

Metals transfer from tobacco to cigarette smoke: Evidences in smokers' lung tissue

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A B S T R A C T

Tobacco use kills millions of people every year around the world. The current level of 11 metals in tobacco was determined and their transfer rate to cigarette smoke was calculated as the difference between the total metal content in cigarettes and the amount present in its ashes. The metals content was also determined in the lung tissue of smokers and non-smokers in order to evaluate the marks that smoking leaves in this tissue. Metals content in tobacco ranged from less than 1 $\mu\text{g/g}$ (Co, Cd, Pb, As and Tl) to several hundreds of $\mu\text{g/g}$ (Al, Mn and Ba). The highest transfer rate from tobacco to cigarette smoke was found for Tl (85–92%) and Cd (81–90%), followed by Pb (46–60%) and As (33–44%). Significantly higher levels of As, Cd and Pb were found in the lung tissue of smokers compared to non-smokers, showing that smoking results in an increase of these metals in the lungs and that they contribute to the carcinogenic potential of cigarette smoke. This study presents important data on current metals content in tobacco and its transference to cigarette smoke and provides evidence of their accumulation in smokers' lung tissue.

Keywords: ICP-MS Lungs Smoking Cigarette Metals

Introduction

Tobacco use continues to kill millions of people around the world, with cigarettes being the most widely used tobacco product. Nowadays, about 12% of all deaths among adults aged 30 years and over are attributed to tobacco [1]. It is estimated that globally tobacco kills around 6 million people each year: five million from direct tobacco smoking while 600,000 deaths are attributable to second-hand smoke effects [2]. On current smoking trends, the annual death toll from tobacco is expected to rise to around 10 million people by 2030 [3].

Smoking negatively impacts on health across the life-course and dramatically reduces both quality of life and life expectancy. The negative impacts of tobacco smoking includes, among others, increased rates of cardiovascular-related death (e.g., ischemic heart disease and stroke) [4], high rates of cancers (especially lung cancer), and death associated with diseases of the respiratory system, including tuberculosis and pneumonia [5].

Direct tobacco smoking and involuntary smoking (exposure to secondhand smoke) have been classified by the International Agency for Research on Cancer (IARC) as agents considered carcinogenic to humans (classified as group 1 exposure circumstances) [6]. Tobacco smoke is a complex and dynamic chemical mixture. Researchers have estimated that tobacco smoke contains more than 7000 chemical compounds from many different classes. Thus, tobacco smoke can be an important source of known toxic compounds such as nitrosamines, polycyclic aromatic hydrocarbons (PAHs), pesticides, heterocyclic and aromatic amines and metals [7]. Among the many harmful and toxic compounds found in tobacco and tobacco smoke, metals seem to play an important role in the overall hazard [8,9]. As with all plants, tobacco plants require several nutrients to sustain their physiological functions, which they acquire from the soil. However, both essential and toxic elements can enter and accumulate in plant tissues [10,11]. Thus, the main source of metals in tobacco plant is the soil.

Among metals, the most commonly associated with health effects are arsenic (As),¹ cadmium (Cd), chromium (Cr), nickel (Ni) and lead (Pb). As, Cd and Ni are classified as Group 1 carcinogens for humans by IARC [12–14]. Chromium, in its hexavalent oxidation state, is also a Group 1 carcinogen [15]. Inorganic Pb have been classified in Group 2A (“probably carcinogenic”) by IARC [16]. Thus, most available literature focus on the content of these elements in both tobacco and cigarette smoke [17–20]. Despite that, several other metals such as cobalt (Co), manganese (Mn), zinc (Zn), antimony (Sb) and beryllium (Be) are also present in tobacco and cigarette smoke and can contribute to the harmful effect of smoking.

Virtually, all metals present in tobacco may transfer at some extent into cigarette smoke, and this transfer into both mainstream and sidestream smoke has been of particular interest in the last decades [8,20,21]. More recently, the hot topic of metal transfer has been also studied in the case of electronic cigarettes, which are promoted as safer alternatives to conventional cigarettes [22,23]. The extent of which a metal transfer from tobacco to cigarette smoke varies greatly depending on several factors such as the metal properties, metal content in the tobacco, cigarette design, rod length and diameter, filter type, ventilation, among others. [7,8,17].

The aims of this study were: (1) to determine the current levels of several metals in tobacco; (2) to calculate the transfer rate of metals from tobacco to cigarette smoke; and (3) to look for evidences of smoking habits in lung tissue (to establish a link between smoking and metals content in lung tissue). The transfer rate of

metals from tobacco to cigarette smoke was calculated as the difference between the total metal content in cigarettes and the amount present in its ashes.

