Polymer templated nickel cobaltate for energy storage

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ABSTRACT *In order to take advantage of the increasing sophistication of technology for harnessing renewable energy resources, serious attention must be paid to how to store and re-access this energy. Electrochemical storage, in the guise of batteries, supercapacitors and pseudocapacitors, has attracted much attention as a viable option for enhanced energy storage applications. But in order for these technologies to be implemented successfully we need to find materials that perform better and are relatively easy to synthesise. Bimetallic transition metal oxides are materials that are readily synthesised and may be multifunctional, i.e. have a role at the electrochemical atomic level as well as the device level. In order for these materials to work efficiently in new generation systems based on sodium and lithium they also need to be mesoporous. This can be achieved by trying to find synthetic techniques that produce specific, highly regulated nanostructures or by adding a 'templating' agent during the bulk synthesis step. We have investigated the simple hydrothermal preparation of a number of nickel cobaltate ($NiCo_2O_4$) materials using polymer templates, eggshell membrane (ESM) and poly methyl methacrylate (PMMA), as potential electrode materials for supercapacitors. The ESM was expected to act as a fibrous, random polymeric template while the PMMA should produce a much more ordered material. Electrochemical testing showed that the different templates have led to changes in material morphology and these have resulted in a difference in electrochemical properties. Templated materials had an increased specific capacitance than non-templated and the choice of template could influence the capacitance by as much as 30 %.*

Keywords: supercapacitor, bimetallic transition metal oxide, polymer template, eggshell membrane

1 INTRODUCTION

The demand for advanced electrochemical energy storage devices with increased power and energy densities is increasing due to sustainable energy and environmental issues. Electrical energy storage and conversion systems such as fuel cells, batteries and supercapacitors will play a significant role in the effective utilisation of clean energy sources (e.g. wind and solar) with intermittent energy output. Supercapacitors have attracted extensive attention for this role due to their ability to convert chemical energy to electrical energy with high efficiency and excellent cyclic stability [1,2]. In addition, supercapacitors have unique properties such as ultrafast charge-discharge behaviour, high power densities and very long-term stability compared to lithium-ion batteries [3-5]. Supercapacitors are typically classified into two main types depending on the charge storage mechanism utilised; electric double layer capacitors (EDLCs) and pseudocapacitors. In an EDLC, the electrical energy, in the form of free ions accumulated on the electrode surface, are stored by ion adsorption. In a pseudocapacitor that electrical energy is stored by fast surface redox reactions. With the acknowledgement that improving the performance of electrode materials is perhaps the best option for improving energy storage overall, a third type of 'hybrid' supercapacitor is emerging stores the charge by both redox reaction and electrostatic phenomena occurring at the electrode/ electrolyte interface. These 3rd wave electrodes promise high cell voltage, high specific capacitance, unmitigated cyclic stability, and improved energy density [6-8].

In developing these new electrode materials researchers have found that enhancing surface area, electrical conductivity, providing short ion-diffusion pathways and having excellent interfacial integrity lead to desirable characteristics for applications ranging from use in electric vehicles to portable electronics [9-12]. Binary transition metal oxides (as opposed to simple transition metal oxides) are one group of materials that can provide all of these characteristics and show particular promise for supercapacitor applications, particularly as they contain mixed metal valencies providing rich redox behaviour for exploitation. Two main challenges exist for the successful utilisation of BMTOs in hybrid electrochemical storage devices; producing particles with sufficient surface area to complement the redox capabilities and producing them from relatively cheap and environmentally benign metals.

The problem of surface area may be tackled by developing synthetic methods that produce intricate nano/meso scale structure or porosity in the BMTO that results in a dramatically increased surface area per particle, keeping in mind that ions still need to be able to access the surface, i.e. there is no point in having surface structure at a scale that is too small for ion migration to the surface [13-16]. One way to achieve this is to add a polymer template during the initial synthesis of the material. Using polymeric materials as templates may also result in improvements in the mechanical flexibility of the electrode, more reliable mesoporosity, and the capability to introduce pore shape and volume versatility depending on the polymer template utilised [17-19]. Many such templating agents exist with two of the more interesting being egg shell membrane (ESM) and poly methyl methacrylate (PMMA). The former has been suggested as a useful template due to its porous structure, high temperature of decomposition (over 200° C), low water uptake and swelling properties [20]. In addition use of ESM could be viewed as re-use/valorisation of a product normally considered a waste. The PMMA has a much more regular (and potentially tunable) structure [19].

