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1 **First-principles study on hydrogen diffusivity in BCC, FCC, and HCP iron**

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16 **Abstract**

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18 The hydrogen diffusion behavior in BCC, FCC, and HCP iron has been investigated by means of first-principles
19 calculations. Diffusion coefficients were estimated quantitatively from the migration energy derived by the Nudged
20 elastic band method, and phonon calculations including the vibrations of all atoms at every stable and metastable site.
21 Our calculations on the BCC structure show good agreement with those in the previous report. In the FCC structure
22 as well, the calculated diffusion coefficients are in good agreement with experimental data. Our results suggest that
23 the consideration of the antiferromagnetic state in FCC is important for the reproduction of experimental results. For
24 the HCP structure, although there was a lack of systematic experimental results, our calculations predict that the
25 diffusion coefficient is smaller than that in the case of the FCC sample. In the HCP lattice, there are two diffusion
26 paths: one parallel to the c -axis and the other in the c -plane. The direction and the diffusion coefficient can be
27 controlled by the tuning of c/a , which is the ratio of the lattice constants.
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32 **1. Introduction**

33 Hydrogen is being increasingly used as a clean energy source that reduces the amount of carbon dioxide emitted
34 into the atmosphere. Therefore, hydrogen-resistant structure materials have been sought for hydrogen-energy-related
35 infrastructure. In this regard, steels are advantageous with respect to the strength and cost performance. Key factors
36 affecting mechanical performance of steels under hydrogen environment are hydrogen content, distribution, and
37 diffusivity. In particular, the diffusivity of hydrogen drastically alters fatigue properties which are the determinant
38 factor for safety structure design. From a viewpoint of hydrogen diffusivity, close-packed structures are believed to
39 show a low diffusivity of hydrogen, i.e. FCC and HCP structures. In steels, austenitic steels actually show a low
40 diffusivity [1-3], which results in high resistance to fatigue crack growth in hydrogen environments [4].

41 On the other hand, there is few systematic knowledge about hydrogen diffusivity for another promising candidate,
42 HCP structure. Because stable and single HCP phase cannot be obtained under ordinary pressure. Instead,
43 deformation-induced HCP-martensitic transformation from austenitic phase was utilized to obtain high resistance to
44 fatigue crack growth in Fe-Mn-Cr-Ni steel, which successfully endowed comparable fatigue crack growth resistance
45 to stable austenitic steels [5]. Correspondingly, some indirect evidences of low diffusivity of HCP-martensite has
46 been reported [6,7]. The utilization of HCP martensite is expected to give a new pathway for alloy design of
47 hydrogen-resistant steels. However, because the effect of lattice defects including its interface on diffusivity cannot
48 be quantified, experimental methodologies such as permeation test cannot determine underlying diffusivity along
49 interstitial sites of HCP. Furthermore, the c/a and volume of HCP change sensitively by composition of alloys and
50 constraint condition with parent phase. Thus, designing HCP-based hydrogen-resistant materials needs systematic
51 and atomic-level insight revealed from the computational simulations.

52 From a theoretical point of view, the diffusion behavior of hydrogen has been studied using several simulation
53 techniques [8-13]. First-principles calculations based on the density functional theory (DFT) have been used to
54 determine the energetics associated with the migration of interstitial H atoms in crystalline metals. Since hydrogen
55 is the lightest chemical element, its diffusion is affected by quantum mechanical effects. Quantum mechanical effects
56 on the diffusion of interstitial atoms can be divided into two contributions. The first is due to the zero point energy
57 (ZPE) of the ground state of a vibrating proton or, more generally, due to the presence of discrete vibrational energy
58 levels and their occupied state at finite temperatures. The second contribution is due to quantum mechanical tunneling.

59 Kimizuka *et al.*[9] and Stefano *et al.*[10] have pointed out that it is difficult to explain the diffusivity of hydrogen
60 with accuracy in BCC materials using the transition state theory (TST) since it cannot account for the quantum effect
61 become prominent at low temperature. On the other hand, the diffusion coefficient calculated using the TST is in
62 good agreement with experimental results for Ni and Ti that have the FCC and HCP closed packed structures,
63 respectively[10,12]. This observation could be due to the quantum effect being negligible in those metals. Therefore,
64 an investigation based on the TST in FCC and HCP irons would be insightful. Comparison of the hydrogen diffusivity
65 in BCC, FCC, and HCP, missing piece of the study, make a significant contribution toward materials design for
66 hydrogen-resistant steels.

67 In this study, we have investigated the hydrogen diffusion behavior in BCC, FCC, and HCP iron by means of first-
68 principles calculations. Our calculation results for the BCC iron show good agreement with those of previous reports
69 [9,10]. In the FCC iron, the calculated diffusion coefficients are compared with the experimental data. Our

70 calculations predict that the diffusion coefficient of the HCP sample which has several c/a is smaller than that of the
71 FCC sample. In the HCP lattice, there are two diffusion paths: one parallel to the c -axis and the other in the c -plane.
72 The direction and the diffusion coefficient can be controlled by the tuning of c/a , which is the ratio of the lattice
73 constants. The low and direction controllability of hydrogen diffusivity in HCP iron suggests a different paradigm in
74 steels under hydrogen environment.

75 2. Computational details

76 Following Wert and Zener [14] the interstitial diffusion coefficient is described as

$$77 D = nL^2\Gamma \quad (1)$$

78 where n is a numerical coefficient, whose value depends on the location of the interstitial positions, L and Γ are the
79 jump distance projected onto the diffusion direction and the jump rate, respectively. In the Kehr approximation [15],
80 the jump rate can be written as follows:

$$81 \Gamma = \frac{k_B T}{h} \exp\left(-\frac{\Delta E}{k_B T}\right) \quad (2)$$

82 where ΔE is the migration energy in the diffusion pathway. k_B , T and h indicate the Boltzmann constant, temperature
83 and Planck constant, respectively. The simplest extension of this classical expression taking into account zero-point
84 energy, the temperature-dependent part of the internal energy and the vibrational entropy is usually done by simply
85 adding the difference in vibrational free energies, ΔG_{vib} , corresponding to the saddle point and the initial state, to
86 the migration energy in the exponential term in Eq.(2). The corrected diffusivity is hence expressed as

$$87 D = nL^2 \frac{k_B T}{h} \exp\left(-\frac{\Delta E + \Delta G_{\text{vib}}}{k_B T}\right) \quad (3)$$

88 where

$$89 \Delta G_{\text{vib}} = G_{\text{vib}}^{\text{saddle}} - G_{\text{vib}}^{\text{interstitial site}} \quad (4).$$

90 The vibrational free energy G_{vib} in several states is explained as

$$91 G_{\text{vib}} = -k_B T \int_0^\infty g(\nu) \ln\left[2 \sinh\left(\frac{h\nu}{2k_B T}\right)\right] d\nu \quad (5)$$

92 where $g(\nu)$ is the phonon density of states and ν is vibrational frequency.

