



UNIVERSITÀ DI PARMA

ARCHIVIO DELLA RICERCA

University of Parma Research Repository

Crystal structure and ferroelectric properties of ϵ -Ga₂O₃ films grown on (0001)-sapphire

This is the peer reviewed version of the following article:

Original

Crystal structure and ferroelectric properties of ϵ -Ga₂O₃ films grown on (0001)-sapphire / Mezzadri, Francesco; Calestani, Gianluca; Boschi, Francesco; Delmonte, Davide; Bosi, Matteo; Fornari, Roberto. - In: INORGANIC CHEMISTRY. - ISSN 0020-1669. - 55:22(2016), pp. 12079-12084. [10.1021/acs.inorgchem.6b02244]

Availability:

This version is available at: 11381/2820383 since: 2021-10-11T12:11:20Z

Publisher:

American Chemical Society

Published

DOI:10.1021/acs.inorgchem.6b02244

Terms of use:

openAccess

Anyone can freely access the full text of works made available as "Open Access". Works made available

Publisher copyright

(Article begins on next page)

Crystal Structure and Ferroelectric Properties of ϵ -Ga₂O₃ Films Grown on (0001)-Sapphire

Francesco Mezzadri,^{*,†,‡} Gianluca Calestani,[†] Francesco Boschi,^{‡,§} Davide Delmonte,[‡] Matteo Bosi,[‡] and Roberto Fornari^{‡,§}

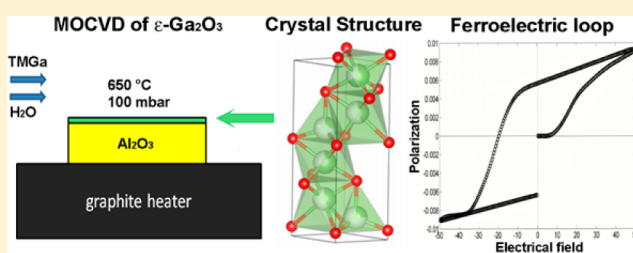
[†]Department of Chemistry, University of Parma, Parco Area delle Scienze 17/A 43124 Parma, Italy

[‡]IMEM-CNR, Parco Area delle Scienze 37/A, 43124 Parma, Italy

[§]Department of Physics and Earth Sciences, University of Parma, Parco Area delle Scienze 7/A, 43124 Parma, Italy

S Supporting Information

ABSTRACT: The crystal structure and ferroelectric properties of ϵ -Ga₂O₃ deposited by low-temperature MOCVD on (0001)-sapphire were investigated by single-crystal X-ray diffraction and the dynamic hysteresis measurement technique. A thorough investigation of this relatively unknown polymorph of Ga₂O₃ showed that it is composed of layers of both octahedrally and tetrahedrally coordinated Ga³⁺ sites, which appear to be occupied with a 66% probability. The refinement of the crystal structure in the noncentrosymmetric space group *P6₃mc* pointed out the presence of uncompensated electrical dipoles suggesting ferroelectric properties, which were finally demonstrated by independent measurements of the ferroelectric hysteresis. A clear epitaxial relation is observed with respect to the *c*-oriented sapphire substrate, with the Ga₂O₃ [10–10] direction being parallel to the Al₂O₃ direction [11–20], yielding a lattice mismatch of about 4.1%.



INTRODUCTION

Gallium oxide (Ga₂O₃) is a well-known sesquioxide with semiconducting properties. It may present five different polymorphs, α , β , γ , δ , and ϵ , each with different structure and physical properties. In the past few years, most research and technological development have focused on β -Ga₂O₃, especially because large single crystals^{1–3} and oriented substrates of this material were made available for epitaxy. These technological developments led to deposition of high-quality homoepitaxial films of β -Ga₂O₃^{4–6} and to fabrication of novel devices for high-power electronics⁷ and UV detection.⁸ Although the major attention is still focused on the β polymorph, there has been an increasing parallel interest also on the other phases, beginning from the pioneering work by Roy et al., who provided the first unambiguous identification of the different polymorphs in 1952.⁹ Among the polymorphs, the ϵ phase is particularly interesting because of its higher symmetry and simpler epitaxial growth conditions with respect to the more popular β phase. Furthermore, it shows a favorable matching to commercial sapphire as well as to other hexagonal or pseudohexagonal substrates.^{10–12} Recently, there were theoretical studies aiming at establishing structure and formation energy, i.e. thermodynamic stability, of the different polymorphs.¹³ While these first-principles calculations were able to give a qualitative description of the formation energies, concluding that the β phase (monoclinic phase) is the most stable polymorph, they failed in reproducing the actual structure of the ϵ phase. Some theoretical papers^{13,14} suggested