Nickel cobaltate ($NiCo_2O_4$) has attracted considerable attention as a BTMO electrode material due to the relatively low cost of Ni and Co, their environmental friendliness, and natural abundance [21,22]. Furthermore, the material possesses rich redox chemistry, electronic conductivity and electrochemical activity when compared to the corresponding simple metal oxides, NiO and Co_3O_4 [15,16,23]. Most syntheses of $NiCo_2O_4$ presented in the literature don't use sacrificial templates as a means of increasing surface area, and particularly not polymeric templates. Researchers have fabricated $NiCo_2O_4$ based on α-MnO_2@$NiCo_2O_4$core-shell heterostructure and hollow $NiCo_2O_4$ nanoparticle/grapheme composite, the capacitance decreases from (991.7 to 384 $F.g^{-1}$) and from (1238 to 462 $F.g^{-1}$) respectively when the current increases from (10 to 50 $A.g^{-1}$) [24,25].
The aim of this work was to determine if the addition of polymeric templates could increase the electrochemical capacitance of hydrothermally synthesised $NiCo_2O_4$.

2 METHODOLOGY

All chemicals were purchased from Sigma-Aldrich or Chem Supply. ESM was prepared by immersing natural eggshells in 2M nitric acid for 15 minutes then separating the thin membrane layer from shell, washing with deionized water twice and drying at 90°C for 2hr.

Non-templated $NiCo_2O_4$ was synthesized via a hydrothermal process by dissolving of $Ni(NO_3)_2.6H_2O$ and $Co(NO_3)_2.6H_2O$ (2 mmol:4 mmol, respectively) into a mixed solution of ethanol and deionized water (DI) (40 ml each) at room temperature. Urea (24 mmole) was then added to the clear pink solution, the reaction mixture heated in an oven at 90°C for 8 hours and cooled to room temperature. Finally, the solution was annealed at 400°C for 3 hours. Templated $NiCo_2O_4$ was synthesised by adding (1g, 1.5g, or 2.5g) of template (eggshell membrane (ESM) or poly-methyl methacrylate (PMMA)) to the above solution immediately prior to heating at (90°C).

$NiCo_2O_4$ materials were characterized by SEM (JEOL JCM-6000) equipped with Energy-Dispersive X-ray spectroscopy (EDS) to determine surface composition. X-ray diffraction (XRD) data was collected (2θ=20°-80°) with a GBC Enhanced Multi-Material Analyser (EMMA). Specific surface area and pore size distribution were evaluated using Brunauere Emmette Teller (BET) nitrogen (N_2) adsorption-desorption isotherms and Barrett-Joyner- Halenda (BJH) method, respectively, on a Micromeritics

Tristar II surface area and porosity analyser. Fourier Transform Infrared Spectroscopy (FT-IR) was conducted using a Perkin Elmer xxx using Spectrum software, version10.4.2.

Electrochemical properties of the prepared samples were investigated by constructing a working electrode consisting of(75 wt.%) active materials, (15 wt.%) activated carbon, (10 wt.%) poly vinylidene fluoride (PVDF) binder, and (250 µL) N-methyl-2-pyrrolidine (NMP). Ingredients were mixed to produce a homogenous paste which was coated onto a ($1cm^2$) graphite sheet. Cyclic voltammetry (CV) experiments were performed in (2M) NaOH electrolyte, using Pt wire and Hg/HgO as the counter and reference electrodes in a three- electrode cell connected to a Princeton Applied Research versa STAT3. Galvanostatic charge-discharge was conducted using a two electrode cell (working electrode and activated carbon) in the potential range of (0.2-1.6 V) at current of (1 mA). Electrochemical behaviour was evaluated using Battery Analyser ((MTI Corp, USA) operated by a battery testing system (BTS). Specific capacitance was calculated from galvanostatic charge-discharge curves using:

$$Cs = \frac{I * \Delta t}{m * \Delta V}$$

Where I is the constant discharge current (A), Δt is the discharge time (s), m is mass of the electroactive materials (g) and ΔV is the potential voltage (V). The measured specific capacitances are shown in (table 1) for blank $NiCo_2O_4$, $NiCo_2O_4$ templated ESM and $NiCo_2O_4$ templated PMMA respectively.