93 In order to calculate ΔE for hydrogen diffusion in iron, the climbing image nudged elastic band method[16] (CI-
94 NEB) was used. The NEB method provides the means to find a minimum energy path (MEP) when the initial and
95 final states of a process are known [17]. The CI-NEB method is an improved version of the NEB method, and it is
96 used to locate the MEP and transition state for the diffusion of hydrogen in bulk Fe. We used the VTST code for the
97 Vienna Ab initio Simulation Package (VASP) code [18,19] developed by Henkelman *et al.* [16]. The supercells of
98 Fe_{54}H ($3 \times 3 \times 3$), Fe_{36}H ($2 \times 2 \times 2$), and Fe_{96}H ($4 \times 4 \times 3$) were used for BCC, FCC, and HCP, respectively in the
99 CI-NEB calculation. We set linear interpolation images between the initial and final states with three or five images,
100 which were converged to the MEP with a $3 \times 3 \times 3$ mesh for k-point sampling in BCC and HCP. We used a $5 \times 5 \times$
101 5 mesh for k-point sampling with respect to FCC.

102 The lattice relaxation and total energy calculations were performed with the VASP code, which is based on DFT.
103 The exchange and correlation functions were given by the generalized gradient approximation, as proposed by
104 Perdew *et al.*[20]. We employed Blochl's projector-augmented wave (PAW) method as implemented by Kresse and
105 Joubert [21, 22]. The Monkhorst-Pack method [23] is used to set the k-point mesh and a cutoff of 334.9 eV for the
106 plane waves was applied. Before the NEB calculations were performed, structure optimizations were performed with
107 a $13 \times 13 \times 13$ mesh for k-point sampling with a cutoff energy of 348.247 eV regarding BCC and FCC. In the
108 calculation of HCP, we used a $15 \times 15 \times 9$ mesh for k-point sampling.

109 In this study, the magnetic properties were assumed to be ferromagnetic in BCC, antiferromagnetic in FCC, and
110 nonmagnetic in HCP samples. In the FCC iron, magnetic states have been investigated and it was suggested that the
111 antiferromagnetic state was stable [24-26]. However, the double-layered antiferromagnetic structure [25] causes the
112 lattice symmetry to change from cubic to tetragonal. Therefore, these two lattice types, namely, antiferromagnetic
113 FCC and antiferromagnetic FCT, were generated in the following manner. First, the lattice parameters were optimized
114 in the nonmagnetic FCC and HCP. The optimized lattice constants were $a = 3.44 \text{ \AA}$ in FCC and $a = 2.46$ and $c = 3.89$
115 \AA in HCP, respectively. These values are consistent with those reported in previous works[8,10,24,26]. The FCC
116 structure, which was optimized for the antiferromagnetic state, resulted in a tetragonal structure, which had $a = 3.46$
117 and $c = 3.75 \text{ \AA}$. The difference in the equilibrium volume between the antiferromagnetic FCT and nonmagnetic FCC
118 is estimated to be 10.3 % , which is expected to affect the diffusivity of hydrogen. Hence, we calculated the migration
119 energy of the antiferromagnetic FCT and FCC, which have the same equilibrium volume, by the NEB method. On
120 the other hand, although HCP has been calculated as antiferromagnetic state[26], the difference in the equilibrium
121 volume between the antiferromagnetic and nonmagnetic state was only 2.7%, which is comparatively lower than that
122 of FCC. Therefore, we have calculated the migration energy of hydrogen in the HCP in the nonmagnetic state since
123 the effect of the volume change is expected to be low.

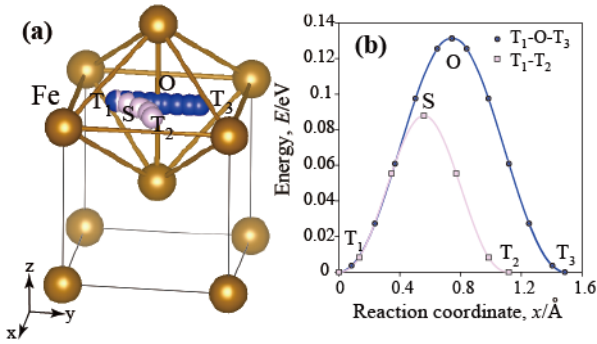
124 Thermodynamic functions are obtained from phonon dispersions, which are calculated by the direct method [27]
125 with ab initio forces from VASP. To estimate the force due to displacement within the DFT calculation, we used the
126 plane wave energy cut off of 267.882eV, the convergence criterion for electric self-consistency loop of 10^{-6} eV, a Γ
127 point centered k-mesh of $2 \times 2 \times 2$, the Methfessel-Paxton smearing method with the width of the smearing in 0.2
128 eV, and a displacement of 0.01 \AA .

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147 **3. Results and discussion**

148 **3.1. BCC**

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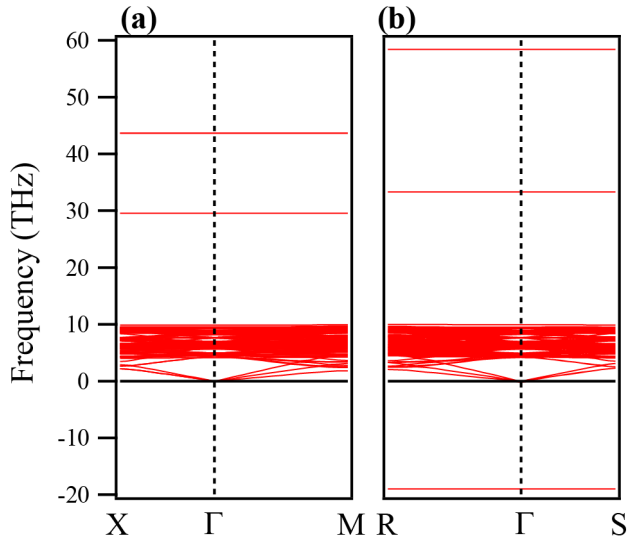
150 Fig.1. (a) The hydrogen migration pathways in BCC. O, T, and S represent the octahedral site, tetrahedral site, and
151 saddle point, respectively. (b) The energy profiles for two paths: T₁-T₂ direct and T₁-O-T₃ indirect.

