

1 **BIOLOGICAL SCIENCES: Microbiology**

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3 **Mechanism for nitrogen isotope fractionation during ammonium assimilation by**

4 ***Escherichia coli* K12**

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26

27 **Abstract**

28 Organisms that use ammonium as the sole nitrogen source discriminate between [<sup>15</sup>N] and  
29 [<sup>14</sup>N]ammonium. This leaves an isotopic signature in their biomass that depends on the external  
30 concentration of ammonium. To dissect how differences in discrimination arise molecularly we  
31 examined a wild-type strain of *E. coli* K12 and mutant strains with lesions affecting ammonium-  
32 assimilatory proteins. We used isotope-ratio mass spectrometry to assess the nitrogen isotopic  
33 composition of cell material when the strains were grown in batch culture at either high or low  
34 external concentrations of NH<sub>3</sub> (achieved by controlling total NH<sub>4</sub>Cl and pH of the medium). At  
35 high NH<sub>3</sub> (≥ 0.89 μM), discrimination against the heavy isotope by the wild-type strain (-19.2‰)  
36 can be accounted for by the equilibrium isotope effect for dissociation of NH<sub>4</sub><sup>+</sup> to NH<sub>3</sub> + H<sup>+</sup>. NH<sub>3</sub>  
37 equilibrates across the cytoplasmic membrane and glutamine synthetase does not manifest an  
38 isotope effect *in vivo*. At low NH<sub>3</sub> (≤ 0.18 μM), discrimination reflects an isotope effect for the NH<sub>4</sub><sup>+</sup>  
39 channel AmtB (-14.1‰). By making *E. coli* dependent on the low-affinity ammonium-assimilatory  
40 pathway, we determined that biosynthetic glutamate dehydrogenase has an inverse isotope effect  
41 *in vivo* (+8.8‰). Likewise, by making unmediated diffusion of NH<sub>3</sub> across the cytoplasmic  
42 membrane rate-limiting for cell growth in a mutant strain lacking AmtB, we could deduce an *in*  
43 *vivo* isotope effect for transport of NH<sub>3</sub> across the membrane (-10.9‰). The paper presents the  
44 raw data from which our conclusions were drawn and discusses the assumptions underlying  
45 them.

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## 54 **Introduction**

55 Fractionation of heavy and light isotopes of nitrogen ( $^{15}\text{N}$  vs.  $^{14}\text{N}$ ), carbon ( $^{13}\text{C}$  vs.  $^{12}\text{C}$ ) and  
56 hydrogen (D vs. H) can provide information about metabolic pathways and reaction mechanisms  
57 within living organisms (1–4). For example, fractionation between  $^{15}\text{N}$  and  $^{14}\text{N}$  during  
58 incorporation of ammonium N in a single-celled organism like *E. coli* is determined by the rate-  
59 limiting step for assimilating it into glutamate, the precursor of 88% of cellular nitrogen-  
60 containing material (5). All nitrogen assimilated into this central metabolic intermediate goes on  
61 to be incorporated into cell material. Transfers from glutamate to other molecules are direct.  
62 Although transfers from glutamine, including that to glutamate, involve deamidation, the  $\text{NH}_3$   
63 released is carried directly to the assimilatory catalytic site through a tunnel and hence cannot be  
64 protonated or diffuse away (6). The overall ratio of  $^{15}\text{N}$  to  $^{14}\text{N}$  in biomass is thus controlled by a  
65 step(s) at or prior to assimilation of ammonium N into glutamate.

66

67 In *E. coli*, the proteins participating in early and potentially rate-determining steps in the  
68 incorporation of  $\text{NH}_4^+$  into glutamate are: 1) AmtB, its only membrane channel for  $\text{NH}_4^+$ (7); 2)  
69 glutamine synthetase (GS), the first enzyme of the high-affinity ammonium-assimilatory pathway;  
70 3) glutamate synthase [glutamine(amide) 2-oxoglutarate amino transferase], GOGAT; and 4)  
71 glutamate dehydrogenase (GDH), the first enzyme of the low-affinity ammonium-assimilatory  
72 pathway (Fig. 1; reviewed in (8)). To study effects of these proteins on the *in vivo* fractionation of  
73 ammonium N we used wild-type and genetic mutant strains in which one or more was lacking or  
74 defective and studied these strains at high or low external concentrations of  $\text{NH}_3$ . To decrease the  
75 concentration of external  $\text{NH}_3$  and still achieve significant cell yield, we lowered both the total  
76 concentration of  $\text{NH}_4\text{Cl}$  and the pH of the medium. Though *E. coli* lives in the human gut, which is  
77 nitrogen-rich, it also survives in fresh and brackish water, in which supplies of available  
78 ammonium can be limited (9, 10). Moreover, it acidifies its own environment by fermentation.

79 Accordingly, the conditions we have chosen to study the behavior of *E. coli* at low external  $\text{NH}_3$  are  
80 pertinent to its normal life cycle.

81

82 Ammonium ( $\text{NH}_4^+ + \text{NH}_3$ ) is the optimal nitrogen source for *E. coli*, *i. e.*, that which yields most  
83 rapid growth. Ammonium [ $\text{pKa} = 9.25$ ] enters cells in two forms (Fig. 1).  $\text{NH}_3$ , which is  $\sim 2\%$  of  
84 total ammonium at pH 7.4, crosses the cell membrane by unmediated diffusion, a process that  
85 cannot be altered genetically. When the pH is decreased to 5.5,  $\text{NH}_3$  is only  $\sim 0.02\%$  of total  
86 ammonium.  $\text{NH}_4^+$ , which is the bulk of total ammonium at both pH 7.4 and pH 5.5, can enter the  
87 cells if and only if the AmtB channel is expressed and functional. Its expression is controlled at the  
88 transcriptional level and is regulated largely by the free-pool concentration of glutamine in the cell  
89 interior (8), whereas its activity is controlled by the regulatory protein GlnK, largely in response to  
90 the free-pool concentration of the precursor metabolite 2-oxoglutarate, which is an intermediate  
91 in the tricarboxylic acid cycle ((11) and references cited therein). Expression of AmtB increases as  
92 the glutamine concentration declines and the channel is activated as the concentration of 2-  
93 oxoglutarate rises.

94

95 Within the cell, N is assimilated into glutamate, the organic precursor of most cellular nitrogen, by  
96 a high-affinity cycle and by a low-affinity enzyme (Fig. 1). The high-affinity cycle is constituted by  
97 the exquisitely regulated GS and by GOGAT. The low-affinity enzyme is biosynthetic (NADPH-  
98 dependent) GDH (8). Use of one mol of ATP per glutamate synthesized in the GS/GOGAT cycle  
99 drives assimilation of N even at extremely low concentrations of ammonium but is apparently  
100 detrimental when ammonium is abundant and energy is limiting (12). Both GS and GDH use  $\text{NH}_3$   
101 as their substrate because bond formation requires the lone pair of electrons on the N of  $\text{NH}_3$  (6,  
102 13). Hence,  $\text{NH}_4^+$  must be dissociated to  $\text{NH}_3 + \text{H}^+$  before either enzyme can use it. An equilibrium  
103 isotope effect associated with this spontaneous process leads to depletion of  $^{15}\text{N}$  in  $\text{NH}_3$  relative to

104  $\text{NH}_4^+$ . Its magnitude is  $-19.2\text{‰}$  (weighted mean, std. error =  $0.4\text{‰}$ ,  $n = 3$ ; (14)). That is, at  
105 equilibrium,  $^{15}\text{N}/^{14}\text{N}$  in  $\text{NH}_3$  is 19.2 parts per thousand lower than that in  $\text{NH}_4^+$ .