3 RESULTS AND DISCUSSION

The prepared materials were identified as predominantly $NiCo_2O_4$ in a spinel conformation by comparison of X-ray diffraction (XRD) patterns with the JCPDF standard (Figure 1) and previously reported data [8,23,26]. There appears to be a small amount of contamination from $Ni(OH)_2$ and Co_3O_4 in the templated materials. The XRD peaks are quite broad indicating a relatively amorphous material.

Energy Dispersive X-ray Spectroscopy (EDS) of all materials (Figure 2) indicated the presence of Co, Ni and O on the surface with a 1:2 Ni to Co atomic ratio, consistent with the stoichiometric ratio of $NiCo_2O_4$ and those previously reported [27]. The chemical composition of the material surface was further elucidated by X-ray Photoelectron Spectra (XPS). This data (Table 1) provided further proof that the materials were $NiCo_2O_4$ and also showed that the Ni and Co are present in multivalent forms consistent with a spinel type structure. The Ni 2p peak was composed of two spin-orbit doublets characteristic of Ni^{2+} (855 eV) and Ni^{3+} (874 eV) [15,28] while the presence of Co^{2+} and Co^{3+} was indicated with major signals in the Co 2p peak at the binding energies of (780eV) and (796eV), respectively [10,29]. The presence of metal-oxygen bonds, consistent with formation of an oxide, was confirmed by a signal at ~ $630cm^{-1}$ in the FT-IR spectrum for each material [18,30].

The increased porosity of the templated materials was easily verified by SEM (Figure 2) and confirmed by calculation of the specific surface area and pore size distribution (Table 1). The composites appear to be deposited as an irregular porous structure (as indicated by the XRD data) as observed in the $NiCo_2O_4$ ESM and PMMA template materials at high magnification. The N_2 adsorption-desorption isotherms exhibited a hysteresis loop and analysis using the Brunauer-Emmett-Teller (BET) method showed that both template materials had a higher specific surface area than the blank material, with the PMMA template material having the largest surface area (50 times greater than the blank). The corresponding pore size distribution was calculated by the Barrett-Joyner-Halenda (BJH) method and confirmed that the ESM and PMMA samples exhibit a large pore volume and well-formed meso-porosity.

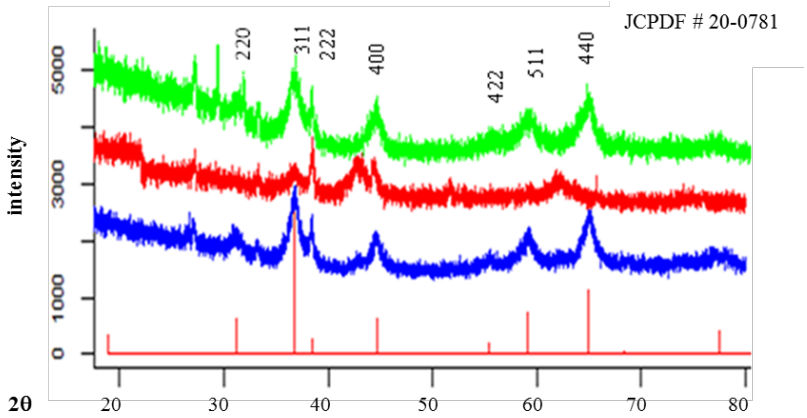

Fig. 1. XRD patterns of $NiCo_2O_4$ blank (blue), ESM templated (red), PMMA templated (green).

