

1 Investigation into Reversed Phase Chromatography Peptide Separation Systems Part II: An 2 Evaluation of the Robustness of a Protocol for Column Characterisation

3 Jennifer K. Field ^a, Melvin R. Euerby ^{a,b}, Patrik Petersson ^{c*}

4 ^a *Strathclyde Institute of Pharmacy and Biomedical Sciences, University of Strathclyde, Glasgow, United*
5 *Kingdom G4 0RE*

6 ^b *Shimadzu UK, Milton Keynes, Buckinghamshire, United Kingdom MK12 5RD*

7 ^c *Novo Nordisk A/S, Måløv, Denmark DK-2760*

8 ** Author for correspondence, +45307 92146, ppso@novonordisk.com*

9

10 Abstract

11 The robustness of the Peptide Reversed Phase Chromatography (RPC) Column Characterisation
12 Protocol was evaluated using reduced factorial design, to ascertain the degree of control required
13 for parameters including temperature, flow rate, dwell volume, a systematic shift in the gradient,
14 amount of formic acid in the aqueous and organic, pH of the ammonium formate and amount of
15 acetonitrile in the strong solvent (%MeCN). All levels were deemed acceptable within reasonable
16 tolerances except the %MeCN in the strong solvent, where a loss of MeCN resulted in an
17 unacceptable variation. Mitigations have been introduced to ensure the integrity of the data to
18 allow RPC columns to be characterised using peptides as probes, with the definitive protocol
19 described. In addition, the instrument and column batch to batch variability were assessed with
20 good reproducibility.

21

22 Keywords

23 Robustness; Factorial Design; Peptides; Column Characterisation; RPC Stationary Phases; Principal
24 Component Analysis

25

26 1 Introduction

27 The selection of appropriate stationary phases is a key element of method development, however,
28 with the plethora of phases available from many different column manufacturers, it is difficult to
29 make a rational decision as to which column to start with. For small molecules, there are various
30 strategies including the Tanaka and its extended protocol, Snyder's Hydrophobic Subtraction Model
31 and Lesellier's Linear Subtraction Energy Relationships (LSER) which have characterised reversed
32 phase (RP) stationary phases, with the results available in databases which are free to access [1-5].
33 These databases allow the end user to establish which phases are chromatographically different for

34 a diverse range of phases used for method development purposes, or alternatively, select phases
35 which are chromatographically similar, in order to select “back-up” columns.

36 Until recently, there has not been a corresponding characterisation protocol for RP columns suitable
37 for peptide separations, however, the Peptide RPC Column Characterisation Protocol was developed
38 to address this deficiency [6]. A range of 26 specifically designed peptides with different physico-
39 chemical properties were synthesised to assess prominent interactions such as hydrophobicity,
40 hydrogen bonding, electrostatic interactions and aromatic character as well as reflect typical paths
41 for degradation of peptides. These 26 peptides were evaluated on 14 stationary phases that
42 possessed different properties based on prior knowledge of the column chemistry using gradient
43 chromatography at low and intermediate pH. Selectivity was measured using the difference in
44 normalised retention (Δt_g^*), where a total of 66 delta values were produced to probe different types
45 of interactions which were critically assessed using principal component analysis (PCA). The mobile
46 phases selected were formic acid (~pH 2.5) and ammonium formate at (~pH 6.45, native pH) to
47 evaluate the purer interactions of the stationary phase. The rationale for mobile phase selection can
48 be observed in reference 6. The stationary phases were categorised into three distinct groupings to
49 describe the prominent interactions; neutral (phases which possess a high degree of ligand density
50 and / or end capping), negative / polar (phases which either contain a negatively charged moiety in
51 the ligand, capable of silanophilic interactions due to a light bonding / lack of end capping, or
52 functionalities which are capable of forming polar interactions such as hydrogen bonding) and
53 positive character (phases which contain a positively charged moiety). The total of peptides required
54 was reduced from 26 to 11 which generated 11 delta values to describe changes in positive /
55 negative charge, hydrophobicity, phenolic character, aromatic character as well as steric changes,
56 racemisation, deamidation and oxidation (Table 1 and Table 2). A smaller study was conducted to
57 observe the effect of TFA instead of formic acid as the low pH additive, as TFA is quite frequently
58 used for peptide analysis. The resulting data caused the distinct categories produced via the formic
59 acid / ammonium formate biplot to lose their structure which suggests a loss of information when
60 using TFA. This is as hypothesised as TFA acts as an ion pair therefore the character of the stationary
61 phase is of less importance, thus emphasising the necessity to conduct a column characterisation
62 protocol using an appropriate additive like formic acid. An iterative study was conducted which
63 appraised the 66 delta values to determine if the total number could be reduced whilst still fully
64 describing the stationary phases.

65 In order for the Peptide RPC Column Characterisation protocol to be truly valuable, the robustness
66 must be considered [7]. Robustness is defined by the ICH guidelines as “a measure of its capacity to
67 remain unaffected by small but deliberate variations in method parameters and provides an

68 indication of its reliability during normal usage” [8,9]. The robustness of the protocol was assessed
69 using a reduced factorial design often referred to as a DoE (design of experiments), which is one of
70 the most commonly used tools for the assessment of robustness both within academia and the
71 pharmaceutical industry. In this methodology, small systematic changes reflecting the expected
72 experimental errors associated with the procedure are explored [9-11]. Alternative methods such as
73 one factor at a time (OFAT) could be used, which looks to vary one variable whilst maintaining all
74 other parameters of interest. However, it can often prove more resource intensive (i.e. time,
75 materials etc) and can fail to estimate the interaction between different variables. A considerable
76 amount of information can be derived from DoE with statistical significance, which cannot be
77 obtained via the OFAT approach [12]. The 11 peptide delta values were used as responses to create
78 the DoE models which highlighted the degree of deviation the systematic changes created from the
79 nominal centre point i.e. the original method conditions. This data informs what variable(s) are
80 statistically significant and their practical relevance on the result, thus highlighting what operating
81 parameters must be carefully controlled in order to maintain a robust methodology. It is integral
82 that the robustness of the procedure be considered before a larger set of stationary phases are
83 evaluated which can be included into a column characterisation database for individuals to rationally
84 select stationary phases for peptide / protein separations. A similar approach was used to assess the
85 robustness of the Tanaka and extended protocols in order to ensure the integrity of the column
86 database [10].

87 This paper aims to gain a greater understanding of the crucial factors which could impact on method
88 robustness, which not only includes performing a DoE on LC parameters used in the protocol, but
89 also establishes the robustness of mobile phase pH switching and sample load. Finally, the practical
90 operating limits of the LC parameters used in the protocol will be established in order to maintain
91 acceptable repeatability and reliability of the integrity of the peptide-based column characterisation
92 database. This paper is part of a series of articles which are designed to interrogate the factors which
93 influence the peptide separation system, where firstly the stationary phases were evaluated before
94 investigating the effect of different mobile phase additives to aid the chromatographer in making
95 rational decisions relating to their chromatographic separation.