152

153 First, we have examined our calculated results for BCC and compared them with those reported in previous
154 studies[8,10]. Figs.1 (a) and (b) show the migration pathways of the hydrogen atom in the BCC and the energy
155 profiles for T₁-T₂ direct and T₁-O-T₃ indirect hopping, respectively. The minimum energy path along T₁-O-T₃ is
156 almost straight, while along T₁-T₂ it seems to curve towards the O-site. This behavior is consistent with those reported
157 in a previous paper[8]. From Fig.1(b), it is revealed that the T₁-T₂ direct hopping is a more stable transition path than
158 T₁-O-T₃ indirect hopping. The “bare” migration energy ΔE can be estimated to be 0.088 eV. The calculated phonon
159 dispersions of several states are given in Figs. 2. In fig.2(a), two branches at 29.6 and 43.7 THz correspond to the
160 vibration of the hydrogen atom at tetrahedral interstitial site. The branches of hydrogen vibration derived from
161 asymmetry of BCC lattice with hydrogen in T-site. The results have good agreement with the calculated values
162 reported by Sakagami *et al.*[28]. From the Fig.2(b) we see that the hydrogen has one imaginary mode of vibrational
163 frequency at the transition state of T₁-T₂. The imaginary mode means that there is dynamical instability at the site.
164 The stable transition state for the diffusion has only one imaginary mode. Therefore, the S-site is a stable transition
165 state. Furthermore, we estimated that the ΔZPE was 0.053 eV from these energy frequencies. The ZPE-corrected
166 migration energy was 0.035 eV, which is in agreement with 0.042eV and 0.044eV presented in previous reports[8,10].

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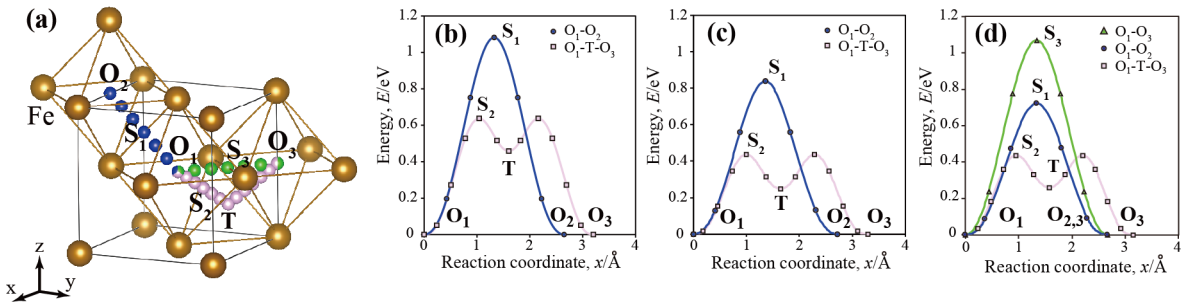
170 Fig. 2. The calculated phonon dispersions of a 56 atom-Fe supercell with hydrogen atom in (a) T-site and (b) the
 171 saddle (S) point of T_1 - T_2 , respectively. The symmetry of the supercell describing the transition state is different
 172 from that of the tetrahedral site, which is reflected in the labeling of the special points in the Brillouin zone.

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175 3.2. FCC

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178 Fig.3. (a) The hydrogen migration pathways in FCC. (b) The energy profiles for O-T-O indirect and O-O direct paths
 179 in nonmagnetic, (c) antiferromagnetic FCC, and (d) antiferromagnetic FCT, where the O_1 - O_2 , and O_1 - O_3 paths are
 180 distinguished.

181

182 Fig. 3(a) shows the migration pathways in the FCC and FCT lattices considered in this study. In the FCC case, the
 183 stable interstitial site is the O-site, which is consistent with the result of Ismer *et al.*[11]. Three types of migration
 184 pathways between the O-sites are indicated, namely, O_1 - O_2 direct, O_1 - T - O_3 indirect, and O_1 - O_3 direct (which is
 185 identical to the O_1 - O_2 path in the case of FCC) paths. The energy profiles for these migration pathways in
 186 nonmagnetic FCC, antiferromagnetic FCC, and antiferromagnetic FCT are shown Fig.3 (b)-(d). In antiferromagnetic
 187 FCT, the two paths of the direct nearest neighbor O-sites are distinguished. In all cases, the energy barrier between
 188 the nearest neighbor O-sites mediated by the T-site are lower than O_1 - O_2 and O_1 - O_3 direct hopping. Therefore, the
 189 O_1 - T - O_3 indirect migration pathway can be considered as a stable one. For antiferromagnetic FCT, there are two O-

190 T-O paths. One is in the same spin layer and another is the path between different spin layers. The more stable path
 191 is found in the same spin layer and the migration energies of all the diffusion pathways are listed in Table 1. The
 192 energy barrier of the antiferromagnetic FCC (0.44 eV) and FCT (0.44 eV) are consistent with the experimental
 193 migration energy, which was reported as 0.46 eV[29].

194 Figure 4 shows the phonon dispersion for each atomic configurations, (a) nonmagnetic FCC supercell with hydrogen
 195 in O-site and T-site, (b) saddle point, (c) antiferromagnetic FCC supercell with hydrogen in the interstitial sites in O-
 196 site and T-site, (d) saddle point, and (e) antiferromagnetic FCT. In Fig. 4(a), (c), and (e), several dispersionless mode
 197 mainly originated from the hydrogen vibration for different configurations are shown in the same figure, because the
 198 phonon dispersion derived from host Fe atoms is slightly changed. The hydrogen vibrations in an FCC Fe-25Cr-20Ni
 199 stainless steel alloy with a low hydrogen content has been reported by neutron spectroscopy [30]. The vibrational
 200 energy has been found to be ~ 130 meV(31 THz). This value has consistent with calculated vibrational energies of
 201 nonmagnetic FCC (139 meV), antiferromagnetic FCC (140 meV) and antiferromagnetic FCT (137 meV),
 202 respectively.