Materials and methods

Cigarette samples

Packs of twenty (20) different cigarette brands were purchased from retail outlets at Oporto city, Portugal, in 2014. These brands represent the 20 best-selling cigarette brands in the country. Samples were kept in their original packaging until analysis.

Lung tissue samples

Lung tissues samples were collected from men (n=37) and women (n=25) not registered in the Portuguese National Registry of Refusal to Organ Donation database and complying with all the current regulations regarding human tissue collection for scientific research purposes.

Samples were obtained from individuals submitted to forensic autopsy exams during the year of 2013 and 2014 at the North Branch (Porto) of the Portuguese National Institute of Legal Medicine and Forensic Sciences. A fragment of ca. 1 cm³ was collected from the superior lobe of the right lung and placed in decontaminated polypropylene tubes. Samples were thoroughly rinsed with ultrapure water to completely remove biological fluids and then stored at –4 °C.

Reagents

High purity HNO₃ (≥69% w/w, TraceSELECT[®], Fluka, France), HF (47–51% v/v, TraceSELECT[®], Fluka, Germany) and boric acid (99.999% trace metal basis, Aldrich, St. Louis, MO) were used as received. Calibration standards were prepared from a 10 mg/L multi-element ICP-MS standard solution (PlasmaCAL SCP-33-MS, SCP Science, Baie-d’Urfé, Quebec, Canada). The internal standards solution was prepared by appropriate dilution of the AccuTrace[™] (AccuStandard[®], New Haven, CT) ICP-MS-200.8-IS-1 solution (100 mg/mL of Sc, Y, In, Tb and Bi). All solutions were prepared using ultrapure water (>18.2 MΩ.cm at 25 °C) obtained with a Sartorius (Goettingen, Germany) Arium[®] pro water purification system.

Tobacco samples preparation

Two cigarettes from each brand were dried in decontaminated plastic containers for a minimum of 24 h at 105 °C. Then, the filter was removed and the dried tobacco was manually crushed, homogenized and stored in tightly sealed polypropylene tubes until further analysis. Tobacco samples were solubilized by closed-vessel microwave assisted acid digestion in a MLS-1200 Mega (Sorisolet, Italy) microwave oven according to Fresquez et al. [19]. Briefly, samples (ca. 200 mg) were directly weighted into the polytetrafluoroethylene (PTFE) microwave oven vessels and 5 mL of high-purity concentrated HNO₃ was added. The microwave heating program was set to reach the temperature of 200 °C in 4 min, followed by a 3 min digestion at 200 °C. After cooling, sample solutions were transferred into 25 mL decontaminated polypropylene volumetric flasks, 5 mL of HF (5% v/v) were added and the volume was made-up with ultra-pure water. Sample blanks were obtained using the same procedure. Each sample was digested in triplicate.

¹ Actually, arsenic is a semi-metal (or metalloid), however for convenience of writing the word “metals” is used to refer all the studied elements.

Cigarettes ash samples preparation

Cigarettes were submitted to a simulated smoking procedure. Each cigarette was burned by drawing air through the cigarette with a pipette bulb. During the procedure, the ashes produced were collected into decontaminated polypropylene tubes. Ashes were then solubilized by closed-vessel microwave assisted acid digestion in the same equipment mentioned above according to Hepp, Mindak and Cheng [24]. Cigarette ashes (ca. 100 mg) were directly weighted into the PTFE microwave oven vessels and 7 mL of high-purity HNO₃ and 2 mL of high-purity HF were added. The samples were heated for 15 min at 130 °C and held at that temperature for three minutes. Then, the temperature was ramped to 200 °C in 15 min and kept at 200 °C for 30 min. The vessels were allowed to cool and 30 mL of 4% (w/v) boric acid solution was added. The samples were heated again in the microwave to 170 °C in 15 min and kept for 10 min at this temperature. After cooling, sample solutions were transferred into 50 mL decontaminated polypropylene volumetric flasks and the volume was made-up with ultra-pure water. Sample blanks were obtained using the same procedure. Each sample was digested in triplicate.