96

97 **2 Experimental**

98 **2.1 Chemicals, Reagents and Instrumentation**

99 All chemicals used are described in reference 6. Unless otherwise stated all LC separations were
100 performed on a Shimadzu Nexera X2 UHPLC system where modules were previously described [6].

101 The base sequence for each peptide can be located in Table 2, and further description of the peptide
102 probes described in reference 6.

103

104 **2.2 Factorial Design**

105 The Peptide RPC Column Characterisation protocol conditions were described in reference 6. The
106 new test mixtures with each specific column load can be found in Table 3. Each test mixture
107 contained the two peptides required to calculate a precise delta value. This sometimes meant that
108 there was duplication of peptides in multiple test mixtures. This was to ensure the delta values were
109 well described and increase method robustness as small degrees of retention drift could impact on
110 subtle interactions such as racemisation. The peptides selected were based on the iterative process
111 of elimination conducted within reference 6, and contain both subtle and more significant changes
112 to evaluate common degradation and specific interactions to describe the column for
113 characterisation. The delta values, which are a measure of the selectivity in gradient elution, are
114 used as the input for the DoE.

115 The factors evaluated in the DoE were summarised in Table 4, including the different ± 1 levels. The
116 factors investigated were selected as the most likely sources of error, where the levels were
117 ascertained to reflect the random variation one could expect (i.e. at least 3x the expected standard
118 deviation). Error propagation calculations based on instrument specifications and qualification data
119 were employed in the estimation of standard deviation [13].

120 Modde Pro software (see Section 2.4) was employed to create and evaluate reduced factorial
121 designs, which utilised eight methods with various +1 or -1 levels for the different variables, and
122 three repeat methods for the central nominal conditions to ascertain the reproducibility of the
123 procedure (Tables 5 and 6 for formic acid and ammonium formate gradients, respectively). Different
124 batches of solvents were produced for the nominal conditions. The software created a random run
125 order to remove any bias within the results.

126

127 **2.3 Instrument Variability**

128 Instrumental variability was compared between three additional LC instruments; Waters H-Class,
129 Waters I-Class (Milford, Ma, USA) and Agilent 1290 binary system (Waldbronn, Germany) using the
130 same batches of mobile and stationary phase. The dwell volumes varied between 300 and 700 μL .
131 The Waters instruments were controlled by Empower 3 (Feature release 3) whilst the Agilent
132 configuration was controlled by OpenLab CDS (Chemstation C.01.07 SR4).

133

134 **2.4 Stationary Phases**

135 The Peptide RPC Column Characterisation protocol was developed using 14 diverse stationary
136 phases [6] and a standardised column format of 150 x 2.1 mm. All columns were new as supplied by
137 the manufacturer. A Phenomenex Kinetex Evo C18 (150 x 2.1 mm, 100 Å, 2.6 µm) was used for the
138 reduced factorial design experiments whilst the instrument variation was performed on a Waters
139 Acquity HSS C18 (150 x 2.1 mm, 130 Å, 1.8 µm). Column batch to batch variability was assessed on
140 six different batches of Ascentis Express C18 with six differing base silicas and four differing silanes
141 (150 x 2.1 mm, 90 Å, 2.7 µm). Loading studies were performed on a Phenomenex Kinetex C18 (150 x
142 2.1 mm, 100 Å, 2.6 µm). The peak apex of a water injection was used as the dead time marker for
143 each column [14].

144

145 **2.5 Software and Calculations**

146 Principal Component Analysis (PCA) was performed using SIMCA (Version 14.1, Umetrics, Umeå,
147 Sweden) and Origin (Version OriginPro 2016, OriginLab, Northampton, MA, USA). The variables
148 within the PCA were all autoscaled, in order to give each variable the same importance. Reduced
149 factorial design was performed using Modde Pro (Version 12.0.1. Umetrics, Umeå, Sweden). The net
150 charges of the peptide probes were calculated at both pH 2.5 and 6.45 using General Protein/Mass
151 Analysis for Windows (GPMAW) software (Version 9.51, Lighthouse Data, Odense, Denmark).

152

153 **3 Results and Discussion**

154 **3.1 Principle Component Analysis and Factorial Design**

155 The robustness of each delta value (Table 4) was assessed by fitting a first order polynomial model to
156 the data obtained for the formic acid and ammonium formate designs (Table 5 and 6). The typical
157 Δt_g and Δt_g^* values obtained on the Kinetex Evo C18 can be seen in Table 1 under the nominal centre
158 point conditions.

159 The quality of the model is measured using a regression coefficient from fitting the model (R2) and
160 one from a cross validation of the model (Q2) [11,15]. For a good model both values are close to 1.
161 However, in a robustness evaluation an ideal outcome would be a poor model, i.e. low R2 and Q2
162 values. The effect of a ± 1 change of the different factors should, for a robust methodology,
163 correspond to the experimental noise at nominal conditions.

164 The average R2 and Q2 for the delta values in formic acid were 0.761 (standard deviation SD 0.208)
165 and -0.133 (SD 0.151), respectively, whilst for the ammonium formate delta values, R2 and Q2
166 measured 0.750 (SD 0.080) and -0.200 (SD 0.000), respectively. The centre point experiments (N9-
167 11) provide an indication into the reproducibility of the procedure, where the average difference for
168 the delta probes was 0.001 (ranged from 0.000 to 0.004 for the relevant delta values). The result
169 gives an early indication that the Peptide RPC Column Characterisation protocol should be robust.

170 The method robustness can be evaluated further by comparing coefficient plots which evaluates the
171 individual effect of each parameter on the robustness (Fig. 1 and 2). Each variable was scaled and
172 centred in the coefficient plots, so that they were comparable. The height of the bar gives the
173 degree of the effect whilst the error bar indicates the 95% confidence interval which highlights
174 statistically significant parameters when the error bar does not cross zero.

175

176 **3.1.1 Temperature**

177 Temperature had no discernible effect on any of the responses in ammonium formate and the
178 majority in formic acid, however, there was a statistical response for $\Delta(8a,1)$ and $\Delta(16,13)$ in formic
179 acid. The height of the bars however would indicate that the practical significance on these two
180 delta results would be minimal. Although for these probes it has a minimal effect, it is recommended
181 that the actual temperature of the column is determined, as it is known that the column oven design
182 and even column position within the oven can change the actual temperature within the column.

183 This can be achieved by injecting caprylophenone (detection at 250 nm) onto a column using a
184 premixed mobile phase of MeCN/H₂O (45:55 w/w) to eliminate instrument to instrument %MeCN
185 variations. The column oven temperature is changed in 5 °C intervals over the range 30-60 °C. The
186 same column is then tested by immersing it in a water bath with 30 cm of 0.12 mm tubing prior to
187 the column for thermostating. This is then assumed to be an accurate measure of the temperature
188 of the column. The temperature of the water bath should be determined using a calibrated /
189 certified thermometer with an appropriate accuracy e.g. ± 0.5 °C between 30 to 60 °C.