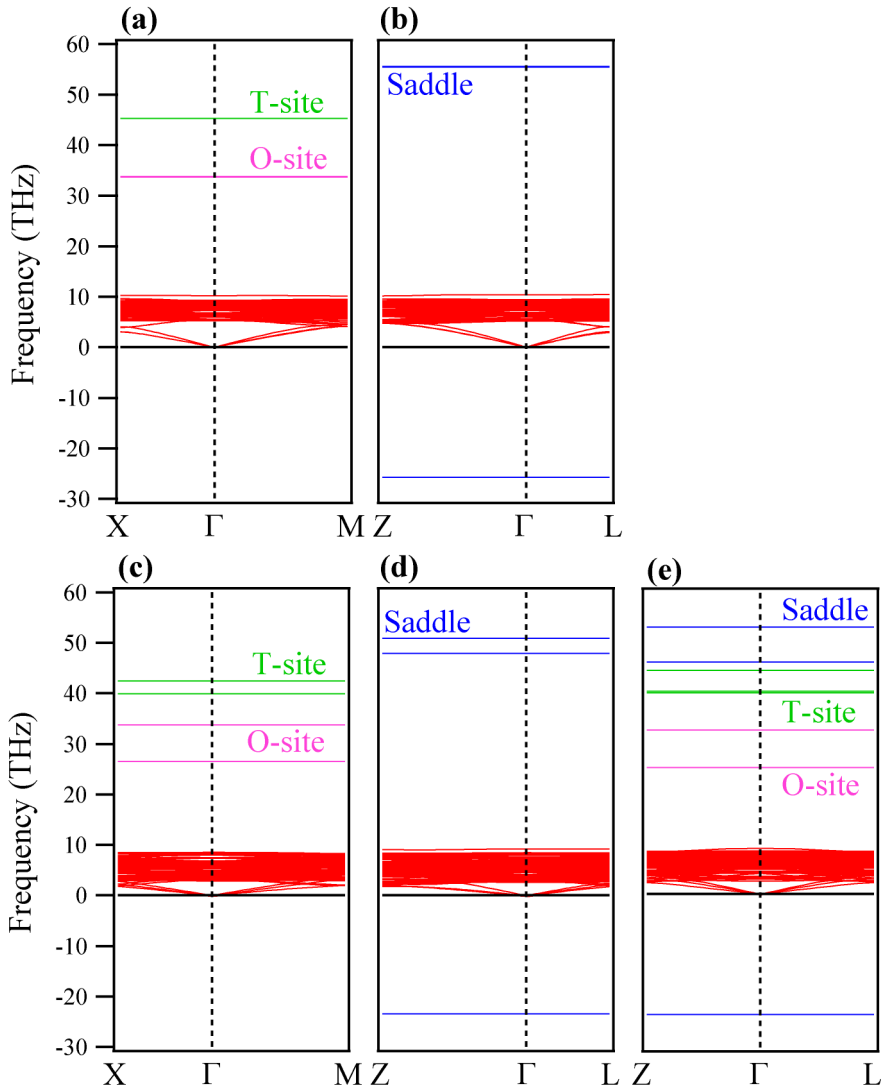
203 For the nonmagnetic FCC with hydrogen in O-site and T-site, there is no imaginary mode. Therefore we would like
 204 to point out that the T-site is not unstable, but a metastable site. On the other hand, the state with hydrogen in saddle
 205 point has one imaginary mode. Thus, the stable diffusion pathway of hydrogen is thought to be between the O-sites
 206 mediated by the T-sites in FCC. A similar behavior observed in antiferromagnetic FCC and FCT is shown in fig. 4
 207 (c) ~ (e). From the figures, O-site and T-site are indicated as a stable interstitial site, and the saddle point between
 208 O₁-T in antiferromagnetic FCC and FCT have one imaginary mode and these states are revealed stable transition
 209 state of hydrogen diffusion. Ismer *et al*[11]. have reported similar results from first-principles calculations. Recently,
 210 Machida *et al.*[31] have deduced that the interstitial site for hydrogen in FCC is not only the O-site but also the T-
 211 site from neutron diffraction patterns under high pressure. Thus, our results are consistent with the calculated and
 212 experimental results.

213
 214

215 Table 1 The calculated migration energy for hydrogen diffusion in nonmagnetic FCC, antiferromagnetic FCT and
 216 FCC, respectively.

| | Nonmagnetic FCC (eV) | Antiferromagnetic FCT (eV) | Antiferromagnetic FCC (eV) |
|----------------------------------|-------------------------|-------------------------------|-------------------------------|
| O ₁ -T-O ₃ | 0.64 | 0.44 | 0.44 |
| O ₁ -O ₃ | 1.08 | 1.07 | 0.84 |
| O ₁ -O ₂ | - | 0.72 | - |

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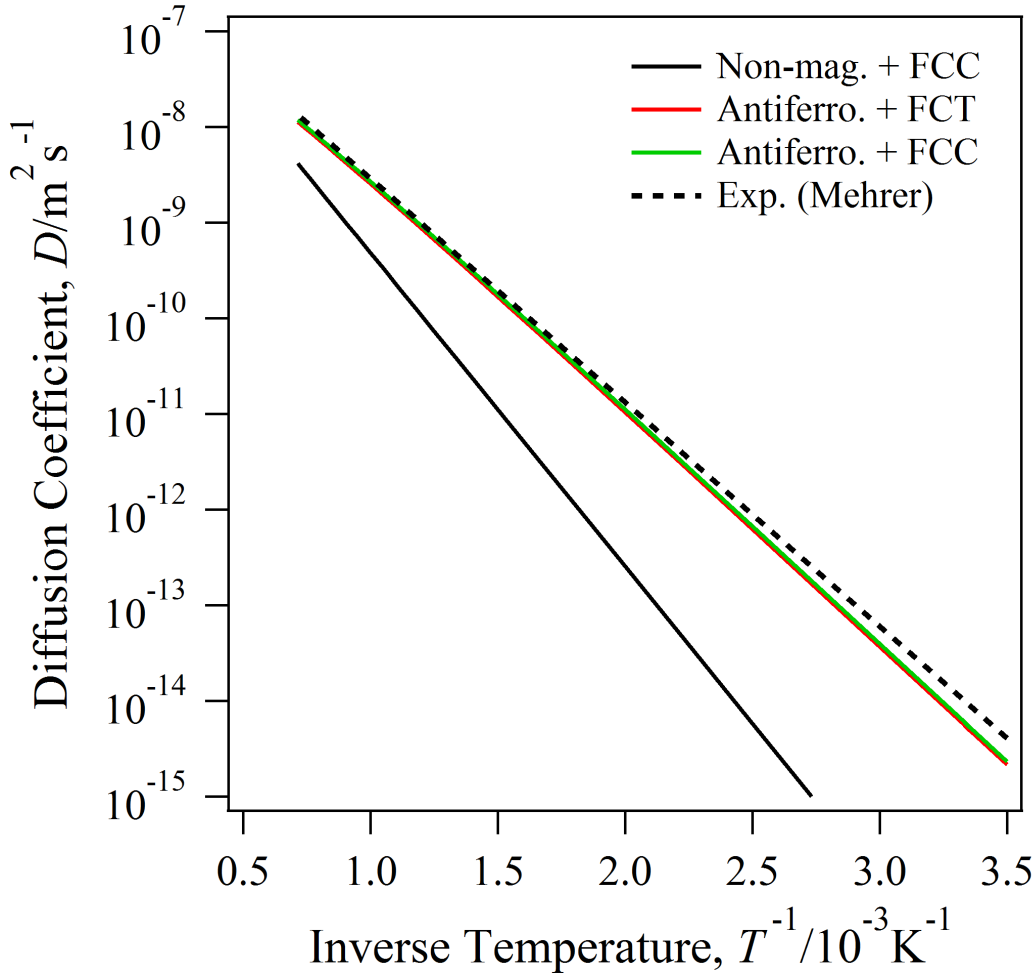
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Fig. 4. The calculated phonon dispersions of a 32-atom Fe supercell with hydrogen atom in (a) O-site and T-site, and (b) saddle point in nonmagnetic FCC, respectively. The calculated phonon dispersions of antiferromagnetic FCC with hydrogen which states in (c) O-site and T-site, (d) saddle point, respectively. (e) the calculated phonon dispersions of antiferromagnetic FCT supercell with hydrogen which occupies several sites. The labeling of the special points in the Brillouin zone depends on the symmetry of several supercells. The pink, green and blue lines indicate the vibrational frequencies of hydrogen in O-site, T-site and saddle point, respectively.



