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Ultrahigh bacterial production in a eutrophic subtropical Australian river: Does viral lysis short-circuit the microbial loop?

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Abstract

We studied trophic dynamics in a warm eutrophic subtropical river (Bremer River, Australia) to determine potential sources of dissolved organic carbon (DOC) and the fate of heterotrophic bacterial production. Sustained high rates of bacterial production suggested that the exogenous DOC was accessible (labile). Bacterial specific growth rates (0.2 h^{-1} to 1.8 h^{-1}) were some of the highest measured for natural aquatic ecosystems, which is consistent with high respiration rates. Bacteria consumed 10 times more organic carbon than that supplied by the daily algal production, a result that implies that terrestrial sources of organic carbon were driving the high rates of bacterial production. Viruses (10^{11} L^{-1}) were 10 times more abundant than bacteria; the viral to bacterial ratio ranged from 3.5 to 12 in the wet summer and 11 to 35 in the dry spring weather typical of eutrophic environments. Through a combination of high bacterial respiration and phage lysis, a continuous supply of terrestrial DOC was lost from the aquatic ecosystem in a CO_2 -vented bacterial–viral loop. Bacterial processing of DOC in subtropical rivers may be contributing disproportionately large amounts of CO_2 to the global carbon cycle compared to temperate freshwater ecosystems.

In freshwater ecosystems, bacterial activity respire terrestrial dissolved organic carbon (DOC) and returns CO_2 to the atmosphere (Cole and Caraco 2001; Cole et al. 2006; Farjalla et al. 2009a). However, there are few reports that consider the quantitative significance of bacterial dynamics in lakes and rivers as major sources of CO_2 to the atmosphere (Richey et al. 2002; Mayorga et al. 2005; Cole et al. 2007).

The interaction of catchment and hydrological processes that supply DOC to rivers regulates microbial structure and metabolism (Sinsabaugh and Findlay 2003). Declines in DOC concentrations along flowing rivers parallel the declines in heterotrophic bacterial activity and dissolved oxygen. These declines are a function of dilution, residence time, and the supply of DOC; and they are linked to variations in microbial metabolism, at least in temperate studies (Sobczak and Findlay 2002). Terrestrial DOC entering an aquatic freshwater ecosystem can be rapidly respired, with most of the water column respiration associated with aerobic heterotrophic bacterial metabolism (Findlay et al. 1998) even if the DOC has persisted in the terrestrial environment for many years (Cole and Caraco 2001). As the DOC moves downstream it supports bacterial respiration, growth, and production, connecting the river with a reservoir of DOC and generating CO_2 to become an active part of the global carbon cycle (Cole et al. 2007). Much of the terrestrial DOC derived from the catchment that enters freshwater rivers is respired before it reaches the ocean (Cole and Caraco 2001; Muylaert et al. 2005).

Several factors complicate this picture of catchment to river to ocean carbon flow. CO_2 in freshwater can also originate from soil microbial respiration of plant organic carbon (Johnson et al. 2008) and subsequent infiltration. Viruses may also indirectly enhance the use of DOC as a

substrate for bacterial respiration and production (Pollard 2007; Bonilla-Findji et al. 2008; Rohwer and Thurber 2009).

Descriptions of global trends in carbon oxidation are biased toward temperate aquatic environment studies. Even in a major review of freshwater global carbon budgets, conclusions were based primarily on studies of temperate aquatic environments (Sobek et al. 2005, 2007). With less than 3% of studies carried out in tropical environments, global generalizations are biased toward temperate waters. Partly for these reasons, the DOC–bacteria pathway is the biggest gap in our knowledge of the importance of the subtropics in the global carbon cycle (del Giorgio and Cole 2000; del Giorgio and Williams 2005).

The fate of the high bacterial production in subtropical streams is also uncertain. Organic carbon is returned to the DOC pool, with bacterial communities turning over every few days because of viral lysis and/or grazing (Weinbauer et al. 2007; Abedon 2008; Boras et al. 2009). Protozoan bacterivores were originally thought to provide the link of bacterial production to higher trophic levels (Azam et al. 1983; Sherr and Sherr 1988; Meyer 1994). However, an alternative view suggests that most terrestrially derived DOC is a major subsidy of pelagic bacterial respiration, with little, if any, of this bacterial carbon passing up the food web (Cole et al. 2006). The DOC in estuaries is also a major subsidy of bacterioplankton respiration; again, little of this bacterial production was seen passing up the food web to higher trophic groups (Sobczak et al. 2005). Increasingly, viral infection and lysis of bacteria are viewed as “sinks” of DOC by increasing the amount of DOC that bacteria process (Weinbauer 2004; Suttle 2005; Bonilla-Findji et al. 2008), thereby preventing bacterial production from passing to higher trophic groups. While freshwater heterotrophic bacteria respire DOC rapidly and efficiently, respiration can be uncoupled from bacterial production

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(Pollard et al. 1997), further diminishing the amount that is transferred to higher trophic groups.

Thus, a recent paradigm shift is emerging that places more emphasis on primary production as the major link to the metazoan production and higher trophic groups in freshwater (Thorp 2002). The major source of this primary production is under debate. On the one hand, riverine food web studies from tropical and arid environments show autochthonous (within the system) algal production as a major source of organic carbon to metazoans (Bunn et al. 2003; Clapcott and Bunn 2003; Bunn et al. 2006), the algae having a high nutritional value (Brett et al. 2009). On the other hand, in temperate environments, allochthonous (outside the system) terrestrial particulate organic carbon is directly subsidizing river and lake food webs via zooplankton (Pace et al. 2004; Cole et al. 2006; Caraco et al. 2010). In either scenario, there is little contribution to higher trophic levels from the heterotrophic bacterial production (Tanaka et al. 2005; Cole et al. 2006; Farjalla et al. 2009b).

We do not know what happens to the high heterotrophic bacterial production in these freshwater ecosystems in which the primary production supports the top-end users. Compared to microbial processes in marine environments and temperate freshwater, little is known about those of subtropical freshwaters, especially the relationship between bacterial production, viruses, and the sources of high CO₂ saturation seen in rivers (Marotta et al. 2009).

Globally, partial pressure of CO₂ supersaturation, as a consequence of bacterial metabolic activity, is directly correlated with freshwater temperatures (Marotta et al. 2009). With climate change causing increases in global freshwater temperatures (Koeve and Ducklow 2001; Acuna et al. 2008), there is a widening gap in our knowledge of the effect of freshwater bacterial activity in warmer climates. This information is vital to predict future trends in global CO₂ budgets, as increases in bacterial production are inversely related to latitude and hence directly related to increasing global temperature. Increases in bacterial activity and changes in its fate could alter the role of rivers in the global carbon cycle, with an increase in the amount of allochthonous DOC that is respired (Acuna et al. 2008). Here, we intend to follow biodegradable DOC inputs into a subtropical river by comparing heterotrophic bacterial production and carbon demand to the rates of primary production, with a view to describing potential pathways for DOC and bacterial production loss from the ecosystem.