190 The sample is injected at 30 to 60 °C in 5 °C intervals in order to construct a plot (retention time –
191 system dead time) vs. temperature plot in order to determine any deviation in temperature for the
192 LC system (i.e. ΔT at a certain retention time). System dead time is subtracted from the retention
193 time to allow for comparison between the ovens and is determined by the retention time of water
194 (detection at 215 nm), where the column is replaced with a union. This procedure should be
195 followed for any new type of column oven design, and once the actual temperature of the column is

196 determined, the Peptide RPC Column Characterisation Protocol can be adapted to compensate for
197 any deviation in temperature.

198

199 **3.1.2 Systematic Shift in the Gradient Composition**

200 The responses in formic acid and ammonium formate were all unaffected by the systematic shift in
201 the gradient. This systematic shift in the gradient assumes the same degree of error will apply across
202 the length of the gradient.

203

204 **3.1.3 Flow rate**

205 A change of flow rate also corresponds to a change in gradient slope. Flow rate only presented a
206 small statistical response for $\Delta(9,1)$ in formic acid, whilst all other responses in both ammonium
207 formate and formic acid were unaffected. Similar to temperature, the actual practical effect of this
208 variable would be quite minimal, thus this parameter can be assumed to be robust within the
209 methodology.

210

211 **3.1.4 Dwell Volume**

212 Dwell volume was statistically insignificant for all responses in either mobile phase, indicating this
213 variable does not impact on the robustness of either the formic acid or ammonium formate
214 gradient. This is to a large extent due to the normalisation of the retention times which removes the
215 effect of dwell volume, allowing direct comparison between different instrumentation. The dwell
216 volume range selected for evaluation (100 - 500 μL) should cover UHPLC instrumentation.

217

218 **3.1.5 Amount of Formic Acid**

219 Differences in formic acid levels could impact on the robustness of the protocol as different levels
220 would result in a different pH which would affect the overall net charge on the peptides. However,
221 differences in formic acid volume were observed to be insignificant, the $\Delta(9,1)$ and $\Delta(16,13)$ values
222 exhibited a very small statistical significance but this was deemed to be of little practical relevance. It
223 is recommended though that formic acid volumes should be dispensed volumetrically from a pre-
224 calibrated pipette which is checked each time a solvent is prepared to ensure the integrity of the
225 chromatographic results.

226

227 **3.1.6 Ammonium Formate Stock Solution pH**

228 The stationary phase environment at intermediate pH is somewhat unpredictable due to the range
229 of pK_a values for the residual silanols [16,17]. It is believed the majority of silanols should be ionised
230 at pH 6.45 (the native pH of ammonium formate), however, this uncertainty can potentially lead to a
231 greater degree of variation in results and hence can contribute to the lack of robustness. The
232 ammonium formate can also be a source of error, where the age of the buffer, its storage
233 environment and its resultant pH range were investigated. The pH was measured for 16 different
234 200 mM solutions where the average pH was 6.45 (SD 0.03). The levels in the DoE (pH 6.45 ±0.06)
235 were set based on a 99% CI based on triplicate determinations of the pH. The age of the ammonium
236 formate did not appear to greatly affect the overall pH of the solution, where the pH measured was
237 within the range tested in the DoE. There was, however, a change in pH based on salts which were
238 inappropriately stored, which resulted in a lower pH for a poorly capped container which indicates a
239 loss of ammonia. This could impact on the degree of silanol ionisation and hence affect retention
240 and the delta values. Ideally, the formate salt should be stored in a desiccator to reduce water
241 uptake and firmly replacing the cap should reduce the risk of ammonia loss.

242 The responses in ammonium formate were all stable within the upper and lower pH limits in the
243 DoE, with no statistical significance. The pH of the stock buffer solution should be measured using
244 appropriately calibrated standards to ensure the pH is within this range to ensure the integrity of the
245 protocol. It is also advised that if the ammonium formate exhibits any considerable signs of
246 hygroscopicity in addition to changes in pH then it should not be used.

247 To avoid microbial growth which could contaminate the LC system and potentially block the column
248 inlet frit, causing split peaks and higher back pressures, it is recommended to limit the storage of
249 stock buffer solution to 4 months at 5 °C.

250

251 **3.1.7 MeCN Composition in the B Solvent**

252 The selectivity of $\Delta(3,1)$, $\Delta(9,1)$, $\Delta(10,9)$ and $\Delta(24,13)$ were particularly susceptible to the change in
253 acetonitrile content in the B solvent in the ammonium formate gradients. The effect was actually
254 significant enough that it would have practical relevance, unlike previous variables, and thus
255 warranted further investigation.

256 As described in the previous publication which defined the Peptide RPC Column Characterisation
257 Protocol, PCA is used in order to visualise similarities and differences between delta values and
258 columns in so called biplots [3,18]. The delta values for the ammonium formate and formic acid
259 DoE's were included in a PCA to evaluate the robustness of the method on the protocols ability to

260 differentiate differing stationary phases from one another and to see what limits of the proposed
261 MeCN levels had on the integrity of the results. As can be seen in the resulting biplot (Fig. 3), most of
262 the DoE runs are clustered close to the origin except for a small subset of runs which are located
263 further down in the lower left quadrant. Further evaluation indicated the subset was pulled away by
264 the -1 level for the MeCN content in the B solvent.

265 A loss of MeCN could possibly be expected by evaporation of solvent in the mobile phase reservoirs
266 over a period of time. An evaporation study was therefore performed to ascertain what could be
267 reasonably lost via the LC solvent caps. An Agilent Valve cap (Waldbronn, Germany) and a SCAT
268 safety cap (Mörfelden-Walldorf, Germany) were compared against a closed cap used for solvent
269 storage. There was 0.00% loss in weight for the closed cap over 30 days, suggesting acetonitrile is
270 not lost during storage, however, it was calculated that losses of 0.04 and 0.03% for the Agilent and
271 SCAT caps, respectively, could be expected per day, which could prove practically problematic.

272 In order to combat this, the approach was taken to change the B solvent from 20 mM ammonium
273 formate in MeCN/H₂O (80:20 w/w) to 100% MeCN. The gradient was adjusted accordingly to achieve
274 the same volume fraction of MeCN and compared against the original method, with similar
275 chromatographic results, regardless of the reduced buffer concentration in the B solvent (the
276 ammonium formate concentration is reduced from 18 to 12 mM during the part of the gradient
277 where peptides typically elute).

278

279 **3.2 Instrument Variability**

280 The ability to successfully translate LC methodologies between differing LC instrumentation is
281 extremely important for the widespread acceptance of the protocol. This is particularly important
282 with respect to gradient chromatography where the contribution of column volume and dwell
283 volume has been shown to cause considerable selectivity differences [19].

284 The three peptide mixtures were injected onto an Acquity HSS C18 column on three different
285 instruments, Agilent 1290, Waters Acquity H- and I-class ranging in dwell volume between 300 and
286 700 μ L, which was larger than the range assessed within the DoE. The same batches of mobile
287 phases and column were used on all the comparisons to remove any variation attributed to mobile
288 phase preparation differences and understand the instrument contribution to the study.