236

237 Fig.5. The diffusion coefficient of hydrogen in FCC. The calculated values of nonmagnetic FCC, and
 238 antiferromagnetic FCT and FCC are denoted by the solid, red, and green lines, respectively. The dashed line
 239 indicates the experimental diffusion coefficient by Mehrner[29].

240

241 Diffusion coefficient of hydrogen between nearest stable O-site in FCC are described as

242
$$D_{FCC} = a^2 \Gamma \quad (6)$$

243 where a is a lattice constant. In this case, however, hydrogen diffusion occurs via metastable T-site. Therefore
 244 diffusion coefficient is modified as follows,

245
$$D'_{FCC} = \frac{1}{2} D_{FCC} (1 + 2 \exp(-\frac{\Delta G_{tet-oct}}{k_B T}))^{-1} \quad (7).$$

246 $\Delta G_{tet-oct}$ is the difference of the free energy between hydrogen in tetrahedral and octahedral sites. Fig.5 shows an
 247 Arrhenius plot of the diffusion coefficient of hydrogen in nonmagnetic FCC, and antiferromagnetic FCC and FCT
 248 along with the experimental diffusion coefficient in pure FCC between 1183K ($1000/T = 0.845$) and 1373K ($1000/T$
 249 $= 0.728$) [29]. Our calculation results of the nonmagnetic state agree with those presented in a previous report[11].
 250 The calculated diffusion coefficients of antiferromagnetic FCC and FCT are approximately same, and closely agree

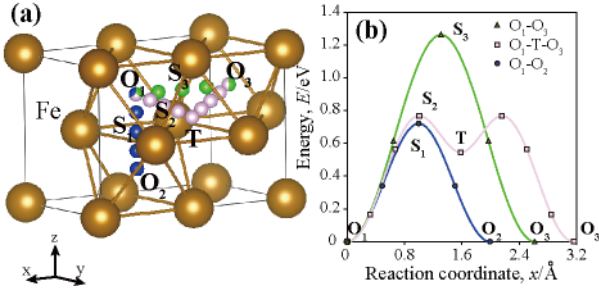
251 with the experimental value for pure Fe. These results are consistent with a fact that the FCC has the antiferromagnetic
 252 state as the ground state.

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255 3.3. HCP

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258 Fig.6. (a) The hydrogen migration pathways in HCP. (b) The energy profiles for several paths in the case of $c/a =$
 259 1.58.

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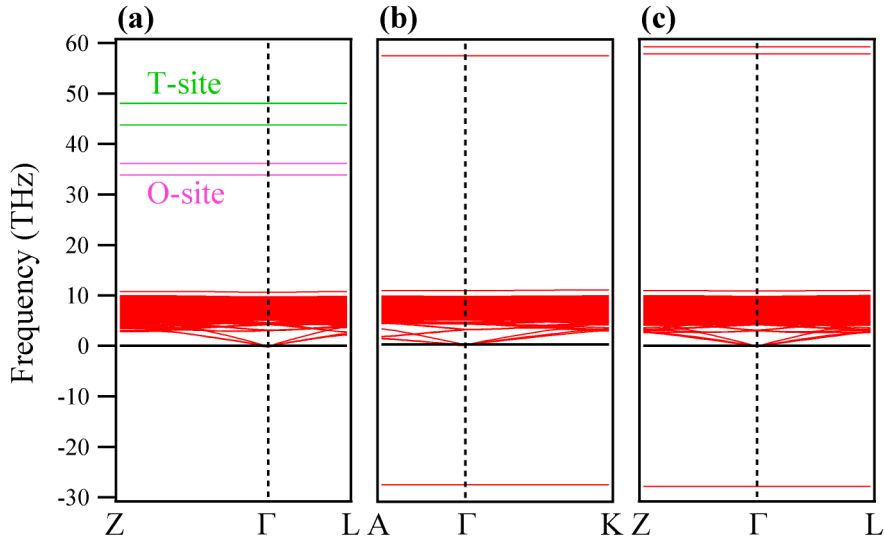
261 Fig. 6(a) shows the hydrogen atoms on migration pathways in the HCP iron. In this study, a fully relaxed lattice was
 262 obtained with $c/a = 1.58$. The migration energies of several diffusion paths are shown in Fig. 6(b). The stable
 263 interstitial site is found to be the O-site, while the T-site is metastable. The migration energies of O₁-O₂ direct, O₁-T-
 264 O₃ indirect, and O₁-O₃ direct hopping are estimated to be 0.72, 0.77, and 1.26, respectively. In case of $c/a = 1.58$, it
 265 is revealed that the O₁-O₂ direct diffusion path along the c -axis is slightly more stable than the diffusion path via the
 266 T-site on c -plane.

267 The calculated phonon dispersion of nonmagnetic HCP supercell with hydrogen in O-site and T-site is shown in
 268 Fig. 7 (a). Fig. 7 (b) and (c) show the calculated phonon dispersions of the saddle points S₁ between O₁-O₂ and S₂
 269 between O₁-T. The hydrogen vibrational energy of dhcp FeH has been reported as 103meV by inelastic neutron
 270 scattering spectrum [32] and calculated value (140 meV, 110meV) with hydrogen in O-site is slightly higher than the
 271 experimental value. The difference of vibrational energy between experimental and calculated values can be caused
 272 by the difference of the metal-hydrogen distance.