Methods

Location and sampling sites—The Bremer River is a major tributary of the upper Brisbane River, South East Queensland, Australia. The region is characterized by winter and spring dry weather and summer wet weather. The study was carried out in both spring dry and summer wet periods. The Bremer receives runoff from a wide range of land uses such as light and heavy industry, residential, agriculture, grazing, and bush land, and it has been deemed highly eutrophic (Greenwood et al. 1999). The lower reaches are tidally influenced up to 17 km from the

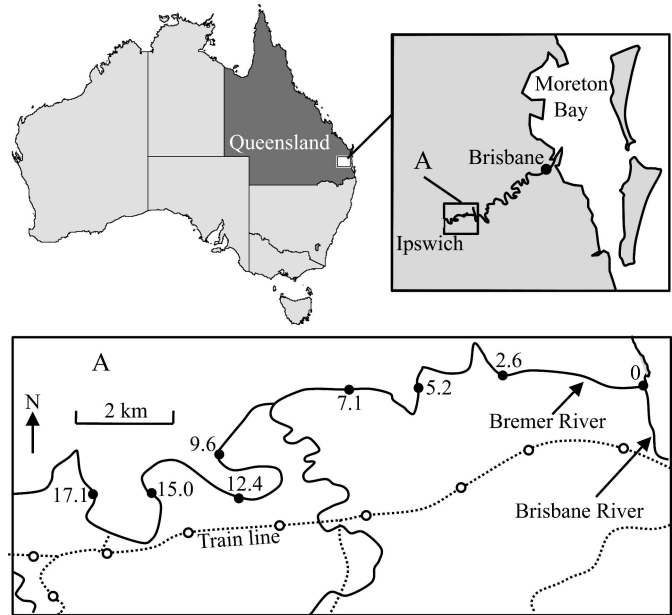


Fig. 1. Bremer River study sites. The eight sites are indicated as the distance (km) from the junction of the Bremer and Brisbane Rivers (AMTD).

junction with the Brisbane River, with little hydrological exchange between them in the dry weather (Pollard et al. 2001). A volume of 3388 ML moves up and down with each tidal cycle; hence it is well aerated and mixed but is minimally exchanged with the Brisbane River downstream. Salinity ranges from 0.13 g L⁻¹ to 0.40 g L⁻¹ through the year. For much of the time, the freshwater inflows to the river were small in the dry weather. There was little tidal flushing, the water column was well mixed, and there was little exchange with the marine environment (Pollard et al. 2001).

Eight sampling sites (Fig. 1), approximately 2.5 km apart, were established within the first 17.1 km at strategic locations—starting at the junction of the Brisbane River (latitude 27°34.946'; longitude 152°51.217'). Sampling was 18 September 2000–19 September 2000 in spring dry weather; in summer wet weather it was on 13 February 2001, 12 d after a major wet weather event in the watershed. For most of the year, the river dry weather physicochemical character persisted; even after rain, it returned to this state within 2 to 3 weeks.

Respiration—A Yellow Springs Instruments 200 biological oxygen demand (BOD) probe (John Morris Scientific) with a built-in stirrer and pretensioned cap membrane was used to measure dissolved oxygen in 300-mL BOD bottles. Bottles were filled with river water and immediately incubated in the dark at ambient temperature for 40 min. The decrease in the dissolved oxygen concentration was used to determine the oxygen uptake rate and the water column respiration rate. The rate of organic carbon respired was calculated from the rate of oxygen used, assuming a respiration quotient of 1:1 for O₂:CO₂ (del Giorgio and Williams 2005) and that aerobic respiration

and the bacterial DOM substrates were carbohydrates, proteins, and some nucleic acids.

DOC—An accredited laboratory (Queensland Health and Scientific Services, Coopers Plains, Brisbane, Australia) measured DOC and total organic carbon (TOC) concentrations. Duplicate 100-mL water samples were taken; one was filtered through a prewashed sterile 0.45- μm filter for DOC, while the other was used for TOC analysis. Both were stored in the dark on ice for transport back to the lab, where they were frozen until analyzed.

Primary productivity—Primary productivity was measured using the incorporation of radioactively labeled ^{14}C - HCO_3^- . Ranges of light (photosynthetically active radiation [PAR]) intensities were used to establish the relationship between primary productivity and PAR at several sites along the Bremer River. Water was sampled 250 mm below the surface in the euphotic zone (750 mm below surface). Shade bags were used to mimic the light profile in the water column. Water samples were added to new 600-mL polycarbonate bottles with 740 MBq (20 μCi) of ^{14}C - HCO_3^- (Millan Panich Biomedical) added in the dark. The incubation was started by mooring the bottle inside the shade bag on the water surface at Cribb Park, Ipswich (Site 15.0 adopted middle thread distance [AMTD]; Fig. 1). To stop the reaction, bottles were placed in the dark on ice (4°C), and the phytoplankton samples were collected on 47-mm-diameter polycarbonate or nylon filters with pore sizes of 0.2, 10, 20, or 180 μm within 45 min. The filters were acid treated overnight, and the following day scintillant was added and the radioactivity measured. A more detailed description of the method can be found in Pollard and Kogure (1993), with the modification that the natural bicarbonate concentration in the water was determined using the alkalinity, pH, and conductivity in the Bremer River according to Strickland and Parsons (1968). This same paper describes the methanol method used to extract and quantify the chlorophyll *a* (Chl *a*) that was also collected on the filters.

The PAR was measured using a Li-Cor L2000 data logger (Li-Cor). This was fitted with two light sensors; one was an underwater spherical quantum sensor, while the other was a planar quantum sensor used to measure air surface radiance. Both sensors measured light in the wavelength range of 400 to 700 nm. PAR was measured simultaneously at the surface and depth to estimate the percentage transmission.

The relationship between photosynthesis and irradiance (PI function) was described using the model originally described by Eilers and Peeters (1988):

$$\text{PP} = \text{PAR} \times (a \times \text{PAR}^2 + b \times \text{PAR} + c)^{-1} \quad (1)$$

where PP is the primary productivity, PAR is the photosynthetically active radiation, and *a*, *b*, and *c* are the equation parameters.

Primary productivity was determined by integrating the Eilers–Peeters 1988 model for the euphotic zone, over an 8-h day. The latter was based on the daily fluctuating light

regime and characteristics of the photosynthesis-light curves of the phytoplankton community.

Bacterial production—In situ heterotrophic bacterial production was determined at all study sites. The rate was directly related to the rate of incorporation of radioactively labeled [methyl- ^3H] thymidine (^3H -Tdr) into the deoxyribose nucleic acid (DNA) of the dividing bacteria (Fuhrman and Azam 1982) as originally described. This is a filtration followed by an acid washing method that separated the DNA precursor from the labeled DNA.

The ^3H -Tdr was purchased from Millan Panich Biomedical at a concentration of 37 MBq mL^{-1} (1 mCi mL^{-1}) in sterile aqueous solutions at a specific activity 74 GBq mmol^{-1} (2 Ci mmol^{-1}). River water was sampled (15 mL), and the incubation was started immediately by adding 50 μL of the stock ^3H -Tdr (1.7 $\mu\text{mol L}^{-1}$). This high concentration was necessary to overcome the abiotic adsorption of the isotope in this rich turbid environment (P. C. Pollard unpubl. data). Each site was sampled and assayed six times ($n = 6$). The assay was stopped with 0.250 mL of 36% (v:v) formaldehyde, and the sample was placed on ice in the dark. Incubation times were to a maximum of 40 min, and a time course was generated for each site as per the isotope dilution study described below. The intercept at time zero represents the control or background label incorporated. The samples were processed in the laboratory within 24 h. No correction to the stock specific activity (2 Ci mmol^{-1}) of thymidine (Tdr) was needed to calculate the bacterial production rates and productivity.

A conversion factor (K_{Tdr}) was used to determine the rate of formation of new cells from the rate of incorporation of labeled Tdr ($r_{\text{v,Tdr}}$; Pollard and Greenfield 1997). Similar values for this factor have been derived both empirically and theoretically, and we have used an average value for K_{Tdr} of 2×10^{18} cells mol Tdr^{-1} (Fuhrman and Azam 1982; Chróst et al. 1988; Bell 1993).

Bacterial and viral abundance—The bacterial and viral abundances at each site were determined using epifluorescence microscopy. For quantifying viral abundance, the bacteria were first removed from the sample on a 0.2- μm -pore-size filter before the viruses were collected on the aluminum oxide filter. There were so many bacteria that they often obscured the viruses; hence bacteria and viruses were counted on separate filters.