that this polymorph is orthorhombic, while recent experimental studies concur that it is hexagonal.^{10–12,15} Although theoretical estimates and experimental observations indicate that ϵ -Ga₂O₃ is metastable, it was recently observed that this phase remains stable up to at least 650 °C and that only after prolonged annealing at $T > 800$ °C does it undergo a complete transition to β .^{10–12} This allows sufficiently high working temperatures for devices based on ϵ -Ga₂O₃, which makes this phase very interesting in view of practical applications and justifies further investigations. Providing new information and enriching the very limited body of knowledge on ϵ -Ga₂O₃ is exactly the motivation of the present work. Here, the main aim is to provide a precise description of crystal structure and lattice parameters. Actually, in the literature we just found one report¹⁵ regarding a structure determination of the ϵ phase by neutron diffraction. In that report, ϵ -Ga₂O₃ was prepared via thermal decomposition of gallium nitrate, which unfortunately never supplied phase-pure materials but rather a mixture of β - and ϵ -Ga₂O₃. Nevertheless, the authors succeeded in solving the structure of the ϵ phase, albeit possibly distorted by the β contamination. They suggested that this polymorph was constituted by an hexagonal close-packed array of oxygen ions with partial filling of octahedral and tetrahedral sites. In the present work, single-crystal diffraction of phase-pure ϵ -Ga₂O₃ provides an accurate description of the structure of this material

Received: September 16, 2016

75 in thin film form, confirming from one side the results of ref 15
76 but showing that the positive and negative charges barycenters
77 do not coincide. This gives rise to uncompensated electrical
78 dipoles and to ferroelectric properties, as demonstrated by
79 independent measurements of the ferroelectric hysteresis.
80 Further, the epitaxial relations between the single-crystalline
81 ϵ -Ga₂O₃ layer and the α -Al₂O₃ substrate were determined along
82 with an estimate of their lattice mismatch.

83 ■ EXPERIMENTAL SECTION

84 A thick ϵ -Ga₂O₃ layer (thickness of about 3 μ m) was grown on c -plane
85 sapphire by MOCVD (metal-organic chemical vapor deposition) at
86 650 °C and 100 mbar using water and trimethylgallium (TMG) as
87 reagents and palladium-purified H₂ as carrier. The use of H₂O was
88 observed to be more favorable, probably due to the higher reactivity of
89 atomic oxygen from water dissociation with respect to molecular
90 oxygen. The H₂O to TMG partial pressure ratio was varied in the
91 range 100–1000 and generally set at 200. More details regarding the
92 growth procedure can be found in ref 11. Under these growth
93 conditions, all layers were reproducibly of ϵ phase. The precursor ratio
94 and relatively low deposition temperature are in our opinion decisive
95 factors in deciding what polymorph of Ga₂O₃ is actually grown. This
96 would indicate that thermodynamics rather than kinetics plays the
97 dominant role. This is supported by the experimental evidence that so
98 far the ϵ phase has been reported in connection with HVPE and
99 MOCVD growth experiments, but never as a result of deposition by
100 pure kinetic methods such as MBE or PLD, although conducted at low
101 temperature. Furthermore, it was demonstrated also recently¹⁶ that
102 MOCVD at temperature higher than 800 °C invariably leads to β -
103 Ga₂O₃. While this can be easily understood remembering that β is the
104 only thermodynamic stable polymorph, the failure in obtaining the ϵ
105 phase by means of kinetic methods, even at low temperatures,
106 probably lies in the higher distance from thermodynamic equilibrium
107 of these methods: i.e., the very energetic species adsorbed at the
108 substrate surface do not promote ϵ but just β nucleation. The sample
109 was then properly prepared for single-crystal X-ray diffraction in order
110 to maximize the diffraction intensities of the film with respect to those
111 of sapphire. The substrate thickness was mechanically reduced with a
112 lapping machine down to about 200 μ m by using 60 grit sandpaper.
113 The sample dimensions were then further reduced by mechanical
114 cleavage, which allowed obtaining small fragments suitable for single-
115 crystal diffraction experiments. X-ray diffraction (XRD) data were
116 collected with Mo K α radiation on a Bruker AXS Smart diffractometer,
117 equipped with an APEX II CCD area detector. The reconstruction of
118 precession images was carried out using the Precession Images plugin
119 available in the Bruker APEX2 software. The ferroelectric properties
120 were tested using the TF-Analyzer 2000E AixACcT system equipped
121 with the Ferroelectric Module (FE-Module). The sample was
122 previously metalized by sputtering 100 nm of gold on both surfaces
123 of a rectangular specimen with surface area of 3.55 mm², in the so-
124 called planar-plate capacitor configuration, and then annealing for 24 h
125 at 100 °C. As a result the studied sample can be idealized as two
126 capacitors in series: 200 μ m thick Al₂O₃ and 3 μ m thick ϵ -Ga₂O₃. The
127 measurements were performed at room temperature, exploiting the
128 dynamic hysteresis measurement (DHM) protocol (see the
129 Supporting Information for more details about the method). An ac
130 triangular bias between 0 and 1 kV, and frequency below 1500 Hz
131 were applied. The current flowing through the dielectric was recorded
132 as a function of the real-time value of the applied voltage during the
133 triangular wave signal. In order to get the maximum contribution from
134 the polar-induced current produced by ϵ -Ga₂O₃ and to reduce the
135 relative weight of the displacement current promoted by sapphire, the
136 current was collected from the gold plate on the film side. Blank
137 measurements were performed also on a sapphire sample. The
138 obtained data were analyzed using the Matlab platform.

