used to construct a twopart rating curve with a percentage error of 14.6% (see Supplementary Information A for further details).

## 3.3.2. Analysis of the proglacial runoff record

Daily hydrographs were categorised into shape categories following Hannah et al. (2000, 1999) and Swift et al. (2005). The hydrograph classes were predetermined and hydrographs classified manually to allow a more direct comparison of results with other workers. The runoff record was divided into 24 h periods ('days') with the beginning of the 'day' being the average time of minimum runoff (for complete days with no rain) of 11:00 in 2010. Daily hydrographs were classified as either:

- i. diurnal (a peak in the middle of the day, and lower flows at the beginning and end)
- ii. rising,
- iii. falling, or
- iv. unknown (hydrograph steady or could not be assigned to one of the

 Table 3

 Details of proglacial stream and water chemistry instruments.

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Quantity	Location	Time period	Manufacturer	Type	Accuracy	e, et a
Stage	Proglacial Proglacial	2010 and June 2011 Aug and Sep 2011	GE Sensing Onset	Druck PTX1830 (vented) HOBO U20 -001-04 (non-vented)	$\pm$ 0.1% full scale (or $\pm$ 0.06% full scale) $\pm$ 0.075% full scale, $\pm$ 0.07 cm	L.
Fluorescence	Proglacial Proglacial	2010 and June 2011 July, Aug, Sep 2011	Seapoint Turner	Rhodamine fluorometer Cyclops-7 Rhodamine	Not stated but minimum detection 0.02 ppb Not stated but minimum detection 0.01 ppb	
Conductivity and temperature	Supraglacial Proglacial	Both years. Both years.	Hanna Campbell	H19033 with HI 76302 W probe CS547A	$\pm$ 1% full scale (excluding probe) $\pm$ 5% of reading 0.005–0.44 mS cm <sup>-1</sup> $\pm$ 0.4 °C	
pH Turbidity	All	Both years.	Hanna Hanna	HI 98140/HI 98128 HI 93703	$\pm$ 0.01 pH $\pm$ 0.5 FTU or 5% whichever is greater	

other categories).

If a hydrograph fell between two categories then the most prominent feature of the hydrograph was used to determine its category. Only complete days with no rain were categorised. Mean hydrographs for categories i) to iii) are in Fig. 2. Following Swift et al. (2005) specific parameters were calculated for each of the hydrographs classified as diurnal, as well as averages for each year, see Table 5.

The 2010 and 2011 melt season hydrographs were delimited into phases based on the dominant hydrograph type (Fig. 3). The percentage of days of each hydrograph type (diurnal, rising or falling, not including unknown hydrographs) was calculated for a running 10 day period, with the mid-day of the 10 day period used as the phase boundary. The early season period is phase 1. Later in the season (from the beginning of July onwards) hydrographs are classified as either Phase 2, when more than 50% of the hydrographs are diurnally-classified, or Phase 3, when less than 50% of the hydrographs are diurnally-classified.

To find the lag time between peak air temperature and discharge, the time between maximum air temperature and maximum discharge was calculated (with the day starting at 11:00). Lag times were also calculated between maximum air temperature and minimum conductivity and between minimum conductivity and maximum discharge. Only diurnally-classified conductivity data (defined as having a minimum conductivity in the middle of the day and higher values before and after this) were used in this analysis. Lag times were also calculated statistically using a lag correlation between hourly discharge and air temperature records from LWS and UWS. The discharge record was moved back in time in hourly increments and the correlation repeated up to a lag of 24 h, with the lag time with the highest correlation coefficient giving the average lag time. The records were also split into 10 day periods and the process repeated, to reveal seasonal changes. Only the 2010 record was used in the statistical analysis, due to its completeness, and all days were included.

#### 4. Results

# 4.1. The influence of debris on supraglacial and proglacial water chemistry

#### 4.1.1. Supraglacial water chemistry

Supraglacial water chemistry results are given in Table 4, with sulphate and bicarbonate concentrations shown in Fig. 4. There is a distinction in bicarbonate concentrations between the two largest streams on the upper glacier (S12 and S14) which demonstrated lower bicarbonate concentrations (and also lower SSC), and most of the other samples which had bicarbonate concentrations around  $300 \, \mu eq \, l^{-1}$ . Note that although S15 was close to S12 and S15 it had a smaller, mainly debris-covered catchment. SP5 (a small pond) had the highest bicarbonate concentrations, with the lower glacier streams S1 and S7 also showing elevated concentrations. The average supraglacial stream bicarbonate concentration was much lower ( $286 \, \mu eq \, l^{-1}$ ) than the proglacial stream values ( $783 \, \mu eq \, l^{-1}$  in 2010 and  $603 \, \mu eq \, l^{-1}$  in 2011).

There was not a distinction in sulphate concentrations between the larger upglacier streams and the smaller streams on the lower glacier. However, the sulphate concentration of SP5 was particularly high (87  $\mu$ eq l<sup>-1</sup>), with the values for the C4 and C8 lakes also high compared to the supraglacial streams. The elevated sulphate concentrations measured at WS5 are likely because the samples were taken in early June 2010 when its catchment was snow-covered, since snow sample 3 demonstrated elevated sulphate concentrations. The mean sulphate concentration of supraglacial stream measurements of  $16 \mu$ eq l<sup>-1</sup> was an order of magnitude lower than gauging station values.

#### 4.1.2. Proglacial water chemistry

The concentration of sulphate ions measured at the proglacial stream was high compared with other clean glaciers (Table 6), with values most similar to the debris-covered Dokriani Glacier outside of

Table 4
Water quality data. Values are means where more than one sample was taken although mean proglacial values are of twice daily values and do not include values from hourly sampling on specific days. Where the number of measurements averaged does not equal the number of samples stated (due to instrument failure/ unsatisfactory errors in laboratory analysis) the number of measurements is given in brackets. 'S' streams were also dye injection streams, whereas 'WS' streams were used only for water chemistry measurements. 'SP' refers to small ponds whereas the lakes were much larger (diameter of tens of metres). Q is the stream discharge (either of the supraglacial or proglacial stream depending on the sample location), and  $u_s$  is the supraglacial stream velocity.