106

107 To control the site of rate limitation, we employed well-characterized mutant strains (Table 1).  
108 One such strain ( $\Delta\text{amtB}$ ) lacked AmtB. A second strain ( $\text{AmtB}\Delta\text{C-term}$ ) had a poorly active AmtB  
109 channel in which the normal membrane pores lacked the usual carboxy-terminal cytoplasmic  
110 extensions (15, 16). A third strain ( $\Delta\text{gdhA}::\text{kan}$ ) lacked GDH, the low-affinity ammonium-  
111 assimilatory enzyme. The last two strains ( $\text{gltD}::\text{kan}$  and  $\text{gltD}::\text{kan}\Delta\text{amtB}$ ) lacked GOGAT (Table 1;  
112 (8)). We did not employ a mutant lacking GS because this strain is auxotrophic for glutamine and  
113 requires that glutamine be added to the medium in high concentrations even in the presence of  
114 high concentrations of ammonium (17). We studied the mutant strains and their parental strain,  
115 which is a physiologically robust *E. coli* K12 wild-type (18, 19), under both ammonium-excess and  
116 limiting conditions. We determined doubling time, cell yield, residual ammonium in the medium,  
117 and the isotopic fractionation associated with incorporation of ammonium into cell material (see  
118 Materials and Methods). When the absence or alteration of one protein increased the doubling  
119 time of the strain (*i. e.*, decreased the growth rate), we could determine the rate-limiting step in  
120 transport or assimilation.

121

## 122 **Results**

123 *Parental wild-type strain.* The doubling time was 50 min. at all values of  $c_0$ , the initial external  
124 concentration of ammonia (*n. b.*,  $\text{NH}_3$ , not  $\text{NH}_3 + \text{NH}_4^+$ ; Table 2, Fig. 2A). For  $0.89 \leq c_0 \leq 280 \mu\text{M}$ ,  
125 measured isotopic fractionations ( $\epsilon_b$ ) ranged from  $-16.1$  to  $-23.8\text{‰}$  (mean and std. dev.  $-19.2 \pm$   
126  $2.6\text{‰}$   $n = 7$ ), with negative values of  $\epsilon$  indicating depletion of  $^{15}\text{N}$  in the biomass relative to the  
127 dissolved inorganic N in the medium (Table 2). At lower concentrations ( $c_0 \leq 200 \text{ nM}$ ), *i. e.*, at 1.0  
128 or 0.5 mM total ammonium and  $\text{pH} = 5.5$ , the isotopic fractionation decreased to  $-8.1$  or  $-5.4\text{‰}$ ,  
129 respectively (Table 2, Fig. 2B). AmtB is highly expressed even when the concentration of external

130  $\text{NH}_3$  is 5 to 10 times higher, namely  $1 \mu\text{M}$  ( $5 \text{ mM}$  total ammonium at pH 5.5; (11)), but its activity is  
131 not needed for optimal growth and it is inhibited by the regulatory protein GlnK (20–22). AmtB is  
132 not expressed when external  $\text{NH}_3$  is  $10 \mu\text{M}$  ( $0.5 \text{ mM}$  total ammonium at pH 7.4) or higher (11, 23).

133

134 *GDH<sup>-</sup> strain.* The GDH<sup>-</sup> strain ( $\Delta\text{gdhA}::\text{kan}$ ) lacks the low-affinity pathway for ammonium  
135 assimilation and hence is completely dependent on the high-affinity pathway for synthesis of  
136 glutamate and glutamine. Its doubling time was indistinguishable from that of wild type under all  
137 conditions of nitrogen availability and its isotopic fractionations were very similar to those of the  
138 wild-type strain (Table 2). For  $0.89 \leq c_0 \leq 280 \mu\text{M}$ ,  $\epsilon_b$  ranged from -18.8 to -25.4‰ (mean and std.  
139 dev.  $-21.3 \pm 2.6\text{‰}$ ,  $n = 7$ ). For  $c_0 = 89 \text{ nM}$ ,  $\epsilon_b$  decreased to -6.1‰. Because the ranges of  $\epsilon_b$  for cells  
140 having or lacking GDH overlap, these observations strongly support earlier reports that the GS-  
141 GOGAT cycle is the primary means for incorporating ammonium into biomass at all concentrations  
142 of  $\text{NH}_3$  (12, 24, 25).

143

144 *GOGAT<sup>-</sup> strains.* The GOGAT<sup>-</sup> strain ( $\text{gltD}::\text{kan}$ ) lacks the high-affinity ammonium-assimilatory cycle  
145 and depends on the linear, low-affinity pathway (biosynthetic, NADPH-dependent GDH) for  
146 synthesis of glutamate. GS converts approximately 12% of the glutamate product of GDH to  
147 glutamine to meet biosynthetic needs. When initial external concentrations of  $\text{NH}_3$  were decreased  
148 from 70 to  $7 \mu\text{M}$ , the doubling time of the strain increased from 50 to 65 min. (Table 2, Fig. S1A)  
149 and hence the activity of GDH [*E. coli*. has only a biosynthetic GDH (26)] was apparently rate-  
150 limiting for cell growth under these conditions. The weighted-mean isotopic fractionation  
151 was -11.2‰ (std. error =  $0.3\text{‰}$ ,  $n = 3$ ) at both concentrations of  $\text{NH}_3$  (Fig. S1D). This strain did not  
152 grow at all at  $c_0 \leq 1 \mu\text{M}$  (23).

153

154 In agreement with its dependence on the low-affinity enzyme, GDH, for synthesis of glutamate,  
155  $\text{gltD}::\text{kan}$  has an abnormally low internal-free-pool concentration of glutamate at low  $\text{NH}_3$  (27).

156 The *gltD::kan* strain also has an unusually high free-pool concentration of glutamine (27–29), the  
157 primary metabolic indicator of nitrogen sufficiency and hence the primary metabolic regulator of  
158 the transcriptional response to nitrogen availability (8, 27). This strain fails to express a number  
159 of proteins under the control of nitrogen regulatory protein C (NtrC), which is active at low  
160 internal concentrations of glutamine (8, 30).

161

162 Although we presumed that AmtB, which is one of the proteins controlled by NtrC, was poorly  
163 expressed (23, 31), we constructed a double-mutant strain (*gltD::kanΔamtB*) to be certain that  
164 AmtB was completely absent. The doubling time of the *gltD::kanΔamtB* strain was slightly longer  
165 than that of the *gltD::kan* strain but the isotopic fractionation was unchanged at -10.3‰ (Table 2,  
166 Fig. S1D).

167

168 *AmtB*- and *AmtB*-defective strains. Finally, the *AmtB*<sup>-</sup> strain ( $\Delta amtB$ ) lacks the NH<sub>4</sub><sup>+</sup> channel. For  
169 acquisition of ammonium it must depend on unmediated diffusion of NH<sub>3</sub> across the cytoplasmic  
170 membrane. At high external concentrations of NH<sub>3</sub>, both the doubling time of the  $\Delta amtB$  strain and  
171 its isotopic fractionation were identical to those of the wild-type and of the  $\Delta gdhA$  strain (Table 2).  
172 In fact, under these conditions, the wild type does not transcribe the *glnKamtB* operon (11, 23,  
173 31). When the external concentration of NH<sub>3</sub> was decreased to 1 μM, the  $\Delta amtB$  strain grew very  
174 slowly; its doubling time was initially 200 min. and growth slowed even more as the strain  
175 consumed ammonium and thereby decreased the external NH<sub>3</sub> concentration further (11, 23).  
176 Under these conditions, the weighted-mean isotopic fractionation was -30.1‰ (std. error = 0.6‰,  
177  $n = 2$ ), markedly larger than that observed in other strains.

178

179 For the strain in which the AmtB protein was modified at the carboxyl terminal (*AmtB* $\Delta C$ -term;  
180 (15, 16)), both the doubling time and isotopic fractionation at  $c_0 = 0.89 \mu\text{M}$  did not differ from  
181 those of the wild-type, GDH<sup>-</sup>, and *AmtB*<sup>-</sup> strains (Table 2). However, when  $c_0$  was decreased to 89

182 nM, AmtB $\Delta$ C-term had a doubling time of 110 min., more than twice as long as that of the wild  
183 type and approximately half as long as that of  $\Delta amtB$ . Under this condition, the activity of AmtB  
184 appeared to be rate-limiting for growth and the weighted-mean  $\epsilon_b$  was -17.6‰ (std. error  
185 = 0.3‰,  $n = 2$ ).