289 The delta values were recorded for each instrument and compared, after which the data was
290 analysed by PCA and placed in the biplot (Fig. 3). The three instruments are circled within a 95%

291 confidence limit. This variation is comparable to what can be expected from the DoE thus indicating
292 results obtained on different type of instruments should be comparable.

293

294 **3.3 Column Batch to Batch Variability**

295 The column batch to batch variability was assessed using six Ascentis Express C18 columns. All
296 columns differed by silica, whilst three columns contained the same batch of silane with three
297 additional silane batches used for the remaining columns.

298 All batch to batch columns were tested using the new protocol with the reduced number of probes
299 (i.e. removal of probes susceptible to changes in MeCN) on the same occasion and mobile phases to
300 remove their contribution to any variability.

301 The batch to batch observations can be seen encircled within the biplot (Fig. 4) where the scatter
302 seen is due to the batch to batch variation as the data was collected on the same day using the same
303 instrument and solvents to eliminate their contribution from the results. The results are also in
304 keeping with previous batch to batch studies performed on other columns using various protocols
305 [9,20-26]. This highlights that any deviation between columns within the biplot is caused by
306 selectivity differences, thus it is feasible to distinguish stationary phases which are
307 chromatographically similar or dissimilar using this approach.

308

309 **3.4 Slow Equilibration**

310 Slow equilibration of certain RP materials when changing from intermediate to low pH has been well
311 documented by Snyder *et al.* It is exhibited by a steady retention time drift on changing the mobile
312 phase pH. It is believed that approximately 40% of all commercially available stationary phases
313 exhibited some form of slow equilibration [27]. The exact mechanism for this phenomenon is not
314 known, but it is speculated that with the advent of modern silica with low surface charge, changes in
315 pH can require significant time to re-equilibrate which is displayed as retention drift for ionisable
316 species.

317 A selection of C18 type phases were assessed using the peptides as probes to determine any
318 practical constraints for the Peptide RPC Column Characterisation protocol, which utilises both low
319 and intermediate pH.

320 The peptide test mixture was repeatedly injected on a C18 stationary phase using the formic acid
321 gradient conditions with consistent results (Fig. 5(A)). The phase was then exposed to the
322 ammonium formate gradient conditions, which saw quick equilibration of the peptide mixture within

323 duplicate injections, suggesting consistent results can be achieved when moving from low to
324 intermediate conditions.

325 However, when the same column was then re-exposed to the formic acid gradient, it failed to yield
326 results comparable to those prior to exposure to intermediate pH (slow reduction in retention – see
327 Fig. 5(B)). The retention times for all peaks had increased, but were consistently decreasing in
328 retention between injections.

329 Literature has suggested static equilibration can restore a column which exhibits slow equilibration
330 [27-29], however, even an overnight static equilibration in low pH conditions failed to restore this
331 stationary phase to its original chromatographic retentivity (Fig. 5(C)).

332 This phenomenon is not exclusive to just formic acid, but between any switch from intermediate or
333 high pH to low pH for ionisable species. The issue would also not be corrected with the use of TFA, as
334 the ion pairing effect of this additive would mean dedicated columns would be required which
335 would be impractical for column characterisation [30]. Although there are stationary phases which
336 have been devised to combat this slow equilibration issue such as the Acquity CSH range of phases
337 [31], there are a number of commercially available columns which do exhibit this phenomenon. As
338 such, the decision was made to characterise each stationary phase initially using the formic acid
339 gradient, before testing at intermediate pH using ammonium formate in order to avoid any
340 detrimental retention drifts or the necessity for excessive equilibration times.

341

342 **3.5 Loadability**

343 Chromatographic efficiency can be highly susceptible to analyte overloading which contributes to
344 poor peak shape, especially for ionisable species. The permissible load before chromatographic
345 performance is affected is often substantially lower for peptides and protein separations [31]. A
346 significant amount of literature is available on the subject of loadability of basic species, and
347 rationalising overloading effects [32-34].

348 A stock solution of [L-Asp³]-Bovine GLP-2 (1-15) (1 mg/mL) underwent a series of dilutions using
349 DMSO/H₂O (80:20 v/v). Each solution was reproducibly injected onto the Kinetex Evo C18 (150 x 2.1
350 mm, 2.6 μm) using the formic acid gradient chromatographic conditions. The low ionic strength of
351 formic acid is a worst-case scenario; thus, it was selected to observe the effects of loadability and
352 overloading.

353 Eight dilutions were made in total (four shown for simplicity in Fig. 6) with the resulting
354 chromatograms overlaid. The hydrophilic peptide, which has a net charge of +1.2 at pH 2.5,

355 displayed the characteristic “shark fin” peak shape with increased load for a positively charged
356 species in acidic conditions i.e. a typical right-angled front and extreme tailing. The apex of the peak
357 is used to measure the retention time decrease with increased load on the column and as such,
358 would affect the normalised retention times used to calculate delta values. The degree to which this
359 effect occurs could be different for each peptide, hence the load for each peptide must be well
360 described in order to maintain consistent delta values. The peak shape for this study is not of critical
361 concern as it is the retention time which must be consistent, hence the necessity to keep the load
362 constant. When devising chromatographic methods, it would be crucial to select mobile phases
363 which can provide better peak shape, with the biopharmaceutical industry typically utilising
364 phosphate salt based systems.

365 The loading profile could also be different depending on the type of stationary phase used. For
366 example, the Acquity CSH range of stationary phases were optimised to provide improved peak
367 shape and efficiency for basic species to provide linear isotherms, rather than typical Langmuir
368 isotherm (Fig. 6). Similar to the slow equilibration effect, overloading behaviour is thought to be due
369 to variations in the surface charge, where the balanced surface charge of the CSH range counteracts
370 that issue to produce symmetrical, efficient peaks [31].

371 The sample solubility also is critical when it comes to sample load. The net charge of the hydrophilic
372 peptides at pH 2.5 and 6.5 are +1.1 to +1.2 and -4.7 to -3.7, respectively. The hydrophobic peptides
373 have a net charge of +2.2 at pH 2.5, whilst at pH 6.5 the net charge is 0.0. A pI of 0.0 may highlight
374 potential solubility problems and the possibility of precipitation and clogging of the inlet frit.

375 Pressure increases and decreased column performance were observed after prolonged exposure to
376 intermediate pH conditions. Replacement of the inlet frit and scanning electron microscopy proved
377 that particulates had been deposited onto the frit. Hence, in order to minimise the likelihood of this
378 happening the load of the peptides was reduced and inline filters installed between the injector and
379 the column.

380

381 **3.6 Mitigating Action: Adjustment of the Peptide RPC Column Characterisation Protocol**

382 The ideal scenario from the reduced factorial design is to find that none of the parameters assessed
383 are statistically or practically relevant. However, if there are parameters which are practically
384 relevant, mitigation must be put in place to reduce the effect.