Based on the dominance of bacteria both in abundance and rates of reproduction (populations of 10^7 doubling every 20 min) over eukaryotic unicellular organisms, which take days if not weeks to replicate in the pelagic environment, we have assumed that free viruses were mostly bacteriophages (Thingstad et al. 2008; Boras et al. 2009).

Bacteria and viruses were stained with SYBR[®] Green-1 (Molecular Probes; Noble and Fuhrman 1998). The stock solution of SYBR[®] Green-1 was diluted 1 : 2.5×10^{-3} with 0.02- μm -filtered, Milli-Q deionized water (working solution). The pH of a 5-mL river water sample was measured and, if below 7, adjusted with Tris-ethylenediaminetetra-

acetic acid buffer. The SYBR® Green-1 (30 μL) working solution was added per 1 mL of the sample. After 30 min in the dark, the sample was filtered through a 0.22- μm -pore-size, 25-mm diameter, black polycarbonate membrane filter (Millipore Australia) to collect the bacteria. The filtrate was collected and then filtered through a 0.025- μm Whatman, Al_2O_3 Anodisc 25 filter (Crown Scientific) on a Millipore manifold at 15 kPa until dry, to collect the viruses.

Filters were transferred to glass slides and covered with a drop of nonfluorescent oil (Cargile B) and a 25-mm coverslip, and another drop was added on top of the coverslip. The bacteria and viruses were counted using an Olympus epifluorescent microscope (model CH30) equipped with a 100/1.25 objective lens and a blue filter set (excitation < 490 nm; dichroic filter, 500 nm; barrier filter, > 515 nm). A 100 \times oil immersion objective was used with a 10 \times ocular that contained a counting graticule. The graticule was calibrated with a 0.1-mm stage micrometer. The numbers of bacteria and viruses in the sample were determined with the graticule in the ocular of the microscope.

Bacterial carbon content—The carbon content of the bacterial cells is an essential parameter for comparing bacterial productivity in trophic studies of aquatic environments (Ducklow 2000). The type of bacteria, its nutrient environment, and its growth state influence the size (volume) and shape of a bacterial cell. Therefore we measured cell volumes of the bacteria in the Bremer River. Cell dimensions were measured on photographs of stained bacteria as von M \ddot{u} nch and Pollard (1997) have described.

Bacteria were measured (length [L_{cell}] and width [W_{cell}]) in the water sample. The cell volume (V_{cell}) of each bacterium was calculated with the following formula:

$$V_{\text{cell}} = \frac{\pi}{4} \times W_{\text{cell}}^2 \times \left(L_{\text{cell}} - \frac{W_{\text{cell}}}{3} \right) \quad (2)$$

This formula is based on bacteria that are straight rods with hemispherical ends, but it works equally well for cocci. On average, the bacteria in the Bremer sample had a cell volume of $0.16 \pm 0.02 \mu\text{m}^3$, length of $1.0 \pm 0.03 \mu\text{m}$, and width of $0.5 \pm 0.01 \mu\text{m}$ (mean \pm SE, $n = 100$).

Using a conservative value for the carbon content per bacterial cell volume of $196 \text{ fg } \mu\text{m}^{-3}$ (Bloem et al. 1995; Ducklow 2000) and the cell volume ($0.16 \mu\text{m}^3$), the bacterial mean biomass in terms of carbon was calculated using $31 \pm 8 \text{ fg carbon per bacterial cell}$. This figure was used to convert the bacterial cell production rates (cells $\text{m}^{-3} \text{ h}^{-1}$) into the bacterial carbon productivity ($\mu\text{g C m}^{-3} \text{ h}^{-1}$).

Bacterial specific growth rates—From the bacterial production and the number of bacteria at each site, the specific growth rates μ were determined as

$$\mu = \frac{dx}{dt} \times \frac{1}{X} = \frac{P}{X} \quad (3)$$

where X is the concentration of bacteria at time (t) and

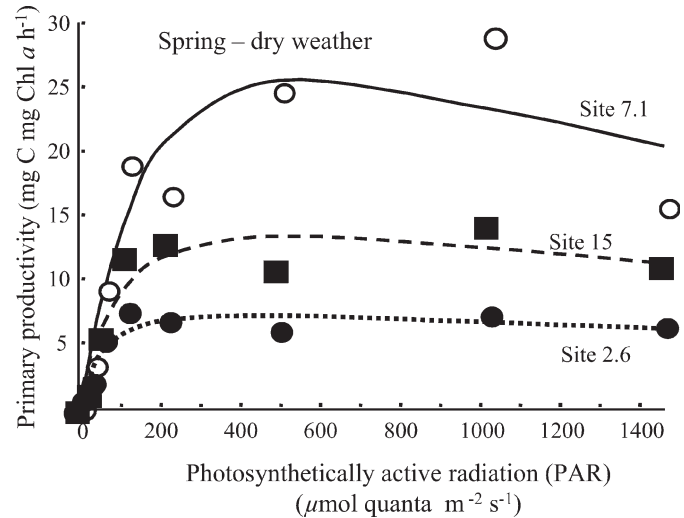


Fig. 2. The relationship between primary production and PAR during spring (dry weather). This relationship was determined for phytoplankton collected on 0.2- μm -pore-size filters using water samples collected from Sites 2.6, 7.1, and 15.0 km AMTD (Fig. 1).

bacterial production rate (P) is the rate of increase in the number of bacteria (dx/dt), i.e., the production rate divided by the number of bacteria.

Data analysis—Wet and dry weather data sets were compared by performing F -tests, generally with $n = 8$. Significance was evaluated at the $p = 0.05$ level.

Results

Primary and secondary production in this ecosystem were measured to determine quantitatively their trophic relevance to each other. The heterotrophic bacterial production was used to determine the bacterial carbon (DOC) demand in combination with respiration measurements.

Primary productivity—The relationships between photosynthesis (primary productivity in terms of carbon) and irradiance (PI function) for Sites 2.6, 7.1, and 15.0 km are shown in Figs. 2 and 3, with their characteristics summarized in Table 1. For dry weather, the maximum rates of primary productivity were between 7 and 26 $\text{mg C mg Chl } a^{-1} \text{ h}^{-1}$ with optimal PAR light intensities of about $500 \mu\text{mol quanta m}^{-2} \text{ s}^{-1}$ (Table 1; Fig. 2). The euphotic zone was determined from the relationship between primary productivity and PAR light and the depth to which PAR penetrated the water column along the river. Most primary productivity was in the top 0.75 m of the water column. Water samples taken within or below this euphotic zone produced similar PI curve characteristics. This reflected the well-mixed nature of the shallow water column.

Wet weather showed a very different picture of phytoplankton productivity (Fig. 3). Rainfall, 12 d prior to sampling, and the subsequent increase in turbidity and

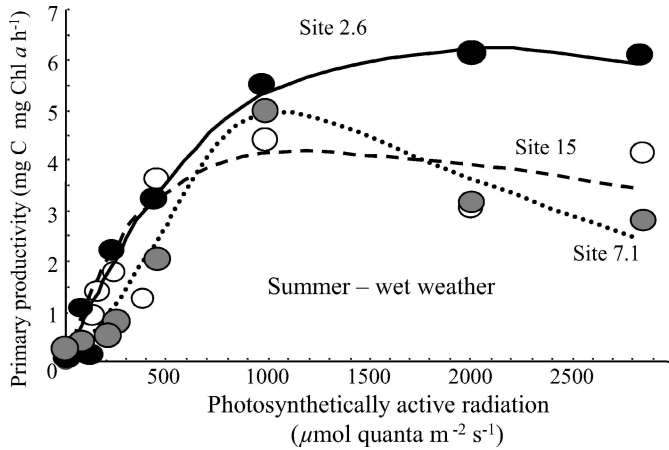


Fig. 3. The relationship between primary productivity and PAR during summer (wet weather). This relationship was determined for phytoplankton collected on 0.2- μm -pore-size filters using water collected from Sites 2.6, 7.1, and 15.0 km AMTD (Fig. 1).