■ RESULTS AND DISCUSSION

139

Single-crystal diffraction experiments, carried out on a 200 μ m
140 thick sample (film plus thinned substrate), as expected showed
141 2D diffraction patterns, constituted by the superimposition of
142 two lattices. The first lattice was characterized by reflections
143 with higher intensity, indexed in an hexagonal lattice with $a = 144$
4.759(1) Å and $c = 12.992(3)$ Å and corresponding to the
145 sapphire substrate,¹⁷ while the weaker reflections could be
146 indexed with a smaller hexagonal cell with $a = 2.906(2)$ Å and c
147 = 9.255(8) Å, corresponding to the Ga₂O₃ film. In Figure 1 a
148 precession image reconstructed on the basis of the collected
149 reciprocal space is reported, allowing identification of the
150 relative orientation of the two lattices. 151

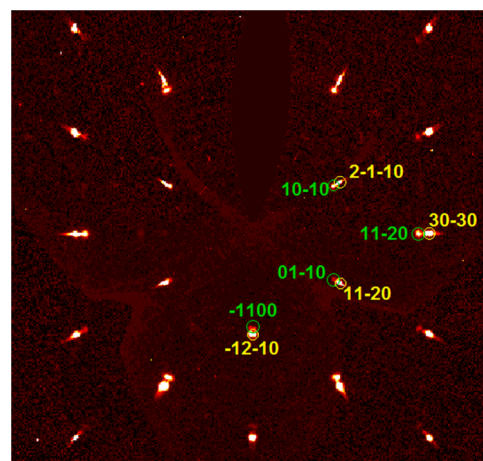


Figure 1. Precession image reconstructed by single-crystal diffraction data, $hki0$ projection; yellow and green labels refer to Al₂O₃ and Ga₂O₃ reflections, respectively.

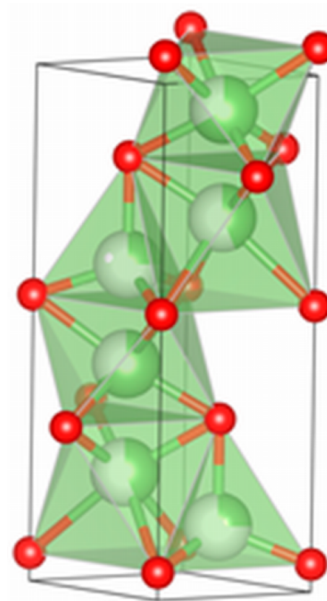
As can be seen, despite the low thickness of the ϵ -Ga₂O₃ film, 152
its excellent crystal quality gives rise to well-defined and 153
relatively intense diffraction peaks. The $hki0$ plane shows the 154
presence of two almost coincident hexagonal lattices with 155
slightly different d spacings. No extra peaks or powder rings 156
were detected, indicating the good quality of the film. The 157
Ga₂O₃ [10–10] direction is aligned with the Al₂O₃ [11–20] 158
direction, and consequently the two hexagonal lattices are 159
rotated in the ab plane by 30° with respect to each other. On 160
the other side, the difference in the c axes gives rise to a more 161
complex pattern in the $1kil$ plane (see Figure S1 in the 162
Supporting Information), where completely different reflection 163
sequences are produced for the two phases. In order to obtain 164
diffraction data suitable for quantitative structural analysis, a 165
small fragment with approximate dimensions 193 × 64 × 16 166
 μ m³ (the first number being the thickness of the film plus 167
residual substrate) was used, which allows for a reduction of the 168
intensity of the reflections from the substrate by several orders 169
of magnitude. At the same time, by increasing the sample to 170
detector distance from 50 to 70 mm, it was possible to isolate 171
the diffraction spots of the Ga₂O₃ phase. In this way, the ϵ - 172
Ga₂O₃ intensities could be accurately determined despite the 173
stronger Al₂O₃ reflections, which in a first instance impeded the 174
precise determination of the gallium oxide related intensities, in 175
particular at low θ values. The presence of Al₂O₃ on one side of 176
the sample, producing asymmetrical absorption, gives rise to 177
large scale factor variations, in the 0.7–1.4 range, and 178
unavoidably produces high residuals in merging reflections 179