Name	No. of samples	Cond. ( $\mu S \text{ cm}^{-1}$ )	SSC $(mg l^{-1})$	pH	Temp. (°C)	$HCO_3^-$ ( $\mu eq l^{-1}$ )	$SO_4^{2-}$ (µeq $l^{-1}$ )	C-ratio	$p(CO_2)$	$Q (m^3 s^{-1})$	$u_{\rm s}~({\rm ms^{-1}})$
Proglacial stream	m samples (averag	ged over time perio	ds)								
2010	54	66.2 (46)	312	8.2 (50)	0.4 (50)	783 (47)	200 (46)	0.80 (40)	-3.5(43)	4.19	-
June 2010	33	70.7	381	8.5 (32)	0.4 (32)	856 (28)	204 (26)	0.82 (22)	-3.8(27)	3.06	-
July/Aug 2010	12	62.4 (6)	242 (12)	7.5 (9)	0.4 (9)	714 (10)	199	0.80 (10)	-3.0(7)	7.84	-
Sep 2010	9	48.4 (7)	153	7.7	0.3	630	189 (8)	0.76 (8)	-3.2	3.51	-
2011	22	55.5 (13)	146 (21)	8.2 (13)	-	603 (21)	215 (21)	0.74 (21)	-3.9(12)	4.64	-
June 2011	9	69.0 (4)	106	8.1 (4)	-	608	224	0.73	-3.5(4)	2.68	-
July/Aug 2011	9	47.2 (5)	172 (8)	7.6 (5)	-	608 (8)	199 (8)	0.76 (8)	-3.4(4)	6.22	_
Sep 2011	4	52.4	186	9.1	-	580	228	0.73	-4.6	5.49	-
Lake and pond	samples (averaged	l for each location)									
C4 Lake	1	-	17	-	-	314	47	0.87	-	_	-
C8 Lake	2	-	8	-	-	292	30	0.92	-	_	-
SP5 (pond)	1	56.8	15	9.79	2.7	658	87	0.88	-5.27	-	-
Snow samples (a	average for all sar	mples)									
Snow	3	-	26 (2)	6.70(1)	-0.3(1)	145	76 (2)	0.66(2)	-2.78(1)	-	-
Supraglacial stre	eam samples (ave	raged for each loca	tion)								
WS1	1	7.6	60	8.39	0.5	254	_	_	-4.28	_	_
WS2	1	14.3	610	8.51	0.4	329	30	0.92	-4.29	_	-
WS3	1	10.2	32	8.48	0.2	299	23	0.93	-4.30	_	-
WS4	1	_	125.6	-	_	299	14	0.95	_	_	-
WS5	2	_	0	-	_	254	57	0.83	_	_	-
S1	1	16.1	266	7.66	0.1	464	2	1.00	-3.29	_	-
S5	4	10.2	54	7.53 (3)	0.6(2)	258	12	0.96	-3.42 (3)	0.03(3)	0.17(3)
S7	4	28.0 (3)	55	8.20	-	408	6	0.98	-3.90 (3)	0.02(3)	0.26(3)
S8	1	_	277	-	-	269	21	0.93	_	_	-
S12	4	4.0 (3)	18	7.75 (1)	-	187	10	0.95	-3.73(1)	0.18	0.46
S14	1	_	1	-	-	135	0	1.00	_	0.53 (5)	1.14 (5)
S15	5	18.4 (3)	32	8.86 (1)		278	6	0.98	-4.71 (1)	0.02(2)	0.32(4)

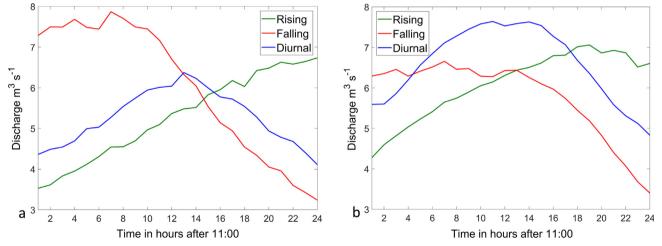


Fig. 2. Mean diurnal, rising and falling hydrographs, a) for 2010 and b) for 2011.

 $\label{eq:table 5} Parameters calculated for each diurnally classified daily hydrograph. Units of all parameters are $m^3 s^{-1}$, except $Q_{dsdamp}$ which is dimensionless.}$ 

Symbol	Parameter
Q <sub>damp</sub>	Diurnal discharge amplitude (maximum minus minimum runoff)
$Q_{dmax}$	Diurnal discharge maximum (runoff peak)
$Q_{dmin}$	Diurnal discharge minimum (either before or after the peak)
Q <sub>dmean</sub>	Diurnal discharge mean
$Q_{dsdamp}$	Diurnal discharge standardised amplitude ((maximum flow-minimum
•	flow)/minimum flow)
$Q_{dstd}$	Diurnal discharge standard deviation

the monsoon season. The highest values of sulphate on Miage Glacier were either when there was a pulse of ion and sediment rich water (09/06/2010, 474  $\mu eq\,l^{-1}$ , corresponding to a fall in discharge and a peak in conductivity, SSC and bicarbonate concentration) or when discharges were low during cold weather (e.g. 367  $\mu eq\,l^{-1}$  on 20/06/2010 and 383  $\mu eq\,l^{-1}$  on 15/09/2011). Even though the highest sulphate concentration measured was on 09/06/10, the delayed-flow sulphate concentrations were still high even in mid-summer (Table 4). The variation between different times of year was small with the difference between mean June, and mean July/August concentrations being only 5  $\mu eq\,l^{-1}$  in 2010, and 25  $\mu eq\,l^{-1}$  in 2011. This contrasts with Haut Glacier d'Arolla and Austre Brøggerbreen where June sulphate

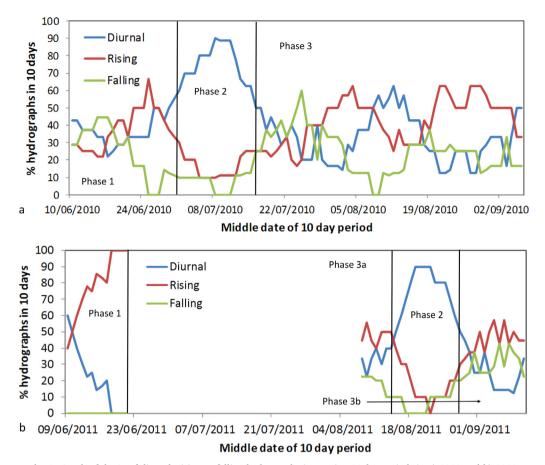


Fig. 3. Graph of the % of diurnal, rising or falling hydrographs in running 10 day periods in a) 2010 and b) 2011.

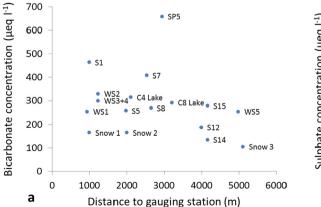
concentrations were at least twice those later in the season (Tranter et al., 1996). The relationship between sulphate and discharge was generally weak, so in 2010 the only relationship with discharge was in September when the sulphate concentration increased as discharge decreased (p-value 0.016,  $R^2 = 37.3$ ), although in 2011 this negative relationship was also found in June (p-value 0.009,  $R^2 = 27.4$ ), and more strongly in July and August (p-value 0.017,  $R^2 = 58.4$ ).

Even taking account of the elevated supraglacial bicarbonate concentrations (Section 4.1), the mean and range of proglacial values are still high compared to clean glaciers (Table 6). The values are most similar to the debris-covered Batura Glacier and Dokriani Glacier during the monsoon season, slightly above the values for Grimsvötn, and below those from the icing at Scott Turnerbreen. As for sulphate,

bicarbonate concentrations were highest either in June 2010 when there was a pulse of sediment and ion rich water  $(913 \, \mu eq \, l^{-1} \text{ on } 09/06/2010$  at 10:05), or during recessional flow, for instance  $1092 \, \mu eq \, l^{-1}$  at 10:30 on 22/06/2010, corresponding with a particularly low discharge of  $1.65 \, \text{m}^3 \, \text{s}^{-1}$ . The bicarbonate concentration decreased over the season in 2010, but this was not mirrored in 2011. Unlike sulphate, the bicarbonate concentration never had a significant relationship with discharge in 2010 or 2011.