DOC in the water column, meant less light was available for photosynthesis. Compared to spring dry weather, primary productivity decreased to 4 and 6 mg C mg Chl $a^{-1} h^{-1}$, while optimal light intensities increased to between 1000 and 1900 $\mu\text{mol quanta m}^{-2} s^{-1}$ (Table 1; Fig. 3). Photoinhibition occurred at the higher light intensities.

We divided the phytoplankton production into the different size fractions that were responsible for primary production. The nanoplankton (2–20 μm) fraction (cyanobacteria and small chlorophyta) was responsible for most of the phytoplankton production in both seasons, but production was 10 times greater in the spring dry weather (Table 2) because the water column was much less turbid.

DOC—More than 90% of the organic carbon in the river was dissolved (difference between TOC and DOC); the

DOC average was $6.12 \pm 0.44 \text{ mg L}^{-1}$ (mean \pm SE, $n = 8$) in dry weather and $12.59 \pm 0.90 \text{ mg L}^{-1}$ (mean \pm SE, $n = 8$) in wet weather (Table 2). DOC was significantly different between wet and dry periods (F -test, $F_{df} = 7$, $p < 0.001$).

Heterotrophic bacterial production and productivity—Figure 4 shows Site 15.0 (midriver; summer dry season) Tdr incorporation with time, using two different concentrations of labeled Tdr. The mean of both slopes was 0.233 ± 0.015 , which equated to a high bacterial productivity of $960 \text{ pmol Tdr L}^{-1} h^{-1}$. Linear kinetics were observed with high correlation coefficients for each concentration of isotope. Slopes were also similar, showing that no isotope dilution occurred, i.e., the rates of isotope incorporation were independent of the concentration used (Pollard and Moriarty 1984). This relationship was reproduced at each study site for dry and wet weather. Tdr incorporation rates were exceptionally high (Table 2). Time course incubations accumulated significant amounts of radioactivity above the zero time control (blanks) after just 5 min and remained linear for at least 40 min (Fig. 4).

On the other hand, bacterial abundance was not exceptionally high ($1\text{--}9 \times 10^9 \text{ L}^{-1}$; Table 2; Fig. 5), given the high incorporation rates. The resulting specific growth rates, derived as explained above, were accordingly high (Table 2).

Despite the difference between summer and spring water temperatures (27.3°C and 19.1°C , respectively), there was little difference between the bacterial productivity— $91 \pm 8 \text{ mg C m}^{-3} h^{-1}$ in the dry weather compared to $111 \pm 17 \text{ mg C m}^{-3} h^{-1}$ (mean \pm SE, $n = 8$) in wet weather (F -test, $F_{df} = 7$, $p < 0.04$).

However, there was a twofold increase in the specific growth rates in the summer wet compared to the spring dry season ($1.33 \pm 0.32 h^{-1}$ summer and $0.69 \pm 0.20 h^{-1}$ spring; mean \pm SE, $n = 8$), primarily because the

Table 1. Phytoplankton photosynthetic characteristics. Data generated from the primary productivity vs. irradiance relationships (PI curves) for upper, middle, and lower reaches of the Bremer River in spring and summer.

Primary production (PP)*	Spring dry weather			Summer wet weather			Remarks
AMTD† (km)	2.6	7.1	15.0	2.6	7.1	15.0	
P_{max} (mg C mg Chl $a^{-1} h^{-1}$)	6.9	25.5	14.2	6.2	4.2	5.1	Maximum rate of production
I_m ($\mu\text{mol quanta m}^{-2} s^{-1}$)	432.6	550.1	528.0	1924.5	1199.7	1131.0	Optimal intensity
I_k ($\mu\text{mol quanta m}^{-2} s^{-1}$)	46.4	136.0	85.9	666.9	330.6	1395.7	Characteristic intensity
S_{\ddagger}	0.149	0.188	0.165	0.009	0.013	0.004	Initial slope
Photoinhibition	7.3	2.0	4.1	0.9	1.6	−1.2	Relative index of photoinhibition potential (smaller values correspond to greater potential)

* Determined from PI function described in Methods.

† Distance from the junction of the Bremer and Brisbane Rivers.

‡ S units = $(\text{mg C mg Chl } a^{-1} h^{-1})/(\mu\text{mol quanta m}^{-2} s^{-1})$.

Table 2. DOC, TOC, bacterial abundance, H³-Tdr incorporation rate, primary and secondary production, bacterial productivity, specific growth rate, number of virions, and the ratio (V : B ratio) between viral and bacterial abundance for each site are shown. While bacterial production was highest in dry weather, the DOC was greatest in wet weather.

Weather season	Bremer River site (AMTD [km])*	DOC (mg L ⁻¹)	TOC (mg L ⁻¹)	Bacterial abundance (cells L ⁻¹ × 10 ⁹)	H ³ -Tdr incorporation rate (pmol L ⁻¹ h ⁻¹)	Bacterial production rate (cells L ⁻¹ × 10 ⁹) †	Bacterial productivity carbon (mg C m ⁻³ h ⁻¹)	Maximum rates of photosynthesis (mg C m ⁻³ h ⁻¹)	Bacterial specific growth rate (h ⁻¹)	Virions (No. L ⁻¹ × 10 ¹⁰)	V : B ratio
Dry spring	0	5	4.4	6.49	2045.8	4.09	128.3		0.63	10.80	16.64
	2.6	5	4.4	4.61	1390.6	2.78	87.2	36.0	0.60	8.50	18.44
	5	5	5.0	3.84	1659.7	3.32	104.1		0.86	13.50	35.16
	7.1	6	5.7	5.06	1444.4	2.89	90.6	20.7	0.57	11.80	23.32
	9.6	7	6.6	9.24	1030.5	2.06	64.6		0.22	9.30	10.06
	12.4	7	6.5	6.15	1098.2	2.20	68.9		0.36	7.14	11.61
	15	7	6.3	5.03	1163.9	2.33	73.0	26.9	0.46	9.60	19.09
	17.1	7	6.6	8.95	1778.4	3.56	111.5		0.40	9.60	10.73
	0	6.0	6.0	2.39	1628.0	3.26	102.1		1.36	2.37	9.90
	2.6	12.6	13.0	1.47	1323.7	2.65	83.0	3.3	1.80	1.76	11.95
Wet summer	5	14.0	14.0	3.53	1562.5	3.13	98.0		0.89	1.22	3.46
	7.1	13.4	14.0	3.41	488.6	0.98	30.6	1.5	0.29	1.37	4.03
	9.6	13.8	14.0	3.26	2870.9	5.74	180.1		1.76	1.20	7.08
	12.4	13.9	14.0	3.12	2594.7	5.19	162.7		1.66	2.52	8.08
	15.1	14.0	14.0	3.71	2422.3	4.84	151.9	1.9	1.31	1.83	4.94
	17.1	13.0	13.0	4.02	1312.2	2.62	82.3		0.65	1.22	3.04

* Distance from the junction of the Bremer and Brisbane Rivers.

† 2×10^{18} bacterial cells mol Tdr⁻¹ is the conversion factor used to calculate bacterial production.