Table 1. Selected Interatomic Distances, Formal Oxidation States ($Q(ij)$), Equivalent Coordination Numbers (ECoN), and Ion Charges

| atoms | bond length (Å) | atoms | bond length (Å) | atoms | bond length (Å) | atoms | bond length (Å) |
|--------------------|-----------------|--------------------|-----------------|--------------------|-----------------|---------|-----------------|
| GA1–O1 (×3) | 1.94(2) | GA2–O1 (×3) | 1.944(18) | GA3–O2 (×3) | 1.778 (10) | GA1–GA3 | 1.797(18) |
| GA1–O1 (×3) | 2.072(15) | GA2–O2 (×3) | 2.23(2) | GA3–O1 | 1.87(5) | GA1–GA2 | 1.947(15) |
| av | 2.006(18) | av | 2.087(19) | av | 1.80 (2) | GA2–GA3 | 1.890(10) |
| $Q(ij)$ | 1.979 | $Q(ij)$ | 1.011 | $Q(ij)$ | 1.01 | | |
| ECoN | 5.721 | ECoN | 4.811 | ECoN | 3.932 | | |
| $Q(ij)/\text{sof}$ | 2.968 | $Q(ij)/\text{sof}$ | 2.889 | $Q(ij)/\text{sof}$ | 3.176 | | |

180 ($R_{\text{int}} = 0.21\%$). Nonetheless, the quality of the data is not
 181 dramatically affected by absorption, allowing reliable structural
 182 solution and refinement, as discussed in the following. The
 183 Sir2011 suite was used for structure solution,¹⁸ while
 184 refinement was carried out using the Shelxl software.¹⁹ The
 185 cell parameters, refined by using the positions of 46 reflections
 186 in the theta range 4.401–19.914°, are $a = 2.9081(7)$ and $c =$
 187 $9.262(3)$ Å. The analysis of the systematic absences, showing
 188 reflection conditions $hh-2hl = 2n$ and $000l = 2n$, is compatible
 189 with the $P6_3/mmc$, $P\bar{6}2c$, and $P6_3mc$ space groups. While the
 190 structure solution process failed in the first two cases, a
 191 plausible result was obtained for the $P6_3mc$ space group. It has
 192 to be noted that the found solution may look wrong at first
 193 sight, since it implies a close-packed layered model in which the
 194 ratio between cationic and anionic sites is reversed with respect
 195 to the expected value (3/2 instead of 2/3). However, the full-
 196 matrix refinement, carried out on 102 unique reflections,
 197 pointed out unusually large values of the atomic displacement
 198 parameters (adps) of the cations that can be interpreted as a
 199 fingerprint of the statistical occupation of gallium sites. When
 200 refined independently as free variables, the gallium site
 201 occupancies converged to values that are in good agreement
 202 with the expected Ga_2O_3 stoichiometry, leading to a normal-
 203 ization of the adps. In particular, the occupancy of the GA1 site,
 204 occupying an octahedral cavity between two adjacent close-
 205 packed layers of oxygens, converged to 2/3. The same happens
 206 when the global occupancy of GA2 and GA3, which occupy in
 207 mutually exclusive ways the octahedral and tetrahedral cavities
 208 between the next adjacent layers, are considered. Therefore, the
 209 site occupancies of Ga atoms within the same layer were fixed
 210 to 2/3 in the last refinement cycles, whereas their anisotropic
 211 thermal parameters were constrained to be equal, in order to
 212 limit the number of structural variables. Indeed, the statistical
 213 occupation of the cationic sites in a cell with a reduced volume
 214 (to which a low density of reflections in the reciprocal space is
 215 related) leads to a noticeable increase of the number of
 216 parameters needed for describing the structure, reducing the
 217 data to variable ratio. Nonetheless the results obtained are
 218 highly reliable and the refinement converged with agreement
 219 indices $R1 = 0.0621$ for $65 F_o > 4\sigma(F_o)$ and $R1 = 0.1123$ for all
 220 101 data, $\text{GOF} = 1.077$. Crystal data and refined parameters are
 221 summarized in Table S1 in the Supporting Information, while
 222 relevant bond lengths and the results of bond strength–bond
 223 length analysis are reported in Table 1.

224 The crystal structure is shown in Figure 2 and consists of a
 225 4H stacking of close-packed oxygen layers, in which, similarly to
 226 what is observed in $\beta\text{-Ga}_2\text{O}_3$,²⁰ both the octahedral and
 227 tetrahedral cavities are occupied by gallium ions. Within two
 228 adjacent oxygen layers, the cavities are partially occupied, so
 229 that the Ga_2O_3 stoichiometry is guaranteed by the presence of
 230 Ga vacancies. Two types of layers alternate along the stacking

**Figure 2.** $\epsilon\text{-Ga}_2\text{O}_3$ structure representation. Green balls are gallium ions and red balls oxygen atoms.

direction: in the first one only the octahedral cavities are
 231 occupied (GA1) in a 2/3 ratio, whereas in the second the
 232 occupancy of both octahedral (GA2) and tetrahedral (GA3)
 233 cavities further complicates the picture.