Samples analysis

Multi-elemental analysis was performed using an iCAP™ Q (Thermo Fisher Scientific, Bremen, Germany) ICP-MS instrument equipped with a MicroMist™ nebulizer, a Peltier-cooled baffled cyclonic spray chamber, a standard quartz torch and a two-cone interface design (sample and skimmer cones). High-purity (99.9997%) argon (Gasin II, Leça da Palmeira, Portugal) was used as the nebulizer and plasma gas. The ICP-MS was operated under the following conditions: RF power, 1550 W; argon flow rate, 14 L/min; auxiliary argon flow rate, 0.8 L/min; nebulizer flow rate, 0.98 L/min. The following elemental isotopes (*m/z* ratios) were monitored for analytical determinations: ²⁷Al, ⁵²Cr, ⁵⁵Mn, ⁵⁹Co, ⁶⁰Ni, ⁶⁶Zn, ⁷⁵As, ¹¹¹Cd, ¹³⁵Ba, ²⁰⁵Tl and ²⁰⁸Pb; the elemental isotopes ⁴⁵Sc, ⁸⁹Y, ¹¹⁵In and ¹⁵⁹Tb were monitored as internal standard. The limits of detection (LOD) were calculated as the concentration corresponding to 3 times the standard deviation of 10 replicate measurements of the digested sample blanks and are presented in Supplementary material (Table S1).

Estimation of metals transfer from tobacco to cigarette smoke

The rate of metals transfer from tobacco to cigarette smoke was calculated according to the following equation:

$$MT = \frac{M_{\text{Cigarette}} - M_{\text{Ash}}}{M_{\text{Cigarette}}} \times 100$$

where MT is the metal transfer in percentage (%), $M_{\text{Cigarette}}$ is the metal content in tobacco (μg/g of cigarette, dry weight) and M_{Ash} is the metal content in the ash (μg/g of cigarette, dry weight).

Analytical quality control

For analytical quality control purposes, the following certified reference materials (CRM) were used: BCR 679 (white cabbage), BCR 176R (fly ash), BCR 667 (estuarine sediment), BCR 723 (road dust)—all supplied by EC Institute for Reference Materials and Measurements (Geel, Belgium). The BCR 679 was analyzed under the same conditions as cigarette samples while BCR 176R, BCR 667 and BCR 723 were analyzed under the same conditions as cigarette ashes. The results obtained in the CRM analysis are presented in Table S2 (Supplementary material).

Table 1

Metals content of tobacco (μg/g, dry weight).

Element	Range ^a	Literature data (range)	Reference
Al	458.6–904.7 (mean: 667.2)	699–1200 333–546	[21] [25]
As	0.08–0.20 (mean: 0.14)	0.22–0.36 0.06–2.07	[19] [17]
Ba	102.3–169.6 (mean: 123.0)	40.7–56.6 0.19–4.67	[21] [17]
Cd	0.49–1.42 (mean: 0.79)	1.0–1.7 1.66–2.96	[19] [25]
Co	0.61–1.08 (mean: 0.84)	0.44–1.11 <0.01–0.94	[19] [21]
Cr	0.74–2.52 (mean: 1.39)	1.3–3.2 <0.1–3.45	[19] [21]
Mn	107.7–179.0 (mean: 148.4)	131–245 155–400	[19] [21]
Ni	1.49–2.65 (mean: 2.10)	2.1–3.9 <2–400	[19] [21]
Pb	0.44–0.72 (mean: 0.55)	0.18–4.63 0.60–1.16	[17] [19]
Tl	0.046–0.104 (mean: 0.074)	0.40–1.39 –	[25] –
Zn	18.6–30.7 (mean: 25.2)	16.8–30.5	[21]

^a Results obtained for mixed samples (two cigarettes) of 20 brands, analyzed in triplicate.

Statistical analysis

Data exploration as well as calculation of descriptive statistics and ANOVA was performed with IBM SPSS Statistics for Windows, version 22.0 (IBM Corp., Armonk, NY). Significant differences were assumed at $p < 0.05$. For mathematical calculations, results below the LOD were imputed as the LOD divided by the square root of 2.

Results and discussion

Metals content in tobacco

Twenty (20) cigarette brands were analyzed for their content in 11 metals. The results are summarized in Table 1. The decreasing order of metal content was Al > Mn > Ba > Zn > Ni > Cr > Co > Cd > Pb > As > Tl. Three groups were formed according to the average metal content. The first group (contents > 100 μg/g) includes Al (667.2 μg/g), Mn (148.4 μg/g) and Ba (123.0 μg/g). In the second group (contents ranging between 1 and 100 μg/g), it is included Zn (25.2 μg/g), Ni (2.10 μg/g) and Cr (1.39 μg/g). The third cluster (contents < 1 μg/g) includes Co (0.84 μg/g), Cd (0.79 μg/g), Pb (0.54 μg/g), As (0.14 μg/g) and Tl (0.074 μg/g). The average metals content obtained in this study is in close agreement with the literature data [17,19,21,25]. See Table 1 for a detailed comparison between the values obtained in this study and data from literature.