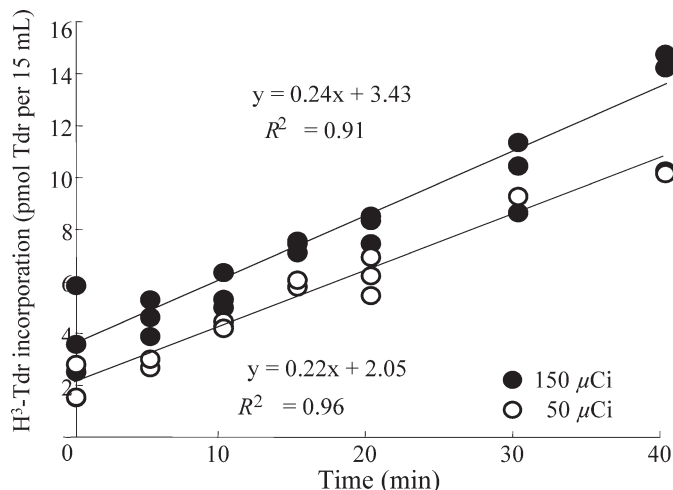


Fig. 4. Typical time course, kinetic, and isotope dilution study of the rate of incorporation of ^3H -Tdr into bacterial DNA. Linear kinetics were observed with high correlation coefficients for each concentration of isotope. This relationship was reproduced at each study site for dry and wet weather. No isotope dilution occurred as the slopes were similar (Pollard and Moriarty 1984; Site 15.0 midriver; dry weather; mean of both slopes was $0.24 = 960 \text{ pmol Tdr L}^{-1} \text{ h}^{-1}$).

abundances were lower (Table 2). Specific growth rates were significantly different between wet and dry periods (F -test, $F_{df} = 7$, $p < 0.019$).

Abundance of bacteria and viruses—The abundance of pelagic bacteria and viruses and the ratio between the two (V : B ratio) can be compared in Table 2 and Fig. 5. The average spring bacterial abundance was twice that of summer— $6.20 \pm 0.70 \times 10^{12} \text{ cells m}^{-3}$ compared to $3.10 \pm 0.34 \times 10^{12} \text{ cells m}^{-3}$ (mean \pm SE, $n = 8$). Viral abundance was 10 times greater than bacterial abundance. In spring dry weather, the average number of viruses was $10.0 \pm 1.1 \times 10^{13} \text{ virus-like particles (VLPs) m}^{-3}$ (mean \pm SE, $n = 8$), which was significantly higher (F -test, $F_{df} = 7$, $p < 0.001$, $n = 8$) than the summer wet weather average of $1.7 \pm 2.5 \times 10^{13} \text{ VLPs m}^{-3}$ (mean \pm SE, $n = 8$).

The association of viral (V) and bacterial (B) abundance can be described with the V : B ratio, which was 19.0 ± 3.5 (mean \pm SE, $n = 8$) in the dry and 6.5 ± 1.1 (mean \pm SE, $n = 8$) in the wet. The V : B ratio was significantly ($p < 0.001$, $n = 8$) higher in spring dry weather compared to the wet summer. Within each season, the trend in V : B ratios followed those of bacterial-specific growth rates (Table 2).

Discussion

Various assumptions are typically made to derive estimates of bacterial production rates (Ducklow 2000). Our assumptions are discussed briefly before exploring the implications of the high production rates for carbon flux in the Bremer River. This study also complements that of Hewson et al. (2001), who determined the VLP distribution in the sediments and water column in the Brisbane River

and Moreton Bay 26 km downstream from the junction of the Bremer and Brisbane Rivers.

Bacterial carbon demand (BCD) is the DOC used as substrate for bacterial metabolism (del Giorgio and Cole 2000). The flows of DOC into bacterial production and respiration are related by the bacterial growth efficiency (BGE) term that is defined as bacterial production / (bacterial production + bacterial respiration) (Farjalla et al. 2009b). Determining quantitatively the amount of readily degradable DOC supplied to the bacteria relies on sensitive and accurate measurements of bacterial production in situ. We achieved this through the measurement of the incorporation of ^3H -Tdr into growing bacteria after taking into account background (nonspecific binding of the isotope) and isotope dilution. This dilution can be the result of the cell's ability to synthesize its own Tdr residues and must be considered in cases in which ^3H -Tdr is used to measure bacterial growth rates (Pollard and Moriarty 1984).

Isotope dilution—Our kinetic study of the rate of incorporation of ^3H -Tdr into bacterial DNA showed that the rate of incorporation of the radiolabeled Tdr into bacterial DNA was linear for at least 40 min ($r^2 = 0.9$; Fig. 4) for the incubation of $50 \mu\text{Ci}$ and $150 \mu\text{Ci}$ ^3H -Tdr, respectively. The rate of incorporation of ^3H -Tdr, as indicated by the slopes, was the same irrespective of the concentration of Tdr used. The bacteria were not likely using the Tdr as a carbon source as there were sufficient other sources of DOC (Table 2). The correlation coefficients were close to 1, and each showed a similar rate of isotope incorporation. Hence we also showed that there was no isotope dilution from the de novo synthesis of Tdr (Pollard and Moriarty 1984). This was the basis for choosing $50 \mu\text{Ci}$ to determine the bacterial production (i.e., final concentration of Tdr was $1.7 \mu\text{mol L}^{-1}$).

The intercept on the x-axis in Fig. 4 represented the zero time control (background). The high backgrounds were due to the nonspecific binding of ^3H -Tdr to suspended colloidal clay particles in the water column. This same suspended material also caused the low light penetration through the water column in the summer wet weather. In spite of the high zero time background, the highly linear time courses establish the precision of the rate estimates.

Conversion factors—We chose conservative conversion factors to determine bacterial production based on those used in marine and freshwater research (see Methods). The high rates of bacterial production were due to the high rate of incorporation of Tdr and were not due to the conversion factors or artifacts (background labeling was taken into account) in the method used. Bacterial cell volumes for the Bremer were measured directly, and a conservative value for cell carbon content was used to determine productivity. These high rates were further supported by the high rates of bacterial respiration (Table 3).

Bacterioplankton production and carbon demand—We reviewed other measurements of bacterial specific growth rates in aquatic systems (Table 4). Generally, bacterial specific growth rates in freshwater were more than 10 times