273 For the state with hydrogen in O-site and T-site, there no imaginary mode. Therefore, these interstitial sites are
 274 stable from the view point of dynamics. On the other hand, the two states of saddle point have one imaginary mode
 275 derived from hydrogen. Therefore, both the O₁-O₂ path along the c -axis and the O₁-T-O₃ path in the c -plane can serve
 276 as stable diffusion paths in HCP.

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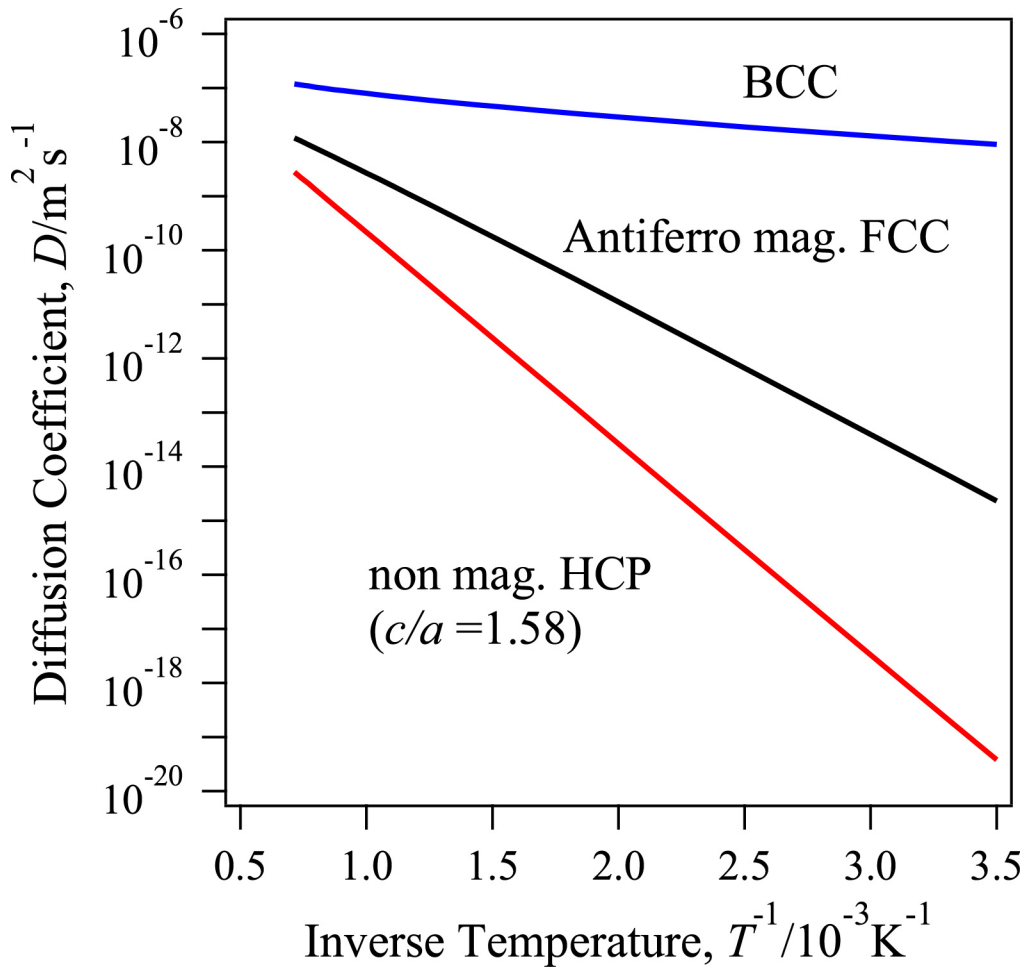
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280 Fig. 7. The calculated phonon dispersions of a 96-atom Fe supercell with hydrogen atom in (a) O-site and T-site,
 281 (b) the saddle point of S_1 between O_1 - O_2 and (c) S_2 between O_1 -T in nonmagnetic HCP, respectively. The labeling
 282 of the special points in the Brillouin zone depends on the symmetry of several supercells. The pink and green lines
 283 indicate the vibrational frequencies of hydrogen in O-site and T-site, respectively.

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289 Fig.8 The Arrhenius plot of the diffusion coefficient of hydrogen in HCP and antiferromagnetic FCC. The red solid
290 line indicates the diffusion coefficient for HCP with $c/a = 1.58$. The black and blue lines indicate the diffusion
291 coefficient for antiferromagnetic FCC and BCC, respectively.

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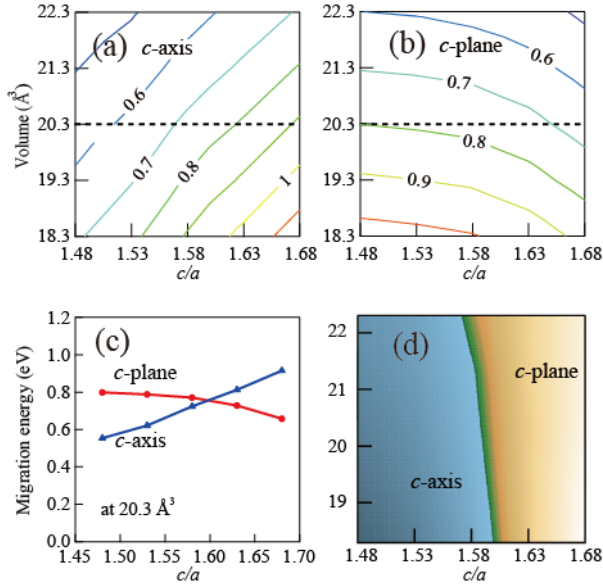
293 Here the diffusion coefficient in BCC, FCC, and HCP are compared. Although the hydrogen diffusion in BCC is
294 isotropic, that in HCP is anisotropic along c -axis. Therefore, hydrogen diffusion in HCP must be considered with a
295 coordination number between nearest stable O-site along c -axis. The diffusion coefficient in BCC and HCP are
296 described as follows, respectively.