186

187 *Process-related summary of isotopic observations.* Table 2 includes 11 different observations of  $\epsilon_b$   
188 for cells equilibrating NH<sub>3</sub> by diffusion and having both the low- and high-affinity pathways for its  
189 assimilation (Fig. 1). These include seven wild-type cultures with  $0.89 \leq c_0 \leq 280 \mu\text{M}$ , three  $\Delta amtB$   
190 cultures with  $0.89 \leq c_0 \leq 70 \mu\text{M}$  and one AmtB $\Delta$ C-term strain with  $c_0 = 0.89 \mu\text{M}$ . Values of  $\epsilon_b$  range  
191 from -16.1 to -23.8‰. The weighted mean is -19.6‰ (std. error = 0.7‰). There were seven  
192 different observations for cells equilibrating NH<sub>3</sub> by diffusion and lacking the low-affinity GDH  
193 pathway for its assimilation. For these  $\Delta gdhA$  cultures, as for wild-type,  $0.89 \leq c_0 \leq 280 \mu\text{M}$ . Values  
194 of  $\epsilon_b$  range from -18.8 to -25.4‰. The weighted mean is -22.2‰ (std. error = 0.9‰). In total,  
195 there are 18 different observations of  $\epsilon_b$  for cells equilibrating NH<sub>3</sub> by diffusion and using the high-  
196 affinity GS-GOGAT cycle to incorporate ammonium N into organic molecules. Values of  $\epsilon_b$  range  
197 from -16.1 to -25.4‰. The weighted mean is -21.1‰ (std. error = 0.6). Variations of  $\epsilon_b$  are not  
198 correlated with  $c_0$  ( $r^2 = 0.13$ ). Three values of  $\epsilon_b$  were obtained for cells relying on the AmtB  
199 channel for transport of NH<sub>4</sub><sup>+</sup> and assimilating ammonium N *via* GS + GOGAT. In those cases, two  
200 wild-type cultures with  $c_0 = 0.089$  and  $0.18 \mu\text{M}$  and one  $\Delta gdhA$  culture with  $c_0 = 0.089 \mu\text{M}$ ,  $\epsilon_b$   
201 ranged from -5.4 to -8.1‰. In four cases, cells incorporated N using GS + GOGAT but either had no  
202 AmtB channel or an AmtB channel that was impaired by deletion of the C-terminal extensions.  
203 When AmtB was entirely absent (two cases,  $c_0 = 0.18$  and  $0.089 \mu\text{M}$ ), weighted-mean  $\epsilon_b$  was  
204 -30.1‰. When AmtB was only modified ( $c_0 = 0.089 \mu\text{M}$ , two cases), the weighted-mean  $\epsilon_b$  was  
205 -17.6‰. Finally, when cells lacked GOGAT (three cultures of  $\Delta gltD$  and one of  $\Delta gltDamtB::kan$ ),  
206 values of  $\epsilon_b$  varied from -10.3 to -11.5‰ with a weighted mean of -10.5‰ (std. error = 0.2‰).

207



208 **Discussion**

209 When external concentrations of  $\text{NH}_3$  exceed  $0.18 \mu\text{M}$ ,  $\text{NH}_3$  can rapidly equilibrate across the  
210 cytoplasmic membrane of many bacteria by unmediated diffusion and  $\text{NH}_4^+$  channels are not  
211 expressed (7, 11, 23, 32–34). The interior pool of  $\text{NH}_3$  (in equilibrium with intracellular  $\text{NH}_4^+$ ) has  
212 only one input, diffusion of  $\text{NH}_3$  from the external medium. Though it has three potential  
213 outputs—assimilation of  $\text{NH}_3$  by GS, assimilation of  $\text{NH}_3$  by GDH, and leakage from the cell—our  
214 results are not significantly affected by the absence of GDH (see Process-related summary), which  
215 reduces the outputs to two. The isotopic budget is summarized in Fig. 3A. Balancing the input  
216 against the outputs, we can write

217

218 [1]  $\delta_{\text{ae}} + \varepsilon_{\text{at}} = g(\delta_{\text{ai}} + \varepsilon_{\text{GS}}) + (1 - g)(\delta_{\text{ai}} + \varepsilon_{\text{at}})$

219

220 where the  $\delta$  values are defined in Fig. 3A,  $\varepsilon_{\text{at}}$  is the isotope effect associated with transport of  $\text{NH}_3$   
221 across the membrane, and  $g$  is the fraction of the input that is incorporated into biomass. When  
222 equilibration of  $\text{NH}_3$  across the membrane is rapid in comparison to the rate of assimilation,  $g \rightarrow 0$   
223 (*n. b.*,  $g$  is not the *amount* of  $\text{NH}_3$  that is assimilated but instead the *fraction* which is assimilated)  
224 and  $\delta_{\text{ae}} = \delta_{\text{ai}}$ . Because  $\delta_{\text{b}} = \delta_{\text{ai}} + \varepsilon_{\text{GS}}$ , the isotopic composition of the external  $\text{NH}_3$  is related to that of  
225 the cells by  $\delta_{\text{ae}} = \delta_{\text{b}} - \varepsilon_{\text{GS}}$ . Since essentially all external N is in the form of  $\text{NH}_4^+$ ,  $\delta_{\text{ae}} = \delta_{\text{e}} + \varepsilon_{\text{h}}$ , where  $\delta_{\text{e}}$   
226 is the measured isotopic composition of the N supplied to the medium and  $\varepsilon_{\text{h}}$  is the equilibrium  
227 isotope effect relating  $\text{NH}_4^+$  and  $\text{NH}_3$ . Finally, recalling that measured values of  $\varepsilon_{\text{b}}$  are equal to  
228  $\delta_{\text{b}} - \delta_{\text{e}}$ , we obtain  $\varepsilon_{\text{GS}} = \varepsilon_{\text{b}} - \varepsilon_{\text{h}}$ . As noted above and summarized in Table 3, 18 experiments yielded  
229  $\varepsilon_{\text{b}} \sim -21.1\text{‰}$ . The 95% confidence interval of that value overlaps with the 95% confidence interval  
230 for  $\varepsilon_{\text{h}}$ . Accordingly, there is no evidence for fractionation by GS.

231

232 An alternative interpretation, with  $g$  appreciably greater than zero, is not tenable. Specifically, the  
233 18 experiments yielding  $\varepsilon_{\text{b}} \approx \varepsilon_{\text{h}}$  could then be explained only if (i)  $g$ ,  $\varepsilon_{\text{GS}}$ , and  $\varepsilon_{\text{at}}$  happened in all

234 cases to have values satisfying the relationship  $g = \epsilon_{GS}/(\epsilon_{GS} - \epsilon_{at})$  and (ii)  $g$  was independent of the  
235 external concentration of  $\text{NH}_3$ . Additionally, a well documented experimental study of nitrogen  
236 and carbon isotope effects associated with glutamine synthetase from *E. coli* found a near-zero  
237 nitrogen isotope effect of  $-0.7 \pm 0.6\text{‰}$  (mean and standard deviation,  $n = 7$ , (35)).

238

239 It is unlikely that GS limits growth because it is synthesized in excess when supplies of  $\text{NH}_3$  are  
240 plentiful and its catalytic activity is regulated downwards by covalent modification (36–38). It is  
241 more likely that the flux of N into glutamate, the most plentiful intermediate in central nitrogen  
242 metabolism, is limited by the capacity of GOGAT. No isotopic fractionation is associated with  
243 GOGAT because practically all of the amide N in gln is transferred to 2-oxoglutarate to produce glu.