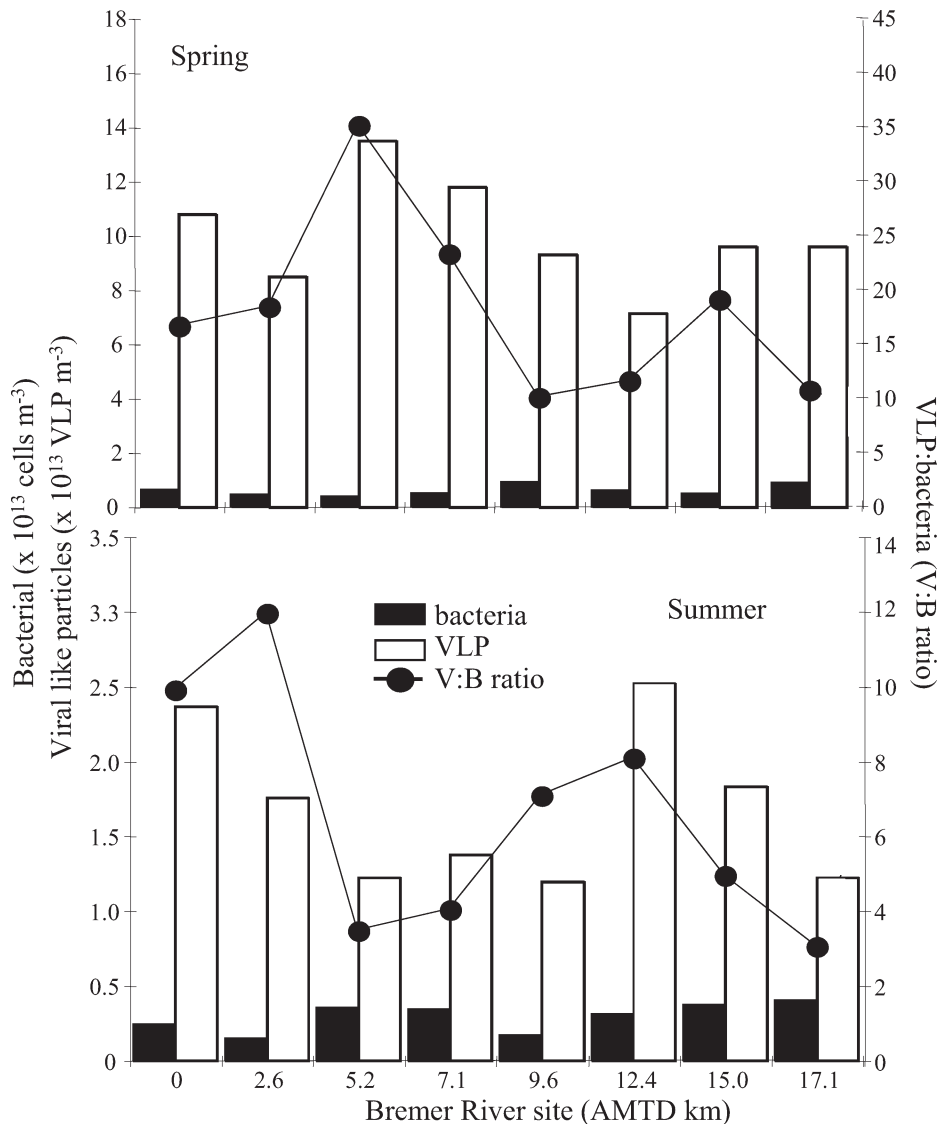


Fig. 5. Bacterial abundance and VLPs. The line shows the trend in the ratio between the two for spring (dry weather) and summer (wet weather).

greater than those measured in marine ecosystems, and our measurements were at least twice those of other eutrophic freshwater ecosystems. Our rates are among the highest recorded for natural tropical aquatic ecosystems. Other freshwater ecosystems were sampled in South East Queensland and confirmed these high rates of bacterial specific growth rates for three freshwater reaches of urban rivers and creeks—the Brisbane and Mary Rivers and Lota Creeks (Table 4; P. C. Pollard unpubl. data).

We estimated the BGE by assuming that most of the measured respiration in the water was bacterial; others that would contribute include algae, plants, protozoans, and invertebrates. This assumption overestimates the actual bacterial respiration and thus provides a conservative estimate of the BGE (Table 5). In the literature, the fraction of the community respiration that bacteria contribute ranges from 11% to 100% (Findlay et al. 1992; del Giorgio and Cole 2000; Warkentin et al. 2007). Our

measurements and assumptions yielded BGE values of 0.60 and 0.46 for the dry and wet seasons, respectively (Table 3). If we assumed that the bacterial respiration was much less than the total measured respiration, the resulting BGE values would be unrealistically high.

In a review of BGE values for aquatic ecosystems, most eutrophic systems plateau near 0.5 (del Giorgio and Cole 1998), consistent with our estimates. Bacterial populations tend to maximize their use of DOC; the energy generated in this process is then allocated between growth and maintenance, depending on the physiological state of the bacteria and the inorganic nutrients available for growth (Pollard et al. 1997). As long as the system is supplied with DOC substrate, the river bacteria continue to respire the DOC. The respiration rate for spring dry weather was lower than the bacterial production, i.e., $BGE > 0.5$ (Table 3). This may have been because bacterial respiration and production were uncoupled, or the critical high initial

Table 3. Summary of the key parameters measured in this study to quantify the trophic dynamics of the ecosystem of the Bremer River. BGE = (bacterial productivity)/(bacterial respiration + bacterial productivity; mean \pm SE).

Measured parameters	Spring dry weather (g C m ⁻³ d ⁻¹)	Summer wet weather (g C m ⁻³ d ⁻¹)
Heterotrophic bacterial productivity	1.50 \pm 0.20 (<i>n</i> =22)	2.6 \pm 0.4 (<i>n</i> =8)
Dissolved organic carbon*	5.7 \pm 0.3 (<i>n</i> =5)	12.6 \pm 1.0 (<i>n</i> =8)
Respiration†	1.04 \pm 0.12 (<i>n</i> =6)	3.2 \pm 0.1 (<i>n</i> =6)
Primary production	0.10–0.52	0.01–0.045
Bacterial growth efficiency (BGE)#	60%	46%

* g C m⁻³.

† Calculated from the oxygen uptake rates in the dark.

% bacterial production/(bacterial production + bacterial respiration)

Table 4. Bacterial specific growth rate and productivity in aquatic environments reported in the literature. Freshwater rates were consistently more than 10 times higher than in marine ecosystems.

Location	Ecosystem type	Bacterial specific growth rate (h ⁻¹)	Bacterial productivity (g C m ⁻² d ⁻¹)	Source
Atlantic Barrier Reef—Belize	Marine—eutrophic	0.04	0.016	Rath et al. 1993
North Atlantic	Marine oceanic—euphotic zone	0.013	0.300	Ducklow 1999
Continental shelf of the Southern East China Sea	Nutrient-rich coastal and upwelling waters	0.010	0.329	Shiah et al. 2000
Danish lakes	Temperate, shallow, and eutrophic	0.1	0.008	Theil-Nielsen and Sondergaard 1999
Han River Estuary and Yellow Sea	Macrotidal eutrophic environment—high tide	0.159	—	Hyun et al. 1999
Wastewater treatment	Engineered	0.20	—	Pollard and Greenfield 1997
Spain	Freshwaters	0.36	—	Iriberry et al. 1990
Bietri Bay (Ebrie Lagoon, Ivory Coast)	Monomictic eutrophic bay of a tropical lagoon	0.41	6.2	Torreton et al. 1989
Lota Creek (Queensland, Australia)	Subtropical freshwater reaches	0.17–0.42	—	P. C. Pollard unpubl. data
Mary River (Queensland, Australia)	Subtropical estuary to freshwater	0.023–0.91	2.2–5.1	P. C. Pollard unpubl. data
Brisbane River (Queensland, Australia)	Subtropical freshwater reaches	0.05–0.62	0.06–0.5	P. C. Pollard unpubl. data
Bremer River (Queensland, Australia)*	Subtropical freshwater—spring dry weather	0.69	4.5	This study
Bremer River (Queensland, Australia)*	Subtropical freshwater—summer wet weather	1.15	7.5	This study

* Mean water depth of 3 m.

Table 5. Comparison of in situ bacterial growth rate efficiencies in estuaries and lakes.

Site	Bacterial growth efficiency	Source
Bremer River (spring)	0.60	This study
Bremer River (summer)	0.46	This study
Florida estuaries	0.11–0.61	Coffin et al. 1993
Hudson River	0.10–0.50	Findlay et al. 1992
Santa Rosa Sound	0.60–0.61	Kroer 1993
Brackish estuary	0.40	Laanbroek et al. 1985
Blackwater river	0.31	Meyer et al. 1987
Frederiksborg Slotsso (eutrophic lake)	0.41	Middelboe and Sondergaard 1993