Metals transfer from tobacco to cigarette smoke

The rate of metals transfer from tobacco to cigarette smoke was calculated based on the metal content in tobacco and cigarette ashes. The results are summarized in Table 2. It was observed that the rate of transference varies greatly depending on the metal. The highest rates were observed for Tl (85–92%) and Cd (81–90%) and the lowest for Al (0.2–2.5%), Ba (0.5–1.6%) and Mn (0.1–2.4%). Both As and Pb showed to pass to cigarette smoke at an intermediate extent (33–44% and 46–60%, respectively). These results are in good agreement with literature data. Nandi et al. [26] studied the Cd partition between cigarette ash and smoke and found that just 16% of Cd stays in the ash while 84% is volatilized. Kazi, Jalbani, Arain, Jamali, Afridi, Sarfraz and Shah [25] also studied the

Table 2
Rate of metals transfer from tobacco to cigarette smoke (%).

Element	Range ^a	Literature data	Reference
Al	0.2–2.5	1.0–2.7	[25]
As	33–44	30–50	[17]
Ba	0.5–1.6	40	[30]
Cd	81–90	–	–
		69–85	[25]
		84	[26]
Co	1.8–9.5	70–80	[17]
		12.5	[34]
Cr	0.7–8.3	20	[29]
		50.2	[34]
Mn	0.1–2.4	5	[29]
		56.3–65.5	[25]
Ni	3.6–15.5	2.8–8.7	[36]
		22.5–41.6	[33]
		54.3–67.1	[25]
		24.3–35.5	[31]
Pb	46–60	37	[32]
		30–50	[17]
Tl	85–92	–	–
Zn	13–21	12.4–18.5	[33]
		7.6	[34]

^a Results obtained for mixed samples (two cigarettes) of 20 brands, analyzed in triplicate.

partition of Cd and other metals (Al, Ni and Pb) between cigarette ash and smoke and found that only a small fraction of Cd is present in the ashes (15–31%), while the remaining (69–85%) passes into the smoke. Piadé et al. [17] also reported that Cd in ashes ranges between 20 and 30% of the total Cd present in the tobacco. For Tl, no studies are available regarding its transfer from tobacco to cigarette smoke. However, the very high rates of Tl transfer to cigarette smoke obtained in this study (83–92%) can be explained by its high volatility. In fact, Tl was included in the group of elements showing high volatility during coal combustion [27,28] and thus it is expected that, during cigarette smoking, Tl greatly passes into the smoke fraction.

In our study, the lowest transfer rates were observed for Al, Ba and Mn, as mentioned above. Kazi, Jalbani, Arain, Jamali, Afridi, Sarfraz and Shah [25] showed that, after smoking, most of the Al in tobacco is present in the ash (97.3–99.0%) and that only a small fraction (1.0–2.7%) actually passes to the smoke, which is in good agreement with our results (0.2–2.5%). Ba and Mn also showed lower rates of transfer to cigarette smoke (0.5–1.6% and 0.1–2.4%, respectively). Comparison with literature data is not possible for Ba since no studies are available. For Mn, only one study reported the rate of transfer to cigarette smoke. In this particular study, most Mn was present in the ashes (95%), and only 5% showed to pass into the smoke [29]. Despite the lack of studies for these metals, it is plausible that most Ba and Mn stays in the cigarettes ash. These metals are in the list of non-volatile elements, being concentrated in the ash residues after coal combustion [27].

For the others metals analyzed, lower rates of transfer to cigarette smoke (<10%) were generally observed. Three elements showed to be notable exceptions: As (33–44%), Pb (46–60%) and Zn (13–21%). Piadé et al. [17] reported that 30–50% of As and Pb present in tobacco passes to the smoke. Other studies reported similar values for As (40%) [30] and for Pb (24–37%) [31,32]. For Zn, two studies reported values ranging from 7.6 to 18.5% [33,34], which is slightly below the range obtained in this study (13–21%).

The temperature at the tip of a cigarette that burns may reach 900 °C, and this temperature is high enough to pass most metals into the gas phase [35]. The chemical species of each element actually present in tobacco greatly affect its volatility (and therefore its transfer rate to the gas phase). For instance, most Cd seems to be present in tobacco as part of organic molecules and thus it is volatilized at lower temperatures, compared to the inorganic

Table 3
Metals content in lung tissue (µg/g, dry weight; mean ± SD).