244

245 When external  $\text{NH}_3$  concentrations are below  $0.89 \mu\text{M}$ , *E. coli* K12 depends on the ammonium  
246 channel AmtB to maintain an optimal growth rate. Cells lacking this channel ( $\Delta\text{amtB}$ ) depend on  
247 uncatalyzed transport of  $\text{NH}_3$  across the membrane. At  $c_0 \leq 0.2 \mu\text{M}$ , growth of the  $\Delta\text{amtB}$  strain is  
248 extremely slow and the rate decreases as the external concentration of  $\text{NH}_3$  declines (Table 2, Fig.  
249 2A). The mass balance described by Eq. 1 applies but the conditions differ from those just  
250 discussed. Instead,  $g \neq 0$  and, because GS imposed no fractionation even when supplies of  $\text{NH}_3$   
251 were abundant, we know that  $\delta_{ai} = \delta_b$  and  $\epsilon_{GS} = 0$ . Making these substitutions and simplifying, we  
252 obtain  $\delta_{ae} = \delta_b - g\epsilon_{at}$ . Substituting  $\delta_{ae} = \delta_e + \epsilon_h$  and  $\epsilon_b = \delta_b - \delta_e$  leads to  $\epsilon_b = g\epsilon_{at} + \epsilon_h$ . At the limit in  
253 which growth is limited by transport of  $\text{NH}_3$  and  $g \rightarrow 1$ ,  $\epsilon_{at} = \epsilon_b - \epsilon_h$ . If the slowest growing cultures  
254 (experiments 27 and 1, Table 2) represent that case,  $\epsilon_{at} = -30.1 + 19.2 = -10.9 \pm 0.7\text{‰}$  (std. error  
255 from combining 0.5 and 0.4 in quadrature). This relatively large value suggests that transport of  
256  $\text{NH}_3$  across the membrane is limited by some process other than simple diffusion. Polar  
257 interactions within the membrane may play a role. Rishavy and Cleland (39) commented that the  
258 isotope effect in a related case “could easily be 2%” (*i. e.*,  $20\text{‰}$ , almost twice that estimated here).

259

260 For cells with a normal AmtB channel (wild-type or  $\Delta gdhA$  strain) and with  $c_0 = 0.089 \mu\text{M}$ ,  $\epsilon_b$  was  
 261 observed as low as  $-5.4 \pm 0.3\text{‰}$ , much lower than that of a strain lacking the channel ( $\Delta amtB$ ). The  
 262 corresponding mass balance is shown schematically in Fig. 3B and expressed mathematically in  
 263 Eq. 2.

264

$$265 \quad [2] \quad \delta_e + \epsilon_{ht} = g\delta_b + (1 - g)(\delta_b + \epsilon_{at})$$

266

267 Here,  $\epsilon_{ht}$  is the isotope effect associated with transport of  $\text{NH}_4^+$  by the AmtB channel and  
 268 substitutions introduced above have been adopted where appropriate ( $\delta_{he} = \delta_e$ ,  $\delta_{ai} = \delta_b$ ).

269 Simplifying gives  $\epsilon_b = \epsilon_{ht} - (1 - g)\epsilon_{at}$ . A recent quantitative study of AmtB function (11) indicates  
 270 that  $g$  is  $\sim 0.2$ . For  $\epsilon_b = -5.4\text{‰}$ , adopting  $\epsilon_{at} \approx -10.9\text{‰}$  (see above), we find  $\epsilon_{ht} = -14.1\text{‰}$ . If an  
 271 uncertainty of 0.05 is assigned to  $g$ , the estimated std. error of  $\epsilon_{ht}$  is  $0.7\text{‰}$ . Notably, the reduced  
 272 fractionation at low values of  $c_0$ , a condition that may be encountered in nature, derives not only  
 273 from fractionations associated with the AmtB channel but from an interplay between  
 274 fractionations associated with  $\epsilon_{ht}$  and  $\epsilon_{at}$ .

275

276 When the wild-type strain was grown with a slightly higher  $c_0 = 0.18 \mu\text{M}$ , the observed  
 277 fractionation increased to  $-8.1\text{‰}$ . The AmtB channel also functions at this  $\text{NH}_3$  concentration  
 278 because an AmtB<sup>-</sup> strain continues to grow suboptimally. Hence eq'n 2 applies. Using  $\epsilon_{ht} = -14.1 \pm$   
 279  $0.7\text{‰}$  and solving for  $g$ , we find  $g = 0.45 \pm 0.1$ . When AmtB functions, the internal ammonium  
 280 concentration is held constant and hence there is less leakage of  $\text{NH}_3$  at higher external  $\text{NH}_3$   
 281 concentrations (11).

282

283 For cells with an altered AmtB channel lacking the cytoplasmic C-terminal extensions (16),  $\epsilon_b =$   
 284  $-17.6 \pm 0.3\text{‰}$  (wt'd. mean and std. error) at  $c_0 = 0.089 \mu\text{M}$ . The fraction of N assimilated by GS is  
 285 0.5 (11) and solving as above yields  $\epsilon'_{ht} = -23.0 \pm 0.7\text{‰}$  (where a prime is used to denote the

286 altered channel). If the uncertainty assigned to  $g$  is doubled (to 0.1), the standard error increases  
287 only to  $\pm 1.2\%$ . Hence the isotope effect for the mutant AmtB channel is significantly larger than  
288 that for the wild-type channel. The mutant channel is known to lack coordination between the  
289 function of its individual monomers and to have other unusual properties (40).

290

291 To make *E. coli* dependent on the low-affinity pathway for assimilation of ammonium we  
292 inactivated GOGAT. This eliminates the GOGAT cycle and makes the organism dependent on  
293 biosynthetic GDH. At an external  $\text{NH}_3$  concentration of  $10 \mu\text{M}$ , GDH activity already limits the  
294 growth rate of the GOGAT<sup>-</sup> strain, and the strain does not grow at all at  $c_0 = 1 \mu\text{M}$ . The fractionation  
295 observed for the GOGAT<sup>-</sup> strains with  $c_0 \geq 7 \mu\text{M}$  is  $-10.5\%$ . Assuming that  $\text{NH}_3$  inside and outside  
296 the cell was in equilibrium, as for the wild-type,  $\Delta\text{gdhA}$ , and  $\Delta\text{amtB}$  strains at the same  
297 concentrations, it follows that internal  $\text{NH}_3$  was depleted in  $^{15}\text{N}$  by  $19.2\%$  relative to external  
298 dissolved inorganic N and, therefore, that the observation of  $\epsilon_b = -10.5\%$  requires inverse  
299 fractionation of  $^{15}\text{N}$  by GDH (*i. e.*, enrichment of the product relative to the reactant) with  
300  $\epsilon_{\text{GDH}} = 8.7 \pm 0.4\%$ . An inverse isotope effect has also been reported for bovine liver GDH (41).

301

302 *Conclusion.* Our studies of *E. coli* K12 have yielded *in vivo* isotope effects as summarized in Table 3.  
303 To our knowledge, the isotope effect for transport of  $\text{NH}_3$  is the first for a biological membrane. A  
304 previous measurement was made *in vitro* with a membrane filter (14).

305

306 That  $\epsilon_b$  for the  $\Delta\text{gdhA}$  strain is the same as that for the wild-type strain at all external  
307 concentrations of  $\text{NH}_3$  confirms the finding of Yuan et al. (24, 25) that the GOGAT cycle is the  
308 major means for assimilation of  $\text{NH}_3$  by *E. coli* K12 not only at low but also at high concentrations.  
309 Although the GOGAT cycle is widespread in bacteria and archaea (8, 42, 43), whereas the  
310 occurrence of biosynthetic GDH appears to be more restricted (44), several important examples of  
311 organisms naturally lacking GOGAT have recently come to light (45). Determining  $\epsilon_b$  for the

312 abundant ocean archaean *Nitrosopumilis maritima* will be particularly interesting not only  
313 because it lacks GOGAT and depends on GDH for ammonium assimilation but also because it  
314 oxidizes ammonium extracellularly as its primary energy source.

315

316 We hope that the present results will help future workers to correlate environmental genomic  
317 data with isotopic variations observed in nature.