## 4.2. The impact of debris on the proglacial hydrograph

The runoff and meteorological conditions in 2010 and 2011 are given in Fig. 5. Average values are calculated for the melt season, with



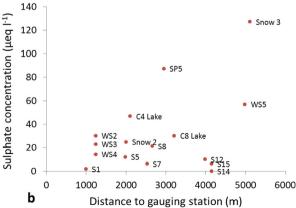


Fig. 4. a) bicarbonate and b) sulphate concentration measured from supraglacial locations plotted against the straight line distance from the gauging station to the sample location. Some values are averages where more than one sample was taken: see Table 4 for details.

Table 6 Comparison of sulphate and bicarbonate ion concentrations between different glaciers. All values are in  $\mu$ eq  $l^{-1}$ , with the mean in brackets and the range giving the maximum and minimum values recorded. Bold indicates debris-covered glaciers with background data of the comparison glaciers in Table 10. \* represents studies cited in Brown (2002). \*\* note that these values include hourly samples, unlike Table 4.

Glacier	Source	Non-snowpack ${\rm SO_4}^{2-}$	HCO <sub>3</sub>
Miage Glacier, Italy in 2010 (all proglacial samples)	This study.	(202) 97-473**	(777) 344-1186 <sup>**</sup>
Miage Glacier, Italy in 2011 (all proglacial samples)	This study.	(215) 128-323	(603) 494-688
Batura Glacier, Pakistan	Hodson et al. (2002)	160	730
Dokriani Glacier, India (pre monsoon)	Hasnain and Thayyen (1999a)	160-418	159-397
Dokriani Glacier, India (monsoon)	Hasnain and Thayyen (1999a)	85-1140	128-1053
Dokriani Glacier, India (post monsoon)	Hasnain and Thayyen (1999a)	137-431	168-384
Gangotri Glacier, India	Kumar et al. (2009)	(673) 333-1186	(1138) 17-4130
Austre Brøggerbreen, Svalbard	Tranter et al. (1996)	10-140	145-520
Bench Glacier, Alaska	Anderson et al. (2000)*	262	427
Chamberlain, USA	Rainwater and Guy (1961)*	29-310	150-200
Engabreen, Norway	Ruffles (1999)*	0-142	51–675
Fjallsjökull, Iceland	Raiswell and Thomas (1984)*	26-66	190-300
Grimsvötn, Iceland	Steinpórsson and Óskarsson (1983)*	132	573
Haut Glacier d'Arolla, Switzerland	Brown et al. (1996)	30-240	180-460
Nigardsbreen, Norway	Brown (2002)	7–40	1.4-8.5
Scott Turnerbreen, Svalbard	Hodgkins et al. (1998)	(130) 96-200	(170) 110-260
Scott Turnerbreen, Svalbard (icing)	Hodgkins et al. (1998)	(830) 0-3200	(1800) 350-4600
Tsanfleuron, Switzerland	Fairchild et al. (1994)*	118	627

2011 meteorological averages found for the same dates as the 2010 record (see Table 2).

#### 4.2.1. Hydrograph shape classification

Hydrograph shape classification (Table 7) revealed that the majority of hydrographs were diurnal in both years, with almost as many rising but fewer falling hydrographs. Of the diurnally classified hydrographs in 2010 (2011), 7 (6) were found in June, 14 (no data) in July, 8 (16) in August and 3 (3) in September. Even diurnally classified hydrographs often did not smoothly rise or fall and could exhibit a flattened or variable peak. Hannah et al. (2000) and Swift et al. (2005) classified daily hydrographs from Haut Glacier d'Arolla into shape categories, with Hannah et al. (1999) conducting similar analysis on Taillon

Glacier runoff. The Haut Glacier d'Arolla and Taillon Glacier runoff records were classified using principle component analysis, meaning the hydrograph classes were statistically derived and unique to the runoff record. A direct comparison with the Miage Glacier classes (although predetermined and conducted manually to aid comparison) should therefore be treated with caution. Nevertheless, Table 7 shows Miage Glacier has fewer diurnal hydrographs over the season than was classified for Haut Glacier d'Arolla in 1998 and 1999. Miage Glacier had a similar number of diurnal hydrographs to Taillon Glacier (and Haut Glacier d'Arolla in 1989), but some of the 'Building/Late peaked' hydrographs (which rise to a peak around 24:00) would have been classed as diurnal for Miage Glacier.

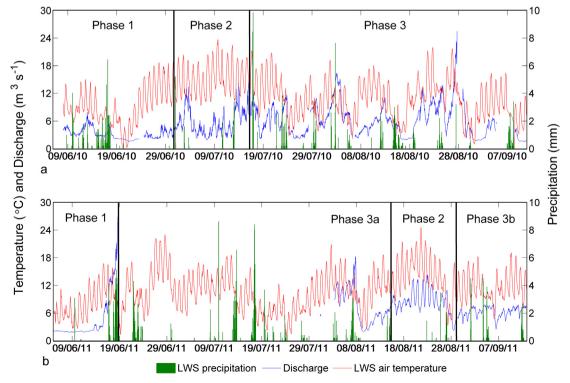


Fig. 5. Hourly air temperature and precipitation at LWS and proglacial discharge for a) 2010 and b) 2011. The left y-axis in b has been constrained to  $30 \text{ m}^3 \text{ s}^{-1}$  to allow comparison between years.

Table 7

Hydrograph shape classification statistics for selected glaciers, with 'N' the number of hydrographs and '%' the percentage of total. Thirteen hydrographs were not classified in 2010 (15%) and 4 in 2011 (7%). Only Miage Glacier is debris-covered. See Table 10 for background data on the comparison glaciers.

Hydrograph shape classifications	Mia	ge Gla	icier		Haut Gla (2005))	cier d'Arolla (	1989: Hannah	et al. (2000)	, 1998 and 19	999): Swift et al.	Taillon Gla	cier (Hannah et al., 1999)
Year	201	0	201	1	1989		1998		1999		1995 + 19	96
_	N	%	N	%	N	%	N	%	N	%	N	%
Rising (*Building/Late Peaked)	26	36	20	38	36 <sup>*</sup>	39*	21	13	30	19	37 <sup>*</sup>	33*
Falling/Recession	15	21	7	13	7	8	27	17	28	18	12	11
Peaked Falling/Peaked Recession					2	2	11	7	5	3		
Peaked/Diurnal	32	44	25	48	47	51	97	62	91	59	56	50
Attenuated					1	1					8	7

**Table 8**Hydrograph statistics for Miage Glacier runoff for individual months and phases. Values given are for diurnally classified hydrographs only except where values are given in brackets which are for all hydrographs.