297
$$D_{BCC} = \frac{1}{12} a^2 \Gamma \quad (8)$$

298
$$D_{HCP} = \frac{1}{4} c^2 \Gamma \quad (9)$$

299 Fig.8 shows an Arrhenius plot of the diffusion coefficient of hydrogen in HCP for $c/a = 1.58$, along with relevant
300 data of the antiferromagnetic FCC and BCC. The diffusion coefficient for HCP is revealed to be lower than that for
301 FCC and BCC. Hence, materials that have the HCP structure can be considered to have the potential for resistance
302 against hydrogen embrittlement that could either equal or even surpass FCC materials. This result has qualitatively
303 agreement with the experimental result: the hydrogen diffusion behavior has not been changed on epsilon martensitic

304 transformation, which is the transformation from FCC to HCP, by Koyama *et al.* [6, 7]. On the other hand, c/a of
 305 FeMn steels with epsilon martensite was reported to be 1.63 from experiments[33]. The different c/a values obtained
 306 from different experiments are assumed to be due to the additive elements present in the alloy. Therefore, we have
 307 investigated the c/a and volume dependence of hydrogen diffusivity in HCP because these values are expected to be
 308 influenced by additive elements in actual materials.
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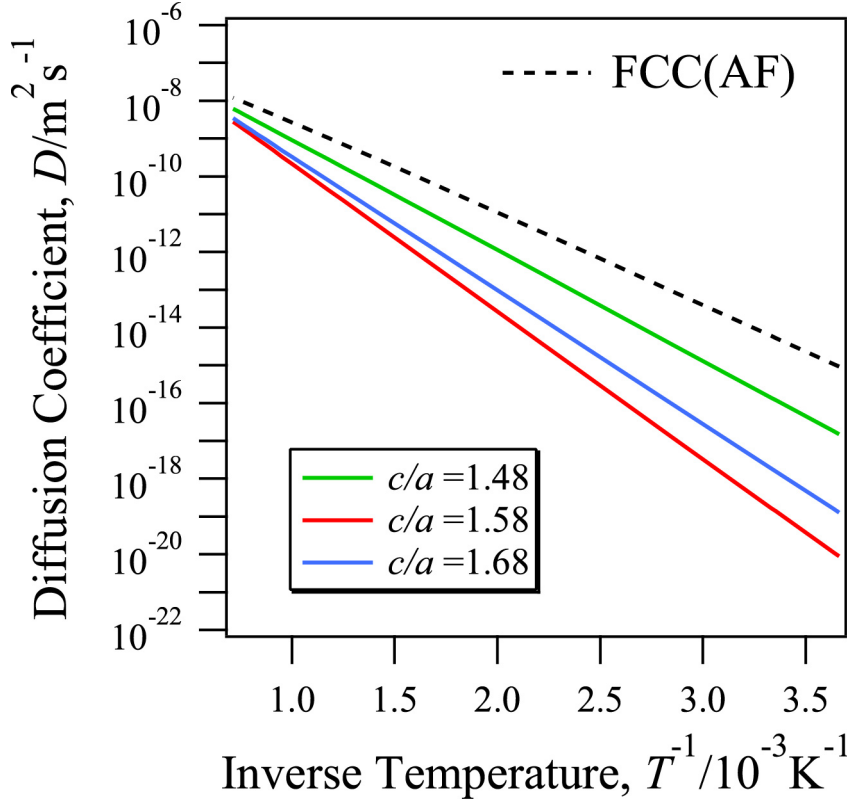
311
 312 Fig.9. The migration energy in eV of (a) O₁-O₂ along c -axis and (b) O₁-T-O₃ on the c -plane with respect to volume
 313 and c/a of HCP. (c) Comparison of the hydrogen migration energies along c -axis and on the c -plane on the cross-
 314 section, denoted by dashed line in (a) and (b). (d) The difference of migration energy between c -axis and c -plane.
 315 The left and right sides indicates c -axis and c -plane diffusion dominated region, respectively.
 316

317 From the calculated energy profiles, we found an interesting behavior of hydrogen in HCP iron. This was the
 318 anisotropic diffusion of hydrogen depending on c/a . Fig.9 shows the relationship between the migration energy,
 319 volume, and c/a of HCP. The migration energies of O₁-O₂ along the c -axis and O₁-T-O₃ on the c -plane are indicated
 320 in Fig.9(a) and (b), respectively. The equilibrium volume and c/a of nonmagnetic HCP are 20.3 Å³ and 1.58,
 321 respectively. The level lines of migration energy of O₁-O₂ are at about 45° and are assumed to be affected equally by
 322 a change in volume and c/a ratio. On the other hand, the lines of migration energy of O₁-T-O₃ seem to shift slightly
 323 with changing c/a ratio and hence they are considered to be affected to a lower extent by the change in c/a .

324 The c/a -dependent migration energies of both the pathways are shown in Fig.9(c). The data corresponds to the cross-
 325 section marked using dashed lines in Fig.9(a) and (b). The diffusion pathway in the c -plane becomes stable with
 326 increasing c/a values whereas the path along the c -axis becomes unstable. The stable pathways are found to be along
 327 the c -axis below $c/a \sim 1.6$ and in the c -plane above $c/a \sim 1.6$. Fig. 9(d) indicates the stable pathways with respect to
 328 volume and c/a . The blue and orange regions indicate which is the stable diffusion paths. It is revealed that the stable
 329 diffusion path is along the c -axis O₁-O₂ below $c/a \sim 1.6$ and the c -plane O₁-T-O₃ becomes stable above $c/a \sim 1.6$.

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333 Fig.10. The Arrhenius plot of the diffusion coefficient of hydrogen in HCP with $c/a = 1.48, 1.58,$ and $1.68,$
 334 respectively. Dashed line indicates the calculated results of antiferromagnetic FCC.

335

336 From the migration energies investigated above, the diffusion coefficient was estimated in the range of $c/a = 1.48-$
 337 $1.68.$ For the calculation of diffusion coefficient in the case of $c/a = 1.68,$ the diffusion path contains the metastable
 338 T-site. Therefore diffusion coefficient is modified as follows,

$$339 \quad D'_{HCP} = \frac{3}{2} D_{c-plane} (1 + 2 \exp(-\frac{\Delta G_{tet-oct}}{k_B T}))^{-1} \quad (10)$$

340 where

$$341 \quad D_{c-plane} = \frac{3}{2} a^2 \Gamma \quad (11).$$

342 The equation (11) was developed by considering a coordination number between stable O-sites in c -plane. The
 343 Arrhenius plots of the diffusion coefficient of hydrogen in HCP at 20.3 \AA^3 are given in Fig.10. Minimum diffusivity
 344 is observed around $c/a \sim 1.6,$ and the diffusion coefficient increases with both increasing and decreasing c/a ratio.
 345 However, the direction of the H diffusion changes with the value of $c/a.$ When c/a is higher than $\sim 1.6,$ the diffusion
 346 path is in the c -plane $O_1-T-O_3.$ On the other hand, when c/a is lower than $\sim 1.6,$ the diffusion path is along the c -axis.
 347 Recently, the diffusion coefficient of hydrogen in HCP has been reported by theoretical calculation for $c/a \sim 1.58$
 348 [13]. On the other hand, actual experimental c/a in FeMn steels was estimated at ~ 1.63 [33]. Our findings regarding
 349 the diffusivity in the HCP lattice has important implications in terms of the possibility of control over the magnitude