234 As discussed before, the independent refinement of GA2 and
 235 GA3 occupancies clearly indicates that the two sites are
 236 occupied in a mutually exclusive way, which allows maintenance
 237 of the stoichiometric cation to oxygen ratio. This situation may
 238 derive from a purely stochastic occupation of the gallium sites,
 239 although the presence of an ordered superstructure cannot be
 240 ruled out. Actually, if the domain size is below the coherence
 241 length of the X-rays or the satellite intensities are too low, only
 242 an averaged structure can be observed in the diffraction
 243 experiment. However, different stacking sequences can
 244 reasonably be hypothesized, a few examples of which are
 245 reported in Figure 3.

246 The refined structure is in good agreement with that,
 247 ascribed to the Ga_2O_3 ϵ phase, recently proposed by Playford et
 248 al.¹⁵ on the basis of neutron powder diffraction data and pair
 249 distribution function analyses of polycrystalline samples
 250 including both the β and ϵ phases. The present study, based
 251 on the refinement of single-crystal data, provides new
 252 knowledge of $\epsilon\text{-Ga}_2\text{O}_3$, adding moreover relevant information
 253 about the epitaxial relation with the sapphire substrate, of great
 254 relevance for application. On comparison to the well-known
 255 and thermodynamically stable $\beta\text{-Ga}_2\text{O}_3$, the average bond
 256 distances of the ϵ phase are in general longer, in agreement with
 257

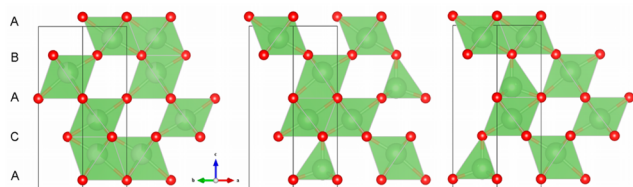


Figure 3. ϵ -Ga₂O₃ 4H structure viewed along the [1-100] direction. The three panels correspond to examples of possible stackings; in each layer 2/3 of the Ga sites are occupied.

vacancy-characterized structures, with the least occupied octahedron showing the longest bonds. The charge distribution analysis, performed with the CHARDIS99 program²¹ and making use of bond strength–bond length relations, indicates formal oxidation states of 1.979, 1.011, and 1.01 for the GA1, GA2, and GA3 sites, respectively. Taking into account their partial occupation, one obtains the oxidation states 2.968 for GA1, 2.889 for GA2, and 3.176 for GA3, quite close to the 3+ value expected on the basis of the refined stoichiometry. It is worth noting that, in agreement with the polar character of the space group, the positive and negative charges do not mutually compensate along the *z* direction, giving rise to nonzero electrical dipoles. Consequently, the phase should be pyroelectric and, on the basis of the structural information, have an electrical polarization of 0.18 $\mu\text{C}/\text{cm}^2$, as estimated using a simple point-charge model. Additional DHM measurements were carried out in order to assess the actual character of the ϵ -Ga₂O₃ polymorph, demonstrating that this material is indeed ferroelectric. As mentioned above, we used triangular pulses of 1 kV amplitude for such measurements. In the whole frequency range, the tested sample behaves as a good dielectric material with negligible presence of leakage currents. The registered value of the relative dielectric constant is around 7.17 at 1 kHz (lower than the value of about 8.6 expected for bulk sapphire,²² measured at 1000 Hz) while the measured resistivity is $1.74 \times 10^{12} \Omega \text{ cm}$. Under 1 Hz bias condition, we succeeded in detecting the electrical polarization loop of ϵ -Ga₂O₃. To do so, raw data had to be elaborated, i.e. by deleting the *V*-invariant displacement current of sapphire and taking into account the very weak leakage currents. The result of this process is shown in Figure 4. The present measurement, obtained by the integration of the current signal shown in Figure S3 in the Supporting Information, clearly indicates the formation of net

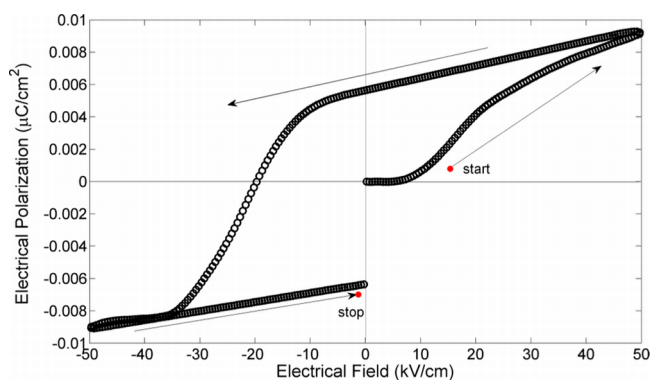


Figure 4. DHM measurement of the electrical polarization performed at room temperature by applying (1 Hz, 1 kV) triangular pulses on a sample constituted by a substrate of sapphire over which ϵ -Ga₂O₃ was grown.

polarization with maximum value of 0.0092 $\mu\text{C}/\text{cm}^2$ arising from a depolarized state. The hysteresis characteristic shows that the polarization is not saturated, at least by applying a field up to 50 kV/cm. It is noteworthy that, in the present measurement configuration, the electric field is applied perpendicular to the film surface and is consequently parallel to the polar *c*-axis of the compound.