Phase/Period	Q <sub>damp</sub> (m <sup>3</sup>	s <sup>-1</sup> )	Q <sub>dsdamp</sub>	$(m^3 s^{-1})$	Q <sub>dmax</sub> /	'Q <sub>dmin</sub>
Year	2010	2011	2010	2011	2010	2011
June	1.66	0.40	0.56	0.20	1.56	1.20
July	5.00	-	1.62	-	2.62	-
Aug	3.03	4.91	0.62	0.83	1.62	1.83
Sep	1.06	3.19	0.32	0.99	1.32	1.99
Phase 1	1.66	0.40	0.56	0.20	1.56	1.20
Phase 2	5.00	4.99	1.62	0.77	2.62	1.77
Phase 3 (2010)	2.67 (3.86)		0.60		1.60	
Phase 3a (2011)		6.03 (4.09)		1.12		2.12
Phase 3b (2011)		3.05 (3.00)		0.89		1.89
Mean	3.41 (3.60)	3.62 (3.39)	1.01	0.70	2.01	1.70

## 4.2.2. The amplitude of diurnal hydrographs

The diurnal runoff amplitude, described by Q<sub>dmax</sub>/Q<sub>dmin</sub>, can be compared to other clean and debris-covered glaciers (Tables 8 and 9). The range for Miage Glacier (from 1.2 in June/phase 1, to 2.12 in phase 3a) is similar to Dokriani Glacier, slightly higher than Gangotri Glacier and always higher than Dunagiri Glacier. The aforementioned glaciers are all debris-covered and their values (including those of Miage Glacier) are all lower than those given for the clean Qiyi and Parlung No. 4. Glaciers. Although the glacier area may alter the discharge ratio, the glacier area of Miage Glacier and Parlung No. 4 are very similar (Table 10). Values of Q<sub>dsdamp</sub> found by averaging values for peaked hydrographs of different magnitudes were also given by Swift et al. (2005) for Haut Glacier d'Arolla. The Swift et al. (2005) values ranged from 0.83–2.40 (equating to  $Q_{dmax}/Q_{dmin}$  of 1.83–3.40), which appears relatively high compared to values for both Miage Glacier and the other debris-covered glaciers. Similarly found values of Q<sub>dsdamp</sub> given by Hannah et al. (1999) for Taillon Glacier ranged from 0.434 to 2.649

(equating to  $Q_{dmax}/Q_{dmin}$  of 1.434–3.649), again giving high values compared to the debris-covered glaciers. This suggests that the diurnal runoff amplitude of debris-covered glaciers is less than clean glaciers.

#### 4.2.3. Lag time analysis

Lag times between air temperature and peak discharge determined by analysis of daily hydrographs are shown in Table 11. The mean and minimum lag times of the debris-covered glaciers are greater than that of the clean glaciers. The maximum lag time was greatest for Parlung No. 4 Glacier, but this value could have been influenced by colder weather in October. Lag times were also derived by conducting lag correlations between discharge and air temperature in 2010. All correlations were statistically significant (p < 0.05), with correlations generally stronger with UWS than LWS. The lag time for the whole discharge time series was between 9 and 10 h (UWS relationship). Table 12 compares statistically derived lag times for a range of glaciers. The two debris-covered glaciers (Miage Glacier and Keqikaer Glacier) have longer minimum lag times than all of the clean glaciers. Baounet Glacier and Haut Glacier d'Arolla are both temperate alpine glaciers, similar to Miage Glacier, and both their maximum and minimum lag times are much lower than the values for Miage Glacier (by 14-15 h and 4-7 h. respectively). Despite the small dataset there is therefore evidence that there is a greater lag time between peak air temperature and peak runoff for the debris-covered glaciers compared to the clean gla-

Lag times from peak air temperature to minimum conductivity (3.47) and (2.75) h in 2010 and 2011 respectively) were shorter than between peak air temperature and peak runoff (6.34) and (2.72) h in 2010 and 2011 respectively). When both diurnally classified proglacial discharge and diurnally classified conductivity data were available (only 8 days in total, see Fig. 6) there was often a delay between the time of lowest conductivity and highest discharge, resulting in discharge and conductivity rising in phase for a few hours. The mean time between minimum conductivity and maximum discharge in 2010 (2011) was 2.5 (4.25) h.

Fig. 6 shows the temporal change in the relationship between conductivity and discharge on individual days. The clearest cyclic

Table 9  $Q_{dmax}/Q_{dmin}$  values for selected glaciers. Debris-covered glaciers are bold with background data of the comparison glaciers in Table 10.

Glacier	Year	Reference	$Q_{dmax}/Q$	dmin					
			May	June	July	Aug	Sep	Oct	Mean
Miage	2010	This study.	_	1.56	2.62	1.62	1.32	-	2.01
Miage	2011	This study.	_	1.2	_	1.83	1.99	_	1.7
Dokriani	1996-1998	Singh et al. (2004, cited in Li et al., 2016)	-	1.3	1.5	2.3	1.7	-	1.7
Dunagiri	1987-1989	(Srivastava et al., 2014)	_	_	1.1	1.1	1.1	_	1.1
Gangotri	2000-2003	(Singh et al., 2006)	1.1	1.2	1.4	1.3	1.2	1.1	1.2
Parlung No. 4	2008/2010-2012	(Li et al., 2016)	2.3	2.2	2.2	2.2	2.4	6.3	2.9
Qiyi	2006	Song et al. (2008, cited in Li et al., 2016)	_	_	_	9	9	_	9

Table 10
Background information on glaciers used for comparison in Tables 1, 6, 7, 9, 11 and 12. Glaciers with a substantial debris cover (which covers the full width of the ablation zone) are bold. \*The original sources used for comparison were either not available or did not provide the necessary information so alternative sources have been used. The glacier areas and elevations may therefore have changed since the hydrological study referred to in other tables.

Glacier	Country or region	Debris-covered?	Basin area (km²)	Glacier area (km²)	Elevation range (m a.s.l.)	Source
Austre Brøggerbreen	Svalbard	n		11	50-720	Tranter et al. (1996)
Batura Glacier	Pakistan	y		365	$> 7000 \text{ to } \sim 3000$	Hodson et al. (2002)
Baounet Glacier	France	n	6	2.5	2800-3300	Jobard and Dzikowski (2006)
Bench Glacier	Alaska	n		8.09	932-1791	RGI Consortium (2017)*
Chamberlain	Alaska	n	3.78	2.41		Rainwater and Guy (1961)
Dokriani Glacier	India	y (75% of ablation area)	9.58	5.76	~4000 to 6632	Hasnain and Thayyen (1999a)
Dunagiri Glacier	India	у	17.9	2.56	4200-5100	Srivastava et al. (2014)
Engabreen	Norway	n		36	111-1544	Elvehøy and Jackson (2018)*
Fjallsjökull	Iceland	n		63.5	24-2030	RGI Consortium (2017)*
Gangotri Glacier	India	y	556	286	3800-7000	Singh et al. (2006)
Grimsvötn (River Skeiðará)	Iceland	n		1561.2	111–1957	RGI Consortium (2017)* (details for Skeiðarárjökull)
Haut Glacier d'Arolla	Switzerland	n	12	6.3	2560-3838	Brown et al. (1996); Tranter et al. (1996)
Kegikaer Glacier	China	y		83.6	3020-6435	Xie et al. (2006)
Khumbu Glacier	Nepal	y (47% of glacier)		41	4900-8848	Irvine-Fynn et al. (2017)
Koxkar Glacier	China	y (83% of ablation area)		83.56	3060-6342	Han et al. (2010)
Miage Glacier	Italy	y (42% of glacier)	22.7	10.5	1740-4640	This Study
Nigardsbreen	Norway	n		46.6	330-1952	Kjøllmoen (2018)* (area from 2013)
Parlung No. 4 Glacier	China	n	24.3	11.71	4650-5964	Li et al. (2016)
Qiyi	China	n	6.8	2.9		Song et al. (2008, cited in Li et al., 2016)
Rongbuk Glacier	China	y	298	203	5070-8848	Liu et al. (2010)
Scott Turnerbreen	Svalbard	n	12.8	3.3	230-680	Hodgkins et al. (1998)
Taillon Glacier	France	n	0.37	0.22	2526-3022	Hannah et al. (1999)
Tsanfleuron	Switzerland	n		3.81		VAW/ETHZ and EKK/SCNAT (2018)* (area for 1973)

hysteresis is anticlockwise, resulting in conductivity increasing with discharge in the evening (or sometimes high but variable discharge and increasing conductivity). For conductivity to increase with discharge, the solute load must be increasing at a greater rate than discharge (Fountain, 1992).

