

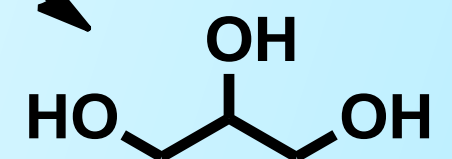
# NON CATALYTIC CONTINUOUS-FLOW TRANSESTERIFICATION OF ORGANIC CARBONATES WITH GLYCEROL ACETALS <sup>[1]</sup>



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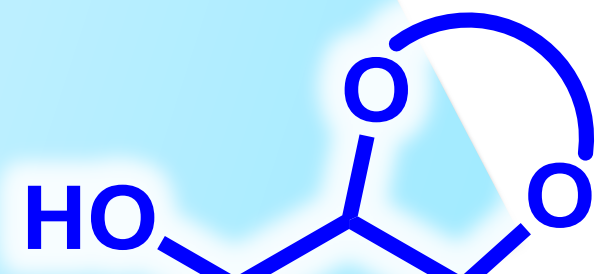
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From Biomass

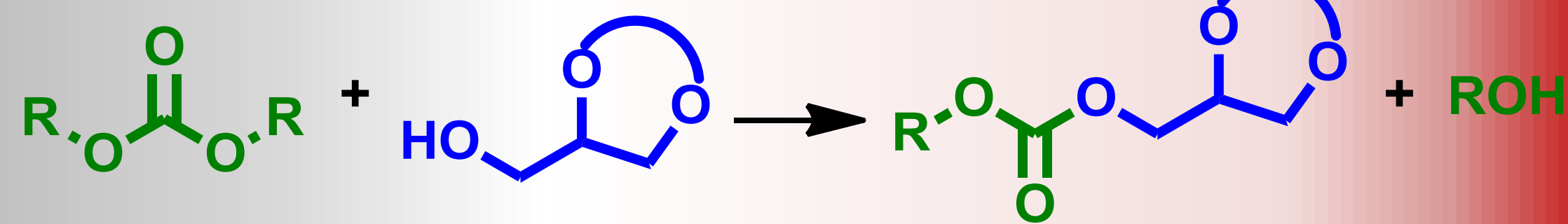


Acetalization

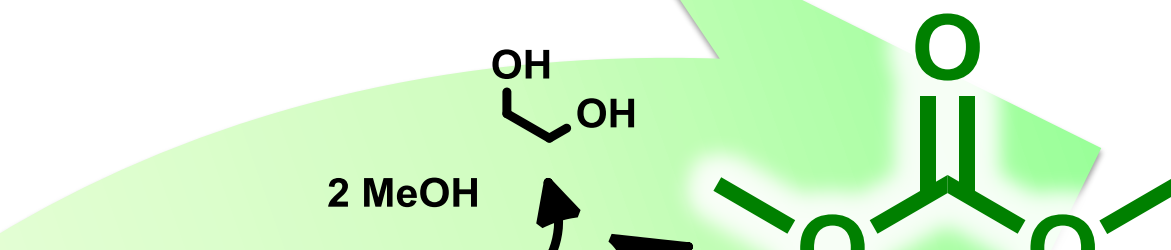
[2]



*Continuous flow transesterification*



- High Atom Economy
- Carbonates act as solvent/carrier
- ROH is the only byproduct



Cat1

Cat2

[3]