Unfortunately, it was not possible to reach the polarization saturation in these ϵ -Ga₂O₃ films, as dielectric breakdown always occurred at electric fields of about 60 kV/cm in different samples. Hence, just a minor loop is covered in the figure. In any case, by taking the polarization sign change one may determine the coercive field: this was $E_C = 20.7 \text{ kV}/\text{cm}$ for the sample reported in Figure 4. In principle, this consideration may justify the consistent difference between the measured $P_{S,\text{rel}}$ and the saturation polarization estimated by calculations on the basis of the structural analysis (e.g., 0.18 $\mu\text{C}/\text{cm}^2$). The coexistence of semiconducting and ferroelectric properties is an unusual occurrence in solid-state physics and might open the route to the application of this material in new technological fields: for instance, by exploiting the charge separation induced by the internal electrical field. This could help to overcome the well-known limits of oxide semiconductors: i.e. the lack of an effective p-type conductivity. The ferroelectricity of ϵ -Ga₂O₃, presented here for the first time, makes this compound an interesting playground for additional experimentations.

A precise epitaxial relation is observed in the diffraction images, and this is not surprising if the ϵ -Ga₂O₃ and Al₂O₃ structures are compared. As shown in Figure 5, sapphire is indeed composed of identical layers of face-sharing AlO₆ octahedra packed with hexagonal symmetry and stacked along the *z* direction of the lattice.

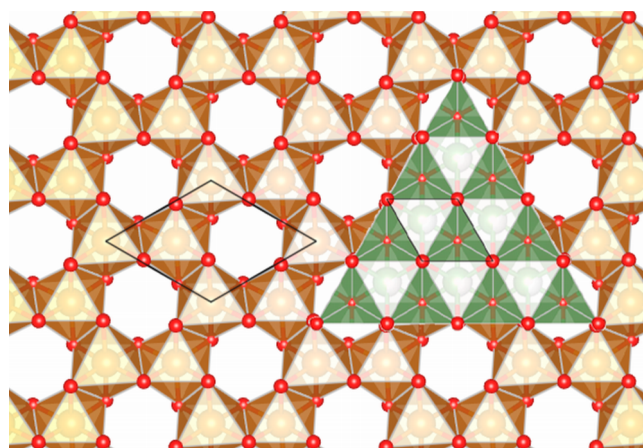


Figure 5. Al₂O₃ (orange) and ϵ -Ga₂O₃ (green) structures projected along the mutual (0001) direction, highlighting the epitaxial relation between the two lattices.

Two-thirds of the sites are periodically unoccupied in each layer, and the voids are shifted along the (1-100) direction in the *z*-adjacent layers. As discussed previously, ϵ -Ga₂O₃ is composed of layers of face-sharing polyhedra: one of the layers is composed solely of the octahedra occupied by the GA1 ion while the second layer contains both octahedrally (GA2) and tetrahedrally (GA3) coordinated ions. In all cases the distance between adjacent cations atoms in the *ab* plane is 2.91 Å, while it is 2.79 Å in Al₂O₃, thus revealing a relatively good condition for epitaxy, with a compressive lattice mismatch of about 4.1%.

333 Within this framework, it is easy to understand the partial
334 occupation of the cation sites in ε -Ga₂O₃, with the need to
335 guarantee both the Ga–O stoichiometric ratio and electrical
336 neutrality of the building blocks. Consequently, in strict analogy
337 with Al₂O₃, 2/3 of the sites are occupied on the GA1 layer and
338 the same applies to the GA2/GA3 layer, where the sum of the
339 occupancies is again 2/3. It is interesting to note that in this
340 layer the smaller GA3 tetrahedral site, occupied with 31.8% of
341 probability (i.e., almost 1/3), is likely used to release the
342 compressive strain produced during the epitaxial growth. The
343 present data are not sufficient, alone, to determine the local
344 structure of the Ga₂O₃ film. Actually, the observed fractional
345 occupation could be related to merely statistical occupation of
346 the sites as well as to short-range ordered domains. However,
347 considering that the growth process first involves nucleation on
348 the sapphire substrate, followed by in-plane growth and
349 coalescence,¹¹ it can be hypothesized that ordered Al₂O₃-like
350 islands with lateral dimensions smaller than the X-ray coherency
351 length initially form, driving the following growth steps. This
352 would be in agreement with the limited strain (or the apparent
353 absence of it) in the deposited Ga₂O₃ film, resulting in well-
354 shaped spherical diffraction spots. Additional high-resolution
355 TEM measurements are in progress to shed some light on the
356 order–disorder question.