4.2.4. Runoff fluctuations and their relation to meteorological variations 4.2.4.1. Phase 1: Early season runoff. During the entire phase 1 period in 2010 and prior to 13/06/11 in the 2011 record discharges generally remained low (average 3.2 m³ s⁻¹ and 2.1 m³ s⁻¹ for 2010 and 2011 respectively), and although diurnal hydrographs dominated they had a small amplitude. In 2010 variations over a few days had a larger magnitude than diurnal fluctuations. In 2011 discharges rose on 13/06/11 but did not rise substantially until 15/06/11, followed by the large flood event on 18/06/11 caused by very heavy rainfall (Fig. 5). In both years air temperatures during Phase 1 periods were lower than average.

4.2.4.2. Phase 2: Diurnal runoff. During phase 2 in both years the

discharge increased, alongside an increase in the percentage of diurnal hydrographs and the mean amplitude of diurnal hydrographs (to  $5.0\,\mathrm{m}^3\,\mathrm{s}^{-1}$  in both years) (Fig. 7a and b). These changes were driven by consistently fine weather with increased air temperatures and only occasional rainfall (mean LWS air temperature in Phase 2 was greater than average by 4.4 °C in 2010 and 2.3 °C in 2011). The flows in 2010 did not increase until around three days after the rise in air temperature, although within the phase 2 period the discharge was more responsive to meteorological changes (e.g. the shape of the hydrograph peak relates to the hourly modelled ablation on 11/07/10, see Fig. 7a).

4.2.4.3. Phase 3: Hydrographs with reverse asymmetry. The phase 3 runoff in 2010 begins around 16/07/10 (Fig. 7c). The overall discharge magnitude was similar to average  $(5.9\,\mathrm{m}^3\,\mathrm{s}^{-1})$ , average  $5.3\,\mathrm{m}^3\,\mathrm{s}^{-1}$ ), with the discharge amplitude larger than average  $(2.7\,\mathrm{m}^3\,\mathrm{s}^{-1})$  for diurnal hydrographs only, with  $3.4\,\mathrm{m}^3\,\mathrm{s}^{-1}$  the average). The lag time to peak discharge for diurnal hydrographs

Table 11
Maximum and minimum lag times for selected glaciers derived using analysis of diurnal hydrographs. Debris-covered glaciers are bold with background data of the comparison glaciers in Table 10. \*Cited in Li et al. (2016). \*Lag time analysis was only conducted for June 2005 so mean, maximum and minimum lag times are from within June.

Glacier	Year	Reference	$\begin{array}{l} \text{Mean lag time } T_{max} \\ \text{to } Q_{dmax} \text{ (h)} \end{array}$	Max. lag time $T_{max}$ to $Q_{dmax}\left(h\right)$	Phase of max. lag time	Min. lag time $T_{max}$ to $Q_{dmax}$ (h)	Phase of min. lag time
Miage Glacier	2010	This study.	6.34	8.43	June/Phase 1	3.63	Aug
Miage Glacier	2011	This study.	7.92	9.67	Sep/Phase 3b	5.5	June/Phase 1
Dokriani Glacier	1996–1998	Singh et al. (2004)*	4.5	6.0	June	3.3	Sep
Gangotri Glacier	2000–2003	Singh et al. (2006)	5.3	6.30-7.30	Oct	4	July
Koxkar Glacier	2005-2008	Han et al. (2010)	7.0	8.80	Oct	5.2	Aug
Rongbuk Glacier	June 2005**	Liu et al. (2010)	10.78	14	9th June	8	10th June
Parlung no. 4 Glacier	2008, 2010-2012	Li et al. (2016)	3.8	14.8	Oct	0.8	July
Qiyi	2006	Song et al. (2008)*	2	-	-	-	

Maximum and minimum lag times for selected glaciers derived using statistical analysis of hydrological and meteorological time series. Debris-covered glaciers are bold with background data of the comparison glaciers

Glacier	Year	Year Reference	Max. lag time T to Q timeseries (h) Phase of max. lag time	Phase of max. lag time	Min. lag time T to Q timeseries (hours) Phase of min. lag time	Phase of min. lag time
Miage Glacier	2010	This study.	24	25/07/10-23/08/10/Aug/Phase 3	7	15/07/10-24/07/10 /July/Phase 2
Kegikaer Glacier	2004	Xie et al. (2006)	25	May	19	June
Austre Brøggerbreen	1991	Hodson et al. (1998)	29	04/07/91-13/07/91/Phase 2	4	06/08/91-16/08/91/Phase 5
Austre Brøggerbreen	1992	Hodson et al. (1998)	10	23/06/92-04/07/92/Phase 1	5	16/07/92-02/08/92/Phases 4,5,6
Baounet Glacier*	2003	Jobard and Dzikowski (2006)	10	01/06/03 Phase 1	1	end of July-early Aug 2003 Phase 3
Baounet Glacier*	2004	Jobard and Dzikowski (2006)	6	12/06/04 Phase 1	2	mid Aug 2004 Phase 3
Haut Glacier d'Arolla	1998	Swift et al. (2005)	10	02/06/98-07/06/98/June/Phase 2	0	05/08/98-09/08/98/Aug/Phase 8
Haut Glacier d'Arolla	1999	Swift et al. (2005)	6	17/06/99–22/06/99/June/Phase 1	3	20/07/99-23/07/99/ July/Phase 5
Scott Turnerbreen	1992	Hodgkins (2001)	24	11/08/92-17/08/92Interval 4	1	21/07/92-27/07/92 Interval 1

(4.46 h) was shorter than average and for phase 2, but including all hydrographs was much longer (8.17 h). The mean air temperature at LWS was within 1  $^{\circ}$ C of the average.

The phase 3 runoff in 2011 occurred both before (phase 3a) and after (phase 3b, Fig. 7d), the phase 2 runoff period. The mean discharge of the phase 3a and 3b runoff for diurnal and all hydrographs was higher (3a) and lower (3b) than average but by not much more than  $1\,\mathrm{m}^3\,\mathrm{s}^{-1}$ . The discharge amplitude was higher than average (using only diurnal hydrographs) in phase 3a but lower than average in phase 3b. Using both diurnal and all hydrographs the lag time of phase 3a and 3b (around 9–10 h) was higher than phase 2 and the seasonal average. The mean air temperature at LWS of the phase 3a and 3b periods was also within 1 °C of the average.

Phase 3 runoff tends to show reverse asymmetry in 2010 and 2011 (see Fig. 7c). The increasingly warm air temperatures between 30/07/10 and 01/08/10 lead to a gradually rising hydrograph, then after the precipitation inputs passed and the weather cooled the runoff decreases. Warmer temperatures lead to gradually increasing discharges, but a decrease in air temperature leads to a sharp runoff decrease. In 2011, Fig. 7d shows the rise in night-time air temperature on 28/08/10 results in a rising hydrograph on 29/08/10. In the following days the discharge does become diurnal but the minimum flow has a clearer trough than the maximum flow has a peak.