385 The tolerances for the systematic shift in the gradient, the flow rate and the dwell volume in both
386 the formic acid and ammonium formate DoE's, and the pH of ammonium formate were all
387 acceptable. Despite being deemed statistically significant, the practical relevance of the temperature

388 limits in both DoE's, and the % formic acid in both the aqueous and organic was negligible, thus the
389 method for characterisation using formic acid can be considered robust within the stated limits.

390 The concentration of acetonitrile in the B solvent for the ammonium formate gradient, however, has
391 been shown to be a statistically significant result with practical implications. The decrease in
392 concentration had ramifications on four of the six delta values, leading to the removal of the
393 sensitive $\Delta(3,1)$, $\Delta(9,1)$ and $\Delta(10,9)$ probes from the characterisation. The removed delta values were
394 probes for racemisation, which was still represented by $\Delta(9,1)$ and $\Delta(14,13)$ in formic acid. In
395 addition, the integrity of the score plot was also assessed without the affected probes, with
396 consistent results for the stationary phases.

397 It is believed the robustness will actually be significantly better than shown in this study since the
398 mitigations introduced (Table 7) will reduce variation further. The updated Peptide RPC Column
399 Characterisation protocol is described in Appendix I.

400

401 **4 Conclusion**

402 The robustness of the Peptide RPC Column Characterisation protocol was assessed using reduced
403 factorial design and PCA, with various factors systematically altered to deduce the impact on subtle
404 changes to the protocol. The results indicate the formic acid gradient can be seen to provide robust
405 results within the given tolerances of this study. The ammonium formate gradient, however,
406 required mitigation to improve robustness in regards to the concentration of acetonitrile in the B
407 solvent. All other parameters assessed did not influence the robustness.

408 The sample load for the columns was also determined and the potential impact on switching
409 between low and intermediate pH for certain commercially available stationary phases was
410 evaluated. Both studies had ramifications for the protocol and mitigation was put in place to address
411 both phenomena.

412 In addition, the instrument variability on three different LC configurations and column batch to
413 batch variability was assessed to ascertain the degree of variability which could be expected. Both
414 the LC and column variability were minimal and highlights that the differences between stationary
415 phases observed using the Peptide RPC Column Characterisation protocol are caused by
416 chromatographic selectivity differences, rather than random error.

417 A number of modifications were suggested in order to improve robustness, thus moving forward,
418 the results should provide even greater reliability and reproducibility than shown in this study. This
419 will offer greater confidence in characterising different stationary phases using peptides as probes

420 and distinguishing their selectivity differences, thus allowing complementary stationary phases to be
421 selected for method development or similar columns for back-up methods.

422

423 **5 Acknowledgments**

424 With thanks to Shimadzu for the supply of the Nexera X2 instrument and mass spectrometer and to
425 Agilent, Fortis, Phenomenex, Supelco and Waters for the supply of stationary phases. Finally, special
426 thanks to Waters for performing the SEM experiments.

427

428

429 **Appendix I: Description of the Peptide RPC Column Characterisation Protocol for 150 x 2.1 mm**
 430 **Column Formats**

431 If different column dimensions are employed, it is recommended that the user employs method translation tools [19].

Parameter	Protocol												
Mobile Phase	A1: 0.1% ($\pm 0.005\%$) v/v formic acid in water – Add 1.000 mL formic acid to 999.0 g (± 0.01 g) water B1: 0.1% ($\pm 0.005\%$) v/v formic acid in acetonitrile – Add 1.000 mL formic acid to 785.2 g (± 0.01 g) acetonitrile A2: 20 mM Ammonium formate in water – Add 100.0 g (± 0.01 g) 200 mM ammonium formate stock solution to 900.0 g (± 0.01 g) water B2: Acetonitrile												
Stock Buffer	200 mM Ammonium formate pH 6.45 (± 0.06) – Dissolve 1.261 g in 100.0 g (± 0.01 g) water and measure the pH using an appropriately calibrated probe												
Gradient	<table border="1"> <thead> <tr> <th>Time</th> <th>%B</th> </tr> </thead> <tbody> <tr> <td>0.0</td> <td>4.5</td> </tr> <tr> <td>40.0</td> <td>50.0</td> </tr> <tr> <td>42.0</td> <td>50.0</td> </tr> <tr> <td>42.1</td> <td>4.5</td> </tr> <tr> <td>54.1</td> <td>4.5</td> </tr> </tbody> </table>	Time	%B	0.0	4.5	40.0	50.0	42.0	50.0	42.1	4.5	54.1	4.5
Time	%B												
0.0	4.5												
40.0	50.0												
42.0	50.0												
42.1	4.5												
54.1	4.5												
Flow Rate	0.3 mL/min (± 0.005 mL/min)												
Column Oven Temperature	40 °C (± 2 °C)												
Autosampler Temperature	Recommend 10 °C												
Wavelength	215 nm Ref 360 nm (band width 8 and 100 nm)												
MS	Selected Ion Monitoring (z=2)												
Dwell Volume	100 – 500 μ L												
Sample Concentration & Diluent	0.25 mg/mL in DMSO/H ₂ O (80:20 v/v)												

432

433 The run order is of great importance and should be first assessed at low pH then intermediate pH.

Test Mixture	Peptide Number	Peptide	Rationale	m/z	Load (μ g)
TM1	1	Bovine GLP-2 (1-15)	Original sequence	820	0.250
	8a	[Met(O)10]-Bovine GLP-2 (1-15)	Oxidation	828	0.250
	9	[L-Asp11]-Bovine GLP-2 (1-15)	Deamidation / Negative charge	820	0.125
	13	Bovine GLP-2 (16-33)	Original sequence	1069	0.250
	15	[Ile26,Leu27]-Bovine GLP-2 (16-33)	Switch in AA sequence	1069	0.075
TM2	8a	[Met(O)10]-Bovine GLP-2 (1-15)	Oxidation	828	0.250
	13	Bovine GLP-2 (16-33)	Original sequence	1069	0.250
	15	[Ile26,Leu27]-Bovine GLP-2 (16-33)	Switch in AA sequence	1069	0.075
	24	[Tyr26]-Bovine GLP-2 (16-33)	Phenolic effect	1076	0.125

	26	[Lys26]-Bovine GLP-2 (16-33)	Positive charge	1094	0.250
	8a	[Met(O)10]-Bovine GLP-2 (1-15)	Oxidation	828	0.250
	13	Bovine GLP-2 (16-33)	Original sequence	1069	0.250
TM3	14	[D-Ser16]-Bovine GLP-2 (16-33)	Racemisation	1069	0.125
	15	[Ile26,Leu27]-Bovine GLP-2 (16-33)	Switch in AA sequence	1069	0.075
	16	[L-Asp21,Gly22]-Bovine GLP-2 (16-33)	Loss of aromatic group / Racemisation	1024	0.125

434

Test Mixture	Delta	Measured in Formic Acid	Measured in Ammonium Formate
	$\Delta(8a,1)$	✓	
TM1	$\Delta(9,1)$	✓	
	$\Delta(15,13)$		✓
	$\Delta(24,13)$		✓
TM2	$\Delta(26,13)$	✓	✓
	$\Delta(14,13)$	✓	
TM3	$\Delta(16,13)$	✓	