318

319 *Comparison to earlier work.* Studying the  $\gamma$ -proteobacterium *Vibrio harveyi*, a close evolutionary  
320 relative of *E. coli* K12, Hoch *et al.* (46) also found that isotopic fractionation between external  
321 ammonium and cell material varied with external ammonium availability. Now, by using AmtB<sup>-</sup>  
322 strains, we have determined that the decrease in  $\epsilon_b$  from  $\sim -20\text{‰}$  to  $-4\text{‰}$  that they observed as  
323 external ammonium was dropped from an initial concentration of  $\sim 500$  to  $\sim 25$   $\mu\text{M}$  (pH 7.4) did, as  
324 they proposed, depend on the activity of an active ammonium channel. Specifically,  $\epsilon_b \sim -5\text{‰}$   
325 results from (i) acquisition of  $\text{NH}_4^+$  rather than  $\text{NH}_3$  by the growing cells, (ii) an isotope effect  
326 associated with transport of  $\text{NH}_4^+$  by the AmtB channel ( $\epsilon_{\text{ht}} \sim -14\text{‰}$ ), (iii) equilibration of  $\text{NH}_4^+$   
327 and  $\text{NH}_3$  inside the cell, (iv) an absence of isotopic fractionation during assimilation of  $\text{NH}_3$  by GS,  
328 and (v) leakage of  $^{15}\text{N}$ -depleted  $\text{NH}_3$  from the cells.

329

330 Given the very large  $\epsilon_b$  characteristic of *E. coli* AmtB<sup>-</sup> strains at low external  $\text{NH}_3$  ( $-30\text{‰}$ ), which  
331 appears to result from rate-limiting diffusion of  $\text{NH}_3$  across the cytoplasmic membrane, it is  
332 tempting to speculate that the fractionation observed in *V. harveyi* as the external concentration of  
333 ammonium decreased from 182 to 107  $\mu\text{M}$  in a single experiment [ $\epsilon_b = -26.5\text{‰}$ ; (46)] may  
334 indicate the precise range of external ammonium concentrations at which unmediated diffusion of  
335  $\text{NH}_3$  becomes limiting in wild-type *V. harveyi*, just prior to activation of its AmtB channel. In *E. coli*  
336 expression of AmtB occurs in response to a decrease in the internal-free-pool concentration of  
337 glutamine, whereas activation requires, in addition, an increase in the pool concentration of its

338 precursor metabolite 2-oxoglutarate (11). The latter occurs at lower external  $\text{NH}_3$  concentrations  
339 than the former. Activation requires release of the inhibitory gating protein GlnK (20–22).

340

341 Finally, we think that the decrease in  $\epsilon_b$  from -21 to -14‰, which Hoch *et al.* observed above 5 mM  
342 external ammonium, is an artifact of growth inhibition (doubling time increased from the optimal  
343 of 84 min. to 138 min. at high ammonium). Whatever the explanation, the  $\epsilon_b$  of -14‰ cannot be  
344 characteristic of GDH, as they proposed, because the NADH-dependent GDH they characterized is  
345 a catabolic enzyme. The genome sequence of *V. harveyi* is now known and it apparently lacks a  
346 biosynthetic, NADPH-dependent GDH. Moreover, we find that the biosynthetic, NADPH-dependent  
347 GDH of *E. coli* contributes very little to ammonium assimilation even at 20 mM ammonium, their  
348 highest concentration and ours.

349

## 350 **Materials and Methods**

351 **Bacterial Strains and Cultures.** NCM3722 (18) was the parental strain for all genetic mutant  
352 strains used in this work (Table 1). Additional details of strain construction are in SI Materials and  
353 Methods. For growth experiments, bacterial cultures were grown on the minimal medium of  
354 Neidhardt *et al.* (47) in MOPS buffered medium, pH 7.4, with 0.1% glucose as sole carbon source  
355 and  $\text{NH}_4\text{Cl}$  as sole nitrogen source. For experiments at pH 5.5, cultures were additionally adapted  
356 to low pH in minimal medium buffered with MES at pH 5.5. Growth and doubling time were  
357 determined by measuring optical density at 420 nm.

358

359 **Ammonia Assay.** Residual ammonium in cell-free supernatants was assayed in a GDH catalyzed  
360 reaction (AA0100 kit, Sigma). In the assay 2-oxoglutarate is reduced to L-glutamate by GDH using  
361 ammonium as substrate and NADPH as the cofactor providing reducing equivalents. Oxidation of  
362 NADPH is measured by a change in absorbance at 340 nm.

363

364 **Sample Preparation and Isotopic Analyses.** Bacterial cell samples were taken at various points  
365 during growth and were removed from the supernatant by high-speed centrifugation. The cell-  
366 free supernatant was frozen at -80°C for later measurement of residual ammonium, glucose, and  
367 final pH. The cells were washed twice in medium without additional glucose or ammonium, and  
368 dried in air overnight. Uniform amounts of 2 mg dry weight yielding 0.8 mg carbon and 0.2 mg  
369 nitrogen were transferred into pre-weighed tin capsules (part number 240-053-00, Costech  
370 Analytical Technologies, Inc.). Capsules containing only the reactant glucose and ammonium  
371 chloride used in the media were also prepared. All samples were analyzed at the UC Berkeley  
372 Center for Stable Isotope Biogeochemistry.  $\delta^{15}\text{N}$  and  $\%N$  were determined by using a PDZ Europa  
373 system consisting of an ANCA-NT carbon/nitrogen analyzer in combination with a 2020 mass  
374 spectrometer (48). The isotopic-abundance parameters are defined as follows:

375

$$376 \quad [3] \quad \delta^{15}\text{N} = 10^3 \left[ \left( \frac{{}^{15}R_{\text{sample}}}{{}^{15}R_{\text{standard}}} \right) - 1 \right]$$

377

378 Where  ${}^{15}R \equiv {}^{15}\text{N}/{}^{14}\text{N}$  and the isotopic standard for nitrogen is  $\text{N}_2$  in air, for which  ${}^{15}R = 0.0036765$   
379 (49). Values of  $\delta^{15}\text{N}$  express the relative difference between the isotope ratio in the sample and in  
380 the standard, expressed in parts per thousand (‰). A value of  $\delta^{15}\text{N} = -12.2\%$ , for example,  
381 indicates that  ${}^{15}R_{\text{sample}}$  is 0.0036316. The precision of the analyses, expressed as a standard  
382 deviation of a single observation and based on five pairs of duplicates and four sets of triplicates  
383 collected during the analyses (thus 13 degrees of freedom) is 0.06‰.

384

385 **Calculations.** The objective of the isotopic analyses is to determine  $\epsilon_b$ , the overall isotope effect  
386 associated with the assimilation of N. This is most simply expressed by the isotopic difference  
387 between the starting pool of inorganic N in the medium and the first increment of biomass formed  
388 following inoculation. In mathematical terms, the isotopic difference is expressed as a ratio of  
389 isotope ratios. Since the isotope ratios are very similar, a notation is used that expresses the

390 difference in terms of parts per thousand:

391

$$392 [4] \varepsilon_b \equiv 10^3 [(^{15}R_{e0}/^{15}R_{b0}) - 1]$$

393

394 where  $^{15}R_{e0}$  is the ratio of  $^{15}\text{N}$  to  $^{14}\text{N}$  in the initial medium and  $^{15}R_{b0}$  is the same ratio in the first  
395 increment of biomass. As growth proceeds, the measured isotopic compositions of both the  
396 medium and the biomass change as a result of preferential transfer of either  $^{15}\text{N}$  or  $^{14}\text{N}$  (depending  
397 on the sign of the isotope effect) from the medium to the biomass. Measurements of  $\varepsilon_b$  must take  
398 this into account.

399

400 Here, we employ the regression of  $\delta_b$  on  $[f/(1-f)] \cdot \ln f(1)$ , thus fitting the observations to a linear  
401 equation of the form:

402

$$403 [5] \delta_b = \delta_0 - [f/(1-f)] \varepsilon_b \cdot \ln f$$

404

405 where  $\delta_b$  is the measured  $\delta^{15}\text{N}$  of the biomass,  $\delta_0$  is the measured  $\delta^{15}\text{N}$  of the medium at  $t = 0$ , and  $f$   
406 is the fraction of ammonium unutilized. If, for example,  $\varepsilon_b = -18.8\text{‰}$ , it indicates that  $^{15}\text{N}$  is  
407 assimilated and used to produce biomass 18.8 parts per thousand more slowly than  $^{14}\text{N}$ .