#### 5. Discussion

#### 5.1. The influence of debris on supraglacial and proglacial water chemistry

The proglacial concentration of sulphate and bicarbonate ions was high for Miage Glacier compared with clean glaciers (Section 4.1.2). The concentration of sulphate ions can be influenced by a) snowpack sulphate derived from deposition of mainly the sea salt aerosol (Tranter et al., 1996); b) contact of water with sediment at the bed resulting in sulphide oxidation (Tranter et al., 1993b); or c) by the influence of groundwater. The influence of supraglacial debris on sulphate concentrations is not known from previous studies.

Sulphate values in snow were low, with a mean concentration of 76  $\mu$ eq  $l^{-1}$ , compared to 215  $\mu$ eq  $l^{-1}$  on average in the proglacial stream in 2011 (Table 4). Most supraglacial streams had very low sulphate concentrations, demonstrating that they mainly contained ice melt. There was not a clear distinction in sulphate concentrations between the upper and lower glacier supraglacial stream samples, suggesting that supraglacial debris does not increase the sulphate concentrations. However, a higher sulphate concentration was measured in the small pond SP5, suggesting that the longer sediment contact was conducive to coupled sulphide oxidation and carbonate dissolution reactions which produce bicarbonate and sulphate ions (Brown, 2002). However, the small ponds were not prevalent enough to be a significant ion source. Similar conditions likely occur in the larger lakes (C4 and C8), although meltwater from their large ice cliffs would lower ion concentrations. The lake ion concentrations were lower than proglacial values, with lakes only providing ions to proglacial runoff upon drainage. Therefore, apart from the early season snowmelt, most proglacial sulphate is not of a supraglacial origin.

The lower glacier streams draining debris-covered catchments had higher bicarbonate concentrations than the upper streams draining mainly clean ice. Contact of meltwater with debris (possibly while within the debris matrix) therefore results in enough water-rock interaction to enrich the bicarbonate concentration through acid hydrolysis. The greater sediment availability may also lead to a higher SSC, allowing in-stream acid hydrolysis and the production of bicarbonate ions, even in ice-walled streams (Tranter et al., 1993a). Ice cliff meltwater may reduce bicarbonate concentrations (e.g. at S5 which is a stream bounded by ice cliffs in a thickly debris-covered catchment). Concentrations were higher in SP5, again suggesting that small ponds may be an ion source. Although debris enriched the supraglacial

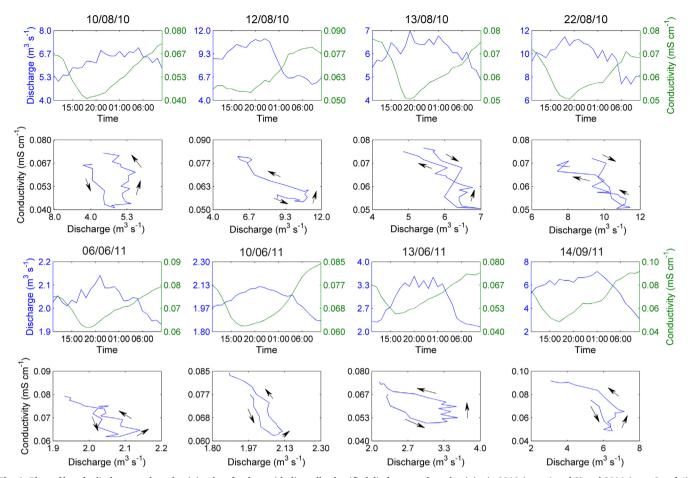


Fig. 6. Plots of hourly discharge and conductivity data for days with diurnally classified discharge and conductivity in 2010 (rows 1 and 2) and 2011 (rows 3 and 4). Note the day starts at 11:00. The temporal relationship between hourly discharge and conductivity is given below the time series plots.

bicarbonate concentration, the average supraglacial stream concentration (286  $\mu eq\,l^{-1}$ ) was still considerably lower than the average gauging station concentration (783  $\mu eq\,l^{-1}$  in 2010 and 603  $\mu eq\,l^{-1}$  in 2011). Given the difference in supraglacial and proglacial concentrations of both bicarbonate and sulphate concentrations there must therefore be a significant proportion of the proglacial concentration produced instream, from groundwater, or at the bed.

It is most likely that the proglacial ions are produced from reactions within a distributed hydrological system which underlies the lower glacier, since there is already evidence of its existence from dye tracing (Fyffe et al., 2019). The ions could also be produced if water spent time in the sedimentary raised bed underlying the lower glacier (Pavan et al., 1999, cited in Deline, 2002). Both of these environments would contain freshly comminuted rock flour and be conducive to long enough residence times to elevate sulphate concentrations (Tranter et al., 1993b; Tranter and Raiswell, 1991). Both bicarbonate and sulphate concentrations remained high throughout the melt season, suggesting that a significant proportion of meltwater passed through the distributed system even in mid-summer (Section 4.1.2).

Groundwater was sampled from a spring next to the valley river, the Dora d'Veny (Fig. 1) and concentrations of sulphate and bicarbonate were very high (2959  $\mu eq\,l^{-1}$  and 1361  $\mu eq\,l^{-1}$ , respectively), with a Cratio of 0.32. However, the C-ratio was much lower than the lowest value measured at the proglacial stream (0.63). Even if the bicarbonate concentration is artificially increased by 286  $\mu eq\,l^{-1}$  to account for the average supraglacial input of bicarbonate ions, the C-ratio is still 0.36. The sulphate concentrations were twice the highest values found from boreholes beneath Haut Glacier d'Arolla, with such high values suggesting long residence times in anoxic conditions, where sulphide

oxidation may be catalysed by microbial activity (Brown, 2002). It is therefore less likely the proglacial ions are derived from groundwater (at least that of a composition similar to the spring measured).

Furthermore, although bicarbonate ions can be produced within post-mixing chemical reactions (Tranter et al., 1993a), the bicarbonate and sulphate ion concentrations tend to vary together (although the only significant relationship was in July/August 2010). They also tend to be high during low flow periods, suggesting that a portion of the bicarbonate concentration is produced subglacially.

# 5.2. The impact of debris on the proglacial hydrograph

5.2.1. The influence of supraglacial debris on proglacial discharge amplitude and lag time

Comparison of the lag time and diurnal discharge amplitude of debris-free and debris-covered glaciers (Sections 4.2.2 and 4.2.3) suggests that glaciers with an extensive supraglacial debris cover have a more subdued proglacial hydrograph with a smaller amplitude and longer lag time between peak air temperature and discharge compared to clean glaciers. This corroborates the findings of Li et al. (2016) in the Tibetan Plateau and corresponds with the lower overall percentage of diurnal hydrographs found when comparing Miage Glacier and Haut Glacier d'Arolla (Section 4.2.1). However, comparable data were only available for 6 debris-covered and 6 clean glaciers for lag time analysis and 4 debris-covered and 4 clean glaciers for diurnal discharge amplitude parameters.