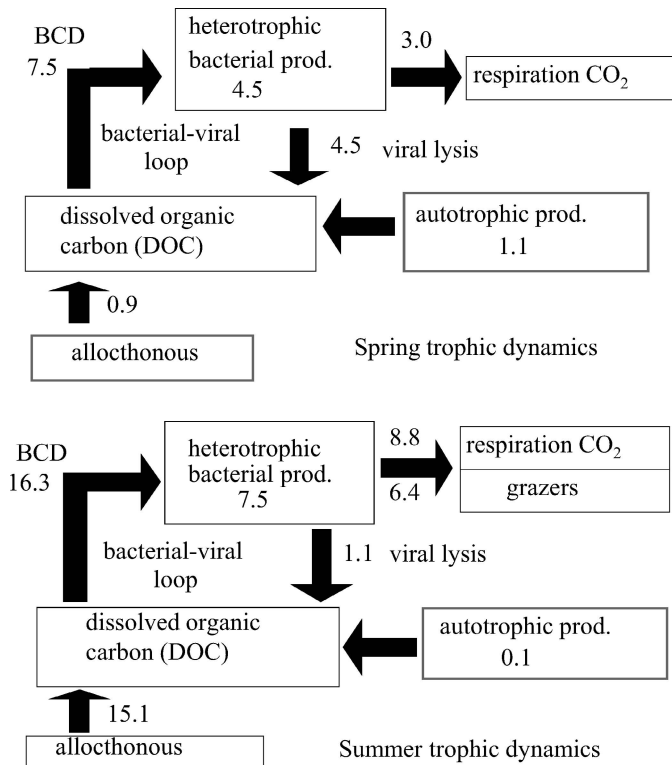


Fig. 6. Trophic relationships of the Bremer. The ecosystem was net heterotrophic in spring and summer with primary production (prod.) a minor contributor to the DOC pool that supported bacterial production (BCD = bacterial carbon demand). All units are $\text{g C m}^{-2} \text{d}^{-1}$. All primary production is assumed to enter the DOC pool over a seasonal time scale in this “best-case” scenario. Allochthonous inputs were adjusted to balance the DOC pool. Average water depth was 3 m.

oxygen decay rate (used to determine respiration) may have been missed because the bacterial respiration and production were so high.

Phytoplankton production and sources of DOC—To help quantitatively distinguish sources of organic carbon to the ecosystem, primary production was also measured and found not to be sufficient to support the high BCD. Benthic microalgae were not considered a major source of primary production, because the euphotic zone was so shallow (0.75 m). In situ heterotrophic bacterial production dominated over primary production in both wet and dry periods (Fig. 6). In spring dry weather, the heterotrophic bacteria were using DOC at a rate of at least $4.5 \text{ g C m}^{-2} \text{d}^{-1}$ (net bacterial production; Fig. 6) plus the $3.0 \text{ g C m}^{-2} \text{d}^{-1}$ of CO₂ that was lost to the atmosphere through respiration using the BGE of 60%. The total BCD was $7.5 \text{ g C m}^{-2} \text{d}^{-1}$. However, on a daily basis, the primary production represented only 15% of this demand. The pool of DOC in the Bremer River was $6.4 \pm 0.5 \text{ mg L}^{-1}$ (mean \pm SE, $n = 8$) and was near a steady state at a concentration consistent with DOC in tropical rivers (Mulholland 2003). DOC was the main fraction of the total pool of organic carbon (Table 2). Based on the rate at which bacteria were using DOC, the water column DOC was turned over every couple of days.

Summer wet weather showed twice as much DOC ($12.6 \pm 1.0 \text{ mg L}^{-1}$; mean \pm SE, $n = 8$); however, the rates of primary production were lower than in the dry spring season, while rates of heterotrophic bacterial production were higher. Primary production represented only 1% of the BCD and, again, could not alone have met the heterotrophic BCD (Fig. 6). In wet weather, twice as much DOC was delivered to the system, and, at the same time, less light was available for photosynthesis based on the PI relationships and PAR measurements in the water column (Table 1; Fig. 3).

In a review of primary and bacterial productivity in the marine environment, Ducklow (1999) found that primary productivity rates ranged from 0.5 to $1.6 \text{ g C m}^{-2} \text{d}^{-1}$. In freshwaters and eutrophic estuaries, the range for primary productivity is higher (0.6 to $4 \text{ g C m}^{-2} \text{d}^{-1}$; Rothlisberg et al. 1994; Kromkamp et al. 1995; Cole et al. 2006). We found that in spring dry weather the primary productivity ranged from 0.4 to $1.5 \text{ g C m}^{-2} \text{d}^{-1}$, and in summer wet weather it was much lower, ranging from 0.05 to $0.13 \text{ g C m}^{-2} \text{d}^{-1}$. The subtropical sites in this study were supplying similar amounts of primary production to the ecosystem as temperate freshwater environments, at least in the spring dry climate.

While terrestrial plant DOC is a major source of organic carbon in estuaries (McCallister et al. 2004) and is a major substrate for pelagic bacterial respiration, little of this bacterial carbon passes up the food web (Cole et al. 2006). The higher trophic groups in freshwater depend on primary production (Pace et al. 2004; Davies et al. 2008; Brett et al. 2009), not bacterial production. The results of this study show that phytoplankton production was only a small part of the source of the DOC driving a substantial amount of heterotrophic bacterial production.

Viral abundance and replication—In another study of a tropic freshwater reservoir, others also observed that the dry season was higher than the wet season in terms of viral abundance and was most favorable for viral infection of bacteria (Peduzzi and Schiemer 2004). While our measures of the bacterial specific growth rates were highest in the wet, the viral infection and lysis were lower as indicated by the lower V : B ratio, with bacterial mortality due to viral infection and lysis lower during the summer wet season. This conclusion was confirmed using a viral decay rate term, as outlined below.

Viral lysis of heterotrophic bacteria is important to the cycling of organic and inorganic nutrients and is a major cause of bacterial mortality in oligotrophic (Boras et al. 2009) and eutrophic environments (Bongiorni et al. 2005). Heldal and Bratbak (1991) report freshwater viral decay rates of 1.1 h^{-1} with an average burst size of 50. Assuming that within a season the abundance of bacteria and viruses remains in a steady state, the following applies:

$$\begin{aligned} \text{Viral specific decay rate} &= \text{viral specific production} \\ &= \mu = 1.1 \text{ h}^{-1} = \ln 2 \times t_d^{-1} \end{aligned}$$

With viral abundance remaining constant in the water column, viral specific production (μ) would be halved every