408

409 Values of  $\delta_0$  vary between experiments, depending on the batch of  $\text{NH}_4\text{Cl}$  that was used. Specific  
410 values are, for experiments 1-5,  $3.25 \pm 0.19\text{‰}$  (mean and standard deviation,  $n = 8$ ); 6-15,  
411  $1.43 \pm 0.06\text{‰}$  ( $n = 3$ ); 16-17,  $1.12 \pm 0.19\text{‰}$  ( $n = 2$ ); 18-19,  $0.96 \pm 0.19\text{‰}$  ( $n = 2$ ); 20-21,  
412  $1.15 \pm 0.21\text{‰}$  ( $n = 2$ ); 22-24,  $1.36 \pm 0.09\text{‰}$  ( $n = 3$ ); and 25-29,  $0.91 \pm 0.08\text{‰}$  ( $n = 5$ ).

413

414 Uncertainties in  $\varepsilon_b$ , calculated from the variance about the regression and expressed as standard  
415 errors of the slope, are reported in Table 2 and range from 0.1 to 2.9‰. Where weighted means



416 are reported, the weighting factor is the inverse variance. The reported standard errors of  
417 weighted means are conventional or dispersion-corrected, whichever is greater. Uncertainties  
418 reported for calculated isotope effects are derived by conventional propagation of errors.

419

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426 former colleague, Wally van Heeswijk. Deceased at 49, Wally discovered GlnK, which is the most  
427 widespread regulatory protein known in the living world.

## References

1. Mariotti A et al. (1981) Experimental determination of nitrogen kinetic isotope fractionation: Some principles; illustration for the denitrification and nitrification processes. *Plant Soil* 62:413–430.
2. Stanley SM (2010) Relation of Phanerozoic stable isotope excursions to climate, bacterial metabolism, and major extinctions. *Proc Natl Acad Sci USA* 107:19185–9.
3. Valentine DL (2009) Isotopic remembrance of metabolism past. *Proc Natl Acad Sci USA* 106:12565–6.
4. Zhang X, Gillespie AL, Sessions AL (2009) Large D/H variations in bacterial lipids reflect central metabolic pathways. *Proc Natl Acad Sci USA* 106:12580–6.
5. Wohlhueter R, Schutt H, Holzer H (1973) in *The Enzymes of Glutamine Metabolism*, eds Prusiner S, Stadtman ER (Academic Press, New York), pp 45–64.
6. Raushel FM, Thoden JB, Holden HM (2003) Enzymes with molecular tunnels. *Acc Chem Res* 36:539–48.
7. Fong RN, Kim K-S, Yoshihara C, Inwood WB, Kustu S (2007) The W148L substitution in the Escherichia coli ammonium channel AmtB increases flux and indicates that the substrate is an ion. *Proc Natl Acad Sci USA* 104:18706–11.
8. Ikeda TP, Shauger AE, Kustu S (1996) Salmonella typhimurium apparently perceives external nitrogen limitation as internal glutamine limitation. *J Mol Biol* 259:589–607.
9. Hartl DL, Dykhuizen DE (1984) The population genetics of Escherichia coli. *Annu Rev Genet* 18:31–68.
10. Van Elsas JD, Semenov A V, Costa R, Trevors JT (2011) Survival of Escherichia coli in the environment: fundamental and public health aspects. *ISME J* 5:173–83.
11. Kim M et al. (2012) Need-based activation of ammonium uptake in Escherichia coli. *Mol Syst Biol* 8:616.
12. Helling RB (1994) Why does Escherichia coli have two primary pathways for synthesis of glutamate? *J Bacteriol* 176:4664–4668.
13. Hartman SC (1973) in *The Enzymes of Glutamine Metabolism*, eds Prusiner S, Stadtman ER (Academic Press, New York), pp 319–330.
14. Hermes JD, Weiss PM, Cleland WW (1985) Use of nitrogen-15 and deuterium isotope effects to determine the chemical mechanism of phenylalanine ammonia-lyase. *Biochemistry* 24:2959–2967.
15. Severi E, Javelle A, Merrick M (2007) The conserved carboxy-terminal region of the ammonia channel AmtB plays a critical role in channel function. *Mol Membr Bio* 24:161–71.
16. Inwood WB, Hall JA, Kim K-S, Fong R, Kustu S (2009) Genetic evidence for an essential oscillation of transmembrane-spanning segment 5 in the Escherichia coli ammonium channel AmtB. *Genetics* 183:1341–55.
17. Kustu SG, McFarland NC, Hui SP, Esmon B, Ames GF (1979) Nitrogen control of Salmonella typhimurium: co-regulation of synthesis of glutamine synthetase and amino acid transport systems. *J Bacteriol* 138:218–234.

18. Lyons E, Freeling M, Kustu S, Inwood W (2011) Using genomic sequencing for classical genetics in *E. coli* K12. *PLoS ONE* 6:1–16.
19. Soupene E et al. (2003) Physiological Studies of *Escherichia coli* Strain MG1655: Growth Defects and Apparent Cross-Regulation of Gene Expression. *J Bacteriol* 185:5611–5626.
20. Van Heeswijk WC et al. (1995) An additional PII in *Escherichia coli*: a new regulatory protein in the glutamine synthetase cascade. *FEMS Microbiol Lett* 132:153–7.
21. Javelle A, Severi E, Thornton J, Merrick M (2004) Ammonium sensing in *Escherichia coli*. Role of the ammonium transporter AmtB and AmtB-GlnK complex formation. *The Journal of biological chemistry* 279:8530–8.
22. Conroy MJ et al. (2007) The crystal structure of the *Escherichia coli* AmtB-GlnK complex reveals how GlnK regulates the ammonia channel. *Proc Natl Acad Sci USA* 104:1213–8.
23. Soupene E, He L, Yan D, Kustu S (1998) Ammonia acquisition in enteric bacteria: Physiological role of the ammonium/methylammonium transport B (AmtB) protein. *Proc Natl Acad Sci USA* 95:7030–7034.
24. Yuan J, Bennett BD, Rabinowitz JD (2008) Kinetic flux profiling for quantitation of cellular metabolic fluxes. *Nat Protoc* 3:1328–40.
25. Yuan J, Fowler WU, Kimball E, Lu W, Rabinowitz JD (2006) Kinetic flux profiling of nitrogen assimilation in *Escherichia coli*. *Nat Chem Biol* 2:529–30.
26. Blattner FR (1997) The Complete Genome Sequence of *Escherichia coli* K-12. *Science* 277:1453–1462.
27. Csonka LN, Ikeda TP, Fletcher S a, Kustu S (1994) The accumulation of glutamate is necessary for optimal growth of *Salmonella typhimurium* in media of high osmolality but not induction of the proU operon. *J Bacteriol* 176:6324–33.
28. Yan D, Ikeda TP, Shauger AE, Kustu S (1996) Glutamate is required to maintain the steady-state potassium pool in *Salmonella typhimurium*. *Proc Natl Acad Sci USA* 93:6527–6531.
29. Yan D (2007) Protection of the glutamate pool concentration in enteric bacteria. *Proc Natl Acad Sci USA* 104:9475–80.
30. Pahel G, Zelenetz AD, Tyler BM (1978) gltB gene and regulation of nitrogen metabolism by glutamine synthetase in *Escherichia coli*. *J Bacteriol* 133:139–148.
31. Van Heeswijk WC et al. (1996) An alternative PII protein in the regulation of glutamine synthetase in *Escherichia coli*. *Mol Microbiol* 21:133–146.
32. Boussiba S, Dilling W, Gibson J (1984) Methylammonium transport in *Anacystis nidulans* R-2. *J Bacteriol* 160:204–210.
33. Andrade SLA, Einsle O (2007) The Amt/Mep/Rh family of ammonium transport proteins (Review). *Mol Membr Bio* 24:357–365.
34. Loqué D, Von Wirén N (2004) Regulatory levels for the transport of ammonium in plant roots. *Journal of experimental botany* 55:1293–305.