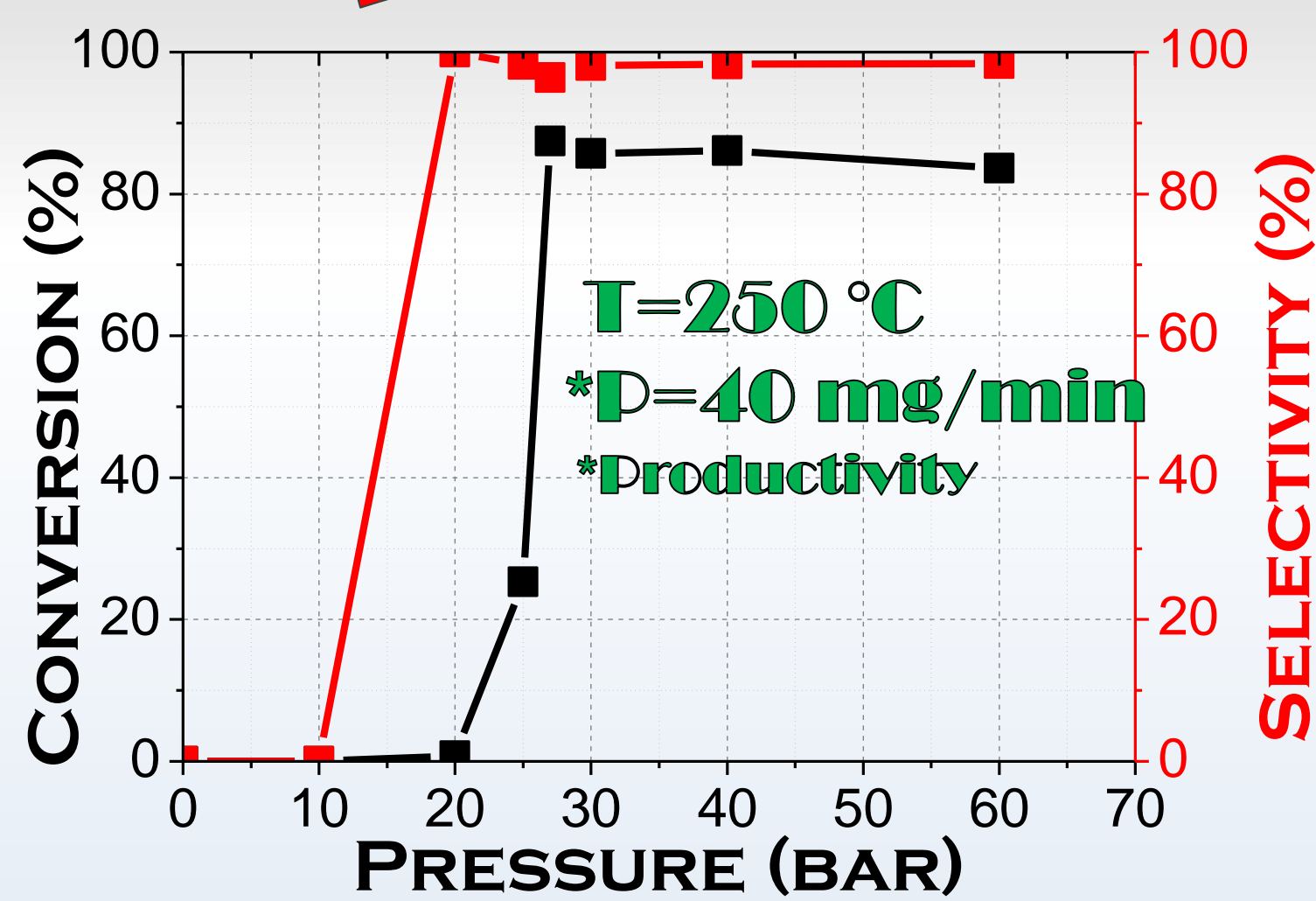


CO<sub>2</sub>

20 ÷ 70 bar



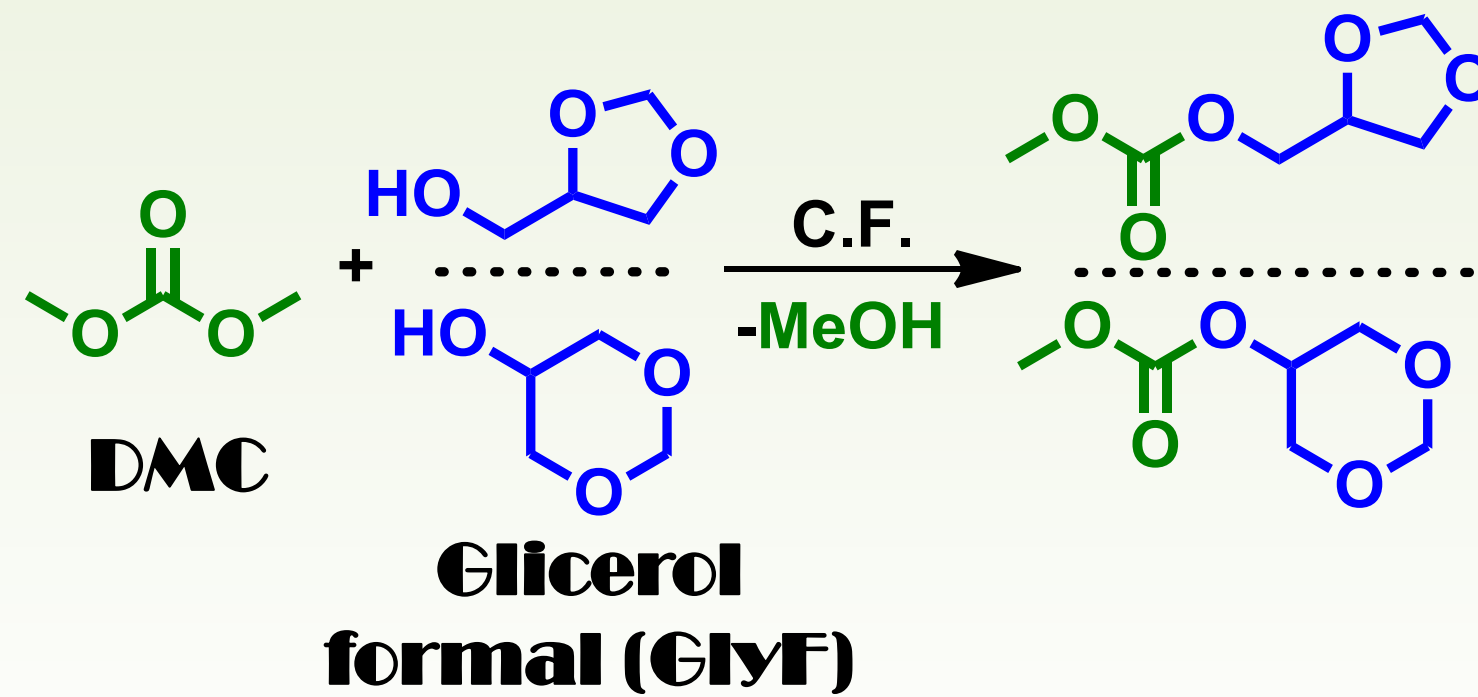
*Reaction profile*



The process plausibly occurs in supercritical phase

For the DMC: T<sub>c</sub>= 284 °C  
P<sub>c</sub>= 48 bar  
d<sub>c</sub>=4 g/mL

*Model reaction*



\*Conversion } ≥ 97%  
\*Selectivity }

\*With a single recycle of the mixture

*The Innovation*

[4]

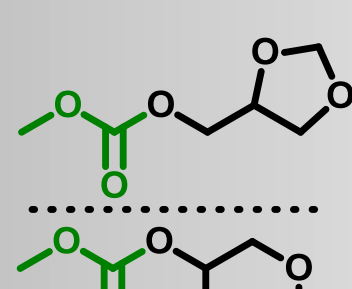


*Catalyst-Free reactions*

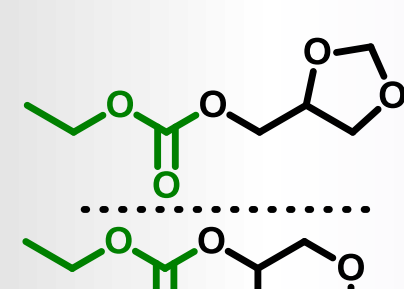
Reactions performed in an empty capillary reactor

*Products*

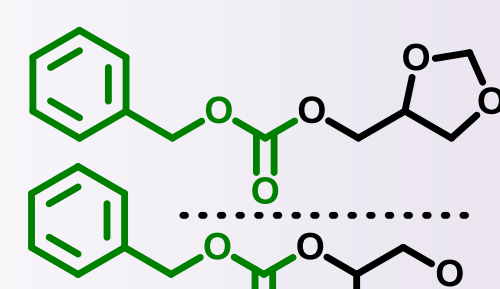
from different acetals and carbonates



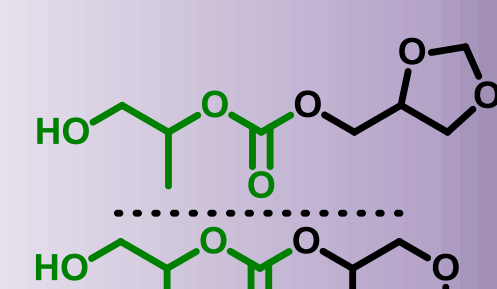
Conv = 86%  
Sele = 98%



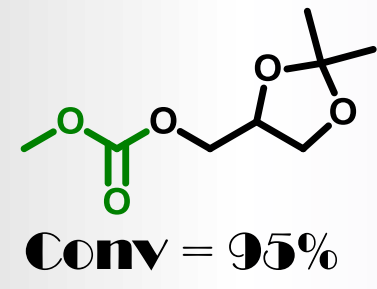