Element	Smokers (n = 24)	Non-smokers (n = 38)
Al	97.8 ± 94.4	78.2 ± 68.9
As [*]	0.089 ± 0.034	0.043 ± 0.022
Ba	0.48 ± 0.45	0.33 ± 0.24
Cd [*]	0.68 ± 0.44	0.074 ± 0.062
Co	0.076 ± 0.064	0.065 ± 0.052
Cr	1.25 ± 1.37	0.63 ± 0.47
Mn	0.55 ± 0.71	0.32 ± 0.19
Ni	0.66 ± 0.93	0.38 ± 0.42
Pb [*]	0.32 ± 0.26	0.11 ± 0.09
Tl	0.002 ± 0.001	0.002 ± 0.001
Zn	30.6 ± 15.2	27.5 ± 8.7

^{*} Statistically different at p < 0.05.

forms of this element. By contrary, As and Pb are usually present in their inorganic and non-volatile form, which explains their higher content in cigarette ashes compared to Cd [17]. The same is also observed for Ni, where most of its content remains in the ashes after cigarette burning [17,36].

Evidences of metals exposure from cigarette smoke in lung tissues

The lung tissue of 77 adults (24 smokers and 38 non-smokers) was analyzed for their content in 11 metals. The results are summarized in Table 3. The lung tissue of smokers showed a higher content of the 11 analyzed elements compared to the lung tissue of non-smokers. Significant higher levels of As, Cd and Pb were found in smokers' lung tissue. The As level was twice as high in the smokers compared to non-smokers (0.089 ± 0.034 vs. 0.043 ± 0.022 µg/g). For Pb, mean level was three times higher in smokers than non-smokers (0.32 ± 0.26 vs. 0.11 ± 0.09 µg/g). For Cd, the difference was even more noticeable. The Cd level was almost ten times higher in the lung tissue of smokers compared to non-smokers (0.68 ± 0.44 vs. 0.074 ± 0.062 µg/g). Thus, it is evident that smoking influences the content of As, Cd and Pb in the lung tissue. These results are in good agreement with literature data. Paakko, Kokkonen, Anttila and Kalliomaki [37] determined the Cd and Cr levels in the lung tissue of 45 individuals and found that Cd was significantly higher in smokers than non-smokers (3.0 ± 2.2 vs. 0.4 ± 0.4 µg/g). More recently, De Palma, Goldoni, Catalani, Carbognani, Poli, Mozzoni, Acampa, Internullo, Rusca and Apostoli [38] determined the levels of several metals, including As, Cd and Pb, in the lung tissue of patients affected by lung cancer and found that the content of Cd and Pb was significantly higher in smokers. As, Cd and Pb are well-recognized toxic elements – As and Cd have been classified in Group 1 (“Carcinogenic to humans”) and inorganic Pb in Group 2A (“probably carcinogenic”) by IARC [12,14,16]. Thus, smokers have an increased risk of developing lung diseases due to exposure to these metals present in tobacco.

Besides As, Cd and Pb, other metals, especially those that are strongly inflammation-inducing and sensitizing, such as Cr, Co, Ni, can play an important role in the etiology of lung diseases. Recently, tobacco smoking has been associated with interstitial lung disease, a large group of disorders characterized by progressive scarring of the lung tissue between and supporting the air sacs [39,40]. In our study, a higher content of Cr, Co and Ni was observed in the lung tissue of smokers compared to non-smokers (1.25 ± 1.37 vs. 0.63 ± 0.47 µg/g for Cr, 0.076 ± 0.064 vs. 0.065 ± 0.052 µg/g for Co and 0.66 ± 0.93 vs. 0.38 ± 0.42 µg/g for Ni), proving that smoking can be an important source of human exposure to toxic metals.

Conclusion

Tobacco contains important amounts of a wide range of metals potentially harmful to human health. The transfer rate of metals to cigarette smoke varies greatly depending on the element. The higher transfer rates were observed for Tl and Cd, followed by Pb and As. It was found that As, Cd and Pb levels were significantly higher in the lung tissue of smokers compared to non-smokers, proving that, besides systemic effects, smoking results in a significant increase of these metals in that tissue. Since it is well-known that these metals are highly toxic, and in particular are known or presumed human carcinogens, this study shows that they may represent an important contribution to the overall carcinogenic potential of cigarette smoke and stresses the importance of public health measures aiming at to reduce smoking habits and exposure to second-hand smoke. This study also provides updated information on metals levels in tobacco and cigarette smoke, and new data for some elements that have not been yet evaluated (ex. Ba and Tl). Finally, strong evidence of the marks that smoking leaves in the lungs of smokers is provided.

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