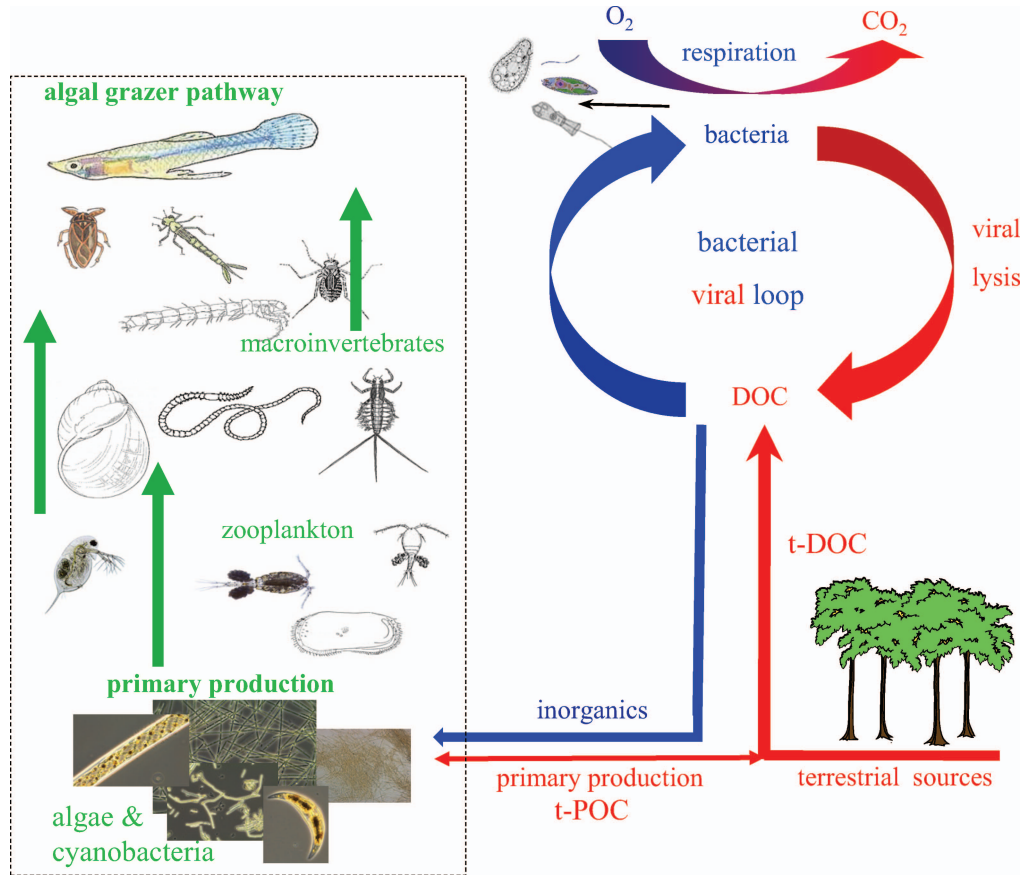


Fig. 7. A bacterial–viral loop prevented bacterial production passing to higher trophic groups. Viral lysis products of new bacterial biomass were passing out of the ecosystem via respiration in this CO_2 -vented bacterial–viral loop. On each pass of the loop, more DOC was respired as CO_2 , with the resulting inorganic compounds supporting the algal grazer pathways. Viral infection and lysis of the bacteria expedited the conversion of DOC to CO_2 . t-DOC and t-POC represent terrestrial DOC and terrestrial particulate organic carbon, respectively.

0.63 h (doubling time = t_d). In the spring dry season, the mean viral abundance was $10.03 \pm 0.72 \times 10^{10}$ virions L^{-1} (mean \pm SE, $n = 8$), which, divided by 0.63, would give a viral production value of $17.02 \pm 1.23 \times 10^{10}$ virions $\text{L}^{-1} \text{h}^{-1}$. Division of this value by the burst size of 50 gives a rough estimate of what the bacterial mortality might have been, i.e., $0.34 \pm 0.03 \times 10^{10}$ bacterial cells $\text{L}^{-1} \text{h}^{-1}$. This bacterial death rate was similar to the rate of bacterial production of $0.29 \pm 0.03 \times 10^{10}$ bacterial cells $\text{L}^{-1} \text{h}^{-1}$ (mean \pm SE, $n = 8$); hypothetically, most, if not all, of the bacterial production was lost through viral lysis. Thus, most bacterial production was lost from the food web (“sink,” Ducklow et al. 1986) through viral lysis of bacteria—little of this bacterial production may have passed to higher trophic levels in the spring dry season. This conclusion is consistent with results from the stable isotope research noted above in which primary production is seen as the basis of freshwater food webs, even in the presence of high rates of bacterial production. Several reports also indicate that viral-induced bacterial mortality is similar to that by heterotrophic nanoflagellates (Heldal and Bratbak 1991; Weinbauer 2004; Tanaka et al. 2005).

Using the same viral decay = production rate (1.1 h^{-1}) and burst size (50) for the summer wet season, viral

production was possibly five times lower than in the spring dry season ($2.7 \pm 0.6 \times 10^{10}$ virions $\text{L}^{-1} \text{h}^{-1}$; mean \pm SE, $n = 8$) that may have been responsible for a bacterial mortality at a rate of $0.05 \pm 0.01 \times 10^{10}$ bacterial cells $\text{L}^{-1} \text{h}^{-1}$. The bacterial growth rate was $0.36 \pm 0.07 \times 10^{10}$ bacterial cells $\text{L}^{-1} \text{h}^{-1}$ (mean \pm SE, $n = 8$), so only 14% of bacterial mortality could be attributed to viral lysis of bacteria. Others (Peduzzi and Schiemer 2004; Boras et al. 2009) have shown bacterial mortality ranged from 15% to 48% that could be attributed to viruses. Heterotrophic nanoflagellate grazers were possibly responsible for the remaining bacterial mortality. These grazers transfer little bacterial production to higher trophic levels, because several trophic transfers are required to move the bacterial carbon to larger grazers, with respiratory losses at each level (Ducklow et al. 1986).

Ecological consequences of viral lysis of bacteria—Viral lysis of host bacteria generates DOC (Middelboe and Lyck 2002) as seen in Fig. 7. This DOC has been shown to contain remnants of bacteria (McCarthy et al. 1998). Other bacteria consume this lysed bacterial DOC; effectively, bacterial biomass is cannibalized (Nagata 2000). For marine ecosystems, Fuhrman (1999) presented a planktonic

viral loop showing a semiclosed loop connecting prokaryotes, viruses, and DOC. The net effect of this loop is to convert organic matter into dissolved inorganic nutrients, including respired CO₂. Products of viral lysis of bacteria have also been shown to be available to bacteria for growth (Noble et al. 1999). The bacterial–viral loop detours bacterial production away from protists that would otherwise consume the bacteria, returning the biomass to the DOC pool (Fuhrman 1999; Thingstad et al. 2008; Fig. 7).

Figure 6 quantitatively shows the bacterial–viral CO₂-vented loop balancing carbon movement through this ecosystem. The concept (Fig. 7) differs from that of Fuhrman (1999) as it emphasizes the quantitative significance of the trophic loss of organic carbon from the DOC pool via bacterial respiration. The net effect of viral lysis of the bacteria is to oxidize organic material and regenerate inorganic nutrients that support primary production (Haaber and Middelboe 2009). New bacterial biomass coming from the products of viral lysis of other bacteria continues to respire CO₂ in the CO₂-vented bacterial–viral loop (Fig. 7). On each pass of the loop, more DOC would be respired as CO₂; the DOC is continuously passing out of this loop as CO₂. Thus little bacterial carbon is passing to higher trophic levels.

In our study (Figs. 6, 7), the DOC pool remained relatively constant within a season despite the high BCD. This suggests that there was a continuous supply of terrestrial DOC to support the high bacterial production. Understandably, this source was greatest in wet weather, when runoff would bring much organic carbon into the river from the catchment. On the other hand, in the dry weather less carbon came from the catchment, but much more DOC was being recycled through viral lysis of bacteria. In the absence of sufficient primary production to support all the BCD, terrestrial DOC was likely returned to the atmosphere as CO₂ gas emissions through bacterial respiration (Cole et al. 2000, 2006) assisted by viral infection and lysis of the bacteria, while phytoplankton sustained the higher trophic groups (Brett et al. 2009).

The subtropical Australian Bremer River was net heterotrophic in dry and wet weather. We demonstrated that the net heterotrophic condition was the result of very high bacterial production and respiration. Further, we demonstrated that primary production was not sufficient to support the high BCD. The work showed that the relatively constant concentrations of DOC (within a season) turned over within days; the abundance of bacteria and viruses accompanied some of the highest rates of bacterial production seen in natural freshwater ecosystems, with trends in high specific bacterial growth rates accompanying V:B ratio trends. Viral-induced bacterial mortality likely consumed most of the new bacterial production in the dry spring season. Bacterial metabolic processes appeared to be removing a continuous supply of terrestrial DOC through the viral lysis of the bacterial biomass that augmented bacterial respiration. The net effect of viral lysis of the bacteria was to oxidize more dissolved organic material and regenerate inorganic nutrients to support primary production. New bacterial biomass coming from the products of

viral lysis of bacteria continued to respire DOC in a CO₂-vented bacterial–viral loop. On each pass of the loop, more DOC was respired as CO₂ and was lost from the ecosystem. These subtropical freshwater ecosystems of Australia appear to contribute a disproportionately large amount of CO₂ to the global carbon cycle. To answer the question in the title, yes—viral lysis can short-circuit the microbial loop.

Acknowledgments

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