Conv = 81%  
Sele = 97%



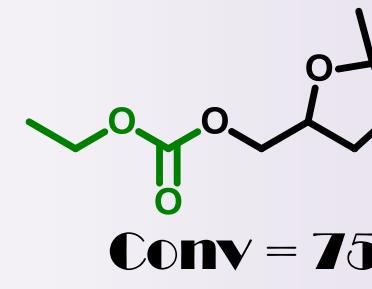
Conv = 91%  
Sele = 98%



Conv = 40%  
Sele = 98%



Conv = 95%  
Sele = 98%



Conv = 75%  
Sele = 98%

**CONCLUSIONS:**

- Continuous-flow transesterification
- Pure thermal reactions (Catalyst-Free)
- Plausibly, a supercritical phase is needed
- Synthetic extension to different acetals and carbonates

REFERENCES:

- [1] Selva, M.; Noe, M.; Guidi, S., *Green Chemistry*, 2014.  
 [2] Vanlaldinpuia, K.; Bez, G., *Tetrahedron Letters*, 2011, 52, 3759-3764.  
 [3] Bhanage, B.; Fujita, S.-i.; He, Y.; Yakushima, Y.; Shirai, M.; Torii, K.; Arai, M., *Catalysis Letters*, 2002, 83, 137-141.  
 [4] Pasiadis, S. A.; Barakos, N. K.; Papayannakos, N. G., *Ind. Eng. Chem. Res.*, 2009, 48, 4266-4273.