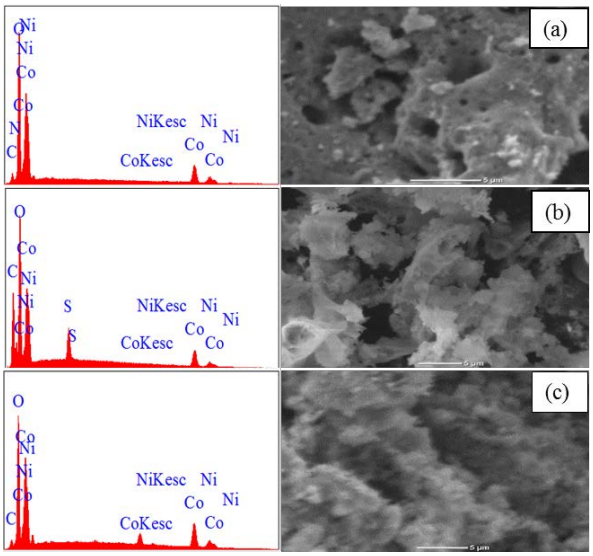

Fig. 2. SEM and EDS of $NiCo_2O_4$ blank (a), ESM template (b), PMMA template (c).

The electrochemical performance of the three synthesized $NiCo_2O_4$ materials was investigated by cyclic voltammetry (CV) in a standard three- electrode cell and charge discharge (CD) methods using a 2 electrode configuration. The CV measurements showed a clear increase in redox behaviour with addition of the PMMA template (Figure 3) as well as a dramatic increase in the peak current density. These results are mirrored in the CD data where the template materials clearly have a longer discharge time. The PMMA template material exhibits the best performance, as indicated by a doubling of the specific capacity compared with the non-templated blank (Table 1).

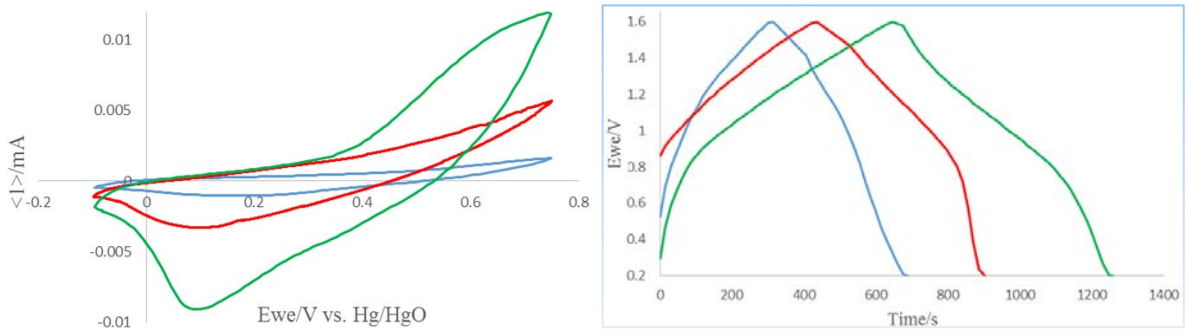

Fig. 3. CV curves (left) and CD behaviour (right) of $NiCo_2O_4$ blank (blue), ESM templated (red), PMMA templated (green)

Superior performance of the PMMA templated material appears to be due to an increase in surface area and a much larger average pore size. Converting the measured capacitance to that expected for a 3 electrodes CD system shows that the PMMA templated material exhibits a specific capacitance comparable with $NiCo_2O_4$ produced using more difficult techniques such as the capacitance for porous $NiCo_2O_4$ reduced graphene (three electrodes cell) is (783 $F.g^{-1}$) [31-33].

Table 1. Summary of surface chemistry and electrochemical data for $NiCo_2O_4$ electrode materials

		Blank		**ESM templated**		**PMMA templated**	
		eV					
XPS	Ni 2p	855	Ni^{+2}				
		874	Ni^{+3}				
	Co 2p	780	Co^{+2}				
		796	Co^{+3}				
	O 1s	529	O1				
		531	O2				
Surface area ($m^2.g^{-1}$)		0.43		7.34		**20.91**	
Pore size (nm)		*Average*	*Range*	*Average*	*Range*	*Average*	*Range*
		14	5-50	15	5-50	**50**	5-50
Capacitance ($F.g^{-1}$)		25.39		28.19		**38.03**	

4 CONCLUSIONS

In summary, polymer templated $NiCo_2O_4$ materials have been synthesised by a simple hydrothermal process followed by annealing at (400° C). The electrochemical properties of the electrode materials are highly affected by their morphology. Therefore, high performance materials have been synthesised with desirable morphology to develop supercapacitors applications. Templating improved electrochemical performance dramatically due to increasing material porosity. It appears that the average size of the pores may be a contributing factor to the electrochemical performance of these bimetallic transition metal oxide materials.

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