435

436

437

438 **6 Reference**

- 439 [1] ACD Column Selection Database. <https://www.acdlabs.com/resources/freeware/colsel/>
440 [Accessed: 23/10/2018]
- 441 [2] U.S. Pharmacopeial Convention <http://apps.usp.org/app/USPNF/columnsDB.html> [Accessed:
442 23/10/2018]
- 443 [3] M.R. Euerby and P. Petersson, Chromatographic classification and comparison of commercially
444 available reversed-phase liquid chromatographic columns using principal component analysis, *J.*
445 *Chromatogr. A* **994** (2003) 13-36
- 446 [4] N.S. Wilson, M.D. Nelson, J.W. Dolan, L.R. Snyder, R.G. Wolcott, P.W. Carr, Column selectivity in
447 reversed-phase liquid chromatography I. A general quantitative relationship, *J. Chromatogr. A* **961**
448 (2002) 171-193
- 449 [5] C. West, E. Lemasson, S. Bertin, P. Hennig and E. Lesellier, An improved classification of stationary
450 phases for ultra-high performance supercritical fluid chromatography, *J. Chromatogr. A* **1440** (2016)
451 212-228
- 452 [6] J.K. Field, M.R. Euerby, P. Petersson, J. Lau, H. Thørsen, Investigation into reversed-phase
453 chromatography peptide separation systems Part I: Development of a protocol for column
454 characterisation, *J. Chromatogr. A* (Submitted for publication Nov 2018)
- 455 [7] B. Dejaegher and Y.V. Heyden, Experimental designs and their recent advances in set-up, data
456 interpretation and analytical applications, *J. Chromatogr. A* **1158** (2007) 158-167
- 457 [8] M.E Swartz and I. Krull, Method Validation and Robustness, *LCGC North America*, **24** (2006) 480-
458 490
- 459 [9] Guidelines for Industry, Text on Validation of Analytical Procedures, ICH-Q2A, November 2005
- 460 [10] P. Petersson and M.R. Euerby, An evaluation of the robustness of the Tanaka characterization
461 protocol for reversed-phase liquid chromatography columns, *J. Sep. Sci.* **28** (2005) 2120-2129
- 462 [11] G.E.P. Box, J.S. Hunter and W.G. Hunter, Statistics for Experimenters: Design, Innovation, and
463 Discovery, 2nd Ed. Wiley, New Jersey 2005
- 464 [12] V. Czitrom, One-Factor-at-a-Time Versus Designed Experiments, *The American Statistician* **53**
465 (1999) 126-131

- 466 [13] J.C. Miller and J.N. Miller, *Statistics for Analytical Chemistry*, 3rd Ed. Ellis Horwood Limited,
467 Chichester 1993
- 468 [14] P. Petersson, B.O. Boateng, J.K. Field, M.R. Euerby, A practical approach to modelling of
469 reversed-phase liquid chromatographic separations: Advantages, principles and possible pitfalls,
470 *LCGC Europe* **31** (2018) 120-143
- 471 [15] D.C. Montgomery, E.A. Peck and G.G Vining, *Introduction to Linear Regression Analysis*, 5th Ed.
472 Wiley & Sons, Inc, New Jersey, 2012
- 473 [16] Uwe D Neue *HPLC Columns: Theory, Technology and Practice*, 1st Ed. Wiley-VCH, Inc 1997
- 474 [17] A. Méndez, E. Bosch, M. Rosés and U.D. Neue, Comparison of the acidity of residual silanol
475 groups in several liquid chromatography columns, *J. Chromatogr. A* **986** (2003) 33-44
- 476 [18] I.T. Jolliffe, *Principal Component Analysis*, Springer, Berlin 2011
- 477 [19] P. Petersson, M.R. Euerby and M.A. James, Translations between differing liquid
478 chromatography formats: Advantages, principles and possible pitfalls, *LCGC North America* **32** (2014)
479 558-567
- 480 [20] M. Kele and G. Guiochon, Repeatability and reproducibility of retention data and band profiles
481 on reversed-phase liquid chromatography columns: I. Experimental protocol, *J. Chromatogr. A* **830**
482 (1999) 41-54
- 483 [21] M. Kele and G. Guichon, Repeatability and reproducibility of retention data and band profiles on
484 reversed-phase liquid chromatographic columns: II. Results obtained with Symmetry C18 columns, *J.*
485 *Chromatogr. A* **830** (1999) 55-79
- 486 [22] M. Kele and G. Guichon, Repeatability and reproducibility of retention data and band profiles on
487 reversed-phase liquid chromatography columns: III. Results obtained with Kromasil C18 columns, *J.*
488 *Chromatogr. A* **855** (1999) 423-453
- 489 [23] M. Kele and G. Guichon, Repeatability and reproducibility of retention data and band profiles on
490 reversed-phase liquid chromatography columns: IV. Results obtained with Luna C18 (2) columns, *J.*
491 *Chromatogr. A* **869** (2000) 181-209
- 492 [24] M. Kele and G. Guichon, Repeatability and reproducibility of retention data and band profiles on
493 reversed-phase liquid chromatography columns: IV. Results obtained with Vydac 218TP C18
494 columns, *J. Chromatogr. A* **913** (2001) 89-112

- 495 [25] M. Kele and G. Guichon, Repeatability and reproducibility of retention data and band profiles on
496 six batches of monolithic columns: IV. Results obtained with Luna C18 (2) columns, *J. Chromatogr. A*
497 **960** (2002) 19-49
- 498 [26] U.D. Neue, C.H. Phoebe, K. Tran, Y.F. Cheng and Z. Lu, Dependence of reversed-phase retention
499 of ionizable analytes on pH, concentration of organic solvent and silanol activity, *J. Chromatogr. A*
500 **925** (2001) 49-67
- 501 [27] D.H. Marchand, L.A. Williams, J.W. Dolan and L.R. Snyder, Slow equilibration of reversed-phase
502 columns for the separation of ionized solutes, *J. Chromatogr. A* **1015** (2003) 53-64
- 503 [28] D.V. McCalley, Overload for Ionized Solutes in Reversed-Phase High-Performance Liquid
504 Chromatography, *Anal. Chem.*, **78** (2006) 2532-2538
- 505 [29] D.V. McCalley, Study of Overloading of Basic Drugs and Peptide in Reversed-Phase High-
506 Performance Liquid Chromatography using pH Adjustment of Weak Acid Mobile Phases Suitable for
507 Mass Spectrometry, *J. Chromatogr. A.*, **1075** (2005) 57-64
- 508 [30] J.W. Dolan, Ion Pairing – Blessing or Curse? *LCGC Europe* **21** (2008) 258-263
- 509 [31] P.C. Iraneta, K.D. Wyndham, D.R. McCabe and T.H. Walter, *A Review of Waters Hybrid Particle*
510 *Technology. Part 3. Charged Surface Hybrid (CSH) Technology and Its Used in Liquid*
511 *Chromatography*, (2011)
- 512 [32] S.M.C. Buckenmaier, M.R. Euerby and D.V. McCalley, Overloading study of bases using
513 polymeric RP-HPLC columns as an aid to rationalization of overloading on silica-ODS phases, *Anal.*
514 *Chem.* **74** (2002) 4672-4681
- 515 [33] M.M. Fallas, S.M.C. Buckenmaier, D.V. McCalley, A comparison of overload behaviour for some
516 sub 2 μm totally porous and sub 3 μm shell particle columns with ionised solutes, *J. Chromatogr. A*
517 **1235** (2012) 49-59
- 518 [34] G.B. Cox and L.R. Snyder, Preparative high-performance liquid chromatography under isocratic
519 conditions: III. The consequences of two adjacent bands having unequal column capacities, *J*
520 *Chromatogr. A* **483** (1989) 95-110
- 521 [35] M.M. Fallas, M. Hadley and D.V. McCalley, Practical assessment of frictional heating effects and
522 thermostat design on the performance of conventional (3 μm and 5 μm) columns in reversed-phase
523 high -performance liquid chromatography, *J. Chromatogr. A* **1216** (2009) 3961-3969