350 and direction of the diffusion behavior. By tuning the c/a ratio, which can be performed by adding some elements
351 into the iron, the magnitude and direction of the H diffusion can be controlled independently in the range $c/a < 1.6$
352 or $c/a > 1/6$. For example, the c/a ratio of HCP-Fe can be changed by loading high pressure [34] and addition of
353 alloying elements [35]. In fact, the change with respect to the c/a ratio of HCP-Co bulk and film has been reported
354 by these methods[36,37]. This diffusion characteristic, which is a general physical property of the HCP lattice, can
355 now be used to produce a new functionality of the HCP alloy system.

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361 **4. Conclusion**

362 Hydrogen diffusivity in HCP iron, which is the promising candidate for high resistance to fatigue crack growth has
363 been investigated by a theoretical calculation and compared to those of BCC and FCC. In BCC and FCC iron, the
364 calculated results with respect to stable interstitial sites are in good agreement with those reported in previous
365 experimental and theoretical investigations. In the FCC case, the stable diffusion path is between the O-sites mediated
366 by the T-site. The calculated hydrogen diffusion coefficient was enhanced and agreed with experimental results,
367 when the antiferromagnetic state was considered. In HCP, the interstitial site was found to be the O-site and the
368 migration energy for the hydrogen diffusion was estimated to be higher than that in FCC, which causes low diffusivity
369 of H in the HCP lattice. The hydrogen diffusion was seen to have an anisotropic behavior that depended on the c/a
370 ratio. At an equilibrium volume of nonmagnetic HCP, the stable pathways were found to be the direct O-site path
371 along the c -axis when $c/a < 1.6$ and the path between the O-sites mediated by the T-site on the c -plane when $c/a >$
372 1.6. The diffusion coefficient of hydrogen in HCP was found to be lower than that in FCC and it can be concluded
373 that the epsilon martensite has a low susceptibility to hydrogen embrittlement. The low and direction controllability
374 of hydrogen diffusivity in HCP iron suggests a different paradigm in steels under hydrogen environment.

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439 **List of figure captions**

440

441 Figure 1. (a) The hydrogen migration pathways in BCC. O, T, and S represent the octahedral site, tetrahedral site,
442 and saddle point, respectively. (b) The energy profiles for two paths: T₁-T₂ direct and T₁-O-T₃ indirect.

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445 Figure 2. The calculated phonon dispersions of a 56 atom-Fe supercell with hydrogen atom in (a) T-site and (b) the
446 saddle (S) point of T₁-T₂, respectively. The symmetry of the supercell describing the transition state is
447 different from that of the tetrahedral site, which is reflected in the labeling of the special points in the
448 Brillouin zone.

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451 Figure 3. (a) The hydrogen migration pathways in FCC. (b) The energy profiles for O-T-O indirect and O-O direct
452 paths in nonmagnetic, (c) antiferromagnetic FCC, and (d) antiferromagnetic FCT, where the O₁-O₂, and
453 O₁-O₃ paths are distinguished.

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456 Figure 4. The calculated phonon dispersions of a 32-atom Fe supercell with hydrogen atom in (a) O-site and T-site,
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458 antiferromagnetic FCC with hydrogen which states in (c) O-site and T-site, (d) saddle point, respectively.
459 (e) the calculated phonon dispersions of antiferromagnetic FCT supercell with hydrogen which occupies
460 several sites. The labeling of the special points in the Brillouin zone depends on the symmetry of several
461 supercells. The pink, green and blue lines indicate the vibrational frequencies of hydrogen in O-site, T-site
462 and saddle point, respectively.

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465 Figure 5. The diffusion coefficient of hydrogen in FCC. The calculated values of nonmagnetic FCC, and
466 antiferromagnetic FCT and FCC are denoted by the solid, red, and green lines, respectively. The dashed
467 line indicates the experimental diffusion coefficient by Mehrer[29].

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470 Figure 6. (a) The hydrogen migration pathways in HCP. (b) The energy profiles for several paths in the case of $c/a =$
471 1.58.

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474 Figure 7. The calculated phonon dispersions of a 96-atom Fe supercell with hydrogen atom in (a) O-site and T-site,
475 (b) the saddle point of S₁ between O₁-O₂ and (c) S₂ between O₁-T in nonmagnetic HCP, respectively. The
476 labeling of the special points in the Brillouin zone depends on the symmetry of several supercells. The

477 pink and green lines indicate the vibrational frequencies of hydrogen in O-site and T-site, respectively.
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 480 Figure 8. The Arrhenius plot of the diffusion coefficient of hydrogen in HCP and antiferromagnetic FCC. The red
 481 solid line indicates the diffusion coefficient for HCP with $c/a = 1.58$. The black and blue lines indicate the
 482 diffusion coefficient for antiferromagnetic FCC and BCC, respectively.
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 485 Figure 9. The migration energy in eV of (a) O_1-O_2 along c -axis and (b) O_1-T-O_3 on the c -plane with respect to volume
 486 and c/a of HCP. (c) Comparison of the hydrogen migration energies along c -axis and on the c -plane on the
 487 cross-section, denoted by dashed line in (a) and (b). (d) The difference of migration energy between c -axis
 488 and c -plane. The left and right sides indicates c -axis and c -plane diffusion dominated region, respectively.
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490
 491 Figure 10. The Arrhenius plot of the diffusion coefficient of hydrogen in HCP with $c/a = 1.48, 1.58, \text{ and } 1.68$,
 492 respectively. Dashed line indicates the calculated results of antiferromagnetic FCC.
 493 ...15

494
 495 Table 1 The calculated migration energy for hydrogen diffusion in nonmagnetic FCC, antiferromagnetic FCT and
 496 FCC, respectively.

| | Nonmagnetic FCC (eV) | Antiferromagnetic FCT (eV) | Antiferromagnetic FCC (eV) |
|-------------|-------------------------|-------------------------------|-------------------------------|
| O_1-T-O_3 | 0.64 | 0.44 | 0.44 |
| O_1-O_3 | 1.08 | 1.07 | 0.84 |
| O_1-O_2 | - | 0.72 | - |

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