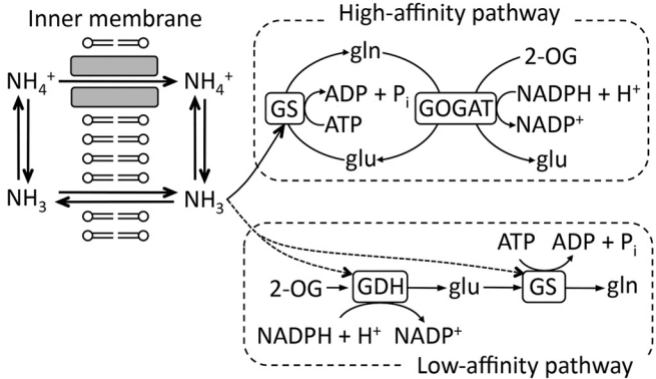
35. Stoker PW (1994) Heavy-atom isotope effects for enzymes of glutamine metabolism. *PhD thesis (The University of Nebraska - Lincoln)*. Available at: <http://digitalcommons.unl.edu/dissertations/AAI9507828> .
36. Jiang P, Ninfa AJ (2011) A source of ultrasensitivity in the glutamine response of the bicyclic cascade system controlling glutamine synthetase adenylation state and activity in *Escherichia coli*. *Biochemistry* 50:10929–40.
37. Okano H, Hwa T, Lenz P, Yan D (2010) Reversible adenylation of glutamine synthetase is dynamically counterbalanced during steady-state growth of *Escherichia coli*. *J Mol Biol* 404:522–36.
38. Rhee SG, Chock PB, Stadtman ER (1989) Regulation of *Escherichia coli* glutamine synthetase. *Adv Enzymol Relat Areas Mol Biol* 62:37–92.
39. Rishavy MA, Cleland WW (1999)  $^{13}\text{C}$ ,  $^{15}\text{N}$ , and  $^{18}\text{O}$  equilibrium isotope effects and fractionation factors. *Canadian Journal of Chemistry* 77:967–977.
40. Inwood WB et al. (2009) Epistatic effects of the protease/chaperone HflB on some damaged forms of the *Escherichia coli* ammonium channel AmtB. *Genetics* 183:1327–40.
41. Weiss PM, Chen CY, Cleland WW, Cook PF (1988) Use of primary deuterium and  $^{15}\text{N}$  isotope effects to deduce the relative rates of steps in the mechanisms of alanine and glutamate dehydrogenases. *Biochemistry* 27:4814–22.
42. Dincturk HB, Cunin R, Akce H (2011) Expression and functional analysis of glutamate synthase small subunit-like proteins from archaeon *Pyrococcus horikoshii*. *Microbiol Res* 166:294–303.
43. Dincturk HB, Knaff DB (2000) The evolution of glutamate synthase. *Mol Biol Rep* 27:141–148.
44. Hudson RC, Daniel RM (1993) L-glutamate dehydrogenases: distribution, properties and mechanism. *Comp Biochem Physiol B* 106:767–92.
45. Walker CB et al. (2010) *Nitrosopumilus maritimus* genome reveals unique mechanisms for nitrification and autotrophy in globally distributed marine crenarchaea. *Proc Natl Acad Sci USA* 107:8818–23.
46. Hoch M, Fogel M, Kirchman D (1992) Isotope fractionation associated with ammonium uptake by a marine bacterium. *Limnol Oceanogr* 37:1447–1459.
47. Neidhardt FC, Bloch PL, Smith DF (1974) Culture Medium for Enterobacteria. *J Bacteriol* 119:736–747.
48. Brooks PD, Geilmann H, Werner RA, Brand WA (2003) Improved precision of coupled  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  measurements from single samples using an elemental analyzer/isotope ratio mass spectrometer combination with a post-column six-port valve and selective  $\text{CO}_2$  trapping; improved halide robustness of the co. *Rapid Commun Mass Spectrom* 17:1924–6.
49. Coplen TB, Krouse HR, Bohlke JK (1992) Reporting of nitrogen-isotope abundances. *Pure Appl Chem* 64:907–908.

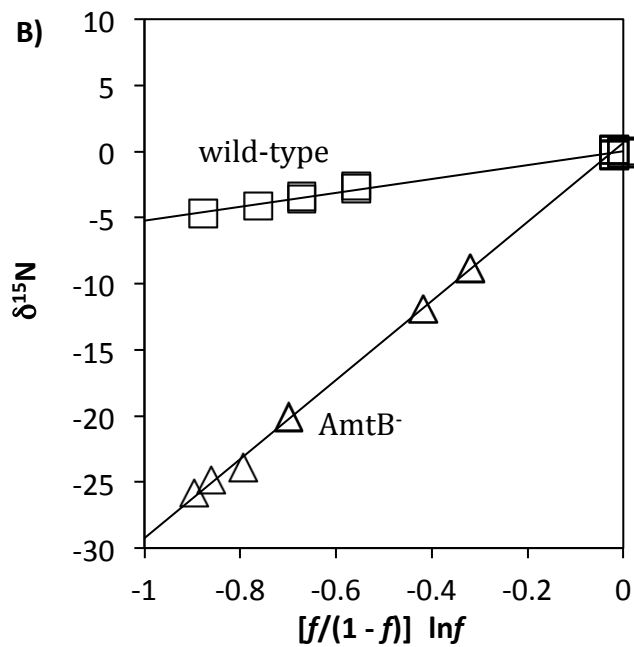
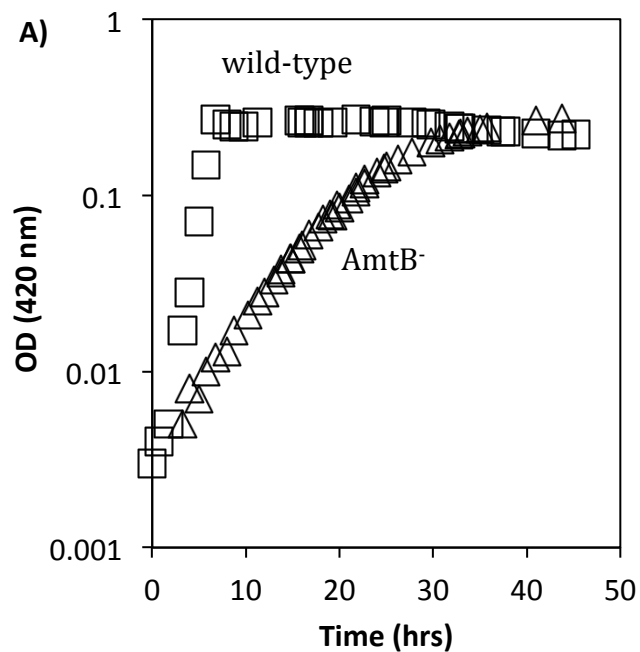
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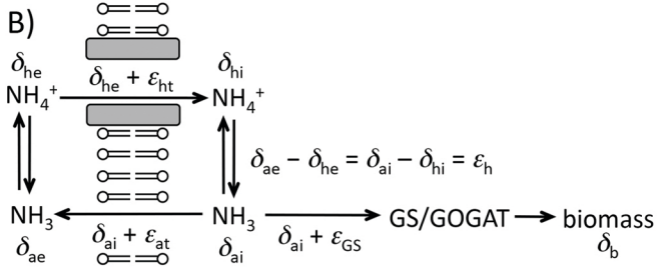
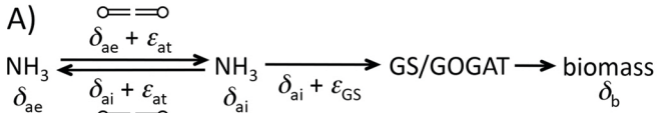
**Fig. 1.** Schematic diagram of nitrogen assimilation from ammonium by *E. coli* K12.

**Fig. 2.** Growth (A) and (B) determination of  $\epsilon_b$  for assimilation of external ammonium ( $c_0 = 0.089 \mu\text{M NH}_3$ , 0.5 mM total  $\text{NH}_4\text{Cl}$ , 0.1% glucose, pH 5.5) into biomass in wild-type (squares) and AmtB<sup>-</sup> strains (triangles). Cultures in panel B were different from those in panel A, which were sampled more frequently.