524 [36] R.G. Wolcott, J.W. Dolan, L.R. Snyder, S.R. Bakalyar, M.A. Arnold and A.J. Nichols, Control of
525 column temperature in reversed-phase liquid chromatography, *J. Chromatogr. A* **869** (2000) 211-230

526

527 **7 Figure Captions**

528 Fig. 1. Coefficient plots for each delta response for the different variables based on formic acid data.

529 Fig. 2. Coefficient plots for each response for the different variables based on ammonium formate
530 data.

531 Fig. 3. A biplot of the columns used to develop the Peptide RPC Column Characterisation protocol in
532 addition to the instrument variability (yellow triangles) and the robustness (light blue diamonds)
533 results to assess the variability of the protocols using the 11 delta value responses.

534 Fig. 4. Biplot of the batch to batch reproducibility performed on six Ascentis Express C18 columns on
535 the reduced number of delta values, indicating the variability which could be expected between
536 different batches of silica and silanes. Fig. 4 is mirrored in comparison to Fig. 3 due to one of the
537 features of PCA where a very small difference in data often result in an axis being mirrored. The
538 pattern and groupings are, however, the same in both figures.

539 Fig. 5. Chromatograms demonstrating the effect of slow equilibration. (A) the original chromatogram
540 in formic acid prior to exposure to ammonium formate at intermediate pH, (B) re-evaluation of the
541 same column in formic acid after exposure to intermediate pH (C) re-evaluation after a static
542 equilibration in formic acid to attempt to restore the original chromatography

543 Fig. 6. Overlaid peak profile of the hydrophilic peptide [D-Asp³]-Bovine GLP-2 (1-15) of differing
544 loads (0.031 – 2.000 µg) using the formic acid gradient conditions. The peak demonstrated the
545 characteristic Langmuir's isotherm with significant peak tailing with increased sample load

Table 1

The peptides used to calculate the delta values in (A) formic acid and (B) ammonium formate, with the accompanying rationale and typical Δt_g and Δt_g^* for the Kinetex Evo C18 under the nominal conditions

(A)

Change	Delta	Peptide Number	Peptide	Rationale	Δt_g	Δt_g^*
[Met10] → [Met(O)10]	$\Delta(8a,1)$	8a 1	[Met(O)10]-Bovine GLP-2 (1-15) [Met10]-Bovine GLP-2 (1-15)	Oxidation	-3.840	-0.286
[L-Asn11] → [L-Asp11]	$\Delta(9,1)$	9 1	[L-Asp11]-Bovine GLP-2 (1-15) [L-Asn11]-Bovine GLP-2 (1-15)	Increase in negative charge	0.729	0.054
[L-Ser16] → [D-Ser16]	$\Delta(14,13)$	14 13	[D-Ser16]-Bovine GLP-2 (16-33) [L-Ser16]-Bovine GLP-2 (16-33)	Steric - racemisation	0.171	0.013
[Phe22] → [Gly22]	$\Delta(16,13)$	16 13	[Gly22]-Bovine GLP-2 (16-33) [Phe22]-Bovine GLP-2 (16-33)	Aromatic – removal of aromatic group	-5.302	-0.395
[Leu26] → [Lys26]	$\Delta(26,13)$	26 13	[Lys26]-Bovine GLP-2 (16-33) [Leu26]-Bovine GLP-2 (16-33)	Increase in positive charge	-7.611	-0.566

(B)

Change	Delta	Peptide Number	Peptide	Rationale	Δt_g	Δt_g^*
[L-Asp3] → [D-Asp3]	$\Delta(3,1)$	3 1	[D-Asp3]-Bovine GLP-2 (1-15) [L-Asp3]-Bovine GLP-2 (1-15)	Steric - racemisation	0.064	0.003
[L-Asn11] → [L-Asp11]	$\Delta(9,1)$	9 1	[L-Asp11]-Bovine GLP-2 (1-15) [L-Asn11]-Bovine GLP-2 (1-15)	Increase in negative charge	1.399	0.063
[L-Asp11] → [D-Asp11]	$\Delta(10,9)$	10 9	[D-Asp11]-Bovine GLP-2 (1-15) [L-Asp11]-Bovine GLP-2 (1-15)	Steric - racemisation	-0.581	-0.026
[Leu26,Ile27] → [Ile26,Leu27]	$\Delta(15,13)$	15 13	[Ile26,Leu27]-Bovine GLP-2 (16-33) [Leu26,Ile27]-Bovine GLP-2 (16-33)	Steric – switch in amino acid sequence	1.053	0.047
[Leu26] → [Tyr26]	$\Delta(24,13)$	24 13	[Tyr26]-Bovine GLP-2 (16-33) [Leu26]-Bovine GLP-2 (16-33)	Aromatic and phenolic – addition of phenolic group	-3.959	-0.179
[Leu26] → [Lys26]	$\Delta(26,13)$	26 13	[Lys26]-Bovine GLP-2 (16-33) [Leu26]-Bovine GLP-2 (16-33)	Increase in positive charge	-6.526	-0.295

Table 2

Peptide sequence of Bovine GLP-2

Amino Acid #	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33
Bovine GLP-2	H	A	D	G	S	F	S	D	E	M	N	T	V	L	D	S	L	A	T	R	D	F	I	N	W	L	I	Q	T	K	I	T	D