The lower discharge amplitude and fewer diurnal hydrographs under average weather conditions could be due to a more subdued input (or recharge) hydrograph (Covington et al., 2012). An attenuated

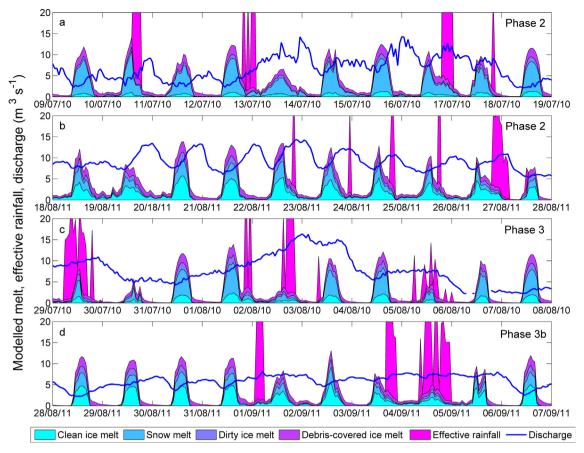


Fig. 7. Close up of proglacial discharge and modelled melt and effective rainfall (precipitation minus evaporation) for each of the different surface types (shown as an area plot) for a 10 day period of a) Phase 2 in 2010, b) Phase 2 in 2011, c) Phase 3 in 2010 and d) Phase 3b in 2011. Note that the y-axis has been constrained to  $20 \text{ m}^3 \text{ s}^{-1}$  to allow discharge fluctuations to be seen more clearly.

input hydrograph from debris-covered areas will be caused by the time taken for the temperature signal to be conducted down through the debris to the ice-debris interface, resulting in melt continuing into the evening (Fyffe et al., 2014 (especially Fig. 7); Nicholson and Benn, 2006; Reid and Brock, 2010). This increases the lag time between peak air temperature and peak melt, which increases with increasing debris thickness (although the runoff delay will be less than that of the average debris thickness since more melt is produced beneath thinner debris) (Fyffe et al., 2014). Thus, supraglacial debris will reduce the amplitude of the input hydrograph, increasing the baseflow and decreasing the diurnal discharge amplitude.

Furthermore, melt must flow within the debris and/or at the ice/debris interface prior to reaching a supraglacial stream. Flow velocities in this system are unknown but it may provide a further delay to water transfer. Flow within streams within the debris-covered region of Miage Glacier is known to be relatively slow (Fyffe et al., 2019) and Miles et al. (2019) reported slow flow velocities between an interconnected pond system on Khumbu Glacier. The relatively inefficient supraglacial transfer of meltwater within the debris-covered region would increase the time between melt production and input into the englacial/subglacial system.

The input hydrograph could also be attenuated by an inefficient englacial or subglacial system, or the existence of a subglacial aquifer (Flowers, 2008; Flowers and Clarke, 2002). An inefficient englacial/subglacial system was found beneath the debris-covered area (Fyffe et al., 2019), with the high proglacial bicarbonate and sulphate concentrations further confirming its existence (Section 5.1). The volume of storage is key to determining whether or not recharge hydrographs are altered by the glacial drainage system (Covington et al., 2012). Flowers (2008) identified that a subglacial aquifer can attenuate

hydrographs, and while Covington et al. (2012) demonstrated that although conduits and a linked cavity system did not normally modify the recharge hydrograph, they could if linked to sufficiently large storage such as crevasses or lakes. The inefficient englacial/subglacial system could therefore alter the recharge hydrograph if the degree of storage in the system was sufficient.

The thickness and hydraulic conductivity of the bed materials could also have a significant influence on the discharge amplitude and lag time (Flowers, 2008; Flowers and Clarke, 2002). The lower Miage Glacier sits on a sediment bed, around 50 m thick (Pavan et al. (1999) cited in Deline (2002)). Flowers (2008) showed that a 'soft' sediment bed would lower the amplitude of diurnal fluctuations, increasing the lag time from peak melt to peak discharge and inhibiting conduit formation, compared to a 'hard' bedded glacier. Flowers (2008) also showed that a subglacial aquifer beneath the sediment bed would further reduce the discharge amplitude, with the effect increasing with aquifer thickness and hydraulic conductivity. This is driven by higher water pressures resulting in flow being transferred from subglacial channels to be stored in the surrounding sediments, with water flowing back to the channels as pressures drop (Hubbard et al., 1995). Debriscovered glaciers often rest on sedimentary beds, since a substantial supraglacial debris cover indicates a high rate of debris supply and high englacial debris concentrations (Maisch et al., 1999). Low sub-debris melt rates and a long, low angled tongue would likely also reduce the debris transport capability of meltwater streams, with debris accumulating at the bed as a result. The impacts of a raised-bed sediment aquifer on the proglacial hydrograph may therefore also apply to other debris-covered glaciers.

Whether the lag time from peak air temperature to peak discharge is increased due to the debris-cover for a particular glacier will depend

upon the debris thickness distribution and its proportional cover over the glacier, since these factors will influence the proportion of sub-debris melt and the average delay due to conduction of heat to the ice-debris interface. If rapidly-routed clean ice melt is a large enough proportion of runoff, then the lag time may not be influenced by the debris cover since the clean ice component will provide the discharge peak (c.f. Miage Glacier in Section 5.3). Furthermore, the bed composition (bedrock or sedimentary) and the thickness and hydraulic conductivity of any raised bed could influence the amplitude and timing of the proglacial hydrograph (Flowers, 2008; Flowers and Clarke, 2002).

# 5.2.2. Runoff fluctuations and their relationship to meteorological conditions

Phase 1 proglacial hydrographs were subdued in both years even though most of the lower glacier was snow free (from 2290 m a.s.l. in 2010 and 2400 m a.s.l. in 2011) and dye tracing suggested the main subglacial channel system was open and efficient (from at least 09/06/ 10 and 10/06/11 (Fyffe et al., 2019)). On the upper glacier the snow cover would have reduced the supraglacial hydrograph amplitude and increased the meltwater travel time (Campbell et al., 2006; Willis et al., 2002). Meanwhile, on the debris-covered lower glacier meltwater inputs would be small and attenuated due to the conduction of heat through the debris (Fyffe et al., 2014), and smaller supraglacial streams (Fyffe et al., 2019). Meltwater from the debris-covered region will therefore have a subdued hydrograph, and if the stream passes through the distributed system underlying the lower glacier (see Section 5.1 and Fyffe et al. (2019)) it may be delayed englacially and subglacially too. This means that a proglacial hydrograph with a smaller amplitude and longer lag from peak air temperature to peak discharge can result even if a large proportion of the lower glacier is snow-free, if the snow-free region is instead debris-covered.

Phase 2 periods were found during warmer than average weather conditions. By mid-summer large areas of clean and dirty ice were snow-free, with the snowline up to approximately the base of the Bionassay Glacier by 31/07/10 and slightly less coverage in 2011. The removal of snowcover from the debris-free area would have increased the magnitude and amplitude of meltwater inputs to the upper glacier moulins (e.g. S12 and S14) (Willis et al., 2002). The main subglacial drainage system remained efficient in both years, and although the drainage system beneath the debris-covered area increased in efficiency, it was still less efficient than that emanating from higher on the glacier (Fyffe et al., 2019). Continued drainage through the distributed system is also confirmed by the proglacial sulphate and bicarbonate concentrations which remained high throughout the summer (Section 5.1). Removal of snow from the clean and dirty ice area is therefore required to increase the amplitude of the upglacier supraglacial hydrograph and therefore lead to a clear proglacial diurnal signal. Phase 2 periods therefore require consistently warm and dry weather as well as the loss of snow cover from the upglacier clean ice.