**Fig. 3.** Flows of N across the cell membrane and within cells. Isotopic compositions are denoted by  $\delta$  terms and isotope effects are denoted by  $\epsilon$  terms. In subscripts, a designates  $\text{NH}_3$ ; b, biomass; e, external; GS, glutamate synthetase; h,  $\text{NH}_4^+$  or protonation; i, internal; and t, transport. Isotopic compositions of the N fluxes are then given by expressions such as  $\delta_{he} + \epsilon_{ht}$ , which indicates the isotopic composition of the ammonium being transported by the AmtB channel. (A) Cells not utilizing the AmtB channel to actively transport  $\text{NH}_4^+$ . (B) Cells in which the AmtB channel is used.









**Table 1. Bacterial strains\***

Strain: Genotype	Phenotype
NCM3722: <i>E. coli</i> K12 wild-type	wild-type
NCM4199: <i>amtB</i> , <i>tesB</i> ::kan	AmtB $\Delta$ C-term <sup>†</sup>
NCM4453: <i>gltD</i> ::kan	GOGAT <sup>‡</sup>
NCM4454: $\Delta$ <i>gdhA</i> ::kan	GDH <sup>‡</sup>
NCM4590: $\Delta$ <i>amtB</i>	AmtB <sup>-</sup>
NCM4701: <i>gltD</i> ::kan $\Delta$ <i>amtB</i>	GOGAT <sup>-</sup> , AmtB <sup>-</sup>

\*All strains were constructed in the background of a physiologically robust *E. coli* K12 wild-type strain, NCM3722 ((18, 19); see Materials and Methods).

<sup>†</sup> Residues from position 382 onward were deleted (15, 16).

<sup>‡</sup> Dalai Yan, Indiana University School of Medicine.

**Table 2. Summary of cultures and measured isotopic fractionations**

Exp.	Strain*	$c_0$ (NH <sub>3</sub> ) <sup>†</sup> $\mu$ M	D.T. <sup>‡</sup> min.	$n$ <sup>§</sup>	$\epsilon_b$ <sup>¶</sup> ‰
17	NCM3722	280	50	8	-20.4 $\pm$ 2.1
23	NCM3722	280	50	7	-16.8 $\pm$ 1.9
16	NCM3722	140	50	8	-16.1 $\pm$ 0.7
22	NCM3722	140	50	7	-18.2 $\pm$ 1.1
8	NCM3722	70	50	9	-19.1 $\pm$ 1.4
10	NCM3722	7	50	5	-23.8 $\pm$ 1.4
5	NCM3722	0.89	50	6	-20.0 $\pm$ 1.0
26	NCM3722	0.18	50	6	-8.1 $\pm$ 0.3
2	NCM3722	0.089	50	5	-5.4 $\pm$ 0.3
9	NCM4590	70	50	9	-19.9 $\pm$ 1.7
11	NCM4590	7	50	4	-22.2 $\pm$ 2.9
3	NCM4590	0.89	50	8	-20.3 $\pm$ 0.5
27	NCM4590	0.18	>100	5	-30.2 $\pm$ 0.7
1	NCM4590	0.089	>200	6	-29.9 $\pm$ 0.9
18	NCM4454	280	50	9	-22.1 $\pm$ 0.7
21	NCM4454	280	50	8	-19.4 $\pm$ 2.2
19	NCM4454	140	50	7	-25.4 $\pm$ 2.9
20	NCM4454	140	50	8	-18.8 $\pm$ 0.5
15	NCM4454	70	50	7	-19.3 $\pm$ 1.4
14	NCM4454	7	50	4	-20.4 $\pm$ 1.4
6	NCM4454	0.89	50	8	-23.9 $\pm$ 0.4
7	NCM4454	0.089	50	5	-6.1 $\pm$ 0.6
13	NCM4453	70	50	9	-10.3 $\pm$ 1.0
12	NCM4453	7	65	5	-11.2 $\pm$ 0.4
29	NCM4453	7	65	5	-11.5 $\pm$ 0.5
28	NCM4701	7	75	5	-10.3 $\pm$ 0.1
24	NCM4199	0.89	50	8	-23.3 $\pm$ 1.2
25	NCM4199	0.089	110	4	-17.2 $\pm$ 0.6
30	NCM4199	0.089	110	4	-17.8 $\pm$ 0.4

\*Strain number (for phenotype see Table 1).

<sup>†</sup>Concentrations of NH<sub>3</sub> determined from total concentrations of NH<sub>4</sub>Cl, which were 71-fold higher than that of NH<sub>3</sub> at pH 7.4 and 5620-fold higher at pH 5.5. Cultures with  $c_0 \leq 0.89$  were grown at pH 5.5.

<sup>‡</sup>Doubling time.

<sup>§</sup>Number of points used in the determination of  $\epsilon_b$ .

<sup>¶</sup>Reported uncertainty is the standard error of the slope derived from the regression described in the section on calculations.

**Table 3. Process-related isotope effects**

Process	Related Experiments	Result*
1. Assimilation of NH <sub>3</sub> by GS For $g \rightarrow 0$ $\varepsilon_{GS} = \varepsilon_b - \varepsilon_h$	Wild type, $c_0 \geq 0.89 \mu\text{M}$ , $n = 7$ AmtB <sup>-</sup> , $c_0 \geq 0.89 \mu\text{M}$ , $n = 3$ AmtB $\Delta$ C-term, $c_0 = 0.89 \mu\text{M}$ , $n = 1$ GDH <sup>-</sup> , $c_0 \geq 0.89 \mu\text{M}$ , $n = 7$	95% Confidence intervals: $\bar{\varepsilon}_b = -21.1 \pm 1.3\text{‰}$ $\bar{\varepsilon}_h = -19.2 \pm 1.7\text{‰}$ $\varepsilon_{GS} \sim 0$
2. Transmembrane transport of NH <sub>3</sub> For $g \rightarrow 1$ $\varepsilon_{at} = \varepsilon_b - \varepsilon_h$	AmtB <sup>-</sup> , $c_0 < 0.2 \mu\text{M}$ , $n = 2$	$\bar{\varepsilon}_b = -30.1 \pm 0.5\text{‰}$ $\varepsilon_{at} = -10.9 \pm 0.7\text{‰}$
3. Transport of NH <sub>4</sub> <sup>+</sup> by AmtB $\varepsilon_{ht} = \varepsilon_b + (1 - g)\varepsilon_{at}$	Wild type, $c_0 = 0.089 \mu\text{M}$ , $n = 1$ GDH <sup>-</sup> , $c_0 = 0.089 \mu\text{M}$ , $n = 1$	$\bar{\varepsilon}_b = -5.5 \pm 0.3\text{‰}$ for $g = 0.2 \pm 0.05$ , $\varepsilon_{ht} = -14.1 \pm 0.8\text{‰}$
4. Transport of NH <sub>4</sub> <sup>+</sup> by altered AmtB $\bar{\varepsilon}'_{ht} = \varepsilon_b + (1 - g)\varepsilon_{at}$	AmtB $\Delta$ C-term, $c_0 = 0.089 \mu\text{M}$ , $n = 2$	$\bar{\varepsilon}_b = -17.6 \pm 0.3\text{‰}$ for $g = 0.5 \pm 0.1$ , $\bar{\varepsilon}'_{ht} = -23.1 \pm 1.2\text{‰}$
5. Assimilation of NH <sub>3</sub> by GDH $\varepsilon_{GDH} = \varepsilon_b - \varepsilon_h$	GOGAT <sup>-</sup> , AmtB <sup>-</sup> , $c_0 = 7 \mu\text{M}$ , $n = 1$ GOGAT <sup>-</sup> , $c_0 \geq 7 \mu\text{M}$ , $n = 3$	$\bar{\varepsilon}_b = -10.5 \pm 0.2\text{‰}$ $\varepsilon_{GDH} = 8.8 \pm 0.4\text{‰}$

\* $\bar{\varepsilon}$  denotes weighted mean. Indicated uncertainties are standard errors except for process 1, where 95% confidence intervals are specified.