Table 3

The peptide test mixtures, rationale and sample load

Test Mixture	Peptide Number	Peptide	Rationale	Load (µg)
TM1	1	Bovine GLP-2 (1-15)	Original sequence	0.250
	8a	[Met(O)10]-Bovine GLP-2 (1-15)	Oxidation	0.250
	9	[L-Asp11]-Bovine GLP-2 (1-15)	Deamidation / Negative charge	0.125
	13	Bovine GLP-2 (16-33)	Original sequence	0.250
	15	[Ile26,Leu27]-Bovine GLP-2 (16-33)	Switch in AA sequence	0.075
	16	[L-Asp21,Gly22]-Bovine GLP-2 (16-33)	Loss of aromatic group / Racemisation	0.125
	24	[Tyr26]-Bovine GLP-2 (16-33)	Aromatic / Phenolic effect	0.125
TM2	8a	[Met(O)10]-Bovine GLP-2 (1-15)	Oxidation	0.250
	9	[L-Asp11]-Bovine GLP-2 (1-15)	Deamidation / Negative charge	0.125
	10	[D-Asp11]-Bovine GLP-2 (1-15)	Deamidation / Racemisation / Negative charge	0.250
	13	Bovine GLP-2 (16-33)	Original sequence	0.250
	15	[Ile26,Leu27]-Bovine GLP-2 (16-33)	Switch in AA sequence	0.075
	26	[Lys26]-Bovine GLP-2 (16-33)	Positive charge	0.250
TM3	1	Bovine GLP-2 (1-15)	Original sequence	0.250
	3	[D-Asp3]-Bovine GLP-2 (1-15)	Racemisation	0.125
	8a	[Met(O)10]-Bovine GLP-2 (1-15)	Oxidation	0.250
	13	Bovine GLP-2 (16-33)	Original sequence	0.250
	14	[D-Ser16]-Bovine GLP-2 (16-33)	Racemisation	0.125
	15	[Ile26,Leu27]-Bovine GLP-2 (16-33)	Switch in AA sequence	0.075

Table 4

Operating parameters investigated in the DoE, including the nominal conditions (0 level) and the expected deviation (± 1 levels)

Parameter	-1 Level	0 Level	+1 Level
Column Temperature ($^{\circ}\text{C}$) ¹	38	40	42
Flow Rate (mL/min) ¹	0.295	0.300	0.305
Systematic Shift in Gradient (%B) ¹	-0.4	0.0	+0.4
Dwell Volume (μL) ¹	100	300	500
Volume Formic Acid in Aqueous (% v/v) ²	0.095	0.100	0.105
Volume Formic Acid in Organic (% v/v) ²	0.095	0.100	0.105
pH of Ammonium Formate ³	6.39	6.45	6.51
MeCN Composition in B (%) ³	79.9	80.0	80.1

¹ Measured using both the formic acid and ammonium formate gradients

² Measured in just the formic acid gradient

³ Measured in just the ammonium formate gradient

Table 5

The DoE design for the formic acid gradient, including the different experimental levels. N1-8 vary the ± 1 levels whilst N9-11 are the nominal central conditions used to assess reproducibility

Experiment Name	Column Temperature (°C)	Systematic Shift in Gradient (%B)	Flow Rate (mL/min)	Volume of Formic Acid in Aqueous (% v/v)	Volume of Formic Acid in Organic (% v/v)	Dwell Volume (µL)
N1	-	-	-	+	+	+
N2	+	-	-	-	-	+
N3	-	+	-	-	+	-
N4	+	+	-	+	-	-
N5	-	-	+	+	-	-
N6	+	-	+	-	+	-
N7	-	+	+	-	-	+
N8	+	+	+	+	+	+
N9	0	0	0	0	0	0
N10	0	0	0	0	0	0
N11	0	0	0	0	0	0

Table 6

The DoE design for the ammonium formate gradient, including the different experimental levels. N1-8 vary the ± 1 levels whilst N9-11 are the nominal central conditions used to assess reproducibility

Experiment Name	Column Temperature (°C)	Systematic Shift in Gradient (%B)	Flow Rate (mL/min)	pH of Stock Ammonium Formate Solution	MeCN Composition in B (%)	Dwell Volume (µL)
N1	-	-	-	+	+	+
N2	+	-	-	-	-	+
N3	-	+	-	-	+	-
N4	+	+	-	+	-	-
N5	-	-	+	+	-	-
N6	+	-	+	-	+	-
N7	-	+	+	-	-	+
N8	+	+	+	+	+	+
N9	0	0	0	0	0	0
N10	0	0	0	0	0	0
N11	0	0	0	0	0	0

Table 7

Mitigation to increase the robustness and reliability of the Peptide RPC Column Characterisation protocol, including the rationale for each action

Mitigating Action	Rationale
Prepare solvents by weight rather than volume	There are greater errors associated with glassware, thus more reproducible mobile phases can be prepared by weight.
Assess the accuracy of the pipette before each use	To ensure the pipette can accurately dispense formic acid.
Measure the pH of the stock ammonium formate solution (6.39-6.51).	The salt container should be carefully capped to avoid loss of ammonia which can result in a lower pH. In addition, measures should be put in place to reduce the effect of hygroscopicity (i.e. use a desiccator, avoid using salt which has significant clump formation).
Use 100% MeCN instead of 20 mM ammonium formate in MeCN/H ₂ O (80:20 w/w) in the B solvent combined with a corresponding change in gradient slope.	The loss of acetonitrile in the B solvent causes significant differences for certain delta values. Changing to 100% MeCN addresses this problem. Even with the change in ammonium formate concentration, the results are still better with this change.
Remove $\Delta(3,1)$, $\Delta(9,1)$ and $\Delta(10,9)$ measured in ammonium formate	Improve the robustness of the procedure as they were sensitive to changes in MeCN. Although these delta values had some influence within the loading plot, they can be removed with minimal effect on the score plot and the remaining probes cover the range of interactions which should be investigated.
Use reference peptides in each test mixture	Allows retention times to be normalised for direct comparison between different batches of solvent, different analysts and removes the contribution from the dwell volume and column volume.
Each test mixture should contain the two peptides used to create the delta value	Removes any random injection to injection variation of retention time in addition to fluctuations in temperature or mobile phase composition.
Characterise the stationary phase in formic acid prior to ammonium formate	Removes the effect of slow equilibration and retention drifts.
Use a specific load for each peptide	Changing the load on the column can cause changes in retention which will impact on the delta value produced.
Assess the actual temperature of the column	Column oven designs can create as much as ± 5 °C difference [35,36], which can impact significantly on selectivity. Obtaining the actual temperature of the oven enables the end-user to adjust the temperature appropriately for direct comparisons of different column oven designs.

Solvent bottles should be stored correctly with a cap and stored at 5 °C. When stored on the system, a good vapour valve should be installed

Reduce load of the peptides and add an inline filter to induce mixing and trapping of particles prior to column

Use a reference column to act as a system suitability test

Storage of capped solvent bottles in the fridge reduces microbial growth and evaporation, whilst the vapour valve prevents dust / microbes entering the chromatographic system and acetonitrile losses.

The hydrophobic peptides have a pI of 0.0 in ammonium formate at pH 6.45, which could cause solubility issues such as precipitation on the frit at the head of the column. This can cause bad peak shapes, increased pressures and reduced column lifetime. By reducing the load and introducing an inline filter, it will reduce the risk for precipitation and increase the robustness of the protocol.

This provides a baseline for the instrument to detect any differences in any asymmetrical shifts in the gradient (as well as other problems).