Phase 3 runoff occurred during average weather conditions, and resulted in close to average proglacial discharges and diurnal amplitudes and a hydrograph with reverse asymmetry where flow increases gradually but falls quickly. Smeared or multiple hydrograph peaks could be caused by multiple input points with different lag times (Covington et al., 2012). Since variations in debris thickness cause variations in the time of the supraglacial hydrograph peak (Fyffe et al., 2014), and the debris morphology leads to many, smaller input points (Fyffe et al., 2019; Miles et al., 2017), this may result in a flattened peak, especially if cooler weather means a larger proportion of runoff is from the debris-covered area.

The reverse asymmetry of the phase 3 runoff hydrograph is somewhat unusual. A possible explanation involves percolation of meltwater into the sediment aquifer beneath the lower glacier. If the input of water into the subglacial system is less than the maximum potential rate of infiltration into the aquifer (which would depend upon the aquifer hydraulic conductivity), the aquifer could essentially capture the

drainage (Flowers and Clarke, 2002). Net loss to the aquifer would be especially noticeable when most meltwater is introduced from further upglacier, and inputs from the lower glacier are limited.

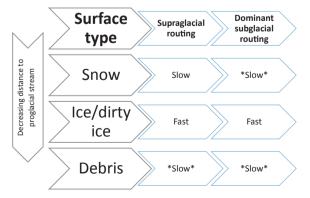
A second possible mechanism involves the overdeepening which occurs just upstream of the bend in the main tongue (Pavan et al. (1999) cited in Deline (2002)). Water pressures close to overburden are required for subglacial water to flow up the adverse bed slope and exit the overdeepening. The rise in the pressure melting point as the water flows up the adverse bed slope can result in supercooling, forming frazil ice which can reduce the hydraulic conductivity and subsequently flow rates (Alley et al., 1998). Whether or not supercooling occurs is dependent upon the riegel slope angle (Alley et al., 1998), how close the water temperature is to the melting temperature, and by water pressure variations (Dow et al., 2014). If the riegel is steep enough, water flow could be restricted, especially when water pressures are lower (Dow et al., 2014), potentially increasing hydrograph recession.

#### 5.3. The flow components of Miage Glacier runoff

By combining the results of the hydrochemistry (Section 5.1) and proglacial discharge analysis (Section 5.2) the main flow components of Miage Glacier runoff can be identified (Fig. 8). Supraglacial hydrochemistry demonstrated that upper glacier clean ice melt does not tend to acquire solutes, with dye tracing revealing that upper glacier meltwater is drained efficiently (Fyffe et al., 2019). The fast transfer of this dilute clean ice melt component suggests its peak at the proglacial stream is indicated by the conductivity minimum which arrives around 3 h after peak air temperature (Section 4.2.4).

Meanwhile, on the debris-covered lower glacier, the small supraglacial streams flow more slowly (Fyffe et al., 2019), with meltwater acquiring bicarbonate ions due to its contact with sediment (Section 4.1.1). Dye tracing indicated some of the lower glacier subglacial drainage pathways were inefficient (Fyffe et al., 2019), with the high proglacial sulphate and bicarbonate concentrations confirming this (Section 5.1). Sub-debris melt will therefore likely have a greater solute concentration and take longer to arrive at the snout. This explains the evening rise of conductivity with discharge shown in Fig. 6 - the delayed 'debris' component becomes a larger proportion of runoff in the evening and likely peaks when the discharge peaks 6-8 h after peak air temperature. However since modelling suggested melt from the debriscovered region provided only 27-30% of total melt (Fyffe et al., 2014), melt from the tributary glaciers very high upglacier may constitute the remainder of the discharge peak. The longer travel times of the 'debris' component explains the observed long lag times between air temperature and peak discharge for debris-covered glaciers (Section 4.2.3).

Presuming that the 'clean ice' component peaks at the time of minimum conductivity, while the 'debris' component peaks at the time of maximum discharge, then the overall delay due to the debris cover



**Fig. 8.** Conceptual diagram of runoff components, and their relative travel time class for a debris-covered glacier. The stars indicate where solute could be principally acquired by the meltwater.

averaged over the glacier can be calculated from the difference between these times (2.5 h in 2010 and 4.25 h in 2011, based on 8 days of conductivity data). Several mechanisms could account for this delay (Section 5.2.1), including attenuation of the melt signal due to conduction of heat through the debris, smaller supraglacial streams, a less efficient englacial/subglacial hydrological system and a sediment bed. The delay due to heat conduction averaged over the glacier is around 1–2 h (Fyffe et al., 2014), with the less efficient englacial/subglacial system resulting in a delay of around 1.9 h (Fyffe et al., 2019) (Supplementary Information B outlines how these delays were derived). Since these delays are roughly equal in duration both should be included in debris-covered glacier runoff models. Supraglacial dye tracing is needed to determine the supraglacial efficiency of debris-covered glaciers, while further runoff modelling would be needed to fully understand the influence of a sediment aquifer.

#### 6. Conclusions

Supraglacial debris influences glacier hydrochemistry in two main ways. Firstly, greater supraglacial water-rock contact on the lower glacier increases the SSC and bicarbonate concentration of supraglacial lakes and streams. The supraglacial debris only influences sulphate concentrations where residence times are long (e.g. in small ponds). Secondly, since the debris cover inhibits the formation of a channelised network beneath thickly debris-covered areas (Fyffe et al., 2019), it indirectly increases the proglacial bicarbonate and sulphate concentrations by increasing the proportion of water which is routed through an inefficient subglacial system.

The Miage Glacier runoff hydrograph is a product of the melt signal and its alteration by the hydrological system. Since the supraglacial debris attenuates the melt signal (Fyffe et al., 2014) and results in an inefficient subglacial system fed by smaller input streams (Fyffe et al., 2019) this means the flow component composed of sub-debris melt has a longer lag time than the flow component from the clean and dirty ice which is routed efficiently from the mid-glacier. This increases the discharge baseflow and increases the lag time between peak air temperature and peak runoff, although the raised sediment bed may also play a role in attenuating the proglacial hydrograph. Discharge and conductivity commonly showed anti-clockwise hysteresis with conductivity and discharge rising in phase for a few hours. This suggests that the dilute melt component likely peaks before the more ion rich 'debris' and 'tributary' components. The delaying effect of the debris also means that both a snow-free upper glacier and particularly warm weather are required for diurnal hydrographs to dominate. Under average weather conditions rising or falling hydrographs dominate and the hydrograph tends to rise more slowly than it falls, resulting in reverse asymmetry in the hydrograph.

These conclusions show that the hydrochemistry and proglacial runoff signal of debris-covered glaciers differs from clean glaciers. Given that data for comparison was from a relatively small sample of glaciers, work to increase this sample size and differentiate the influence of specific catchment and debris characteristics on the proglacial runoff signal would be welcomed. These findings will be relevant to other catchments with debris-covered glaciers, and those that are likely to become more debris-covered in the future (e.g. Bhambri et al., 2011; Bolch et al., 2008; Lambrecht et al., 2011). Since receding glaciers may become more debris-covered, the influence of supraglacial debris on the runoff hydrograph should be accounted for within future runoff models. The present study is a pioneering first step towards quantifying the influence of debris and glacier characteristics on supraglacial and proglacial hydrographs. It will inform future work involving the direct comparison of hydrographs from multiple glaciers using a consistent methodology and the construction of a physically-based hydrological model which takes into the account the influence of the debris on runoff. This would allow the prediction of runoff given changes to the debris cover.

#### **Declaration of Competing Interest**

None.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.jhydrol.2019.06.023.

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