357 ■ CONCLUSIONS

358 The structural properties of hexagonal ε -Ga₂O₃ deposited on *c*-
359 oriented sapphire by low-temperature MOCVD were thor-
360 oughly investigated. The performance of single-crystal X-ray
361 diffraction experiments on a “composite” ε -Ga₂O₃/sapphire
362 sample allowed the accurate determination of the crystal
363 structure of the ε -Ga₂O₃ phase, revealing at the same time a
364 well-defined epitaxial relation with the substrate, the ε -Ga₂O₃
365 [10–10] direction being parallel to the α -Al₂O₃ [11–20],
366 yielding a lattice mismatch of about 4.1%. Structure solution
367 and refinement were carried out in the polar noncentrosym-
368 metric *P*6₃*mc* space group. The structure consists of a 4H
369 stacking of close-packed oxygen layers, in which Ga atoms
370 occupy both octahedral and tetrahedral sites, as in the stable β -
371 phase. Two types of cationic layers are alternatively stacked
372 along the *c* direction, one involving octahedral sites occupied by
373 the gallium ions with a 66% probability, while in the second
374 both octahedral and tetrahedral sites are present, once again
375 with a global occupancy limited to 2/3. This result is in
376 agreement with the expected Ga/O ratio for the Ga₂O₃
377 composition, while bond strength–bond length calculations
378 confirm the 3+ oxidation state for all of the gallium ions. These
379 findings may be explained by a random occupation of the
380 gallium sites, even if some ordering phenomena (ordered
381 superstructure) are likely. Unfortunately, due to experimental
382 limitations (e.g., small coherence domains, low intensities,
383 Al₂O₃ substrate absorption effects), it is not possible to make a
384 decisive statement here about the order–disorder topic. In
385 addition, the structure refinement showed that the gallium and
386 oxygen ion pattern is such that it gives rise to uncompensated
387 electrical dipoles and indeed the unambiguous ferroelectric
388 hysteresis minor loop was measured by the DHM technique.
389 This is the first time that ferroelectricity has been
390 unambiguously detected in a Ga₂O₃ polymorph. The
391 coexistence of semiconducting and ferroelectric properties is
392 rather unusual and makes this material worthy of further
393 studies, aiming in particular at developing novel application
394 fields.

■ ASSOCIATED CONTENT

■ Supporting Information

The Supporting Information is available free of charge on the
ACS Publications website at DOI: 10.1021/acs.inorg-
chem.6b02244.

Reconstructed precession images in *1k1l* projection,
complete structural information, ferroelectric measure-
ments details and raw current data obtained from the
DHM experiment (PDF)
Crystallographic data (CIF)

■ AUTHOR INFORMATION

Corresponding Author

*E-mail for F.M.: francesco.mezzadri@unipr.it.

Funding

The Ph.D. scholarship of F. Boschi was provided by
Fondazione Cariparma.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

The authors are indebted to Prof. M. Solzi for providing access
to his FE measurement facilities.

