| Reaction | Rate constants | Source* |
| :--- | :--- | :--- |
| O -> O + LpxA | $0.00007 \mathrm{~s}^{-1}$ | K |
| LpxA + S -> ES | $883 \mathrm{mM}^{-1} \mathrm{~s}^{-1}$ | K |
| ES -> LpxA + S | $717 \mathrm{~s}^{-1}$ | K |
| ES -> P + LpxA | $7.17 \mathrm{~s}^{-1}$ | E |
| P + LpxA -> ES | $883 \mathrm{mM}^{-1} \mathrm{~s}^{-1}$ | K |
| Q -> Q + LpxC | $0.000096 \mathrm{~s}^{-1}$ | K |
| LpxC + P -> EP | $7973 \mathrm{mM}^{-1} \mathrm{~s}^{-1}$ | K |
| EP -> LpxC + P | $0.015 \mathrm{~s}^{-1}$ | K |
| LpxC -> degrade | $0.00023 \mathrm{~s}^{-1}$ | K |
| EP -> P2 + LpxC | $1.5 \mathrm{~s}^{-1}$ | E |

* $\mathrm{E}=$ experimental $\mathrm{K}=$ from our kinetic model