■ REFERENCES

- (1) Villora, E. G.; Shimamura, K.; Yoshikawa, Y.; Aoki, K.; Ichinose, N. Large-size β -Ga₂O₃ single crystals and wafers. *J. Cryst. Growth* **2004**, *270*, 420–426.
- (2) Aida, H.; Nishiguchi, K.; Takeda, H.; Aota, N.; Sunakawa, K.; Yaguchi, Y. Growth of β -Ga₂O₃ Single Crystals by the Edge-Defined Film Fed Growth Method. *J. Appl. Phys.* **2008**, *47*, 8506–8509.
- (3) Galazka, Z.; Uecker, R.; Irmscher, K.; Albrecht, M.; Klimm, D.; Pietsch, M.; Brützmam, M.; Bertram, R.; Ganschow, S.; Fornari, R. Czochralski growth and characterization of β -Ga₂O₃ single crystals. *Cryst. Res. Technol.* **2010**, *45*, 1229–1236.
- (4) Oshima, T.; Arai, N.; Suzuki, N.; Ohira, S.; Fujita, D. Surface morphology of homoepitaxial β -Ga₂O₃ thin films grown by molecular beam epitaxy. *Thin Solid Films* **2008**, *516*, 5768–5771.
- (5) Sasaki, K.; Higashiwaki, M.; Kuramata, A.; Masui, T.; Yamakoshi, S. Size control of self-assembled InP/GaInP quantum islands. *J. Cryst. Growth* **2013**, *378*, 591–595.
- (6) Wagner, G.; Baldini, M.; Gogova, D.; Schmidbauer, M.; Schewski, R.; Albrecht, M.; Galazka, Z.; Klimm, D.; Fornari, R. Homoepitaxial growth of β -Ga₂O₃ layers by metal-organic vapor phase epitaxy. *Phys. Status Solidi A* **2014**, *211*, 27–33.
- (7) Higashiwaki, M.; Sasaki, K.; Kuramata, A.; Masui, T.; Yamakoshi, S. Development of gallium oxide power devices. *Phys. Status Solidi A* **2014**, *211*, 21–26.
- (8) Guo, D.; Wu, Z.; Li, P.; An, Y.; Liu, H.; Guo, X.; Yan, H.; Wang, G.; Sun, C.; Li, L.; Tang, W. Fabrication of β -Ga₂O₃ thin films and solar-blind photodetectors by laser MBE technology. *Opt. Mater. Express* **2014**, *4*, 1067–1076.
- (9) Roy, R.; Hill, V. G.; Osborn, E. F. Polymorphism of Ga₂O₃ and the system Ga₂O₃–H₂O. *J. Am. Chem. Soc.* **1952**, *74*, 719–722.
- (10) Oshima, Y.; Villora, E. G.; Matsushita, Y.; Yamamoto, S.; Shimamura, K. Epitaxial growth of phase-pure ε -Ga₂O₃ by halide vapor phase epitaxy. *J. Appl. Phys.* **2015**, *118*, 085301–1–5.
- (11) Boschi, F.; Bosi, M.; Berzina, T.; Buffagni, E.; Ferrari, C.; Fornari, R. Hetero-epitaxy of ε -Ga₂O₃ layers by MOCVD and ALD. *J. Cryst. Growth* **2016**, *443*, 25–30.
- (12) Xia, X.; Chen, Y.; Feng, Q.; Liang, H.; Tao, P.; Xu, M.; Du, G. Hexagonal phase-pure wide band gap ε -Ga₂O₃ films grown on 6H-SiC substrates by metal organic chemical vapor deposition. *Appl. Phys. Lett.* **2016**, *108*, 202103–1–5.

- 456 (13) Yoshioka, S.; Hayashi, H.; Kuwabara, A.; Oba, F.; Matsunaga,
457 K.; Tanaka, I. Structures and energetics of Ga₂O₃ polymorphs. *J. Phys.:
458 Condens. Matter* **2007**, *19* (11pp), 346211.
- 459 (14) Maccioni, M. B.; Fiorentini, V. Phase diagram and polarization
460 of stable phases of (Ga_{1-x}In_x)₂O₃. *Appl. Phys. Express* **2016**, *9*, 041102–
461 1–4.
- 462 (15) Playford, H. Y.; Hannon, A. C.; Barney, E. R.; Walton, R. I.
463 Structures of Uncharacterised Polymorphs of Gallium Oxide from
464 Total Neutron Diffraction. *Chem. - Eur. J.* **2013**, *19*, 2803–2813.
- 465 (16) Baldini, M.; Albrecht, M.; Fiedler, A.; Irmscher, K.; Schewski, R.;
466 Wagner, G. Si- and Sn-Doped Homoepitaxial β-Ga₂O₃ Layers Grown
467 by MOVPE on (010)-Oriented Substrates. *ECS J. Solid State Sci.*
468 *Technol.* **2017**, *6*, Q3040–Q3044.
- 469 (17) Genzel, C. A self-consistent method for X-ray diffraction
470 analysis of multiaxial residual-stress fields in the near-surface region of
471 polycrystalline materials. II. Examples. *J. Appl. Crystallogr.* **1999**, *32*,
472 779–787.
- 473 (18) Burla, M. C.; Caliendo, R.; Camalli, M.; Carrozzini, B.;
474 Cascarano, G. L.; Giacobazzo, C.; Mallamo, M.; Mazzone, A.; Polidori,
475 G.; Spagna, R. SIR2011: a new package for crystal structure
476 determination and refinement. *J. Appl. Crystallogr.* **2012**, *45*, 357–361.
- 477 (19) Sheldrick, G. M. Crystal structure refinement with SHELXL.
478 *Acta Crystallogr., Sect. C: Struct. Chem.* **2015**, *71*, 3–8.
- 479 (20) Åhman, J.; Svensson, G.; Albertsson, J. A Reinvestigation of β-
480 Gallium Oxide. *Acta Crystallogr., Sect. C: Cryst. Struct. Commun.* **1996**,
481 *52*, 1336–1338.
- 482 (21) Nespolo, M.; Ferraris, G.; Ohashi, H. Charge distribution as a
483 tool to investigate structural details: meaning and application to
484 pyroxenes. *Acta Crystallogr., Sect. B: Struct. Sci.* **1999**, *55*, 902–916.
- 485 (22) McConnell, R. D.; Wolf, S. *Science and Technology of Thin Film*
486 *Superconductors*; Springer: Berlin, 1989.